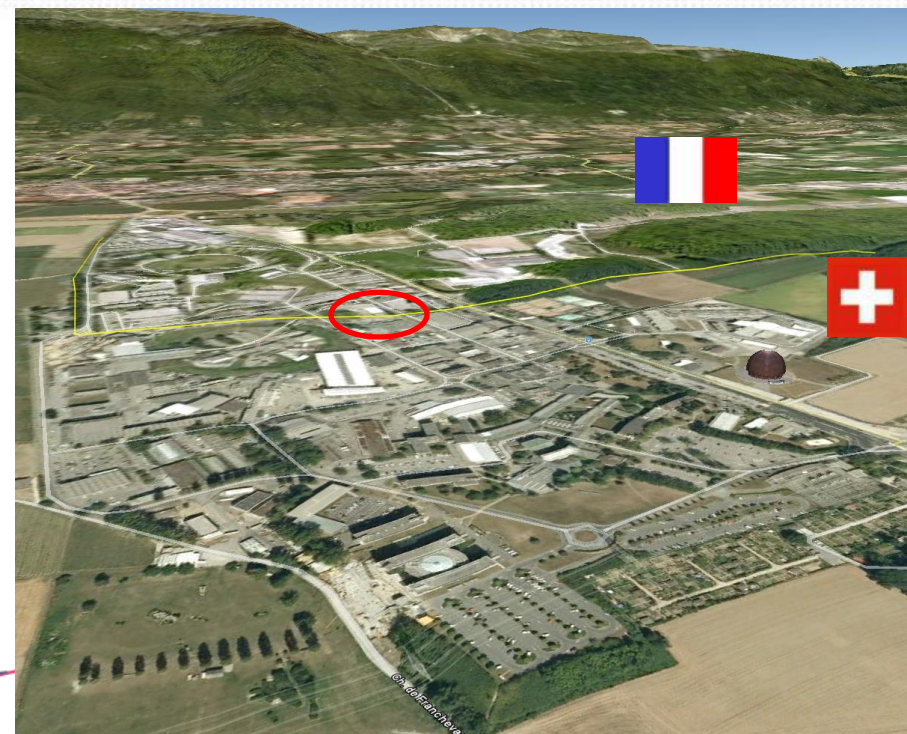
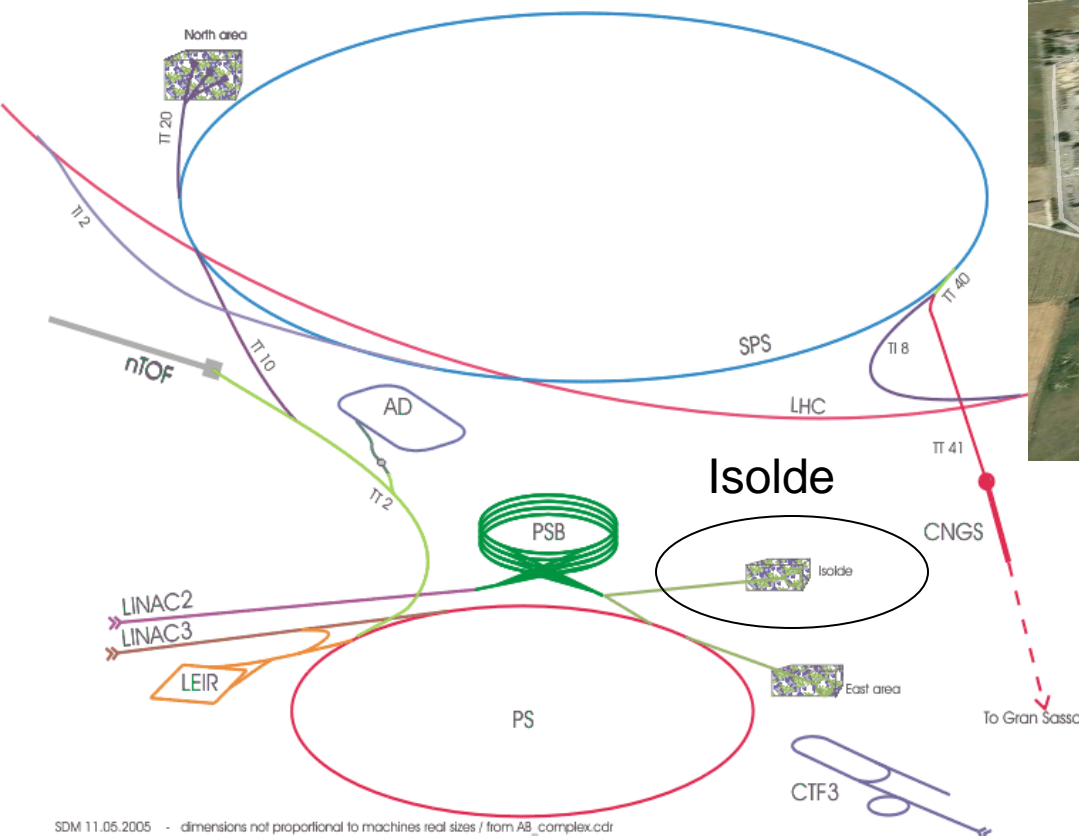


# Beams available at (HIE)-ISOLDE

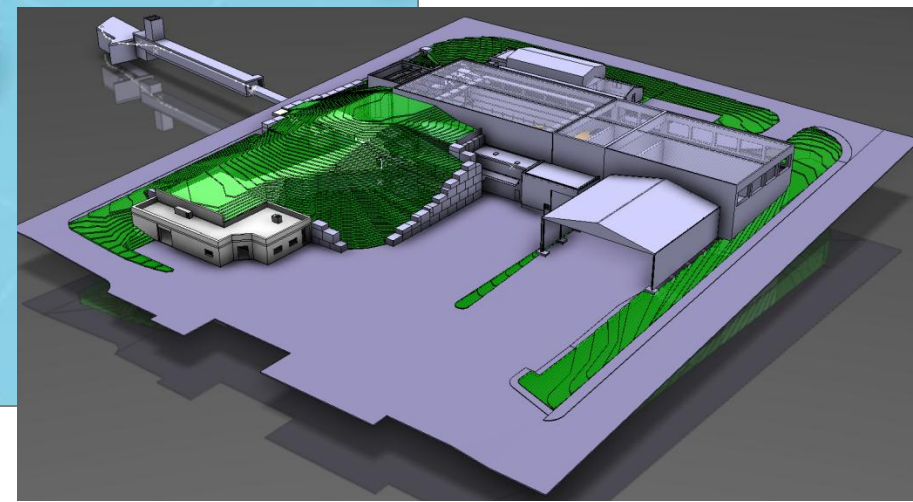
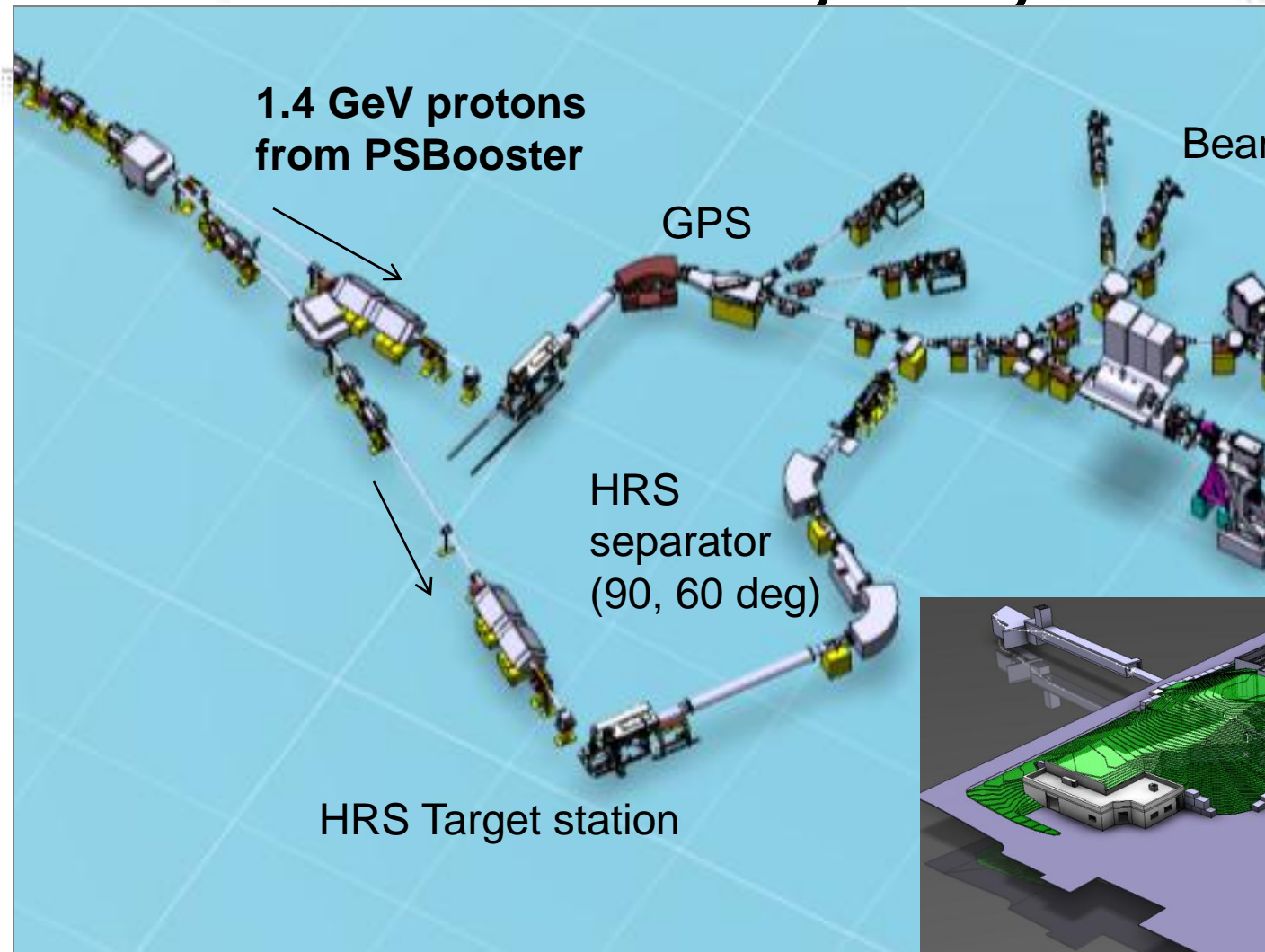
Thierry.stora@cern.ch

Target and Ion Source Development  
Team Leader  
CERN-ISOLDE

# The CERN Accelerator Complex



# Facility Layout



# Beams at ISOLDE and requested for HIE-ISOLDE

Isotopes produced from an element independent on target

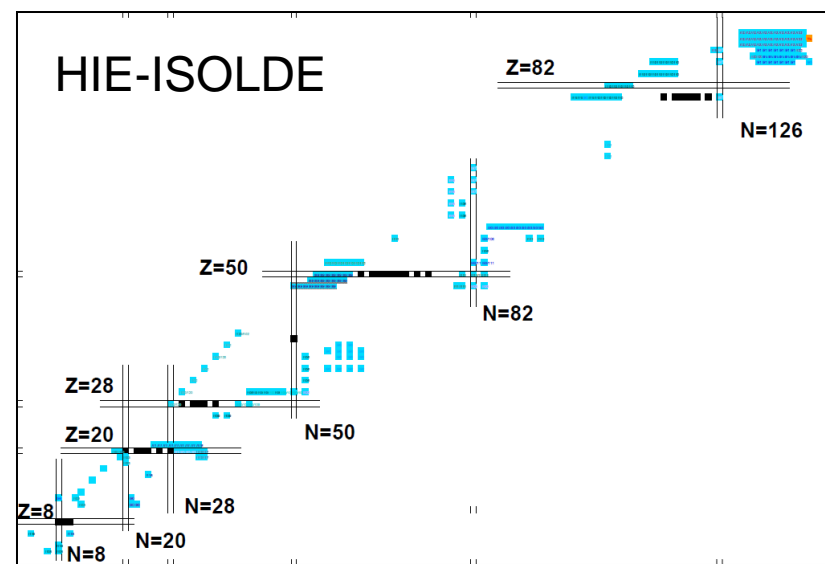
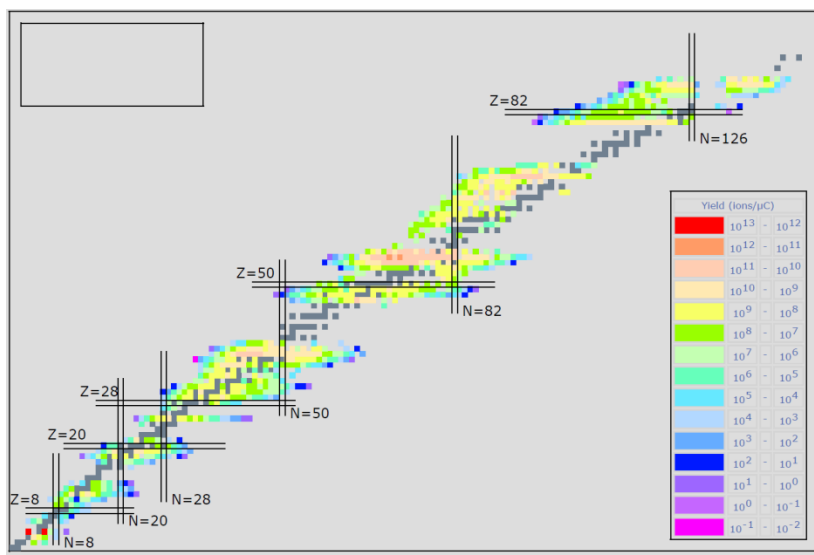
Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	1A	2A	3B	4B	5B	6B	7B	8B			1B	2B	3A	4A	5A	6A	7A	8A	
Period																			
1	1																		
2	2	3	4																10
3	11	12											13	14	15	16	17	18	
4	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	
5	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	
6	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	
7	87	88	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	104	

**Ion source:**

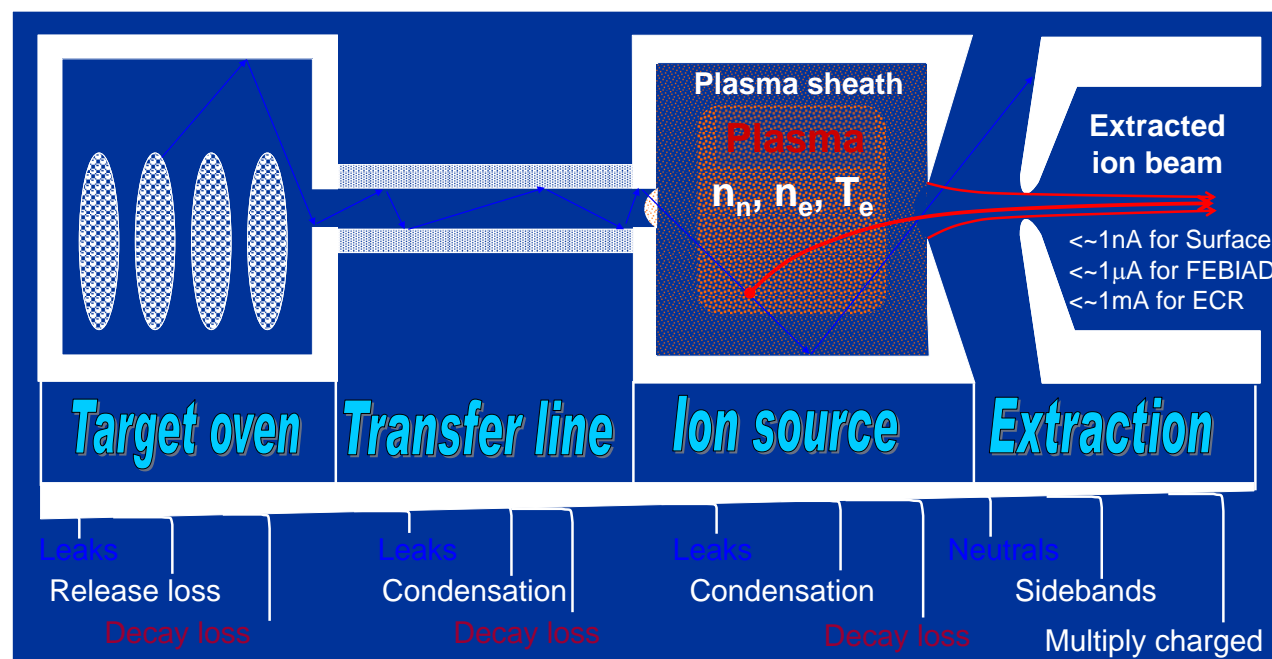
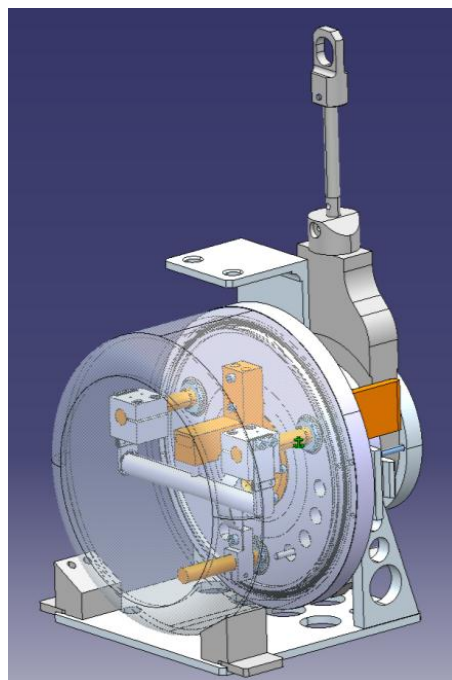
+	Surface	-
hot	Plasma	cool
	Laser	

\* Lanthanides

\*\* Actinides



# How to produce a radioactive ion beam with the "ISOL" technique



# 2238U&232Th targets for ISOL radioisotope beams

RIB intensity  
[s<sup>-1</sup> μA<sup>-1</sup>]

Proton beam  
Intensity  
[s<sup>-1</sup> μ A<sup>-1</sup>]

Avogadro  
Numb.  
N/A

Diffusion+  
Effusion  
Efficiency  
ε<sub>diff + eff</sub>

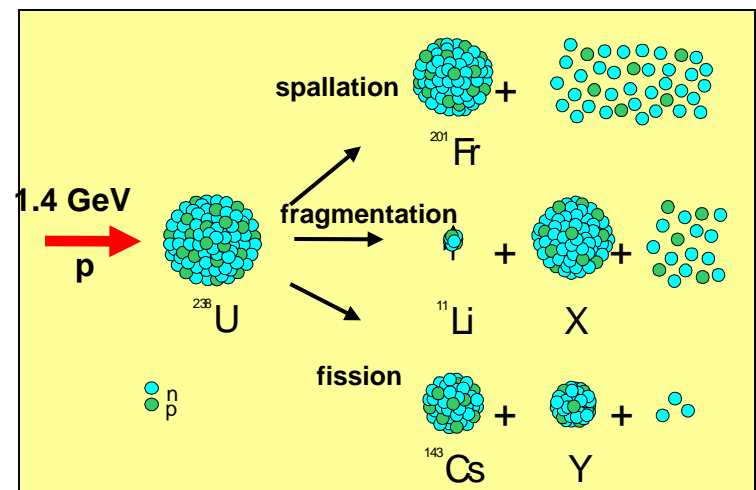
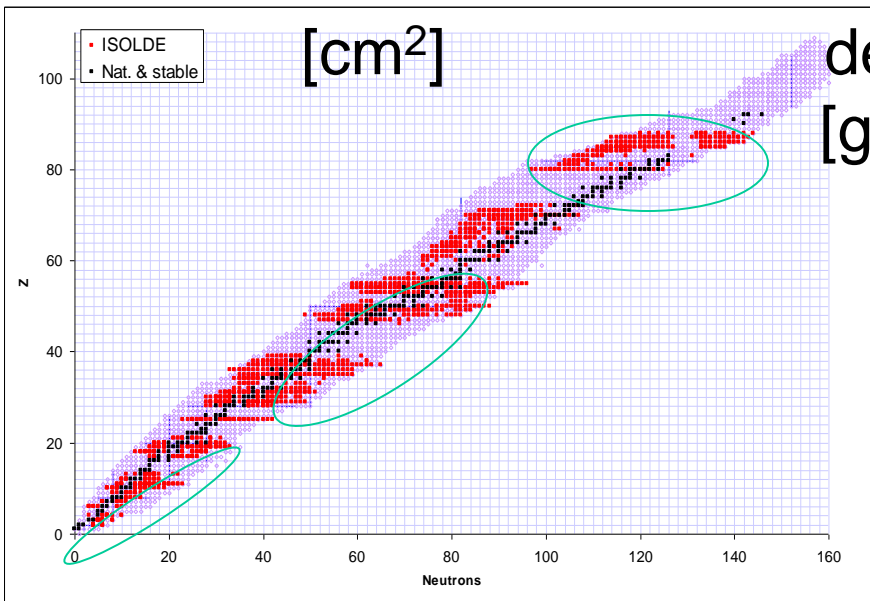
$$I = \int \sigma(E) \Phi(E, \mathbf{x}) \rho(\mathbf{x}) N/A \, dx \quad \varepsilon_{\text{diff + eff}} \quad \varepsilon_{\text{ion}}$$

Cross section  
[cm<sup>2</sup>]

Target  
density  
[g cm<sup>-3</sup>]

Atomic Mass  
[g]

Ionization  
Efficiency  
ε<sub>ion</sub>



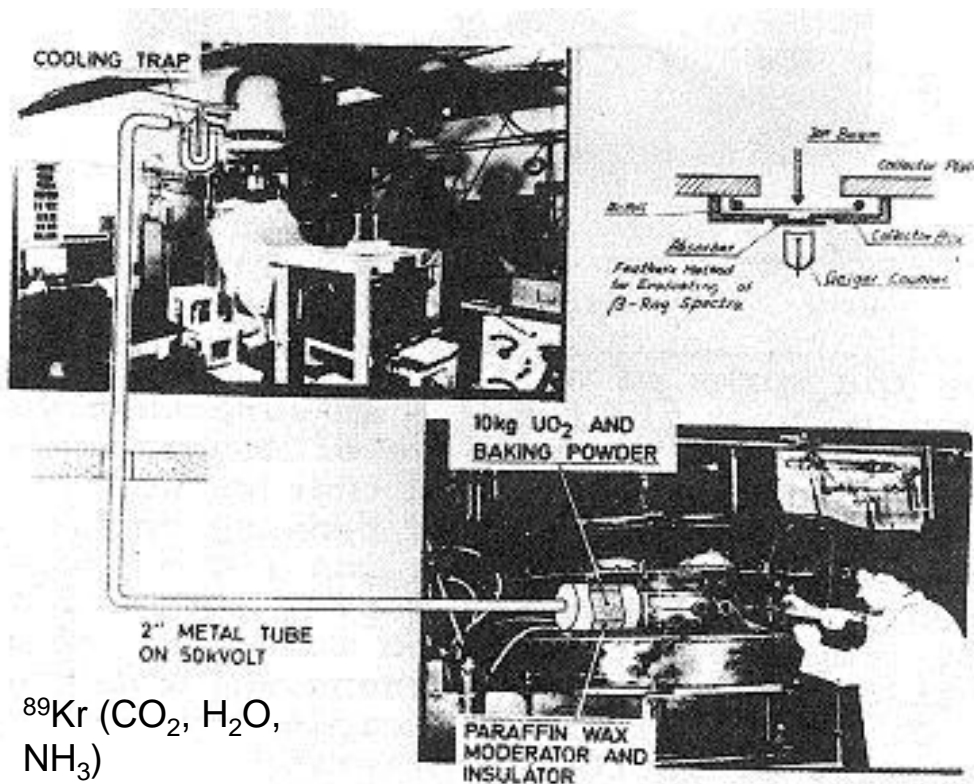
# THE BIRTH OF ON-LINE ISOTOPE SEPARATION

## ISOLDE "0"

**O.Kofoed-Hansen**

**K.O. Nielsen**

*Dan. Mat.Fys.Medd. 26, no. 7 (1951)*



$^{89}\text{Kr}$  ( $\text{CO}_2$ ,  $\text{H}_2\text{O}$ ,  $\text{NH}_3$ )

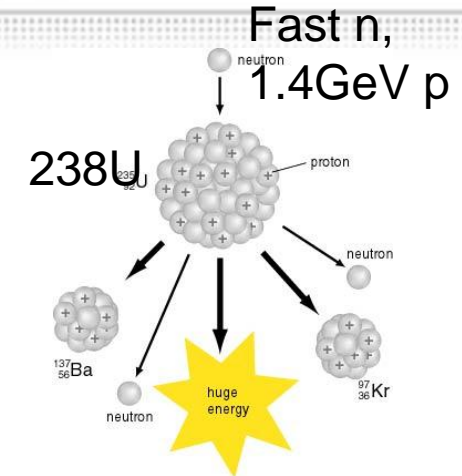
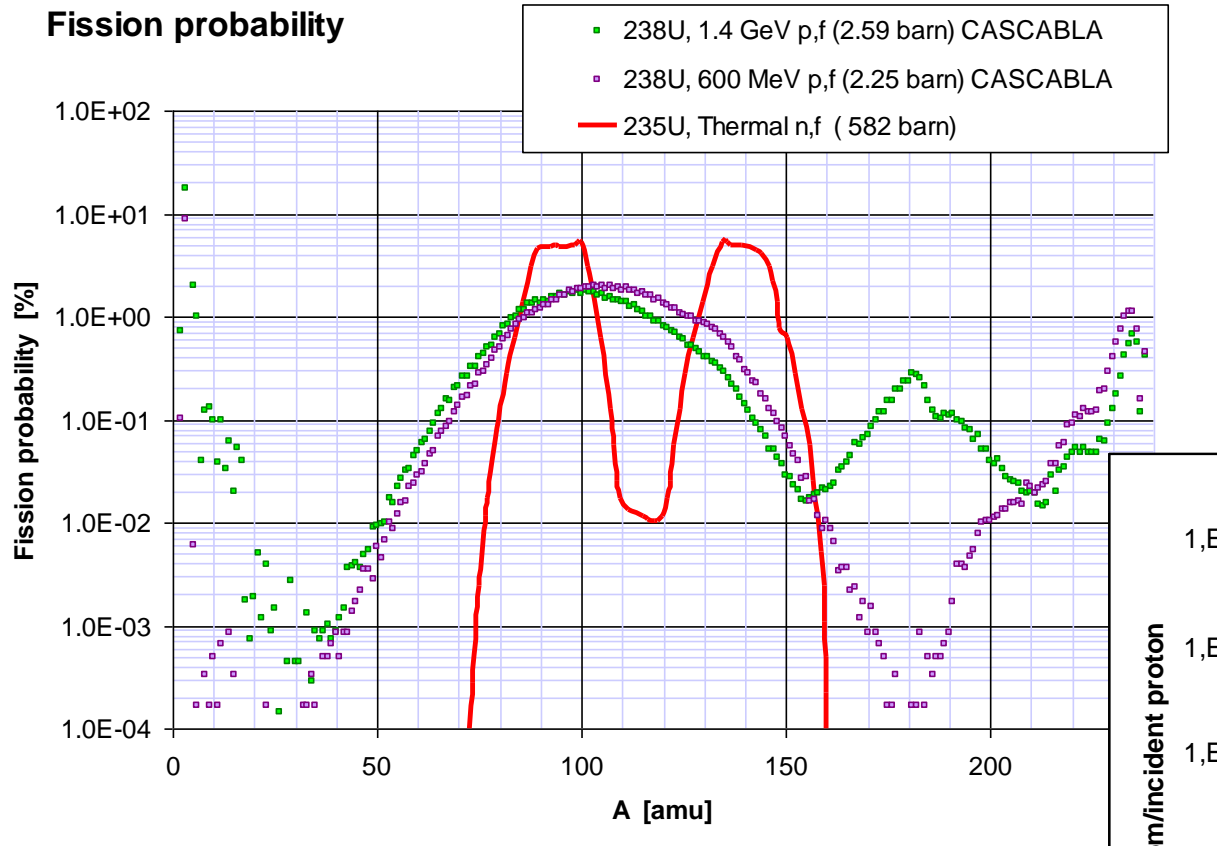


10 MeV deuterons  
 d-to-n converter (Be)  
 n moderator (wax)  
 $\text{UO}_2$  (10 kg)  
 Baking powder

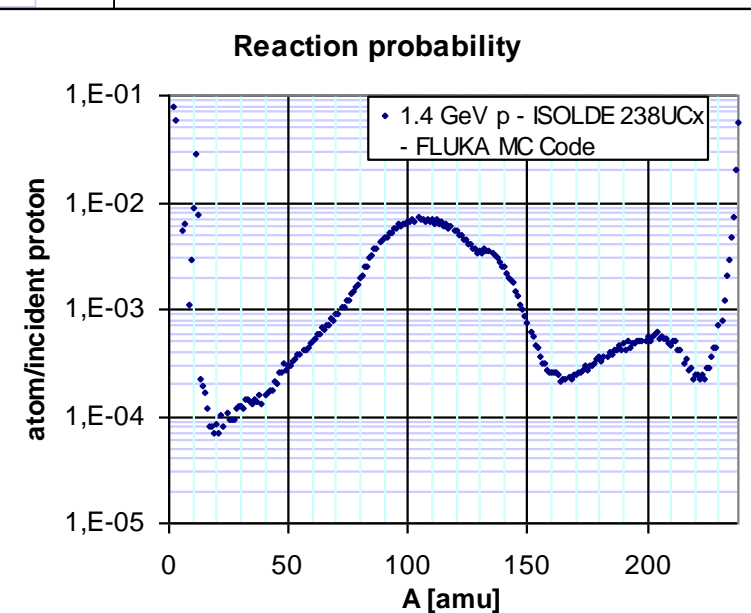
*From CERN 76-13, 3<sup>rd</sup> conf. nuclei far from stability*

# Uranium targets for ISOL beams

**Fission probability**



**Reaction probability**





Bundesministerium für Bildung und Wissenschaft

Forschungsbericht K 70-28

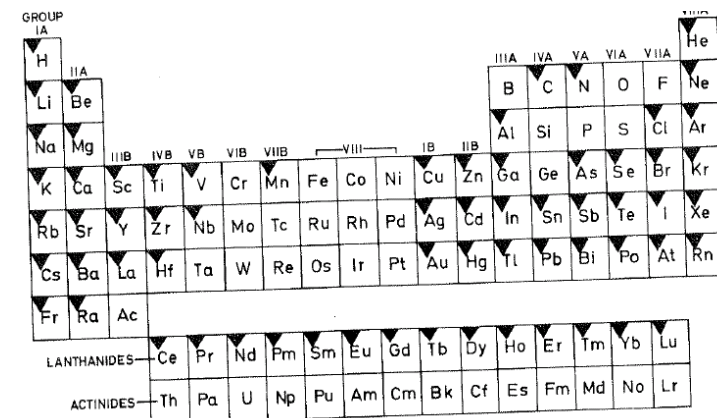
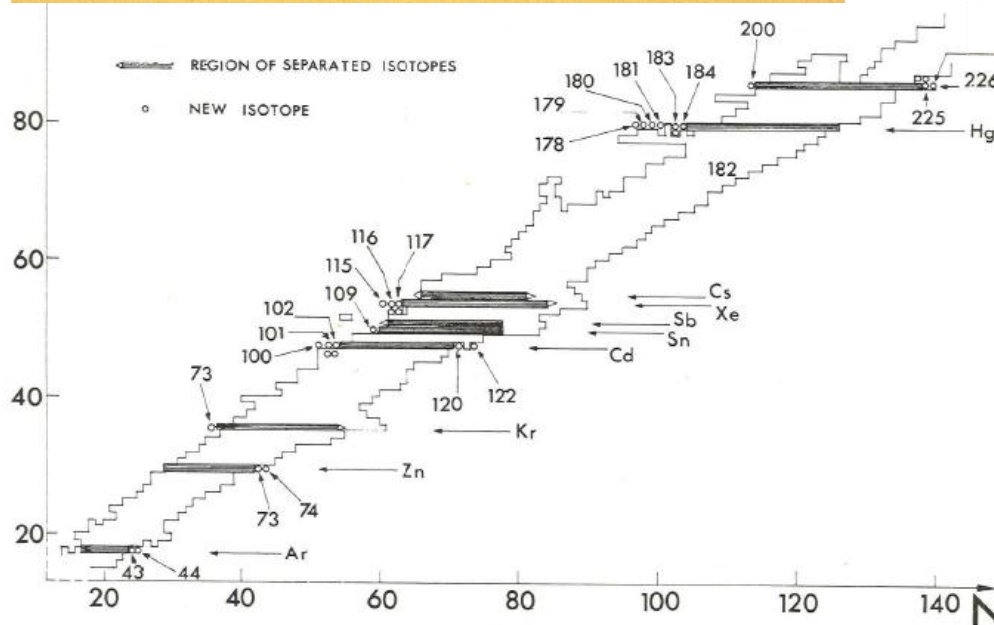
Kernforschung

PROCEEDINGS OF THE INTERNATIONAL CONFERENCE ON  
ELECTROMAGNETIC ISOTOPE SEPARATORS AND THE  
TECHNIQUES OF THEIR APPLICATIONS

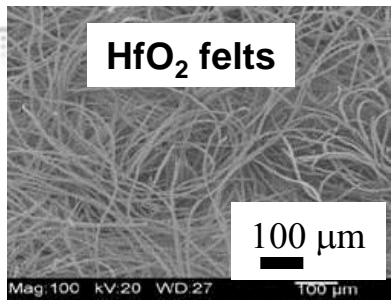
Marburg, Sept. 7 to Sept. 10, 1970

Target Material	Activities released	Target Temp. °C	Line Temp. °C	Comments
TiO <sub>2</sub> ·xH <sub>2</sub> O	Ar	-	-	
ZnO <sub>x</sub> ·xH <sub>2</sub> O	Kr	-	-	
CoO <sub>2</sub> ·xH <sub>2</sub> O	Xe	-	-	
TiO <sub>2</sub> ·xH <sub>2</sub> O	Rn	-	-	
Ge(metal)	Zn	1200	600	Air inlets disastrous
Sn(metal)	Cd	1000	600	
La(metal)	Cs	1300	600	Surface ionization source
	Xe	1300	-	Normal ion source, not as good as Co target
Pb(metal)	Hg (spallation)	760	-	
	Xe (fission)	760	-	
TaCl <sub>4</sub>	Sb	130	-	
	Sn	130	-	Transport line coated with SbCl <sub>3</sub>

CERN 86-05  
Experimental Physics  
Division  
18 July 1986



# ISOL targets and ion sources

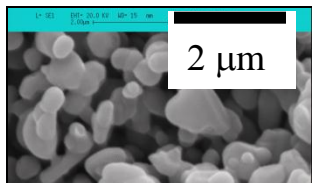
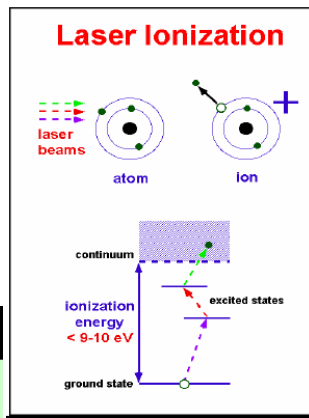


## Target materials (30):

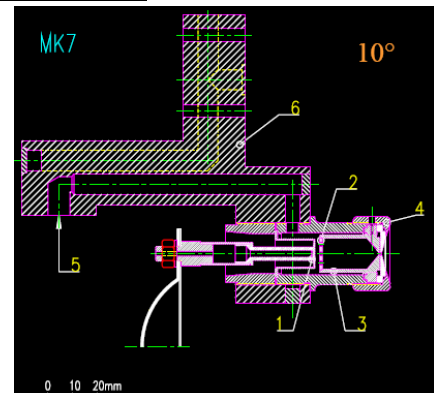
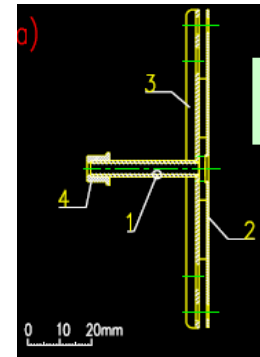
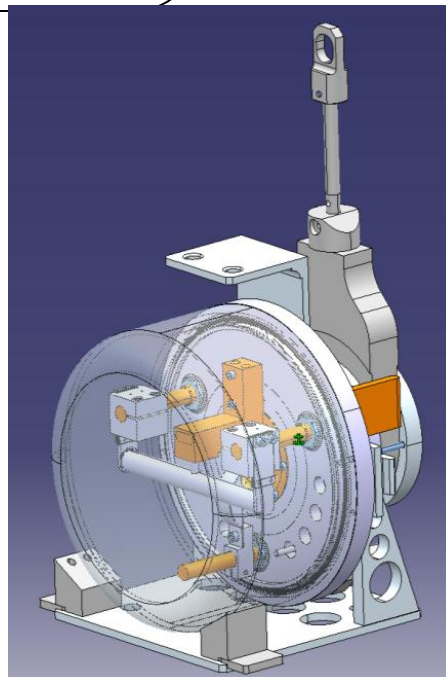
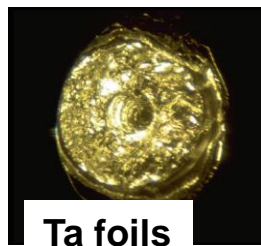
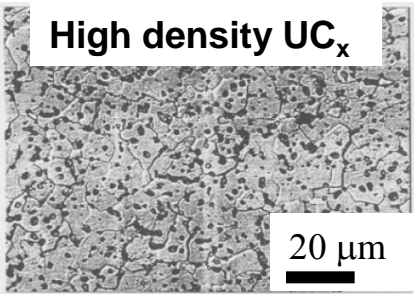
- Refractory oxides and carbides (Al<sub>2</sub>O<sub>3</sub>, SiC)
- Solid metals (Ta, Nb, Mo)
- Molten metals (Pb, La, Sn).

## Ion sources (>5):

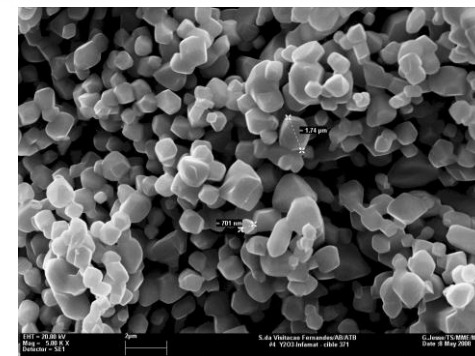
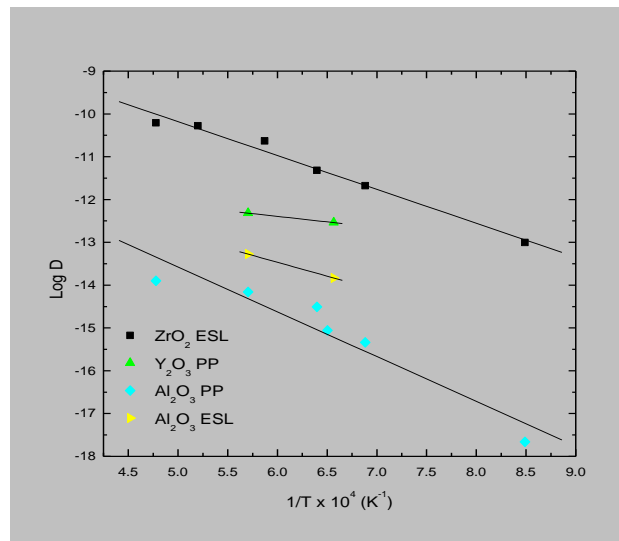
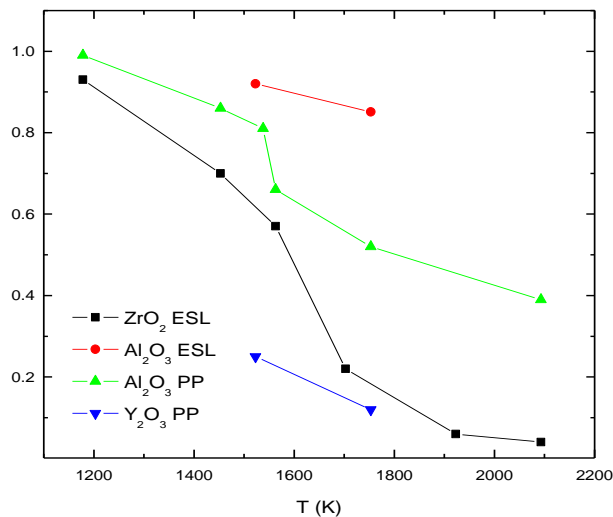
- Surface
- FEBIAD, Plasma
- RILIS



nm/sub-μm SiC



# Release properties of Kr isotopes

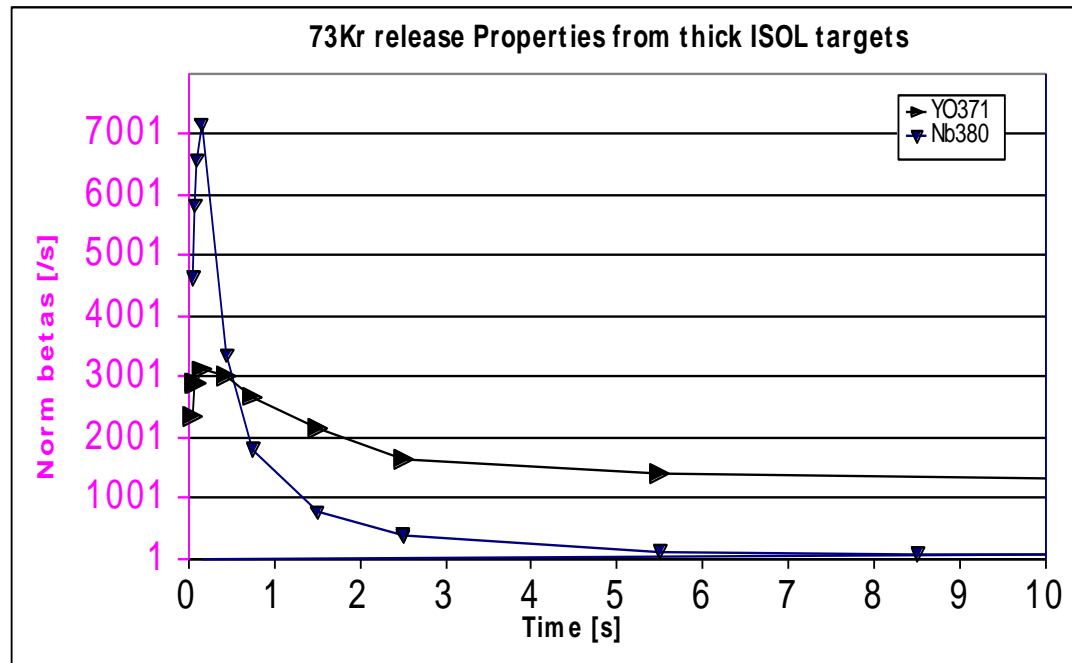


$$(\epsilon_{\text{target}})_{\text{off}} = \frac{3}{\pi} \sqrt{\left(\frac{\mu_s}{\lambda}\right)}$$

Nuclide	$t_{1/2}$ (s)	$\lambda$ (s <sup>-1</sup> )	$\mu_s$ (s <sup>-1</sup> )	$\epsilon(\text{target})_{\text{off}}$
<sup>72</sup> Kr	17.2	0.0405	1.766E-3	0.199
<sup>73</sup> Kr	27.0	0.0265		0.246

# Online yield of Kr

## Release curve Y2O3 sub-micron target vs Nb foils (30μm)



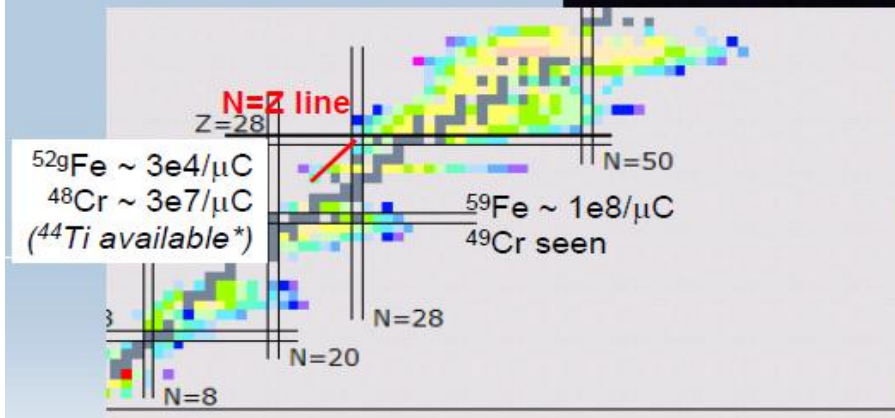
Yields of  $^{72}\text{Kr}$  have been improved by  $\times 10$  from  $2 \cdot 10^3 / \mu\text{C}$  to  $2 \cdot 10^4 / \mu\text{C}$  (combining prod cross section, target thickness, release efficiency and ion source efficiency)

# Direct Fe beams at ISOLDE

YO422- VD5 (2010)

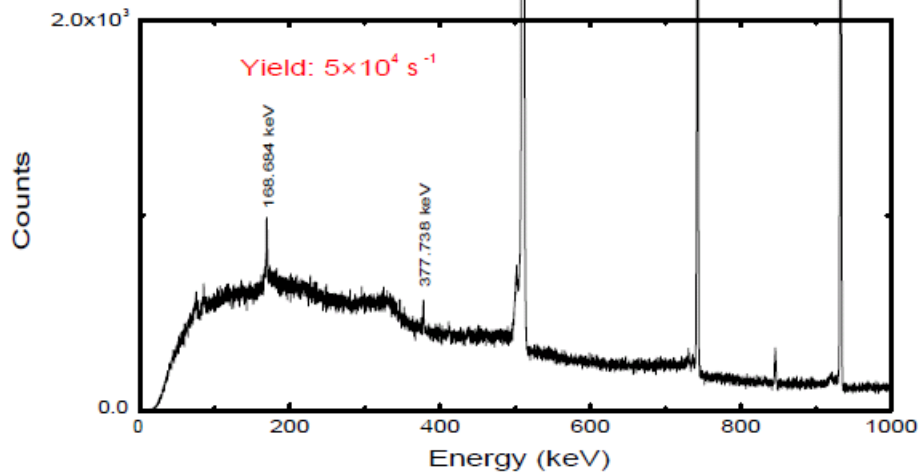
Young Scientist Award 2010

(PhD of Sandra Fernandes)

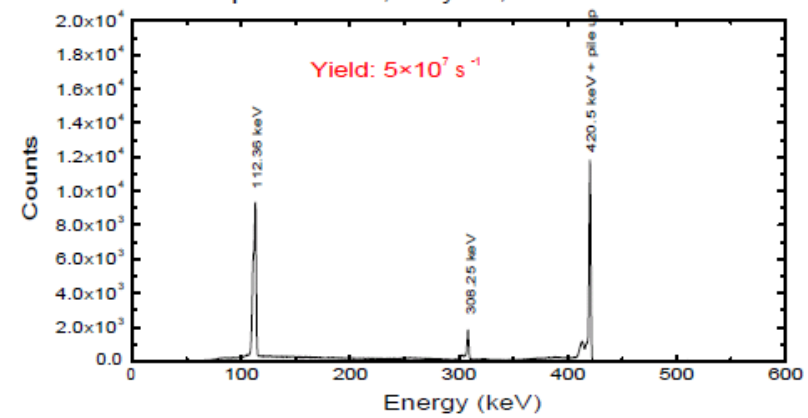


Courtesy of K. Johnston, M. Deicher et al.

Isotope:  $^{52}Fe/^{52}Mn$ , May 15, 2010 16:30



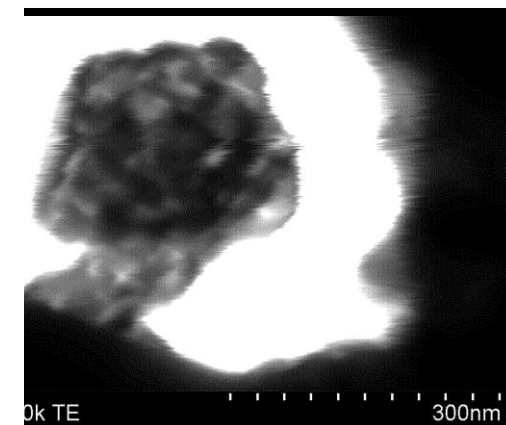
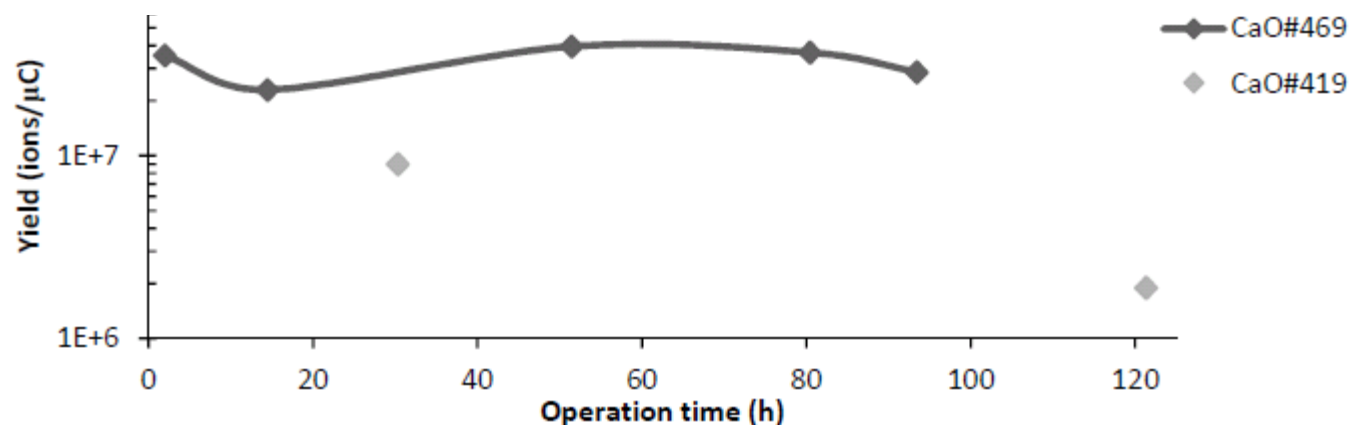
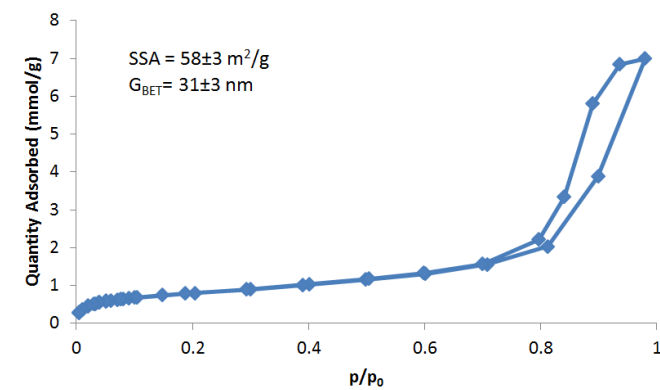
Isotope:  $^{48}Cr/^{48}V$ , May 14, 2010 03:00



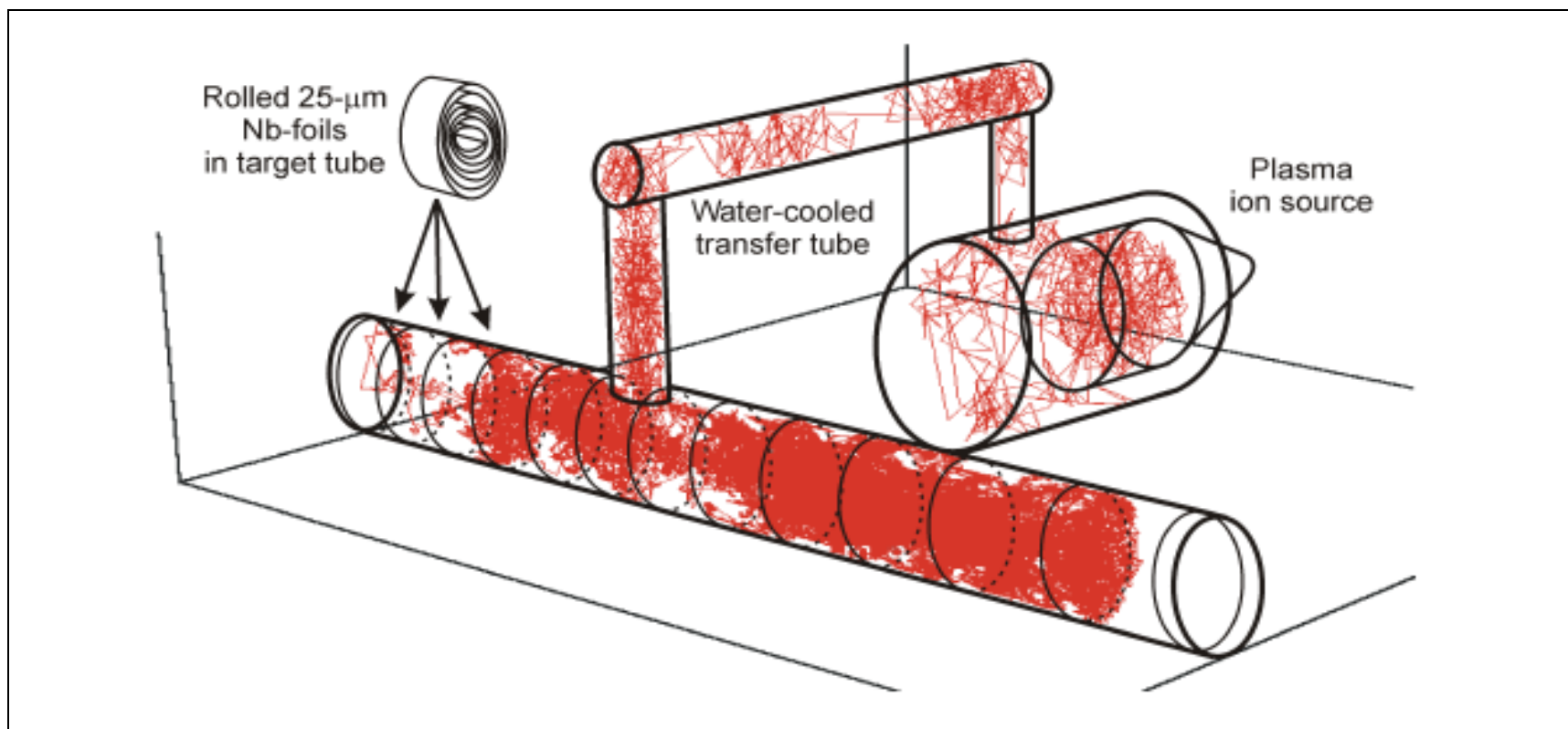
# CaO nanomaterials

JP Ramos et al.

Target	Isotope	Temperature (°C)	$\epsilon_{is}$ (%)	$Y_{prod}$ (ions/ $\mu$ C)	$Y_{obs}$ (ions/ $\mu$ C)	$\epsilon_{rel}$ (%)
CaO#469	$^{31}\text{Ar}$	650		$3.2 \times 10^4$	$4.2 \times 10^1$	0.9
	$^{32}\text{Ar}$			$1.1 \times 10^6$	$2.4 \times 10^3$	1.6
	$^{35}\text{Ar}$		14.0	$2.0 \times 10^7$	$1.2 \times 10^5$	4.3
	$^{35}\text{Ar}$	520	7.2	$3.4 \times 10^9$	$8.9 \times 10^6$	3.6
		650	8.0		$3.5 \times 10^7$	12.9
		730	6.1		$3.9 \times 10^7$	19.2
840		19.4	$2.0 \times 10^8$		29.8	
CaO#123	$^{31}\text{Ar}$	1050	9	$1.8 \times 10^4$	$1.5 \times 10^0$	0.1
	$^{32}\text{Ar}$			$6.2 \times 10^5$	$3.3 \times 10^3$	5.9
	$^{33}\text{Ar}$			$1.1 \times 10^7$	$3.8 \times 10^4$	3.8
	$^{35}\text{Ar}$			$1.9 \times 10^9$	$4.3 \times 10^7$	25.1
CaO#408	$^{31}\text{Ar}$	>1050	21	$2.0 \times 10^4$	$5.0 \times 10^0$	0.1
CaO#419	$^{35}\text{Ar}$	950 (run begin)	8	$1.8 \times 10^9$	$9.0 \times 10^6$	6.2
		950 (run end)			$1.9 \times 10^8$	1.3



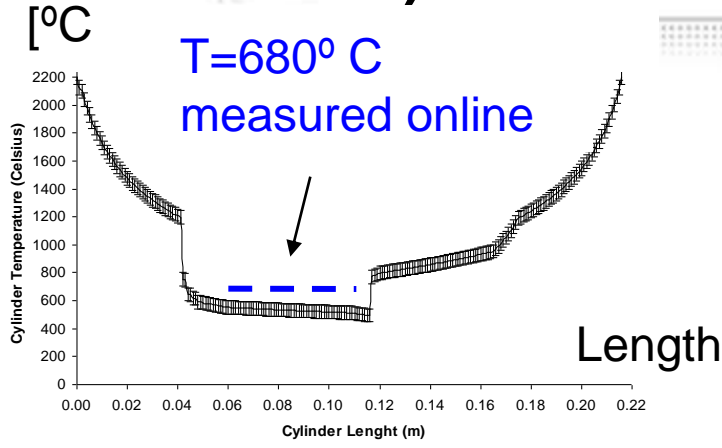
# Monte Carlo code to compute trajectories



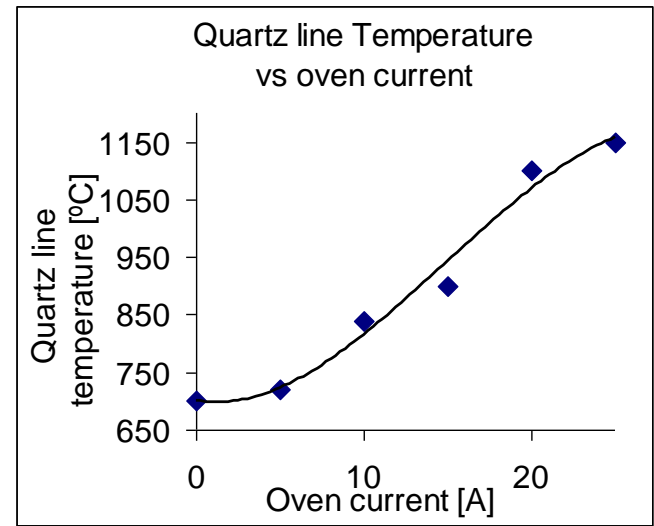
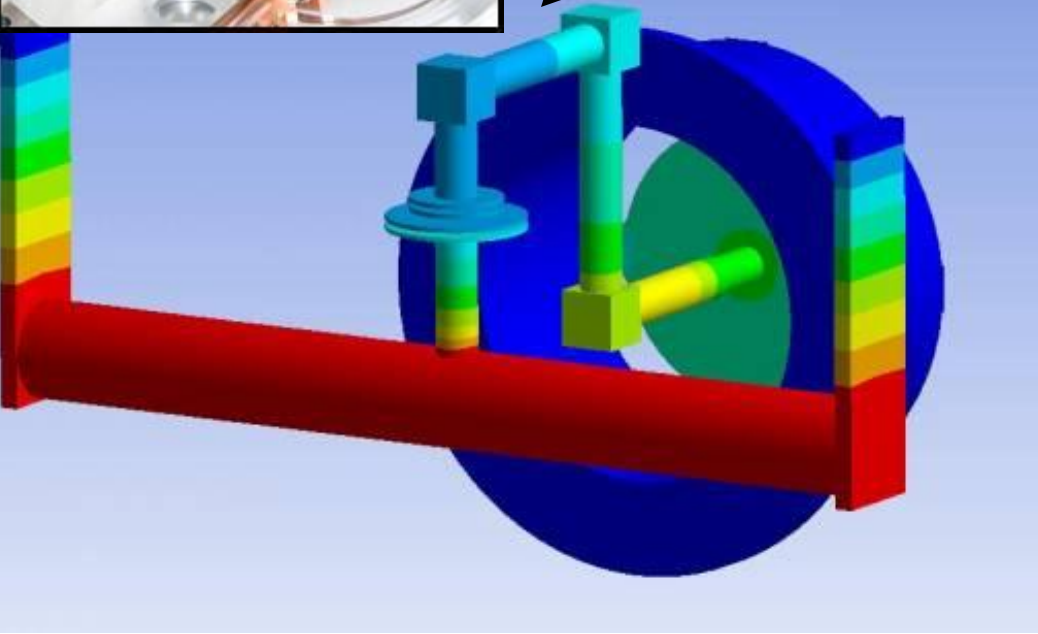
Monte Carlo numerical simulation of isotope trajectory (RIBO code)

# UCx – 314 (Quartz Insert)

Progress: AUKAUIS (U.S. 17/16)

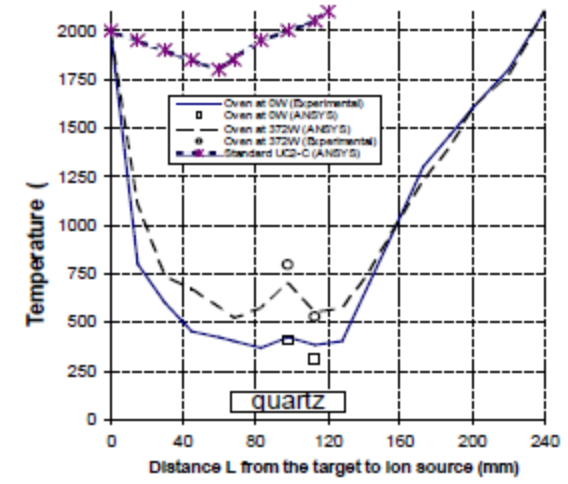
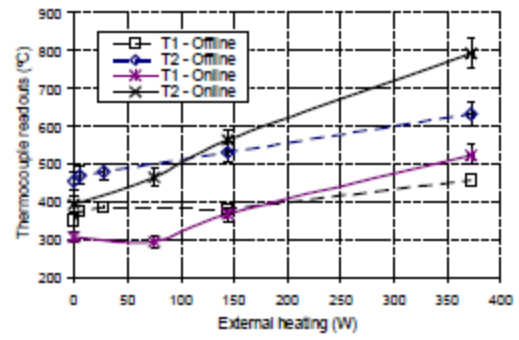
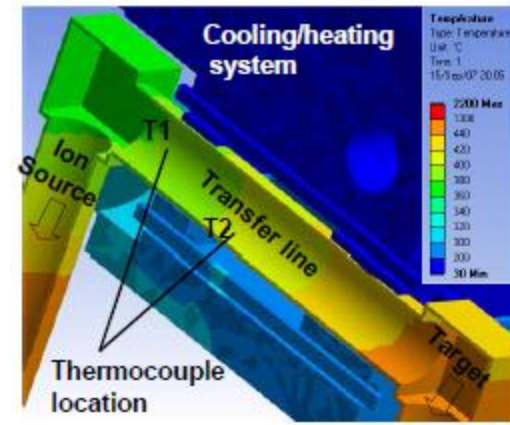
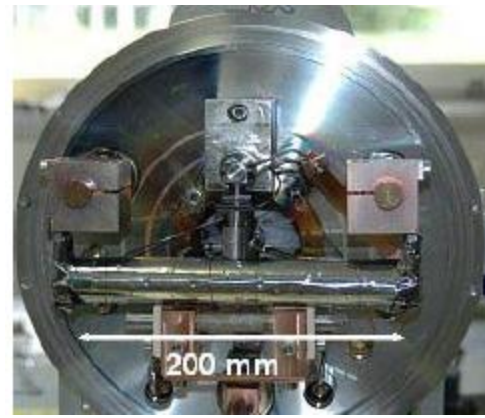
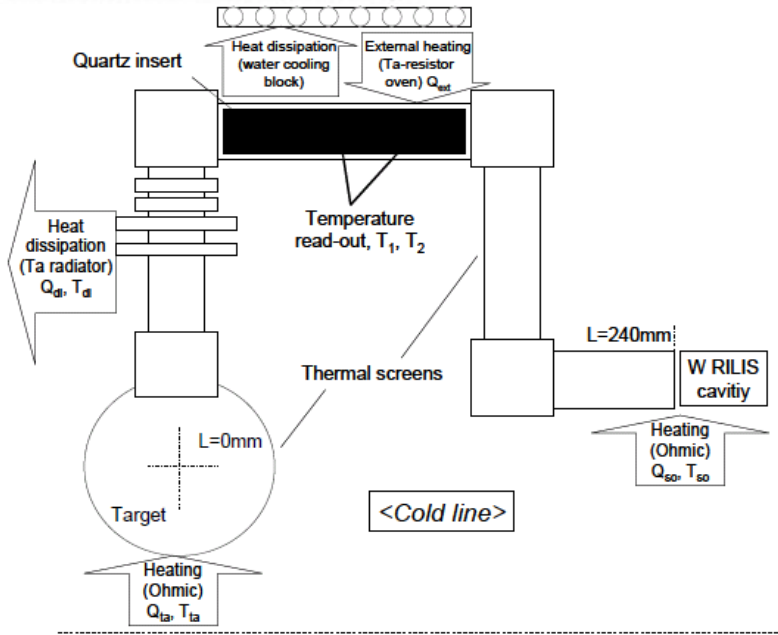


Quartz inserted here

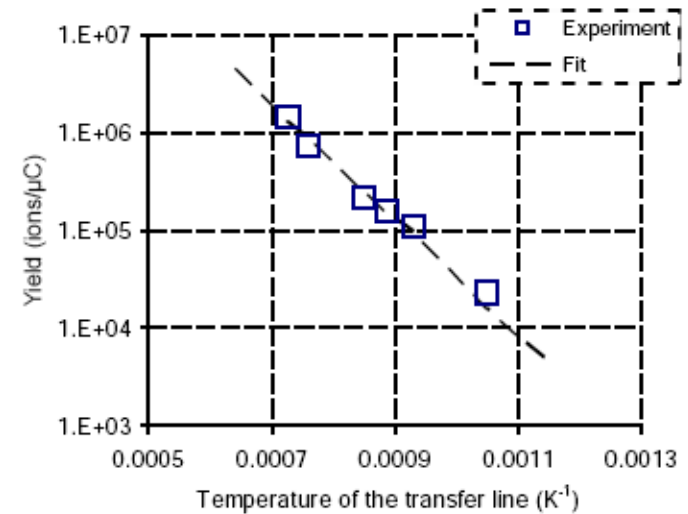
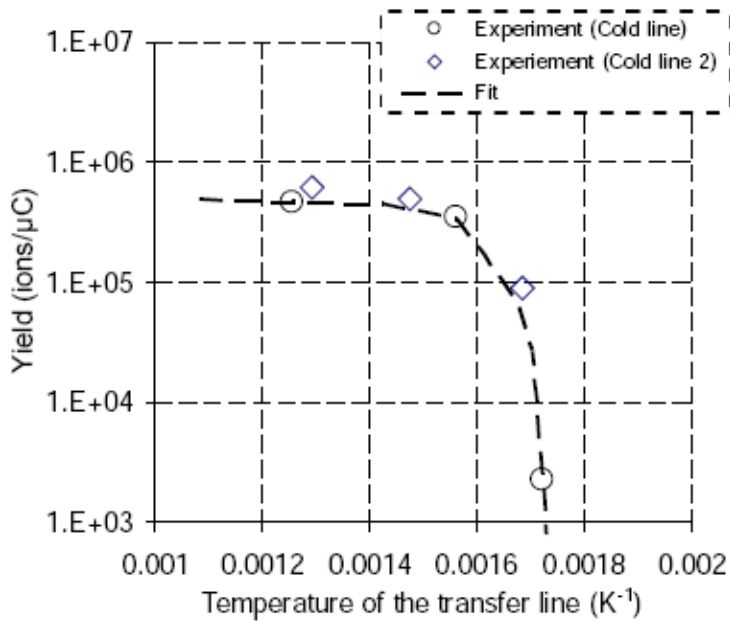




# Heating, temperature profiles...



# Purification of $^{80}\text{Zn}$ & $^{130}\text{Cd}$ beams

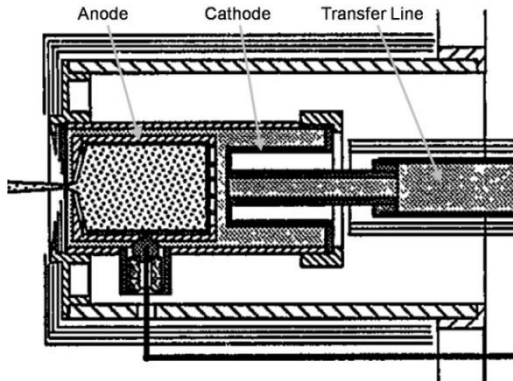


$^{126}\text{Cs}$  yield function of quartz temp  
 Fit with  $\Delta H_{\text{ads}} = -145 \pm 20 \text{ kJ/mol}$   
 as only free parameter  
 Isothermal vacuum chromatography  
 is ca  $-180 \text{ kJ/mol}$

$^{80}\text{Rb}$  yield function of quartz temp  
 Fit with  $\Delta H_{\text{ads}} = -242 \pm 20 \text{ kJ/mol}$   
 as only free parameter  
 Isothermal vacuum chromatography  
 is ca  $-270 \text{ kJ/mol}$

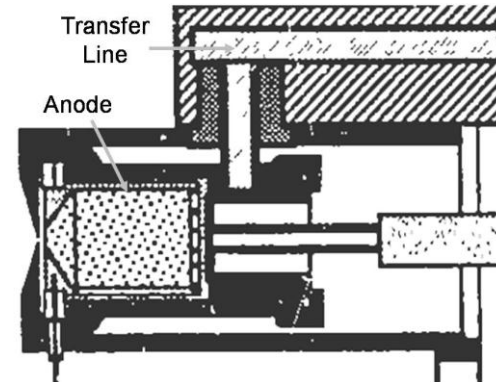
# The 3 most “recent” ISOLDE FEBIADs up to 2009

MK5



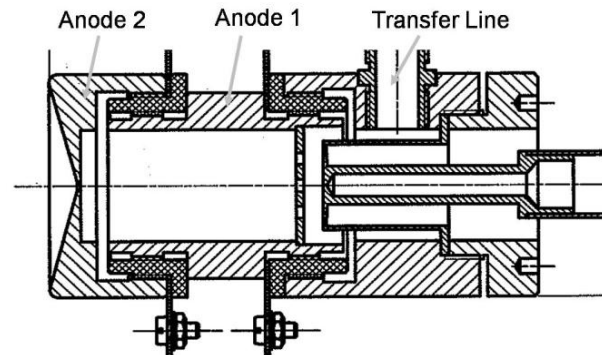
- Hot transfer line
- Employed for the ionization of the condensable elements

MK7



- Water-cooled transfer line
- Employed for the ionization of the noble gases and molecular compounds

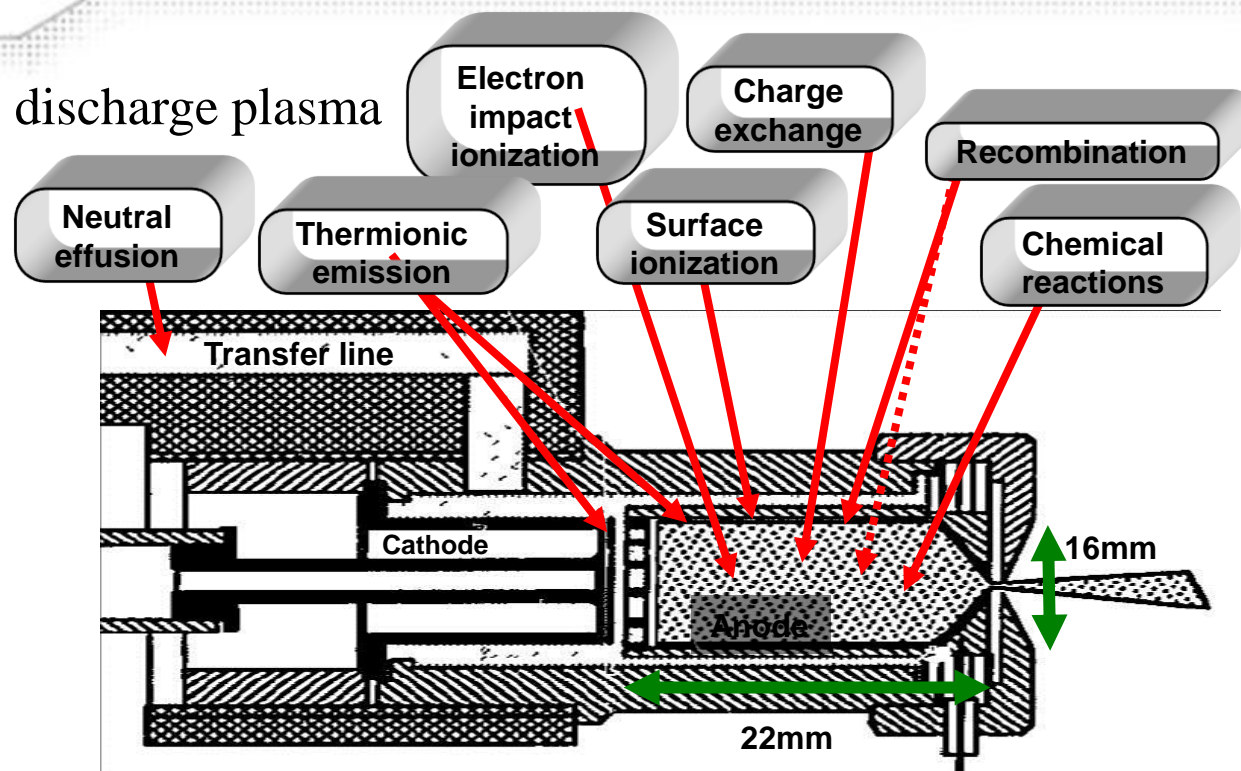
MK3



- Temperature controlled transfer line
- Coupled to the molten metal targets

# Investigation tools: Analytical

Modeling of the arc discharge plasma



- Full cocktail of possible phenomena.
- Not all appearing all over the variation range of the operation parameters.
- Some of them can be neglected at the nominal parameters.
- Application range has been investigated (experiment vs. theory).
- Performance limitations could be pointed out, justified and removed

# Ionization efficiency modeling

$$\varepsilon = f \times \frac{R_{ioniz}}{n_{n\_in}}$$

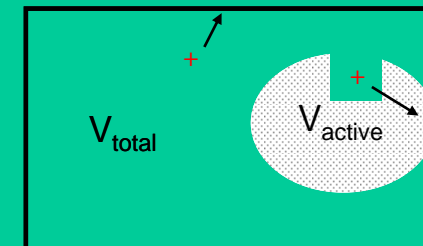
=>

$$\varepsilon = f \times V_{source} \times \frac{n_e \times n_n \times \sigma_{ioniz} \times v_{rel}}{n_{n\_in}}$$

## The extraction factor, $f$

$f$  = the fraction of the produced ions that are extracted before losing their charge on the ion source walls or being pumped.

$f$  (geometrical) = the fraction of the source volume where the generated ions are extracted from, due to favorable field distribution.

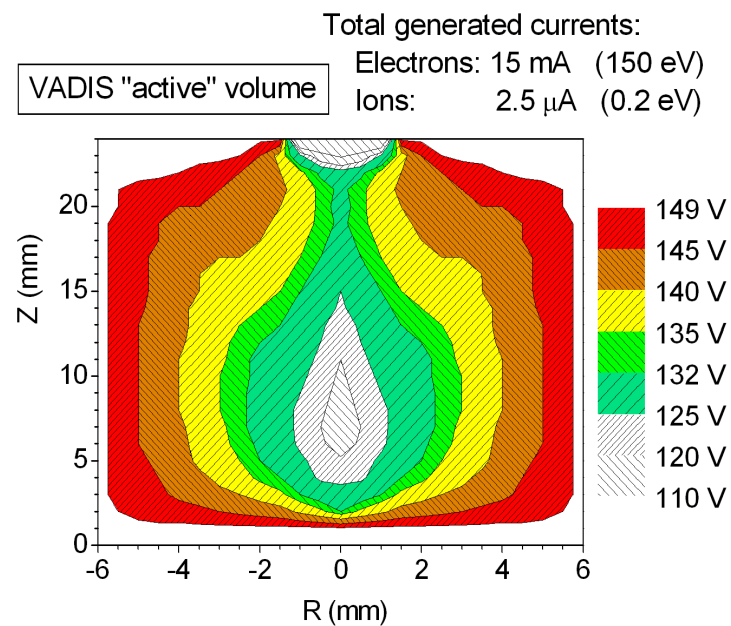
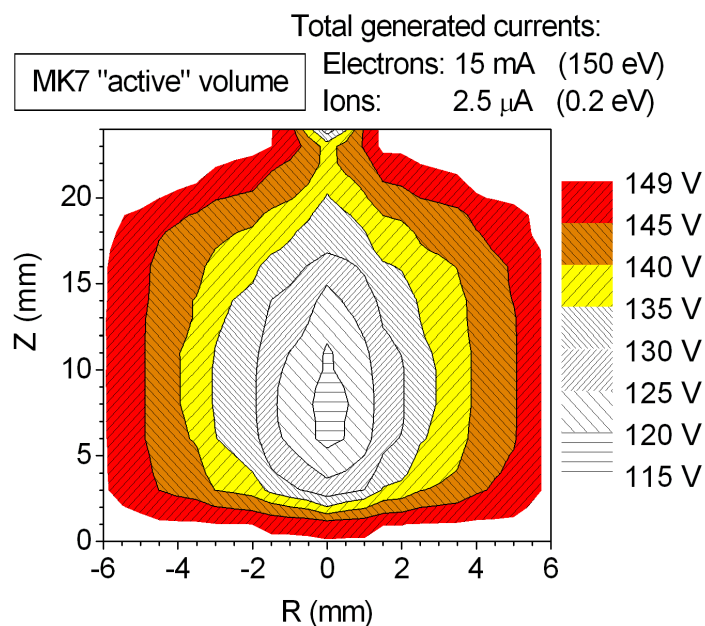


$$f \equiv \frac{V_{active}}{V_{total}}$$

- 1 electron passage;
- no ion trapping;
- $T_e = 150$  eV ( $e \cdot V_{anode}$ , initial energy);
- $T_i = 0.17$  eV (2300 K, thermal energy);
- $n_e$  = temperature dependent (cathode emission given by Richardson Dushman);
- $n_n$  = dep. on pressure,  $n_{n\_in}$ ,  $C_{out}$ .

# 1<sup>st</sup> prototype: simulation (2/2)

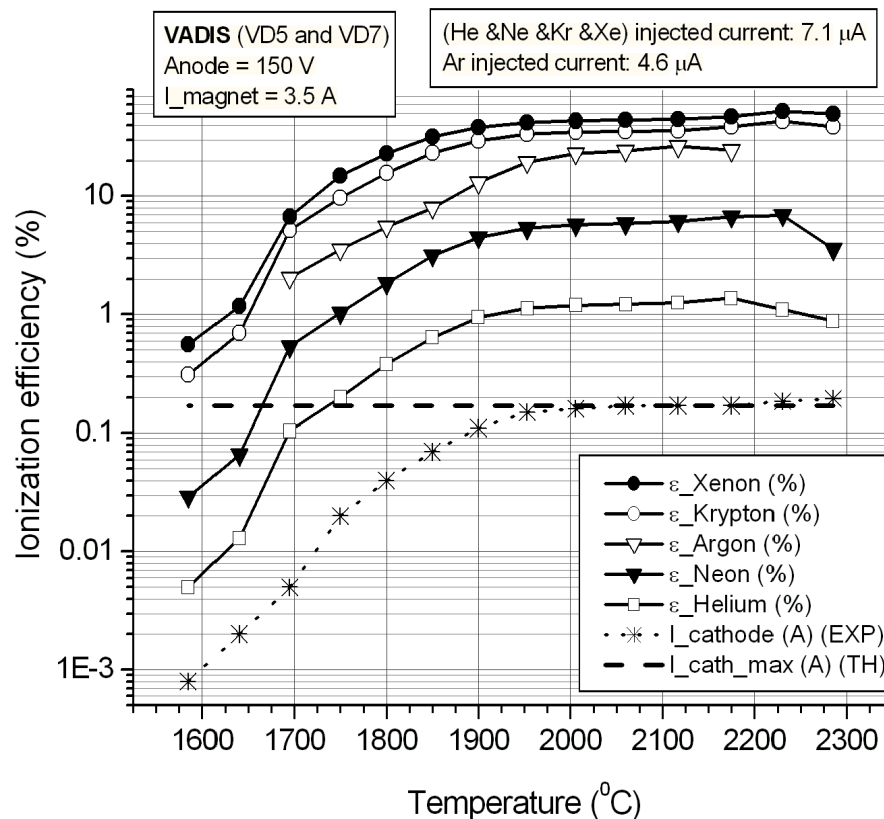
CPO simulation of the internal electrical field distribution



- With electrical charges (1 passage through the volume)
- Active volumes (in color): (132V; 149.8V) for MK7;  
(130V; 149.8V) for MK5

=> The difference in active volumes can justify the efficiency difference.

# 2<sup>nd</sup> prototype: results



=> Better efficiencies, with understood limitations.

# Outcome & perspectives

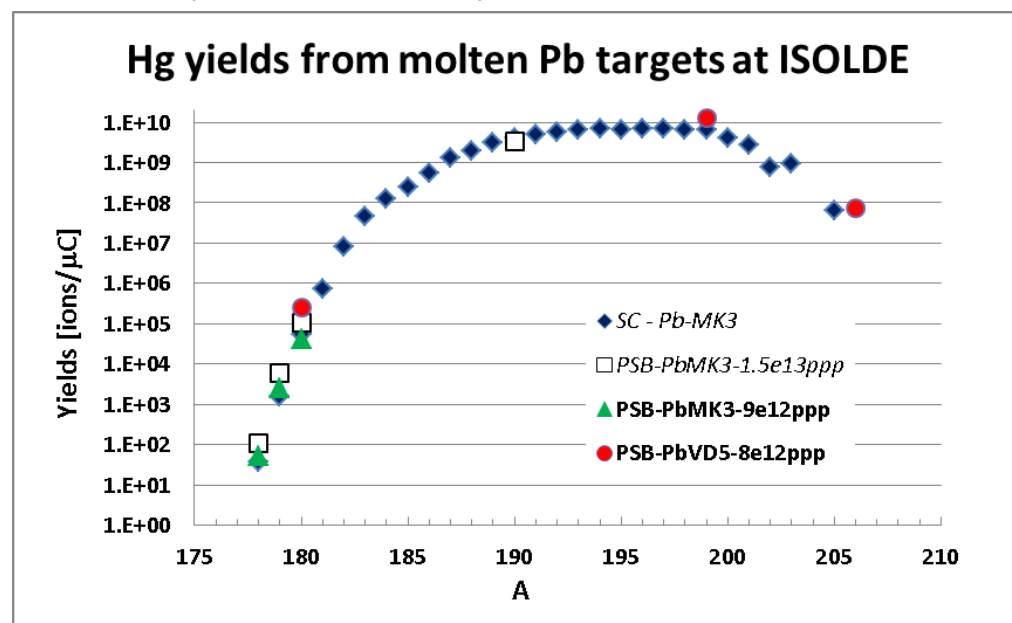
Novel VADIS ion sources

yields on noble gases: x5-10 vs previous figures

$^{229}\text{Rn}$ , D. Neidherr et al., Phys Rev Lett 102, 112501 (2009)

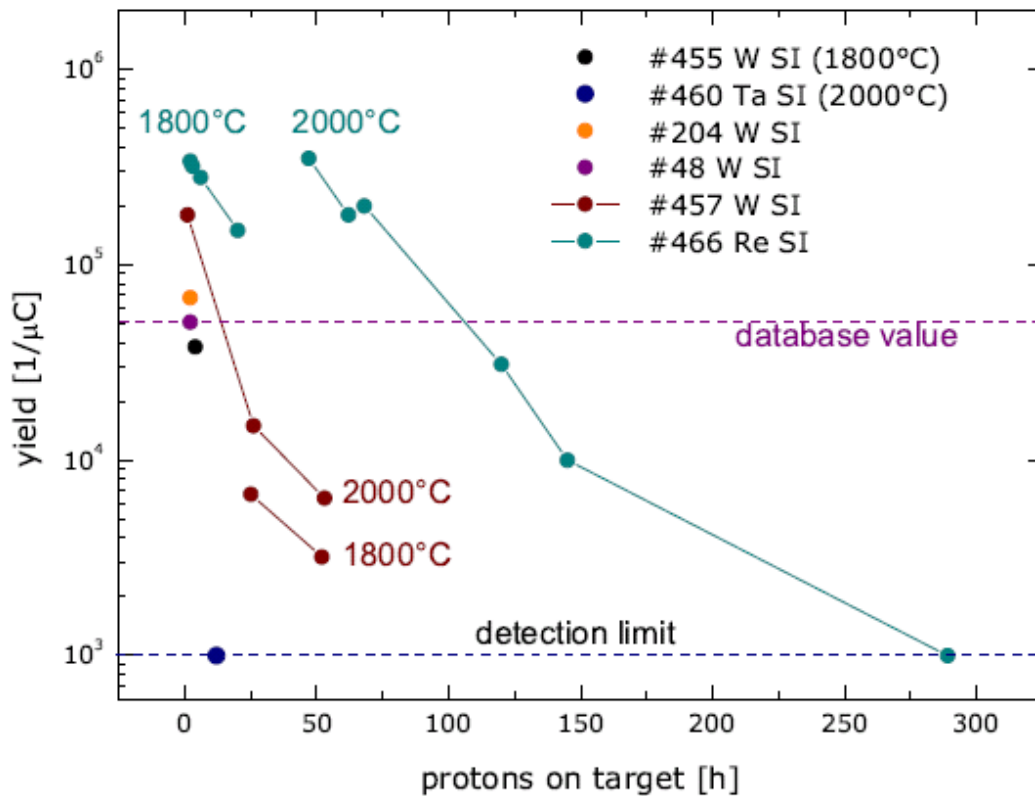
Other elements : improvements, eg Hg beams : x5

Ongoing tests of laser ion source in VADIS cavity for refractory elements

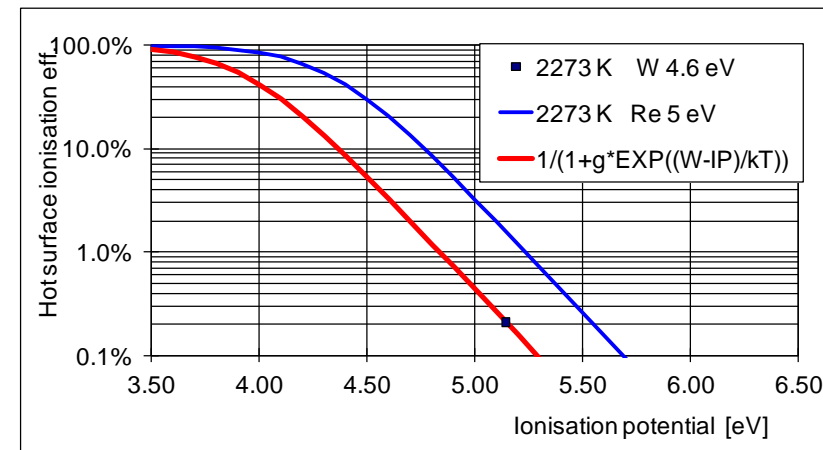




# Recover of historical $^{30}\text{Na}$ yields



Manufacturing of a bulk Rhenium Cavity as hot surface ionizer



Evolution of yields of  $^{30}\text{Na}$  from UCx targets over time under irradiation at ISOLDE

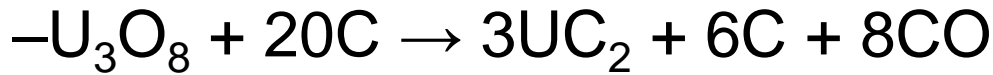
• Uranium Carbide Pills



• Target of  $\sim 50gcm^{-2}$



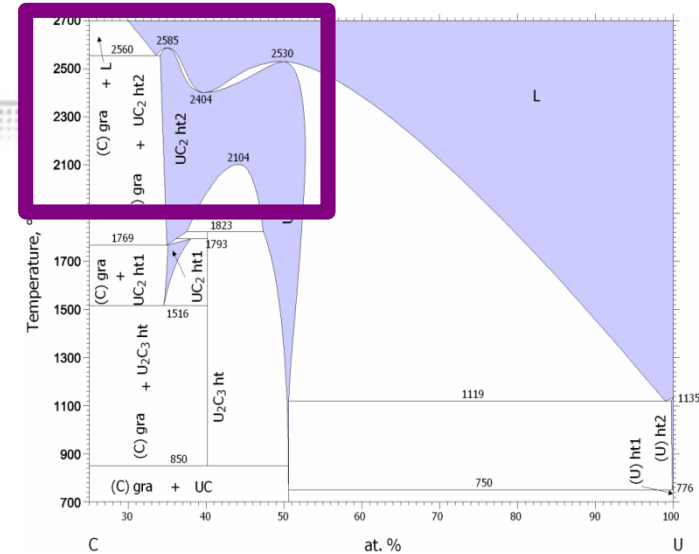
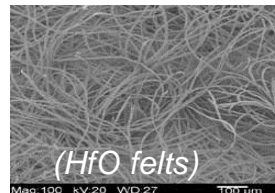
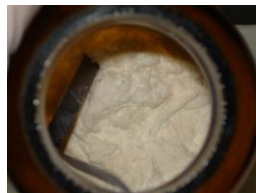
• Uranyl Nitrate  $UO_2(NO_3)_2$



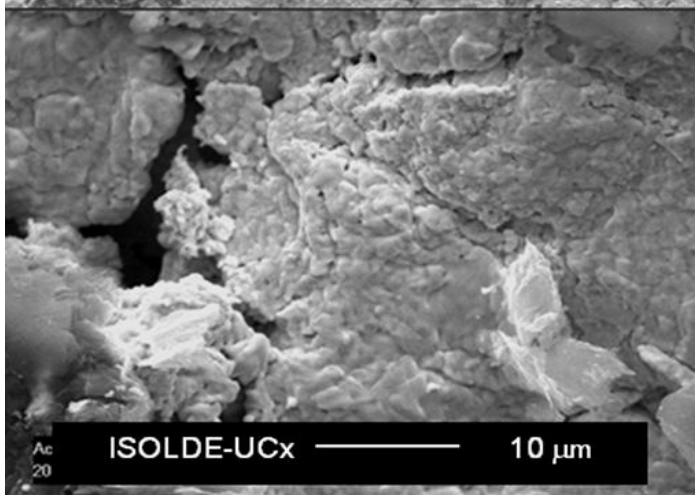
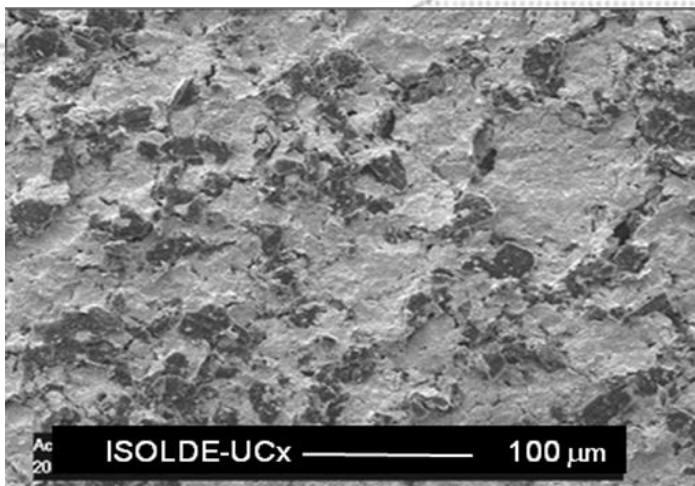
• Thorium Carbide



• Thoria felt

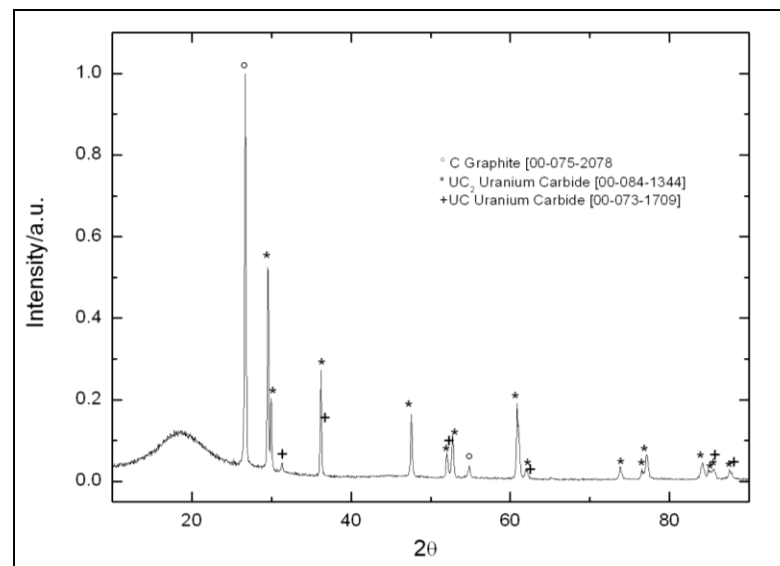


# Characterization UCx ISOLDE



$\rho_{\text{bulk}} = 3.5 \text{ g/cm}^3$

L. Biassetto et al.



How to do release modeling from this material ?

A Gottberg et al.  
FP7 ENSAR ActILab

# 1<sup>st</sup> Targets used at CERN-PS for alkali metals (p 10-24 GeV)

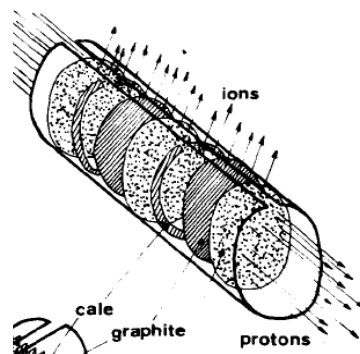
Target preparation:

5cm long, 6mm diameter.

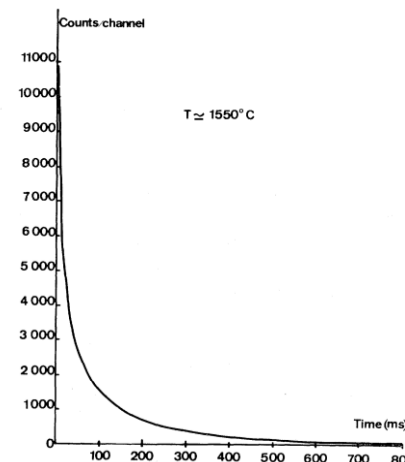
36x 70 $\mu$ m C, 1-10 $\mu$ m (1-8mg/cm<sup>2</sup>) U compound, 100 $\mu$ m gap: tot 0.3g/cm<sup>2</sup> U

Operated at ca 1500 $^{\circ}$ C

UO<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub>.6(H<sub>2</sub>O) layer, converted to UO<sub>3</sub> at 200 $^{\circ}$ C  
 Heated further to obtain U<sub>3</sub>O<sub>8</sub> / UC / UC<sub>2</sub> / oxycarbide



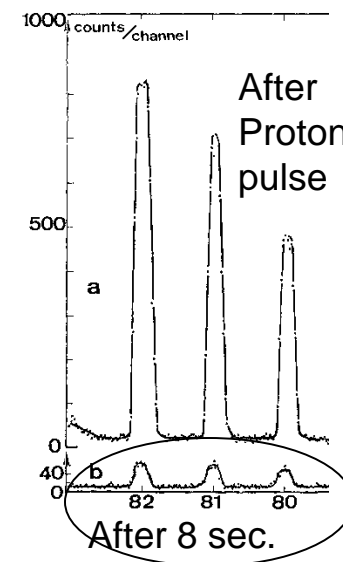
R. Klapish et al.  
 (UCx at CERN-PS&IPNO/CSNSM, 1967)



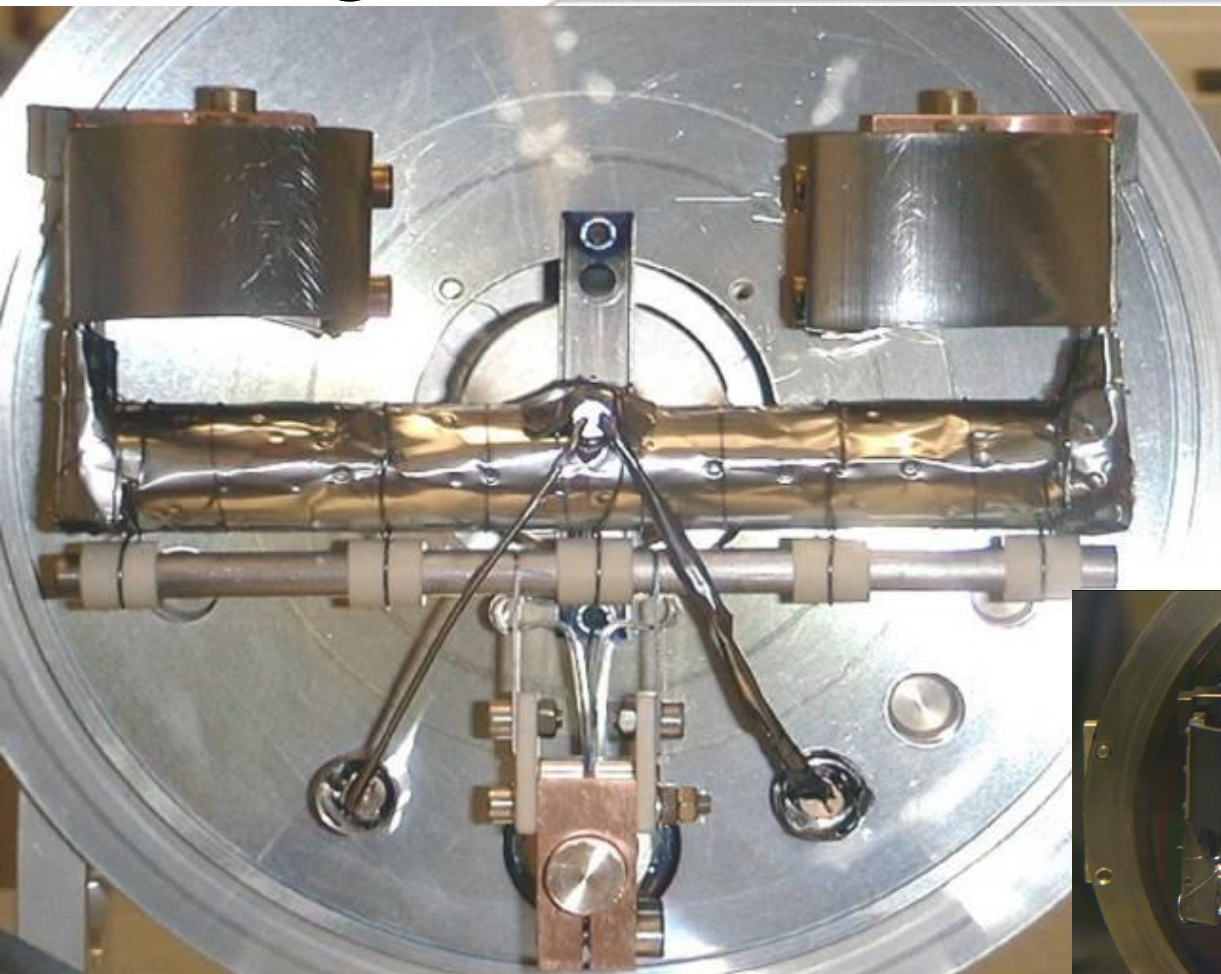
Na from Ir/C target

Fission  
 (10.5GeV p on ThCx)  
 Rb release

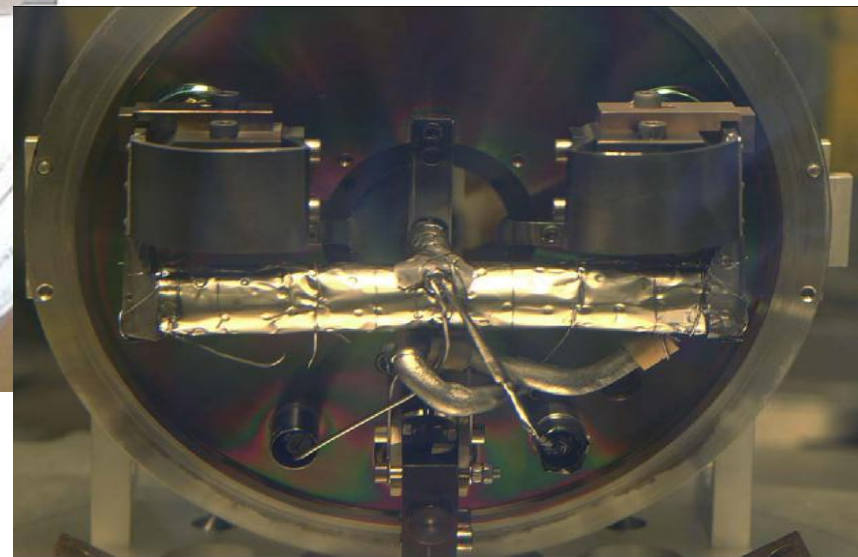
*Phys Rev Lett*, 1968



# Target-converter unit at ISOLDE



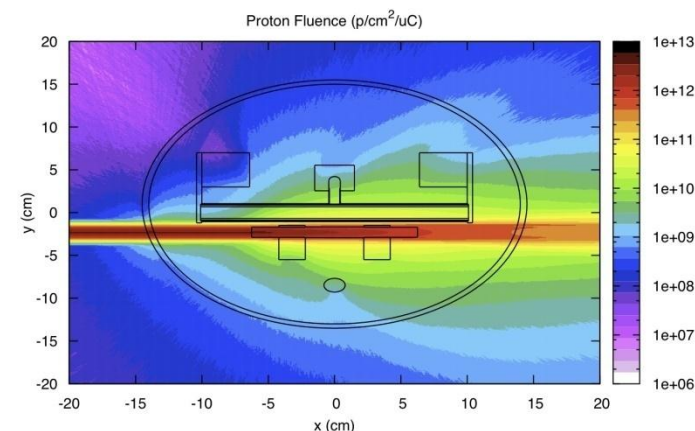
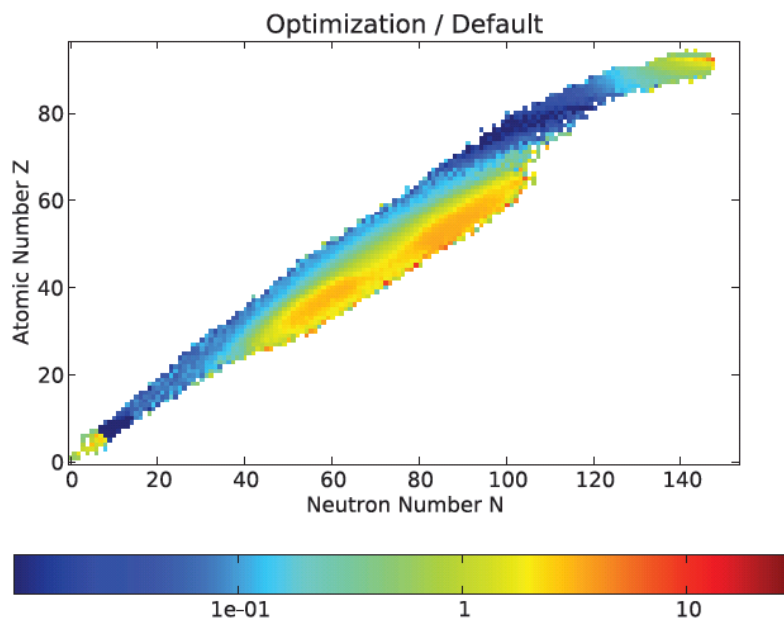
ISOLDE n-spallation source: Ta(W)-rod mounted below the UC target (before and after irradiation)



# Developments of ISOLDE n-converter

Improvement of fission yields (for ex.  $^{80}\text{Zn}$ ,  $^{130}\text{Cd}$ )

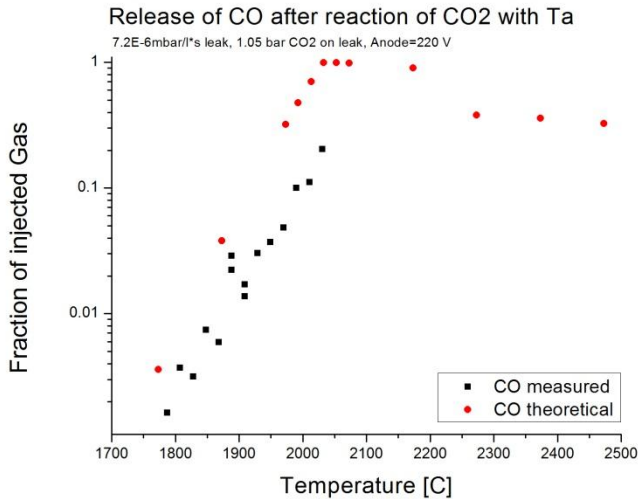
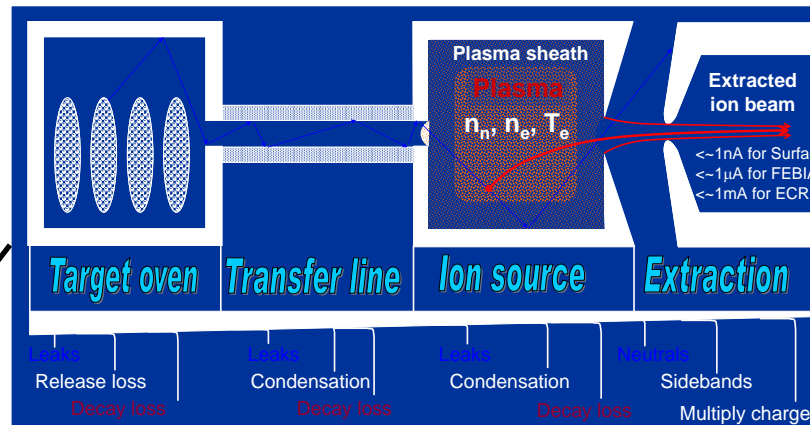
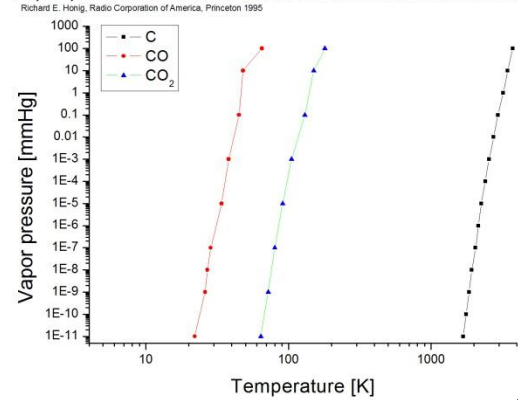
and further reduction of isobaric contaminants ( $^{80}\text{Rb}$ ,  $^{130}\text{Cs}$ )



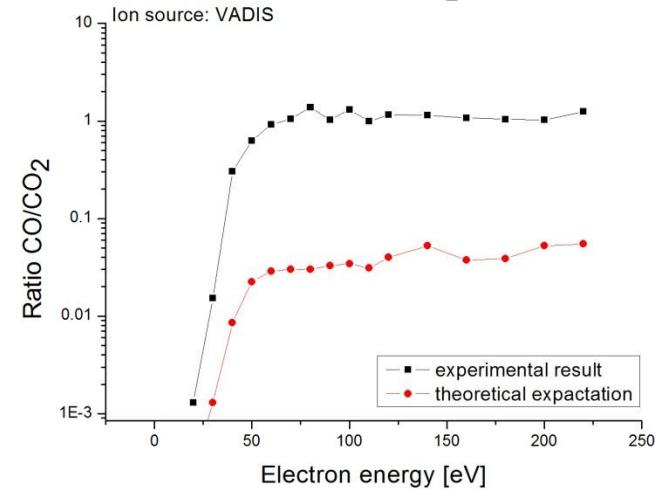
R. Luis et al.  
Eur. Phys. J. A  
Subm.

# C beams as $\text{CO}^+$ , $\text{CO}_2^+$

Vapor pressure of atomic Carbon and different Carbon molecules



Ratio of dissociated over ionized  $\text{CO}_2$  due to electron impact



C. Seiffert et al.

# 92 elements will be produced @ ISOLDE ?

Beam evolution in the past 5 years

17C as CO+  
Helicon  
Ion source

Group	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	
	1A	2A	3B	4B	5B	6B	7B	8B			1B	2B	3A	4A	5A	6A	7A	8A	
Period	X10 <sup>20,21</sup> Mg sub- $\mu$ m SiC								Ion source: + Surface - hot Plasma cool Laser			Purification of <sup>80</sup> Zn, <sup>130</sup> Cd with quartz ( $\Delta$ Hads)			9Be(n, $\alpha$ ) <sup>6</sup> He				
1	1 H	2 He																	
2	3 Li	4 Be																	
3	11 Na	12 Mg																	
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr	
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe	
6	55 Cs	56 Ba	* 71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn	
7	87 Fr	88 Ra	** 103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg								
* Lanthanides			* 57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb			
** Actinides			** 89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No			

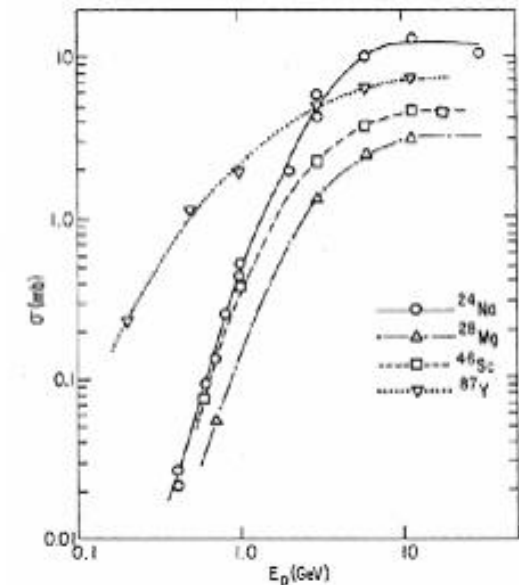
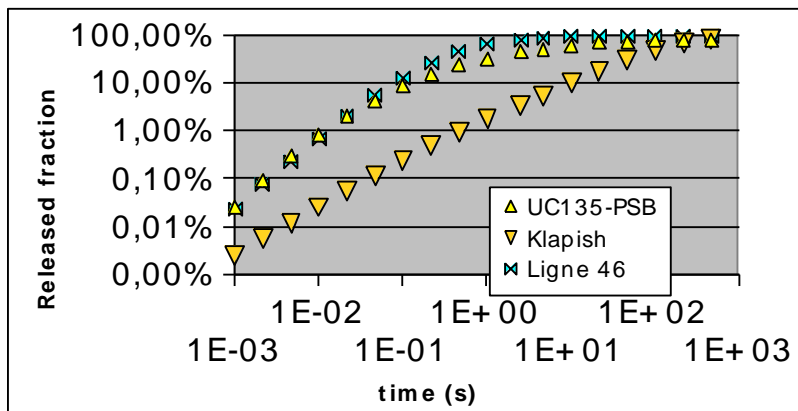
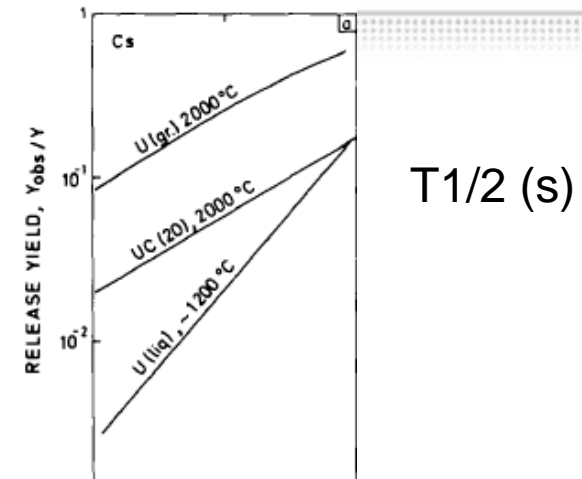
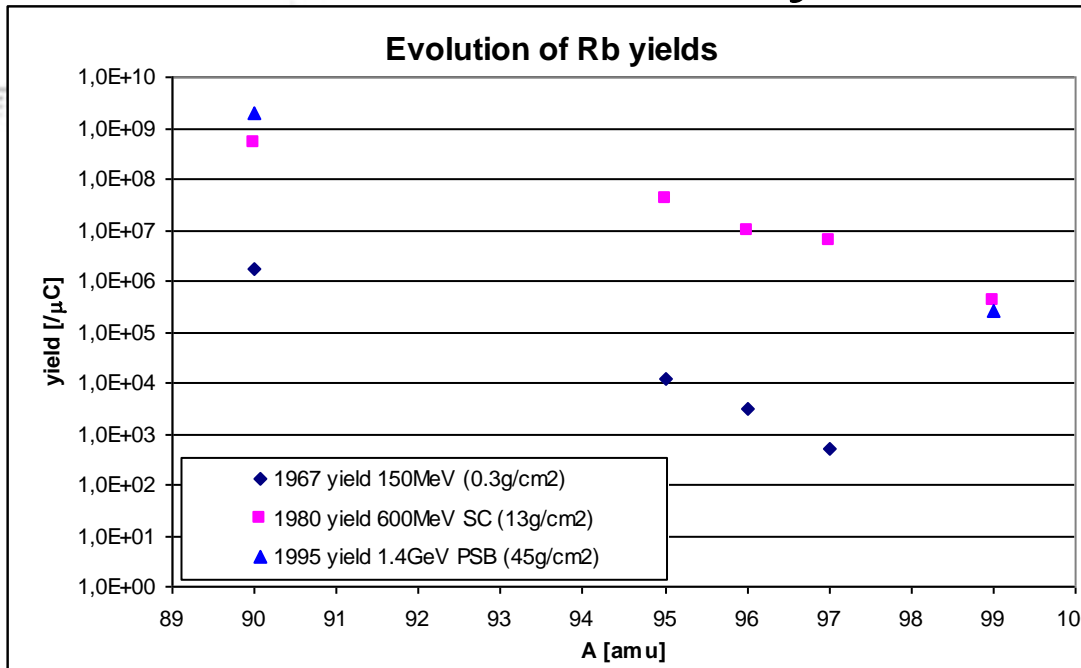
Purification of lanthanide beams :  
GdB<sub>6</sub> ion source cavity + RILIS

X3-10  
Ion source  
Nano  
CaO Y<sub>2</sub>O<sub>3</sub>

Au beams  
by laser ions. X5 Ion source



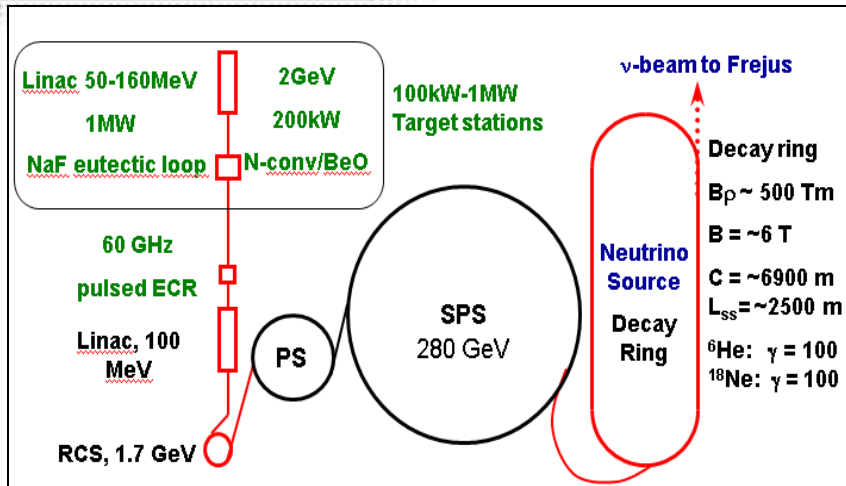
# Evolution of yields over years



**2 GeV, 10kW ideal for HIE-ISOLDE !!!**

# Ions for $\beta$ beams

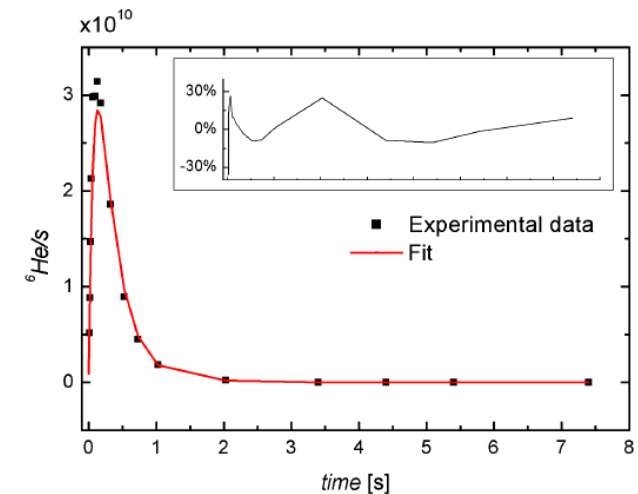
T. Mendonca



${}^9\text{Be} (n, \alpha) {}^6\text{He}$



$$p(t) \propto p_{eff}(t) \star p_{diff}(t)$$



Temperature [°C]	$t_{eff1}$ [ms]	$t_{eff2}$ [ms]	$t_{diff}$ [ms]	Release efficiency [%]	${}^6\text{He}$ production ( $N_0$ )
700	5.5	32	320	59	$2.7 \cdot 10^{10}$
800	5.6	28	150	71	$2.6 \cdot 10^{10}$
1000	4.7	28	1600	51	$4.1 \cdot 10^{10}$
1130	3.3	27	190	79	$3.1 \cdot 10^{10}$
1400	1.8	24	270	82	$2.9 \cdot 10^{10}$

# Team & Collaborations

- Dr. M. Kronberger : ion sources
- C. Seiffert : molecule evaporation
  
- R. Luis : neutronics (ITN, Lisboa)
  
- Dr. A. Gottberg : target materials, incl. Uranium (ENSAR-FP7, ActiLab).
- J. P. Ramos : Target nanomaterials (Univ. Aveiro)
- M. Czapski : material analysis support (CATHI ITN Marie-Curie program)
  
- Dr. T. Mendonca : High power targetry for neutrino physics
- S. Cimino : High power targetry

GANIL, IPNO, INFN, PSI (Uranium, ENSAR “ActiLab”) + ORNL, TRIUMF  
ITN, PSI (neutronics)  
EPFL, Aveiro, ITN (materials)  
ESS, CEA, SCKCEN-Myrrha, SINP (high power targetry)

Selective adsorption to trap impurities : isothermal chromatography

PhD of E. Bouquerel and related publications

Improved isotope release from porous high T nano(sub- $\mu\text{m}$ ) ceramics

PhD of S. Fernandes and related publications (+ CERN patent)

New plasma ion source VADIS

PhD L. Penescu, discovery of  $^{229}\text{Rn}$ , related publications

High power targetry

EURISOL and beta beams