

# **Synthesis of Nanoscale Magnesium Diboride Powder**

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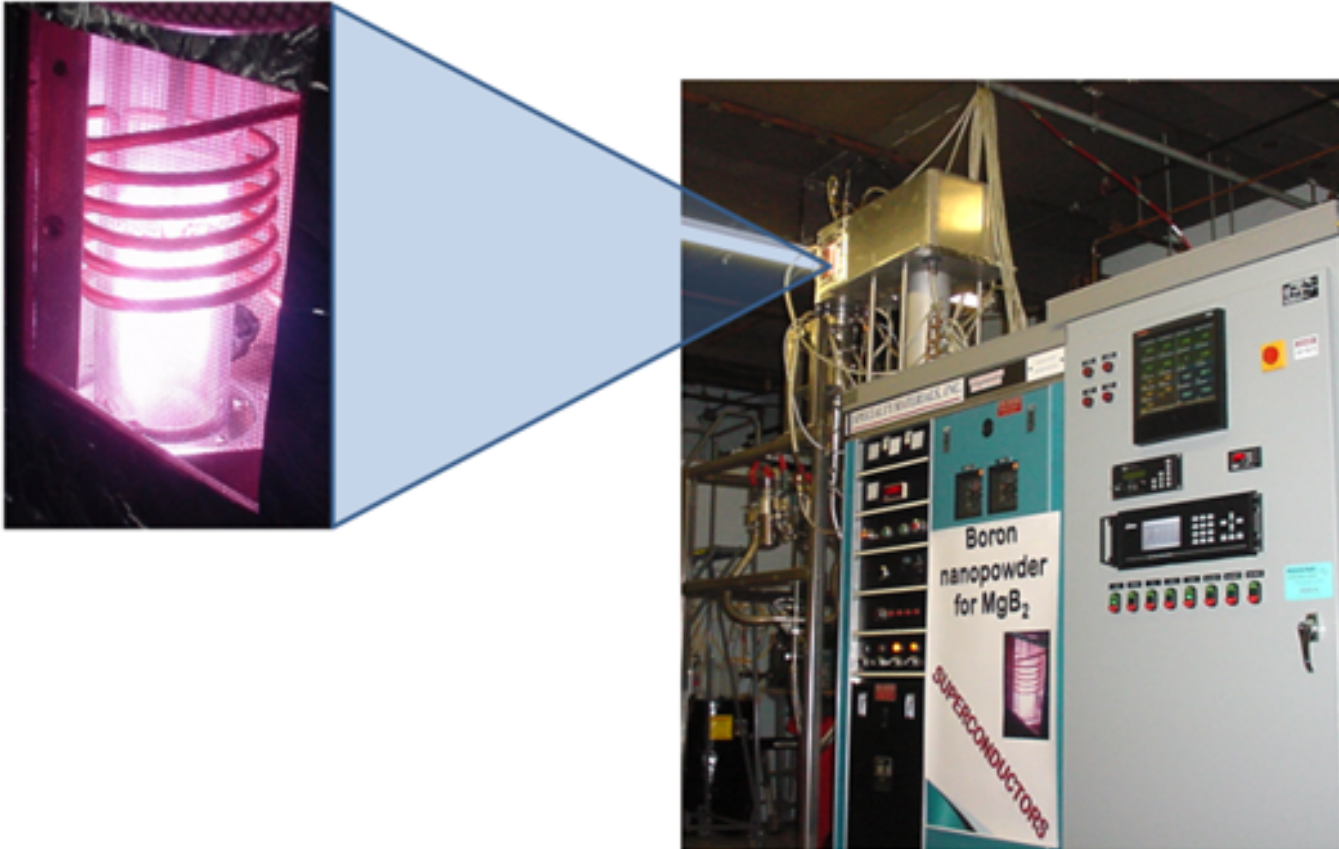
Specialty Materials Inc.

# Collaborator – Made Pure Boron

- James V. Marzik of Specialty Materials Inc. developed the **plasma synthesis** method to make high purity boron nanopowders having particle size in the range of 20 nm to 200 nm.
- [jmarzik@Specmaterials.com](mailto:jmarzik@Specmaterials.com)



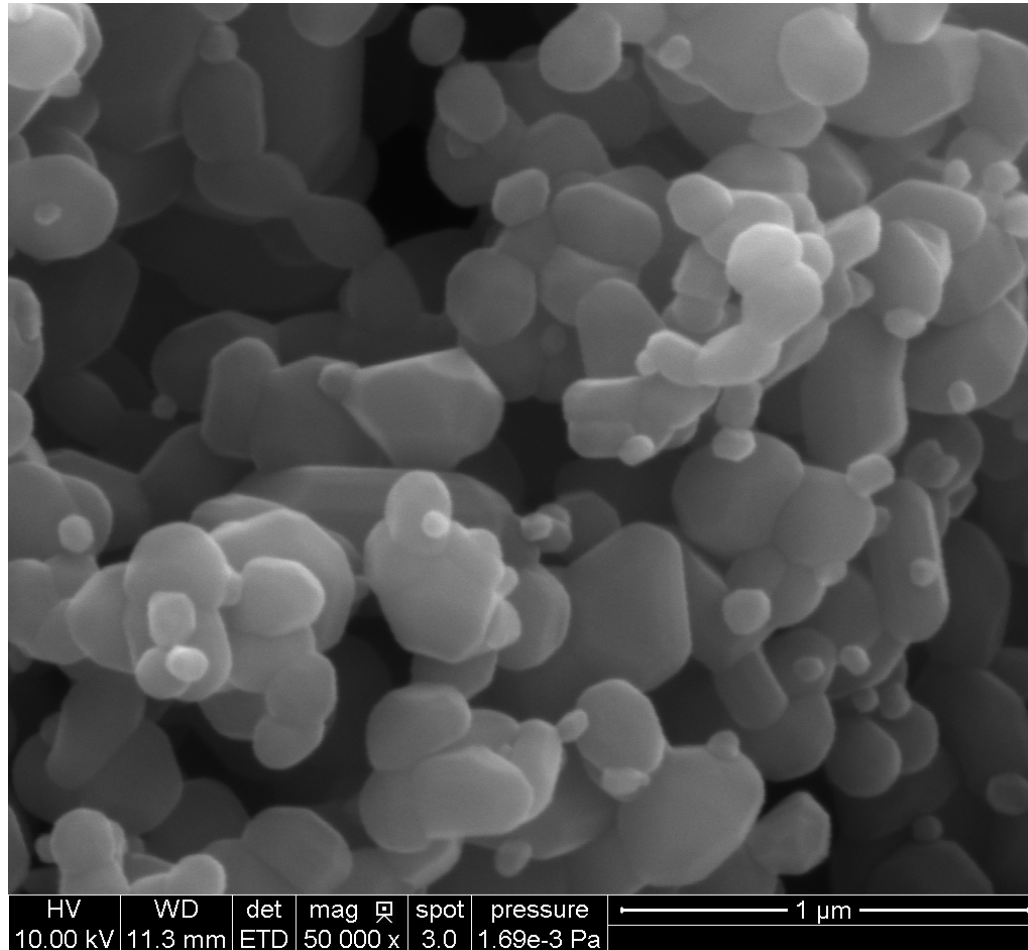
# Plasma Synthesis rf Argon Torch



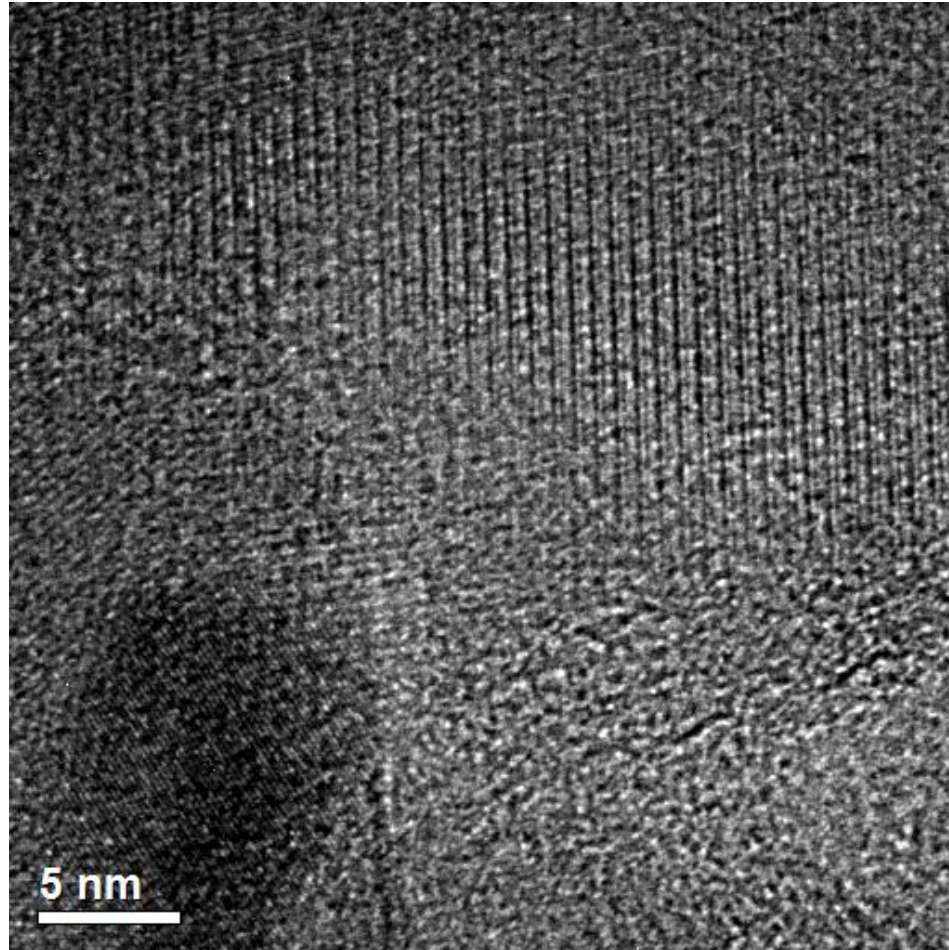
# Gas feed into the rf torch

- $\text{BCl}_2 + \text{H}_2 \rightarrow \text{B} + 2 \text{HCl}$
- Add methane,  $\text{CH}_4$ , for carbon doping
- Add titanium tetrachloride,  $\text{TiCl}_4$ , for  $\text{TiB}_2$  precipitates
- In this work, we use either undoped boron or boron 2% carbon.
- The point of adding carbon is to raise  $H_{c2}$  of the resulting  $\text{MgB}_2$  to 35 Tesla

# SEM of undoped boron powder from Specialty Materials Inc.

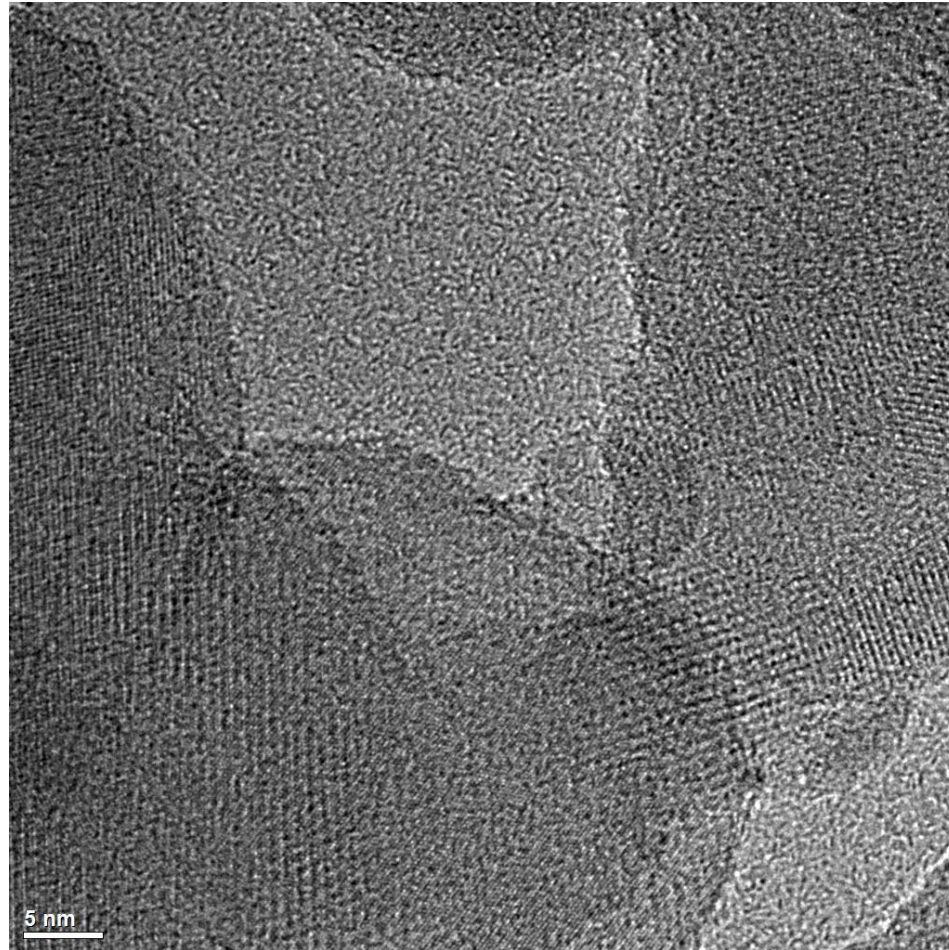


Boron contains both amorphous and crystallized boron – TEM micrograph

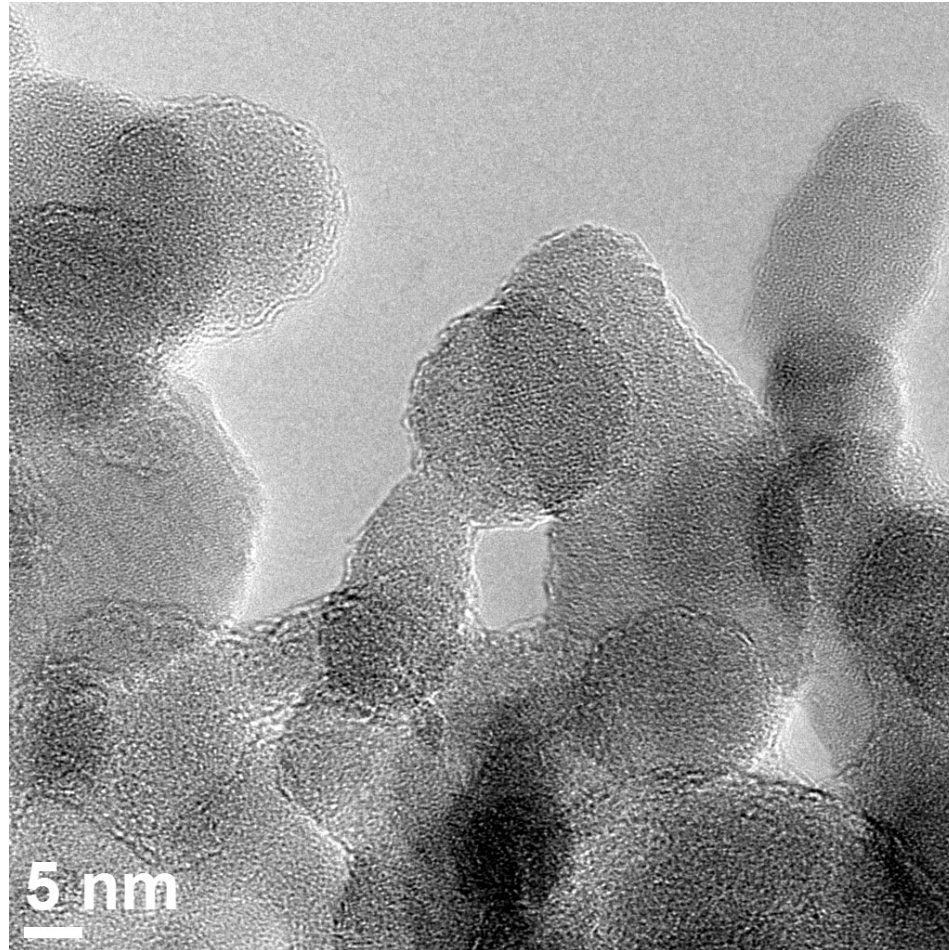




# TEM of mix of amorphous & crystalline boron particles

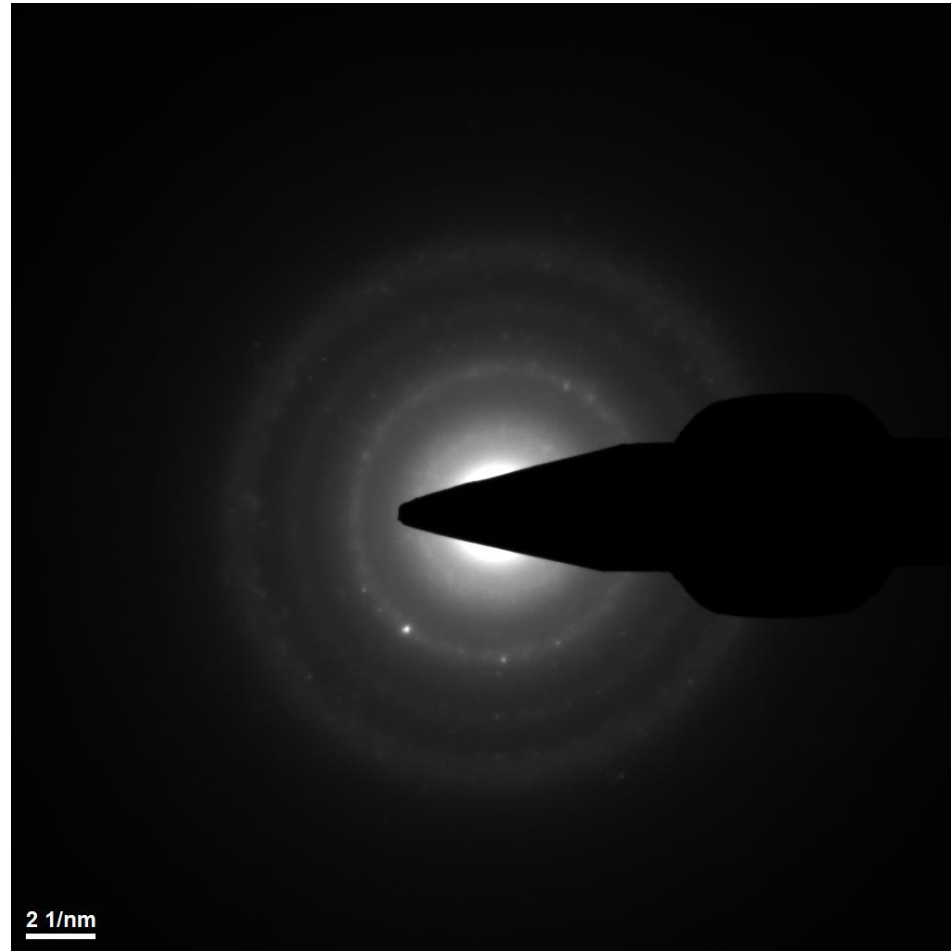


# Amorphous boron particles

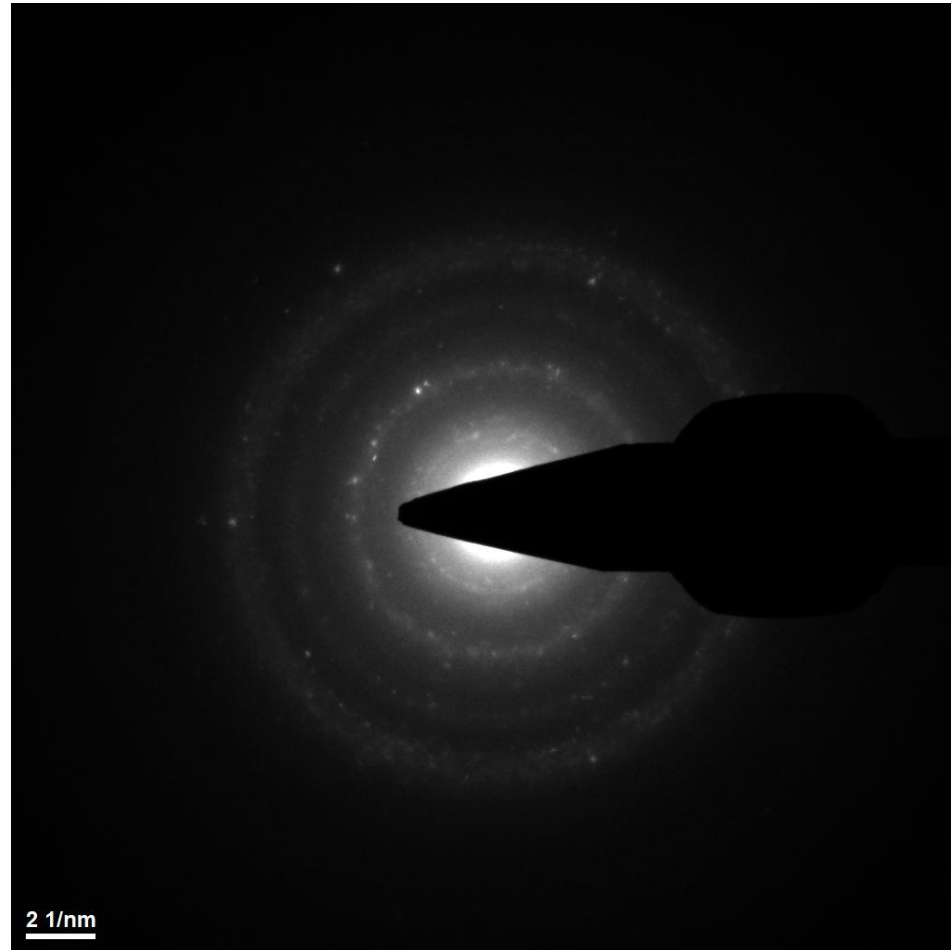




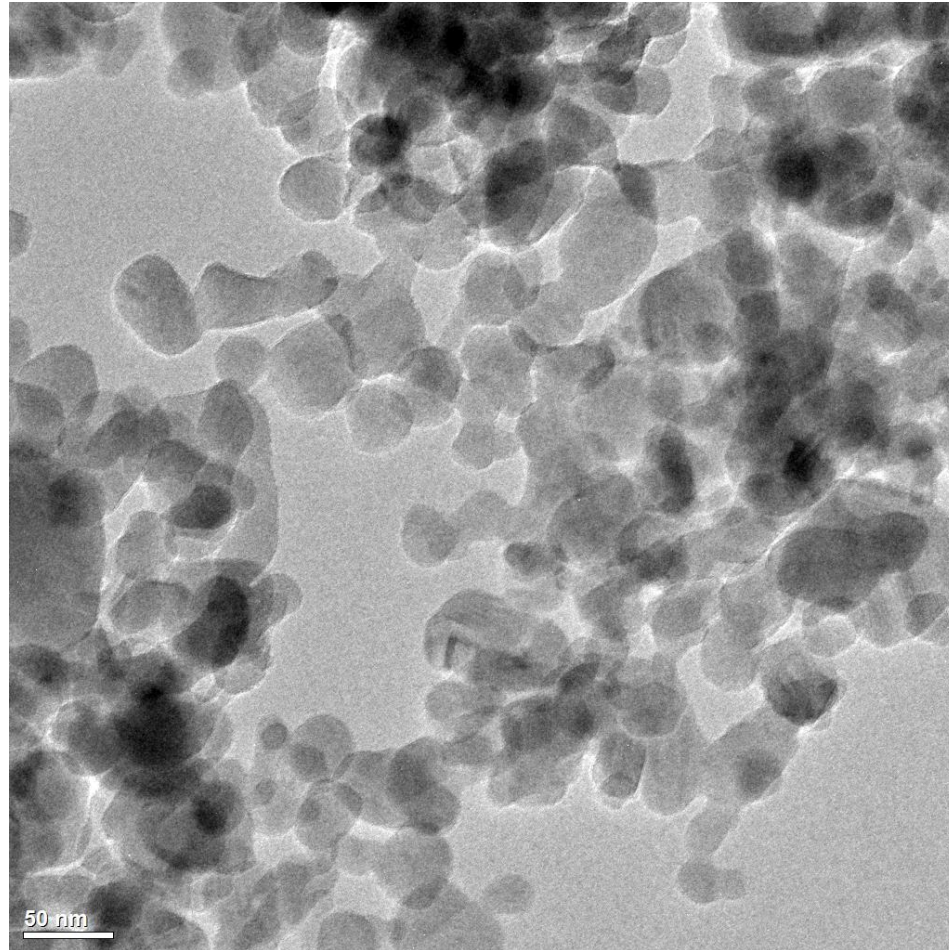
# Selective Area Diffraction (SAD)



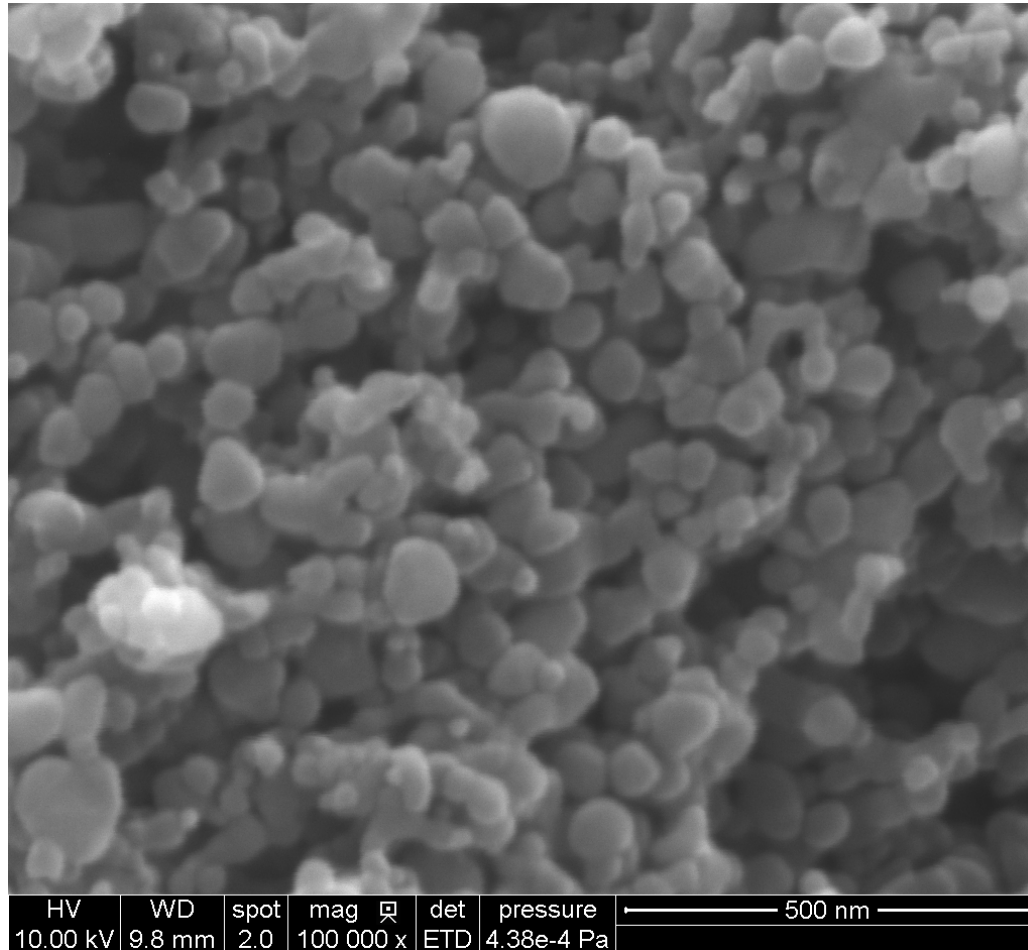
# TEM -- SAD shot



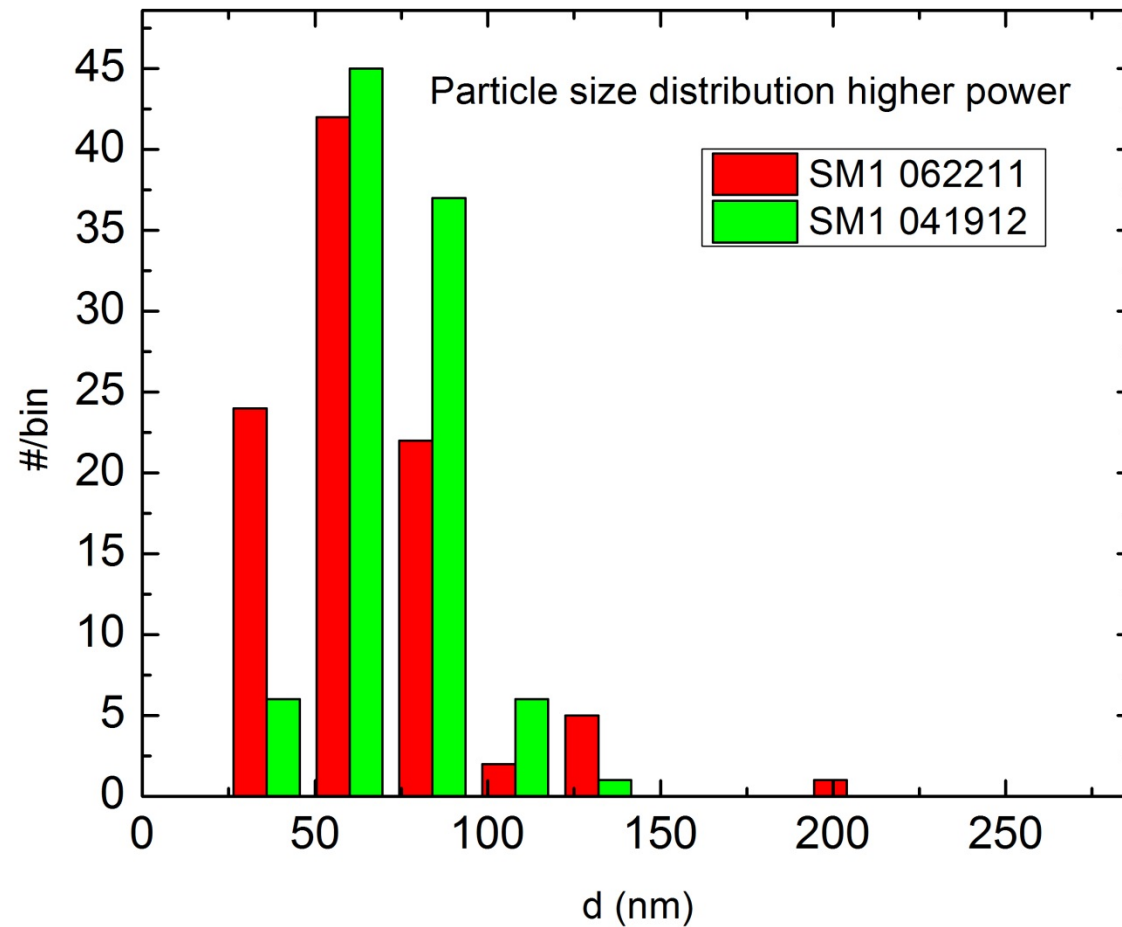
# TEM to get particle size distribution



# SEM of starting boron powder



# Boron powder particle size



# Broad descriptors of boron powder

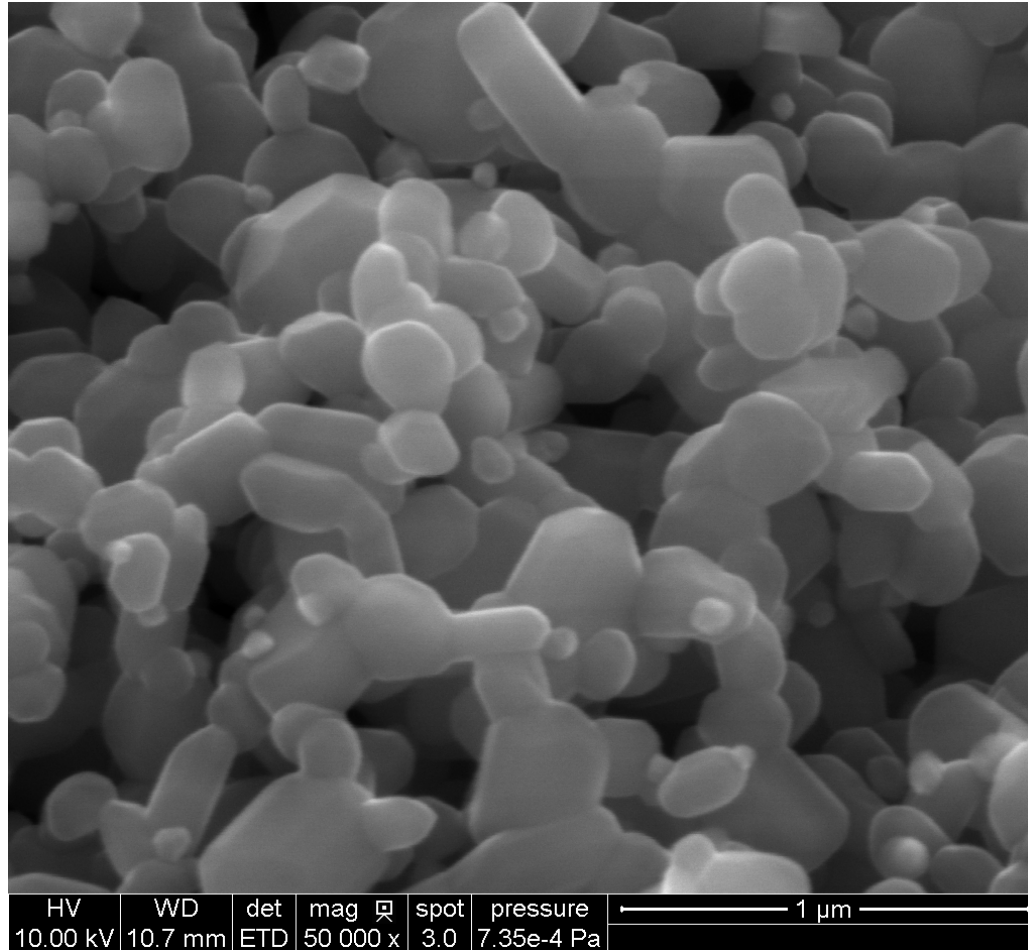
- Fluffy & often electrostatically charged.
- Lacy agglomeration – easily can flatten with spatula.
- Difficult to pack evenly.
- Traces of chlorine in EDS/SEM analysis ~0.2%
- We always handle this powder in dry N<sub>2</sub> in glove box or transport it in Mason jars.



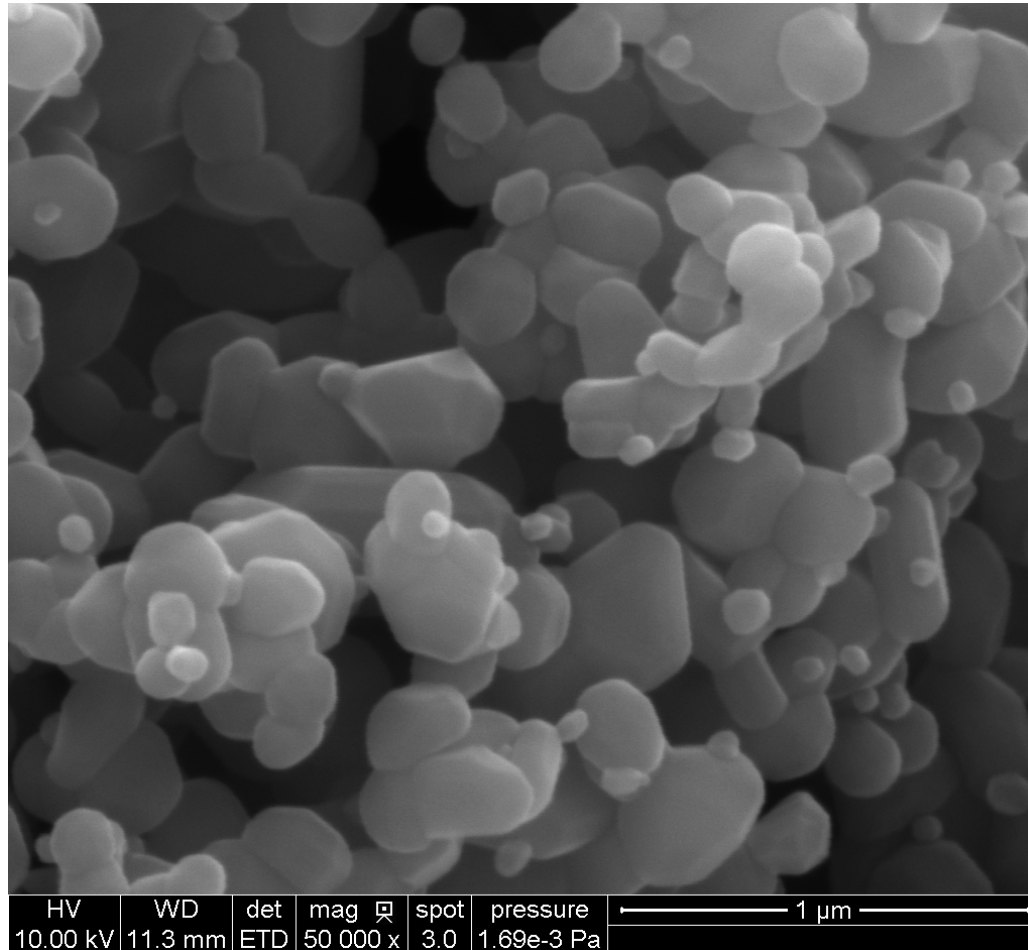
# Convert B powder to $\text{MgB}_2$ powder

- Use 3 mm size Mg chunks
- Place Mg & B in Mason jar under  $\text{N}_2$  gas and mix by rolling for 1 min. [50 g batch]
- Transfer mixed powders to Nb reaction vessel 150 mm long by 28 mm OD by 1 mm wall.
- Evacuate the vessel and e-beam weld shut.
- React Nb vessel under Ar gas at  $830^\circ\text{C}$  for 24 h in horizontal furnace. Open under  $\text{N}_2$  in glove box. Temperature & Time can be adjusted.

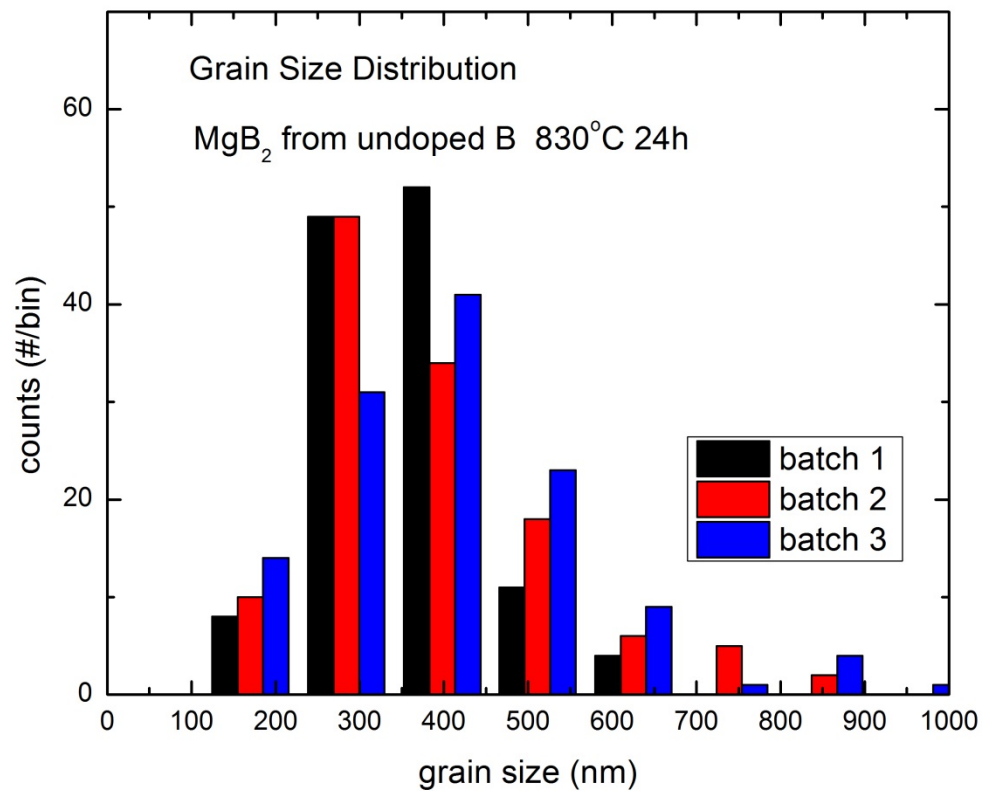
# SEM of $\text{MgB}_2$ Powder 830°C, 24 h



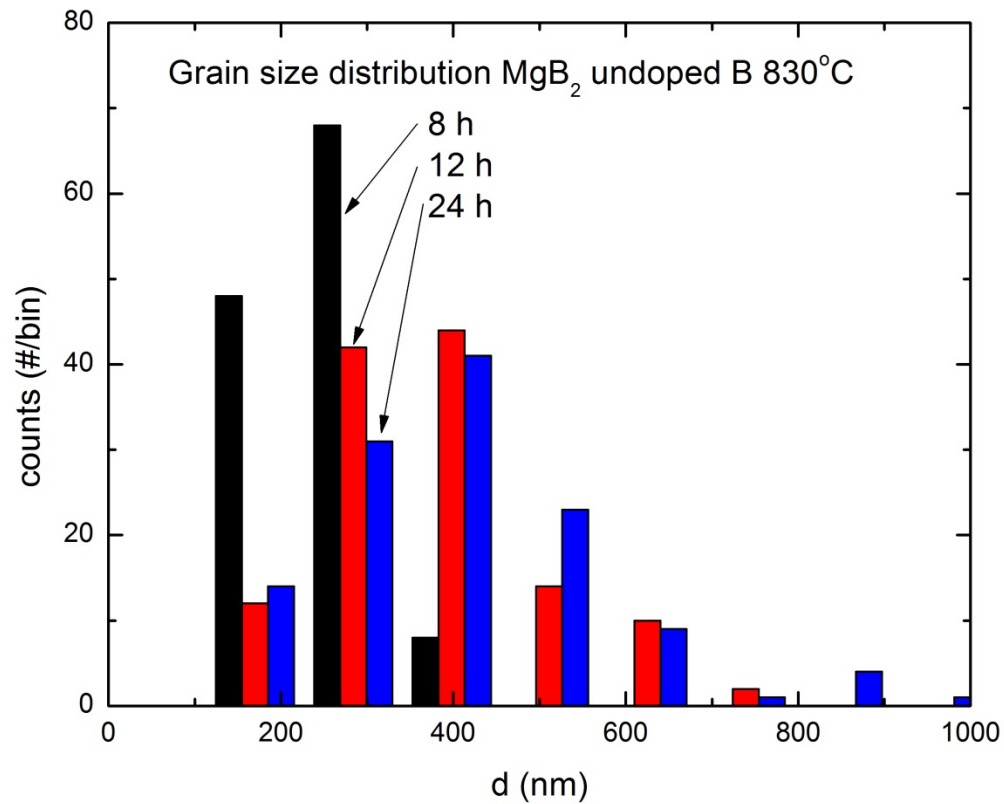
# SEM of $\text{MgB}_2$ Powder 830°C 8h



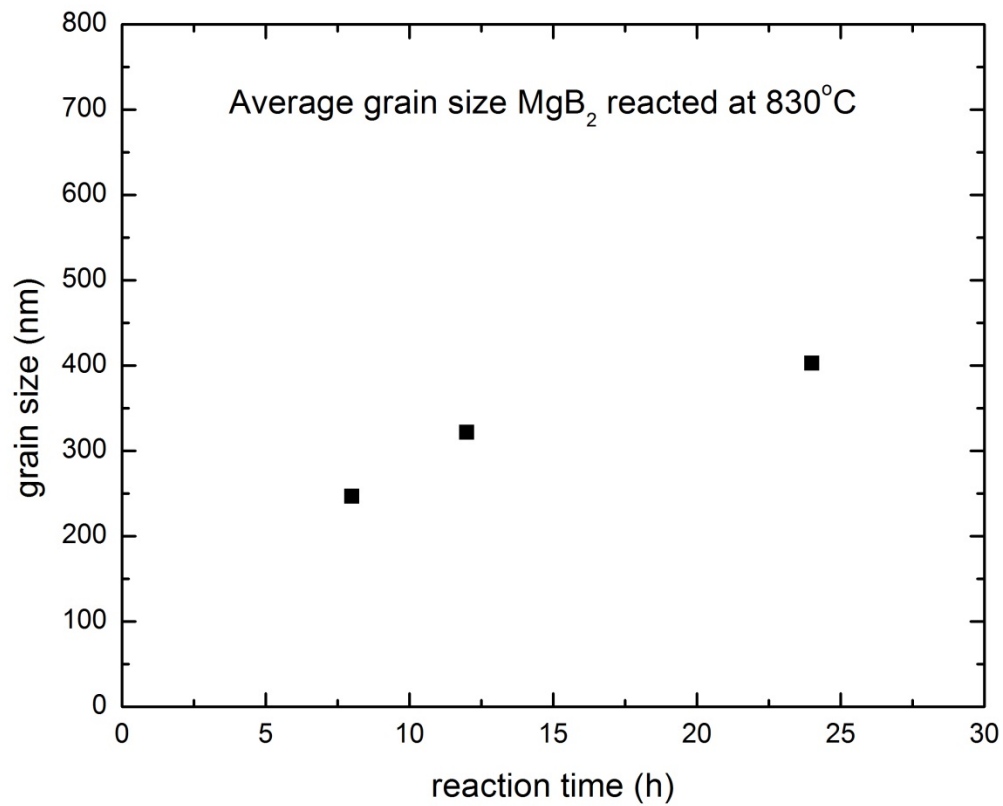
# Grain size for 3 batches 830°C – 24 h



# Grain size at 830°C for 8h, 12h, 24h

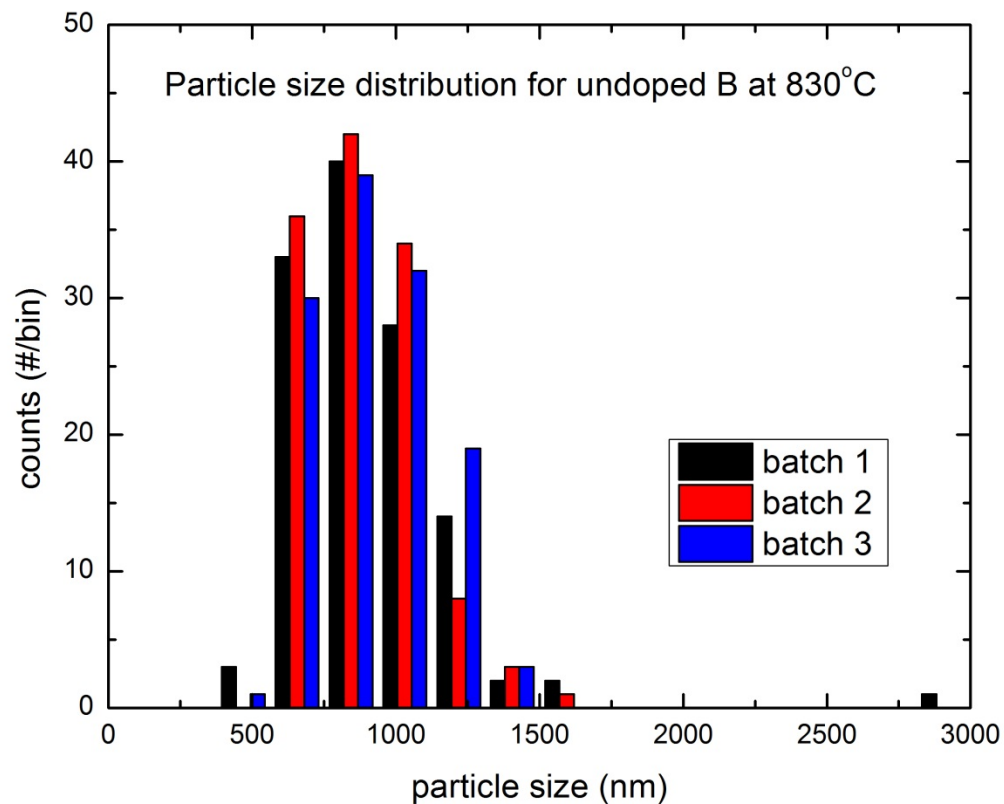


# Grain size at 830°C for 24h, 12h & 8h

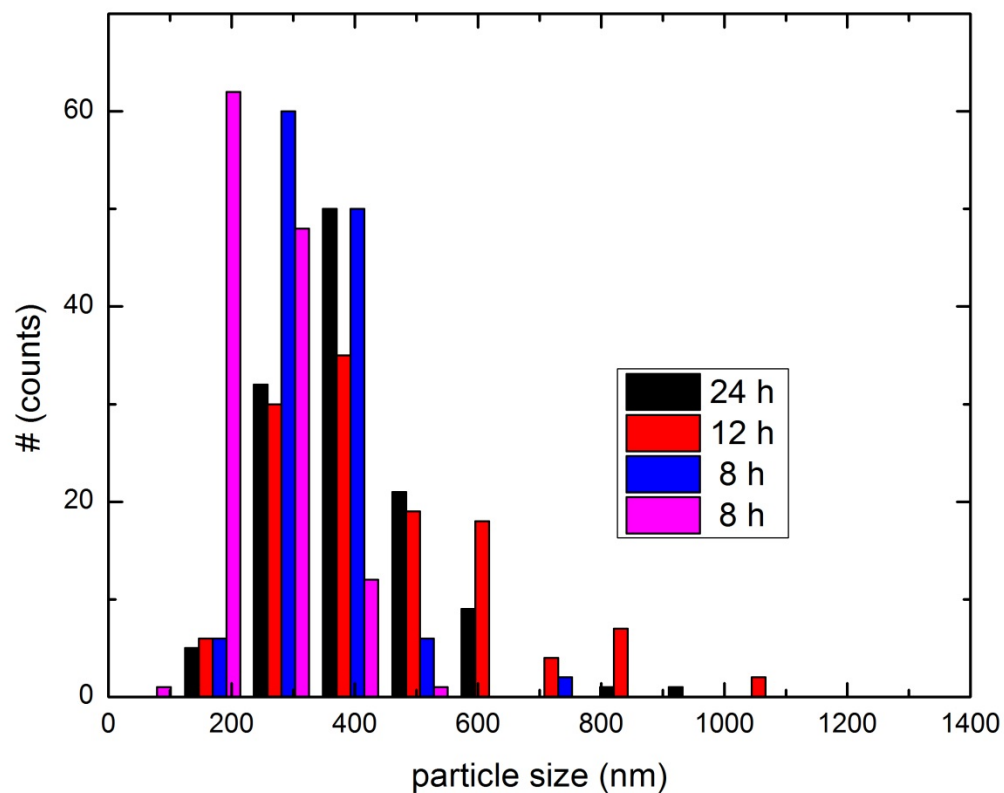




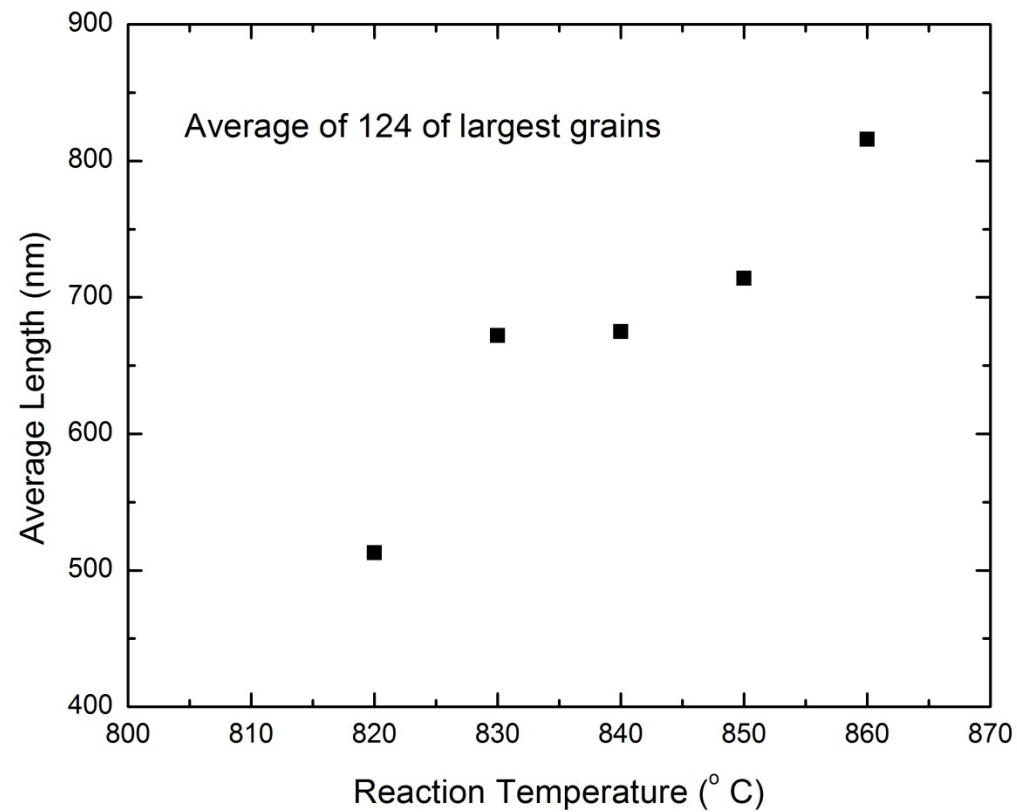
Particles may have several grains  
Look at particle size



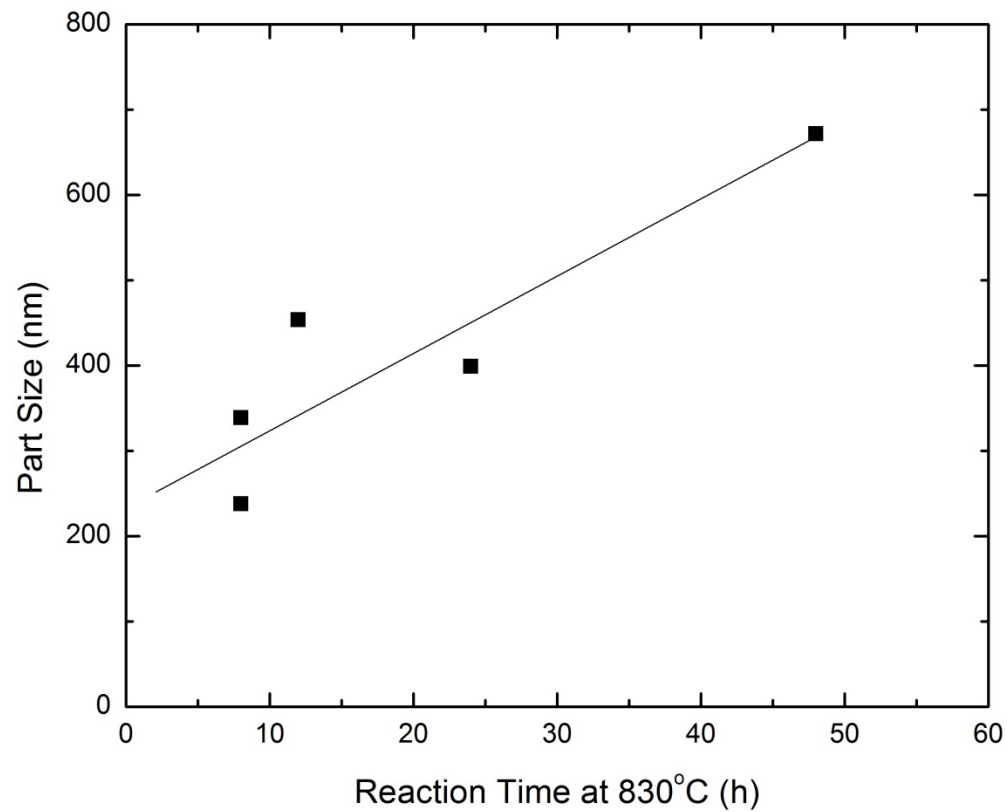
# Change in particle size 830°C for 24h, 12h & 8h



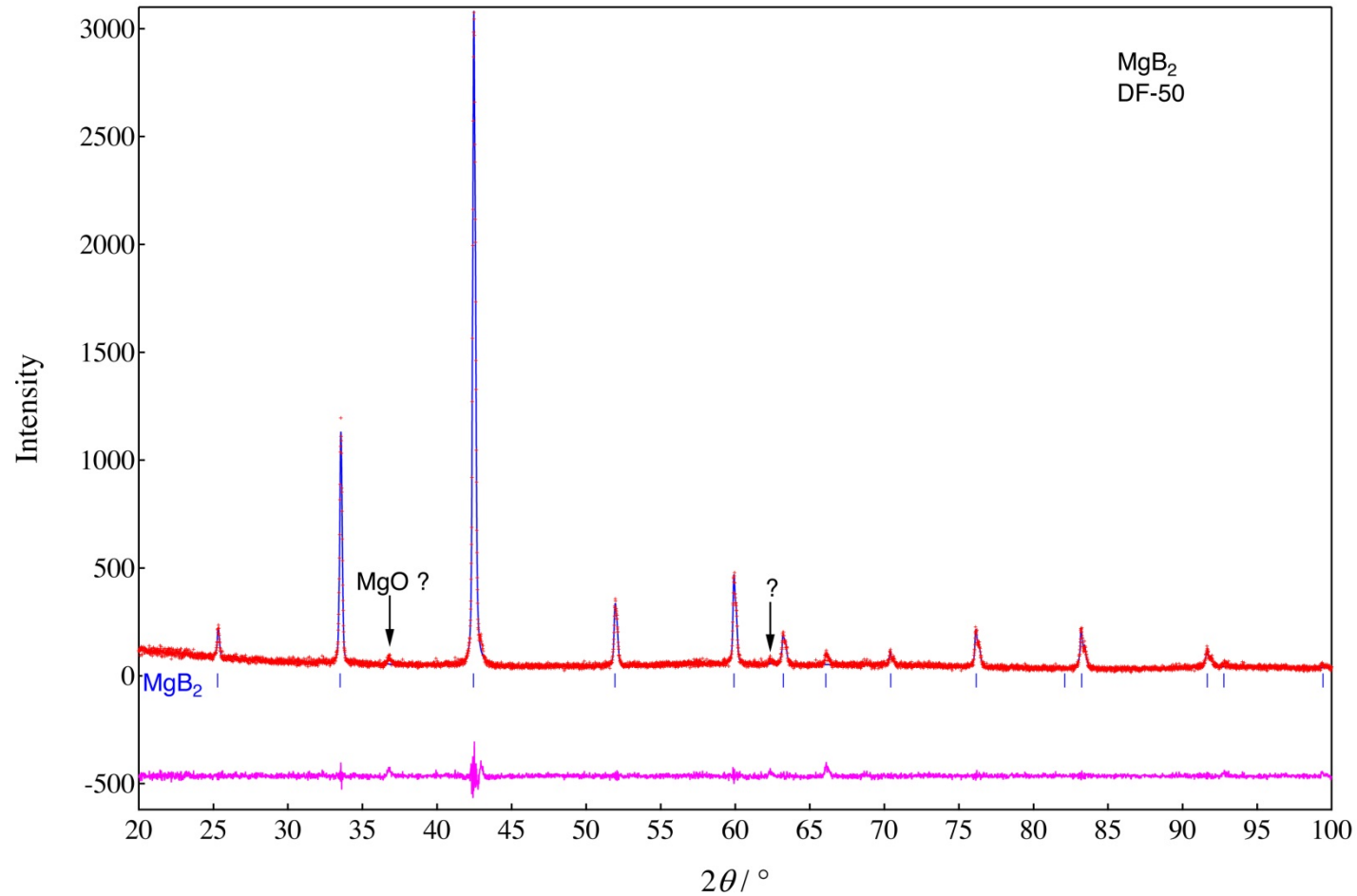
# Average particle size for 48 h



# Average particle size vs time 830°C



# X-ray spectrum for $\text{MgB}_2$ powder



# Conclusions

- 1) 830°C for 24h gives particle size under a micron.
- 2) average grain sizes tend to run about  $\frac{1}{2}$  the average particle size
- 3) reducing the temperature to 800°C and/or the time to 8h substantially reduces particle and grain size.



# Errors of Judgment

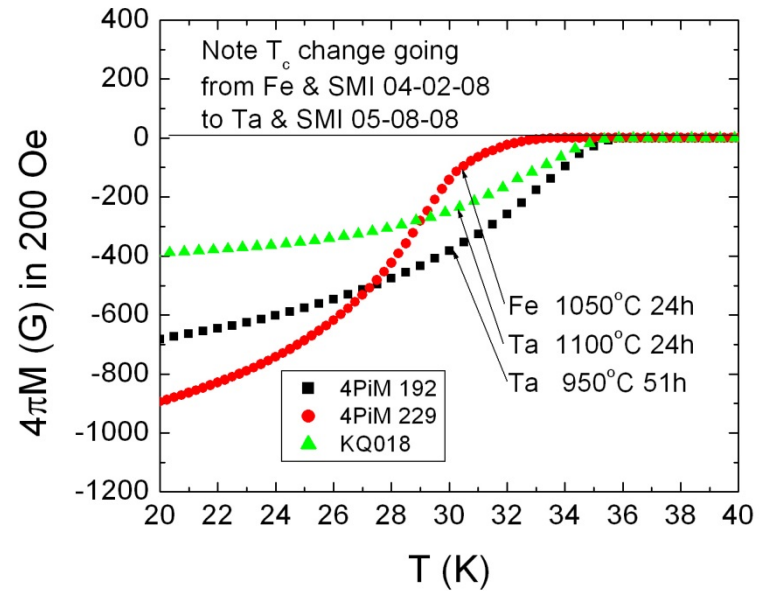
## Things we did wrong

- 1) Use an Fe reaction vessel
- 2) Seal the Nb reaction vessel under Ar and use arc welding.

# Mistakes that we made.

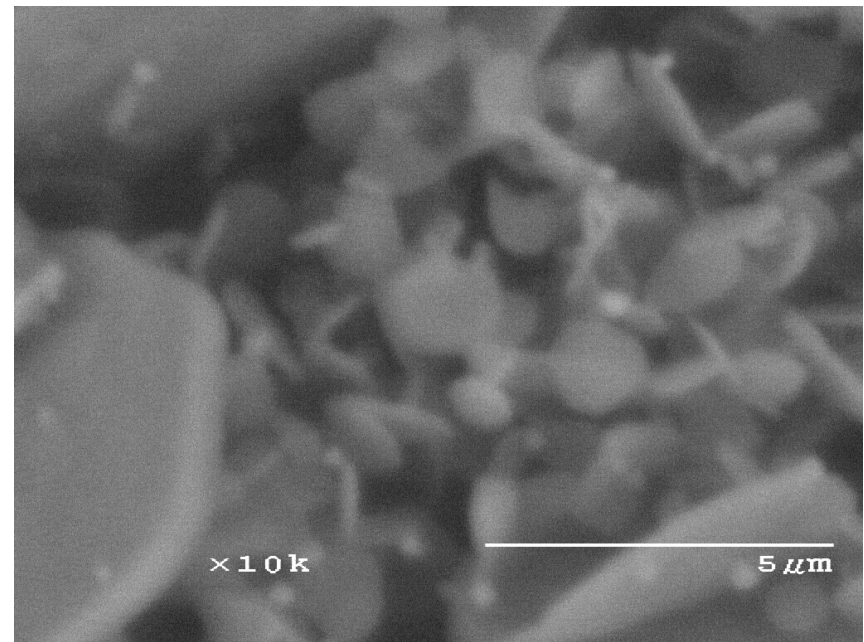
## Problem #1 – Fe depresses $T_c$

- Fe reaction vessel suppresses  $T_c$  about three degrees Kelvin.
- Nb give the same result as Ta so we used Nb for cost reasons.



## Problem # 2, Fe promotes a platy growth, often $10\ \mu\text{m}$ across

- In Fe reaction vessel we saw many large hexagonal plates, many in the 5 micrometers to 10 micrometers across



## Problem #3, must evacuate the Nb reaction vessel

- We initially sealed the Nb reaction vessels under  $\frac{3}{4}$  atmospheres of Ar which gives about 2 atmospheres of Ar at 830°C.
- Reaction is slow and not reproducible
- The Ar impedes the diffusion of Mg vapor through the packed boron powder.
- The Ar slows the conversion of boron to  $\text{MgB}_2$