

Quadrupole moment of ^{223}Ra from molecular spectroscopy of ^{223}RaF

4 of September 2024

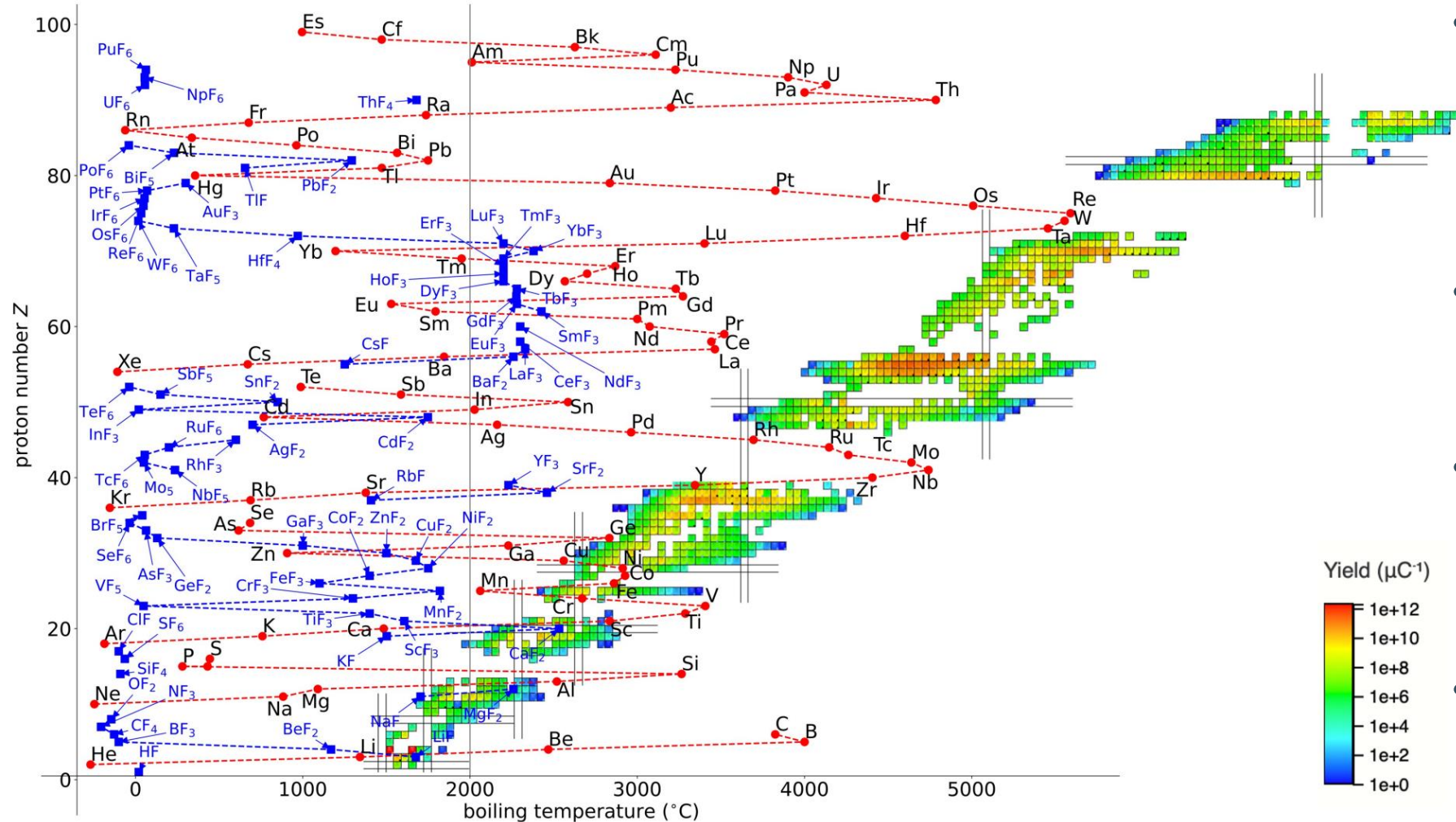
PhD student: Carlos M. Fajardo-Zambrano



Production of radioactive nuclei at ISOLDE and RaF



Produced isotopes at ISOLDE



- Molecules will have different chemical properties than its atomic counterpart → different release and ionization properties.
- Molecules could be used to study poorly produced elements (Ac from AcF 2022)
- Mainly studied to provide high-purity beams in regions with high isobaric contamination.
- Molecules can be more sensitive to new observables (RaF).

Fluorine molecules – laser coolable

Best EDM measurements use laser cooling on **stable** fluorine molecules.

2010: SrF Nature 467

2011: YbF Nature 473

...

2017: SrF Nature Physics 13

2018: YbF PRL 120, 123201

2023: HfF⁺ Science, 381(6653).

- **Octupole deformed nuclei** further enhance the sensitivity to EDM ! (N. Auerbach, et al., Phys. Rev. 4316 (1996)).
- Some **Ra isotopes are octupole deformed** ! (Gaffney et al., Nature 497 (2013)).

RaF: ideal molecule, but RADIOACTIVE !

- Heavy nuclei
- Octupole deform nuclei
- Polar molecule

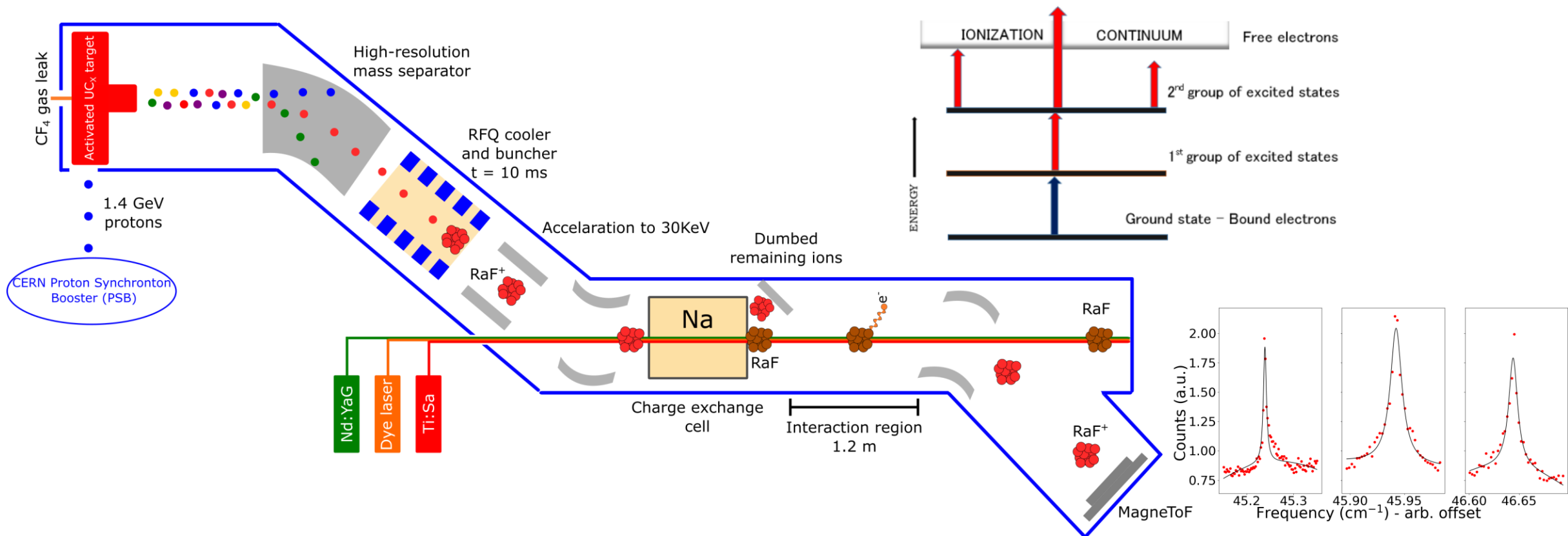
Laser spectroscopy and CRIS



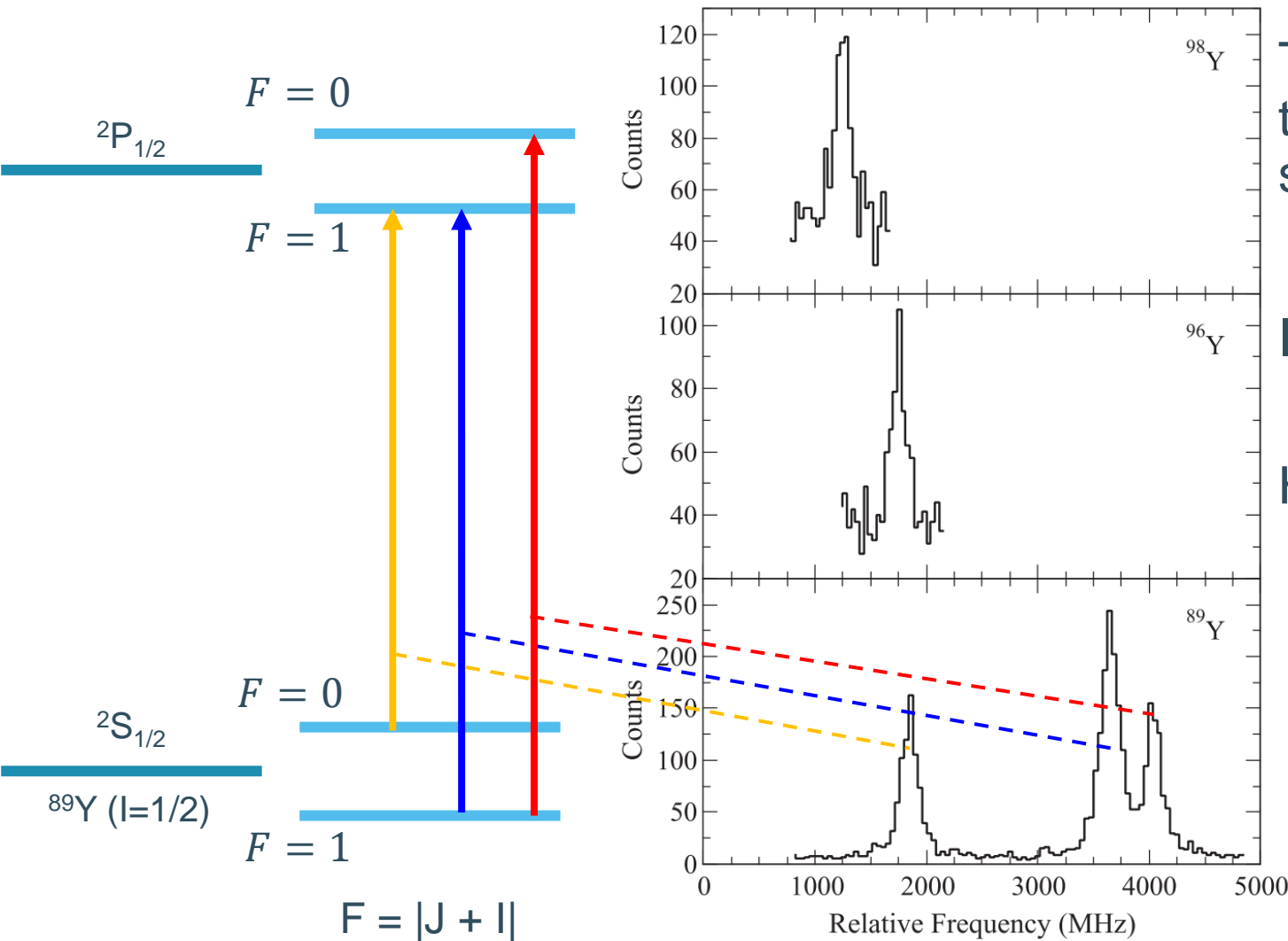
Collinear Resonance Ionization Spectroscopy (CRIS)

Laser beams are overlapped with an accelerated ion beam, reducing doppler broadening.

Resonance ionization spectroscopy is very efficient and selective: background reduction by deflecting the resonant ions from the neutrals → beam intensities of ~10-100 ions per second.



High resolution laser spectroscopy observables



The electron-nuclear interaction leads to the appearance of the Hyperfine structure (H_{hfs}).

Isotope shifts

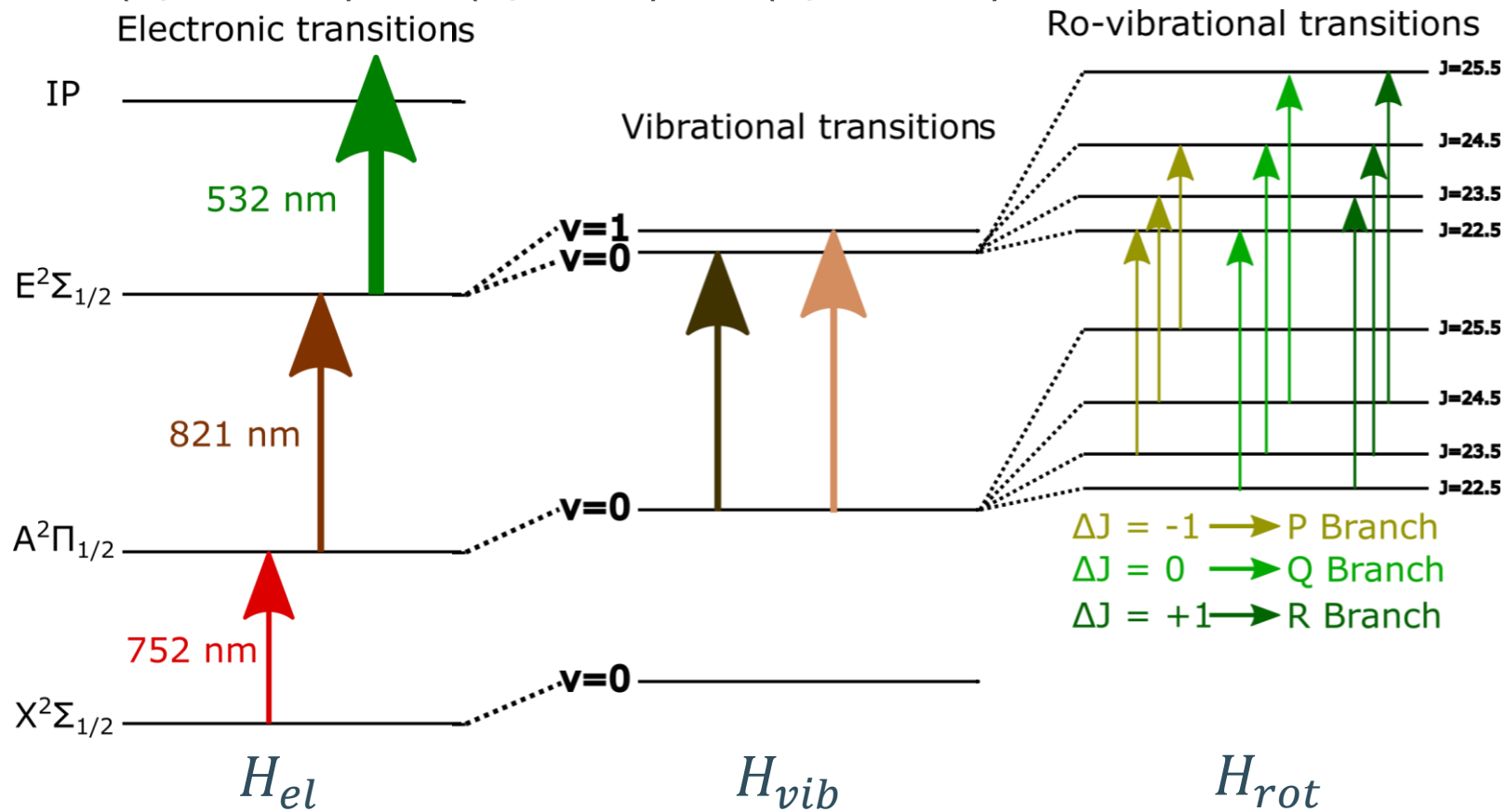
- Change in rms charge radii.

Hyperfine structure

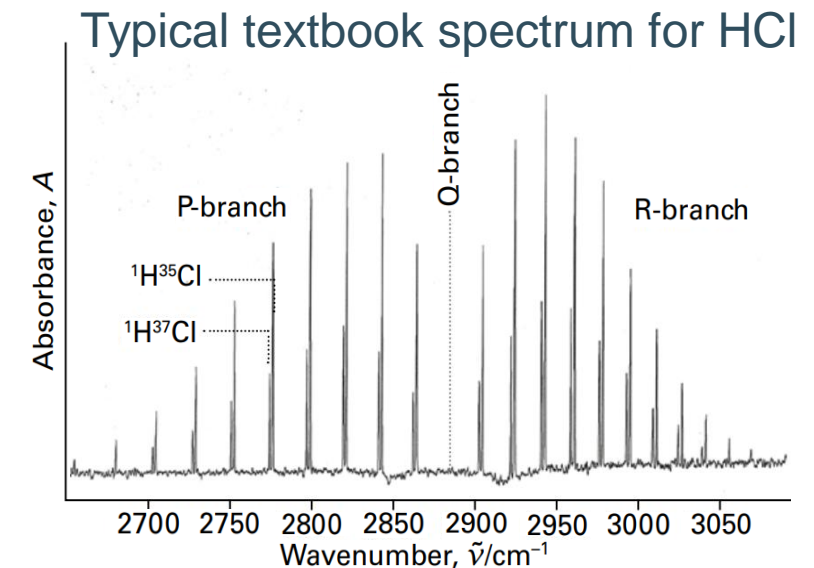
- Nuclear spin.
- Magnetic dipole moment ($I > 0$).
- Electric quadrupole moment ($I > 1/2$)

Laser spectroscopy on molecules

The spectrum of a rovibrational transition is divided in three different branches:
 R ($\Delta J = +1$), Q ($\Delta J = 0$), P ($\Delta J = -1$).

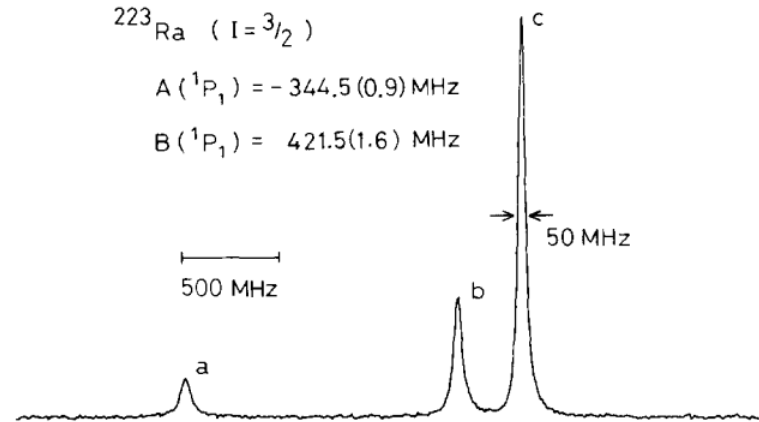
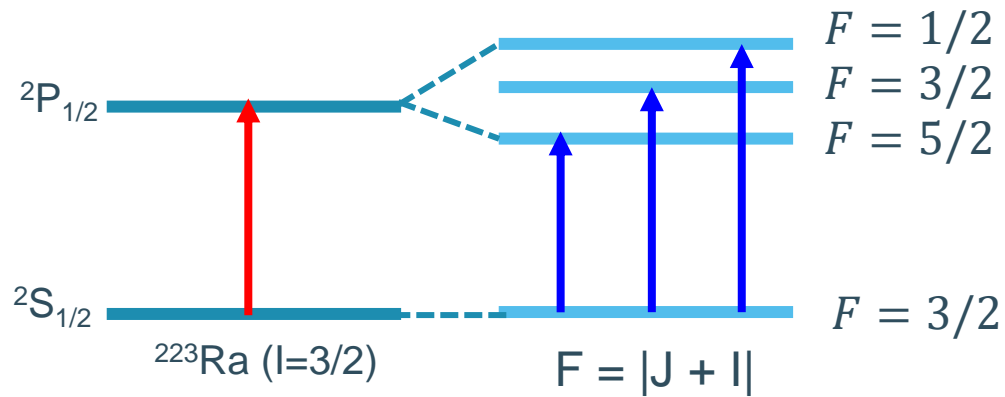


Each rotational line is further split due to the electron-nuclear interaction (H_{hfs}).

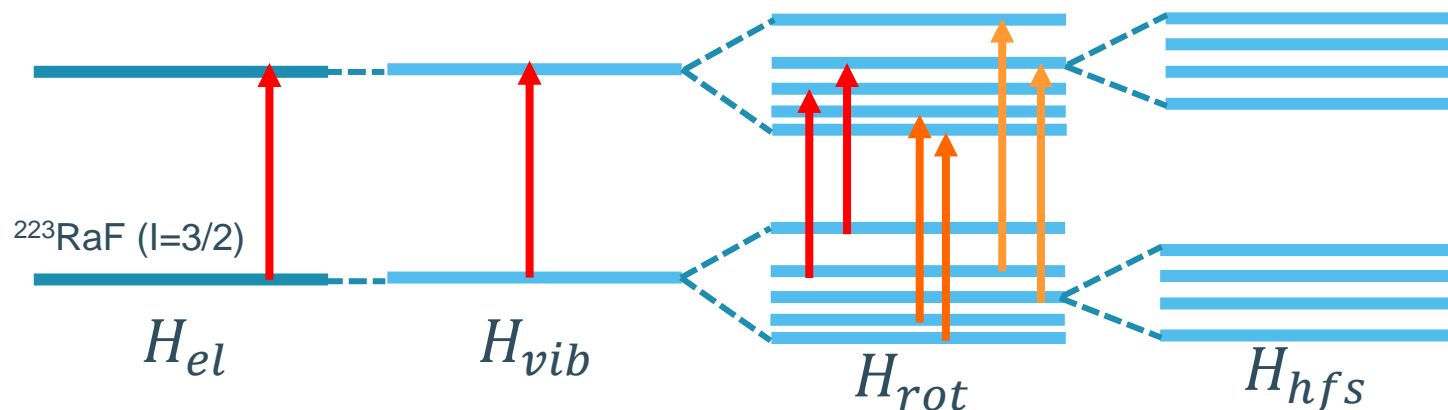


High-resolution in atoms and molecules

$$H_{hfs} = H_m + H_Q = A(I, J) + B(I, J)$$

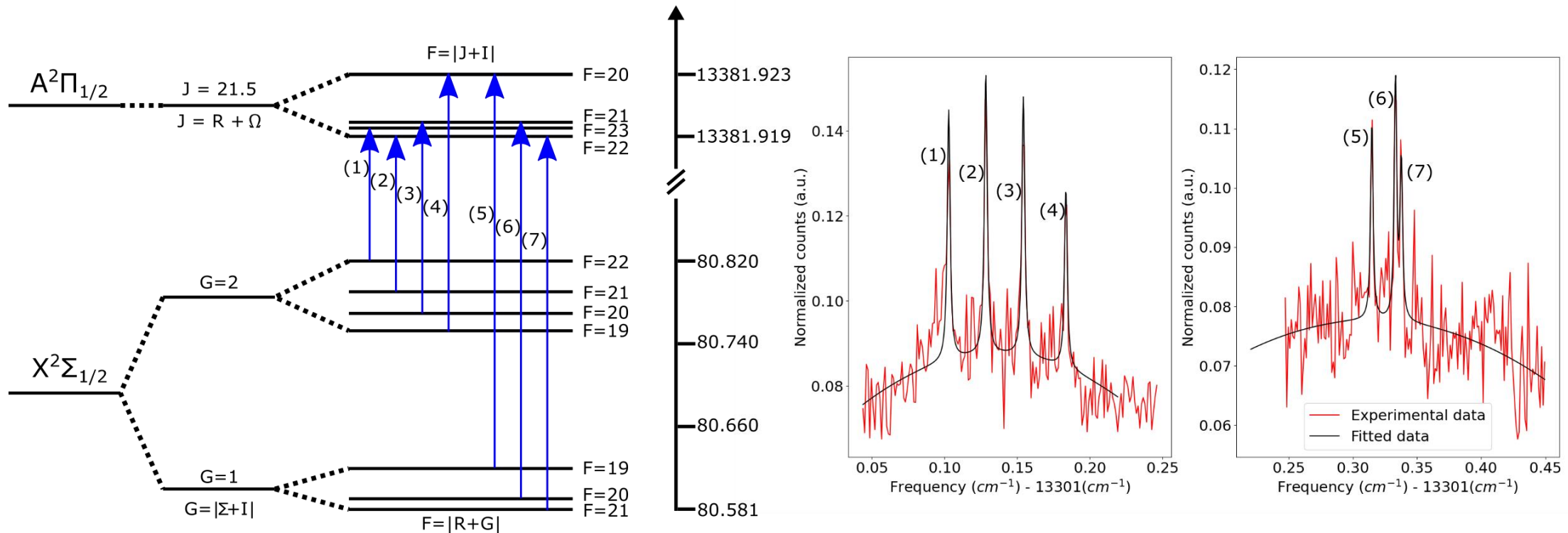


$$H_{hfs,R} = H_m + H_Q = A_{\parallel}(I_z, J_z) + A_{\perp}(I_{(x,y)}, J_{(x,y)}) + eq_0 Q(I_z, J_z) - eq_2 Q(I_{(x,y)}, J_{(x,y)})$$



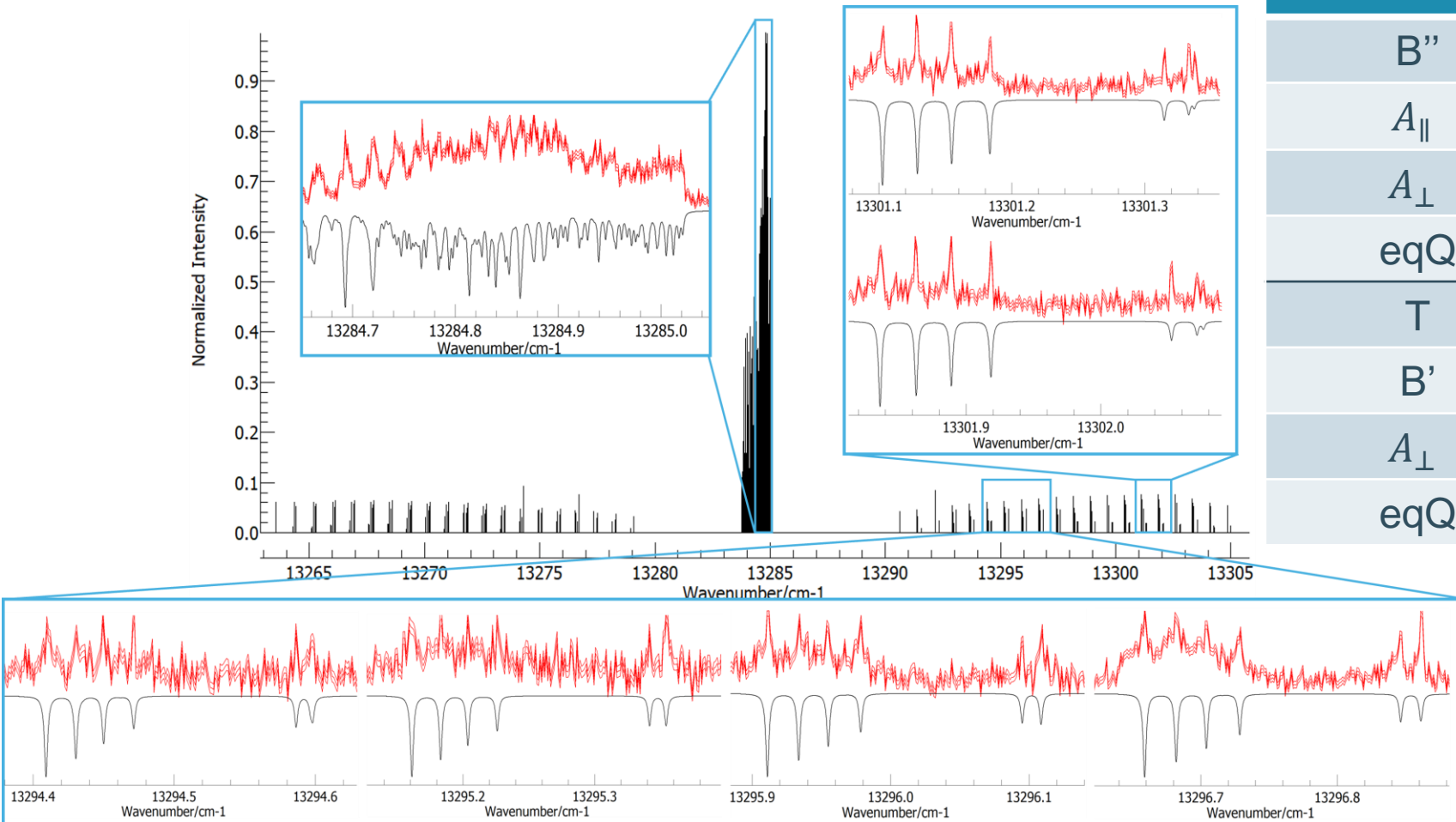
Each rotational transition is one HFS spectrum of ^{223}RaF like the one above

High resolution ^{223}RaF ($I = 3/2$) spectrum (R-branch)



- The HFS of ^{223}RaF was initially fitted using SATLAS2 to include the background on the spectra.
- The obtained centroids were used for the final fit with PGOPHER.

Fitting of the ^{223}RaF spectra (PGOPHER).



Parameter	This work	Theory
B''	5762.32(3)	5730
$A_{ }$	2084(7)	2095
A_{\perp}	2025(2)	2016
eqQ	-1480(29)	-1473
T	13284.6959(1)	-
B'	5732.56(3)	5709
A_{\perp}	255(2)	272
eqQ	-934(23)	-912

Extracting the nuclear moments

The ab initio magnetic dipole (A_{\parallel} , A_{\perp}) and electric quadrupole coupling (eqQ) constants are defined as,

$$A_{\parallel}(A) = \frac{\mu}{I} f(r_{iA}, \Sigma_z) \quad A_{\perp}(A) = \frac{\mu}{I} f(r_{iA}, \Sigma_{x/y}) \quad eqQ(A) = Q(A) 0.2349647 \cdot q(A)$$

Magnetic dipole moment
Magnetic field at the nucleus
Electric quadrupole moment
EFG at the nucleus

Extract the magnetic dipole and electric quadrupole moment **using the known moments of an isotope (^{225}RaF)**

$$\mu_2(A) = \frac{I_2 A_{2(\parallel,\perp)}}{I_1 A_{1(\parallel,\perp)}} \mu_1(A)$$

$$\mu(^{223}\text{RaF}) = 0.27095(85) \mu_n$$

$$\mu(^{223}\text{Ra}) = 0.2703(6) \mu_n$$

Benchmark the theoretical predicted field, e.g., electric field gradient (EFG)

$$q(A) = \frac{eqQ(^{223}\text{RaF})}{Q(^{223}\text{Ra}) 0.2349647}$$

$$q(^{223}\text{RaF})_{\text{exp}} = 5.02(2) \text{ a.u.}$$

$$q(^{223}\text{RaF})_{\text{theo}} = 5.02(2) \text{ a.u.}$$

Only statistical error bars

[9] Wilkins, S., et al. (2023). arXiv preprint arXiv:2311.04121.

13 [10] Lynch, K. M., et al. (2018). Physical Review C, 97(2), 024309.

[11] Petrov, A. N., & Skripnikov, L. V. (2020). Physical Review A, 102(6), 062801.

Reference moments from molecules

Periodic Table of the Elements

Legend:

Atomic Number	Valence
Symbol	
Name	
Atomic Mass	

Lanthanide Series:

57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb	71 Lu
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Actinide Series:

89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No	103 Lr
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The measurement of nuclear moments using molecules is not new.

Elements highlighted have their current reference moments from molecular measurements.

Discussion

- Given its different chemical properties, molecules can be used for the study of isotopes that are poorly produced or are in a region with high isobaric contamination.
- Molecules are the most sensitive system to the presence of an eEDM.
- The magnetic dipole and electric quadrupole moments of ^{223}RaF have been successfully extracted and are in great agreement with their atomic counterpart (work in progress).
- Molecular spectroscopy is a promising tool to study unknown nuclear moments for nuclear and beyond the standard model physics.

References

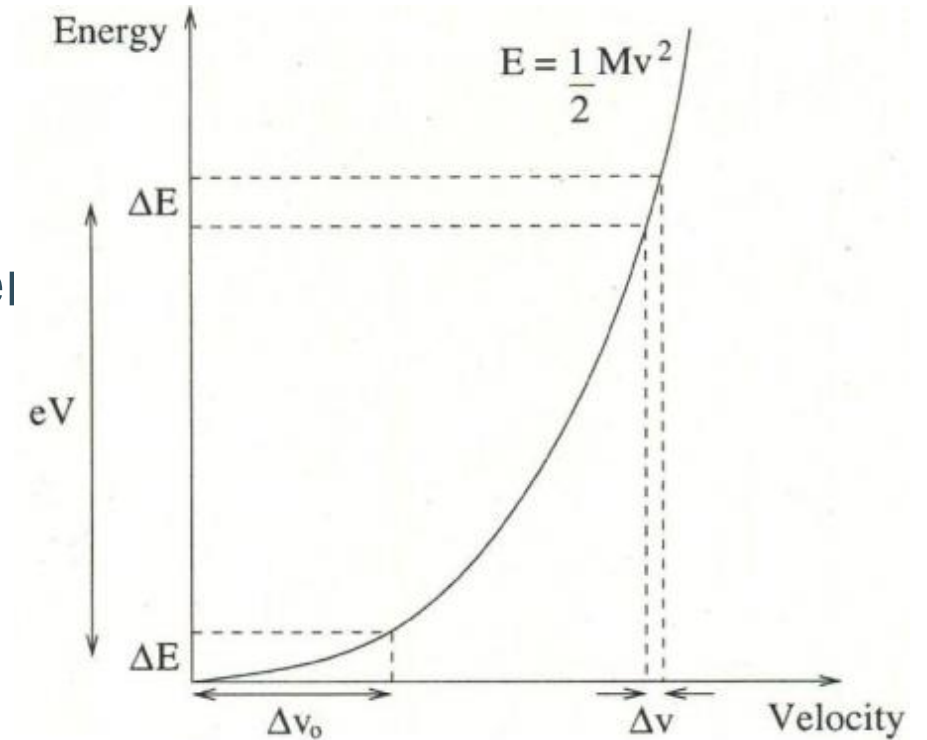
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Thanks for your attention

Backup slides

Collinear geometry

- Ion beam is accelerated to 30-60 keV, reducing the velocity spread.
- Use of a gas-filled radiofrequency cooler-buncher to produce short (μs) bunches.
- Overlapped (anti-)collinearly with two or more lasers.



$$v - v_0 = v_0 \frac{\Delta E}{c \sqrt{2eV_{acc}m}}$$

Reduction in spectrum broadening from ~GHz to ~MHz!!

Background present on the spectra

The line strength of a molecular dipole transition between HFS states is proportional to,

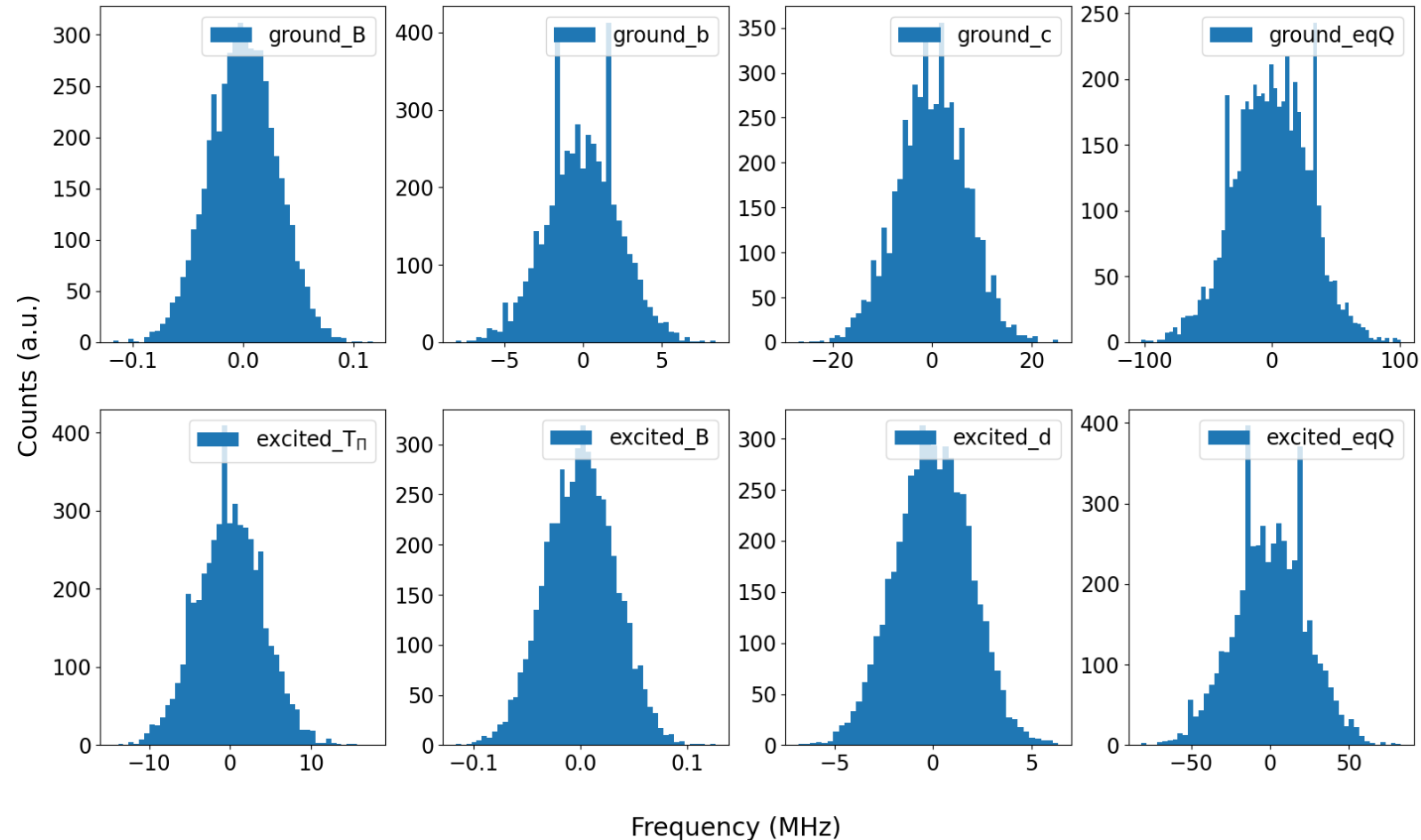
$$\langle N'SJ'IF' | \mu^{(1)} | NSJIF \rangle$$
$$\propto [(2J+1)(2J'+1)(2F+1)(2F'+1)]^{1/2} \left\{ \begin{matrix} N' & J' & S \\ J & N & 1 \end{matrix} \right\} \left\{ \begin{matrix} J' & F' & I \\ F & J & 1 \end{matrix} \right\}$$

Where the terms in $\{ \}$ is the Wigner 6j symbols. For a transition between...(R-branch) the intensity of a $\Delta F = +1$ is ~ 18 and ~ 666 times stronger than a $\Delta F = 0$ and a $\Delta F = -1$ transition, respectively.

If the transition energies the $\Delta F = 0$ are calculated (based on the final fit), the position of the peaks are mainly within the quadruplet observed.

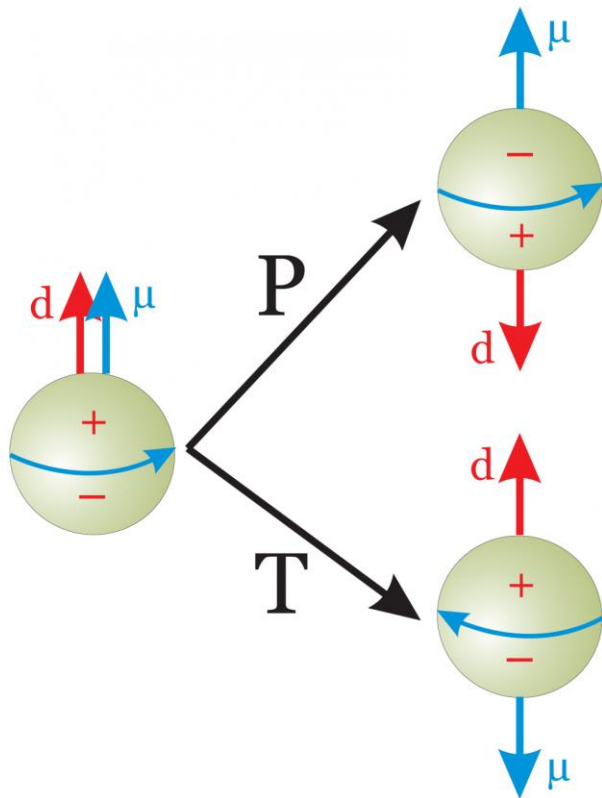
Error bars on the molecular constants of ^{223}RaF

- Each HFS peak was used to create a Gaussian distribution with centroid and width equal to the peak centroid and uncertainty.
- Re-fitting of the spectra with new peak centroids taken from the Gaussians.
- Store the fitted molecular constants and re-fit the spectra.
- The retrieved distribution from each constant gives its nominal value and uncertainty.



Electric dipole moment (EDM)

The EDM changes sign with parity and time reversal transformations.



EDM is odd under parity (P) and time-reversal (T) transformation.

A permanent EDM implies violation P,T

EDM's are theoretically predicted to be extremely small, therefore, not yet observed

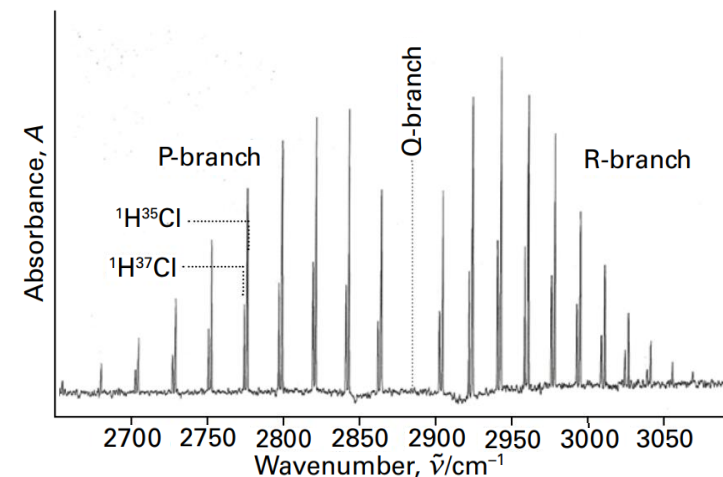
Intensity of a rotational-vibrational transition

This intensity increases with J , passing through a maximum and tailing off as J becomes even bigger.

- Population distribution given by the Boltzmann distribution.
- Degeneracy of each rotational level ($g_J = 2J + 1$).

A bigger population distribution is allowed with J . But the population of a state decreases exponentially as its energy increases.

$$N_J = g_J e^{-E_J/kT}$$



HFS precision limits with laser spectroscopy

Resolution:

- Laser bandwidth
- Lifetime (atomic and nuclear)
- Ion beam properties

Sensitivity to nuclear properties

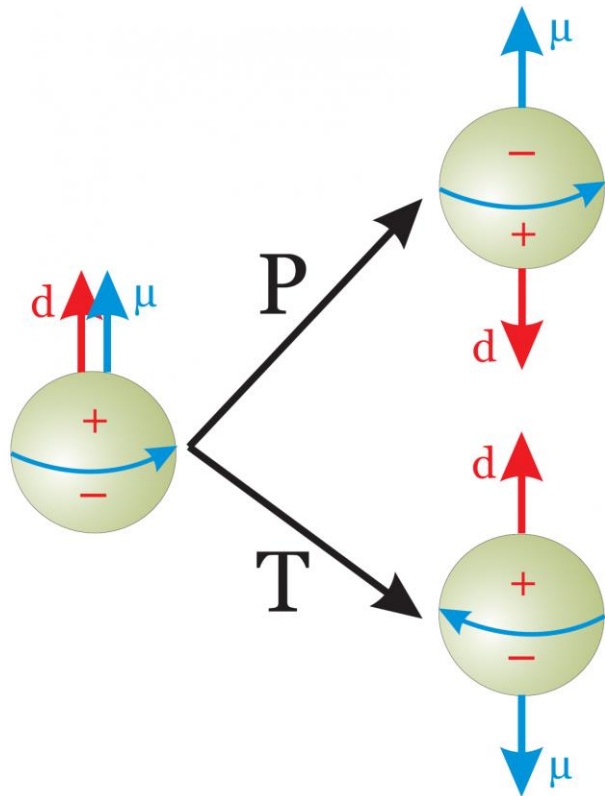
Instrumentation for measurements

- Wavelength meter
- Voltage/laser scanning

Statistics

EDM and why RaF has been studied?

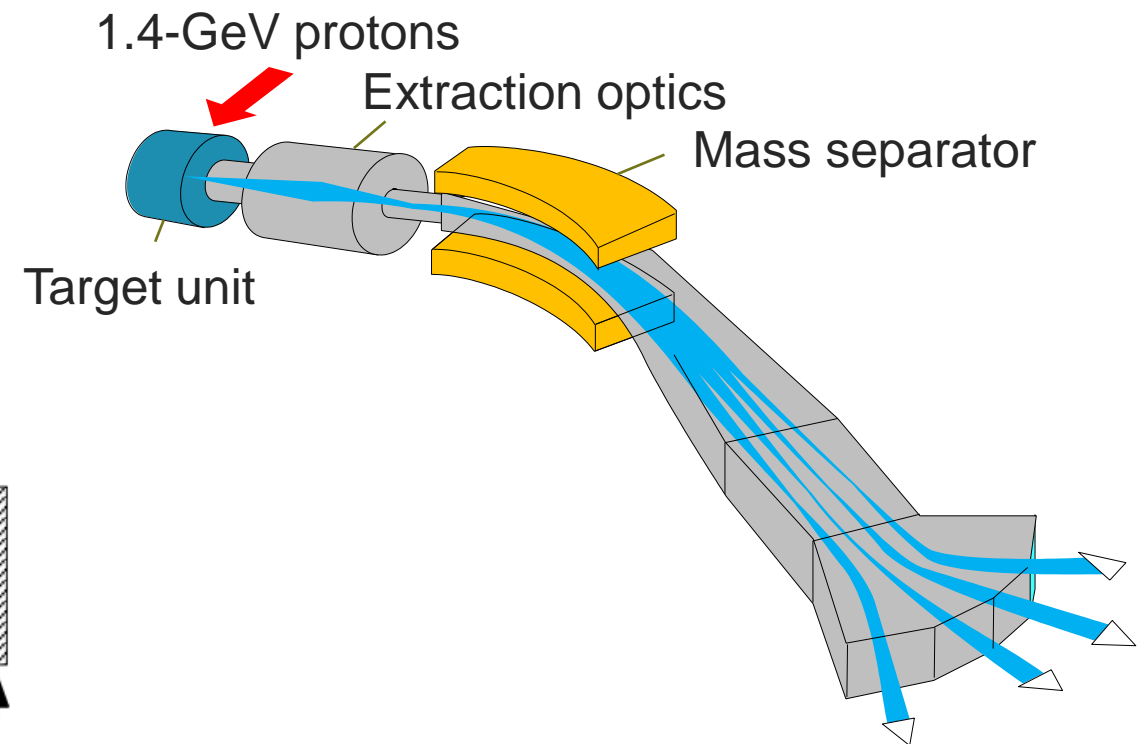
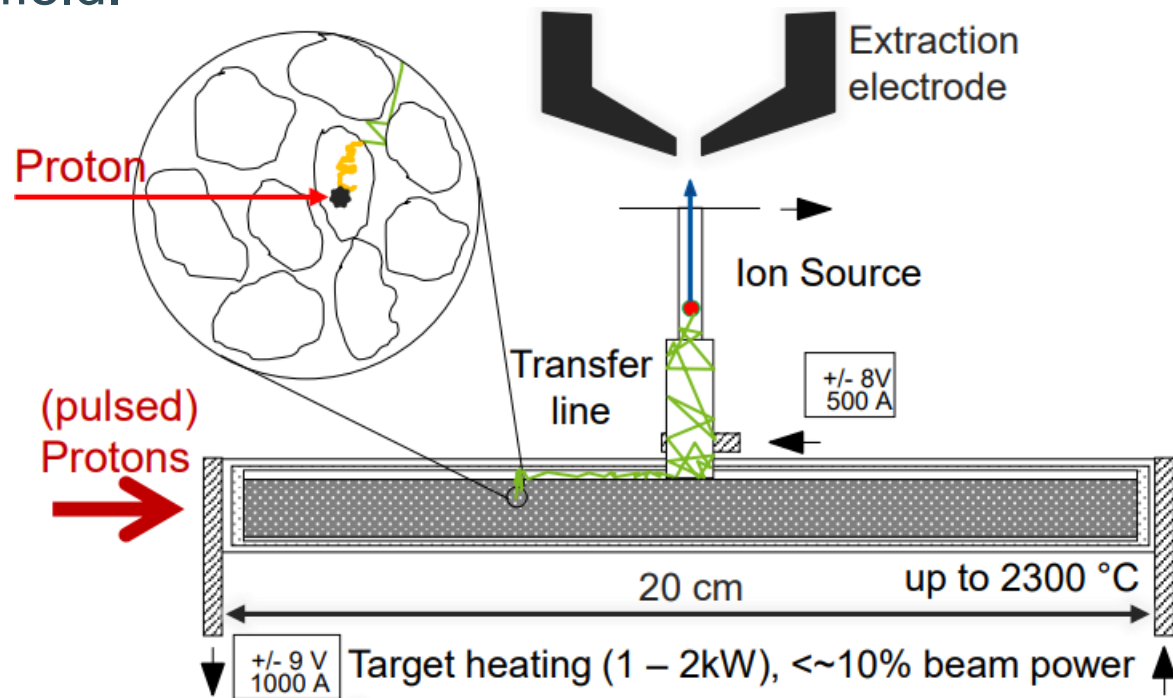
The measurement of a permanent electric dipole moment (EDM) implies P,T violation → to answer the baryogenesis problem and benchmark beyond the standard model theories.



- Diatomic polar molecules have higher sensitivity to the EDM than atoms (Internal effective electric field). Current best measuring of eEDM is **three order** of magnitudes more sensitive than atoms.
- Heavy octupole deformed nuclei are more sensitive to a nuclear EDM, but they are all radioactive.
- RaF as the ideal case: diatomic polar molecule, with a laser cooling scheme and a heavy octupole deformed nuclei.

The isotope separation on-line (ISOL) method

Use of thick targets allows a large total cross-section. Main reactions: spallation, fragmentation and fission. The produced atoms/molecules are ionized in an ion source. Ions of a specific mass (atomic or molecular) are separated from others using a magnetic field.



Molecular HFS Hamiltonian

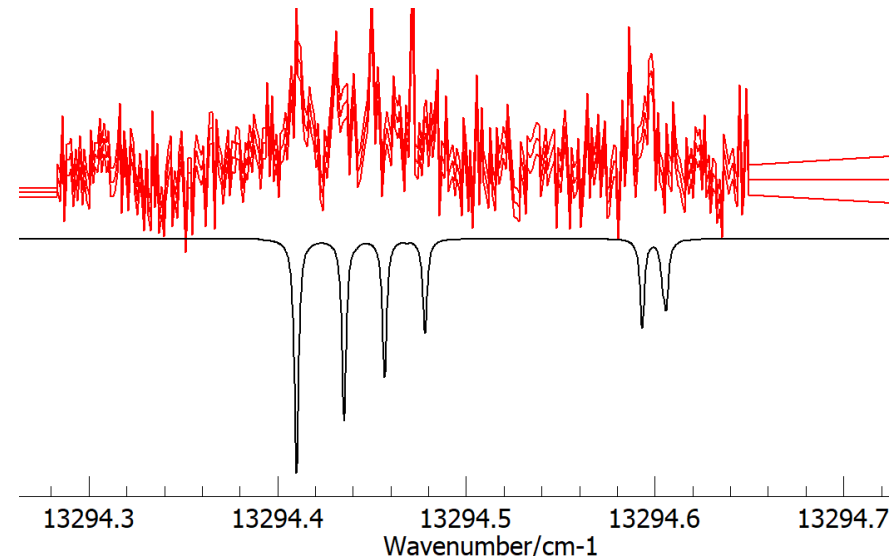
$$H_{hfs} = H_m + H_Q = a(I, L) + b_F(I, S) + c(I, S) - d(S_+I_+, S_-I_-) + eq_0Q(I, S) - eq_2Q(S_+I_+, S_-I_-)$$

- a , is the nuclear spin-electron orbit interaction, b_F is the Fermi contact interaction, c and d are the nuclear spin-electron spin axial and perpendicular dipole interaction
- eq_0Q and eq_2Q are the nuclear quadrupole axial and perpendicular coupling constants.

SPFIT/SPCAT

- The fit does not include all constants. Error on the assignment of transitions.
- All the constants are “close” but not fully reproduce the spectra.
- The “quantum numbers” are arbitrary selected. They might be miss assigned.

NEW PARAMETER (EST. ERROR) -- CHANGE THIS ITERATION				
1	55	E(0)	231518(255)	0
2	66	E(0)	265173(1000)	0
3	77	E(0)	265173(1000)	0
4	88	E(0)	265173(1000)	0
5	99	E(0)	265173(1000) PGOPHER	-0
6	100	B	5767.49(99) 5762.32	-0.00
7	155	B	5725.37(98) 5732.67	0.00
8	200	-D	-3.876(300)E-03	-0.000E-03
9	255	-D	-4.058(300)E-03	0.000E-03
10	10000000	gamma	181.32(271) 175.38	0.02
11	10040055	-.5*p	6157.15(300) 6159.23	-0.00
12	120000000	b_F	2062.1(159) 2044	-0.3
13	120010055	c	-44(69) 52	-0
14	220010000	1.5*eQq	-2305(86) -2259	-0
15	120040055	-.5*d	-164.0(150) 128	-0.1
16	220010055	1.5*eQq	-1210(76) -1417	0
MICROWAVE AVG =		-0.394633 MHz,	IR AVG =	0.00000
MICROWAVE RMS =		24.173334 MHz,	IR RMS =	0.00000



Background present on the spectra

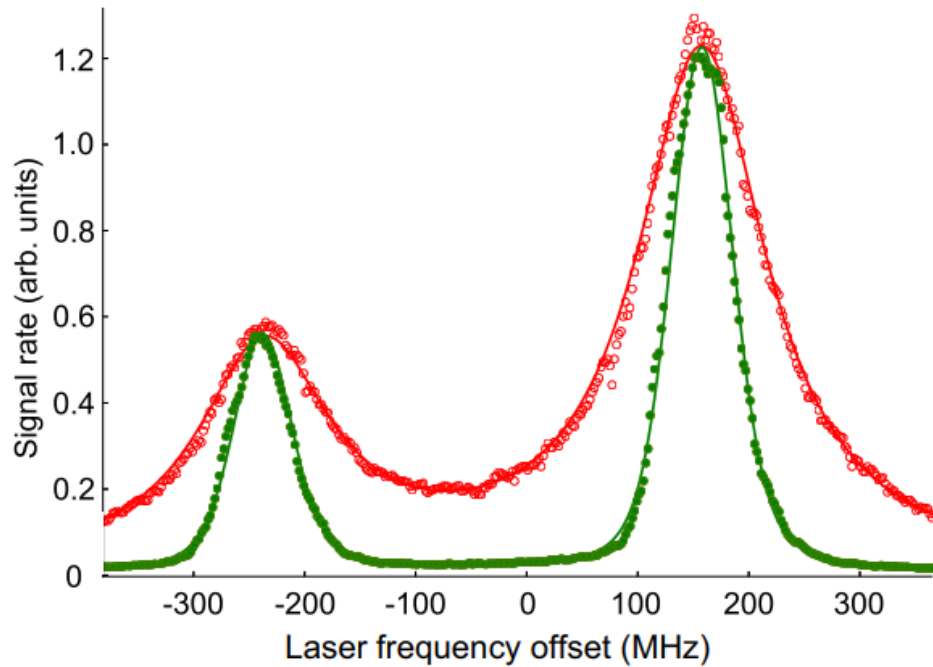
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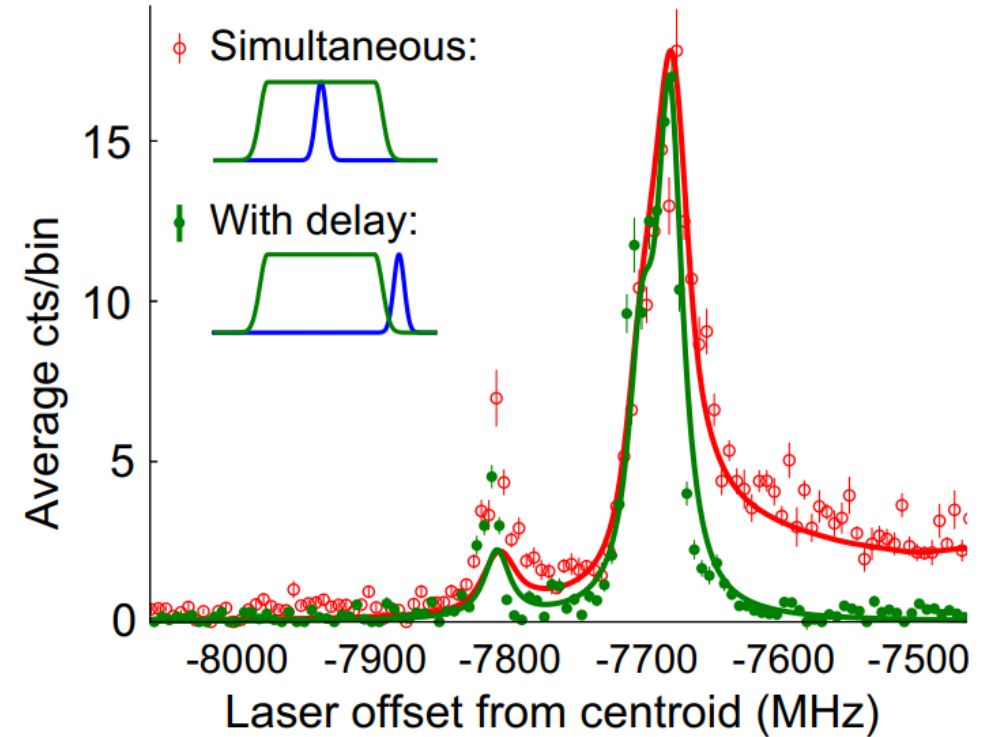
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AC stark effect and power broadening

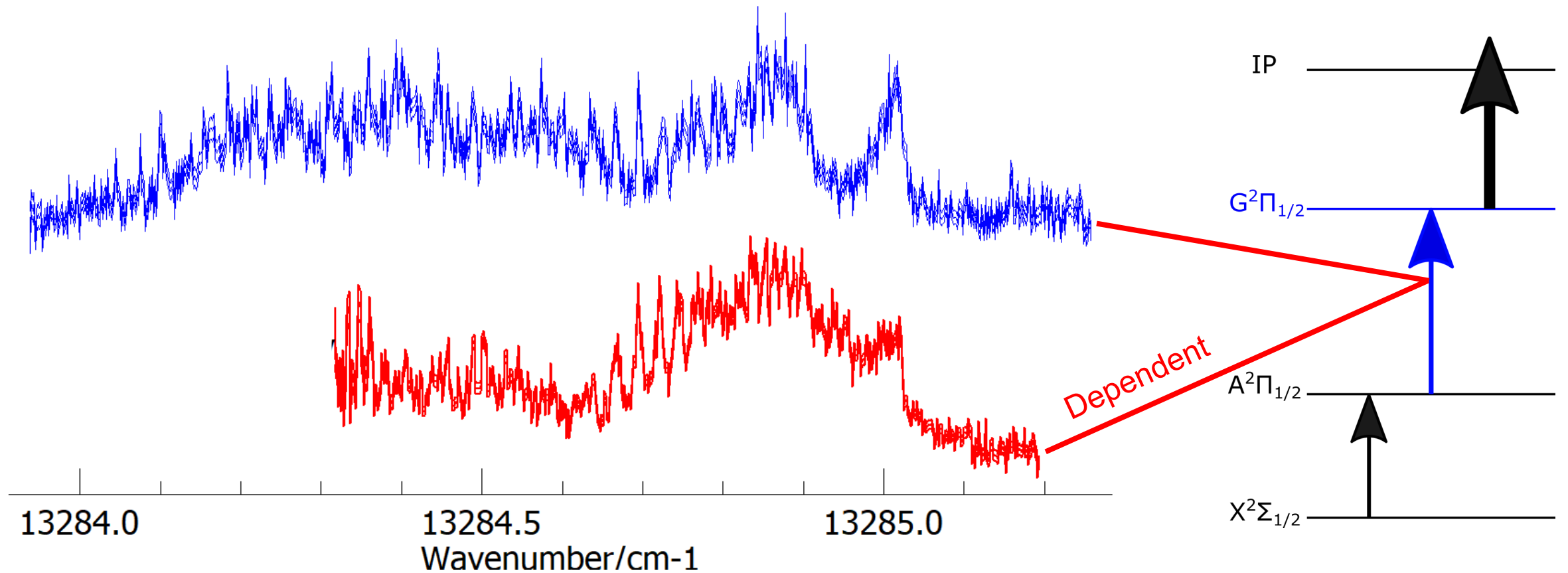


Power broadening (red)



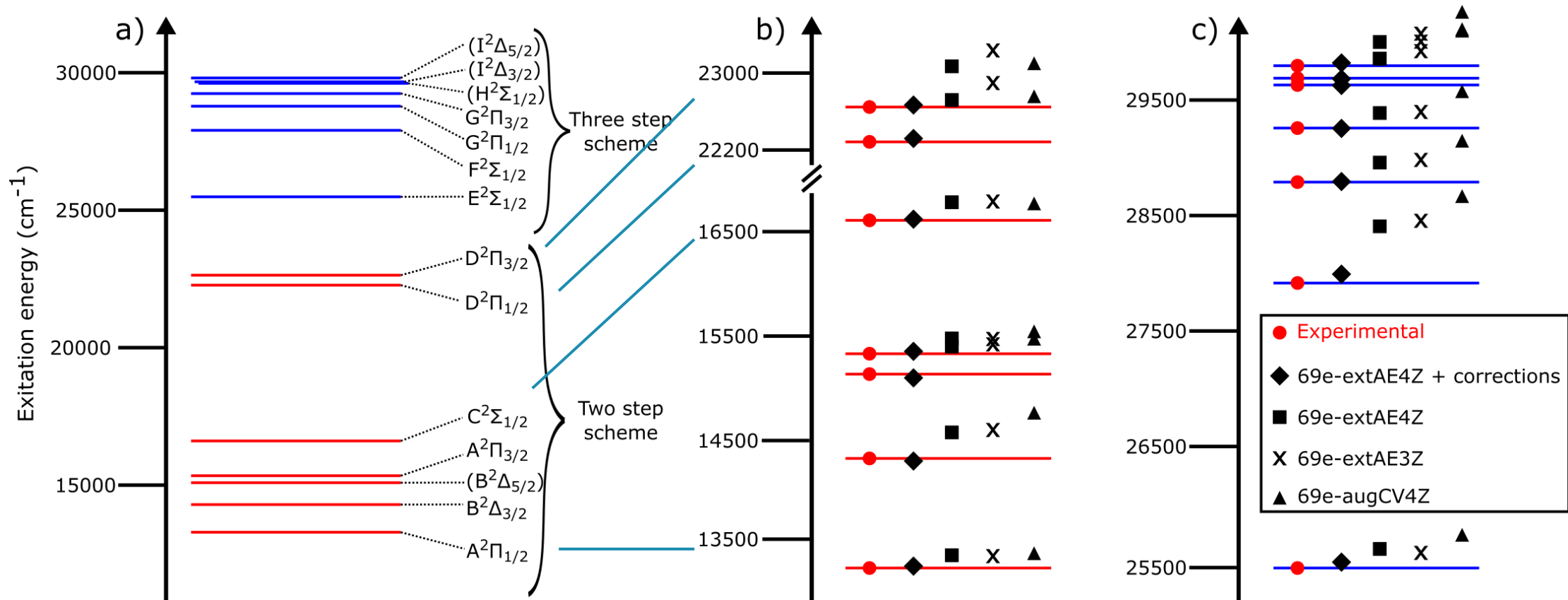
AC stark effect (red)

High resolution ^{223}RaF spectrum (Q-branch)



- Measured changing the wavelength of the second step (J-selectivity).
- Allows the access of different rotational lines in the Q-branch (e.g., low and high J)

Final level scheme of RaF – all levels below 30.000 cm⁻¹



- All predicted electronic levels below 30,000 cm⁻¹ have been measured and analyzed.

Nuclear moments from molecules

The measurement of nuclear moments using molecules is not new, and several isotopes have their current reference moments from molecular measurements:

Magnetic dipole moments

- ^{35}S
- ^{79}Se
- ^{21}Ne

Electric quadrupole moments

- ^2H , HD
- ^{33}S , CS and SiS
- ^{37}Cl , BrCl
- $^{39-41}\text{K}$, KF and KCl, KBr
- ^{75}As , AsP
- ^{79}Br , HBr
- $^{115}\text{In}?$, InF and InI
- ^{121}Sb , SbX, X=N,P,F,Cl
- ^{131}Xe , Six molecules
- ^{139}La , LaX, X=F,I
-