# Setting up and running the photoluminescence laboratory at ISOLDE 

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## Outline

- What is photoluminescence (PL)?
- Setting up the lab
- Experimental setup
- Results
- Why PL at CERN?


## Photoluminescence

- Emission of light from a material under optical excitation
- Each material has its own excitation and emission spectra
- Extremely sensitive technique of electronic states
- No chemical identification of the source of the emitted light
- Solution: implantation of radioactive isotopes



## Before and after!



## Experimental Setup



## Experimental Setup



## He-Cd Laser

- Length: 1.42 m
- Power (max): 200 mW
- Wavelength: 325 nm (near UV)
- Type: continuous wave (cw) metal-vapor
- Lasing medium: Cadmium
- He:Cd ratio: 100:1
- He excitation by 4 kV electric discharge
- Laser Class: 3B

- Hazardous for direct eye exposure
- Protective glasses requires



## Spectrometer and detectors

## Spectrometer



- 3 blazed diffraction gratings:
- $600 \mathrm{~g} / \mathrm{mm}(500 \mathrm{~nm})$
- $900 \mathrm{~g} / \mathrm{mm}(1500 \mathrm{~nm})$
- $2400 \mathrm{~g} / \mathrm{mm}$ (400 nm)
- Speed:160 nm/s


## InGaAs Detector

Quantum Efficiency at $25^{\circ} \mathrm{C}$


- Liquid Nitrogen cooled detector
- operating temperature: -103.3 C


## CCD Detector

- High QE to visible spectrum
- operating temperature: -33 C


## Results

## Identification of impurities in semiconductors

ZnO implanted with radioactive ${ }^{117} \mathrm{Ag}(73 \mathrm{~s})->{ }^{117} \mathrm{Cd}(2.5 \mathrm{~h})->{ }^{117} \mathrm{In}(43 \mathrm{~m})->{ }^{117} \mathrm{Sn}$





GaN implanted with radioactive ${ }^{197} \mathrm{Hg}$ (64.14 h) -> ${ }^{197} \mathrm{Au}$

## Why PL at ISOLDE CERN?

- No Higgs boson, no tetraquark, no pentaquark BUT...


## Why PL at ISOLDE CERN?

- No Higgs boson, no tetraquark, no pentaquark BUT...
- More than 1200 isotopes of 72 different elements


Isotopes of this element used for solid state physics or life science

# Special thanks to my supervisor 

Thank you for your attention

