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[602] Influence of the tungsten surface chemical state on the ammonia formation using low-pressure N2-H2 plasmas

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In this contribution, we discuss the catalytic formation of NH3 using radiofrequency (RF) N2-H2 plasmas and a tungsten (W) foil. A parametric study was done at various RF powers (30–300 W), discharge pressures (2–5 Pa), gas compositions (10–90% N2) and discharge temperatures (< 673 K). NH3 generally increased with the electron density and/or electron temperature, whereas the discharge temperature had a minor effect. We observed no impact on NH3 after contaminating the W surface with H2, N2 and O2. Instead, coating the W foil with a nm-thick boron layer led to a reduction of ammonia by half, possibly due to the formation of B-H bonds.

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