

BEAMS OF REFRACTORY ELEMENTS

Laser break-up molecules and ionization

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Refractory elements

- melting points ~ 2000 K

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	1A	2A	3B	4B	5B	6B	7B	8B			1B	2B	3A	4A	5A	6A	7A	8A	
Period																			
1	1 H Hydrogen																		2 He
2	3 Li	4 Be											5 B	6 C	7 N	8 O	9 F	10 Ne	
3	11 Na	12 Mg											13 Al	14 Si	15 P	16 S	17 Cl	18 Ar	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6	55 Cs	56 Ba	* 71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
7	87 Fr	88 Ra	** 103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg								
* Lanthanides			* 57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb			
** Actinides			** 89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No			

Ion source:

+	Surface	-
hot	Plasma	cool
	Laser	

Molecular beams

□ ZrF_3^+ @ ORSAY

by isocele2 Z. Phys. A 309, 185 (1982)

SrF_2 powder + CF_4 gas
 ^3He beam



ZrF_3^+ was extracted by Plasma ion source

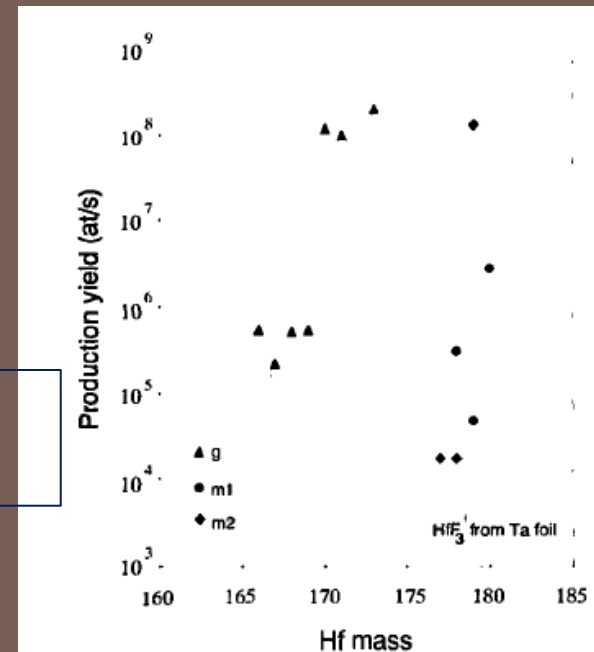
□ HfF_3^+ @ isolde

NIMB 62(1992)535

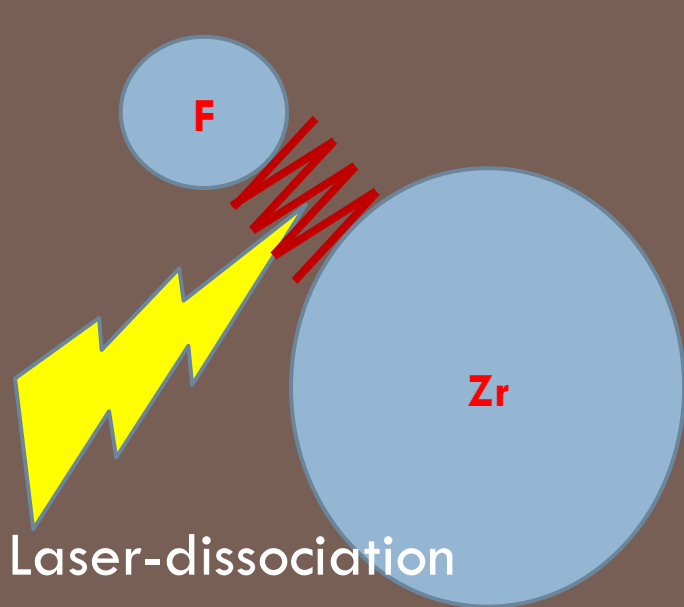
Ta foil + CF_4 gas
p beam



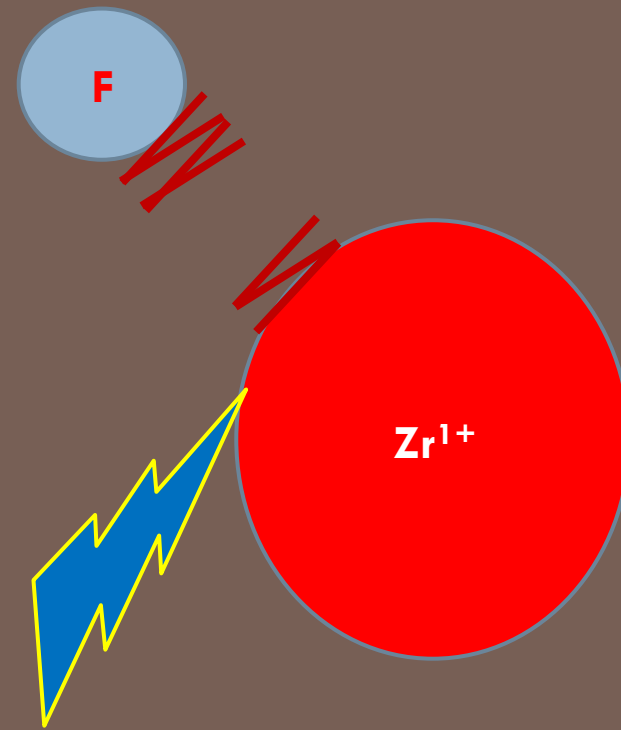
HfF_3^+ was extracted by
Plasma ion source



Laser-induced breakup + ionization



Laser-dissociation



Resonant-ionization

Feasibility study

1. Producing ZrF^+

S. Soorkia et al.,
J. Phys. Chem. A, **2011**, *115* (34), pp 9620–9632,

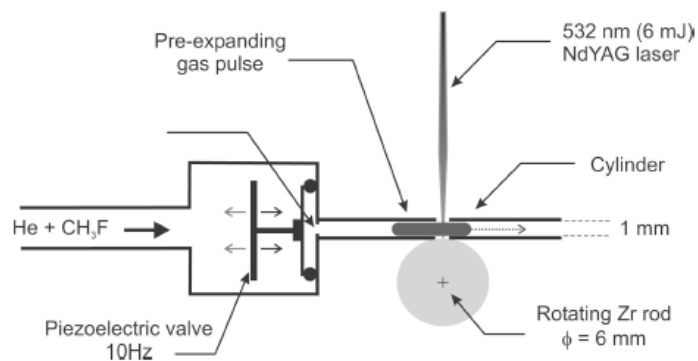


Figure 1. Outline of the laser ablation - molecular beam experimental setup used to obtain cold ZrF radicals.

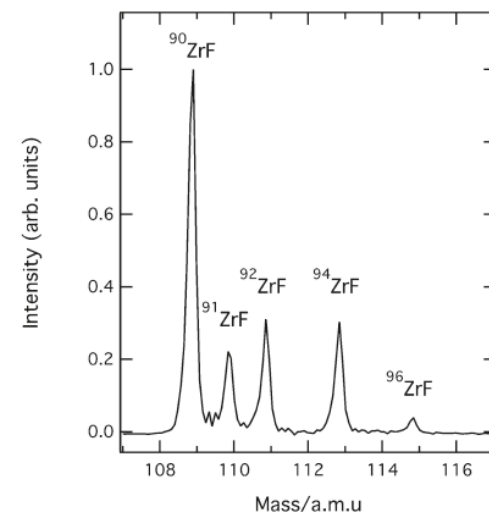
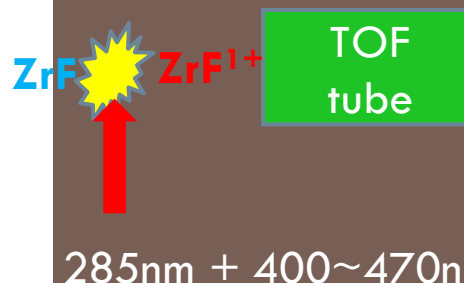


Figure 2. Mass spectrum of the zirconium fluoride radical. The relative intensities of the five naturally occurring isotopologues, i.e., ^{90}ZrF , ^{91}ZrF , ^{92}ZrF , ^{94}ZrF , and ^{96}ZrF , are in the ratio of the isotopic abundances of Zr.

2. Laser radiation of 2 steps: breaking up molecules, ionizing Zr

TOF spectrum
w/He + CH_3F gas (2%)

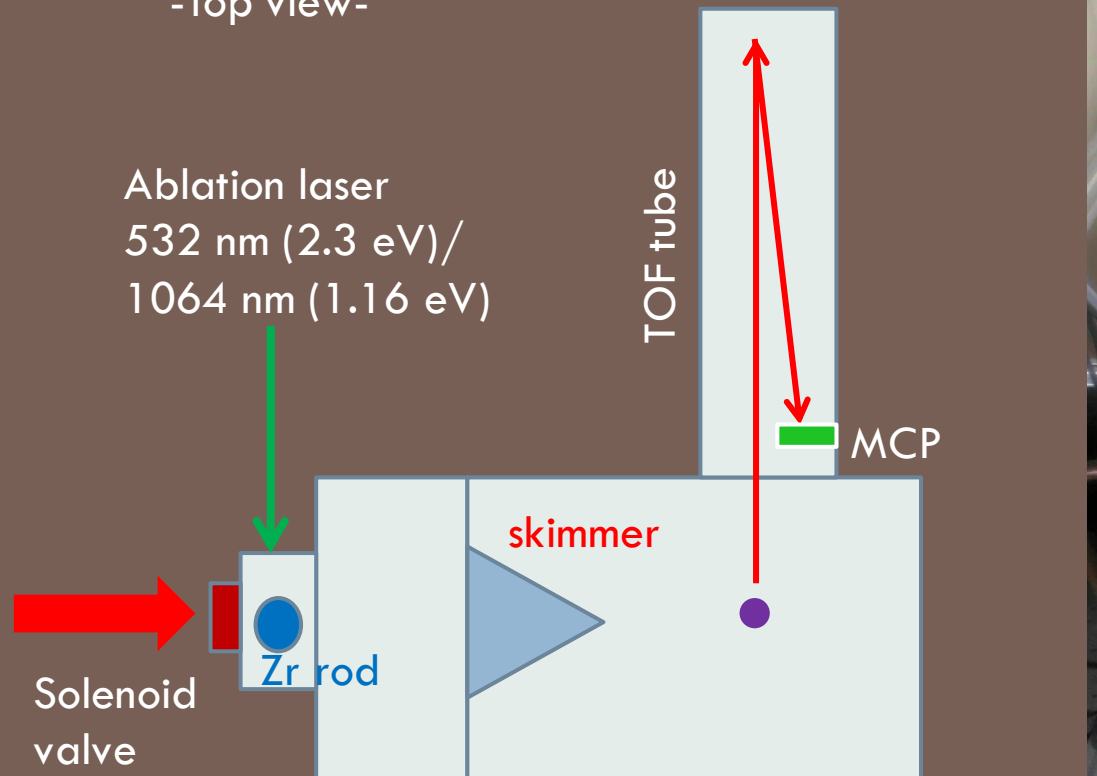
Basic parameters

- Ionization of Zr : $\text{Zr} \rightarrow \text{Zr}^+ + 1\text{e}^-$
Ionization energy Zr(I)
 $= 6.63390 \text{ eV} \sim 2 \times 355\text{nm} (3.5 \text{ eV})$
cross section $\sim 4 \times 10^{-17} \text{ cm}^2 (\text{Fe})$

- Bonding energy of ZrF : $\text{ZrF} \rightarrow \text{Zr} + \text{F}$
 $= 6.3 \text{ eV} (!?) \sim 2 \times 355\text{nm} (3.5 \text{ eV})$
cross section $\sim 5 \times 10^{-19} \text{ cm}^2 (\text{SF}_6) !?$

Test with 355nm@ LARIS lab.

-Top view-



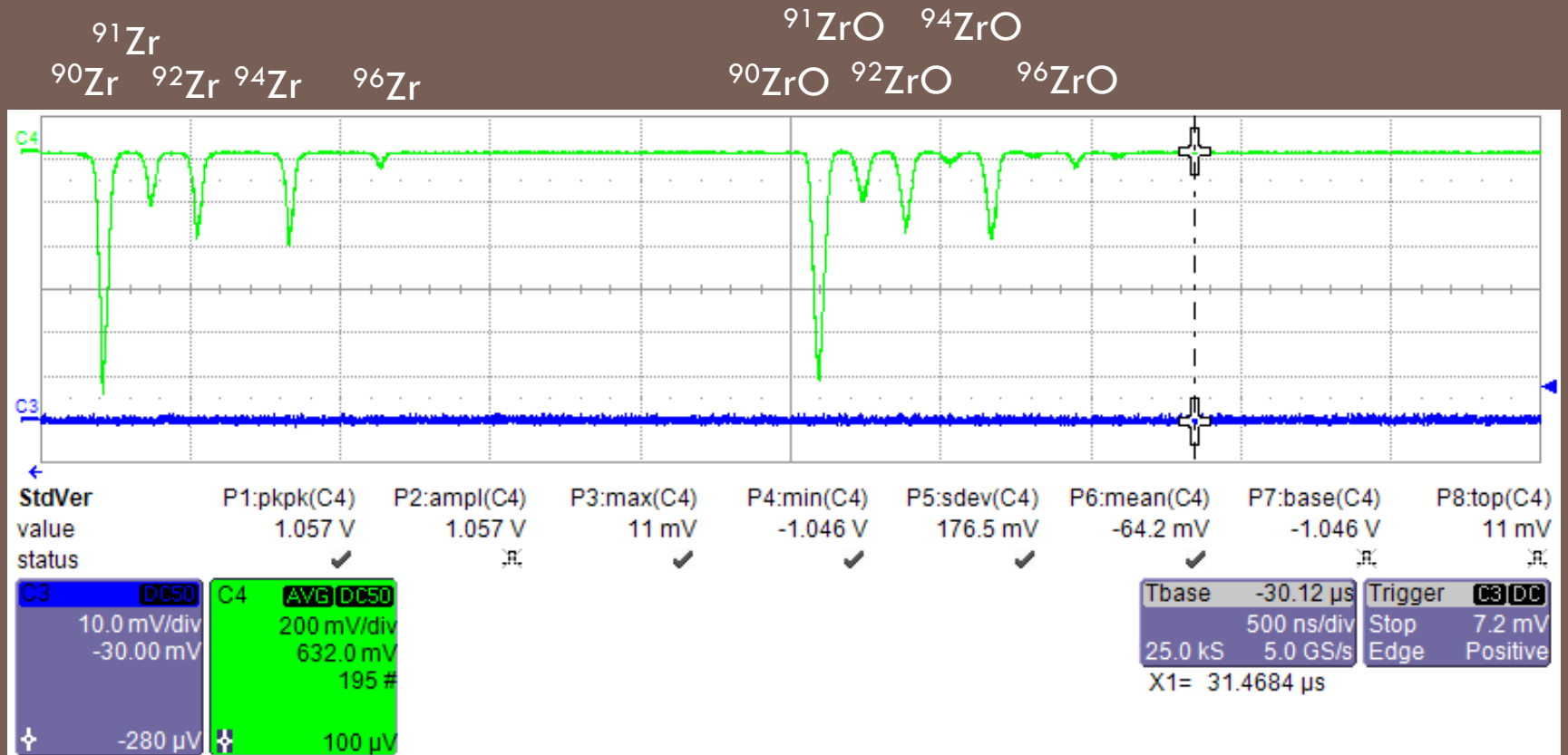
Pulsed gas flow
Ar
+
CF₄ gas

UV laser: 355 nm (3.5 eV)
of Nd/Yag



Valve and lasers are operated with 10 Hz. 2/Feb./2012

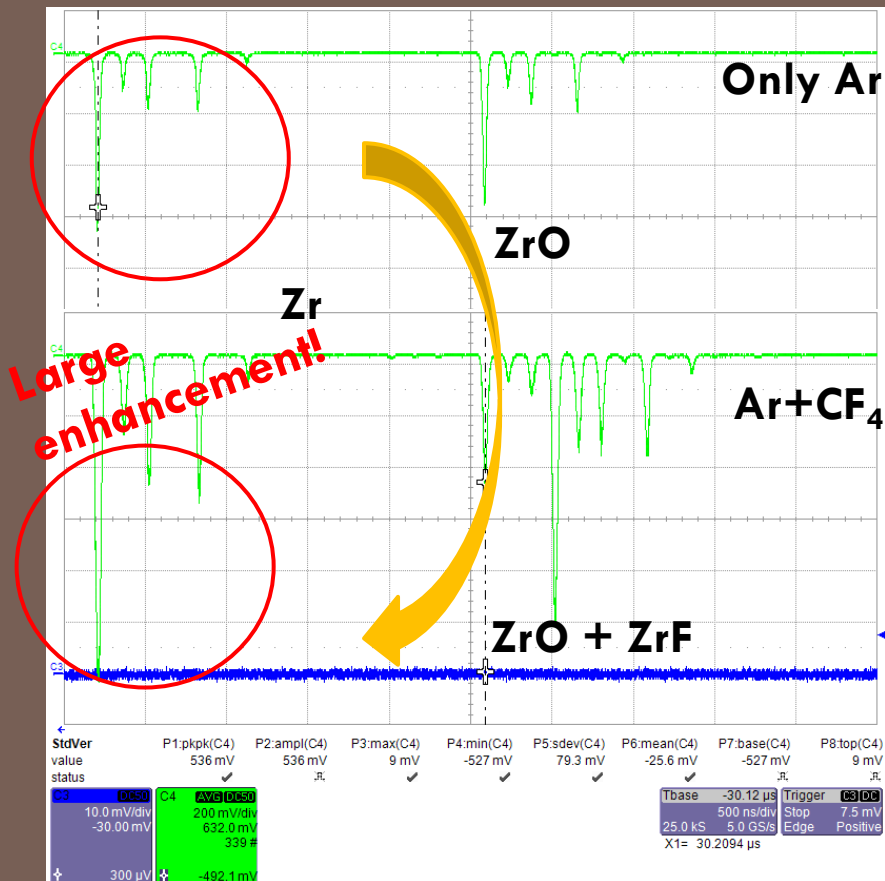
TOF spectrum w/only Ar



Ablation laser: 355 nm 8 mJ

UV(non-resonant ionization): 355 nm 2.3 mJ

TOF spectrum w/Ar + CF₄(6%)



- ZrF⁺ peaks were seen but disappeared in several minutes.
- Zr⁺ peaks were enhanced and seemed stable.
- We couldn't distinguish the breakup and ionization effect.

Ablation laser: 532 nm 8 mJ

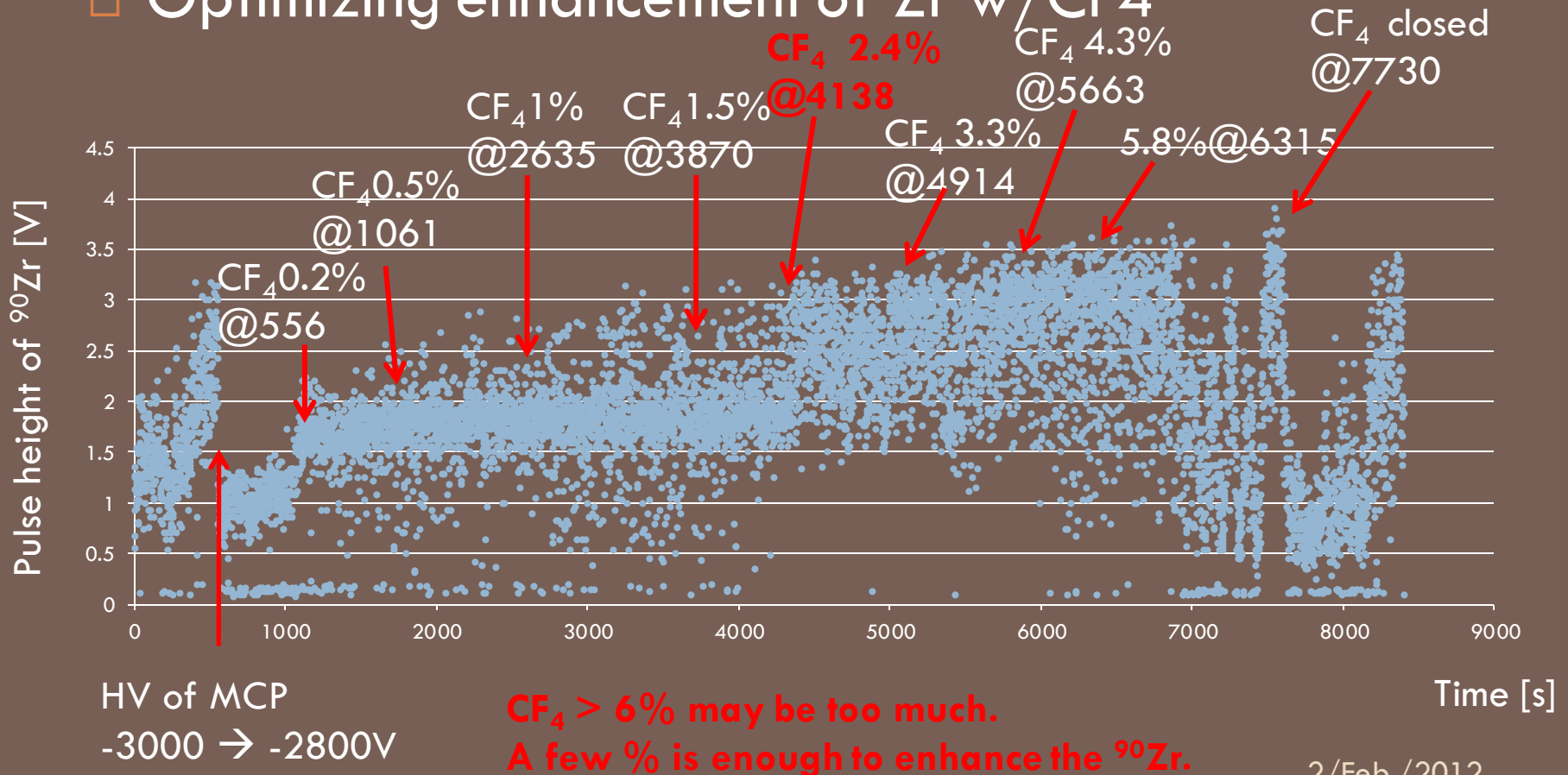
UV(non-resonant ionization): 355 nm 2.3 mJ

2/Feb./2012

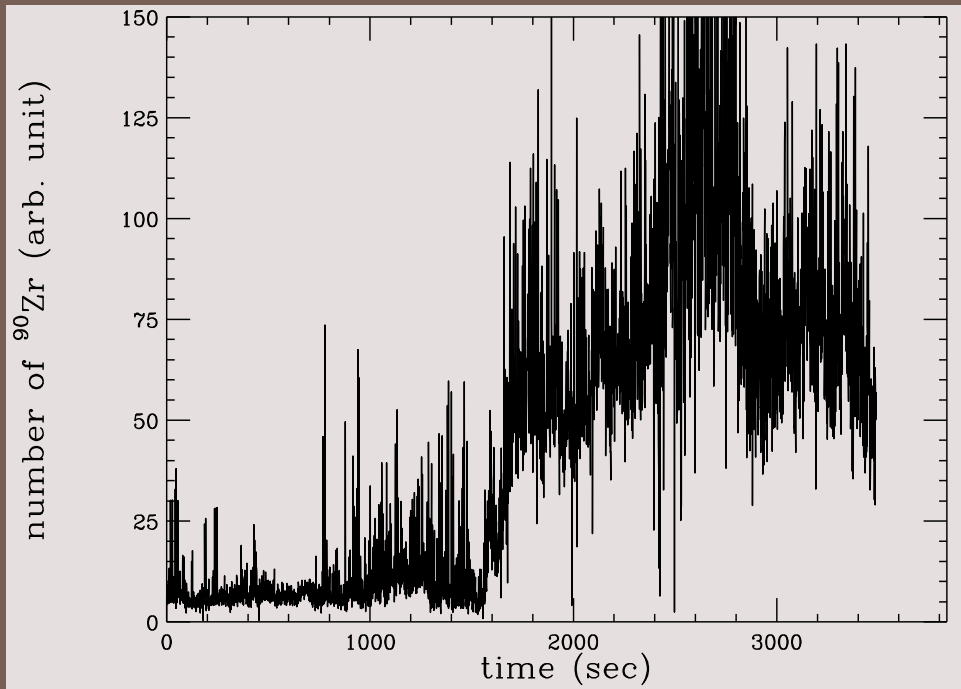
UV as dissociator and excitator

Confirmation of ZrF

Optimizing enhancement of Zr w/CF4



Non resonant ionization of Zr



Ablation laser:
1064 nm 3mJ

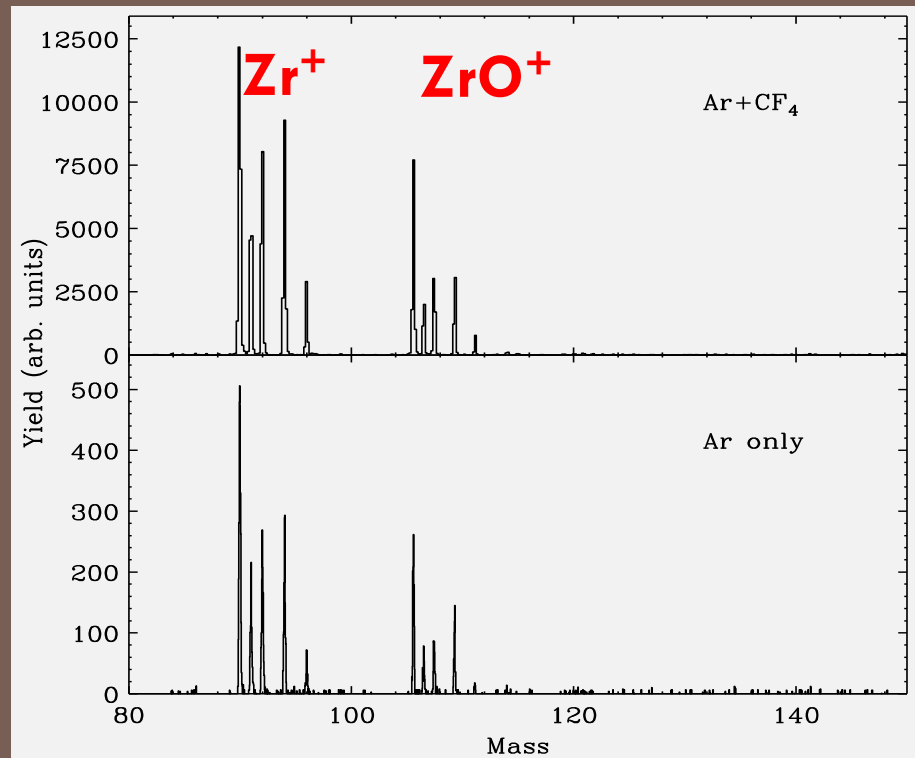
Dissociation + Ionization:
355 nm 2mJ

Only Ar

Ar + CF_4 (2.4%)

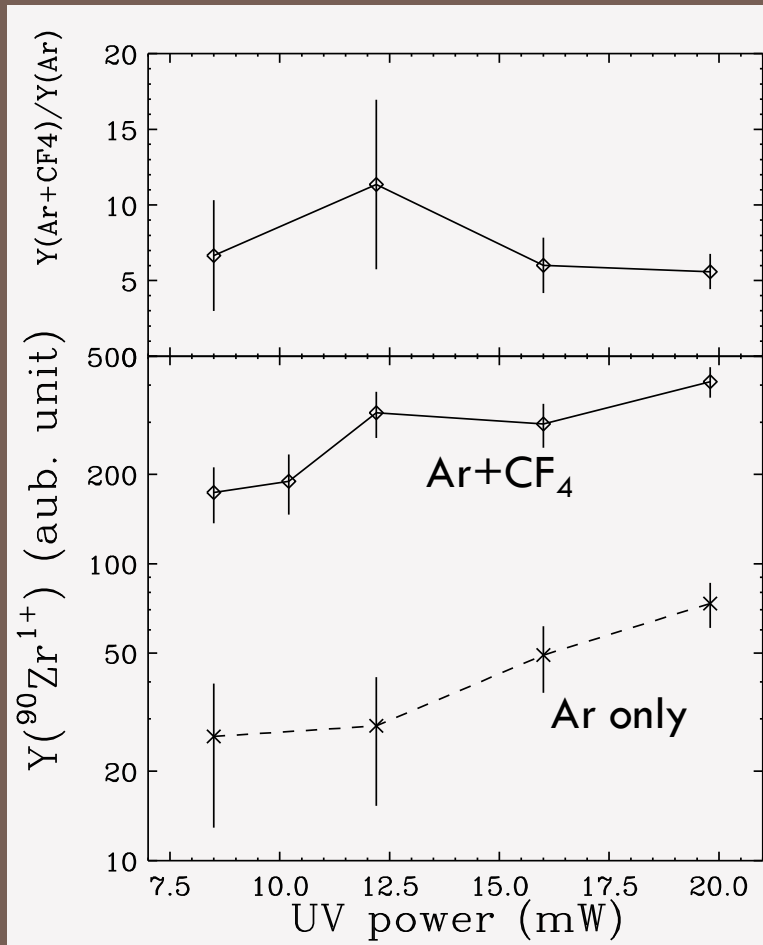
Yield of $^{90}\text{Zr}^+$ was enhanced **by 1 order of magnitude!**

Mass spectrum



- No ZrF⁺ was seen w/Ar +CF₄
- Ratio of Zr/ZrO doesn't change so much

UV (355nm) power dependence



1. Only Ar gas flow



2. Ar + CF₄ (2.4%) gas flow



and



No difference between Ar and Ar+CF₄



Breaking-up molecules seems saturated.

Possible mechanism

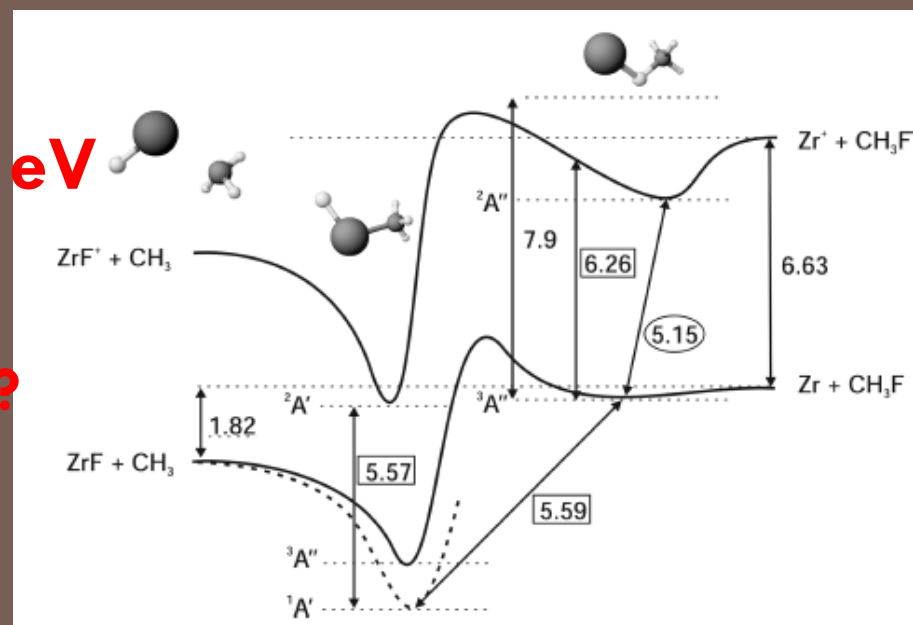
- $\text{ZrF} = \text{Zr} + \text{F} - 6.3 \text{ eV}$
- $\text{ZrF} + \text{CH}_3 = \text{Zr} + \text{CH}_3\text{F} - 1.82 \text{ eV}$

S. Soorkia et al.,
J. Phys. Chem. A, 2010, 114 (18), pp 5655

- $\text{ZrF} + \text{CF}_3 = \text{Zr} + \text{CF}_4 - 0.83 \text{ eV}$

0.83 eV = 1520 nm

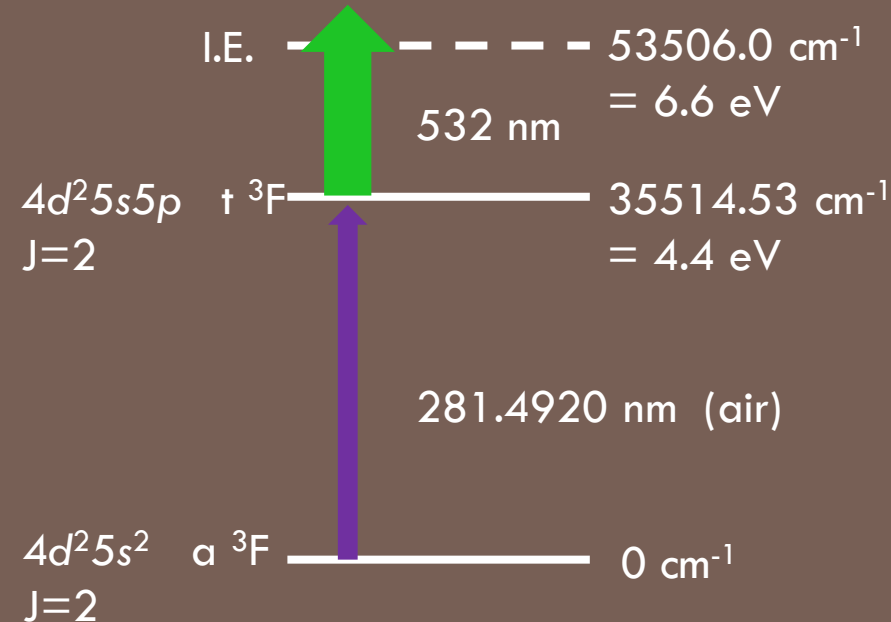
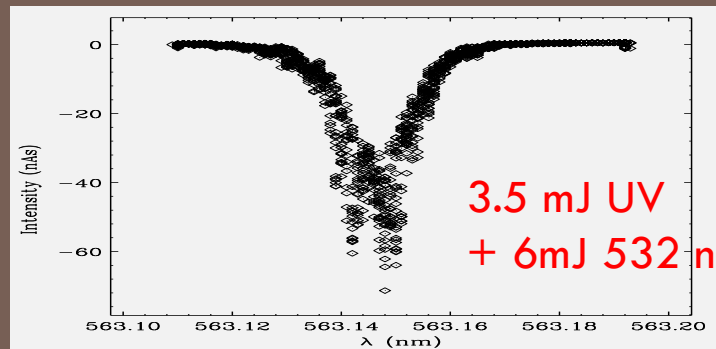
1064 nm Nd:Yag is enough ?



On-going works...

✓ Resonant ionization:

1st step 281.4920 nm + 532 nm
= 543218 cm⁻¹



□ Molecular breakup:

3rd hw (355nm), 2nd hw (532nm), 1064 nm of Nd:Yag
power dependence/time difference will be checked.

□ Removing of ZrO from Zr rod

Etching the Zr rod with acid prior to ablation

How to implement at ISOLDE?

□ VADIS + RILIS

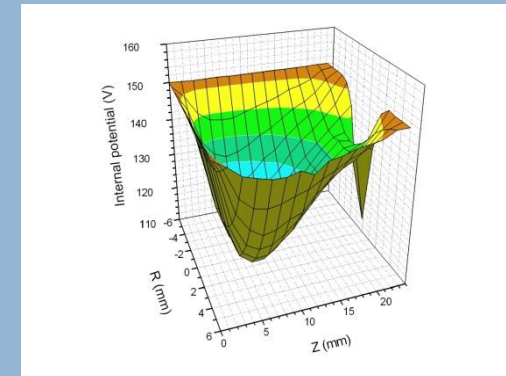
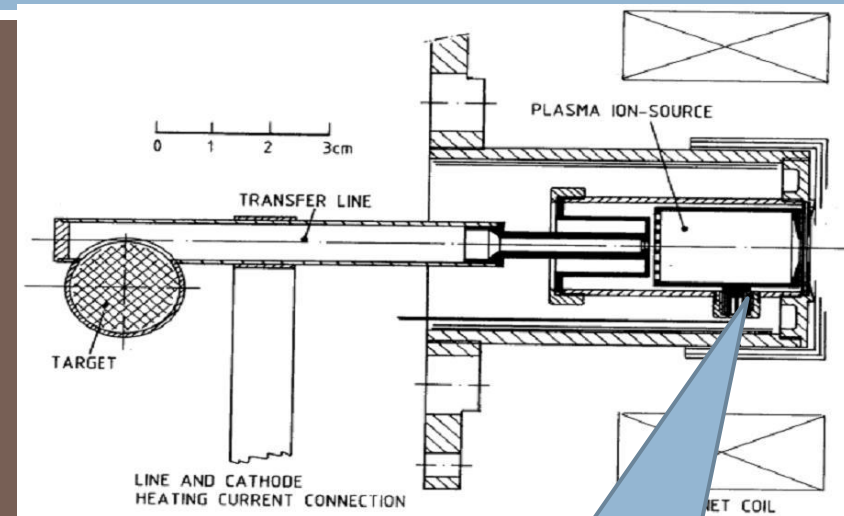
- Breaking-up molecule:

VADIS cavity w/several eV ?

or breakup laser ?

- Resonant ionization: RILIS

The resonant ionization scheme hasn't been established yet.



Summary

- Feasibility study of breaking-up refractory element fluoride is on-going with ablation + TOF spectrometer of Zr.
- In the case of non-resonant ionization, with small injection of CF_4 in Ar gas, large enhancement of Zr^+ peaks were observed.
- Both laser for resonant ionization of Zr and break-up molecules are ready.
- Further investigation of the phenomenon provides promising experimental opportunity.

Outlooks

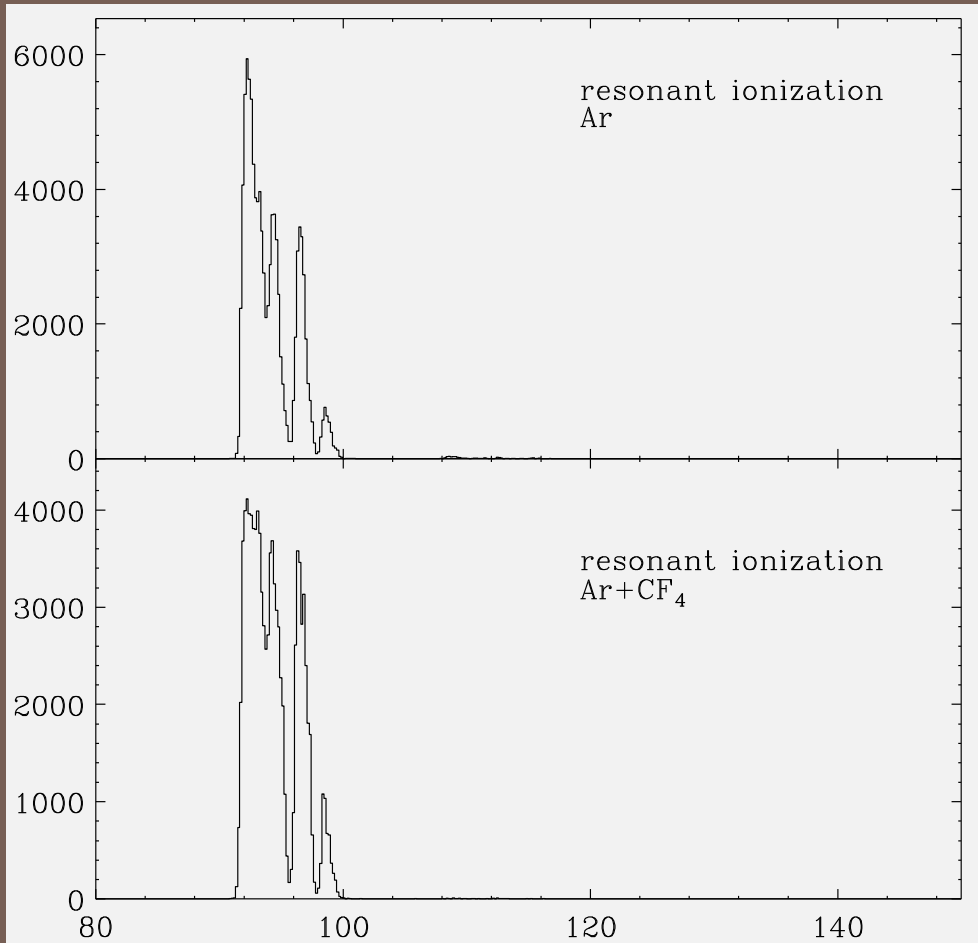
- Implementation with VADIS + RILIS

- Establishment of resonant-ionization:
 hopefully 3 steps-resonant ionization

La Fin



Mass spectrum of resonant ionization



In both cases, only Zr peaks were observed.