# Setting up a $^{14}C$ dating method with solid targets of bulk sediments at the Dating and Ionizing Radiation Laboratory

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- Objetives
- Introduction
- 4 Measurement techniques of  $^{14}C$ .
- CARBONUS facility
- 6 Preparation
- Set-up and measurement
- Results
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# **Objetives**

- $\bullet$  Setting up a  $^{14}C$  dating method for organic matter in marine sediments using solid targetss
  - Prepare
  - Dating
  - Comparison

- Optimisation of the method
  - Obtain replicable results
  - Reduce the background

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Source:

$$n +_{7}^{14} N \rightarrow_{6}^{14} C + p$$

• *C* natural isotopes:

ls	otope	Protons	Neutrons	Proportion (%)	$T_{1/2}$
	$^{12}C$	6	6	99	Estable
	$^{13}C$	6	7	1	Estable
	$^{14}C$	6	8	$10^{-10}$	5700(30) años

Table: Natural abundance of different carbon isotopes

# Carbon cycle

ullet This carbon  $^{14}C$  is distributed throughout the earth's reservoirs via the carbon cycle, but it takes time to reach the deep waters: Age of the reservoir.

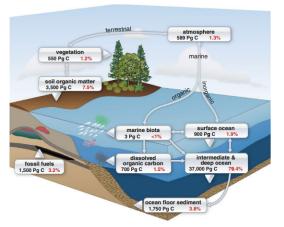


Figure: Carbon cycle

# $^{14}C$ assimilation

 $\bullet\ ^{14}C$  is assimilated by living organisms through photosynthesis, plant consumption and animal consumption.

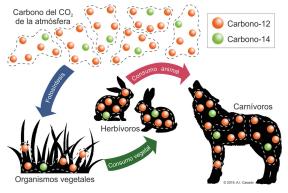


Figure: Incorporation of  $^{14}C$  into living things.

# Age calculation

- Developed by Willard Libby in 1950 for which he won the Nobel Prize in Chemistry in 1960.
- Living things stop exchanging  $^{14}C$  when they die, which allows us to use it for dating:  $A=A_0e^{-\lambda t}\to T=-\frac{1}{\lambda}\cdot \ln\frac{A}{A_0}$

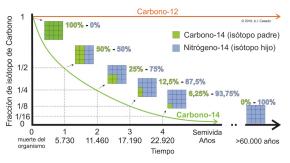


Figure:  $^{14}C$  decay.

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# $\Delta^{14}C$ over time

• The concentration of  $^{14}C$  has varied over time:

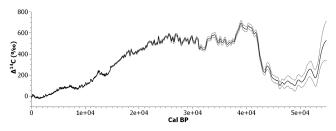


Figure: Variation of  $^{14}C$  with time. CAL BP means years before 1950.

$$\Delta^{14}C(\%) = \left({}^{14}C/{}^{12}C \cdot e^{\lambda(1950 - Y_C)} - 1\right) \cdot 1000$$

# $\Delta^{14}C$ over time

Anthropogenic effects:

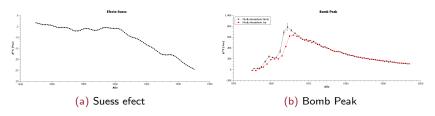


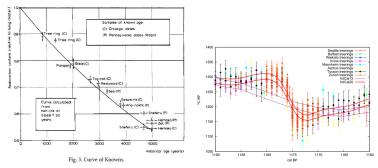
Figure: Variation of  $^{14}C$  due to human activity

- $\bullet$  The fraction of  $^{14}C$  is defined as:  $F^{14}C=\frac{A_f}{A_0}=\frac{^{14}C_f/^{12}C}{^{14}C_0/^{12}C}$
- The year 1950 is 0 B.P:  $F^{14}C\ (1950)\ = 1$
- $^{14}C$  Conventional age  $^{14}C$ :  $T=-\frac{1}{\lambda}ln$   $\left(F^{14}C\right)$



#### Calibration curves

- The  $^{14}C$  content of samples of known ages is measured.
- A calibration curve shows the calendar age on the x-axis and the conventional radiocarbon age on the y-axis.



(a) The first calibration curve reported (b) Modern atmospheric calibration by Willard Libby in 1950.

curve (IntCal20)

Figure:  $^{14}C$  calibration curves

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# Radiometric techniques and AMS

 There are two techniques: Radiometric (A) and Accelerator Mass Spectrometry (N):

$$A = \lambda N$$

• Accelerator mass spectrometry (AMS) is capable of obtaining good statistics in one hour, whereas radiometric techniques take days or even weeks.

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# **AMS**



Figure: Photo of the AMS MICADAS at the University of Salamanca

# $^{14}C$ sample preparation laboratory



Figure:  $^{14}C$  sample preparation laboratory.



Figure: EA and AGE.

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# Samples

- Background: without <sup>14</sup>C. We will use PhA.
- Standard: ratio  $^{14}C/^{12}C$  known. We will use OxII, whose  $F^{14}C\ (OxII)=1.34057(45).$

 Marine sediments: known age (Iberian Margin) and unknown age (Antarctica).

 Bone: Bone of an ovicaprine from the Roman Villa of Almenara-Puras in Valladolid.

# Preparation: factors to consider

• Different samples, different preparations (wood, sediment, textile, bone...)

• Origin of the carbon to be dated.

ullet C% sample: We need 1 mg of final carbon.

# Pretreatment of marine sediments: Fumigation

- Marine sediments are formed by: Eroded rocks, minerals, and inorganic (foraminifera) and organic remains of living things (foraminifera and other organisms).
- We want to date the organic matter.
- The objective of fumigation is to isolate organic carbon by removing inorganic carbon.
- A chemical attack is performed that dissociates the inorganic carbon, which combines with oxygen to form gaseous  $CO_2$ .
- ullet Fumigation: 4 days of HCl + 4 days of NaOH at  $60^{\circ}C$





Figure: Fumigation method

# Combustion and Graphitization

• The fumigated samples are combusted in an elemental analyzer to obtain carbon from the samples in the form of  $CO_2$ . Subsequently, they are reduced to carbon in the form of graphite according to  $CO_2 + 2H_2 \rightarrow C + 2H_2O$ .

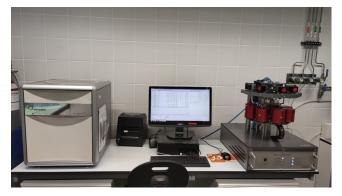


Figure: EA and AGE.

# Pressing

 The graphite is pressed into metal cathodes, which will be introduced into the AMS.







(a) Pneumatic machine

(b) Cathodes

(c) Cathodes on a magazine

Figure: Pressing of samples into cathodes and how they are introduced into the AMS

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#### Measurement

- The AMS must be tuned before each measurement.
- Background and standard samples are measured in each magazine
- Ten passes are performed on each sample, with 10 cycles of 30 seconds each per pass (each sample is measured for a total of 3000 seconds).

Table: Samples measured on each magazine.

Magazine	Measured samples			
221028LRI	Standard, Background, IM, Swiss Soil, Shale, TG-03 POWELL 2020			
221031LRI	Standard, Background, IM, Swiss Soil, Shale, TG-03 POWELL 2020			
221104LRI	Standard, Background, IM, Swiss Soil, Shale, TG-03POWELL 2020			
221109LRI	Standard, Background, TG-01, TG-02 POWELL 2020			
221110LRI	Standard, Background, TG-01, TG-02 POWELL 2020			
221111LRI	Standard, Background, TG-01, TG-02 POWELL 2020			
230203LRI	Standard, Background, Bone			
230208LRI	Standard, Background			
230210LRI	Standard, Background			

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# Results: AMS Background

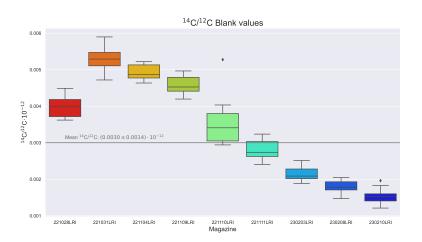


Figure: Evolution of background values over time

# Results: Standard values

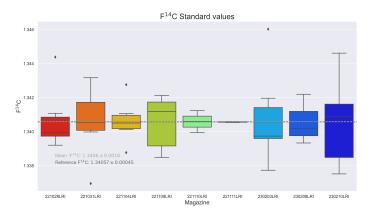


Figure: Evolution of the values of the standard samples over time



# Results: Iberian Margin JC089

 Adequate values have been replicated for marine sediment samples dated at the ETH.

Table: Mean values of  $F^{14}{\cal C}$  of the samples analyzed.

Muestra	Tipo	$F^{14}C_{ETH}$	$F^{14}C_{CARBONUS}$
Shale	Sediment	0,0180(10)	0,01751(15)
Swiss Soil	Sediment	1,0660(10)	1,05356(90)
JC089 10 cm	Sediment	0,7381(61)	0,7051(10)
JC089 70 cm	Sediment	0,3302(51)	0,33089(51)
JC089 110 cm	Sediment	0,2153(29)	0,21834(41)

#### Results: POWELL 2020

• Dating of the TG-01 core POWELL 2020:

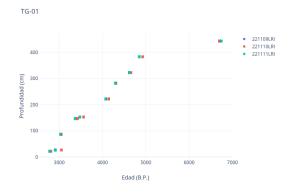


Figure: Results of the dating of core TG-01 from the POWELL 2020.

#### Results: POWELL 2020

Dating of the TG-02 core POWELL 2020:

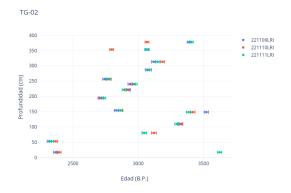


Figure: Results of the dating of core TG-02 from the POWELL 2020.

#### Results: POWELL 2020

• Dating of the TG-03 core POWELL 2020:

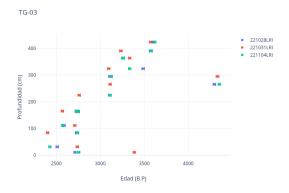


Figure: Results of the dating of core TG-03 from the POWELL 2020.

#### Results: Bone

• The dating results are consistent with the archaeological record:

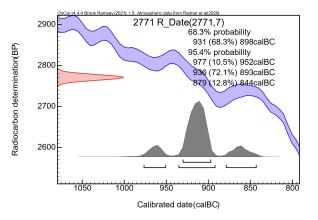


Figure: Calibration curve of the bone with a conventional mean age of  $2771(7)\ yr\ B.P.$ 

02/10/2023

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#### Conclusions

• It has been possible to replicate the appropriate values for standard samples while maintaining low background values.

• The fumigation method has been developed, obtaining comparable results.

• The first dating of samples of unknown age has been achieved.

#### **Future**

• Improve background values.

Repeat sediment measurements from the POWELL 2020 campaign.

• Perform further intercomparison of fumigated samples with the ETH.

Thank you very much for your attention!