



# **Penning effects in Xenon gas mixtures:**

## **survey of the homonuclear associative ionisations**

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# Measuring the transfer probabilities

- ❖ Townsend coefficient adjustment

$$G = \exp \int_{\text{tube}}^{\text{anode}} dr \alpha(E(r)) \frac{\sum v_i^{\text{ion}}(E(r)) + \sum r_i v_i^{\text{exc}}(E(r))}{\sum v_i^{\text{ion}}(E(r))}$$

- ❖  $r_i$  transfer probabilities: assuming  $\alpha$  proportional to the sum of  $v_{\text{ion}}$ ,
- ❖  $\alpha, v_i$  : gas properties (pressure, temperature ...)
- ❖ calculated by Magboltz [S.F. Biagi, *NIM A* **421** (1999) 234–240.]

## Gain calibration

- ❖ uncertainty on the absolute gain,
- ❖ work function,
- ❖ calibration of the equipment.

$$G := g G$$

## Photon feedback

- ❖ secondary avalanches,
- ❖ at high gain,
- ❖ almost uncorrelated, free parameter.

$$G := G / (1 - \beta G)$$

# Energy transfer processes

❖ The following can happen for an  $A^*$ :

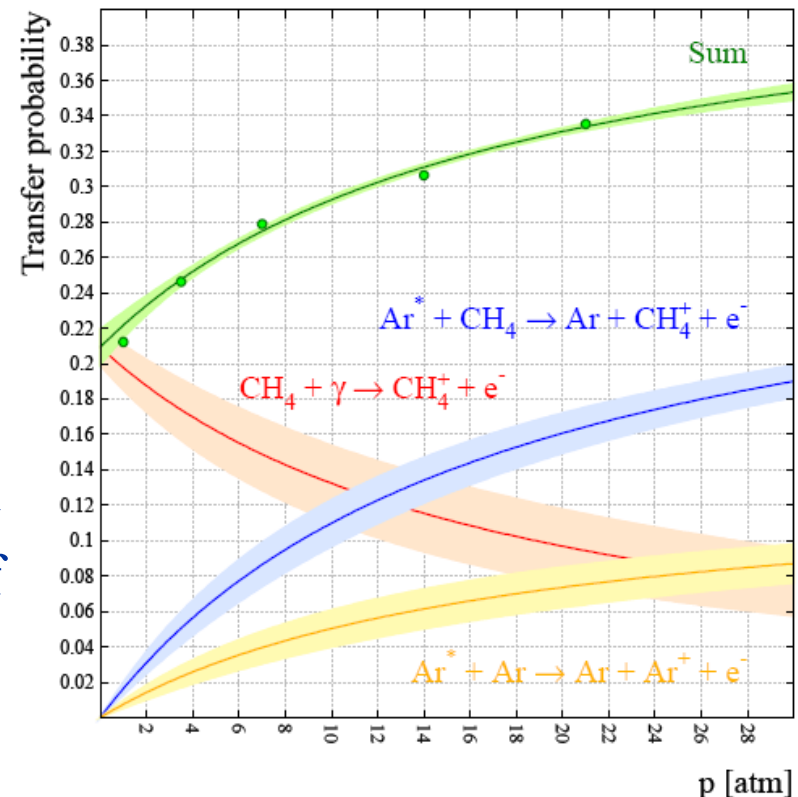
- ❖  $A^* + B \rightarrow A + B^+ + e^-$  : collisional ionisation,
- ❖  $A^* + A \rightarrow A_2^+ + e^-$  : homonuclear associative ionisation,
- ❖  $A^* \rightarrow A + \gamma$  : radiative decay

$$r = \frac{pc \frac{f_{B^+}}{\tau_{A^*B}} + p(1-c) \frac{f_{A^+}}{\tau_{A^*A}} + \frac{f_{rad}}{\tau_{A^*}}}{pc \frac{f_{B^+} + f_{B^-}}{\tau_{A^*B}} + p(1-c) \frac{f_{A^+} + f_{A^-}}{\tau_{A^*A}} + \frac{1}{\tau_{A^*}}}$$

$A^*-B$        $A^*-A$        $A^*-\gamma$

❖ Separations of the processes with pressure and concentration dependence of the transfer rates.

❖  $\lim p \rightarrow 0, r \rightarrow$  radiative transfer



# Experimental Gain Data

## ❖ Single wire proportional counters:

1- Xe – CH<sub>4</sub>

2- Xe – CO<sub>2</sub>

3- Xe – Ar

4- Xe – Ar – CH<sub>4</sub>

5- Xe – C<sub>2</sub>H<sub>2</sub>

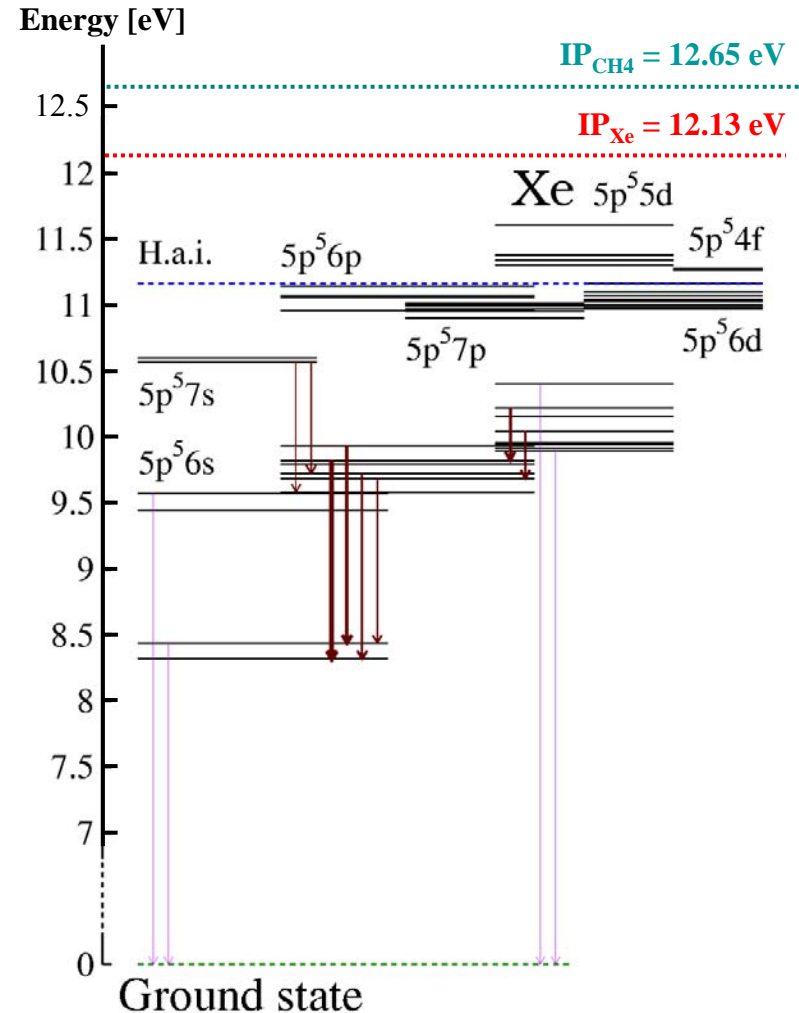
6- Xe – iC<sub>4</sub>H<sub>10</sub>

7- Xe – iC<sub>2</sub>H<sub>4</sub>

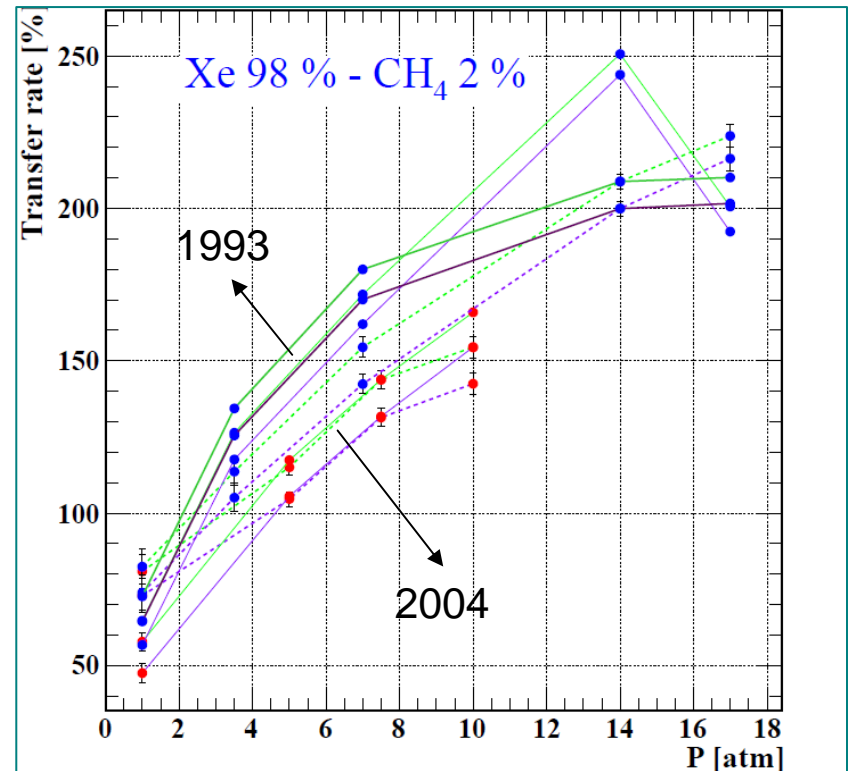
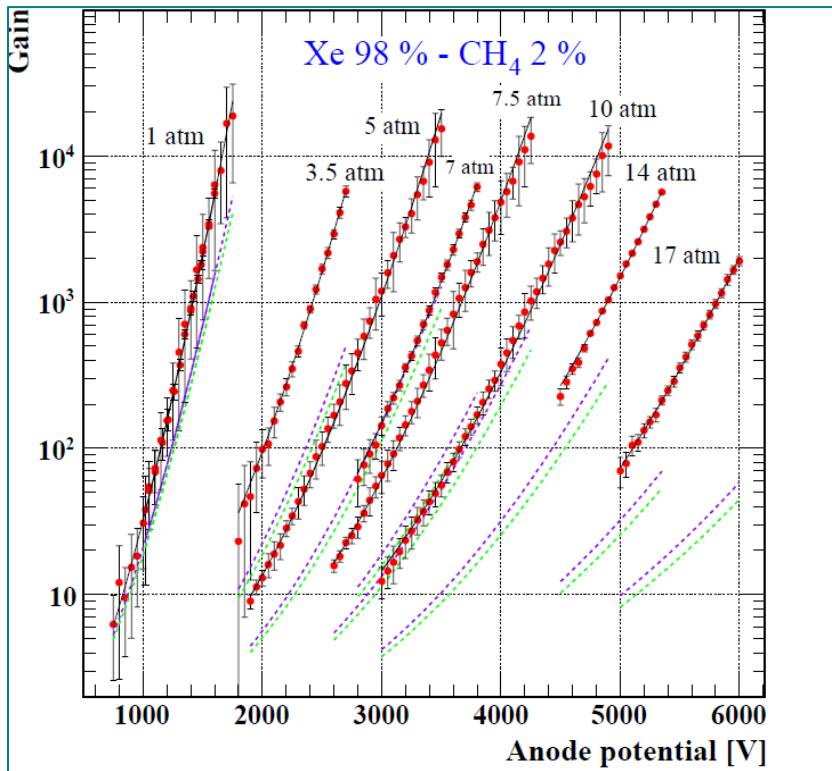
- ❖ R.W. Hendricks,  
*Nucl. Instr. And Meth. A* **102** (1972) 309–312.
- ❖ Z. Ye et al.,  
*Nucl. Instr. and Meth. A* **329** (1993) 140–150.
- ❖ R.K. Sood et al.,  
*Nucl. Instr. and Meth. A* **344** (1994) 384–393.
- ❖ D.J. Grey et al.,  
*Nucl. Instr. and Meth. A* **527** (2004) 493–511.
- ❖ R.K. Manchanda et al.,  
*Nucl. Instr. and Meth. A* **595** (2008) 605–615.

# Xenon – Methane Mixtures

- ❖  $\text{Xe}^* + \text{CH}_4 \rightarrow \text{Xe} + \text{CH}_4^+ + \text{e}^-$   
NOT possible process energetically
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + \text{e}^-$  can happen  
 $\varepsilon > 11.162 \text{ eV}$   $\text{Xe}^*$  states can transfer



# Xenon – Methane Mixtures



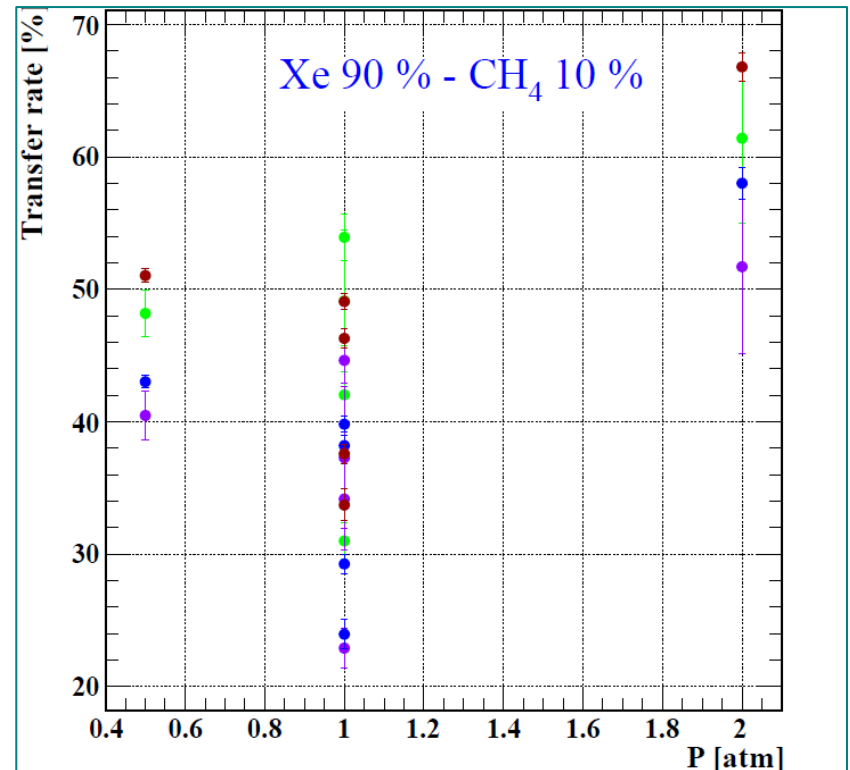
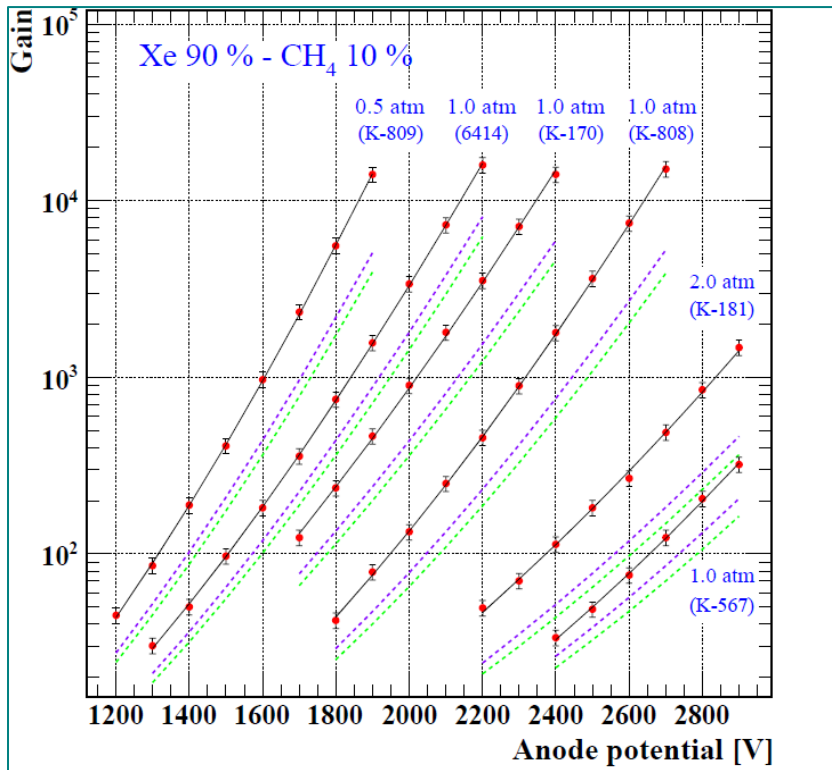
- ❖  $r_a = 12.5 \mu\text{m}$ ,  $r_c = 1.5 \text{ cm}$ ,
- ❖ No feedback

- ❖ Magboltz 8.9.3:
  - ❖  $g_{w1} = 1.02 - 1.10$ ,  $g_{w2} = 0.89$
- ❖ Magboltz 8.9.1:
  - ❖  $g_{w1} = 1.05 - 1.10$ ,  $g_{w2} = 0.91$

[Z. Ye et al., (1993), D.J. Grey et al., (2004)]

- ❖ Difference on transfer rates  $\approx 12 \%$

# Xenon – Methane Mixtures



❖ No feedback

Ser. No	$r_c$ [cm]	$r_a$ [ $\mu\text{m}$ ]
K-809	1.27	25.4
6414	1.27	25.4
K-170	2.54	25.4
K-808	1.27	50.8
K-181	2.54	25.4
K-567	2.54	76.2

[R.W. Hendricks, 1972]

❖ Magboltz 8.9.3:  $g_w = 0.973$

❖ Magboltz 8.9.1:  $g_w = 0.996$

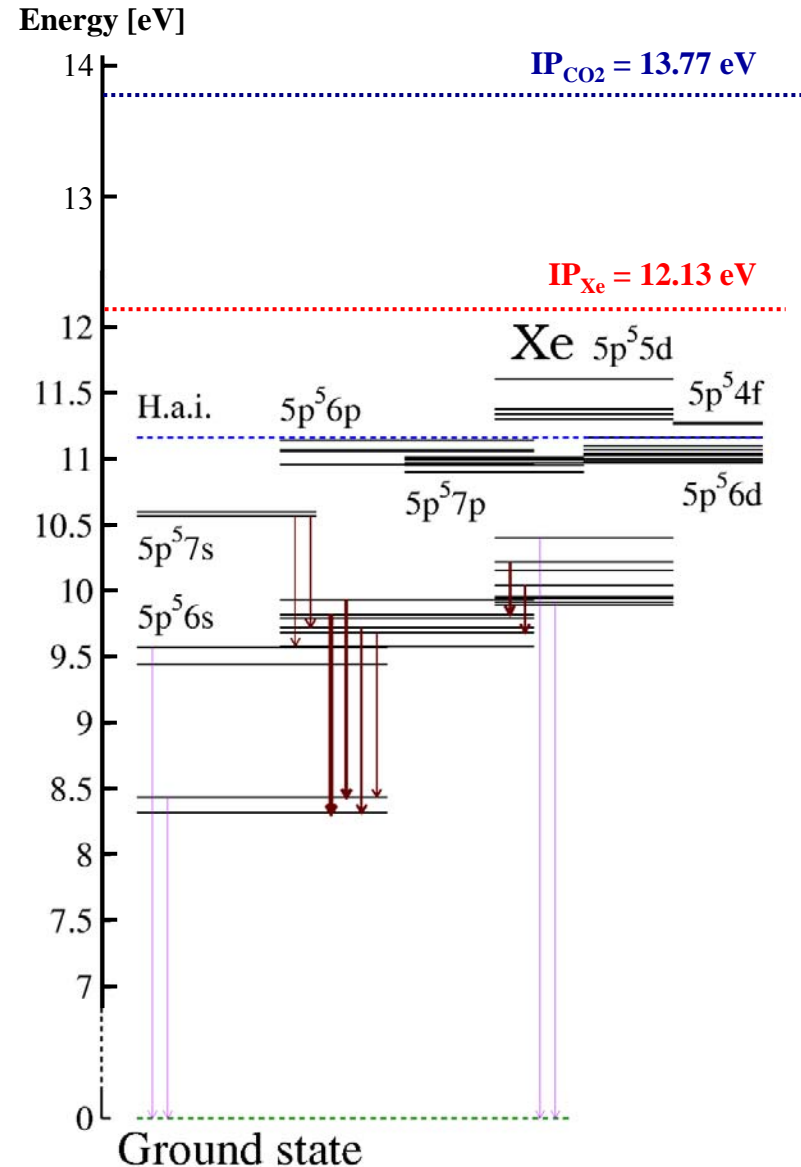
❖ Difference on transfer rates  $\approx 8\%$

❖  $r \approx 15\text{-}20\%$  differences at 1 atm ?

❖ applied voltages

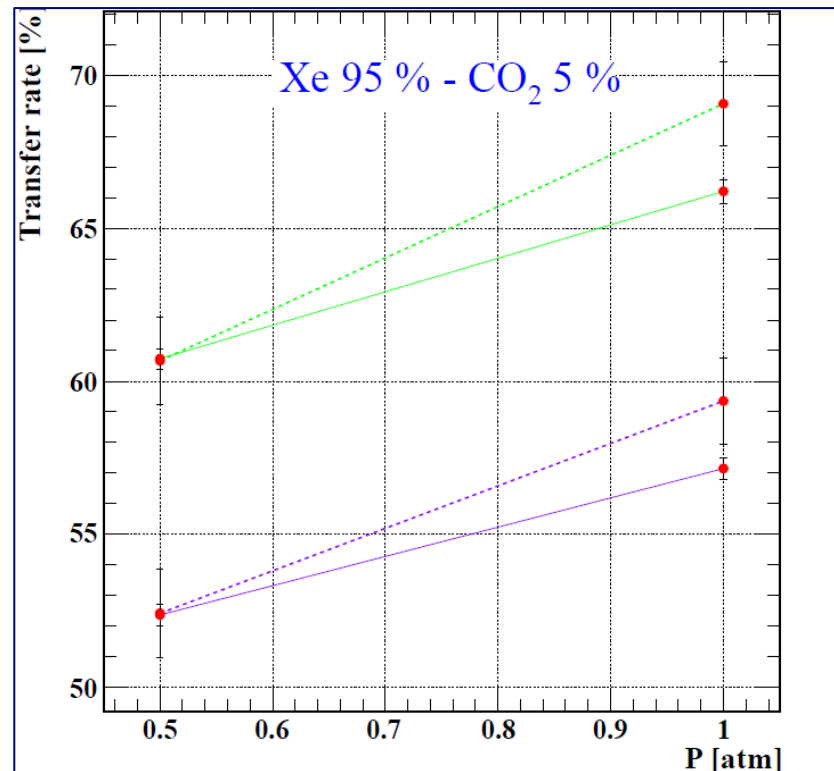
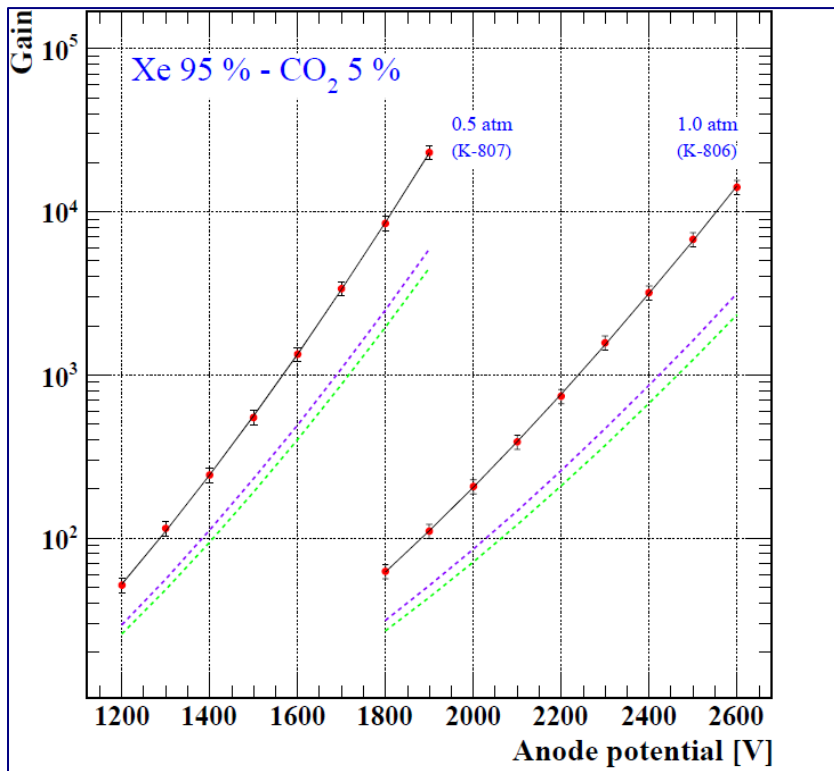
# Xenon – CO<sub>2</sub> Mixtures

- ❖  $\text{Xe}^* + \text{CO}_2 \rightarrow \text{Xe} + \text{CO}_2^+ + \text{e}^-$   
NOT possible process energetically
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + \text{e}^-$   
 $\varepsilon > 11.162 \text{ eV}$   $\text{Xe}^*$  states can transfer





# Xenon – CO<sub>2</sub> Mixtures



❖ No feedback

Ser. No	$r_c$ [cm]	$r_a$ [ $\mu\text{m}$ ]
K-807	1.27	50.8
K-806	1.27	50.8

❖ Magboltz 8.9.3:  $g_w = 0.973$

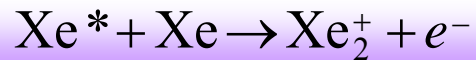
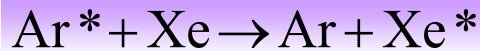
❖ Magboltz 8.9.1:  $g_w = 0.996$

❖ Difference on transfer rates  $\approx 11 \%$

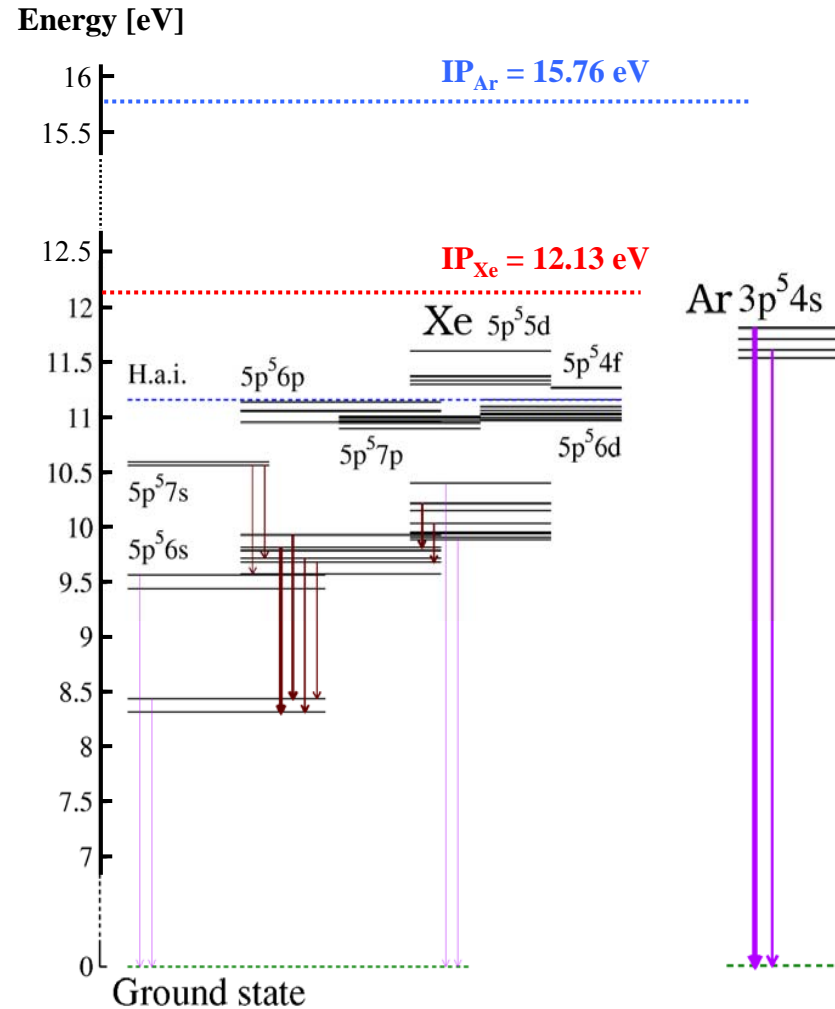
[R.W. Hendricks, *Nucl. Instr. And Meth. A* **102** (1972) 309–312.]

# Xenon – Argon Mixtures

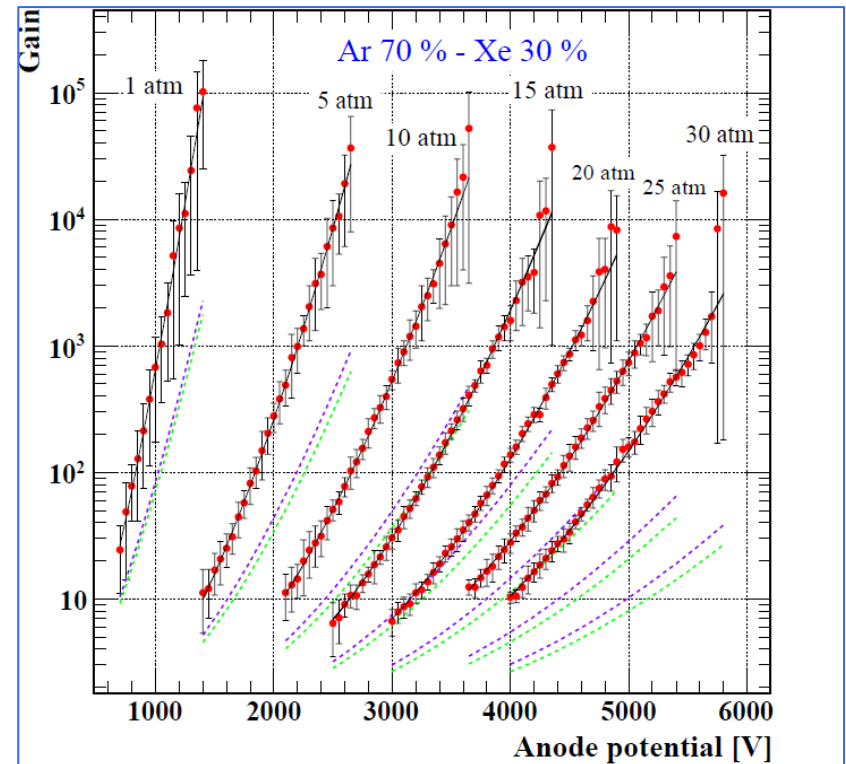
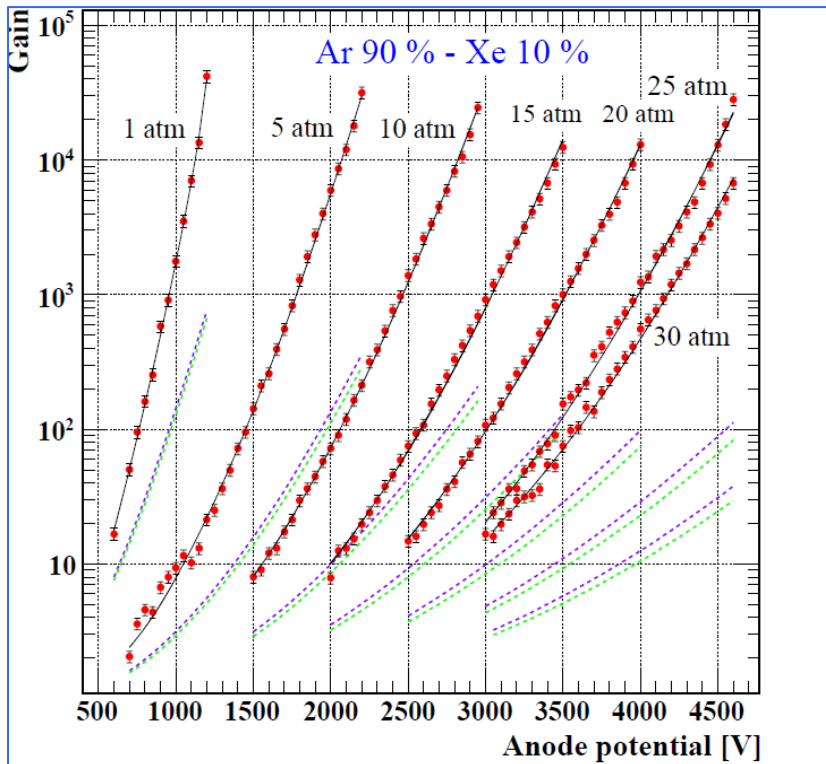
- ❖  $\text{Ar}^* + \text{Xe} \rightarrow \text{Ar} + \text{Xe}^+ + e^-$   
 $\varepsilon > 12.13 \text{ eV}$   $\text{Ar}^*$  states can transfer
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + e^-$   
 $\varepsilon > 11.162 \text{ eV}$   $\text{Xe}^*$  states can transfer
- ❖ Also  $\text{Ar}^*$  induced transfers are possible



[Ö. Şahin et al. *JINST P05002* (2010) 1–30.]



# Xenon – Argon Mixtures

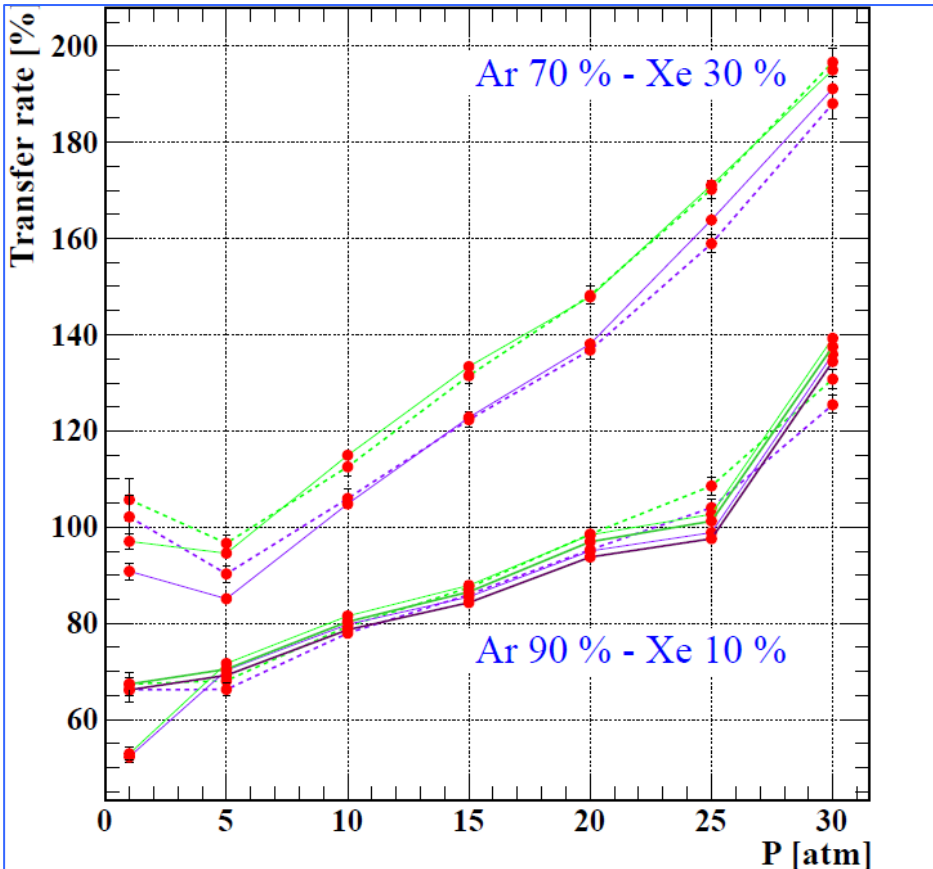


- ❖  $r_a = 12.5 \mu\text{m}$ ,  $r_c = 1.5 \text{ cm}$ ,
- ❖ Feedback at 1 atm for 10 % Xe
  - ❖  $1.05 \cdot 10^{-5} \pm 0.30 \cdot 10^{-5}$  (g free)

- ❖ Larger errors at high gains
- ❖ No feedback

[D.J. Grey et al., *Nucl. Instr. and Meth. A* **527** (2004) 493–511].

# Xenon – Argon Mixtures



Ar 90 % - Xe 10 %

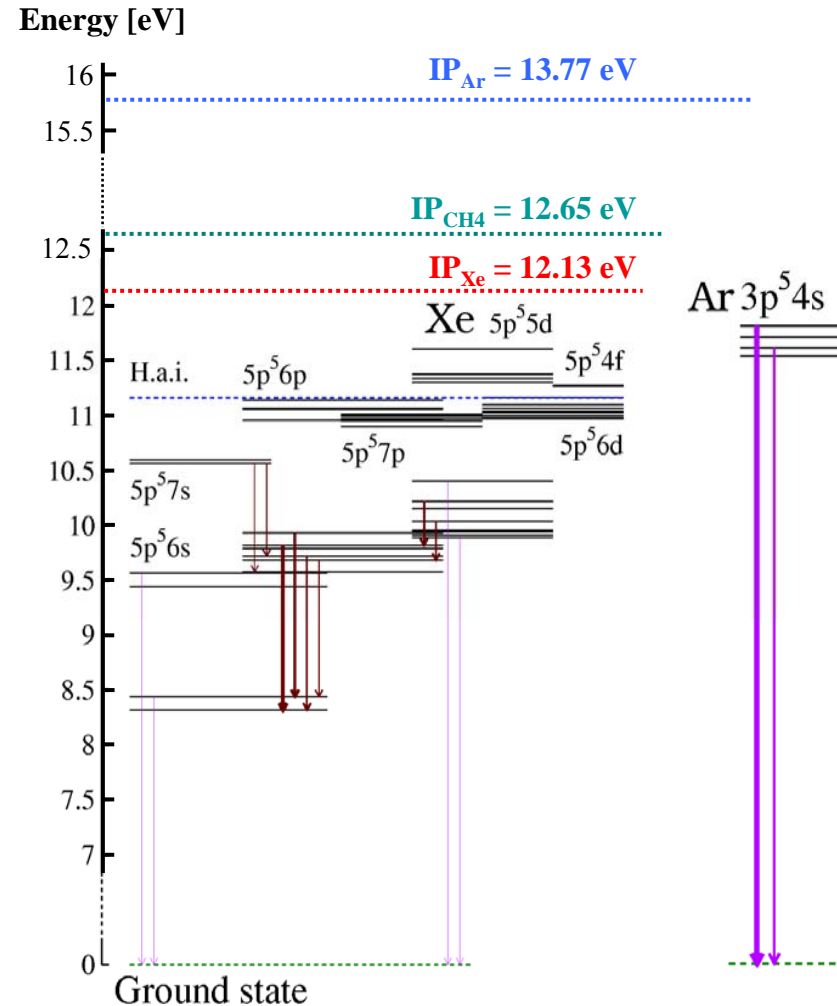
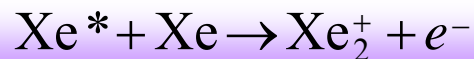
- ❖ Magboltz 8.9.3:
  - ❖  $g_{w1} = 0.88, g_{w2} = 0.93$
- ❖ Magboltz 8.9.1:
  - ❖  $g_{w1} = 0.92, g_{w2} = 0.97$
- ❖ Difference on transfer rates  $\approx 2 \%$

Ar 70 % - Xe 30 %

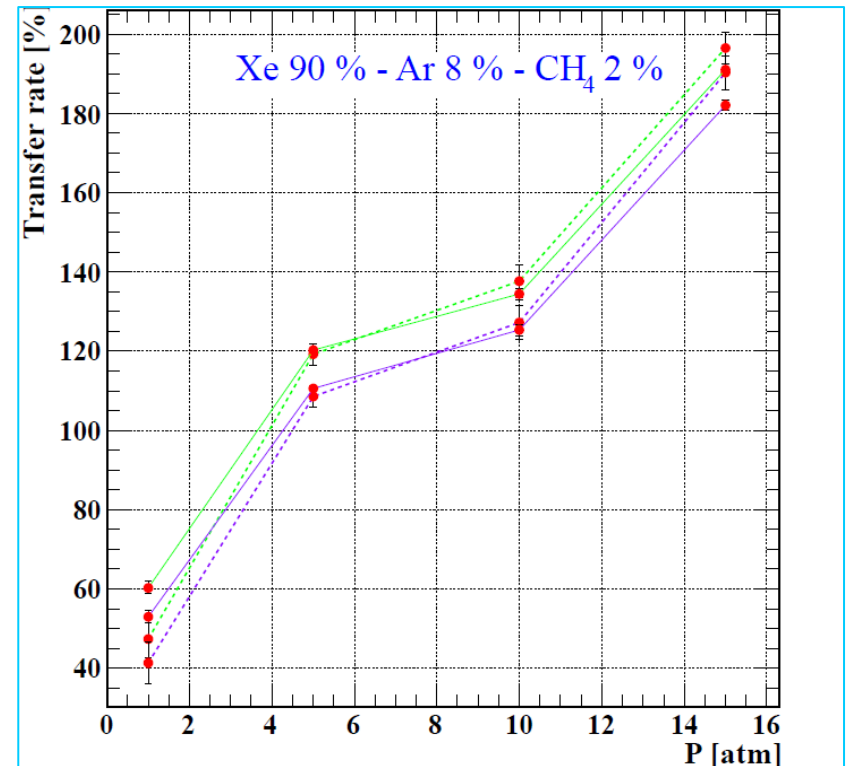
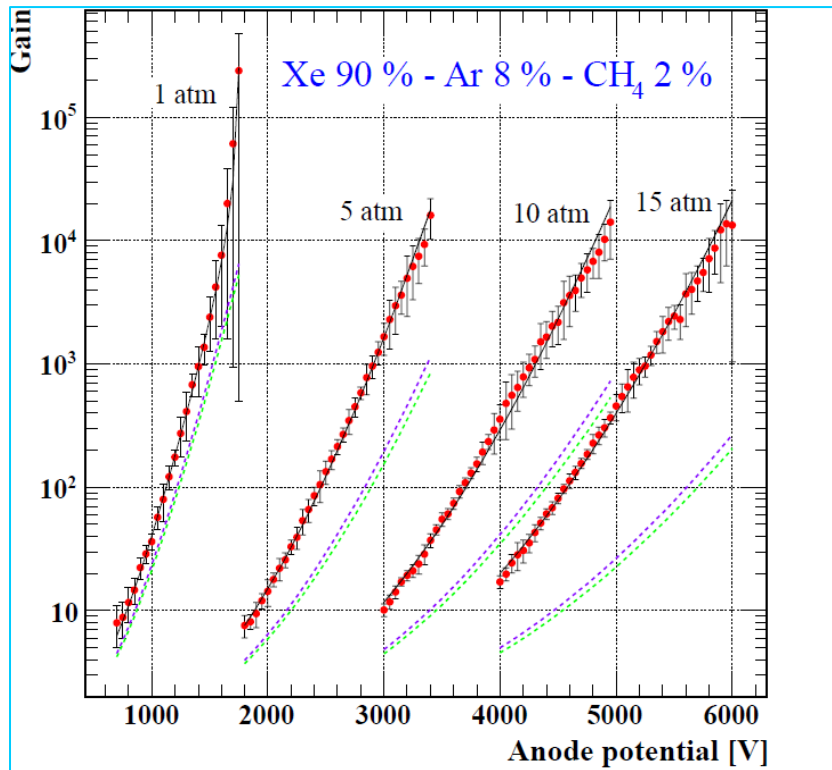
- ❖  $g_w = 0.86, g_w = 0.83$
- ❖ Difference on transfer rates  $\approx 10 \%$

# Xenon – Argon – Methane Mixtures

- ❖  $\text{Ar}^* + \text{CH}_4 \rightarrow \text{Ar} + \text{CH}_4^+ + e^-$   
 $\varepsilon > 12.65 \text{ eV}$  Ar\* states can transfer
- ❖  $\text{Ar}^* + \text{Xe} \rightarrow \text{Ar} + \text{Xe}^+ + e^-$   
 $\varepsilon > 12.13 \text{ eV}$  Ar\* states can transfer
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + e^-$   
 $\varepsilon > 11.162 \text{ eV}$  Xe\* states can transfer
- ❖ Ar\* induced transfers are possible



# Xenon – Argon – Methane Mixtures



## ❖ Feedback at 1 atm

❖  $4.87 \cdot 10^{-5} \pm 0.74 \cdot 10^{-5}$  (g free)

❖ Larger feedback term than Xe-Ar ?

❖ Big errors at 1 atm

❖ Magboltz 8.9.3:  $g_w = 0.973$

❖ Magboltz 8.9.1:  $g_w = 0.996$

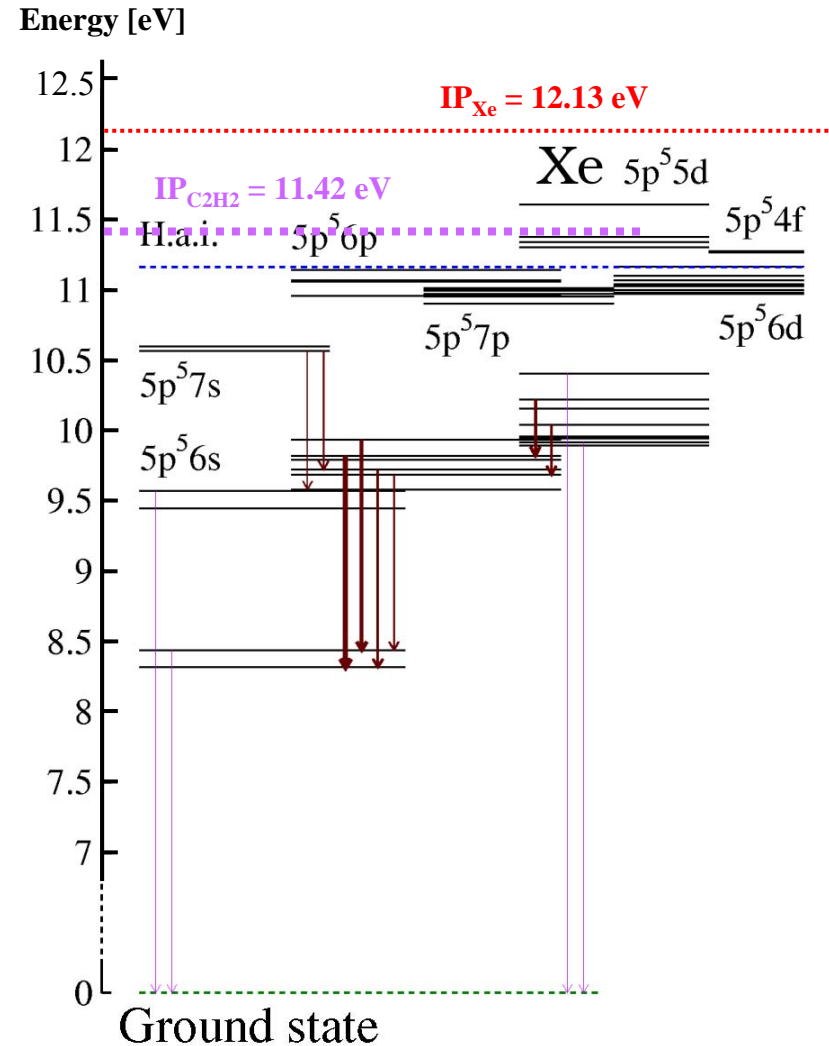
❖ Difference on transfer rates  $\approx 11\%$

❖ Larger transfer rates than Xe-Ar mixtures at high pressures ?

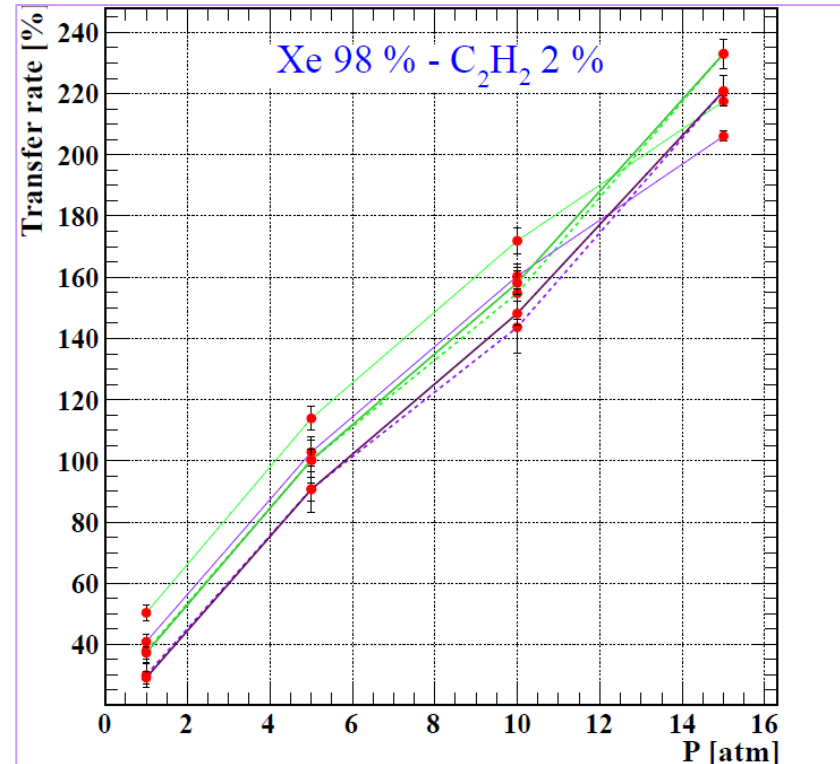
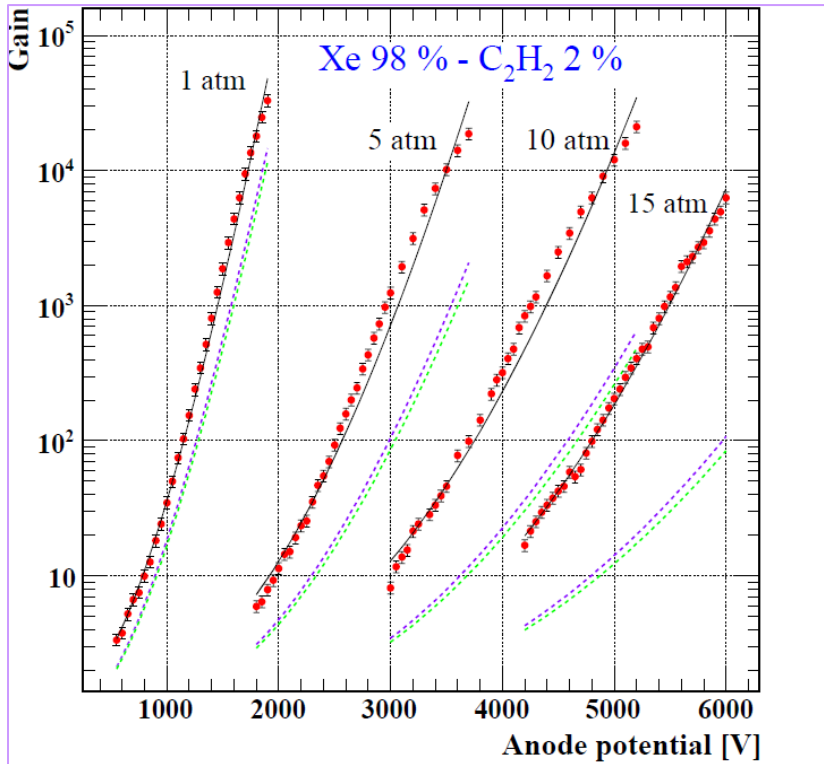
[D.J. Grey et al., (2004)].

# Xenon – Acetylene Mixtures

- ❖  $\text{Xe}^* + \text{C}_2\text{H}_2 \rightarrow \text{Xe} + \text{C}_2\text{H}_2^+ + \text{e}^-$   
 $\varepsilon > 11.42 \text{ eV}$   $\text{Xe}^*$  states can transfer
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + \text{e}^-$   
 $\varepsilon > 11.162 \text{ eV}$   $\text{Xe}^*$  states can transfer



# Xenon – Acetylene Mixtures



- ❖ No feedback
- ❖  $r \approx 76\%$  in Ar 98 % - C<sub>2</sub>H<sub>2</sub> 2% at 1 atm  
[Ö. Şahin et al. *JINST P05002* (2010) 1–30.]

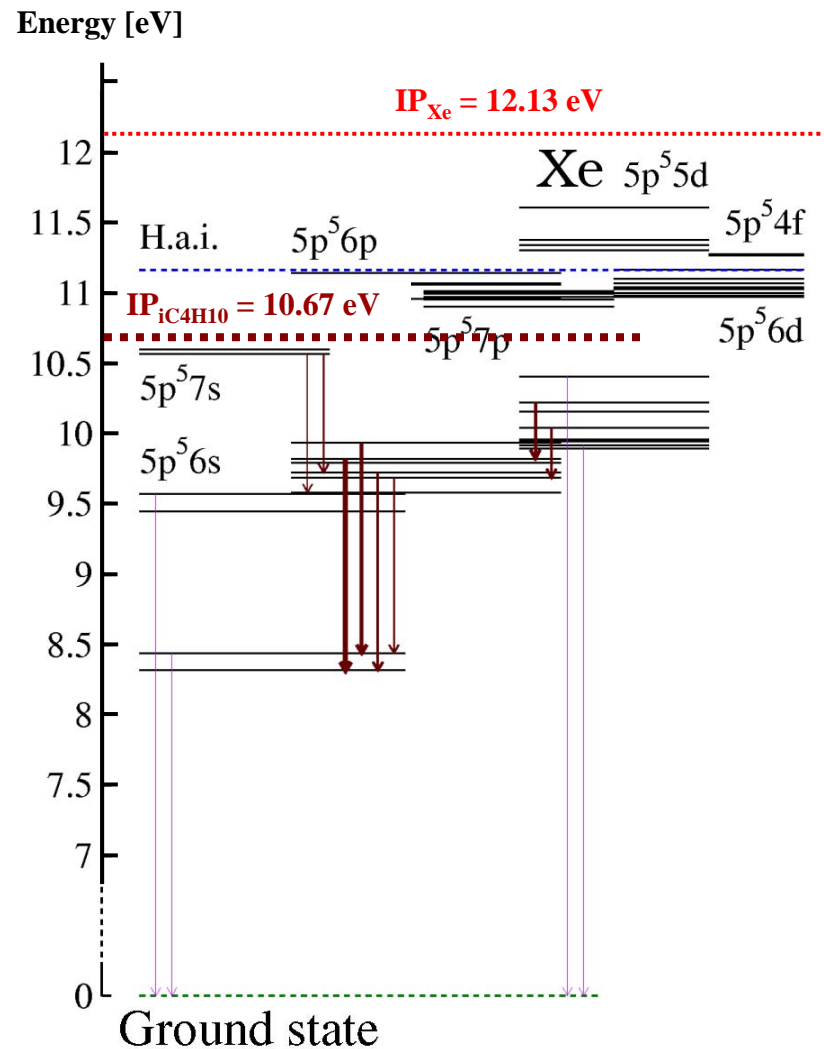
[R.K. Manchanda et al., 2008].

- ❖ Magboltz 8.9.3:  $g_{w1,2} = 1.21, 1.51$
- ❖ Magboltz 8.9.1:  $g_{w1,2} = 1.19, 1.54$
- ❖ Difference on transfer rates  $\approx 10\%$
- ❖ The largest transfer rates at high pressures

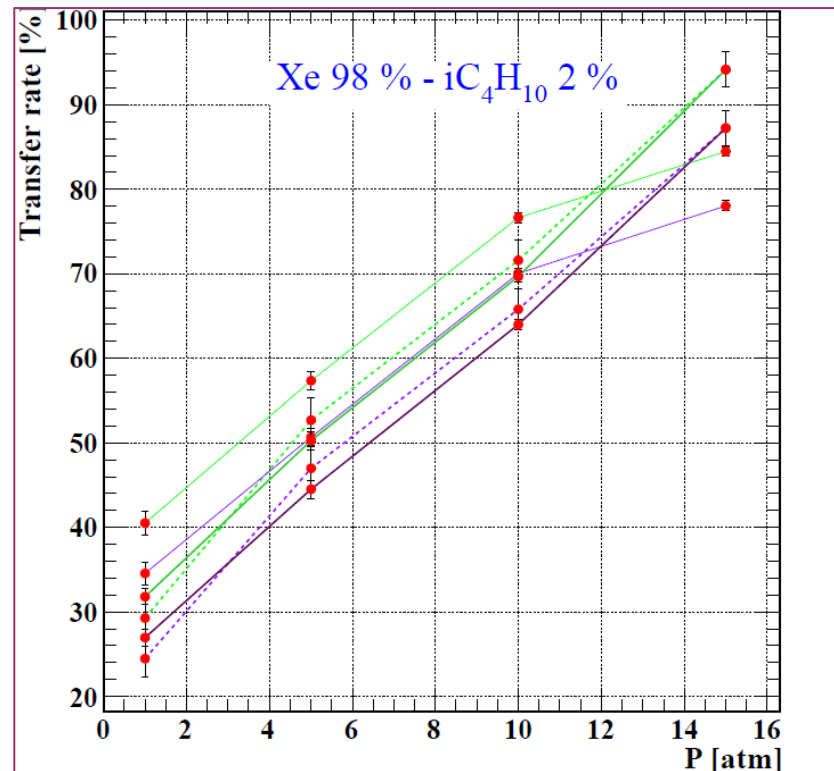
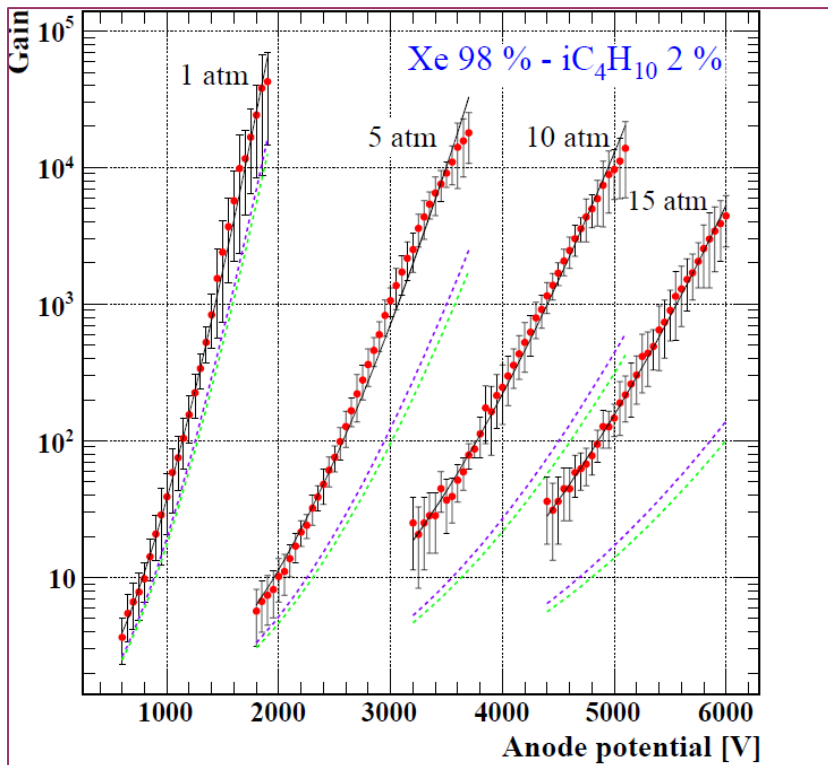


# Xenon – Isobutane Mixtures

- ❖  $\text{Xe}^* + \text{iC}_4\text{H}_{10} \rightarrow \text{Xe} + \text{C}_4\text{H}_{10}^+ + \text{e}^-$   
 $\varepsilon > 10.67 \text{ eV}$   $\text{Xe}^*$  states can transfer
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + \text{e}^-$   
 $\varepsilon > 11.162 \text{ eV}$   $\text{Xe}^*$  states can transfer



# Xenon – Isobutane Mixtures



❖ No feedback

❖ Magboltz 8.9.3:  $g_{w1,2} = 0.94, 1.18$

❖ Magboltz 8.9.1:  $g_{w1,2} = 0.91, 1.19$

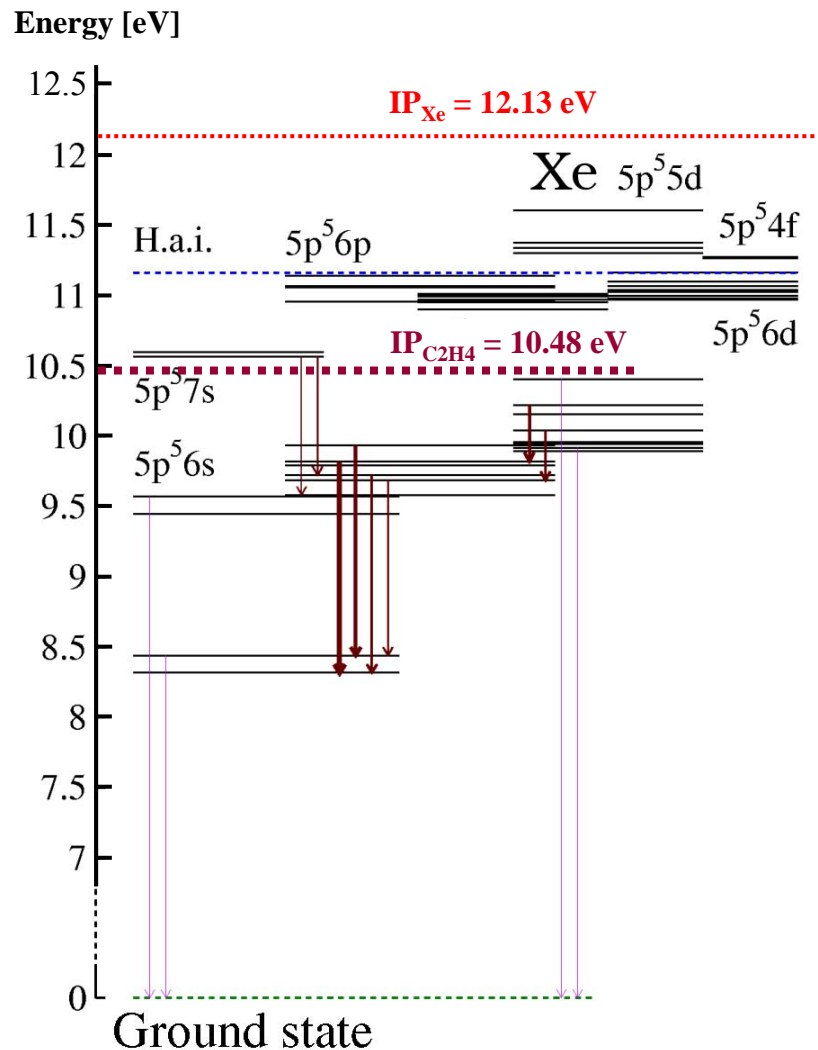
❖ Difference on transfer rates  $\approx 7\%$

❖  $r < 100\%$  !!!

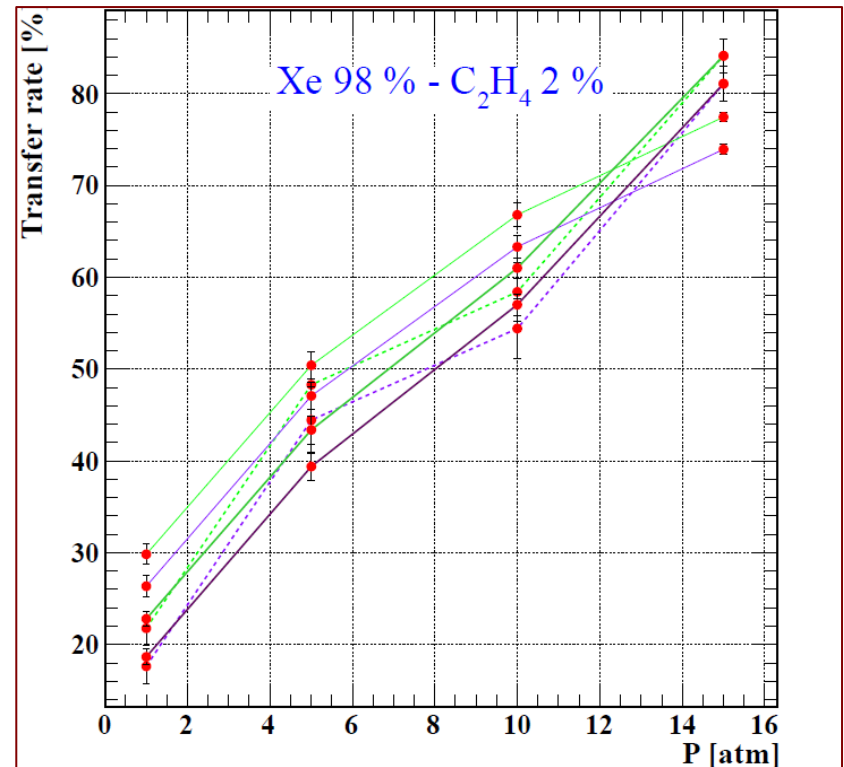
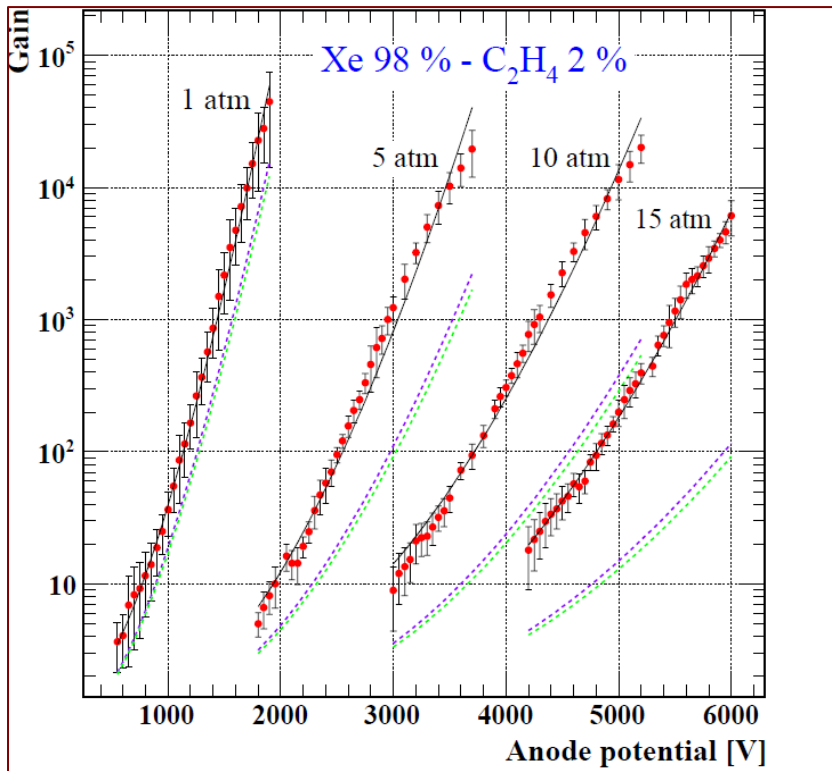
[D.J. Grey et al., 2004].

# Xenon – Ethylene Mixtures

- ❖  $\text{Xe}^* + \text{C}_2\text{H}_4 \rightarrow \text{Xe} + \text{C}_2\text{H}_4^+ + \text{e}^-$   
 $\varepsilon > 10.48 \text{ eV}$   $\text{Xe}^*$  states can transfer
- ❖  $\text{Xe}^* + \text{Xe} \rightarrow \text{Xe}_2^+ + \text{e}^-$   
 $\varepsilon > 11.162 \text{ eV}$   $\text{Xe}^*$  states can transfer



# Xenon – Ethylene Mixtures



❖ No feedback

- ❖ Magboltz 8.9.3:  $g_{w1,2} = 1.04, 1.40$
- ❖ Magboltz 8.9.1:  $g_{w1,2} = 1.07, 1.41$
- ❖ Difference on transfer rates  $\approx 4 \%$
- ❖ The lowest transfer rates !!!
  - ❖ The lowest ionisation potential of C<sub>2</sub>H<sub>4</sub>

[D.J. Grey et al., 2004 ].

# Conclusions

- ❖ Pressure dependence of transfer rates,
- ❖ Large transfer rates ( $r > 100\%$  many times)
  - ❖ homonuclear associative ionisations,
- ❖ Next
  - ❖ need to understand the rates with transfer model,
  - ❖ separation of the transfer processes.

*Thank you ...*