



RADIUM  
INSTITUTE  
ROSATOM

# CONFIRMATION OF A NEW ISOMERIC STATE IN THE $^{186}\text{Re}$ NUCLEUS

V.V. Karasev ; V.V. Koltsov

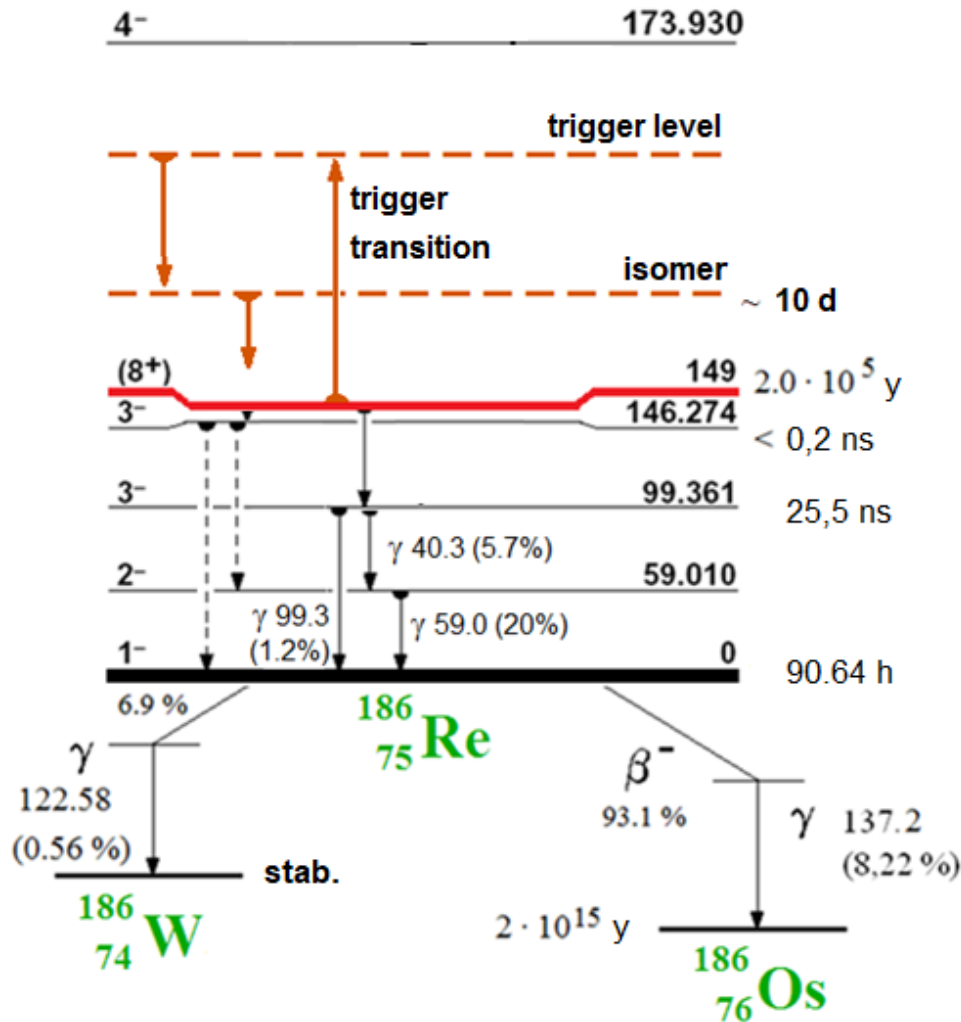
*JSC “Khlopin Radium Institute”, Sankt-Petersburg, Russia.*

Vladimir Koltsov

Khlopin Radium Institute, Saint-Petersburg, Russia

*E-mail: [vladimir-koltsov@yandex.ru](mailto:vladimir-koltsov@yandex.ru)*

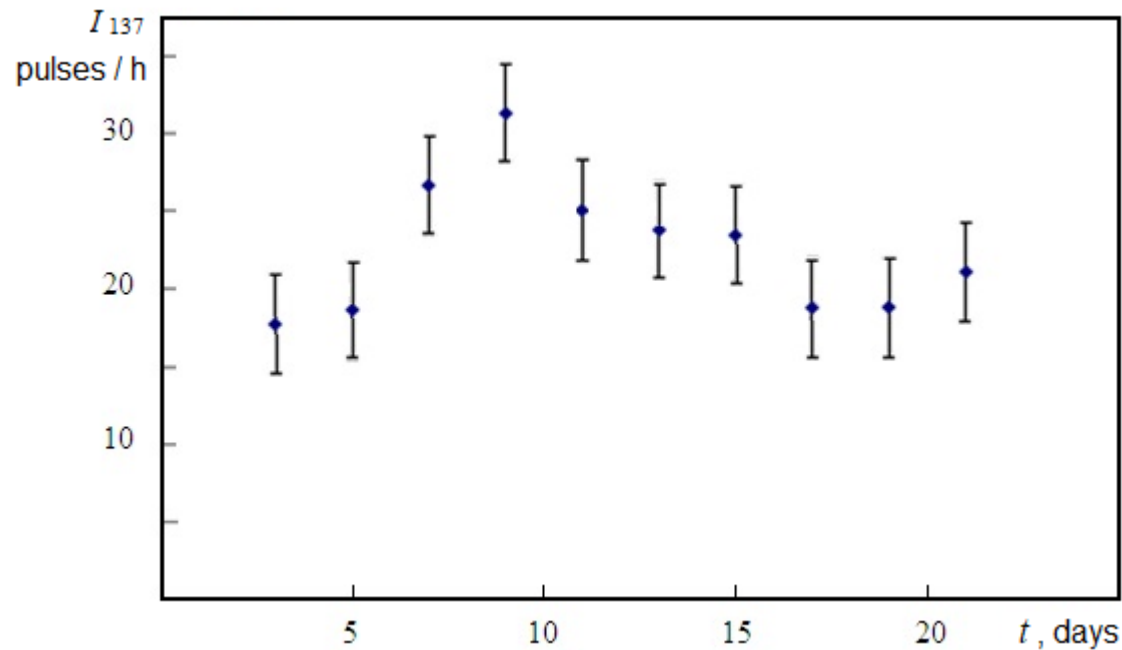
# $^{186}\text{Re}$ nucleus level diagram



Up arrow – the possibility of exciting the trigger level.

- In reactor isomer of 200000 y half-life is populated in 0.3% of cases of neutron capture by  $^{185}\text{Re}$  nuclei.
- Recently de-excitation of the isomer has been stimulated in laser plasma of 1 keV temperature.
- The supposed mechanism of stimulation is the excitation of the isomer to the trigger level, from where it decays to the ground state of the nucleus.
- Detection of stimulation by the decrease in the intensity of 137 keV  $\gamma$ -quanta after a jump-like population of the ground state.
- Now the trigger level parameters are unknown.

## Intensity of the 137 keV line in the $\gamma$ -spectrum of $^{186m}\text{Re}$ irradiated in a laser plasma



Intensity decay  
corresponds to  
 $T_{1/2} = 112 \pm 10$  hours

If the effect is due to the stimulated discharge of  $^{186m}\text{Re}$  nuclei,  
then the discharge of  $\sim 10^{-5} \%$  of the  $^{186m}\text{Re}$  nuclei is stimulated.

We have to assume that there is an unknown excited state in the  $^{186}\text{Re}$  nucleus  
with  $T_{1/2}$  for several days, which is populated upon stimulated discharge of the  
 $^{186m}\text{Re}$  isomers.

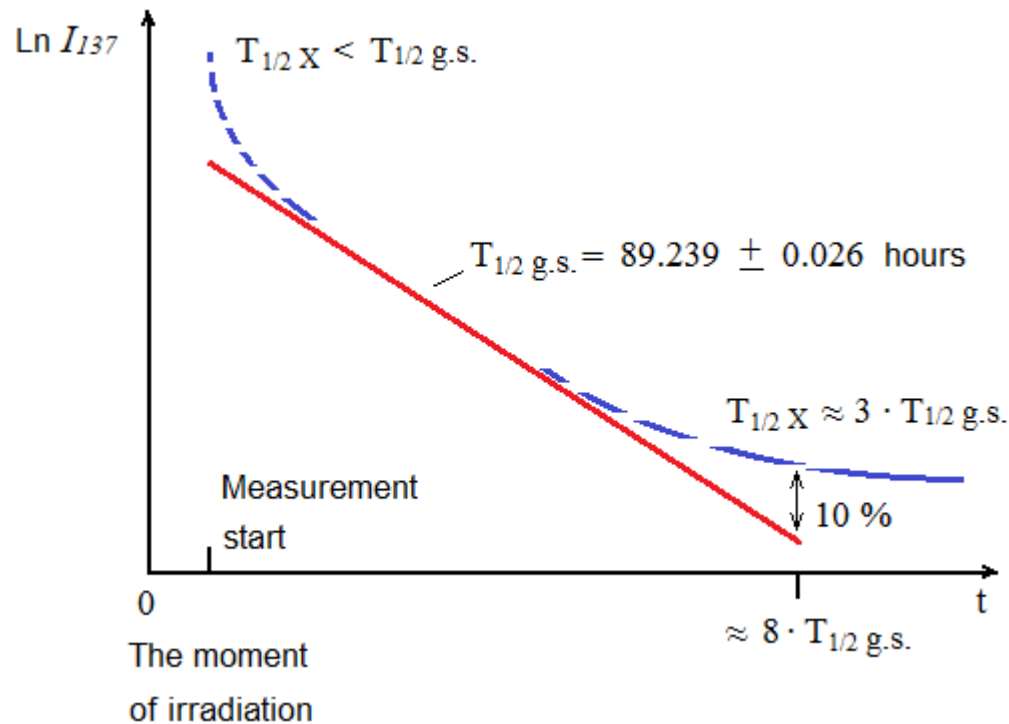
## The difference in the known results of measuring the $^{186}\text{Re}$ half-live

Measurement technique			$T_{1/2, g}$ , hours
$^{186}\text{Re}$ production method	Method measurements	Measurement duration, days	
$^{185}\text{Re} (n, \gamma) ^{186}\text{Re}$	$4\pi$ $\gamma$ -ionization chamber	33	$89.239 \pm 0.026$ <i>(Schonfeld et al, 1994)</i>
$^{185}\text{Re} (n, \gamma) ^{186}\text{Re}$		21	$89.25 \pm 0.07$ [6] <i>(Coursey et al., 1991)</i>
$^{185}\text{Re} (n_{th}, \gamma) ^{186}\text{Re}$	$\gamma$ – spectra	18	$90.600 \pm 0.024$ <i>(Abzouzi et al., 1989.)</i>
$^{186}\text{W} (p, n) ^{186}\text{Re}$		36	$88.35 \pm 0.16$ <i>(Our previous work, 2018)</i>

Accuracy is shown at the one standard deviation level.

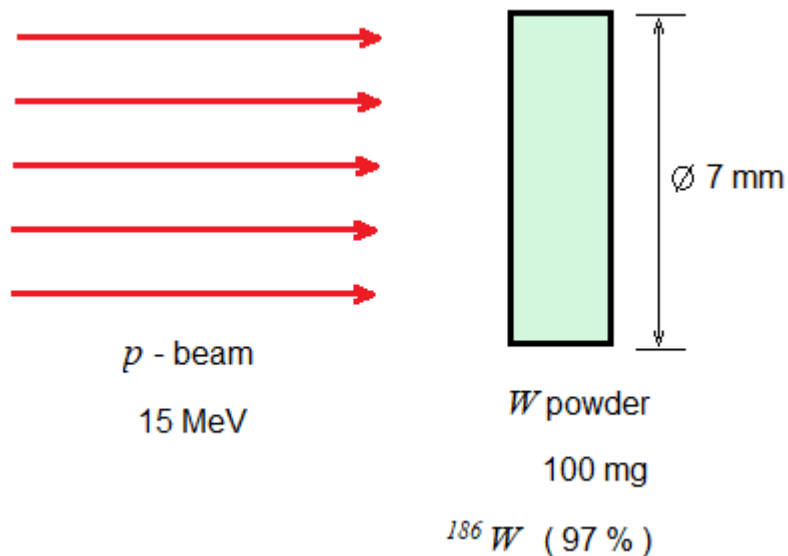
- The discrepancy in the  $T_{1/2, g}$  measurement results is another indication of the possibility of the existence of an unknown isomeric level in the  $^{186}\text{Re}$  nucleus.
- The measurement technique did not take into account the possible dependence of the observed  $T_{1/2, g}$  on the method of obtaining  $^{186}\text{Re}$  nuclei, on the duration of measurements, and on the time of the beginning of measurements after the end of target irradiation.

# The idea of the experiment



- It is possible that the new isomer can be populated in the  $(p, n)$  reaction of formation of the  $^{186}\text{Re}$  nucleus from  $^{186}\text{W}$ .
- Population of a new isomer in this reaction would lead for  $\gamma$ -quanta from an irradiated source to the dependence of the  $\gamma$ -intensity  $I_{137}(t)$  on time  $t$  after the formation of  $^{186}\text{Re}$  nuclei, which differs from the simple exponent associated with the decay of the ground state of  $^{186}\text{Re}$  with a half-life of  $T_{1/2, g} = 89.239 \text{ hours}$

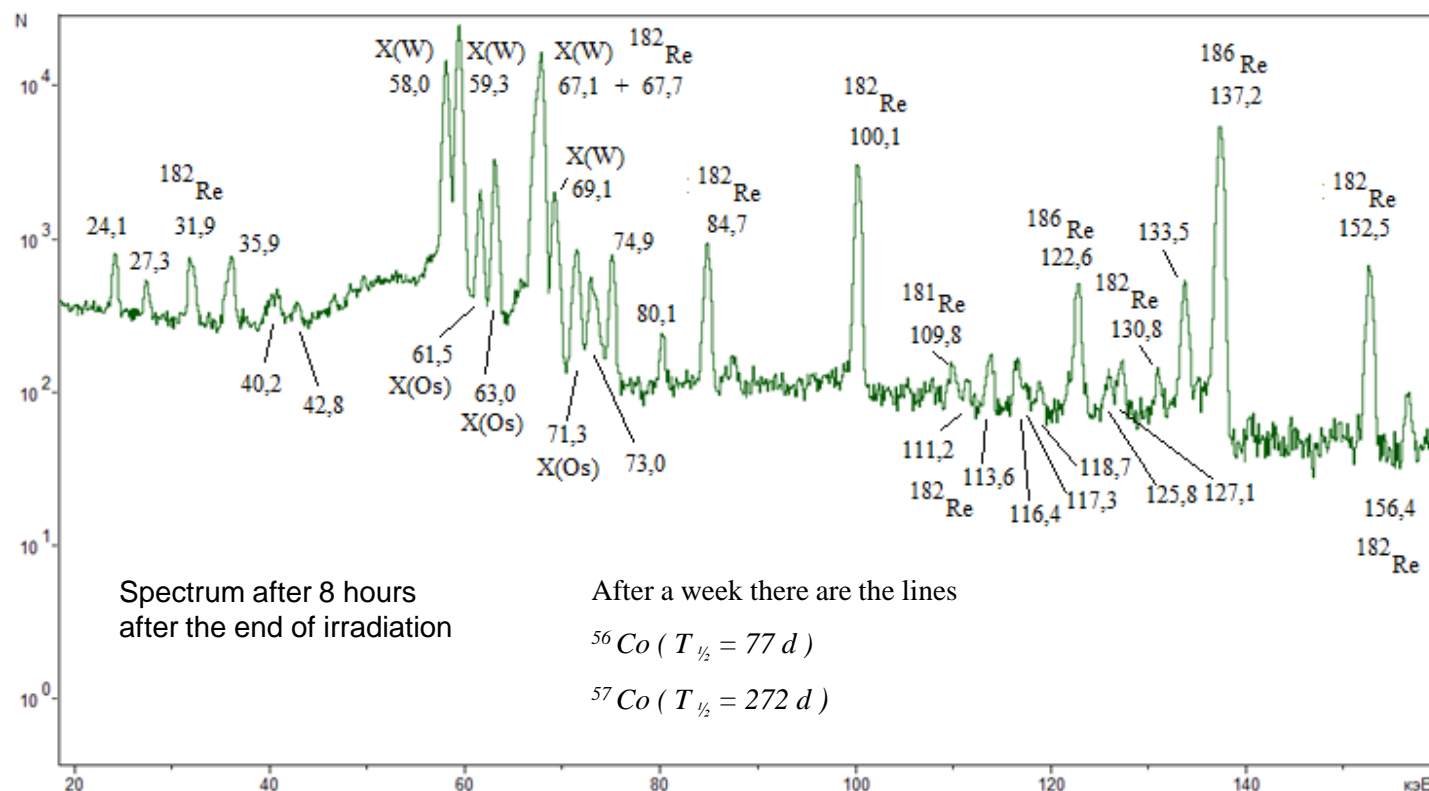
# Experimental method



Irradiation parameters	Our work (2018)	Present work
$p$ - current , $\mu\text{A}$	4	4
Exposition, min	15	100

Measurement method	Our work (2018)	Present work
Measurement time after the end of irradiation, days	0,2 - 36	30 - 105
HPGe detector	$\varnothing 25 \times 15 \text{ mm}$	150 $\text{cm}^3$ with the well
Measurement geometry	10 mm above the detector	In the well
Initial loading of the spectrometer, $\text{s}^{-1}$	$\sim 10^3$	$\sim 10^3$

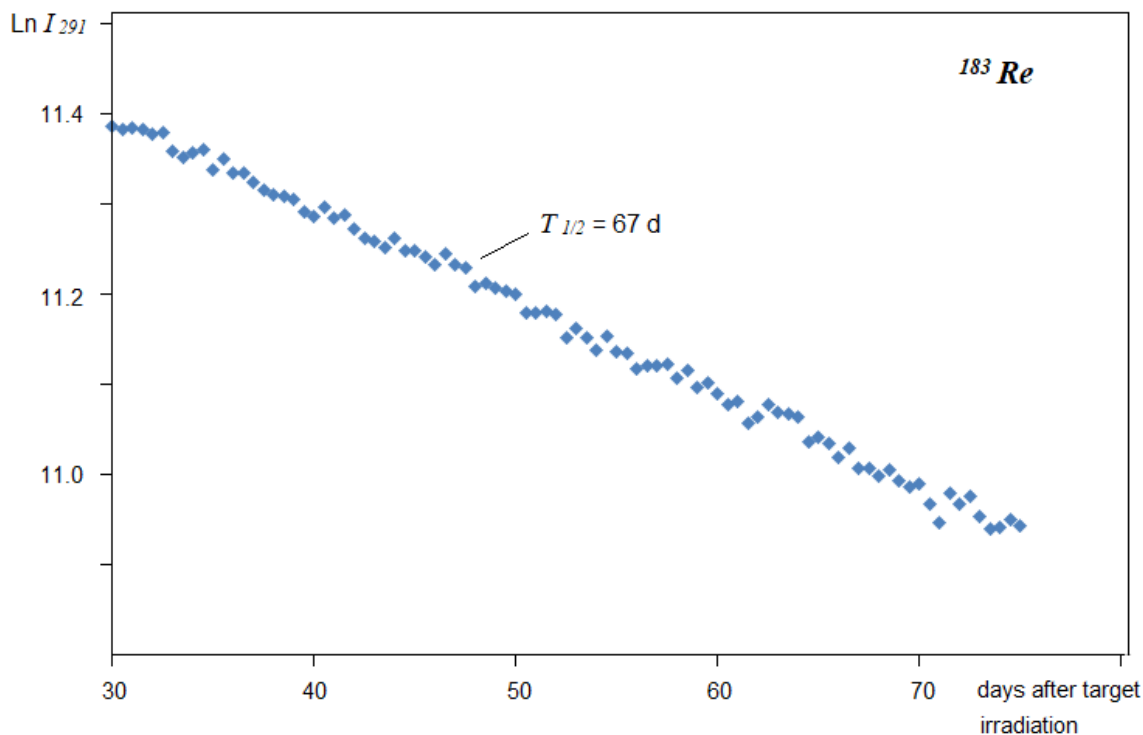
# $\gamma$ -spectrum of the sample



Stable isotope	$^{180}\text{W}$	$^{182}\text{W}$		$^{183}\text{W}$	$^{184}\text{W}$		$^{186}\text{W}$
Nature content, %	0,13	26,3		14,3	30,67		28,6
Reaction products ( $p, n$ ) , ( $p, 2n$ )	$^{180}\text{Re}$	$^{181}\text{Re}$	$^{182}\text{Re}$	$^{183}\text{Re}$	$^{184}\text{Re}$	$^{184m}\text{Re}$	$^{186}\text{Re}$
$T_{1/2}$	2,4 min	20 h	64 h	70 d	38 d	165 d	90, 64 h
Decay mode	$e^-$ capture						$\beta$

# Method for processing the results of $\gamma$ -spectra measurements

- $\gamma$ -spectra were measured sequentially in time with an exposure of 12 hours.
- The areas of  $\gamma$ -lines in the spectra were determined by fitting Gaussian peaks by the least squares method; in this case, the center of the line, its half-width, and area were determined.
- Half-lives were determined by fitting the decay curve with one or two exponents. The exponents were inscribed using the least squares method.
- The  $\gamma$ -lines of radionuclides in the spectrum remained in place with an accuracy of 0.2 keV.

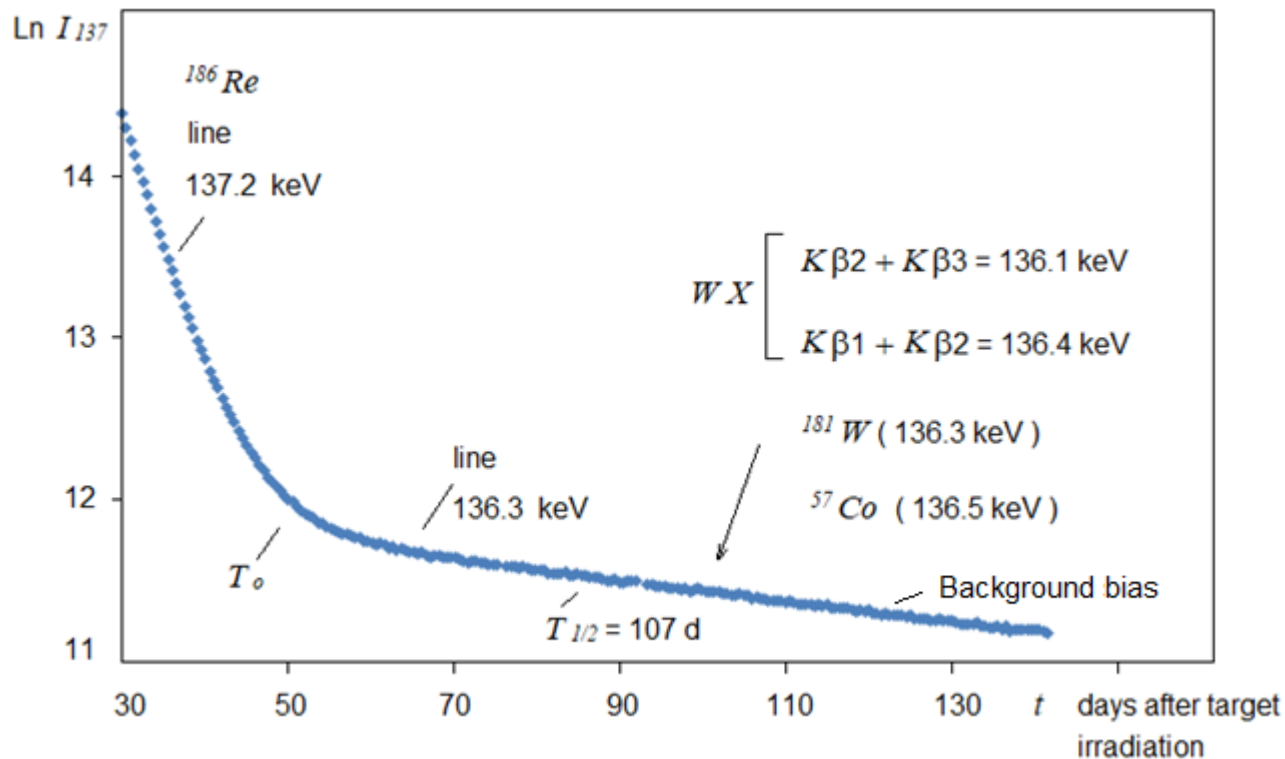


Determination of  $T_{1/2}$  for  $^{183}\text{Re}$  is a measurement correctness test.

The decay curve is plotted by measuring the intensity of  $\gamma$ -quanta 291 keV from  $^{183}\text{Re}$ .



# Time variation of the intensity of the 137 keV $\gamma$ -line



Background bias model when

the measurement time is  $t < T_o$ .

Constant  $\lambda = 0.0064 \text{ d}^{-1}$ .

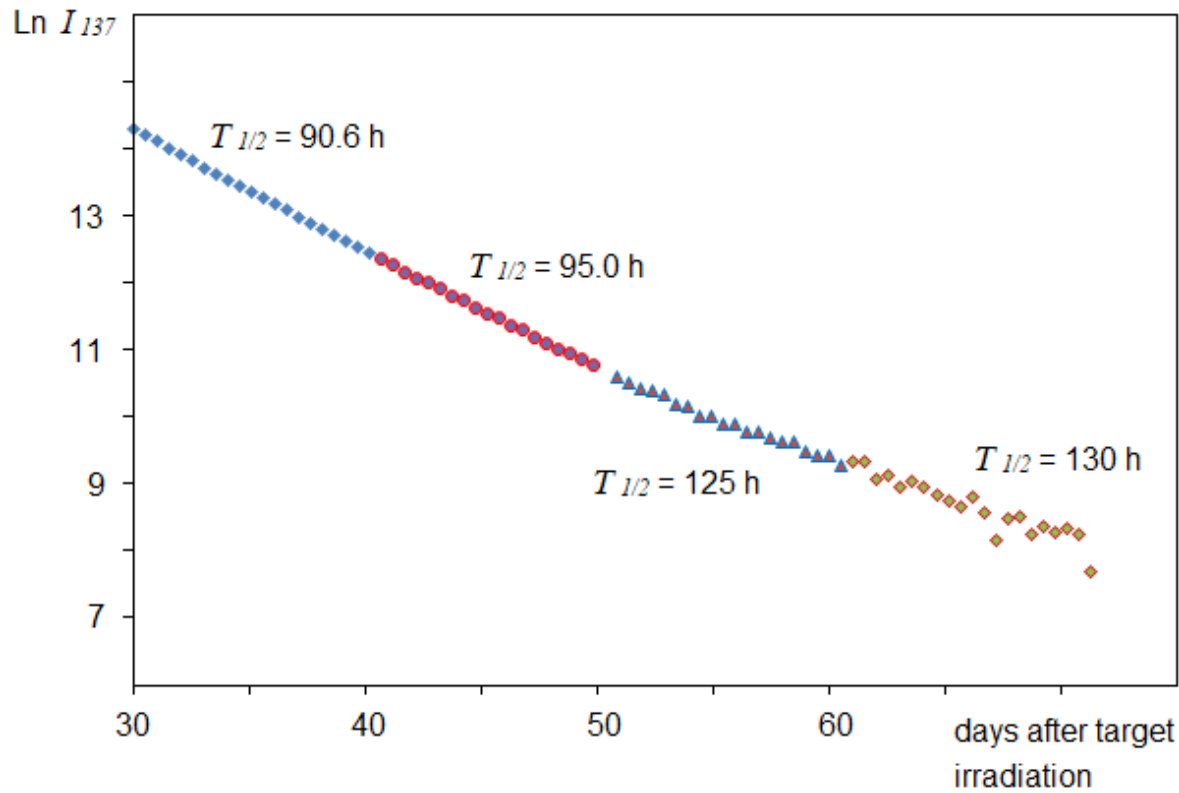
$$I_{bac.}(t) = (107900 \times 0.7) \times e^{\lambda(T_o - t)} + (107900 \times 0.3)$$

The model

takes into account:

- The value of the intensity  $I_{137}$  at the time  $T_o$ .
- Exponential decay of  $I_{137}$  at  $t > T_o$ .
- Requirement for the decay curve to match the half-life of  $^{186}\text{Re}$  at the start of measurements.

## Intensity of 137 keV $\gamma$ -quanta emitted by $^{186}\text{Re}$



Half-life, calculated from different parts of the decay curve, increases over time after the formation of  $^{186}\text{Re}$ .

The time dependence of the  $I_{137}$  intensity corresponds to the existence of the new isomer in the  $^{186}\text{Re}$  nucleus. However, for complete unambiguity of such a conclusion, it is necessary to carry out similar measurements again using an improved technique:

- The target's  $^{186}\text{W}$  isotope enrichment should be much better than 97%.
- The target must be free of impurities, in particular iron.
- Measurements should be carried out in geometry without summation of X-ray lines in the spectrum.

**Thanks !**