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# Production, extraction and mass-separation of medical <sup>43,44,47</sup>Sc radionuclides at the CERN-MEDICIS facility

Author: Edgars Mamis

Supervisors: T. Stora (CERN); E. Pajuste (UoL); M. Radziņa (UoL)

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### **Theranostics**

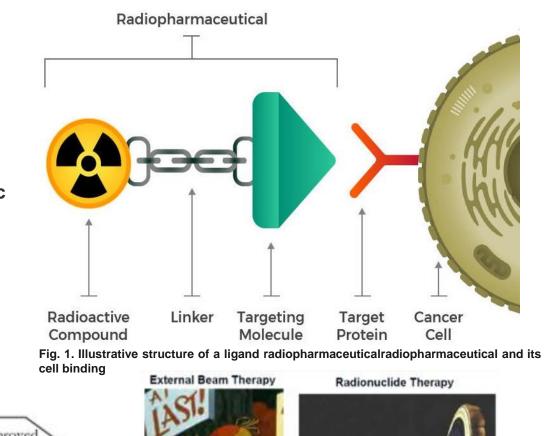
Radiopharmaceuticals - group of pharmaceutical drugs containing radioactive isotopes used for:

- SPECT Single Photon Emission Computed Tomography;
- PET/CT Positron Emission Tomography/Computed Tomography;
- Radionuclide therapy.

Biological molecules or sometimes artificial building blocks for specific targets, labelled with radioactive positron ( $\beta^+$ )/gamma or alpha and  $\beta^-$  emitters;

Theranostics - derived from <u>Therapy and Diagnostics</u>, and refers to the strategy of using radioactively labelled drugs for both purposes

So called "Treat what You see" technique



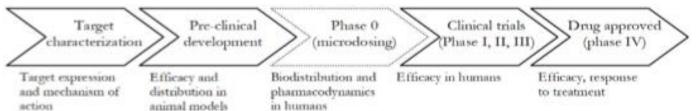


Fig. 2. Illustrative structure of drug develoment Current Radiopharmaceuticals, 2012, 5(2) 90-98.





# Why Scandium ?

#### Cost and efficiency:

- Sc radionuclides can be obtained from natural Ti and V targets cost efficient;
- > No expensive, enriched, low abundancy isotopes required;
- Sc from natural Ti can be obtained in sufficiently large quantities for medical applications – GBq of radioactivity;
- Good production cross-sections for medium energy (12-30 MeV) cyclotrons no nuclear reactors needed.

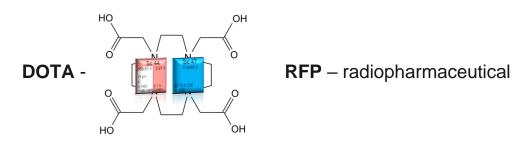
### <sup>43,44g</sup>Sc have diagnostic and <sup>47</sup>Sc therapeutic application decay properties – perfect for so called "matched pair" RFP's;

Same chemistry, different application;

(STI)

Labelled with DOTA chelator combined with peptides

### Scandium radionuclides can be produced and decay to the most biocompatible stable chemical elements such as Ca and Ti;



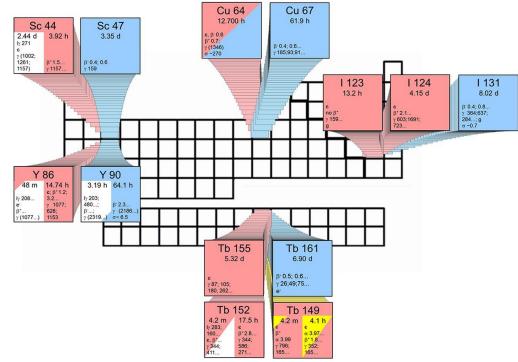


Fig. 3. Theragnostic principle: matched pairs of radionuclides for PET and SPECT imaging and for therapeutic application in nuclear medicine.

C. Müller, et. All. Promising Prospects for 44Sc-/47Sc-Based Theragnostics: Application of 47Sc for Radionuclide Tumor Therapy in Mice, *The Journal of Nuclear Medicine*, October 2014, 55 (10) 1658-1664; DOI: https://doi.org/10.2967/jnumed.114.141614







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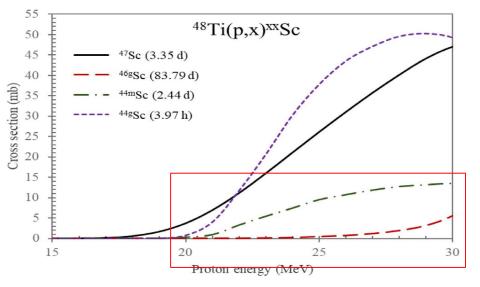
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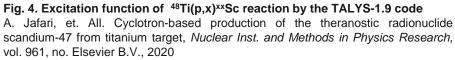
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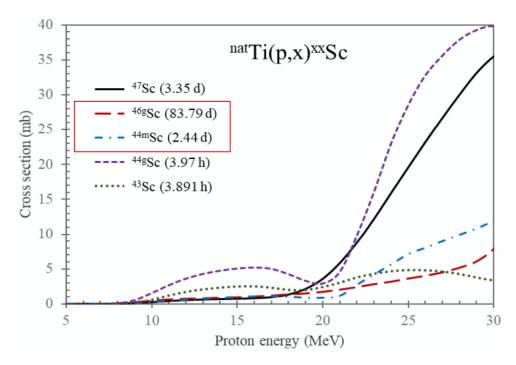
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## Nat-Ti as target material for <sup>4x</sup>Sc production

- Relatively cheap metal titanium is light, corrosion resistant, refractory and nontoxic;
- Natural titanium has five stable isotopes with different percentages, i.e. <sup>46</sup>Ti (8.25%), <sup>47</sup>Ti (7.44%), <sup>48</sup>Ti (73.72%), <sup>49</sup>Ti (5.41%), and <sup>50</sup>Ti (5.18%);
- Favorable theoretical nuclear reaction cross-sections of <sup>44g</sup>Sc and <sup>47</sup>Sc production compared to other Sc isotopes







**Fig. 5. Excitation function of** <sup>nat</sup>**Ti(p,x)**<sup>xx</sup>**Sc reaction by the TALYS-1.9 code** A. Jafari, et. All. Cyclotron-based production of the theranostic radionuclide scandium-47 from titanium target, *Nuclear Inst. and Methods in Physics Research,* vol. 961, no. Elsevier B.V., 2020





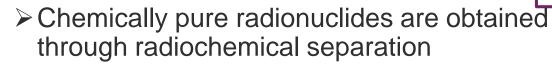
### **Radionuclide separation**

- Isotopically pure target material or ...
- **CERN-MEDICIS**  $\succ$ operates its massseparator to obtain isotopically pure medical radionuclides.
- Combination with cyclotron produced target yields in high molar activity, high purity medical radionuclides

**Mass-separator** 



Radiochemistry







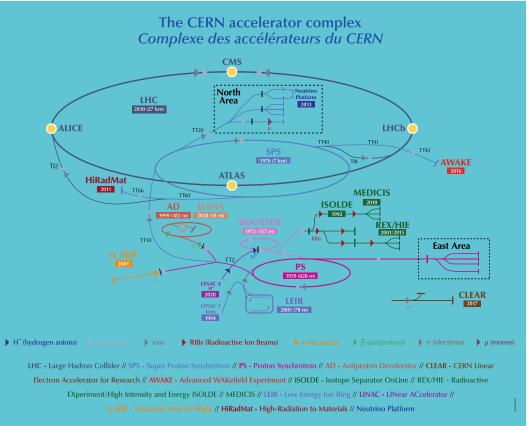


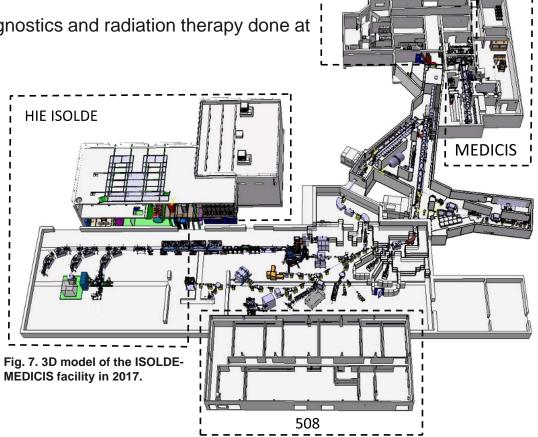
### **CERN-MEDICIS**

**MEDICIS** - MEDical Isotopes Collected from Isolde;

**ISOLDE** - Isotope mass Separator On-Line facility;

Production of non-conventional radionuclides for R&D in cancer imaging, diagnostics and radiation therapy done at partner institutes;





29:2:6:6

Class A labs

Located after the PS Booster, it receives protons with an energy of 1.4 (possibly 1.7 or 2 GeV in upgrade scenarios).

#### Fig. 6. CERN-MEDICIS in CERN accelerator complex



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### **CERN-MEDICIS** – view inside the bunker

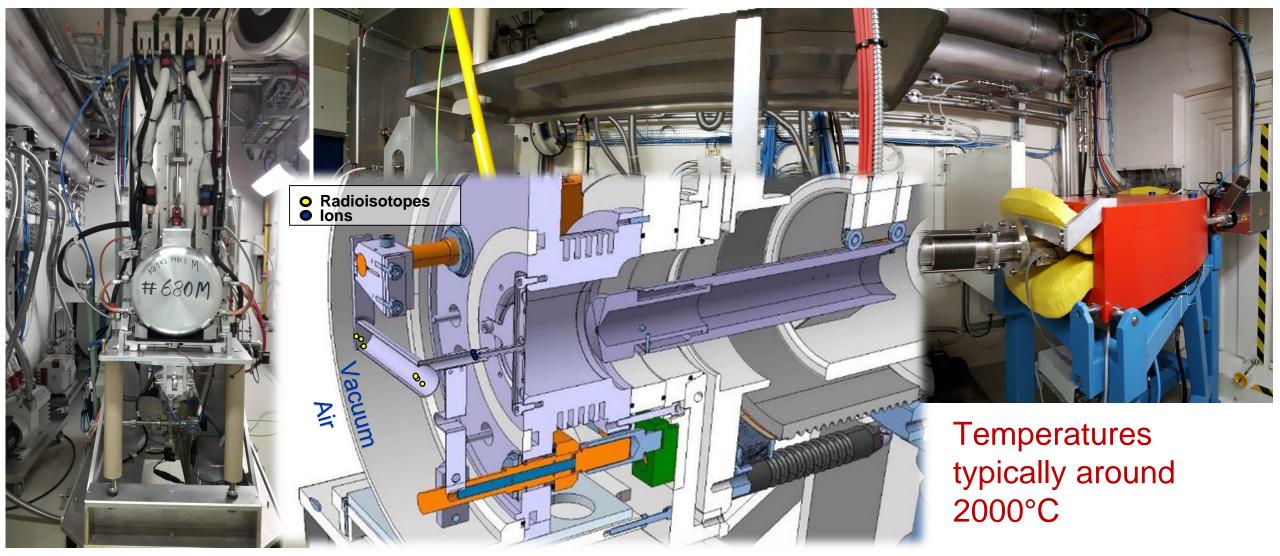


Fig. 8. Animated MEDICIS mass-separator radionuclide extraction principle



# History of Sc<sup>+</sup> and ScF<sub>x</sub><sup>+</sup> extraction at ISOLDE

- Sc extraction from irradiated <sup>nat</sup>Ti rolls as molecular halide beams was previously reported with W surface ion source at ISOLDE (1991);
  - > Sc<sup>+</sup> and ScF<sup>+</sup> beams were observed, but no ScF<sub>2</sub><sup>+</sup> molecular ions alongside evaporation of Ti target
  - ➢ W surface source is not very suitable for molecule ionization

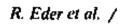
<u>Note:</u>  $Ti^+$  beam current rose from 0.1 to 5  $\mu$ A when fluorinating gas was added (Ti ionization efficiency is up to 1 % with surface source);

> During Sc radionuclide production contaminants such as <sup>46</sup>Sc ( $t_{1/2}$ =83d; E $\gamma$ = 889 keV; 1120 keV) sometimes cannot be avoided, even with enriched target materials;

Such high contaminant energy is bad for shielding and patient dosimetry;

> Mass-separation step therefore is mandatory

No prior data of natural vanadium as target material for Sc production.



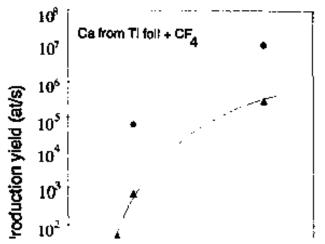


Fig. 9. Production yields of Sc<sup>+</sup> and ScF<sup>+</sup> ions from a 40 g/cm2 Ti foil target with a W surface ionizer. Irradiation 600 MeV protons.

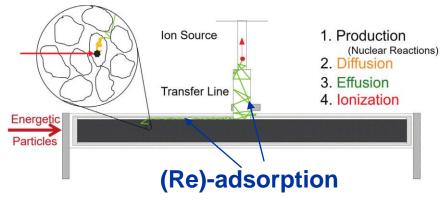
R. Eder, et al. The production yields of radioactive ion-beams from fluorinated targets at the ISOLDE on-line mass separator, *Nuclear Instruments and Methods in Physics Research B62 (1992)* 535-540, North-Holland







### **Molecular beams**





**Fig. 10.** Schematic of the ISOL method production, diffusion, effusion and ionization steps. J. P. F. P. Ramos, *Titanium carbide-carbon porous nanocomposite materials for radioactive ion beam production: processing, sintering and isotope release properties,* Lausanne: ÉCOLE POLYTECHNIQUE FÉDÉRALE DE LAUSANNE, 2017. CERN-THESIS-2016-247 ; EPFL-Thesis-n<sup>o</sup>7363 Fig. 11. Periodic table representing the temperature at which each element has a vapour pressure of 0.01 mbar.

J. P. F. P. Ramos, *Titanium carbide-carbon porous nanocomposite materials for radioactive ion beam production: processing, sintering and isotope release properties,* Lausanne: ÉCOLE POLYTECHNIQUE FÉDÉRALE DE LAUSANNE, 2017. CERN-THESIS-2016-247 ; EPFL-Thesis-nº7363

> Desired isotope extraction efficiency in most cases is enhanced by increasing temperature;

#### Up to a point where target material melts! Lower specific surface.

- Extraction of rare earth refractory metals such as Ti and Sc is challenging due to high boiling points and low vapour pressures;
- > Refractory metals react to target materials, structures and make stable bonds;
- Formation of volatile molecules and beams such as halides (ScF<sub>x</sub>; ScCl<sub>x</sub>; ScBr<sub>x</sub>; ect. x=1-3) are used for effective results and extraction of desired isotopes;
- Collection of desired element/molecule can be shifted to an atomic mass region without increased background (isobars) s contamints;
- Re-adsorbtion (sticking) of Sc and ScF<sub>x</sub> to Ta target structures hinders the extraction to ion source and Fig. 12. Intact (left) and molten (right) Ti roll target material. collection.







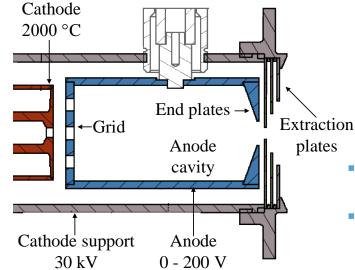


### **Versatile ionization - VADIS VD-5**

### Previousely surface and laser ion sources were used to ionize Sc elements and sometimes molecules;

- lonizing to q = +1 ions, molecules can dissociate in the ion source.
- The cross section for dissociation can exceed the cross section for direct ionization by orders of magnitude.

### By using surface ionizer Sc molecular beams release delay from target unit may be in order of several hours to days;



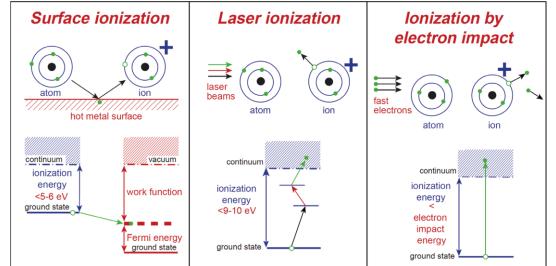


Fig. 13. Ionization mechanisms used in ISOLDE: surface, laser and electron impact ionization. J. P. F. P. Ramos, Titanium carbide-carbon porous nanocomposite materials for radioactive ion beam production: processing, sintering and isotope release properties, Lausanne: ÉCOLE POLYTECHNIQUE FÉDÉRALE DE LAUSANNE, 2017. CERN-THESIS-2016-247 ; EPFL-Thesis-nº7363

- VD-5 is an electron impact ion source with thermionic cathode as electron source and anode cavity for electron acceleration up to couple hundreds of volts with applied axial magnetic field.
- VD-5 can ionize elements/molecules with high ionization potentials (>10 eV).
  - Limiting factor is specific matrix, molecule/element ionization efficiency

#### Fig. 14. Geometry of ISOLDE FEBIAD ion source (VADIS).

Y. M. Palenz, Characterization and optimization of a versatile laser and electron-impact ion source for radioactive ion beam production at ISOLDE and MEDICIS, Belgium: KU Leuven - Faculty of Science, 2019. CERN-THESIS-2019-032

FEBIAD - Forced-Electron Beam Induced Arc-Discharge Ion Source VADIS - Versatile Arc Discharge Ion Source

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# Stable ScF<sub>x</sub><sup>+</sup> and TiF<sub>x</sub><sup>+</sup>;TiOF<sub>x</sub><sup>+</sup> molecular beams (x=1-3)

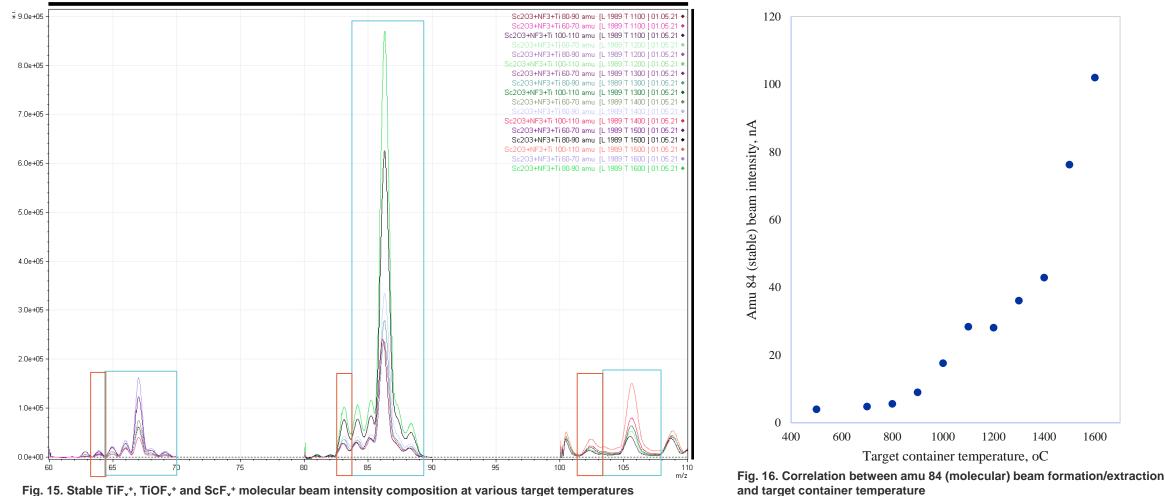


Fig. 15. Stable TiF<sub>x</sub><sup>+</sup>, TiOF<sub>x</sub><sup>+</sup> and ScF<sub>x</sub><sup>+</sup> molecular beam intensity composition at various target temperatures

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### Sc molecular beam target development and constraints

Main total beam contaminants - Be, C, Ta. Large portion of  $\geq$ total ion beam consists  $BeF_x$ ,  $TaF_x$  and  $TaO_xF_y$ ;

 $\triangleright$  Reduced the background of TaF<sub>x</sub> and TaOF<sub>x</sub> molecular beam contribution to total beam by installing additional cooling of gas leak.

- > Reactive gas reaction with target structures mitigated at calibrated gas leak and Ta tube connection.
- > Maximising production of radionuclides with double foil rolls.
- Target temperature cap at 1550 1600 °C.

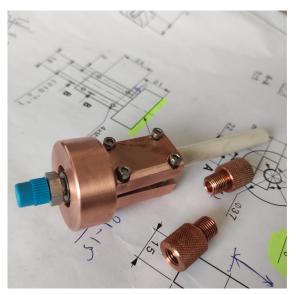


Fig. 18. ISOLDE/MEDICIS target gas supply leak modification (cooling)



Fig. 19. MEDICIS separated isotope collection foils



Fig. 20. natTi double rolls used for Sc radionuclide production

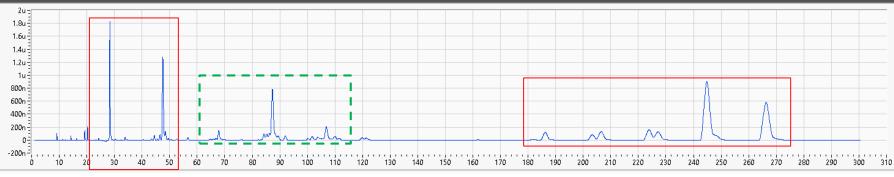


Fig. 17. nat-Ti +  $Sc_2O_3$  + (x)NF<sub>3</sub> system full range mass scan – beam current distribution



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# 44m,44gSc and 47Sc fluoride collection

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- Sc extraction from irradiated <sup>nat</sup>Ti rolls as molecular halide nA beams here was done with hot plasma VADIS VD-5 ion source target;
- ScF<sub>x</sub> molecular beams were always acompanied/masked 90
  by TiF<sub>x</sub> isobars due to reactivity and volatility of Ti halides;
- Di-fluoride (TiF<sub>2</sub><sup>+</sup>, ScF<sub>2</sub><sup>+</sup>) molecular beams are (and were expected) most dominant from the species, with <sup>48</sup>TiF<sub>2</sub><sup>+</sup> reaching ~1 µA beam intensities;

Dissociation of ScF<sub>3</sub> rather than direct ionization

- Conditions to separate ScF<sub>2</sub><sup>+</sup> in a hot plasma ion source were met, keeping TiF<sub>x</sub> currents under control in order not to pollute/damage separator;
- Successful separation from long lived <sup>46</sup>Sc and <sup>48</sup>Sc contaminants was achieved with MEDICIS massseparator;
- Identified strong 511keV (from positron annihilation) line during operation with target 741M
- > Collected radionuclide activity in range of **few kBq**.

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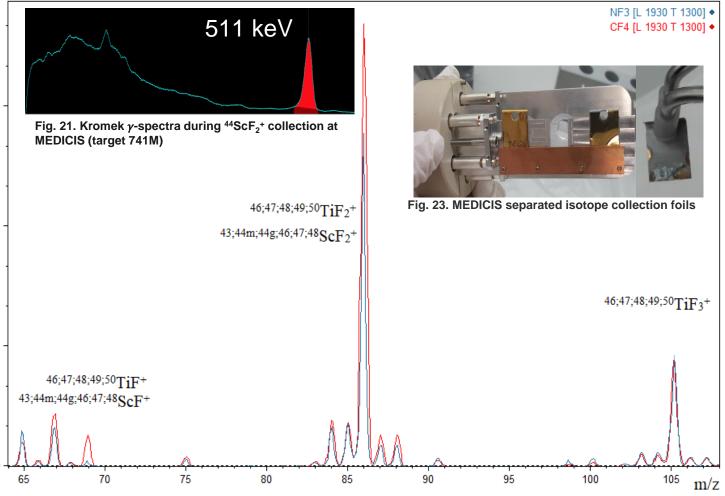


Fig. 22. Mass scan fragments from medical Sc extraction by different fluorinating gases at same operation conditions, identifying strong TiF<sub>x</sub> beams – MEDICIS, target #741M

# <sup>51</sup>V for targets

- >  $^{51}V(p,\alpha p)^{47}Sc$  reaction with cyclotron
- Melting point 1910 °C
- Can be achieved with <sup>nat</sup>V targets (<sup>51</sup>V: 99.750%, <sup>50</sup>V: 0.250%)
- <sup>nat</sup>V material is commercially available in metallic form, different geometries and purity up to 99,8%
- Does not require enriched target material
- Reported that at 18–30 MeV the production of <sup>47</sup>Sc is calculated to be 31 MBq/µA and 82 MBq/µA for 24 h and 80 h irradiation runs respectively.
  - Co-production of <sup>46</sup>Sc can be considered negligible 0.01 MBq/µA and 0.03 MBq/µA.

Energy <sup>nat</sup> V foil (MeV)	<sup>47</sup> Sc cross sec- tion (mb)	<sup>46</sup> Sc cross sec- tion (mb)	Reference radionuclide	Energy monitor foil (MeV)	Monitor cross section (mb)
26.5±0.8	4.3± 0.2		<sup>57</sup> Ni	26.3±0.9	180.2±7.5
29.2±0.8	7.4± 0.5		<sup>57</sup> Ni	29.0±0.8	162.9± 6.8
31.2±0.7	9.0±0.5	0.6± 0.1	<sup>57</sup> Ni	31.1±0.7	136.6± 5.7
33.3±0.6	10.9± 0.6	2.6± 0.2	<sup>57</sup> Ni	33.5±0.6	113.9± 4.8
39.5±0.6	8.3±0.5	18.6± 1.1	<sup>57</sup> Ni	39.3±0.6	85.6±3.6
48.1±0.8	4.6± 0.3	30.8±2.4	<sup>24</sup> Na	48.1 ± 0.8	4.8± 0.3
51.0± 0.7	4.1±0.3	27.8± 2.3	<sup>24</sup> Na	50.9±0.7	6.4± 0.5
53.6± 0.6	4.1±0.3	25.1 ± 2.0	<sup>24</sup> Na	53.7±0.5	8.0± 0.6
55.7±0.8	$4.4 \pm 0.4$	23.0± 1.9	<sup>24</sup> Na	55.6±0.8	9.0±0.6
58.3±0.7	$5.0 \pm 0.4$	21.6± 1.8	<sup>24</sup> Na	58.2±0.7	10.0± 0.7
60.7±0.6	5.1±0.3	18.6± 1.2	<sup>24</sup> Na	60.7±0.5	10.8± 0.5
65.5±0.8	$6.8 \pm 0.4$	16.7± 1.0	<sup>24</sup> Na	65.5±0.8	11.4± 0.5
67.8± 0.7	7.7±0.5	16.7± 1.0	<sup>24</sup> Na	67.7±0.7	11.6± 0.5
70.0±0.6	8.0±0.4	16.4± 0.9	<sup>24</sup> Na	70.0± 0.5	11.6± 0.4

Table 1. Data of the  $^{nat}V(p,x)^{47}Sc$ ,  $^{46}Sc$  nuclear cross sections and the values of the IAEA reference cross sections used.

Gaia Pupillo, et.all. Production of 47Sc with natural vanadium targets: results of the PASTA project, Journal of Radioanalytical and Nuclear Chemistry, 2019, https://doi.org/10.1007/s10967-019-06844-8







### **CERN FLUKA Monte-Carlo calculations of Sc production from Ti and V**

Irradiation mode	Irradiatio n time	Decay time	Sc-44m activity	Sc-44g activity	Sc-47 activity
			Bq	Bq	Bq
Direct	1 hour	1 hour	1.176e+08 +/- 0.82%	1.389e+09 +/- 0.81%	2.244e+08 +/- 1.40%
Indirect			7.999e+06 +/- 1.47%	9.446e+07 +/- 1.46%	1.702e+07 +/- 5.80%
Direct	4 hours	1 hour	4.622e+08 +/- 0.82%	4.456e+09 +/- 0.79%	8.862e+08 +/- 1.40%
Indirect			3.131e+07 +/- 1.47%	3.018e+08 +/- 1.41%	6.720e+07 +/- 5.80%
Direct	14 hours	1 hour	1.527e+09 +/- 0.82%	8.824e+09 +/- 0.73%	2.973e+09 +/- 1.40%
Indirect			1.038e+08 +/- 1.47%	6.001e+08 +/- 1.30%	2.255e+08 +/- 5.80%

Irradiation mode	Irradiation time	Decay time	Sc-43 activity Bq	Sc-44m activity Bq	Sc-44g activity Bq	Sc-47 activity Bq
Direct		1 hour	4.691e+08 +/- 1.50%	8.133e+07 +/- 0.70%	9.604e+08 +/- 0.69%	9.202e+07 +/- 0.80%
Indirect	1 hour		5.069e+07 +/- 4.10%	8.206e+06 +/- 2.10%	9.689e+07 +/- 2.06%	9.839e+06 +/- 2.40%
Direct	4 hours	1 hour	1.465e+09 +/- 1.50%	3.196e+08 +/- 0.70%	3.081e+09 +/- 0.67%	3.634e+08 +/- 0.80%
Indirect	4 hours		1.583e+08 +/- 4.10%	3.225e+07 +/- 2.10%	3.109e+08 +/- 2.02%	3.886e+07 +/- 2.40%
Direct	14 hours 1	1 hour	2.638e+09 +/- 1.50%	1.056e+09 +/- 0.70%	6.102e+09 +/- 0.62%	1.220e+09 +/- 0.80%
Indirect			2.850e+08 +/- 4.10%	1.065e+08 +/- 2.10%	6.156e+08 +/- 1.86%	1.304e+08 +/- 2.40%

Table 2. Results of CERN FLUKA 4.1 natTi(p,x)Sc-44m, Sc-44g and Sc-47 production with 1.4 GeV proton direct impact (C. Duchemin)

Table 3. Results of CERN FLUKA 4.2.2 natV(p,x)Sc-43, Sc-44m, Sc-44g and Sc-47 production with 1.4 GeV proton direct impact

- > In both cases direct irradiation with 6.25E+12 primary/second (1  $\mu$ A) have been considered.
- > Titanium target's density considered is 0.8 g.cm<sup>-3</sup> and its volume is **30.17 cm<sup>3</sup>** (mass of 24 g).
- $\blacktriangleright$  Vanadium target's density considered is 0.87 g/cm<sup>-3</sup> and its volume is **52.16 cm<sup>3</sup>** (mass of **45,29 g**).
- Calculations were done also for 1.7 GeV proton irradiation with less radionuclide production and favouring Ti as production material.

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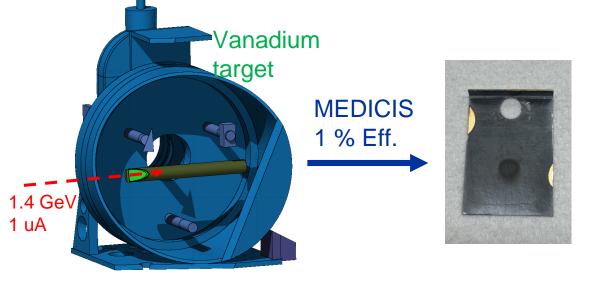




### **Element composition on MEDICIS collection foil**

Element	Quantity (if OF are also collected), mol	Quantity (if no OF are collected), mol	Quantity with OF (if 1% efficiency), mol	Quantity no OF (if 1% efficiency), mol
Са	4,05E-09	4,00E-09	4,05E-11	4,00E-11
Sc	3,46E-09	1,42E-09	3,46E-11	1,42E-11
Ti	2,97E-09	3,95E-10	2,97E-11	3,95E-12
V	3,57E-16	0,00E+00	3,57E-18	0,00E+00

Table 4. Sc-44 collection foil element composition contaminants



Element	Quantity (if OF are also collected), mol	Quantity (if no OF are collected), mol	Quantity with OF (if 1% efficiency), mol	Quantity no OF (if 1% efficiency), mol
AI	5,42E-14	5,42E-14	5,42E-16*	5,42E-16 <sup>*</sup>
Si	1,80E-09	1,80E-09	1,80E-11*	1,80E-11*
Са	5,55E-11	5,55E-11	5,55E-13	5,55E-13
Sc	2,05E-09	2,05E-09	2,05E-11	2,05E-11
Ті	1,47E-08	6,39E-09	1,47E-10	6,39E-11
V	1,28E-08	3,57E-16	1,28E-10	3,57E-18
Cr	4,67E-10	0,00E+00	4,67E-12	0,00E+00

Table 5. Sc-47 collection foil element composition contaminants. Al and Si compounds ure very unlikely to be extracted and expected.

> OF – Oxyfluorines. Oxyfluorines can form in presence of oxygen.

- Crude 1% Extraction efficiency presumed from a directly irradiated target.
- Not taken into account ratio of molecule species







# **Maximising output**

- MEDICIS Double collection slit system installed and tested
- ➤ Collection of Sc-44 and Sc-47 silmutanously
- Successful Tm-165 and Tm-167 collection (100 MBq range)



Fig. 24. Collection chamber for double collection with reduced sputtering

(STI)

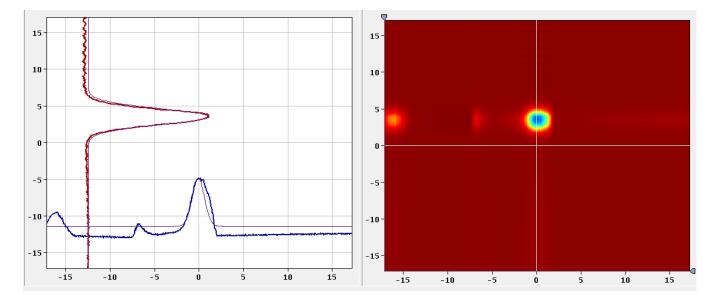


Fig. 25. Double beam collection profile of Tm-165 and Tm-167 silmutanously



Fig. 26. Kromek γ-spectra during double beam collection of Tm-165 and Tm-167 silmutanously



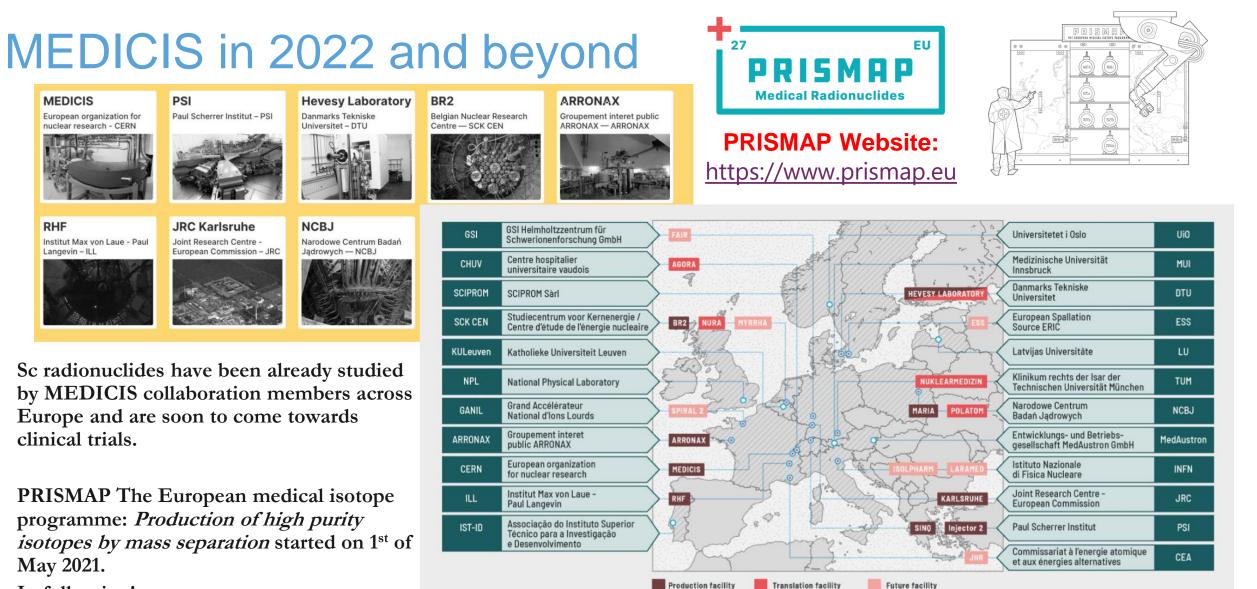




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In full swing! PRISMAP will create a single-entry point for a fragmented user community distributed amongst universities, research centres, industry and hospitals.



#### SY Accelerator Systems







