

**NEW COMPOSITE MATERIALS FOR DECREASING OF  
RADIOACTIVE MOLECULAR IODINE IN THE WATER  
COOLANT ON THE WORKING AND NEW DEVELOPED NPPs.**

**Prof. Sergey A. Kulyukhin  
Dr. Lubov' V. Mizina  
Eng. Anna A. Tishina**

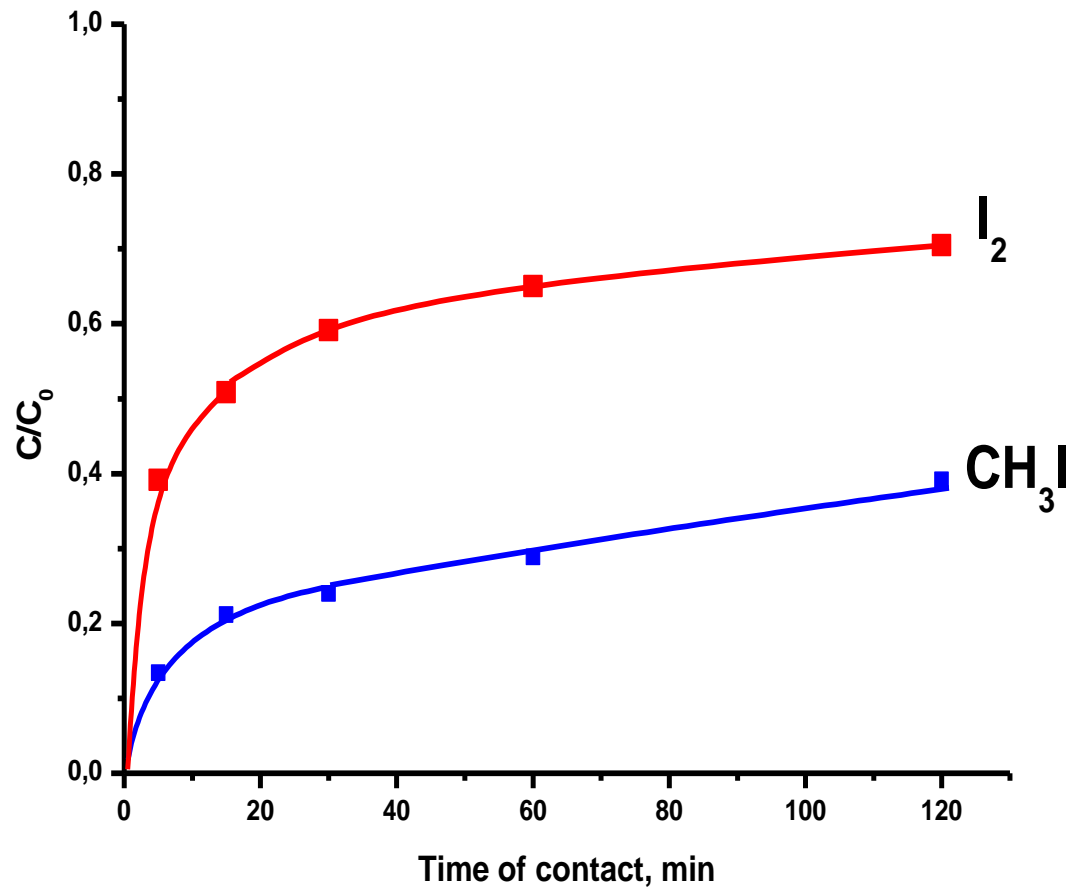
**Institute of Physical Chemistry and Electrochemistry  
Russian Academy of Sciences  
Moscow, Russia**

**$I^-$ ,  $IO_3^-$ ,  $I_2$ ,  $CH_3I$**

**Anion- and cation-  
ion-exchange resins**

**KU-2 and AV-18**

# The absorption of $I_2$ and $CH_3I$ on anion-exchange resin AV-18 from the water coolant of WWER-type NPPs\*



\* The water coolant of the WWER-type NPPs:  
[NaCl] = 1.17 mg/dm<sup>3</sup>; [NH<sub>4</sub>Cl] = 29.96 mg/dm<sup>3</sup>; [H<sub>3</sub>BO<sub>3</sub>] = 5 g/dm<sup>3</sup>; pH 7 - 8.5

## Distribution coefficients $K_d$ for $I_2$ and $CH_3I$ on anion-exchange resin AV-18 from the water coolant of WWER-type NPPs

( $V/m = 100$ ;  $[CH_3^{131}I] = 10^{-5} M$ ;  $[I_2] = 9 \cdot 10^{-4} M$ )

NN	Time of contact between solid and liquid phases	$K_d, cm^3/g$	
		$I_2$	$CH_3I$
1	15 min	$103.5 \pm 3,0$	$12.9 \pm 0,6$
2	60 min	$145.1 \pm 4,5$	$30.2 \pm 1,4$
3	120 min	$244.5 \pm 6,1$	$41.9 \pm 1,7$
4	24 h	$>10^3$	$182.7 \pm 3,4$

\* The water coolant of the WWER-type NPPs:

$[NaCl] = 1.17 mg/dm^3$ ;  $[NH_4Cl] = 29.96 mg/dm^3$ ;  $[H_3BO_3] = 5 g/dm^3$ ; pH 7 - 8.5

**Aim of the work** is to develop a new effective material to absorb molecular and organic iodine from water solutions

**It is desirable to employ materials that are used on working NPPs to coolant decontamination**

**Ion-exchange resin - KU-2**

## Some characteristics of the new materials based on cation-exchange resin KU-2

Material	Quantity of d-elements on cation-exchange resin KU-2		Colour
	mg/g	% wt.	
<b>IPCE-C1</b>	79.93	8.0	brown with metal gloss
	81.28	8.1	
<b>IPCE-C2</b>	55.69	5.6	brown with metal gloss
	54.68	5.5	
<b>IPCE-CS</b>	99.82	10.0	grey
	103.02	10.3	

**IPCE-C1** - about 8.0 wt.% of Cu

**IPCE-C2** - about 5.5 wt.% of Cu

**IPCE-CS** - the total content of metals about 10.0 wt.%  
at molar relation Cu : Ag = 4 : 1

## Distribution coefficients $K_d$ for $^{131}\text{I}^-$ , $^{131}\text{IO}_3^-$ , $\text{CH}_3^{131}\text{I}$ and $\text{I}_2$ on the modified resin KU-2 from a water coolant of the WWER-type NPPs\*

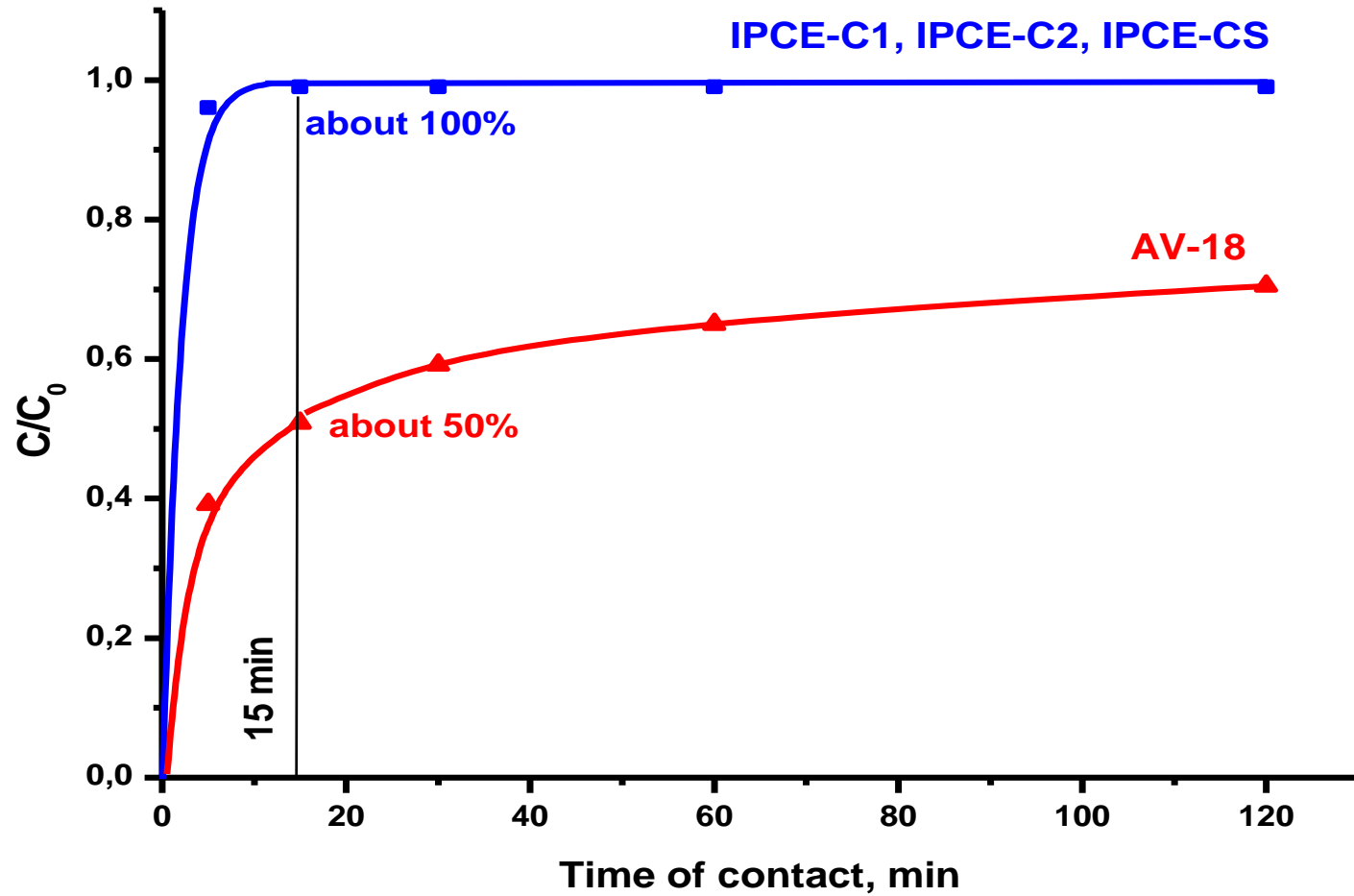
(V/m = 100;  $[\text{K}^{131}\text{IO}_3] = 10^{-5} \text{ M}$ ;  $[\text{K}^{131}\text{I}] = 10^{-5} \text{ M}$ ;  $[\text{CH}_3^{131}\text{I}] = 10^{-5} \text{ M}$ ;  $[\text{I}_2] = 9 \cdot 10^{-4} \text{ M}$   
Time of contact between liquid phase and material is 120 min)

NN	Material	$K_d, \text{ cm}^3/\text{g}$			
		$^{131}\text{I}^-$	$^{131}\text{IO}_3^-$	$\text{I}_2$	$\text{CH}_3^{131}\text{I}$
1	KU-2	-	-	$1.5 \pm 0.4$	$1.5 \pm 0.4$
2	IPCE-C1	$21.2 \pm 2.0$	$36.4 \pm 1.6$	$(2.1 \pm 0.9) \cdot 10^3$	$8.5 \pm 1.3$
3	IPCE-C2	$150.0 \pm 5.5$	$41.2 \pm 2.7$	$(2.3 \pm 0.8) \cdot 10^3$	$8.5 \pm 1.3$
4	IPCE-CS	$136.0 \pm 5.0$	$35.2 \pm 1.9$	$(2.3 \pm 1.0) \cdot 10^3$	$10.6 \pm 1.4$

\* The water coolant of the WWER-type NPPs:

$[\text{NaCl}] = 1.17 \text{ mg}/\text{dm}^3$ ;  $[\text{NH}_4\text{Cl}] = 29.96 \text{ mg}/\text{dm}^3$ ;  $[\text{H}_3\text{BO}_3] = 5 \text{ g}/\text{dm}^3$ ; pH 7 - 8.5

# Dependence of I<sub>2</sub> sorption on the time of contact between solid and liquid phases





# THANK YOU !

