

Electron Beam Ion Sources

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and

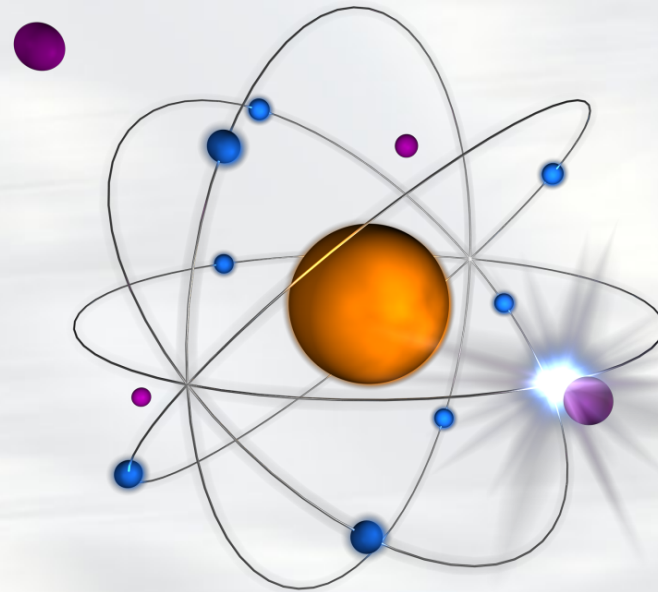
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Department of Physics

and

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Institute of Ion Beam Physics and Materials Research

Why Highly Charged Ions ?

Exciting properties of
highly charged ions



Properties of Highly Charged Ions

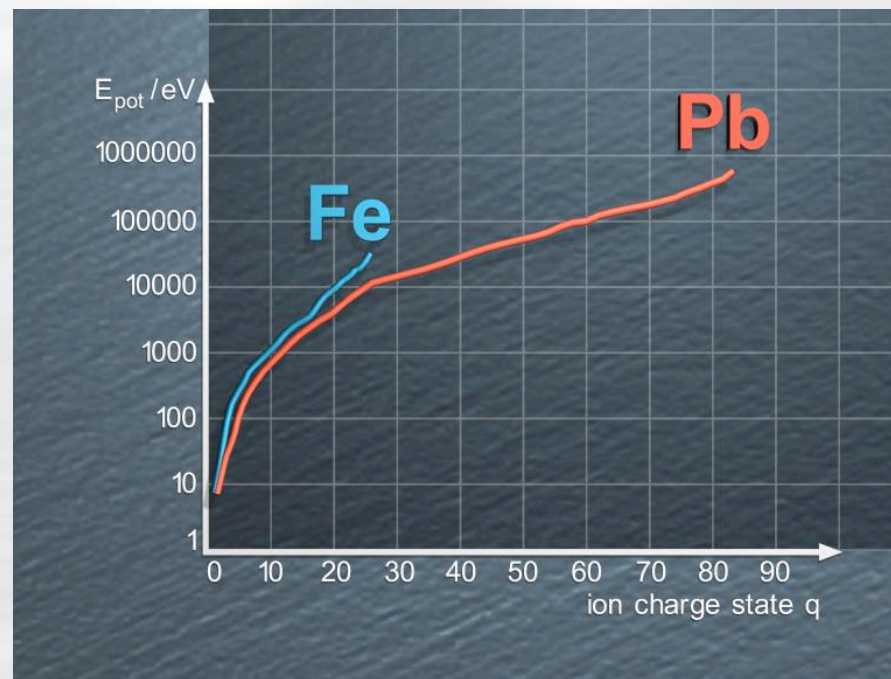
Potential Energy

The potential energy of an ion increases with the degree of ionization.

Example:

Xe^{44+} has a potential energy that is **4200 times higher** than that of Xe^{1+}

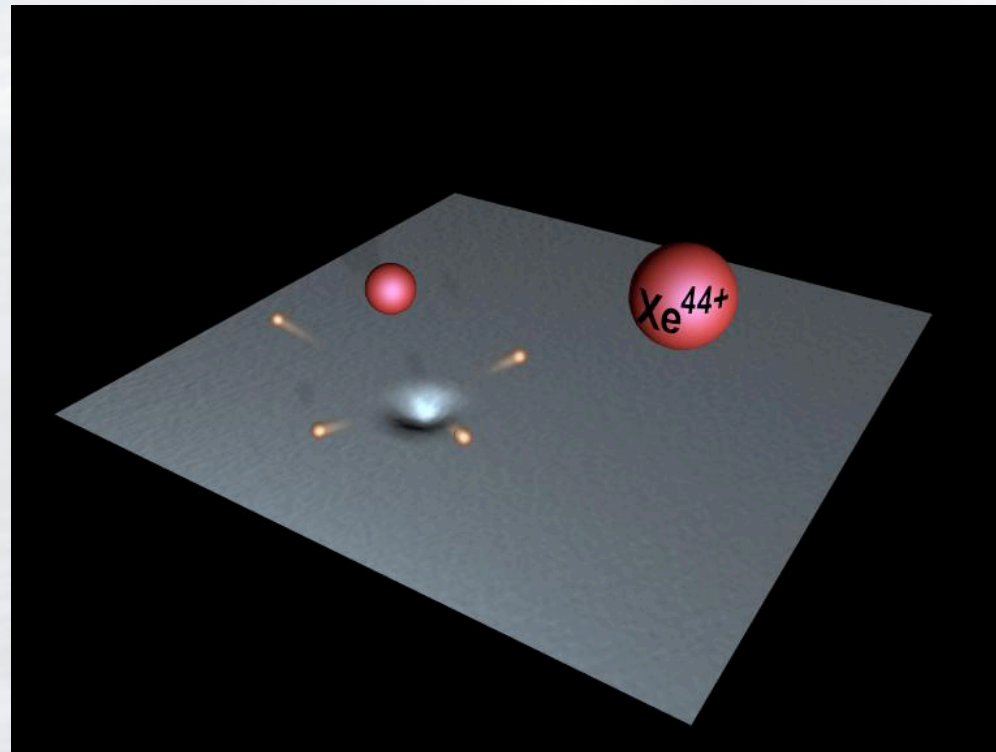
Xe^{54+} has a potential energy that is **16700 times higher** than that of Xe^{1+}



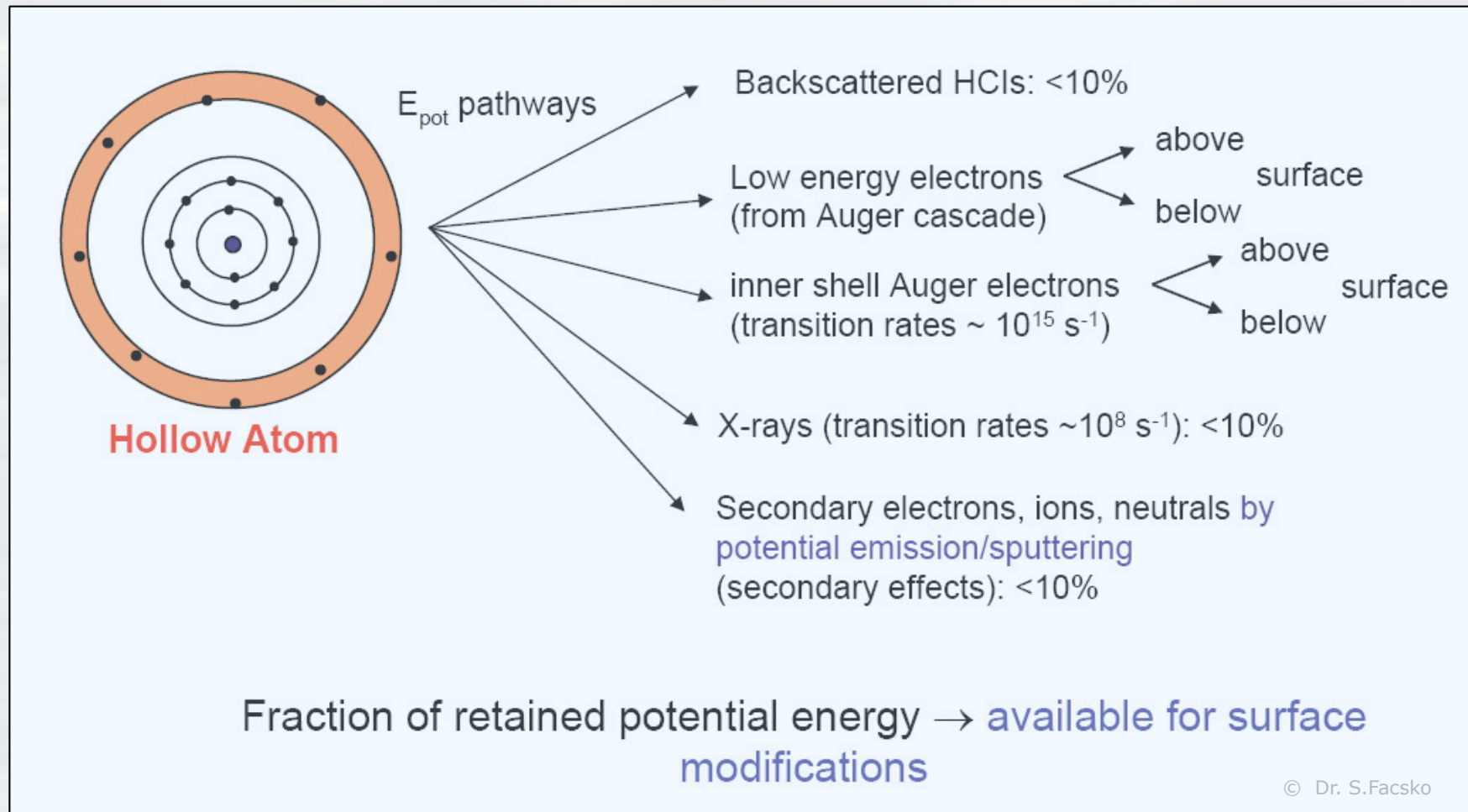
Properties of Highly Charged Ions

High power Deposition into the Surface

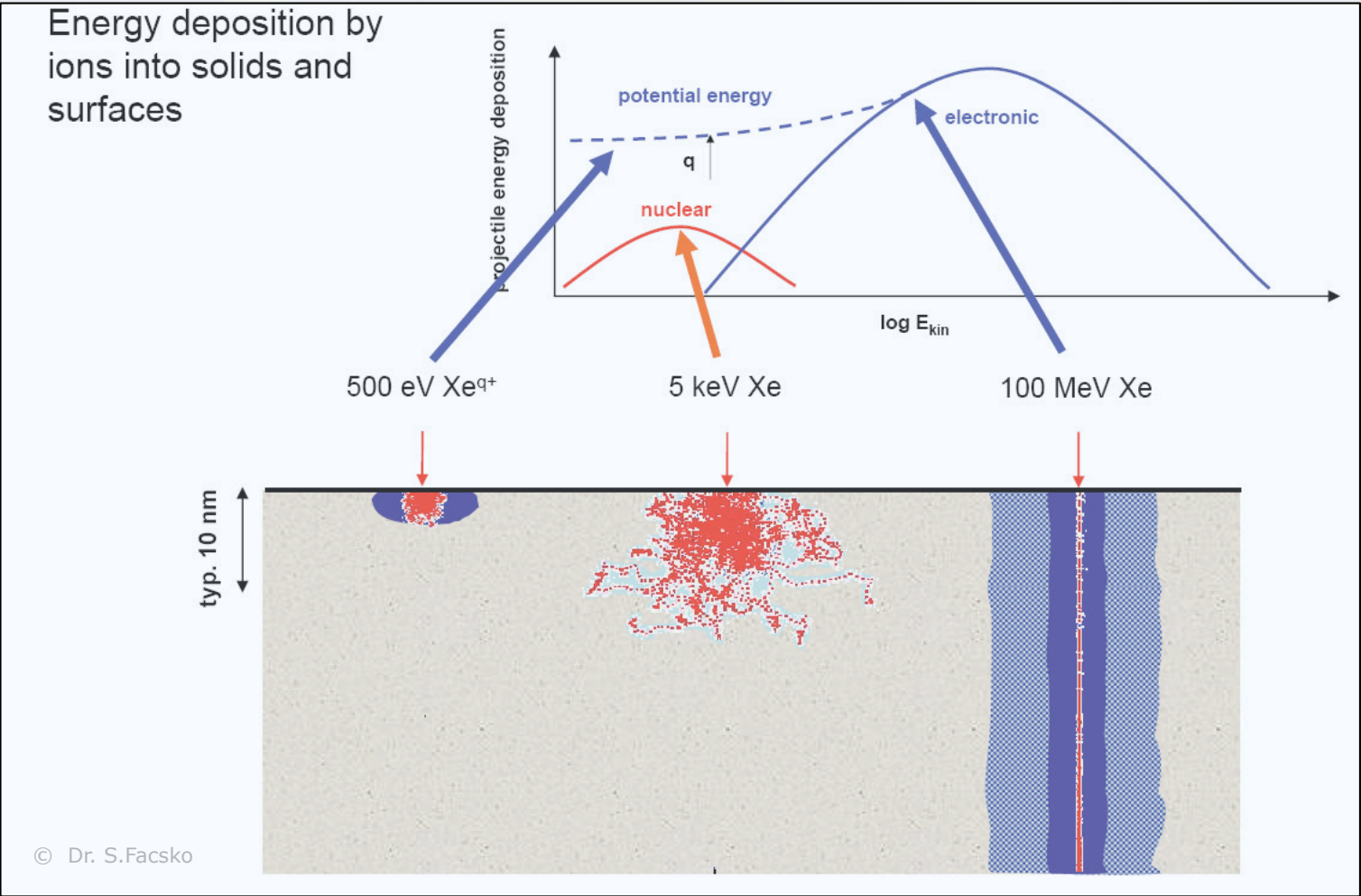
The deposition of potential energy leads to ultrafast intense electronic excitations up to: **$10^{12} \dots 10^{14} \text{ W/cm}^2$**



Pathways of Potential Energy

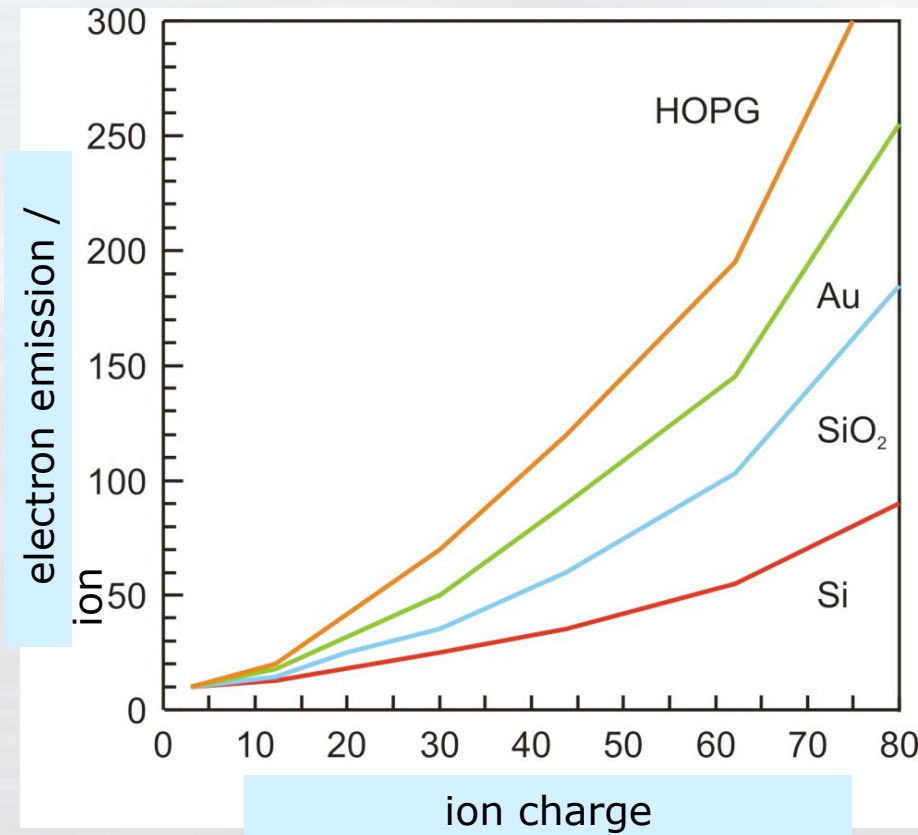


Energy Deposition into Surface



Highly Charged Ions give higher Yields of Secondary Ions and Secondary Electrons

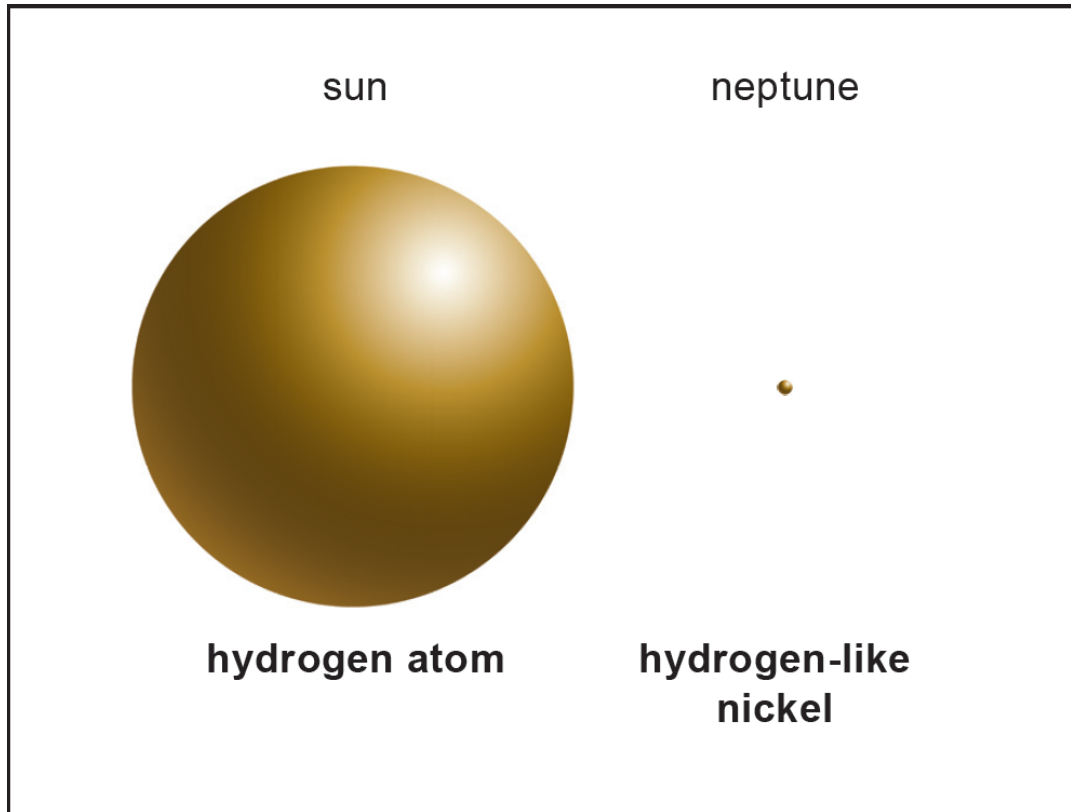
J.W.McDonald et al: NIM B 240, 829 (2005)



Total electron yields vs ion charge state q

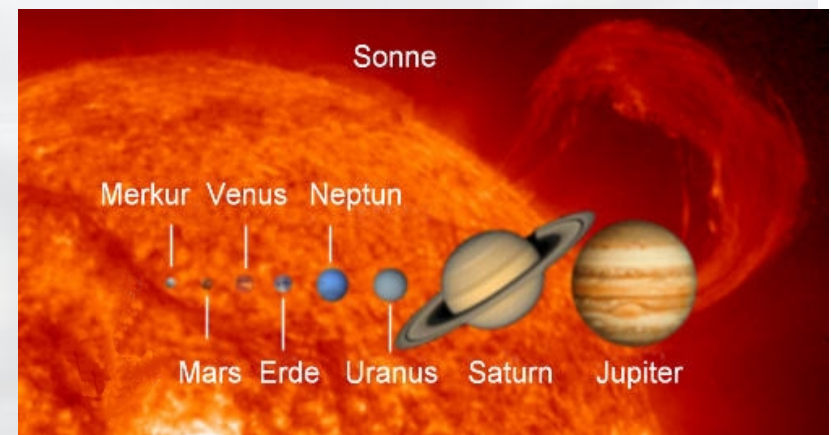
Properties of Highly Charged Ions

Small Spatial Extent of the Projectiles



(Idea by J.Gillaspy, NIST)

The size ratio of a hydrogen-like nickel ion to a neutral hydrogen atom is approximately equal to the size ratio of the planet neptune to the sun.



Properties of Highly Charged Ions

Extremely Compact Accelerator Structures are possible

Due to their high charge q ions can be accelerated very effectively

$\sim q$ for linear accelerators

$\sim q^2$ for ring accelerators

Example:

Xe¹⁺ and Xe⁴⁴⁺ acceleration at $\Delta U = 20$ kV



$\Delta U = 20$ kV

Xe¹⁺

Xe⁴⁴⁺

linear accelerator

20 keV

880 keV

ring accelerator

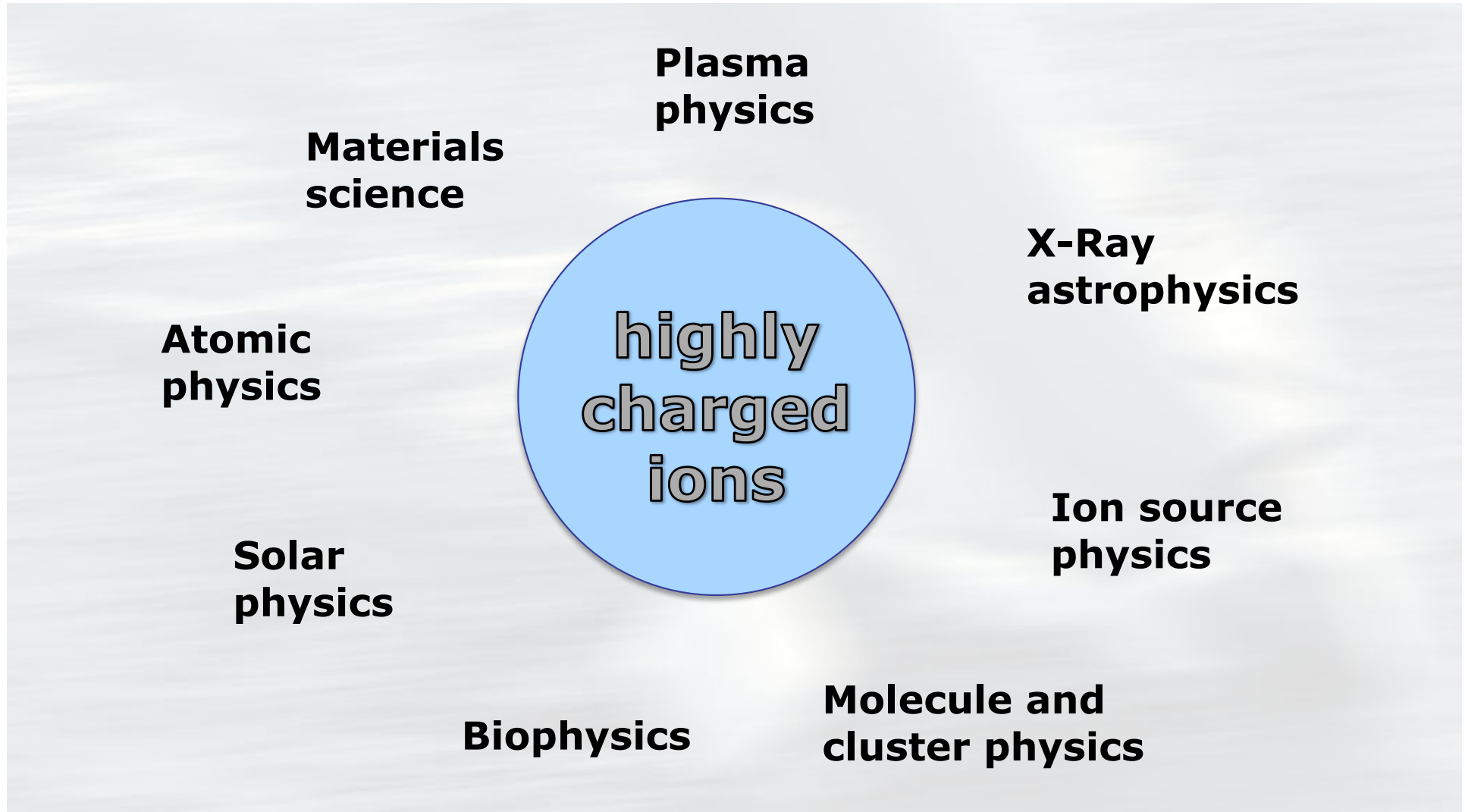
20 keV

38720 keV = 38,72 MeV

(energy gain about factor 2000!)

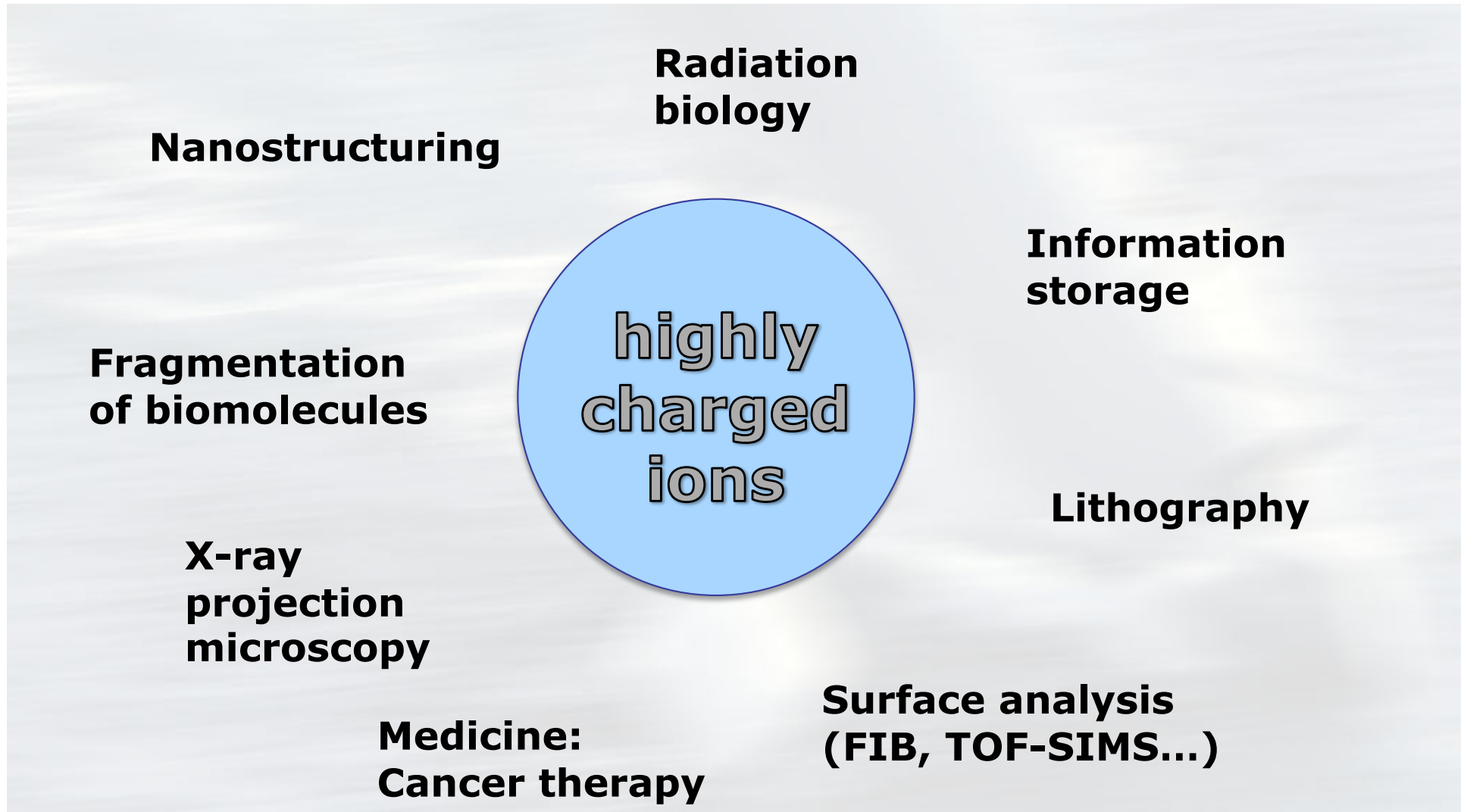
Applications of HCI

Highly Charged Ions in Basic Research and Industry



Applications of HCI

Highly Charged Ions in Basic Research and Industry

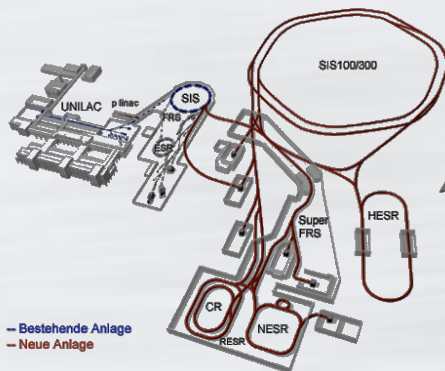


How to Produce Highly Charged Ions?

Ion Accelerators (GSI, TSR HD)

Stripping

→ Up to bare nuclei
at high projectile
energies

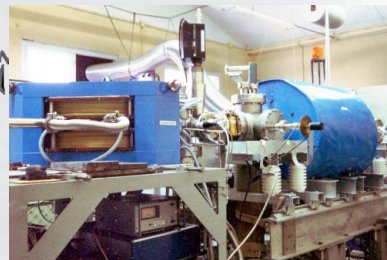


up to U^{92+}

ECR Ion Sources

Electron Cyclotron
Resonance (ECR)

→ Heating of a
magnetically
confined plasma



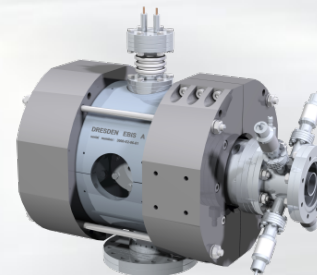
Ar^{16+} , Ta^{38+} , Au^{41+}

Electron Beam Ion Sources/Traps

→ Ionization in high-
dense electron beams

→ Electron beam
compression in strong
magnetic fields

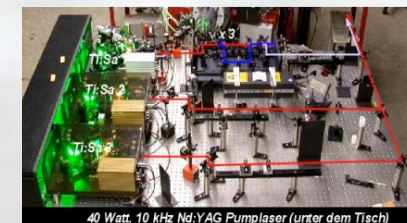
→ Superconducting or
permanent magnets



up to small
amounts of U^{92+}

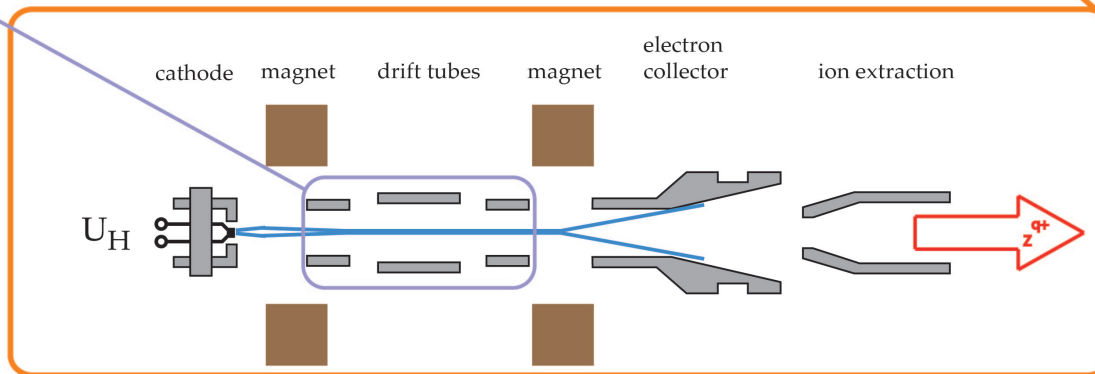
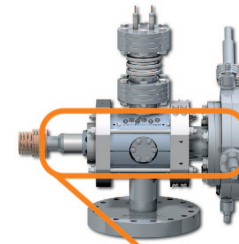
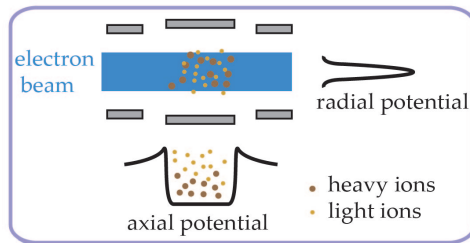
Laser Ion Sources

→ Pulsed laser
irradiation of
selected targets

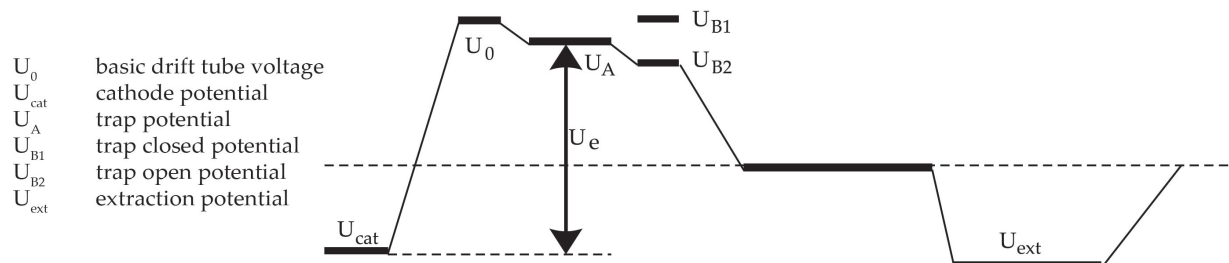


Pb^{27+} etc.

Electron Beam Ion Trap – Basic Idea



quantity	range
Electron beam energy	up to 200 keV
Electron beam current	up to A (typically up to some hundreds of mA)
Source vacuum	10^{-8} mbar up to 10^{-12} mbar



EBIS/T – Short History

Selected Milestones



Year	Place/ Name	Device	Ions	Source type (B, trap length)
1968	Dubna (USSR) Donets	IEL I, IEL II	Au ¹⁹⁺	warm EBIS 0.4 T, 16 cm
1971	Dubna (USSR) Donets/Pikin	KRION I	C ⁶⁺ , N ⁷⁺ , O ⁸⁺ , Ne ¹⁰⁺	SC 1.2 T, 1.2 m
1974	Dubna (USSR) Ovsyannikov/ Donets	KRION 2	Ar ¹⁸⁺ , Kr ³⁶⁺ , Xe ⁵⁴⁺	SC 2.2 T, 1.2 m
1981 1986	Orsay (France) Arianer	CRYEBIS 1 CRYEBIS 2	C ⁶⁺ , N ⁷⁺ , Ne ¹⁰⁺ , Ar ¹⁸⁺	SC, 3 T, 1.66 m SC, 5 T, 1.66 m
1984	Saclay (France) Faure	DIONE	Ar ¹⁶⁺ , Kr ³⁰⁺ , I ⁴¹⁺	SC, 6 T, 1.2 m

EBIS/T – Short History

Selected Milestones



Year	Place/Name	Device	Ions	Source type (B, trap length)
1988 1990	LLNL (USA) Levine Marrs/Knapp	EBIT-I EBIT-II (birth of EBIT!)	Xe ⁵⁴⁺ , U ⁸⁸⁺	SC, 3 T, 2 cm (E _(e,max) = 29 keV)
1990	LLNL (USA) Marrs/ Schneider	S-EBIT	U ⁹²⁺ , Cf ⁹⁶⁺	SC, 3 T, 2 cm (E _(e,max) = 215 keV)
1999	Freiburg (Germany) Crespo	F/HD-EBIT	Xe ⁵⁴⁺	SC, 9 T, 4-30 cm
2009	Brookhaven (USA) Beebe/Pikin	RHIC-EBIS	Xe ³⁶⁺ high current EBIS	SC 6 T, 1.5 m

EBIS/T – Short History

Selected Milestones



Year	Place/ Name	Device	Ions	Source type (B, trap length)
1999	TU Dresden (Germany) Ovsyannikov/ Zschornack	Dresden EBIT	Ar ¹⁸⁺ , Xe ⁴⁴⁺ , Ir ⁶⁷⁺	warm EBIT 0.25 T, 2 cm (E _(e,max) = 15 keV)
2005 2008	Dreebit GmbH (Germany) Ovsyannikov/ Zschornack	Dresden EBIS Dresden EBIS-A	Ar ¹⁸⁺ , Xe ⁴⁸⁺ , Ir ⁶⁷⁺	warm EBIS, 0.4/0.6 T, 6 cm (E _(e,max) = 25 keV)
2009	Dreebit GmbH (Germany) Ovsyannikov/ Zschornack	Dresden EBIS-SC (medical applications and R&D)	C ⁶⁺ , Ar ¹⁸⁺ , Xe ⁴⁸⁺	SC, 6 T, 4-30 cm (E _(e,max) = 20 keV)

„classical“ cryogenic EBIT

- superconducting coils
→ (3... 8) T magnetic field

⇒ $j_e > 1000 \text{ A/cm}^2$
- highest charge states
 $\text{Xe}^{(52...54)+}$, up to $\text{U}^{(90...92)+}$
- large devices,
liquid helium cooling
- latest developments:
Refrigerator cooling

room-temperature EBIT

- permanent magnets (SmCo, NdFeB)
(250...620) mT at the axis

⇒ $j_e = (200... 600) \text{ A/cm}^2$
- bare ions up to $Z=28$,
 Kr^{34+} , $\text{Xe}^{(44...48)+}$, Ir^{67+}
- compact, transportable,
low initial and maintenance
costs,
short setup times

EBIS/T – Short History

Selected Milestones



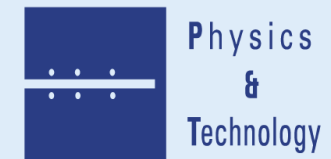
There are actually about 60 EBIS/EBIT around the world.
(For a list see R.Becker, O.Kester; RSI 81(2010) 02A513)

Most of them are special laboratory constructions.

Worldwide there are only two commercial offerers:

1. Physics and Technology Livermore (USA)

REBIT (Refrigerated Electron Beam Ion Trap)



2. DREEBIT GmbH Dresden (Germany)

Dresden EBIT

Dresden EBIS

Dresden EBIS-A

Dresden EBIS-SC

} Room-Temperature
EBIS/T

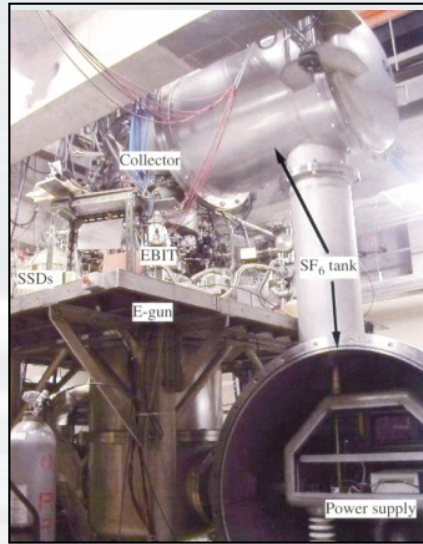
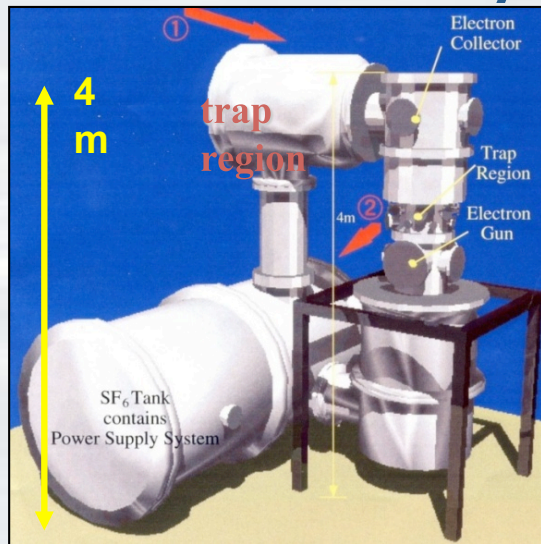
(Refrigerated Electron Beam Ion Trap)



EBIS/T

Different solutions

Tokyo EBIT



Shanghai EBIT

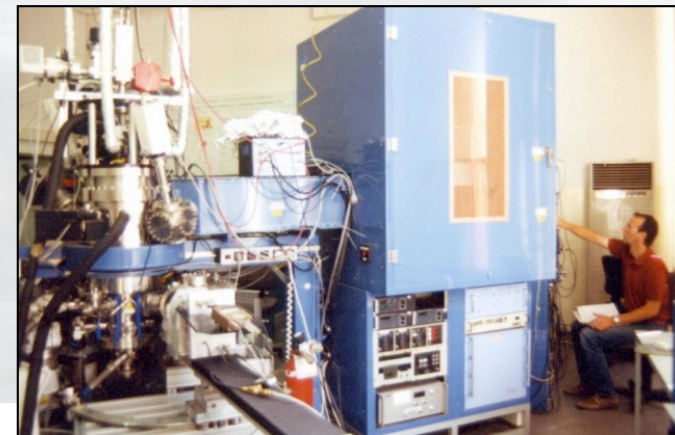


Dresden EBIT

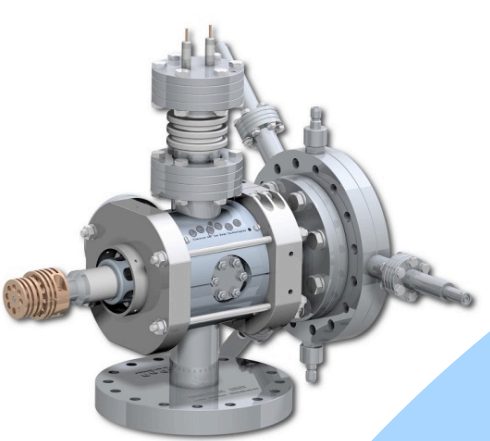
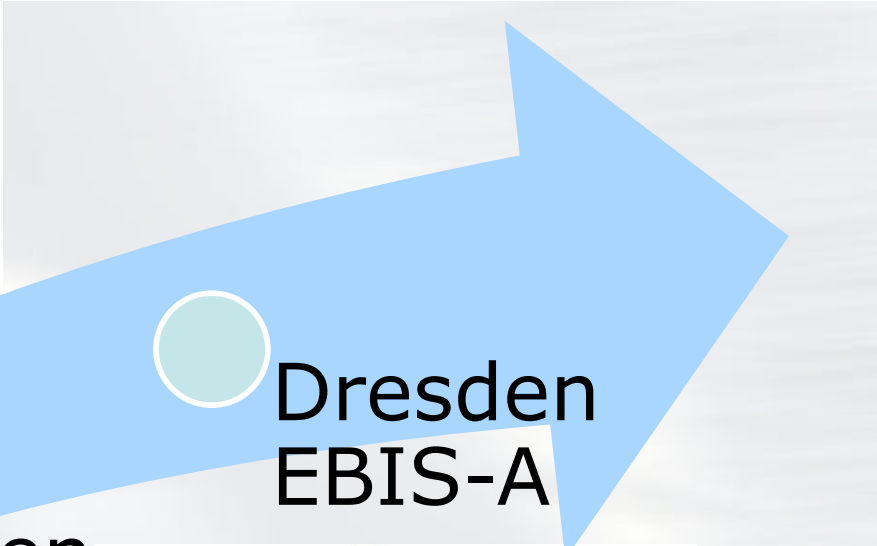
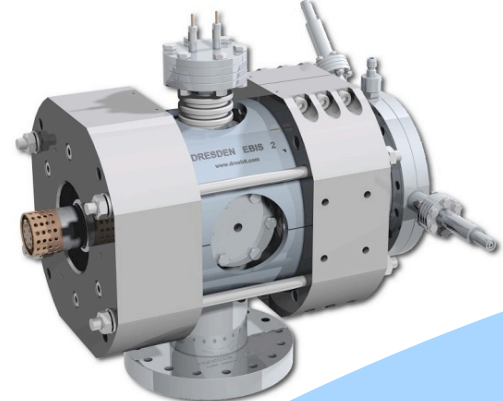


LLNL EBIT

(at the MPI for Plasma Physics Berlin)



Room-Temperature EBIS



Dresden
EBIS2

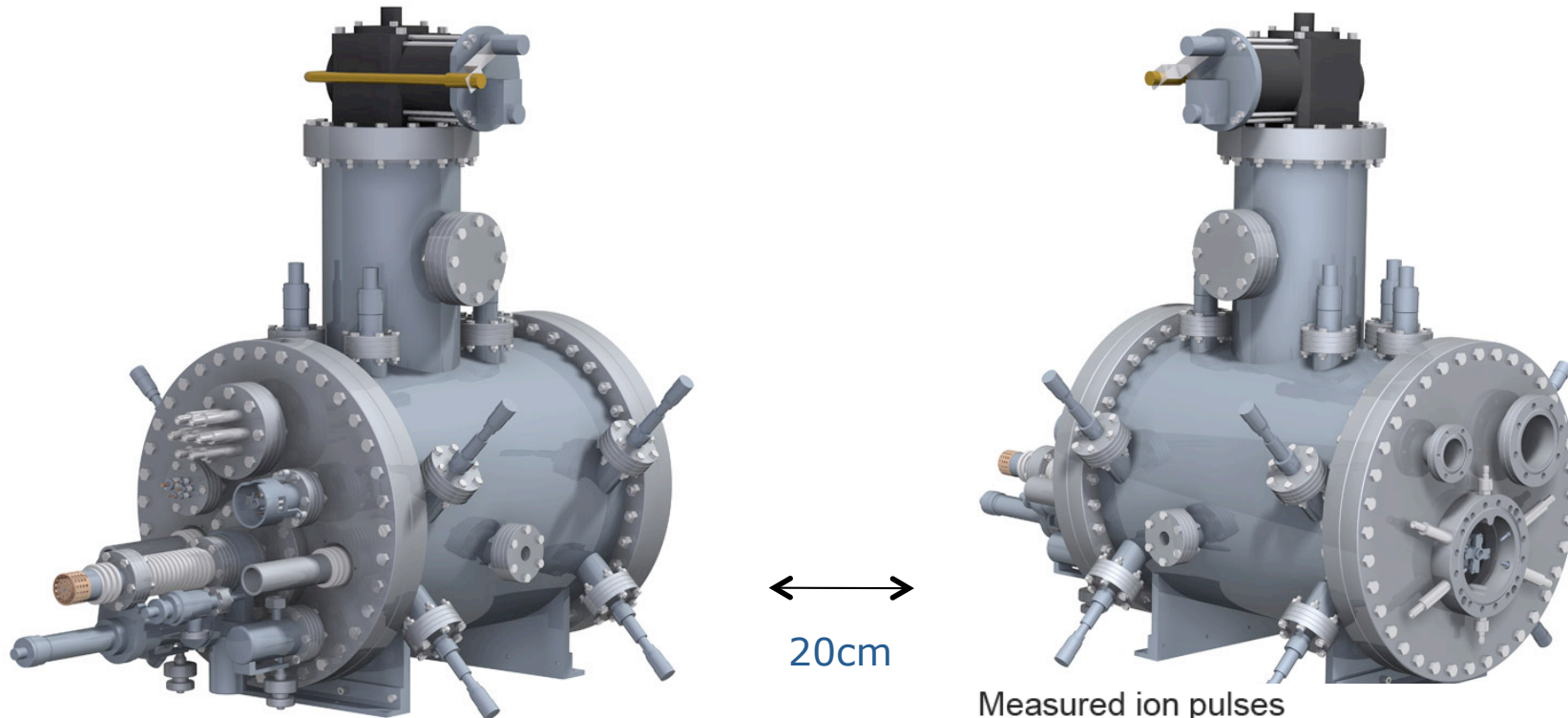
Dresden
EBIS-A

Dresden
EBIS3



20cm

Dresden EBIS-SC – A superconducting EBIS



- L-He free at 4.2K
- electron beam energy up to 30 keV
- electron beam current up to 700 mA
- magnetic field on-axis 6T

Measured ion pulses

Ion	Max. Ions/pulse	Max. pulse rate/Hz
H ⁺	3·10 ⁹	500
H ₂ ⁺	3·10 ⁹	1000
C ⁴⁺	8·10 ⁸	10
C ⁶⁺	4·10 ⁸	10
Ar ¹⁶⁺	2·10 ⁷	2
I ⁴³⁺	1·10 ⁶	1

The intended purpose of an EBIS is to produce highly charged ions. For a certain ionisation stage q two opposite processes take place in the electron beam:

Charge-generating processes

- Ionisation (ion)
- Charge Exchange (ce)
- Radiative Recombination (RR)

Charge destructive processes

- Ionisation
- Charge Exchange
- Radiative Recombination
- Ion loses from the trap

Charge Balance of Ions with the Charge State q

Rate equations

$$\frac{dn_q}{dt} = n_e v_e \left[\sigma_{q-1 \rightarrow q}^{ion} n_{q-1} - \left(\sigma_{q \rightarrow q+1}^{ion} + \sigma_{q \rightarrow q-1}^{RR} \right) n_q + \sigma_{q+1 \rightarrow q}^{RR} n_{q+1} \right] - n_0 v_{ion} \left[\sigma_{q \rightarrow q-1}^{ce} n_q - \sigma_{q+1 \rightarrow q}^{ce} n_{q+1} \right] - f_q^{col} \frac{e^{-\frac{qeU_t}{kT_{ion}}}}{\frac{qeU_t}{kT_{ion}}} n_q$$

Charge-generating processes (sources for A^{q+})

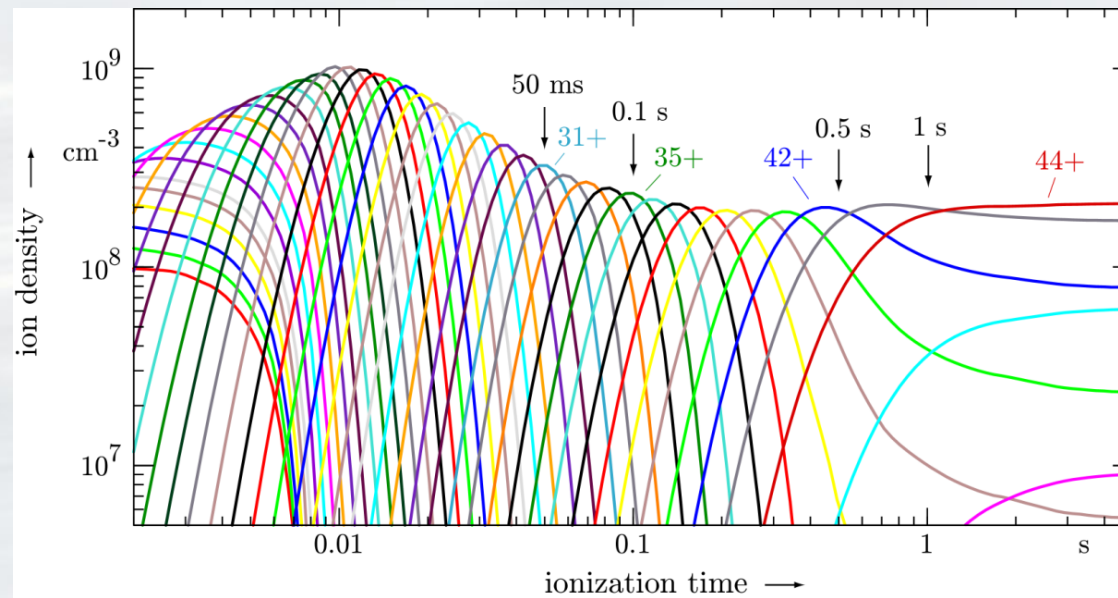
Charge destructive processes (sinks for A^{q+})

U_t - depth of the potential wall
(radial and axial) trap potential
 kT_{ion} - ion energy

→ We should consider

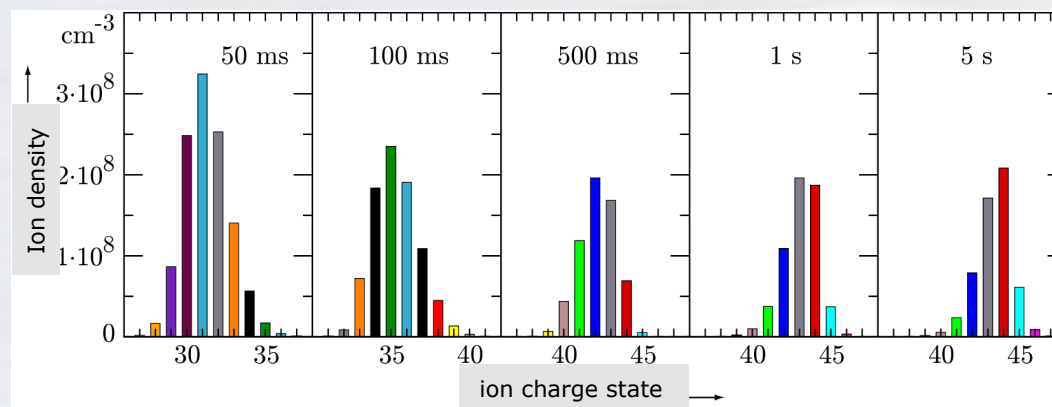
- the electron beam energies
- the vacuum in the ionization region
- the excitation functions of individual processes

Example: Ionization of Xenon in a Dresden EBIS-A



Pressure: 2×10^{-9} mbar

Electron beam energy: 14 keV




Production of HCI - Ionization Factor

The basic process in the electron beam of an EBIS is **successive electron impact ionization** with an average ionization time for the ionization of ions with the charge state q of

$$\tau_{q \rightarrow q+1} = \frac{n_q}{f_{q \rightarrow q+1}} = \frac{1}{n_e v_e \sigma_{q \rightarrow q+1}} = \frac{e}{j_e \sigma_{q \rightarrow q+1}}$$

with the collision frequency

$$f_{q \rightarrow q+1} = n_e n_q v_e \sigma_{q \rightarrow q+1}$$


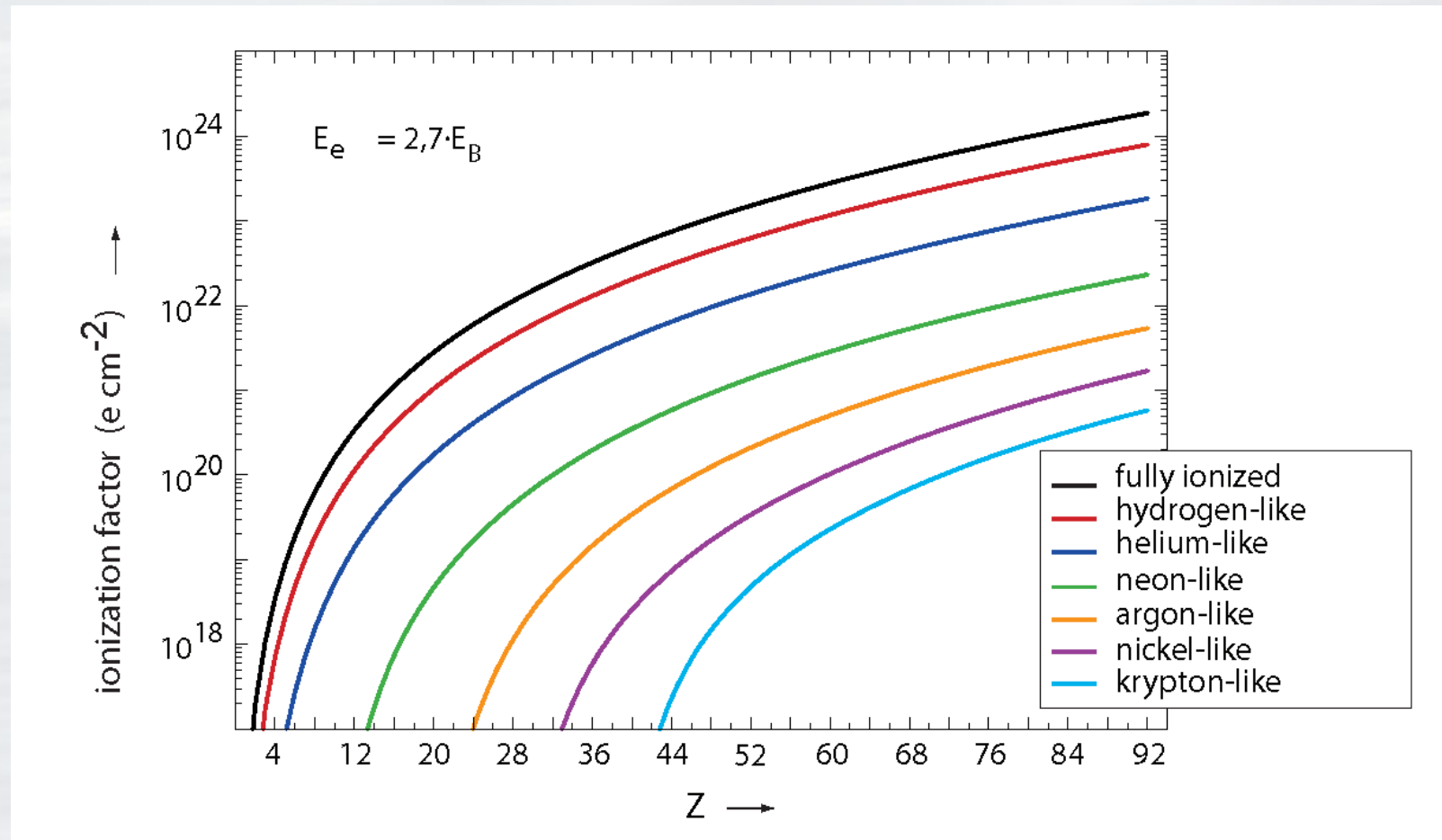
This expressions lead to the **ionization factor**

$$j_e \tau_{q \rightarrow q+1} = \frac{e}{\sigma_{q \rightarrow q+1}}$$

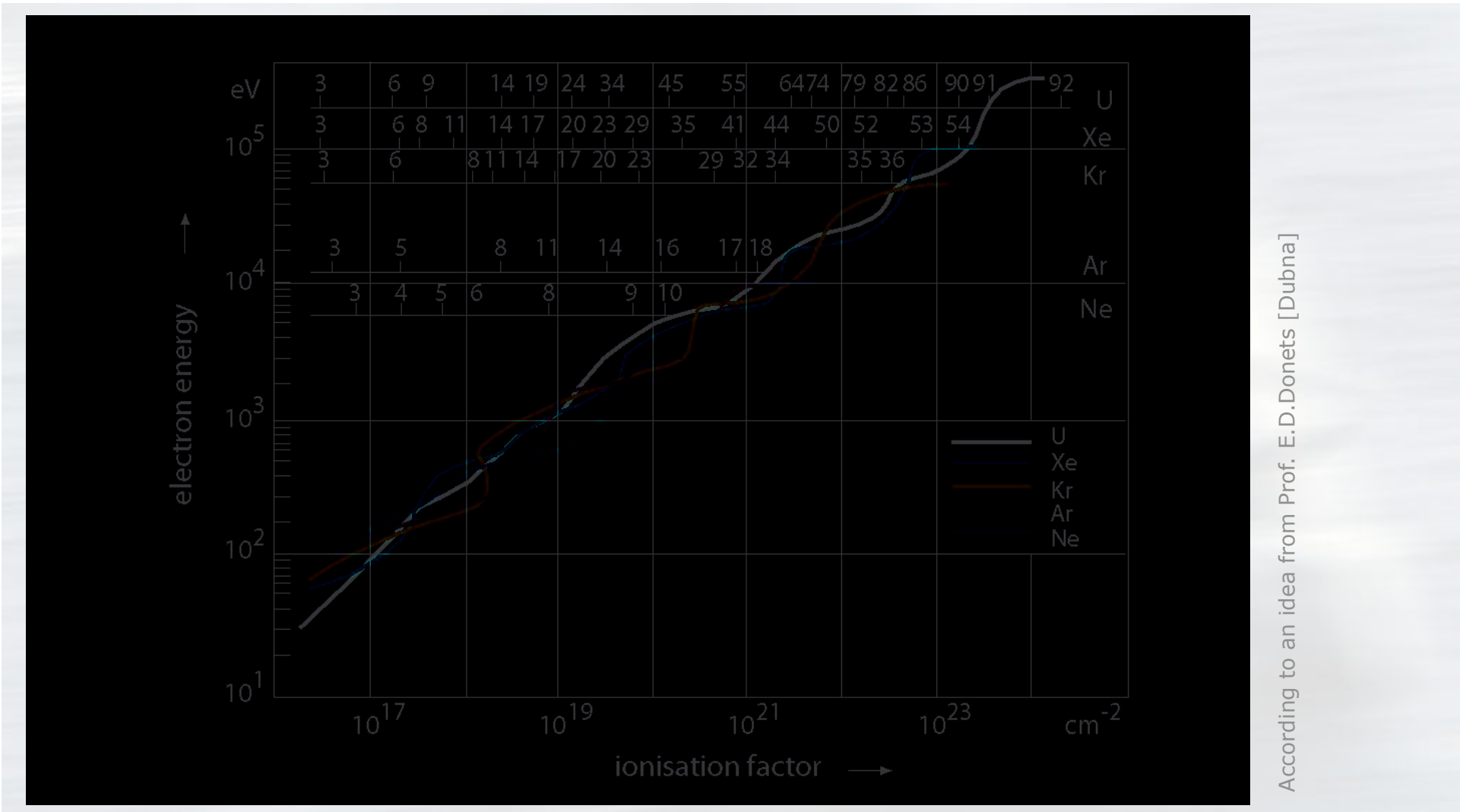
i.e. ionization is possible if we have

$$j_e \tau_{q \rightarrow q+1} \geq \frac{e}{\sigma_{q \rightarrow q+1}}$$

Ionization Factor vs. Atomic Number and Degree of Ionisation



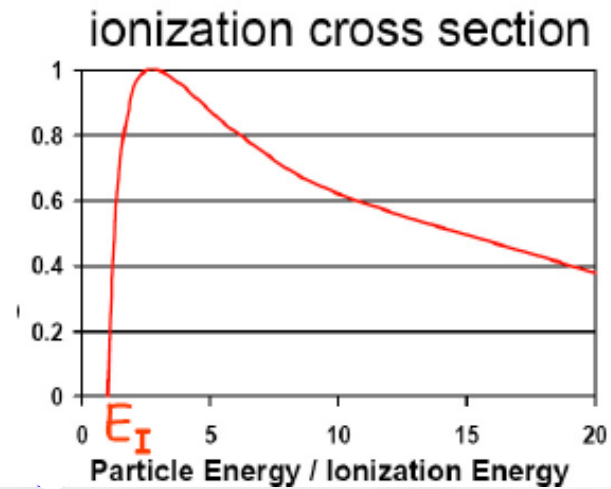
Ionization Factor vs. Atomic Number, Degree of Ionization and Electron Energy



According to an idea from Prof. E.D.Donets [Dubna]

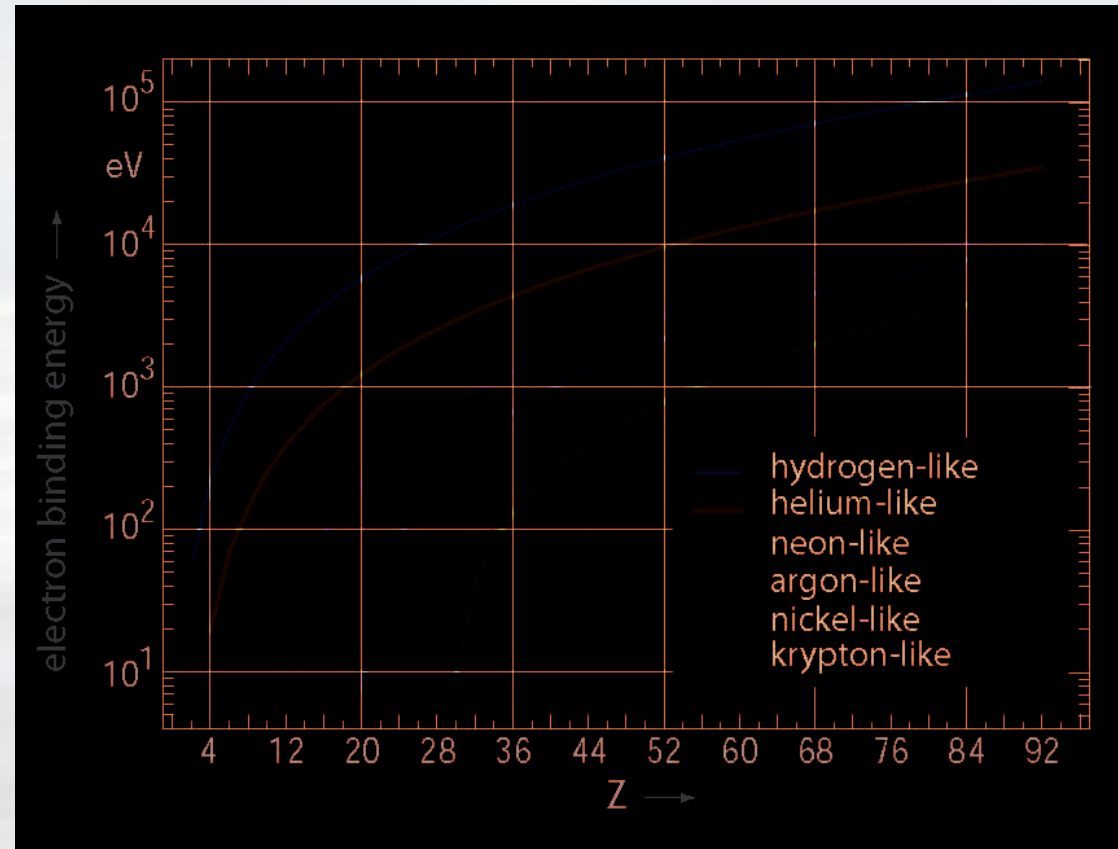
Electron Binding Energies

Threshold values for ionization



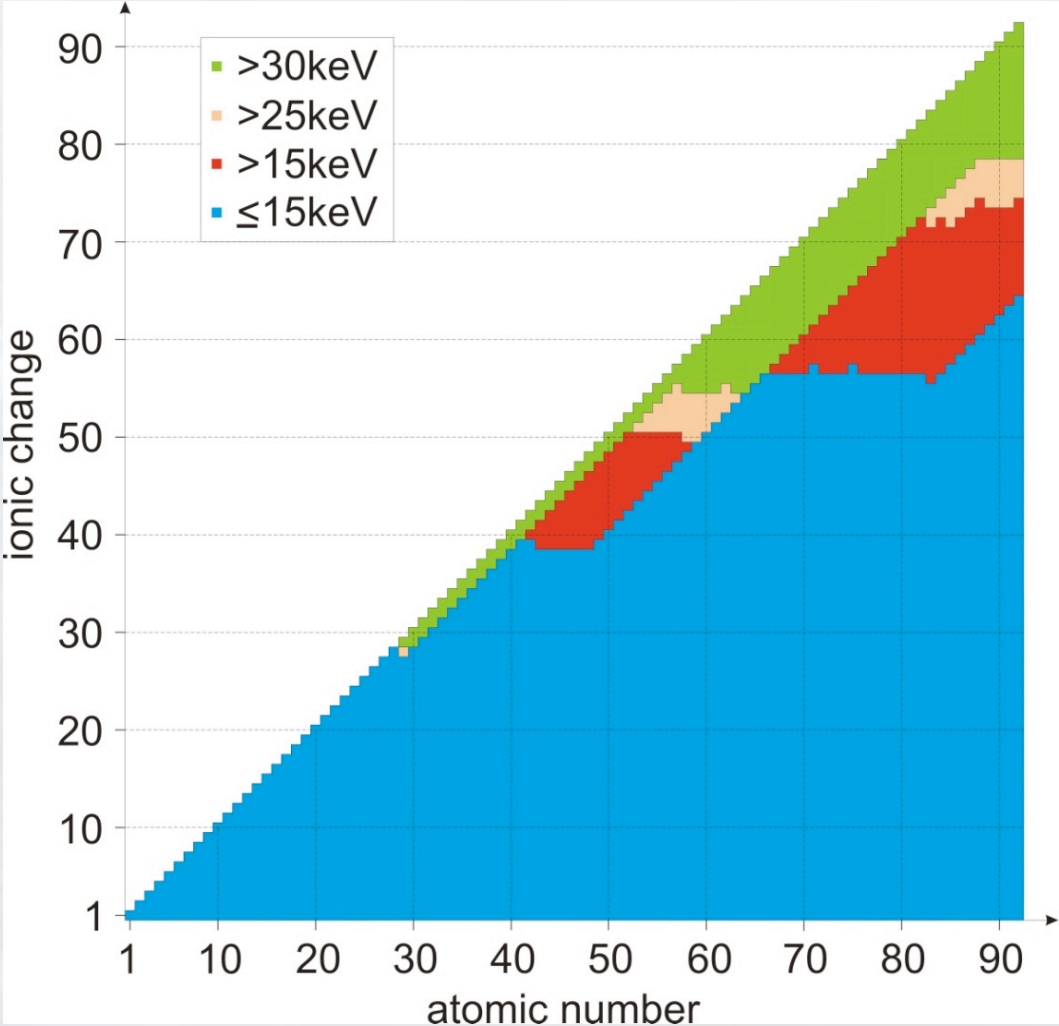
The optimal energy for ionizing an ion from the charge state q to $q+1$ is nearly **e-times** the ionization energy of the weakest bound electron.

Ionization starts at the ionization threshold E_I .



Electron Binding Energies

Threshold values for ionization



Let`s have a look onto the most important physical processes in the electron beam:

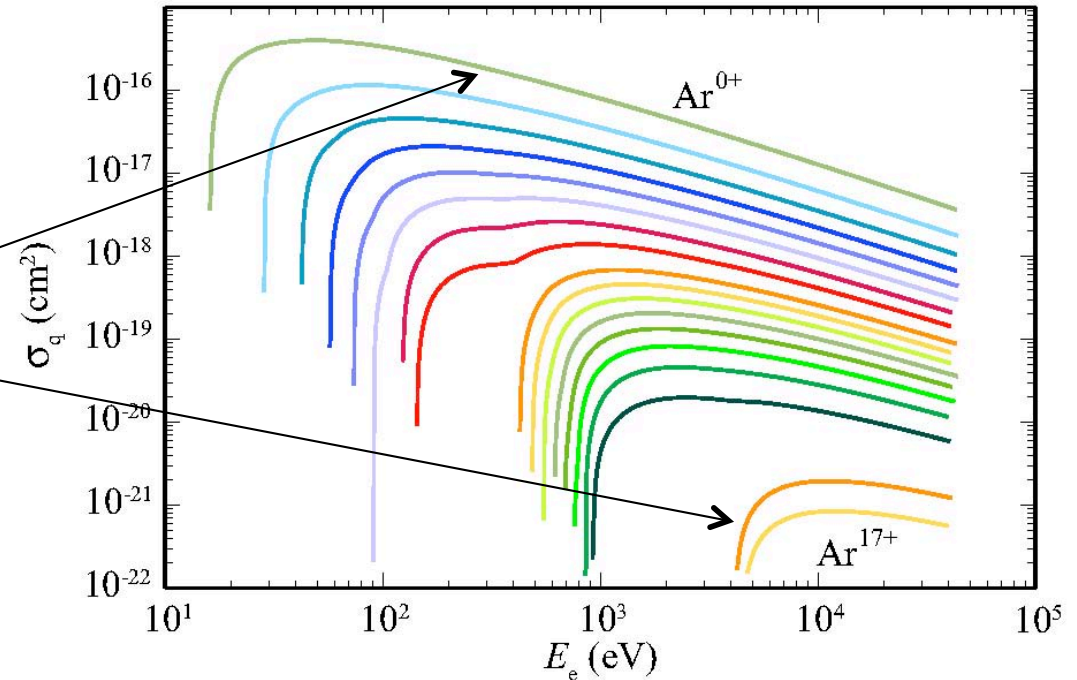
- Single ionization
- Double ionization
- Single charge exchange
- Double charge exchange
- Radiative recombination

Basic Physics

Single Electron Impact Ionization



The higher the ion charge state, the smaller is the ionization cross section

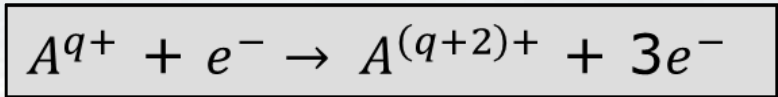


$$\sigma_{q \rightarrow q+1} = 4.5 \cdot 10^{-14} \sum_{i=1}^N \frac{\ln \frac{E_e}{E_{nl}}}{E_e \cdot E_{nl}} [cm^2]$$

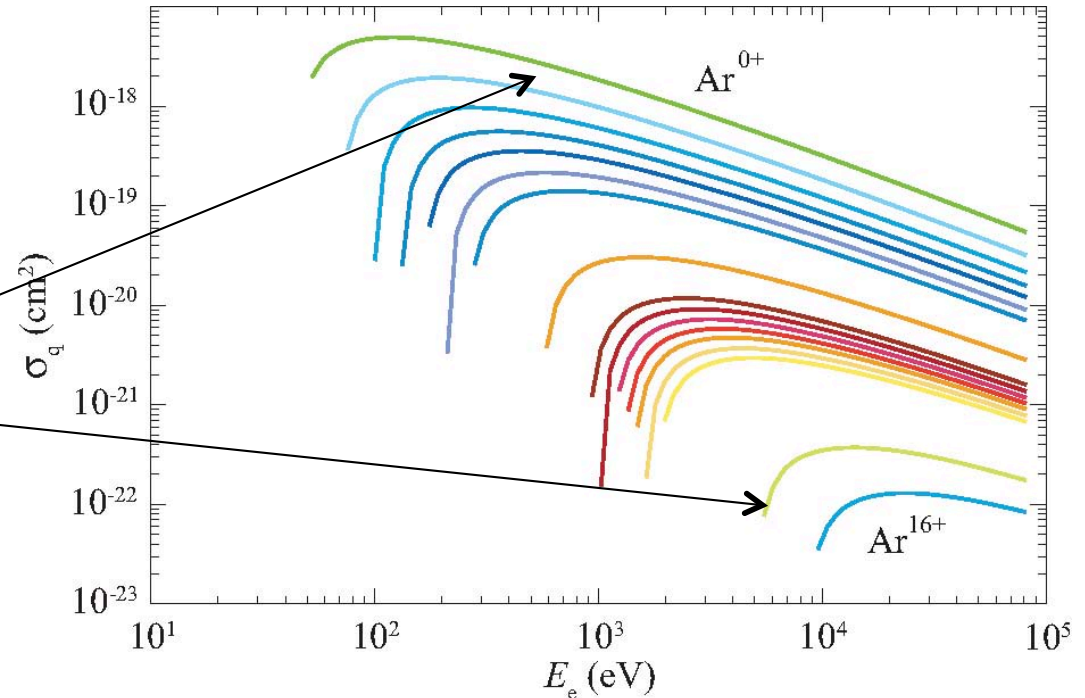
Lotz formula for $E_e \gg E_{nl}$
(estimated error: up to 10%;
N: number of subshells)

Basic Physics

Double Electron Impact Ionization



The higher the ion charge state, the smaller is the ionization cross section



Shevelko formula

$$\sigma_{q \rightarrow q+2} = 1.4 \cdot 10^{-19} \frac{N^{1.08}}{\left(\frac{E_q [\text{eV}]}{13.6 \text{ eV}}\right)^2} \left(\frac{U}{U+1}\right)^c \frac{\ln(U+1)}{U+1} [\text{cm}^2]$$

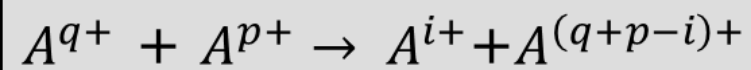
with

$$U = \frac{E_e}{E_q} - 1$$

$c=1$ for neutrals and $c=0.75$ for ions
 E_q - sum of the ionization potentials of both weakest bound electrons
 N - number of electrons in the atom/ion

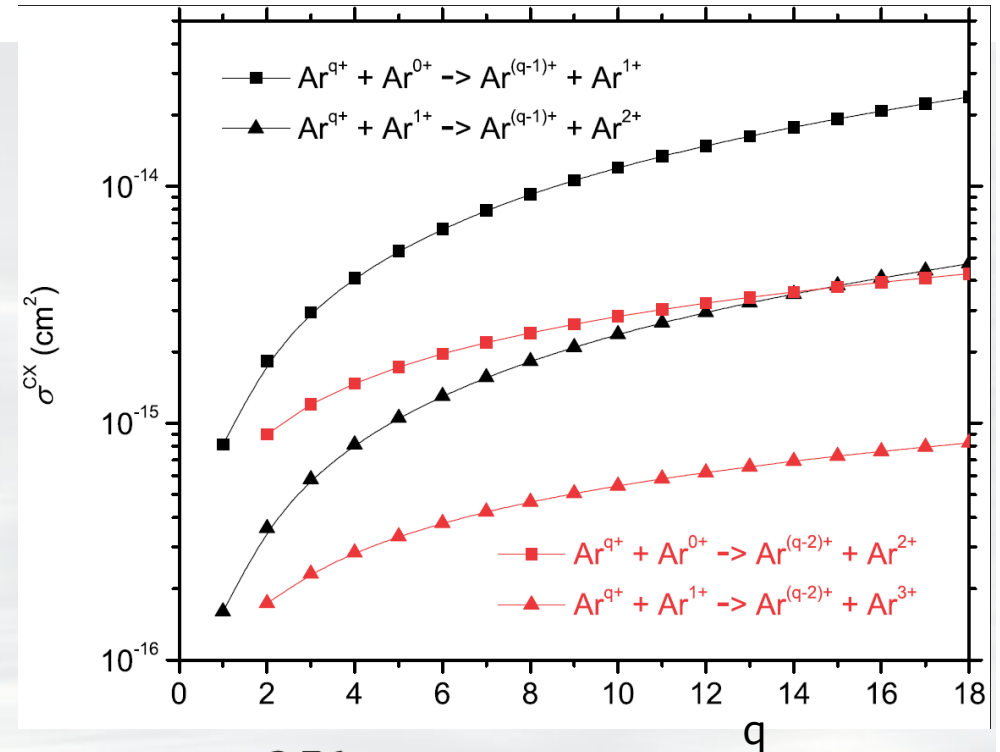
Basic Physics

Charge Exchange



Cross sections are independent on the electron energy.

Charge exchange with neutrals is dominant and the main loss process for highly charged ions.



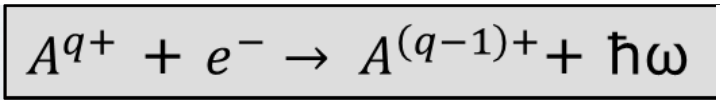
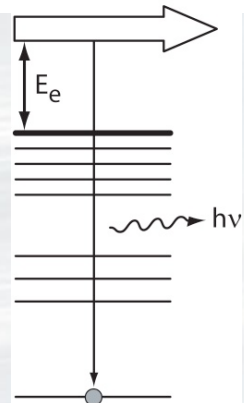
$$\sigma_{q \rightarrow q-1} \approx (1.43 \pm 0.76) \cdot 10^{-12} q^{1.17} (E_q [eV])^{-2.76} [cm^2] \quad \text{Single charge exchange (Müller and Salzborn)}$$

$$\sigma_{q \rightarrow q-2} \approx 1.08 \cdot 10^{-12} q^{0.71} (E_q [eV])^{-2.8} [cm^2] \quad \text{Double charge exchange}$$

Basic Physics

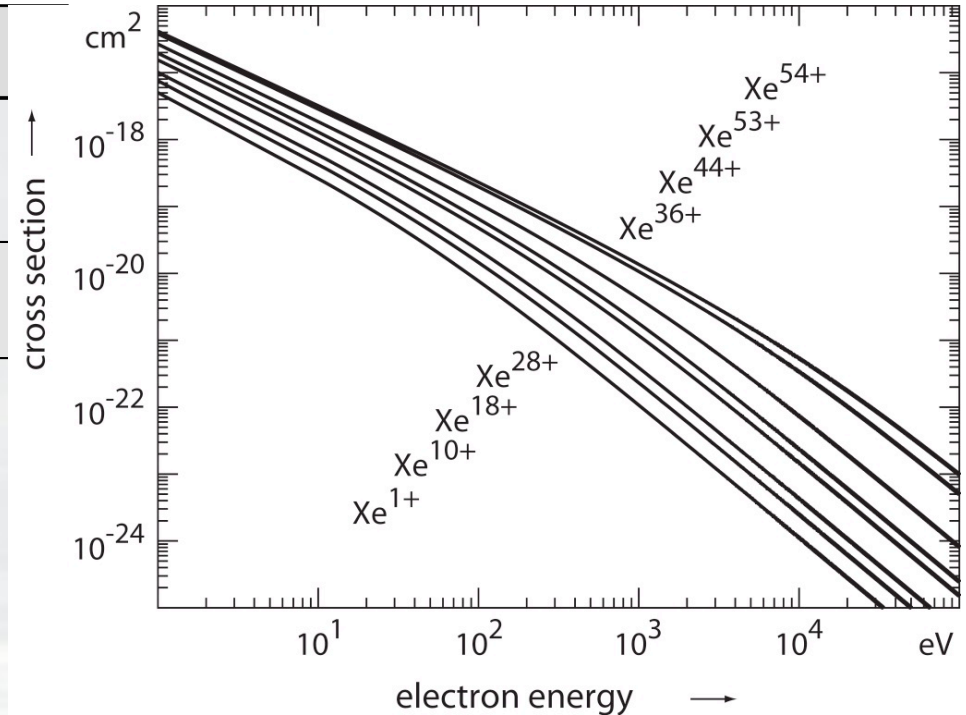
Radiative Recombination (RR)

electron beam



Charge exchange is strong at low electron energies.

Due to RR processes ionization in an EBIS is more efficient at higher electron energies.



$$\sigma_{q \rightarrow q-1}^{RR} = 2.10 \cdot 10^{-22} \frac{E_0^2}{n E_{cm} (E_0 + n^2 E_{cm})} [cm^2]$$

$$\sigma_{q \rightarrow q-1}^{RR} = \frac{8\pi}{3\sqrt{2}} \alpha \lambda_e^2 \chi_q(E_e) \ln \left(1 + \frac{\chi_q(E_e)}{2(n + (1 - W_n) - 0.3)} \right)$$

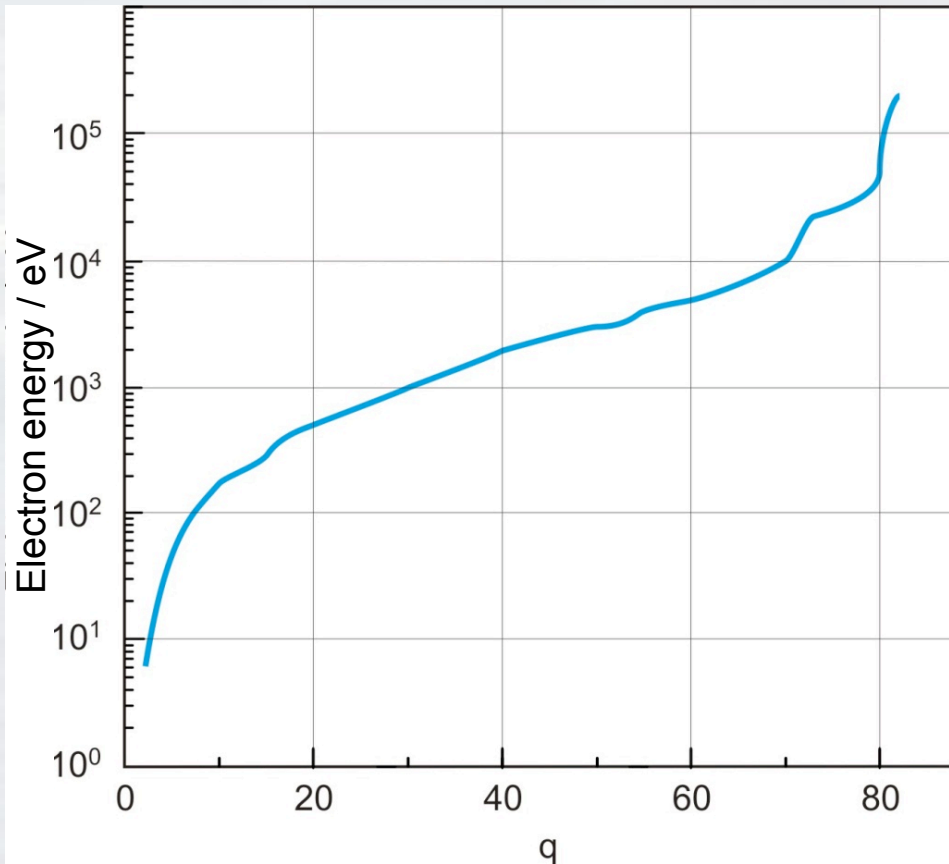
with $\chi_q(E_e) = (Z + q)^2 \frac{13.6 \text{ eV}}{4E_e}$

Theory from Stobbe (for fully ionized atoms)
 E_0 - binding energy of the hydrogen-like ground-state ion
 n - main quantum number of the shell where the electron is captured
 E_{cm} - CM collision energy between electrons and ions

Theory from Kim and Pratt (for all ions)
 W_n - ratio of the number of unoccupied states to the total number of states in the subshell
 λ_e - Compton-wavelength

Basic Physics

Radiative Recombination (RR)



Balance between
ionization and radiative
recombination for lead
ions

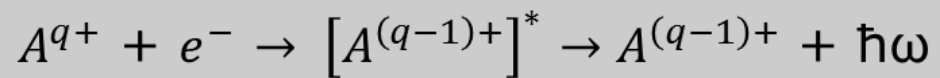
(after: R.Becker; ICIS 2009,
Gatlinburg)

Ion loss ratio:

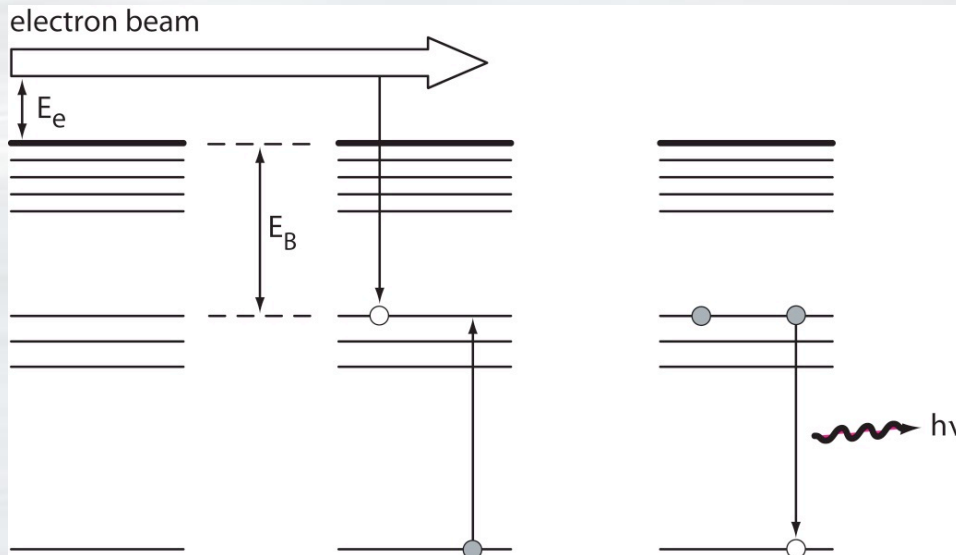
$$R = 2 \cdot 10^{-13} n_e [cm]^{-3} \frac{q^2}{\sqrt{E_e [eV]}} \left[\frac{cm^3}{s} \right]$$

Basic Physics

Dielectronic Recombination (DR)



Capture of an electron from the continuum at simultaneous excitation of a second electron and following deexcitation (photon emission).



$$R_{DR} = 6 \cdot 10^{-10} N \left(\frac{q}{E_e} \right)^{\frac{3}{2}} \sqrt{E_q} e^{-\frac{E_q}{E_e}} \left[\frac{cm^3}{s} \right]$$

N – number of electrons in the outermost occupied shell
 E_q – ionization energy of the ion with the charge q

Ion loss ratio

EBIS: Basic Properties

Electrical trap capacity

The total number of ions stored in an EBIs is determined by the electrical trap capacity C_{el} .

Assumption:

Homogeneous electron beam passing an ion trap of the **length L** with an **electron beam current I_e** . The **electron energy is E_e** .

With

$$I_e = \frac{dQ}{dt}, \quad v_e = \frac{dx}{dt}, \quad v_e = \sqrt{\frac{2E_e}{m_e}} \implies \Delta Q = \frac{I_e \Delta x}{v_e} = \frac{I_e L}{\sqrt{\frac{2E_e}{m_e}}}$$

follows

$$C_{el} = 1.05 \cdot 10^{13} \frac{I_e [A] L [m]}{\sqrt{E_e [eV]}}$$

EBIS: Basic Properties

Electrical trap capacity

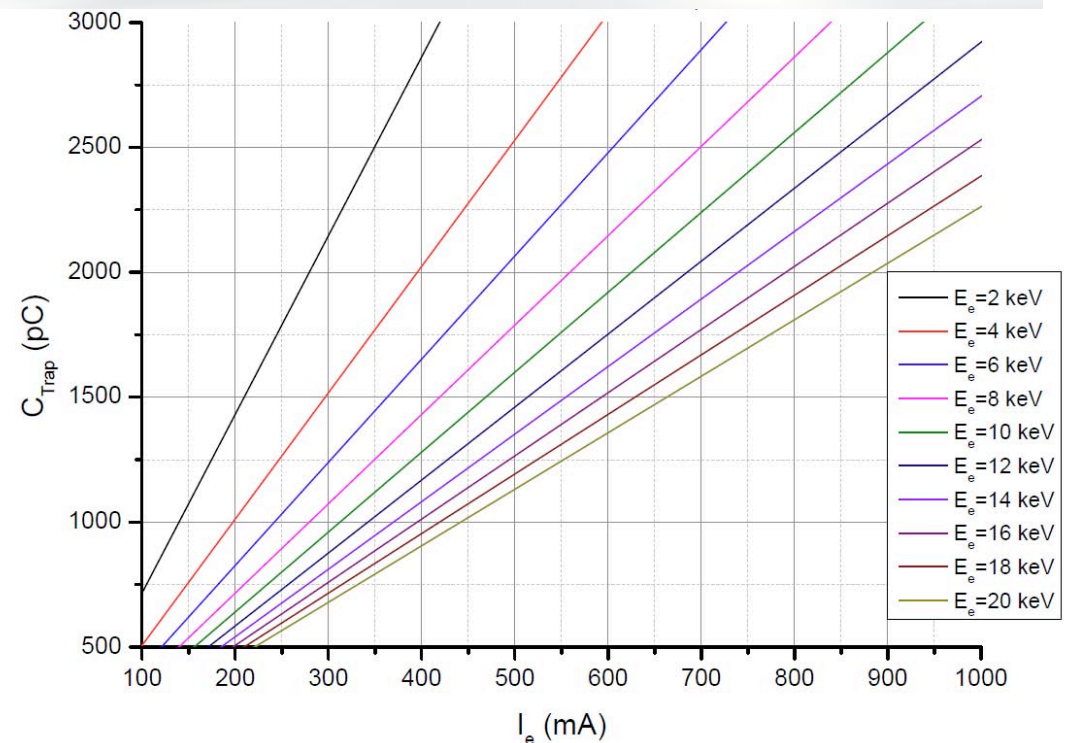
$$C_{el} = 1.05 \cdot 10^{13} \frac{I_e [A] L [m]}{\sqrt{E_e [eV]}} \cdot f \cdot \alpha$$

For practical purposes we must consider

- the charge compensation f ($f < 1$) of the electron beam,
- the fraction α of ions with a certain ion charge state in the ion charge state spectrum of the produced ions.

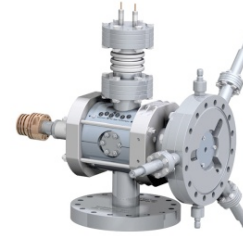
Example:

Trap capacity of the Dresden EBIS-SC at different electron beam currents and different electron beam energies.



EBIS: Basic Properties

Electron beam: space charge potential



Radial trap potential

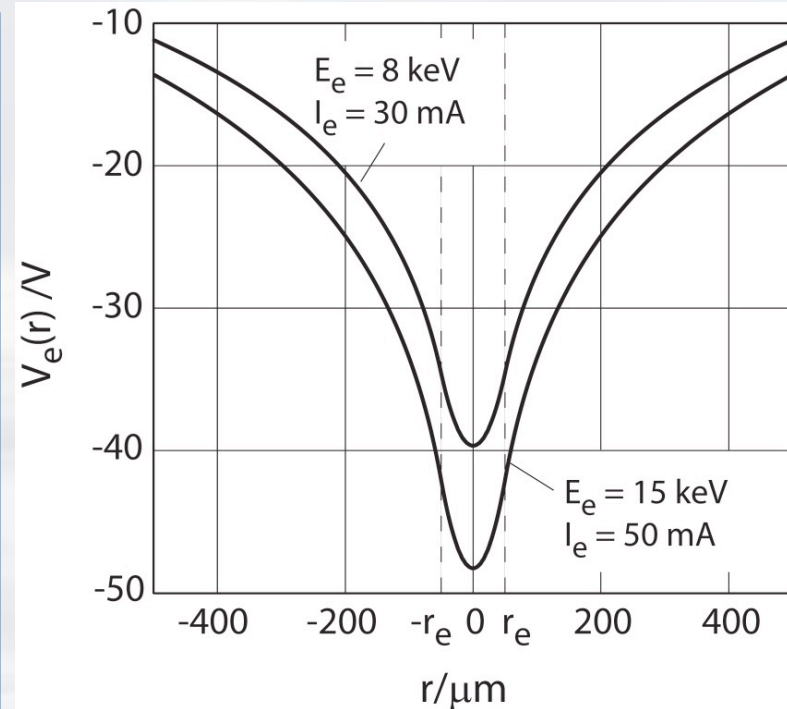
$$V_e(r) = \begin{cases} U_e \left(\frac{r}{r_e}\right)^2 & \text{für } r < r_e \\ U_e \left(2 \ln \frac{r}{r_e} + 1\right) & \text{für } r > r_e \end{cases}$$

with

$$U_e = \frac{I_e}{4\pi\epsilon_0 v_e} = \frac{1}{4\pi\epsilon_0} \cdot \sqrt{\frac{m_e}{2}} \frac{I_e}{\sqrt{E_e}}$$

For estimations we get:

$$U_e = \frac{30 I [A]}{\sqrt{1 - \left(\frac{E_e [keV]}{511} + 1\right)^{-2}}} [V]$$



Example:

Radial trap potentials in a Dresden EBIS.

The potential of the drift tubes is superimposed by U_e in the center of the electron beam.

EBIS: Basic Properties

Electron beam: equation of motion and beam radius



Equation of motion for r

$$\frac{d^2r}{dt^2} = \frac{eI_e}{2\pi\epsilon_0 v_z r m_e} + \frac{e^2}{4m_e^2} \left(\frac{B_c^2 r_c^4}{r^3} - B_z^2 r \right)$$

B_c – B-field at the cathode

r_c – cathode radius

Assuming $B_c = 0$ exists a stationary solution of the above equation. The solution corresponds to an equilibrium flow of the electrons with constant radius, the so- called **Brillouin-Flow**.

We obtain:

$$B = B_B = \frac{1}{r} \left(\frac{2I_e m_e}{\pi\epsilon_0 v_z |e|} \right)^{1/2} \quad \text{and} \quad r_B = \frac{1}{B_B} \left(\frac{2I_e m_e}{\pi\epsilon_0 v_z |e|} \right)^{1/2}$$

For a Brillouin flow all electrons have a constant distance to the beam center. Thereby the Lorentz force caused by the magnetic field is compensated by the space charge and the centrifugal force of the rotating electrons.

EBIS: Basic Properties

Electron beam: beam radius and Herrmann theory *



Electron beam dynamics, considering

- a magnetic field at the cathode,
- thermal effects at the cathode due to filament heating up to the temperature T_c ,
- interactions between the electrons

lead to a corrected electron beam radius (smaller than r_B)

→ Herrmann Theory

* (G.Herrmann; J.Appl.Phys.,
29 (1958) 127)

EBIS: Basic Properties

Electron beam: beam radius and Herrmann theory*



Electron beam radius, enclosing 80% of the beam

$$r_e = r(0) \cdot \sqrt{\left(1 - \frac{r_0}{r(0)}\right)^2 + \frac{2}{1 + \frac{B_C^2 r_C^4}{B_Z^2 r_0^4}} \left(\frac{v_e \tan \gamma}{\frac{e}{m_e} B_Z} \gamma\right)}$$

and

$$r_0 = r_B \left(\frac{1}{2} + \frac{1}{2} \left[1 + 4 \left(\frac{8kT_C r_C^2 m_e}{e^2 B_Z^2 r_B^4} + \frac{B_C^2 r_C^4}{B_Z^2 r_B^4} \right) \right]^{1/2} \right)^{1/2}$$

with

$r(0)$ – beam radius at the cathode

γ – angle deviation from the source axis

* (G.Herrmann; J.Appl.Phys.,
29 (1958) 127)

EBIS: Basic Properties

Electron beam: experimental results

- Electron beam density determines the ionization rate → investigation necessary for understanding the ionization process

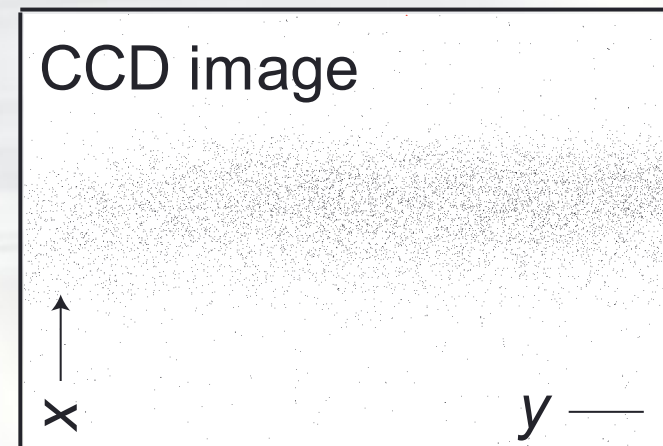
Generally:
electron beam diameter

(40...200) μm

- Dresden EBIT:

$$r_{80\%} = \mathbf{89 \pm 4 \mu\text{m}}; j_e = 96 \pm 9 \text{ A/cm}^2$$

$$\text{@ } E_e = 7.8 \text{ keV}; I_e = 30 \text{ mA}$$



e- beam

aperture

CCD

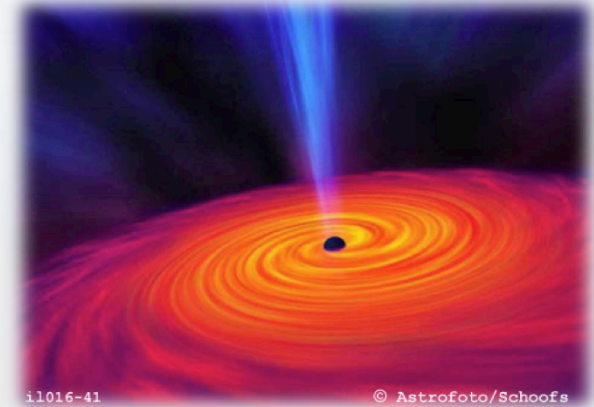
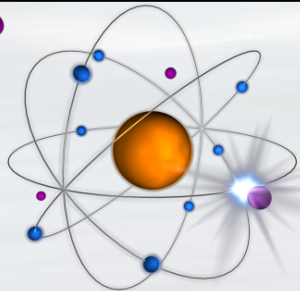


Electron Beam Ion Sources

Production of HCI

Electron-impact ionization:
It is hot in the ion trap

$> 100.000.000 \text{ K}$



Conditions as at the border of a black hole ...

The electron beam energy of an EBIS
can be adjusted with a precision of eV

→ Selective excitation and preparation
of atomic states and ion charge states



Electron beam scalpel

EBIS: Basic Properties

ion beam
properties

pulse form
and width

emittance

energy
spread

extracted
ions
(see later)

Three operation modes:

1. Permanently opened trap – transmission mode

The trap is permanently open and ions are produced in the electron beam without axial trapping.

This mode delivers **high currents of the lowest charged ions (nA ... μ A)**.

2. Partially closed trap – leaky mode

Selecting a low axial potential wall a certain amount of ions with adequate kinetic energy can surpass the potential wall and are extracted continuously.

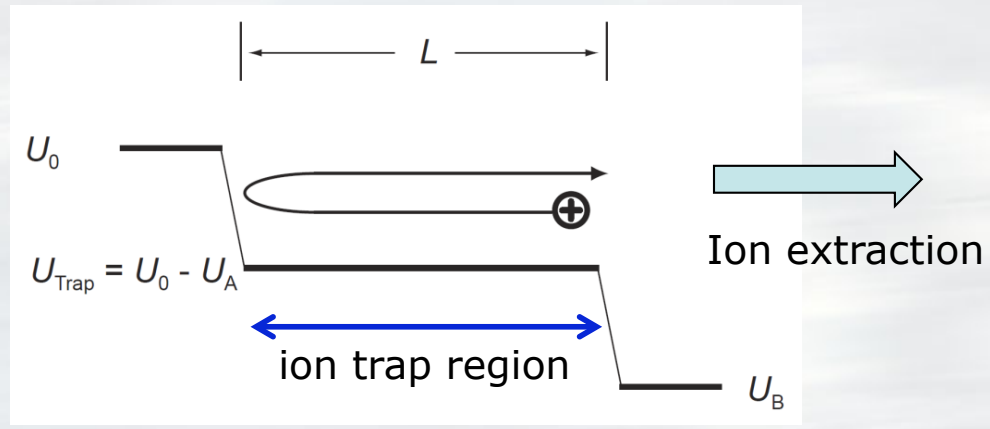
This mode delivers **ions with preferently low up to intermediate ion charge states (up to nA) and a low fraction of higher ion charge states**.

3. Periodically opened and closed trap – pulsed mode

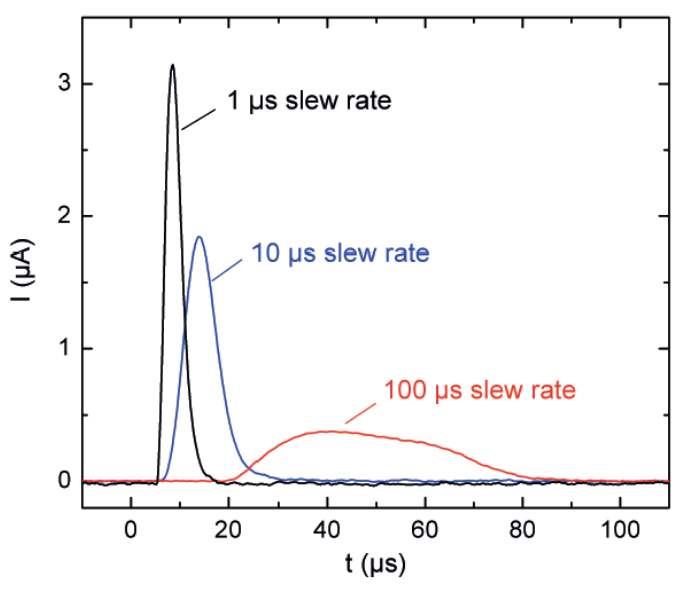
The potential wall is high enough to trap all ions axially. Periodical opening of the trap releases pulses of ions extracted with typical pulse lengths in the order of some microseconds and allow to produce **highest currents of highly charged ions (up to μ A per pulse)**.

Particularities of EBIT/EBIS puls form – classical extraction

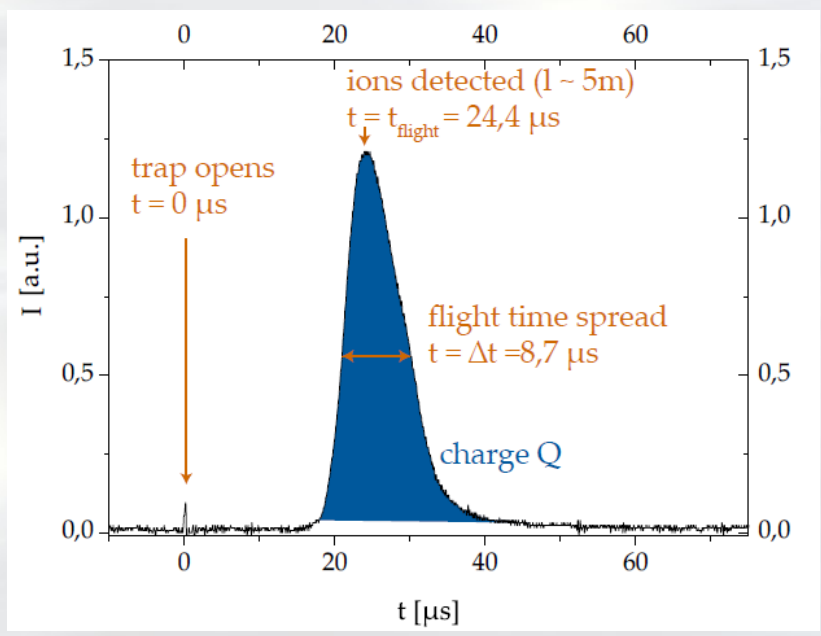
FWHM is in the order of μs



Ion paths in the moment of ion extraction



Ion pulses from the Dresden EBIS-A
Trap length – 6 cm



Particularities of EBIT/EBIS puls form – ns ion extraction

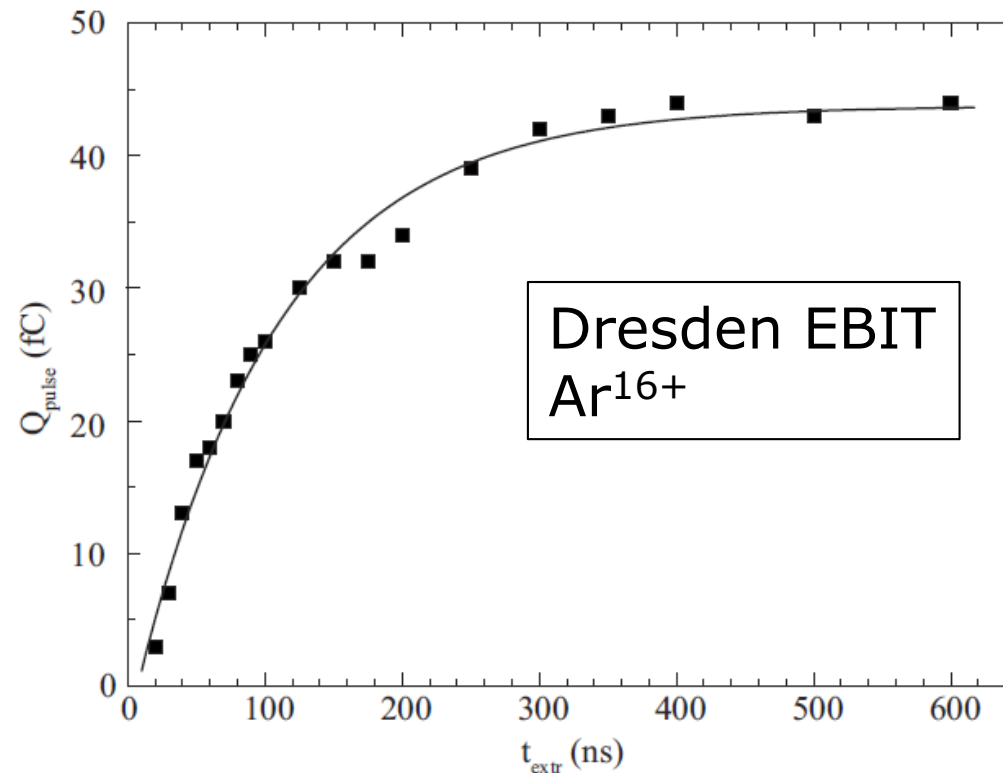
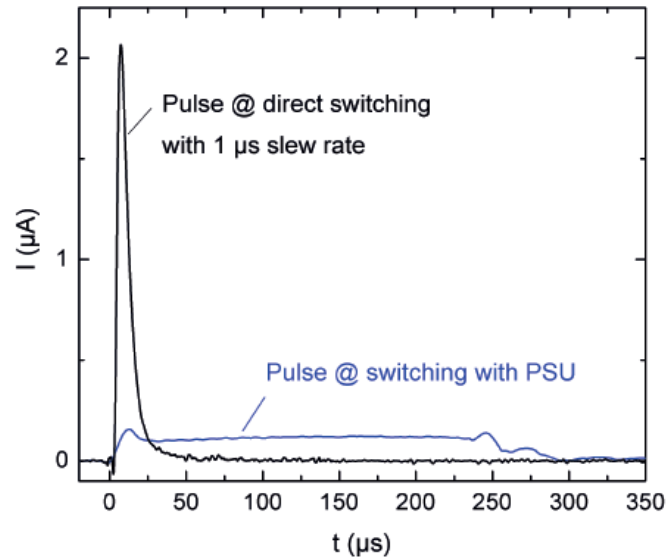


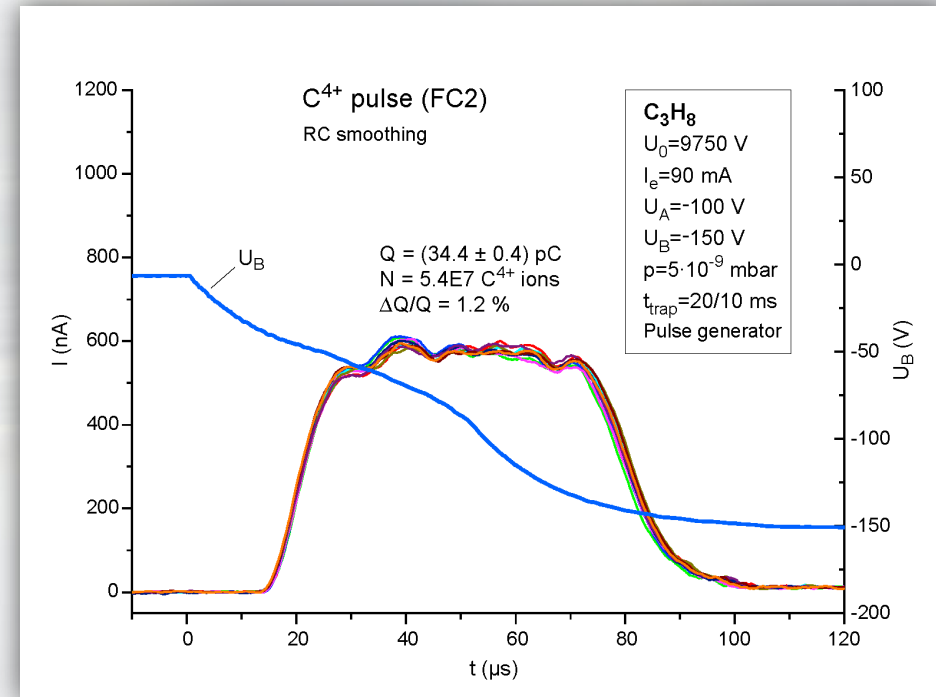
FIG. 3. Extracted ionic charges per Ar¹⁶⁺ pulse in dependence on the extraction time t_{extr} ($U_0=4.0$ kV, $I_e=24$ mA, $t_{\text{cyc}}=100$ μs , $t_{\text{wait}}=1$ s, $p=3.1 \times 10^{-9}$ mbar). The solid line is a guide to the eye.

REVIEW OF SCIENTIFIC INSTRUMENTS 81, 02A507 (2010)

Particularities of EBIT/EBIS puls form – flat top pulses



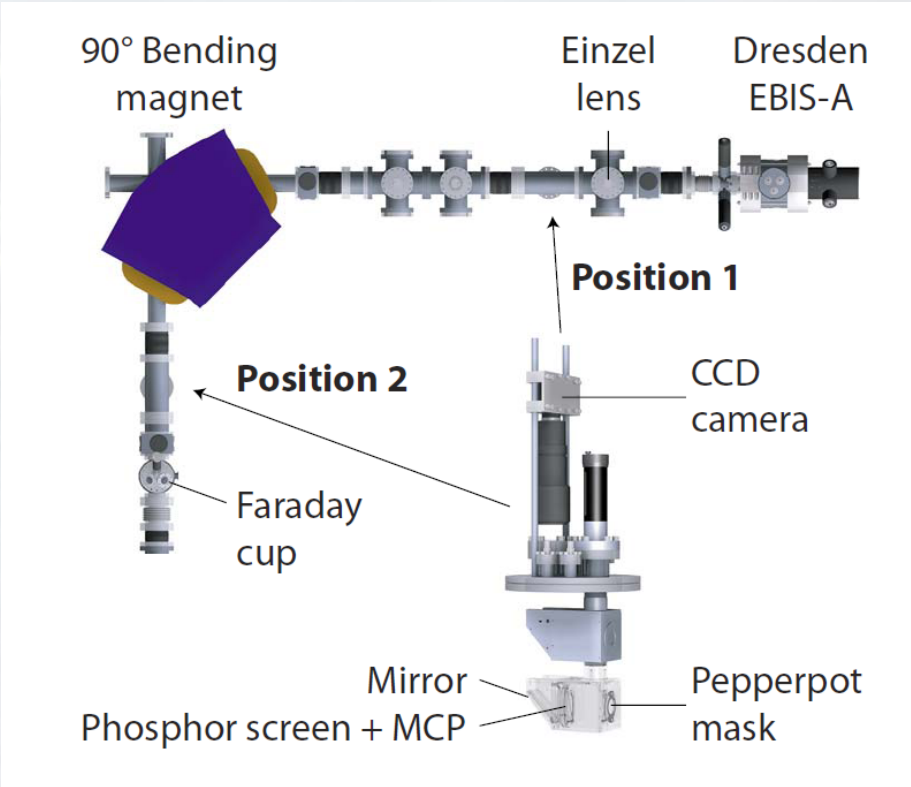
Proton pulse from the Dresden EBIS-A after direct trap opening and after switching with the PSU forming a flat-top pulse shape



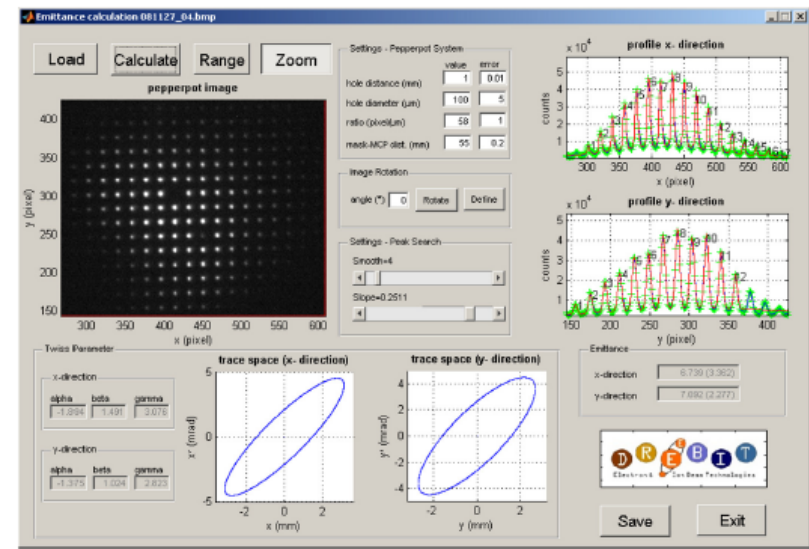
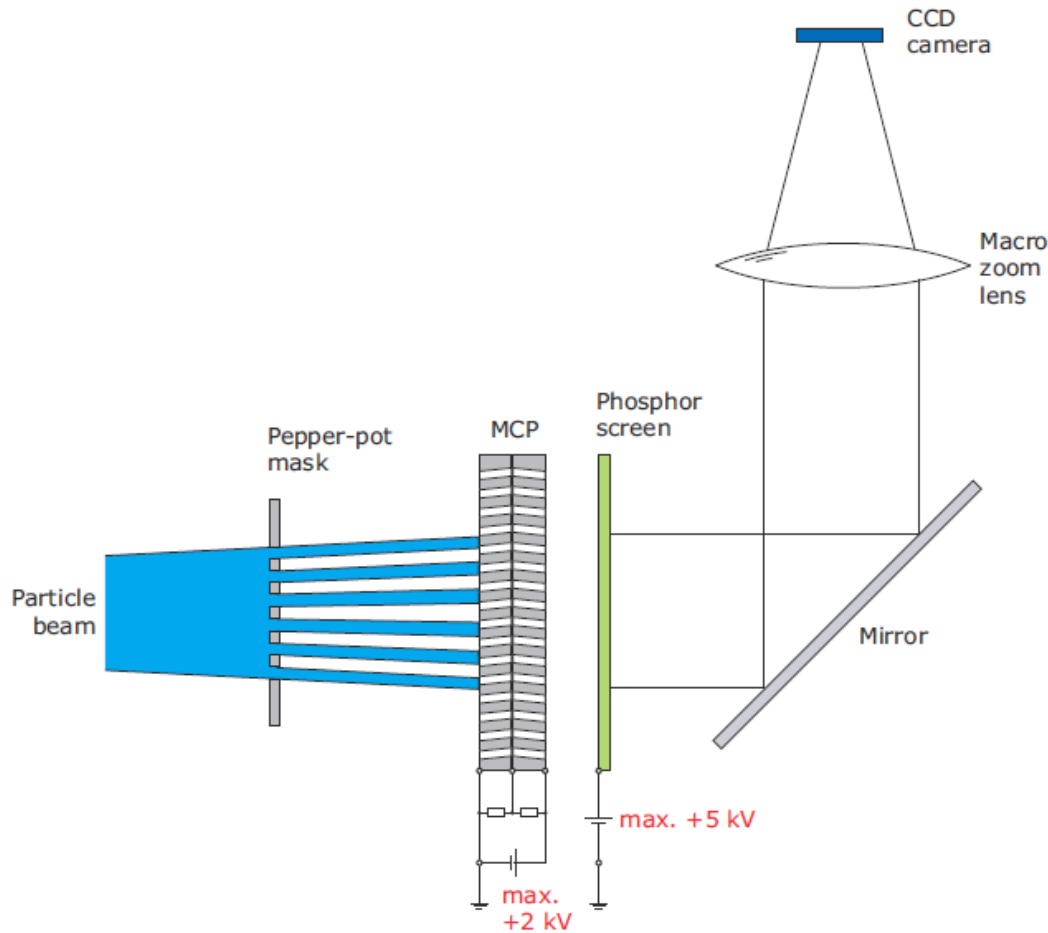
One of the requirements for the applications of EBIS with synchrotrons are flat-top pulses.

Controlling properly U_B flat-top pulses with FWHM to at least 100 μ s can be formed.

Particularities of EBIT/EBIS Emittance



Particularities of EBIT/EBIS Emittance

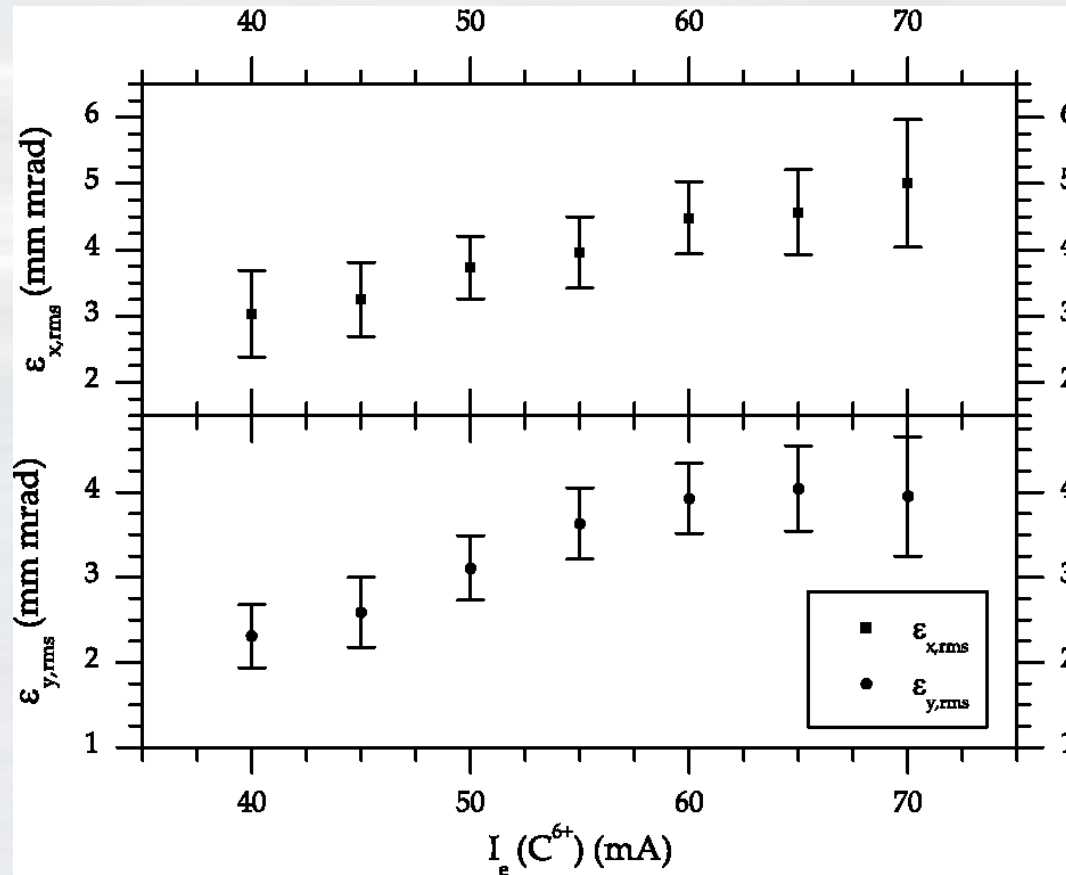


Display of the results after calculation

The scheme of the Pepper-Pot Emittance Meter is pictured in figure 2. The incoming particle beam passes the Pepper-Pot mask and is separated into several beam spots. The particles hitting the MCP create an electron current which is amplified passing the two micro channel plates. The electrons are then accelerated towards the phosphor screen. The visible light spots created at the phosphor screen are detected after 90° deflection by a CCD camera.

The emittance of the beam can be determined from the position, the size, and the shape of the light spots.

Particularities of EBIT/EBIS Emittance



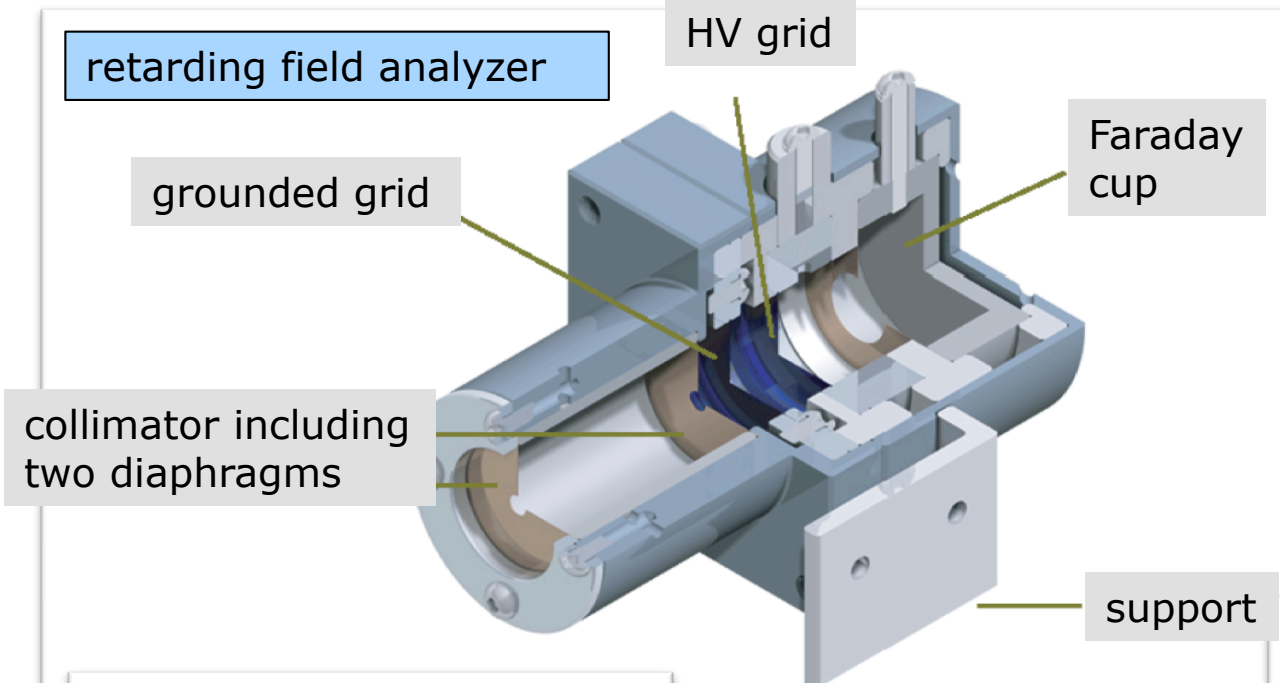
Dresden EBIS-A

For comparison:

Dresden EBIS-SC
about 30 mm mrad
for $I_e = 300$ mA

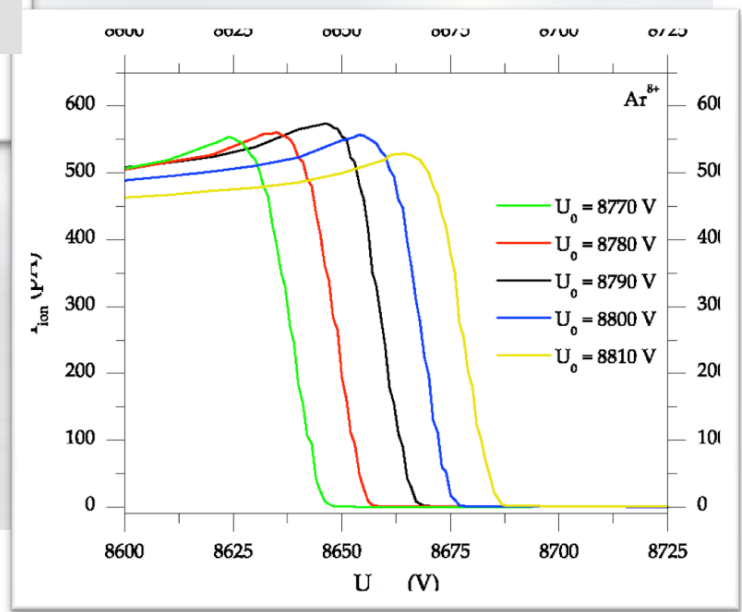
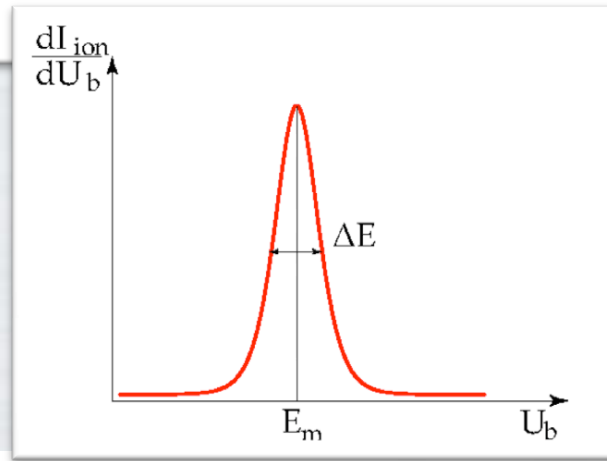
Particularities of EBIT/EBIS

Energy spread



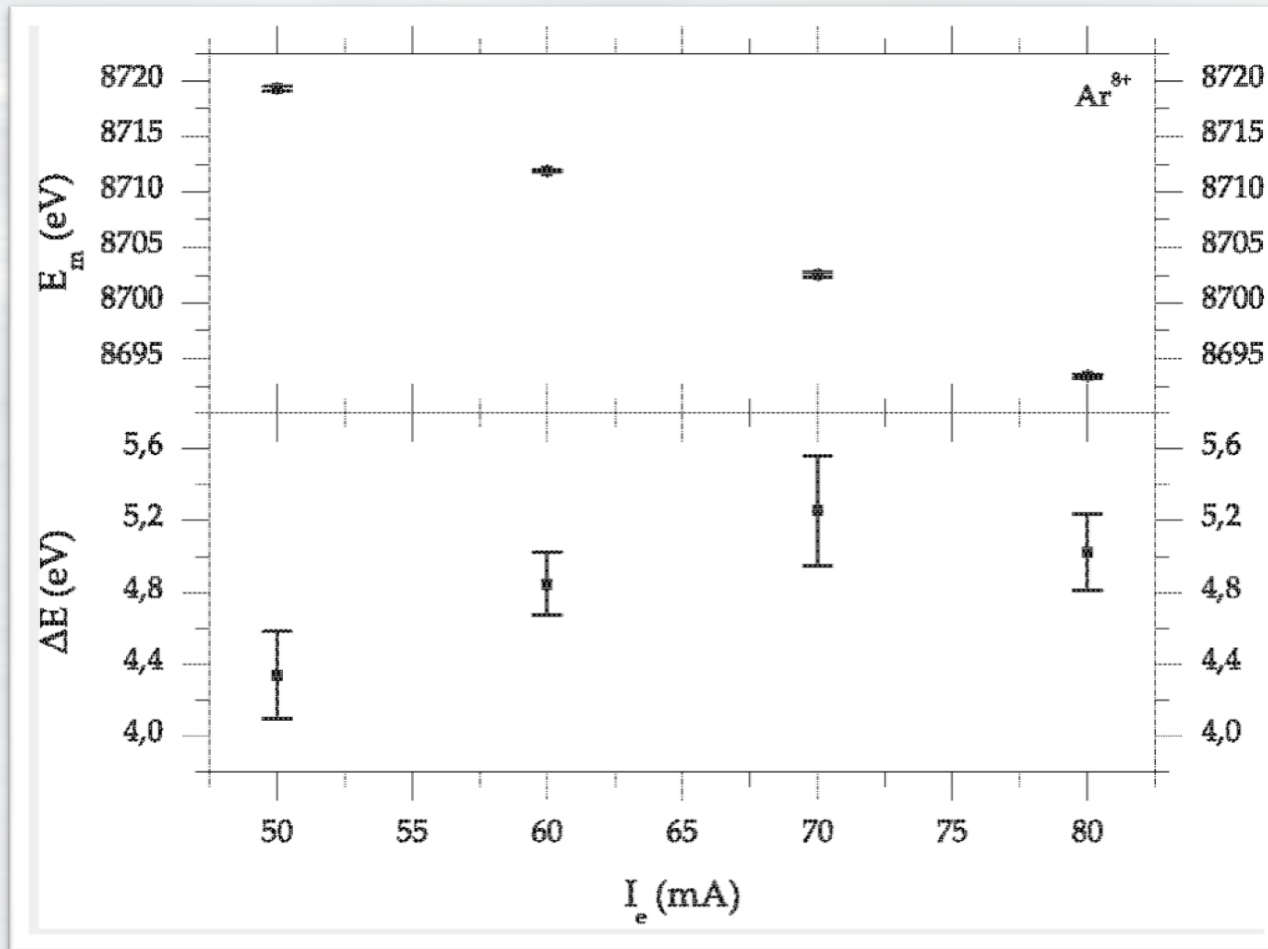
The differentiation of the measured curves gives

- The energy spread
- The total beam energy of the analyzed ion beam.



Particularities of EBIT/EBIS

Energy spread



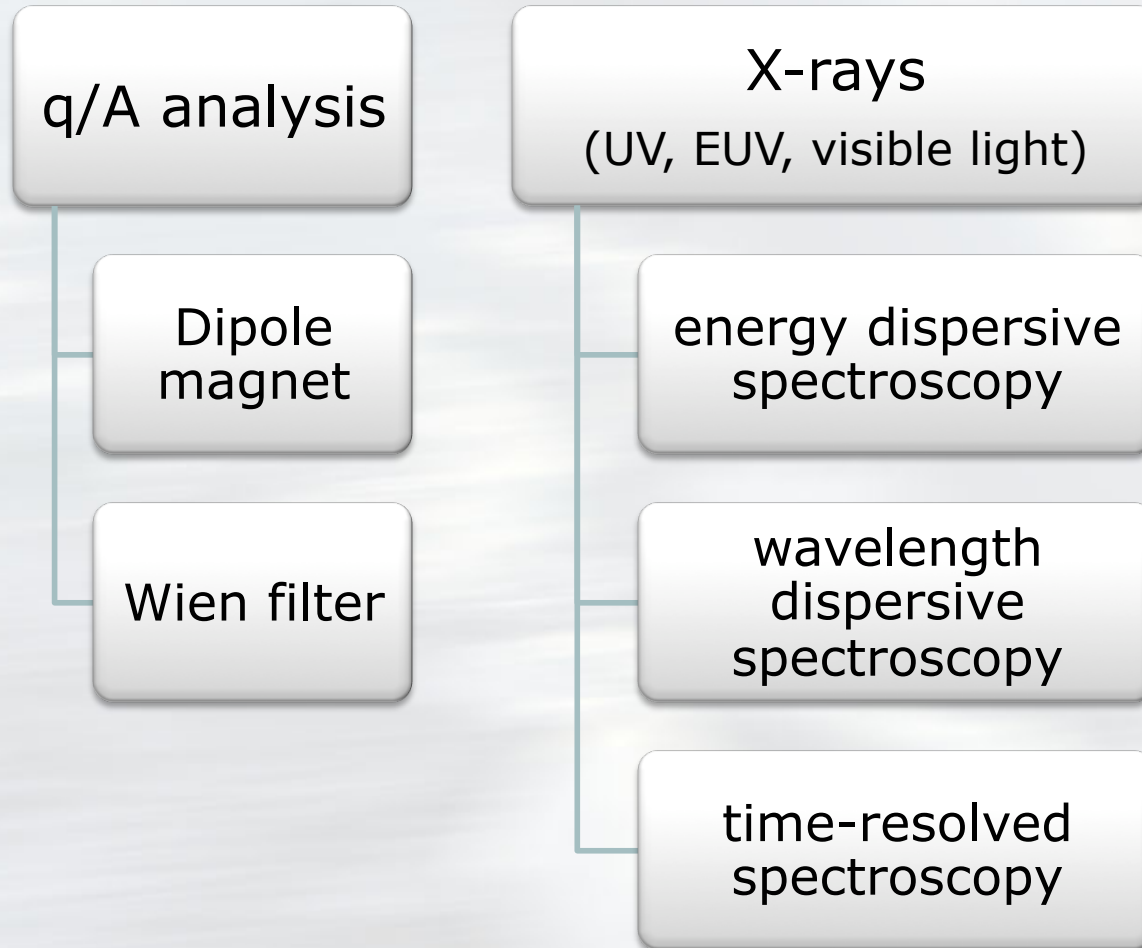
Dresden EBIS-A Ar^{8+}

Total ion beam energy
(shifts due to different
depths of the beam
Coulomb potential)

Energy spread of Ar^{8+}
(the energy spread is
below 1 eV/u in any case)

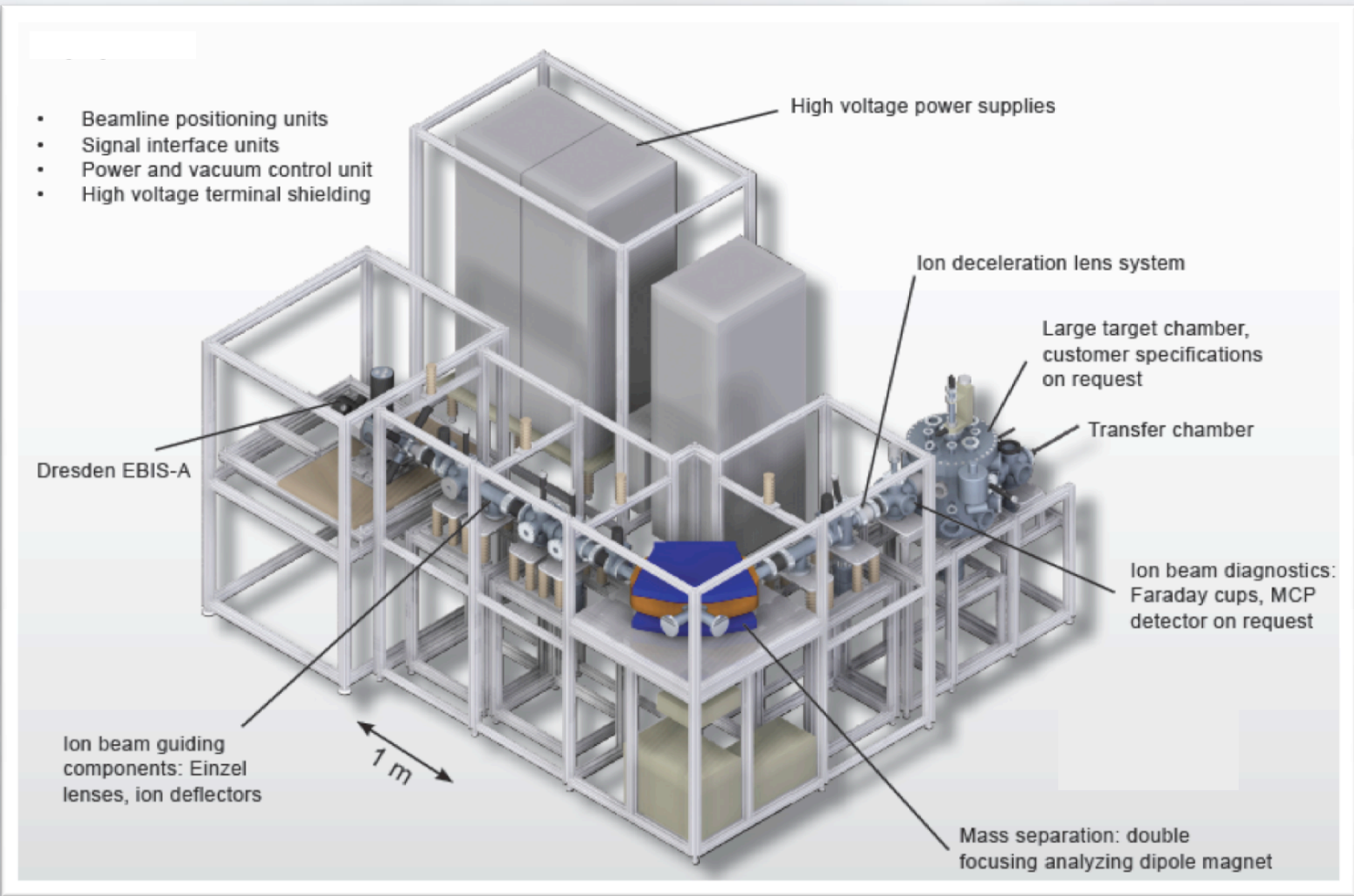
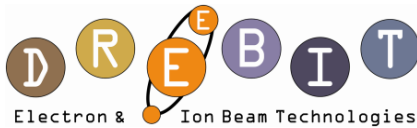
EBIS: Diagnostics

Processes in the ion source



EBIS: Diagnostics

q/A Analysis



q/A analysis: dipole magnet

Objective: charge state separated ion beam

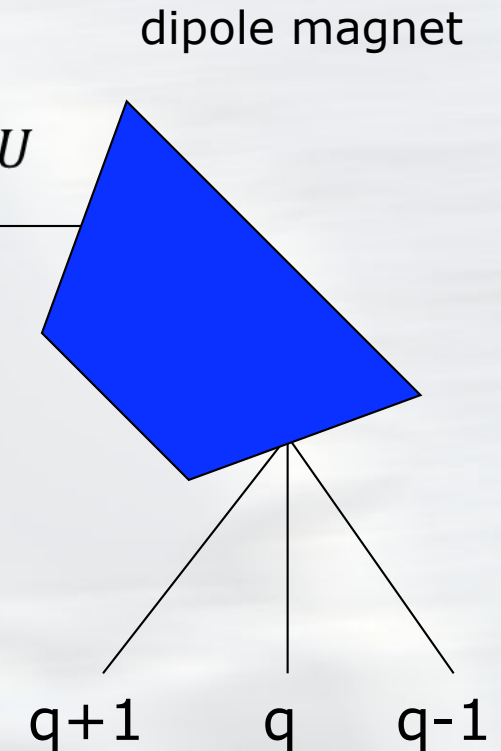
Lorentz force = centripetal force

$$q \cdot v \cdot B = \frac{mv^2}{r}$$

- Ion charge state separation

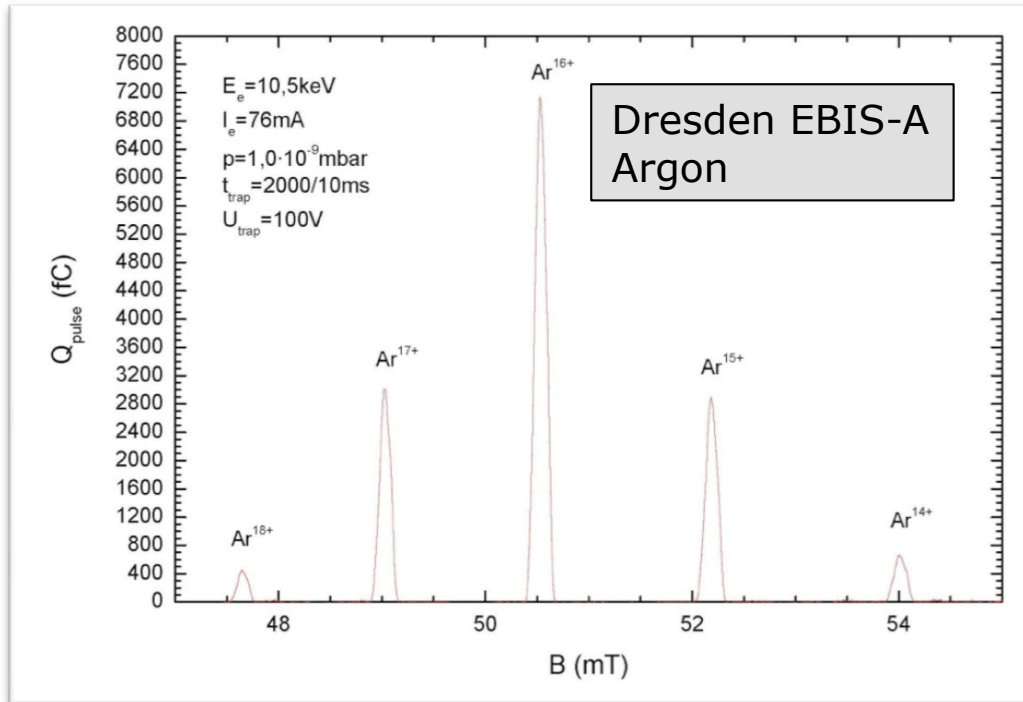
$$\frac{q}{A} = \frac{2 \cdot U}{r^2 B^2}$$

$$E_{kin} = q \cdot U$$

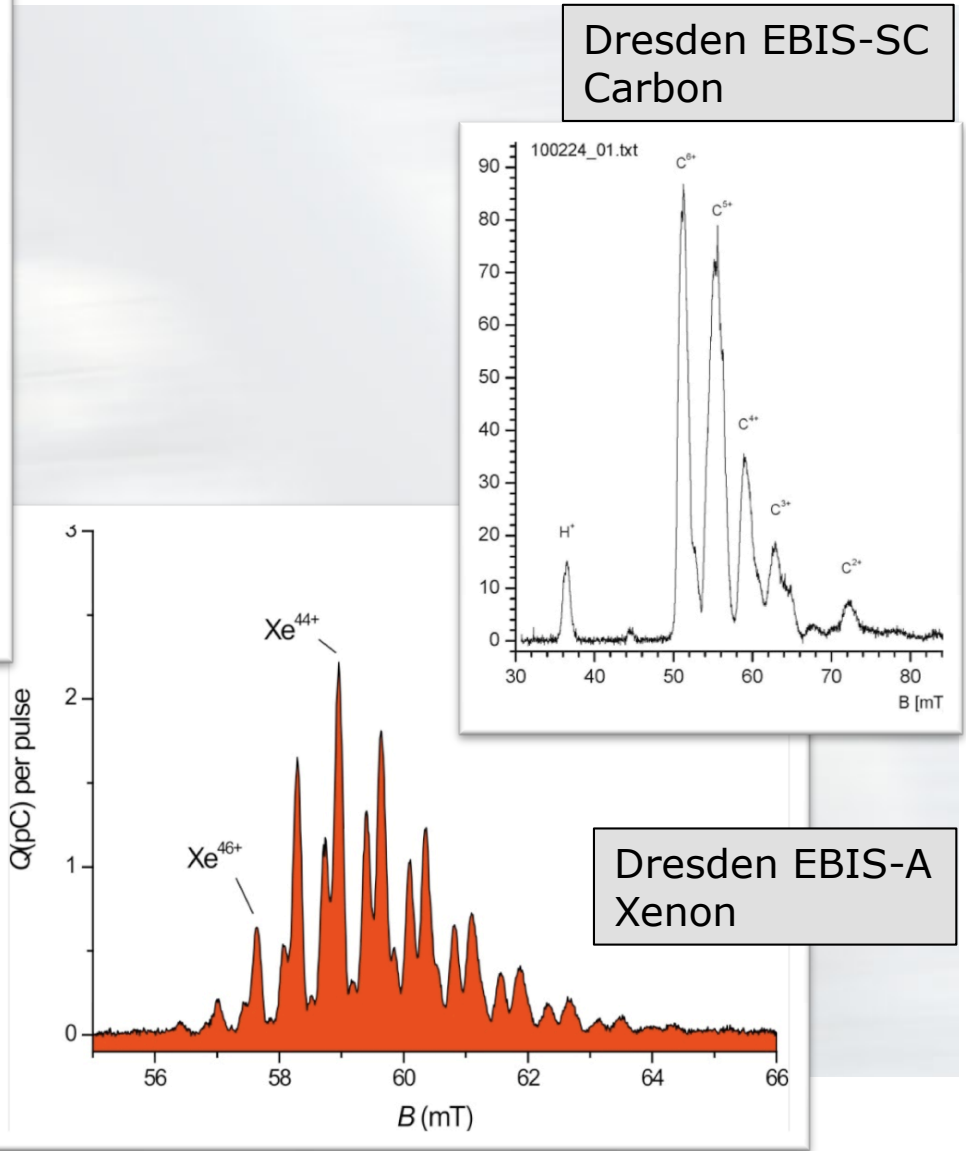


q/A analysis: dipole magnet

Examples for ion charge state spectra



Ion extraction spectra measured with a Faraday cup after ion charge state separation.

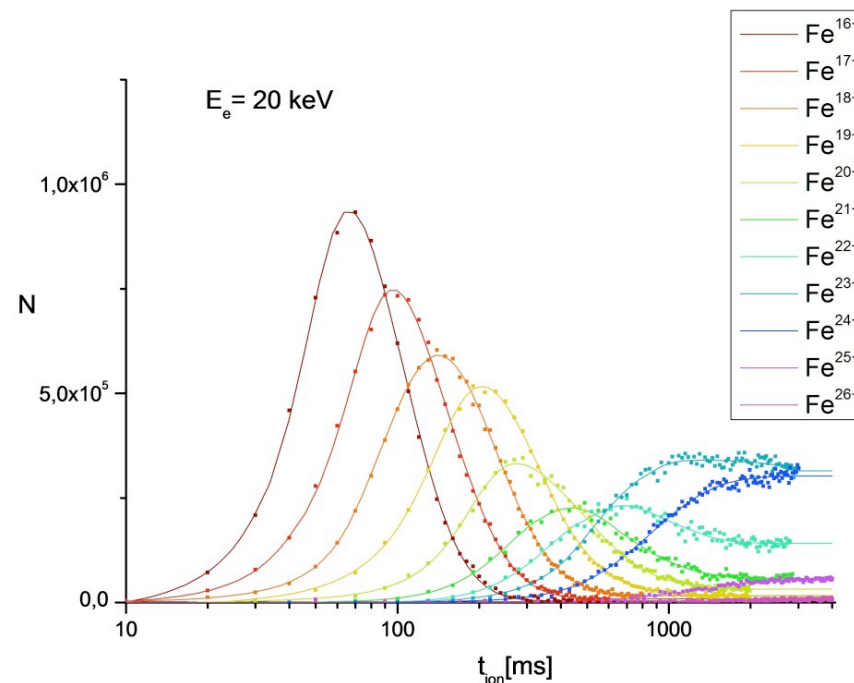


q/A analysis: dipole magnet time-resolved ion charge state spectra

Signal intensity of individual ion charge states measured at different ionization times → reveals the evolution of charge states in the trap

Further analysis allows for

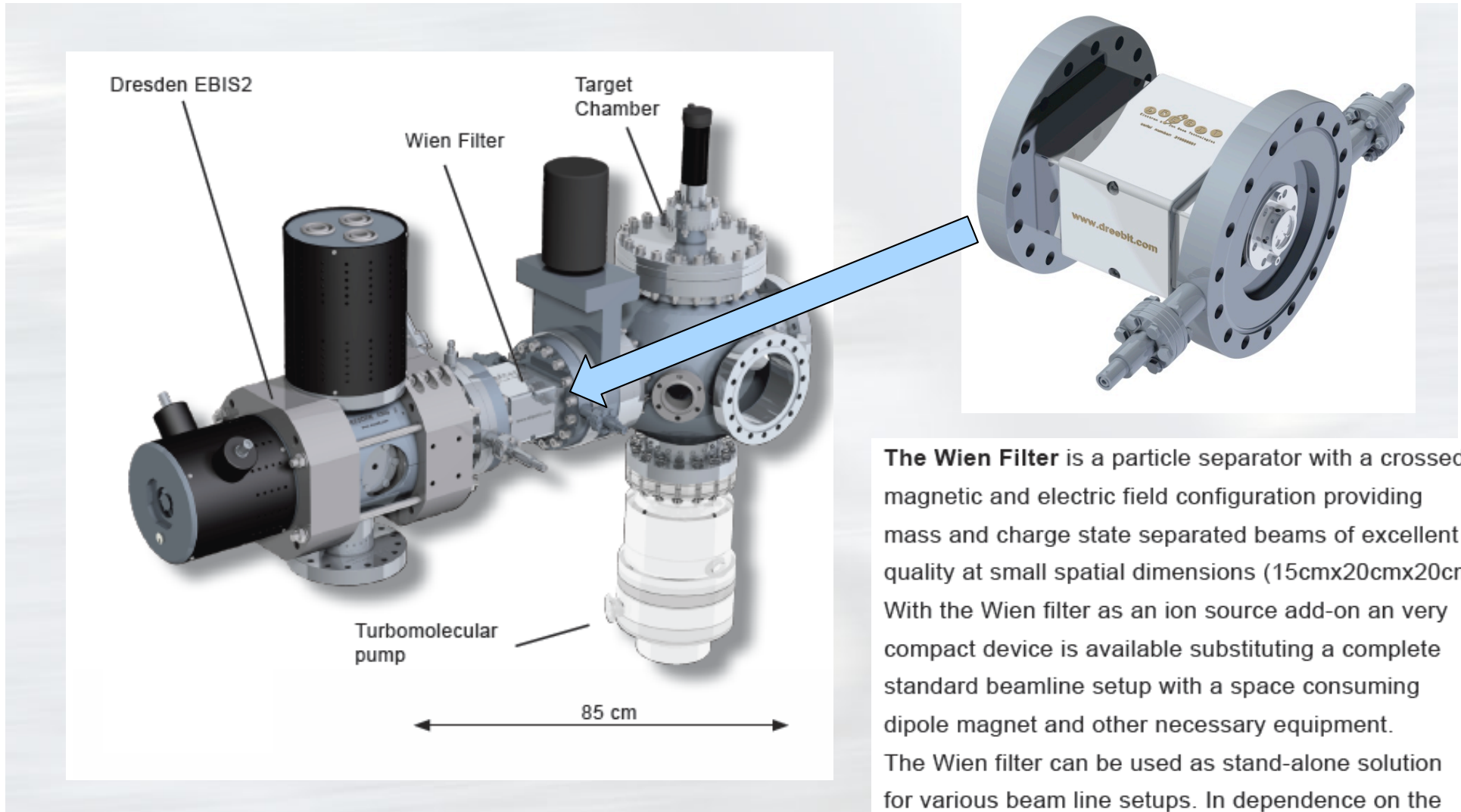
- characterizing the charge balance inside the trap
- estimating the ionisation factor of the source
- determining electron impact ionisation cross sections.



Dresden EBIT Fe^{q+}

EBIS: Diagnostics

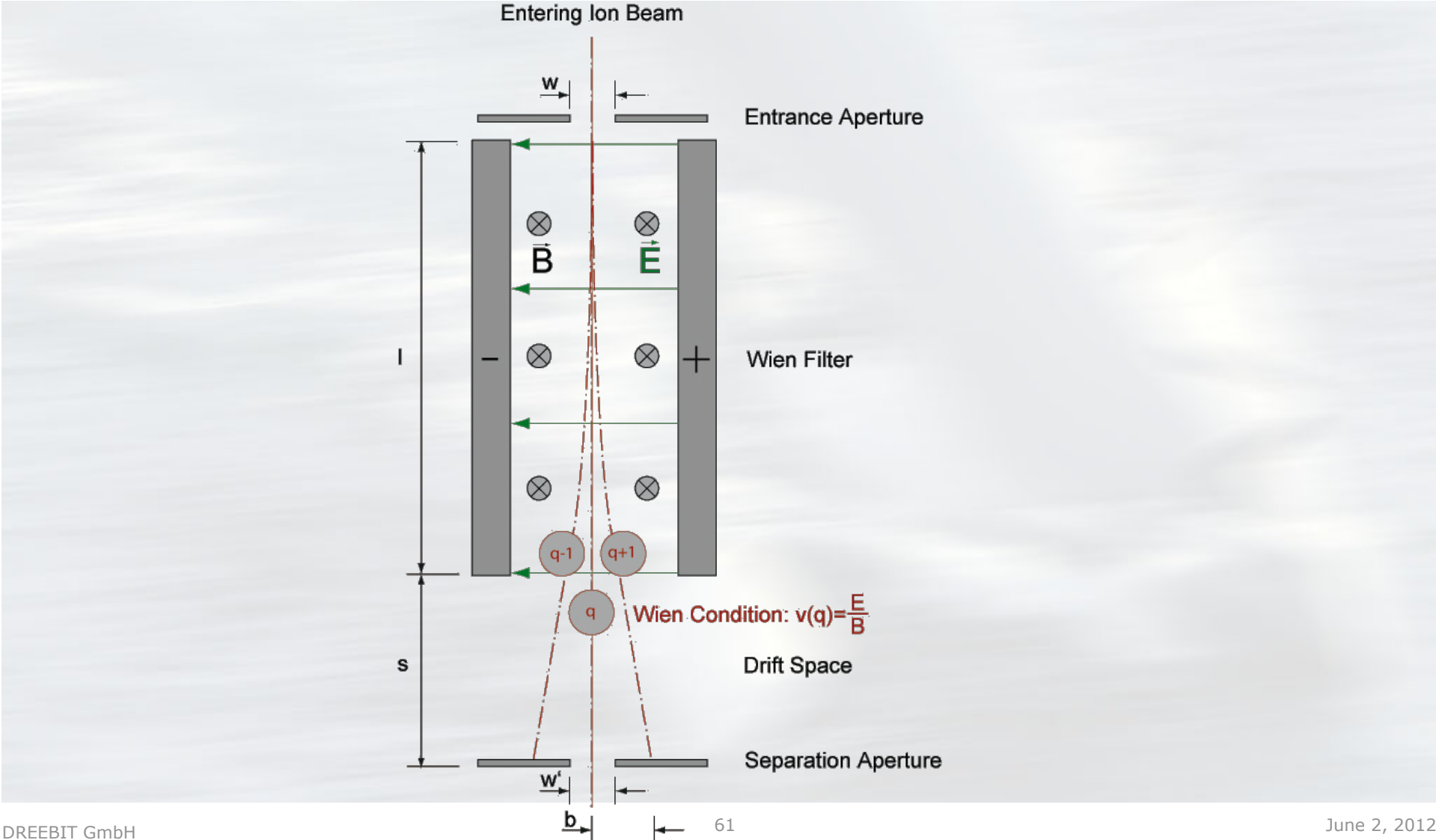
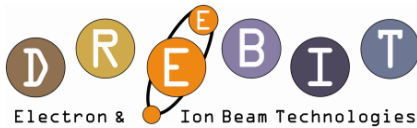
q/A Analysis with a Wien filter



The Wien Filter is a particle separator with a crossed magnetic and electric field configuration providing mass and charge state separated beams of excellent quality at small spatial dimensions (15cmx20cmx20cm). With the Wien filter as an ion source add-on an very compact device is available substituting a complete standard beamline setup with a space consuming dipole magnet and other necessary equipment. The Wien filter can be used as stand-alone solution for various beam line setups. In dependence on the installed mass- and charge state-separating aperture a resolution of better than 80 is available.

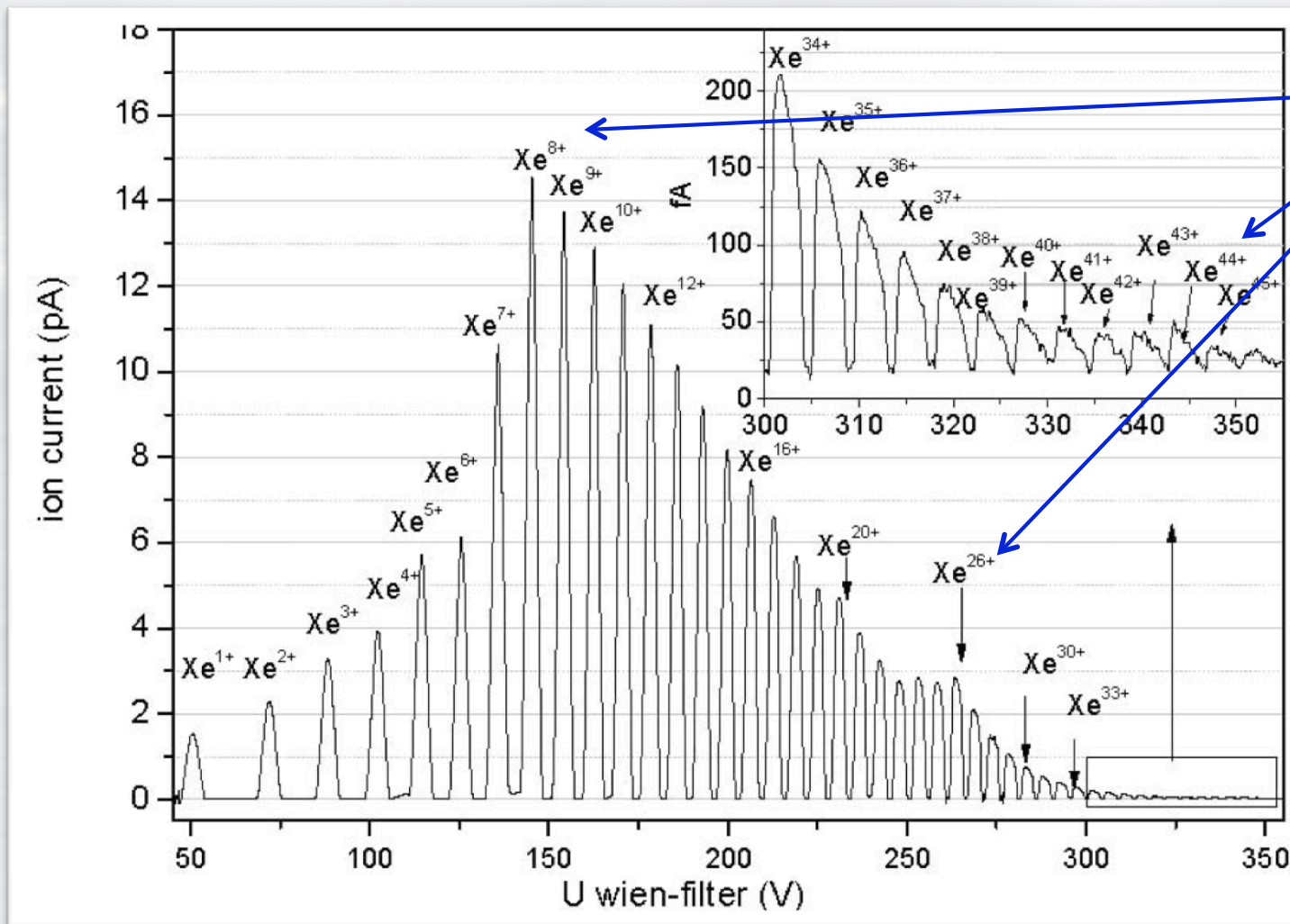
EBIS: Diagnostics

q/A Analysis with a Wien filter



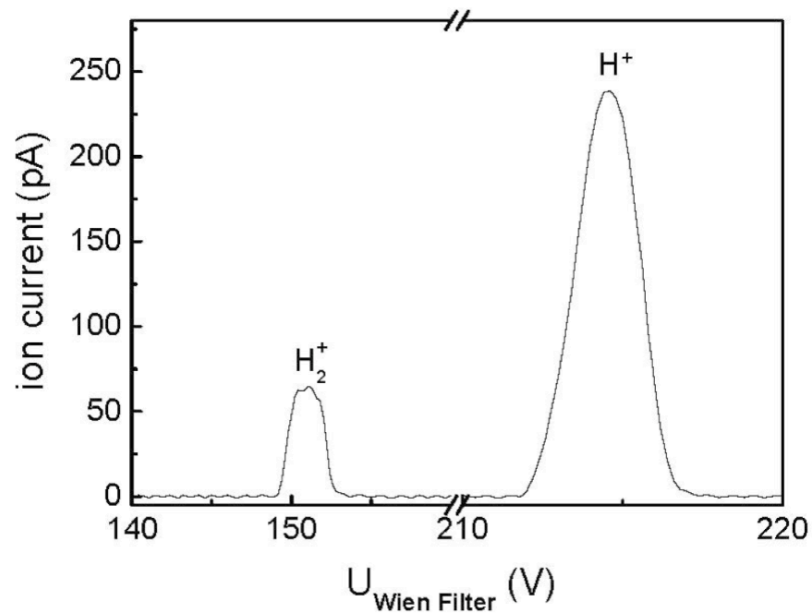
q/A analysis: Wien filter

Examples for ion charge state spectra

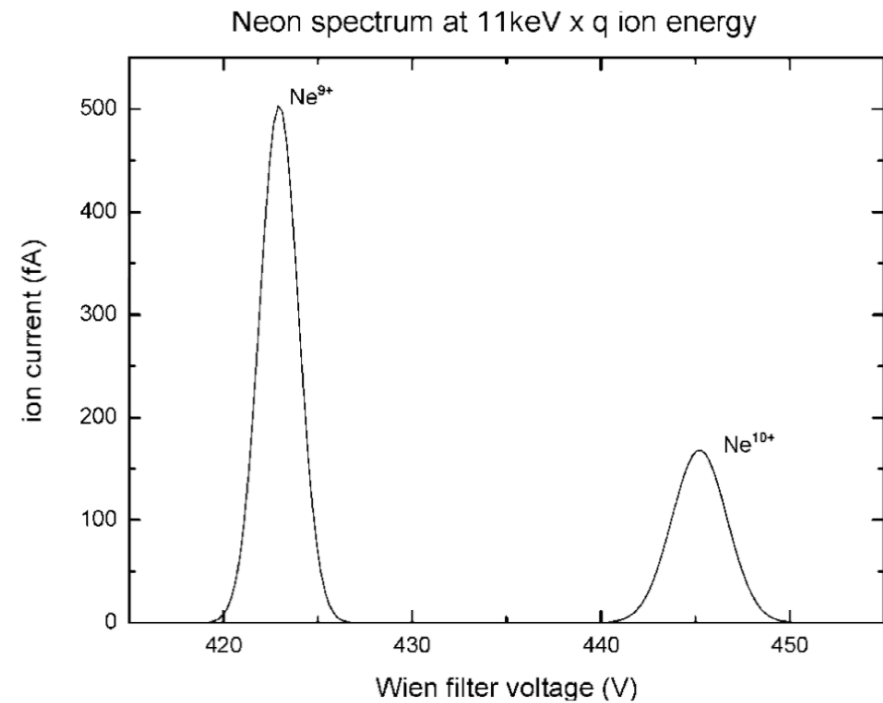


q/A analysis: Wien filter

Examples for ion charge state spectra



Dresden EBIT
Hydrogen



Dresden EBIT
Neon

DREEBIT Ion Sources

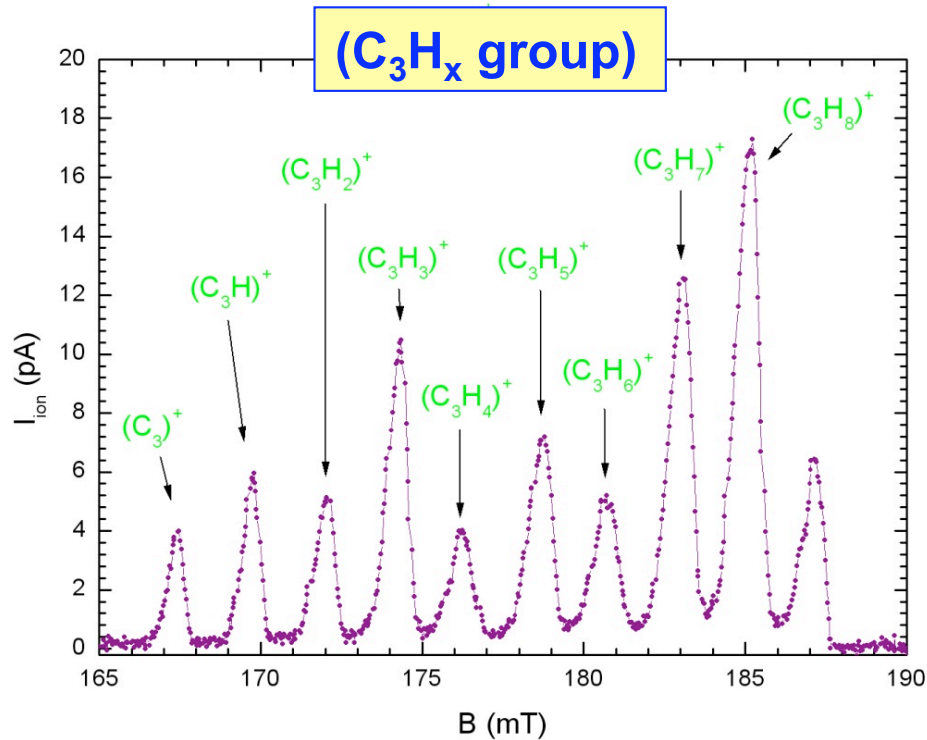
Pulsed mode



Pulsed mode (ions/pulse)

Ion	EBIT	EBIS-A	EBIS-SC	EBIT:EBIS-A:EBIS-SC
C⁴⁺	24.000.000	80.000.000	900.000.000	1 : 3 : 38
C⁶⁺	10.000.000	30.000.000	400.000.000	1 : 3 : 40
Ar¹⁶⁺	900.000	7.800.000	250.000.000	1 : 9 : 278
Ar¹⁷⁺	45.000	1.400.000	22.000.000	1 : 31 : 489
Ar¹⁸⁺	6.000	90.000	1.500.000	1 : 15 : 250
Xe⁴⁴⁺	10.000	700.000	10.000.000	1 : 70 : 1000

Beams of molecular fragments



Propane C_3H_8

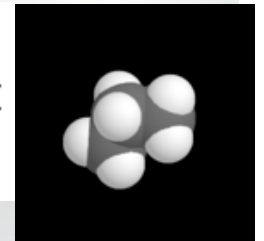
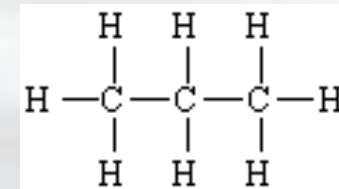
Extraction of all
molecular fragments

C_xH_y

$x = 1 \dots 3$

$y = 1 \dots 9$

$y = 9$: protonation



A unique possibility to form beams of exotic molecular fragments

EBIS: X-Ray Diagnostics

X-ray output from EBIS

For an x-ray detector the following count rate can be expected

$$\dot{N}_q = \varepsilon \frac{\Omega}{4\pi} V j_e \omega_q n_q \sigma_q^{exc}$$

$$\dot{N}_q = \varepsilon \frac{\Omega}{4\pi} I_e \omega_q n_q \sigma_q^{exc}$$

with

ε	-	detector efficiency
$\Omega/4\pi$	-	solid angle
V	-	apparent beam volume
L	-	apparent beam length
ω_q	-	x-ray fluorescence yield
n_q	-	number of ions with the charge q
E_{if}	-	transition energy
Σ_q^{exc}	-	excitation cross section

The emitted radiation power can be estimated as

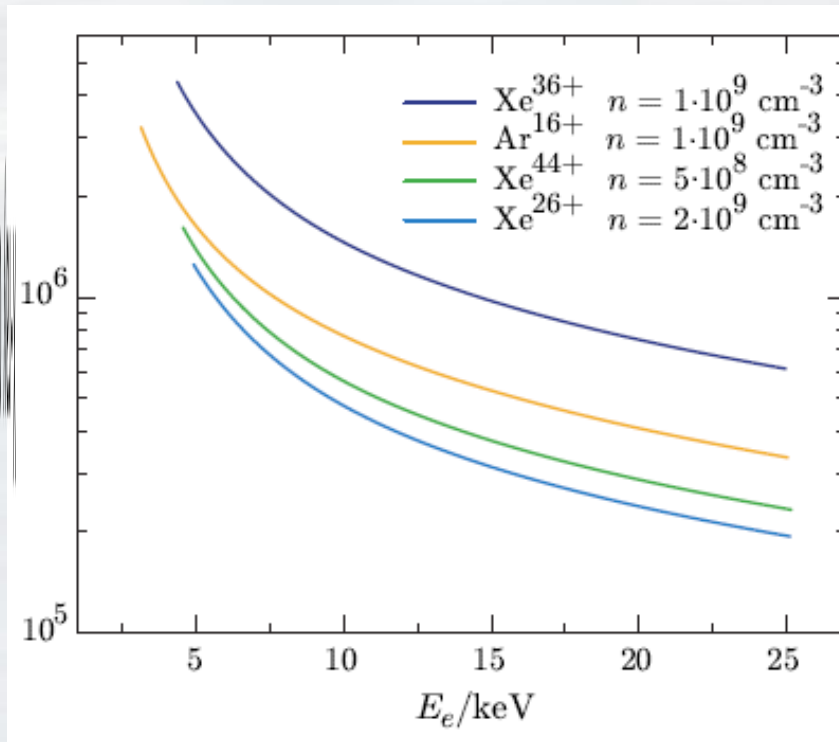
$$P = \dot{N}_q E_{if} e$$

For individual dipole lines radiation power on the order of nW was recorded

EBIS: X-Ray Diagnostics

X-ray output

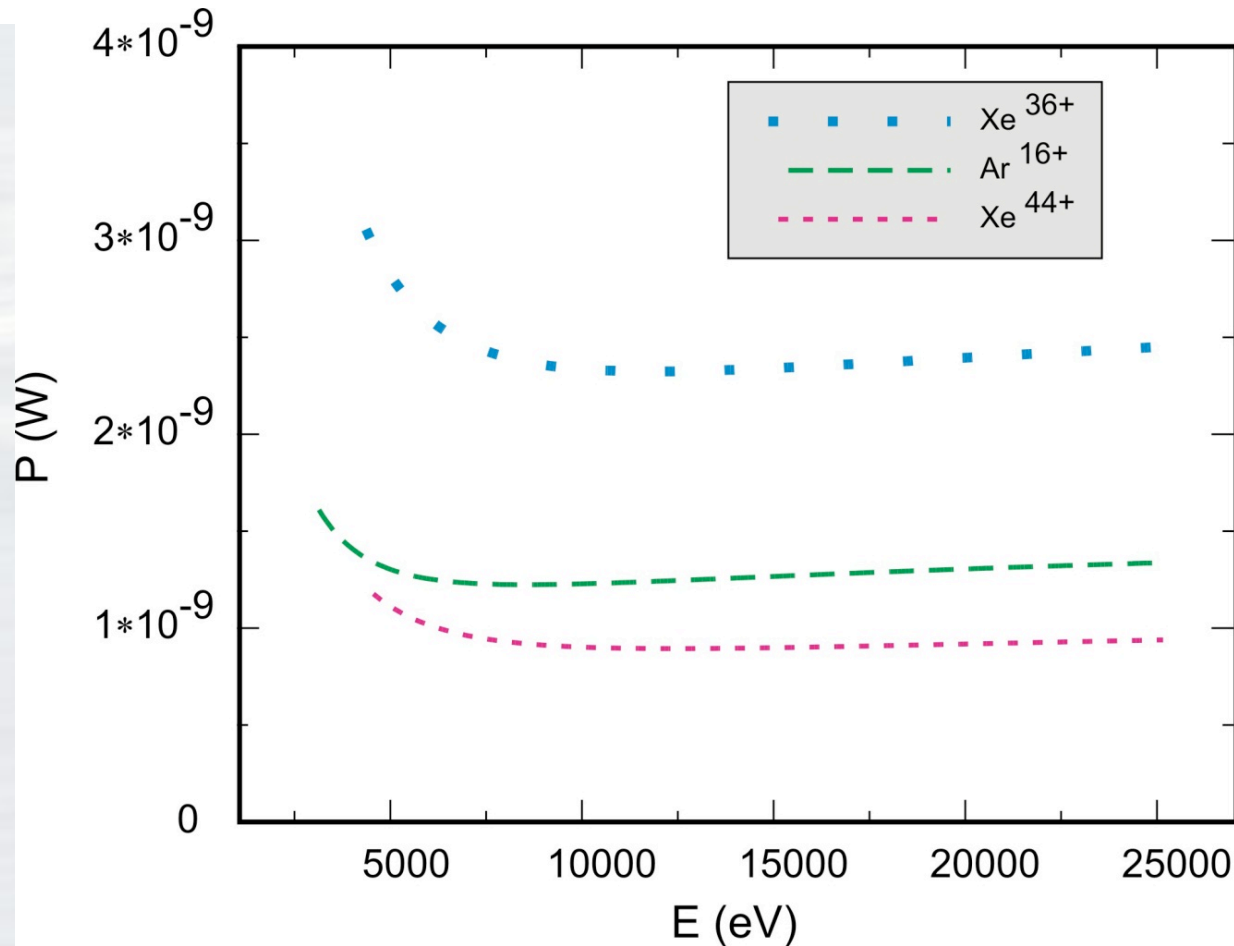
EBIS are excellent sources of X-rays from highly charged ions.



ion	transition $i \rightarrow f$	E_{if}/eV	$A_{if}/\text{eV/h}$
Ar^{16+}	$2p(^1P_1) \rightarrow 1s(^1S_0)$	3138,8	0,073
Xe^{26+}	$3d(^1P_1) \rightarrow 2p(^1D_2)$	4159.6	0.208
Xe^{36+}	$3d(^1P_1) \rightarrow 2p(^1S_0)$	4366.8	0.372
Xe^{44+}	$3d(^1P_1) \rightarrow 2p(^1S_0)$	4558.0	0.321

X-ray output from the Dresden EBIT

X-rays from highly charged ions radiation power of the Dresden EBIT



The graph shows the most dominant dipole transitions for DE in case of 3 different ion species

EBIS-A: $\times 10$

EBIS-SC: $\times 200$

→ higher transition power

$$P[W] = \frac{\text{photons}}{s} E[eV] e[As]$$

Z-dependence allowed and forbidden transitions

HFS, QED, parity violation



E1 ($\Delta n = 0$)	Z
E1 ($\Delta n \neq 0$)	Z^4
M1 ($\Delta n = 0$)	Z^3
M1 ($\Delta n \neq 0$)	Z^6
M1(within fine structure)	Z^{12}
E2 ($\Delta n = 0$)	Z
E2 ($\Delta n \neq 0$)	Z^6
E2 (within fine structure)	Z^{16}
2E1	Z^6
E1M1	Z^6
Hyperfine splitting	Z^3
QED effects	Z^4
E_{SO}	Z^4
Parity violation	Z^5

Table III. *The Z-dependence of the probabilities of allowed and forbidden transitions, hyperfine interaction, QED effect, relativistic effects and parity violation effect along the Hydrogen iso-electronic sequence.*

With higher atomic number the intensity of otherwise weak transitions increases.

For highly charged ions otherwise forbidden transitions can become dominant.

Physica Scripta. Vol. T120, 47–52, 2005

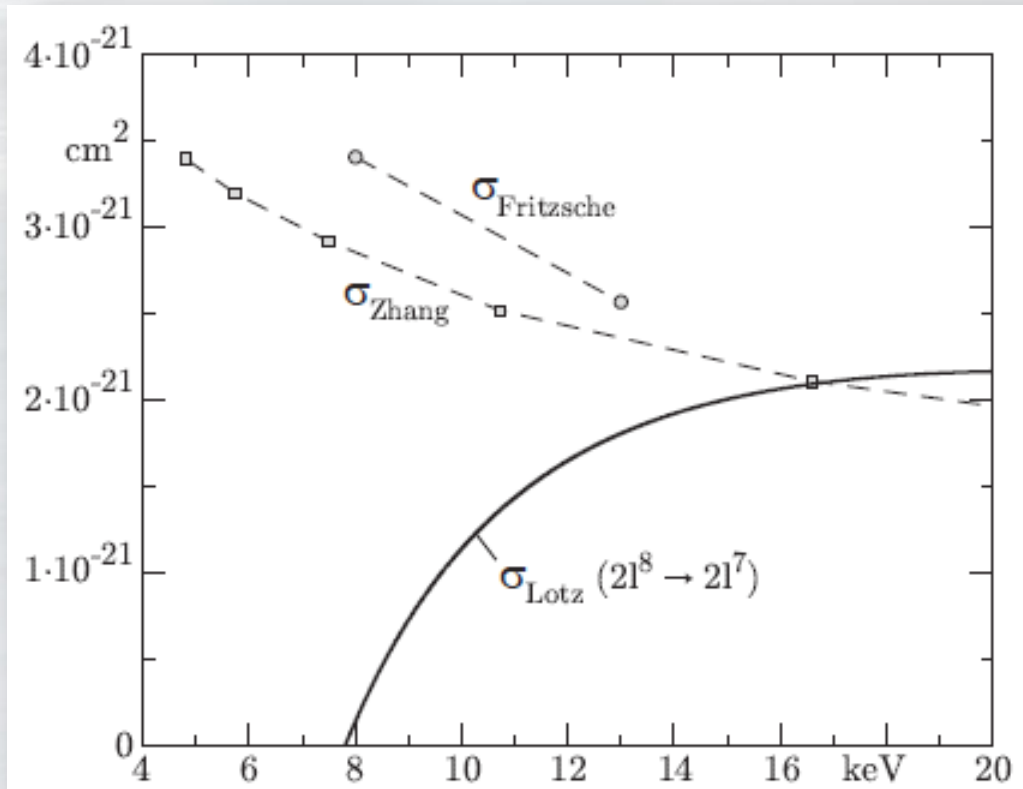
Physics Based on Electron Beam Ion Traps*

Yaming Zou^{1**} and Roger Hutton²

¹The Key Lab of Applied Ion Beam Physics, The Ministry of Education, China
Modern Physics Institute, Fudan University, Shanghai 200433, China

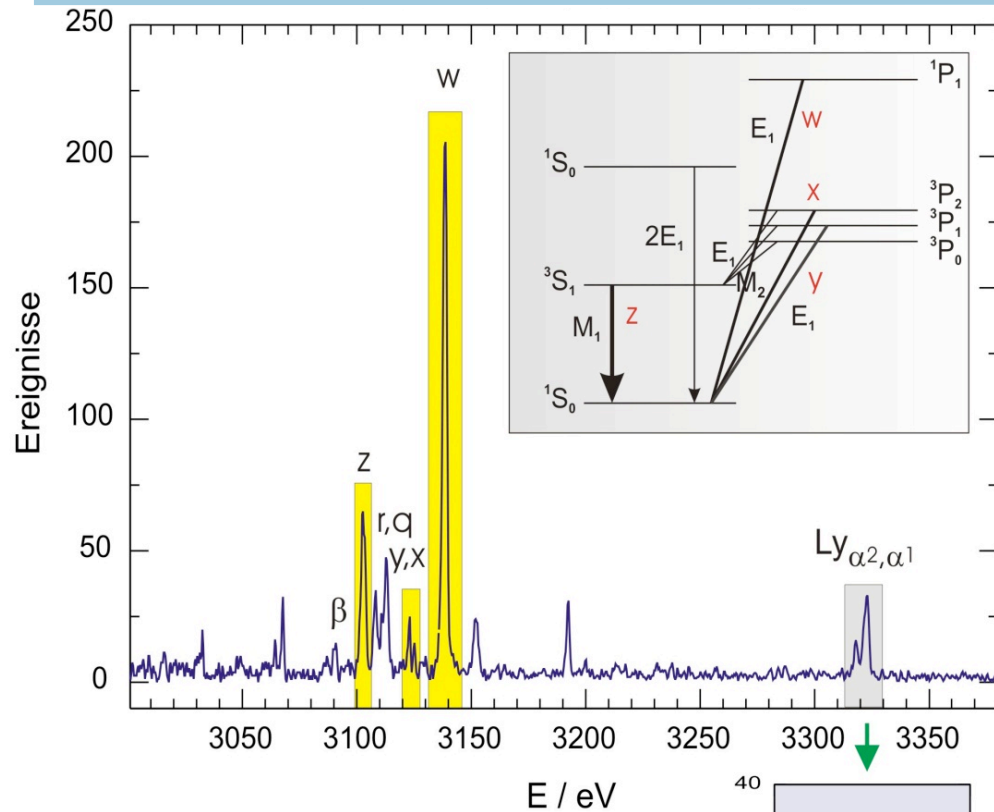
²Astronomy Department, Lund University, BOX 43, SE 221 00 Lund, Sweden

X-rays: excitation vs. ionisation



excitation cross-section for
 $2p(^1S_0) \rightarrow 3d(^1P_1)$
vs.
L-shell ionisation cross-section
in Xe^{44+}

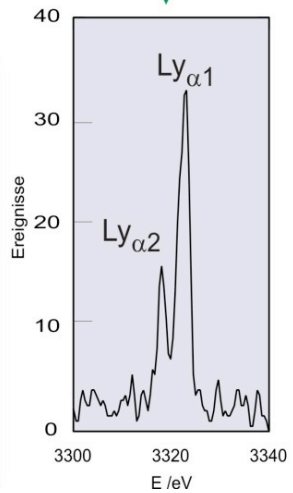
Wavelength X-ray spectroscopy: Argon



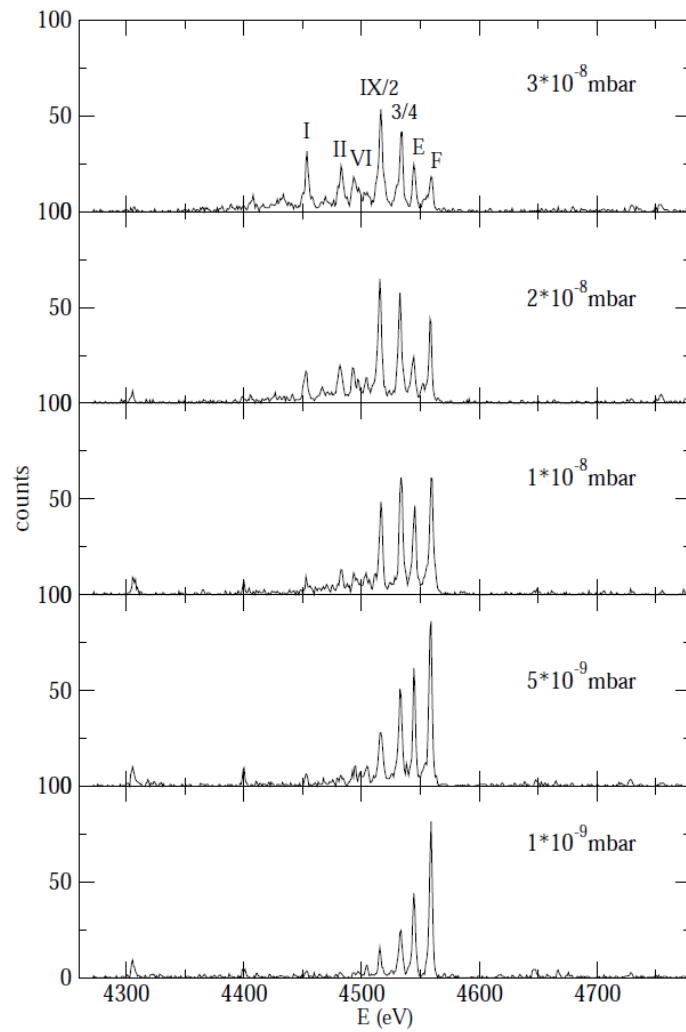
Transition energies in one- and two-electron systems can be calculated very precisely.

Therefore hydrogen-like ions are excellent sources for well known x-ray transitions: **Lyman lines**

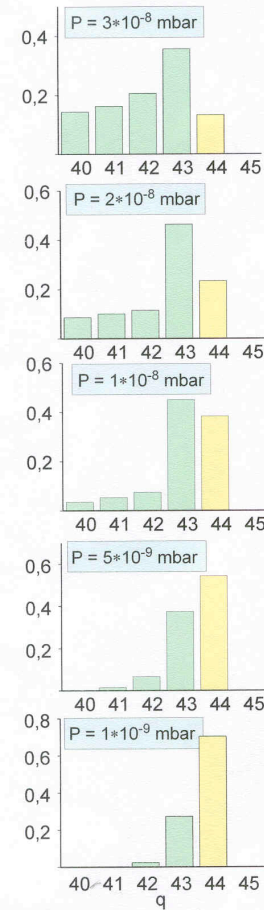
Ar ¹⁶⁺ : w	$2p(^1P_1) - 1s(^1S_0)$	$E = 3139,6(0,25) \text{ eV}$
z	$2s(^3S_1) - 1s(^1S_0)$	
y	$2p_{1/2}(^3P_1) - 1s(^1S_0)$	$E = 3123,6(0,25) \text{ eV}$
x	$2p_{3/2}(^3P_2) - 1s(^1S_0)$	$E = 3126,4(0,4) \text{ eV}$
<small>J.P.Biland et al. Phys.Rev., A28 (1983) 1413</small>		
Ar ¹⁵⁺ : q	$1s2p(^3P)2s(^2P_{3/2}) - 1s^22s(^2S_{1/2})$	
r	$1s2p(^3P)2s(^2P_{1/2}) - 1s^22s(^2S_{1/2})$	
Ar ¹⁴⁺ : beta	$2p(^1P_1) - 1s(^1S_0)$	



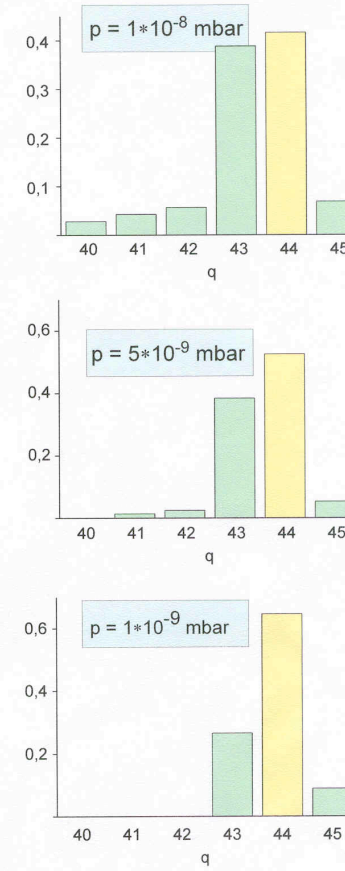
Wavelength X-ray spectroscopy: Xenon



electron energy 8 keV

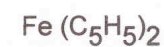
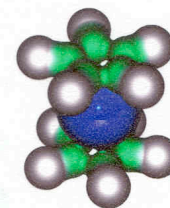
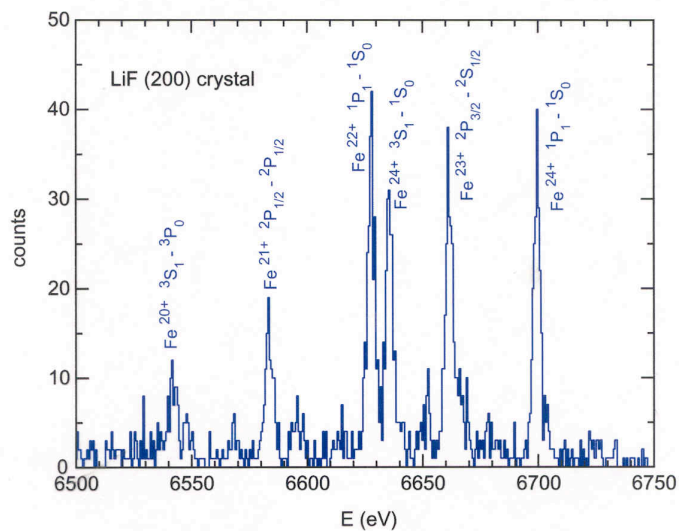
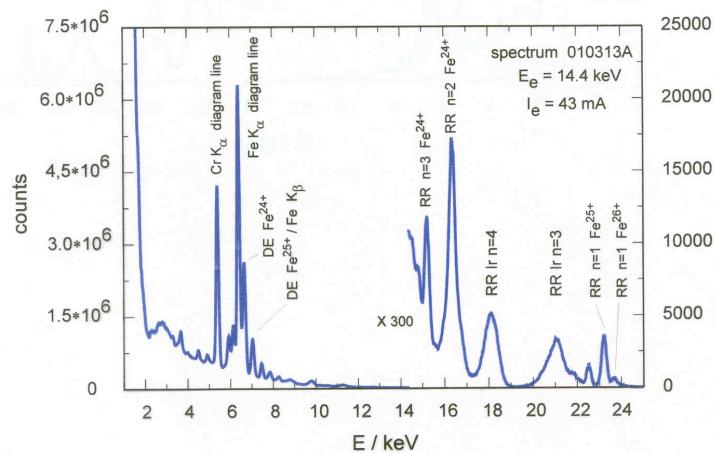


electron energy 13 keV



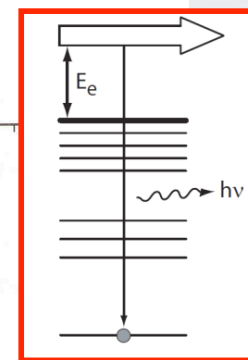
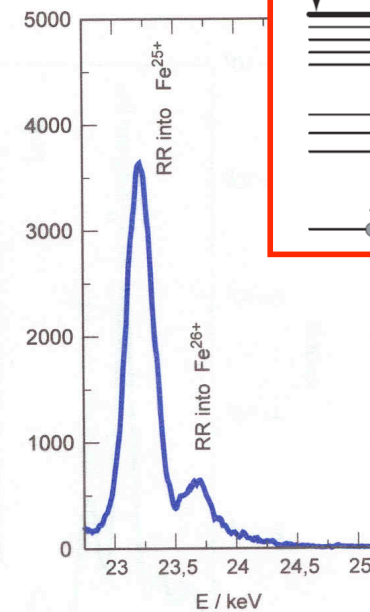
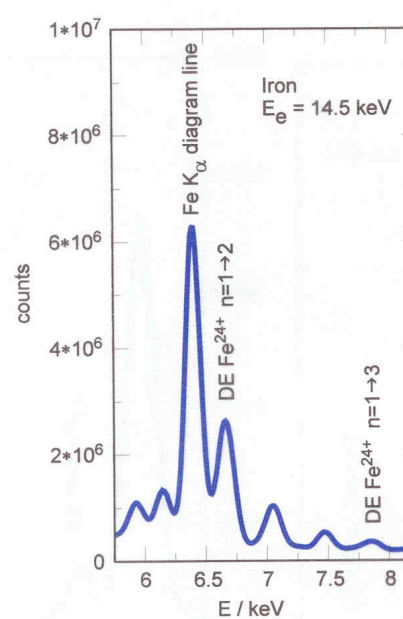
E: 3s - 2p
F: 3d - 2p

Energy and Wavelength X-ray spectroscopy: Iron

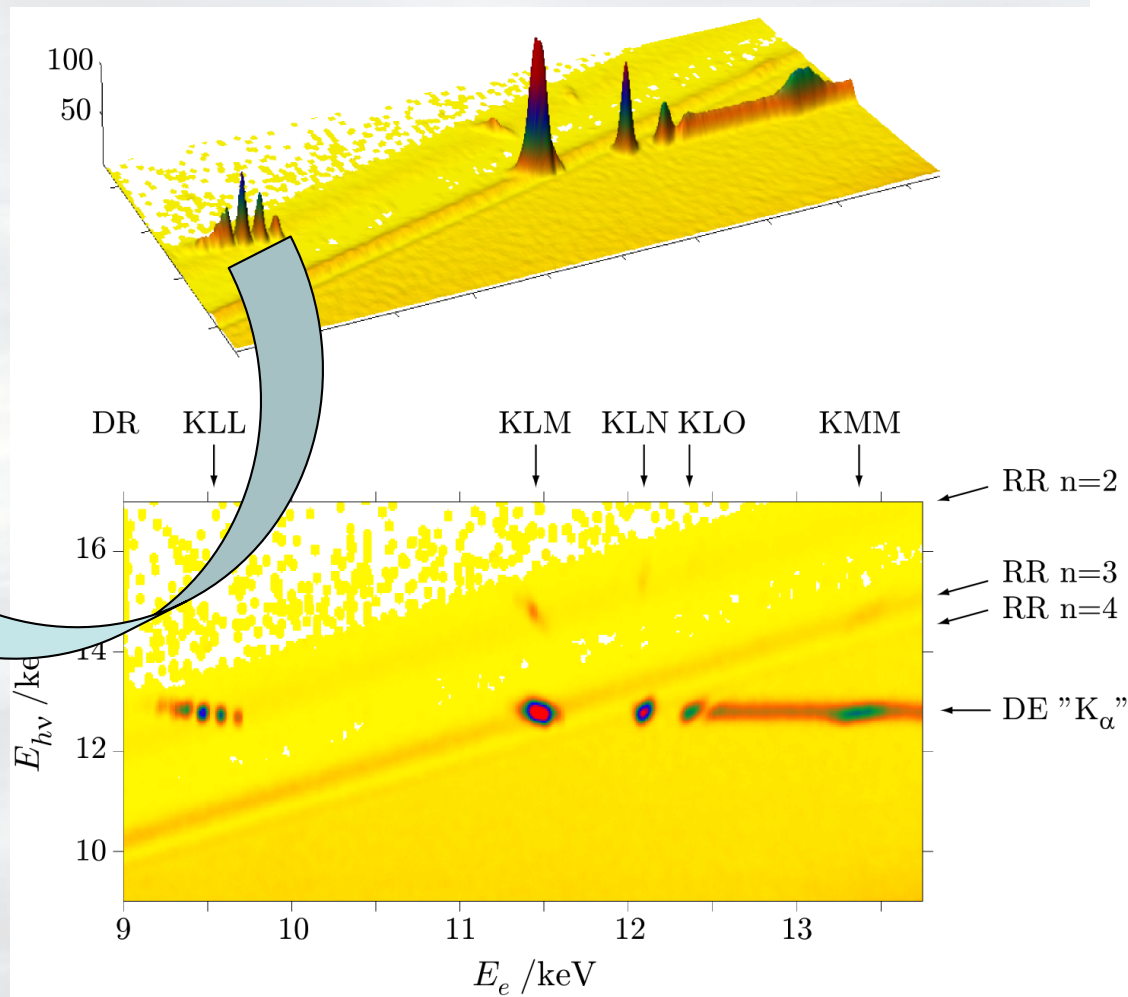
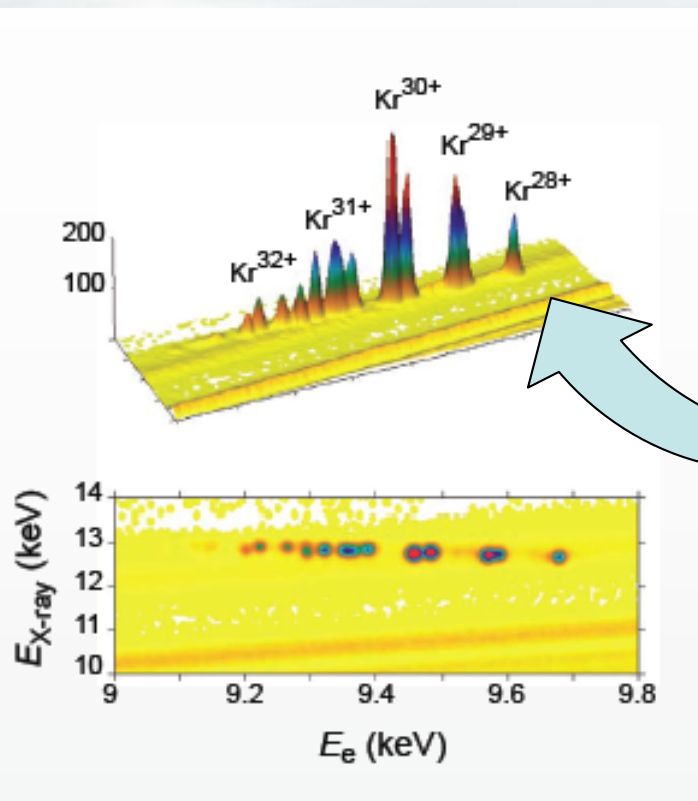


- H
- C
- Fe

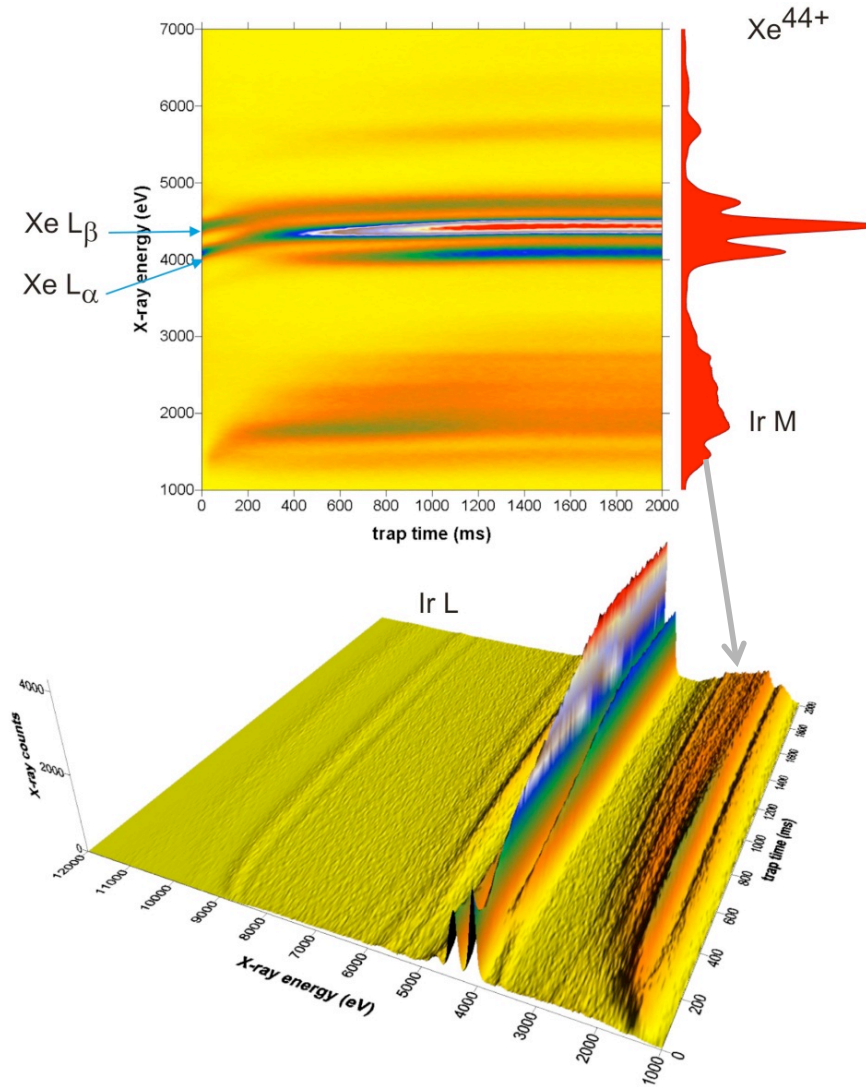
Di (cyclopentadienyl) iron
Ferrocene



X-Ray Spectroscopy: Scatterplot



Time-resolved x-ray spectroscopy

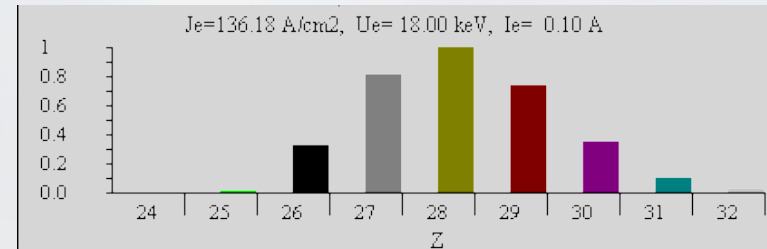


Time-resolved
energy dispersive
x-ray spectroscopy
on xenon ions

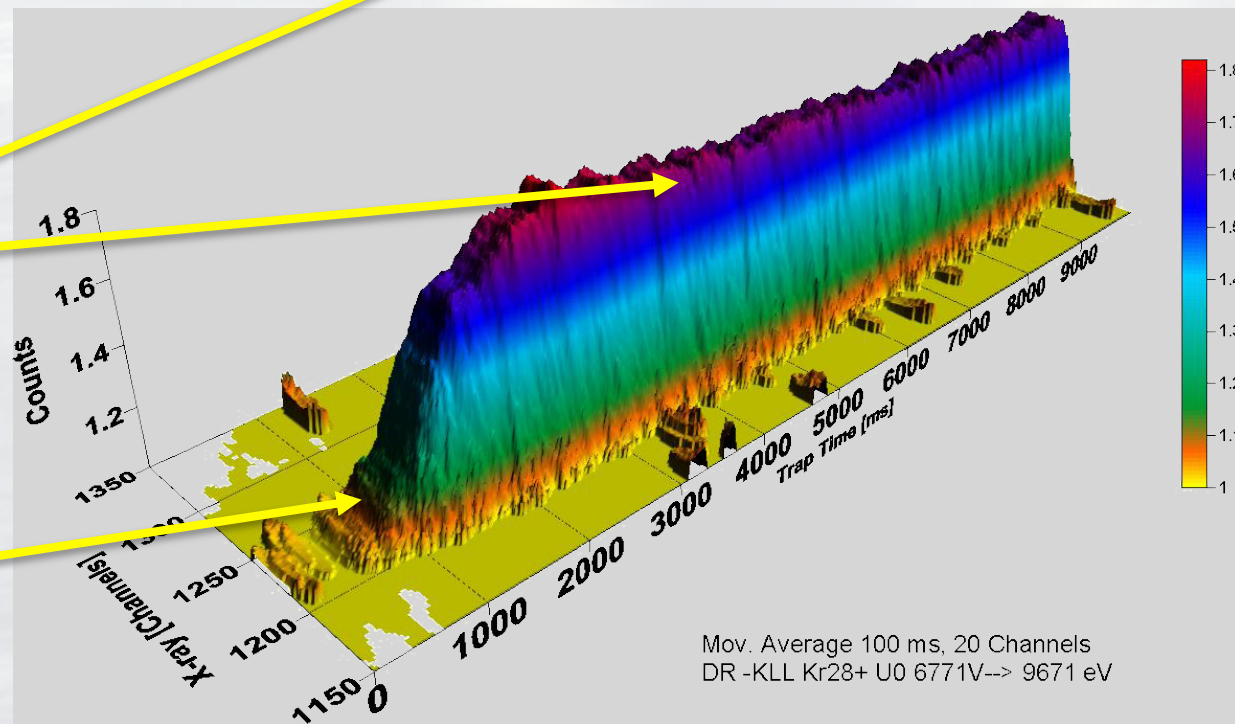
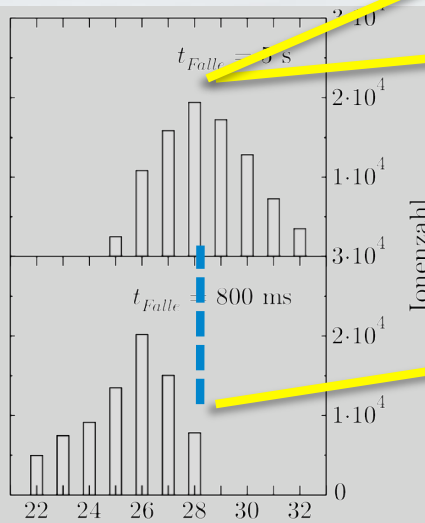
Time-resolved KKL-DE x-ray spectroscopy

KLL Kr^{28+}

(fixed electron beam energy,
but x-rays as a function of the ionization time)



results of ion extraction

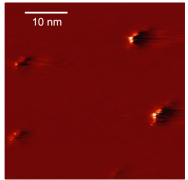


Applications of highly charged ions (examples)

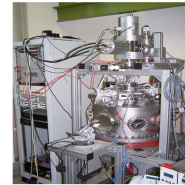


Applications of HCI

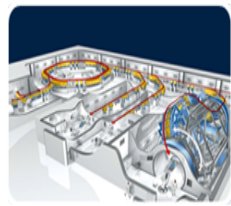
Applications of highly charged ions (examples)



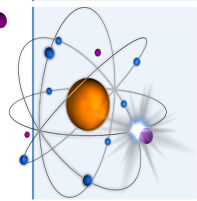
Nanostructuring



FIB, Surface
analysis

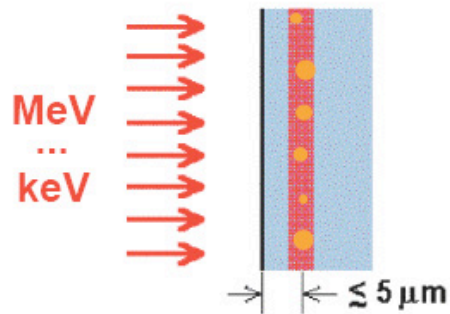


Particle Therapy



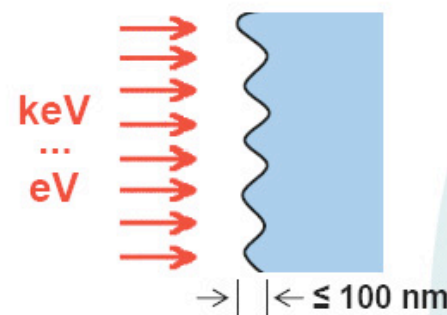
Charge Breeding

Nanostructuring with highly charged ions



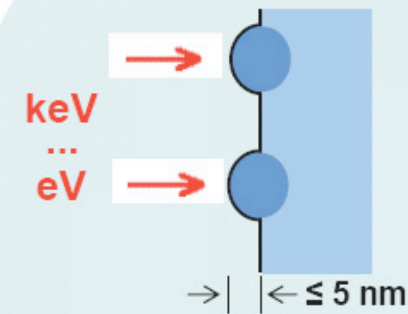
Ion Implantation

Nanocrystals by Implantation and Annealing
Focused Ion Beam Synthesis of Nanostructures
Magnetic Nanostructures



Ion Beam Sputtering

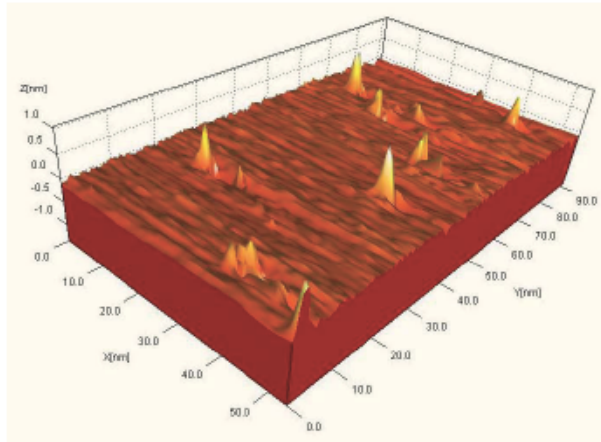
Focused Ion Beam Direct Writing
Self-Organized Periodic Nanopatterns



Highly Charged Ions

Local Modification of Surface Structure by Electronic Excitation
Nanodots/-pits by Single Ion Impact

Surface Modifications Induced by Potential Energy



HOPG, 150 eV Ar⁹⁺, E_{pot} = 1000 eV

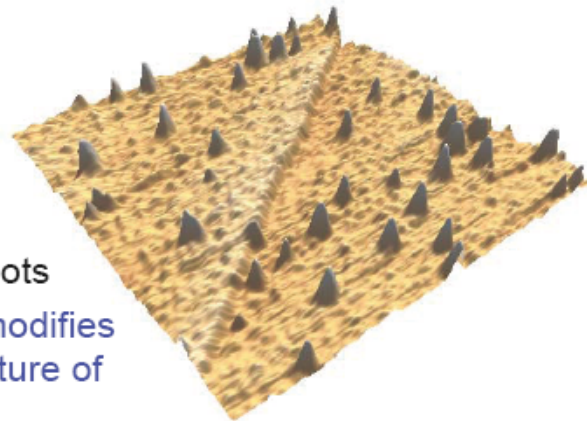
HOPG

- Conductor
- AFM: flat surface
- STM: nanodots

→ HCI impact modifies *electronic* structure of surface

New properties:

- morphologic
- electric
- optic



CaF₂

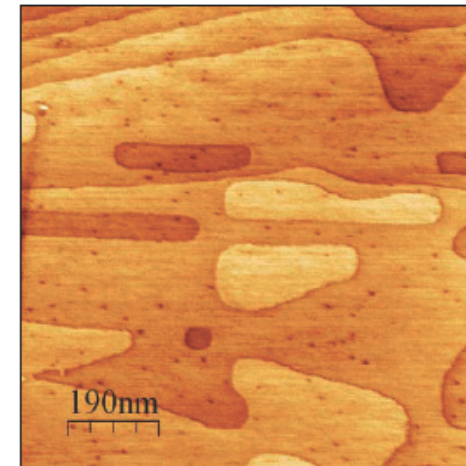
- Insulator
- AFM: nanodots

→ HCI impact modifies *crystal* structure of surface

CaF₂, 5.4 keV Xe³⁶⁺, E_{pot} = 27.8 keV

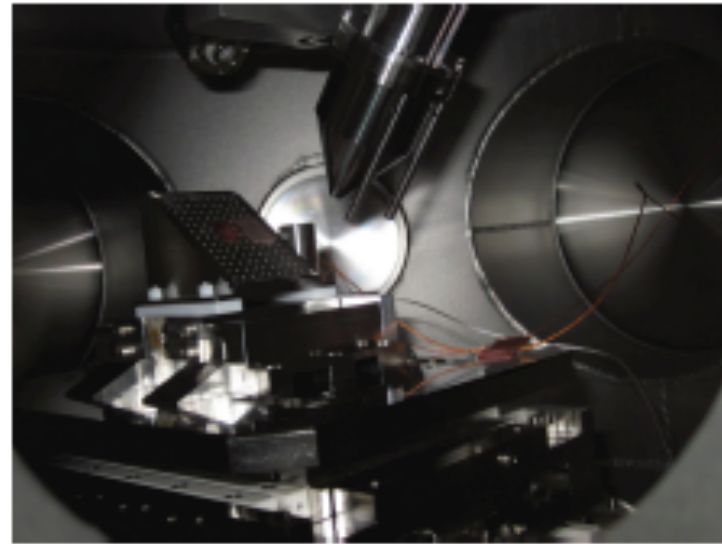
KBr

- Insulator
 - Nanopits
- HCI impact induces desorption



KBr, Xe³⁴⁺, 24 keV, E_{pot} = 20 keV

FIB
(Focused Ion Beam)

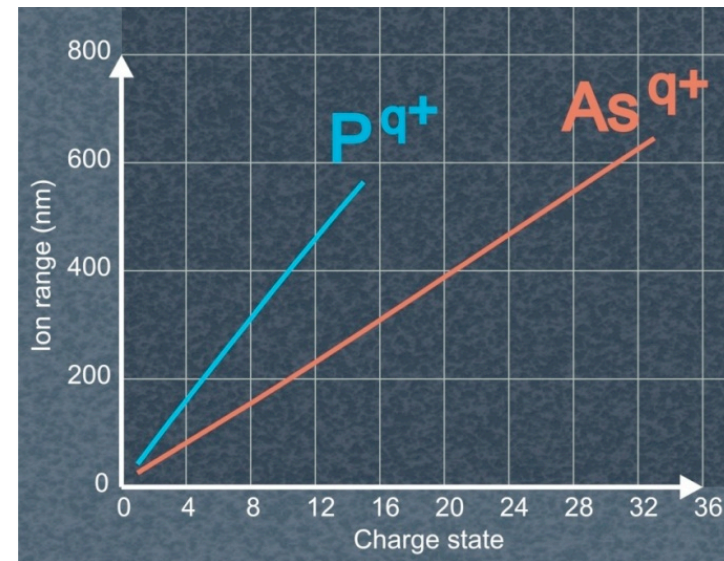


A new class of FIB

Feature	New Advantage
Projectiles	Almost all elements of the periodic table, in particular noble gases
Charge State	Free choice of projectile charge state
Sputter Yield	Variable, according to the kinetic and potential energy
Implantation	Variable implantation depth, according to the kinetic projectile energy

Implantation in Si:

Realization of different implantation depths due to different ion charge states at a fixed ion acceleration potential



A new class of FIB

By using DREEBIT ion sources

Production of ion beams with different ion charge states with diameters in the micrometre up to nanometre region.

Applications

Lithography

Nano Engineering

Photonic Structures

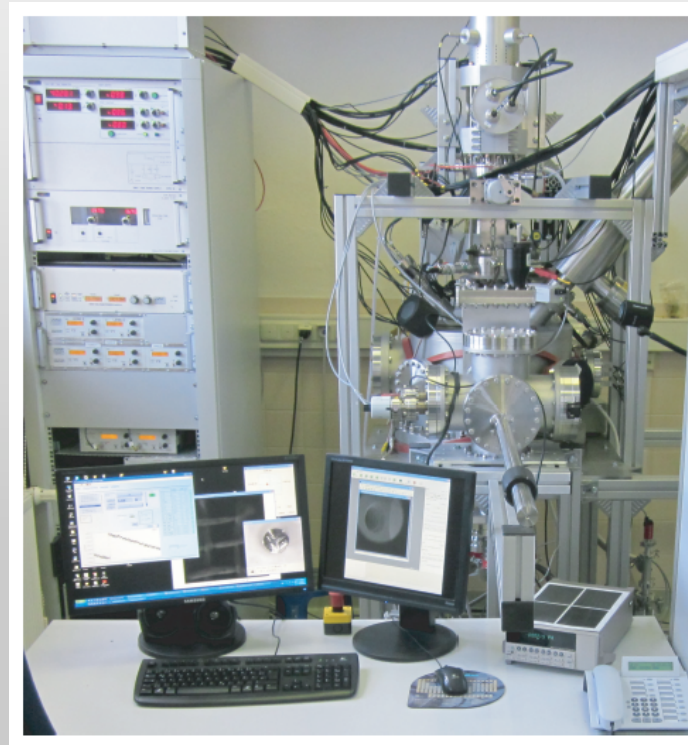
Materials Characterization

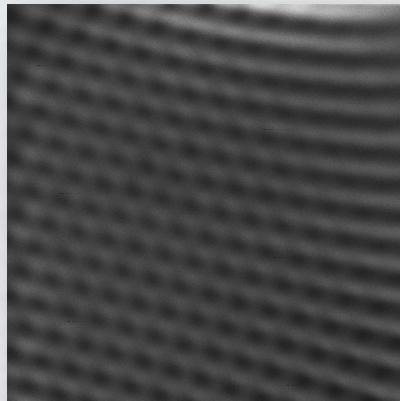
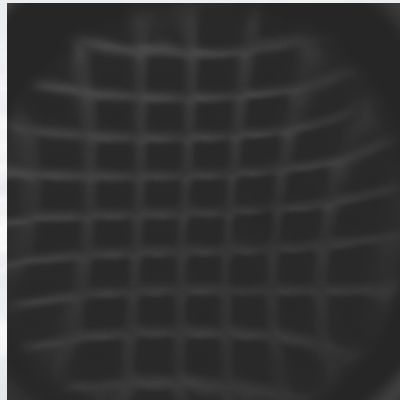
Micro-Machining

Quantum Dots

Radiation Biology

Surface Analytics



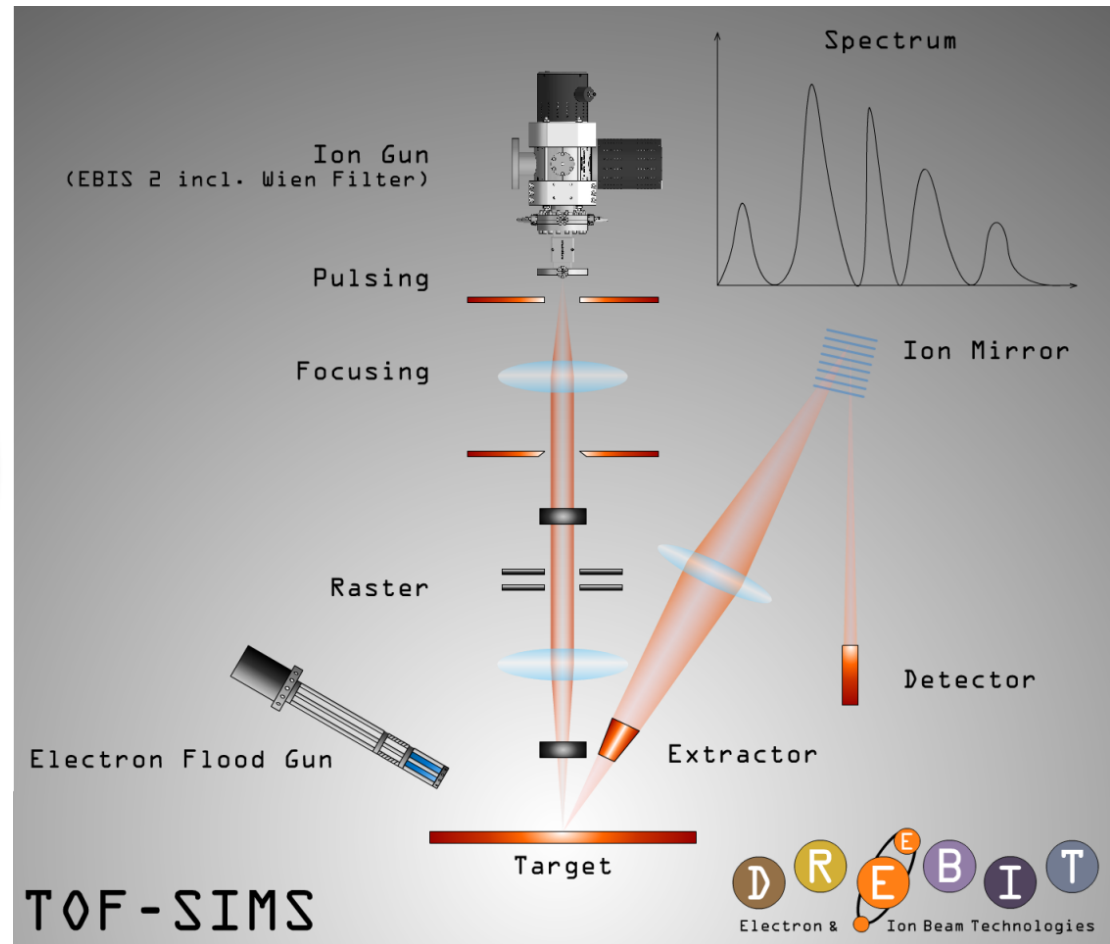


lattice width 2 μm

Worldwide first SEM-figures
produced with a Xe ion beam!

Time-of-Flight Secondary Ion mass Spectrometry

TOF-
SIMS



Time of Flight Secondary Ion Mass Spectroscopy

Anwendungen

Semiconductor industry

Surface analysis

"Soft matter" applications (bio materials, polymers, ...)

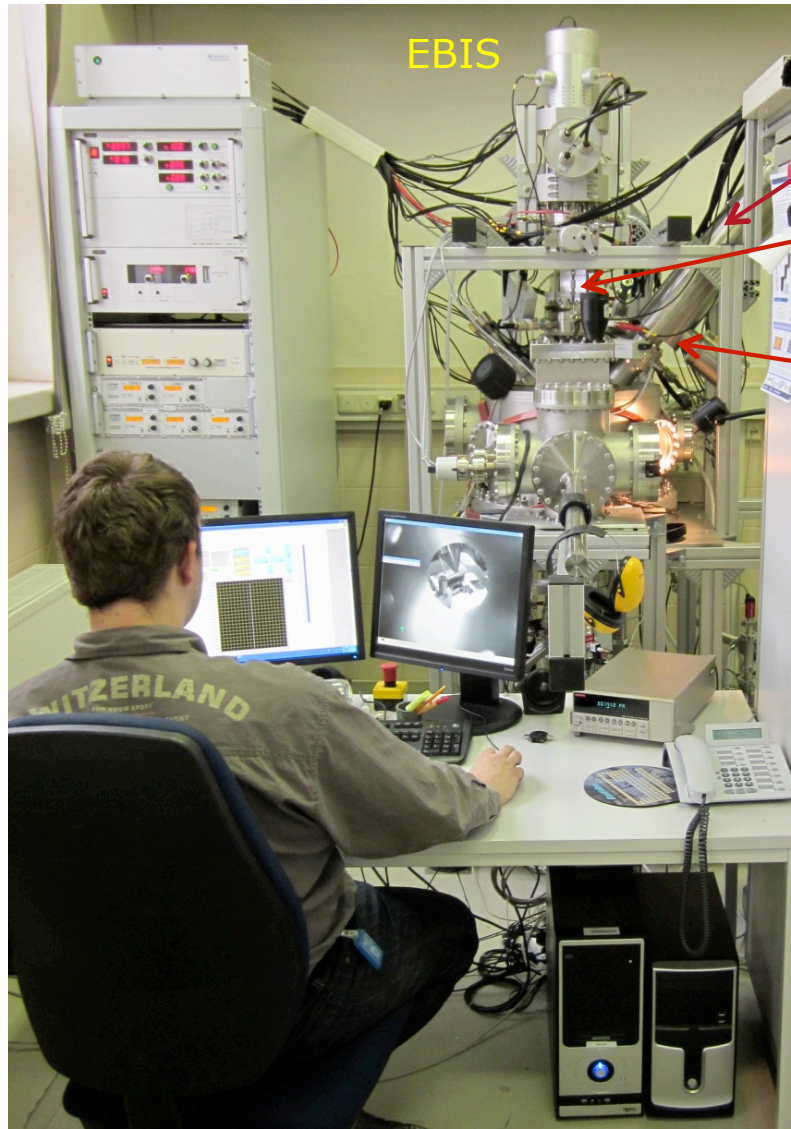
Materials science

Basic research

Classical industry (glass, paper, metal, ceramics, ...)

Analysis of contaminations, adhesion, friction, corrosion, diffusion, cell chemistry, bio compatibility

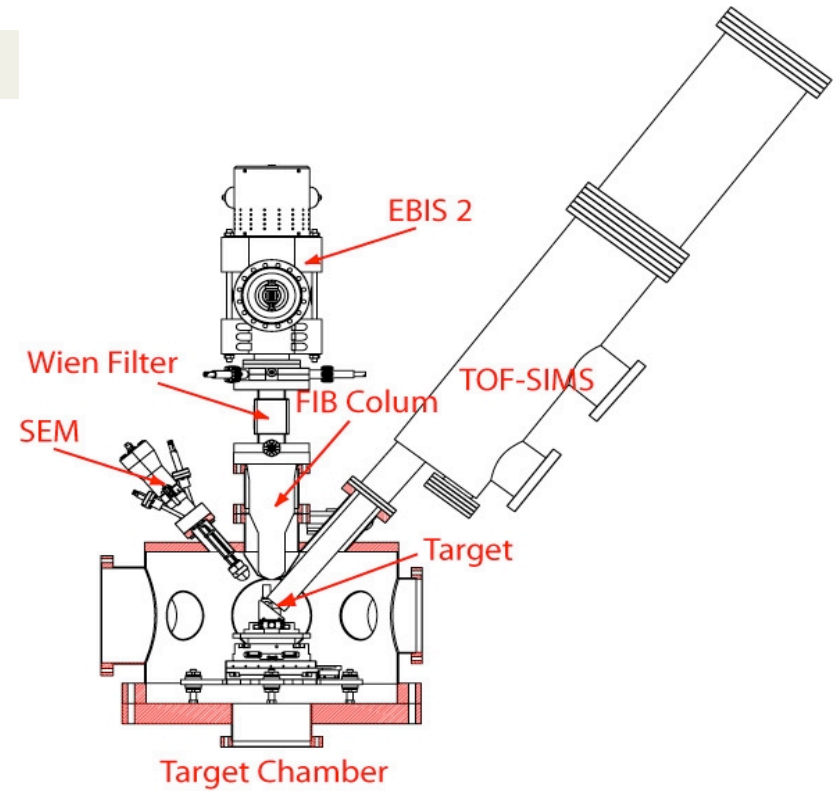
Applications of HCI



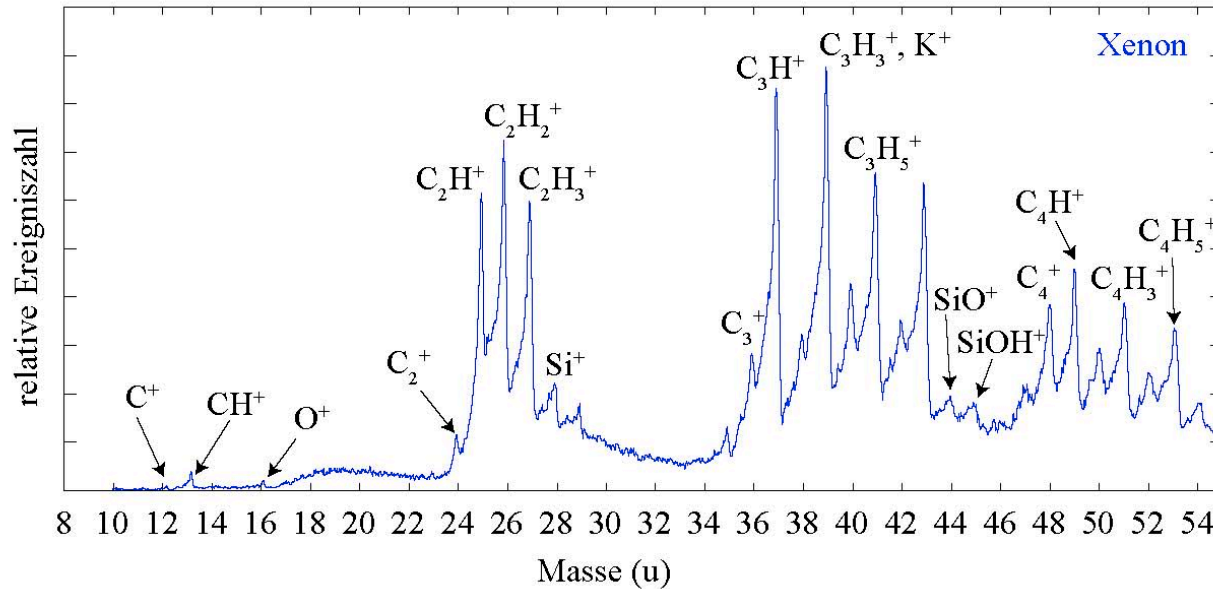
TOF-SIMS

Wien filter

FIB



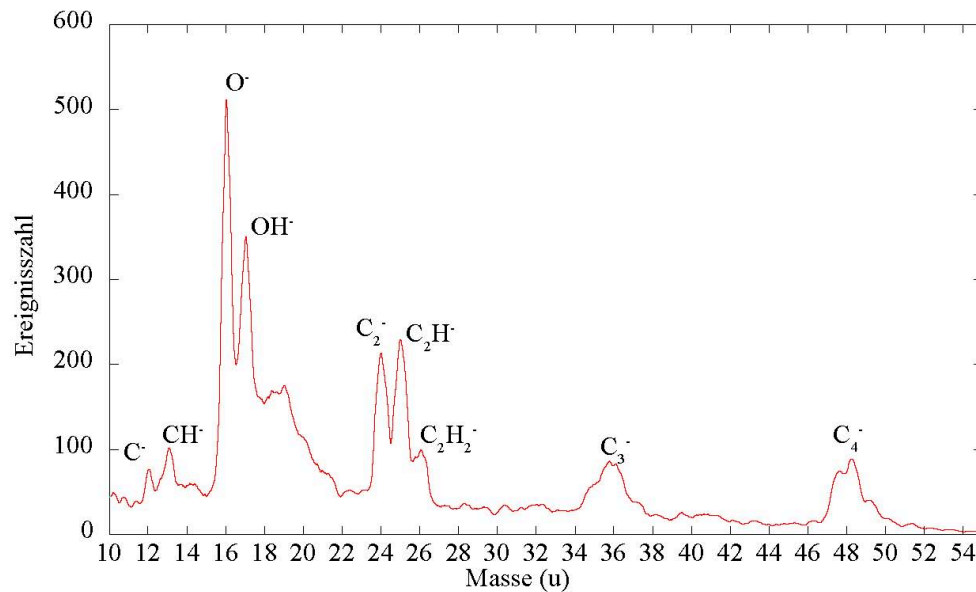
Applications of HCI TOF-SIMS



Non-cleaned Si-surface

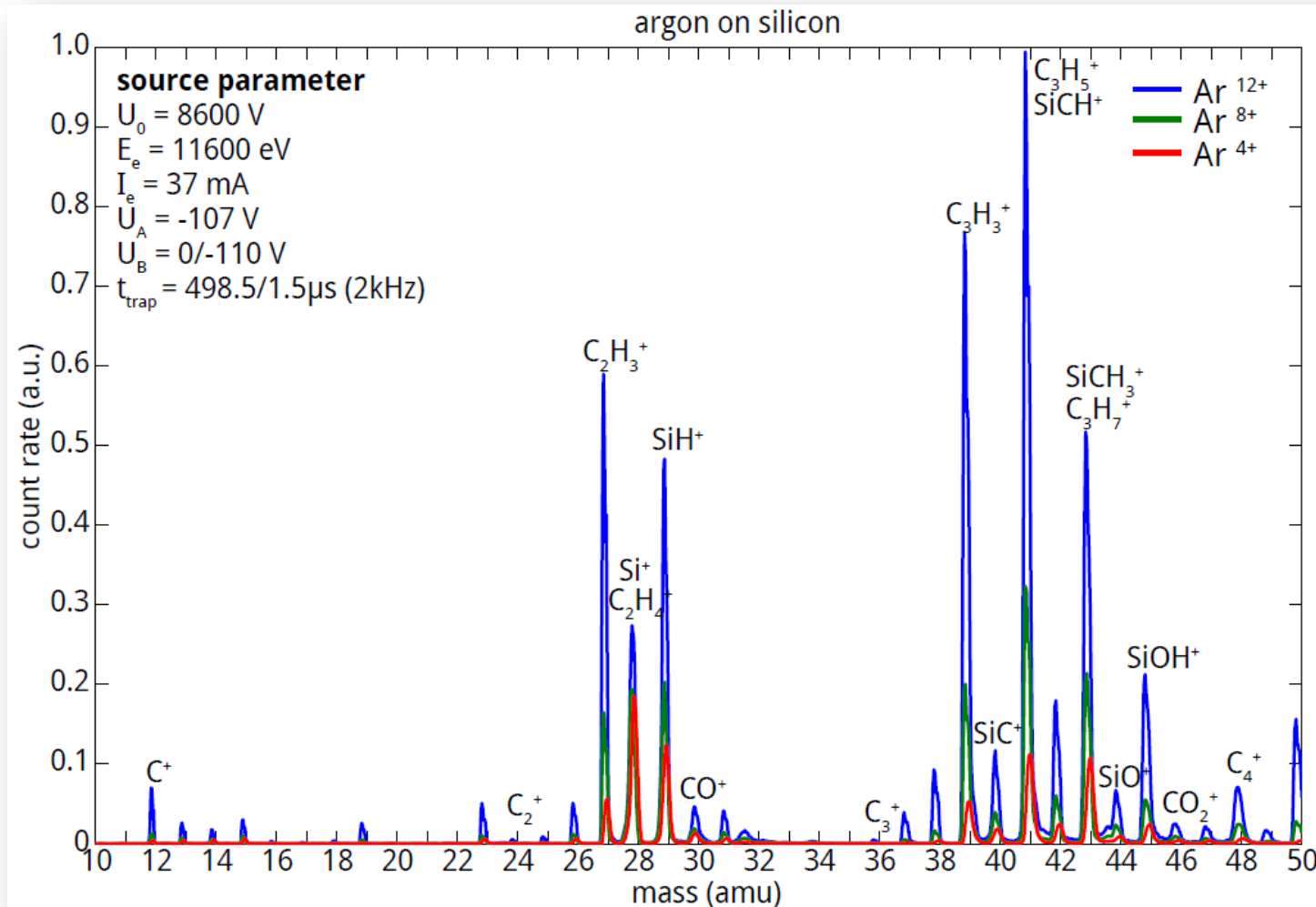
Excitation with xenon ions

Positive ions

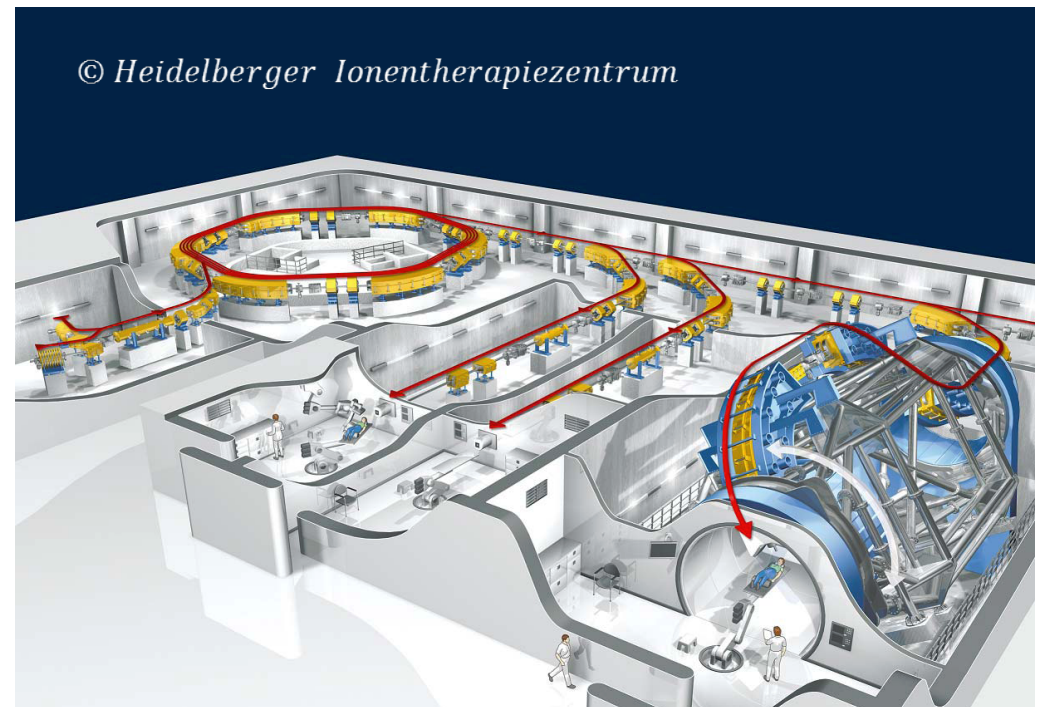


negative ions

Applications of HCI: TOF-SIMS



Hadron Therapy



Cancer - a Worldwide Problem



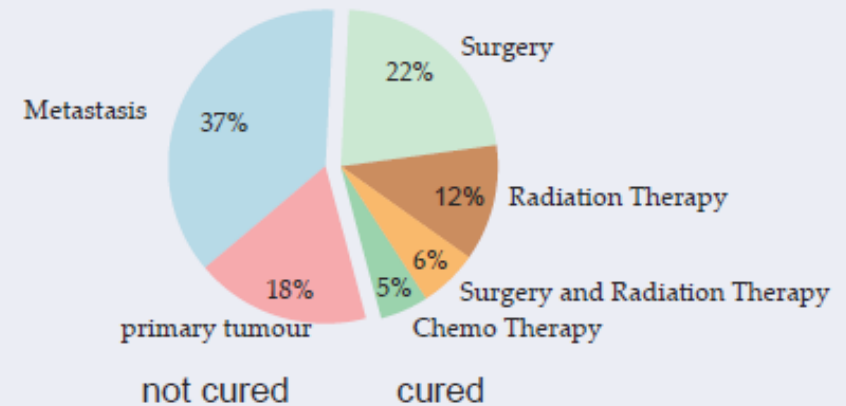
- **Cancer is the the second most common cause of death and about 33% of all inhabitants of the EU will confront some kind of cancer in their life**
- **About 45% of cancer patients can be treated, mainly by surgery and / or radiation therapy [S.Peggs, PAC07, June 25'07]**
- **Hadron therapy with protons and carbon ions is - taken its success rate - the second most successful technique in cancer treatment, outmatched only by surgery**
- **Until 2005 about 40.000 patients worldwide were treated by particle therapy at 22 PT centers (Europe, USA, Japan, China, South Africa). The number of treated patients is constantly increasing.**

Advantages of Therapy with Ion Beams

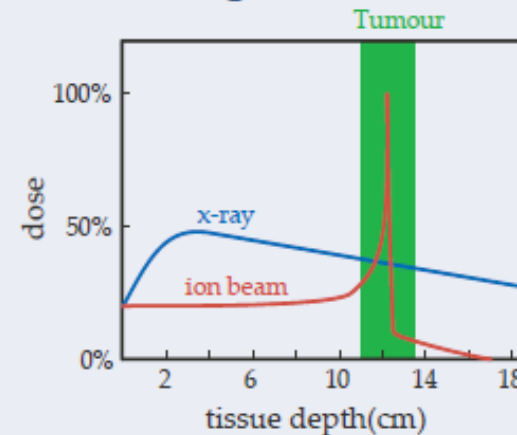
It is possible to focus carbon ions with great precision directly onto the tumor. Therefore, only the tumor is damaged irreversibly but the healthy tissue remains intact.

Another advantage is the high biological efficiency of carbon ions, causing more damage in the tumor cells than other kinds of irradiation.

Cancer Therapy

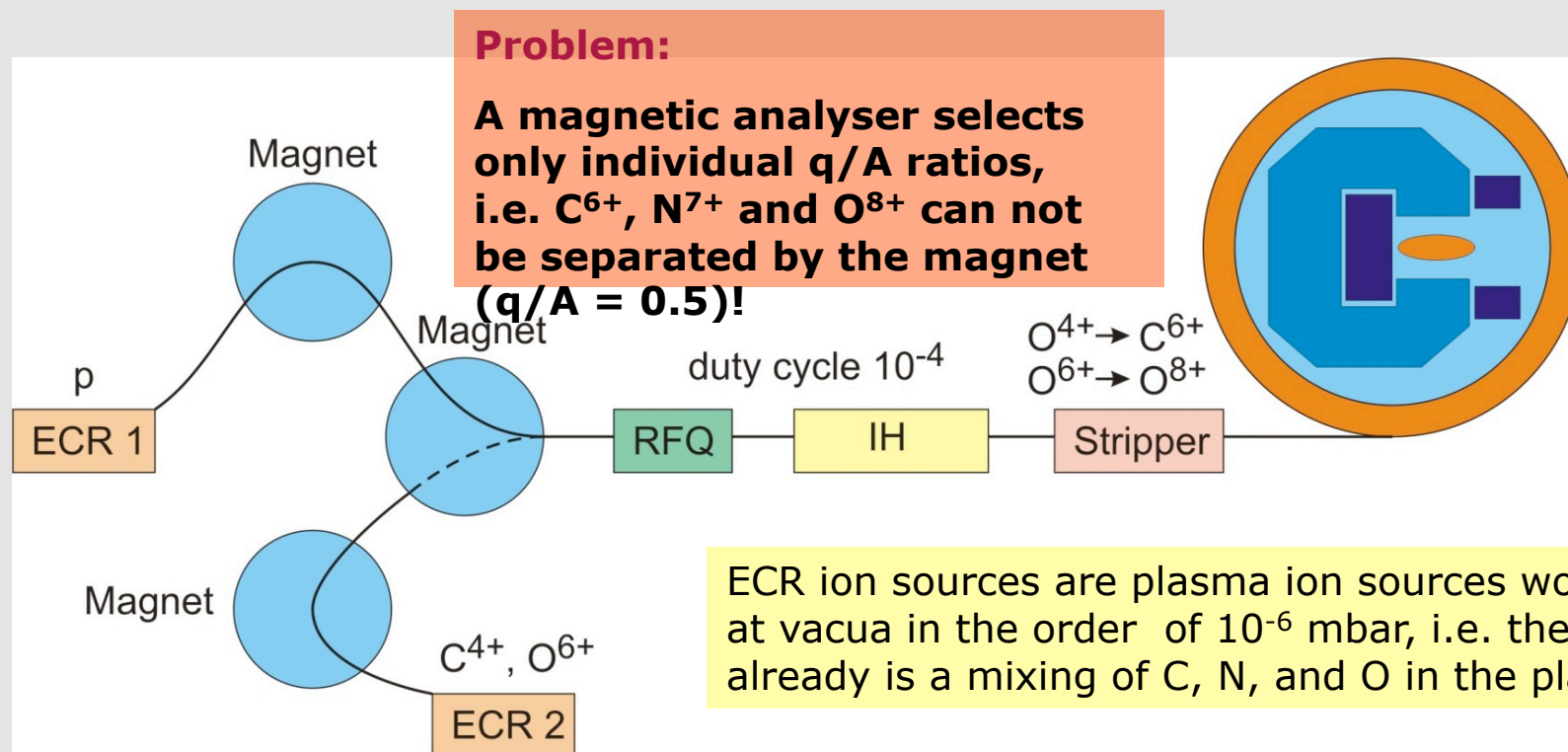


Advantages of Ion Therapy



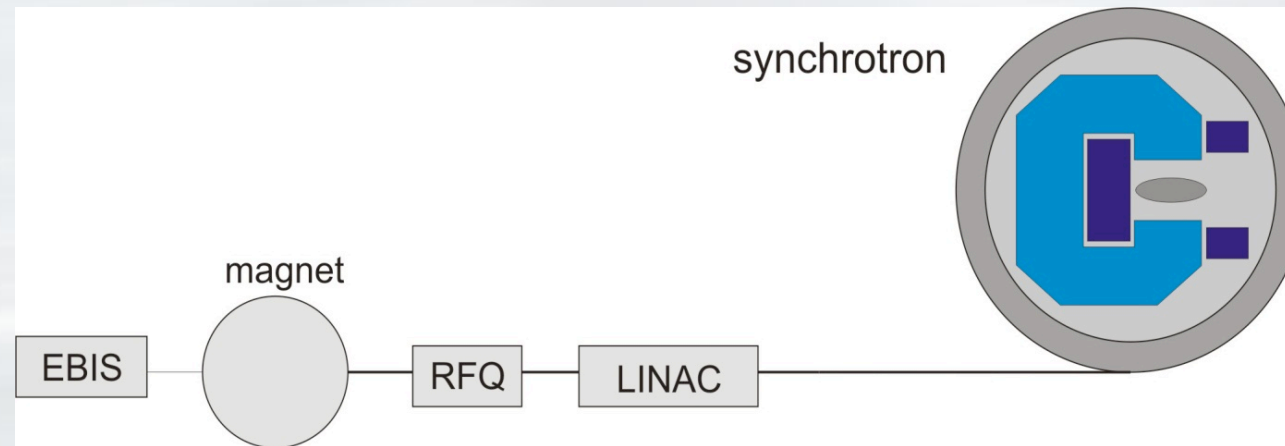
Basic Structure of a Synchrotron-based Irradiation Facility

Heidelberg Hadron Therapy Facility HIT:



R.Becker, ICIS-05 PA9/RSI MS # C05005

Simplification of Therapy Facilities by using a New Kind of Ion Source



Advantages:

- only one ion source
- one separation magnet
- shorter LINAC
- no stripper
- lower injection energy
- single-turn injection (at 4 MeV/u)
- smaller synchrotron magnets
- lower power consumption

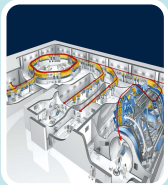
the complexity of the irradiation facility decreases,
the beam quality is improved,
costs can be reduced

only



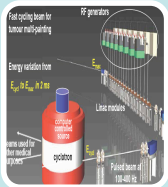
Cyclotrons

- IBA (Belgium)
- SIEMENS (Germany)
- HITACHI (Japan)
- MITSUBISHI (Japan) a.o.



Synchrotrons

- SIEMENS (Germany)
- HITACHI (Japan)
- MITSUBISHI (Japan) u.a.



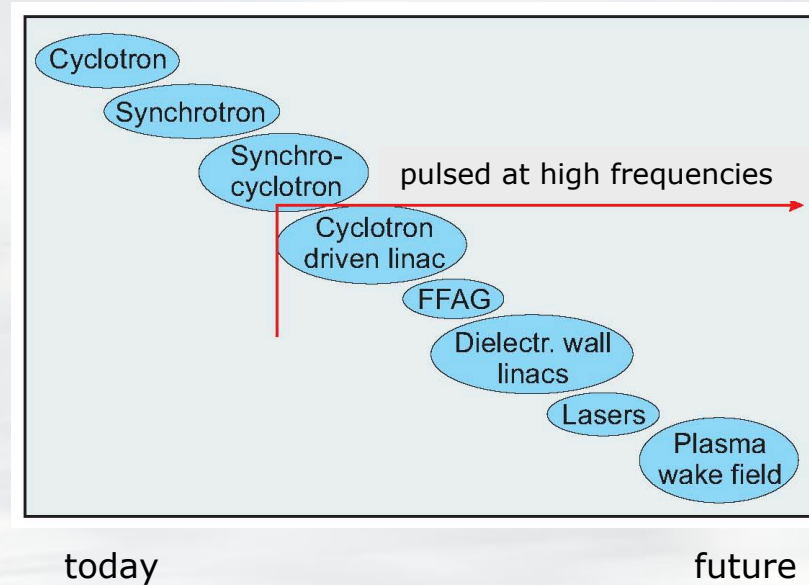
CYCLINACs

- ADAM (Switzerland; CERN)

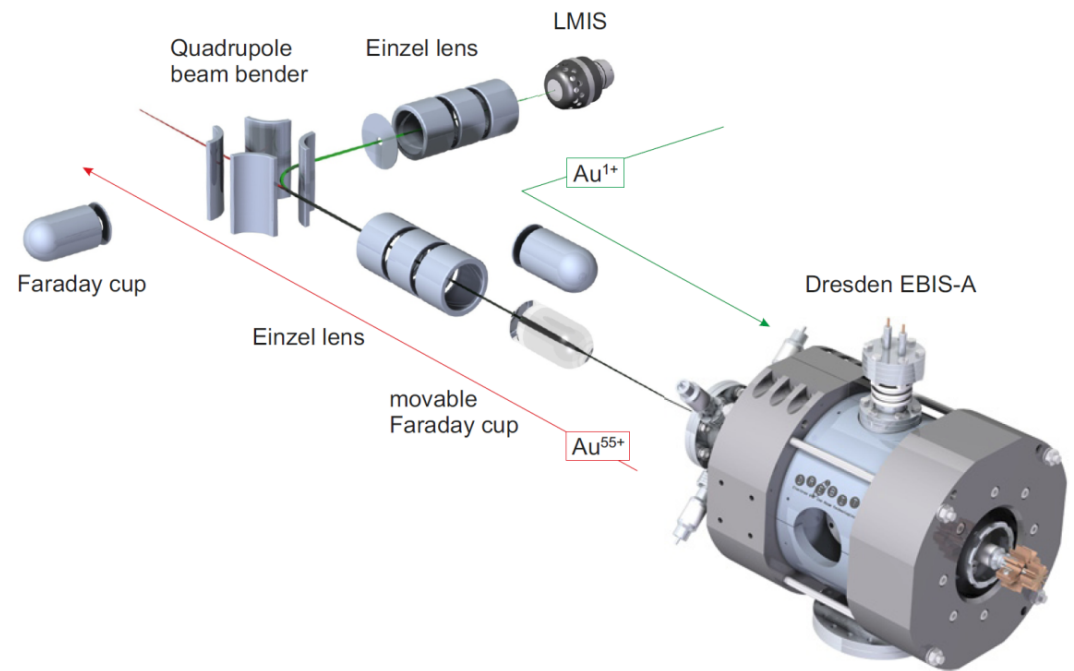


DDA, DWA

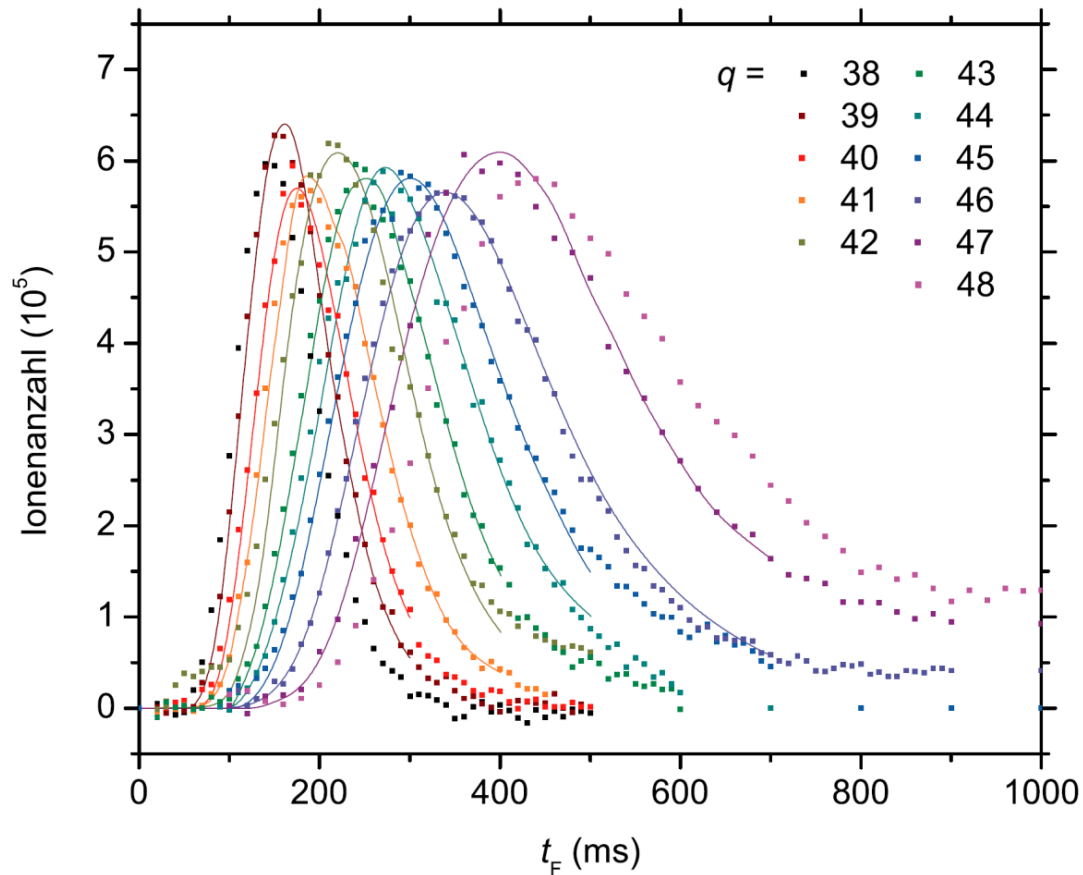
- SIEMENS (Germany)
- some instituts (USA. Japan)



Charge Breeding



Charge Breeding



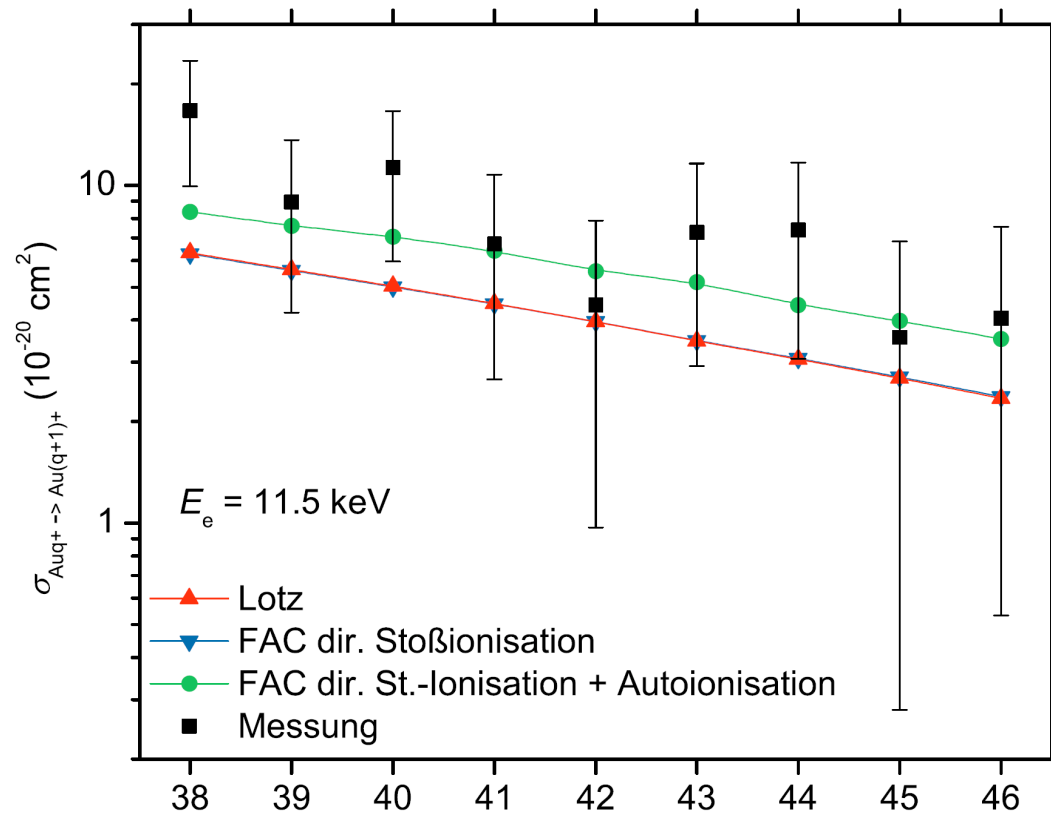
q/A analysis

→ Evolution of the ion charge states Au^{38+} to Au^{48+}

Description:

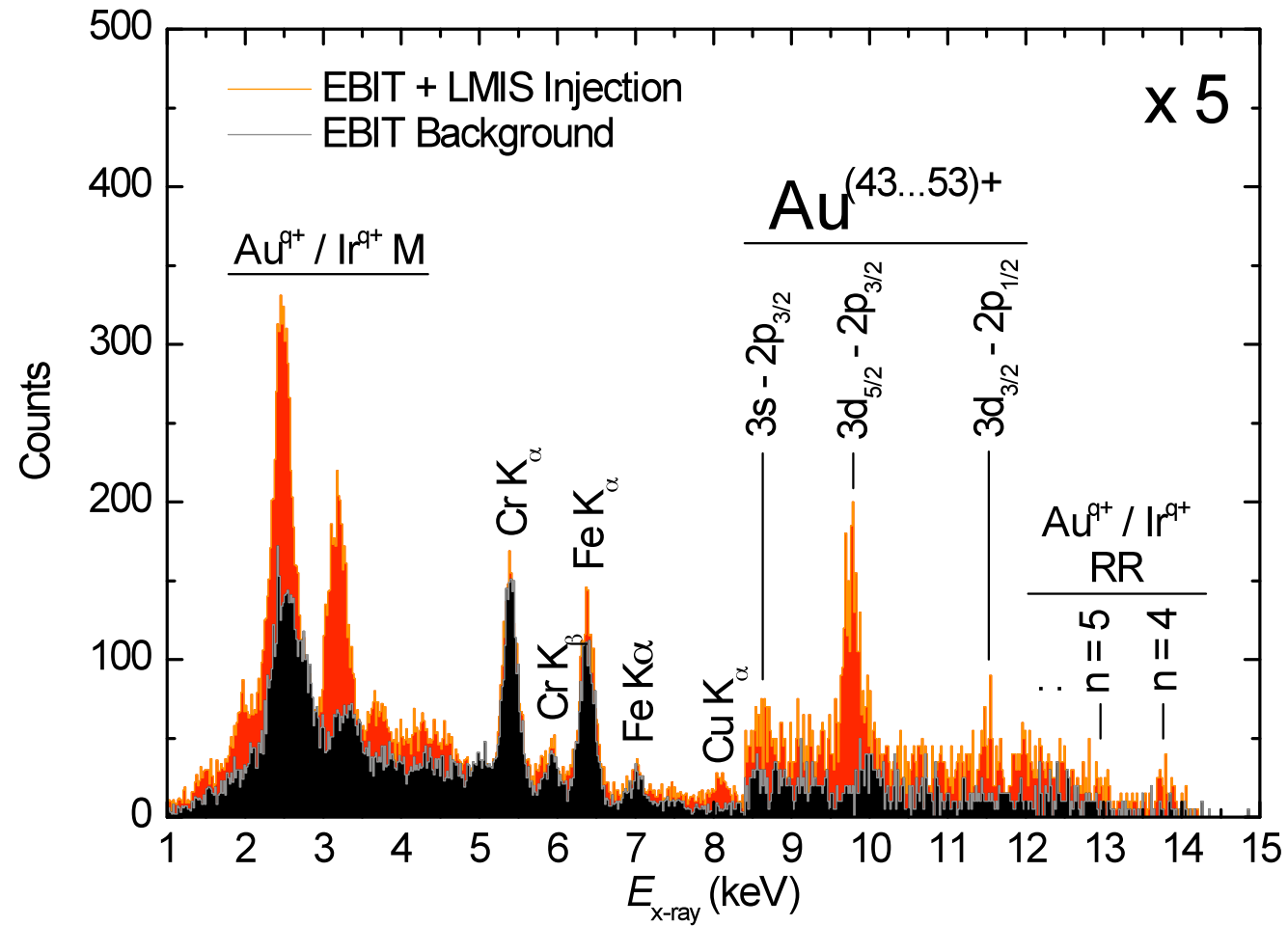
$$dN_{q+}/dt = \lambda_{q-1} \cdot N_{q-1} - \lambda_q \cdot N_q + \lambda_{q+1} \cdot N_{q+1}$$

Charge Breeding



Electron impact ionisation cross-sections for charge bred gold ions

Charge Breeding: Gold



Thank you ... and thanks to the team!



Dr. G. Zschornack



R. Mertzig



U. Kentsch



E. Ritter



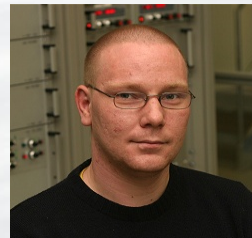
<http://www.tu-dresden.de>



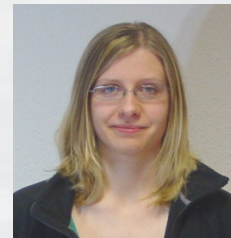
Dr. V. P. Ovsyannikov



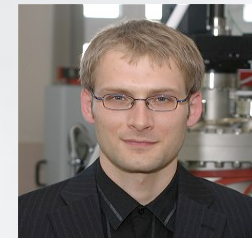
Dr. F. Grossmann



Dr. R. Heller



Dr. A. Thorn



M. Schmidt



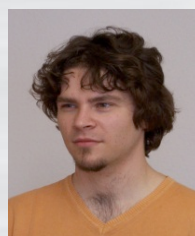
M. Hartig



Dr. F. Ullmann



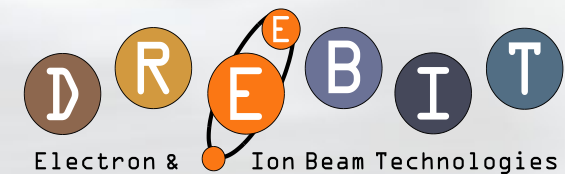
A. Schwan



J. König



M. Kreller



<http://www.drebit.com>

Founding and cooperation



10



Europa fördert Sachsen.

