

EP-DT Detector Technologies



Possible alternatives to SF₆ for Resistive Plate Chambers

Beatrice Mandelli, R. Guida, G. Rigoletti

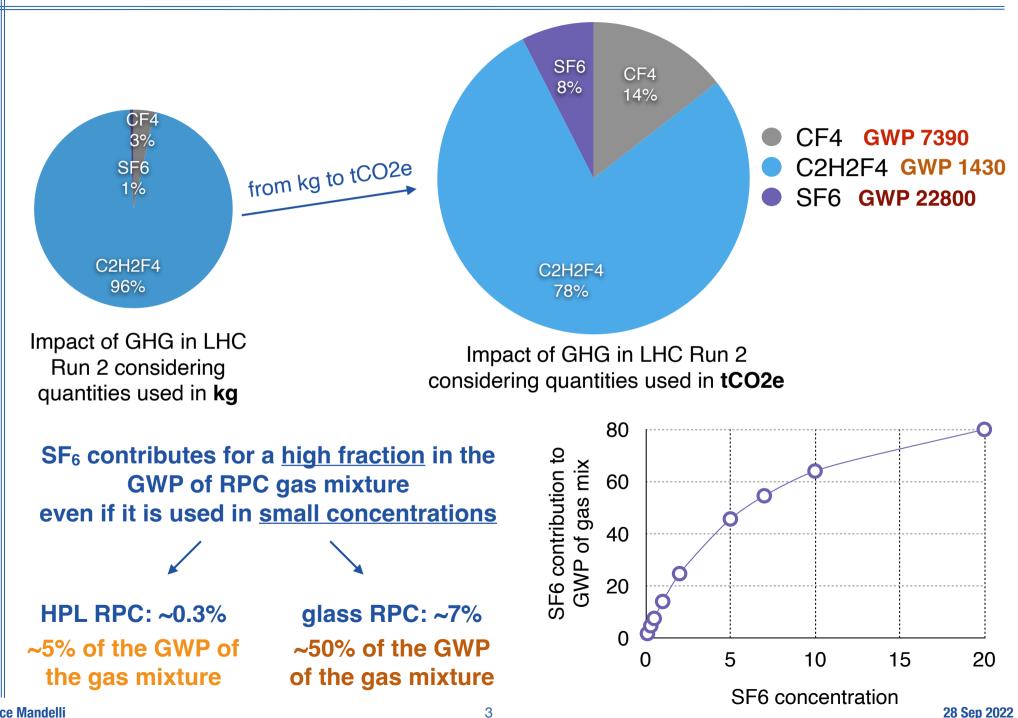
CERN

The XVI Workshop on Resistive Plate Chambers CERN, 28th September 2022 SF₆ impact on GHG emissions and possible alternatives

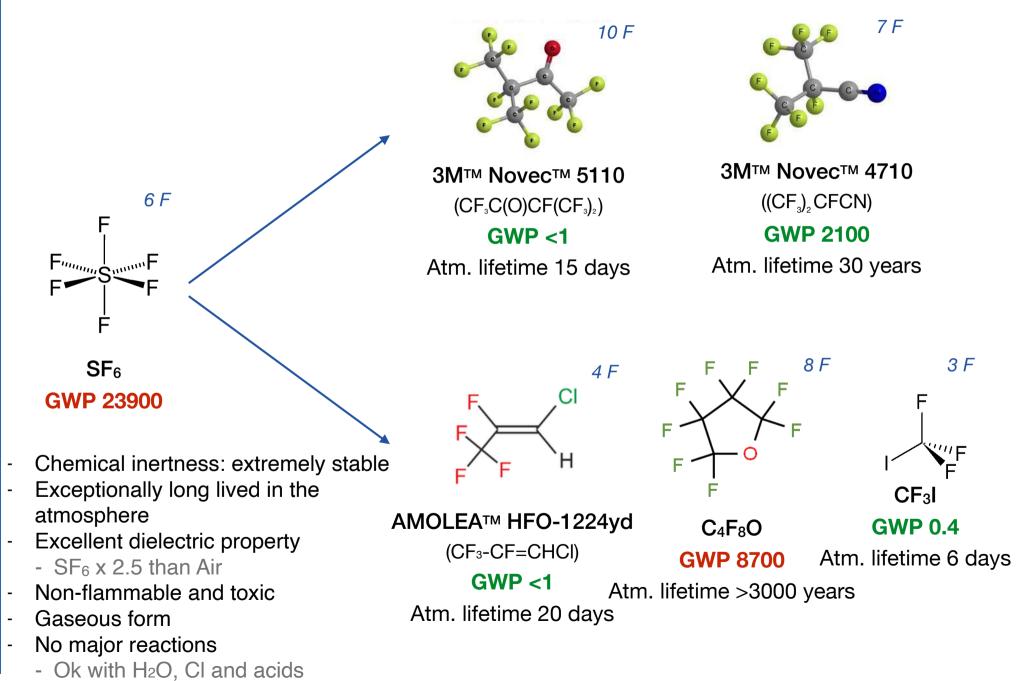
RPC performance with new gases as substitutes of SF₆

Thoughts and tests on properties of new gases

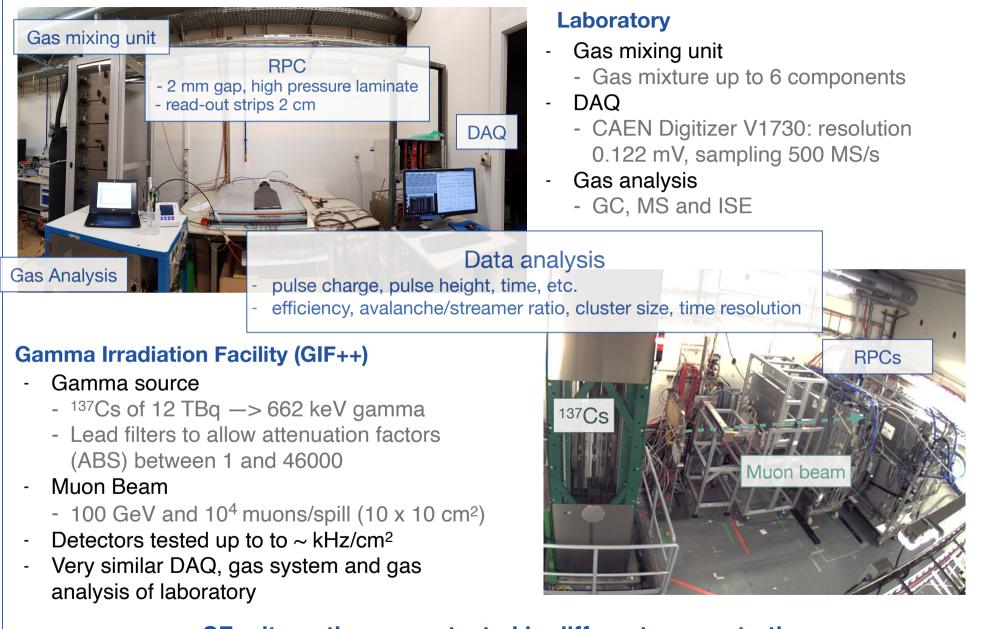
Use of SF₆ in RPC detectors at CERN



Alternatives to SF₆

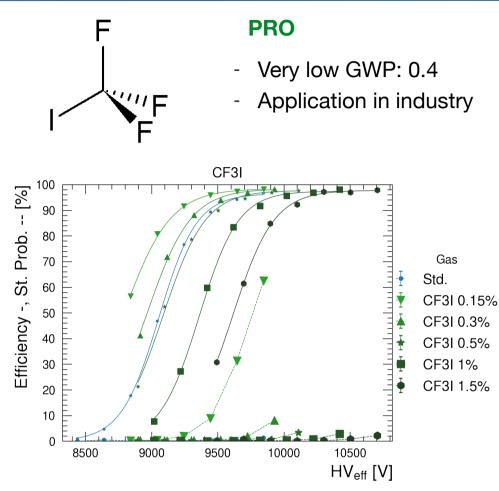


Set-ups: laboratory and GIF++



SF₆ alternatives were tested in different concentrations as substitute of SF₆ in RPC standard gas mixture

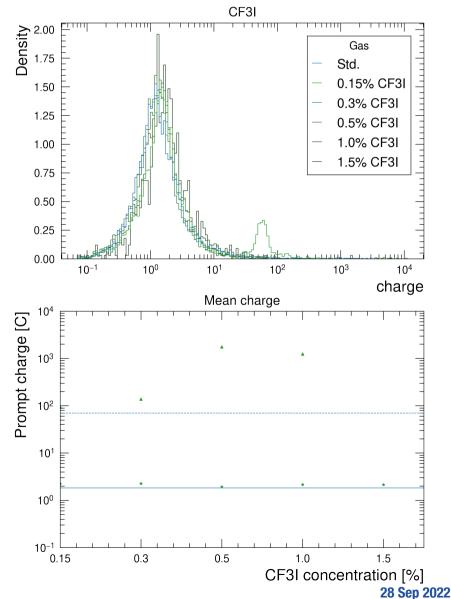
CF₃I



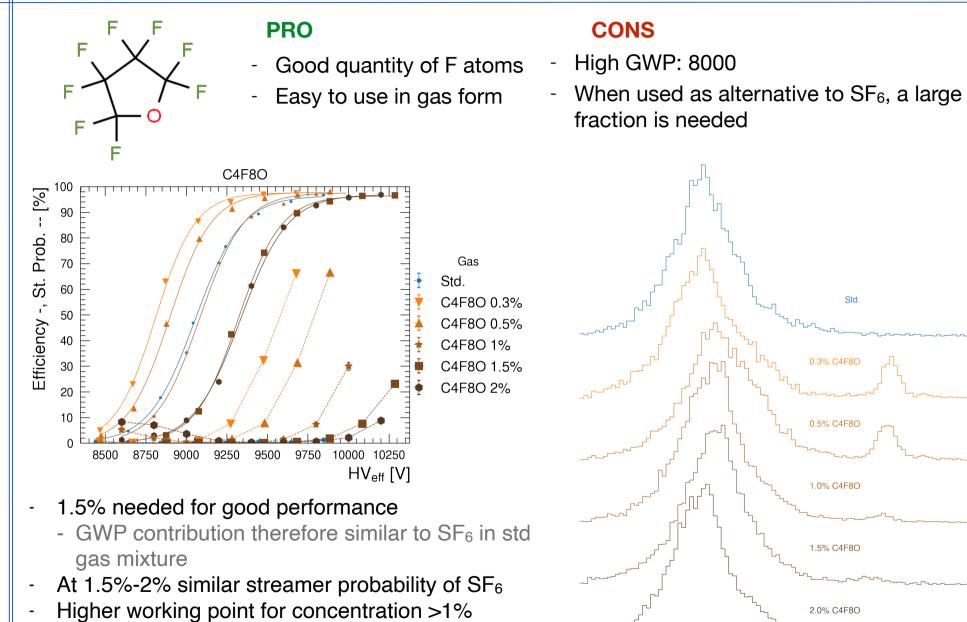
- Low concentration (0.3-0.5%) enough for good streamer suppression
- Working point similar to std gas mixture with 0.3-0.5% of CF₃I
- Avalanche charge similar to std gas mixture
 - Streamer charge much affected by error due to low statistics

CONS

- Toxic (inhalation), low human genotoxic risk
- May react with H₂O



C_4F_8O



- Avalanche charge a bit higher than std gas mixture
 - Streamer charge similar

Charge [pC]

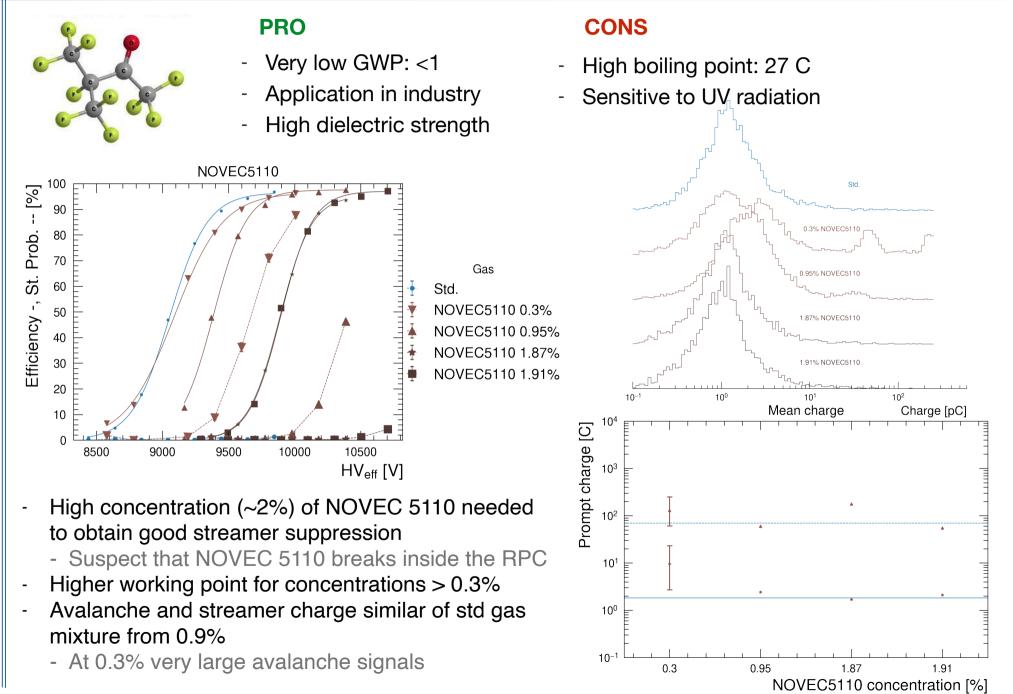
10²

10-1

10⁰

10¹

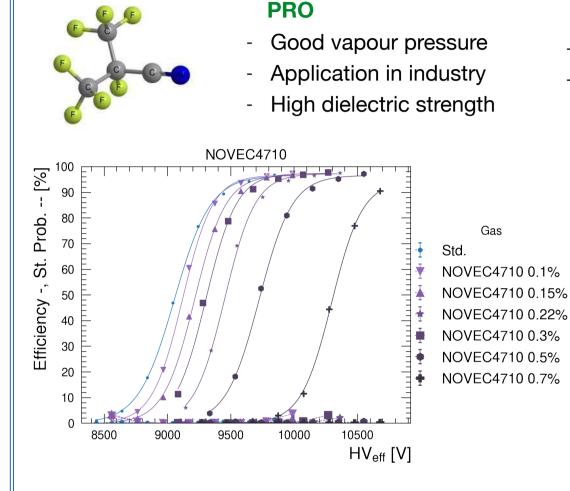
Novec 5110



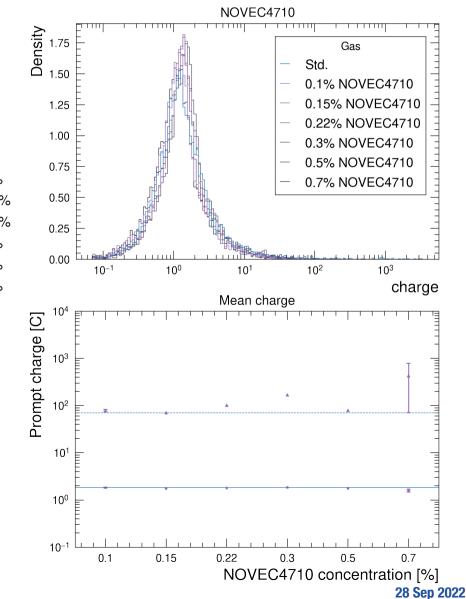
Beatrice Mandelli

28 Sep 2022

Novec 4710

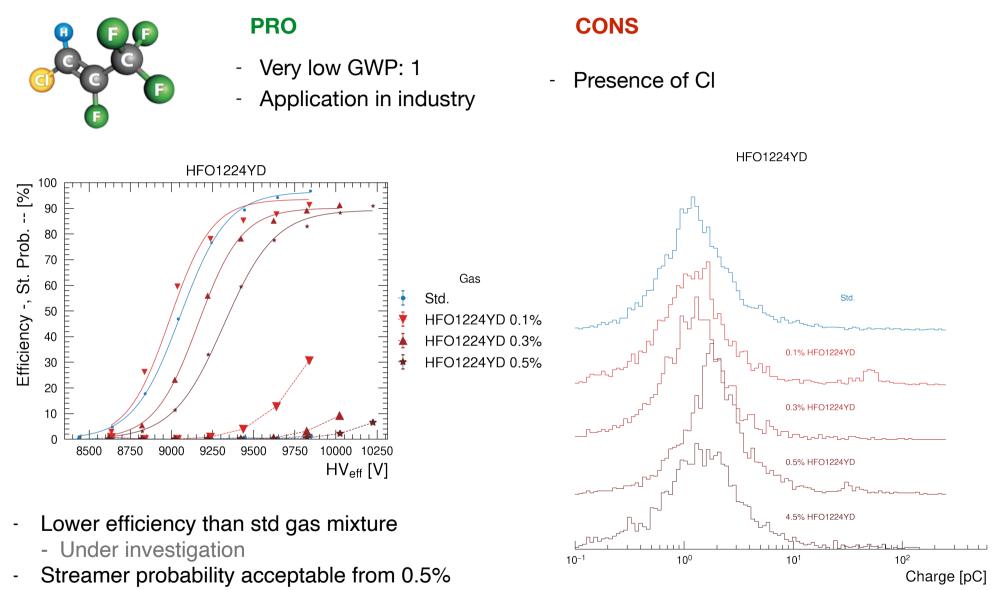


- CONS
- GWP of 2200
- It may react with H₂O



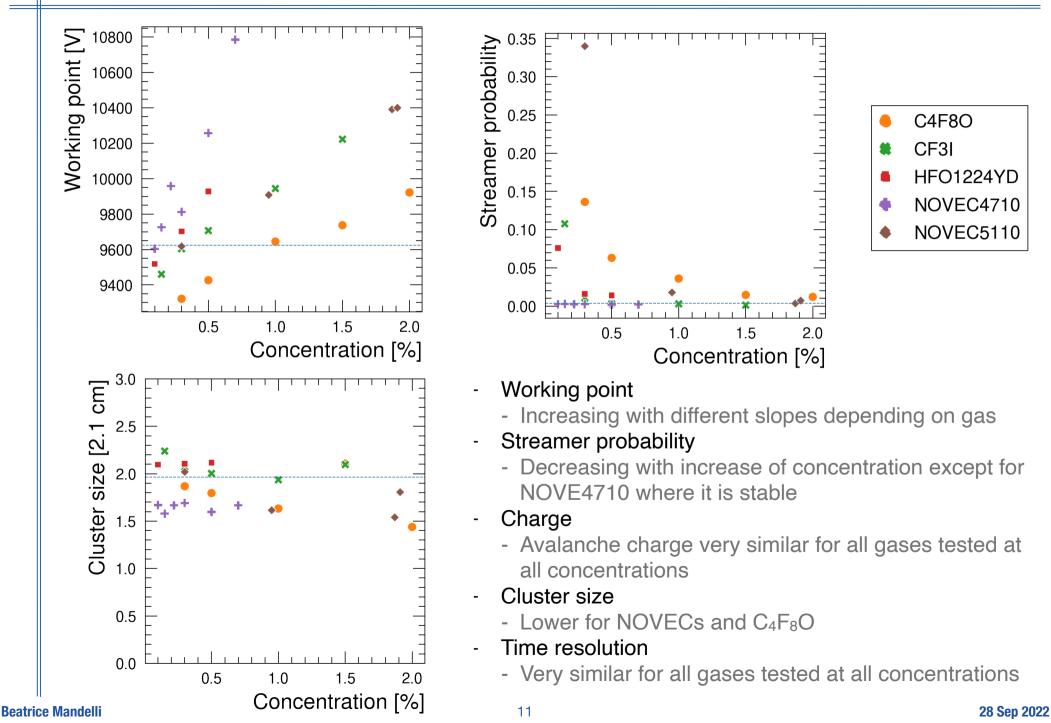
- Streamer probability always lower than std gas mixture
 - 0.1% of NOVEC 4710 already enough!
- Avalanche charge and cluster size lower than std gas mixture
- Higher working point for concentrations > 0.1%

HFO-1224yd

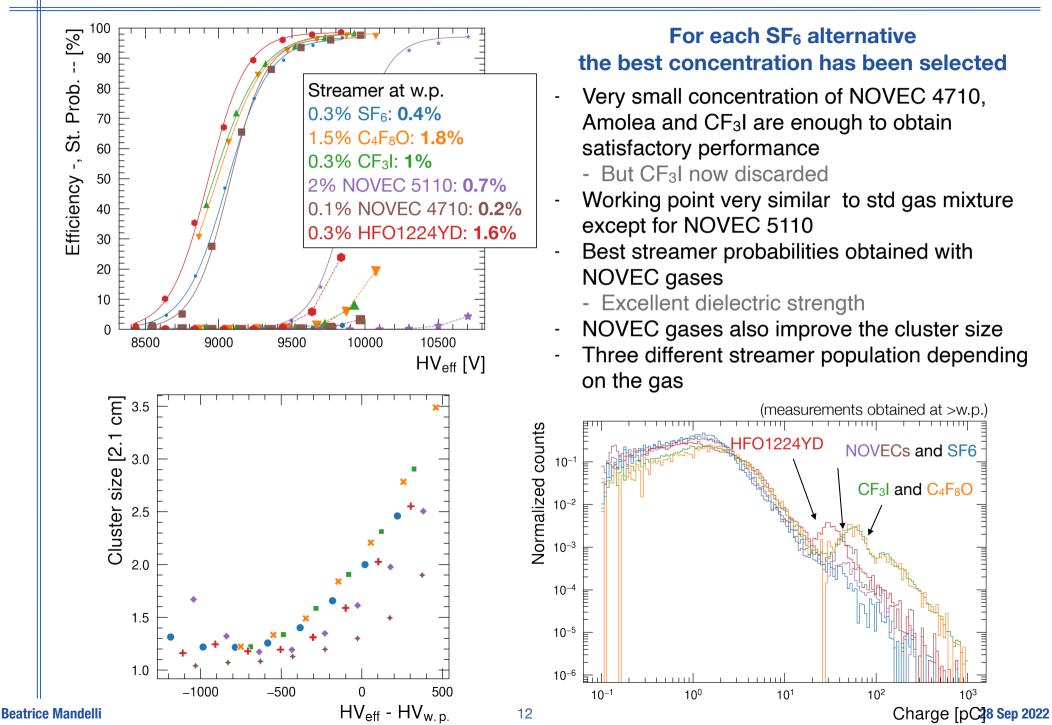


- Avalanche charge a bit higher than std gas mixture
- Higher working point for concentrations > 0.3%

Comparison of all tested gases



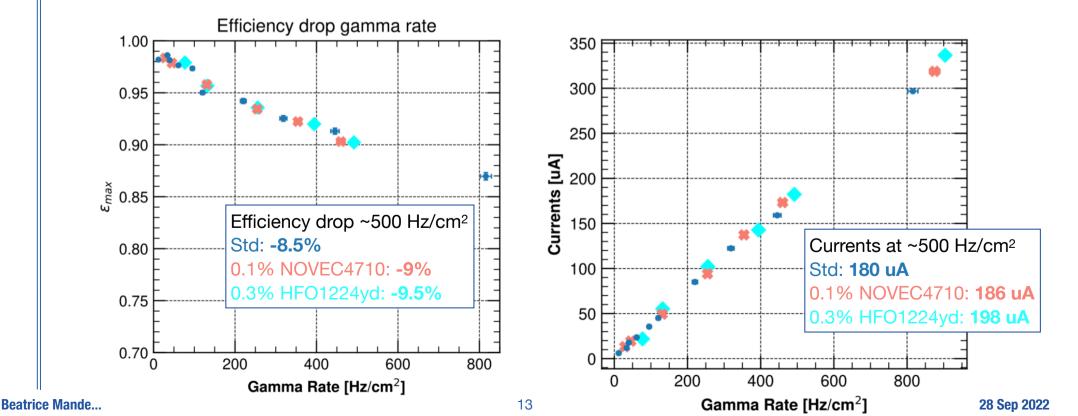
Comparison between best concentration



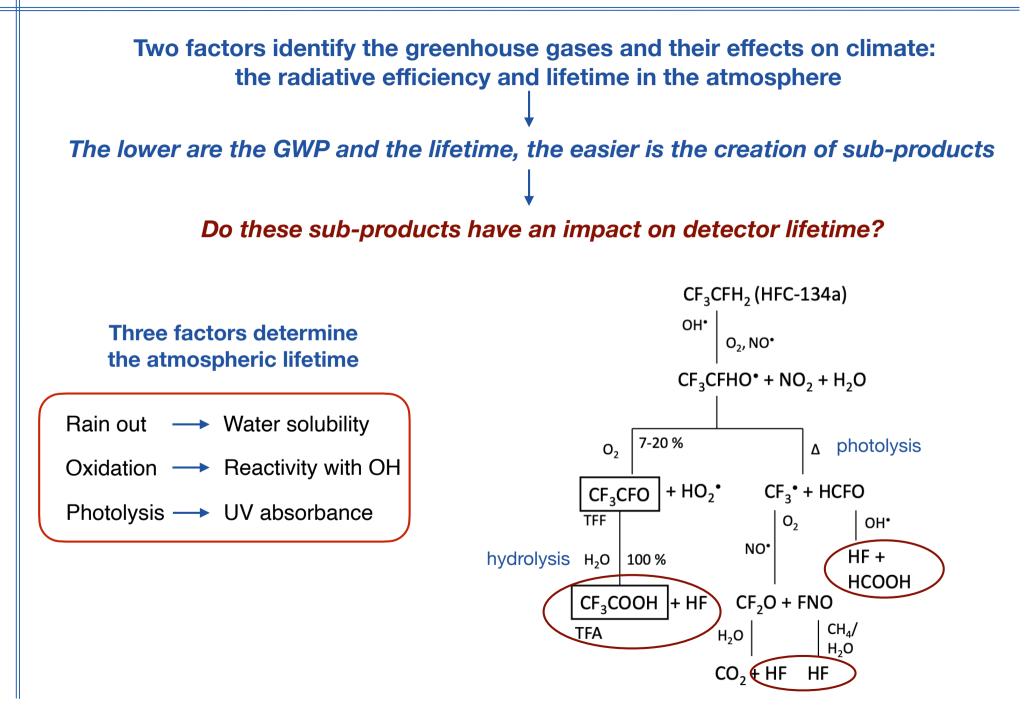
Testbeam with best candidates at GIF++

RPC tested at high gamma background rate and muon beam

- Selection of 0.1% NOVEC4710 and 0.3% AMOLEA HFO1224yd for testing at GIF++
 - Best compromise between streamer probability, working point and avalanche and streamer charge
 - NOVEC5110 to further test
- Small increase of currents and efficiency drop at high gamma rates
 - A bit worse for HFO1224yd
- Best option seems NOVEC4710 but undergoing studies since it reacts with H₂O
- HFO1224yd is also very promising
 - Streamer probability still a bit higher

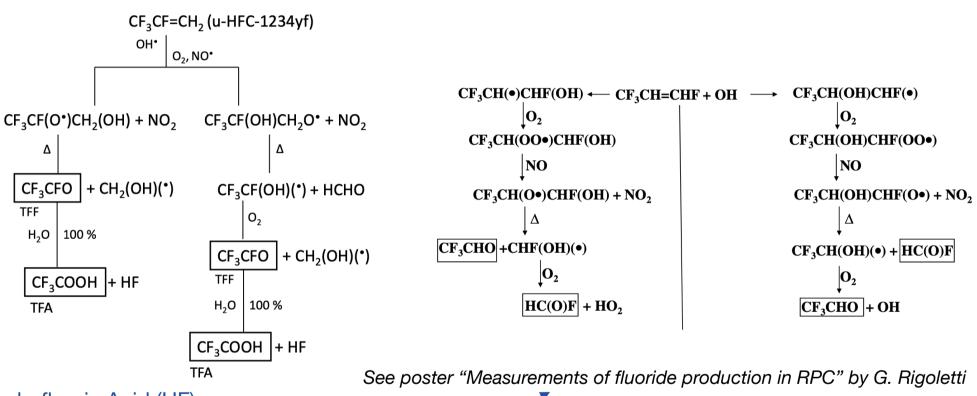


Not only detector performances....



HFO degradation

Atmospheric lifetime of HFO1234yf is 11 days Atmospheric lifetime of HFO1234ze is 18 days Atmospheric lifetime of R134a is 500 years



Hydrofluoric Acid (HF)

- It has already been measured that HFO produces much more HF than R134a in RPC detectors

Trifluoroacetic acid (TFA)

- HFO1234ze is estimated to break down into TFA at less than 10%, whereas R-1234yf will break down into TFA at 100% (R134a at 21%)
- TFA highly soluble: no formation of insoluble salts
- Phytotoxic

NOVEC 5110

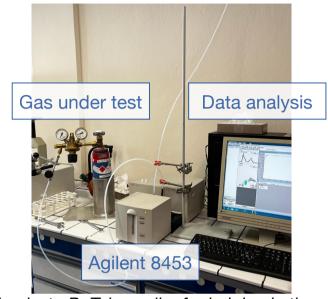


Rain out — water solubility (1ppmw)

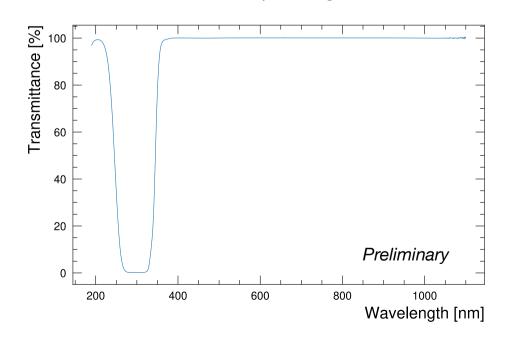
Oxidation --- unreactive with OH

Photolysis ---> strong absorbance in near UV (wavelength > 300 nm)

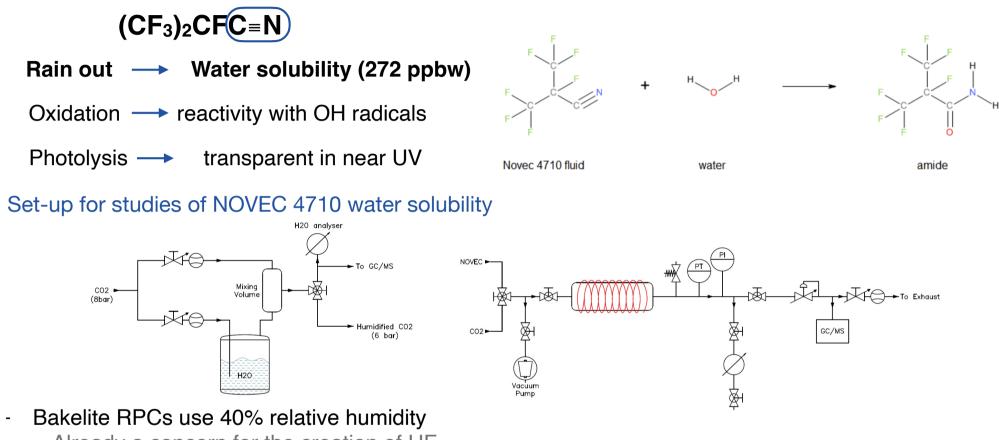
- Several gases tested: NOVEC 5110, NOVEC 4710, iC₄H₁₀, HFO, N₂, CO₂
- Agilent 8453 UV-visible Spectroscopy system
 - Wavelength range: 190-1100 nm
- First measurements in agreement with 3M company
- Works on-going to go to lower wavelengths
 - No commercially available instruments (custom made from companies >500 kCHF)



Thanks to B. Teissandier for helping in the measurements and providing the instrument

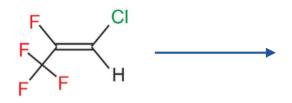


NOVEC 4710



- Already a concern for the creation of HF
- 3M recommend to use NOVEC gases in a dry environment
- Production of an amide from NOVEC 4710 + H₂O
 - Sub-products in the order of ppb
 - Solid at room T with a melting point of 49°C
 - The amide has appreciable vapor pressure at 60°C, it remains in gas phase at low concentrations
- Tests on-going in laboratory
 - Try to reproduce 3M tests
 - Analysis at the output of an RPC

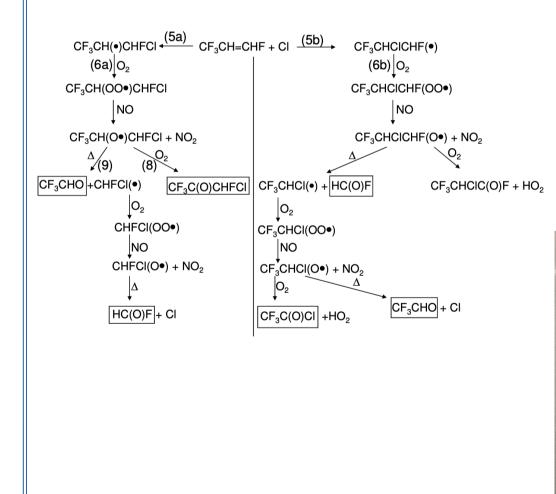
HFO-1224yd



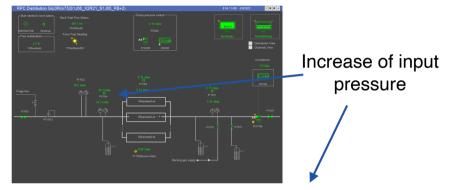
Degradation products: CF₃C(O)OH, CO₂, HF, HCI

TFA yield from atmospheric degradation very low

Reactions of HFO with CI



The CI case of CMS RPC system



presence of white dust at detector input!



24 24 35	2 31 32 33 V 14 V 14 V	* AL	
外市市	·告告之 · ································	A COLUMN	
		Float	balls
1	Element	Ball #1	Ball #2
18	С	5.3	6.1
	0	4.6	7.9
	F	0.0	0.0
	Na	0.0	0.1
	AI	0.0	0.0
	Si	0.7	1.8
	Cl	53.0	28.1
	Cr	0.0	8.6
1	Fe	36.5	47.2
15	Ni	0.0	0.3

Conclusions

Search for alternatives to SF₆

- It has a very high GWP and even if used in small quantities it accounts for a large fraction of the gas mixture GWP
- Not yet a problem as in industry SF6 is largely employed for HV insulating plants but new alternatives start to be used

Detector performance

- Several alternatives were tested in concentrations from 0.1% to 2% in the standard RPC gas mixture
- Very promising candidates with excellent streamer suppression
- Test performed in laboratory and at GIF++

Properties of new gases

- The new low-GWP gases can easily breaks in well-defined conditions (UV radiation, presence of OH, presence of humidity)
- Studies on-going to characterise these gases
- Detailed studies needed to understand possible effects on RPC long-term operation

Back-up slides

Alternatives to SF₆: search in HV industry

(a) 3	· · · ·	▲ 3 C-at	ome			Dielect	ric	GWP	Lifeti	me (y	vear)	٦	「 _b (∘C)
•		• 4 C-ate	oms		SF ₆	1		23900	ć	3200			-63.8
2					N ₂	0.36	6	0		-			-198
					CO ₂	0.3		1	30	0-100)0		-78
		; – –			CF ₄	0.4		6300	5	0000)		-128
					CF₃I	1.2		0.4	6	days	6		-21.8
					C ₄ F ₈ O	1.2		8000	>	3000)		-
0	<u> </u>			C ₄ F	7N (Novec 4710)	2		1490		30			-4.7
200 3	00 _ 1/1	0 50	0	C₅F₁	₀ O (Novec 5110)	1.5 -	2	1	15	5 day	S		27
	T _B [K]			C ₃ F ₄ I	HCI (HFO1224yd)	-		0.88	20) day	S		15
	Ļ					Nr		SMILES		E,	T _B	T _B L	
group	< <i>E</i> r> [rel. SF ₆]	<7 _B > [K]	< <i>z</i> >	n				C3-compounds					
C₃-Ketones C₃-Aldehydes C₃-Acyl Fluorides	0.39 0.56 0.98	319 320 293	2.00 1.89 3.15	17 18 13		(1) (2) (3) (4)	0=C(C(F	X(C(F)(F)F)(F)F X(F)F)C(F)(F)F =0)F)(F)F)F C(F)(F)F		1.34 1.03 1.11 1.31	255 262 263 278	280 280 288 302	
C₄-Ketones C₄-Aldehydes C₄-Acyl Fluorides	0.69 0.77 1.14	352 345 326	2.58 2.39 3.36	138 117 110		(5) (6) (7)	FC(C(C(F	C₄ –compounds (C(C(F)(F)F)F)(F)F F)(F)F)(C(=0)F)F)F)(F)F)(C(=0)F)F		1.63 1.82 1.91	270 277 295	317	
C₅-Ketones C₅-Aldehydes C₅-Acyl Fluorides	0.94 1.00 1.35	375 370 356	3.30 2.91 3.85	923 685 590		(8) (9) (10)	0=C(C(C	C ₅ –compounds (C(F)(F)F)(C(F)(F)F)C(F (F)(F)F)(C(F)(F)F)F)C(F (F)(F)F)(C(F)(F)F)(C(=0)F)F	-)(F)F -)(F)F	2.77 1.93	283 293	303 310	
SF ₆	1	209 [18]	6	1]	(10) (11) (12)	0=C(C(C	~(~(F)(F)(F)(F)(C)(=0)(F) >(C(F)(F)(F)(F)(F)(F)(F)(F) >(C(C(F)(F)(F)(F)(F)(F)(F)(F)(F)(F)(F)(F)(F)(- F)(F)F	2.28 2.01 2.67	296 302 304	322	

M. Rabie, C. Franck, predicting the electric strenght of proposed sf6 replacement gases by means of density functional theory

By-products under Arcing Conditions

SF_6

Data provided by 3M Company

Compound	CAS no.	Molecular formula		
Bispentafluorosulfur oxide	42310-84-9	S ₂ OF ₁₀		
Carbon tetrafluoride	75-73-0	CF ₄		
Carbonyl fluoride	353-50-4	COF ₂		
Fluorine	7782-41-4	F ₂		
Hydrogen fluoride	7664-39-3	HF		
Hydrogen sulfide	7783-06-4	H ₂ S		
Nitrogen trifluoride	7783-54-2	NF ₃		
Oxygen difluoride	7783-41-7	F ₂ O		
Silicon tetrafluoride	7783-61-1	SiF ₄		
Sulfur dioxide	7446-09-5	SO ₂		
Sulfur fluoramide fluorine	81625-84-5	(SF ₅) ₂ NF		
Sulfur fluoride fluorosulfate	81439-35-2	$S_2O_3F_6$		
Sulfur fluoride peroxide	12395-41-4	$S_2O_2F_{10}$		
Sulfur pentafluoride	5714-22-7	S2F10		
Sulfur tetrafluoride	7783-60-0	SF ₄		
Sulfur tetrafluoride oxide	13709-54-1	SOF ₄		
Sulfuryl fluoride	2699-79-8	SO_2F_2		
Thionyl fluoride	7783-42-8	SOF ₂		
Trifluoromethyl sulfur pentafluoride	373-80-8	SF5CF3		

3M NOVEC 4710

Compound	Concentration (ppm)
CO ₂	935018
(CF ₃) ₂ CFCN (Novec 4710)	40600
СО	24000
CF ₂ =CFCN	130
CNCN	65
C ₂ F ₅ CN	60
CF ₃ CN	58
(CH ₃) ₂ SiF ₂	52
COF ₂	14
(CF ₃) ₂ CHCN	1.9
$(CF_3)_2C=CF_2$	1.3

Calculated LC_{50} value for arced-gas mixture = 39000 ppmv

3M NOVEC 5110

Compound	Concentration (ppm)
CO ₂	61000
СО	6200
HF	690
CF ₄	100
(CF ₃) ₂ CFC(O)CF ₃ (Novec 5110)	39
C ₃ F ₈	19
C ₃ F ₇ H	<15
C ₃ F ₆	14
C ₂ F ₆	<10
C ₄ F ₈ O C ₄ F ₁₀	5
C ₄ F ₁₀	2

Calculated LC_{50} value for arced-gas mixture = 122000 ppmv