



Canadian Association
of Physicists

Association canadienne
des physiciens et physiciennes

Contribution ID: 3486 Type: **Poster not-in-competition (Graduate Student) / Affiche non-compétitive (Étudiant(e) du 2e ou 3e cycle)**

(POS-16) Lattice Dynamical Study of High-Entropy Oxides

Tuesday, 7 June 2022 17:42 (2 minutes)

High-entropy oxides (HEOs) comprise an equimolar mixing of metal cations combined into a single-phase crystal structure. First synthesized in 2015 [1], HEOs have garnered much attention as candidates for high-efficiency batteries and heat shields [2, 3]. HEOs composed of four and five binary oxides have been previously investigated by infrared [4] and Raman spectroscopy [5] and lattice dynamical simulations. The IR spectra consist of a strong, reststrahlen mode at 350 cm^{-1} and a much weaker mode at 150 cm^{-1} not predicted by group theory. The absence of spin-phonon splitting in the reststrahlen band below the Neel temperature (T_N), despite appearing in the parent oxides CoO and NiO, has been attributed to a high rate of static disorder scattering. The Raman spectra are composed of five peaks which have been assigned to TO , LO , $LO + TO$, $2LO$ modes, as well as a two-magnon mode. Fits of the spectra to the Lorentz oscillator model revealed a temperature-dependent damping parameter which was ascribed to anharmonic effects. The phonon density of states will be simulated using GULP [6] in order to understand the IR and Raman spectra.

[1] Christina Rost et al. "Entropy-stabilized oxides". In: Nature Communications 6 (Sept. 2015), p. 8485. doi: 10.1038/ncomms9485.

[2] Abhishek Sarkar et al. "High Entropy Oxides for Reversible Energy Storage". In: Nature Communications 9 (Aug. 2018). doi: 10.1038/s41467-018-05774-5.

[3] Joshua Gild et al. "High-entropy fluorite oxides". In: Journal of the European Ceramic Society 38.10 (2018), pp. 3578–3584. issn: 0955-2219. doi: <https://doi.org/10.1016/j.jeurceramsoc.2018.04.010>. url: <https://www.sciencedirect.com/science/article/pii/S0955221918302115>.

[4] Tahereh Afsharvosoughi and D. A. Crandles. "An infrared study of antiferromagnetic medium and high entropy rocksalt structure oxides". In: Journal of Applied Physics 130.18 (2021), p. 184103. doi: 10.1063/5.0070994. eprint: <https://doi.org/10.1063/5.0070994>. url: <https://doi.org/10.1063/5.0070994>.

[5] Tahereh Afsharvosoughi. "Structural, Magnetic and Vibrational Studies of Entropy Stabilized Oxides". Brock University, 2021.

[6] Julian D. Gale. "GULP: A computer program for the symmetry-adapted simulation of solids". In: J. Chem. Soc., Faraday Trans. 93 (4 1997), pp. 629–637. doi: 10.1039/A606455H. url: <http://dx.doi.org/10.1039/A606455H>.

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Session Classification: DCMMP Poster Session & Student Poster Competition (8) | Session d'affiches DPMCM et concours d'affiches étudiantes (8)

Track Classification: Technical Sessions / Sessions techniques: Condensed Matter and Materials Physics / Physique de la matière condensée et matériaux (DCMMP-DPMCM)