

# **The uncertainty principle, virtual particles and real forces.**

([http://teachers.web.cern.ch/teachers/archiv/HST2005/bubble\\_chambers/BCwebsite/articles.htm](http://teachers.web.cern.ch/teachers/archiv/HST2005/bubble_chambers/BCwebsite/articles.htm))

(An introduction to quantum fluctuations)

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# Wave-particle duality: a quick review

Photoelectric effect

$$E = hf$$

Two-slit experiment with electrons

$$p = h/\lambda$$

[\* Why do particle physicists need high energy accelerators?]

# Why do atoms have energy levels?

- \* Confined waves:  $f_1, f_2, f_3, \dots, f_n, \dots$
- \* Assume electrons in H-atoms have wavelike properties (cf. 2-slit)
- \* Then H-atom 'is' a confined electron wave, with allowed frequencies  $f_1, f_2, f_3, \dots, f_n, \dots$
- \*  $E = hf \implies$  allowed energies  $hf_1, hf_2, hf_3, \dots$ , or energy levels  $E_1, E_2, E_3, \dots$
- \* **So, atoms have energy levels because they are confined electron waves.**

# Measuring the frequency of a wave

\* Demo

\* Key idea: to measure frequency with an

accuracy  $\Delta f$  one needs a time  $\geq \frac{r}{\Delta f}$

where  $r$  is a positive number ( $\frac{1}{4\pi}$ ).

\* Assume true for mysterious QM waves too.

\* Then time  $\geq \frac{1}{4\pi\Delta f} = \frac{h}{4\pi h\Delta f} = \frac{h}{4\pi\Delta E}$

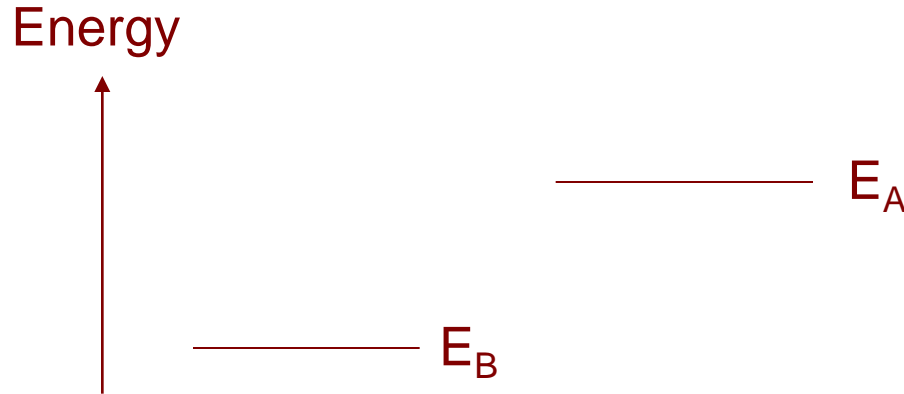
**(Heisenberg Uncertainty Principle)**

# Energy-time uncertainty principle in words:

to measure the energy of a quantum system (something we want to discuss using QM – an electron, for example) with an accuracy of  $\Delta E$ , we need a time greater than  $\hbar/4\pi\Delta E$

- \* Contains `magic loophole': can consider processes that violate energy conservation.
- \* Will see: Exchange Model of Forces

**Master:**



How well would you have to measure these energies?

**Pupil:**

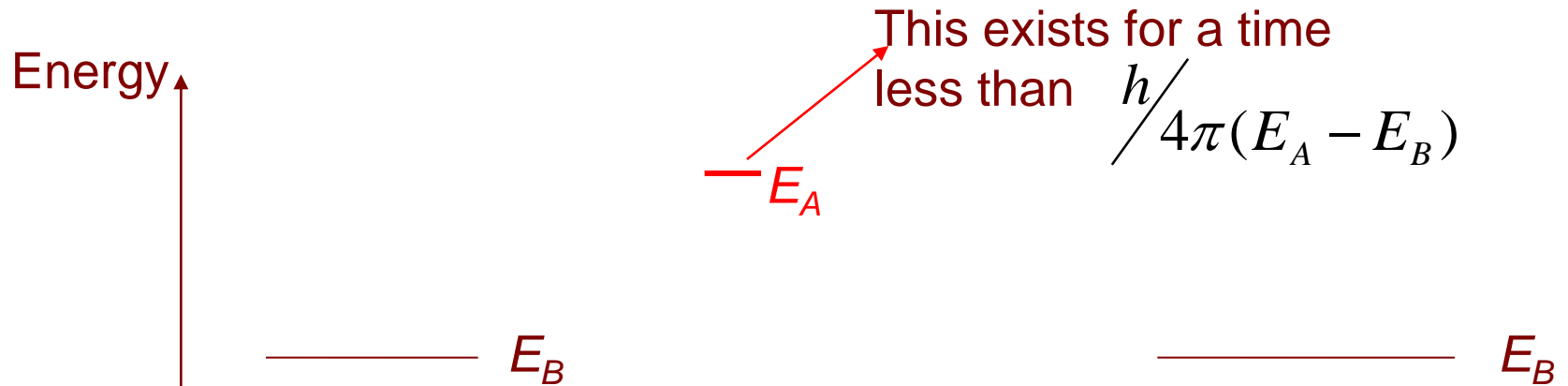
If error  $< (E_A - E_B)$ , can tell if  $E_A = E_B$  or not.

**Master:** how long would it take you to make your  
make your measurement with accuracy  $(E_A - E_B)$ ?

**Pupil:** According to the Heisenberg Uncertainty  
Principle, I'd need a time at least  $\frac{h}{4\pi(E_A - E_B)}$

**Master:** ... moves goalposts ...

## Master (contd):

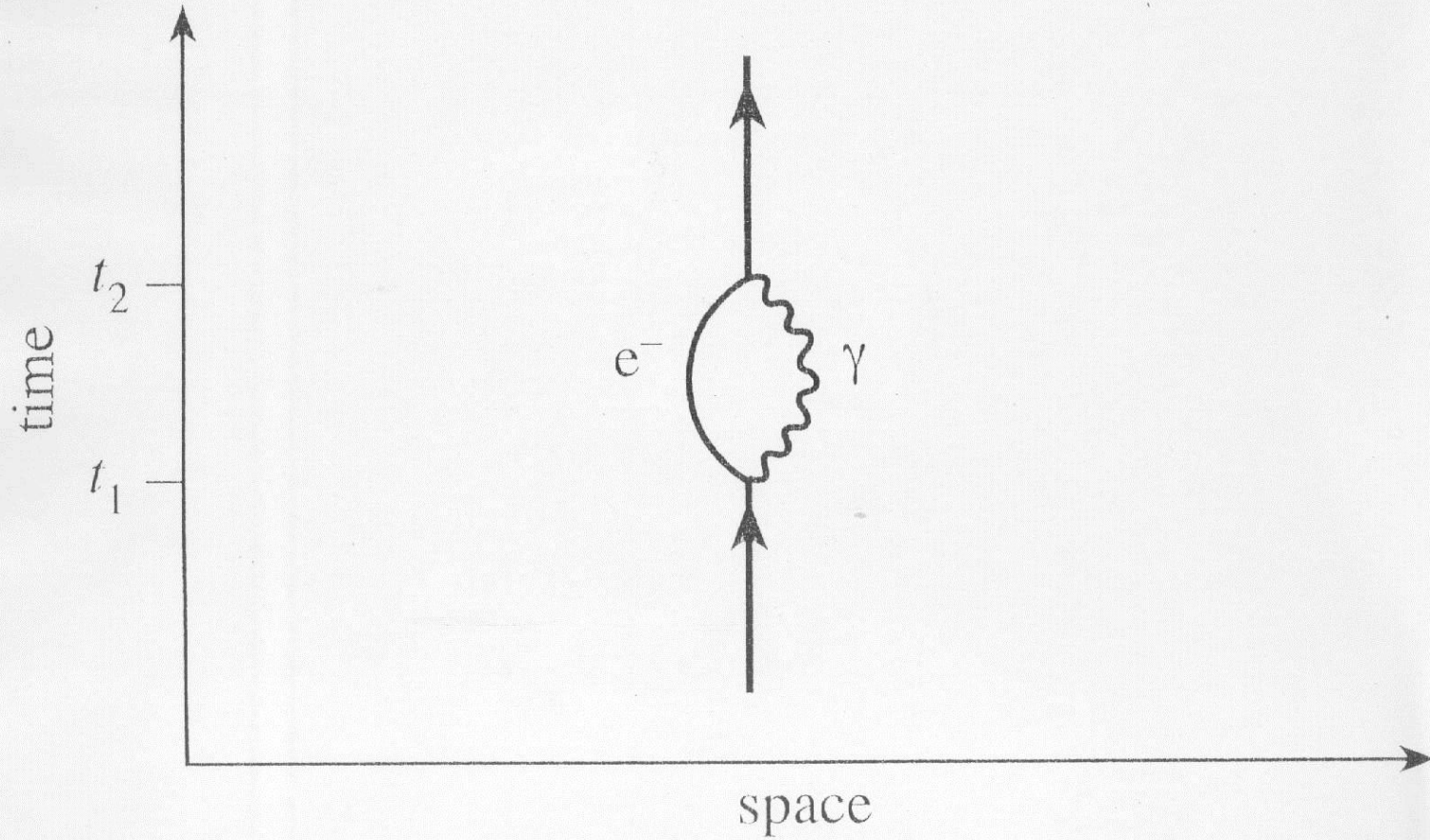


Tell me: how could you show that this sequence of processes could not occur?

**Pupil:** Problem ... not enough time ...  
Very abstract – example?



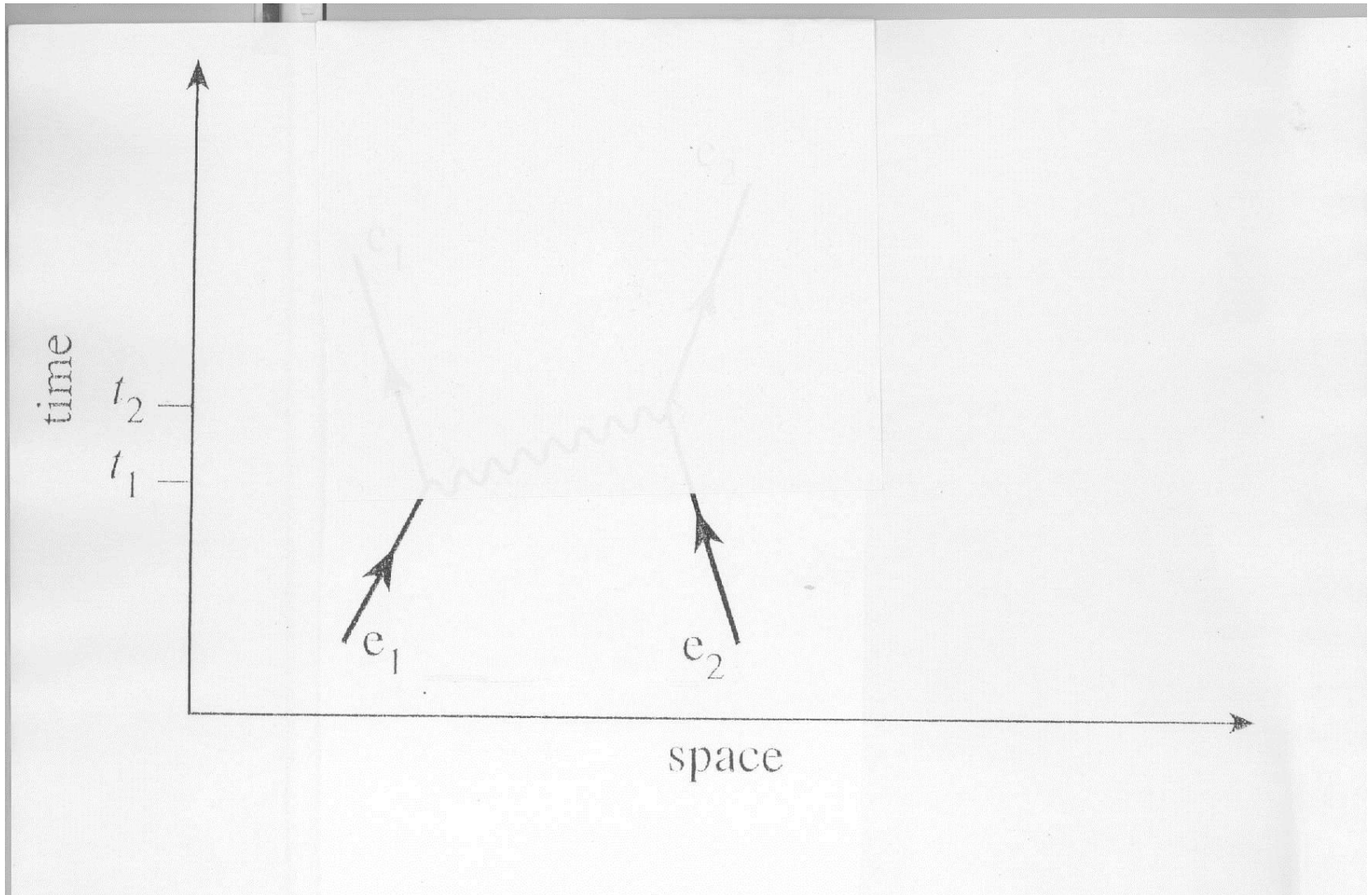
# Feynman diagrams are space-time graphs



Stationary **real**  $e^-$  undergoing one transition to a **virtual state** -  $e^- \gamma$

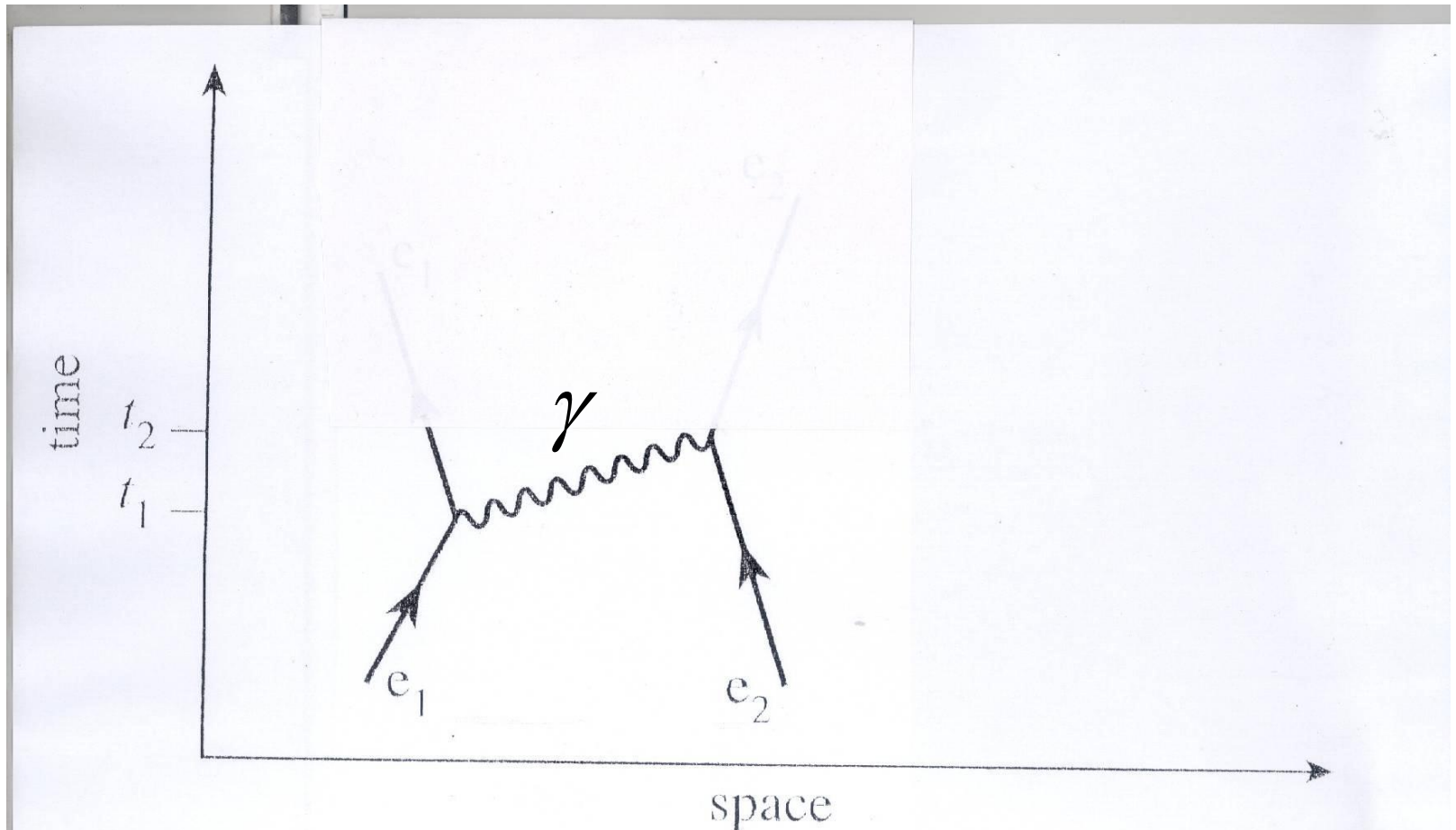
# The exchange model of forces

Consider 2 electrons  $e_1$  and  $e_2$  approaching each other.



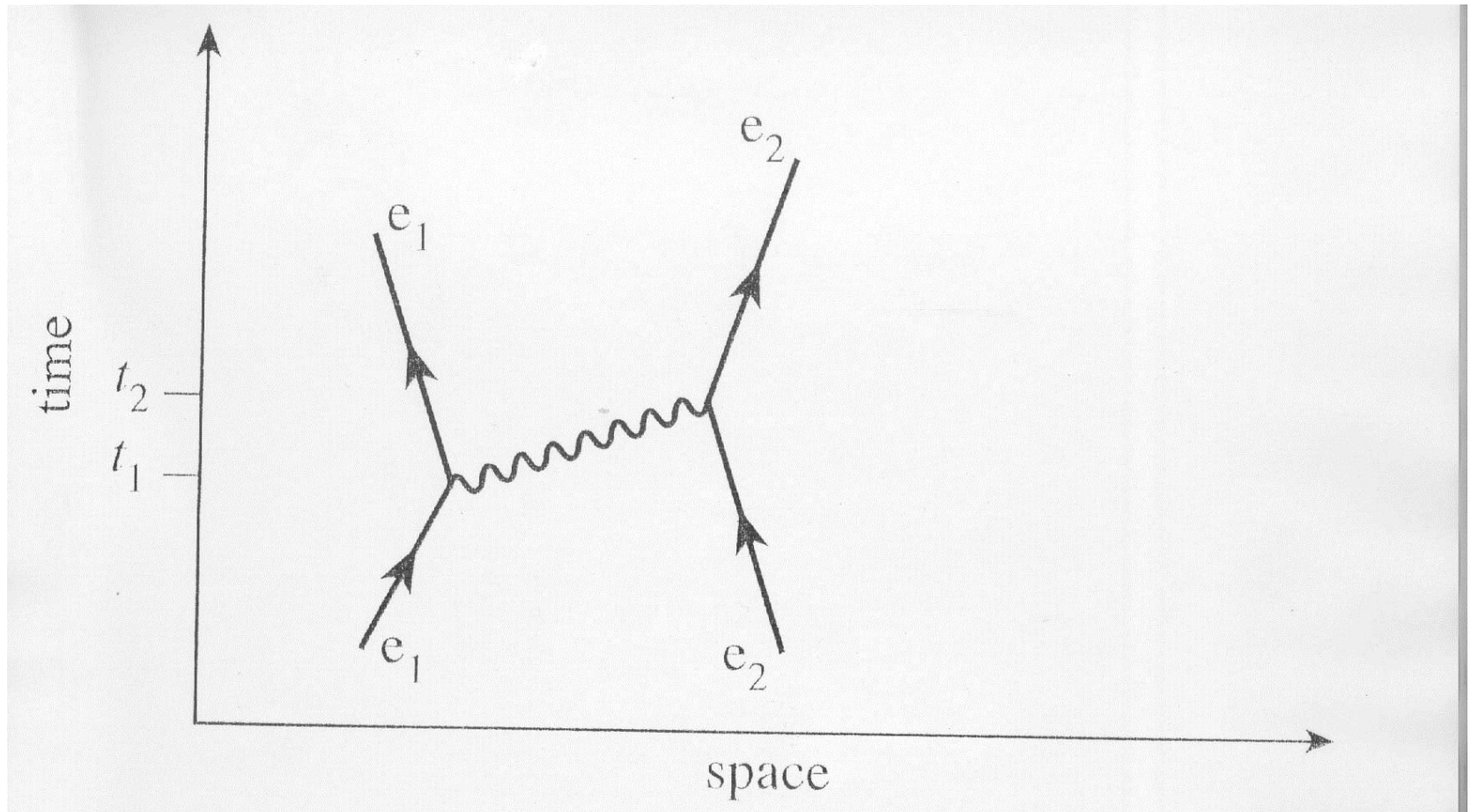
# The exchange model of forces

At time  $t_1$ , one of them emits a virtual photon  $\gamma$  which is absorbed by the other electron at time  $t_2$  in such a way that the total final energy is the same as the total initial energy.



# The exchange model of forces

The effect of this `exchange' of a **virtual** photon is that the electrons are moving away from each other - they have repelled each other.



# Discussion

By the Heisenberg Uncertainty Principle, the **virtual particle** of mass  $m$  can only exist for a time  $t \leq \frac{h}{4\pi mc^2}$

This is a finite time, during which the particle can only travel a finite distance, which we could define as the **range**  $R$  of the force.

The maximum conceivable value for  $R$  would be for a particle moving at the speed of light:

$$R = c \times \frac{h}{4\pi mc^2} = \frac{h}{4\pi mc}$$

**A remarkable formula!**

$R = \frac{h}{4\pi mc}$  gives the range  $R$  in terms of

\*  $h$  – fundamental constant of QM

\*  $c$  – fundamental constant of relativity

\*  $m$  – the mass of the exchanged particle:  $R \propto \frac{1}{m}$

So now we can visualise short-ranged forces.

**Pupil** (getting excited): We know  $R$  for nuclear forces; it is about  $10^{-15} m$ ; so we should be able to calculate  $m$ . Let's do it!

$$h = 6.63 \times 10^{-34} \text{ m}; \text{ and } c = 3 \times 10^8 \text{ ms}^{-1} \quad \Rightarrow$$

$$m \approx \frac{h}{4\pi cR} = \frac{6.63 \times 10^{-34}}{4\pi \times 3 \times 10^8 \times 10^{-15}} = 0.18 \times 10^{-27} \text{ kg}$$

So,  $m \sim 1/9 \times m(\text{proton})$ . What does this tell me?

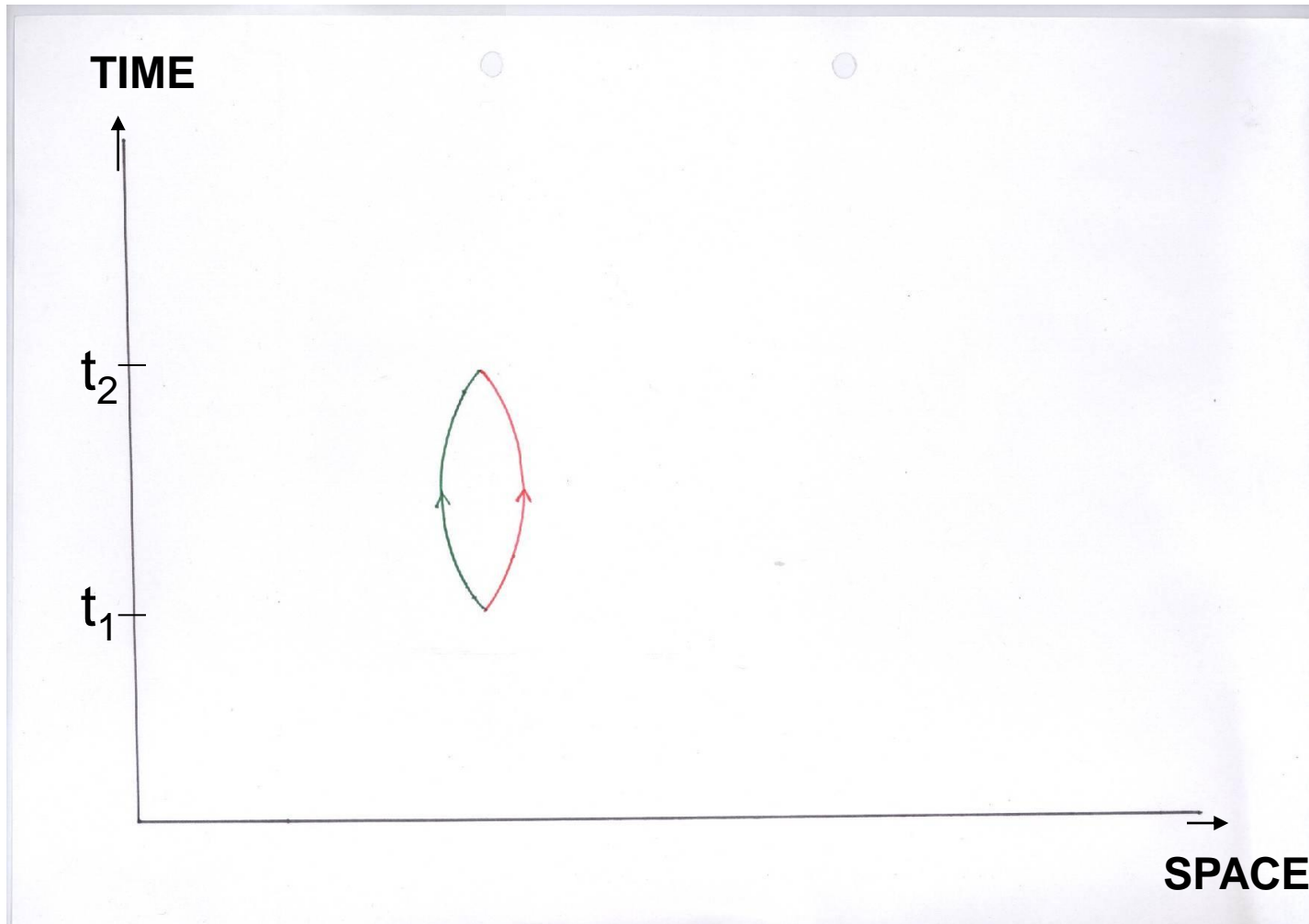
**Master:** When this idea was first put forward in 1934 by Yukawa, no particle of anywhere near this mass was known.

Now the job of theorists is to explore what might be hidden in equations. So one might speculate ...



# Antimatter

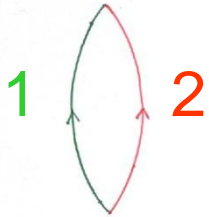
Describe this Feynman diagram:



**TIME**

- Can exist for time  $< h/4\pi E(\text{violation})$   
where

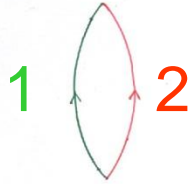
$$E(\text{violation}) = E_1 + E_2$$



**SPACE**

Could 1 and 2 be electrons?

**TIME**



If particle 1 were an electron, what properties would particle 2 need to have?

**SPACE**

What is empty space?