

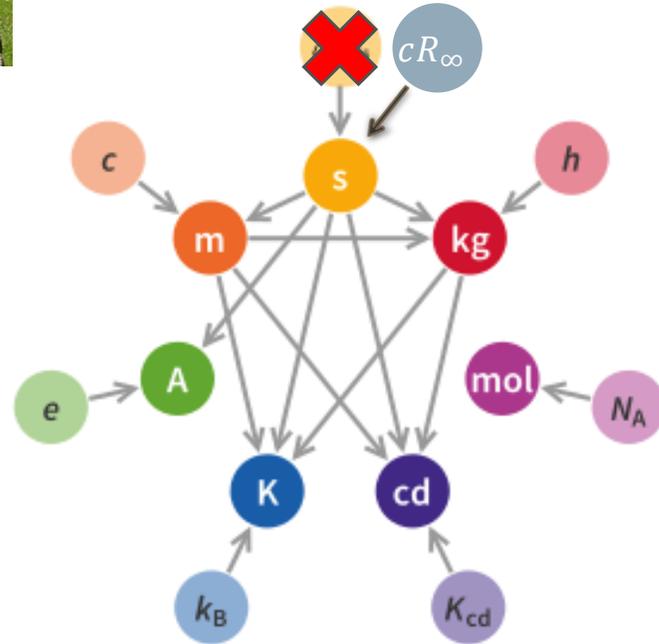
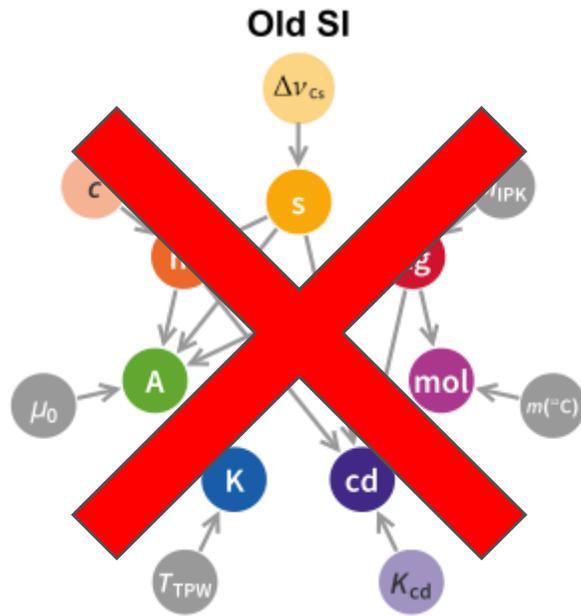
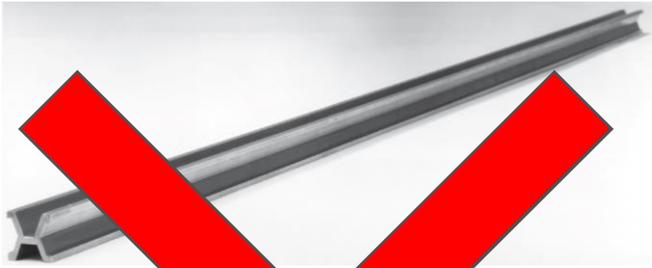
# Towards a hydrogen optical clock

Dr. Omer Amit

Laser spectroscopy division

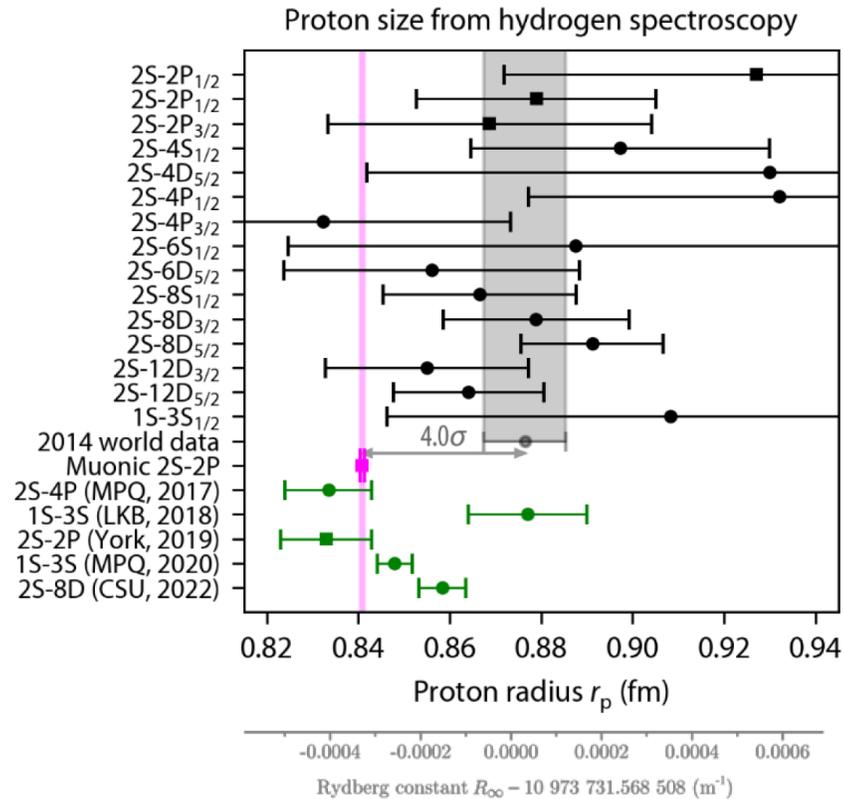
Max-Planck-Institute for Quantum Optics, Garching

23.03.2022



# Goals

- The 1S – 2S transition in hydrogen can be calculated exactly.
- Define the “computable second” based on fundamental constants.



# Goals

- The 1S – 2S transition in hydrogen can be calculated exactly.
- Define the “computable second” based on fundamental constants.
- Reduce uncertainties in the 1S – 2S measurement.

$$E_{nlj} = hcR_\infty \left[ -\frac{1}{n^2} + f_{nlj} \left( \alpha, \frac{m_e}{m_p}, \dots \right) + \delta_{l,0} \frac{C_{NS}}{n^3} r_p^2 \right]$$



# Outline

- Why trap hydrogen? Why hydrogen optical clock?
- Why trapping hydrogen is hard?
- How to trap hydrogen?
- What is being done right now?



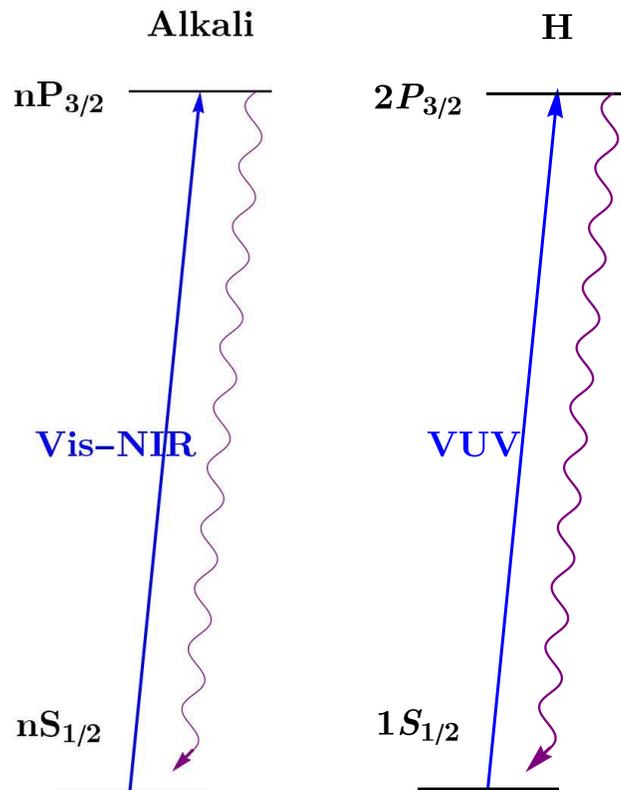
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# Why trapping hydrogen is hard?

- First excited level is very energetic.
- Small mass.



	H	Rb
Mass	1 u	87 u
Wavelength	121 nm	780 nm
Linewidth	100 MHz	6 MHz
Doppler Temp.	2.3 mK	145 $\mu$ K
Recoil Temp.	1.2 mK	360 nK
Recoil velocity	3.2 m/s	5.6 mm/s
Doppler shift	27 MHz	7.5 kHz

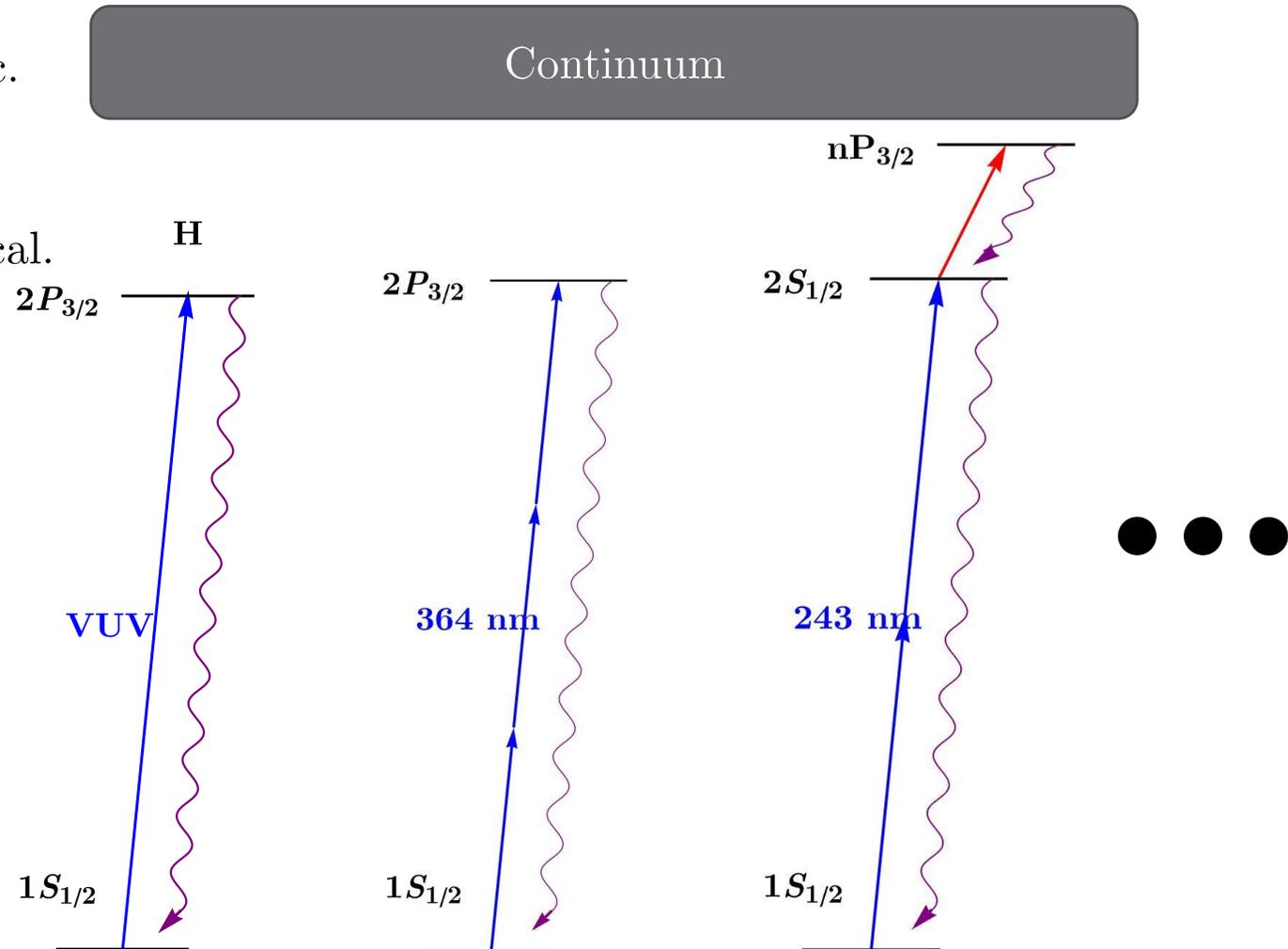
“Alkali”

1 IA 1A	2 IIA 2A	3 IIIB 3B	4 IVB 4B	5 VB 5B	6 VIB 6B
1 H Hydrogen 1.008	4 Be Beryllium 9.012				
3 Li Lithium 6.941	12 Mg Magnesium 24.305				
11 Na Sodium 22.990	20 Ca Calcium 40.078	21 Sc Scandium 44.956	22 Ti Titanium 47.867	23 V Vanadium 50.942	24 Cr Chromium 51.996
19 K Potassium 39.098	38 Sr Strontium 87.62	39 Y Yttrium 88.906	40 Zr Zirconium 91.224	41 Nb Niobium 92.906	42 Mo Molybdenum 95.94
37 Rb Rubidium 85.468	56 Ba Barium 137.328	57-71 Lanthanide Series	72 Hf Hafnium 178.49	73 Ta Tantalum 180.948	74 W Tungsten 183.84
55 Cs Cesium 132.905	88 Ra Radium 226.025	89-103 Actinide Series	104 Rf Rutherfordium [261]	105 Db Dubnium [262]	106 Sg Seaborgium [266]
87 Fr Francium [223.02]					
		57 La Lanthanum 138.905	58 Ce Cerium 140.116	59 Pr Praseodymium 140.908	
		89 Ac Actinium 227.028	90 Th Thorium 232.038	91 Pa Protactinium 231.036	



# Why trapping hydrogen is hard?

- First excited level is very energetic.
- The 2S level can be easily ionized.
- Standard laser cooling is impractical.





# Outline

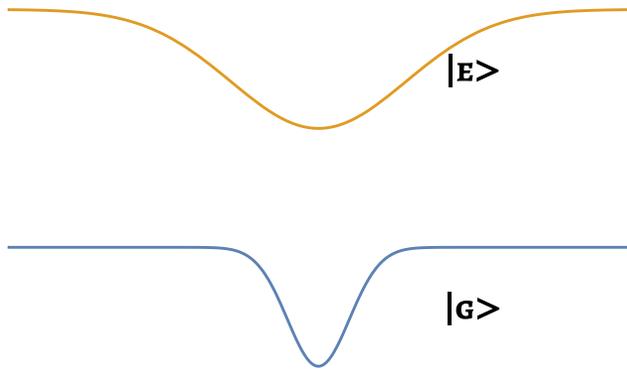
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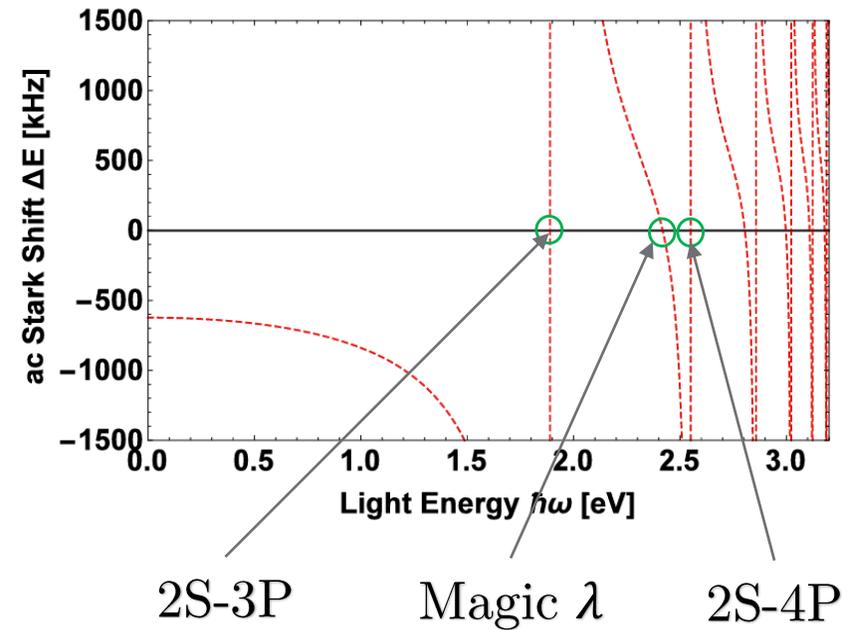
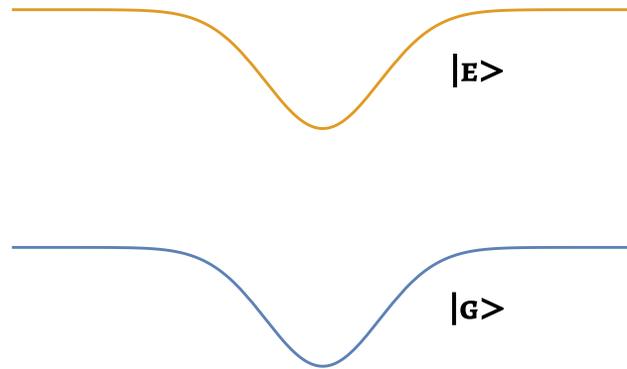
# Magic wavelength dipole trap

- Magic wavelength  $1S \rightarrow 2S$ ,  $\lambda = 514.6 \text{ nm}$

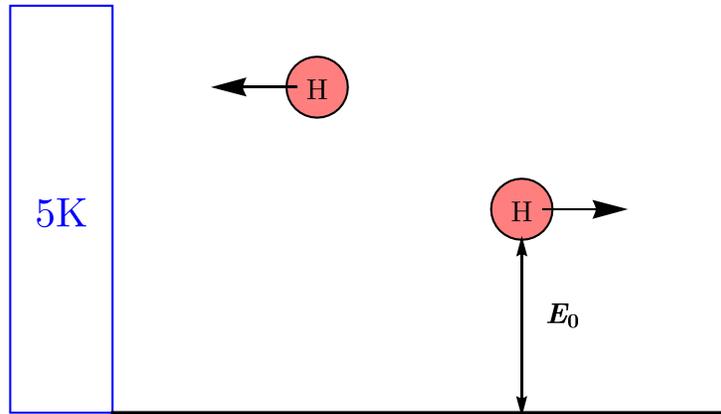
Non-magic



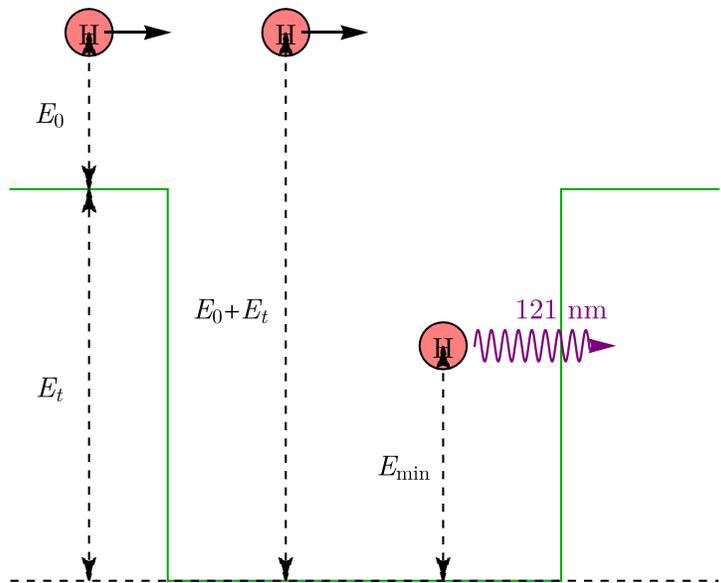
Magic



# How to trap hydrogen?

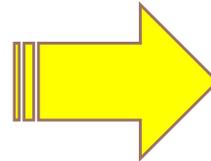


Thermalization to 5 K

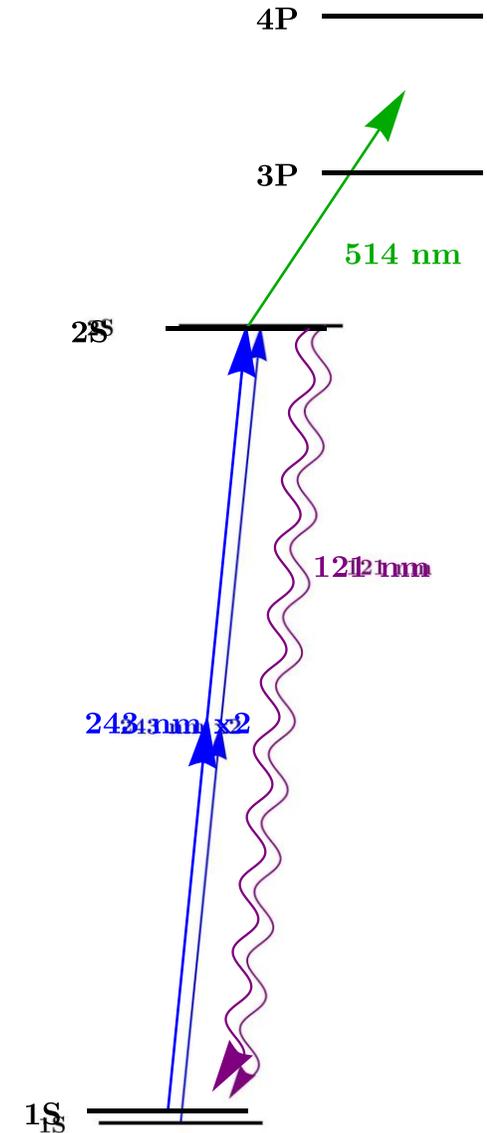
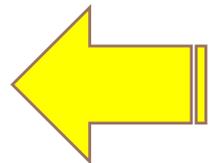


Quenching and momentum kick

$\sim 10^{19}$  atoms/s



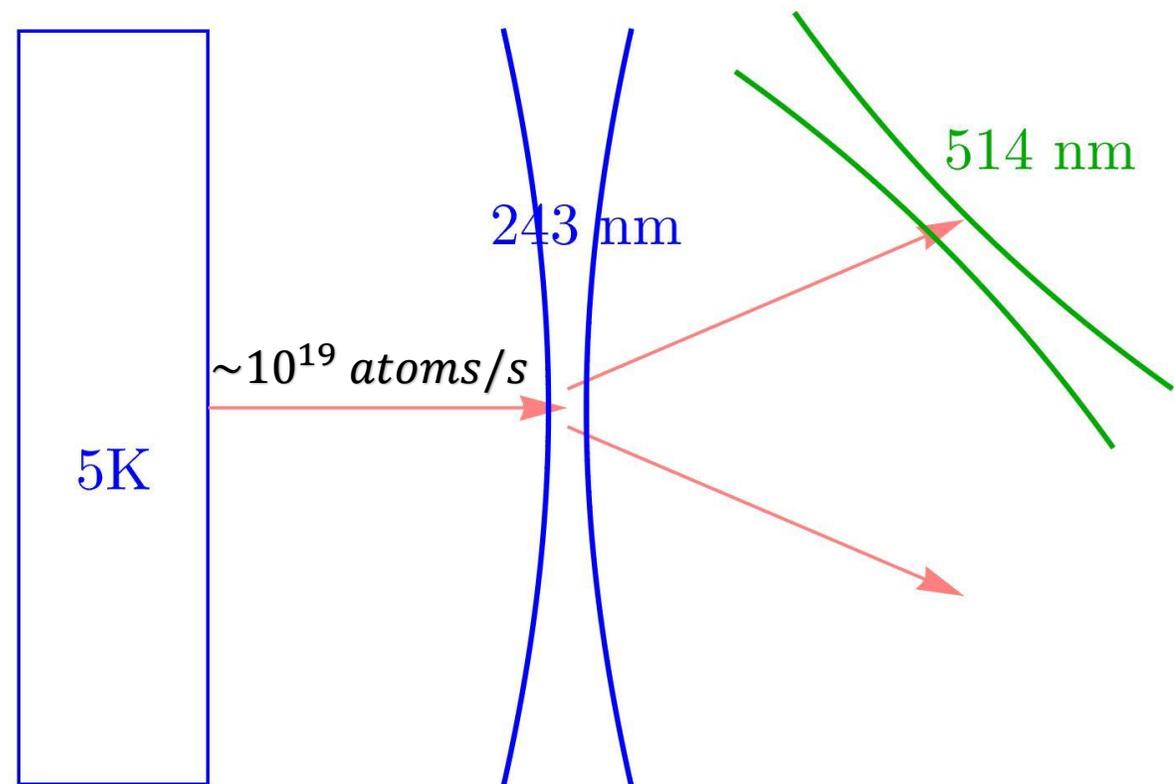
2-photon excitation  
1S - 2S





# How to trap hydrogen?

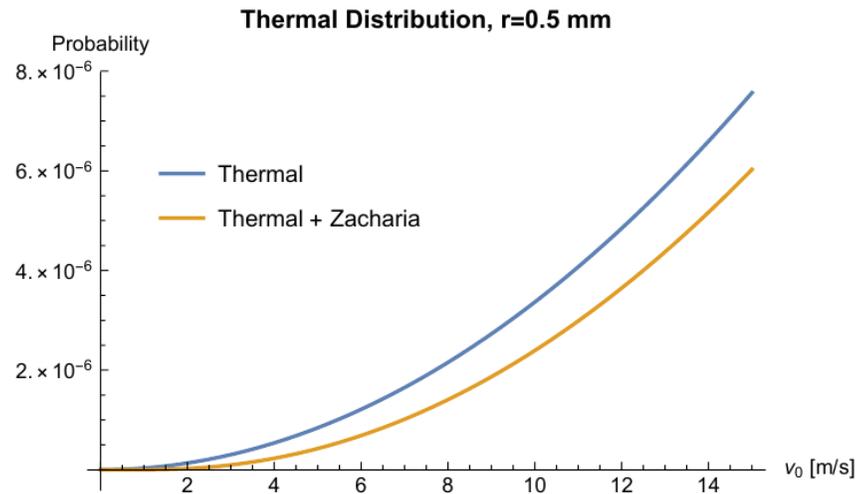
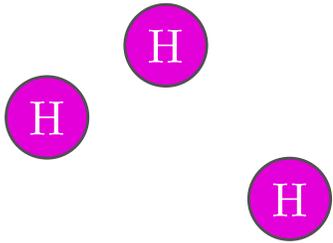
- Using the doppler-sensitive 2-photon transition.
- In a FP cavity, two beams are produced.
- Deflection angle is velocity-dependent.
- Produce cold, low-velocity atomic beam.
- Interesting for 2S-nP spectroscopy.





# Zacharias effect and the tale of the missing slow atoms

- Slow atoms are being pushed out of the atomic beam by fast moving atoms.

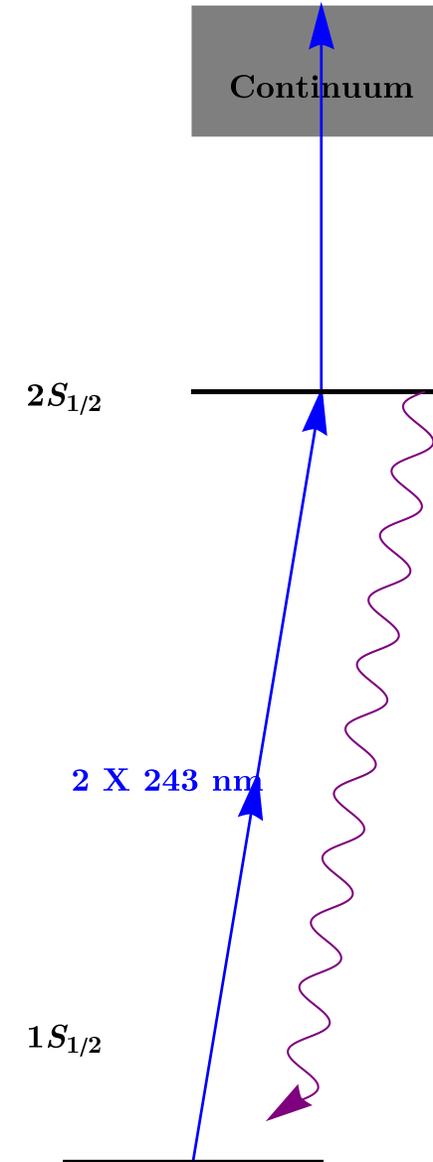
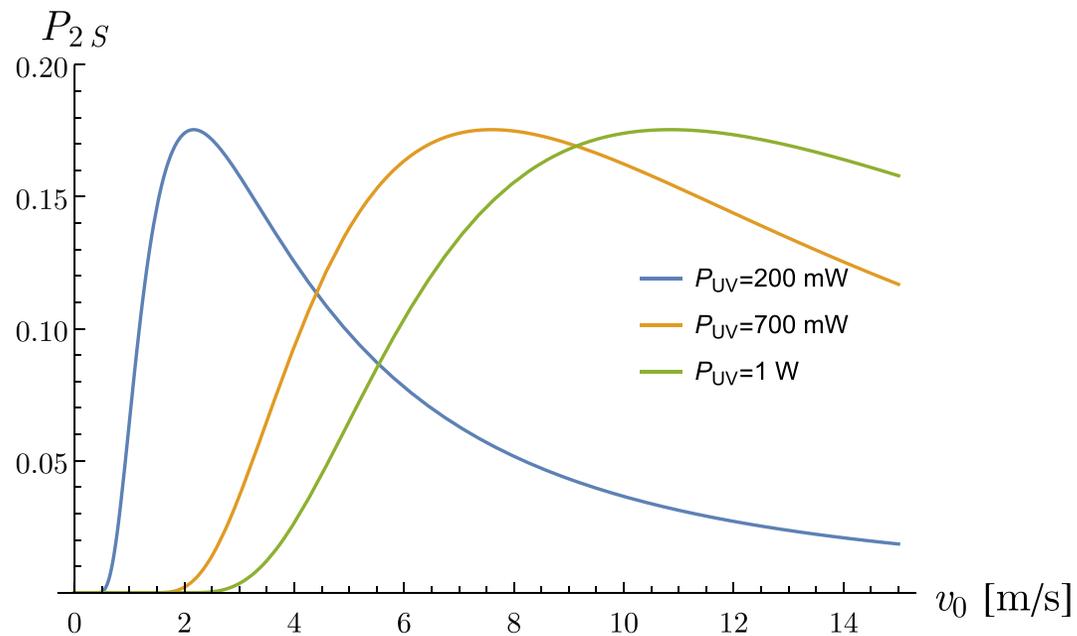


I. Estermann et al. Physical Review **71**, 4 (1947)



# Excitation probability

- Excitation using 2-photon transition  $1S - 2S$ .
- 3-photon ionization limits the efficiency of the process.
- Max efficiency around 17%.
- Starting from  $10^{19}$  atoms/s, still leaves  $10^{18}$  atoms/s.

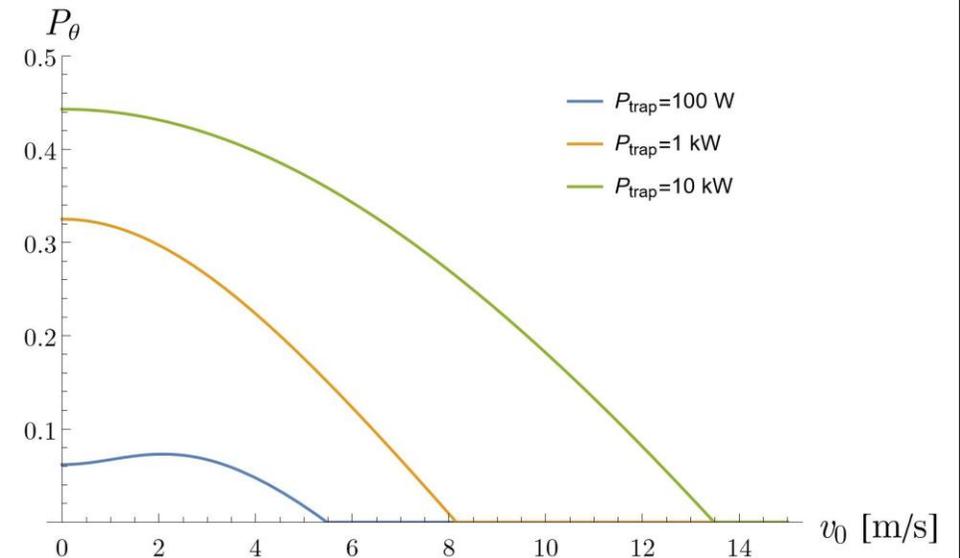
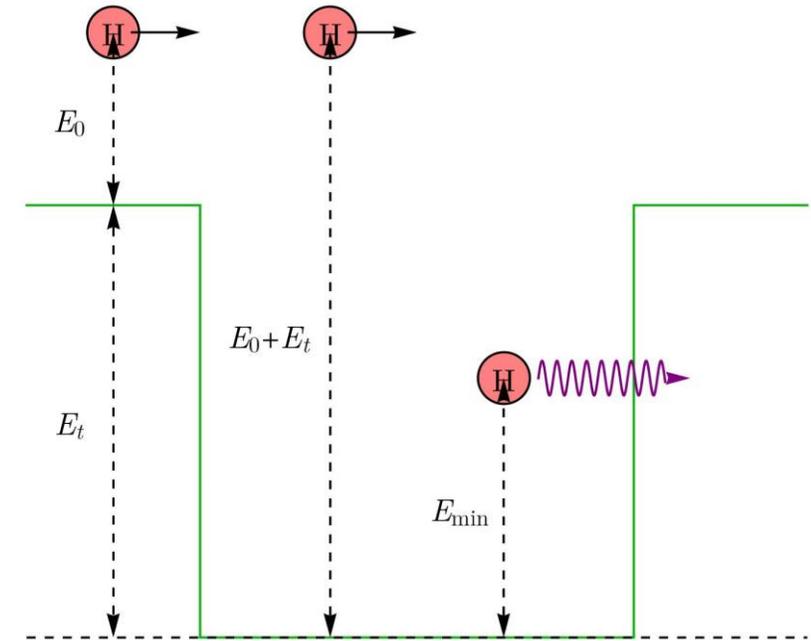




# Trapping probability

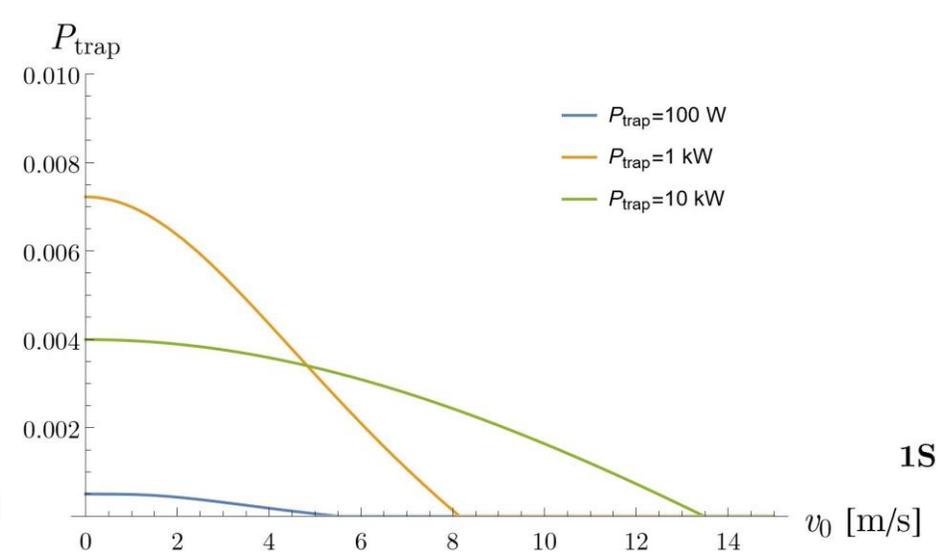
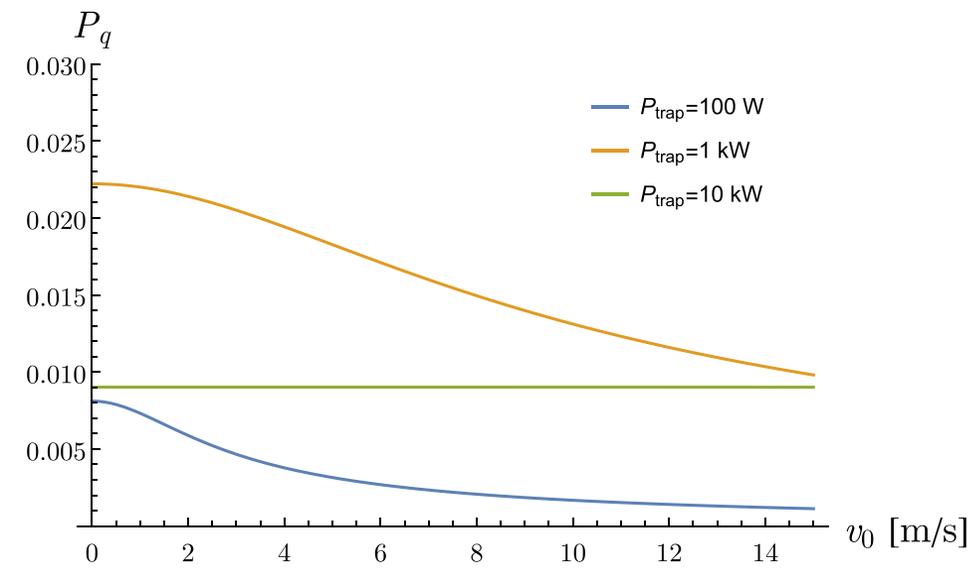
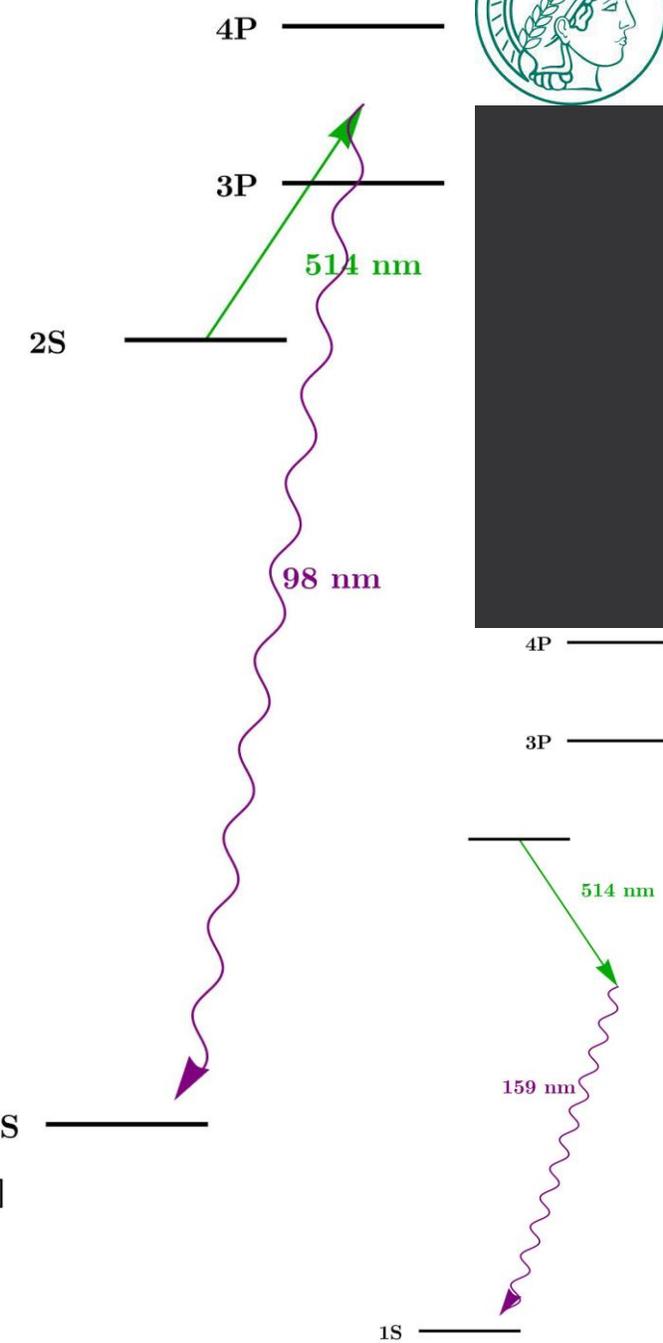
- Initial kinetic energy  $E_0 = \frac{m}{2} v_0^2$
- Maximal velocity  $v_{max} = \sqrt{\frac{2(E_0 + E_t)}{m}}$
- Velocity after recoil  $\vec{v}_f = (v_{max} - v_{rec} \cdot \cos \theta, v_{rec} \cdot \sin \theta)$
- Energy after recoil  $E_{min} = \frac{m}{2} |v_f|^2 < E_t$
- Trapping condition  $\theta_{max} = \cos^{-1} \left( \frac{E_0 + E_{rec1} + E_{rec2}}{2\sqrt{E_{rec2}(E_0 + E_t + E_{rec1})}} \right)$
- Probability  $P_\theta = (1 - \cos \theta_{max})/2$

**Only the slow atoms are**  
**trappable**



# Trapping probability

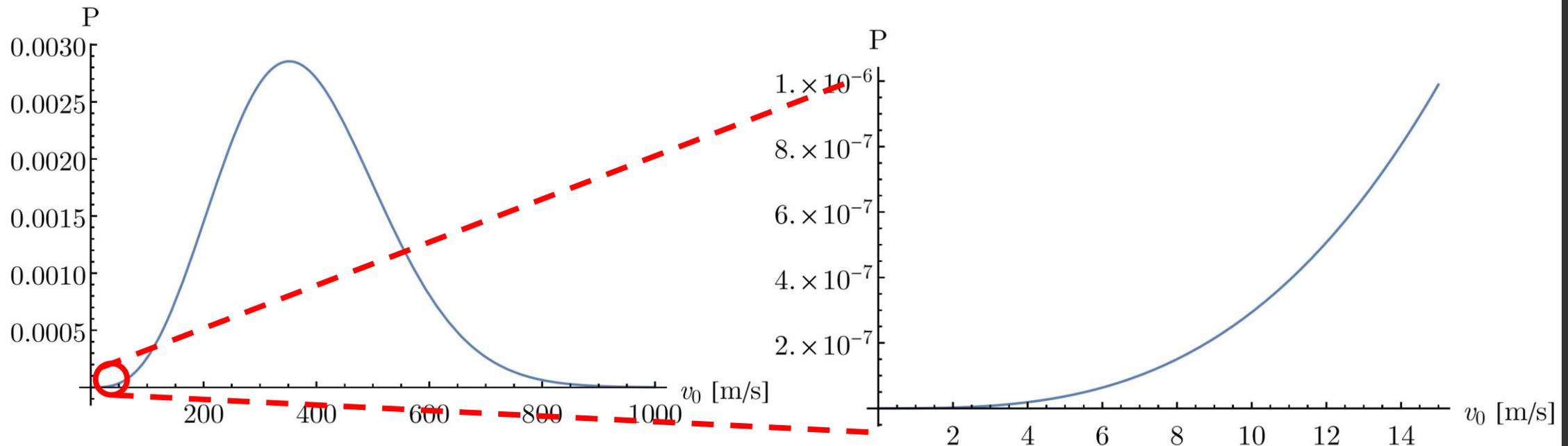
- Quenching of the 2S state is done by mixing 3P and 4P using the trapping laser.
- 2-photon ionization limits the maximum efficiency of this process.
- $P_{trap} = P_{\theta} \cdot P_q$
- $10^{18}$  atoms/s, still leaves  $10^{15}$  atoms/s.





# Thermal flux distribution

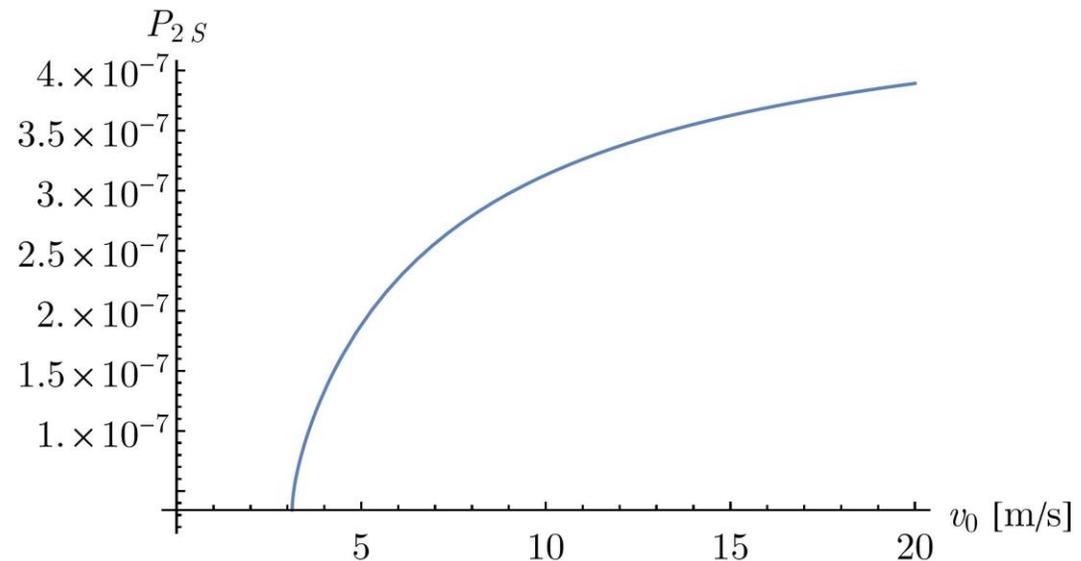
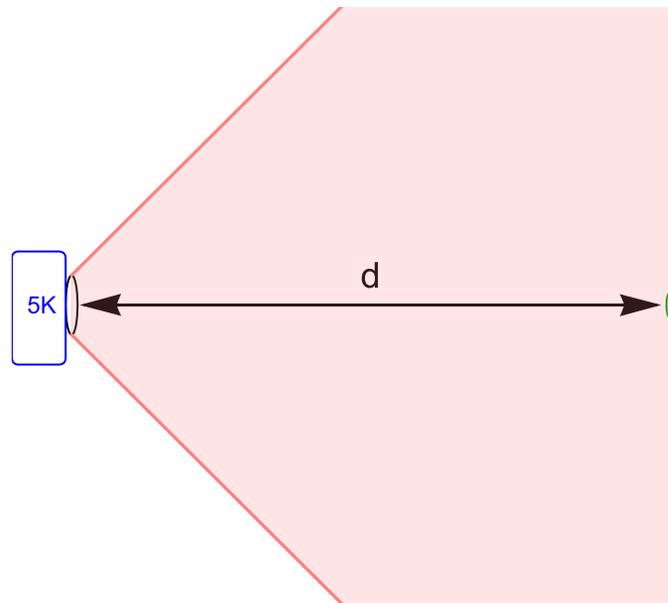
- High energy photon + Small mass = single recoil is enough to trap an atom.
- Trap only the tail of the thermal distribution.
- $10^{15}$  atoms/s, still leaves  $10^8$  trappable atoms/s.





# Transport

- Atoms leaving the nozzle need to hit the trapping volume.
- $P_{hit} = A/d^2$
- Very small fraction of atoms reach the trap.  $P_{hit} \approx 5 \cdot 10^{-7}$  for  $d = 0.5$  m
- Atoms need to survive in the 2S state (0.12 second lifetime).
- $10^8$  atoms/s, still leaves  $10^1$  trappable atoms/s.





# Outline

- Why trap hydrogen? Why hydrogen optical clock?
- Why trapping hydrogen is hard?
- How to trap hydrogen?
  - Some extra tricks for trapping hydrogen
- What is being done right now?



# Pre-deacceleration using magnetic fields

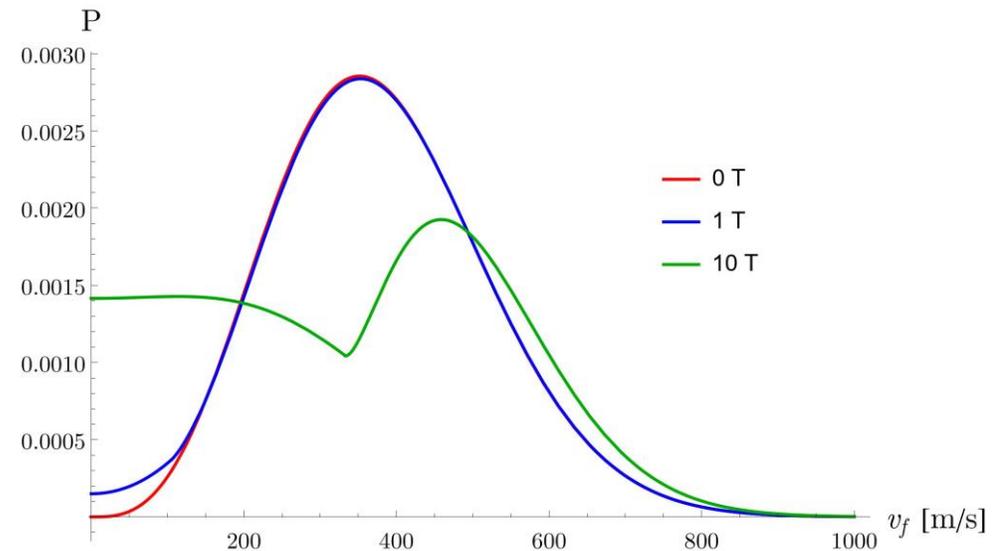
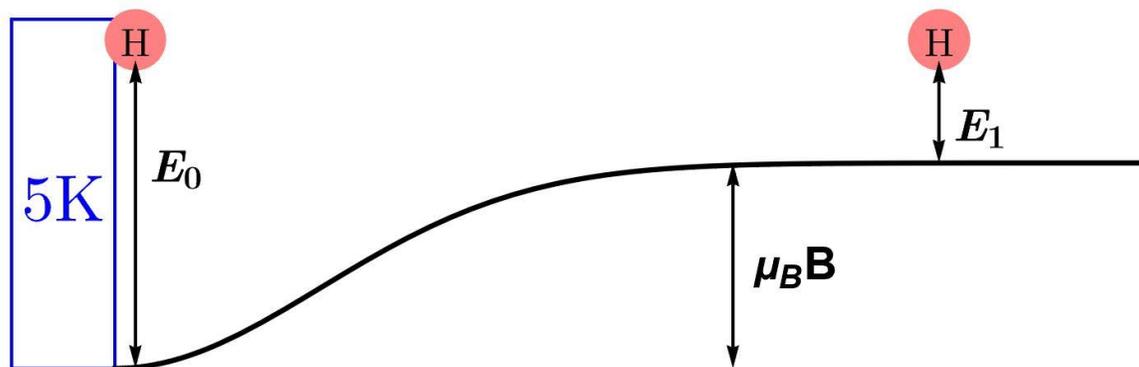
- If an atom is born in a potential well, it will have to trade away its kinetic energy to reach the trap.

- Final velocity  $v_1 = \sqrt{v_0^2 \pm 2\mu_B B/m}$

- Low-field seeking atoms are accelerated.

- High-field seeking atoms are deaccelerate.

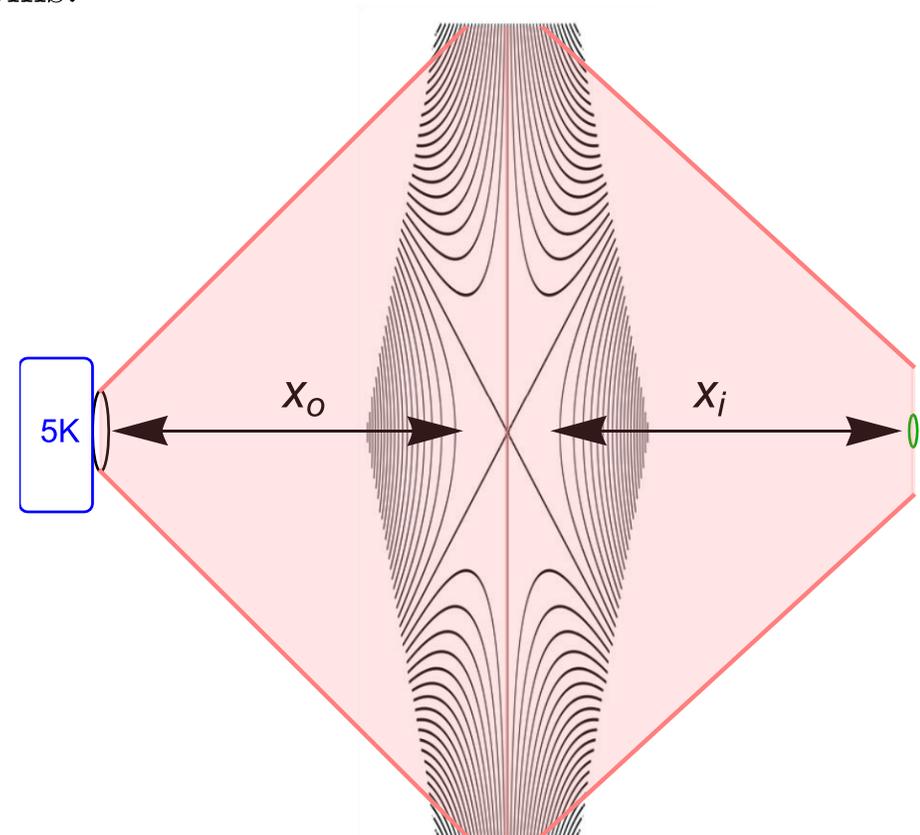
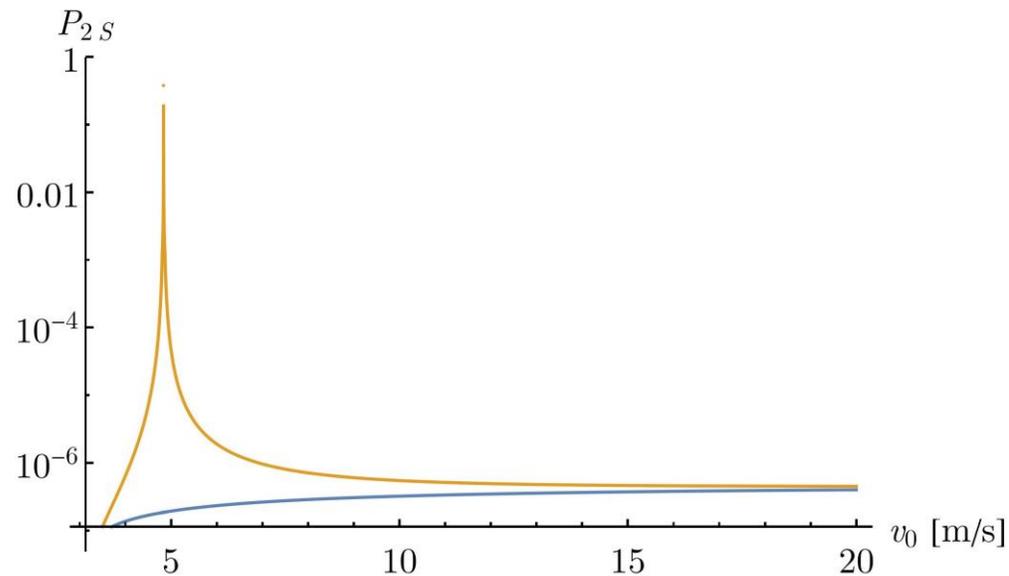
- $10^1$  atoms/s, leaves  $10^4$  trappable atoms/s.





# $P_{hit}$ with magnetic focusing

- $P_{hit} = \frac{A}{r_i(v_0)^2}$
- $r_i(v_0) = x_i + (x_o + r) \left(1 - \frac{x_i}{f(v_0)}\right)$ ,  $f(v_0) = \frac{\pi m R^3}{6 \mu_0 \mu_B I L} v_0^2$
- Velocity dependent. More slow atoms less fast atoms.
- Only focus low-field-seeking atoms.
- $10^4$  atoms/s, leaves  $10^8$  trappable atoms/s.





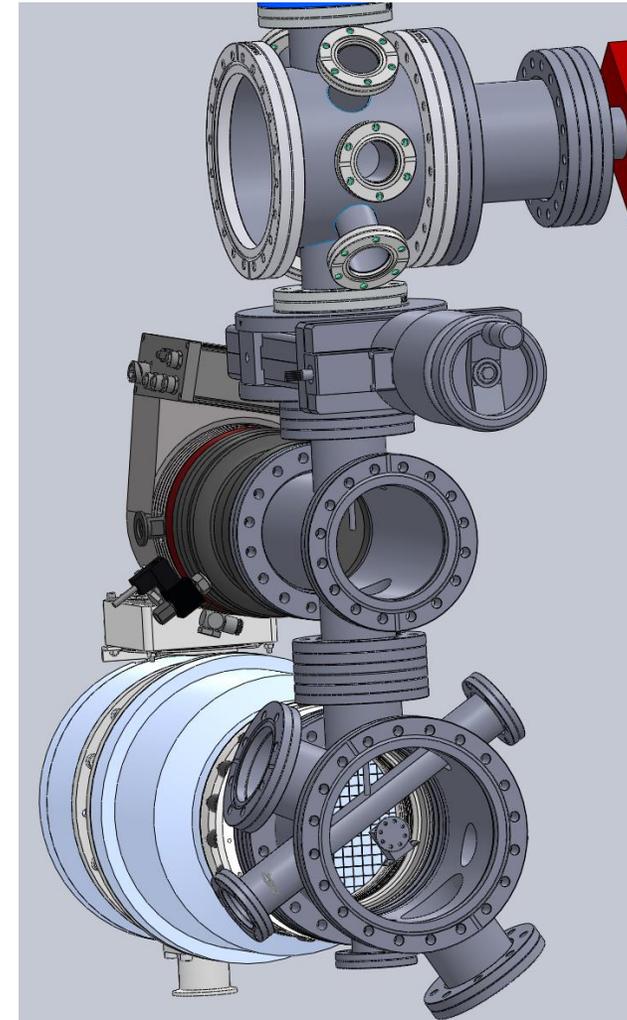
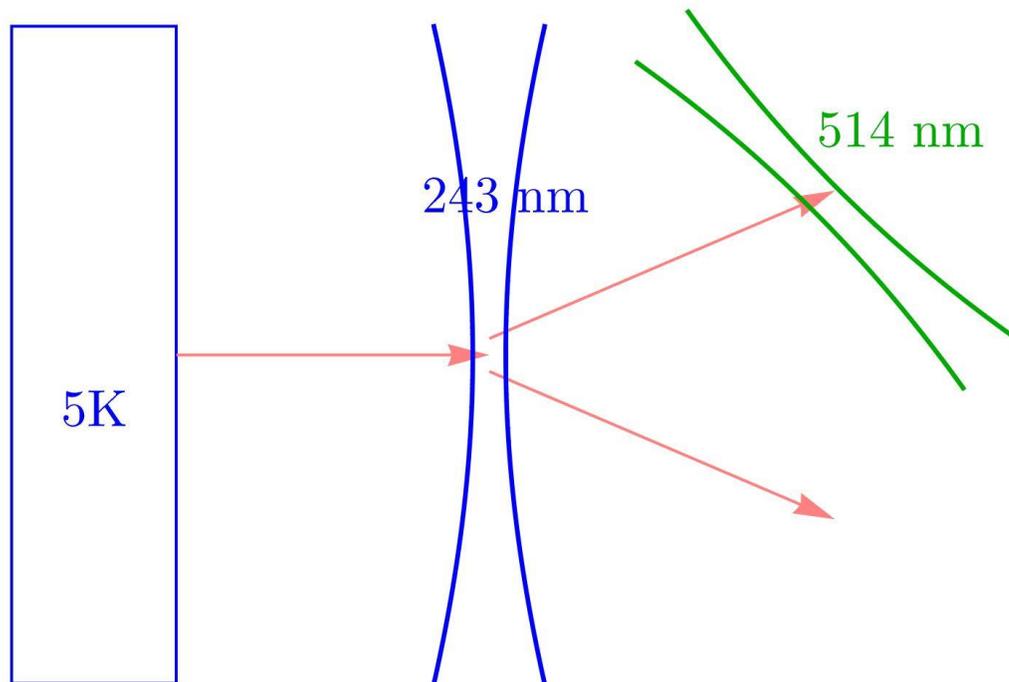
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# What is being done right now?

- Construction of the system.
- Construction of the lasers.
- More simulations.
- First Goal: Measure a low-velocity atomic beam.



Thank you