

Setting up and running the photoluminescence laboratory at ISOLDE

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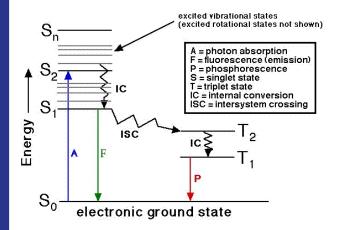
Outline

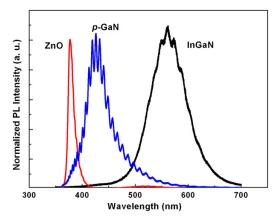
- What is photoluminescence (PL)?
- Setting up the lab
- Experimental setup
- Results
- Why PL at CERN?

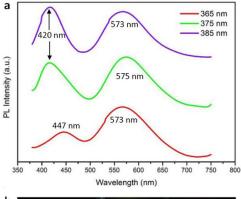


Photoluminescence

- Emission of light from a material under optical excitation
- Each material has its own excitation and emission spectra
- Extremely sensitive technique of electronic states
- No chemical identification of the source of the emitted light
 - Solution: implantation of radioactive isotopes





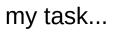






Before and after!





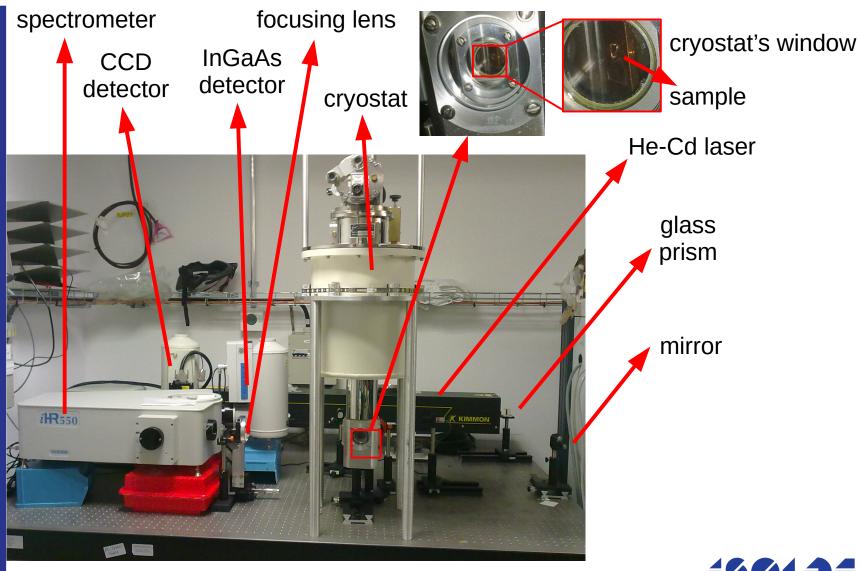




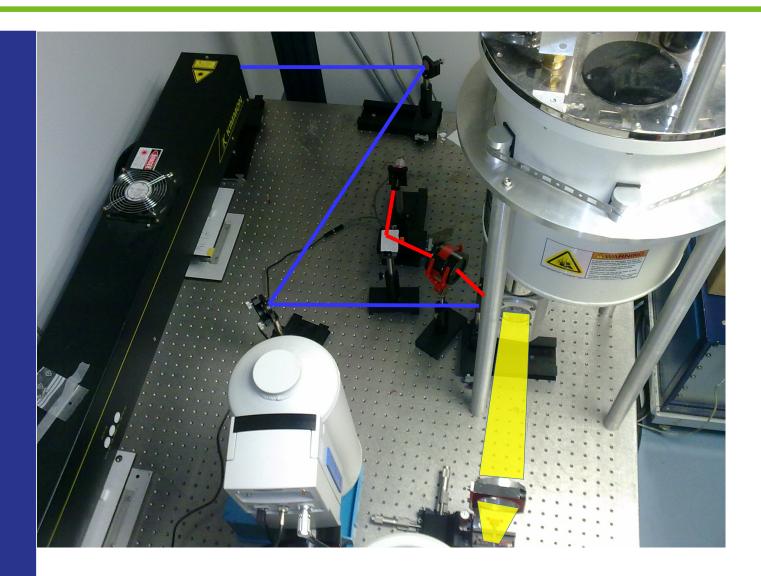




Experimental Setup



Experimental Setup



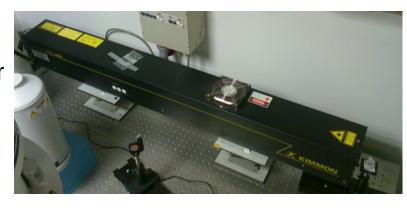


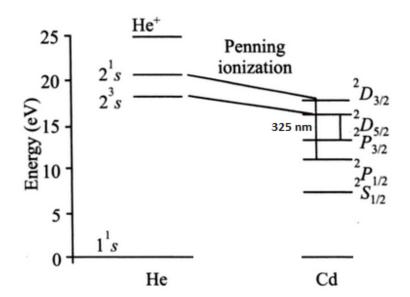
He-Cd Laser

- Length: 1.42 m
- Power (max): 200 mW
- Wavelength: 325 nm (near UV)
- Type: continuous wave (cw) metal-vapor
- Lasing medium: Cadmium
- He:Cd ratio: 100:1
- He excitation by 4 kV electric discharge
- Laser Class: 3B
 - Hazardous for direct eye exposure
 - Protective glasses requires





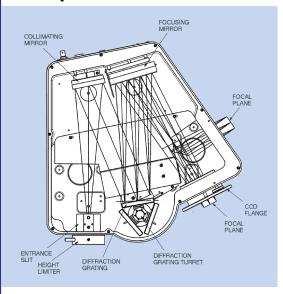






Spectrometer and detectors

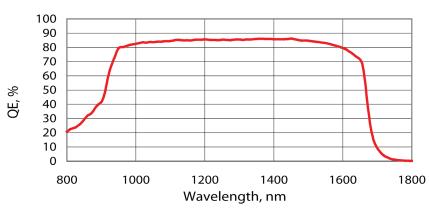
Spectrometer



- 3 blazed diffraction gratings:
 - 600 g/mm (500 nm)
 - 900 g/mm (1500 nm)
 - 2400 g/mm (400 nm)
- Speed:160 nm/s

InGaAs Detector





- Liquid Nitrogen cooled detector
- operating temperature: -103.3 C

CCD Detector

- High QE to visible spectrum
- operating temperature: -33 C

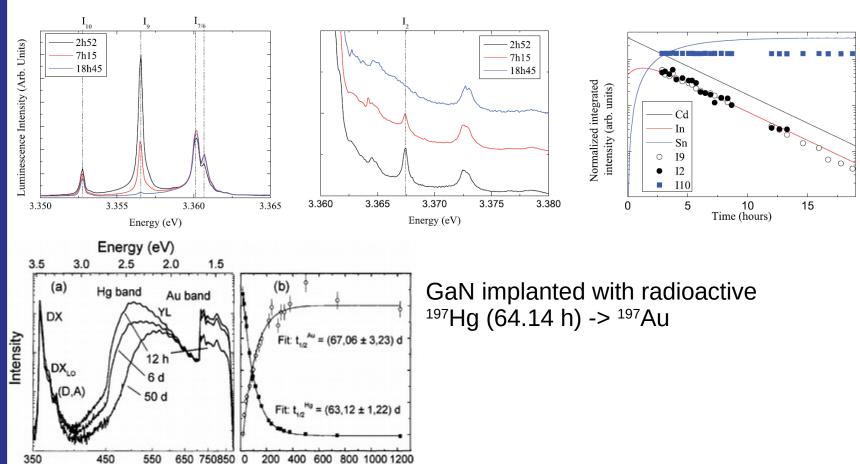


Results

Wavelength (nm)

<u>Identification of impurities in semiconductors</u>

ZnO implanted with radioactive 117 Ag (73 s) -> 117 Cd (2.5 h) -> 117 In (43 m) -> 117 Sn



Time (hours)

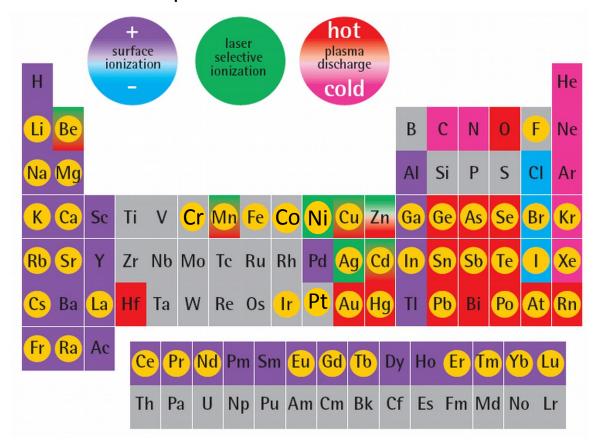
Why PL at ISOLDE CERN?

• No Higgs boson, no tetraquark, no pentaquark BUT...



Why PL at ISOLDE CERN?

- No Higgs boson, no tetraquark, no pentaquark BUT...
- More than 1200 isotopes of 72 different elements



Isotopes of this element — used for solid state physics or life science



Special thanks to my supervisor

Thank you for your attention

