



Exploring N₂ Capturing in Liquid Argon using Li-FAU Mol Sieve in the Iceberg Cryostat

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Background

The DUNE

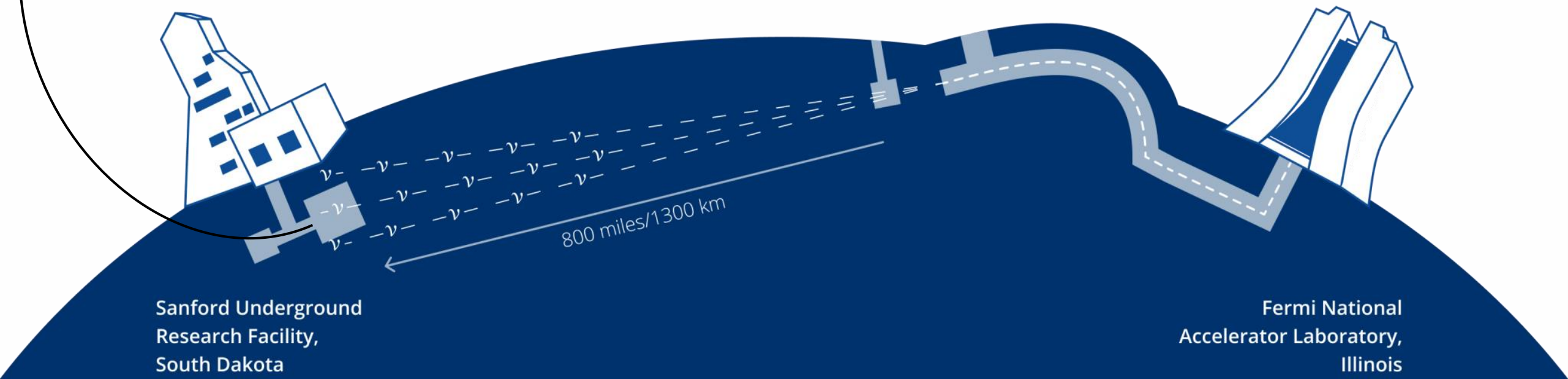
The **Long-Baseline Neutrino Facility (LBNF)**, located at the Sanford Underground Research Facility in Lead, South Dakota, hosts the **Deep Underground Neutrino Experiment (DUNE)**.

- the cryostats contain 17,500 metric tons of ultra-pure liquid argon (LAr) each (~ 5 Olympic pools)

- How pure should the argon be for the experiment?

Detector 1 → < ~ 100 ppt (or ~ 1.4 ml of oxygen equivalent)

Detector 2 → < ~ 50 ppt (or ~ 0.7 ml of oxygen equivalent)



Sanford Underground
Research Facility,
South Dakota

Fermi National
Accelerator Laboratory,
Illinois

Motivation

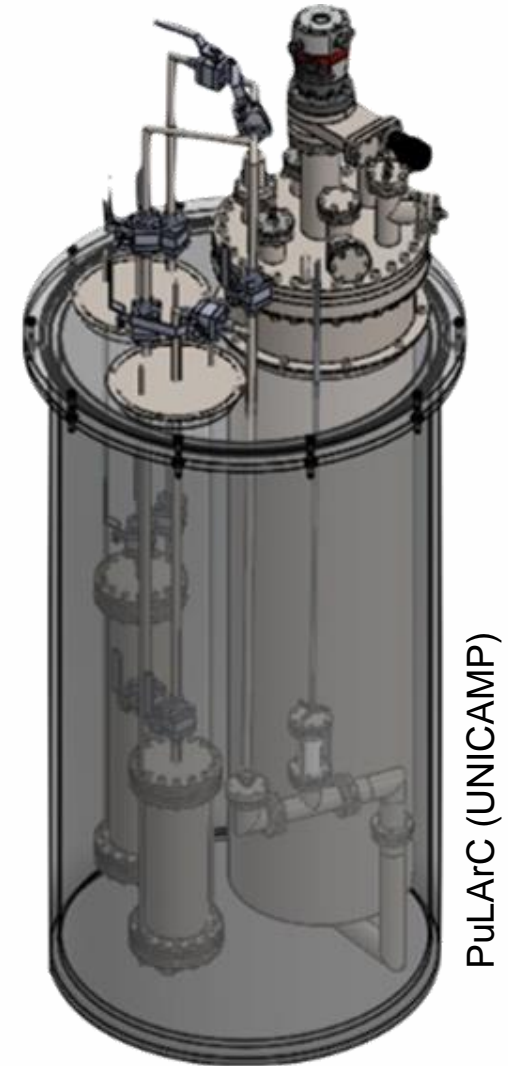
How do you reach such pure argon?

Contaminants in argon **significantly affect** experimental results:

- N_2 , O_2 and H_2O
- The argon is purified using a **filtering media** (which captures O_2 and H_2O);
- As the typical filters can't remove N_2 , it must be kept low in the argon purchasing specification.

An effective removal of N_2 impurities during liquid argon circulation was not known and it is desirable.

- A systematic study incorporating collaborative materials and methods, and combining numerical and experimental analysis;
- **Li-FAU zeolite.**



PuLArC (UNICAMP)

Experiments

Liquid argon purification

Experiments at Unicamp – Purification Liquid Argon Cryostat (PuLArC)

- ~ 1.2 kg of Li-FAU zeolite adsorbent;
- N₂ contamination reduction: 20-50 ppm to 0.1-1.0 ppm (1-2 hours of active circulation)

Experiments at Fermilab – Integrated Cryogenic Experiment for Beam Energy Research and Generation (ICEBERG) Cryostat

- ~ 3 kg of Li-FAU zeolite adsorbent;
- N₂ contamination reduction: ~5 ppm to <1 ppm (96-hour cycles without active circulation).
- calibrated gas analyzer – monitor N₂ dissolved in LAr.

Measure concentration of N₂ dissolved in LAr with passing time:
Recontamination after each cycle.



PuLArC (UNICAMP)



ICEBERG (Fermilab)

Mathematical Model

How was the system modeled?

Transient mass balance in fluid phase, intra-particle, solid phase, and storage tank modeling.

Hypotheses:

- isothermal system;
- constant physical properties;
- plug flow;
- constant mass flow;
- uniform bed porosity;
- well-mixed tank.

Fluid-phase

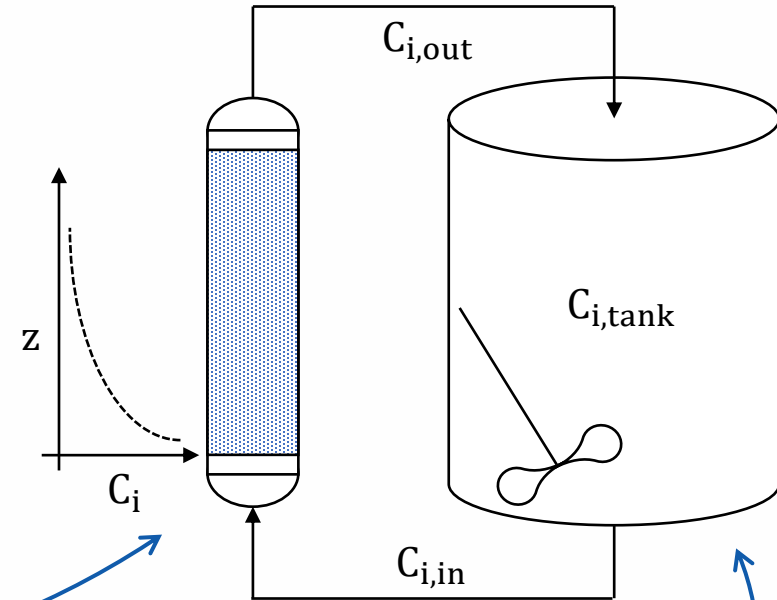
$$\frac{\partial C_i}{\partial t} = D_L \frac{\partial^2 C_i}{\partial z^2} - v_z \frac{\partial C_i}{\partial z} - \frac{(1 - \epsilon_b)}{\epsilon_b} K_L a_V (C_i - Y_i)$$

Intra-particle fluid-phase

$$\frac{\partial Y_i}{\partial t} = \frac{K_L a_V}{\epsilon_p} (C_i - Y_i) - \rho_s \frac{(1 - \epsilon_p)}{\epsilon_p} \frac{\partial q}{\partial t}$$

Solid-phase

$$\frac{\partial q_i}{\partial t} = \sum_{i=1}^{NC} \frac{\partial f_{eq}(Y_i)}{\partial Y_i} \frac{\partial Y_i}{\partial t} \quad f_{eq}(Y_i) = \frac{K_{RP} Y_i}{1 + a_{RP} Y_i^g}$$



Well-mixed tank

$$\frac{\partial C_{i,tank}}{\partial t} = \frac{C_i Q}{V} - \frac{C_{i,in} Q_{in}}{V} - \frac{C_{i,in}}{V} \cdot \frac{dV}{dt}$$

Mathematical Model

Closure equations

Adsorption isotherm:

Redlich-Peterson:

$$f_{eq}(Y_i) = \frac{K_{RP} Y_i}{1 + a_{RP} Y_i^g}$$

K_{RP} constant related to adsorption capacity (m³/kg)
 a_{RP} related to the max amount adsorbed in the first layer
 g adsorption intensity

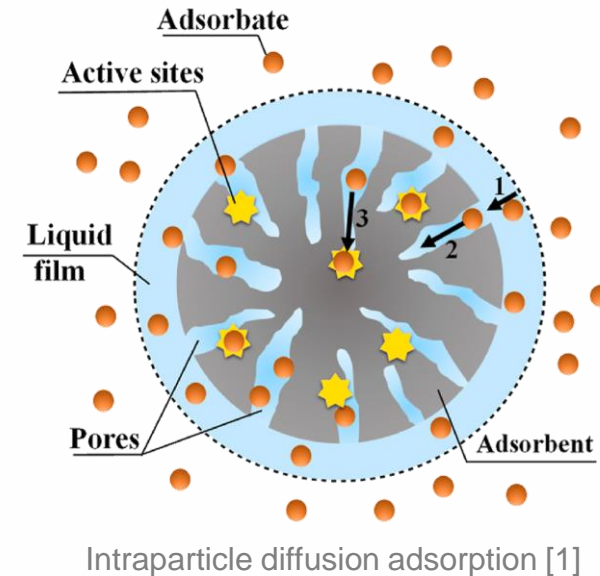
Predicting the filter's behavior is important to determine its capacity – specifying the saturation time and number of cycles needed to reach the required purity

Mass transfer coefficient :

$$\frac{1}{K_L} = \frac{1}{k_e} + \frac{1}{\epsilon_p k_i}$$

ϵ_p intraparticle porosity
 k_e molecular diffusivity of N₂ in Ar
 $k_i = \frac{10 D_L}{\tau d_p}$ particle tortuosity

$$Sh = \frac{k_e d_p}{D_L} \quad \frac{Sh}{Sc^{1/3} Re_m} = 0.91 Re_m^{-0.51} \quad Sc = \frac{\mu_m}{\rho_L D_L}$$



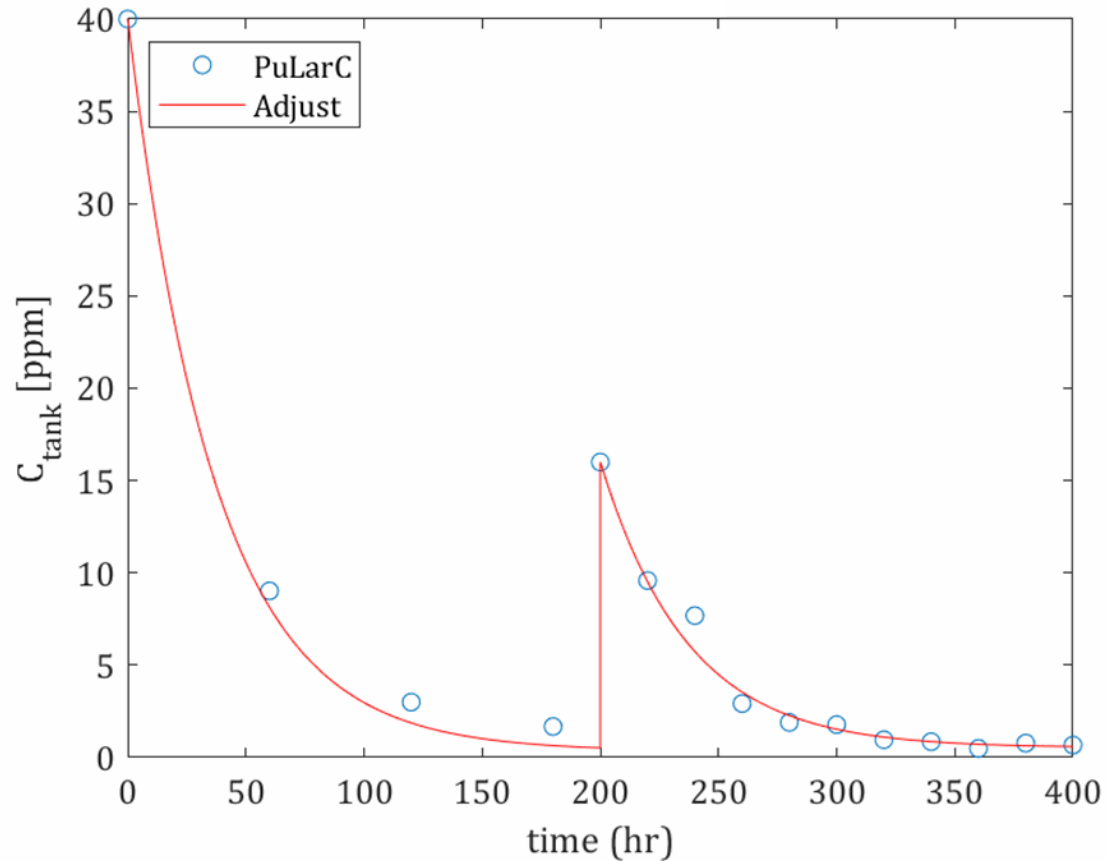
Model Solution

PuLarC parameters optimization

PuLarC simulation conditions:

- simulation – 2 cycles (media is not regenerated before the second cycle);
- $C_{N_2,1} = 40$ ppm
- $C_{N_2,2} = 16$ ppm
- $Q = 4$ L/min

- tank volume: 0.14 m³
- media level: 29 cm
- bed diameter: 95.5 cm
- bed porosity: 0.38
- $\rho_S = 936$ kg/m³
- $d_p = 2$ mm



$$f_{eq}(Y_i) = \frac{K_{RP} Y_i}{1 + a_{RP} Y_i^g}$$

Isotherm parameters

K_{RP}	$1.69 \cdot 10^6$
a_{RP}	$3.47 \cdot 10^7$
g	0.911

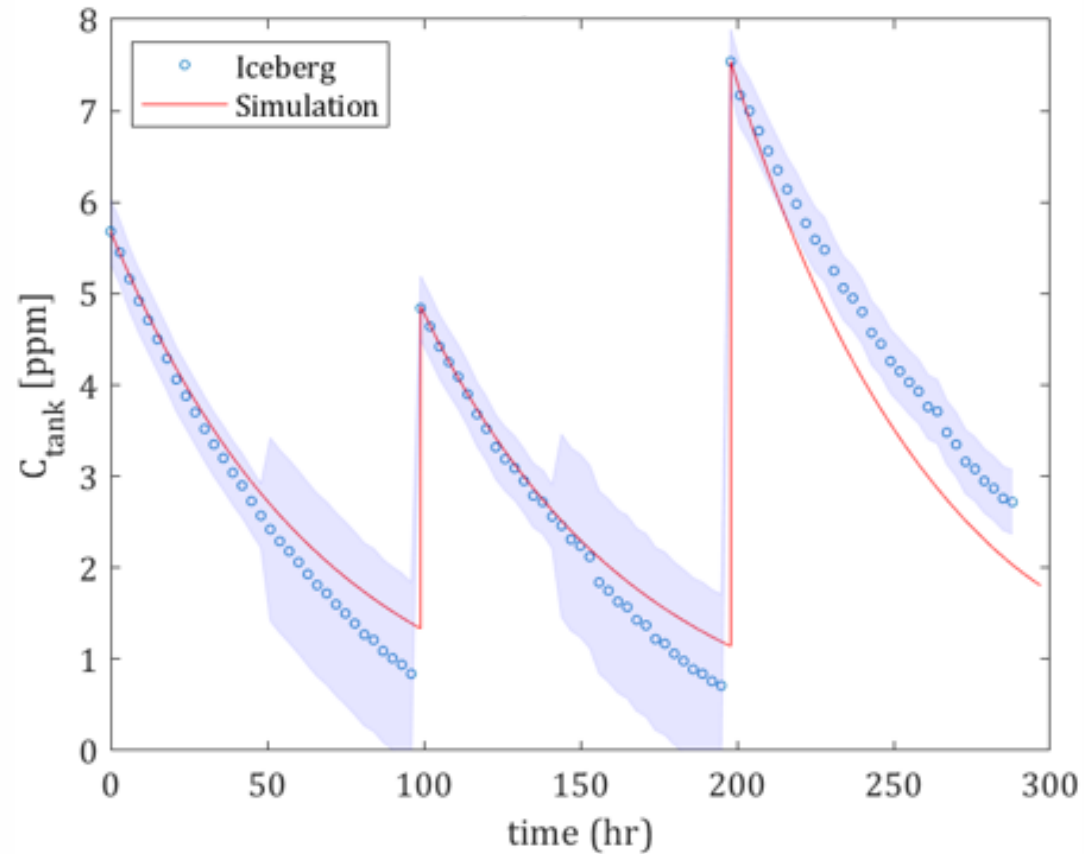
Model Solution

Iceberg simulation with PuLarC parameter

Iceberg simulation conditions:

- simulation – 3 cycles (media is not regenerated before the second and third cycles);
- $C_{N_2,1} = 5.68$ ppm
- $C_{N_2,2} = 4.48$ ppm
- $C_{N_2,1} = 7.54$ ppm
- $Q = 0.61$ L/min

- tank volume: 2.5 m^3
- media level: 47 cm
- bed diameter: 12 cm
- bed porosity: 0.38
- $\rho_S = 936 \text{ kg/m}^3$
- $d_p = 2 \text{ mm}$



$$f_{\text{eq}}(Y_i) = \frac{K_{\text{RP}} Y_i}{1 + a_{\text{RP}} Y_i^g}$$

Isotherm parameters

K_{RP}	$1.69 \cdot 10^6$
a_{RP}	$3.47 \cdot 10^7$
g	0.911

Model Solution

Iceberg simulation with PuLarC parameter

Iceberg simulation:

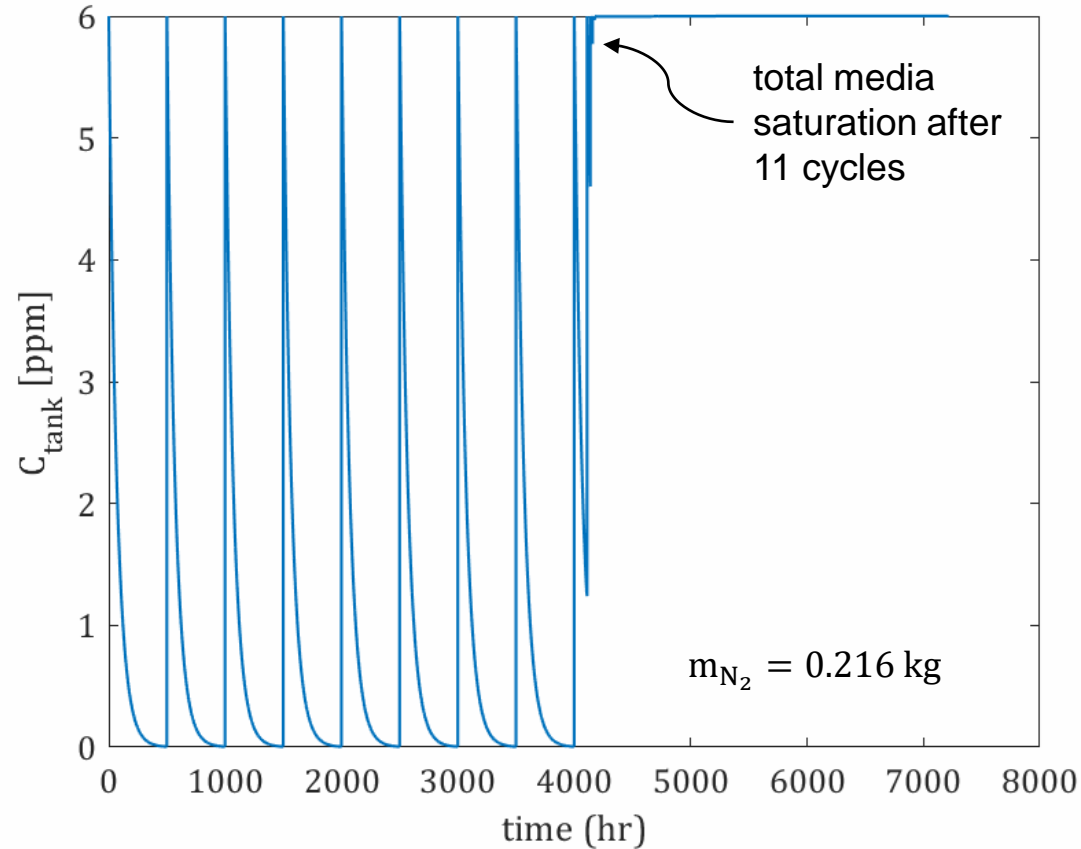
- simulation – 20 cycles (media is not regenerated before each cycle);

- $C_{N_2, in} = 6 \text{ ppm}$
- $Q = 0.61 \text{ L/min}$

- tank volume: 2.5 m^3
- media level: 47 cm
- bed diameter: 12 cm
- bed porosity: 0.38
- $\rho_S = 936 \text{ kg/m}^3$
- $d_p = 2 \text{ mm}$

- Simulation stops when:

$$\frac{dC_{N_2}}{dt} < 10^{-8} \left(\frac{\text{kg}}{\text{m}^3 \text{s}} \right)$$



$$f_{\text{eq}}(Y_i) = \frac{K_{\text{RP}} Y_i}{1 + a_{\text{RP}} Y_i^g}$$

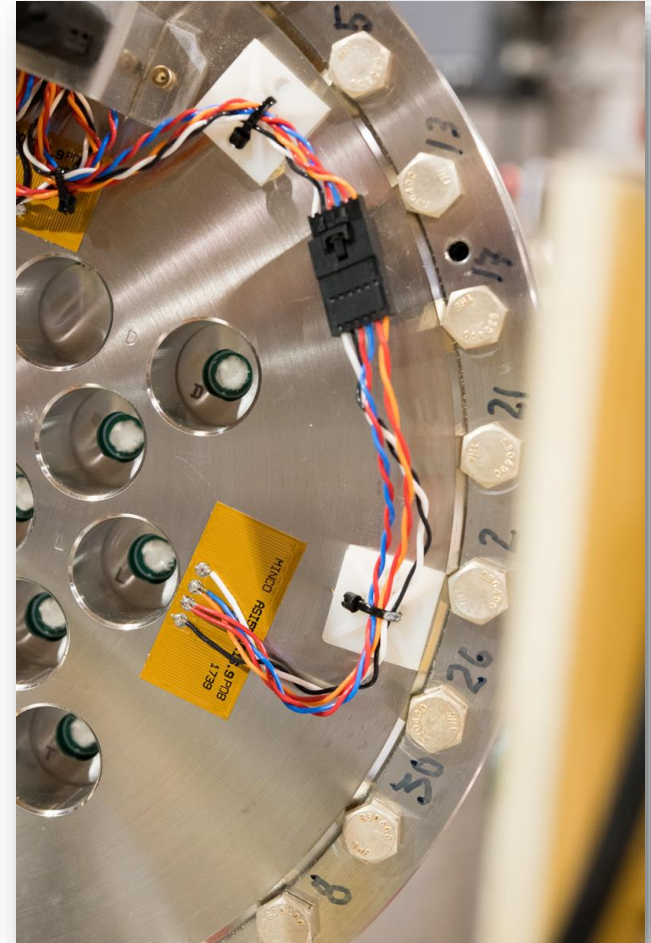
Isotherm parameters

K_{RP}	$1.69 \cdot 10^6$
a_{RP}	$3.47 \cdot 10^7$
g	0.911

Final Remarks

Conclusion and Outlook

- The successful testing of Li-FAU as an **effective adsorbent** for N_2 impurities marks a significant advancement for neutrino experiments, improving LAr quality. (see recent article at <https://news.fnal.gov/2024/08/brazilian-researchers-discover-new-way-to-purify-liquid-argon-for-neutrino-experiments/>)
 - Promising results from the PuLArC and Iceberg cryostats – Li-FAU could be a viable alternative to traditional purification (molecular sieve 4A) in large-scale LAr cryostats.
 - The purification media will improve the quality of the LAr used in experiments + allow for removing N_2 contaminants if air is accidentally introduced into the cryostat due to equipment malfunction.
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- A mathematical model was proposed for the adsorption process of nitrogen using the Li-FAU in liquid;
 - The proposed model was able to match the experimental results obtained with the ICEBERG cryostat;



ICEBERG (Fermilab)

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Thank You for your attention!