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Correlation of critical current density of bulk MgB₂ having high connectivity between SC grains with concentration and distribution of higher magnesium borides and oxygen-enriched inhomogeneities, the role of Ti additions

T. Prikhna ¹, M. Eisterer ², W. Goldacker ³, V. Sokolovsky ⁴,

W. Gawalek ⁵, H.W. Weber ³, A. Kozyrev ¹, V. Moshchil ¹

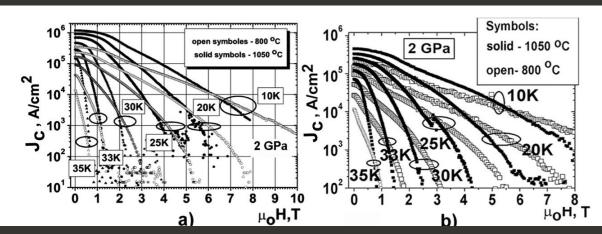
- ¹ Institute for Superhard Materials of the National Academy of Sciences of Ukraine, Kiev 04074, Ukraine
- ² Atominstitut, Vienna University of Technology, Stadionallee 2, 1020 Wien, Austria
- ³ Karlsruhe Institute of Technology (KIT), 76344 Eggenstein, Germany
- ⁴ Ben-Gurion University of the Negev, P.O.B. 653, Beer-Sheva 84105 Israel
- ⁵ Magnetworld AG, Buchaer Straße 6, 07745 Jena, Germany

Introduction

Motivation

Critical current density, j_c, vs. magnetic field, µ_oH, of MgB₂ materials synthesized at 2 GPa for 1 h from Mg+2B at 800 and 1050 °C;

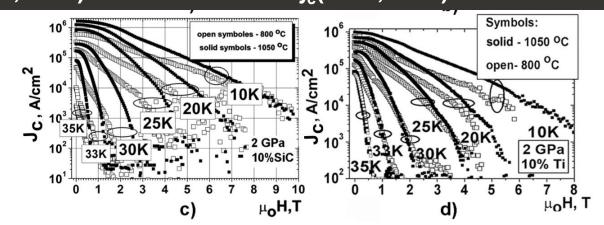
From B (1) with grain size <5 µm, 0.66 wt% O, 0.31 wt% C, 0.48 wt% N, 0.32 wt % H



From B (2) with grain size 4 µm, 1.5 wt% O, 0.47 wt% C, 0.40 wt% N, 0.37 wt% H

 $j_c(0-1 \text{ T}, 20 \text{ K})=0.9-0.7 \text{ MA/cm}^2$ $j_c(0-1 \text{ T}, 20 \text{ K})=0.4-0.3 \text{ MA/cm}^2$

From B(1)
with
10 wt% SiC

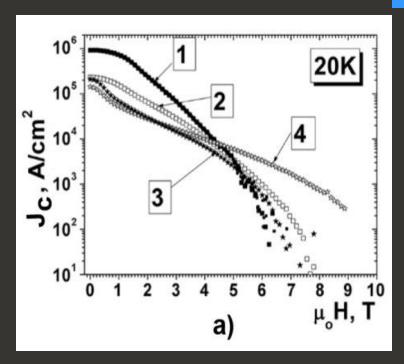


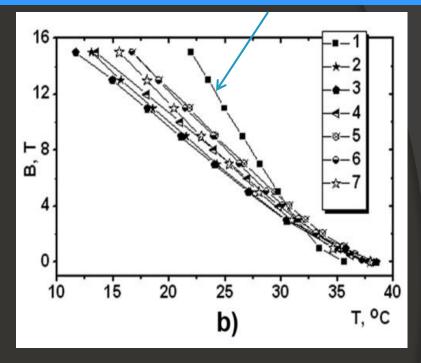
From B(2) with

 $j_c(0-1 \text{ T}, 20 \text{ K})=1.2 - 1.0 \text{ MA/cm}^2$ $j_c(0-1 \text{ T}, 20 \text{ K})=0.7-0.5 \text{ MA/cm}^2$

Introduction

 B_{c2} (22 K)=15 T (No 1) and extrapolated B_{c2} (0 K)=42.1 T, as well, as irreversibility fields: B_{irr} (18.5 K)=15 T and B_{irr} (0 K)=32.5 T





- (a) Comparison of J_c at 20 K in bulk MgB₂ prepared under various temperatures at 2 GPa:
- 1, 2: samples made of B(I) at 1050 °C, (k=0.43) and 800 °C, (k=0.37), respectively;
- 3, 4 of B(II) at 800 °C and 600 °C, (k=0.32), respectively.

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(b) B_{c2} of bulk MgB<sub>2</sub> prepared from Mg:2B: 1:B(II) at 2 GPa, 600 °C, 1 h, (HP); 2:B(III) at 30 MPa, 800 °C, 2h, (HotP); 3: B(III) at 50 MPa, 600-1050 °C, 1 h, (SPS); 4: B(III) at 2 GPa, 900 °C, 1 h (HP); 5:B(V)+10 wt.% Zr at 2 GPa, 800 °C, 1 h (HP); 6:B(V)+10 wt.% Ti at 2 GPa, 800 °C, 1 h (HP); 7:B(I)+10 wt.% SiC at 2 GPa, 1050 °C, 1 h (HP).
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Introduction

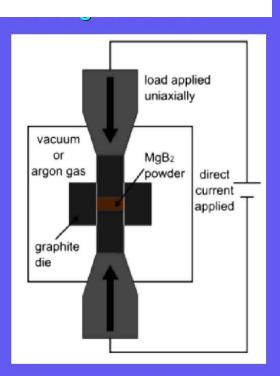
Pinning and j_c in MgB₂ can be influenced by:

- Material density
- Sizes of MgB₂ grains (due to intergrain boundaries can be pinning centers and higher surface area of grain boundaries is preferable)
- Connectivity between SC grains
- Shielding fraction (correlates with amount of SC phase)
- ▶ Presence of inclusions of secondary phases with different (from MgB₂) Tc or nonsuperconductive (sizes of pinning centers should be commensurable with coherence length in MgB₂)
- Dopants which can be introduced into MgB₂ lattice and cause lattice distortion (carbon)

Methods of preparation

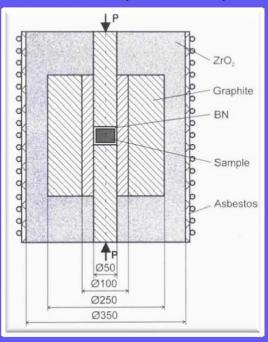
SPS = Spark plasma sintering 16-96 MPa





Scheme of SPS-system

Hot - P = Hot-pressing30 MPa (300 atm)



Possibility to press blocks up to 200 mm in diameter

(in cooperation with Dr. J. Noudem, CRISMAT, Caen, France)

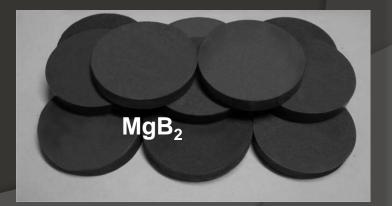
Methods of preparation



High pressure – high temperature 2 GPa, 1050 °C, 1 h

MgB₂ items up to 150 mm in diameter

14 000-ton press 140 MN effort

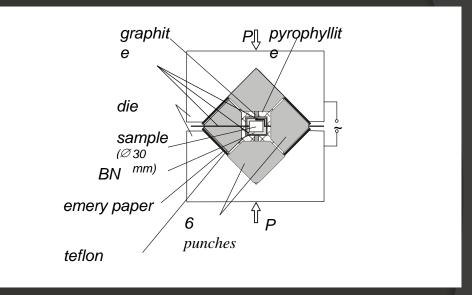


Methods of preparation Cube type high pressure apparatus





The scheme of reaction cell of the cube type high pressure apparatus.



Scheme of load distribution

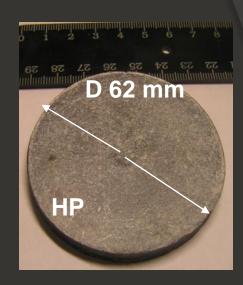


Methods of preparation

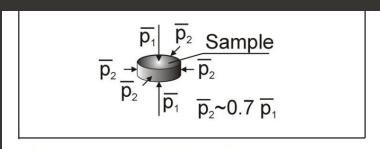
The recessed-anvil type high pressure apparatus with working volume 330 cm³ (under 2 GPa pressure)



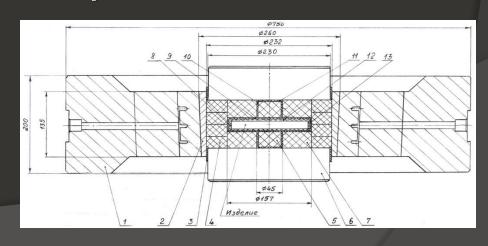




HPA for manufacturing blocks up to 150 mm in diameter



Recessed-anvil-type high-pressure appatatus (HPA)



HPA cylinder-piston(V_p = 6300 cm³) P=1-2 GPa, T=600-1100 °C, Press ASEA 140 MN effort



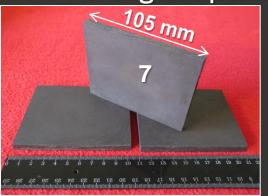
40 mm x 30 mm

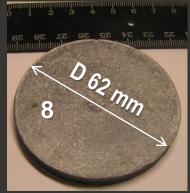
Items from MgB₂

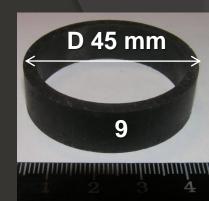
MgB₂-based superconductors

working temperature 20-30 K

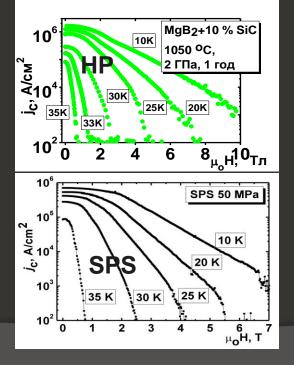


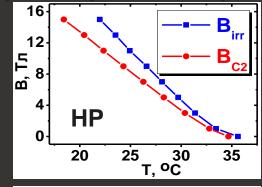


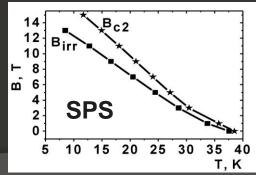


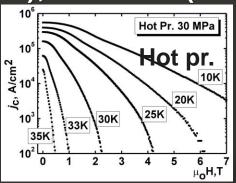


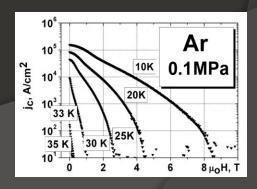
1-4, 8, 9 - High pressed (2 GPa); 5 – HIP (0.1 GPa); 7- Hot Pr. (30 MPa)



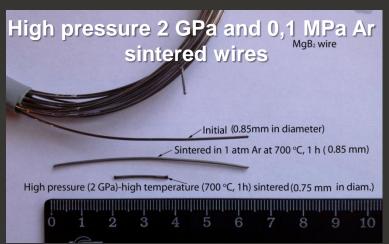


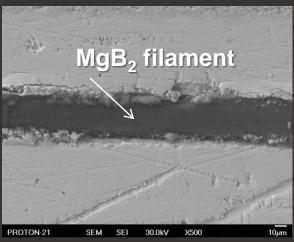






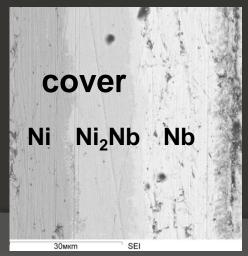
High-pressure (2 GPa, 800 °C, 1 h) treated wire from MgB₂

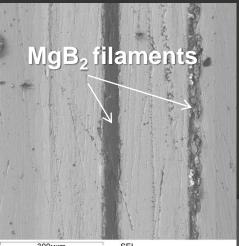


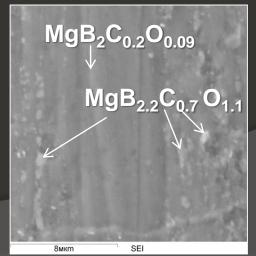


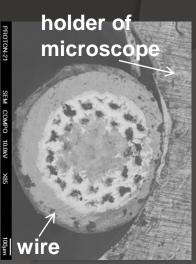


Panorama of MgB₂ filament after high pressure-high temperature treatment









Estimation of MgB₂ density

Table 2. Characteristics (critical current density- j_c , maximal pinning force- $F_{p(max)}$, amount of MgB₂, MgO and MgB₄ estimated from Rietveld refinements, and density- ρ) of MgB₂-based materials prepared under different P-T conditions from Mg and B taken in MgB₂ stoichiometry or from MgB₂ powder using different methods: under high quasihydro-static pressure –HP, by hot pressing HotP, by spark plasma sintering – SPS, by pressureless sintering under 0.1 MPa of Ar – PL.

No	Preparation	Туре	P, MF	Ра <i>Т</i> ,	j₀, MA/cn	n^2 , $F_{p(max)}$ /	MgB ₂ /MgC)/ Den-
		of B		° C	at 0-1 T,	10 ⁹ (N/m ³)) MgB₄,	sity, ρ
					at 20 K	at 20 K	wt%	
1.	in-situ, HP		2000	1050	0.9 - 0.7	7.6	94/6/0	99%
2.	in-situ, HP	1	2000	800	0.2 - 0.15	1.6	91/5.5/0	98%
3.	in-situ, HP	Ш	2000	1050	0.4 - 0.3	2.3	87/13/0	99%
4.	in-situ, HP	Ш	2000	1000	0.36-0.23	2.4	86/14/0	99%
5.	in-situ, HP	Ш	2000	950	0.3 - 0.2	2.1	-	98%
6.	in-situ, HP	Ш	2000	900	0.12 - 0.06	1.1	71/13/0	98%
7.	in-situ. HP	Ш	2000	800	0.12 - 0.07	8.0	73/12/0	97%
8.	in-situ, HP	Ш	2000	600	0.14 - 0.05	0.6	64/30/0	83%
9.	in-situ, SPS	Ш	50	1050	0.5 - 0.4	4.6	83/4.5/12.5	94%
10.	in-situ, SPS	Ш	50	800	0.4 - 0.36	2.7	-	74%
11.	ex-situ, SPS	-	50	1050	0.4 - 0.3	3.3	83/6.5/10.5	96%
12.	in-situ, HotP	Ш	30	1050	0.08 - 0.016	0.2	46/8.5/45.5	99%
13.	in-situ, HotP	Ш	30	800	0.3-0.2	0.6	-	72%
14.	in-situ, PL	IV	0.1	800	0.08-0.03	1.9	90/10/0	55%
15.	in-situ,	Ш	0.1	600	0.26-0.13	1.3	94.5/5.5/0	65%
	cold dens. 2 GP	a,						
	PL							

Estimation of grain sizes in MgB₂

The average crystallite sizes were calculated from line broadening in X-ray diffraction pattern by the standard program in accordance with the following:

Crystallite size =
$$\frac{K \cdot \lambda}{W_{\rm size} \cdot \cos \theta}$$
 with $W_{\rm size} = W_{\rm b} - W_{\rm s}$

where W_{size} is the broadening caused by small crystallites; W_{b} is the broadened profile width; W_{s} is the standard profile width (0.08°); K is the form factor; λ is the wavelength.

Table. Critical current density, j_c , vs. relative average grain size of crystallites of high-pressure sintered from MgB₂ and synthesized from Mg and B taken in 1:2 ratio materials

High pressed	average crystal	lattice pa	rameters		Vcm² 0 K	<i>j</i> _c , kA at 2	
at 2 GPa	sizes	<i>a</i> (nm)	c (nm)	0 T	1 T	0 T	1 T
for 1 h							
at T _s , °C		MaiD (40 -		0.0/ -			
		MgB₂ (10 μ	1277 - 27 1	8 % 0	(0)		
700	19.7 nm	0.30805	0.35188	-	-	-	-
800	18.8 nm	0.30822	0.35212	-	-	-	-
900	18.5 nm	0.30820	0.35208	56	14	36	8
1000	25.0 nm	0.30797	0.35200	28	8	19	5
From Mg cl	hips and B	(4 μm, 1.5	% O) mix	ed an	d mille	ed in 1:2	2 ratio
800	15.0 nm	0.30747	0.35188	245	142	138	79
900	21.0 nm	0.30819	0.35174	205	136	128	61
1000	37.0 nm	0.30808	0.35192	485	364	360	237

Estimation of connectivity, $A_{\rm F}$

$$A_{\mathbf{F}} = \frac{9 \,\mu\Omega \cdot \mathrm{cm}}{\rho_{300} - \rho_{40}}$$

$$9 \mu\Omega \cdot cm$$

perfect connectivity for MgB₂

$$\rho = \frac{V \cdot \sigma}{I \cdot \Delta l}$$

resistivity
$$\rho = \frac{V \cdot \sigma}{I \cdot \Delta l} \quad \rho_{300} = \frac{V_{300} \cdot \sigma}{I \cdot \Delta l} \quad \rho_{40} = \frac{V_{40} \cdot \sigma}{I \cdot \Delta l}$$

 Δl - distance between voltage contacts

 $\sigma = a \cdot b$ - cross-section area of the sample, $a \cdot b \cdot c$ - sample dimensions

 V_{300} - arithmetic average of all V data at room temperature

 V_{40} – voltage at 40 K

I = 100 mA - current

ρ - by four probe method

Estimation of pinning mechanism

M. Eisterer "Calculation of the volume pinning force in MgB₂ superconductors" (PHYSICAL REVIEW B 77, 144524 2008)

Identification of the dominant pinning mechanism from the peak position of pinning volume force:

$$K = B_{\text{peak}}/B_{\text{n}}$$
;
 $B_{\text{peak}} - F_{\text{p(max)}}$, $B_{\text{n}} - F_{\text{p(max)}}/2$

The dominant pinning mechanism was determined from the volume pinning force j_cB according to a scaling procedure proposed by M.Eisterer , which eliminates the influence of the intrinsic anisotropy of MgB_2 on the field dependence of the volume pinning force. The field B_{peak} , where the maximum of the volume pinning force occurs, was normalized by the field B_{peak} , at which the volume pinning force drops to half its maximum (on the high field side), instead of the upper critical field, B_{c2} , in the original approach for isotropic superconductors . For typical values for the anisotropy (4) and the percolation threshold (0.25) in bulk MgB_2 samples, the position of the peak, B_{peak}/B_n , is expected to be at 0.34 and 0.47 for grain boundary and point pinning, respectively. Changes in the anisotropy and the percolation threshold only slightly change the peak position, in particular in the realistic range of these parameters. Another advantage of this scaling approach is that B_n is easily available and no separate experiment for determining B_{c2} or any extrapolation to estimate the irreversibility (or Kramer) field is necessary to define the scaling field.

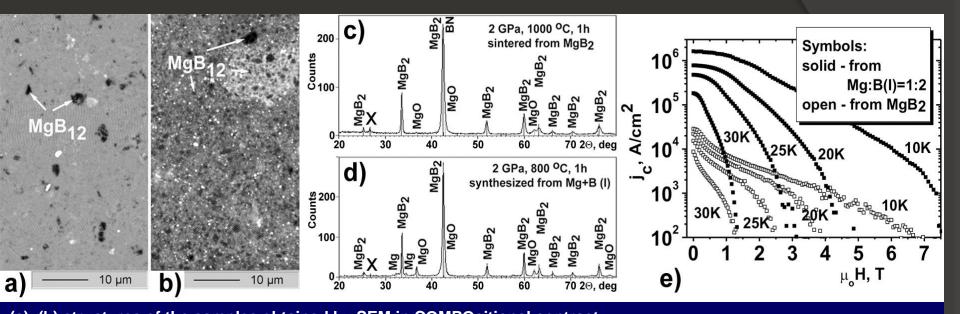
Estimation of connectivity, A_r, amount of shielding fraction, S, and type of pinning, k.

Connectivity, A_F , k-ratio: $k = B_{peak}/B_n$ and amount of shielding fraction, S, of MgB₂ materials prepared under different conditions. The numbering in tables 1 and 2 is the same.

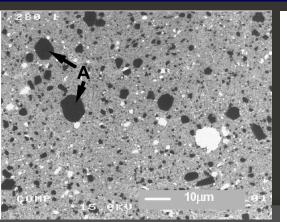
No	Preparation	Туре	P, MPa		k=	A _F , %	S, %
		of B		° C	B_{peak}/E	3 _n	
1.	in-situ, HP	I	2000	1050	0.43	-	_
2.	in-situ, HP	I	2000	800	0.37		
3.	in-situ, HP	III	2000	1050		79	94
5.	in-situ, HP	III	2000	950	-	59	81
7.	in-situ, HP	III	2000	800		57	91-100
8.	in-situ, HP	Ш	2000	600	0.32	18	90
9.	in-situ, SPS	III	50	1050		98	91
11.	ex-situ, SPS	-	30	1050	-	80	100
12.	in-situ, HotP	III	30	1050	-	32	-
13.	in-situ, HotP	Ш	30	800	-	73	
15.	in-situ,	III	0.1	600	-	-	75
	cold dens. 2 GPa	a, PL					
16.	in-situ,	III	0.1	800	-	-	84
	cold dens. 2 GPa	a, PL					

With synthesis temperature increase the shift from grain boundary pinning towards point pinning occurs (about what k-ration increase witnessed)

CONTRADICTIONS BETWEEN X-RAY DATA AND SEM OBSERVATIONS

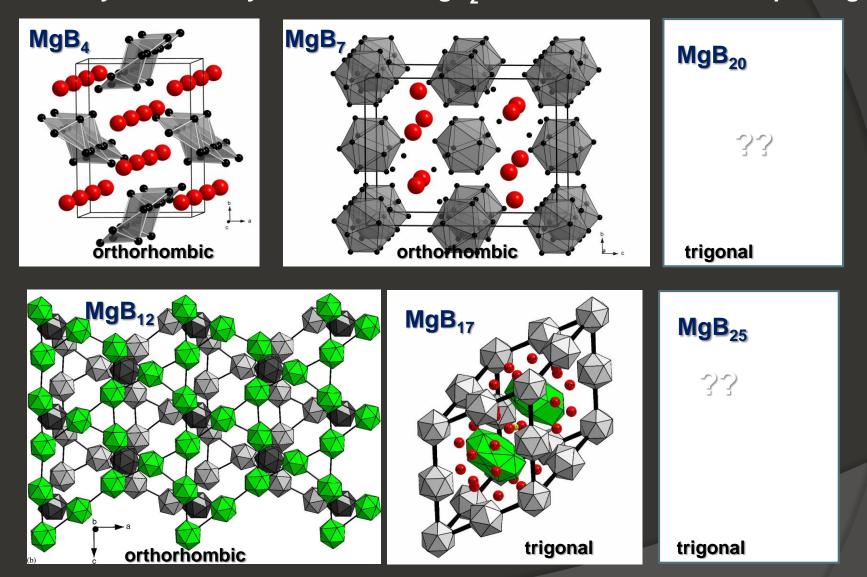


- (a), (b) structures of the samples obtained by SEM in COMPOsitional contrast:
- (a) sintered from MgB₂ at 2 GPa, 1000 °C, 1 h;
- (b) synthesized from Mg and B taken into 1:2 ratio at 2 GPa, 800 °C, 1 h;
- (c), (d) -X-ray patterns of the samples shown in Figs. 5a, b;
- (e) dependences of critical current density, j_c , on magnetic fields, μ oH, at different temperatures of the samples shown in Fig. e: open symbols sintered from MgB₂ material and solid symbols synthesized from Mg and B taken in 1:2 ratio



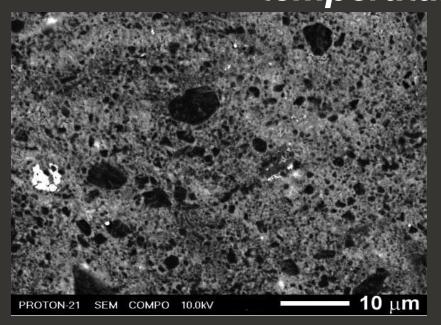
Structure of MgB₂-based material synthesized from amorphous boron and magnesium chips taken in Mg:B=1:2 ratio in contact with Ta foil at 2GPa, 800°C, 1h (BEI SEM image) shows that inclusions of black phase with near MgB₁₂ stoichiometry may crystallize in hexagonal habit or have near hexagonal cross-section (see, for example, inclusion marked "A")

Each of the following non-superconducting higher magnesium borides in the Mg-B system can crystallize in the MgB₂ matrix and can influence pinning



The method of preparation (pressure-temperature-time conditions) can affect the stoichiometry of the MgB_x inclusions in MgB_2 . (2 GPa – MgB_{12})

The size and amount of higher magnesium borides (black inclusions with near MgB₁₁₋₁₃O_{0.2-0.3} stoichiometry according to Auger study) decreasing with manufacturing temperature increase



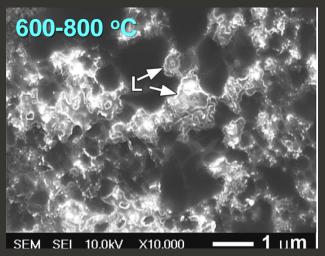
PROTON-21 SEM COMPO 10.0kV — 10 μm

Mg:2B, 2 GPa, 800 °C, 1 h

Mg:2B, 2 GPa, 1050 °C, 1 h

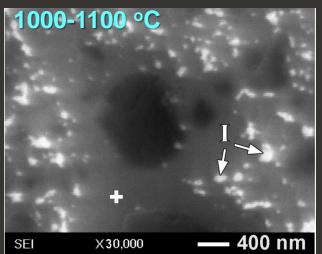
The Berkovich nanohardnes and Young modulus of such inclusions as estimated under the 10-60 mn-load were 32.2±1.7 and 385±14 GPa, respectively, what were even higher than that for sapphire: 31.1±2.0 and 416±22 GPa.

Typical structures of MgB₂ prepared at 0.1 MPa – 2 GPa at low temperatures (a) - 600-800 °C and high (b) -1000-1100 °C and their compositions

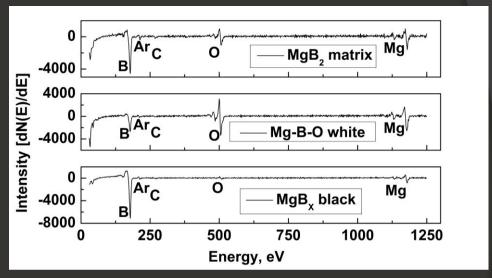


a)

b)



"L" - MgB_{0.9-3.5}O_{1.6-2} white layers, thickness (15-20 nm)



Auger spectrum

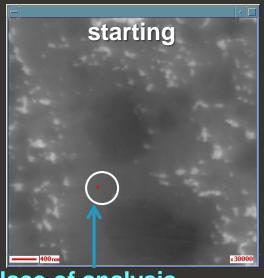
"I" - $MgB_{0.5-0.8}O_{0.8-0.9}$ white inclusions $MgB_{11-13}O_{0.2-0.5}$ black inclusions $MgB_{2.2-1.7}O_{0.3-0.6}$ gray matrix.

Auger study by JAMP-9500F

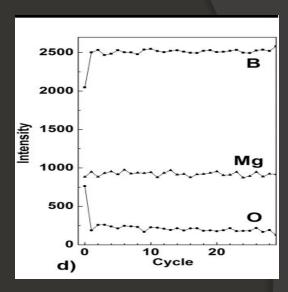
Zone of excitation being 10 nm in diameter and two lattice parameters in depth etching in Ar allows to exclude analyzing of oxidized layer on the surface!

Auger study by JAMP-9500F

zone of excitation being 10 nm in diameter and two lattice parameters in depth Mg(I):2B(III), 1200 °C, 2 GPa, 1h







Place of analysis

 $MgB_{0.5-0.8}O_{0.8-0.9}$ white inclusions $MgB_{11-13}O_{0.2-0.5}$ black inclusions $MgB_{2.2-1.7}O_{0.3-0.6}$ gray matrix.

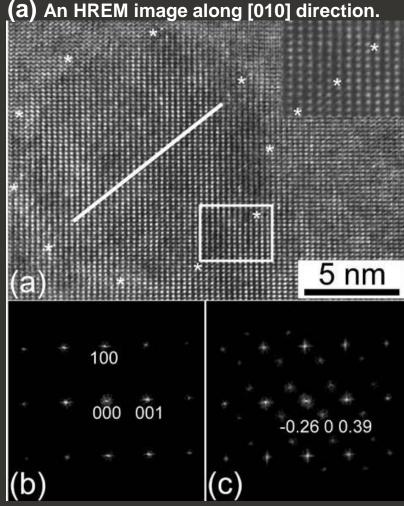
Spectrums of gray matrix

Depth profile of gray matrix

Comparing these results with that of X-ray structure analysis we can conclude that boron incorporated into the MgB₂ structure.

Repeated etching in Ar (in the JAMP-9500F chamber during study) and quantitative Auger analysis showed practically the same oxygen concentration in MgB₂ matrix of the structure after 30 cycles, which points to the oxygen incorporation into the MgB₂ lattice.

Mg(B,O)₂ precipitation in MgB₂ (X. Z. Liao et al. J. Appl. Phys. 93, 6208 (2003)



(b) Fourier transformation of the matrix.

(c) Fourier transformation of the precipitate showing satellite spots with a vector of [-0.26, 0, 0.39] MgB_2 grains to form coherently ordered $MgB_{2-x}O_x$ precipitates from about 5 up to 100 nm in size and that such precipitates can act as pinning centers

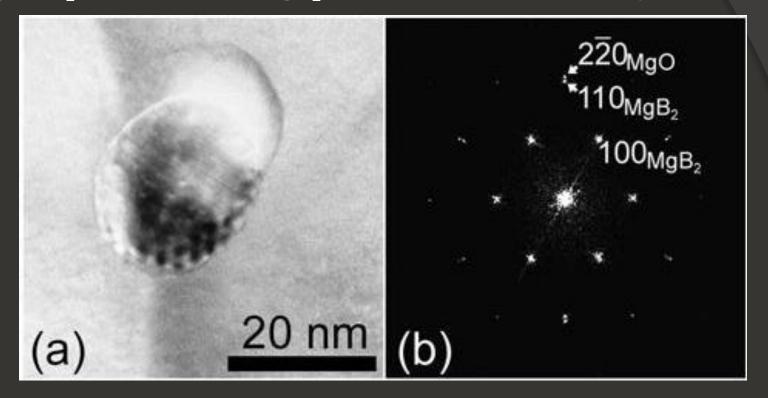
Ordered replacement of B atoms by O and have the same basic structure as the MgB₂ matrix but with composition modulations.

No difference in the lattice parameters between the precipitates and the matrix can be detected in conventional electron diffraction patterns.

However, extra satellite diffraction spots were revealed by HREM study in some directions implying a structural modulation caused by the precipitates.

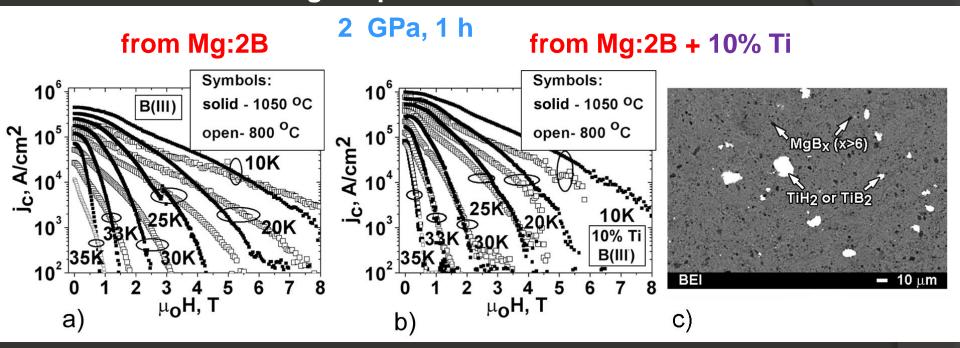
The periodicity of O atom ordering depends on the concentration of O atoms in the precipitate and primarily occurred in the (010) plane.

Mg(B,O)₂ precipitation in MgB₂ (X. Z. Liao et al. J. Appl. Phys. 93, 6208 (2003)



Longer exposure to trace amount of O at high temperatures also results in the transformation of the precipitates from the hexagonal $Mg(B,O)_2$ phase with compositional modulations to the face-centered cubic MgO phase, implying $Mg(B,O)_2$ is not a stable phase at high temperature in an environment containing trace amount of oxygen. The orientation relationship between the resulting MgO precipitates and the MgB_2 matrix revealed by an HREM image and its Fourier transformation in Fig. is: $[001]MgB_2$ //[111]MgO, and $(110)MgB_2$ //(1-10)MgO (note that planar distance $d(110)MgB_2 = 0.154$ nm while d(220)MgO = 0.149 nm. The lattice mismatch between $d(110)MgB_2$ and d(220)MgO is only 3.2%).

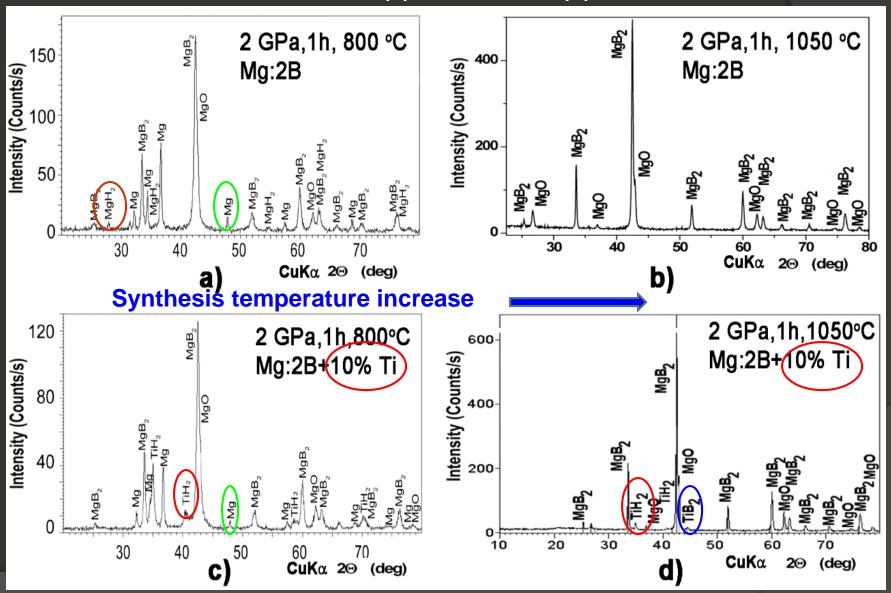
Increase of j_c in low and medium magnetic fields with increase of manufacturing temperature and effect of Ti addition



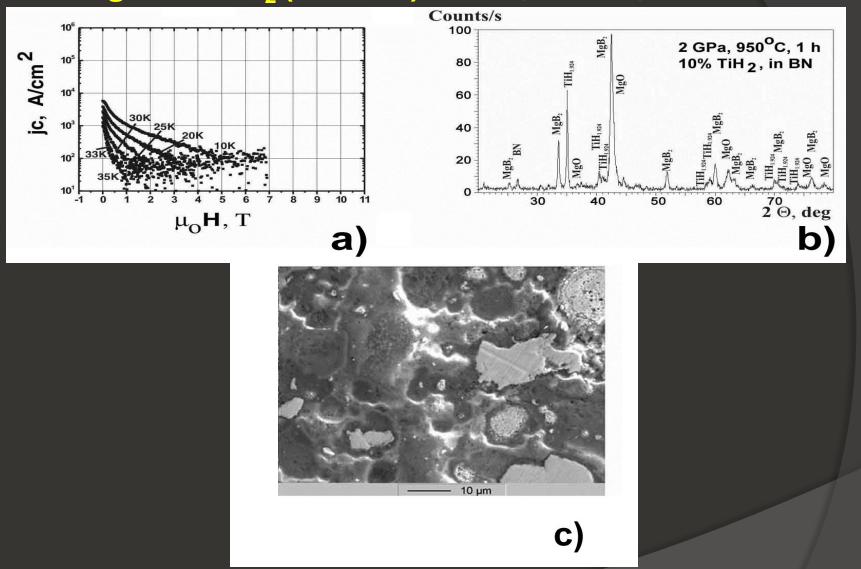
Boron amorphous: grain size 4 μ m, 1.5 wt% O, 0.47 wt% C, 0.40 wt% N, 0.37 wt% H

- The critical current density of MgB_2 based materials usually increases at low and medium magnetic fields with increasing manufacturing temperature, but lower manufacturing temperatures can result in a j_c increase at high magnetic fields.
- >A shift from grain boundary pinning to point pinning was observed
- \triangleright A positive effect in enhancing the critical currents of MgB₂ is played,, by dopants, such as Ti, for example.

X-ray diffraction patterns of the materials prepared from Mg:2B under 2 GPa for 1 h without additions at 800 °C (a) and 1050 °C (b) and with 10 wt. % Ti at 800 °C (c) and 1050 °C (d).

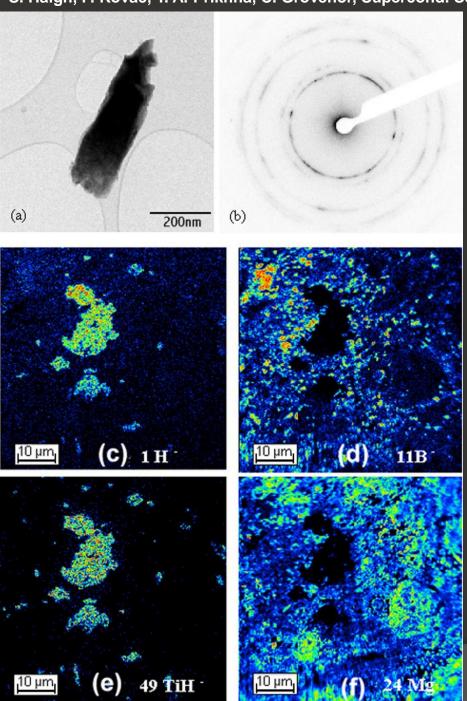


Mg:2B + TiH₂ (10 wt.%) 2 GPa, 950 °C, 1 h



- (a) Dependences of J_c on μ_o H, of the high-pressure synthesized MgB₂ –based materials from Mg:B=1:2 + 10 wt % TiH₂; at 2 GPa, 950 °C, 1 h; (b) X-ray pattern of the sample;
- (c) structure of the sample obtained by SEM (c) SEI (secondary electron image)

S. Haigh, P. Kovac, T. A. Prikhna, C. Grovenor, Supercond. Sci. Technol. 18, 1190 (2005).



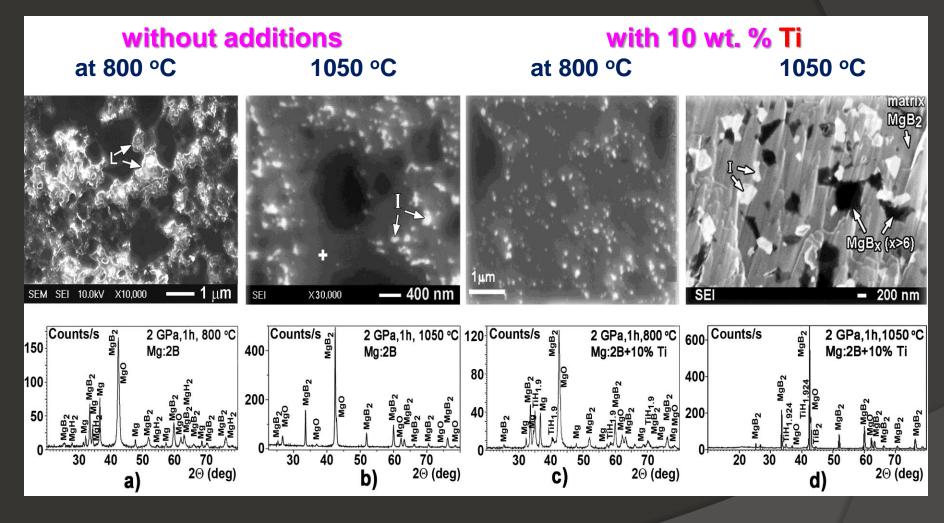
The sizes of the Ti-containing inclusions are rather big to be pinning centers by itself and randomly distributed in MgB₂

Enthalpies of formation values for a range of common Ti compounds

	Enthalpy of formation (kJ mol ⁻¹)
Ti ₃ O ₅	-2459.4
Ti_2O_3	-1520.9
TiO_2	-944.057
TiN	-336.6
TiB ₂	−150 to −314
TiH ₂	-15.0

- a) Bright–field TEM image of a particle from a powdered sample
- b) Electron diffraction pattern (contrast inverted) from the Ti-rich particle shown in (a)
- (c-f) NanoSIMS ion maps of the distribution of (c) mass 1 H−, (b) mass 11 B−, (d) mass 49 TiH₂−, (e) mass 24 Mg− (f)

Evolution of MgB₂ structure with synthesis temperature increase at 2 GPa, 1 h without and with Ti additions



Mg-B-O nanolayers (L)



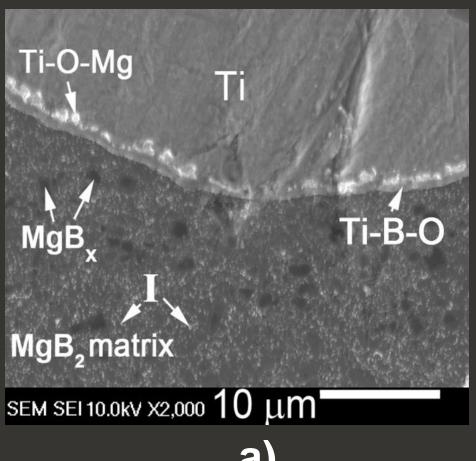
Mg-B-O inclusions (I) the most white

Mg-B-O inclusions

Mg-B-O inclusions

Modeling experiment

SEM images SEI (a) and BEI (b) of the synthesized at 2 GPa for 1 h from Mg:2B with (10 wt. %, coarse grains) under 2 GPa for 1 h at 800 °C



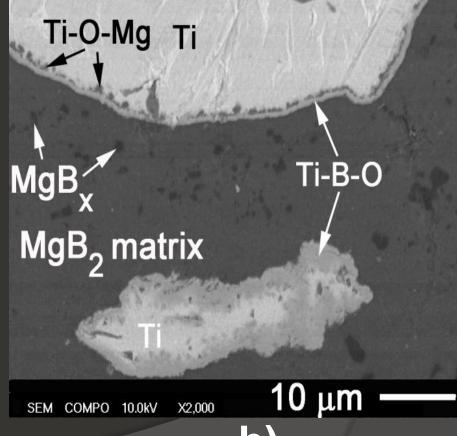
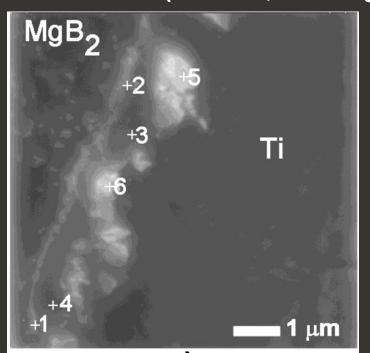


Table 1. Results of quantitative Auger analysis (atomic %) made in points marked 1-6 in Fig. 2g and located at the boundary

SEM images SEI of the synthesized at 2 GPa for 1 h from Mg:2B with Ti (10 wt. %, coarse grains) under 2 GPa for 1 h at 800 °C

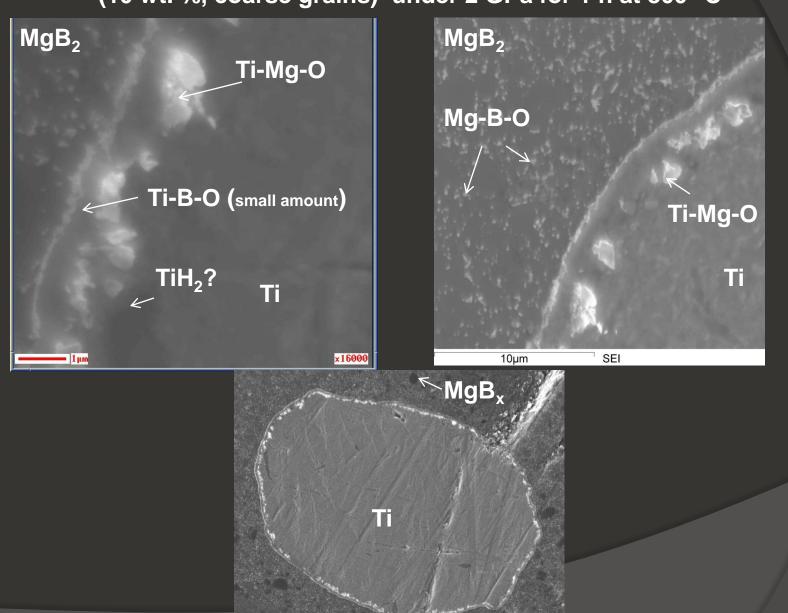


Results of quantitative Auger analysis (atomic %) made in points marked 1-6 and located at the boundary between MgB₂ and big Ti grain of the sample prepared at 2 GPa, 800 °C, 1 h. The sample was etched in Ar in the JAMP-9500F chamber before the study.

Element/ point No	В	Ti	O	Mg	C
1	22.2	10.5	33.1	22.6	11.7
2	24.5	10.8	33.1	21.2	10.3
3	41.4	31.4	8.2	3.4	15.5
4	41.3	28.2	9.1	3.2	17.7
5	4.1	13,9	44.9	29.1	8.0
6	6.5	10.1	44.8	29.1	9.4

Magnesium and oxygen enriched inclusions are observed inside the Ti grain, which were formed as a result of the Mg and O diffusion. The layer of Mg-B-O with somewhat smaller amount of oxygen (Nos 1, 2) then in inclusions (Nos. 5, 6) are well observed near to the grain boundary between the grain contained Ti. Magnesium defused into titanium more intensively than boron (compare Nos 3, 4 with Nos 5, 6, Table). So, the redistribution of boron and oxygen in MgB₂ can be explained by a higher affinity of Ti to Mg and O than that to B. Because of this Mg and O diffuse deeper to the Ti grain then B). We should not exclude the presence of a titanium hydride TiH₂ layer inside the Ti grain.

SEM images SEI of the synthesized at 2 GPa for 1 h from Mg:2B with 11 (10 wt. %, coarse grains) under 2 GPa for 1 h at 800 °C

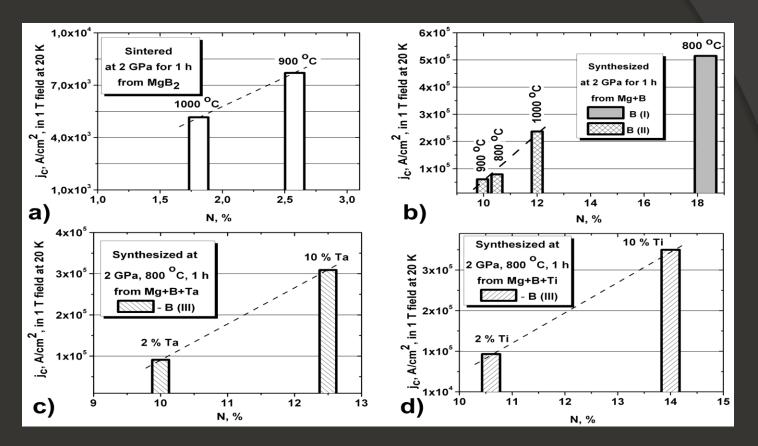


PROTON-21

SEM

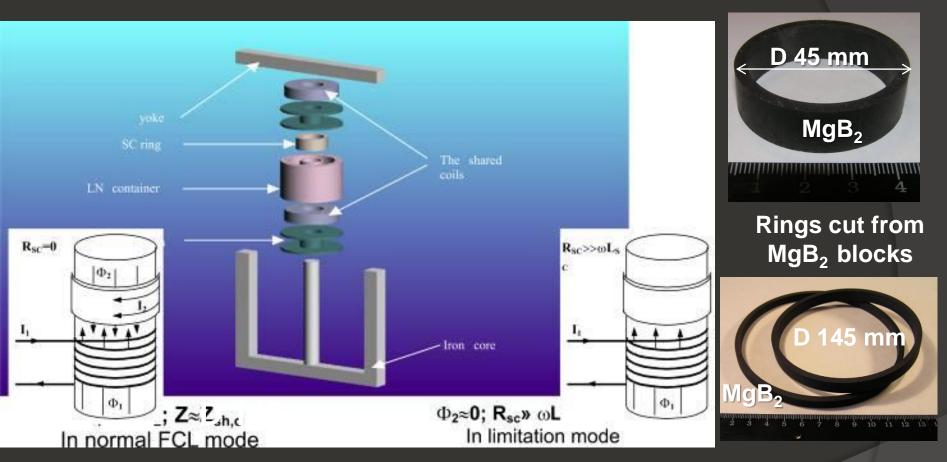
10.0kV

X1,300



Dependences of critical current density, jc, estimated in 1T field at 20 K on the amount of "black" Mg-B (MgB12) inclusions, N, for high pressure-high temperature manufactured MgB2 samples without additions: (a) - sintered from MgB2 (7MB - 0.8 % oxygen and 10-μm grains at 1000oC and 8MB – 98% purity Alfa Aesar at 900 oC), (b)- synthesized from boron 6B (type I -1.9 % oxygen, 1.4 - μm grains) and from boron 2B (type II - 1.5 % oxygen and 4-μm grains) and (c, d) - synthesized from boron 1B (type III – 1-μm grains, MaTecK, 95-97% purity) with additions of Ta and Ti, respectively. Amount of "black" inclusions, N, was calculated as a ratio of the area occupied by "black" inclusions at the COMPO image obtained at 1600x magnification to the total area of this image

MgB₂ for fault current limiter

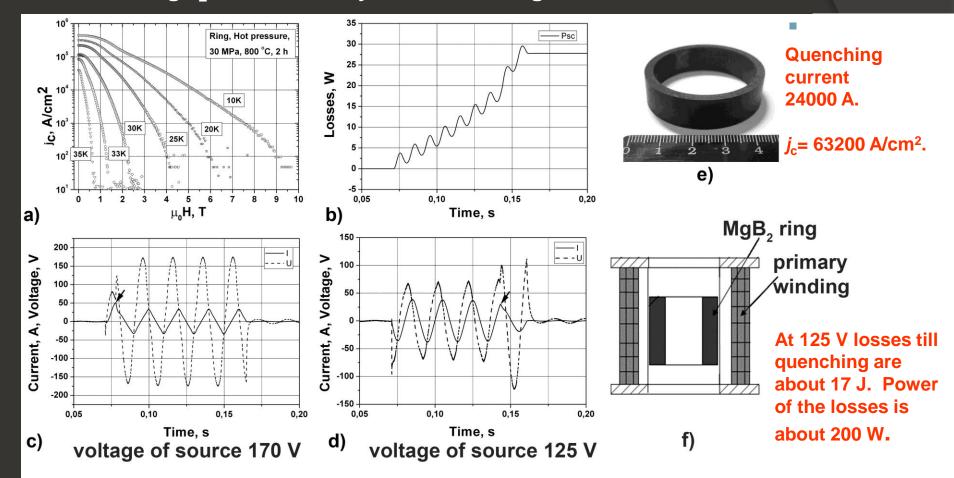


Transformer- type fault current limiter (Budapest University of Technology and Economics)

At fault event an increase in the circuit current causes a growth of the current in a superconductor higher than the critical value. The SC-or passes into the resistive state introducing a required for limitation impedance in the circuit (2-4 ms and 1 ms).33

Smart application of HP and hot-pressed MgB₂-based materials

Characteristics of the MgB2-based material synthesized at 30 MPa, 800 °C, 2 h from amorphous boron (III) (H.C. Starck, 1.5 % O, 4 µm grains) and magnesium (I) taken in the MgB₂ stoichiometry and of the ring cut from this material:



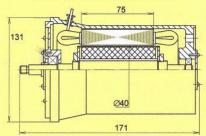
Typical oscilloscope traces of the current (solid lines) and voltage drop across the primary normal-metal winding (dashed lines) at 4.2 K. Arrows show the quenching current. Voltage of source was 170 V (c) and 125 V (d); e) ring: outer diameter – 45 mm, height - 11.6 mm, wall thickness – 3.3 mm.

First MgB2-based 1.3 kW motor by MAI (Moscow), IPHT (Jena), ISM NASU, Kiev

Experimental HTS Motor for 20 K Temperature

Cross-section of HTS Motor

General View of HTS Motor





Compound HTS-Ferromagnetic Rotor With YBCO Bulk Elements

General View of YBCO Samples

Rotor With YBCO Bulks





Compound HTS-Ferromagnetic Rotor With MgB₂ Bulk Elements

General View of MgB2 Bulk Samples

Rotor With MgB₂ Samples



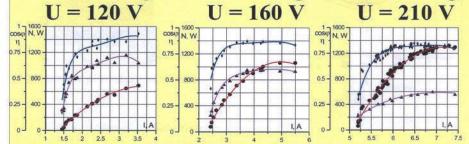


Test results of reluctance motor with MgB₂ blocks



Temperature: $T = 15 \dots 20 \text{ K}$

Phase voltage: Phase voltage:



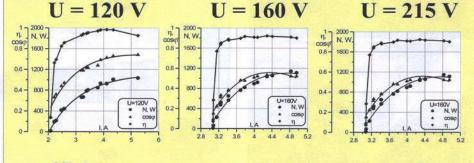
- efficiency, • - output power, ▲ - power factor

Test results of reluctance motor with YBCO blocks



Temperature: $T = 15 \dots 20 K$

Phase voltage: Phase voltage: Phase voltage:



◆ - efficiency,
 ◆ - output power,
 ▲ - power factor

Conclusions

- The distribution of nanostructural inhomogeneities, such as areas with a high concentration of B and impurity O, plays a key role in the variation of j_c in MgB₂ materials.
- ➤ Auger and SEM studies show that with increasing manufacturing temperature oxygen enriched 15-20 nm thick nano-layers transform into distinct dispersed Mg-B-O inclusions.
- ▶In parallel, the Ti addition results in a further increase in j_c , because Ti fosters the localization (or segregation) of oxygen and of higher magnesium borides and facilitates the formation of a homogeneous MgB_2 matrix with low oxygen content, but with Mg-B-O and MgB_x pinning centers.
- ➤ At low synthesis temperature Ti can absorb hydrogen forming titanium hydrides, thus preventing the formation of MgH₂ and promoting the material densification.
- ➤ The positive effect of Ti addition results from the high ability of Ti to absorb H, O, and Mg.

Thank you for your kind attention!

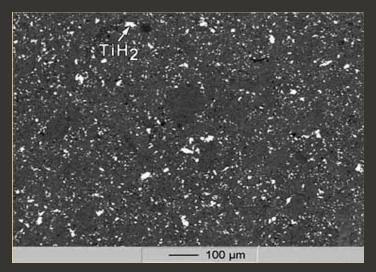


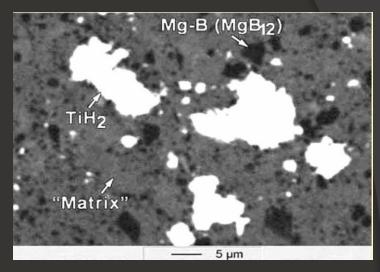


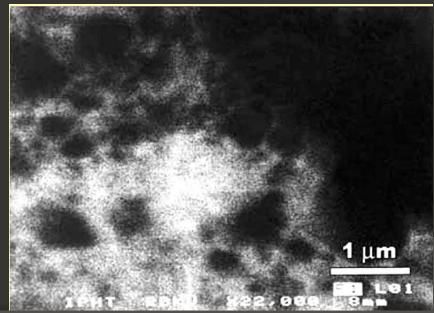




Stucture of MgB₂ with 10 wt% of Ti synthesized at 2 GPa, 800 °C, 1 h

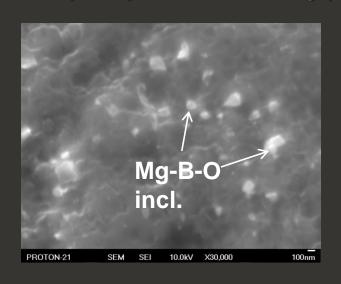


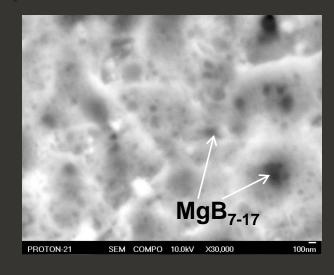


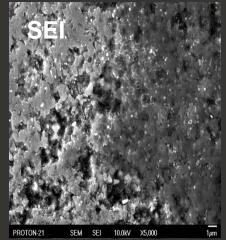


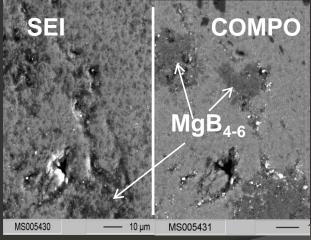
Pressure 30- 50 MPa can not totally suppress the volatility of Mg, thus MgB₄₋₆ porous regions are forming

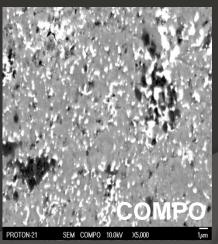
Spark plasma sintering (SPS): 50 MPa, 1050 °C, 32 min





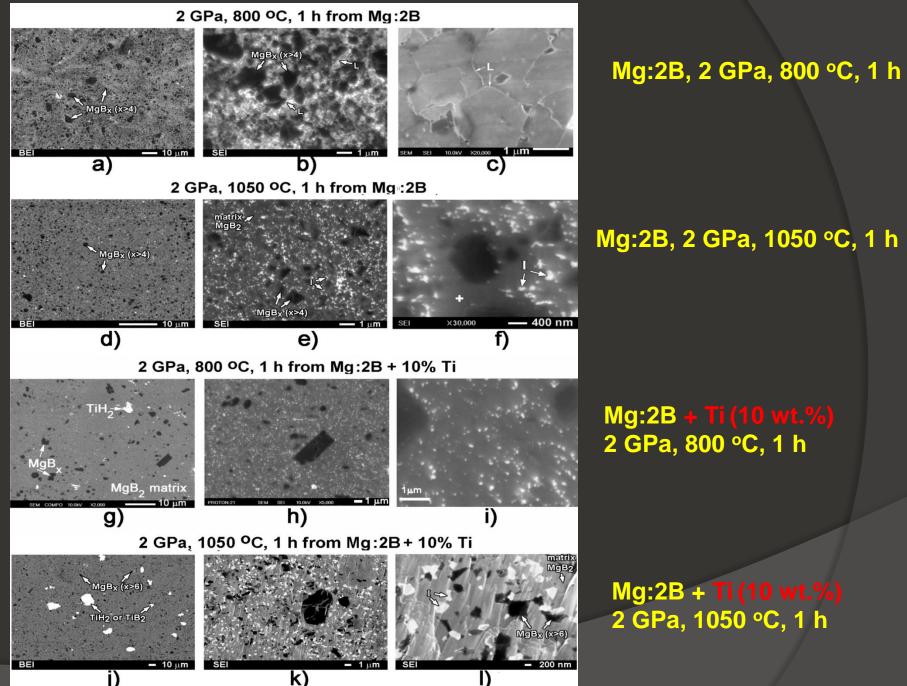






in-situ form Mg:2B

ex-situ form MgB₂



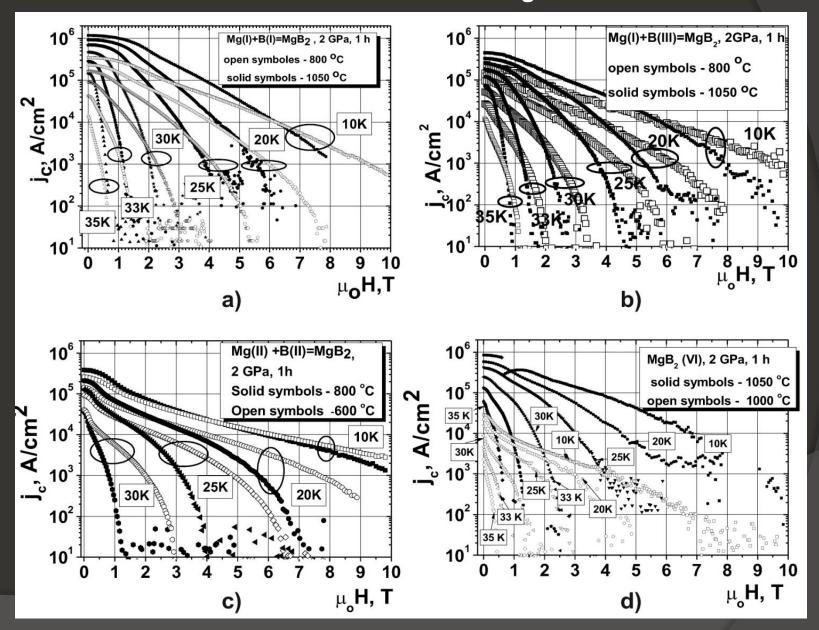
The amount of "black" MgB_x inclusions, N, for high pressure synthesized MgB_2 without additions and with additions of Ta and Ti N,%, was calculated as a ratio of the area that is occupied by "black" inclusions in the image of the structure obtained at 1600x magnification to the total area of the image obtained by SEM in the BEI mode.

The amount of "black" MgB_x inclusions, N, decreased by 6% by increasing the temperature from 800 up to 950 °C (from 1.8 to 8%)

The addition of Ti led to a further increase of ${\rm MgB_x}$ inclusions. If the Ti addition increased from 2 to 8 wt. % the amount of "black" ${\rm MgB_x}$ inclusions increased by about 6% as well .

Despite the essentially different structures and properties, the MgB_x (x>4) inclusions are practically "invisible" for a traditional x-ray diffraction analysis. This can be caused by their fine dispersion in the material structure and the large number of atoms in the unit cell of low symmetry resulting in many "reflecting planes", which essentially reduce the intensities of the x-ray reflections as compared to those of the MgB_2 matrix having a simple hexagonal unit cell

With manufacturing temperature increase the critical current density Increased in low and medium magnetic fields



Experimental

- 2 GPa HP, 30 MPa- HotP, 50 MPa SPS, 0.1 MPa Ar PL Initial materials
 - (I) in-situ from Mg:B=1:2 (MgB₂), from Mg:B=1:7 and 1:12 mixing and milling in a high speed activator for 3-5 min
- ✓ Boron (B) amorphous

Characteristics of initial boron powders.

Type of B	Grain size	O, wt%	C, wt%	N, wt%	H, wt%
1	<5 µm	0.66	0.31	0.48	0.32
II	<1 µm	-	3.5	1.02	0.87
III	4 μm	1.5	0.47	0.40	0.37

- ✓ Magnesium (Mg)
 Mg(I) turning (Technical Specifications of Ukraine 48-10-93-88)
 Mg(II) powder 325 mesh (HyperTec, USA)
- (II) ex-situ from MgB₂ powder
- ✓ Magnesium diboride (98% purity)

Intermediate conclusions

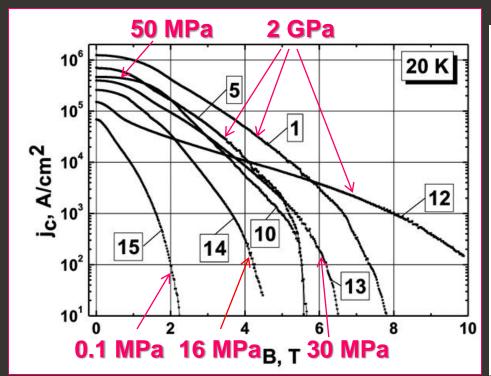
With the manufacturing temperature increase the oxygenenriched nanolayers "L" (15-20 nm thick) transformed into separate inclusions "I" and the sizes and amount of MgB_x inclusions are essentially reduced, what was observed in the range of 0.1 MPa - 2 GPa pressures.

But in the case of processing under 0.1 MPa (PL) the X-ray diffraction patterns showed that materials manufactured at 1050 °C for 15 min have MgB_4 structure and after 2 h – MgB_7 .

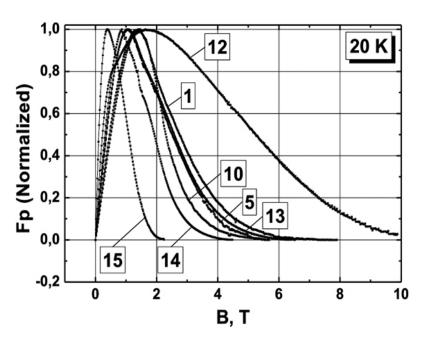
The contradiction between the results of X-ray diffraction analysis and SEM –Auger study can be explained by formation of the solid solutions of boron in MgO and of oxygen in MgB₂ and what due to the similarity of the scattering ability cannot be detected by traditional X-ray diffraction.

Pressure effect on MgB₂

Dependencies of critical current density, j_c , vs. magnetic field, B



Corresponding dependencies of volume pinning force Fp(n), vs. magnetic field, B



```
Curves: (1) Mg(I):2B(I)+10% SiC, 2GPa, 1050 °C,);

(5) Mg(I):2B(III)+10% Ti, 2 GPa, 1050 °C, 1 h,);

(10) Mg(I):2B(III, 50 MPa (SPS), 600°C- 0.3 h, 1050 °C-0.5 h,

(12) Mg(II):2B(II) with 3.5 % C, 2 GPa, 600 °C, 1 h,

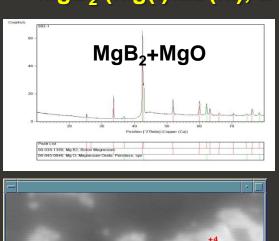
(13) Mg(I):2B(III)+10% Ti, 30 MPa,550°C- 1 h, 1000 °C- 0.2 h);

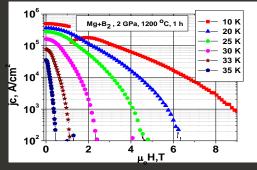
(14) MgB<sub>2</sub>, 16 MPa (SPS), 1150 °C, 0.3 h,);

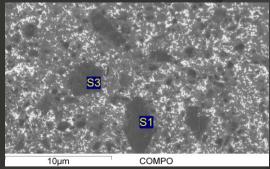
(15) Mg(I):2B(III), 0.1 MPa, Ar, 800 °C, 4 h,
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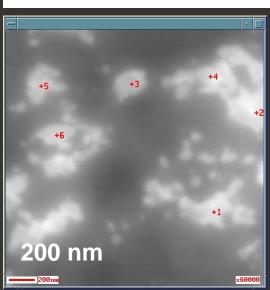
The distribution of boron and impurity oxygen in the ${\rm MgB_2}$ structure is of high importance for attaining high jc . Rather high amount of impurity oxygen can be present in the materials with high SC characteristics.

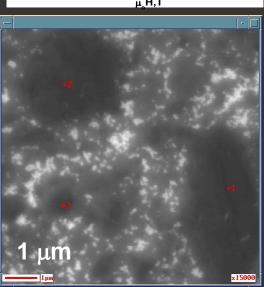
MgB₂ (Mg(I):2B(III), 2 GPa, 1200 °C, 1h), 76 wt % MgB₂, 24 wt % MgO











10µm	СОМРО
	·
	+4
	- 1
30	+1 +3
+5	+2
1 μm	.33
— 1 µ111	x20000

Atom %						
S-007	В	0	Mg	Total		
1	46,7	26,2	27,1	100,0		
2	45,4	27,2	27,4	100,0		
3	51,6	24,6	23,7	100,0		
4	49,1	26,2	24,7	100,0		
5	43,2	28,7	28,1	100,0		
6	46,1	28,1	25,8	100,0		

Atom %						
S-009	В	0	Mg	Total		
1	91,6	1,3	7,0	100,0		
2	91,8	2,1	6,1	100,0		
3	90,9	3,6	5,5	100,0		

<u>6</u>				
В	0	Mg	Total	B/Mg
63,6	7,8	28,6	100,0	2,22
55,6	15,5	28,9	100,0	1,92
51,7	17,9	30,4	100,0	1,70
56,0	15,1	28,9	100,0	1,93
53,5	16,9	29,6	100,0	1,81
	B 63,6 55,6 51,7 56,0	B O 63,6 7,8 55,6 15,5 51,7 17,9 56,0 15,1	B O Mg 63,6 7,8 28,6 55,6 15,5 28,9 51,7 17,9 30,4 56,0 15,1 28,9	B O Mg Total

