

# Zinc and Cadmium: XPS Chemical State Determination Using Auger Parameters and Auger Peak Curve-Fitting Procedures

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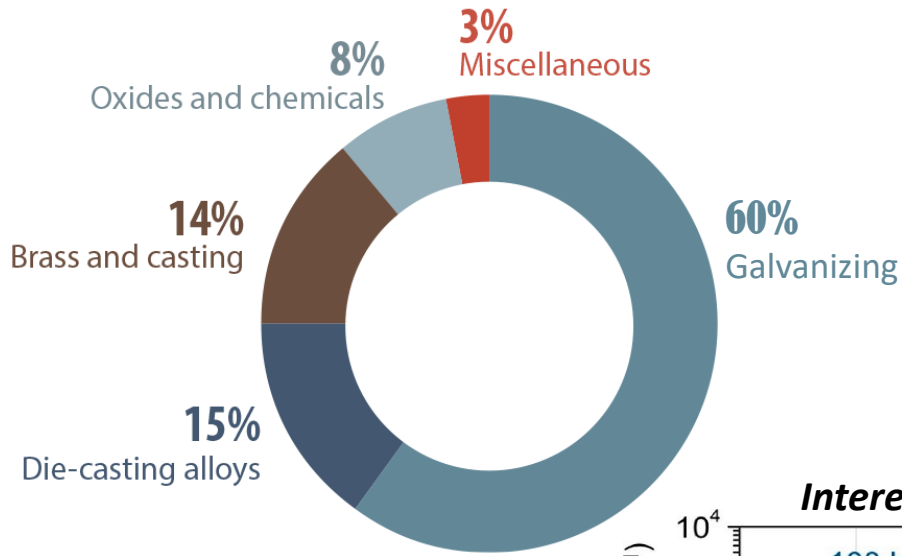
**May 27<sup>th</sup>, 2024**



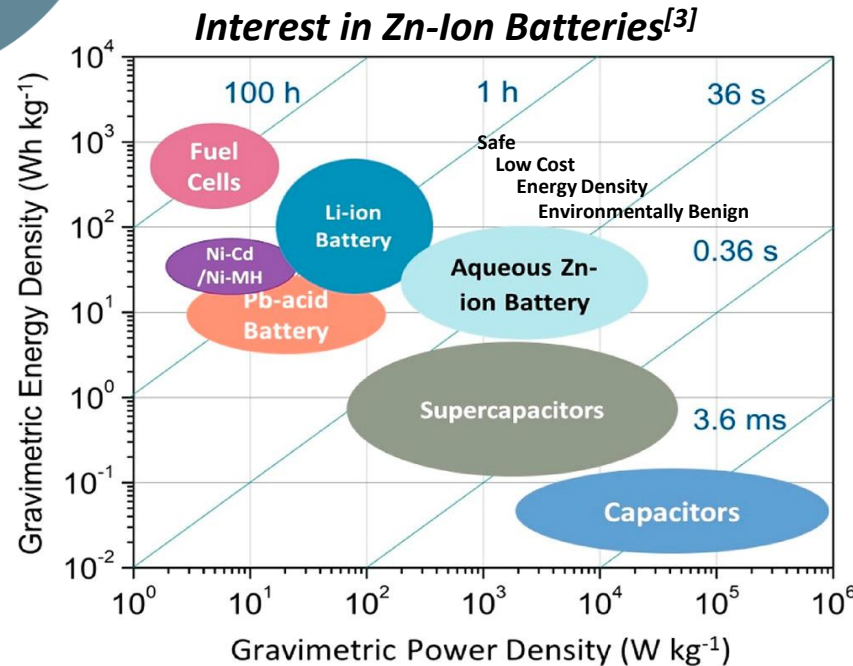
Canadian Association  
of Physicists



# Why Zinc and Cadmium?



**Global Uses of Zinc<sup>[1]</sup>**



## Zinc

- ~ **13 million metric tons** produced globally in 2022<sup>[1]</sup>
- **4<sup>th</sup> most produced metal** in the world<sup>[2]</sup>
- Market demand expected to **exceed production capabilities** in the coming years!

## Cadmium

- ~ **24 000 metric tons** produced globally in 2022<sup>[1]</sup>
- **>5% of the global solar cell market**<sup>[4]</sup>

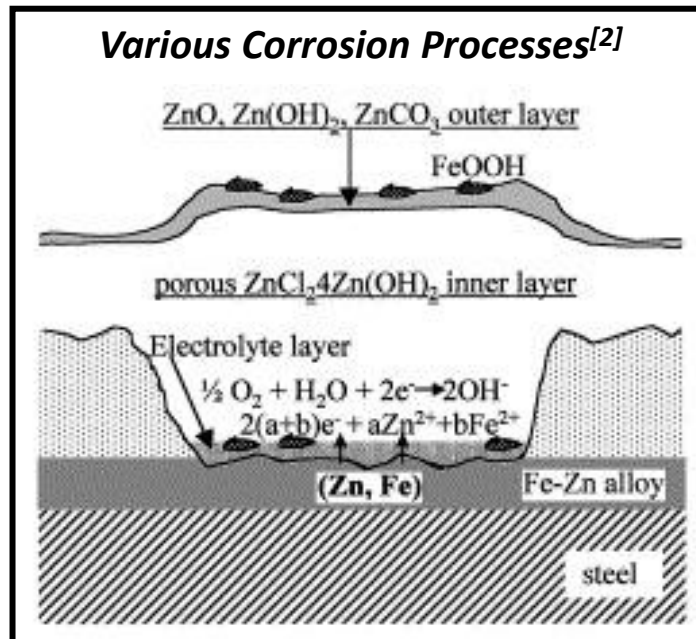
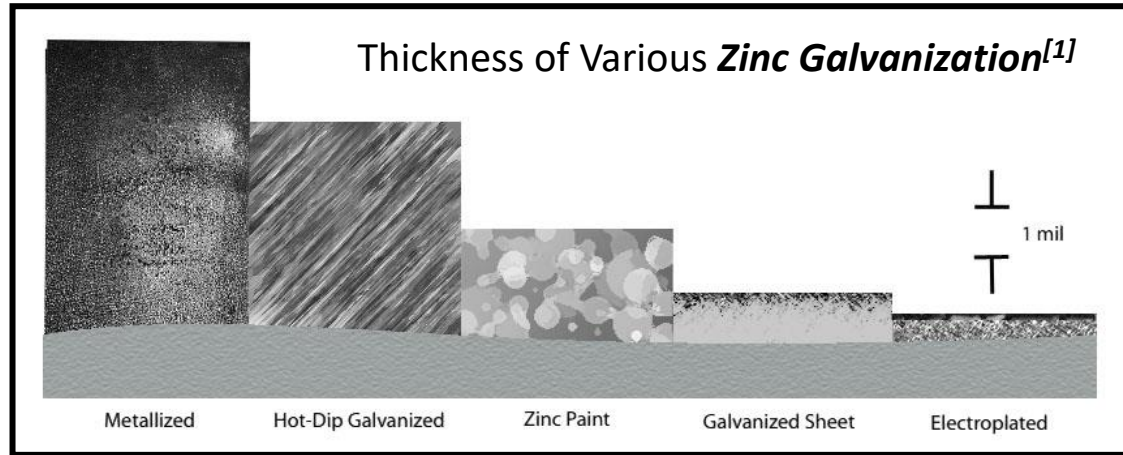
[1] Mineral Commodities Summaries 2023. United States Geological Survey <https://pubs.usgs.gov/periodicals/mcs2023/mcs2023.pdf>

[2] Zinc Statistics and Information | U.S. Geological Survey (2021). <https://www.usgs.gov/centers/national-minerals-information-center/zinc-statistics-and-information>.

[3] Jia, X. et al. Chem. Rev. 120 (15) 2020, 7795-7866.

[4] Solar Technologies Office – Cadmium Telluride. Office of Energy Efficiency and Renewable Energy. (2023) <https://www.energy.gov/eere/solar/cadmium-telluride>

# Zinc on Surfaces



## Forms of Zinc on Surfaces

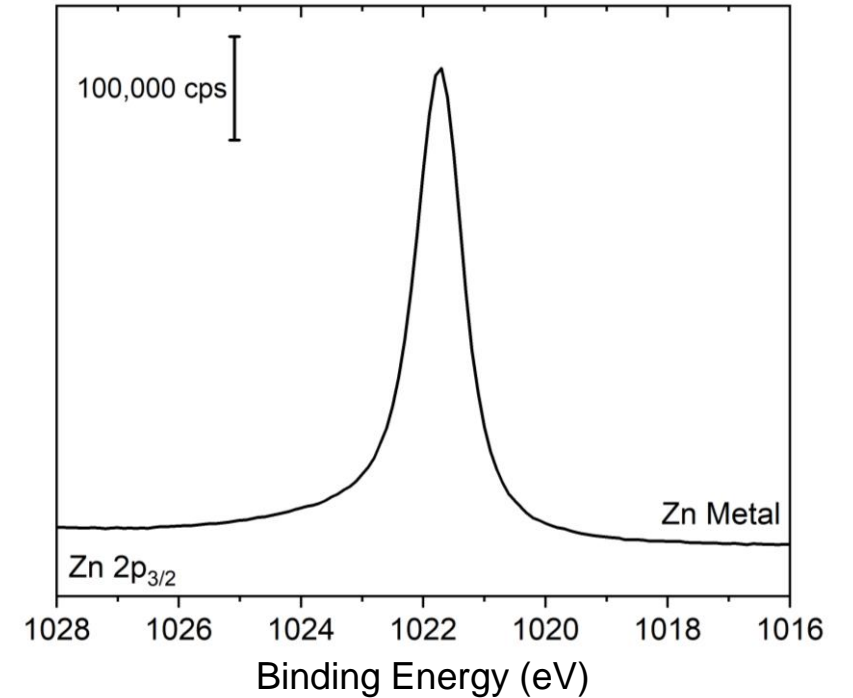
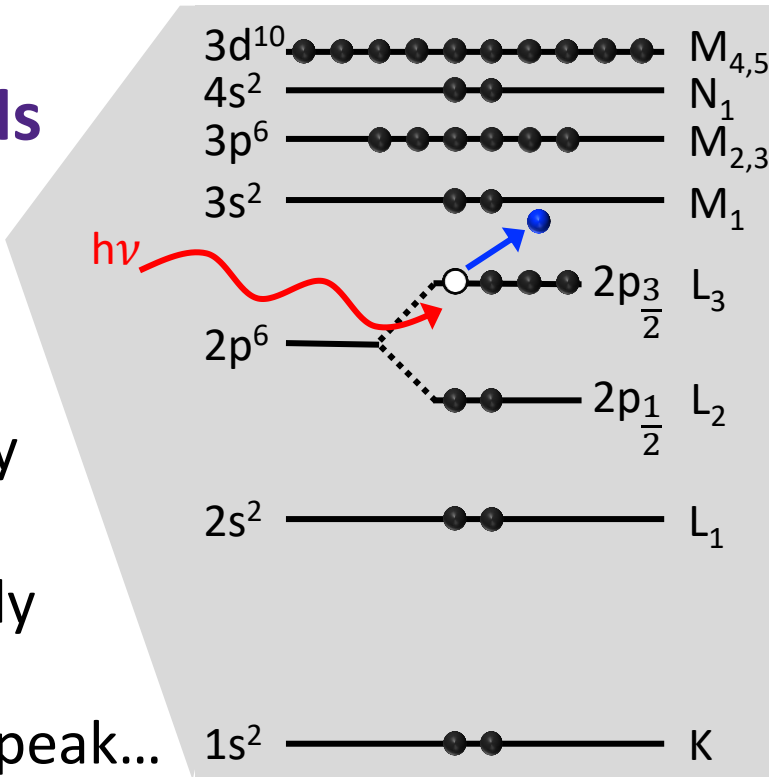
- Different forms of **zinc coatings**
- **Corrosion** by-products
- **Contaminants**
- **Zinc alloys**

**A similar story is true for Cadmium**

# Speciation by XPS

## Photoelectron Signals

- **One-electron** process
- **For many elements...** speciation achieved by consideration of the **photoelectron line** only
- Consider the Zn  $2p_{3/2}$  peak...



Sample	Zinc $2p_{3/2}$			#
	Binding Energy / eV			
Zn Metal	1021.7	±	0.3	15
ZnO	1021.8	±	0.5	13
Zn(OH) <sub>2</sub>	1022.4	±	0.5	4
ZnF <sub>2</sub>	1022.2	±	0.7	3
ZnS	1021.9	±	0.2	4
Zn <sub>5</sub> (CO <sub>3</sub> ) <sub>2</sub> (OH) <sub>6</sub>	1022.2	±	0.4	2

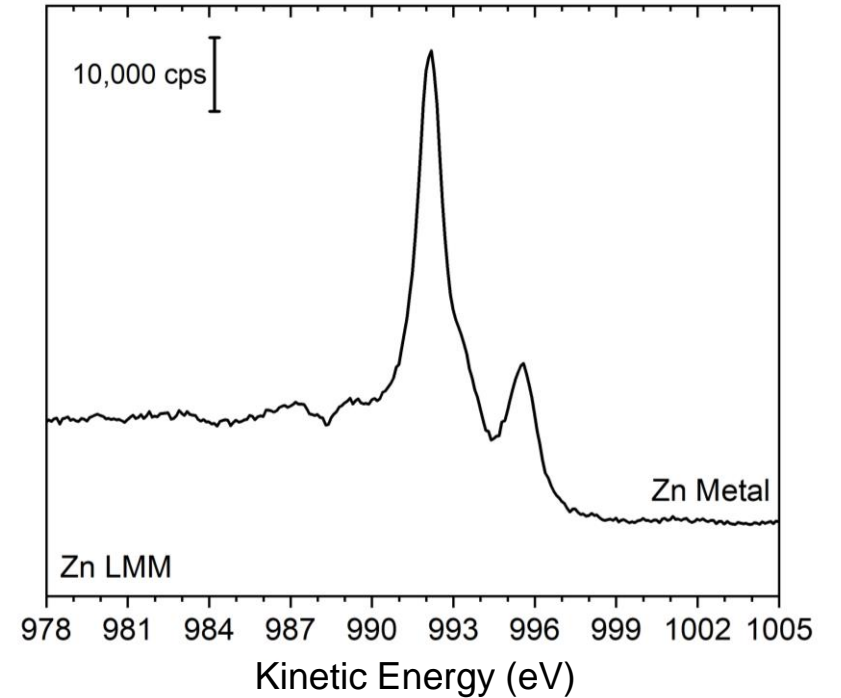
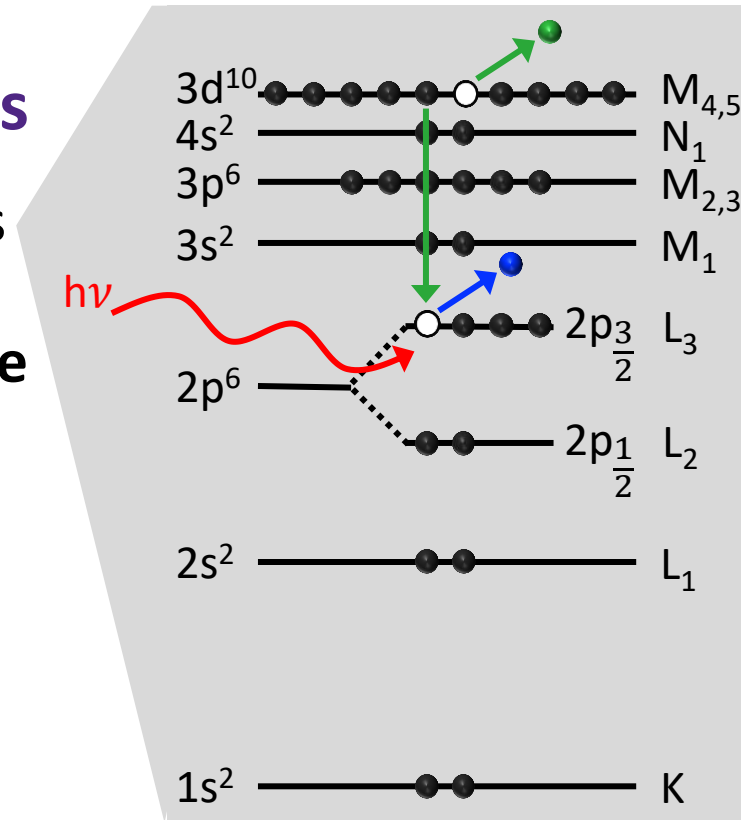
*Spread of only 0.7 eV (1021.7 – 1022.4 eV)*

Speciation made difficult due to overlap.  
Cannot use this peak alone to differentiate.

# Speciation by XPS

## Auger Electron Signals

- **Three-electron process**
- **Auger lines** have **unique shapes and positions**
- Consider the Zn LMM signal...



More information available when considering Auger signals!

Sample	Zinc L <sub>3</sub> M <sub>4,5</sub> M <sub>4,5</sub> Kinetic Energy / eV			#
Zn Metal	992.2	±	0.2	15
ZnO	988.4	±	0.6	13
Zn(OH) <sub>2</sub>	987.2	±	0.5	4
ZnF <sub>2</sub>	986.3	±	0.4	3
ZnS	989.7	±	0.4	4
Zn <sub>5</sub> (CO <sub>3</sub> ) <sub>2</sub> (OH) <sub>6</sub>	987.5	±	0.6	2

Spread of 5.9 eV (986.3 – 992.2 eV)

# Speciation by XPS

## Auger Parameter, $\alpha'$

- Originally proposed by Dr. Charles Wagner<sup>[1]</sup> and later modified

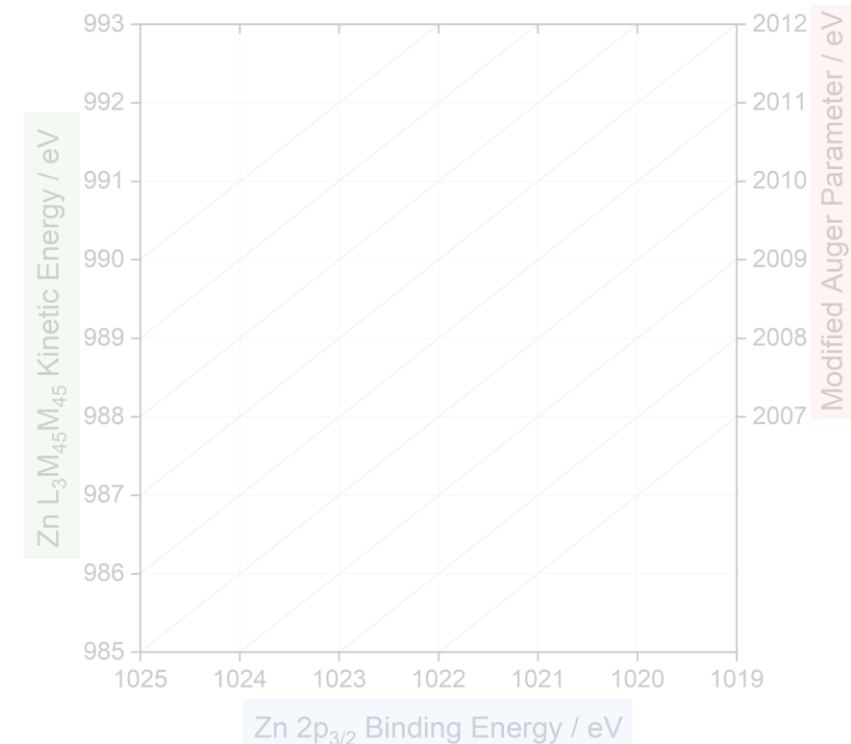
$$\alpha' = E_K(\text{Auger}) + E_B(\text{Photoelectron})$$

- Useful for chemical state analyses
- Avoids interference of surface charging (*i.e.*, same magnitude / opposite in direction)

Sample	Auger Parameter Kinetic Energy / eV			#
Zn Metal	2013.9	±	0.2	15
ZnO	2010.2	±	0.2	13
Zn(OH) <sub>2</sub>	2009.5	±	0.3	4
ZnF <sub>2</sub>	2008.4	±	1.1	3
ZnS	2011.6	±	0.3	4
Zn <sub>5</sub> (CO <sub>3</sub> ) <sub>2</sub> (OH) <sub>6</sub>	2009.7	±	0.1	2

## Wagner (Chemical State) Plots

- Highlights the  $E_K$  of Auger peak,  $E_B$  of photoelectron peak, and the Auger parameter in a compact format.
- Useful tool to understand trends for a series of related compounds

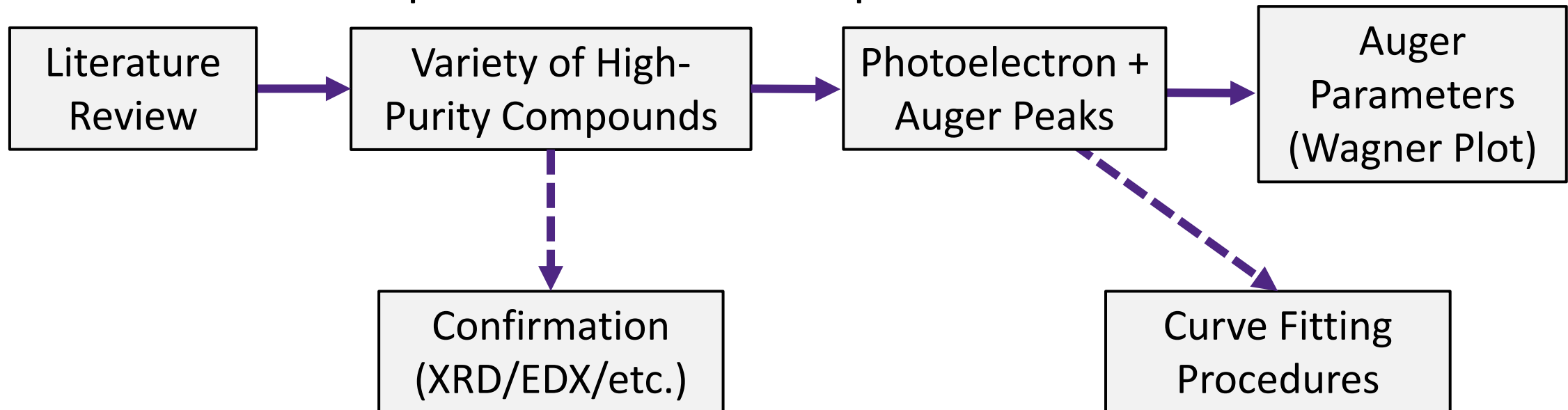


[1] Wagner, C.D., *Anal. Chem.* 44 (6) 1972, 967.

# Project Motivation

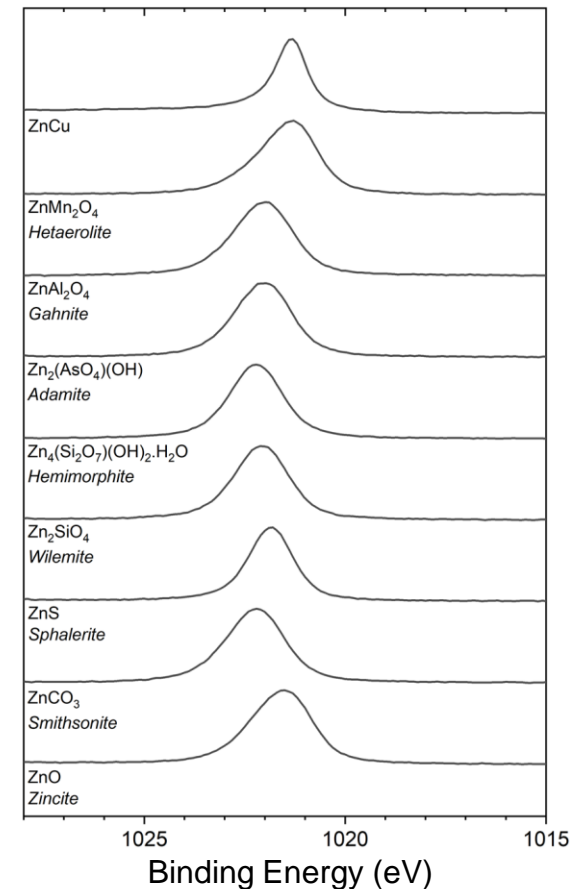
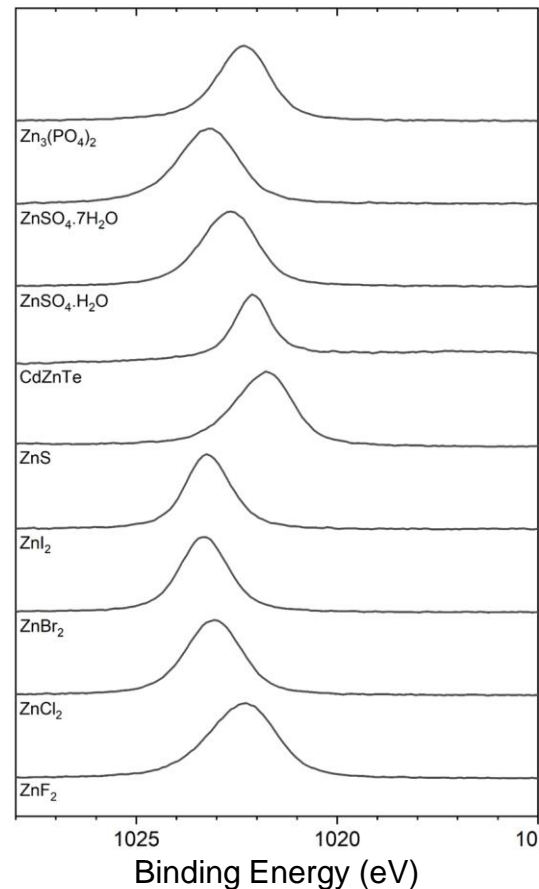
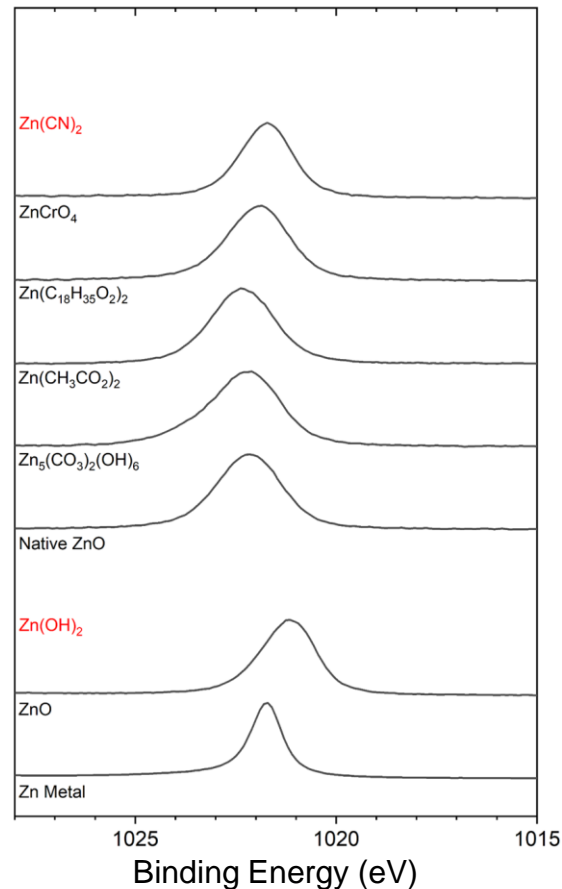
- Issues of reproducibility, consistency, and completeness in the literature
  - Only good data for **Zn metal** and **Zn oxide**

**Goal:** To provide **accurate, reproducible, and comprehensive** reference data to aid in the speciation of zinc compounds and minerals.



# Standard Samples – Zn 2p<sub>3/2</sub>

- 27 zinc compounds and 13 cadmium compounds considered
- Limited Information available from this signal alone



\* To be completed

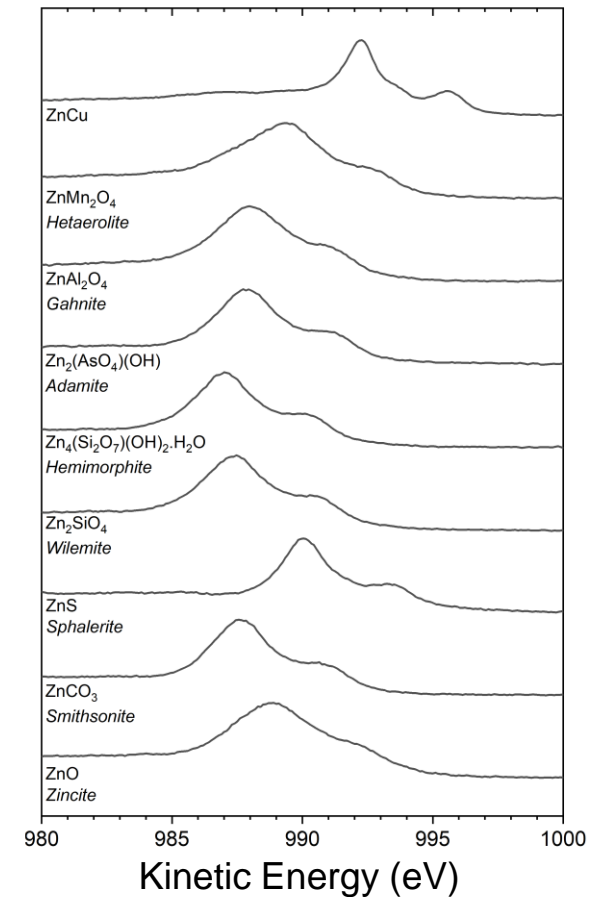
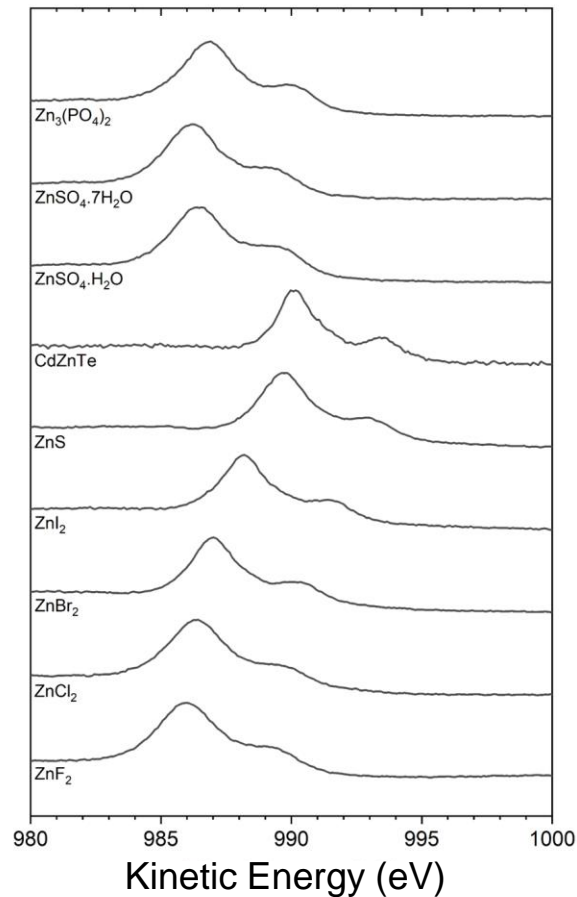
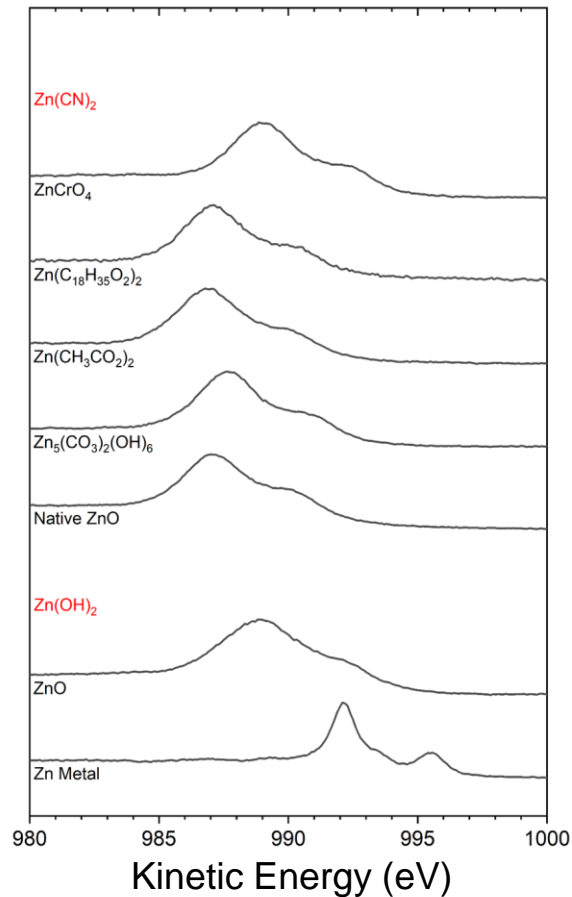


# Standard Samples – Zn LMM

- Larger amount of information available from this signal!

Wider energy range than  $2p_{3/2}$

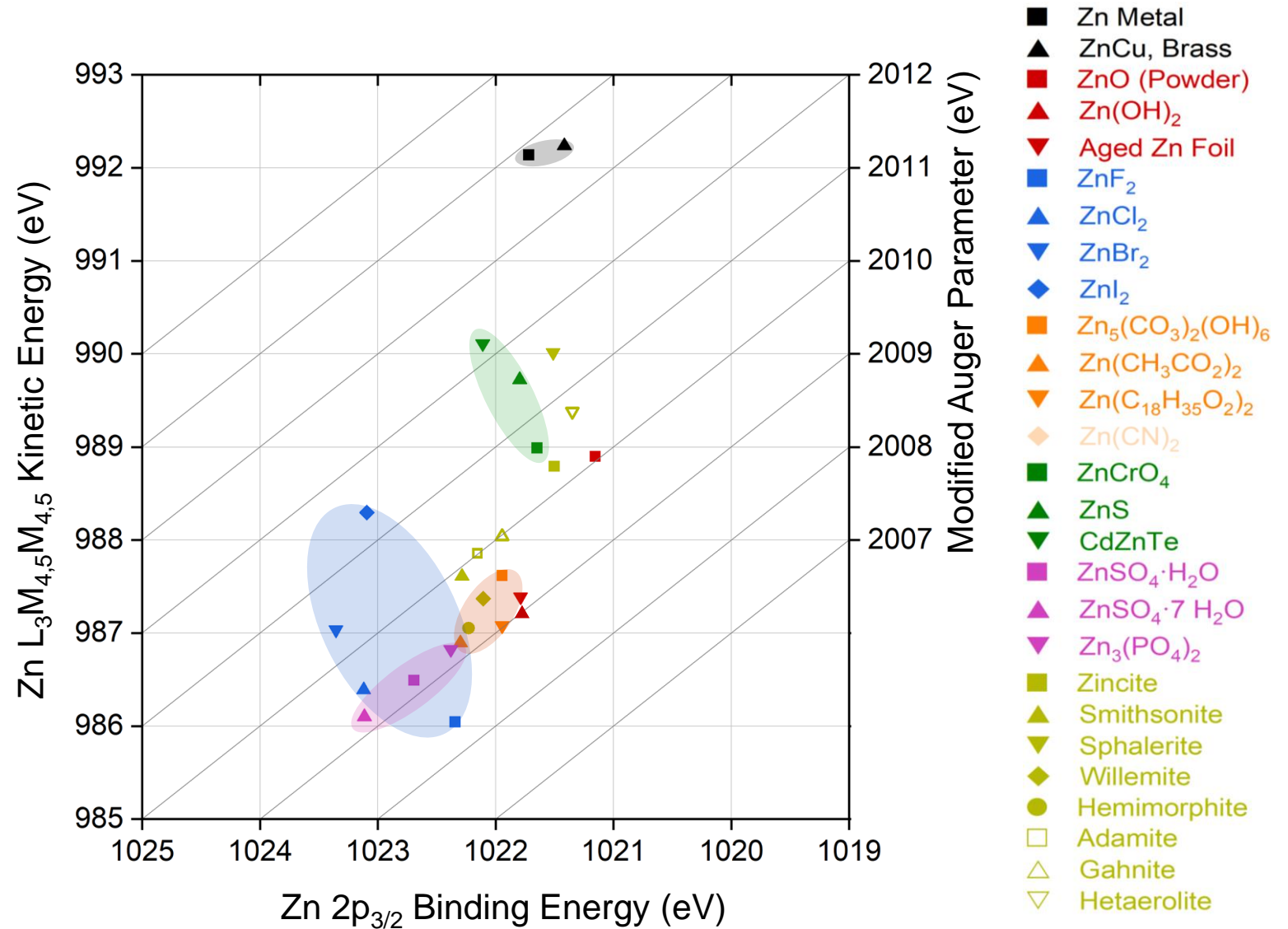
Characteristic shape → Deconvolution



\* To be completed

# Wagner Plot

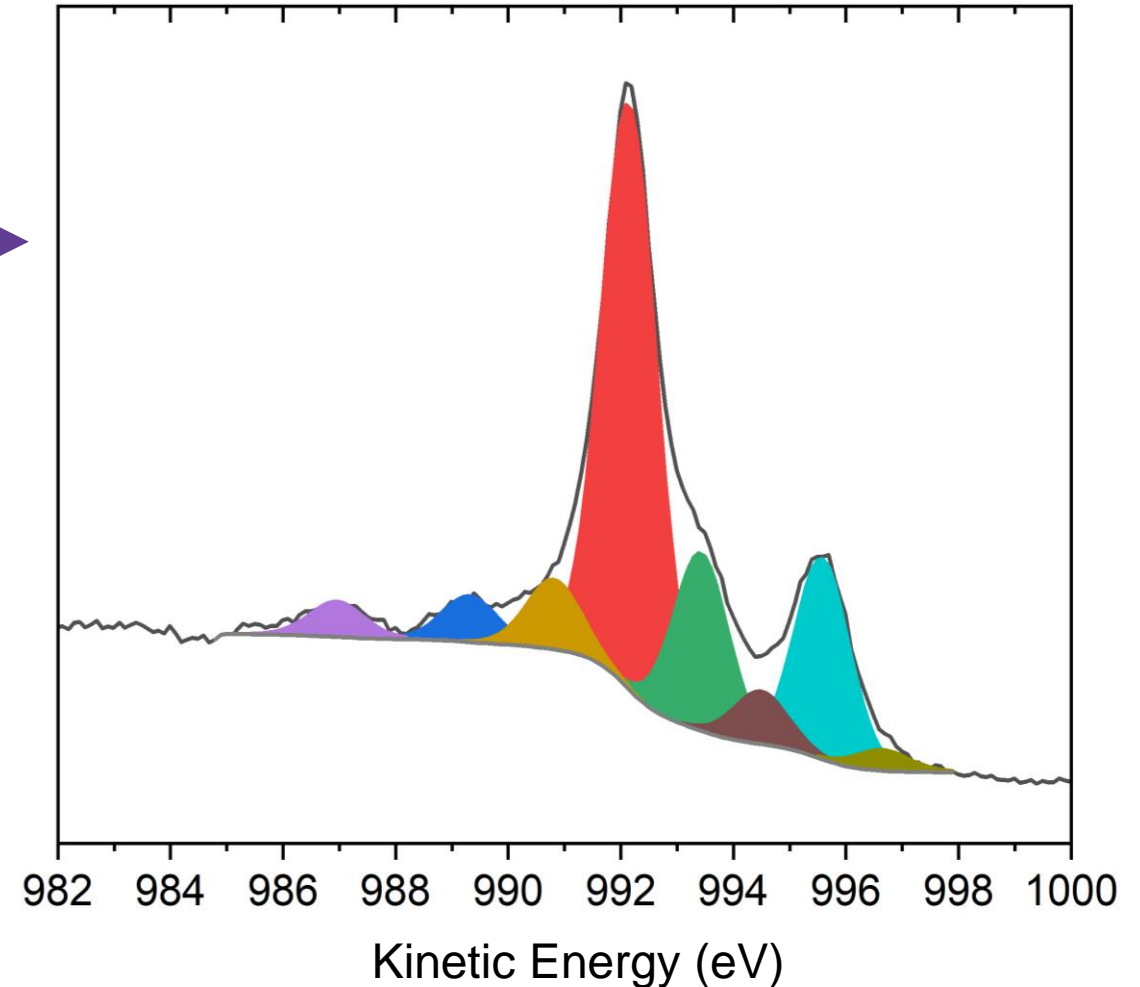
- Consider the zinc Wagner plot
- Each data point represents average over three **triplicate measurements**.
- Trends become easily observed



# Mixed Species Systems

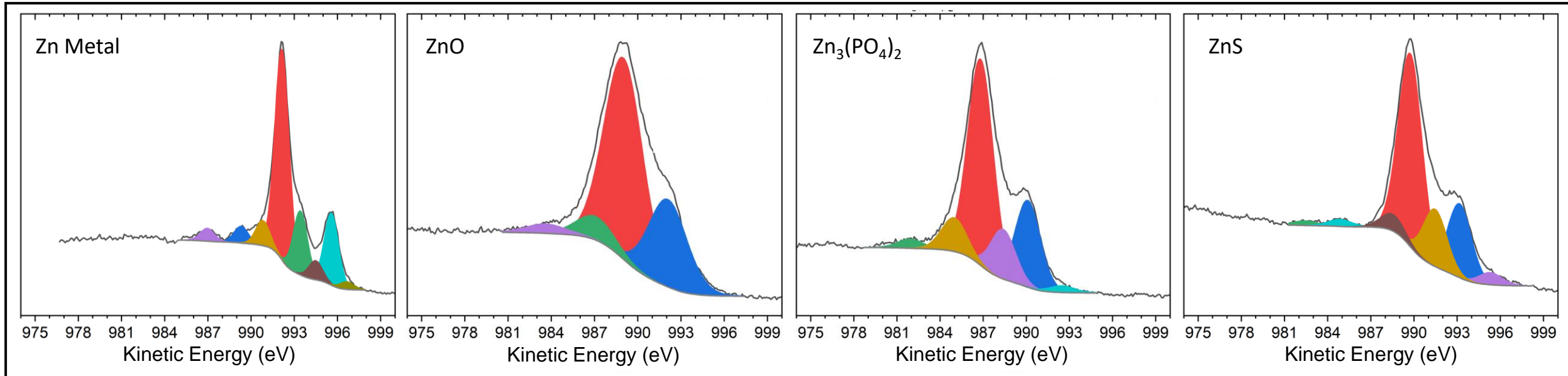
- Establishing method of curve fitting
- Consider the Zn  $L_3M_{4,5}M_{4,5}$  signal for **Metallic Zinc** →
- Series of individual peaks used to reproduce the  $L_3M_{4,5}M_{4,5}$  envelope
- Proper peak **constraints** must be defined in order to maintain the integrity of this characteristic shape.

**Constraints are key!**



# Curve Fitting Procedures

- Consider the  $L_3M_{4,5}M_{4,5}$  signal for a series of Zn species



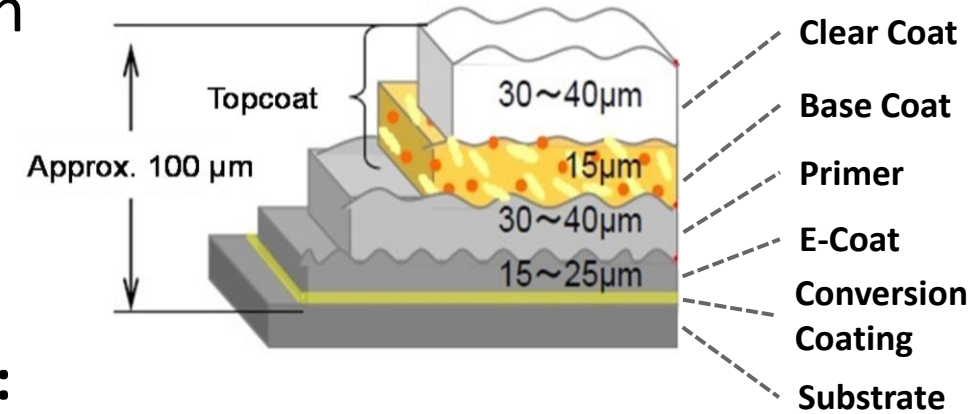
- Development of the necessary information to replicate LMM peak shapes
- Information from counter ions and stoichiometry must also be considered to increase confidence!

# Quantifying Changes in Surface Chemistry

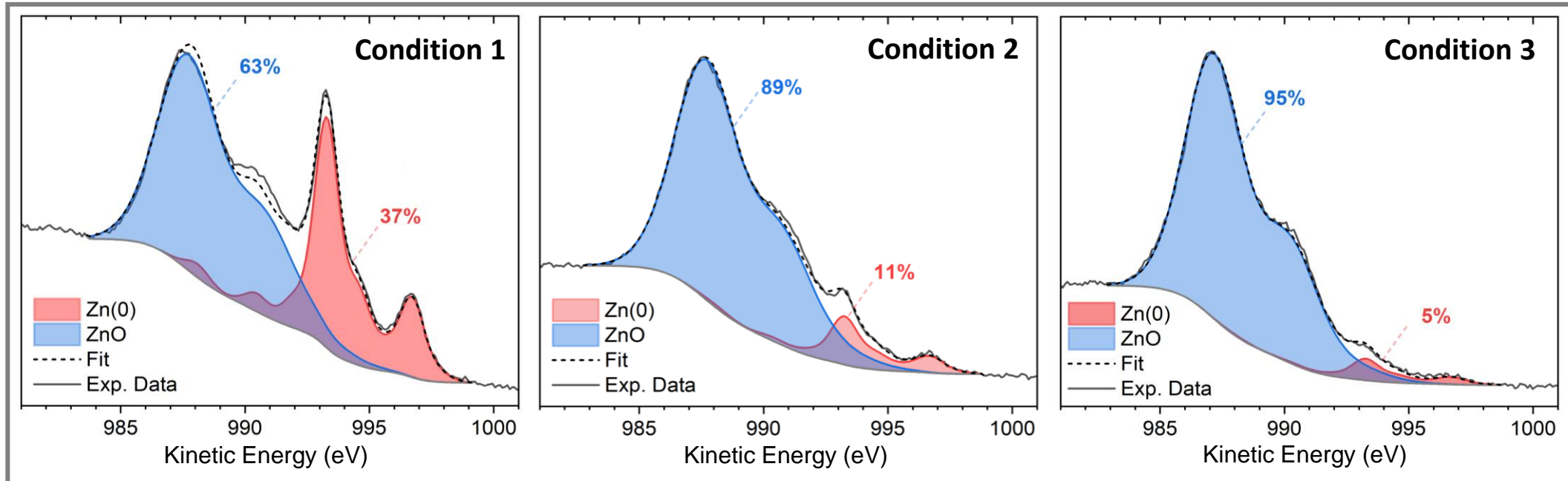
- Company experiencing **cohesive failure** between coating layers due to excess oxide growth

**Goal:**

*Modify conditioning stages to minimize oxide growth before paint and coating stages.*



- Consider the surface of **hot dipped galvanized steel**:



# Summary & Acknowledgments

- Accurate speciation of Zn and Cd with XPS should consider:
  - Binding Energy of **Zn 2p<sub>3/2</sub> / Cd 3d<sub>5/2</sub>**
  - Kinetic Energy of **Zn LMM / Cd MNN**
  - **Modified Auger Parameters**
  - Information from **Counter Ions**
  - Survey quantification and **stoichiometry**
- Mixed species systems can be quantified using careful **peak fitting procedures**

