

Shining light on neptunium

Laser spectroscopy for probing atomic structure

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02.09.2024

Neptunium

- Radioactive actinide
- Long half-life - ^{237}Np - $2.14 \cdot 10^6$ y
- High radiotoxicity



- Ionization potential $50\,535(2)\text{ cm}^{-1}$ [1]
 $6.26554(25)\text{ eV}$
- 462 atomic levels known [2]
- ^{237}Np
 - Magnetic dipole moment $+3.14(4)\mu_N$ [3]
 - Electric quadrupole moment $+3.886(4)\text{ eb}$ [3]
- ^{239}Np
 - Nuclear moments unknown

[1] S. Kohler et al. Spectrochim. Acta B **52**, 717 – 726, (1997)

[2] V. Kazakov et al. Phys. Scr. **92**, 10, (2017)

[3] N.J. Stone, At. Data Nucl. Data Tables **90**, 1, 75-176, (2005)

Neptunium production and trace analysis

Pu 237 45.2 d	Pu 238 87.74 a α 5.499; 5.456... sf; Si; Mg; γ ; e ⁻	Pu 239 2.44·10 ⁵ a α 5.157; 5.144... sf; γ ; e ⁻ ; m	Pu 240 6545 a α 5.168; 5.124... sf; γ ; e ⁻ ; g	Pu 241 14.35 a β^- ; g; α 4.896... γ (149); e ⁻
Np 236 22.5 h ϵ ; β^- ; e ⁻ γ (642...); g	Np 237 2.14·10 ⁶ a α 4.790...; 29...; e ⁻	Np 238 2.102 d β^- 1.2... γ 984...; e ⁻ ; g	Np 239 2.355 d β^- 0.4... γ 106...; e ⁻ ; g	Np 240 7.22 m 65 m β^- 2.2...; β^- 0.9... e ⁻ ; γ ; g; g
U 235 0.7200 26 m lf (0.07); e ⁻	U 236 7.038·10 ⁸ a 120 ns lf 1783; 642... sf	U 237 2·10 ⁷ a 6.75 d β^- 0.2... γ 60; 208...; e ⁻	U 238 99.2745 0.3 μ s lf 2514; 1879... sf	U 239 4.468·10 ⁹ a 23.5 m β^- 1.2; 1.3...; γ 75; 44...

→ neutron capture
← β^- decay



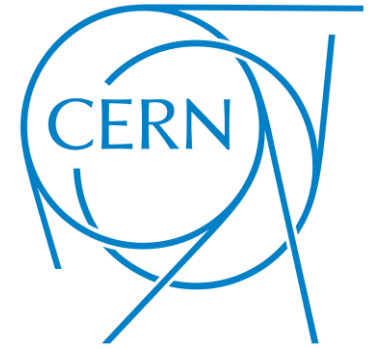
It is essential to monitor nuclear waste repositories and their long-term performance

Trace analysis of environmental samples is of high relevance

- ²³⁹Np as a tracer for precise quantification
- Identification of efficient ionization schemes:
 - High elemental selectivity
 - Isotope related effects like hyperfine structures (HFS) and isotope shift (IS)
 - large splitting and shifts observed in neptunium

Neptunium - on-line studies

- On-line studies of neptunium at ISOLDE
 - In-source laser resonance ionization spectroscopy of neptunium and plutonium (*Letter of Intent INTC-I-243*)
- Access to other isotopes
- Nuclear moments are only known for ^{237}Np
- Validation of theoretical estimates

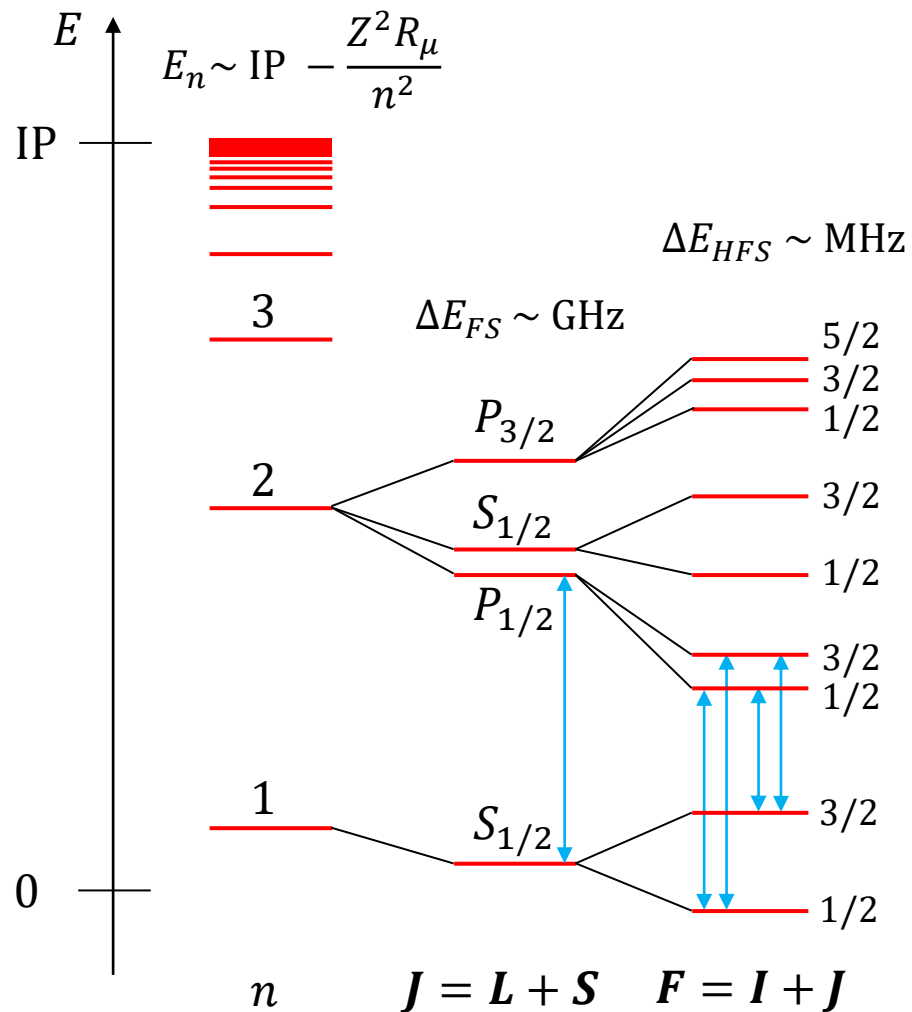


Need for efficient and selective ionization schemes

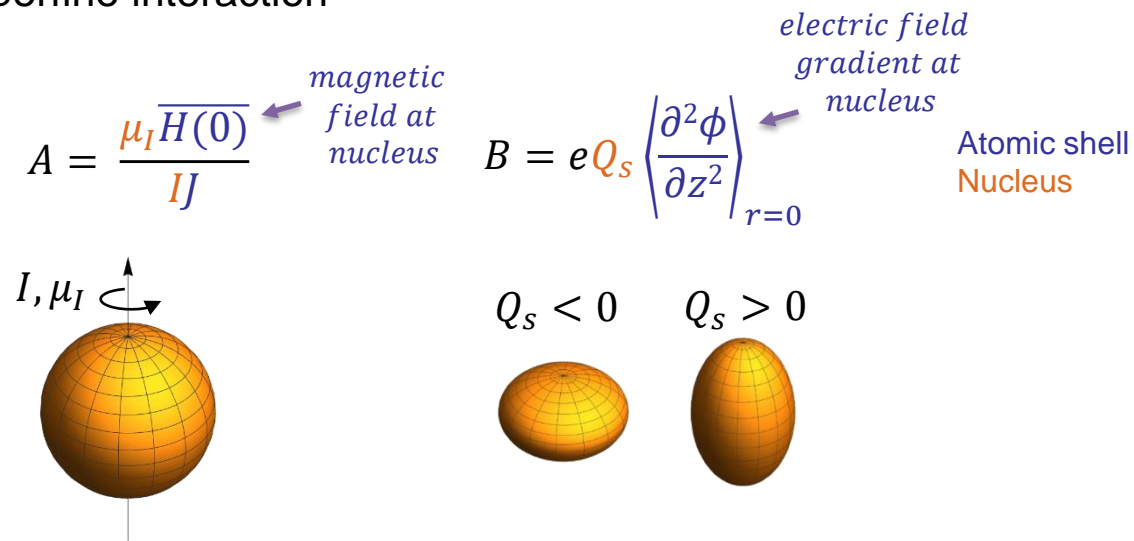
Technique and setup

Resonance ionization spectroscopy
The RISIKO mass separator
Laser ion source

Probing atomic structure with lasers



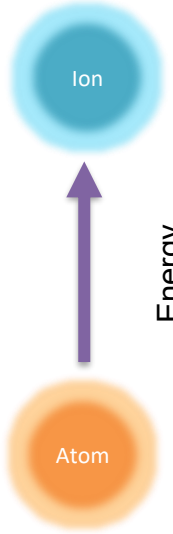
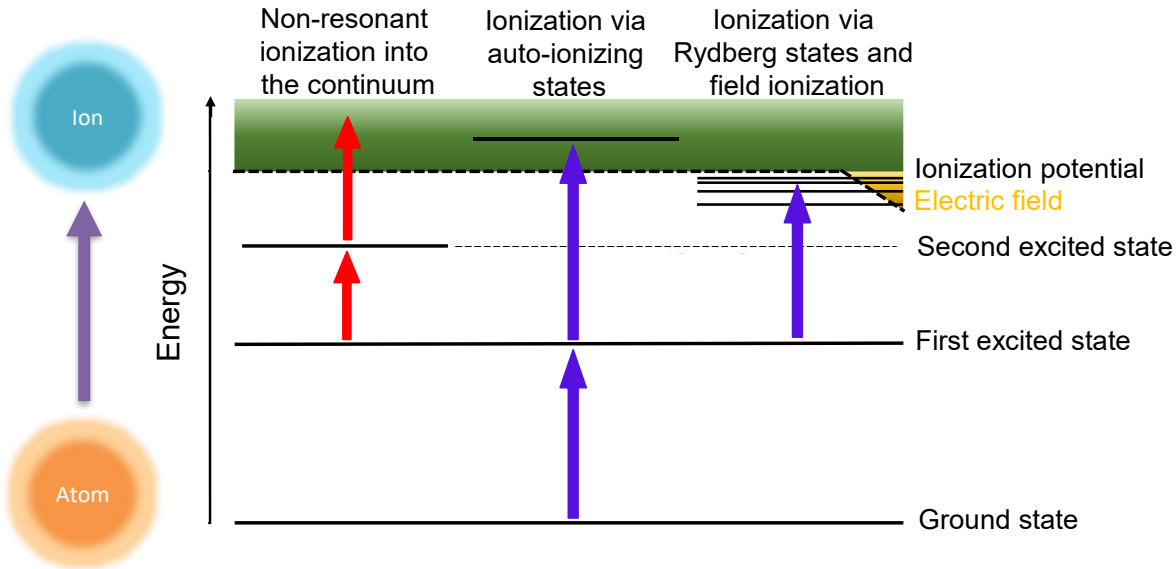
- Atomic level structure through optical transitions
- Ionization limit through Rydberg convergences
- Nuclear spins and electromagnetic moments through hyperfine interaction



n – principal quantum number
 Z – atomic number
 S – total spin angular momentum.
 I – nuclear spin
 μ – magnetic dipole moment

IP – ionization potential
 R – Rydberg constant
 L – total orbital angular momentum
 J – total angular momentum
 Q – electric quadrupole moment

Resonance ionization mass spectrometry



Mass spectrometry

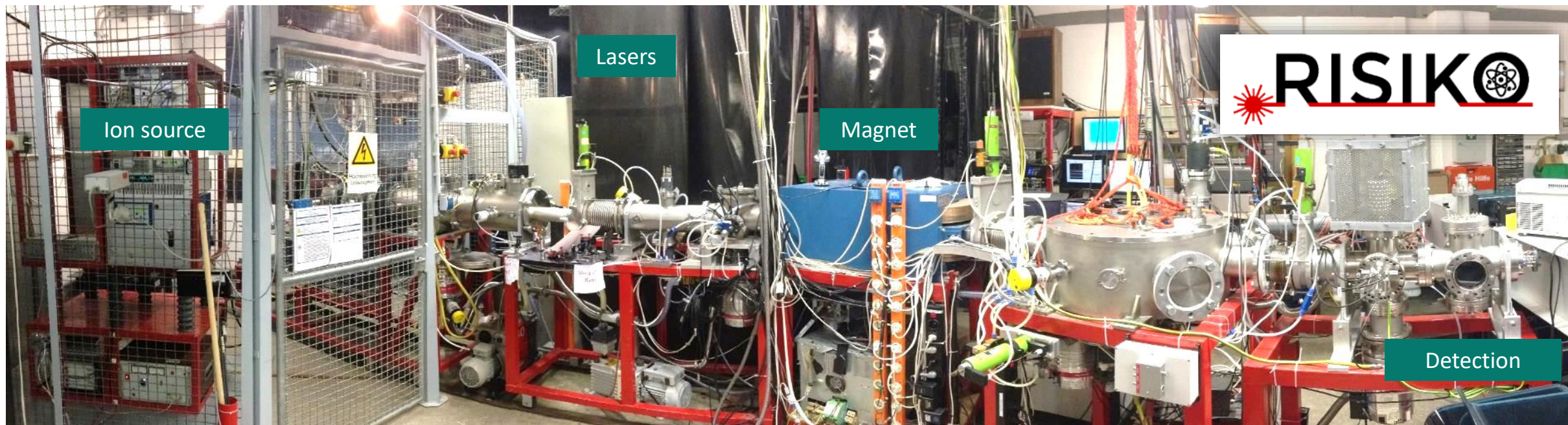
- Separation of m/q
- High detection efficiency

Laser ionization and spectroscopy

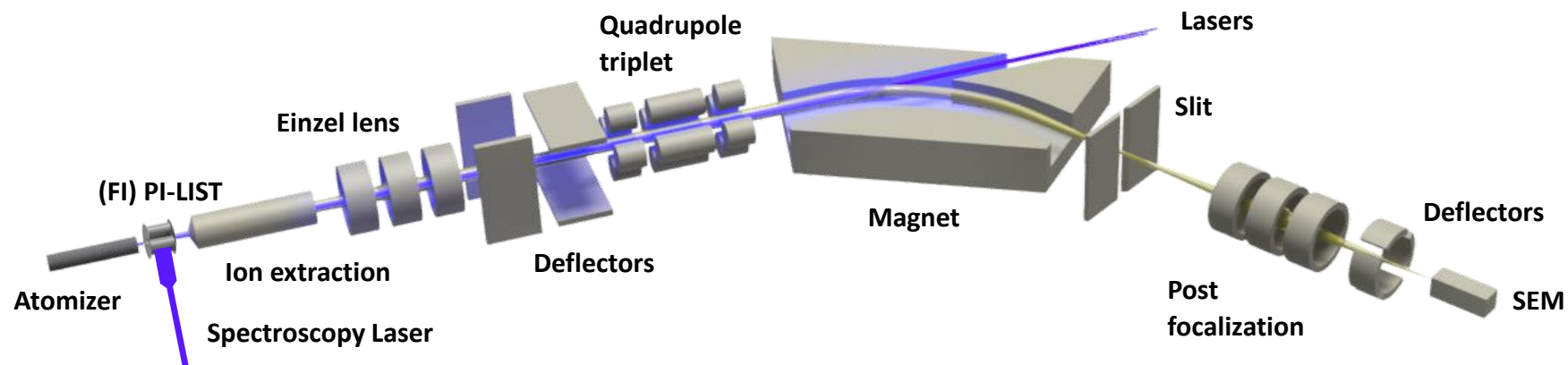
- Highly efficient process
- High element selectivity

	Pu 237 45.2 d	Pu 238 87.74 a α 5.499; 5.456... sf; Si; Mg; γ ; e	Pu 239 $2.44 \cdot 10^5$ a α 5.157; 5.144... sf; γ ; e; m	Pu 240 6545 a α 5.168; 5.124... sf; γ ; e; g	Pu 241 14.35 a β^- ; g; α 4.896... γ (149); e
	Np 236 22.5 h ϵ ; β^- ; e γ (642...); g	Np 237 $1.6 \cdot 10^5$ a ϵ (89%); β^- (12%); α ; γ 160...; e	Np 238 2.102 d β^- 1.2... γ 984...; e; g	Np 239 2.355 d β^- 0.4... γ 106...; e; g	Np 240 7.22 m 65 m β^- 2.2...; β^- 0.9 γ ; e; γ ; g; g
	U 235 0.7200 26 m $7.038 \cdot 10^8$ a β^- (0,07) e α 4.398...; sf Ne; γ 186	U 236 120 ns $2 \cdot 10^7$ a β^- 1783; 642... sf α 4.494...; sf; γ (49...); e	U 237 6.75 d β^- 0.2... γ 60; 208...; e	U 238 99.2745 0.3 μ s $4.468 \cdot 10^9$ a β^- 2514; 1879... sf α 4.198...; sf 26; γ (50); e	U 239 23.5 m β^- 1.2; 1.3...; γ 75; 44...

RISIKO mass separator



- Hot cavity laser ion source
- Pulsed multistep laser ionization
- 30 keV ion extraction
- Mass separation in dipole magnet
- Single ion detection



RISIKO laser ion source

- High operation temp. up to about 2000 °C
- Contamination caused by:
 - black body radiation
 - collisions
- Doppler broadening limiting spectral resolution

$$\Delta\nu = \nu_0 \sqrt{\frac{8k_B T \ln 2}{mc^2}} \sim \text{GHz}$$

$\Delta\nu$ - linewidth

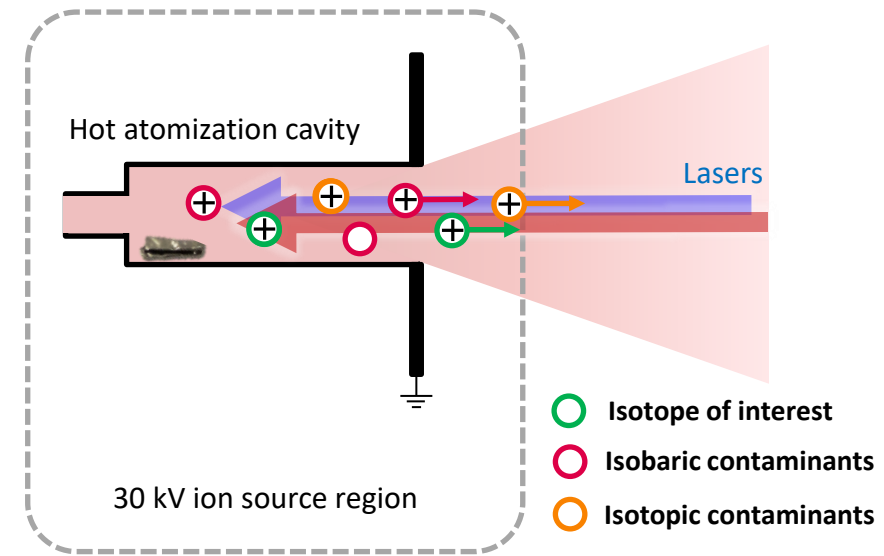
k_B - Boltzmann constant

m - atom mass

ν_0 - resonance freq. for a particle at rest

T - temperature

c - speed of light



Sample



Probing atomic structure

Scheme development
Line profiles
Ionization potential determination
Development and characterization of FI-LIST

Np 237

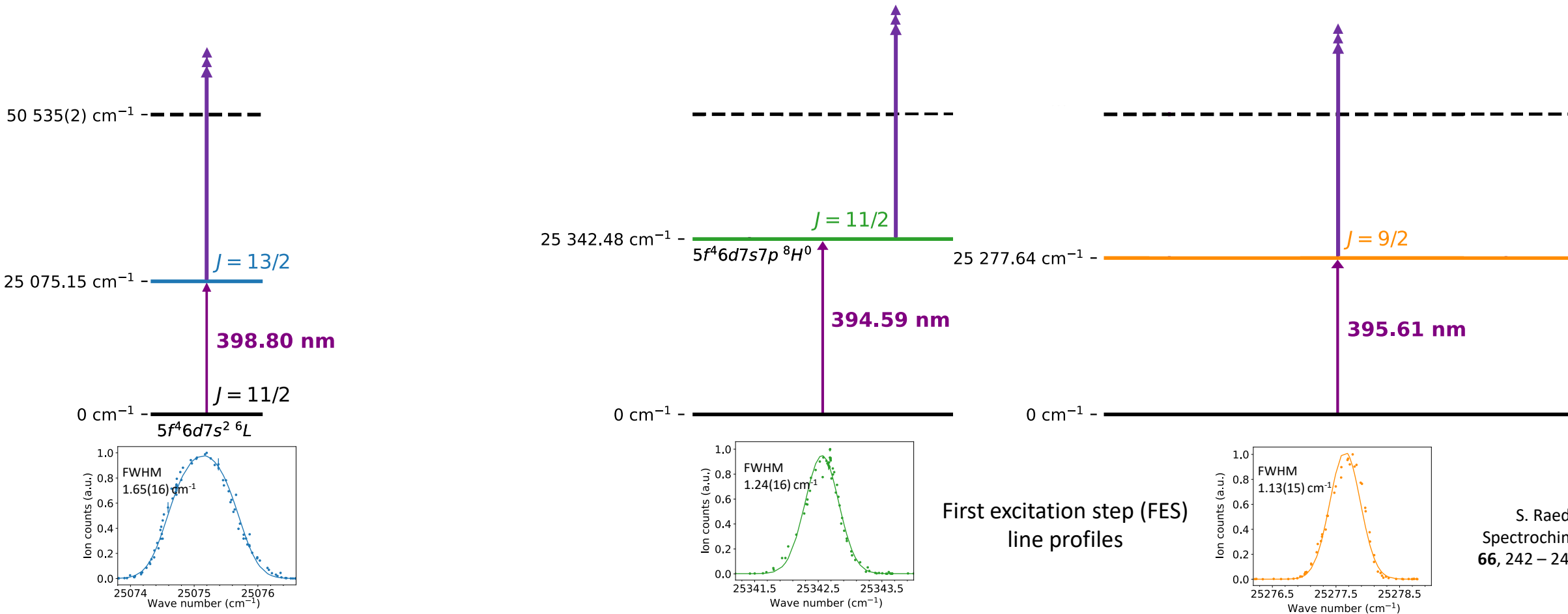
$2.14 \cdot 10^6$ a

α 4.790...;

γ 29...; e^-

Ionization scheme development

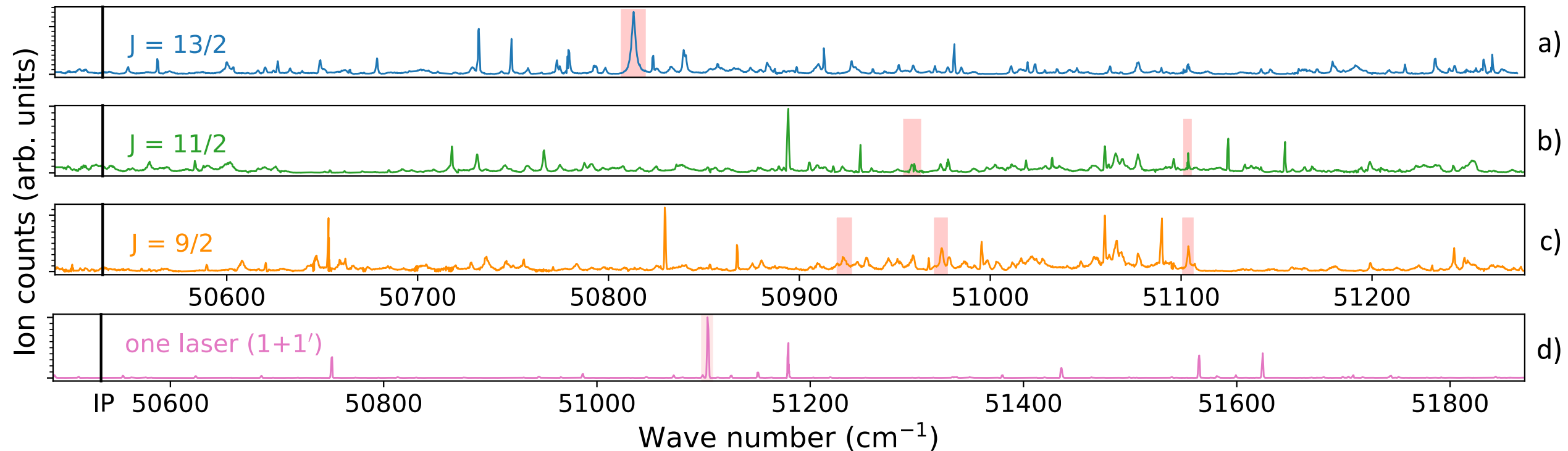
[1] M. Kaja et al., Eur. Phys. J. D **78**, 50 (2024)



All FESs:
S. Raeder et al.,
Spectrochim. Acta B
66, 242 – 247 (2011)

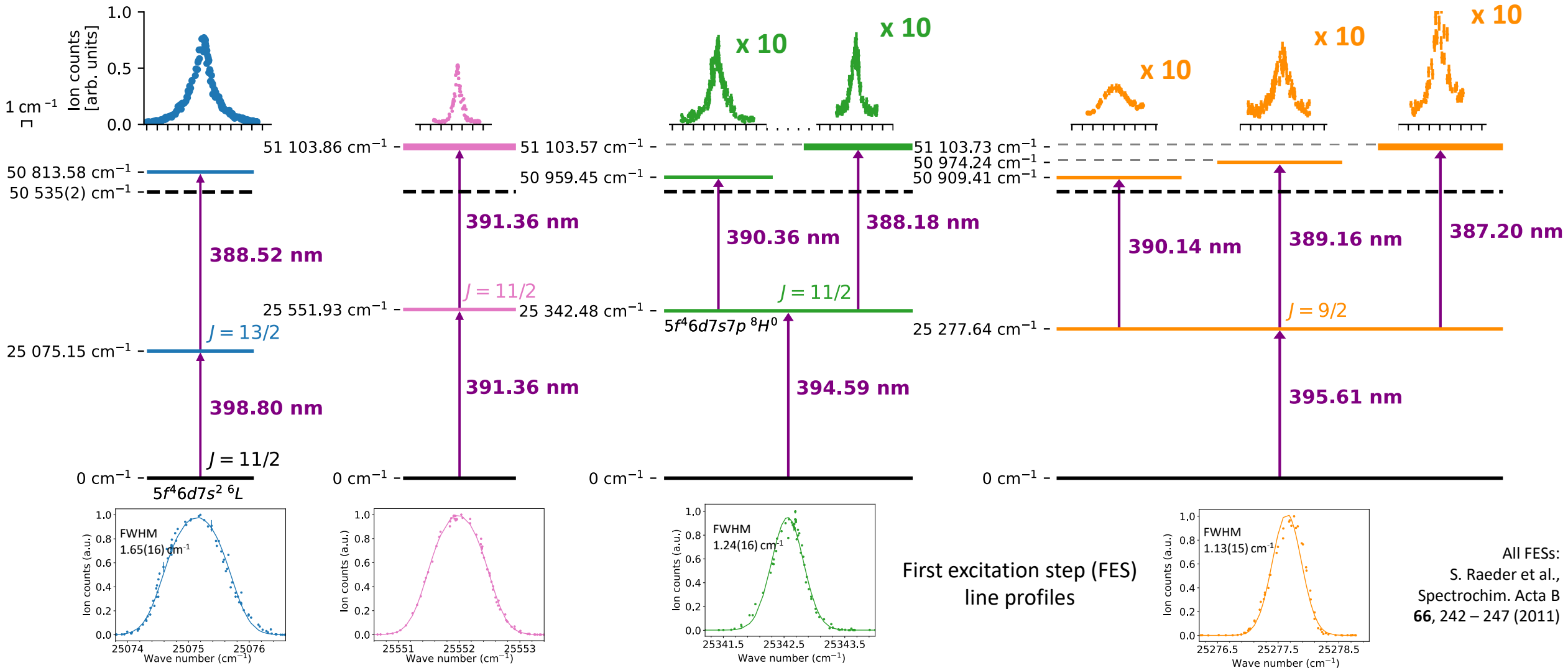
Ionization scheme development

[1] M. Kaja et al., Eur. Phys. J. D, **78**, 50 (2024)



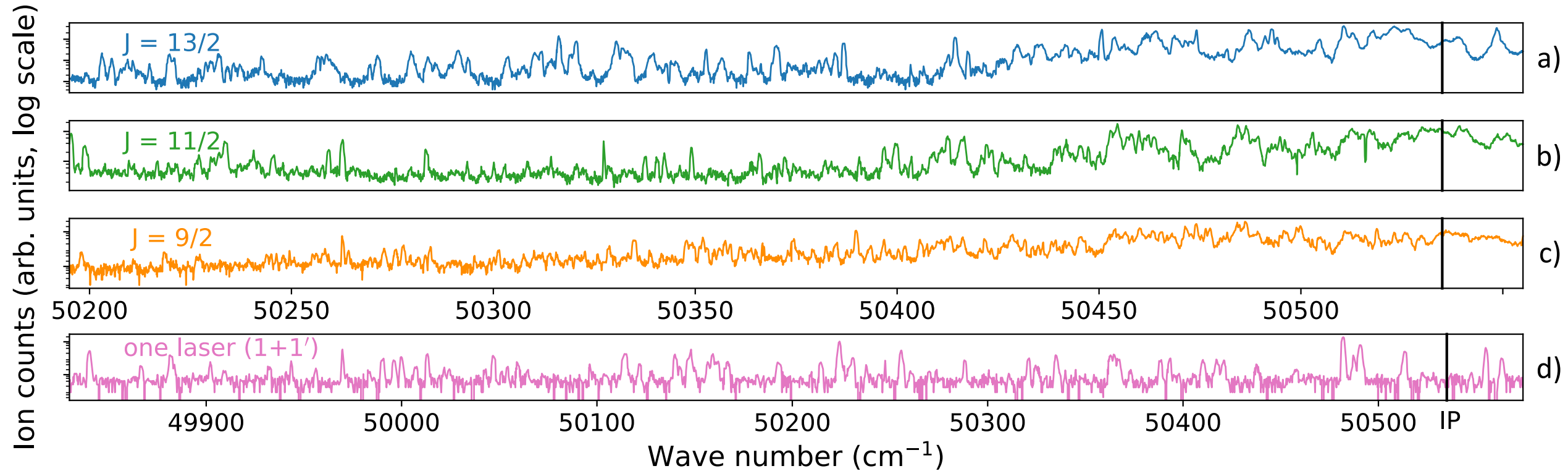
Ionization scheme development

[1] M. Kaja et al., Eur. Phys. J. D **78**, 50 (2024)



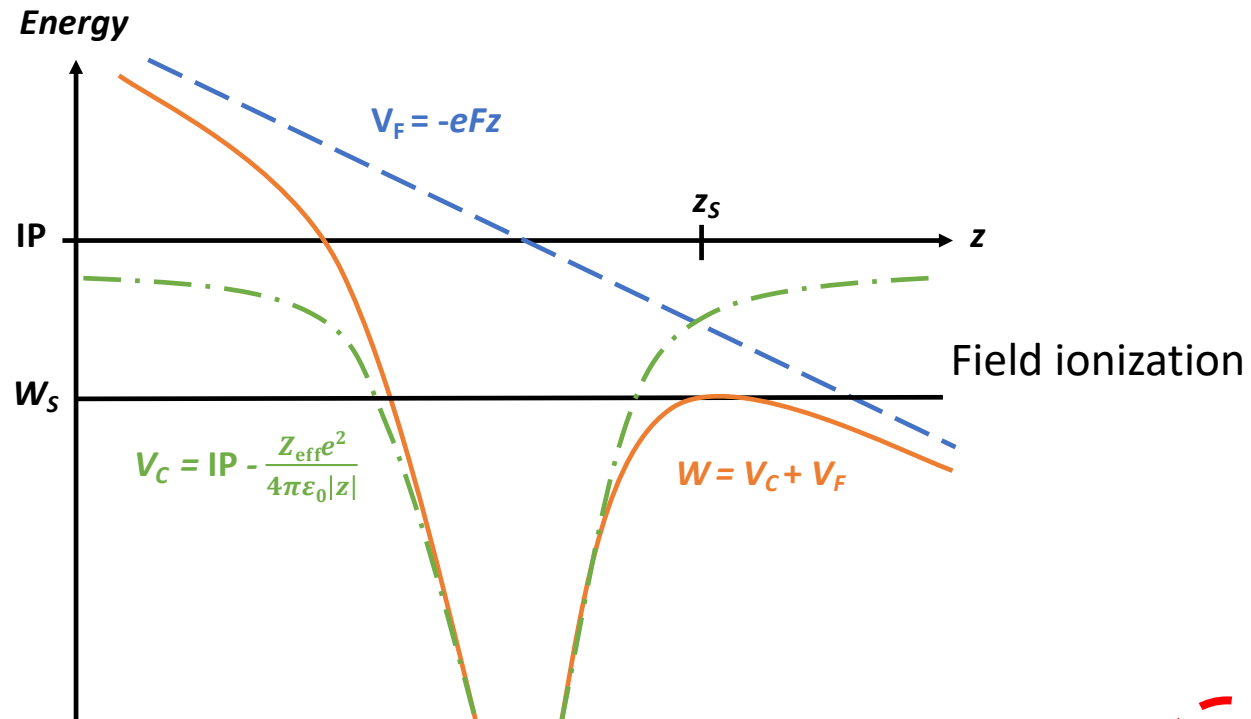
Spectra below the ionization potential

[1] M. Kaja et al., Eur. Phys. J. D, **78**, 50 (2024)



Assignment of levels or analysis of Rydberg Convergences fails for such a complex atomic systems

Field ionization - saddle point model

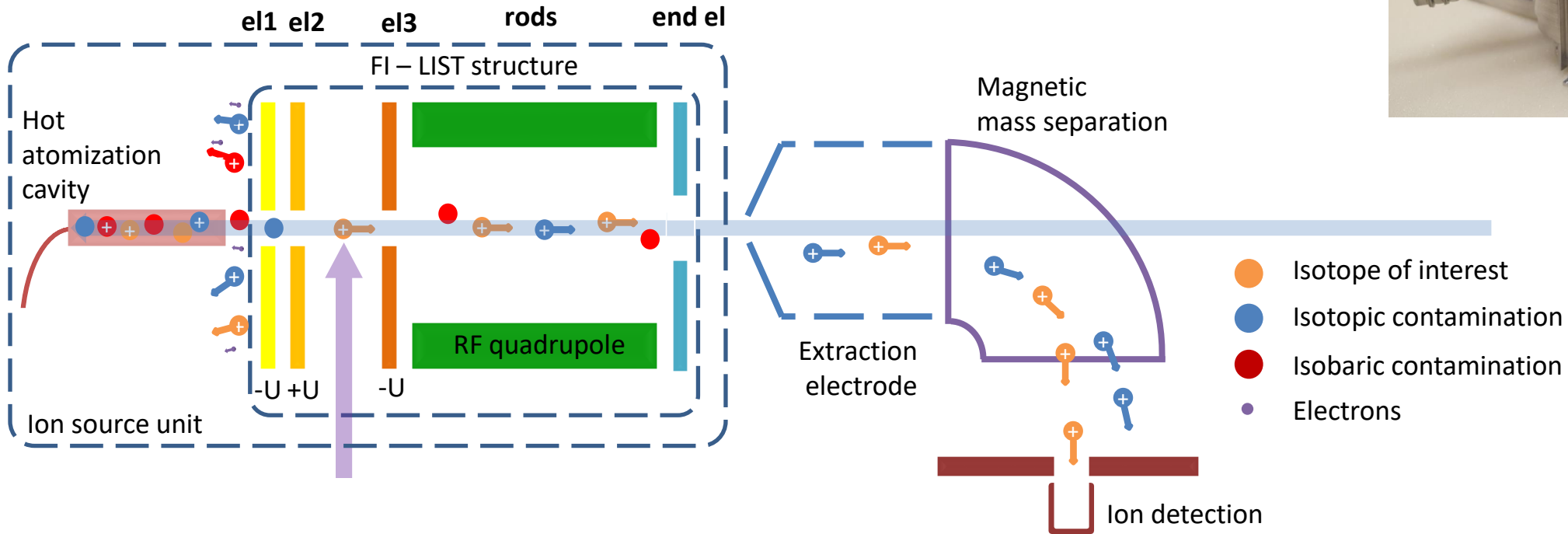
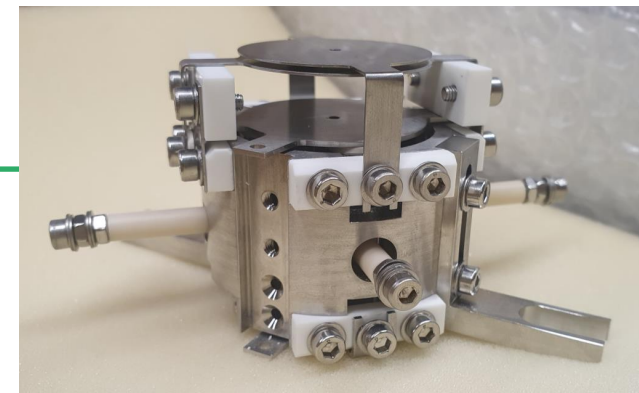


- Ionization threshold according to the saddle point model
- Extrapolation to $F = 0$ V/cm yields IP

$$W_s = IP - 2 \sqrt{\frac{Z_{\text{eff}} e^3}{4\pi\epsilon_0}} \sqrt{F} \quad \text{const}$$

Z_{eff} - effective charge
 F - external field strength
 ϵ_0 - vacuum permittivity

FI-LIST – Field Ionization LIST



- Well controllable electric field for ionization between el2 and el3
- Missing energy to ionize an atom is from the electric field $F = (el2 - el3)/1\text{ cm}$

[1] M. Kaja et al., NIM B **547**, 165213 (2024)

Characterization of FI-LIST – ytterbium measurements

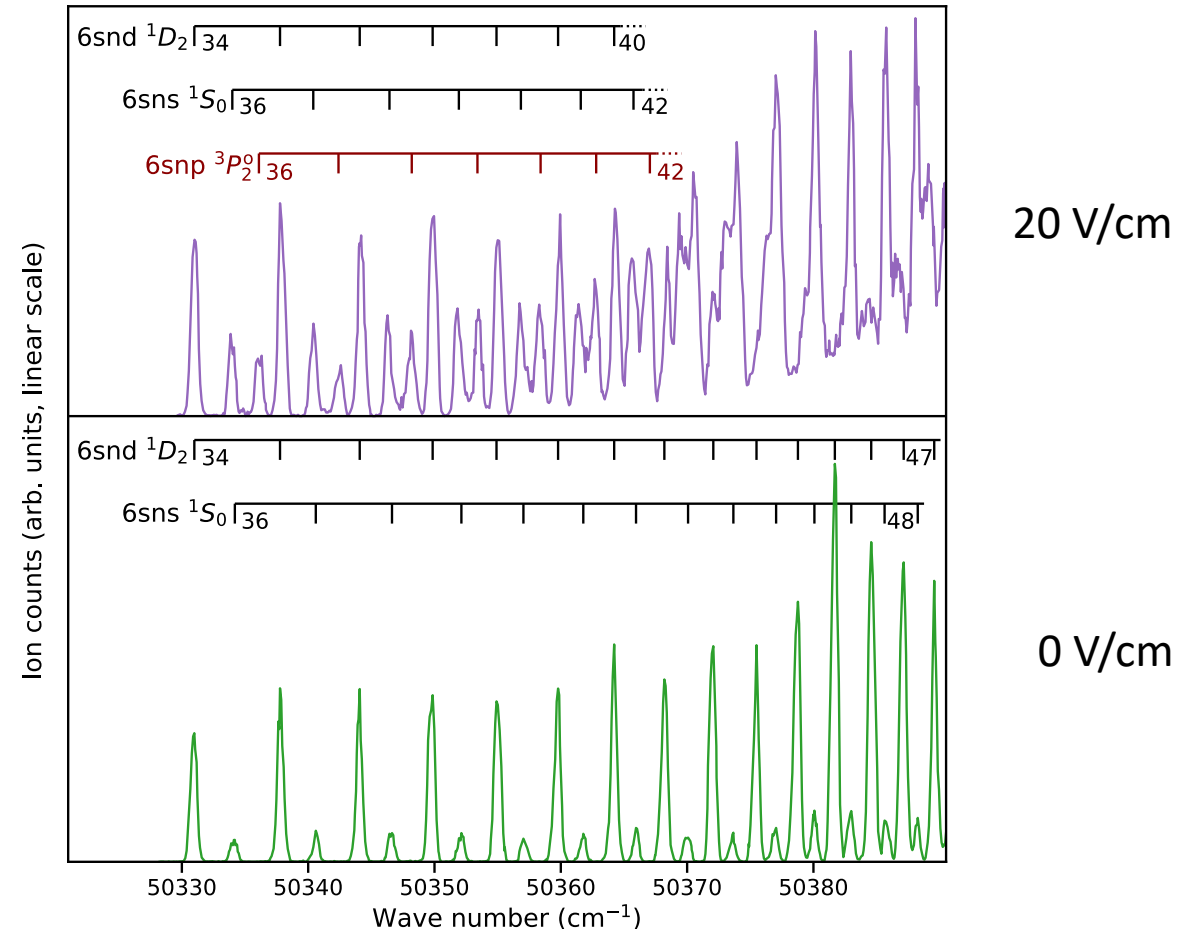
Why Yb?

Precisely known IP

$$\text{IP} = 50\,443.07041(25) \text{ cm}^{-1} \text{ [1]}$$

Yb in electric field

levels shift, broaden and split
 $6snp$ series shows up [2]
due to Stark effect



[1] H. Lehec et al., Phys. Rev. A **98**, 062506, (2018)

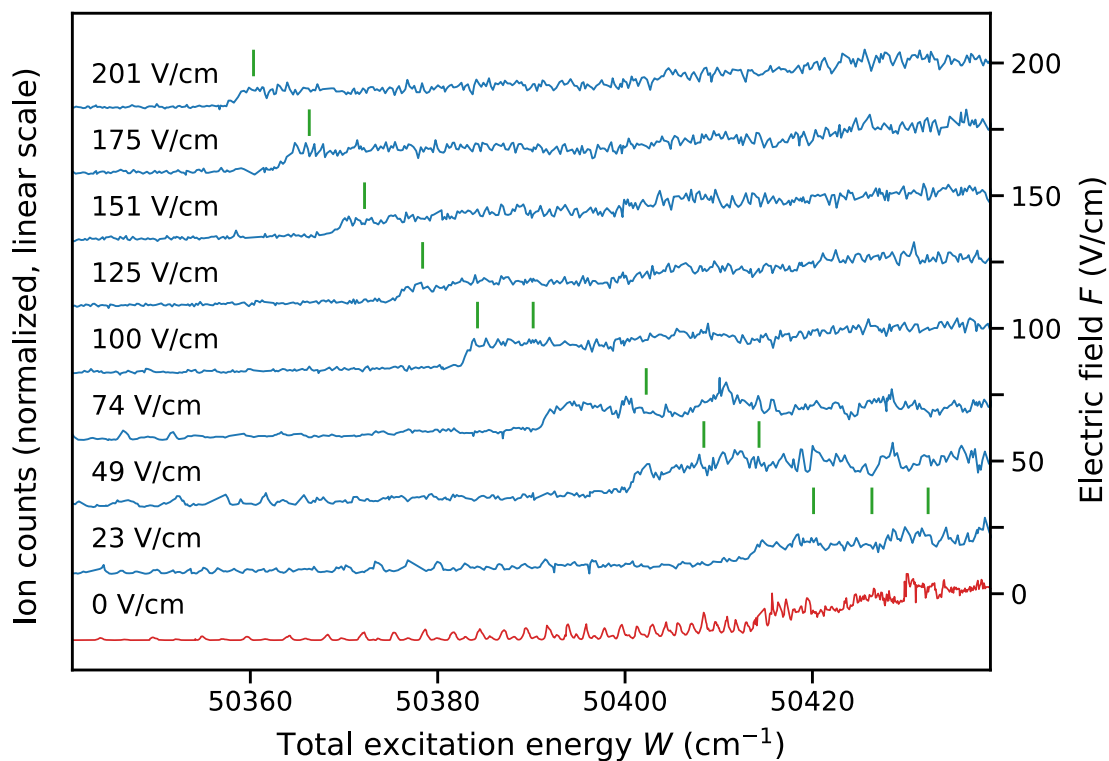
[2] R. Li et al., Spectrochim. Acta B **174** (2020)

[3] M. Kaja et al., NIM B **547**, 165213 (2024)

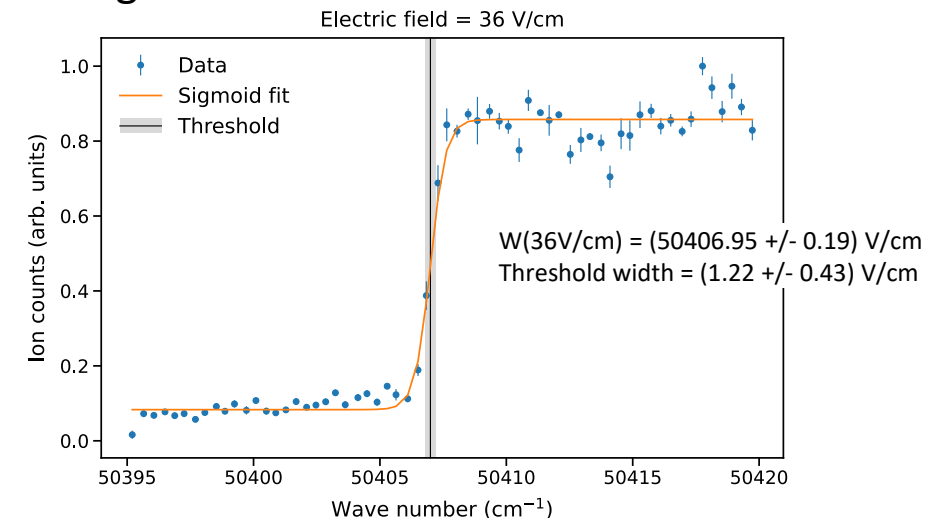
Ytterbium Ionization Potential

Determination of ionization thresholds

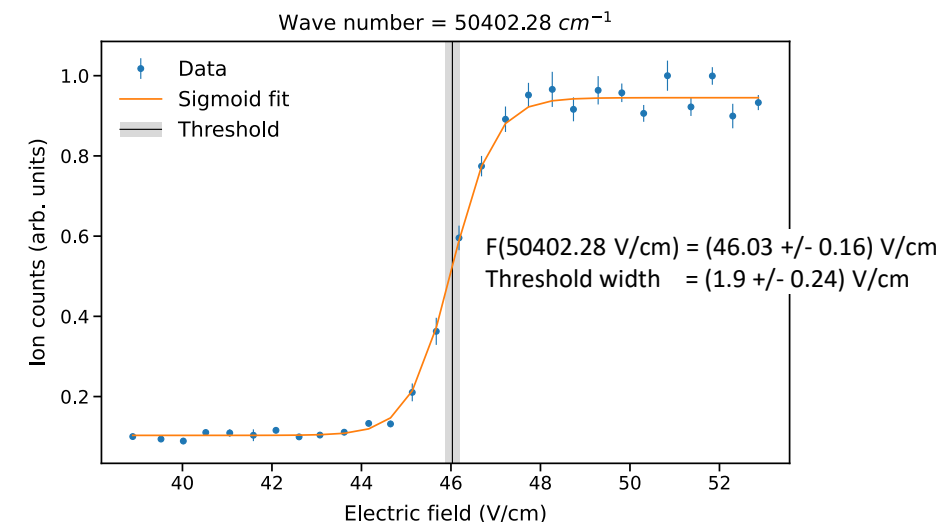
- 28 Wavelength scan for 14 different fixed e. fields
- 24 E. fields scans for 12 different fixed wavelengths



Wavelength scan



Field scan



[1] M. Kaja et al., NIM B **547**, 165213 (2024)

Ytterbium Ionization Potential

$$W_S^F = 50\,443.21(13) \text{ cm}^{-1} - 6.050(7) (\text{Vcm})^{-\frac{1}{2}} \sqrt{F}$$

$$W_S^\lambda = 50\,443.05(15) \text{ cm}^{-1} - 6.034(10) (\text{Vcm})^{-\frac{1}{2}} \sqrt{F}$$

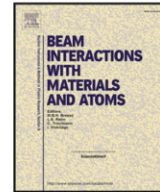
$$\text{IP} = 50\,443.07041(25) \text{ cm}^{-1} [1]$$

Nuclear Instruments and Methods in Physics Research B 547 (2024) 165213

Contents lists available at ScienceDirect

Nuclear Inst. and Methods in Physics Research, B

journal homepage: www.elsevier.com/locate/nimb



Characterization of the field ionization extension for the laser ion source and trap: Measurement of the ionization potential of ytterbium

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^a Institute of Physics, Johannes Gutenberg University Mainz, Germany

^b GSI Helmholtzzentrum für Schwerionenforschung GmbH, Darmstadt, Germany

^c Helmholtz Institute Mainz, Germany

^d STI group, SY Department, CERN, Switzerland

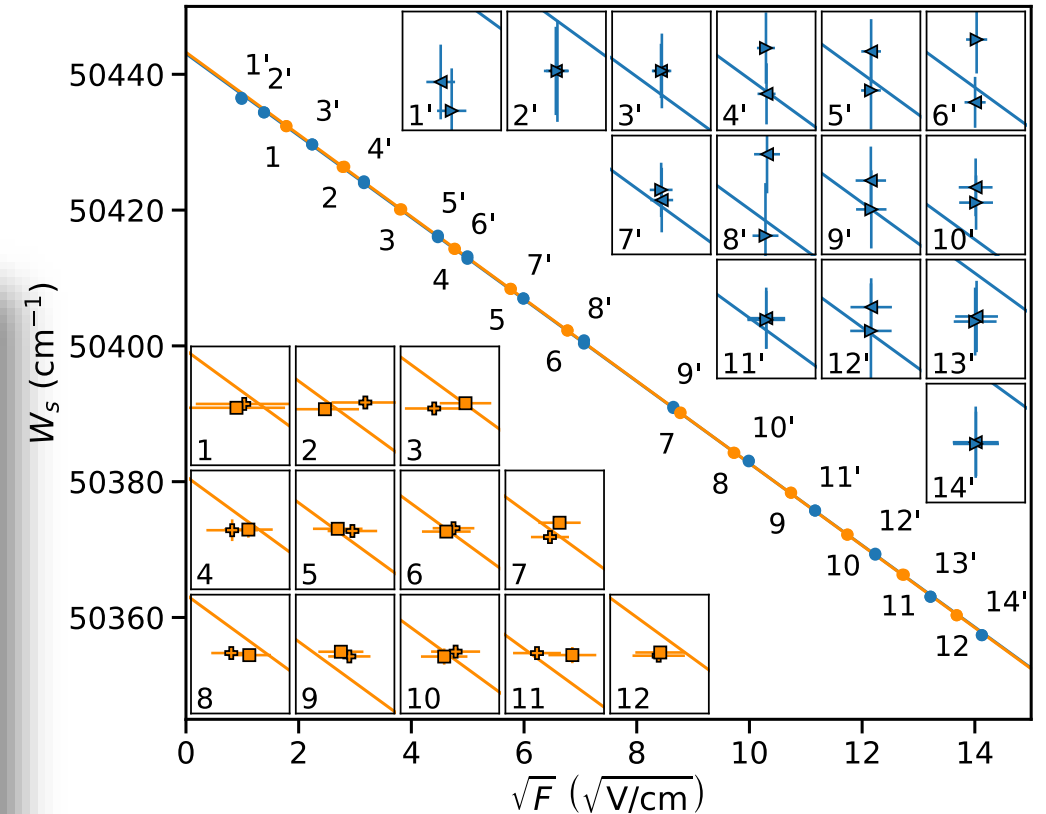
ARTICLE INFO

Keywords:

Resonance ionization laser ion source
Laser spectroscopy
Electric field ionization

ABSTRACT

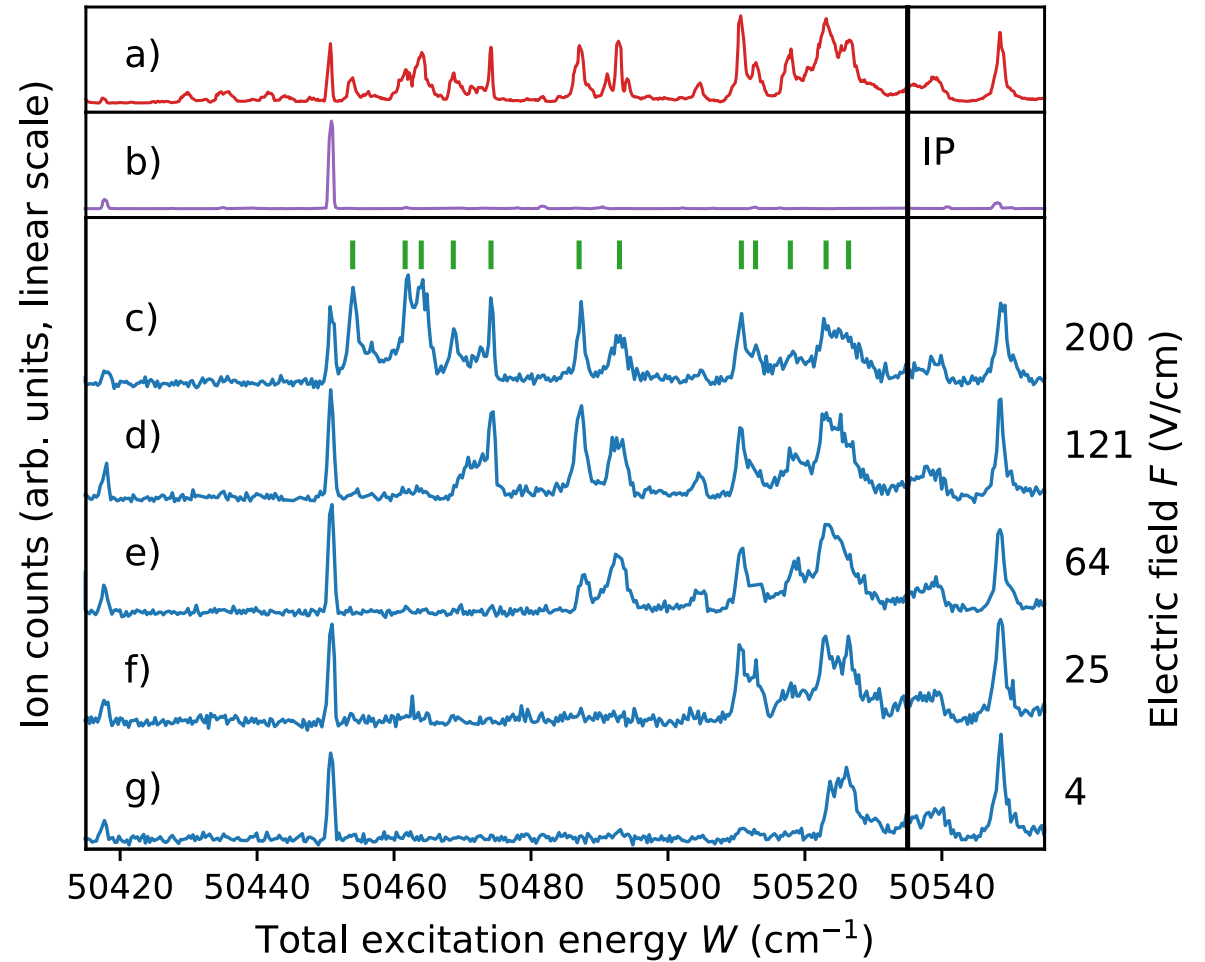
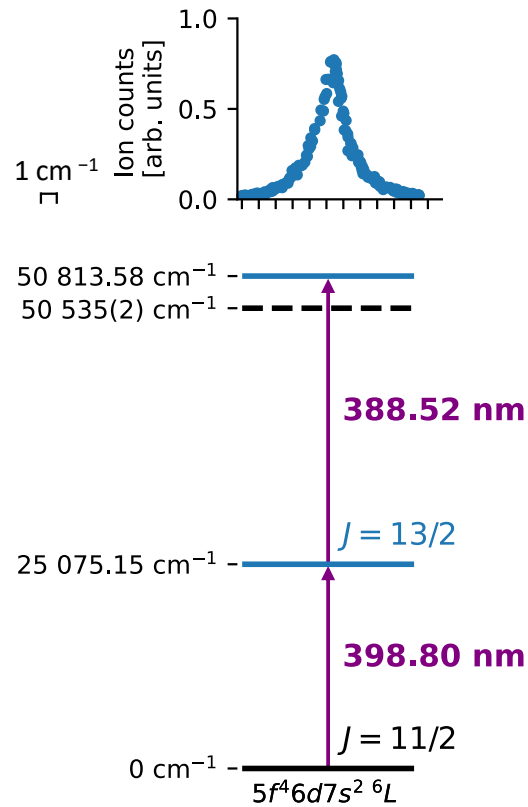
We report on the development and application of the Field Ionization Laser Ion Source and Trap (FI-LIST) at the RISIKO mass separator at Mainz University. The FI-LIST is an adaptation of the well-established LIST unit developed at Mainz and successfully adapted to CERN-ISOLDE. It is specifically designed for field ionization of highly excited atoms in a homogeneous electric field. To assess the performance of the device for future use on radioactive species of e.g. actinides, we conducted ionization potential (IP) measurements on ytterbium, for which the IP is precisely known. The IP value was derived by applying the saddle-point model with a relative precision of $3 \cdot 10^{-6}$ and in very good agreement with the literature value.



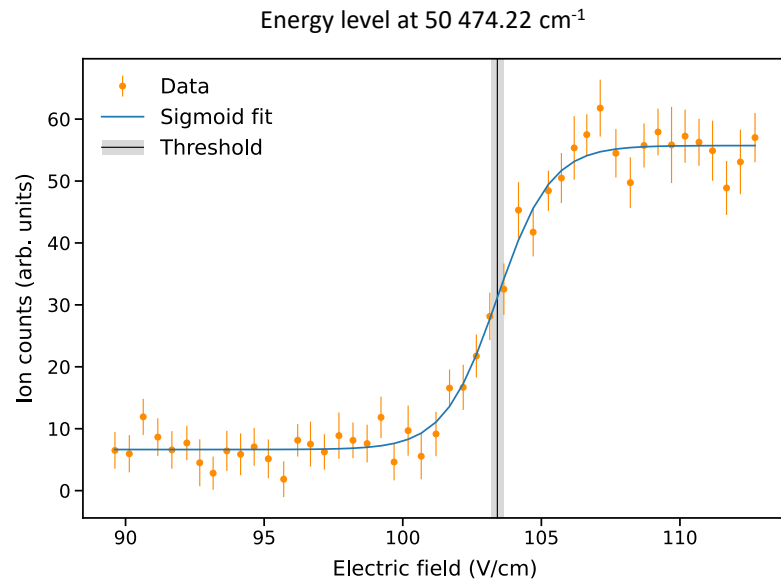
- [1] H. Lehec et al., Phys. Rev. A **98**, 062506, (2018)
[2] M. Kaja et al., NIM B **547**, 165213 (2024)

Ionization potential

[1] M. Kaja et al., Eur. Phys. J. D **78**, 50 (2024)



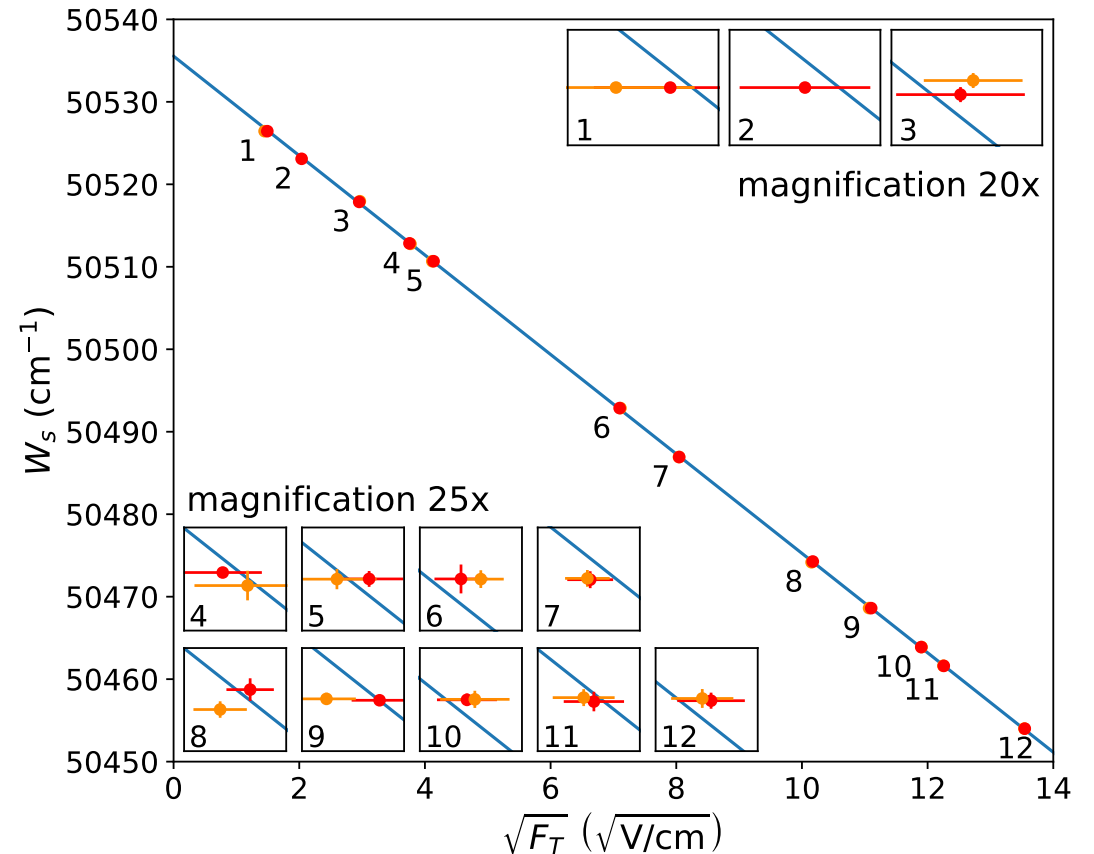
Ionization potential



$$W_s^F = 50\,535.54(15) \text{ cm}^{-1} - 6.029(9) (\text{Vcm})^{-\frac{1}{2}} \sqrt{F}$$

$$\text{IP} = 50\,535(2) \text{ cm}^{-1} [1]$$

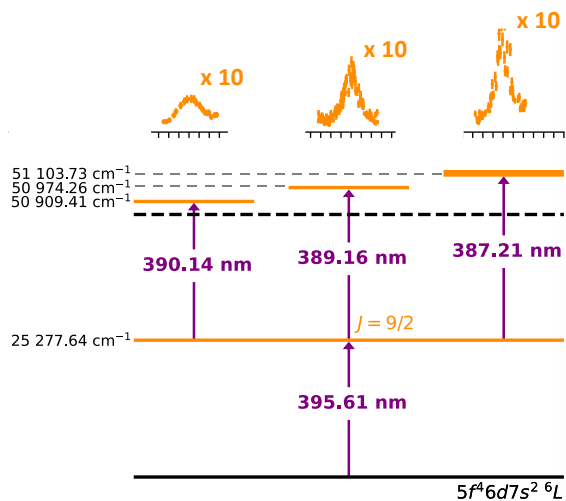
More than 10 times higher precision



[1] S. Köhler et al., Spectrochim. Acta B **52** (6) (1997)
 [2] M. Kaja et al., Eur. Phys. J. D **78**, 50 (2024)

Ionization potential

Scheme investigation



Ionization potential

$$IP = 50\,535.54(15) \text{ cm}^{-1}$$

Eur. Phys. J. D (2024) 78:50
<https://doi.org/10.1140/epjd/s10053-024-00833-7>

THE EUROPEAN
 PHYSICAL JOURNAL D



Regular Article - Atomic Physics

Resonant laser ionization of neptunium: investigation on excitation schemes and the first ionization potential

Magdalena Kaja^{1,a}, Dominik Studer^{2,3}, Felix Berg⁴, Sebastian Berndt^{1,4}, Christoph E. Düllmann^{2,3,4}, Nina Kneip¹, Tobias Reich⁴, Mitzi Urquiza-González^{5,6}, and Klaus Wendt¹

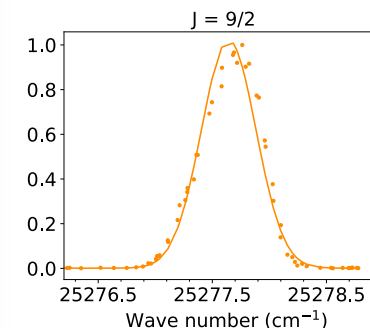
- ¹ Institute of Physics, Johannes Gutenberg University Mainz, 55099 Mainz, Germany
- ² GSI Helmholtzzentrum für Schwerionenforschung GmbH, 64291 Darmstadt, Germany
- ³ Helmholtz Institute Mainz, 55099 Mainz, Germany
- ⁴ Department of Chemistry - Nuclear Chemistry, Johannes Gutenberg University Mainz, 55099 Mainz, Germany
- ⁵ Division Hübner Photonics, Hübner GmbH & Co. KG, 34123 Kassel, Germany
- ⁶ Department of Physics, University of Gothenburg, 41296 Gothenburg, Sweden

Received 30 January 2024 / Accepted 20 March 2024 / Published online 4 May 2024
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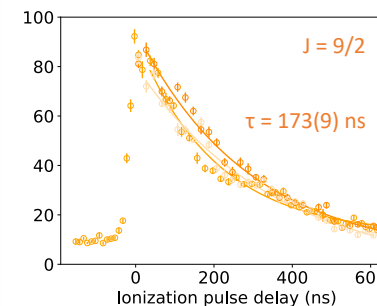
Abstract.

The atomic structure of neptunium (Np) was investigated by two-step resonance ionization spectroscopy. The study involved exploring ground-state transitions as well as following transitions to high-lying states just below the ionization potential (IP) or auto-ionizing states above the IP. That resulted in the identification of two-step ionization schemes, suitable for trace analysis and nuclear structure investigations. The lifetimes of two excited states located at $25,342.48 \text{ cm}^{-1}$ and $25,277.64 \text{ cm}^{-1}$ were determined as $230(12) \text{ ns}$ and $173(9) \text{ ns}$, respectively. Because of the absence of Rydberg series in wide-ranging spectra recorded, the first IP was determined through the field ionization of high-lying, weakly-bound states using a well-controlled static electric field. By applying the saddle-point model, an IP value of $50,535.54(15) \text{ cm}^{-1}$ [$6.265608(19) \text{ eV}$] was derived. This value agrees with the current literature value of $50,535(2) \text{ cm}^{-1}$, while providing a more than ten times higher precision.

Line profiles



Lifetimes measurements



High-resolution spectroscopy

Hyperfine structure measurements
Extraction of nuclear moments

Np 237

$2.14 \cdot 10^6$ a

α 4.790...;

γ 29...; e⁻

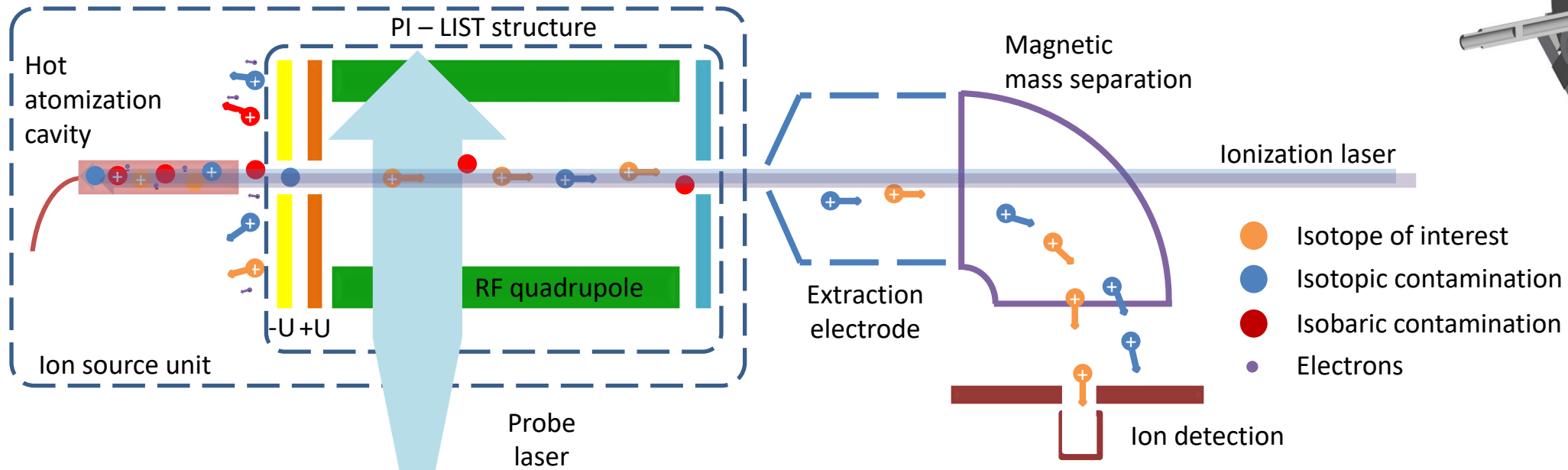
Np 239

2.355 d

β -0.4...

γ 106...; e⁻; g

PI-LIST - Perpendicularly Illuminated LIST

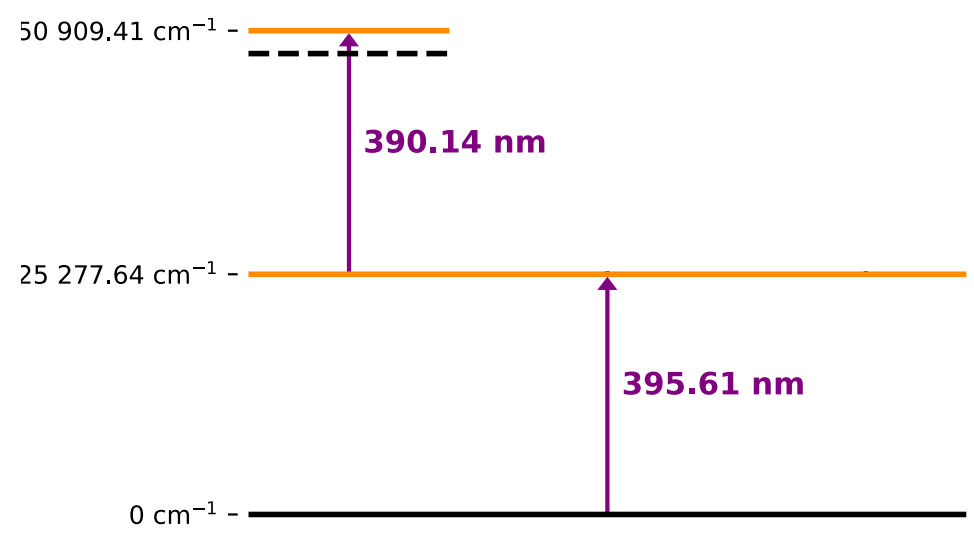
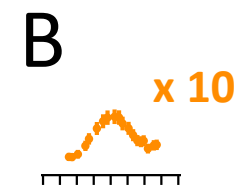
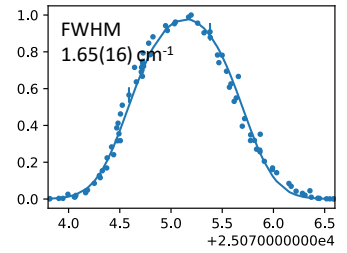
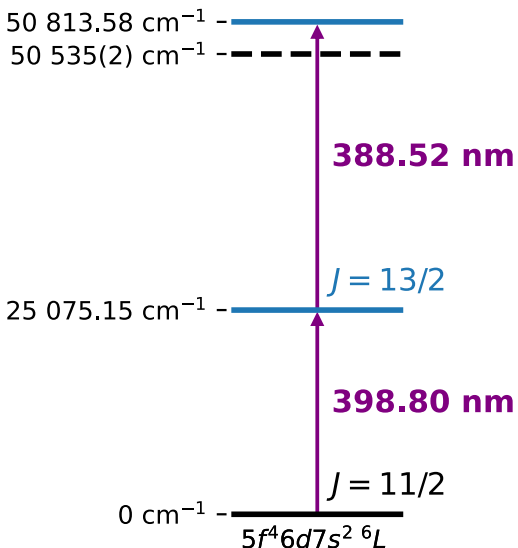
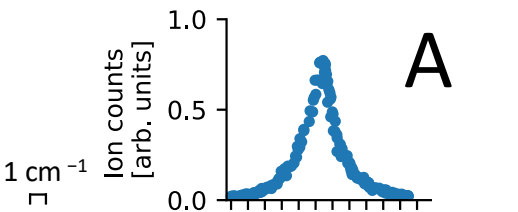


- **Separation** of hot cavity evaporation from laser ionization volume
- **Suppression** of surface ionized species
- **Pulsed laser ionzation** inside RF quadrupole structure
- **Reduced Doppler width** in laser transversal interaction

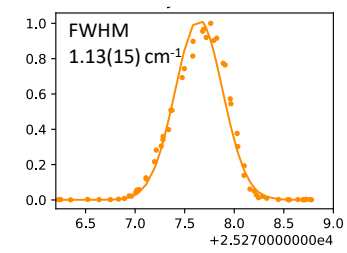
[1] R. Heinke et al., *Hyperfine Interact* **238**, 6 (2017)
 [2] R. Heinke, ... , M. Kaja et. al., *NIM B*, **541**, 8-12, (2023)

Ionization scheme development

[1] M. Kaja et al., Eur. Phys. J. D **78**, 50 (2024)



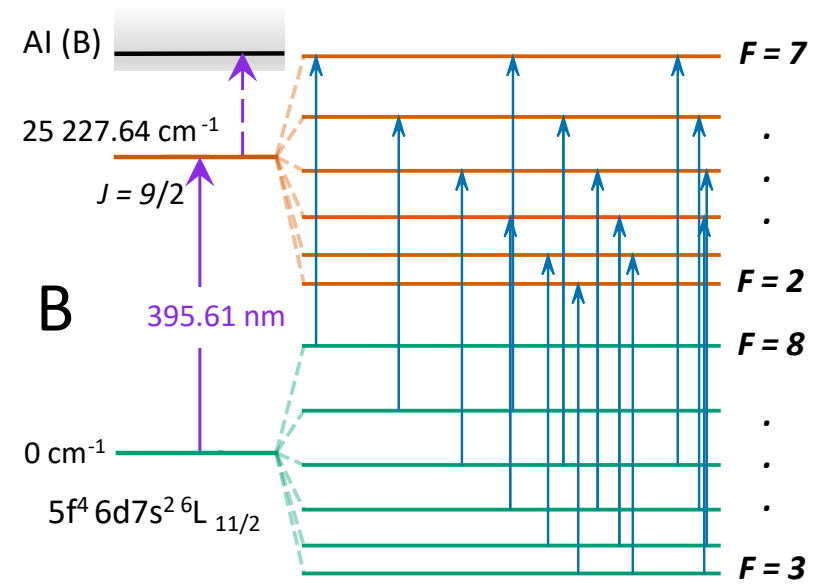
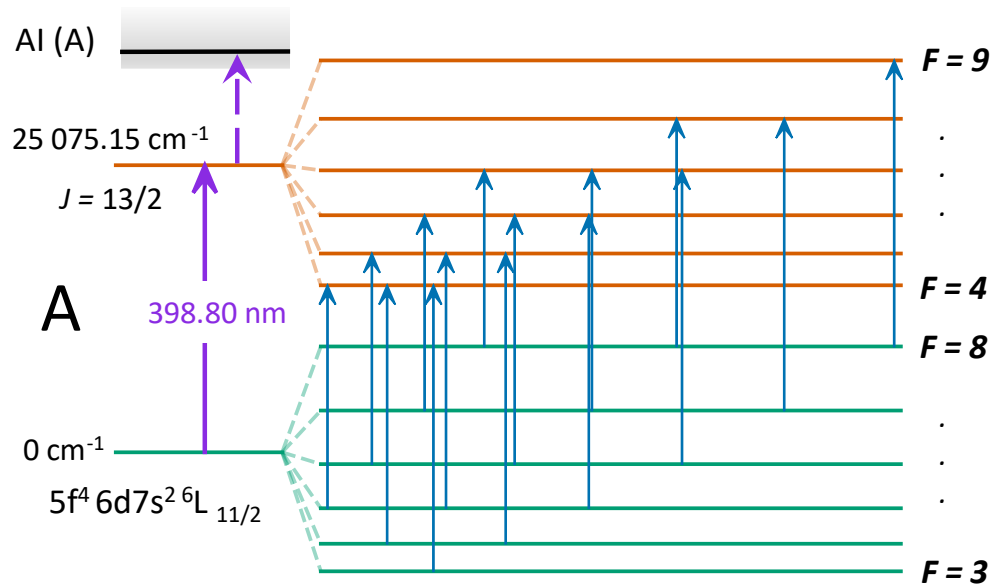
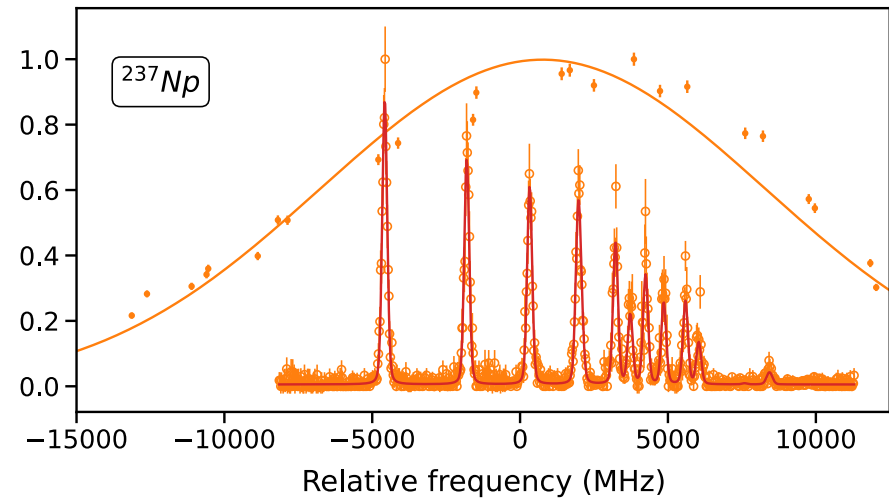
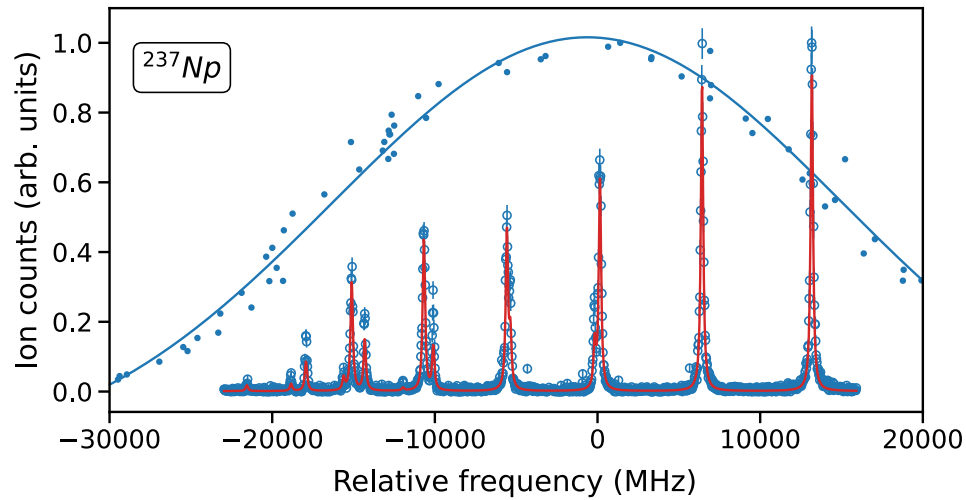
FES line profiles



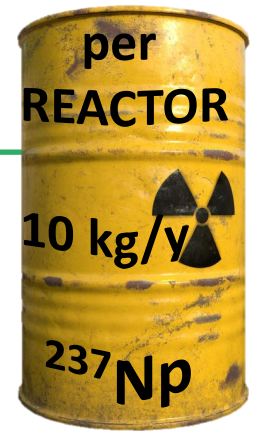
All FESs:
S. Raeder et al.,
Spectrochim. Acta B,
66, 242 – 247 (2011)

Hyperfine spectroscopy in ^{237}Np ($I = 5/2$)

[1] M. Kaja et al., Eur. Phys. J. A **60**, 140 (2024)



Neptunium production and trace analysis



Pu 237 45.2 d α 5.499; 5.456... sf; Si; Mg; γ; e ⁻	Pu 238 87.74 a α 5.157; 5.144... sf; γ; e ⁻ ; m	Pu 239 2.44·10 ⁵ a α 5.168; 5.124... sf; γ; e ⁻ ; g	Pu 240 6545 a β ⁻ ; g; α 4.896... γ (149); e ⁻	Pu 241 14.35 a
Np 236 22.5 h e ⁻ ; β ⁻ ; e ⁻ ; γ (642...); g	Np 237 2.14·10 ⁶ a α 4.790...; 29...; e ⁻	Np 238 2.102 d β ⁻ 1.2... γ 984...; e ⁻ ; g	Np 239 2.355 d β ⁻ 0.4... γ 106...; e ⁻ ; g	Np 240 7.22 m 65 m β ⁻ 2.2...; β ⁻ 0.9 γ; e ⁻ ; h _γ ; g; γ; g
U 235 0.7200 26 m h _γ (0,07) e ⁻	U 236 7.038·10 ⁸ a 120 ns h _γ 1783; 642... sf	U 237 2·10 ⁷ a 6.75 d β ⁻ 0.2... γ 60; 208...; e ⁻	U 238 99.2745 0.3 μs h _γ 2514; 1879... sf	U 239 4.468·10 ⁹ a 23.5 m β ⁻ 1.2; 1.3...; γ 75; 44...

→ neutron capture
← β⁻ decay

Sample:

- 10¹³ atoms of ²³⁷Np
- 10¹¹ atoms of ²³⁹Np

²³⁹Np production:

- irradiation of ²³⁸U at
- reactor TRIGA Mark II Mainz

²³⁷Np – stock solution

Trace analysis of environmental samples is of high relevance

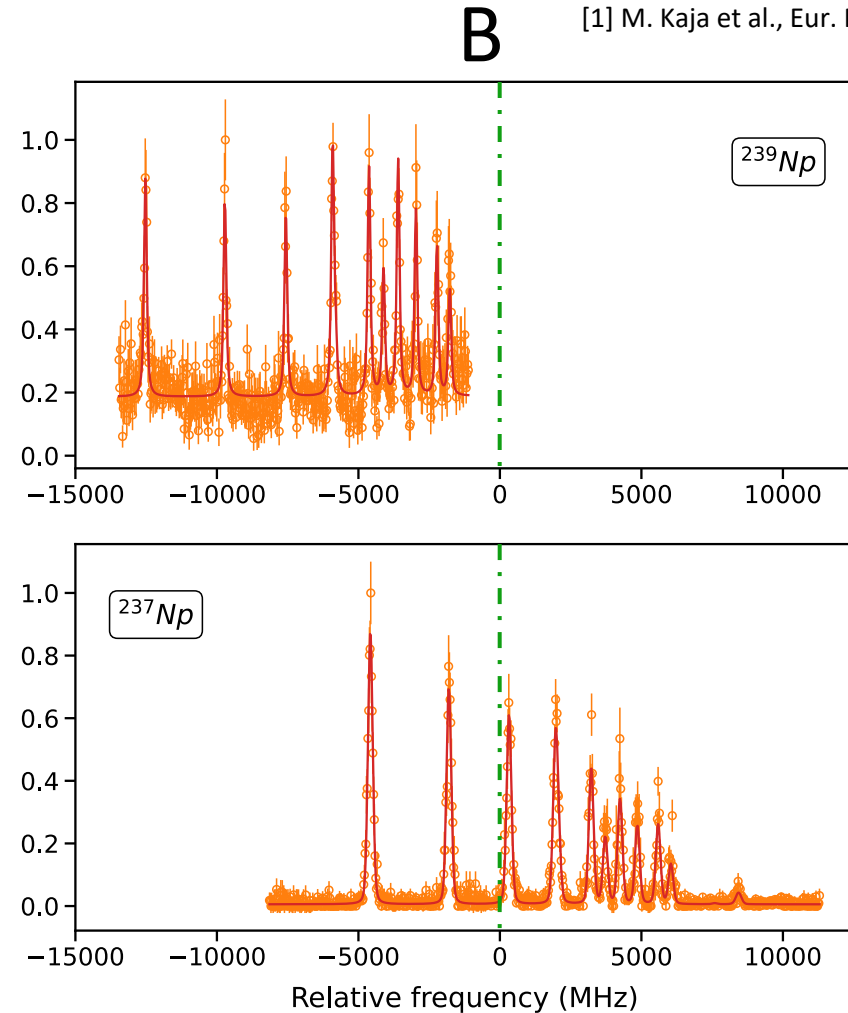
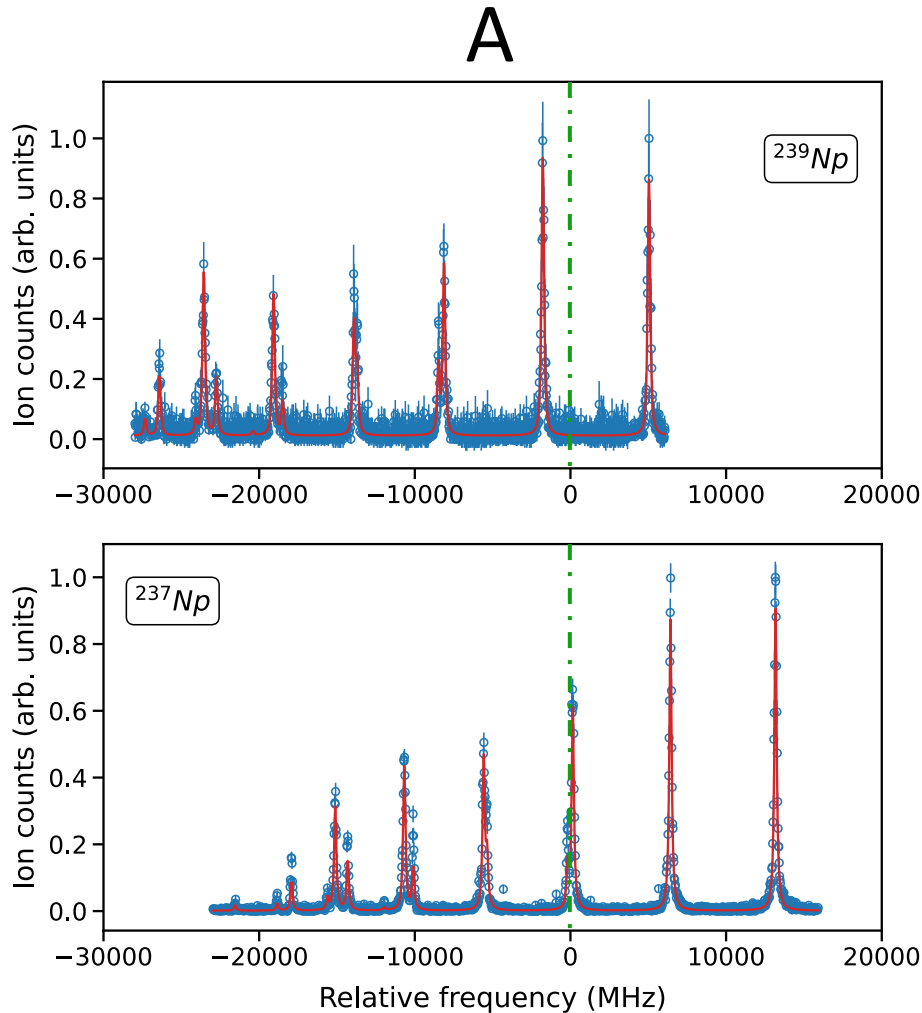
- ²³⁹Np as a tracer for precise quantification
- Identification of efficient ionization schemes:
 - High elemental selectivity
 - Isotope related effects like hyperfine structures (HFS) and isotope shift (IS)
 - large splitting and shifts observed in neptunium

Hyperfine spectroscopy in ^{237}Np and ^{239}Np

[1] M. Kaja et al., Eur. Phys. J. A **60**, 140 (2024)

Np 239
 2.355 d
 β -0.4...
 γ 106...; e; g
 $I = 5/2$

Np 237
 2.14·10⁶ a
 α 4.790...;
 γ 29...; e
 $I = 5/2$



$$\delta\nu_i^{A,A'} = \nu_i^{A'} - \nu_i^A$$

$$\delta\nu_{399\text{ nm}}^{237,239} = -8\,168(17)\text{ MHz}$$

$$\delta\nu_{396\text{ nm}}^{237,239} = -7\,892(13)\text{ MHz}$$

Hyperfine parameters and nuclear moments

State	E (cm ⁻¹)	J	²³⁷ Np		
			\mathcal{A}_{exp}	\mathcal{B}_{exp}	$\mathcal{A}_{lit}^{[2]}$
GS	0	11/2	776.08(20)	928(10)	778
FES _A	25 075.15	13/2	1470.09(18)	323(8)	1470
FES _B	25 277.64	9/2	570.08(22)	-302(9)	-

Determination of nuclear moments

$$\frac{A}{A_{ref}} = \frac{I_{ref}}{I} \frac{\mu_I}{\mu_{I,ref}} \quad \frac{B}{B_{ref}} = \frac{Q_s}{Q_{s,ref}}$$

measured literature

Isotope	I	μ_I [μ_N]	Q_s [eb]
²³⁷ Np	5/2	+3.14(4) [3]	+3.886(6) [4]
²³⁹ Np	5/2	+3.18(4)	+4.05(2)

reference

The European Physical Journal

volume 60 · number 7 · july · 2024

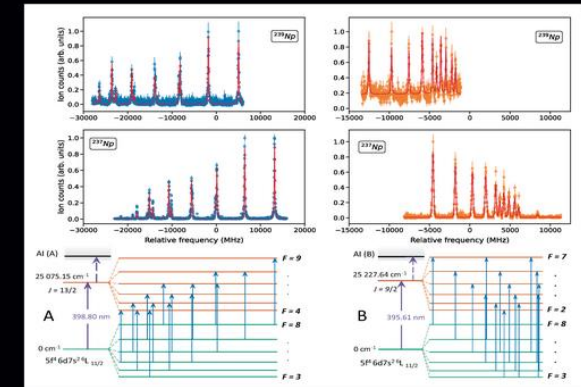
EPJ A



Recognized by European Physical Society

Hadrons and Nuclei

High-resolution laser spectroscopy on the hyperfine structure and isotope shift of ^{237,239}Np by Magdalena Kaja et al.



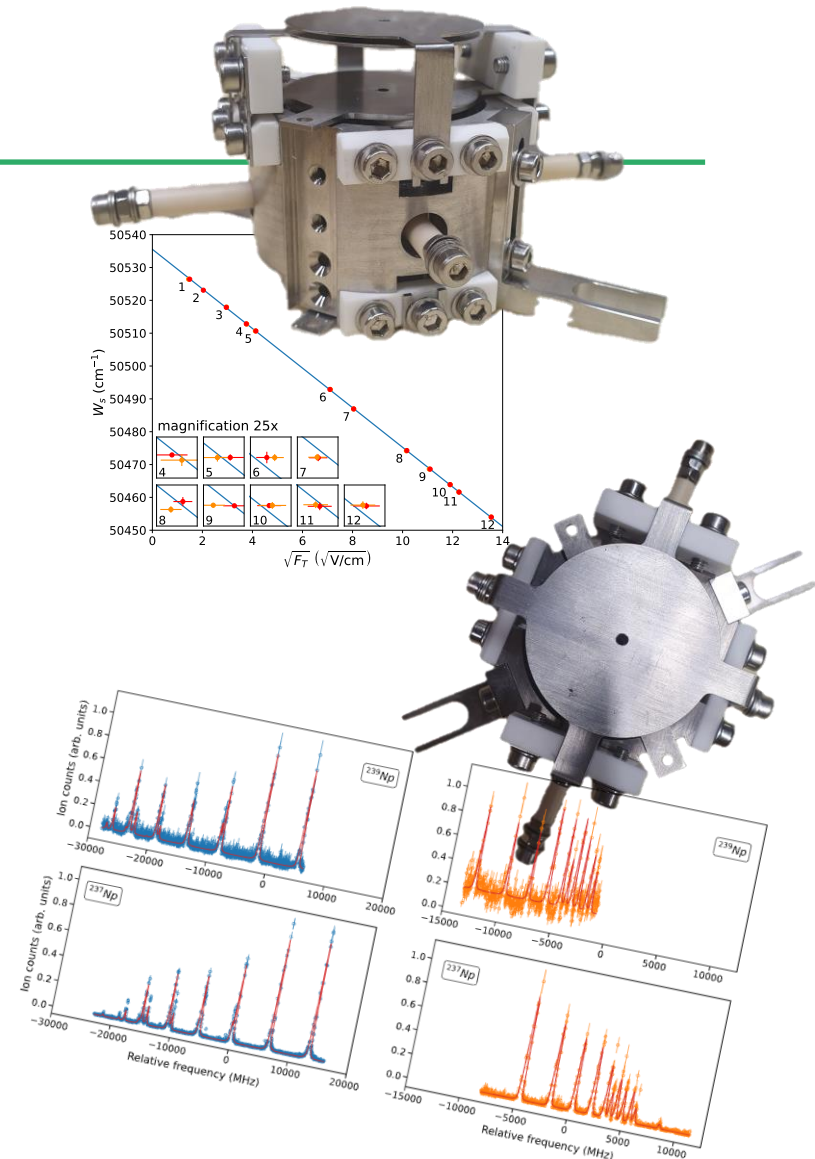
The hyperfine spectra of ^{237,239}Np isotopes in the ground state transition of scheme (A) at 399 nm on the left side and scheme (B) at 396 nm on the right. The hyperfine splitting of the atomic ground and respective first excited state are given below indicating the permitted hyperfine transitions by blue arrows.

Conclusion

- Atomic structure investigations of neptunium
 - Ionization schemes, atomic energy levels, lifetimes
 - Precise IP determination of $50\,535.54(15)\text{ cm}^{-1}$
 - *M. Kaja et al., Eur. Phys. J. D 78, 50 (2024)*
- The FI-LIST – a new useful type of the PI-LIST
 - Dedicated tool for IP studies on rare elements with complex structure
 - Suitable off-line as well as on-line applications
 - Capable of measurements at very low electric fields of $\sim 1\text{ V/cm}$
 - *M. Kaja et al., NIM B 547, 165213 (2024)*
- High resolution spectroscopy in ^{237}Np and ^{239}Np
 - Isotope shifts
 - Nuclear moments for ^{239}Np
 - *M. Kaja et al., Eur. Phys. J. A 60, 140 (2024)*

Outlook

- Off- and on-line IP determination of other elements with complex spectra, e.g. Fm
- Extension of the high-resolution spectroscopy to other isotopes of neptunium

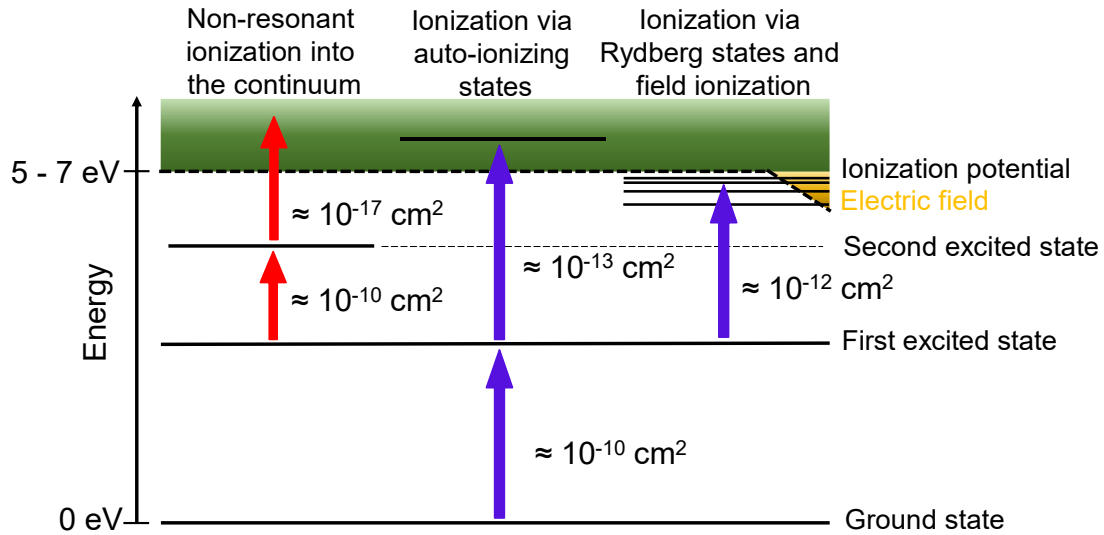




Thank you!



Technique: Resonance ionization mass spectrometry



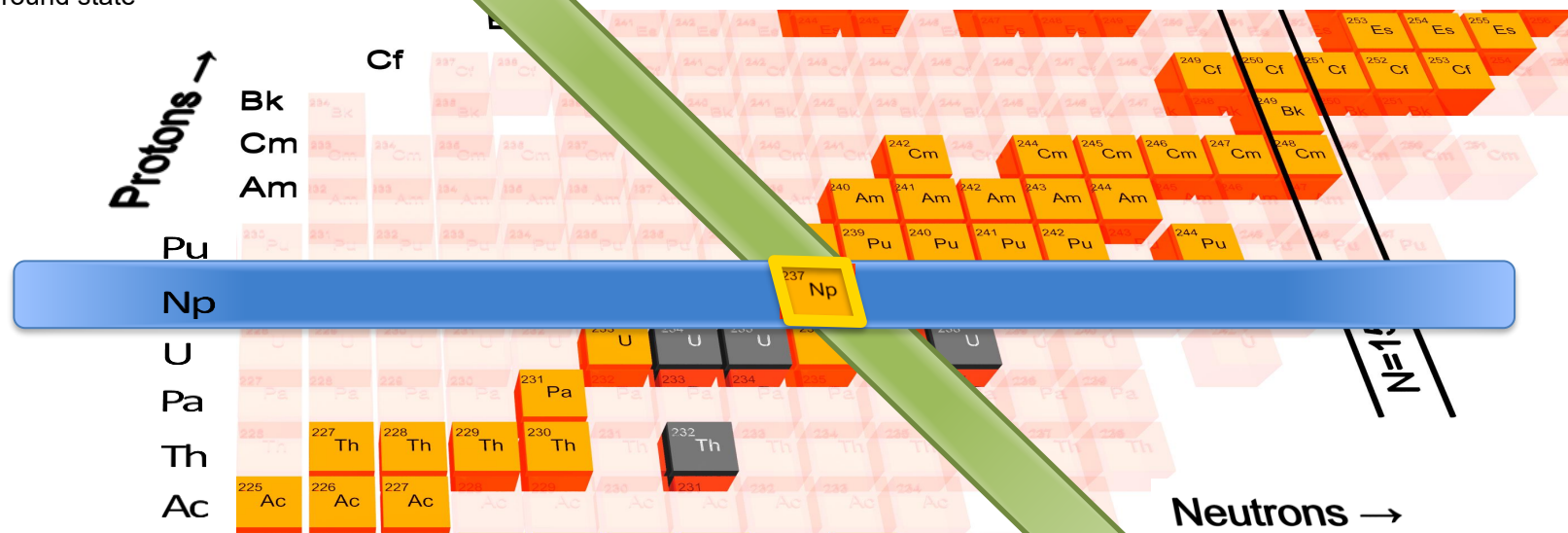
Mass spectrometry

- Separation of m/q
- High detection efficiency

Laser ionization and spectroscopy

- Highly efficient process
- High element selectivity

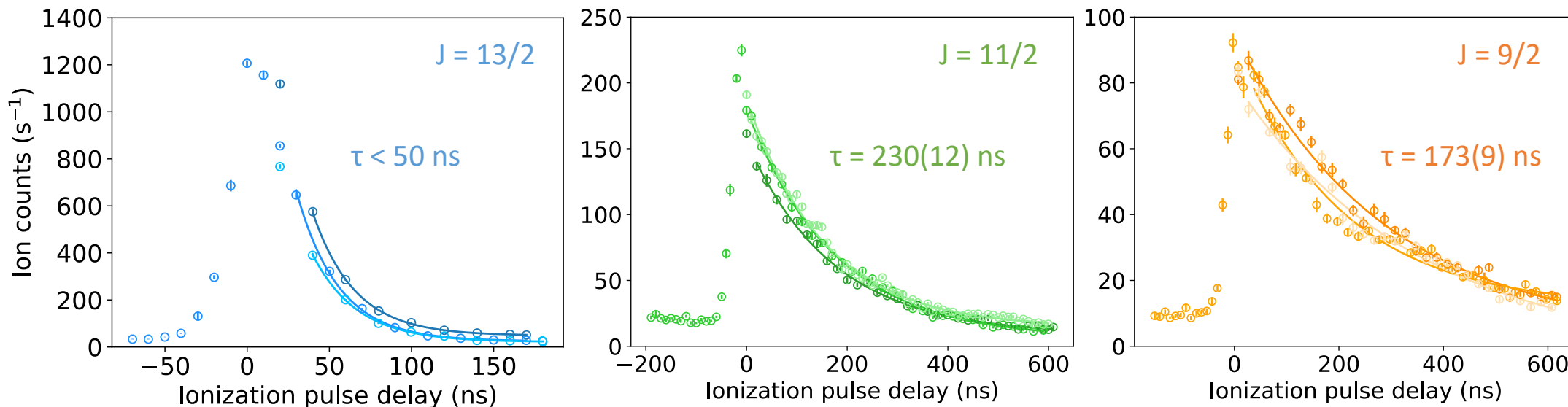
*Indicated isotopes:
long lived, studied by laser spectroscopy
offline at RISIKO/MABU*



Lifetime investigations of first excited states

[1] M. Kaja et al., Eur. Phys. J. D **78**, 50 (2024)

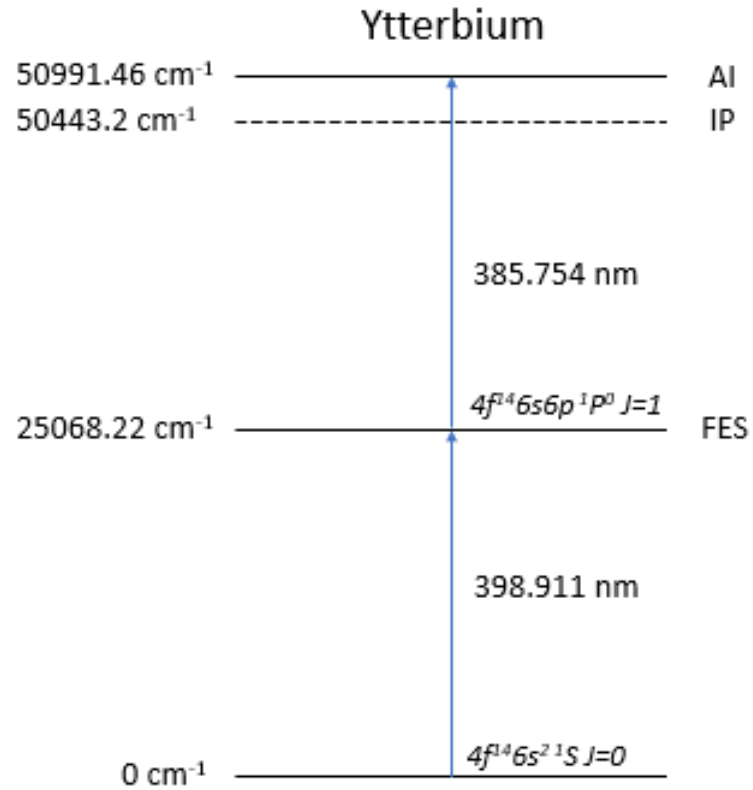
The excited-state population decay as a function of the ionization-pulse delay.



Population development in the "dark" time between pulses corresponds to an exponential distribution.

This method is applicable to lifetimes much longer than the laser-pulse duration (≈ 50 ns) and much shorter than the collision lifetime of the excited atoms within the laser beam in the atomizer tube ($\approx 3 \mu$ s)

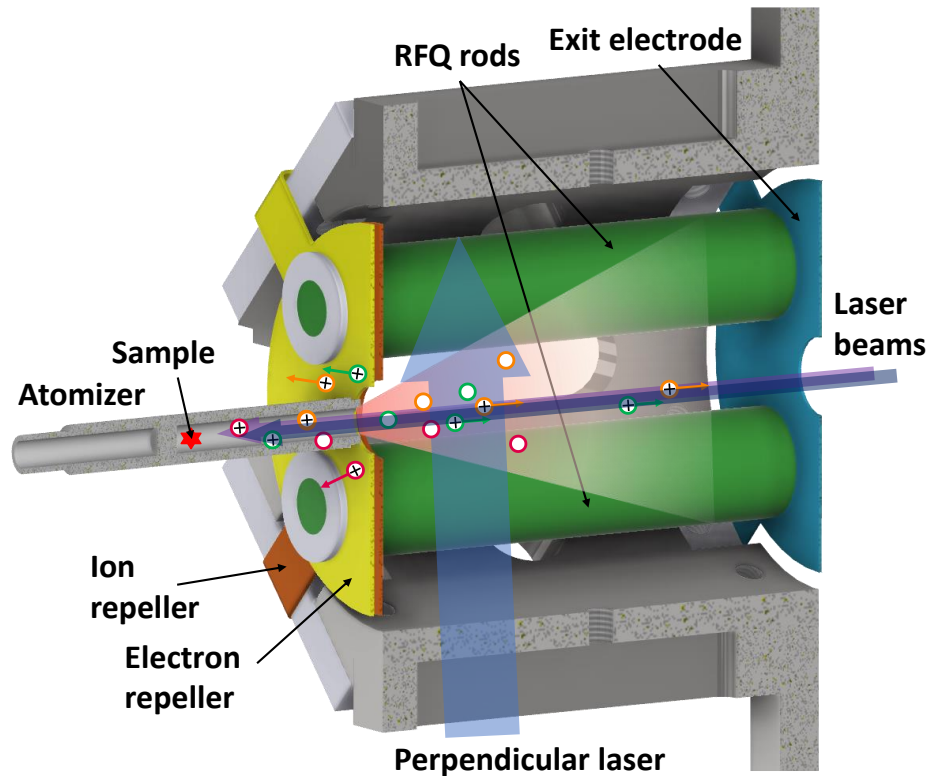
Ytterbium



Laser Ion Source and Trap (LIST)

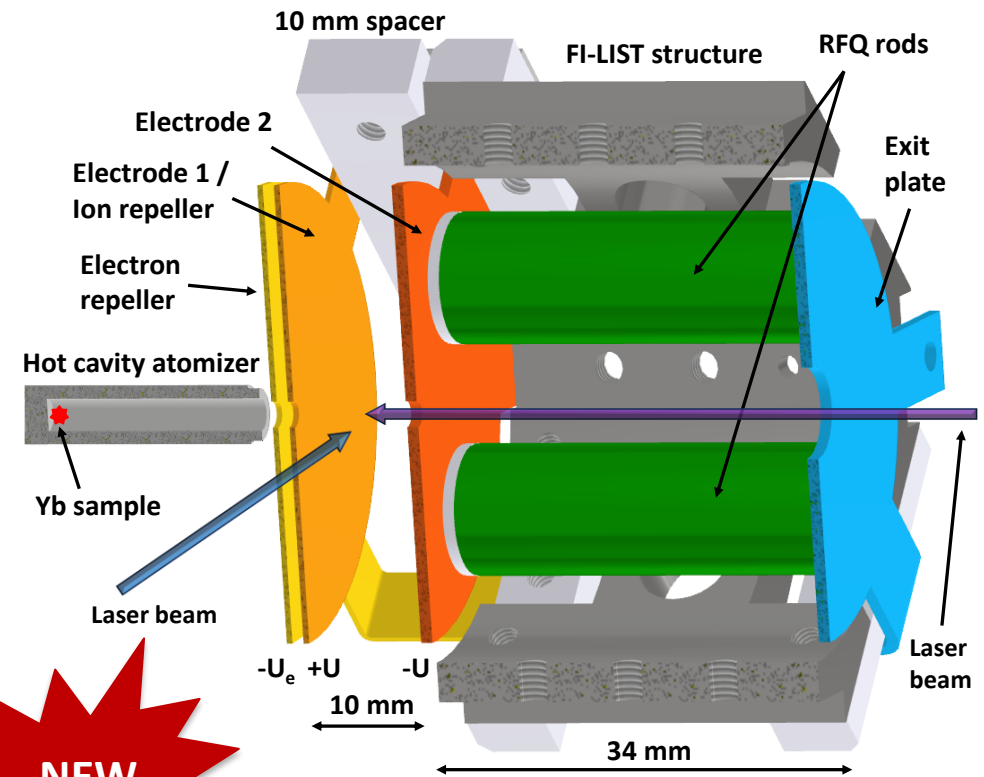
Perpendicularly Illuminated LIST

PI-LIST



Field Ionization LIST

FI-LIST

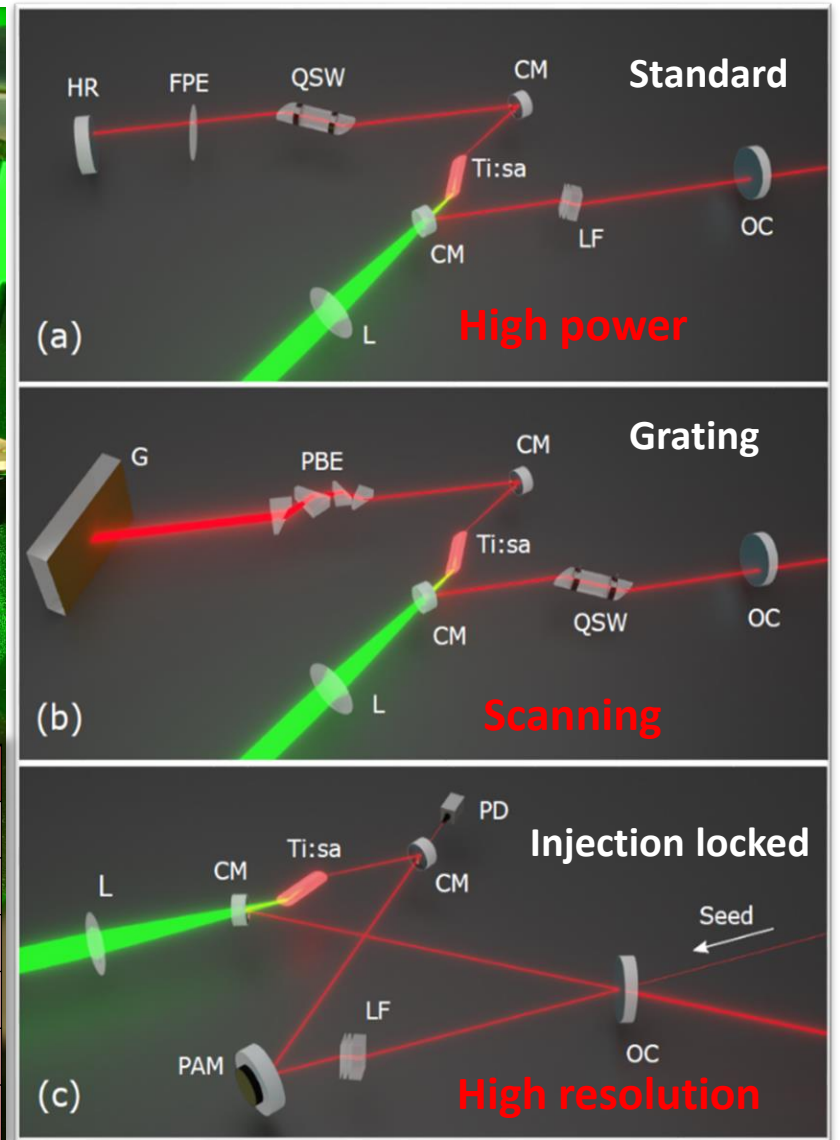


NEW

Ti:Sapphire lasers system



	Standard	Grating-tuned	Injection-seeded
Repetition rate	7 to 15 kHz		
Pulse width	40 to 60 ns		
Average power	3 to 5 W	1 to 2 W	3 to 5 W
Output range	700 to 1020 nm		seed source
Tuning range	100 GHz	700 to 1020 nm	10 to 20 GHz
Spectral bandwidth	1 to 10 GHz	1 to 3 GHz	20 MHz



Neptunium

- Radioactive actinide
- Long half-life - ^{237}Np - $2.14 \cdot 10^6$ y
- High radiotoxicity



Neptunium production

Pu 237 45.2 d	Pu 238 87.74 a α 5.499; 5.456... sf; Si; Mg; γ ; e	Pu 239 $2.44 \cdot 10^5$ a α 5.157; 5.144... sf; γ ; e; m	Pu 240 6545 a α 5.168; 5.124... sf; γ ; e; g	Pu 241 14.35 a β^- ; g; α 4.896... γ (149); e
Np 236 22.5 h ϵ ; β^- ; e γ (642...); g	Np 237 $2.14 \cdot 10^6$ a α 4.790...; γ 29...; e	Np 238 2.102 d β^- 1.2... γ 984...; e; g	Np 239 2.355 d β^- 0.4... γ 106...; e; g	Np 240 7.22 m 65 m β^- 2.2...; β^- 0.9 γ ; e; γ ; g; γ ; g
U 235 0.7200 26 m β^- (0.07) e	U 236 $7.038 \cdot 10^8$ a 120 ns α 4.398...; sf Ne; γ 186 γ 1783; 642... sf	U 237 6.75 d β^- 0.2... γ 60; 208...; e	U 238 99.2745 0.3 μ s $4.468 \cdot 10^9$ a β^- 2514; 1879...sf α 4.198...; sf 2 β^- ; γ (50); e	U 239 23.5 m β^- 1.2; 1.3...; γ 75; 44...

neutron capture
 β^- decay

The development of efficient and selective laser ionization schemes plays an important role for Np spectroscopy and trace analysis.

It is important to take into account the isotope-related effects in ionization schemes coming from hyperfine structure (HFS) and isotope shift (IS).

- ^{239}Np
 - Moments unknown

[1] Kohler, S; et al. Spectrochim. Acta B,52, 717 – 726, (1997)
 [2] Kazakov, V; et al. Phys. Scr., 92, 10, (2017)
 [3] Stone, N.J. At. Data Nucl. Data Tables, 90, 1, 75-176, (2005)

LARISA LAB

Laser development system 3
including seeding and direct laserdiode pumping

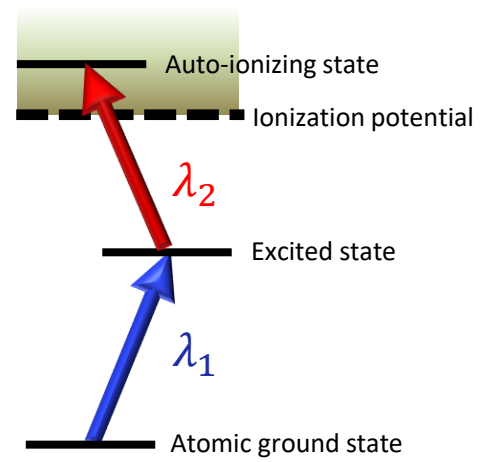
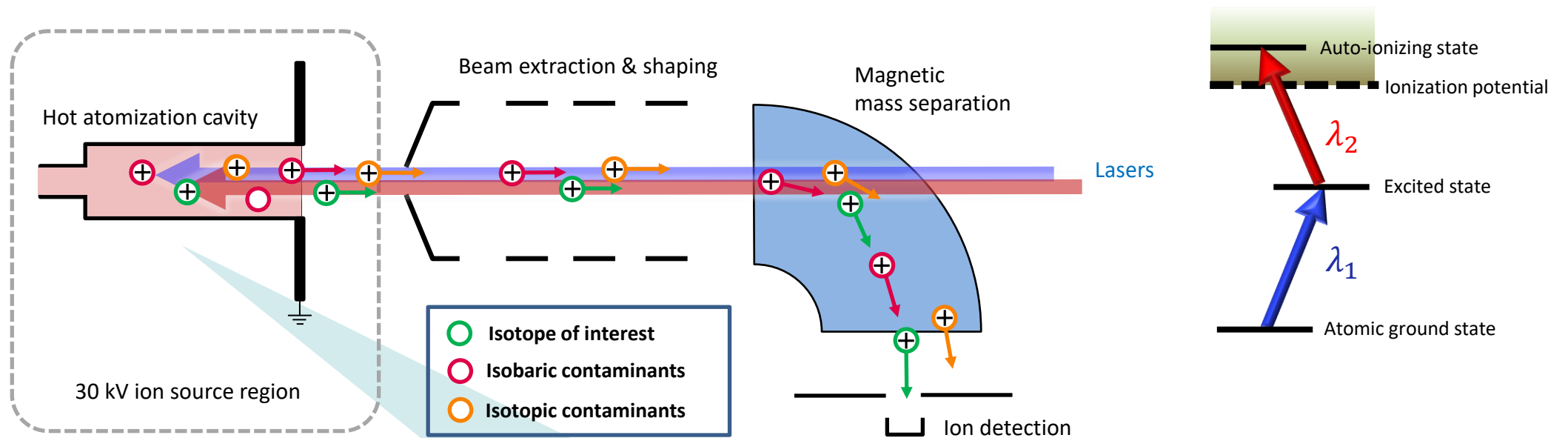
RISIKO –
High voltage sector field MS
for high efficiency, high sensitivity
spectroscopy and ion implantation

Ti:Sa Laser
system 2

Ti:Sa Laser
system 1

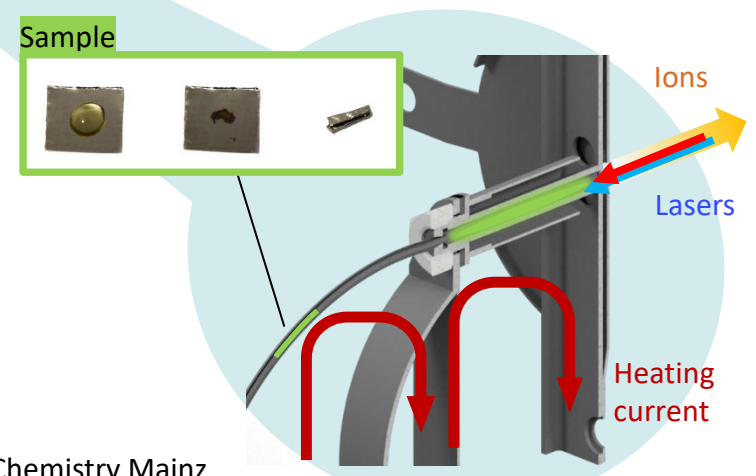
MABU –
low voltage quadrupole MS
for laser spectroscopy

RISIKO mass separator

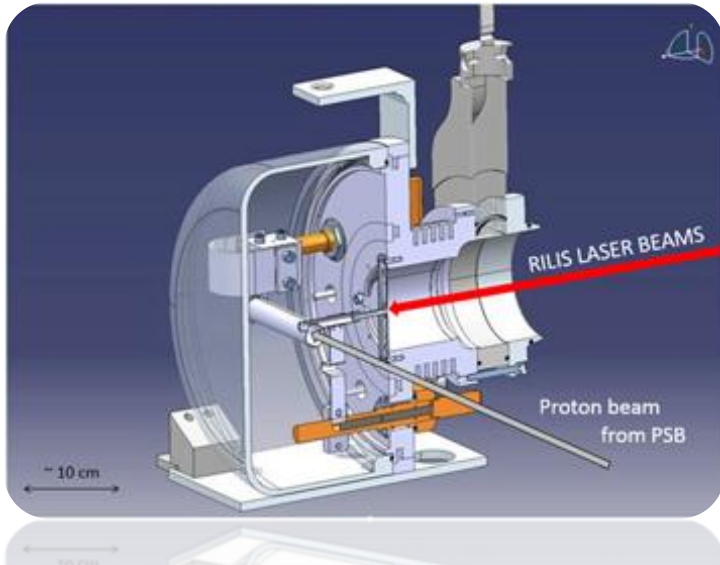


- Chemical sample preparation*
- Sample evaporation in hot cavity
- Multi-step photoionization by pulsed lasers
- Mass separation in dipole magnet $\frac{m}{\Delta m} \approx 800$
- Single ion detection

*in collaboration with Dpt. of Nuclear Chemistry Mainz

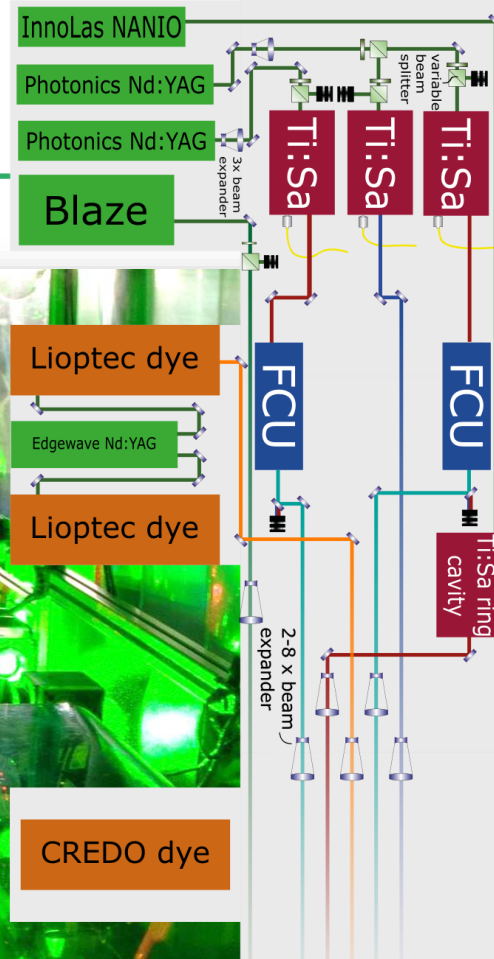
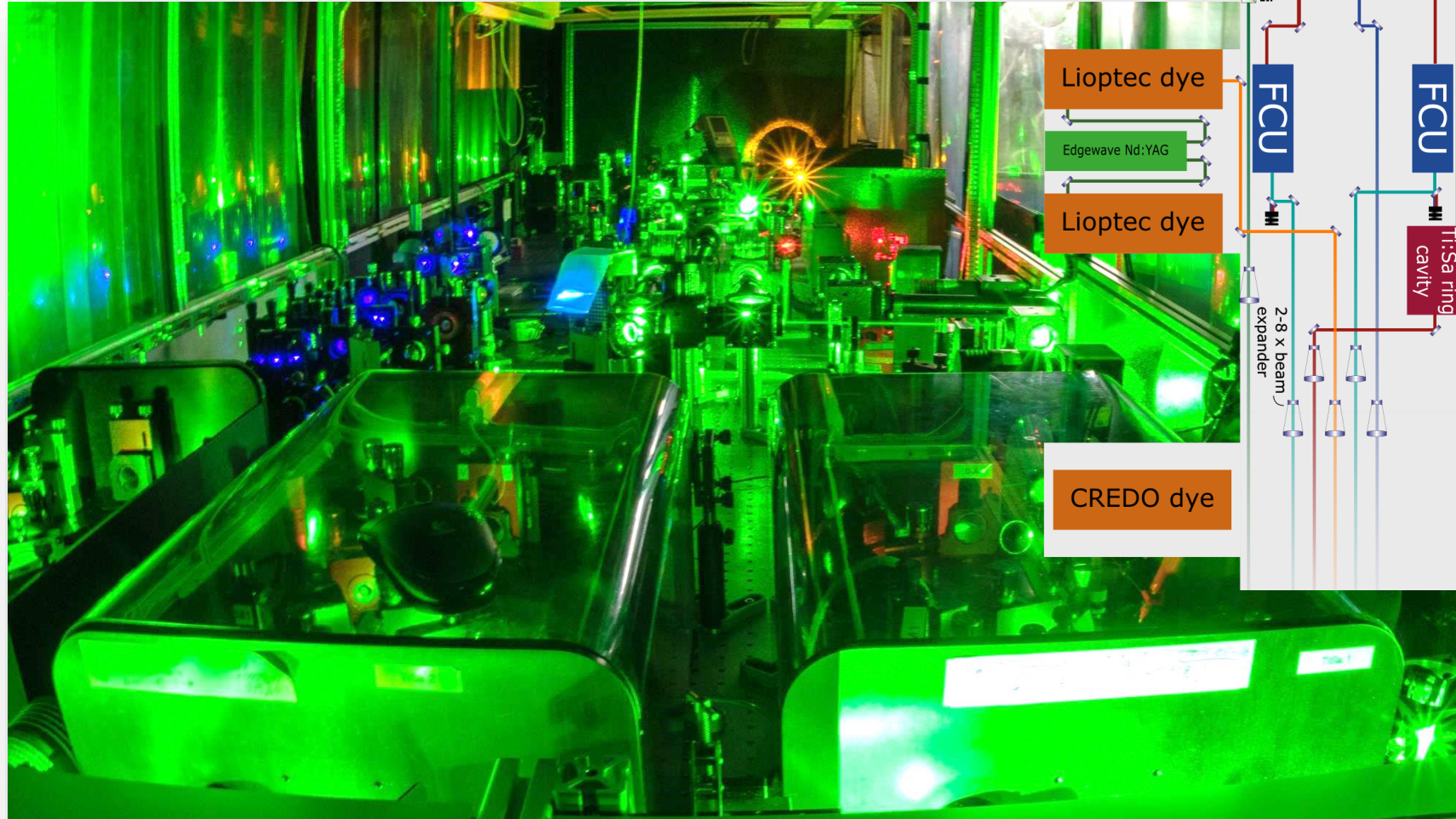
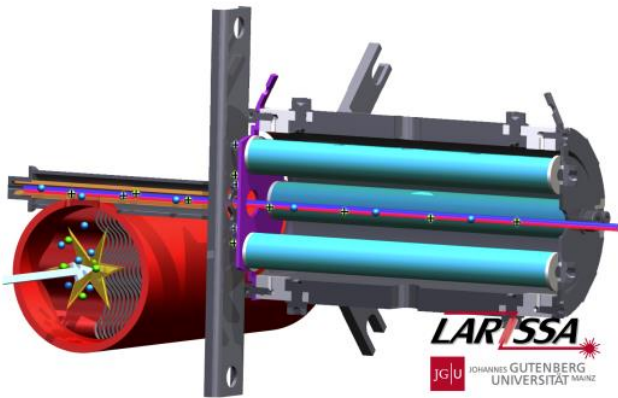


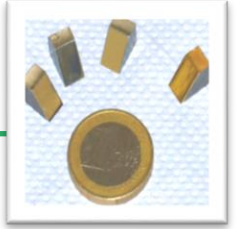
ISOLDE RILIS



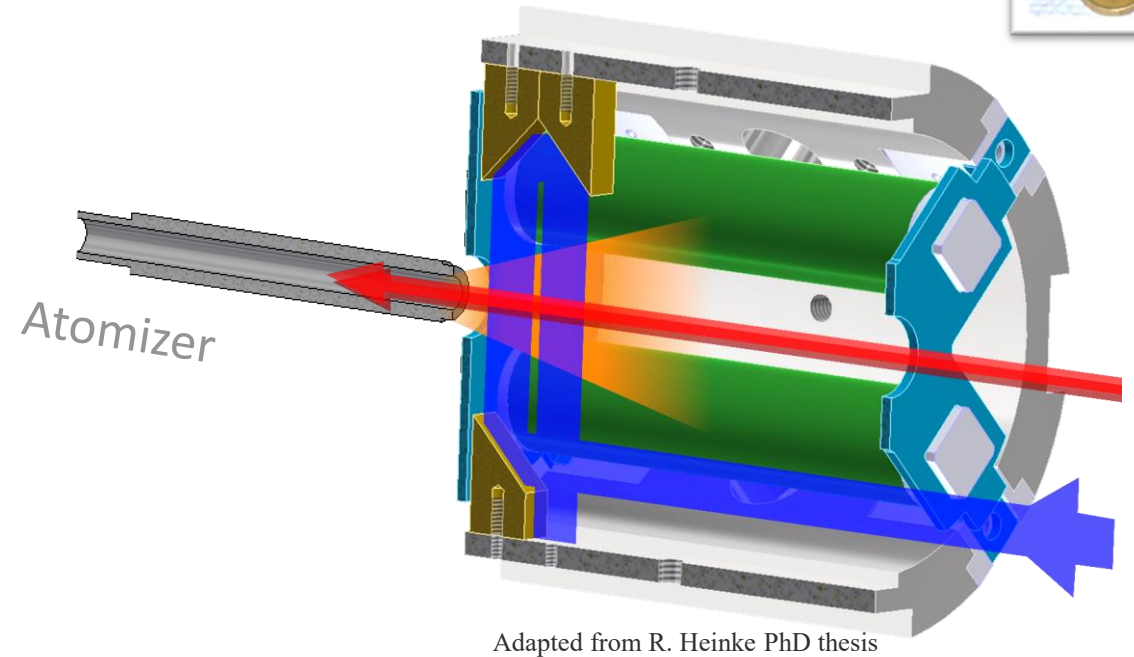
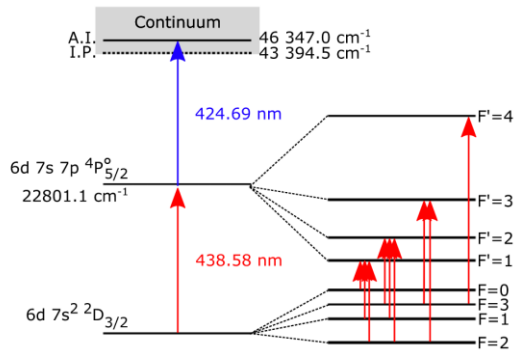
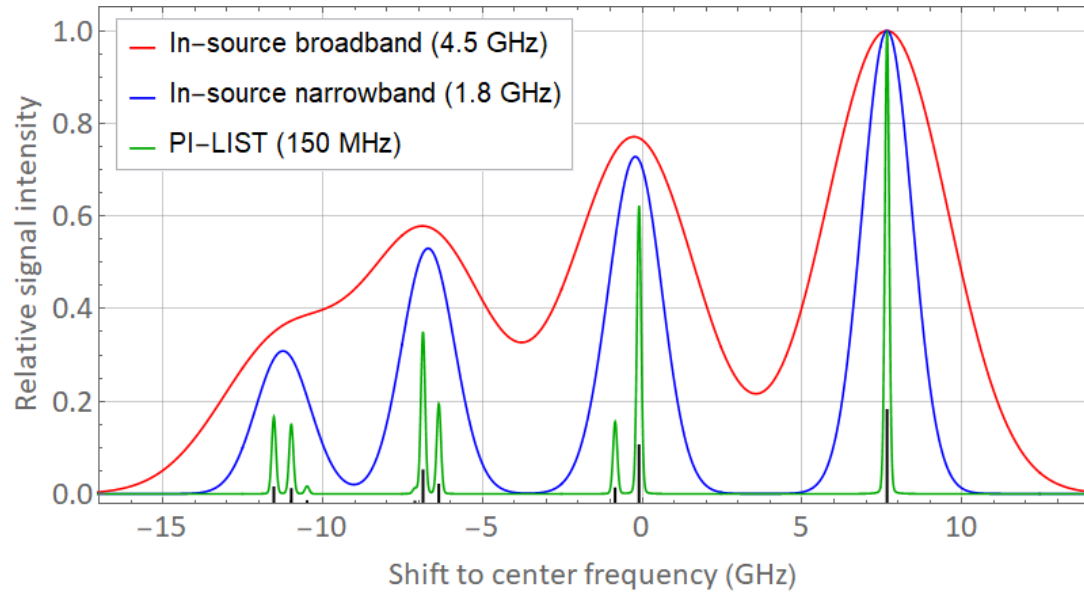
Valentin Fedosseev et al 2017 J. Phys.: G: Nucl. Part. Phys. 44 084006

LIST



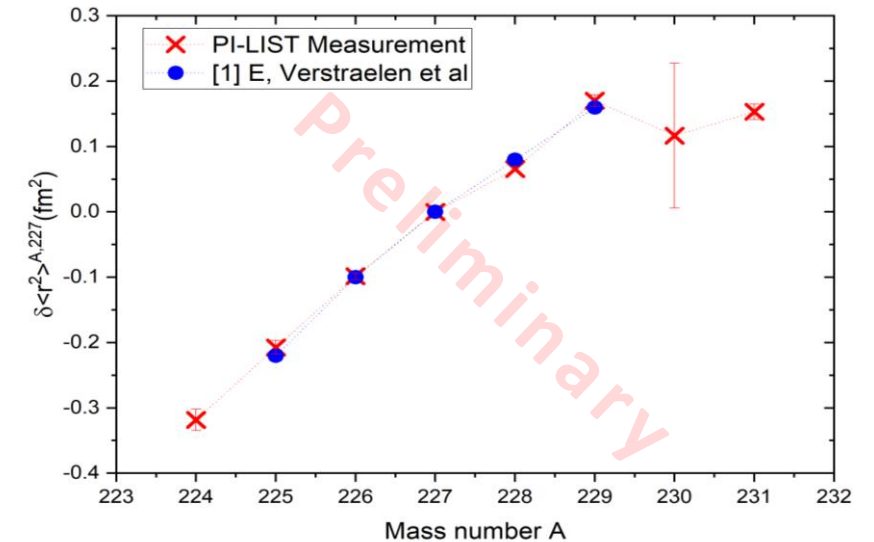
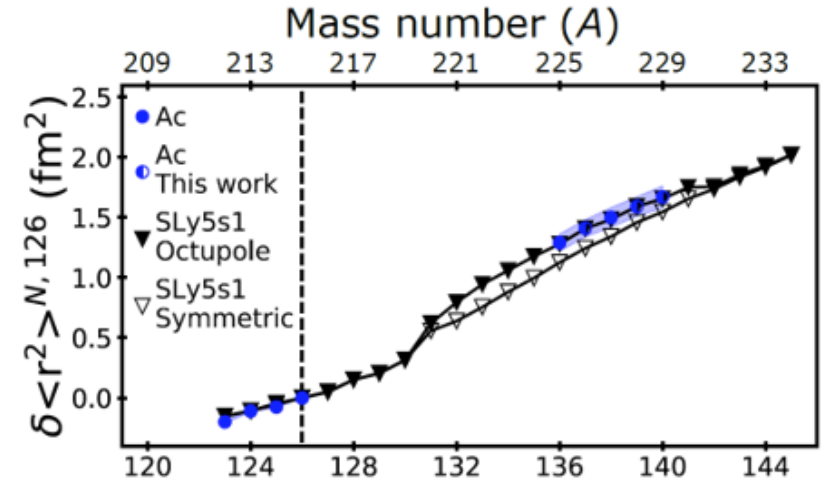
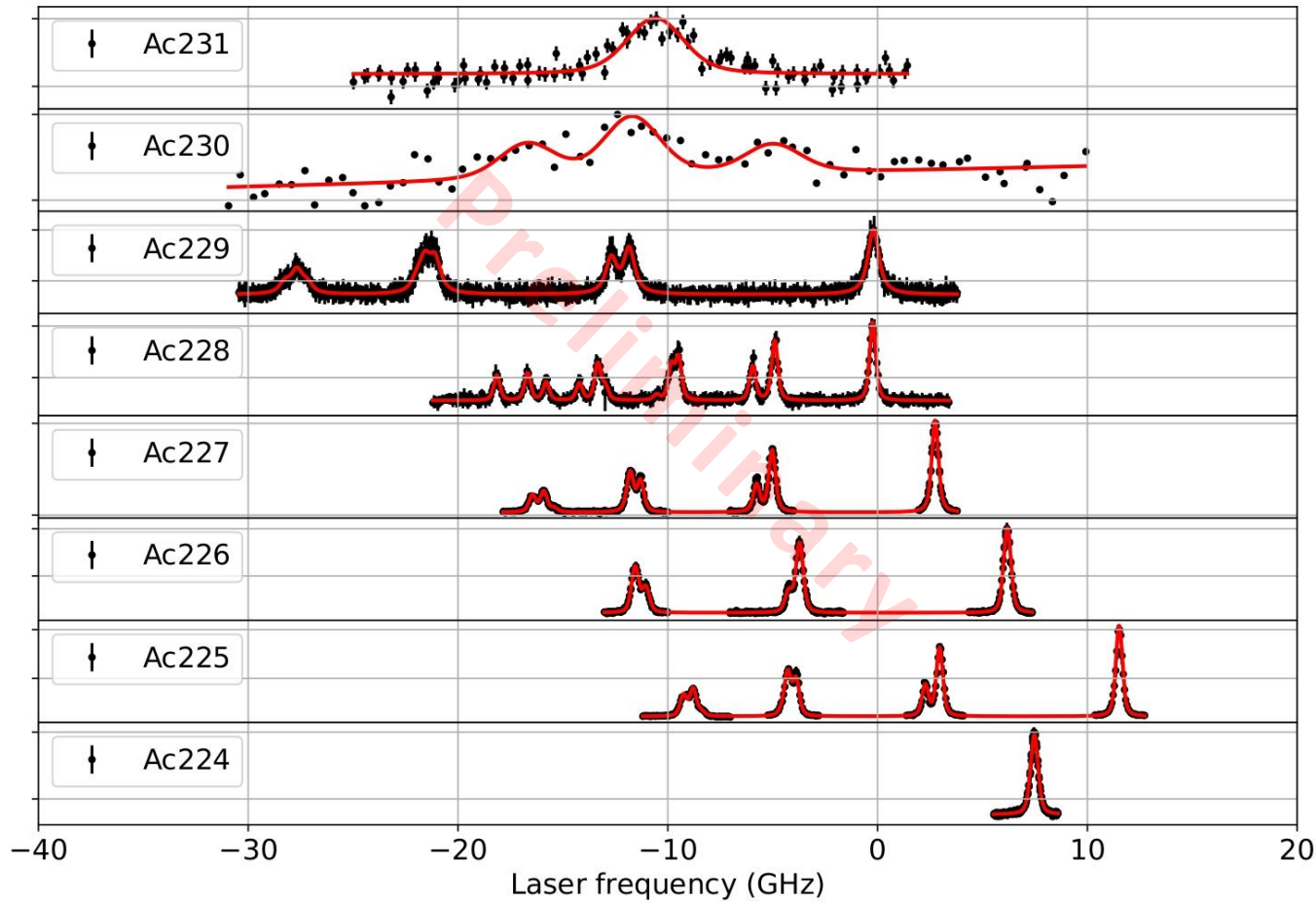


“Sub-Doppler” Hot Cavity In-source Spectroscopy



- Crossed atom beam / laser geometry in LIST structure
- Selection of **reduced Doppler ensemble** in laser intersection volume
- Suitable **narrow-band laser**

Actinium high resolution spectroscopy

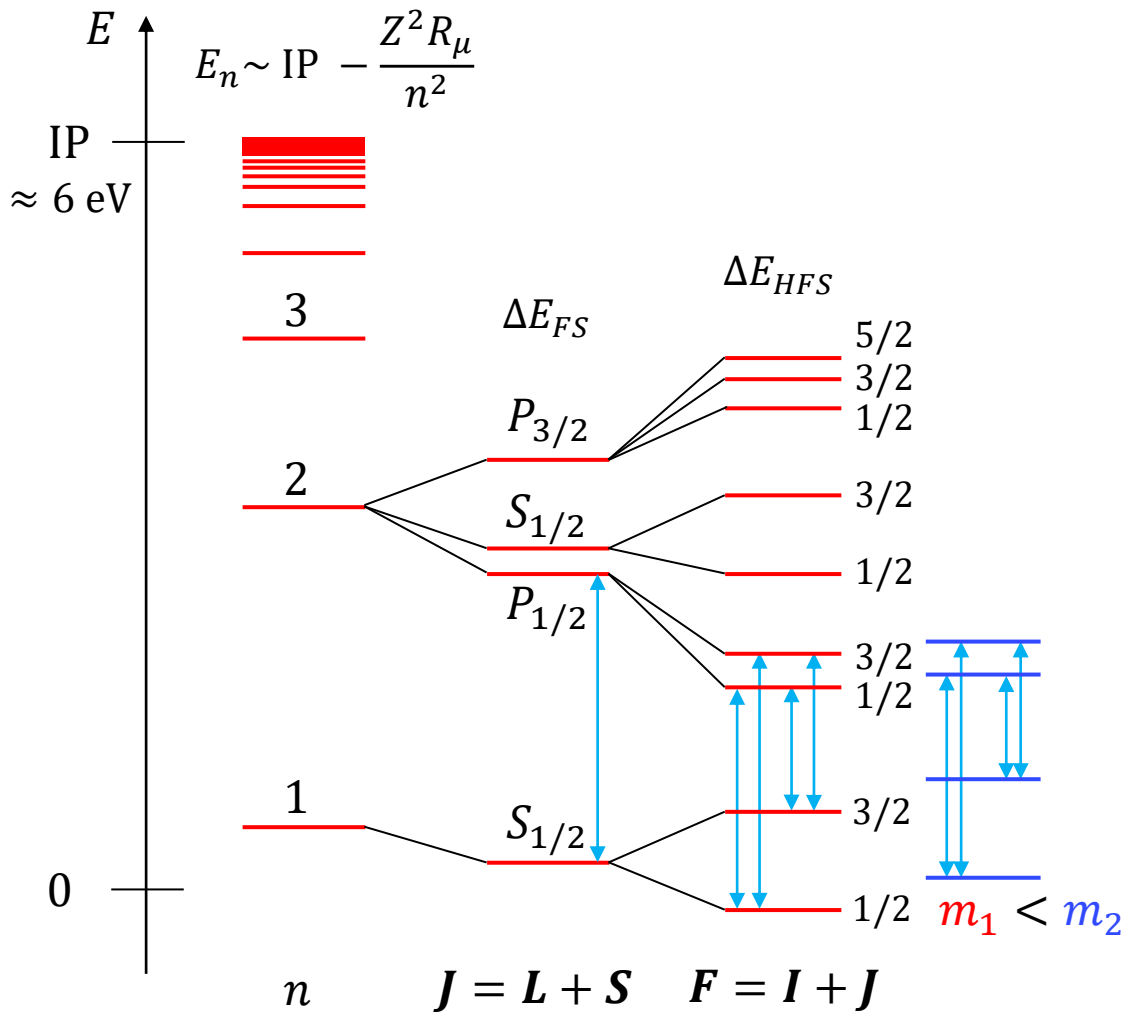


E. Verstraelen et al., PHYSICAL REVIEW C 100, 044321 (2019)

analysis by Michael Heines



Probing atomic structure with lasers



- Atomic level structure through optical transitions
- Ionization limit through Rydberg convergences
- Nuclear spins and electromagnetic moments through hyperfine interaction

$$A = \frac{\mu_I \overline{H(0)}}{IJ}$$

magnetic field at nucleus

$$B = eQ_s \left. \left(\frac{\partial^2 \phi}{\partial z^2} \right) \right|_{r=0}$$

electric field gradient at nucleus

Atomic shell
Nucleus

$Q_s < 0$ $Q_s > 0$

I, μ_I

ΔE_{FS}

n – principal quantum number S – total spin angular momentum
 L – total orbital angular momentum I – nuclear spin
 J – total angular momentum μ – magnetic dipole moment
 Q – electric quadrupole moment

Neptunium production



- Radioactive actinide
- Long half-life
- High radiotoxicity

Pu 237	Pu 238	Pu 239	Pu 240	Pu 241
			6545 a	14.35 a
			168; 5.124... β ⁻ ; g	β ⁻ ; g; α 4.896... γ (149); e ⁻
			239	Np 240
			5 d	7.22 m 65 m
			e ⁻ ; g	β ⁻ 2.2...; β ⁻ 0.9... γ; e ⁻ ; γ; g; g
			238	U 239
			745	23.5 m
			68·10 ⁹ a	β ⁻ 1.2; 1.3...; γ 75; 44...

→ neutron capture
← β⁻ decay



LISA

LASER IONISATION AND SPECTROSCOPY OF ACTINIDES

@LISA_ITN

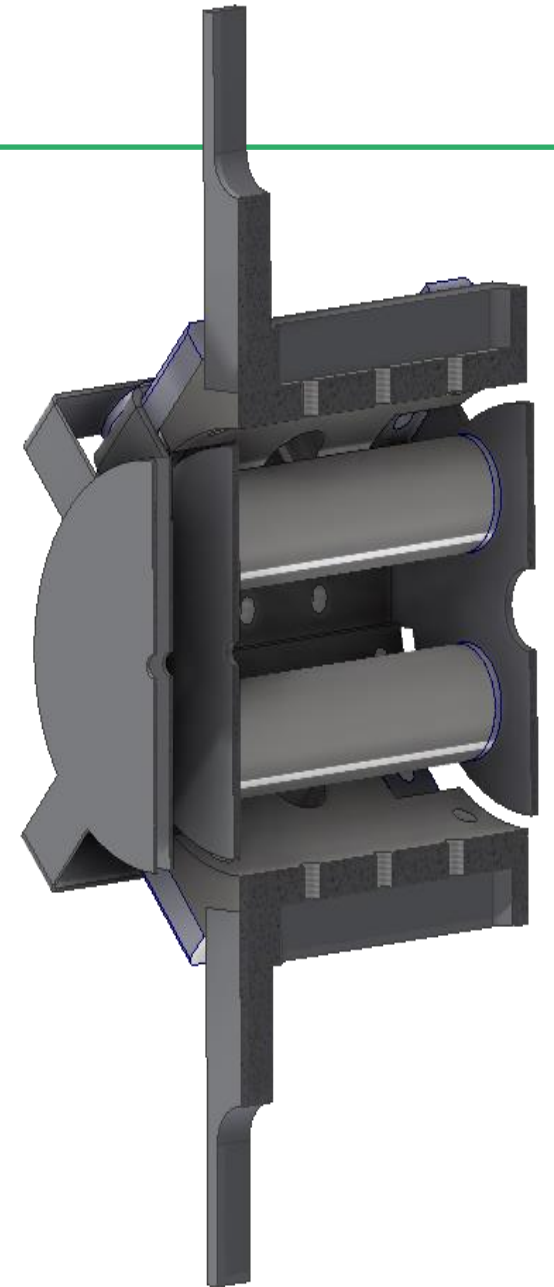
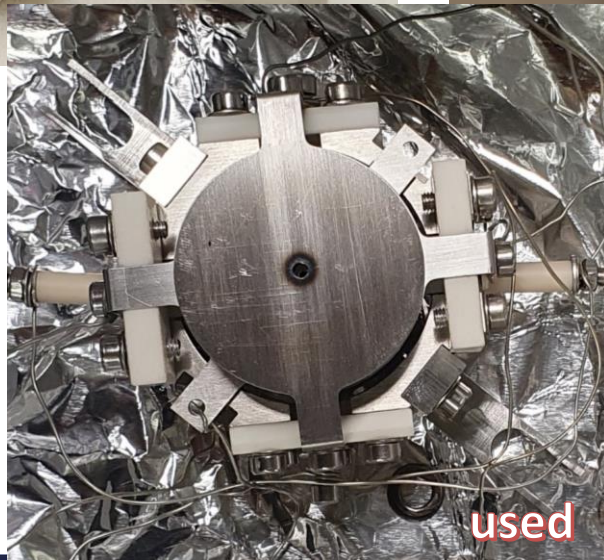
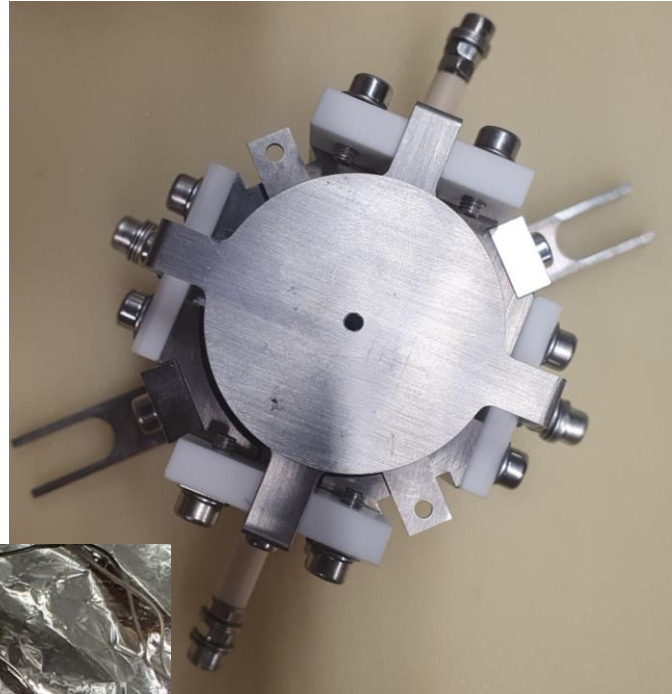
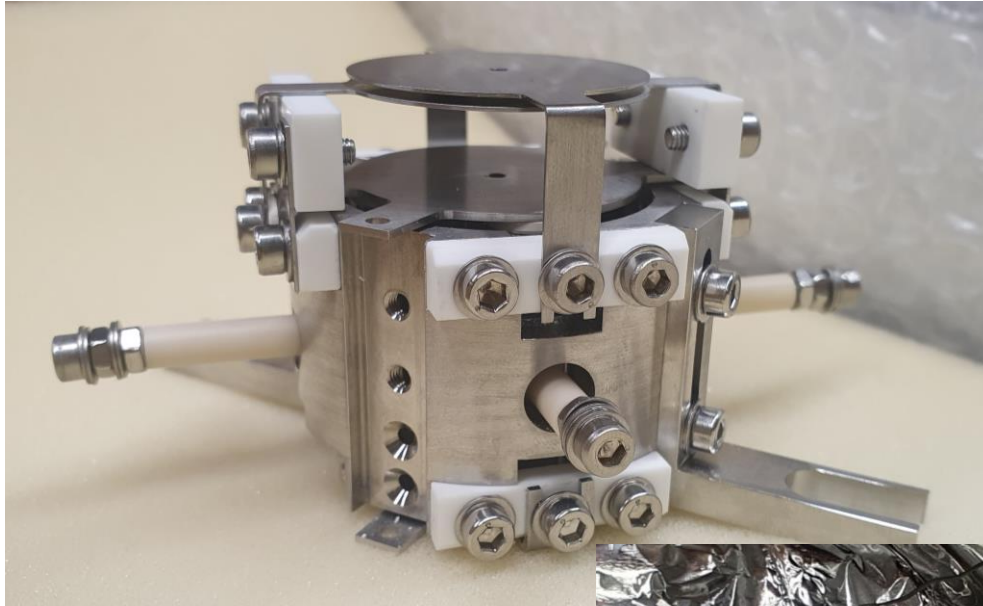
<https://lisa-itn.web.cern.ch/>

ion schemes
e analysis.

count
ion schemes
coming from hyperfine structure (HFS) and isotope shift (IS).

[1] Kohler, S; et al. Spectrochim. Acta B,52, 717 – 726, (1997)
 [2] Kazakov, V; et al. Phys. Scr., 92, 10, (2017)
 [3] Stone, N.J. At. Data Nucl. Data Tables, 90, 1, 75-176, (2005)

FI-LIST Field Ionization LIST



HFS result and nuclear moments in Np

Np 237
 2.14·10⁶ a
 α 4.790...;
 γ 29...; e⁻; g

[1] Fred, M; et al, J. Opt. Soc. Am., 67, 1 (1977)

FES [cm ⁻¹]	Fit		Literature [1]		J	μ_I [μ_N]	Q_S [barn]
	A [MHz]	B [MHz]	A [MHz]	B [MHz]			
0	776.10(18)	929(5)	778(10)	645(100)	11/2	+3.14(4) [2]	+3.886(4) [2]
25 075.1	1470.02(10)	327(5)	1470(10)	264(100)	13/2		
25 277.64	570.08(14)	-307(4)	X	X	9/2		

[2] Stone, N.J. At. Data Nucl. Data Tables, 90, 1, 75-176, (2005)

Np 239
 2.355 d
 β-0.4...
 γ 106...; e⁻; g

FES [cm ⁻¹]	Fit		J
	A [MHz]	B [MHz]	
0	785.26(39)	949(10)	11/2
25 075.15	1487.06(25)	336(9)	13/2
25 277.63	576.74(40)	-319(11)	9/2

Determination of nuclear moments

$$\frac{A}{A_{\text{ref}}} = \frac{I_{\text{ref}}}{I} \frac{\mu_I}{\mu_{I,\text{ref}}} \quad \frac{B}{B_{\text{ref}}} = \frac{Q_S}{Q_{S,\text{ref}}}$$

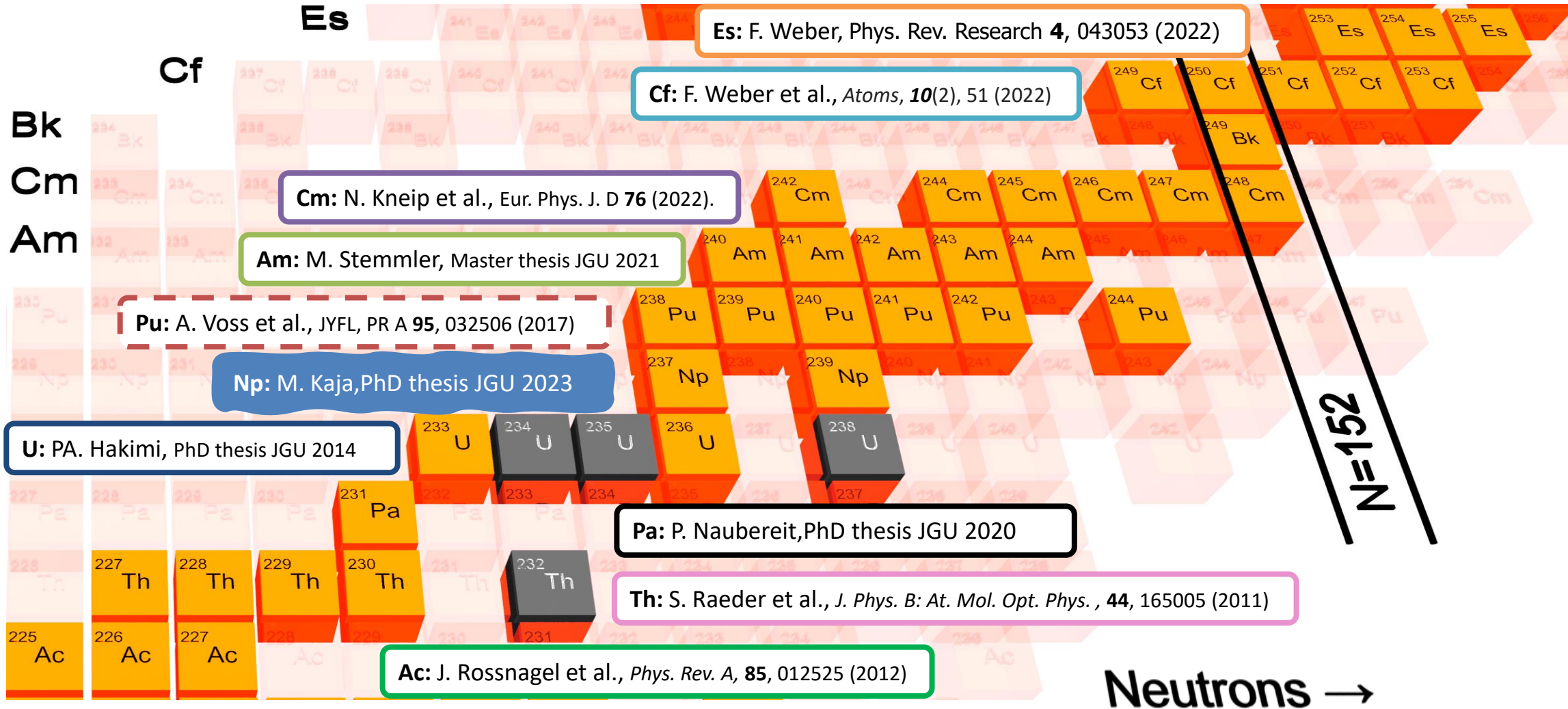
measured
literature

μ_I [μ_N]	Q_S [barn]
+3.18(3)	+3.98(8)

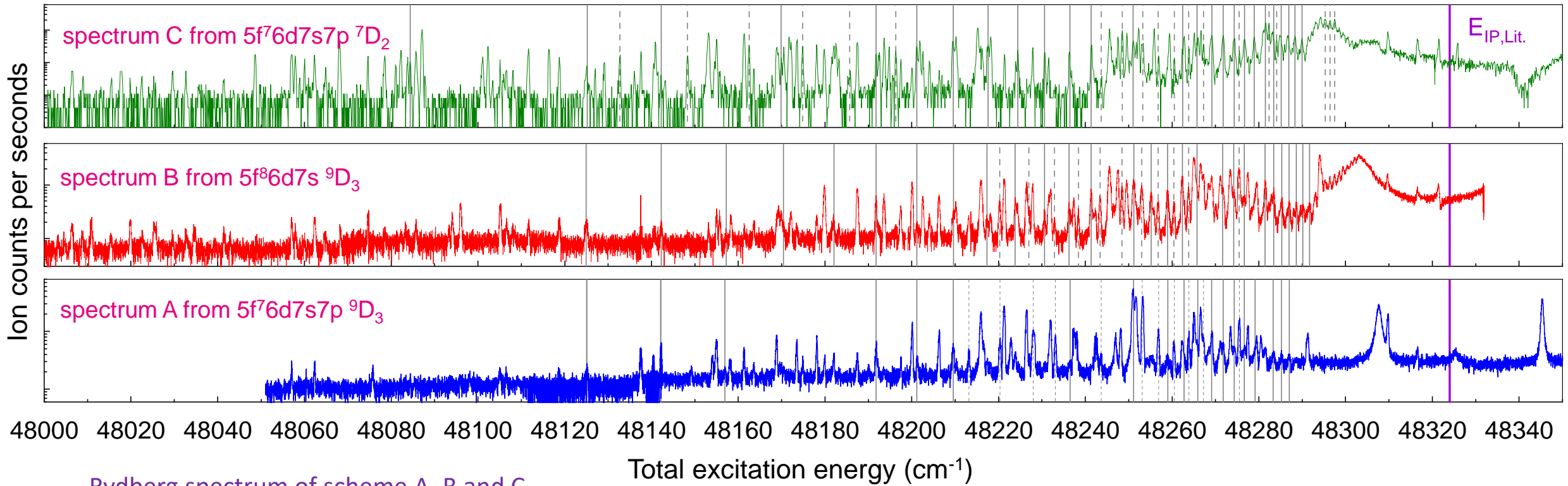
Presented results are preliminary, to be published by M. Kaja et al.

Larissa activities on laser spectroscopy in the actinides

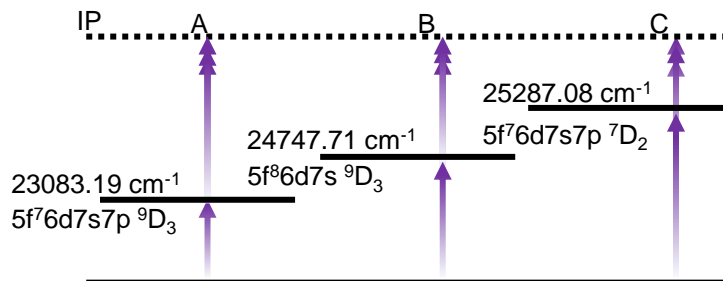
Protons ↑



Ionization potential determination of curium



Rydberg spectrum of scheme A, B and C

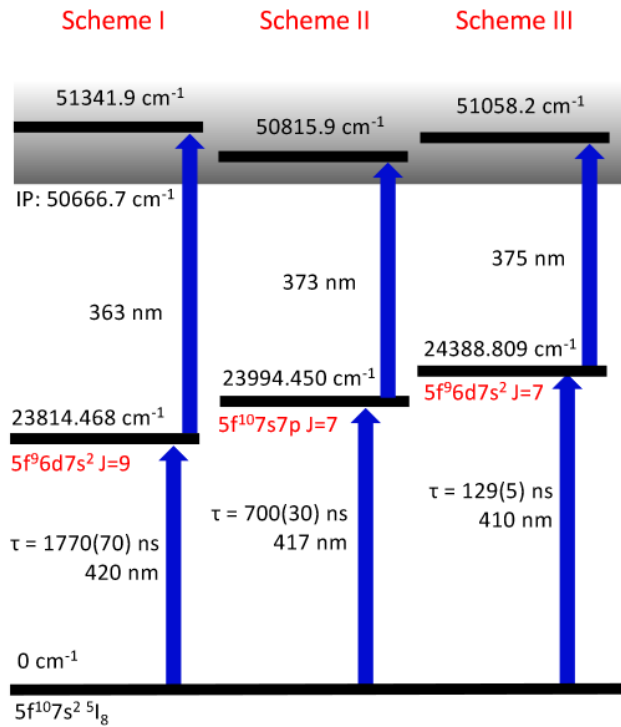


Investigation of the Rydberg spectrum A, B and C

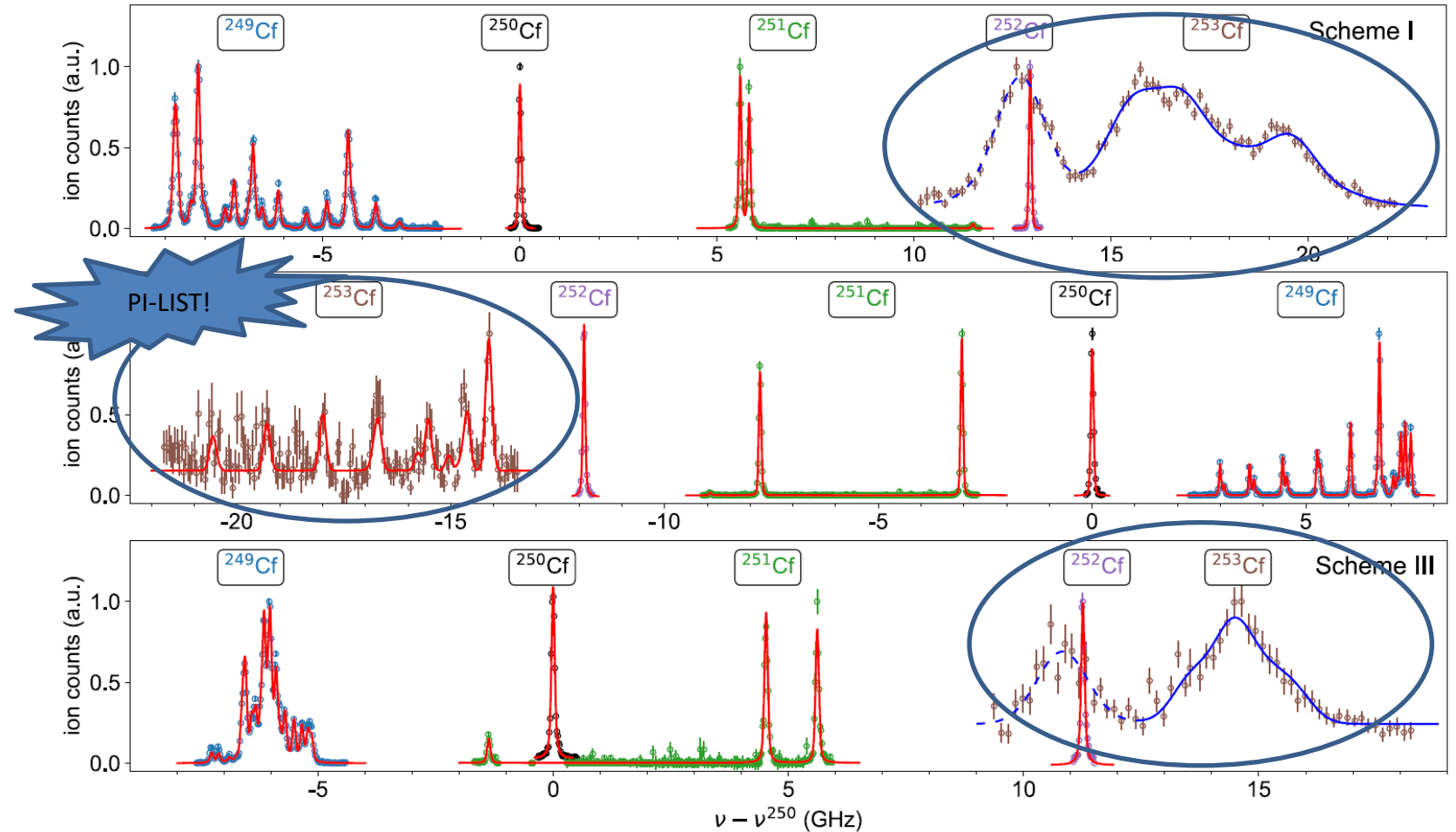
- Spectral scan range 400 cm^{-1}
- IP = $48330.68(12) \text{ cm}^{-1}$

N. Kneip,.., M.K., et al., Eur. Phys. J. D **76** (2022)

High resolution spectroscopy of californium



Cf 252
2.65 y
I=0 1e11 atoms (40 pg)
 α -spectrometry
 Ratio $\sim 3000:1$
Cf 253
17.81 d
I=7/2 $<5e7$ atoms (20 fg)
 mass spectrometry



F. Weber,.., M.K., et al., Phys. Rev. C **107**, 034313 (2023)