

Measurements of F^- in irradiated straw tubes

Mattia Verzeroli¹, Maria Cristina Arena², Katerina Kutznezova⁴, Beatrice Mandelli⁵, Gianluca Rigoletti⁵

UNIVERSITÉ
DE LYON



EP-DT
Detector Technologies

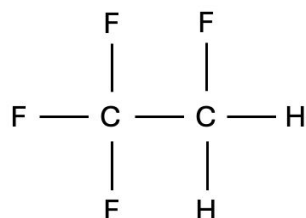
1. Université Claude Bernard Lyon I
2. Università degli Studi di Pavia
3. CERN
4. UF/INP

- Introduction:
 - F- Gases at LHC experiments;
 - F- production.
- Ion Selective Electrode measurements:
 - Technique introduction;
 - Previous tests.
- Dedicated Setup @904;
- Measurements:
 - Overview of the measurements;
 - Data selection and data analysis;
 - Measurement results.
- Conclusion.

F-gases at LHC experiments

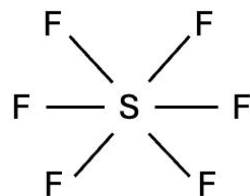
Fluorinated gases are used in several LHC experiment gaseous detectors

R134a
GWP: 1430



ATLAS
RPC

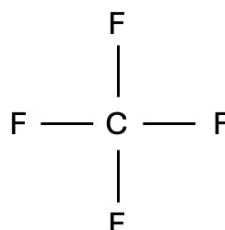
SF₆
GWP: 23500



ALICE
TOF, MID

CMS
RPC

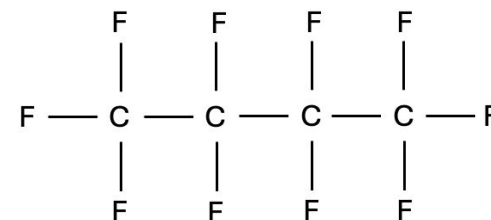
CF₄
GWP: 6630



CMS
CSC

LHCb
MWPC, RICH2

C₄F₁₀
GWP: 8860



LHCb
RICH1

EU fluorinated gases regulation:

(2014)

- Limiting the availability;
- Banning the use of F-gases where alternatives are available.

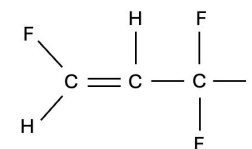
The **research on new eco friendly gas mixture** is ongoing, together with other strategies.

However

some of the promising alternatives still F-gases

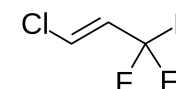
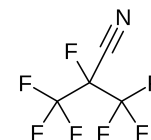
[2023, ECOgas@GIF++](#)
[2024, M. Verzeroli et al.](#)
[2024, G. Proto et al.](#)

...



HFO-1234ze
GWP: 7

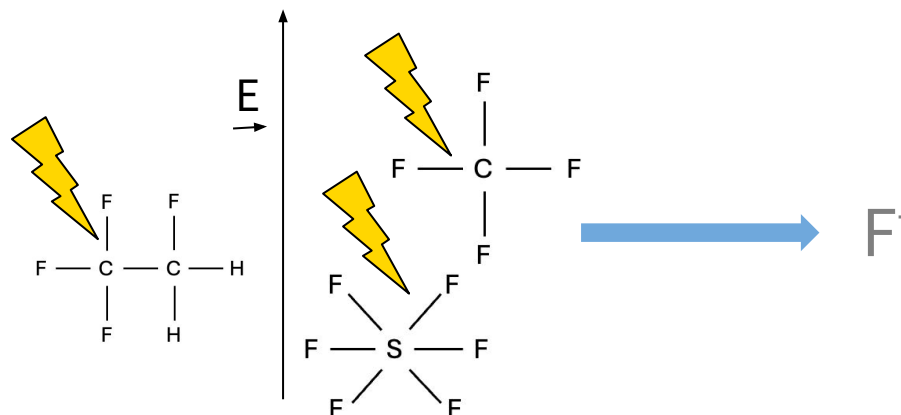
NOVEC 4710
GWP: 2100



HFO-1233zd
GWP: 2100

It is essential to investigate the production of F⁻ impurities not only in the newly proposed environmentally friendly gases but also in the ones currently in use, to establish a clear and direct comparison between them.

The combination of the electric field and high irradiation fragments F-gases in the detector's gas gap.



The **effect of F⁻ impurities** is strictly connected to the detector's design:

RPC

Free F⁻ can react with the water vapour forming hydrofluoric acid:
-> Detector's gap and gas system damages.

Increasing ageing effects

CSC

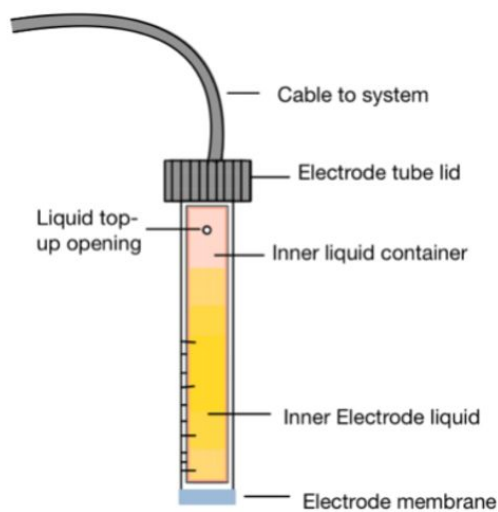
Free F⁻ can react with the wire deposit, and remove it:
-> Wire cleaning effect.

Reducing ageing effects

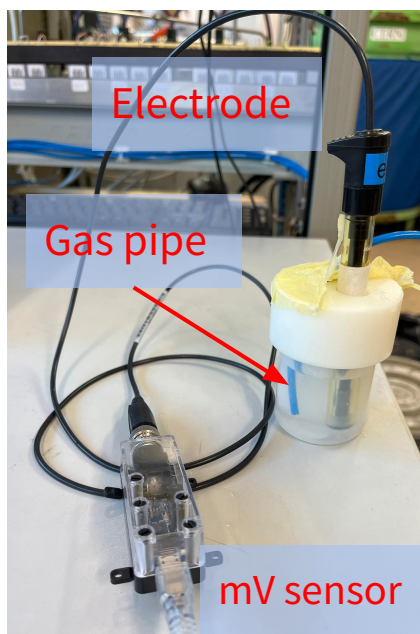
Ion Selective Electrode Measurements

Potentiometric technique that allows to measure the concentration of F⁻ in a specified solution of water + TISAB (Total Ionic Strength Adjustment Buffer)

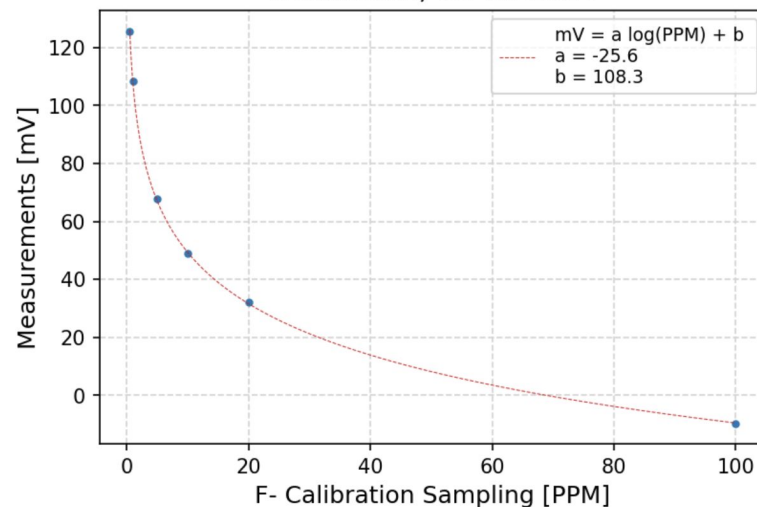
- Real-time measurements;
- Wide concentration measurements range;
- Inexpensive and easy to operate.



Example of ISE setup



Calibration, 02-08-2024



This measurements are not trivial, several parameters need to be fixed:

- Gas flow to the sampling bottle;
- Gas pressure;
- Volume of the sampling solution;
- Periodic calibration;
- ...

And you need to operate with an high irradiation!

Ion Selective Electrode Measurements

Where is it possible to find such high background irradiation conditions?

LHC experiments

GIF++ Facility

**Dedicated set up
@ 904**



- “Parasitic” measurements;
- Long sampling time needed (~months).

- Quite low statistic;
- -> less detectors;
- No free access to setup;
- Long and repeated measurements needed.

- Lower irradiation background;
- Long and repeated measurements needed.

- F^- as a function of Luminosity and Integrated Charge;
- Comparison with previous run (starting from 2017 RUN2);
- Excellent statistic
- -> sampling all detectors.

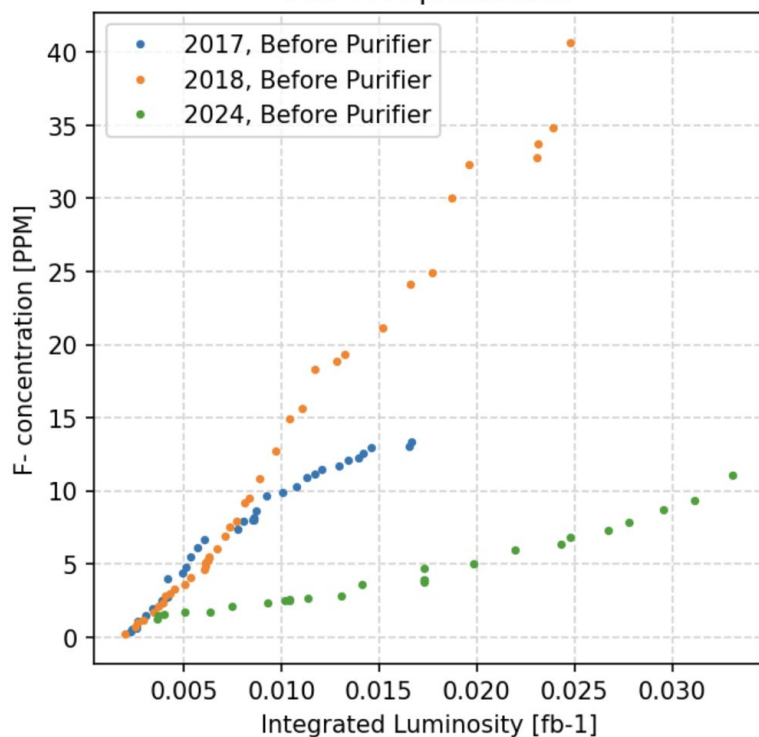
- F^- as a function of Gas Mixture, Integrated Charge and Background Irradiation;
- Setup could be used to improve the actual measurements method.

- F^- as a function of Gas Mixture and Integrated Charge;
- Short Pipes and Controlled Environment;
- Used with Straw detectors, but possibility to extend it to other technologies.

Ion Selective Electrode Measurements

LHC experiments: ALICE MID example

MID PPM production

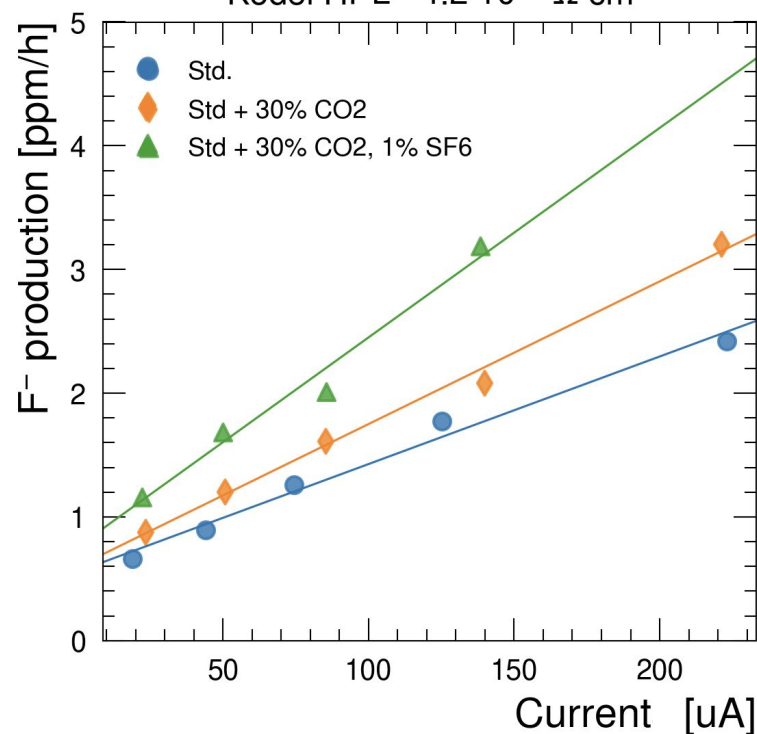


- Accumulated F⁻ over months of irradiation
- Good accuracy: sampling of the common exhaust of the 72 RPCs

[M. Verzeroli, RPC2024](#)
[B. Mandelli, JINST](#)

GIF++ facility EPDT RPC example

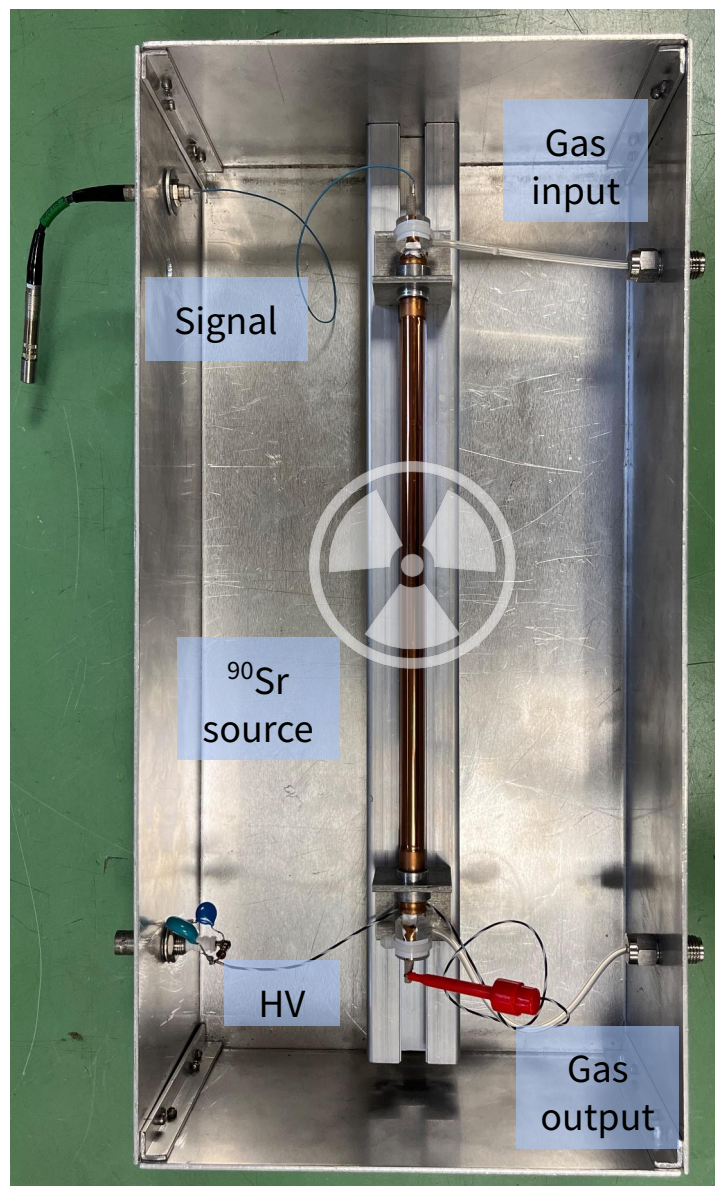
Kodel HPL - $4.2 \cdot 10^{10} \Omega \text{ cm}$



- Measurements for different background irradiation
- Measurements for different gas mixtures

[M. C. Arena, Stability 2023](#)
[G. Rigoletti, NIMA](#)

Dedicated setup @ Bld. 904



Straw detector:

- Selected for small gas volume and thin, transparent envelope to the source;
- \varnothing 1 cm, Length 21 cm;
- Provided by StrawTrackerRD team;
- Operated at 2140 V for the studied 40% Ar-based gas mixture.

^{90}Sr source:

- 235 MBq.

Sampling bottle:

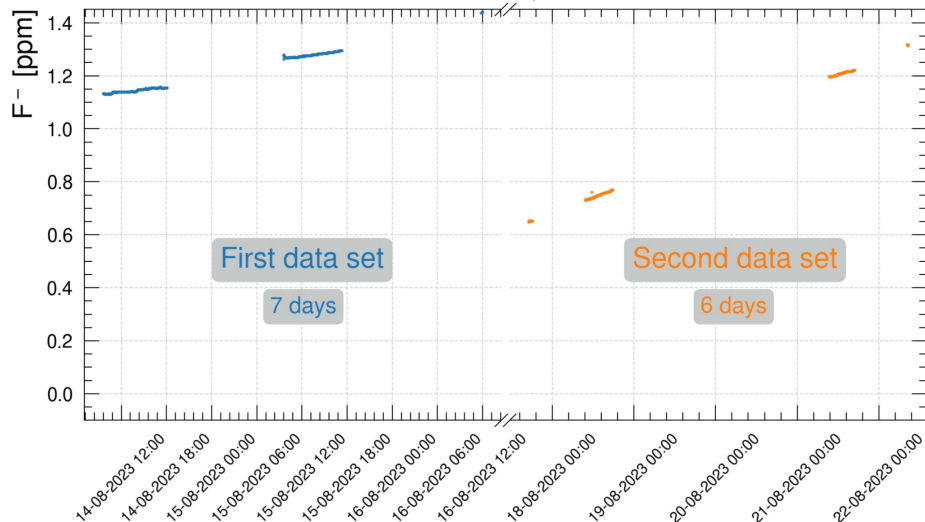
- 20 ml solution;
- Fixed 0.6 ln/h flow;
- Pipes length short as possible.

Measurements performed in collaboration with EP-DT-FS and CMS CSC group in the contest of the Summer Students project.

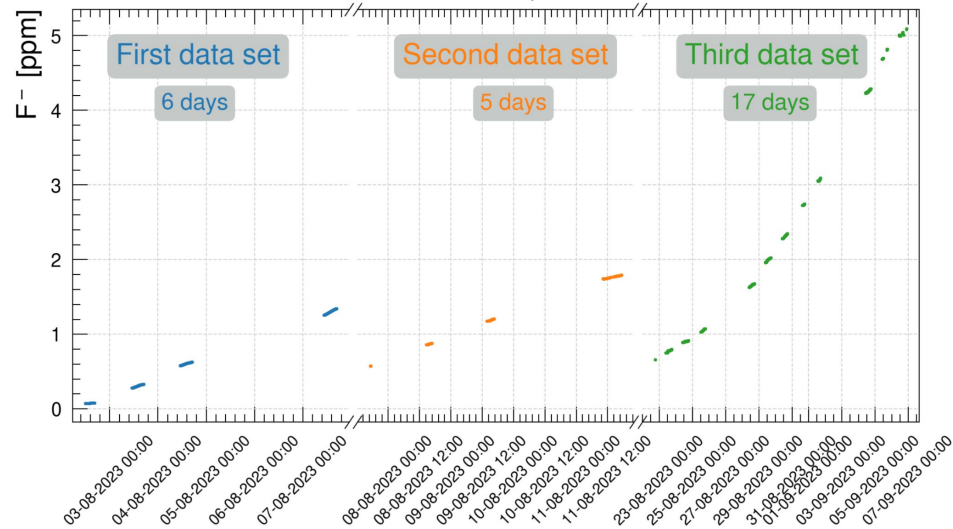
The **goal** is to use the straw detector to investigate the F^- production in CMS CSC gas, focusing on the CF_4 contribution to this production, and to prepare for the comparative measurements within searches for CF_4 alternatives.

Overview of the measurements

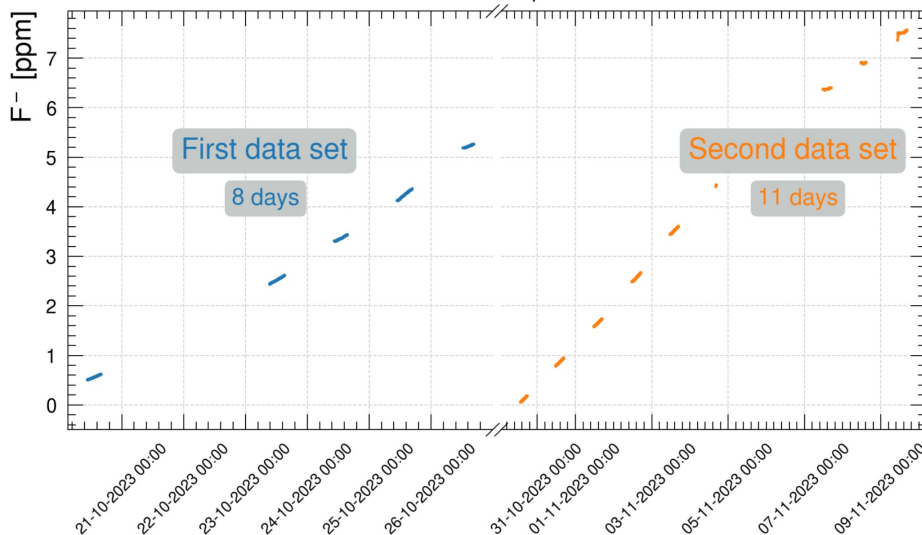
7% CF₄



10% CF₄



40% CF₄



7% CF₄: 2 data sets;
10% CF₄: 3 data sets.

EP/DF/FS + CMS CSC
Summer Students

40% CF₄: 2 data sets.

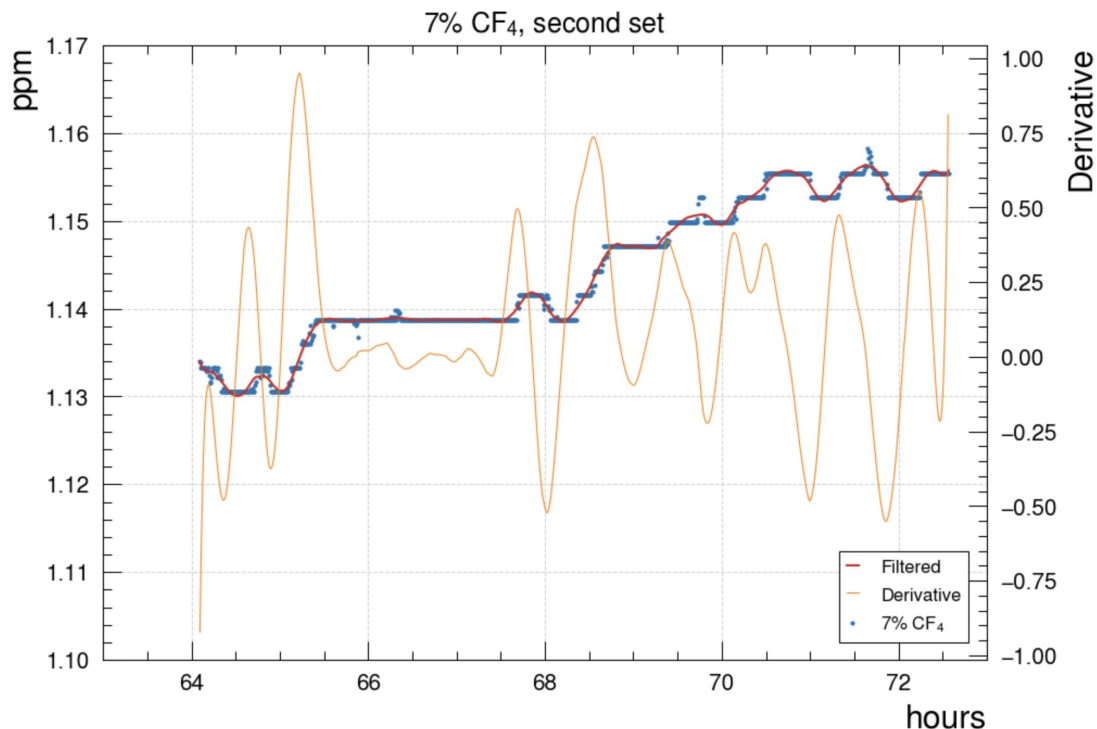
June - November
2023

CMS CSC
Student

Gas mixture:
Ar/CO₂/CF₄

With Ar 40% and variable
%CO₂ and %CF₄

Data selection and analysis



After ~8 hours of measurements,
not stable ppm increase:

- Change in gas flow;
- Change in temperature and pressure;
- Evaporation of the solution bottle;
- Electrode calibration non optimal;
- Difference in bubble surface;
- ...

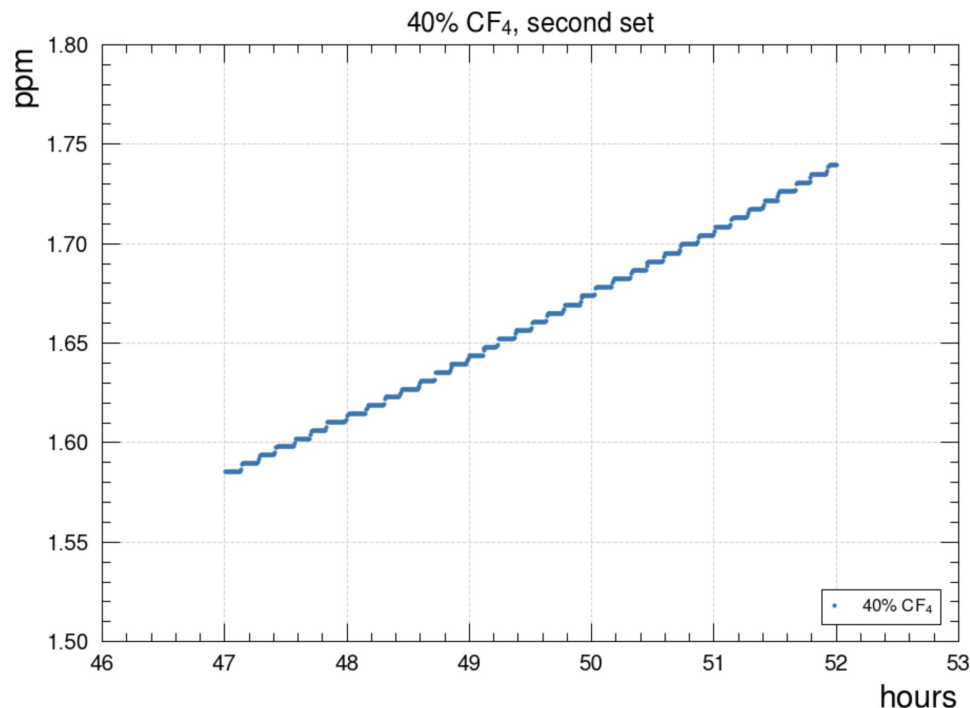
-> Not all the data shows a good reliability and a data selection need to be performed.

26 independent measurements -> 13 selected

Quite tricky measurements due to the amount of possibly systematic and environmental related errors...

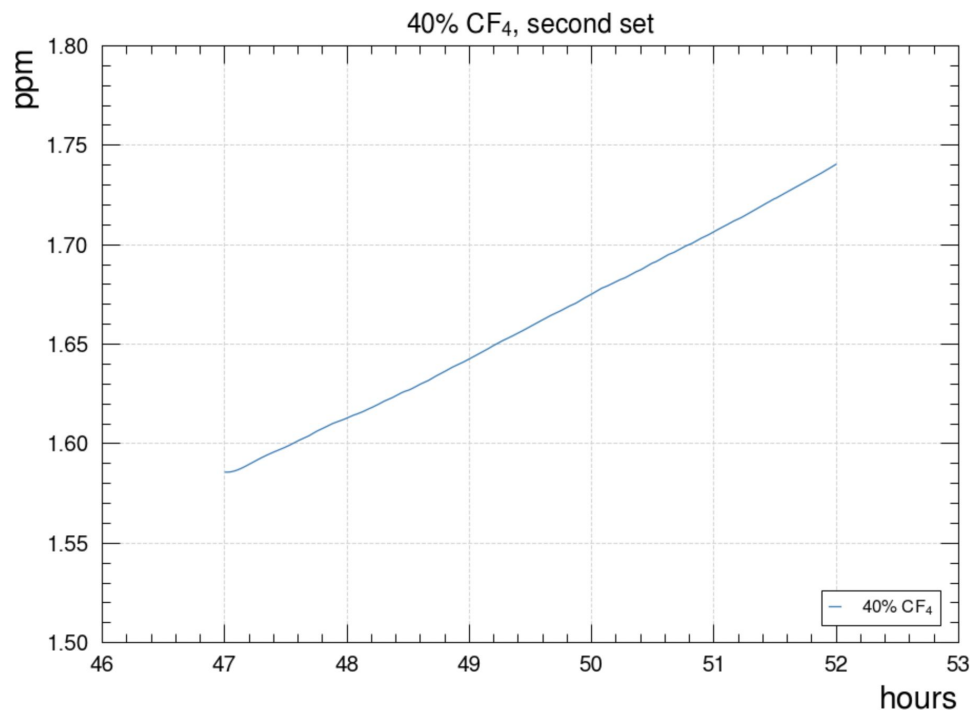
For the data selection, the following steps were considered:

1. Data smoothing with Savitzky-Golay filter (python-based);
2. Computation of the data derivative;
3. Selection of the bigger data interval where the derivative is stable (± 0.1);
4. Linear fit performed on this interval.



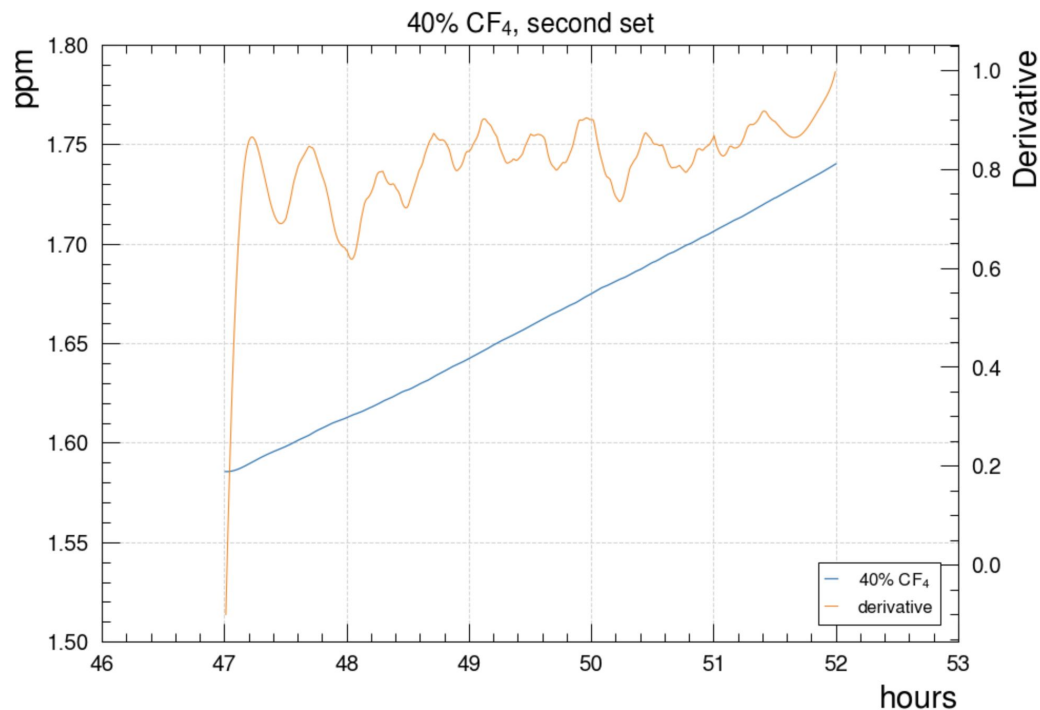
For the data selection, the following steps were considered:

1. Data smoothing with Savitzky-Golay filter (python-based);
2. Computation of the data derivative;
3. Selection of the bigger data interval where the derivative is stable (± 0.1);
4. Linear fit performed on this interval.



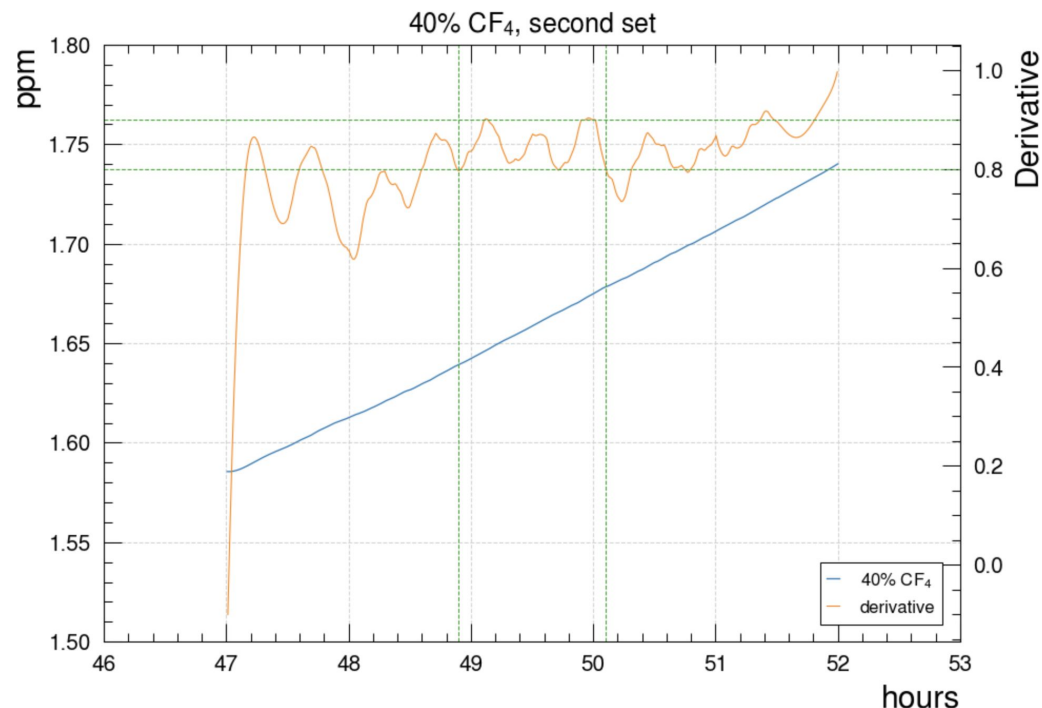
For the data selection, the following steps were considered:

1. Data smoothing with Savitzky-Golay filter (python-based);
2. Computation of the data derivative;
3. Selection of the bigger data interval where the derivative is stable (± 0.1);
4. Linear fit performed on this interval.



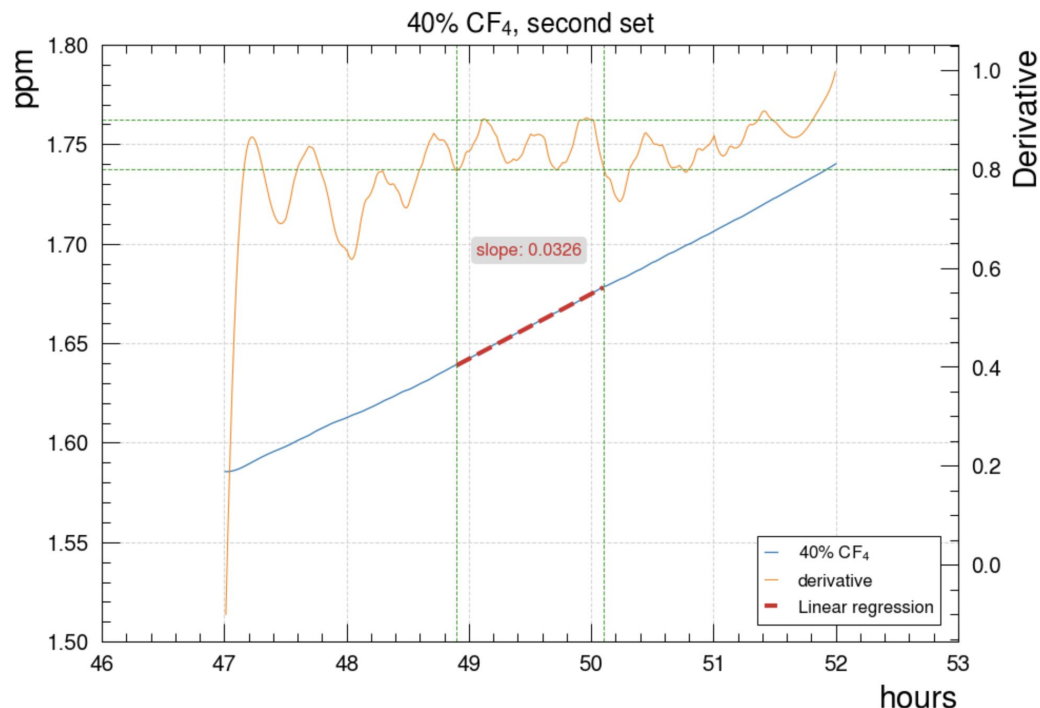
For the data selection, the following steps were considered:

1. Data smoothing with Savitzky-Golay filter (python-based);
2. Computation of the data derivative;
3. Selection of the bigger data interval where the derivative is stable (± 0.1);
4. Linear fit performed on this interval.



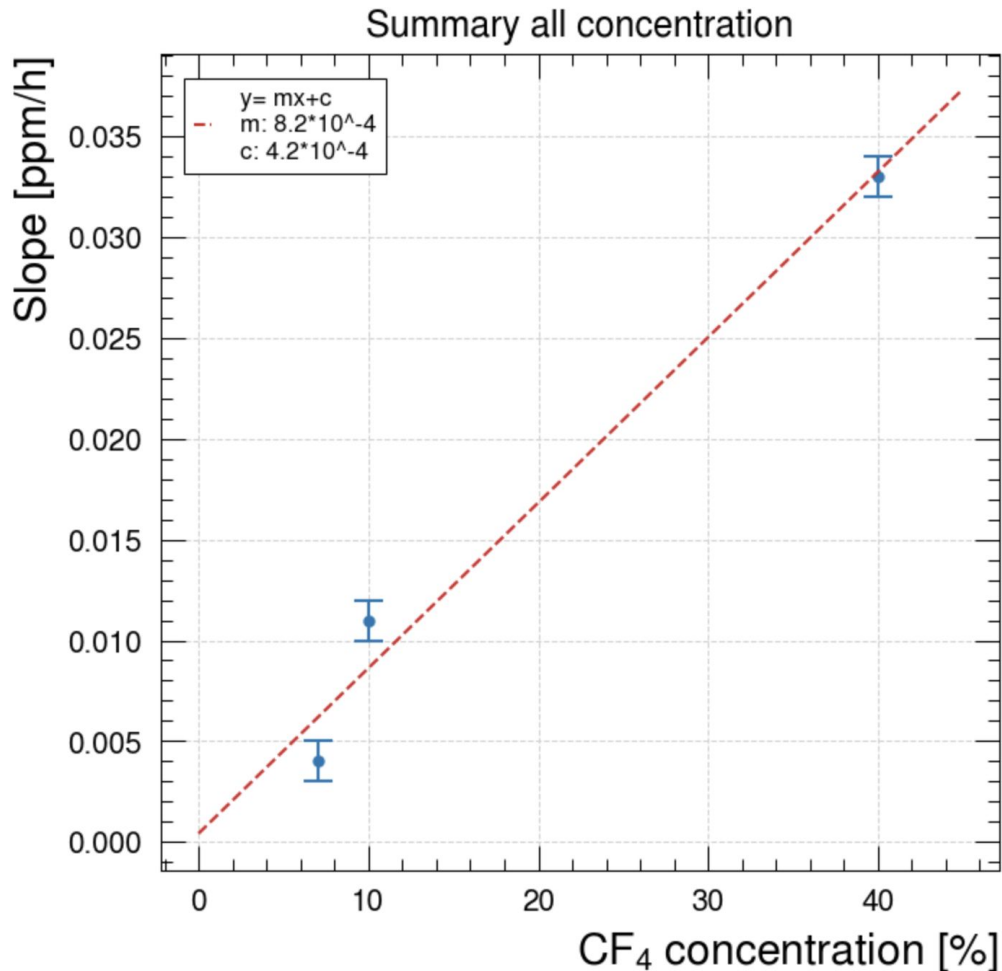
For the data selection, the following steps were considered:

1. Data smoothing with Savitzky-Golay filter (python-based);
2. Computation of the data derivative;
3. Selection of the bigger data interval where the derivative is stable (± 0.1);
4. Linear fit performed on this interval.



The slope is the ppm/h F⁻ production, and it is the result of the mean of the slope for fixed CF₄ concentration. Uncertainty is taken as mean standard deviation.

Measurements results



The increase of CF₄ brings to an increase of the ppm rate production:

The measurements suggest a linear trend, but more measurements are needed to confirm it.

- The dedicated new set up was commissioned during Summer 2023;
- Measurements show promising results for the study of the F⁻ impurities production:
 - Suggest a linear trend in the F⁻ production as a function of the CF₄ concentration;
- The set up is very flexible: possibility to study also other fluorinated gases (e.g. HFO) and other type of detectors.

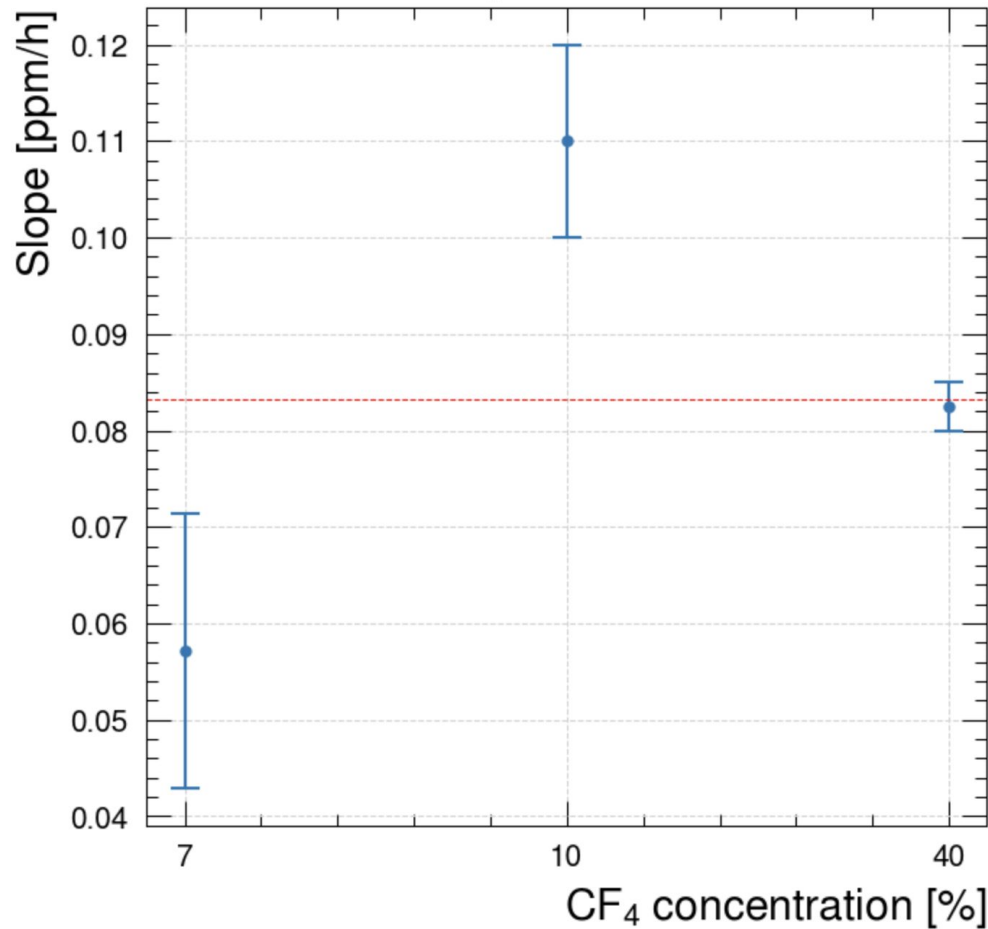
Future improvements:

- Increase the sensitivity adding a second straw detector;
- For CSC purpose: more measurements on the range 0-10% CF₄ concentration;
- Study of error sources (such as variation of flow)

Backup

Measurements results

Summary all concentration



$$F^{-}(\%CF_4) : F^{-}(\text{total } CF_4) = \%CF_4 : 100$$



(0.08 ± 0.02) ppm/h

if the detector operated
with 100% CF₄