

Model development and optimization of cryogenic mixed-refrigerant cycles with phase separators

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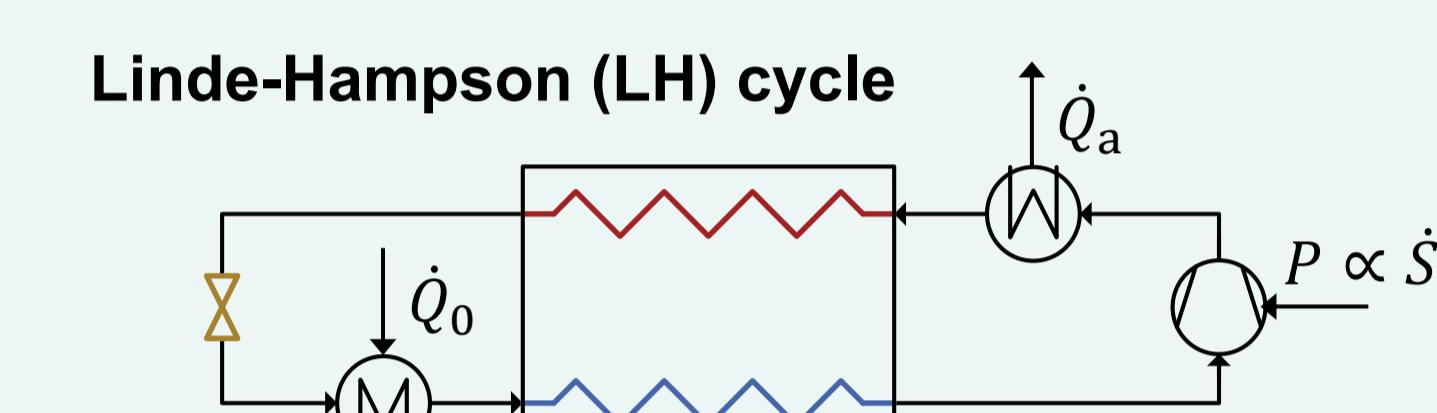
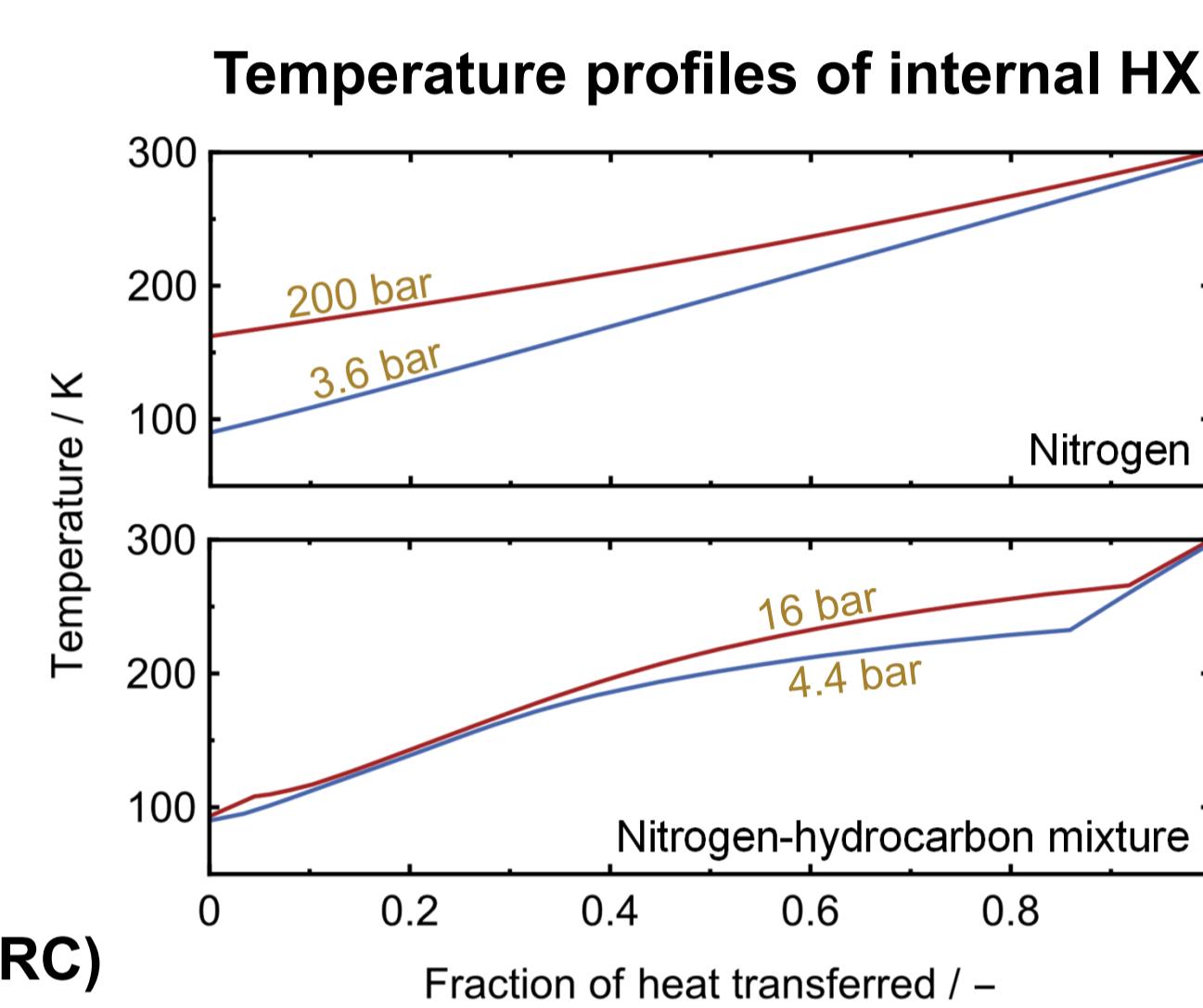
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Motivation

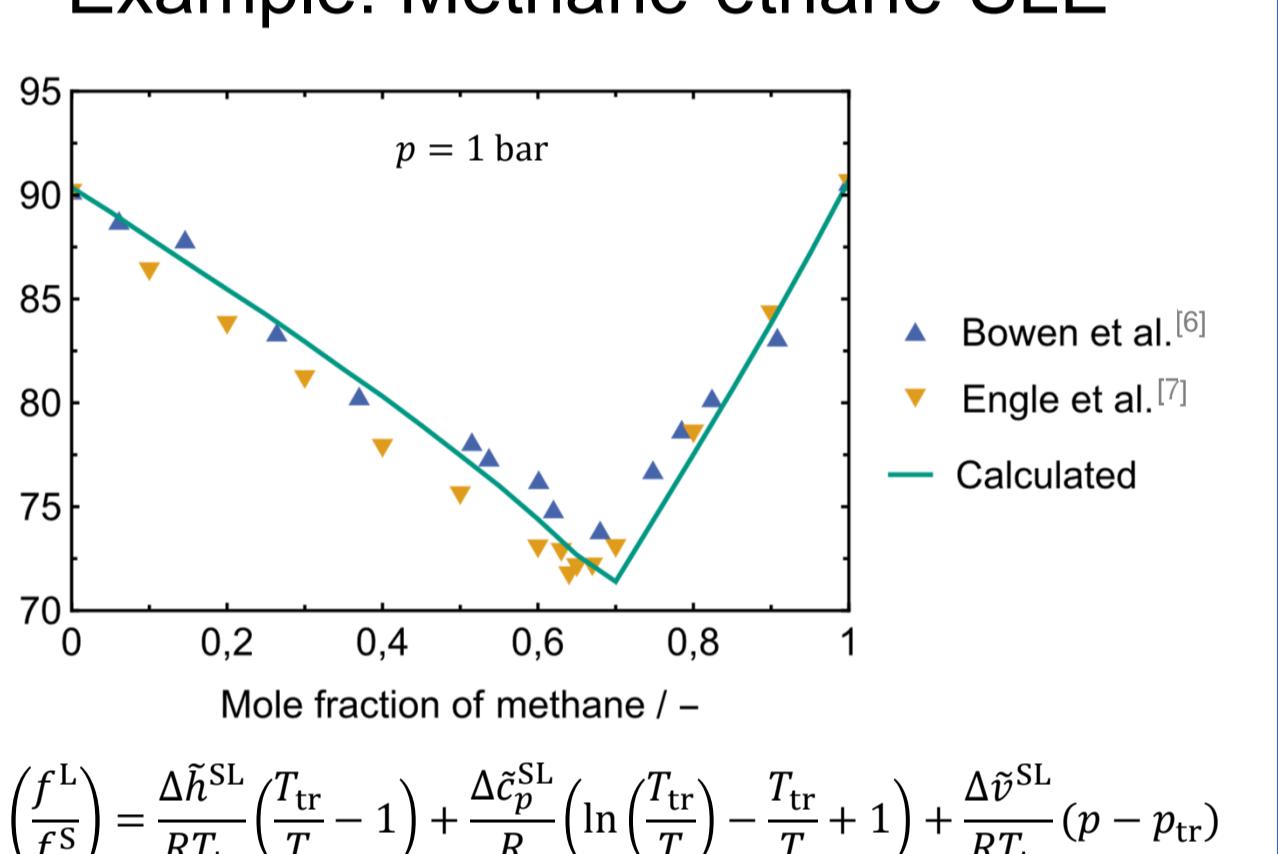
- Efficient cooling at low temperatures
 - ULT-freezers, $T_0 \leq 200$ K
 - HTS cooling, $T_0 \leq 80$ K
- Wide-boiling refrigerant mixtures
- Freezing of high-boiling components limits efficiency at low temperatures
 - CMRC cascade
 - Auto-cascade refrigeration cycle (ARC)



Properties of refrigerant mixtures

- Cubic equations of state
 - Peng-Robinson^[1]
 - α -function by Twu et al.^[2,3]
- Open source tool CoolProp^[4]
- Freezing point estimation
 - Solid-liquid-equilibrium: $f_i^L = f_i^S$
 - Assume pure solid: $f_i^S = f_{0i}^S$
 - Calculation of f_{0i}^S with f_{0i}^L according to Prausnitz et al.^[5]

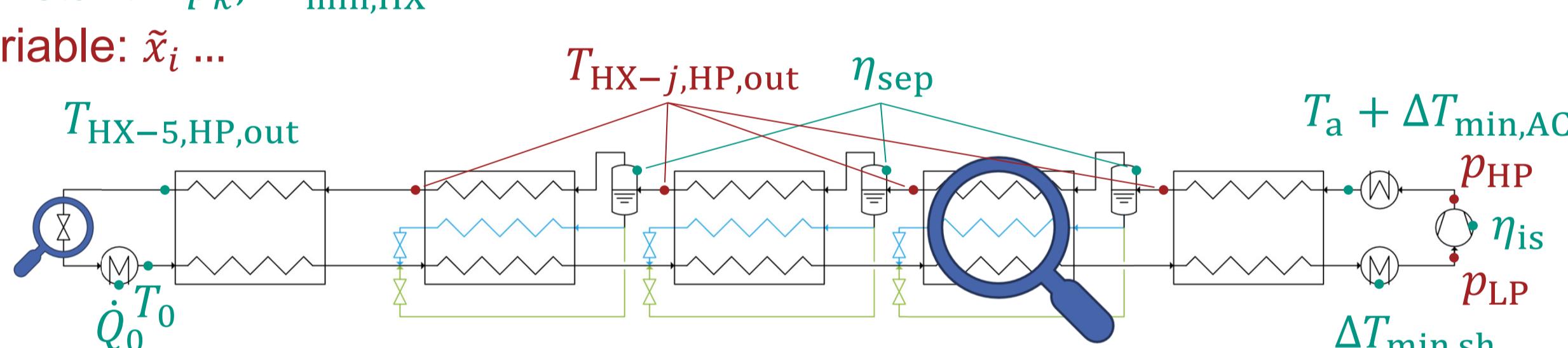
Example: Methane-ethane-SLE



Process modeling

 Constant: $\Delta p_k, \Delta T_{\min, HX} \dots$

 Variable: $\tilde{x}_i \dots$

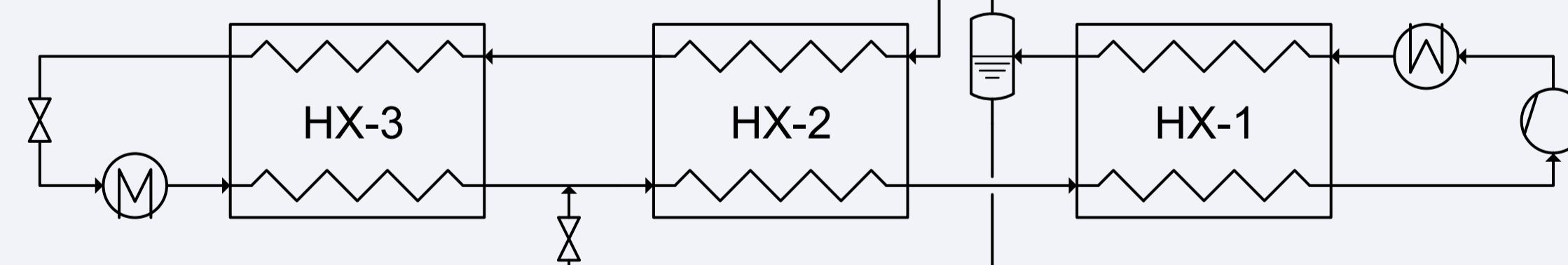
 Model built in Wolfram Mathematica^[8]


General assumptions:

- Steady state
- Adiabatic components and pipes
- Constant pressure losses

- Objective: Calculation of power demand
- Consistency checks
 - Temperature profiles
 - Freezing point estimation

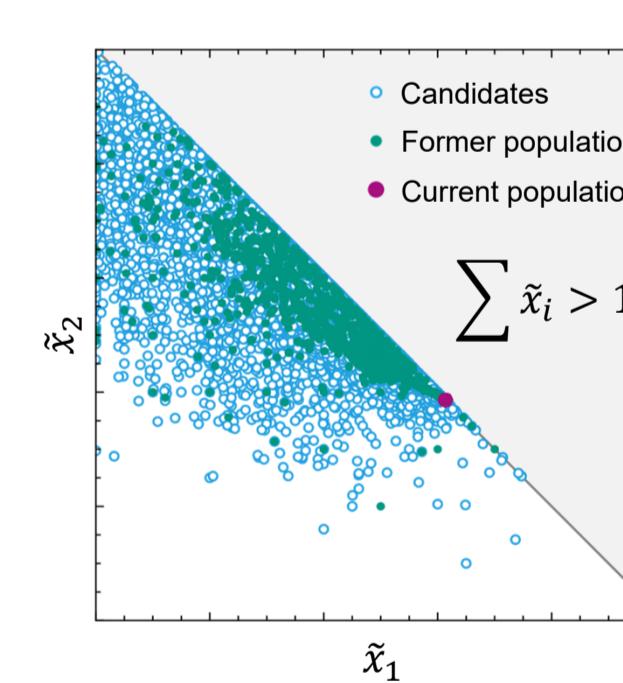
ARC-13 with 1 phase separator and 3 internal heat exchangers (HX)



Optimization

Differential Evolution^[9,10]

- Evolutionary algorithm
- No derivatives required
- Global optimization^[11]



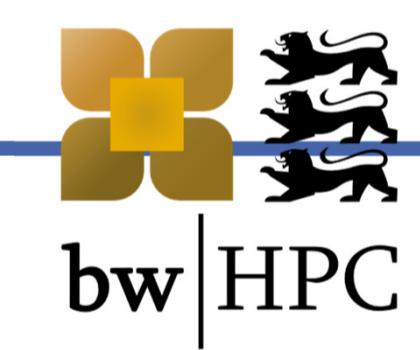
Initialization

Mutation

Crossover

Selection

Optimum

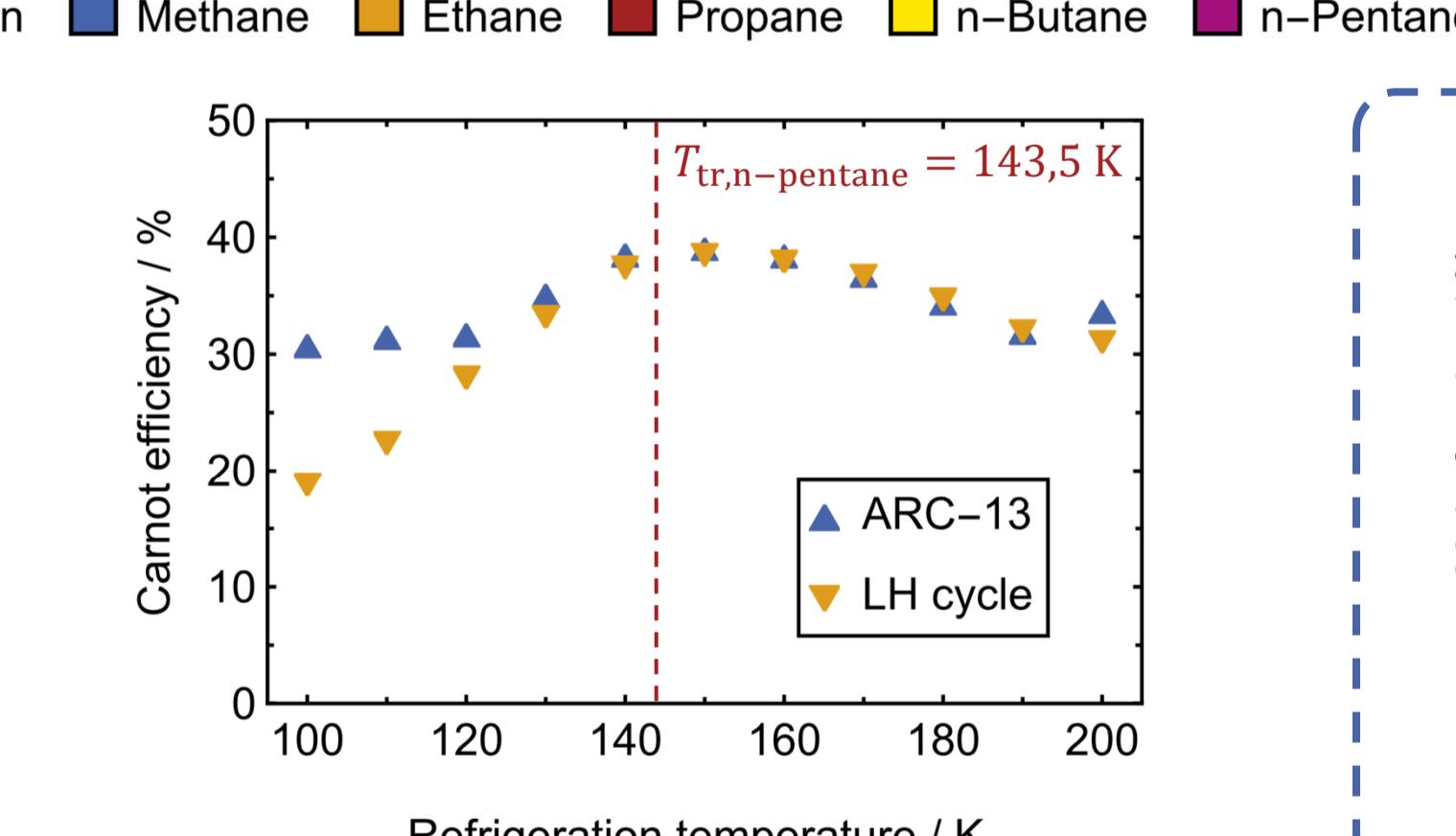
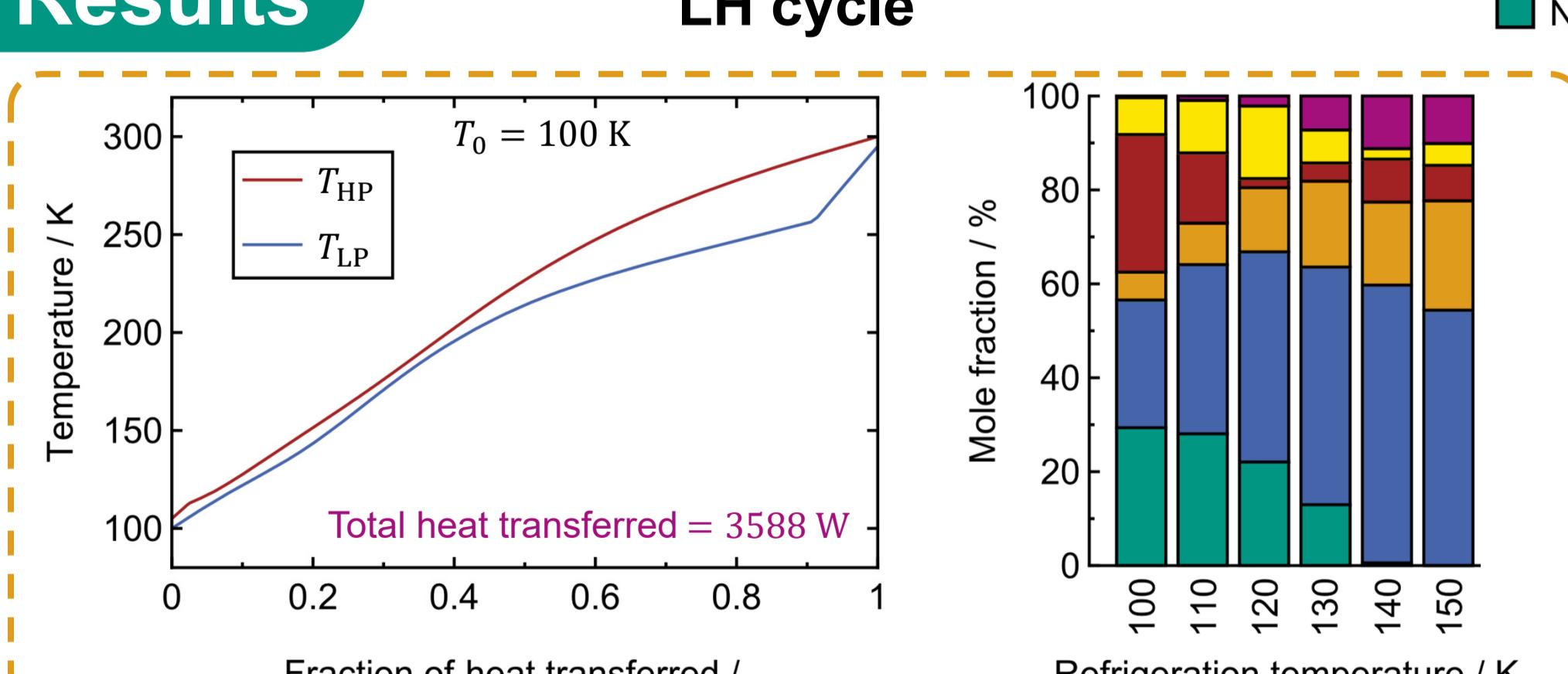


bwHPC [12]

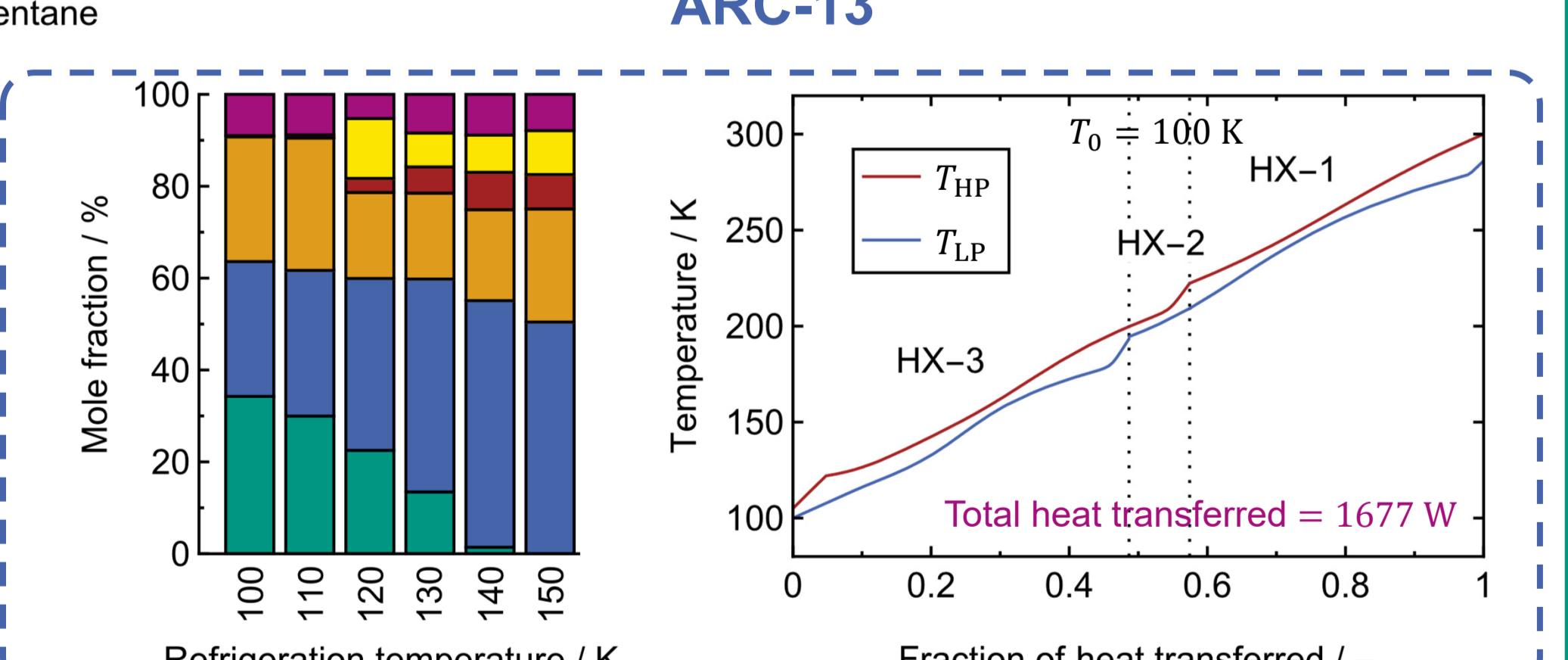


Results

LH cycle



ARC-13



[1] D. Peng and D. Robinson, "New Two-Constant Equation of State," Ind. Eng. Chem. Fundam., vol. 15, 1976.

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[4] I. H. Bell, J. Wronecki, S. Quoilin, and V. Lemort, "Pure and Pseudo-pure Fluid Thermophysical Property Evaluation and the Open-Source Thermophysical Property Library CoolProp," Ind. Eng. Chem. Res., vol. 53, no. 6, 2014.

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[6] L. Bowen et al., "Experimental measurement and model prediction of solid-liquid equilibrium for methane-ethane binary system," J. Chem. Thermodyn., vol. 172, 2022.

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[12] bwHPC, bwUniCluster 2.0. URL: <https://wiki.bwhpc.de/e/BwUniCluster2.0> (last checked: 05.05.2025)