



Vacuum characterization capabilities

Ivo Wevers

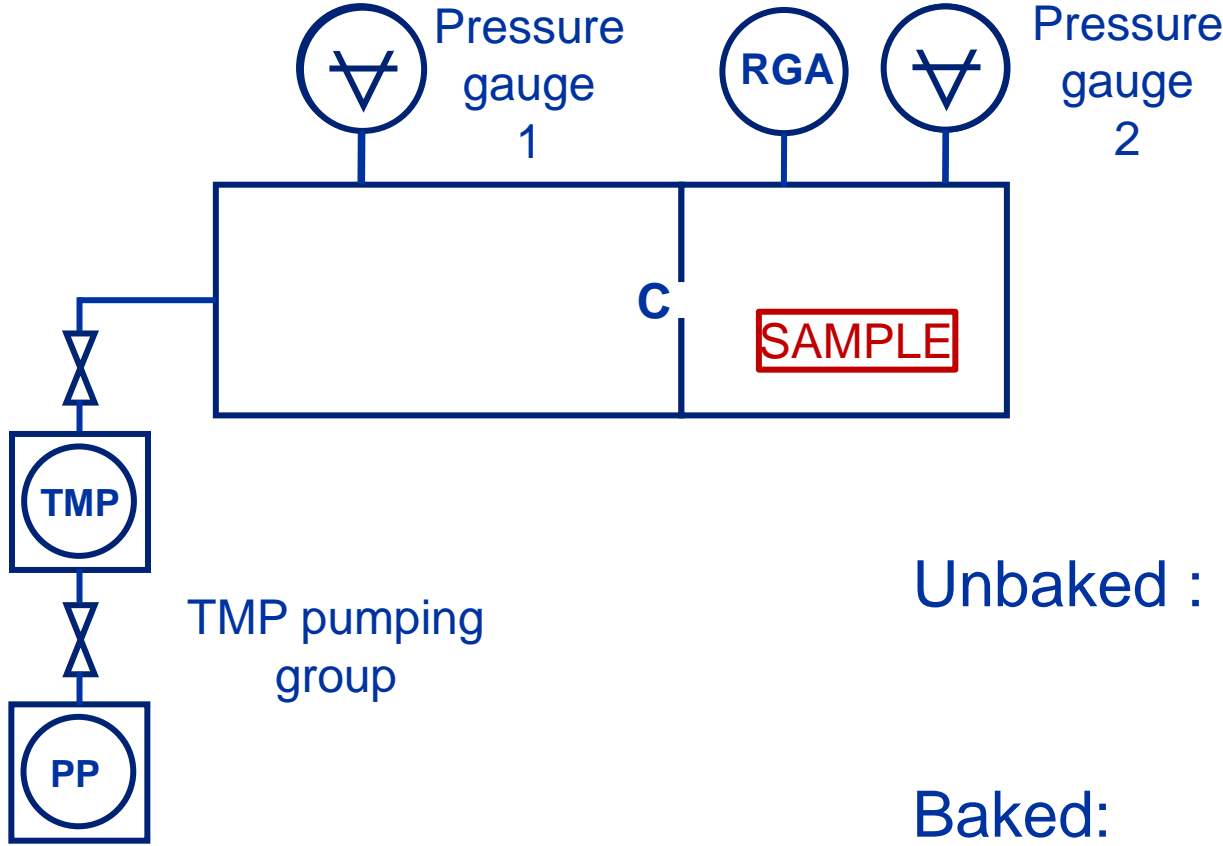
April 2022

Physics Beyond Colliders workshop

Content

- Outgassing:
 - ❖ Throughput method
 - ❖ Coupled method
- Thermal Desorption Spectroscopy - TDS
- Total Mass Loss
- Electron Stimulated Desorption - ESD
- Pumping speed
- Permeation

Outgassing measurement – throughput method



$$Q = \left(\frac{\overbrace{C (P_2 - P_1)}^{\text{Sample \& background}} - \overbrace{C (P_{2bg} - P_{1bg})}^{\text{background}}}{A_{\text{sample}}} \right)$$

Q in mbar l s⁻¹ cm⁻²

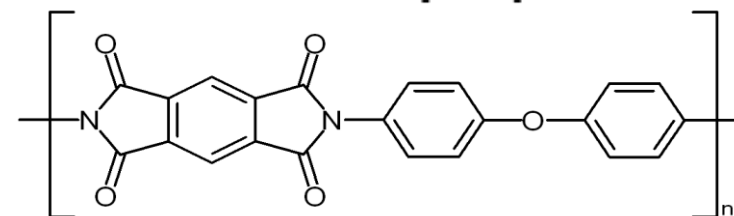
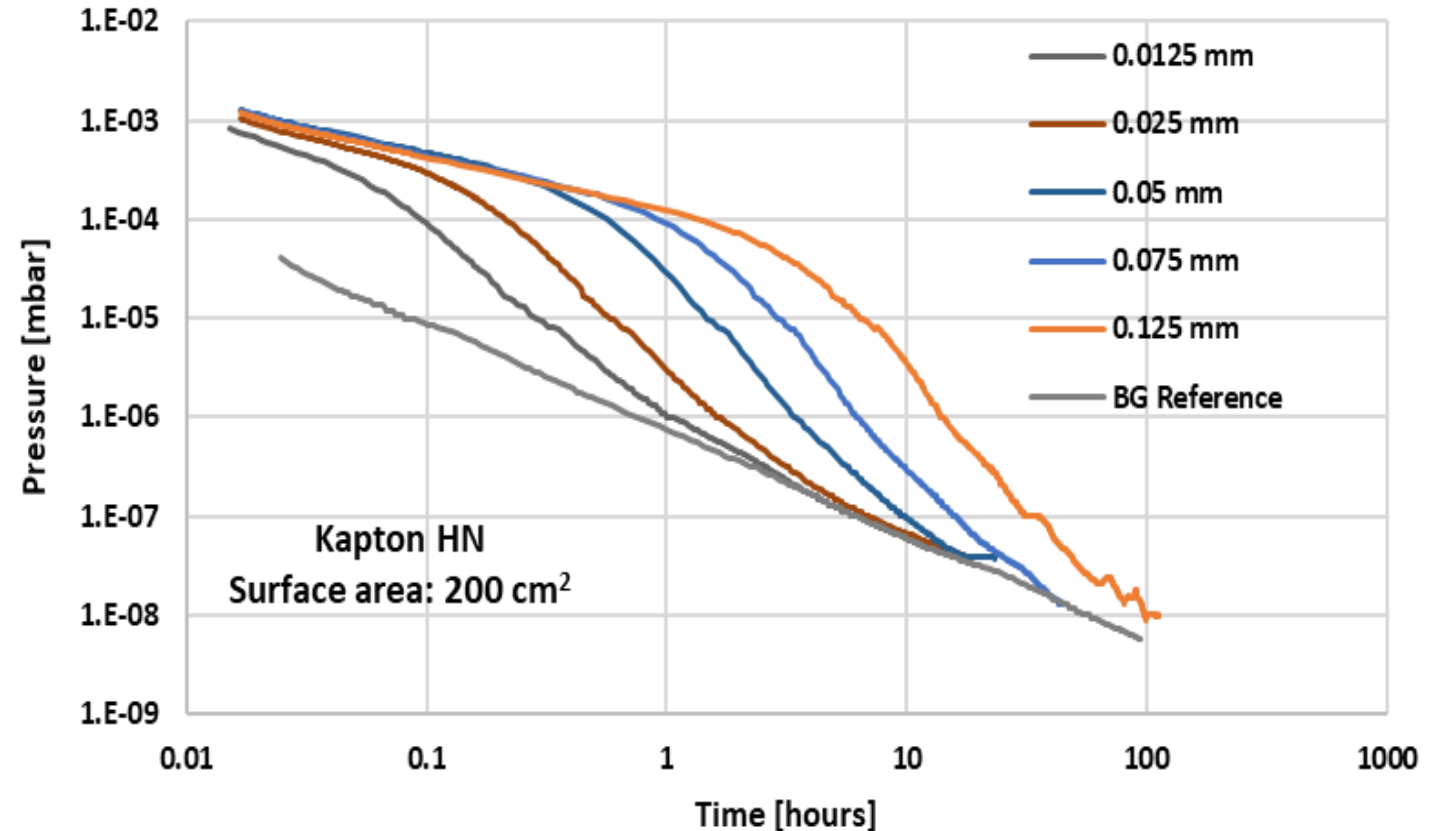
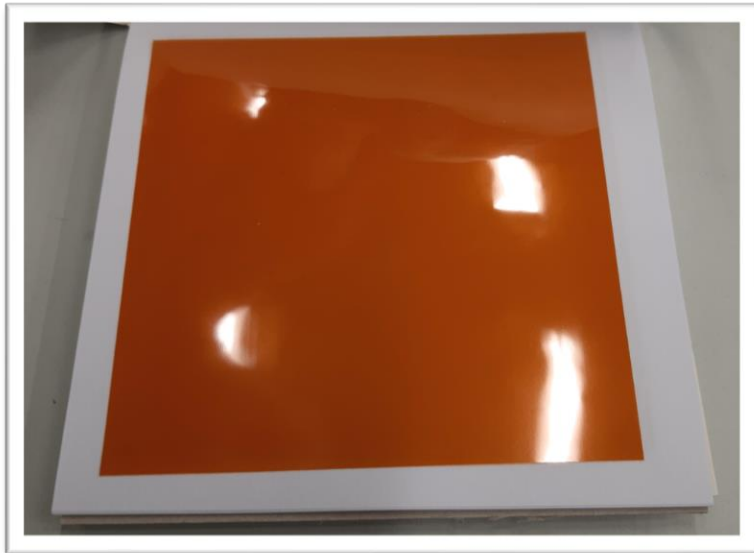
- Unbaked : outgassing of water – time dependent
Cold ionization gauges
- Baked: outgassing of (mainly) hydrogen
cold/hot ionization gauges

Outgassing measurement - throughput method

Kapton HN

Polyimide - Melting temperature: none
Expected water content: 0.96%

Samples: square sheets side 100 mm
Thicknesses: 0.0125-0.125 mm

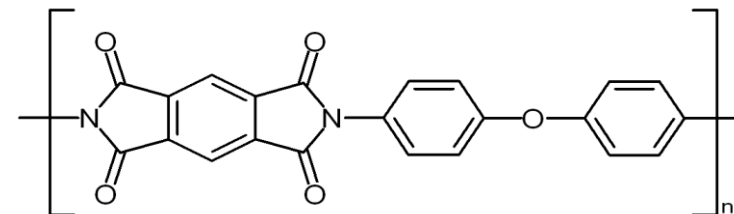
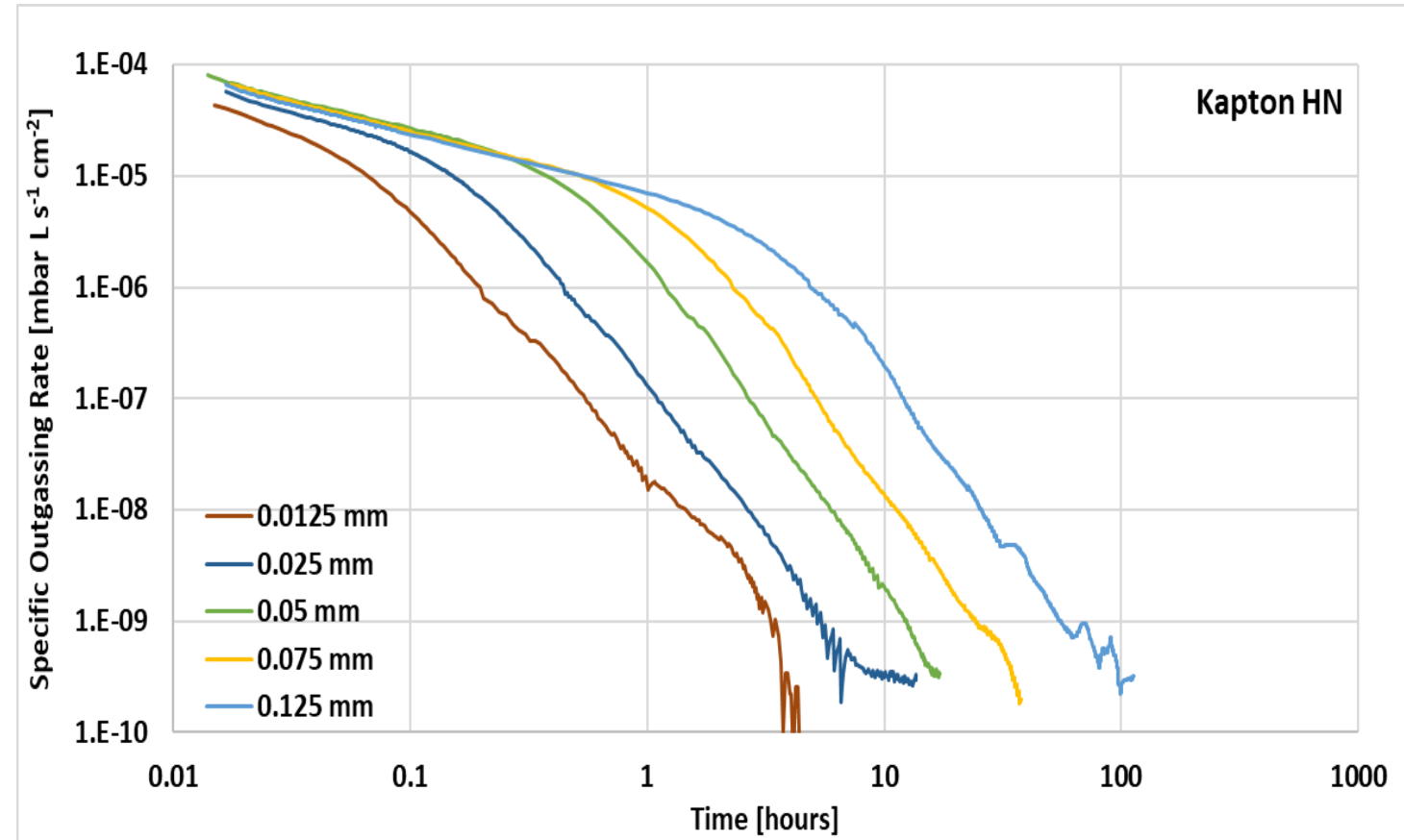
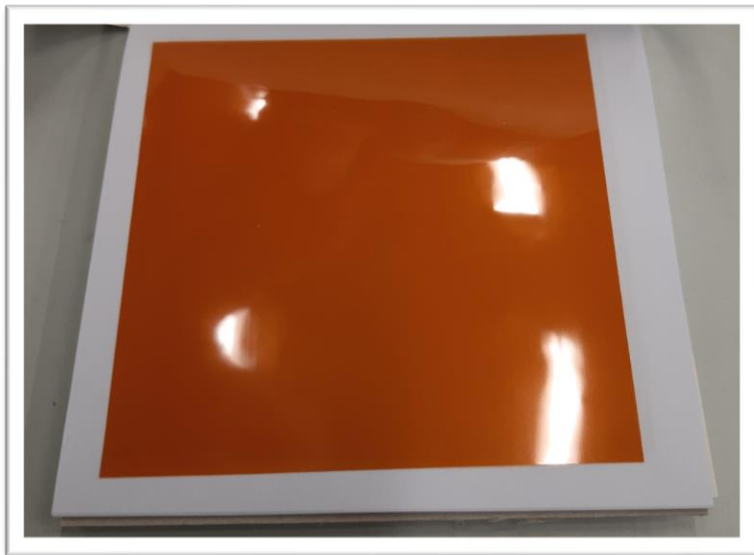


Outgassing measurement - throughput method

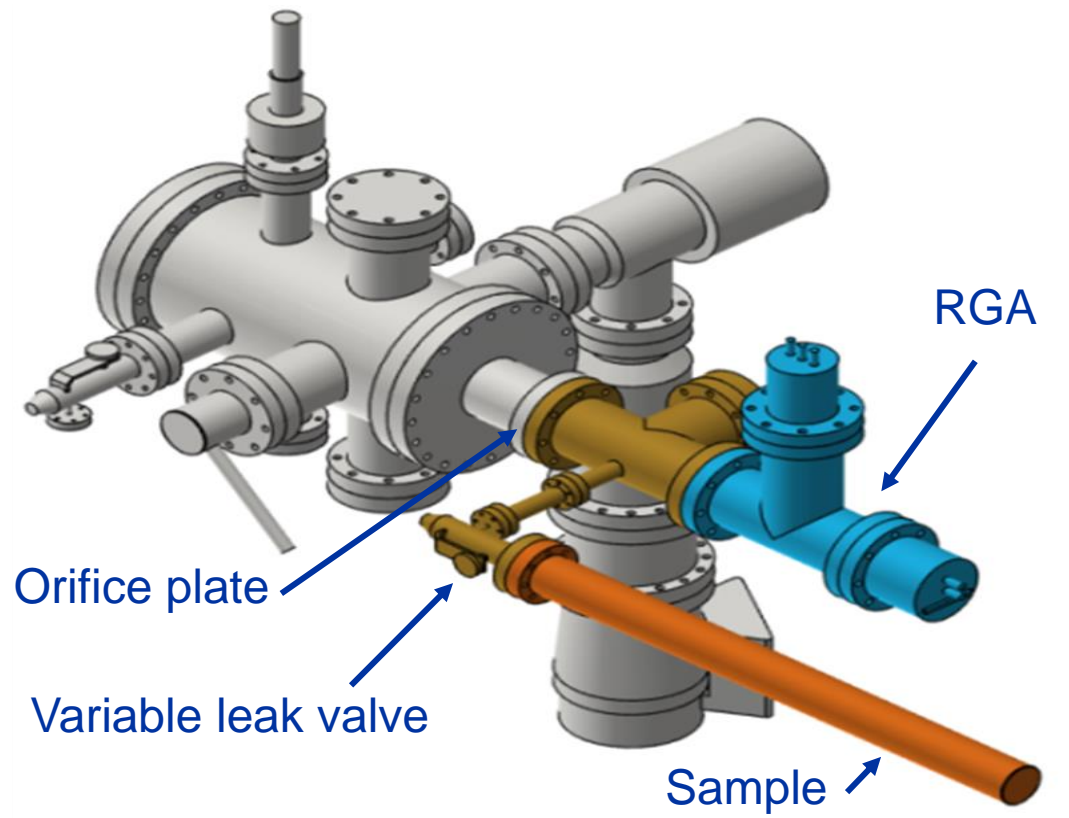
Kapton HN

Polyimide - Melting temperature: none
Expected water content: 0.96%

Samples: square sheets side 100 mm
Thicknesses: 0.0125-0.125 mm



Outgassing: Coupled-method measurements



All system components are baked to temperatures ranging from 200°C to 350°C

Sample and Variable leak valve are heated to requested temperature and duration

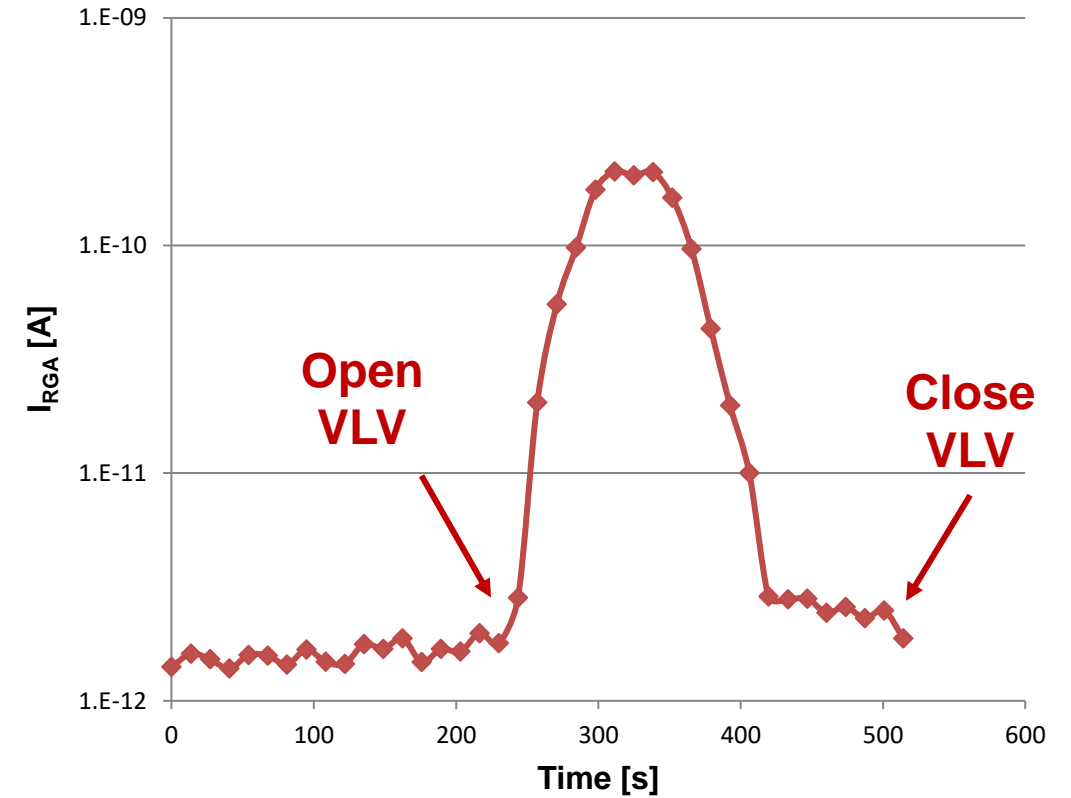
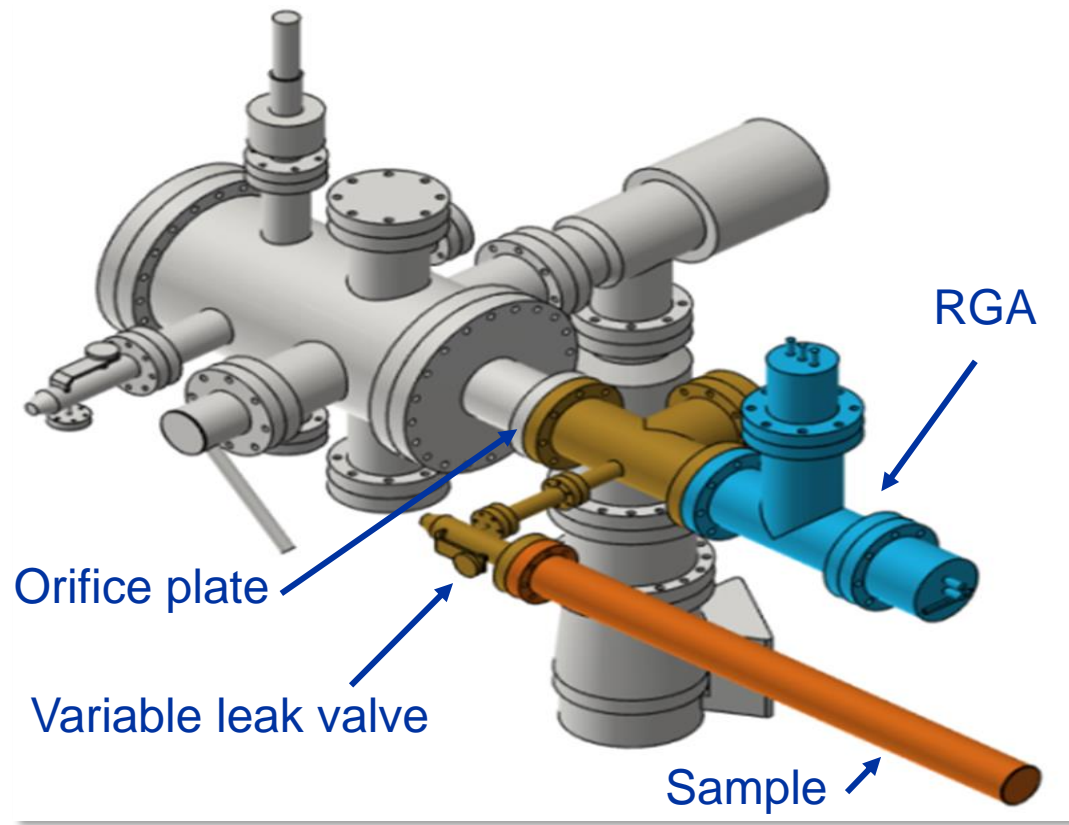
Prior to cooldown all instrumentation is degassed

Variable leak valve body has been vacuum fired to minimize contribution

24 hours at room temperature before the start of the first measurement

RGA has been calibrated to an in-house calibrated hot ionization gauge

Outgassing: Coupled-method measurements



Outgassing: Coupled-method measurements

$$Q \text{ [mbarl/s]} = \frac{S_c \times \int_{\Delta\tau} I_{RGA} \times \alpha_{RGA} d\tau}{\Delta t_a}$$

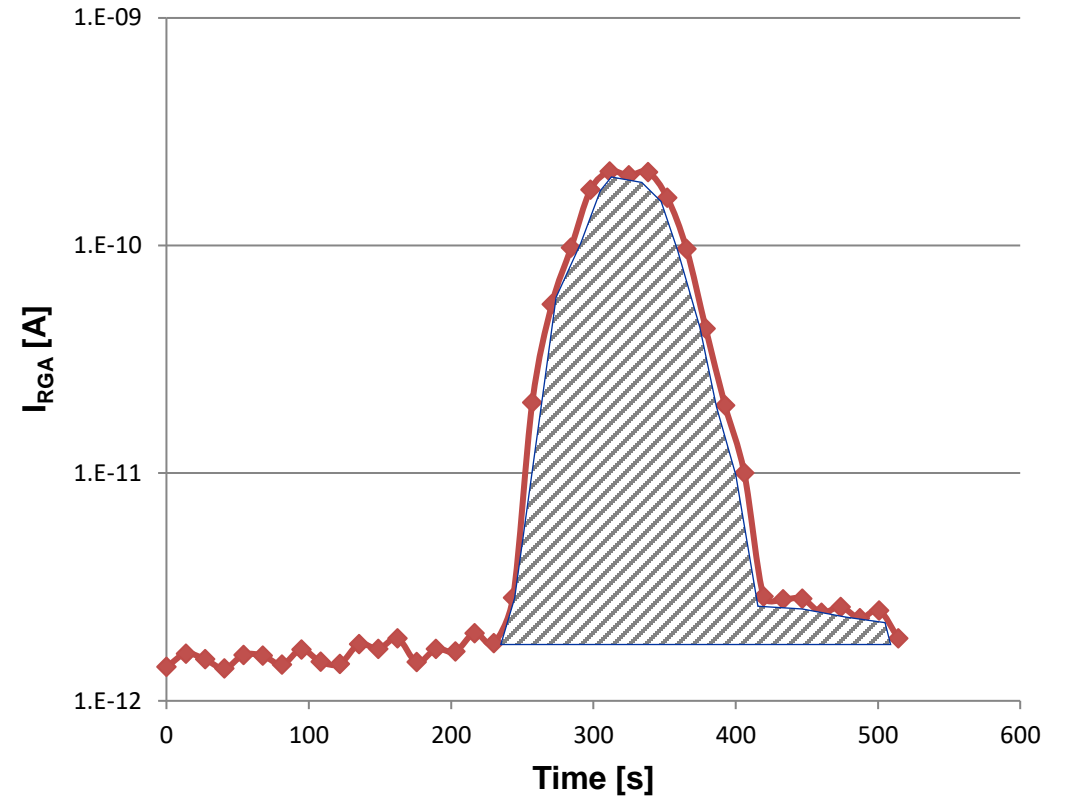
Δt_a : duration of the accumulation [s]

I_{RGA} : current recorded for the gas of interest [A]

S_c : conductance of the orifice for the gas of interest [l/s]

α_{RGA} : calibration factor for the gas of interest [A/mbar]

$\Delta\tau$: duration of the RGA recording [s]



Outgassing: Coupled-method measurements

$$Q \text{ [mbarl/s]} = \frac{S_c \times \int_{\Delta\tau} I_{RGA} \times \alpha_{RGA} d\tau}{\Delta t_a}$$

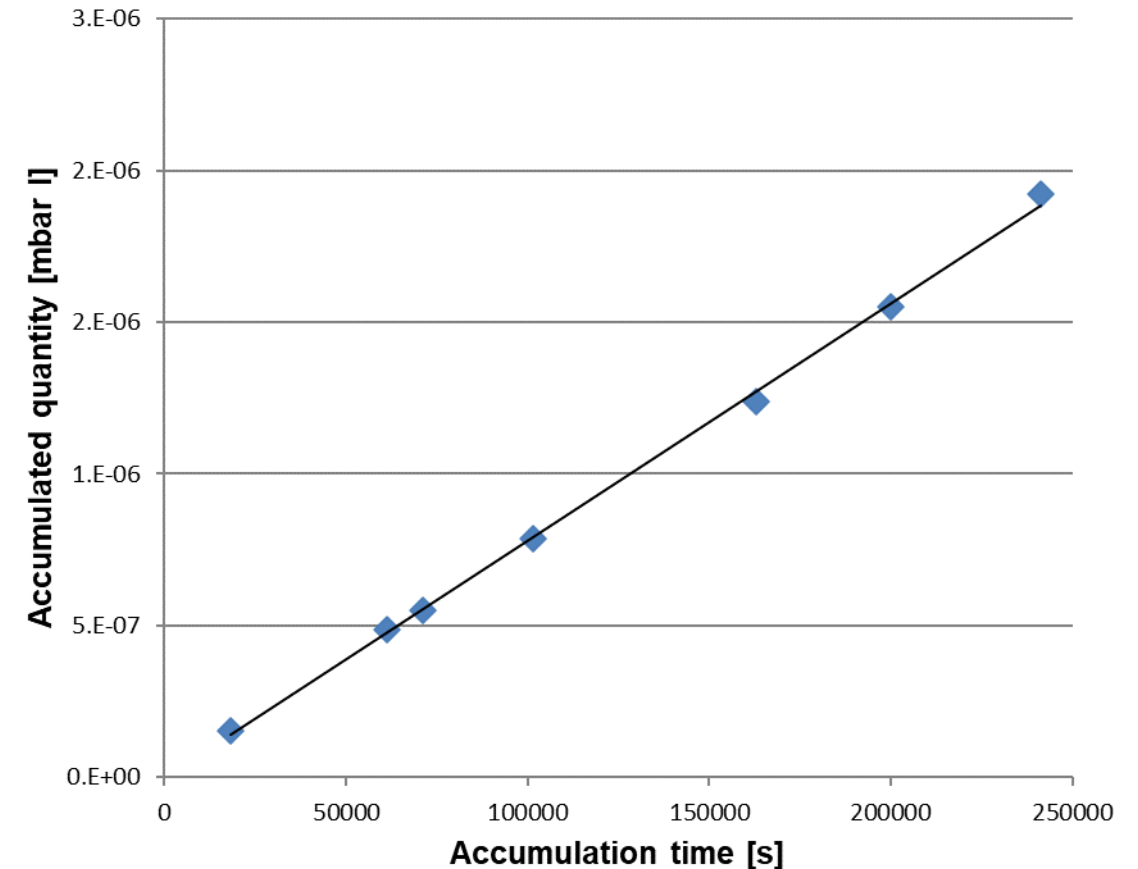
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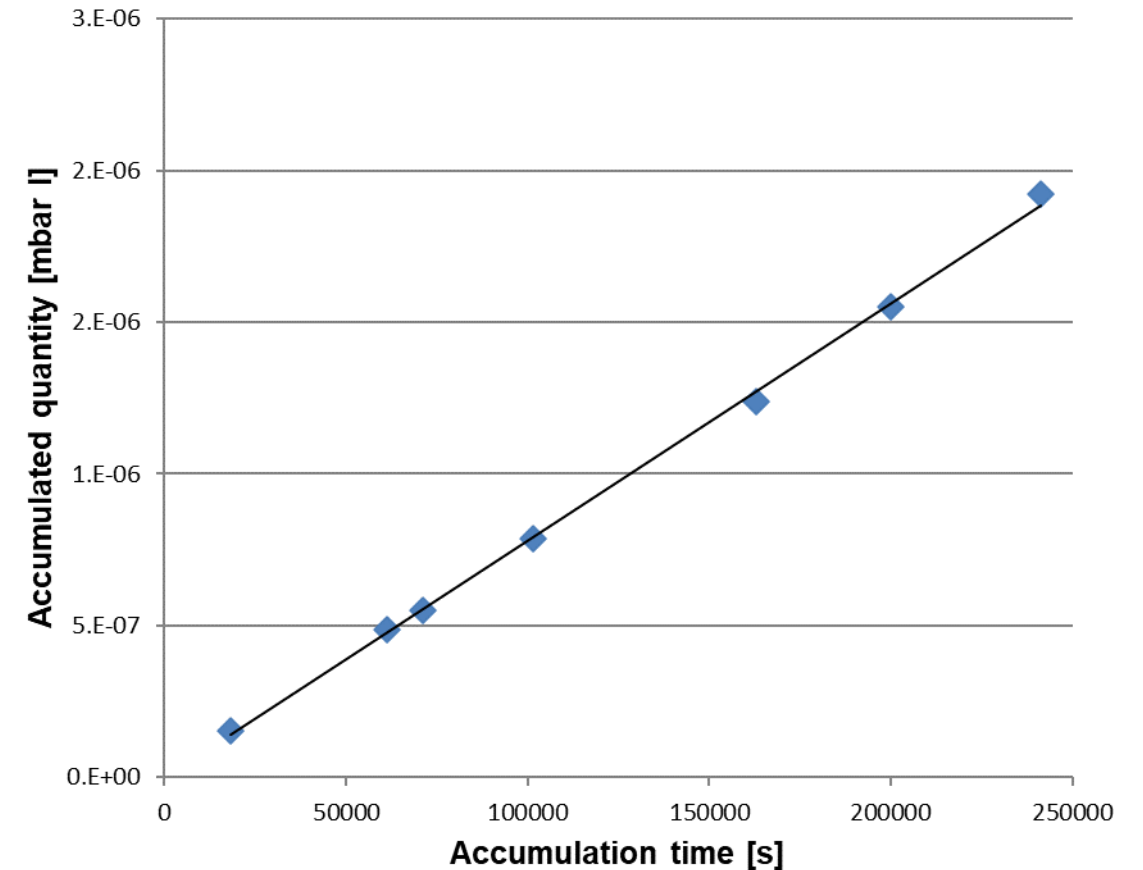
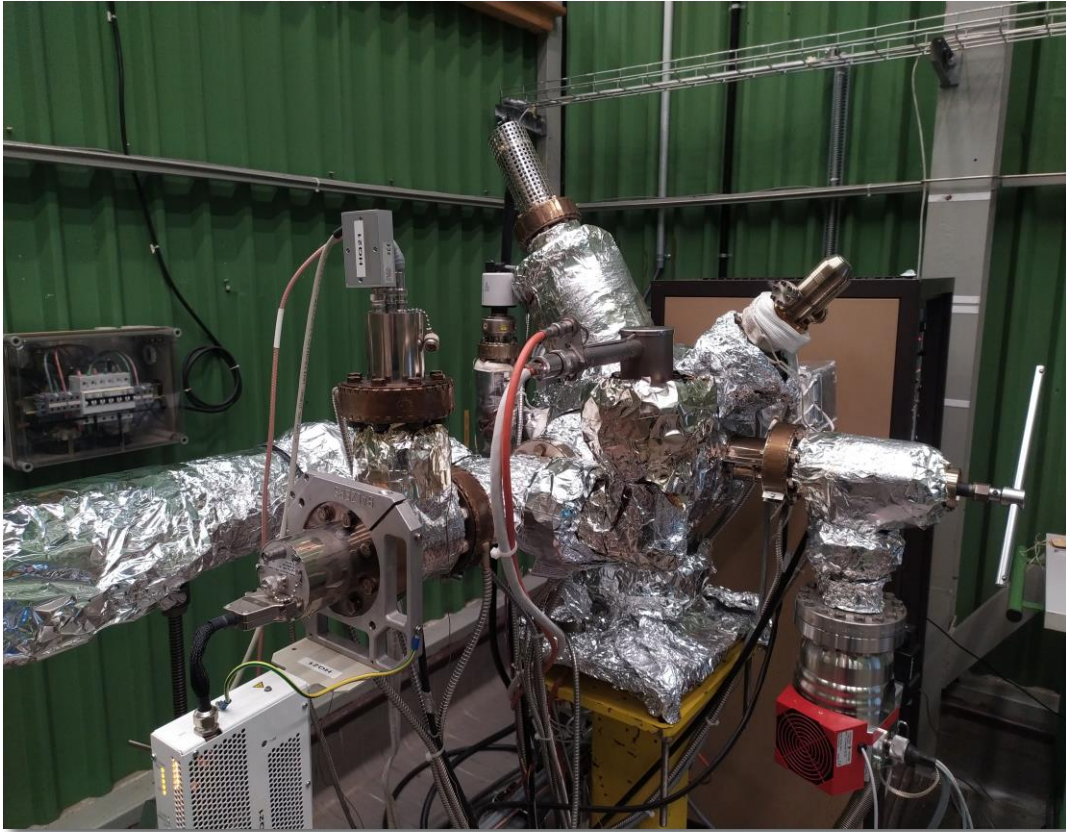
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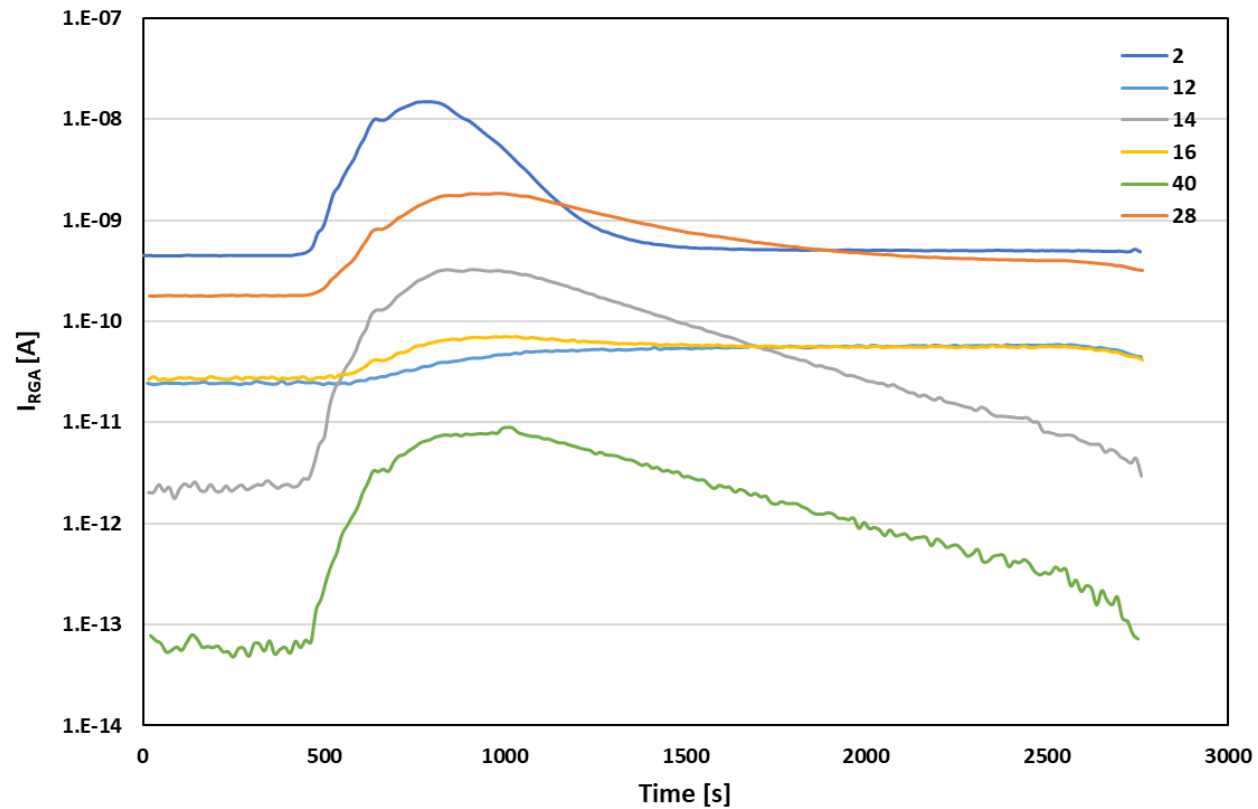


Outgassing: Coupled-method measurements

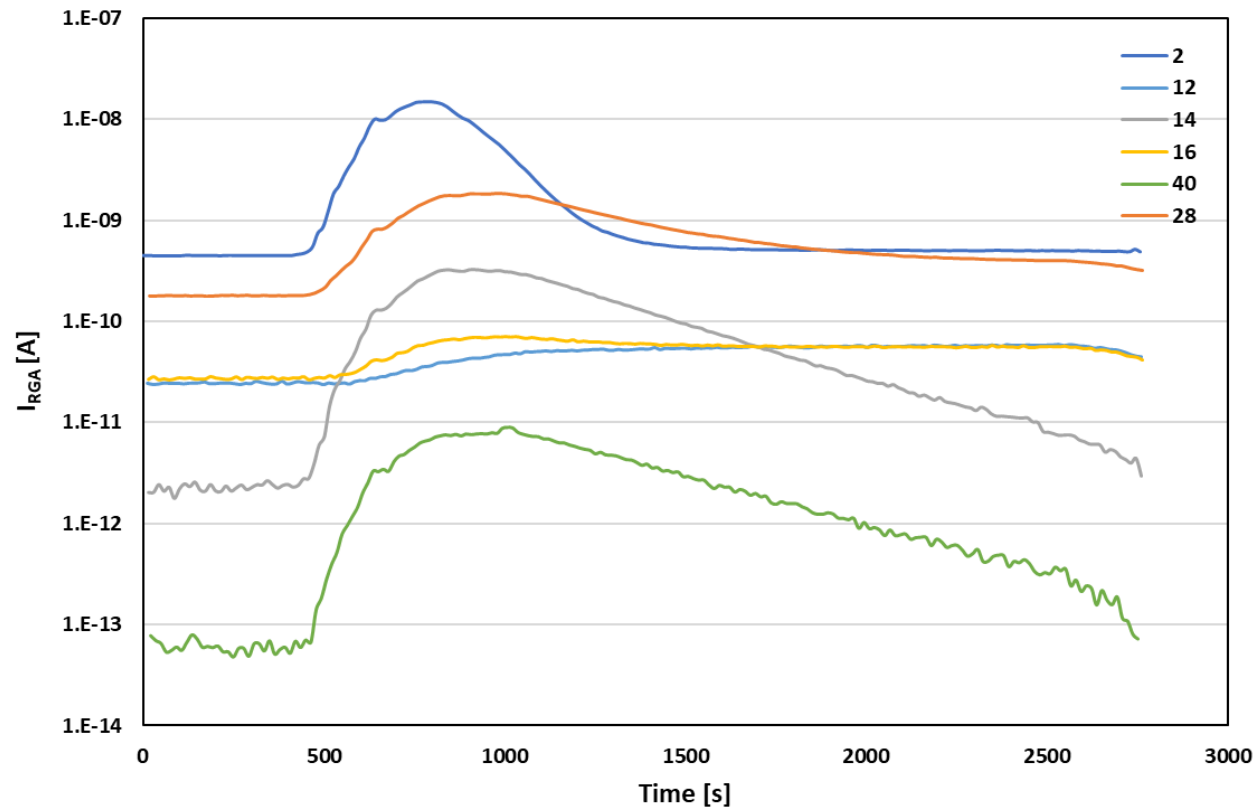
For a description of the method, see p. 11 of <https://arxiv.org/ftp/arxiv/papers/2006/2006.07124.pdf>



Outgassing: Coupled-method measurements

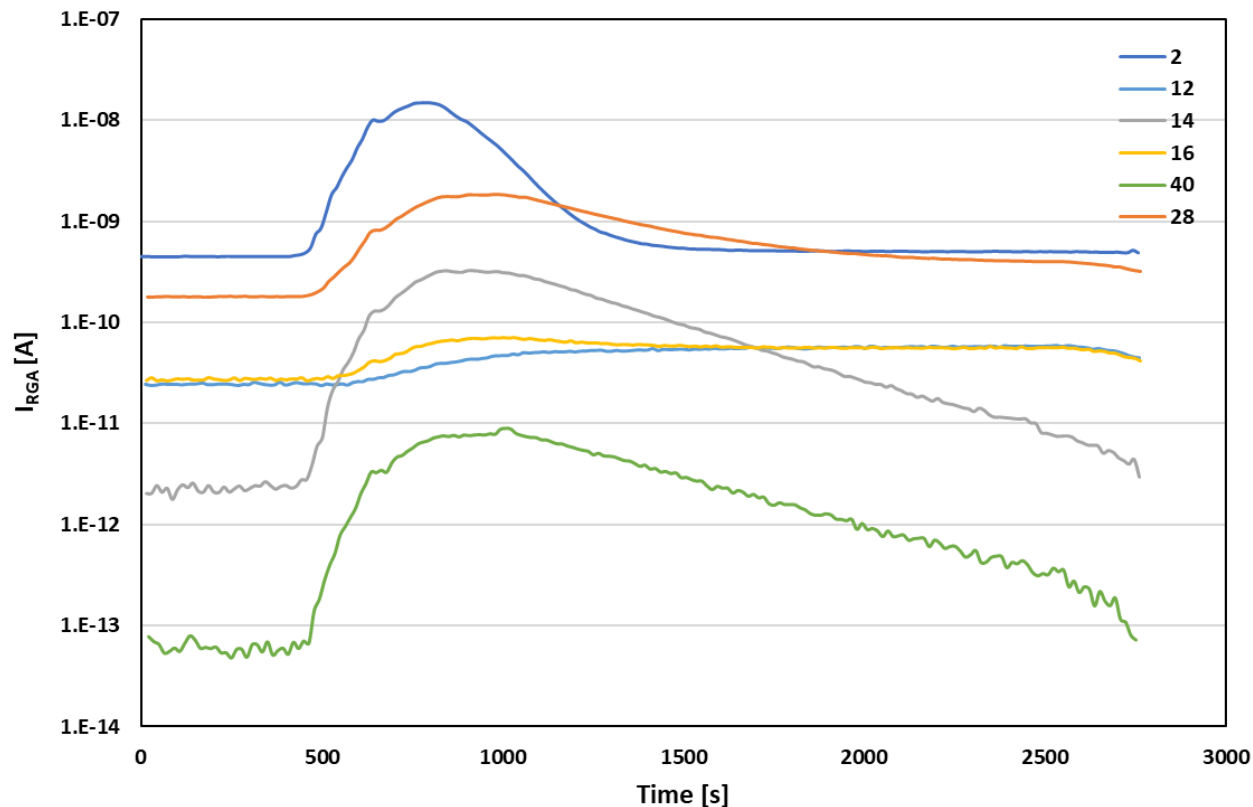


Outgassing: Coupled-method measurements



Mass 12, 16 & 28 do not show a similar behavior not CO

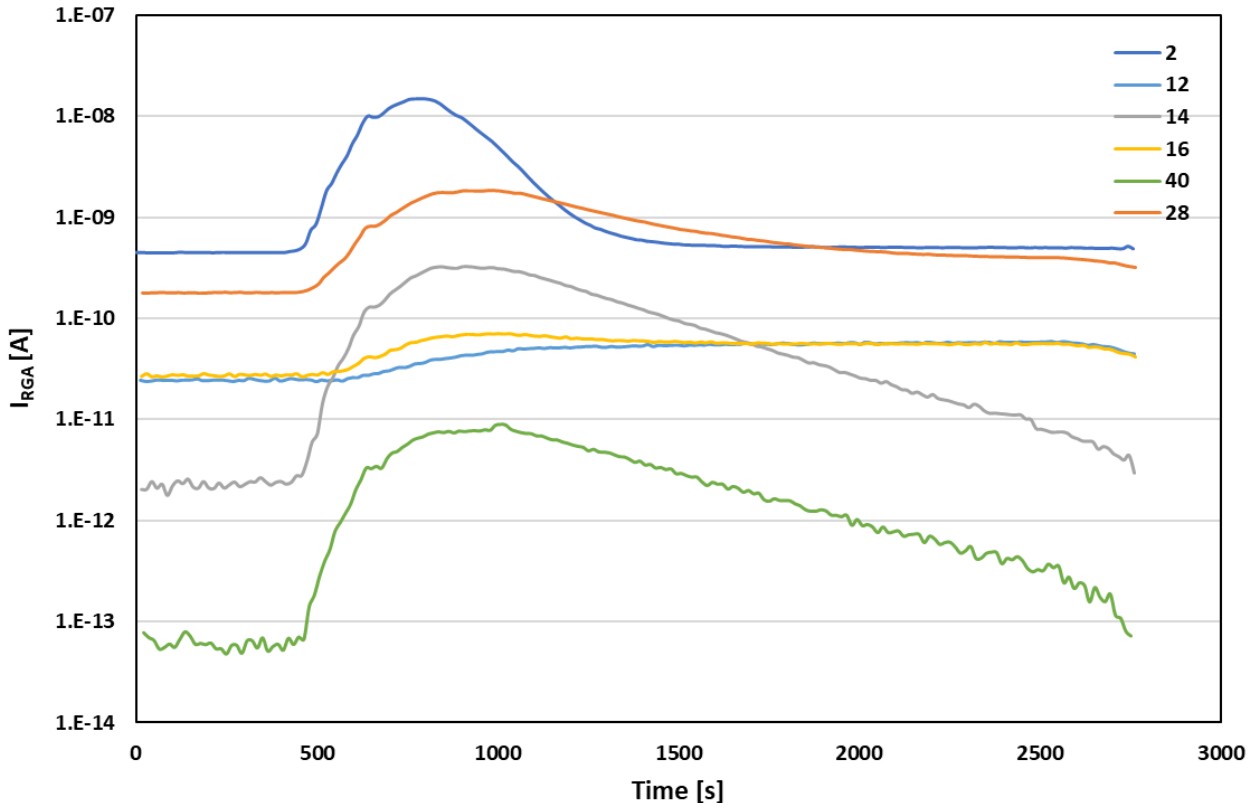
Outgassing: Coupled-method measurements



Mass 12, 16 & 28 do not show a similar behavior
Mass 14 & 28 show a similar behavior
Mass 40 (Argon) also in line with N₂

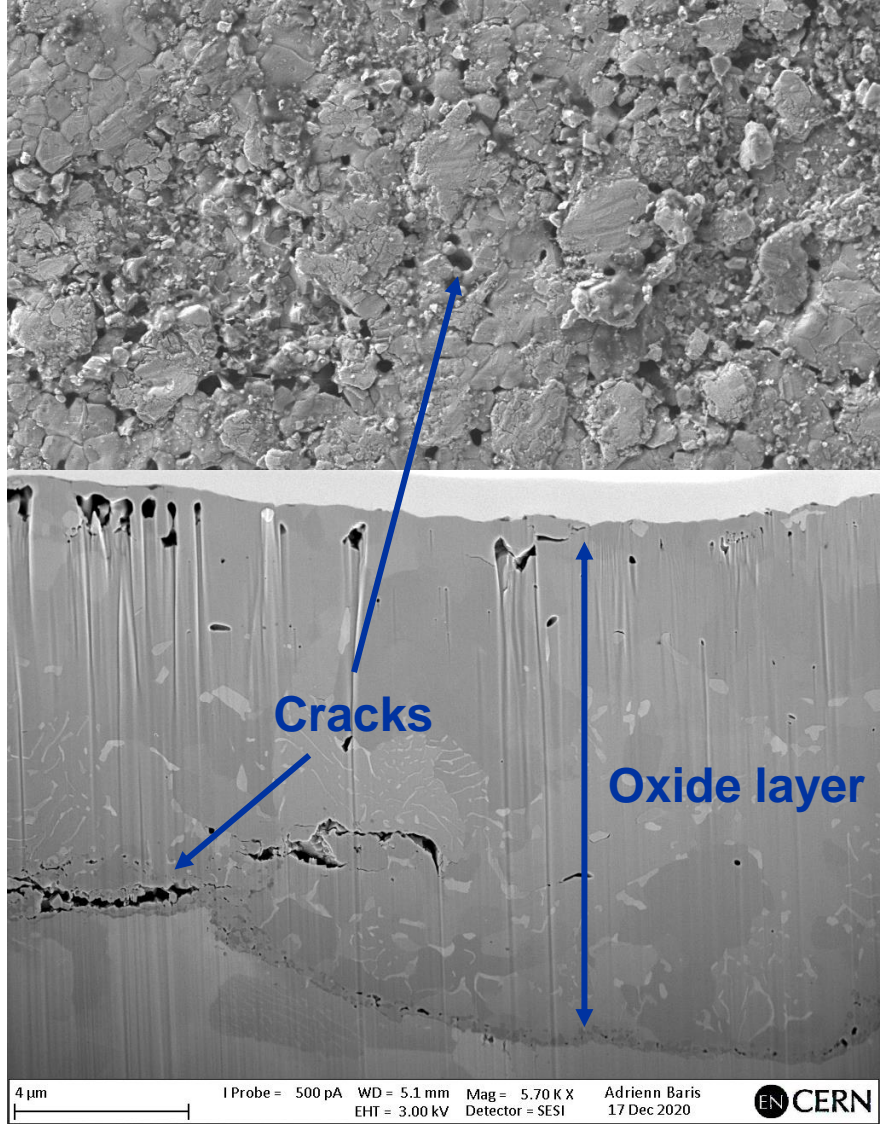
not CO
N₂ } AIR

Outgassing: Coupled-method measurements



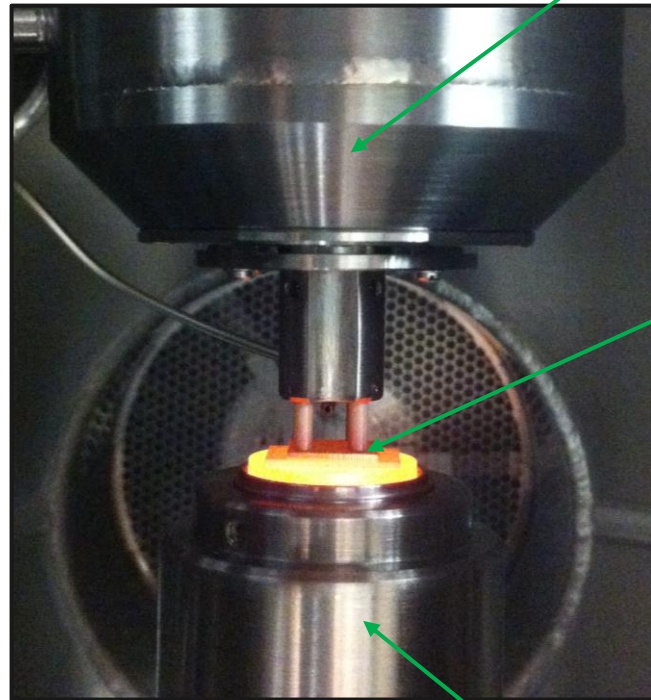
Mass 12, 16 & 28 do not show a similar behavior
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 Mass 40 (Argon) also in line with N₂

not CO
 N₂ } **AIR**



Thermal desorption spectroscopy

Residual Gas Analyser

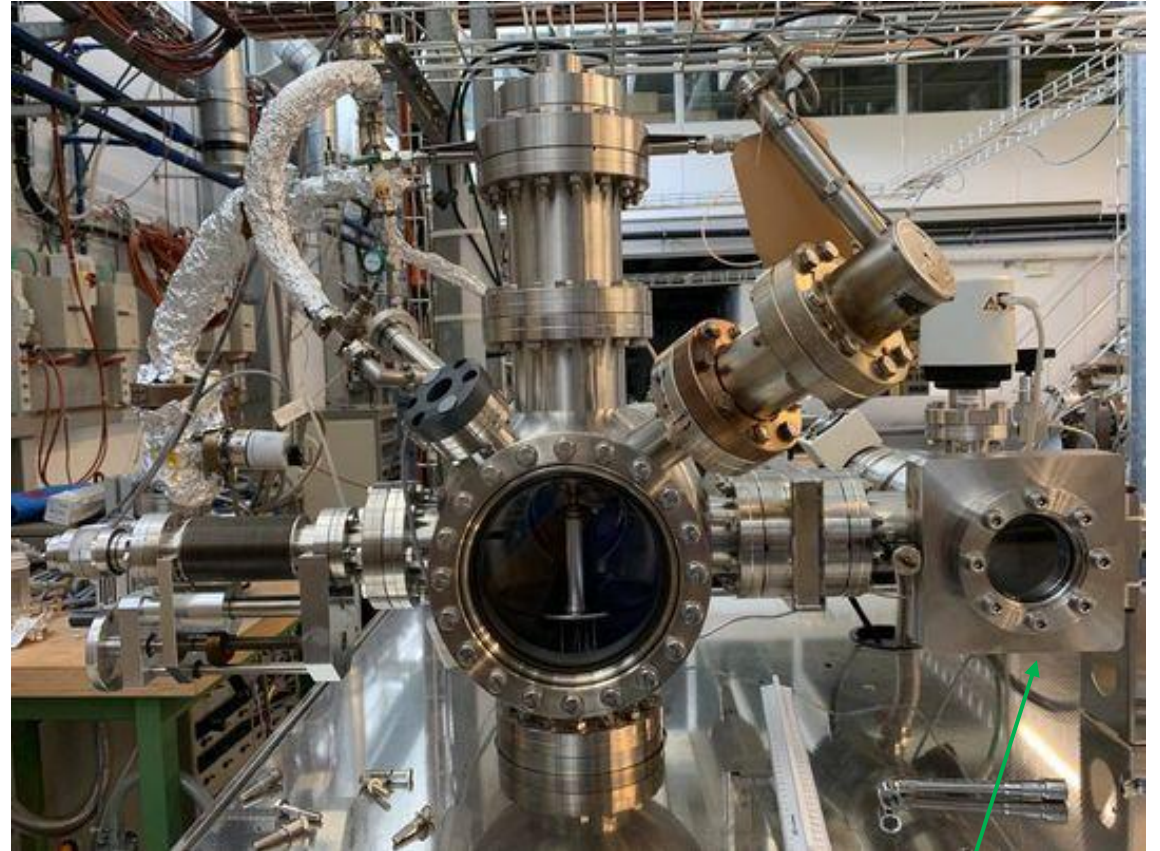


Sample

Heater

Heating rate: 5K/min

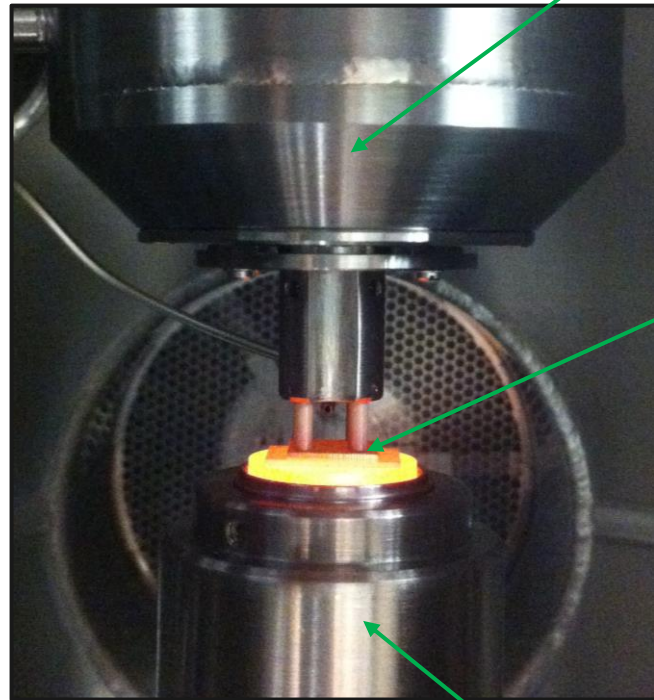
Dimensions of sample: 10mm x 10mm x 1mm



Loadlock

Thermal desorption spectroscopy

Residual Gas Analyser



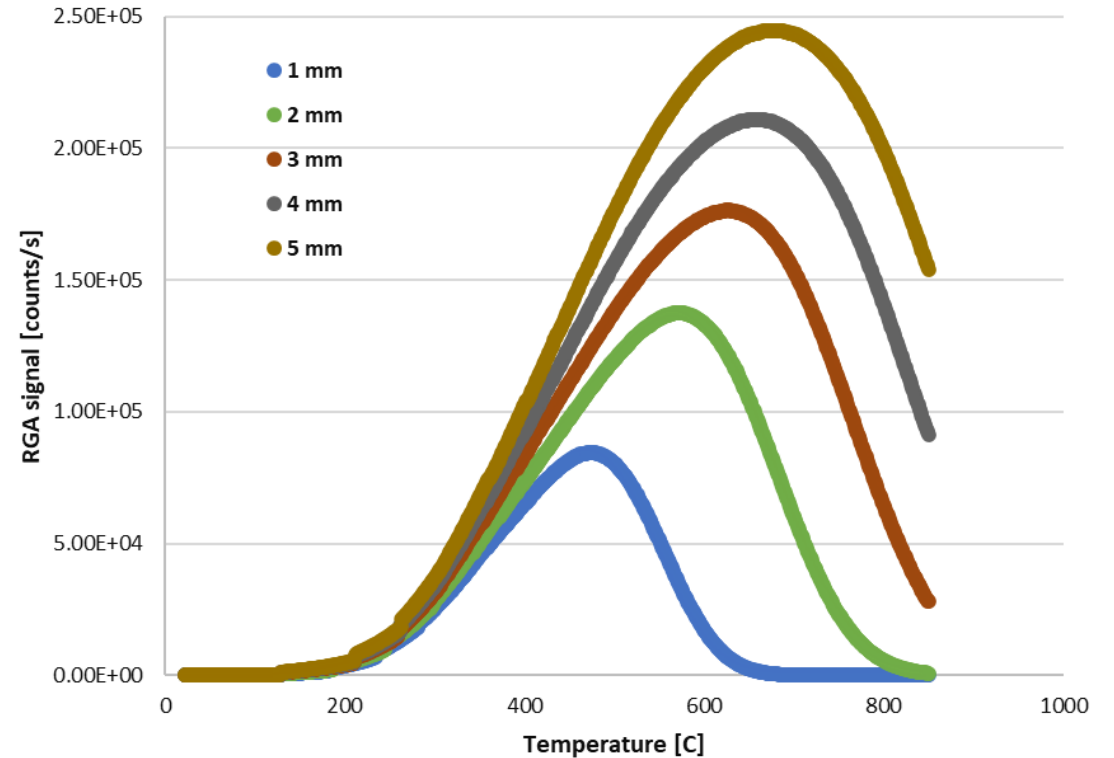
Sample

Heater

Heating rate: 5K/min

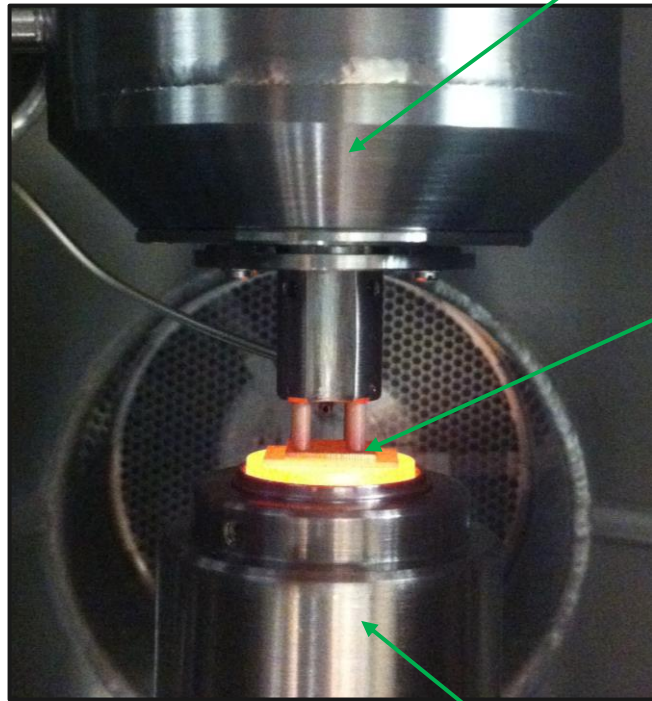
Dimensions of sample: 10mm x 10mm x 1mm

Mass 2



Thermal desorption spectroscopy

Residual Gas Analyser

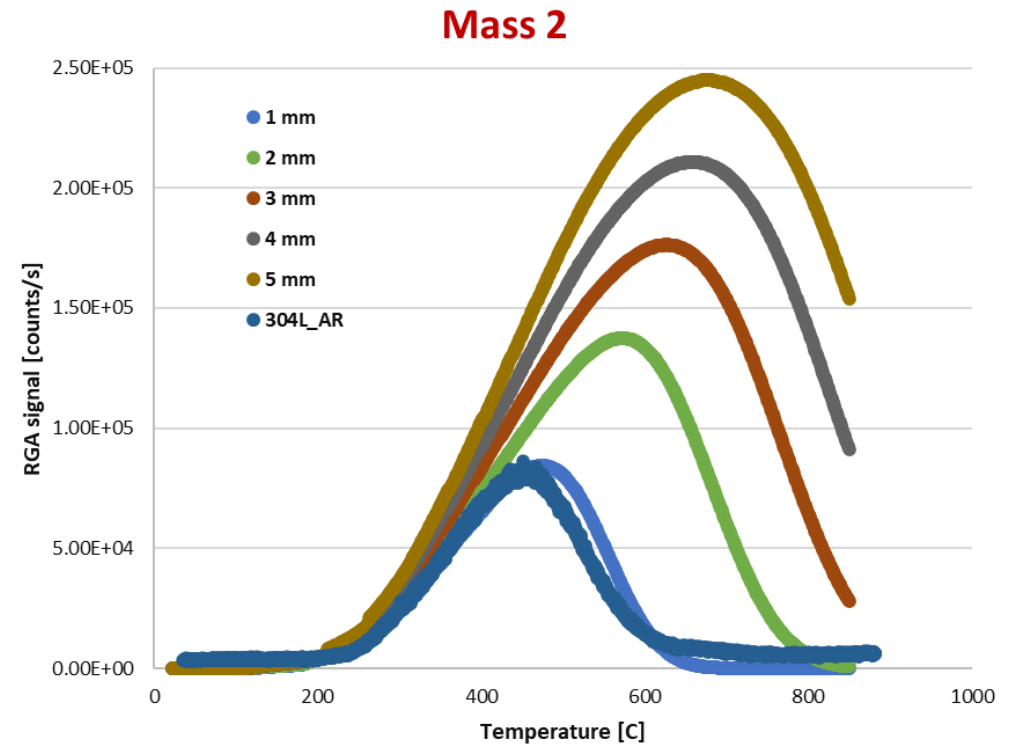


Sample

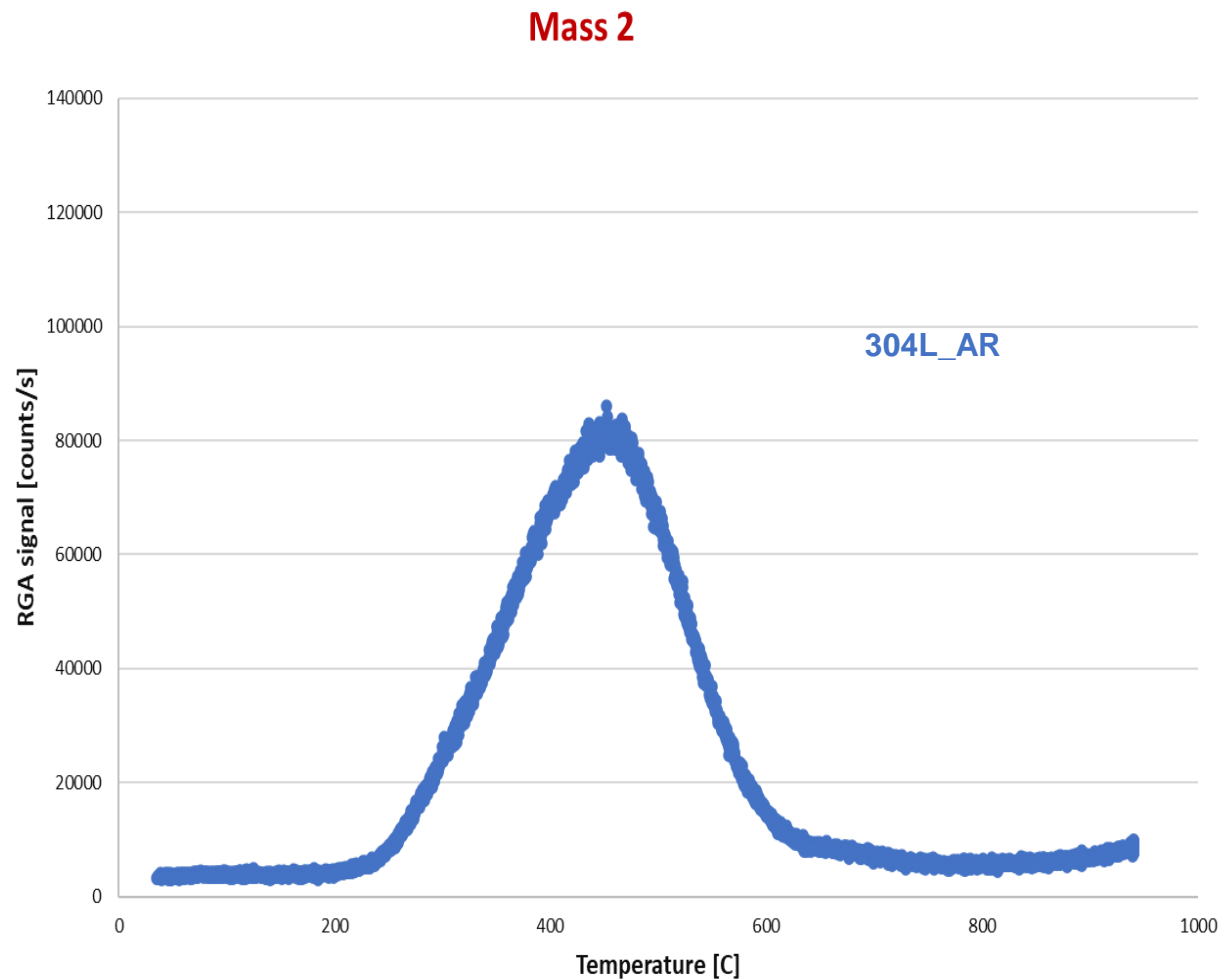
Heater

Heating rate: 5K/min

Dimensions of sample: 10mm x 10mm x 1mm



TDS measurement

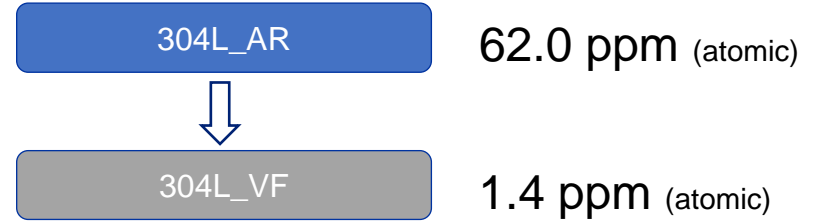
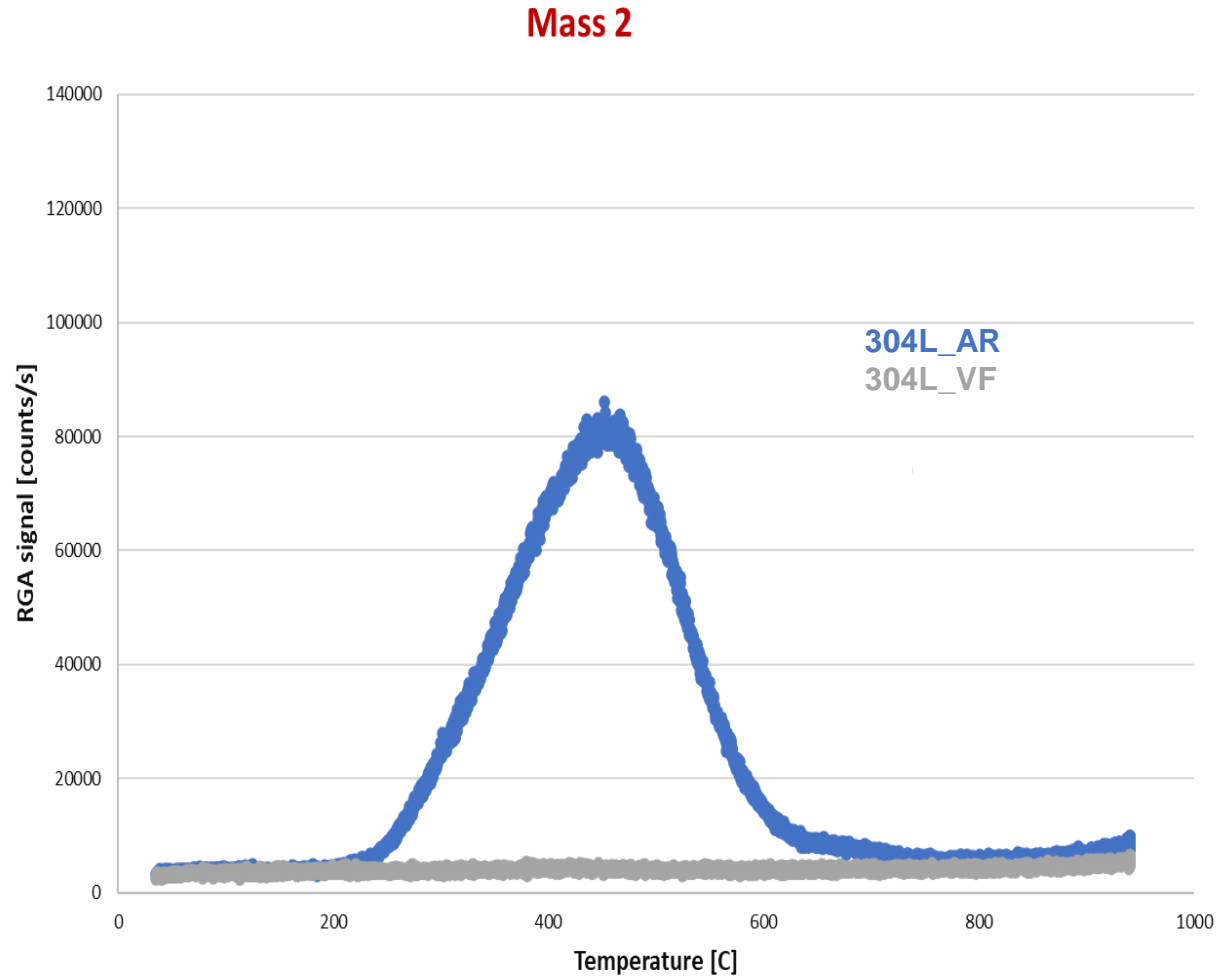


304L_AR

62.0 ppm (atomic)

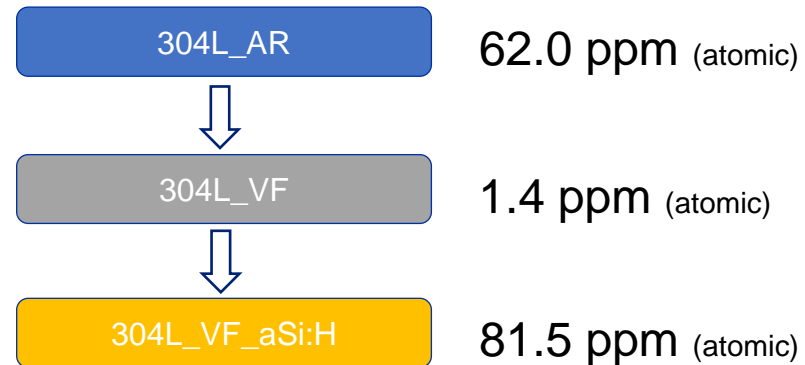
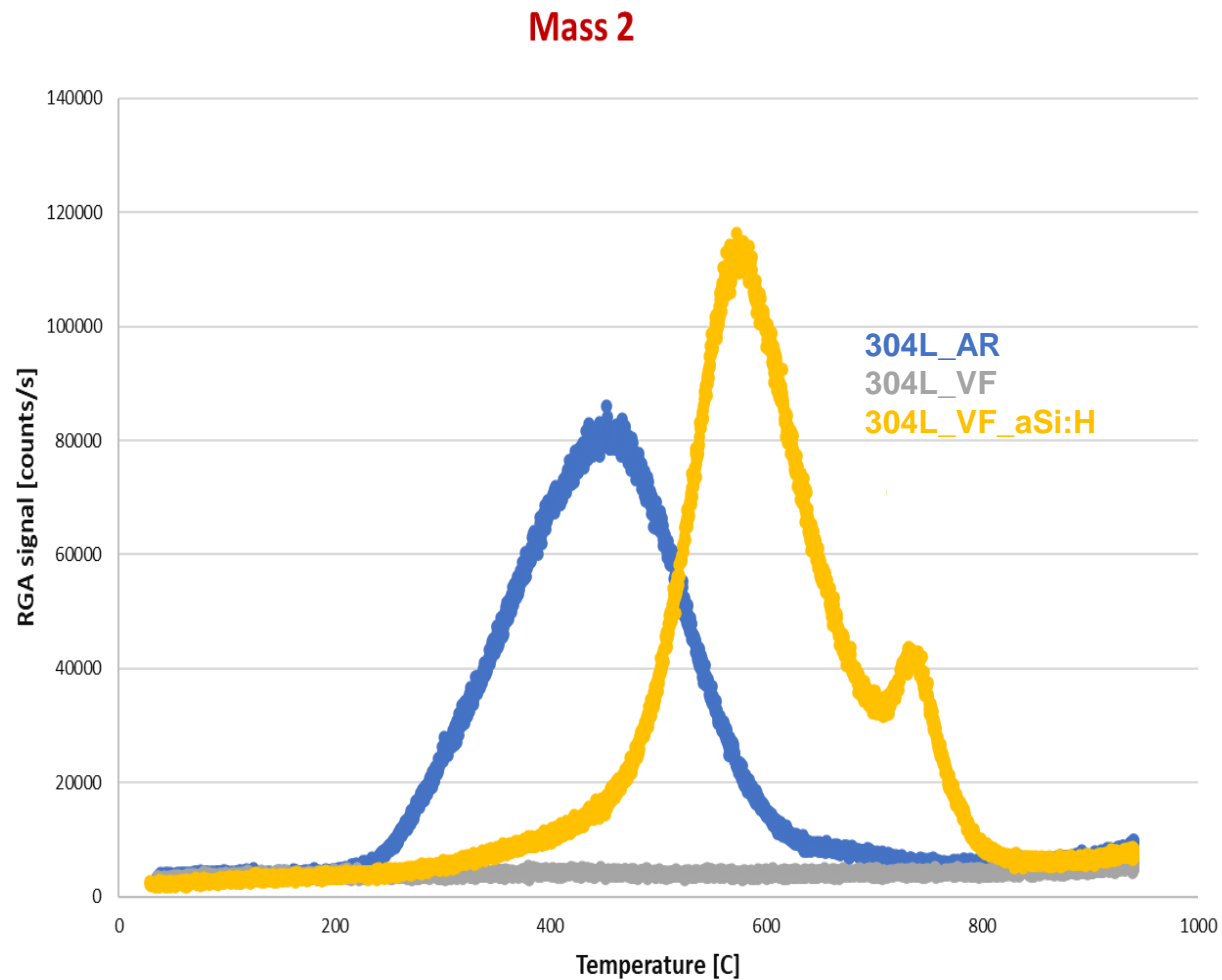
AR : As Received
VF : Vacuum Fired (950C/2h)
aSi:H CVD coating

TDS measurement



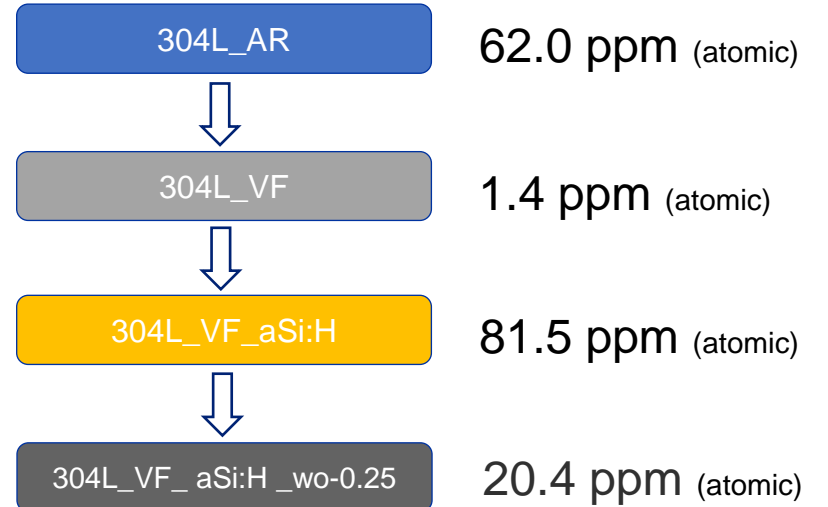
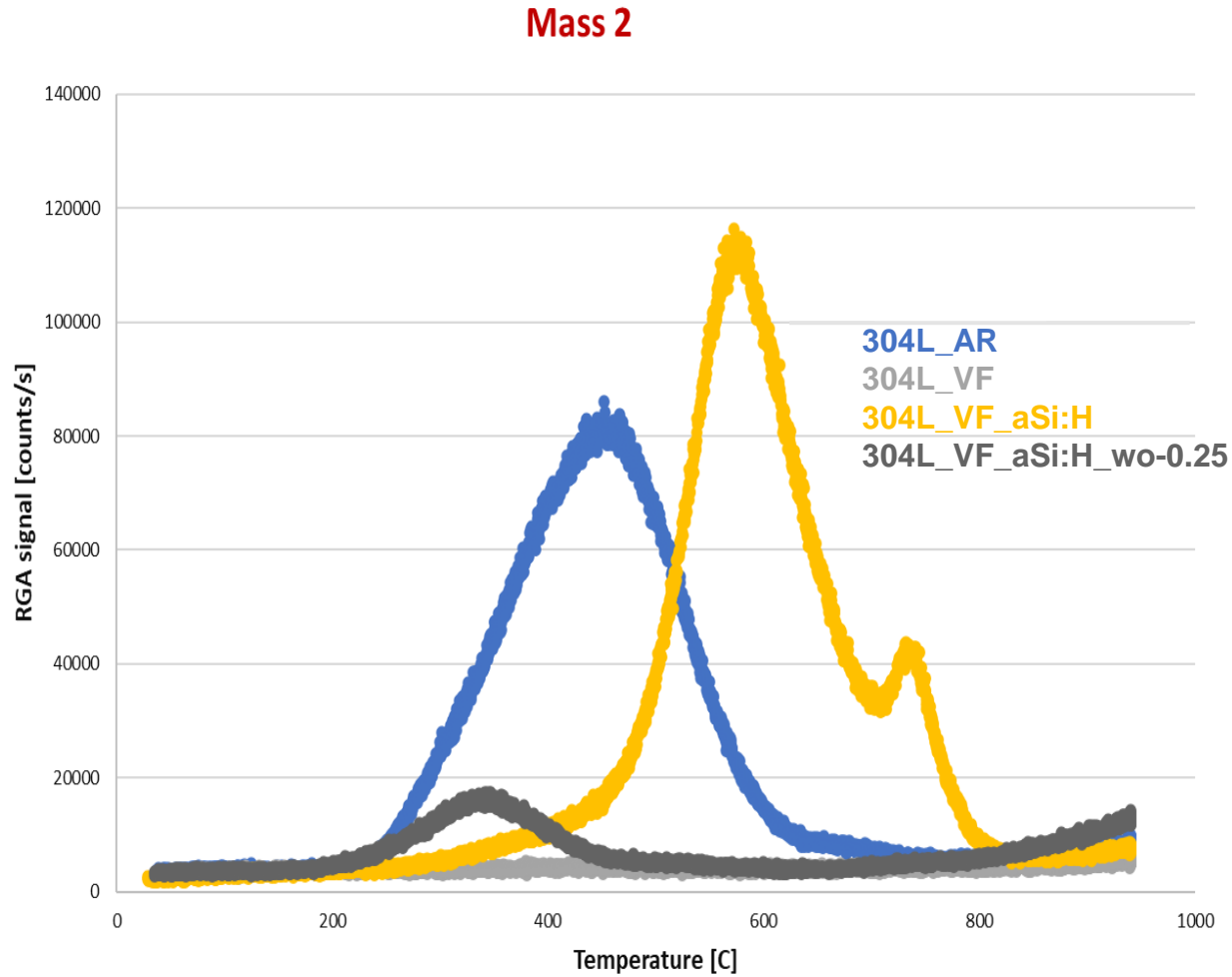
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TDS measurement



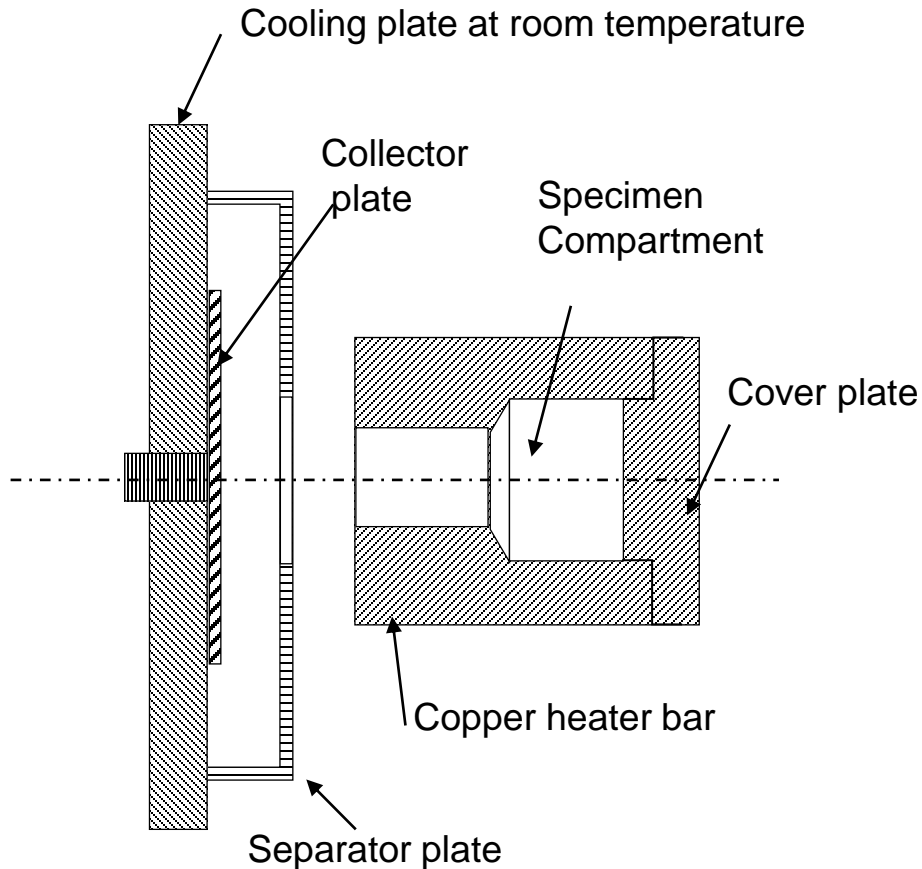
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TDS measurement



AR : As Received
VF : Vacuum Fired (950C/2h)
aSi:H CVD coating

Total Mass Loss



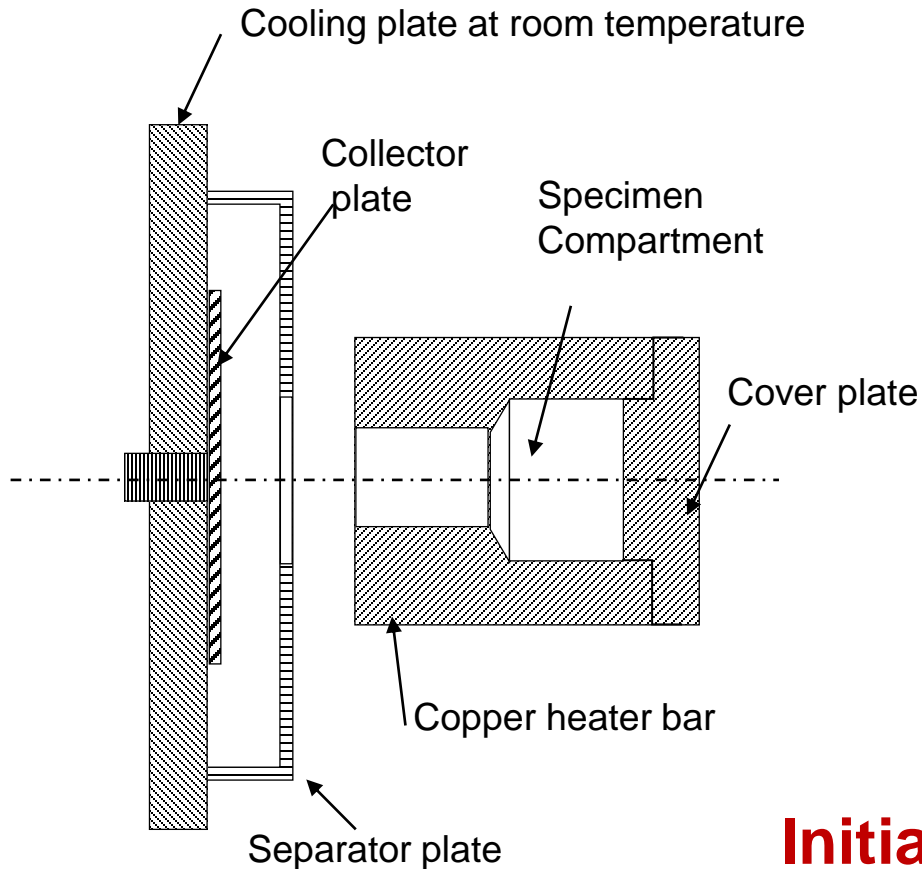
- Polymer specimen is conditioned in a 50% RH environment for 24 hours
- Weight of specimen and collector plate is measured
- Under vacuum the specimen is heated up to 125°C for 24 hours
- Weight of specimen and collector plate is measured

$$TML[\%] = \frac{S_{init} - S_{final}}{S_{init}} \times 100$$

$$CVCM[\%] = \frac{C_{init} - C_{final}}{S_{init}} \times 100$$

ASTM E595-93 : Standard Test Method for Total Mass Loss and Collected Volatile Condensable Materials from Outgassing in a Vacuum Environment

Total Mass Loss



- Polymer specimen is conditioned in a 50% RH environment for 24 hours
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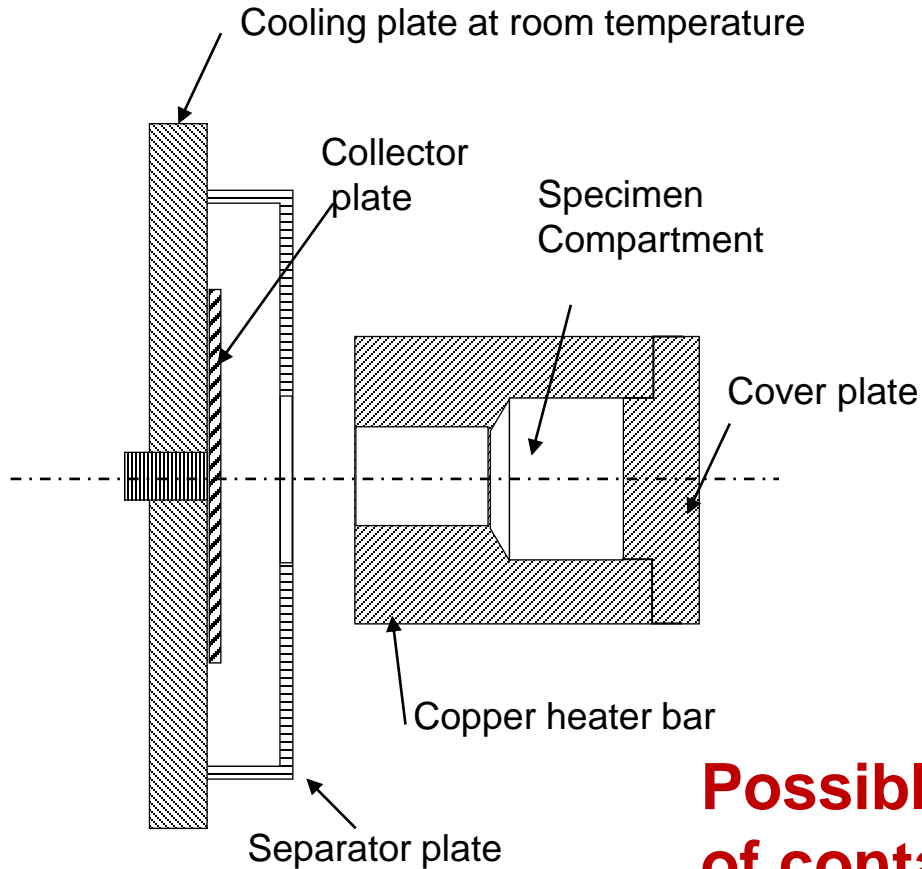
$$TML[\%] = \frac{S_{init} - S_{final}}{S_{init}} \times 100$$

$$CVCM[\%] = \frac{C_{init} - C_{final}}{S_{init}} \times 100$$

Initial concentration

ASTM E595-93 : Standard Test Method for Total Mass Loss and Collected Volatile Condensable Materials from Outgassing in a Vacuum Environment

Total Mass Loss



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- Weight of specimen and collector plate is measured
- Under vacuum the specimen is heated up to 125°C for 24 hours
- Weight of specimen and collector plate is measured

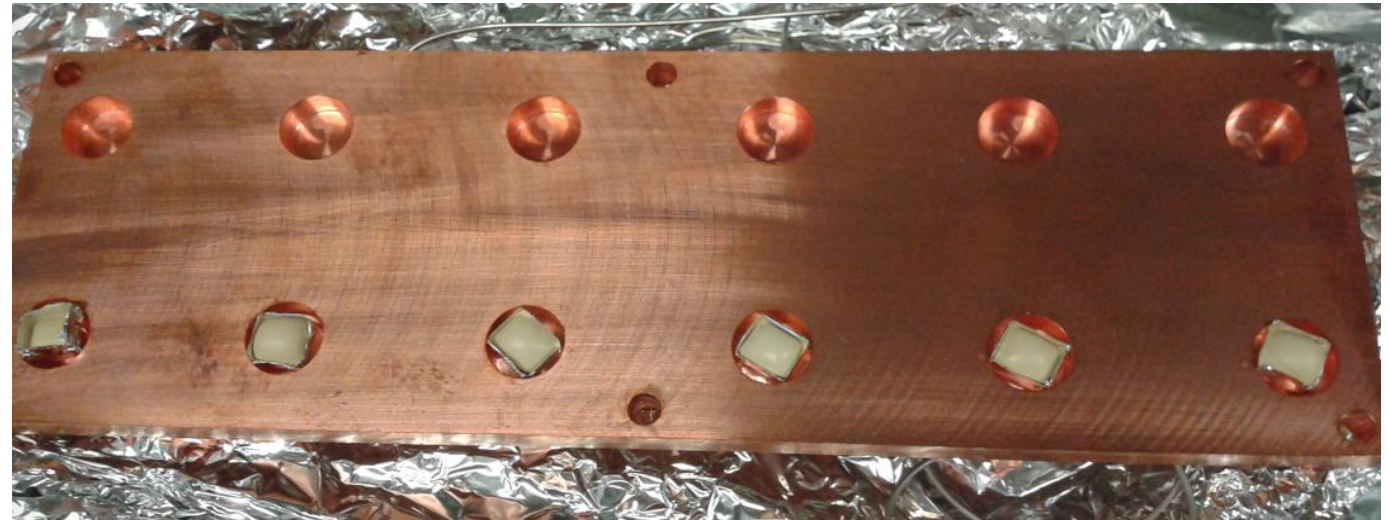
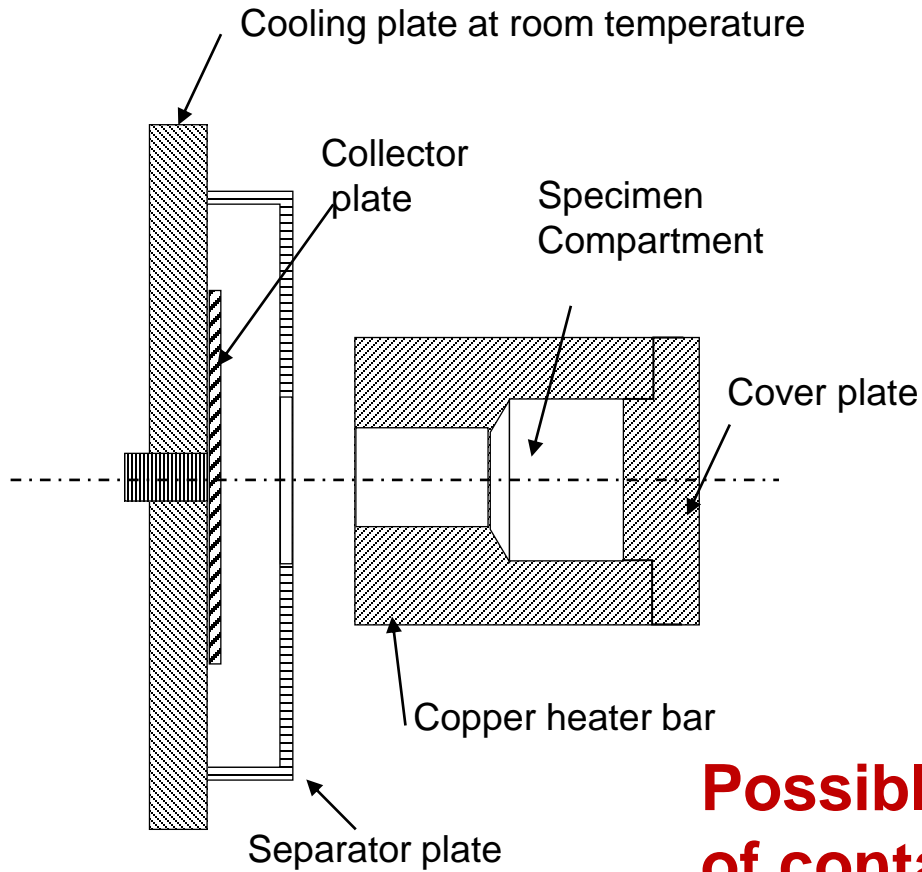
$$TML[\%] = \frac{S_{init} - S_{final}}{S_{init}} \times 100$$

**Possible source
of contamination**

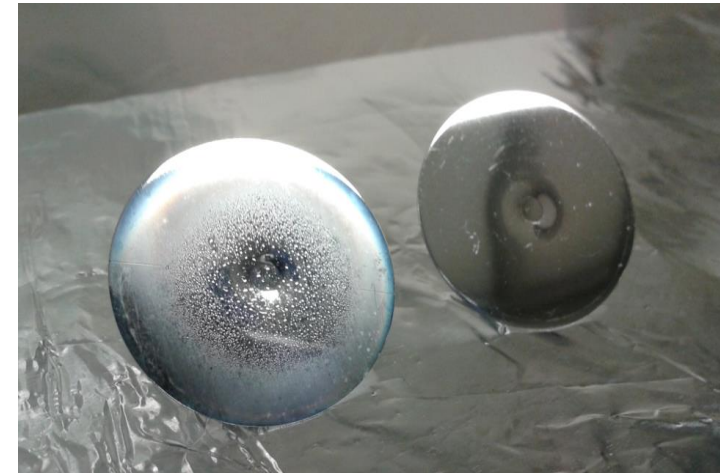
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ASTM E595-93 : Standard Test Method for Total Mass Loss and Collected Volatile Condensable Materials from Outgassing in a Vacuum Environment

Total Mass Loss

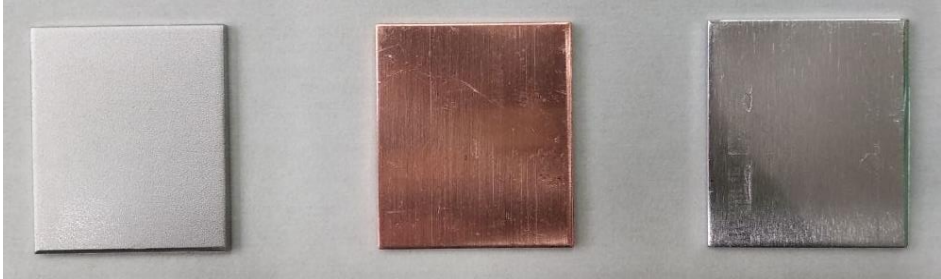
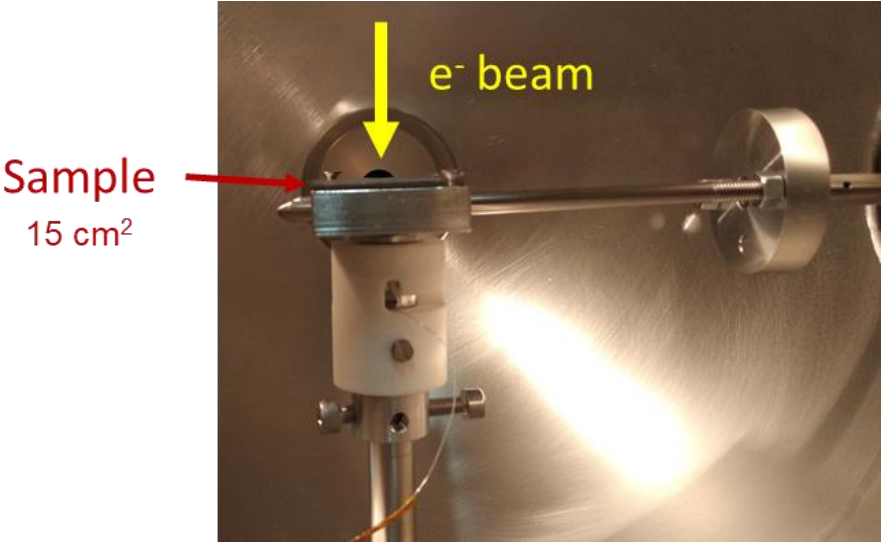
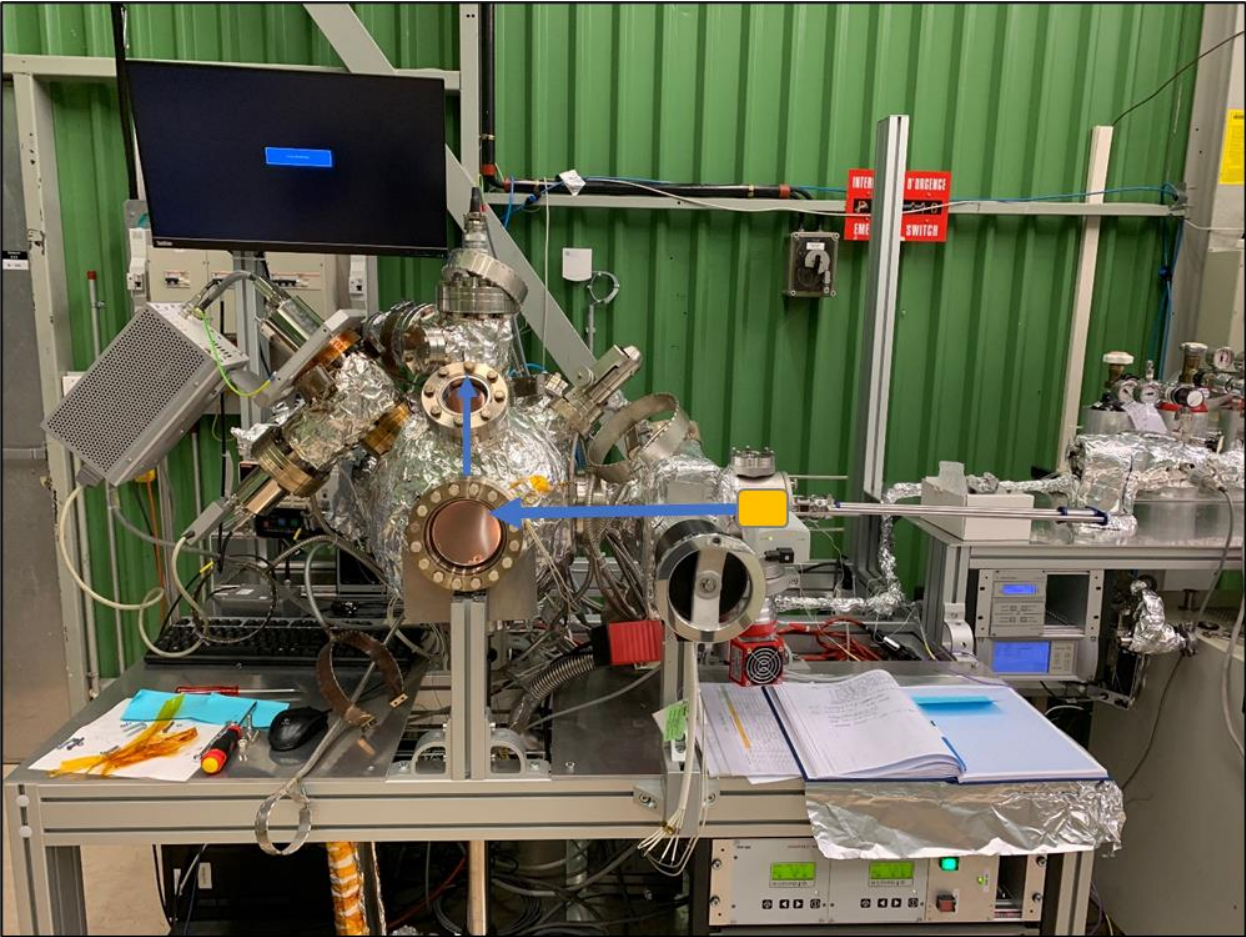


Possible source of contamination →



ASTM E595-93 : Standard Test Method for Total Mass Loss and Collected Volatile Condensable Materials from Outgassing in a Vacuum Environment

Electron Stimulated Desorption

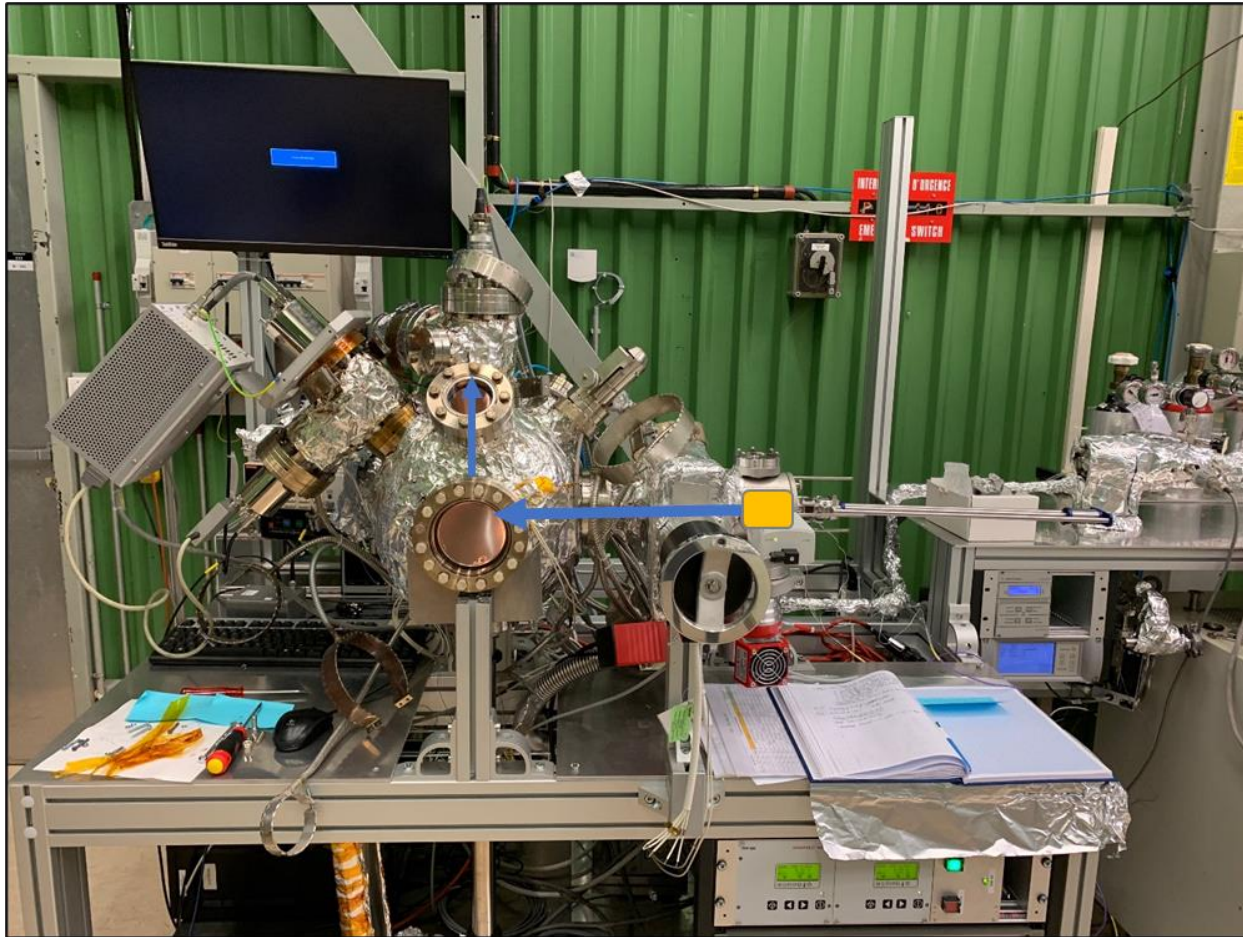


316 LN

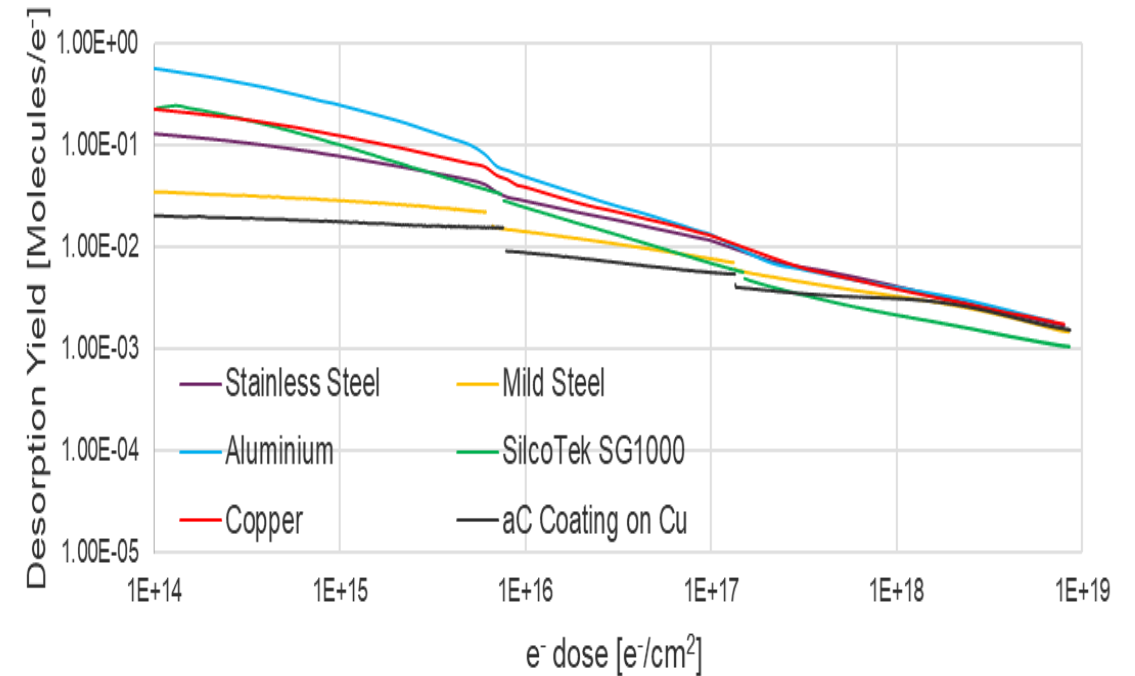
Cu-OFE

Al

Electron Stimulated Desorption



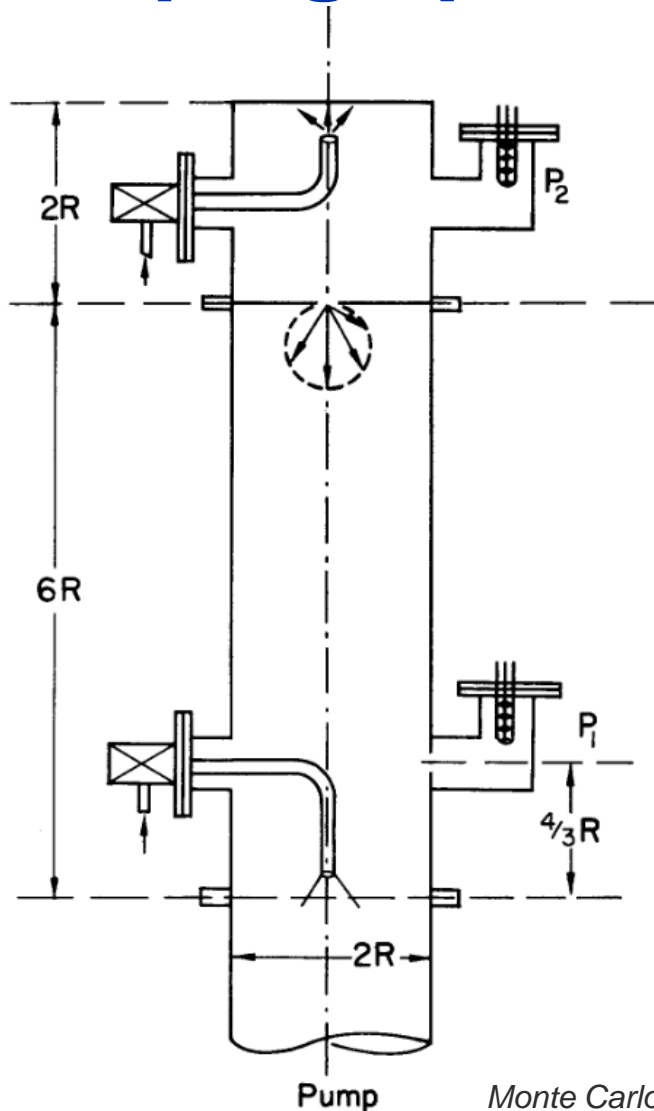
ESD of H₂ for different materials at 295 K



$$\text{Desorption yield} = \frac{\text{molecules of desorbed gas}}{\text{incident electrons}}$$

Dependent on: Energy – material – gas - dose

Pumping speed: Fischer-Mommsen dome

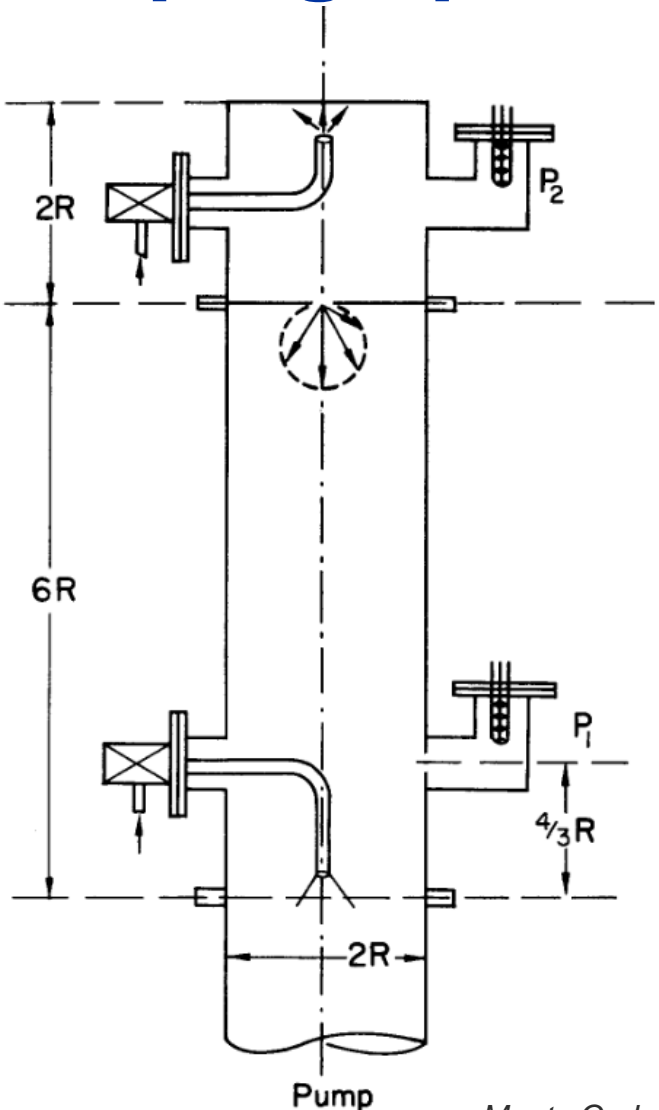


- Specific dimensions and gauge locations must be respected
- Orifice with conductance C separating 2 chambers

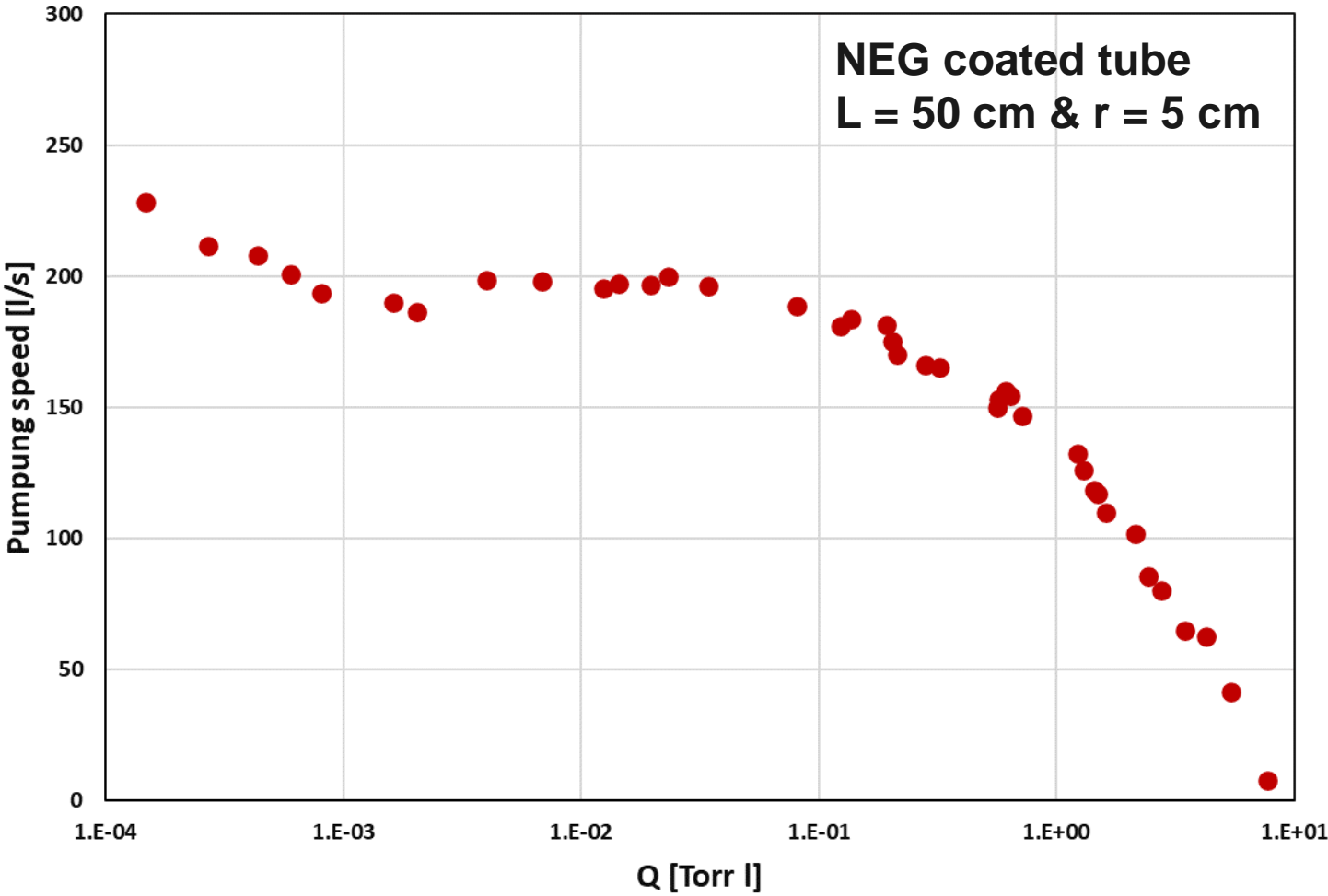
$$\hookrightarrow S = C \left(\frac{P_2}{P_1} - 1 \right)$$

Monte Carlo computations on molecular flow in pumping speed test domes E. Fischer, H Mommsen - Vacuum June 1967 p 309-315

Pumping speed: Fischer-Mommsen dome

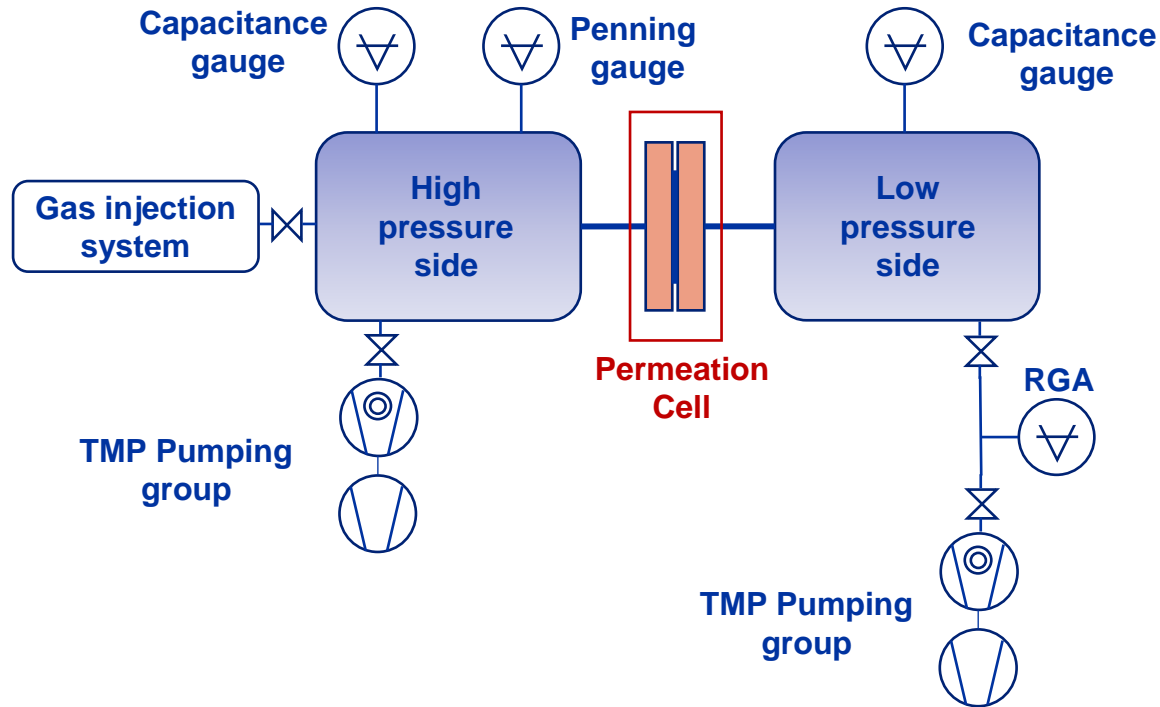


Hydrogen saturation for PAX experiment



Monte Carlo computations on molecular flow in pumping speed test domes E. Fischer, H Mommsen - Vacuum June 1967 p 309-315

Permeation



- Permeation is a combination of dissolution and diffusion

- ❖ Dissolution – Henry's Law

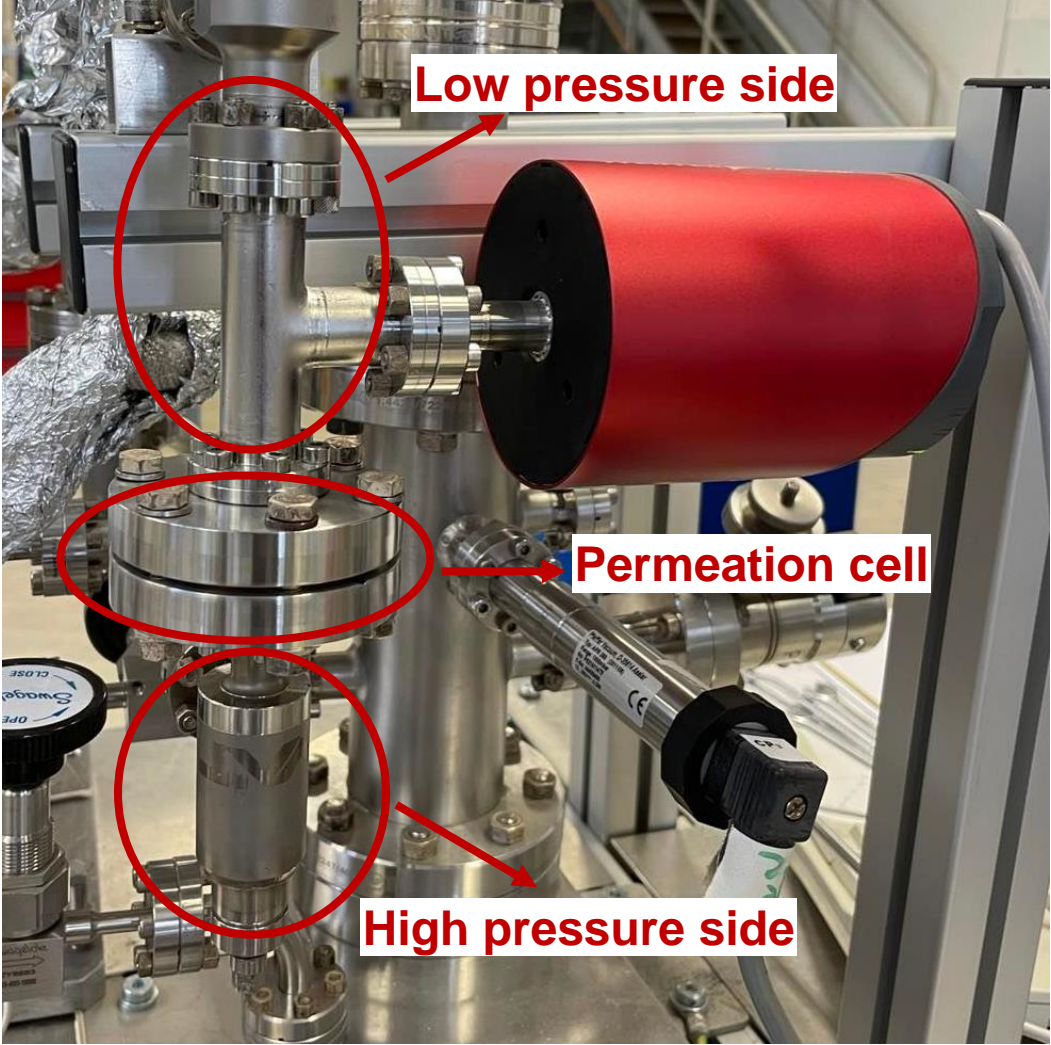
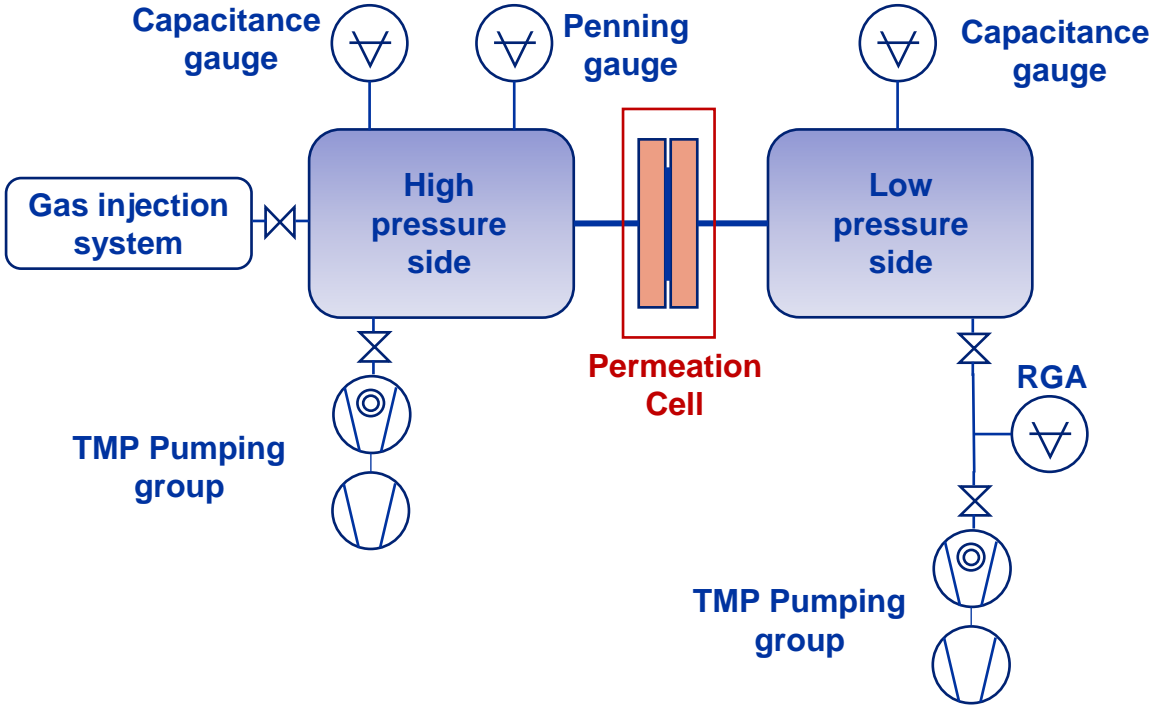
$$C = sP^n$$

- ❖ Diffusion – Fick's first Law

$$Q = -D \frac{dc}{dx}$$

- Dynamic mode
- Static mode

Permeation

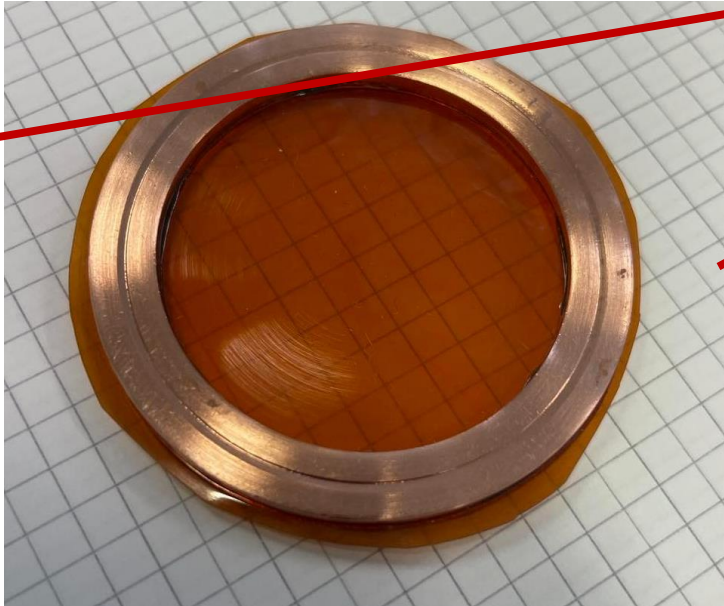
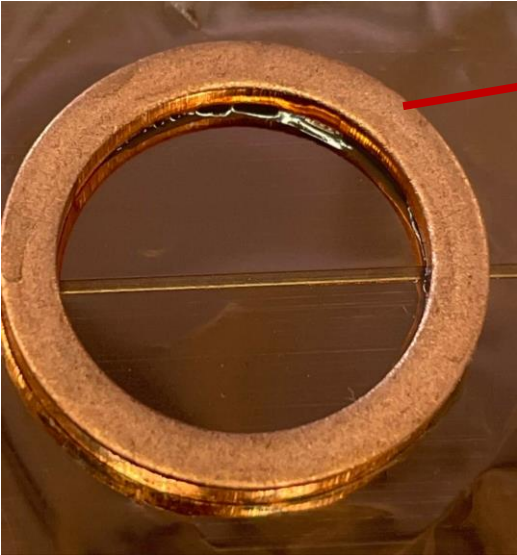


Permeation

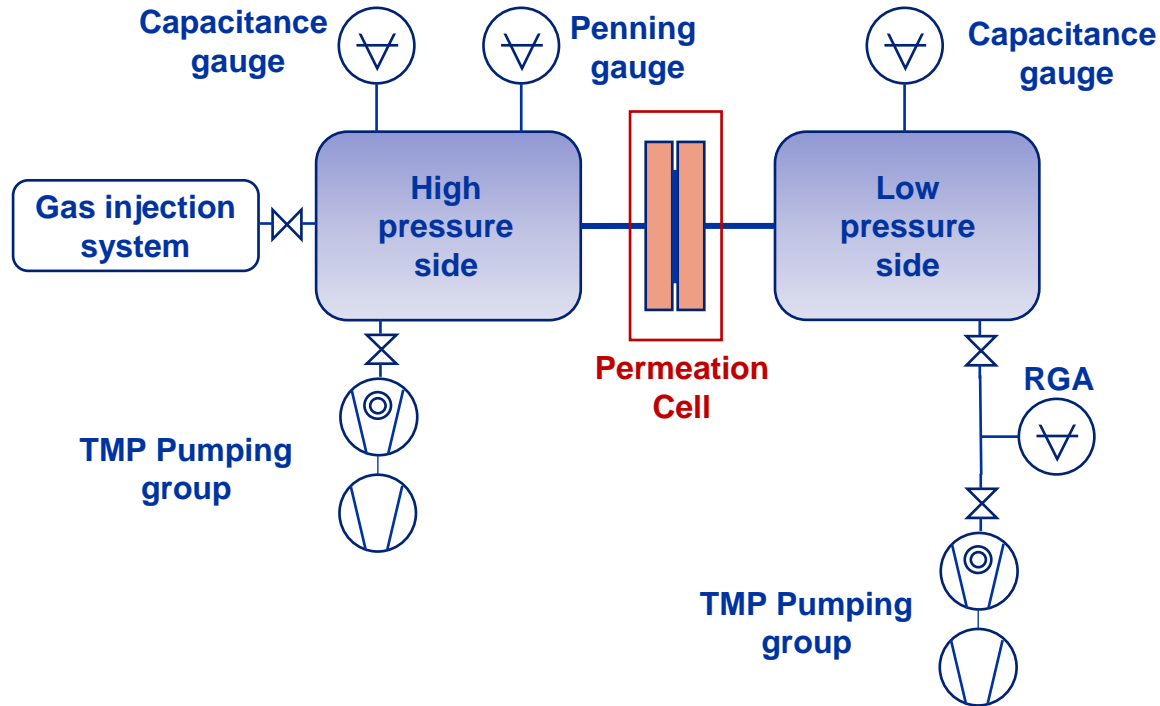
Polymer glued in between 2 copper gaskets

Kapton copper coated
DN16CF gaskets

Kapton
DN40CF gaskets



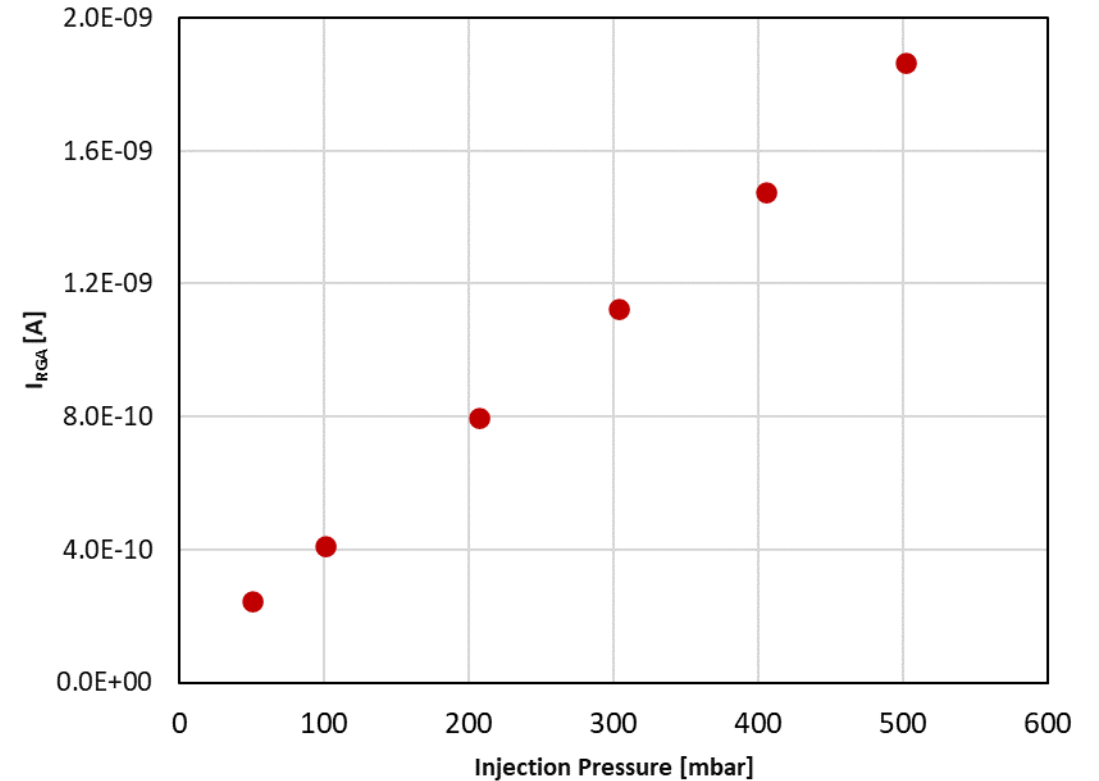
Permeation



$$Permeation = \frac{S_{eff} \Delta I_{RGA} l}{\alpha_{RGA} A P_{inj}}$$

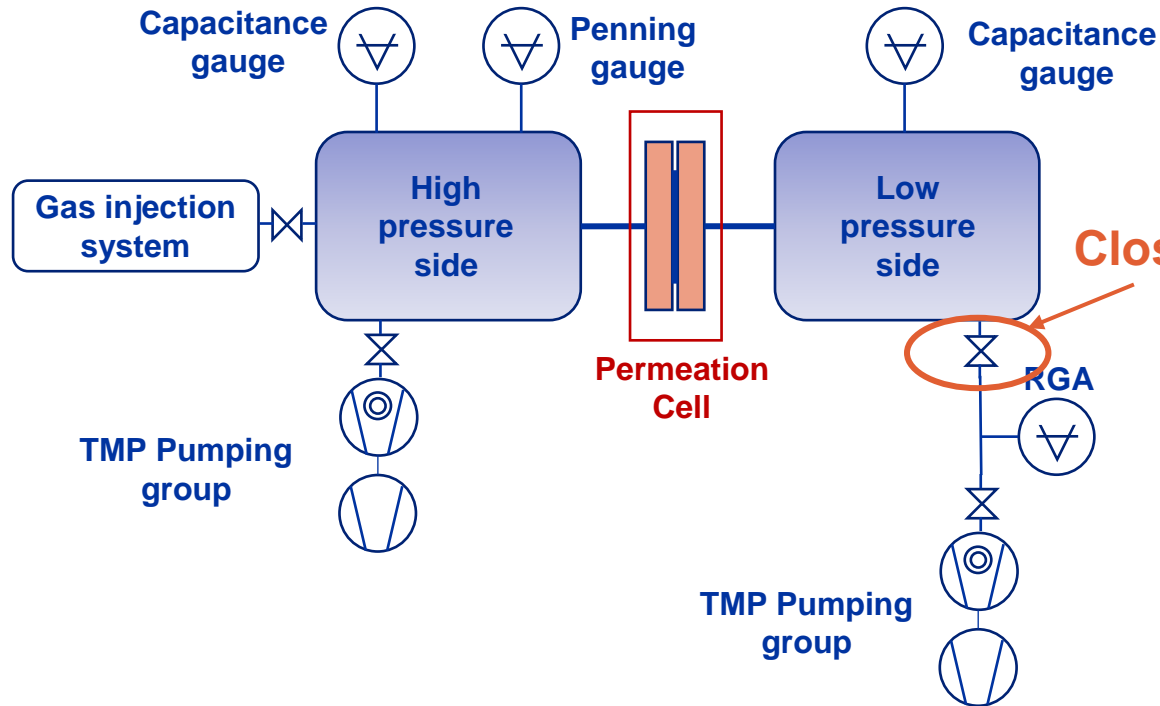
With: S_{eff} pumping speed - ΔI_{RGA} RGA signal for the injected gas - l sample thickness
 α_{RGA} Calibration factor - A sample area - P_{inj} injection pressure

Dynamic mode



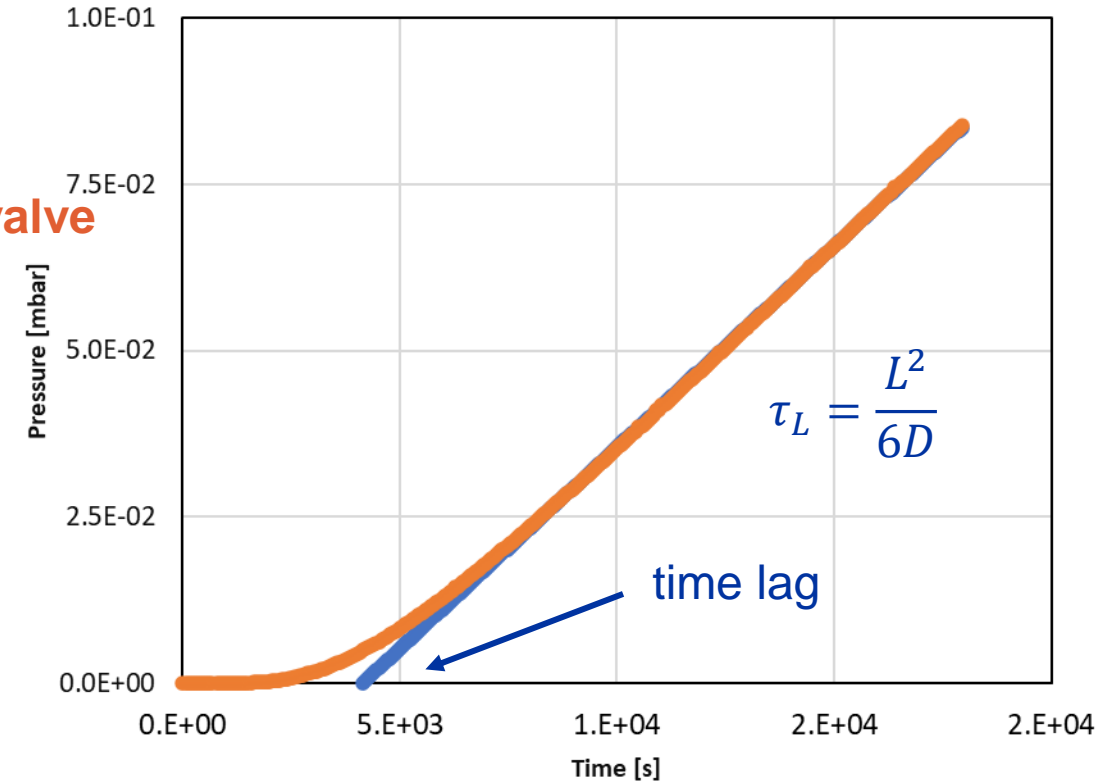
Permeation

Static mode



$$Permeation = \frac{\Delta P V_{lp} l}{\Delta t A P_{inj}}$$

With: $\Delta P / \Delta t$ slope at steady state - V_{lp} Volume low pressure side - l sample thickness
 A sample area - P_{inj} injection pressure





Thank you for your attention