

# Annealing of the vacancy- oxygen and divacancy- oxygen complexes in silicon

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Special thanks to:  
The Norwegian Research Counsel (NFR) for Financial Support



# Background



- During irradiation we get:
  - $V + O_i \rightarrow VO$
  - $V + V \rightarrow V_2$( $V_2$  is also produced directly)

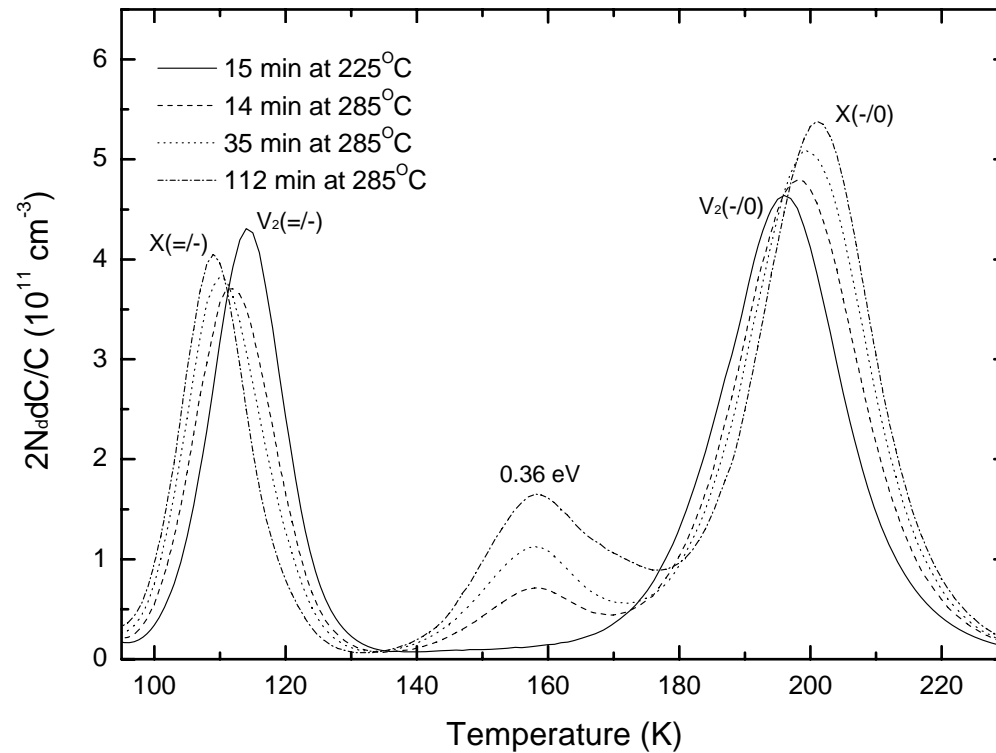
- For higher doses:
  - $V + VO \rightarrow V_2O$

Basis for *I*-level identification

- Another channel for  $V_2O$  formation:
  - $V_2 + O_i \rightarrow V_2O$

Basis for *X*-level identification

# Background



- X was identified as  $V_2O$
- $V_2$  migrates:
  - $V_2 + O_i \rightarrow V_2O$

*M. Mikelsen, E. V. Monakhov, G. Alfieri, B. S. Avset, and B. G. Svensson  
Phys. Rev. B 72, 195207 (2005)*

# Experiment

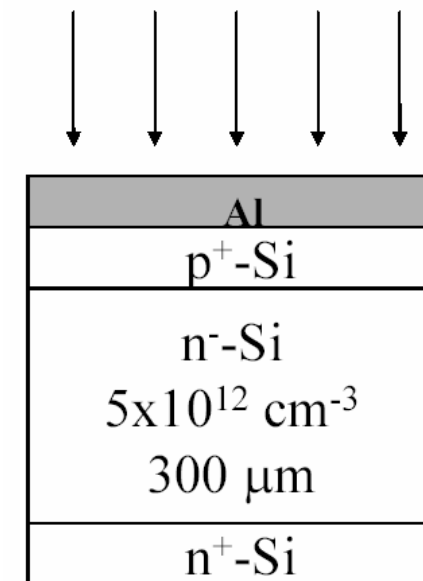


Survey of the samples used in the study:

Sample	Doping (P/cm <sup>3</sup> )	Carbon (cm <sup>-3</sup> )	Oxygen (cm <sup>-3</sup> )
MCZ-Si	$5.5 \times 10^{12}$	$\leq 10^{16}$	$(5-10) \times 10^{17}$
DOFZ-Si	$5.0 \times 10^{12}$	$(2-4) \times 10^{16}$	$(2-3) \times 10^{17}$

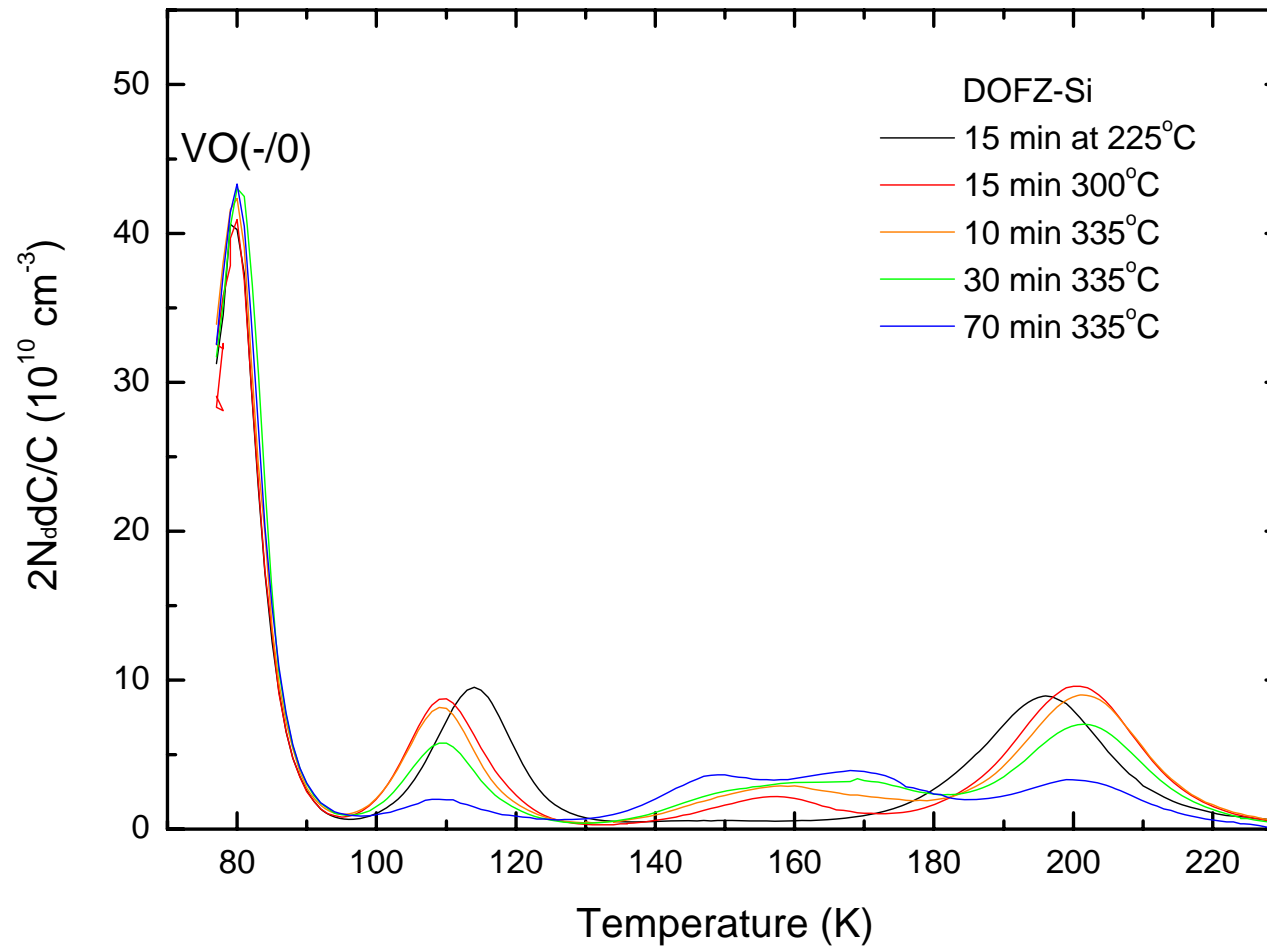
- DOFZ = Diffusion-Oxygenated Float-Zone Si
- MCZ = Magnetic Czochralski
- **Experimental method:**
  - DLTS
  - Pre-annealing to insure a  $V_2 \rightarrow V_2O$  transformation.
  - Isothermal annealing in the range 275-355 °C.

6 MeV electrons

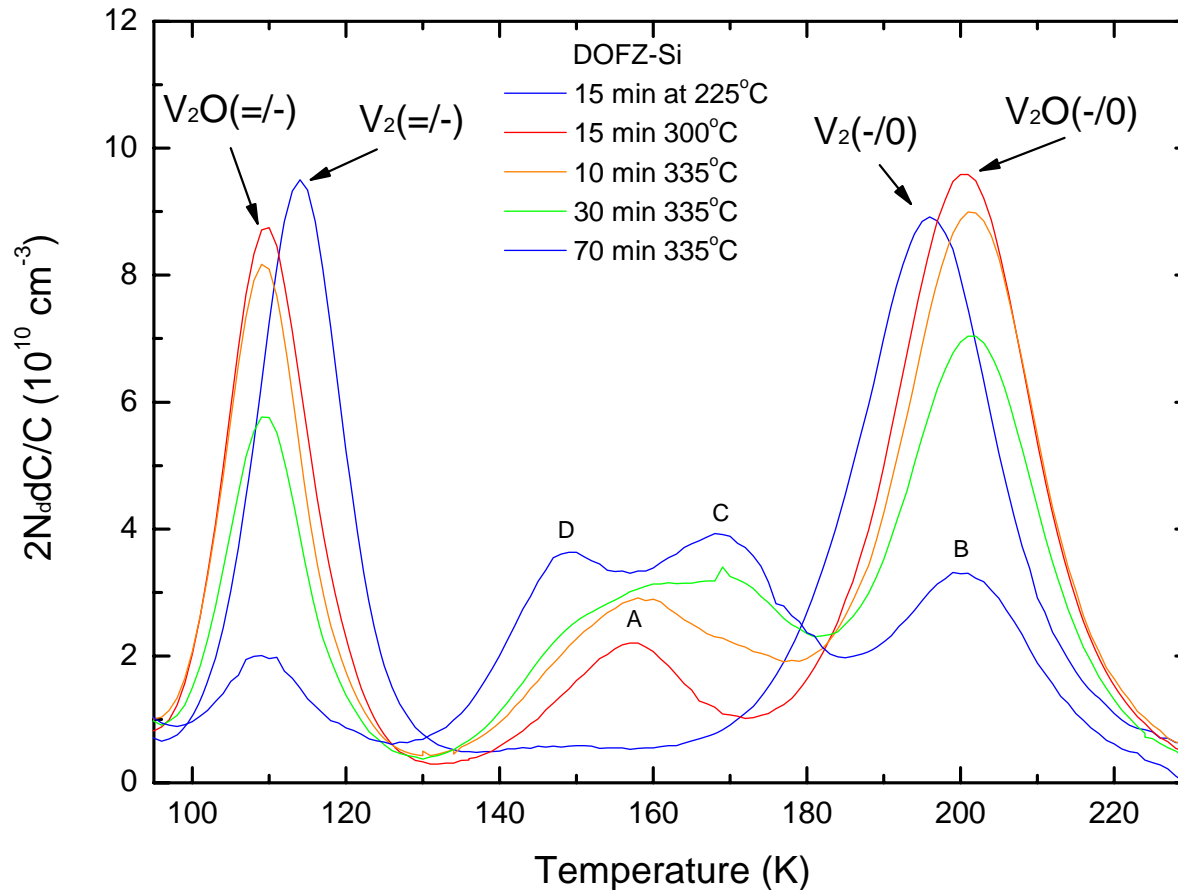


Dose:  $2 - 5 \times 10^{12} \text{ cm}^{-2}$

# Results: DOFZ - Si



# Results: DOFZ - Si



## Other levels observed:

- □ A at ~157 K (\*)
  - Similar electrical parameters as  $V_2O(-/0)$ .
- □ B at ~200 K
- □ C at ~175 K
- □ D at ~157 K

(\*) To be discussed in:

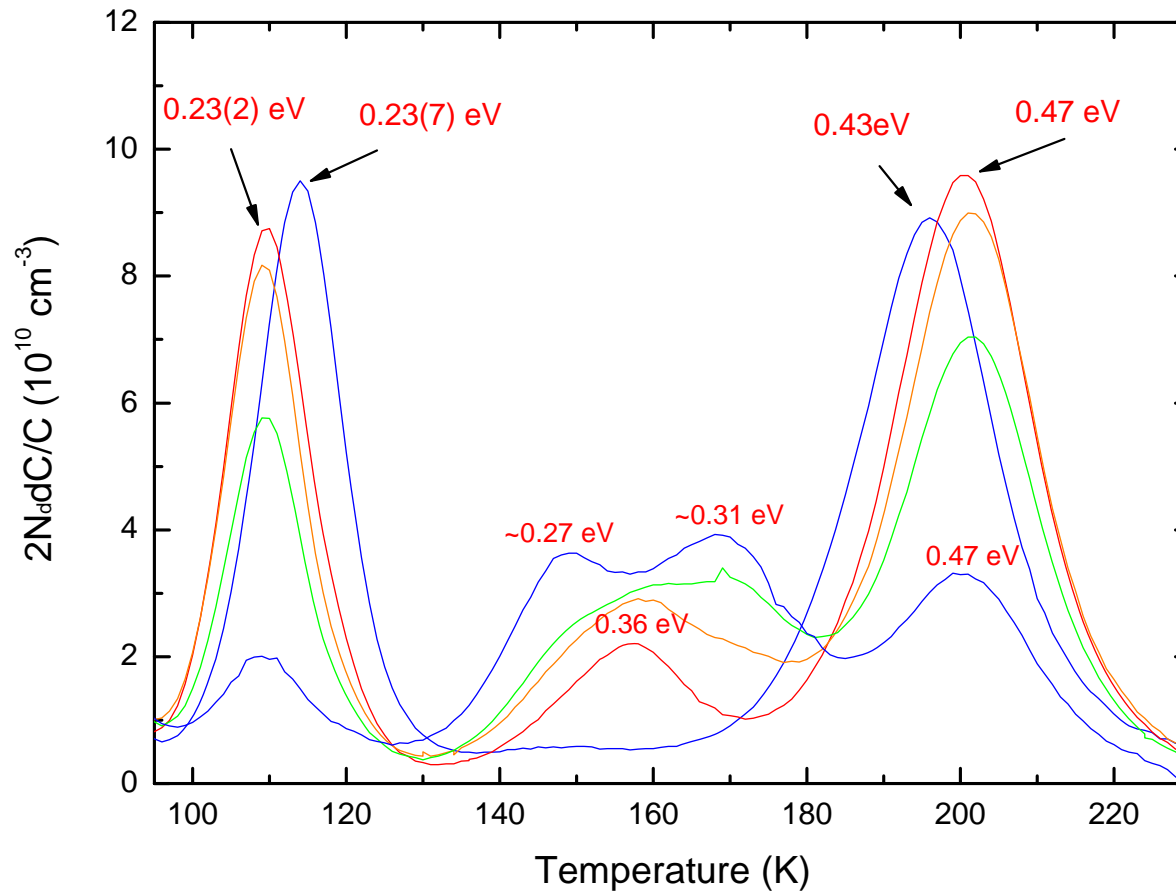
"On the formation of the L-center in silicon during heat treatment in the temperature range 205 to 285°C"

M. Mikelsen, E. V. Monakhov, B. S. Avset, B. G. Svensson

To be published in "Physica scripta"

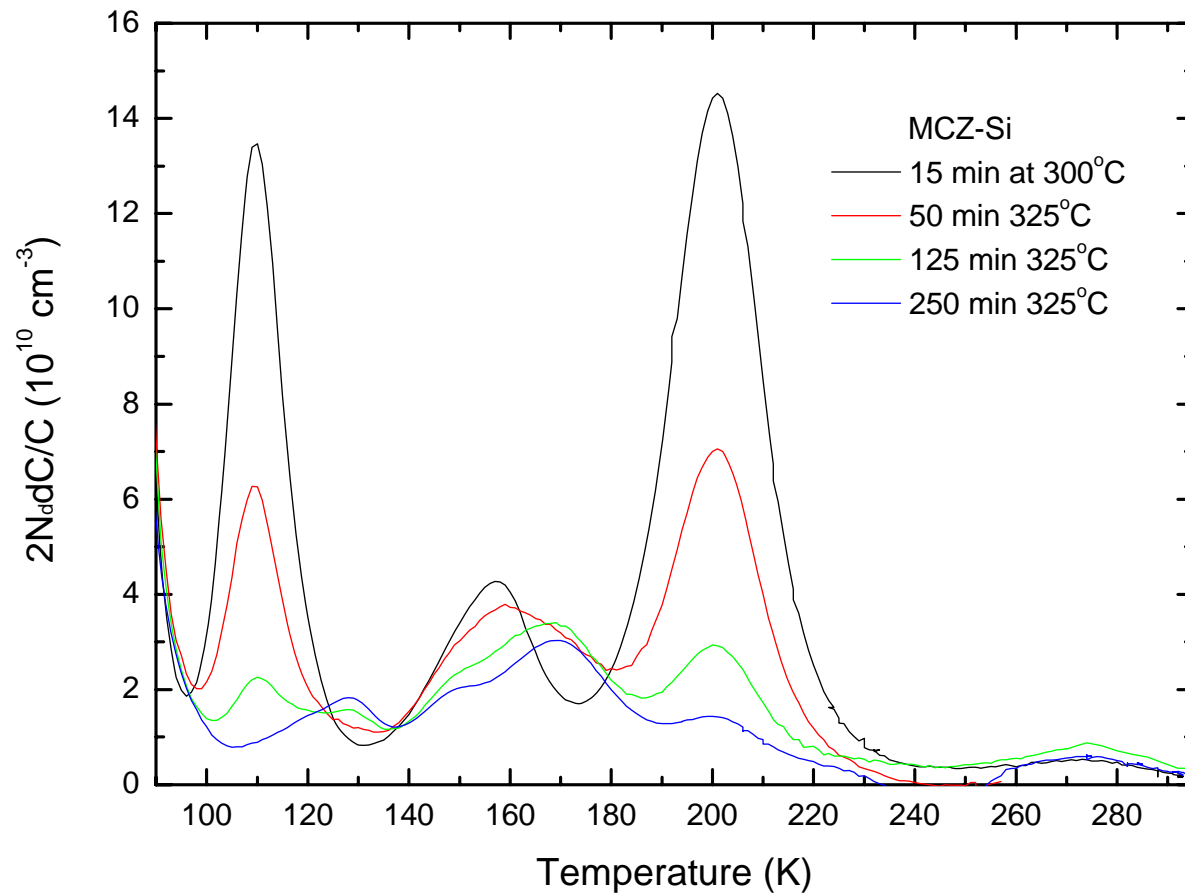
madmike@fys.uio.no

# Results: DOFZ - Si



- The activation enthalpies for peak C and D were obtained by using a computer program that fits the spectrum to a sum of single peaks.

# Results: MCZ Si



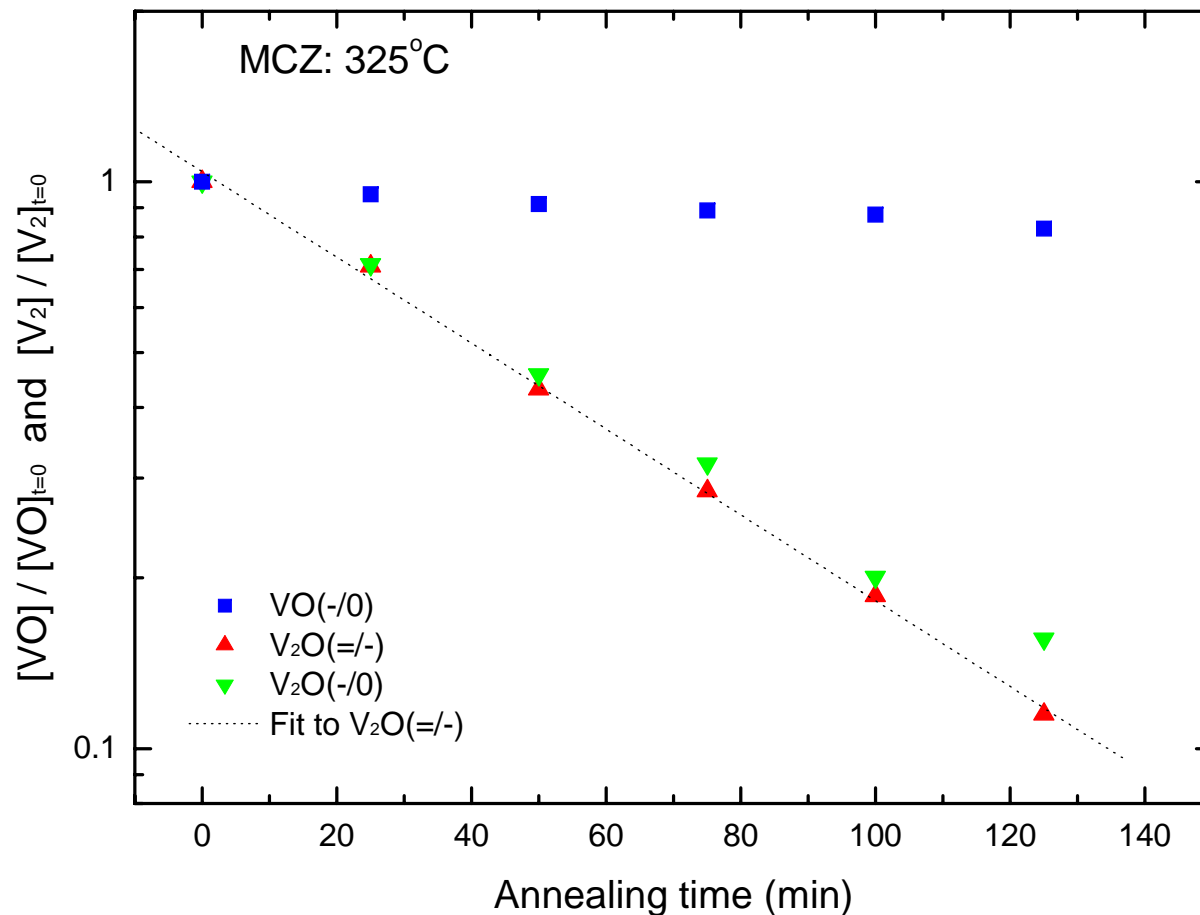
- MCZ-Si behaves similar to DOFZ
  - But the area between the two  $V_2O$  peaks looks a little bit different. (No D-peak)



# Results: $V_2O$ annealing



Example:

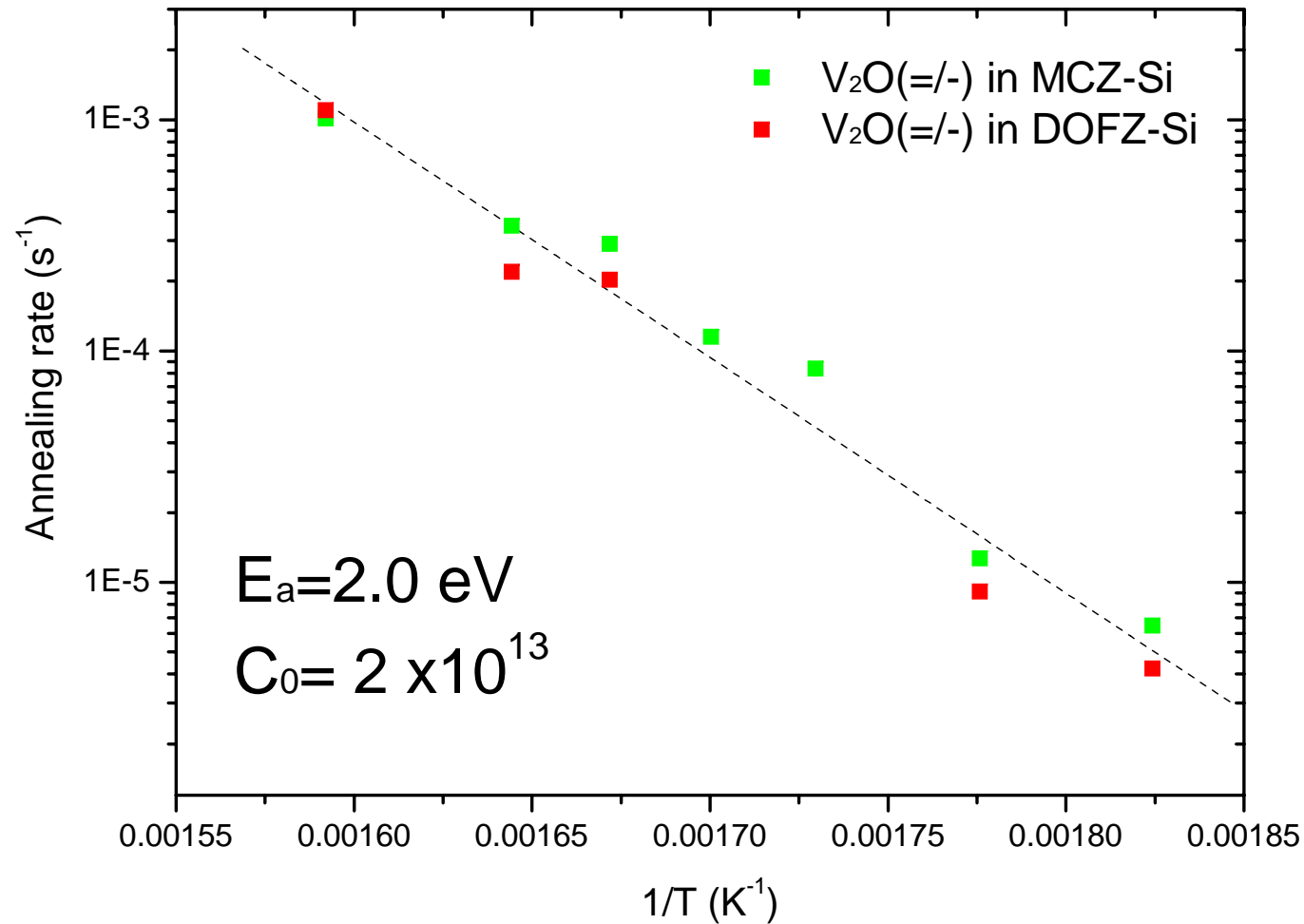


■ The annealing of  $V_2O(=/-)$  exhibits first order kinetics.

- For both MCZ and DOFZ.
- For all annealing temperatures in the range 275 - 355 °C.
- The kinetics for  $V_2O(-/0)$  is affected by overlapping level.

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# Results: $V_2O$ annealing



# The $V_2O$ annealing mechanism



- It is improbable that  $V_2O$  anneals by migration and trapping.
  - Since  $O_i$  is the main impurity, one would expect  $V_2O_2$  to form.
  - $V_2O_2$  is believed to be electrically active with two acceptor centers similar to  $V_2$  and  $V_2O$ .
- $V_2O$  does not anneal by interacting with migrating oxygen dimers.
  - The diffusivity of  $O_2$  is too low by a factor  $10^4$ .

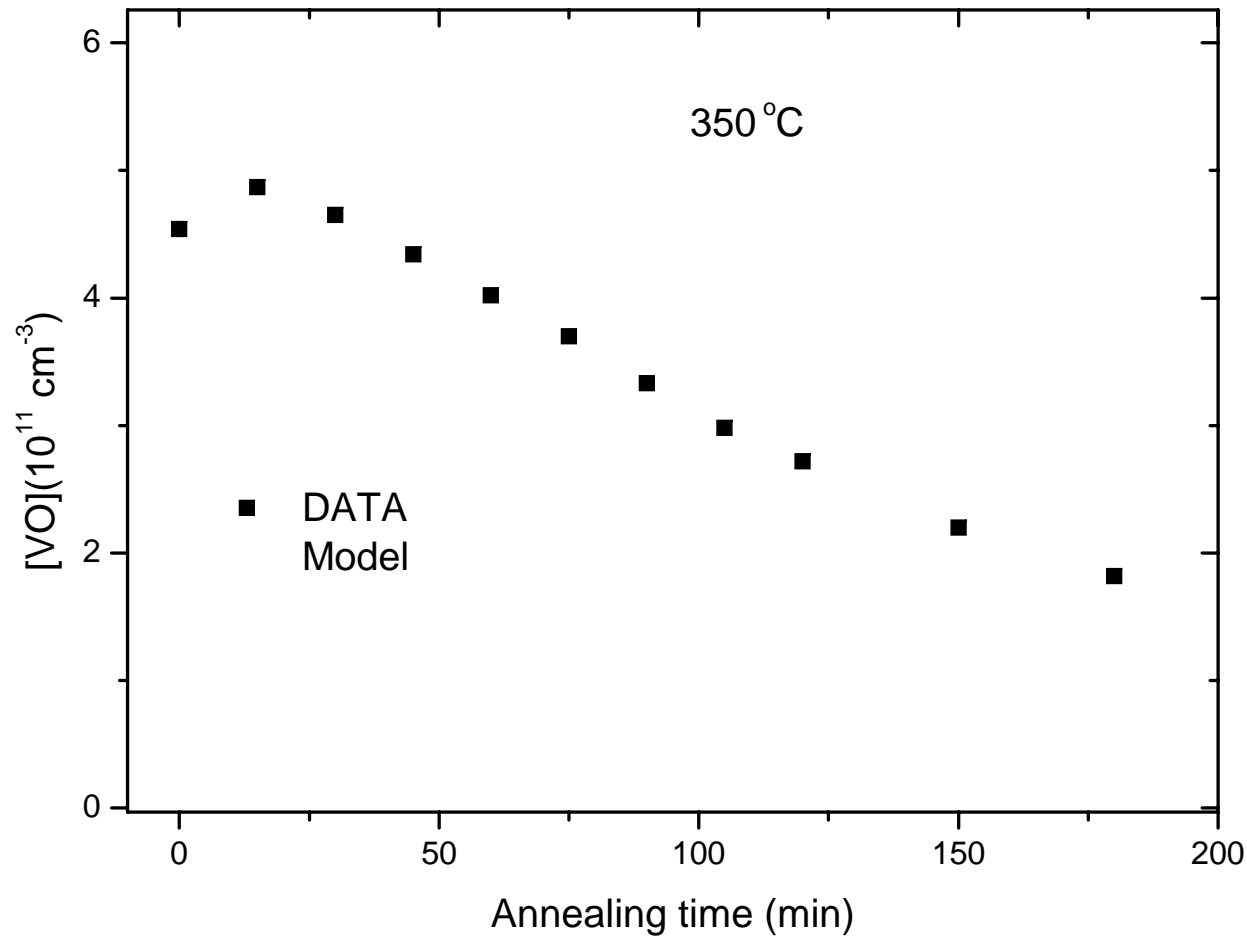
# The $V_2O$ annealing mechanism



- A prefactor,  $c_0$ , in the  $10^{12} - 10^{13}$  range is expected for dissociation.
  - The experimental value is in this range taking in regard the uncertainty.
  - If dissociation were to occur we would expect an increase in VO;  
 $V_2O \rightarrow V + VO$

.... So we must investigate the VO annealing, to check if it really is dissociation.

# Results: VO annealing (DOFZ)



- After the initial increase, [VO] decreases with 1st. order kinetics.

# A model for defect annealing



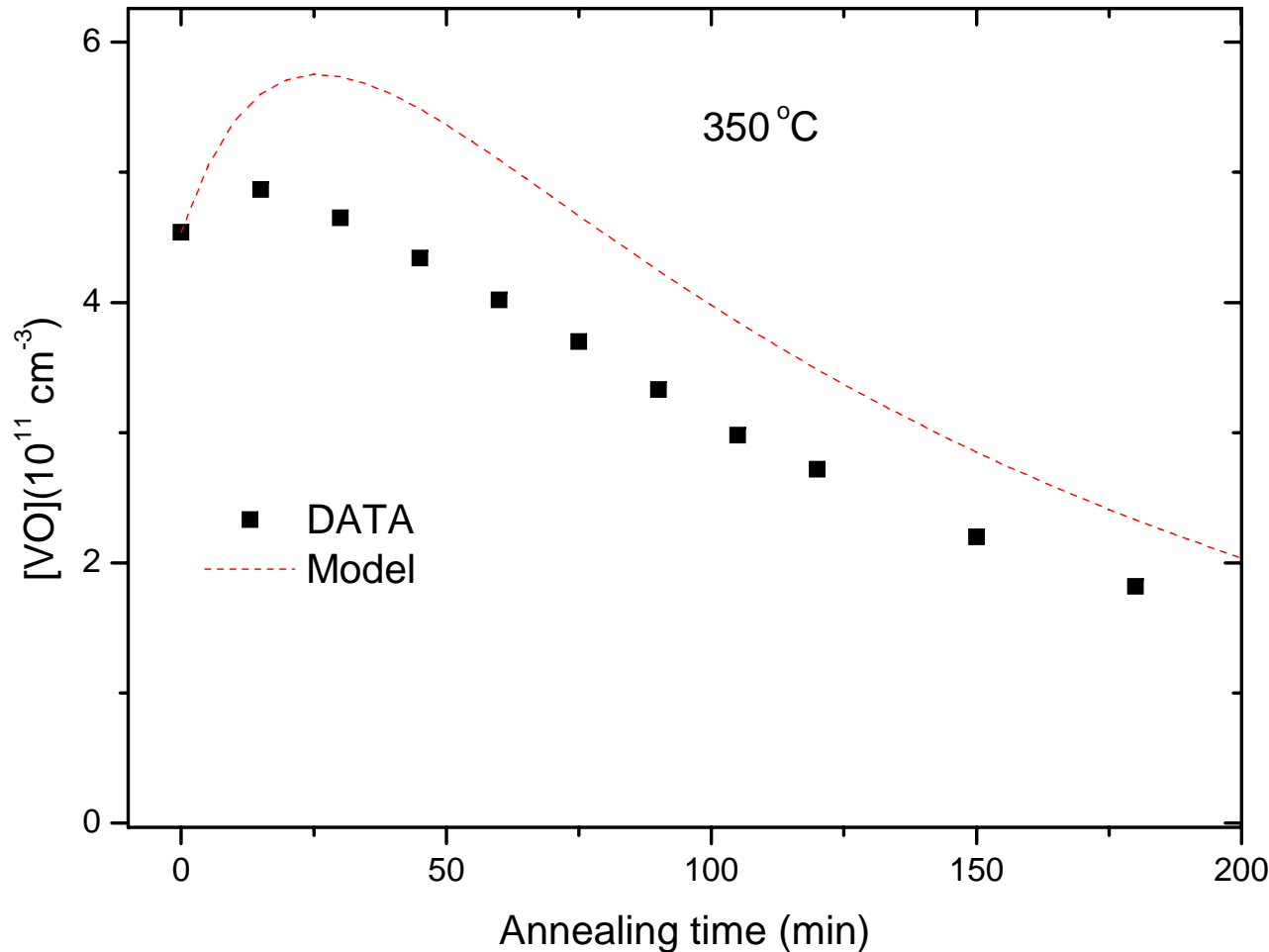
## ■ The reactions

- $V_2O \rightarrow VO + V$  (dissociation)
- $V + O_i \rightarrow VO$  (migration of V)
- $VO + O_i \rightarrow VO_2$  (migration of VO)
- $VO \rightarrow V + O_i$  (dissociation)
- $VO + V \rightarrow V_2O$  (migration of V)

## ■ The diff-equations

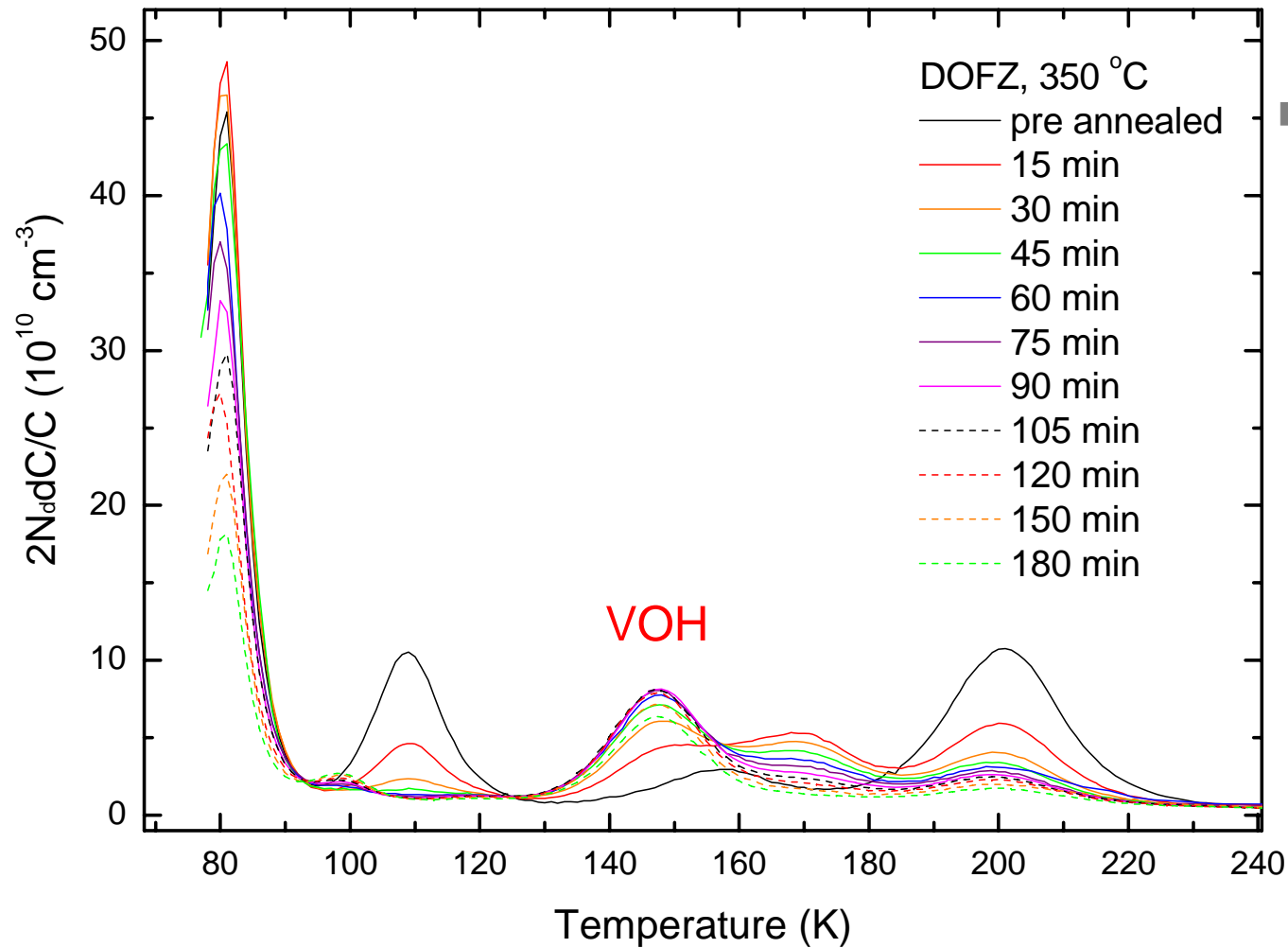
- $d[VO]/dt = - C_{\text{diss\_V2O}}(T) [V_2O] + 4\pi R D_V(T) [V] [VO]$
- $d[VO]/dt = C_{\text{diss\_V2O}}(T) [V_2O] + 4\pi R D_V(T) [V] [O_i] - 4\pi R D_{VO}(T) [VO] [O_i] - C_{\text{diss\_VO}}(T) [VO] - 4\pi R D_V(T) [V] [VO]$
- $d[V]/dt = C_{\text{diss\_V2O}}(T) [V_2O] + C_{\text{diss\_VO}}(T) [VO] - 4\pi R D_V(T) [V] [O_i] - 4\pi R D_V(T) [V] [VO]$

# Results: VO annealing (DOFZ)



- The model is highly sensitive to the  $[O_i]$  concentration. We used  $[O_i] = 2.5 \times 10^{17}$ .

# Results



■ Hydrogen may interact with VO.



# Summary



- The  $V_2O$  defect anneals with an activation energy,  $E_a$ , of  $\sim 2.0$  eV and a prefactor  $\sim 2 \times 10^{13} \text{ s}^{-1}$ .
- Dissociation is the likely annealing mechanism for  $V_2O$ .
- A model that includes VO and  $V_2O$  dissociation and VO migration and trapping describes the observed annealing kinetics well.