

# R&D of water-based liquid scintillator as a reactor anti-neutrino detector

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and

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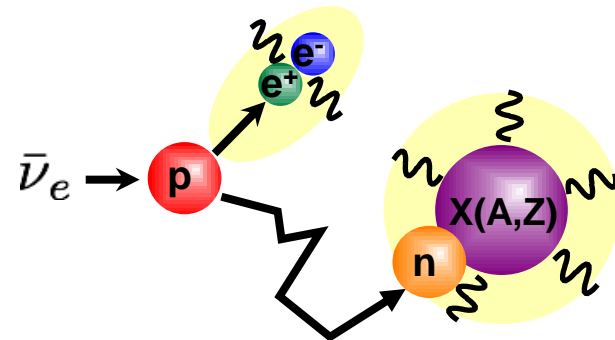
# Why water-based ?

Neutrino energy measurement is important for

- neutrino oscillation experiments (currently organic solvent scintillators are used in DoubleChooz, DayaBay, and RENO).
- **reactor monitor requested by IAEA (International Atomic Energy Agency) as one of the safeguards.**

**Requirements for a reactor monitor:**

1. nonflammable and nonvolatile for the use near a reactor  
→ **water-based scintillator**
2. Gd is soluble to identify neutrino interactions by delayed coincidence measurement.



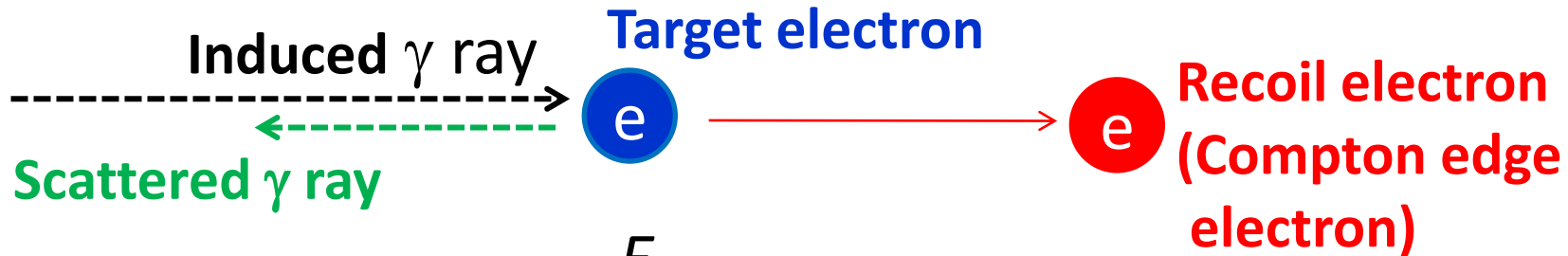
# Principle for R&D

**We try to dissolve following things into the water:**

- 1. Luminescent agent (PPO) with a surfactant**
2. Aromatic molecules with hydrophilic group as a material taking energy transmission, and luminescent agent (PPO)
3. Water-soluble luminescent agent

# Light Yield Measurement by $^{60}\text{Co}$

We use the Compton edge energy  
(maximum energy of recoil electrons) .



$$E_e^{max} = E_\gamma - \frac{E_\gamma}{1 + \frac{2E_\gamma}{m_e c^2}}$$

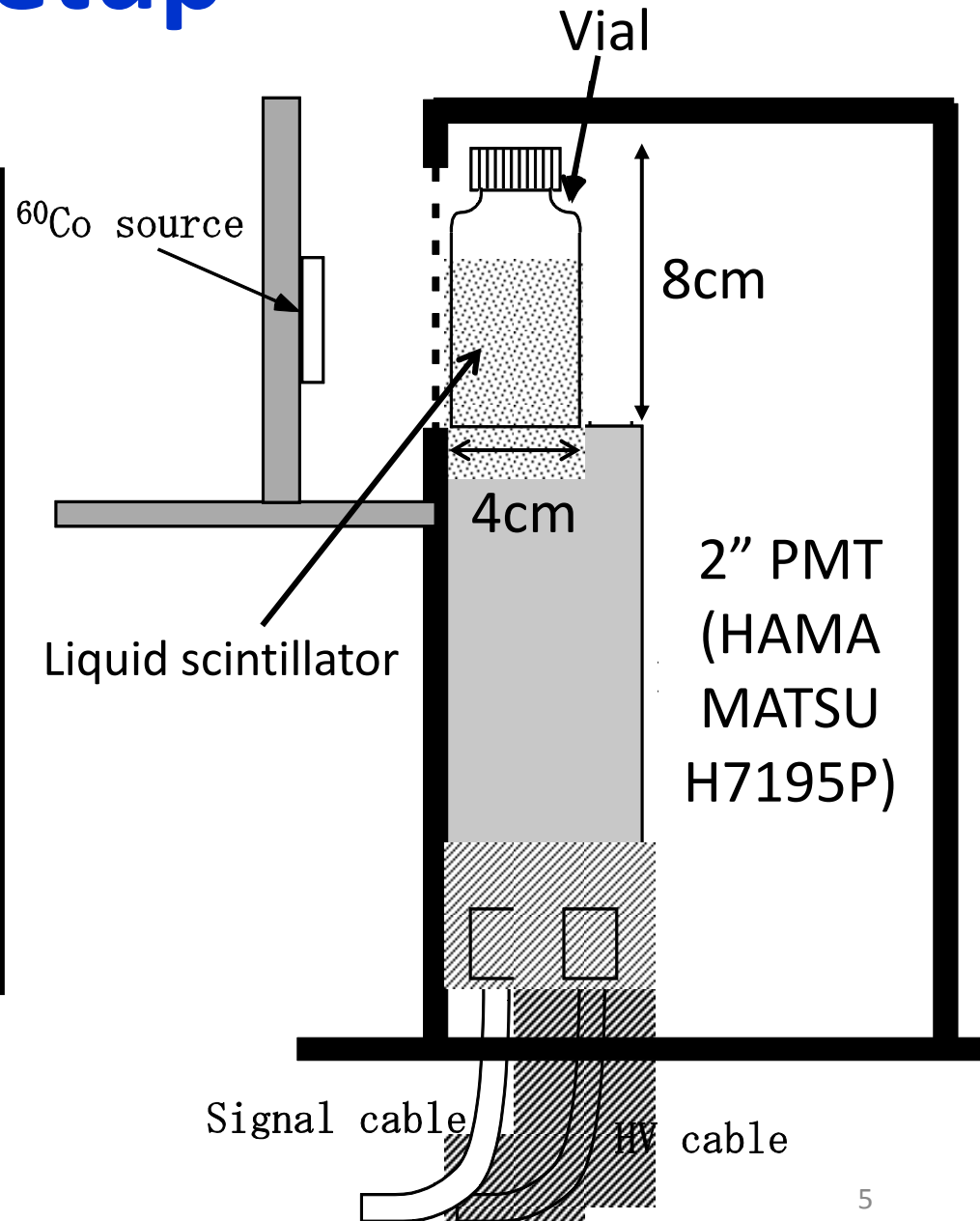
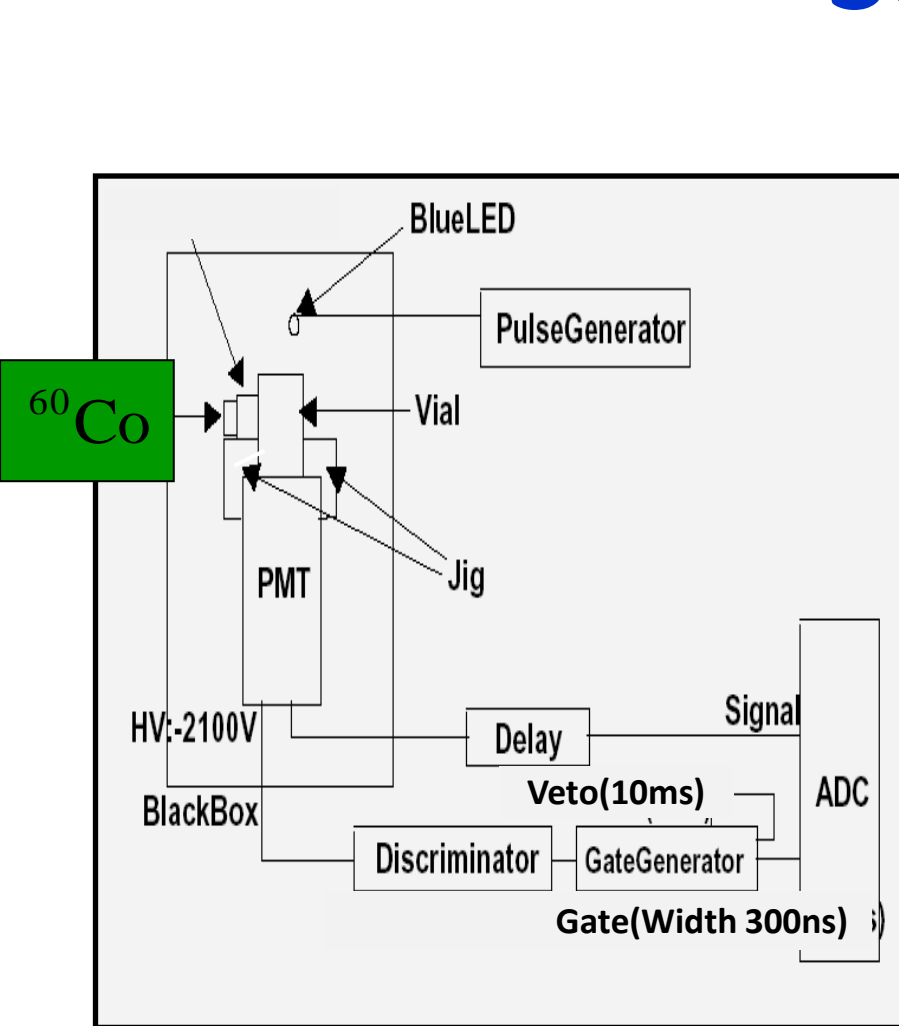
$E_e^{max}$  : maximum energy of the Compton electron

$E_\gamma$  : energy of induced  $\gamma$  ray

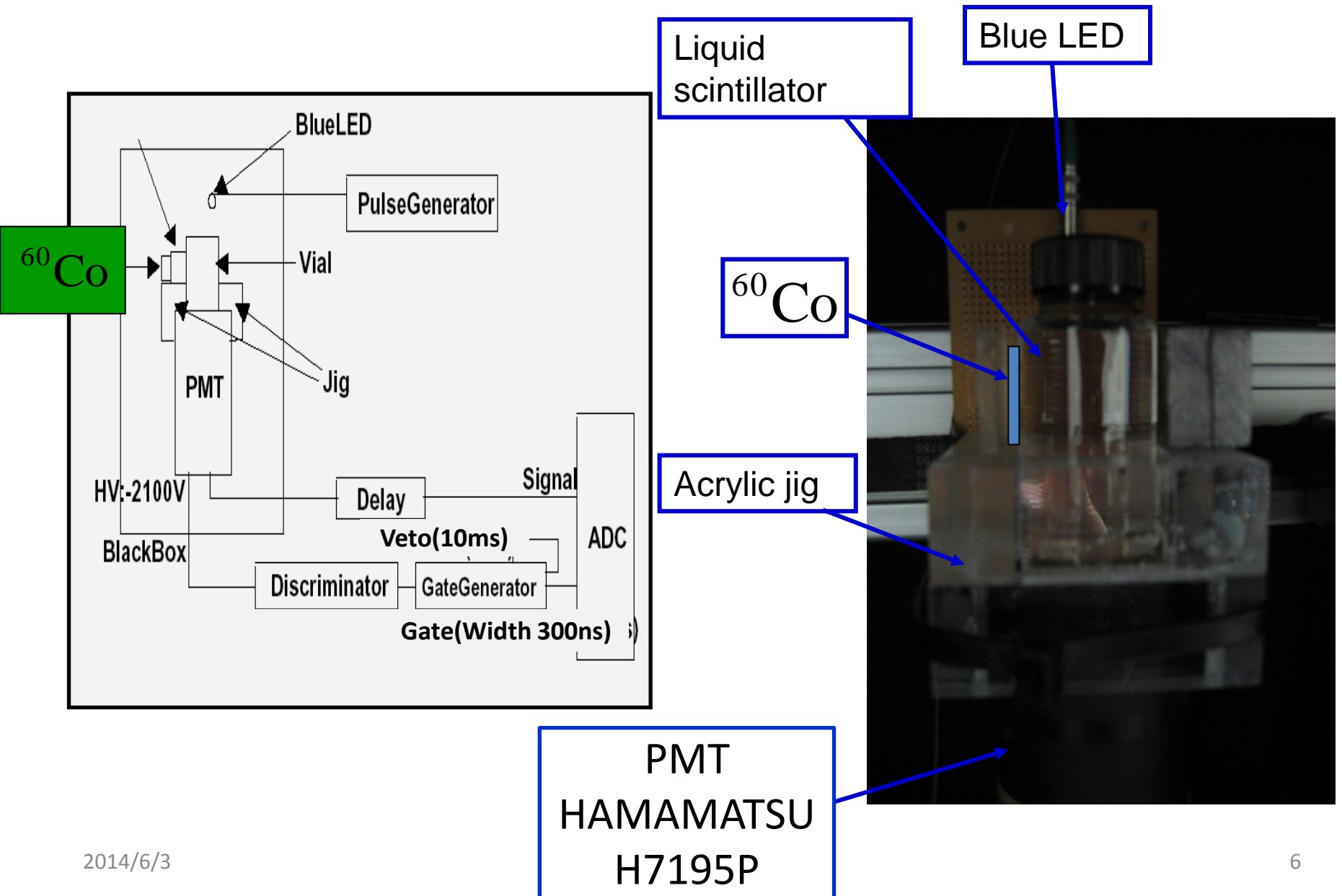
$m_e c^2$  : rest mass energy of electron

- $^{60}\text{Co}$
- 1.17MeV  $\gamma$  ray  $\rightarrow$  0.96MeV
  - 1.33MeV  $\gamma$  ray  $\rightarrow$  1.12MeV

# Setup

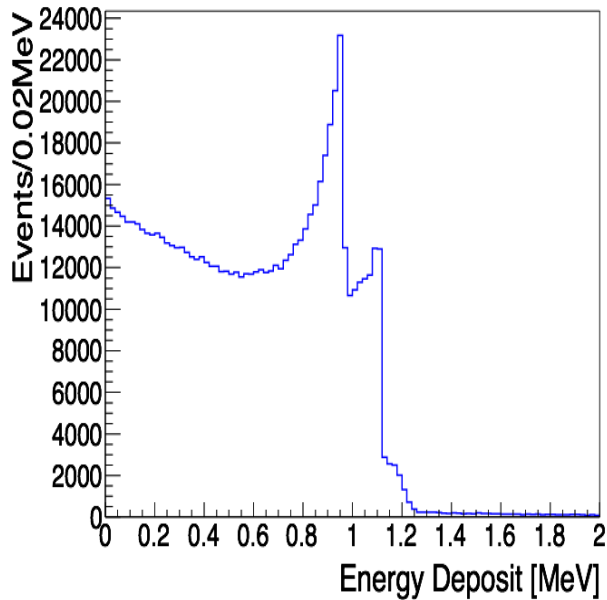


# Setup



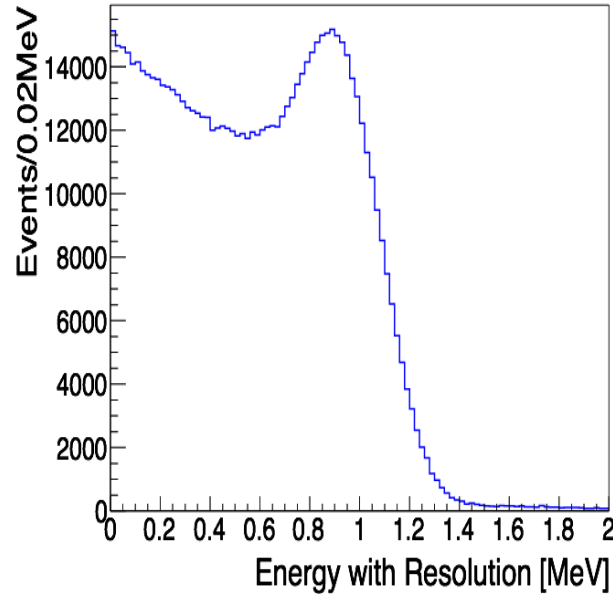
# Energy Deposit by $^{60}\text{Co}$ irradiation

## Energy distribution



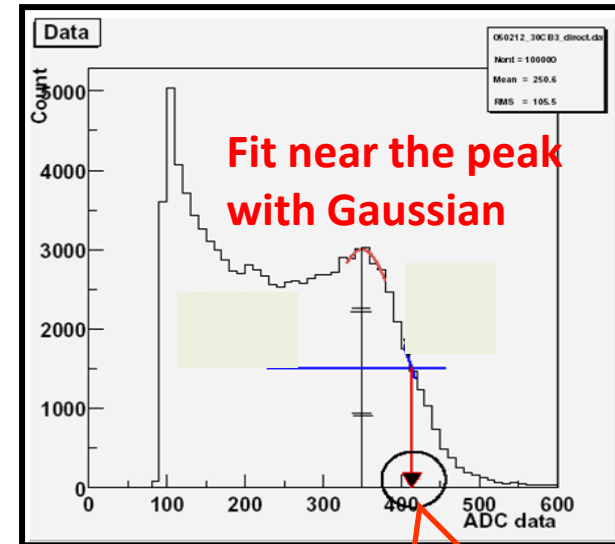
Compton edges of  
1.17MeV and 1.33MeV  
 $\gamma$  rays

## Simulation



In the case of energy  
resolution 10%/ $\sqrt{E}$ [MeV]

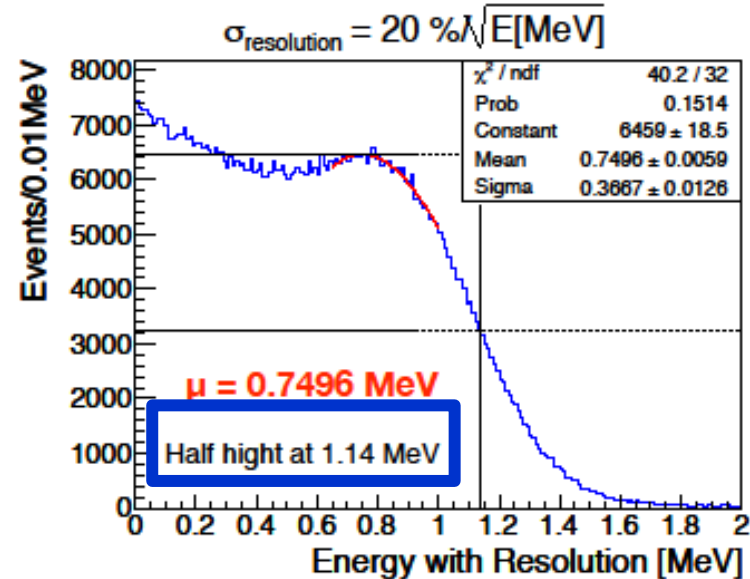
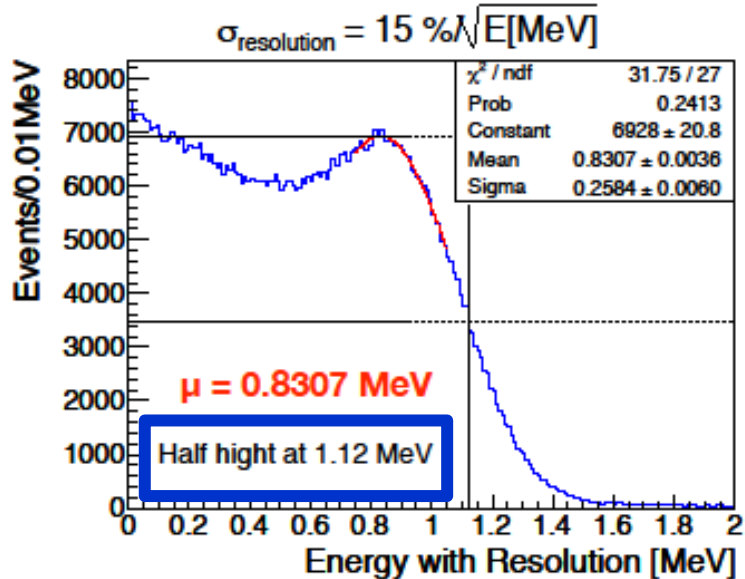
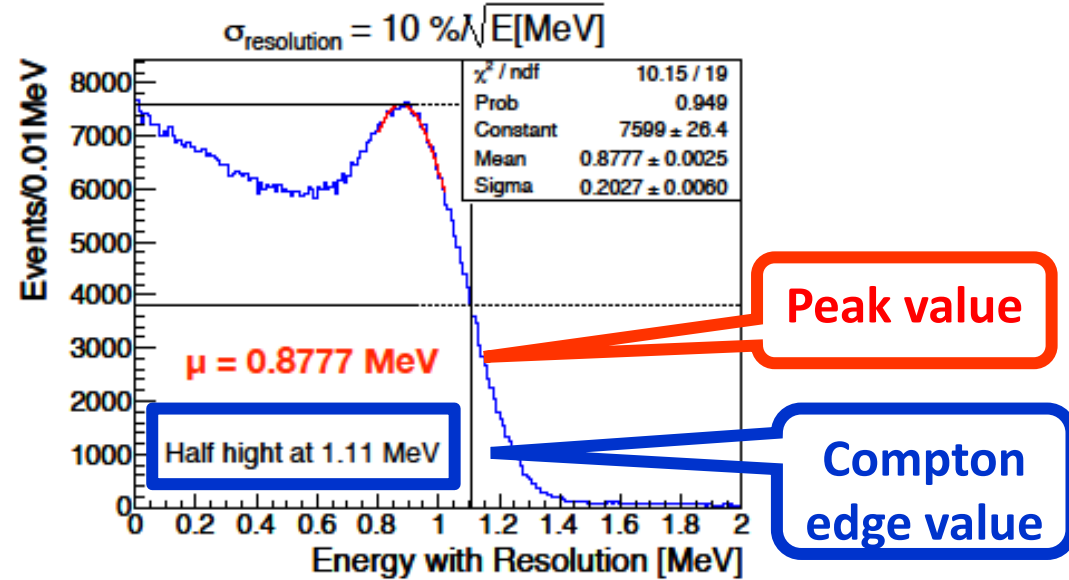
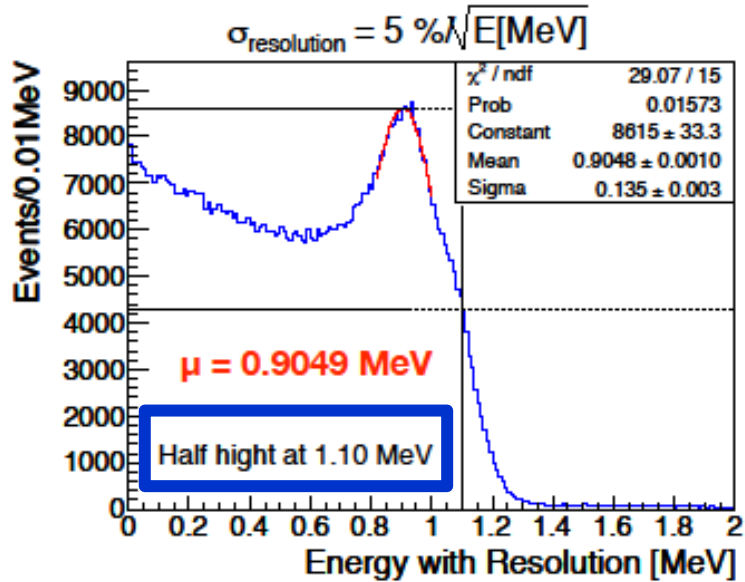
## Measurement



Compton edge !

ADC value at the half  
maximum

# Compton edge value as a function of energy resolution



**Compton edge is constant !**



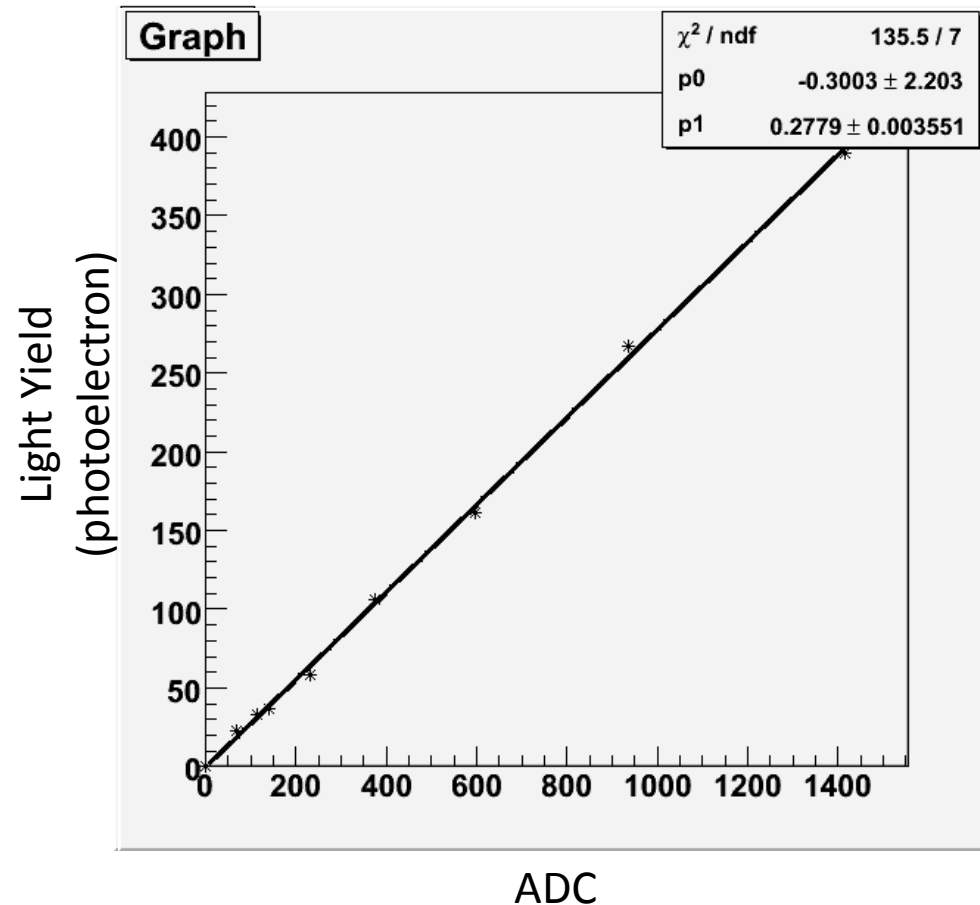
# LED calibration (ADC → light yield)

- Suppose the light yield follows Poisson distribution
- Calculate the light yield in photoelectrons from the mean and standard deviation of LED light yield distribution using the following formula:

$$\text{Light Yield} = \left( \frac{\text{Mean}}{\text{Sigma}} \right)^2 [\text{pe}]$$

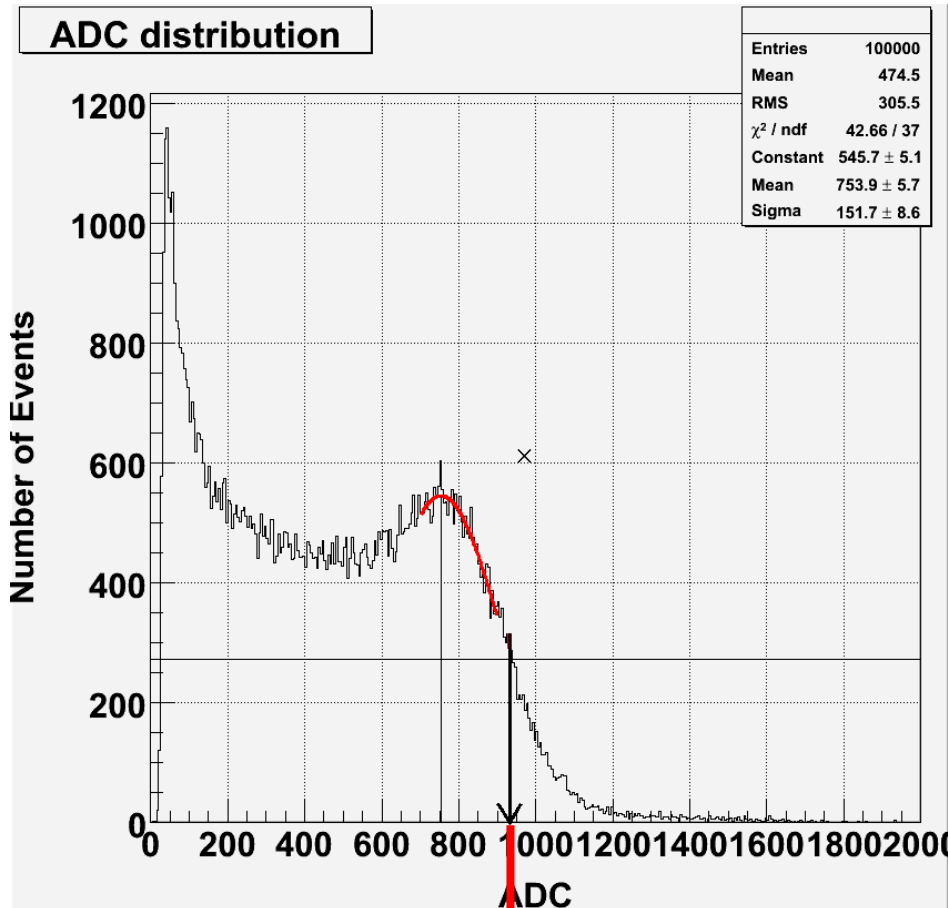
- Obtain the relation between light yield and ADC value like the right figure varying the brightness of LED.

Light Yield [pe] vs ADC



Inclination : 0.278

# Light yield of the benchmark scintillator (pseudocumene + PPO (3g/l))

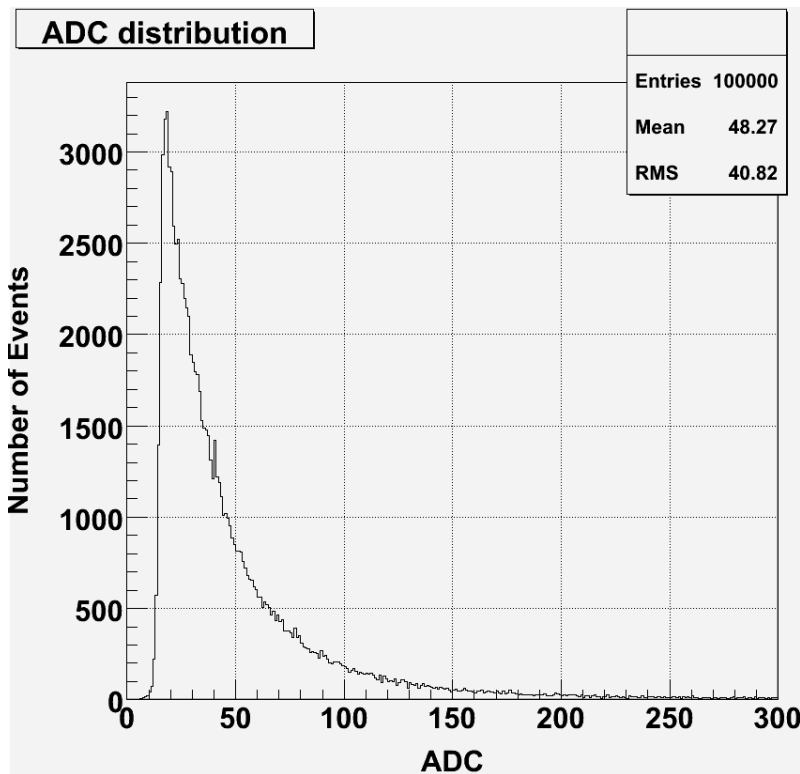


ADC value=932

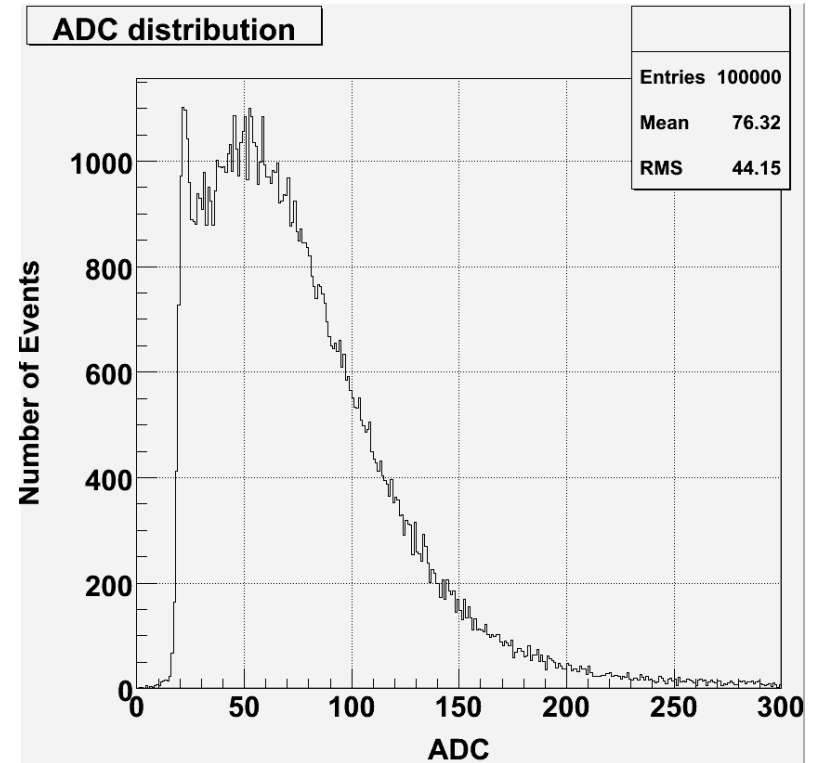
Light yield= $0.278 \times 932 = 259$  [pe]

# Light yield measurement of water-based scintillator

**Sodium dodecyl sulfite (SDS)** is a hopeful surfactant.



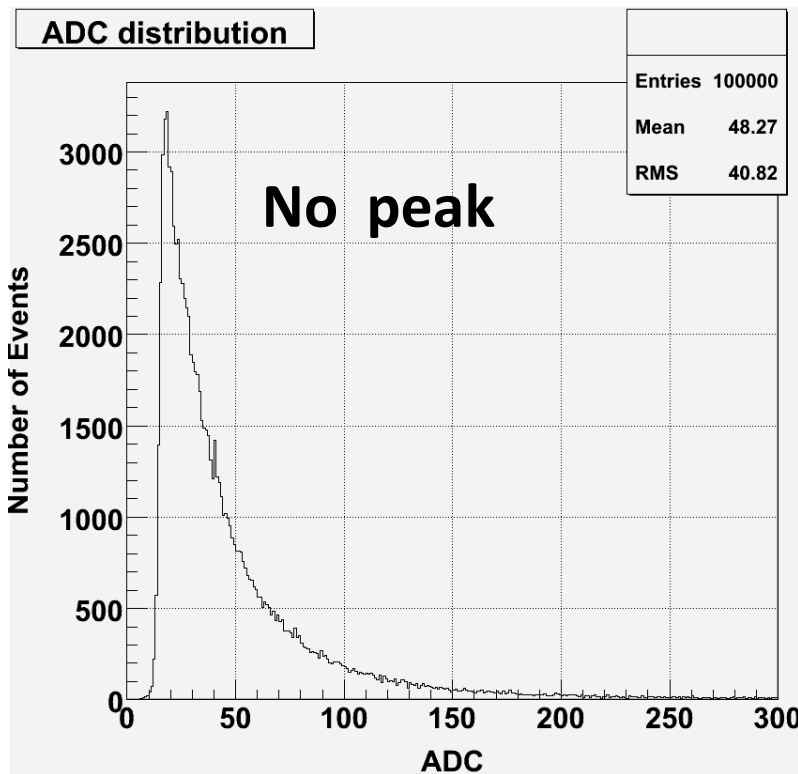
**30% SDS + 70% H<sub>2</sub>O**



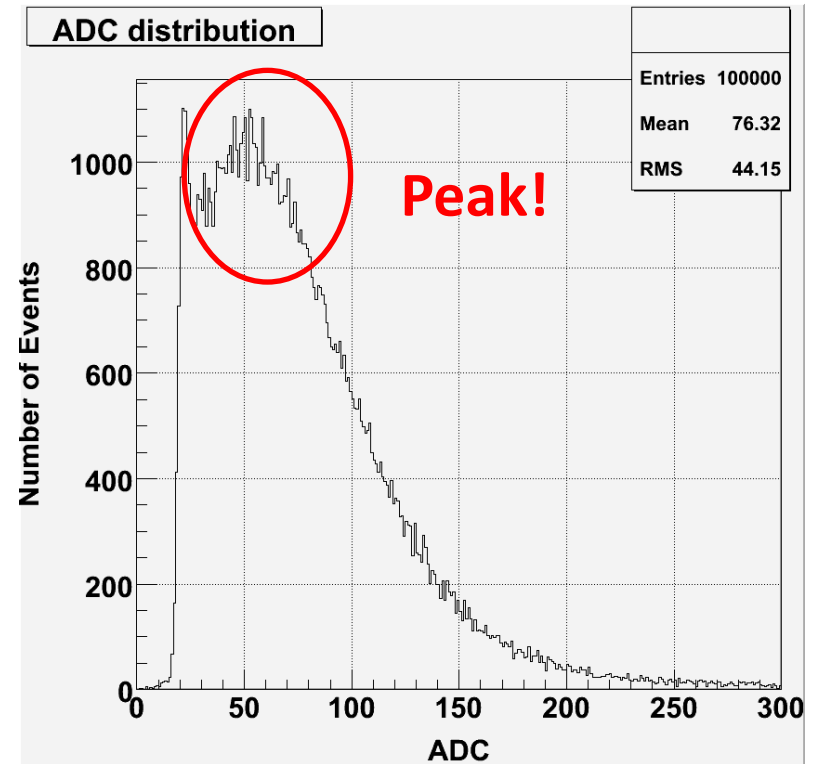
**30% SDS + 70% H<sub>2</sub>O + 30g/l PPO**

# Light yield measurement of water-based scintillator

**Sodium dodecyl sulfite (SDS)** is a hopeful surfactant.

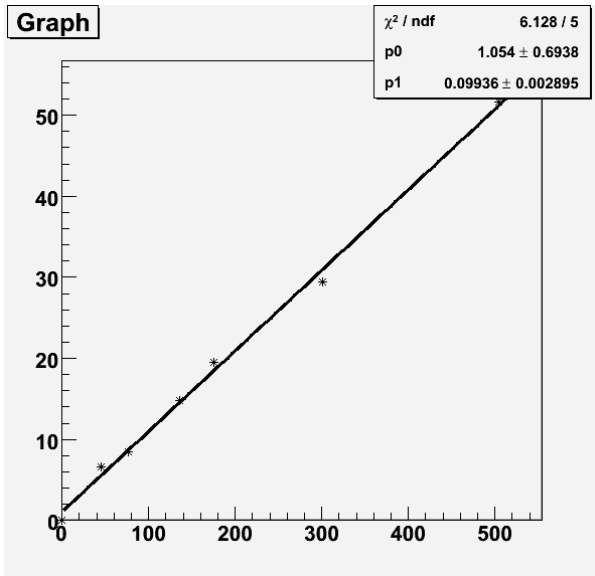


**30% SDS + 70% H<sub>2</sub>O**

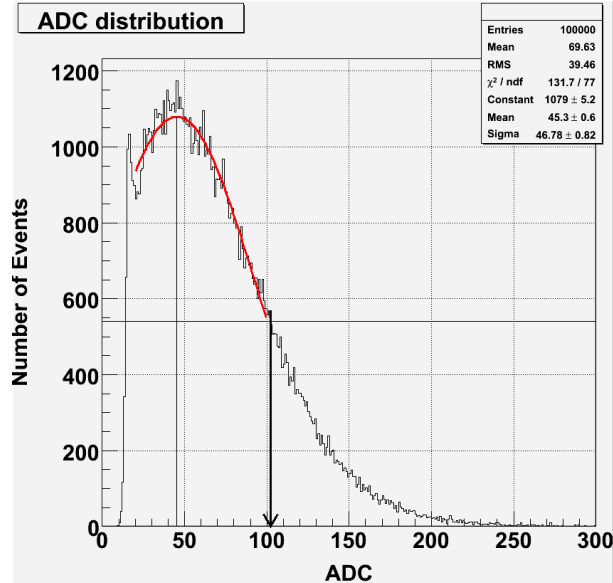


**30% SDS + 70% H<sub>2</sub>O + 30g/l PPO**

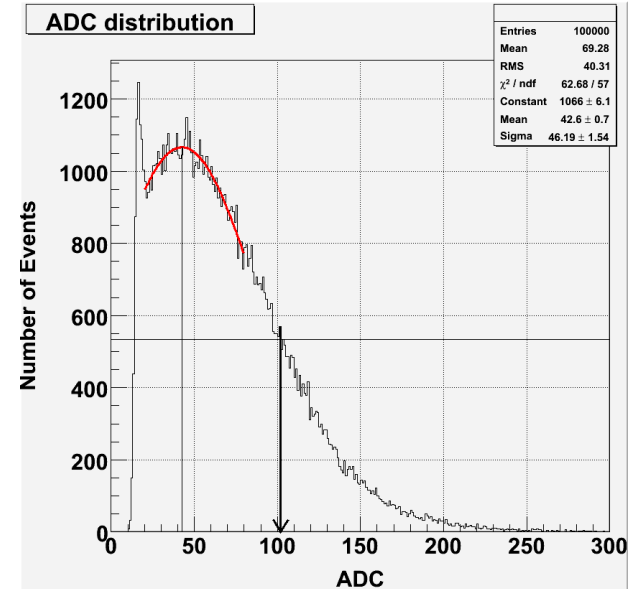
# Light yield measurement: 30%SDS+70%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)



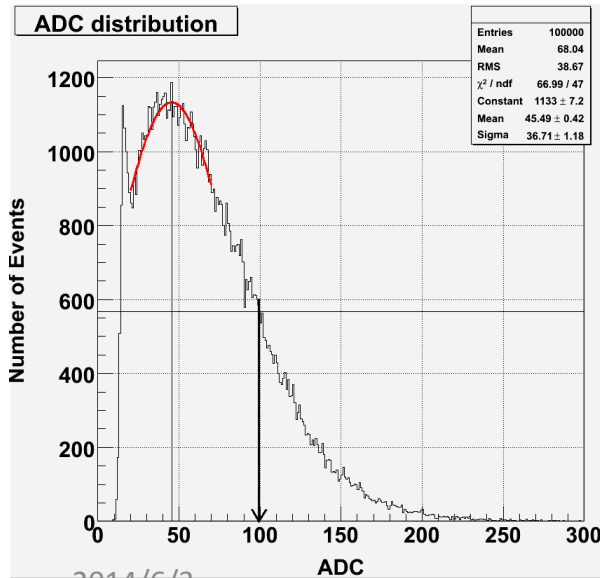
LED calibration  
Inclination: 0.099



10.2[pe]

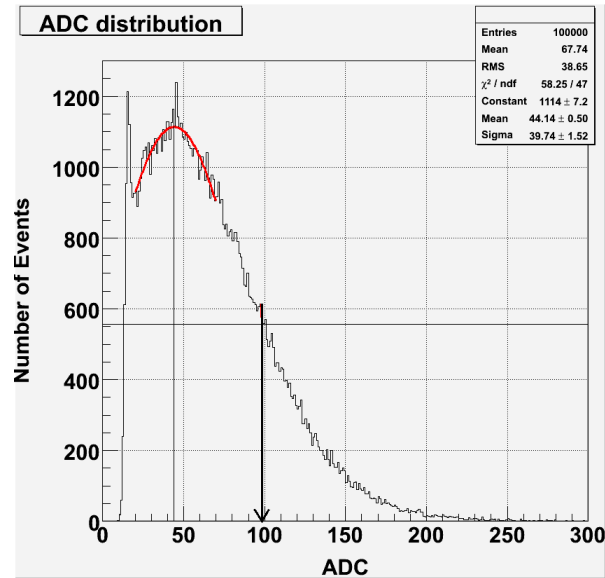


10.1[pe]

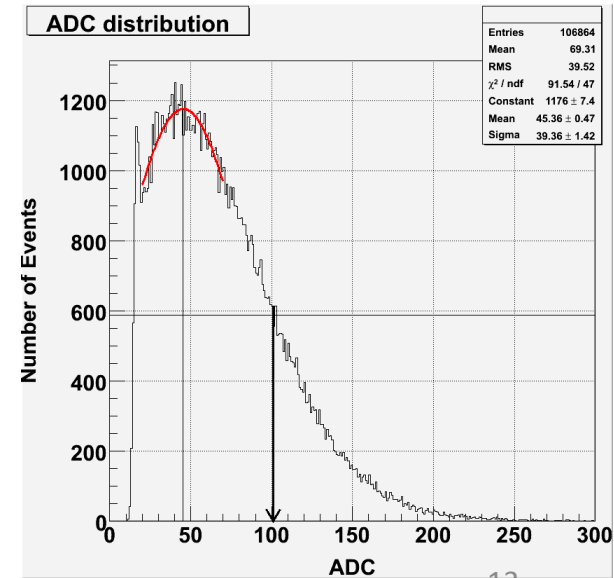


2014/6/3

9.88[pe]

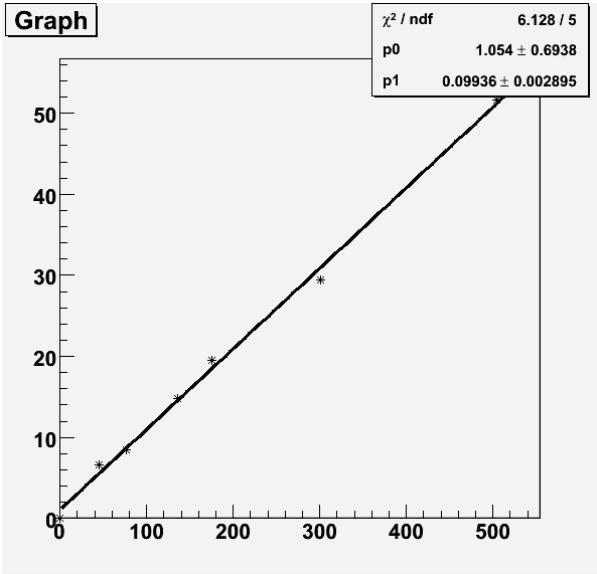


9.76[pe]

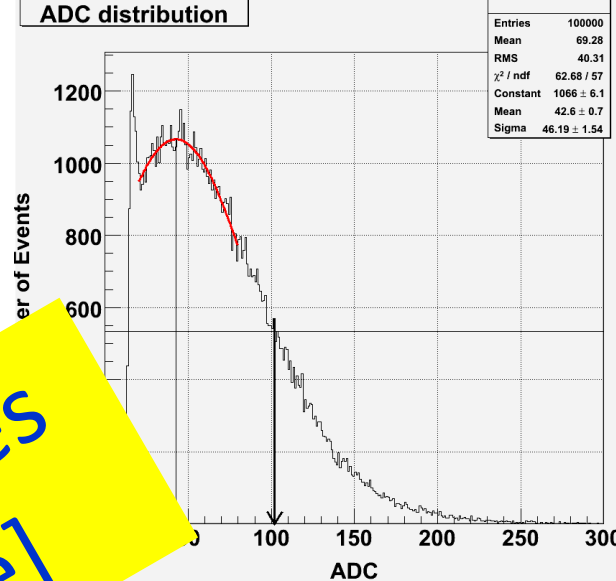
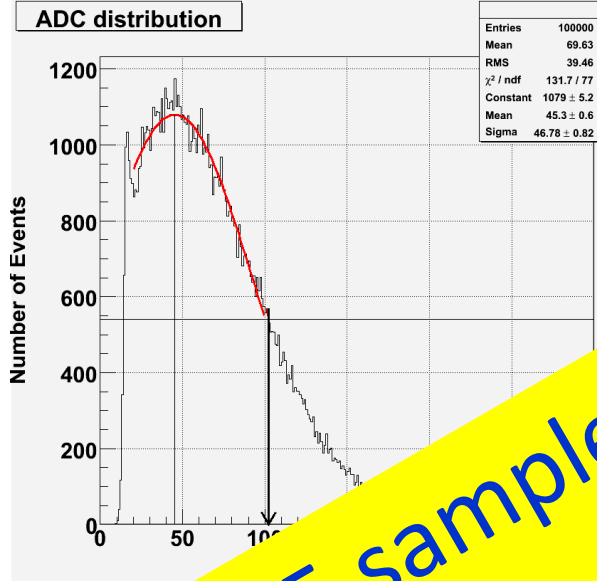


10.0[pe]

# Light yield measurement: 30%SDS+70%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)

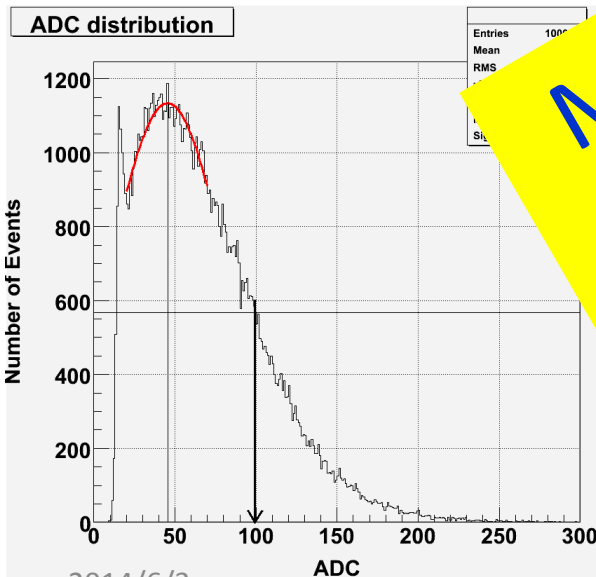


LED calibration  
Inclination: 0.099



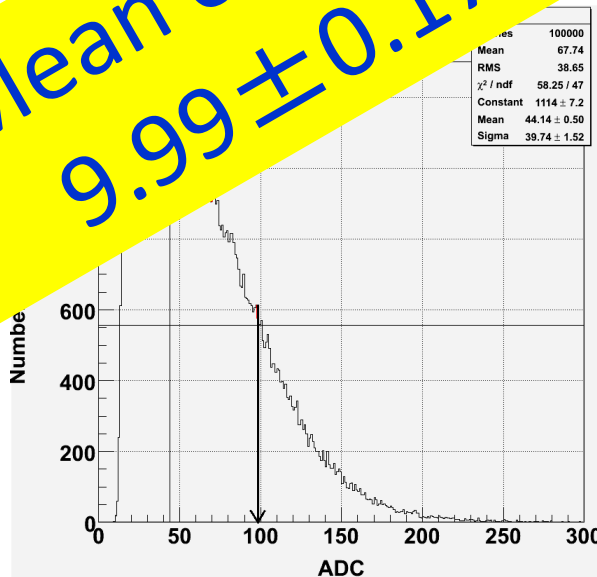
10.1[pe]

Mean of 5 samples  
9.99 ± 0.17 [pe]

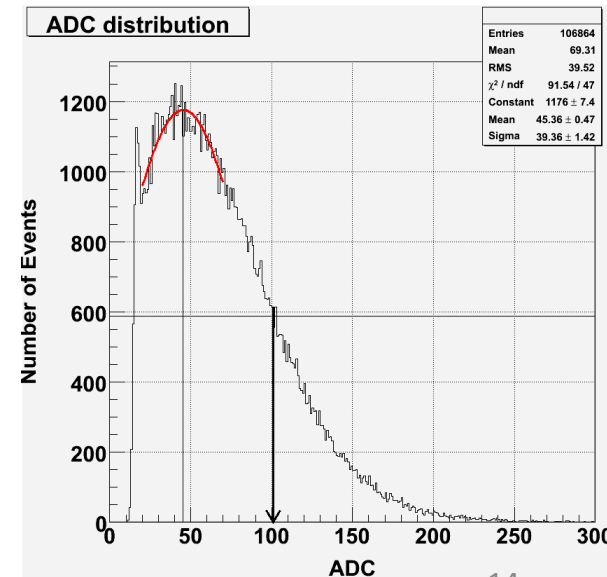


2014/6/3

9.88[pe]



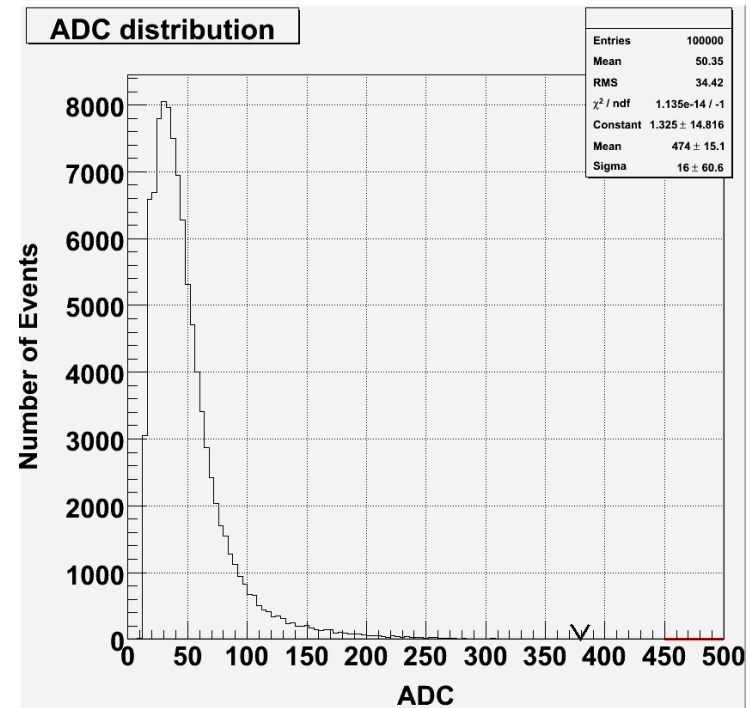
9.76[pe]



10.0[pe]

# Aging measurement of 30%SDS+70%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)

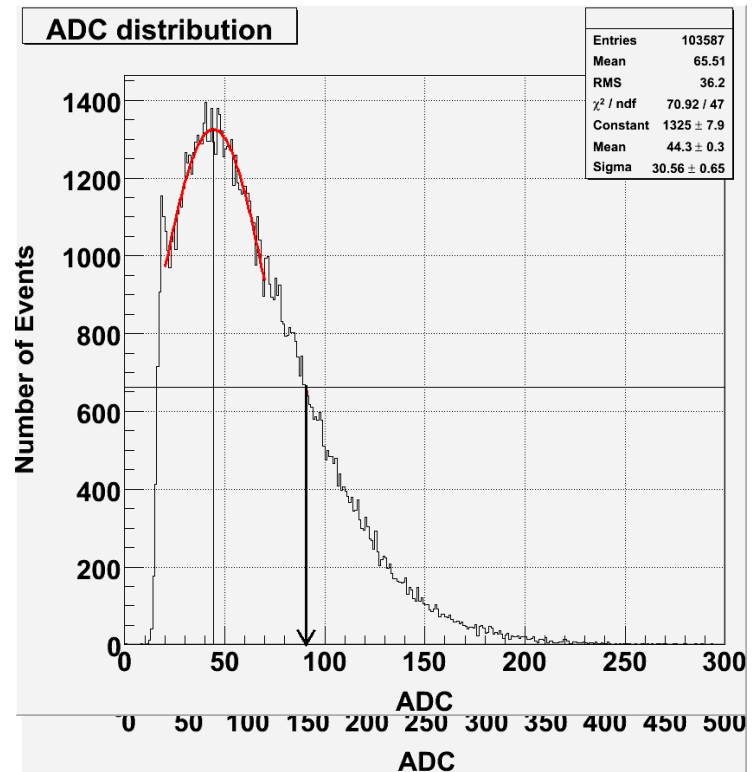
- Crystal formation  
after 1 year



No significant  
light yield

# Aging measurement of 30%SDS+70%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)

- It easily dissolved by shaking



9.18[pe]

(11.6 [pe] 1 year ago)



# Next steps

(1) We tried a **liquid detergent** (35% fatty acid potassium+65% water) as a surfactant.

- Commercially available
- Safe and ecological

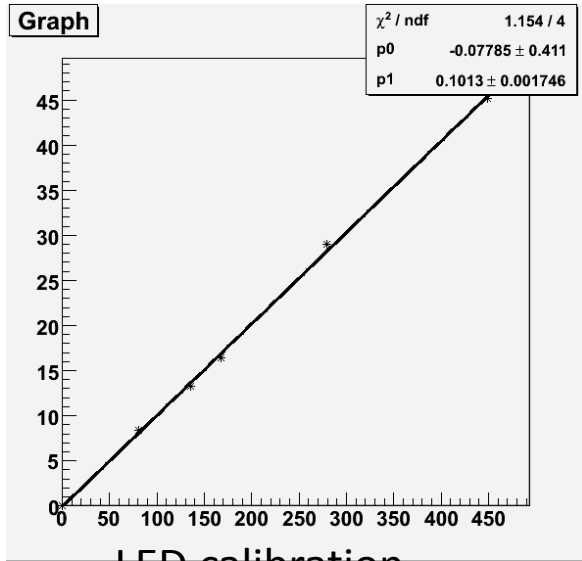
(2) Addition of **Gd** ( $(\text{Gd}_2(\text{SO}_4)_2)$ )

- Soluble ?

target concentration: 0.2% weight ratio  
(enough concentration for neutron capture measurement)

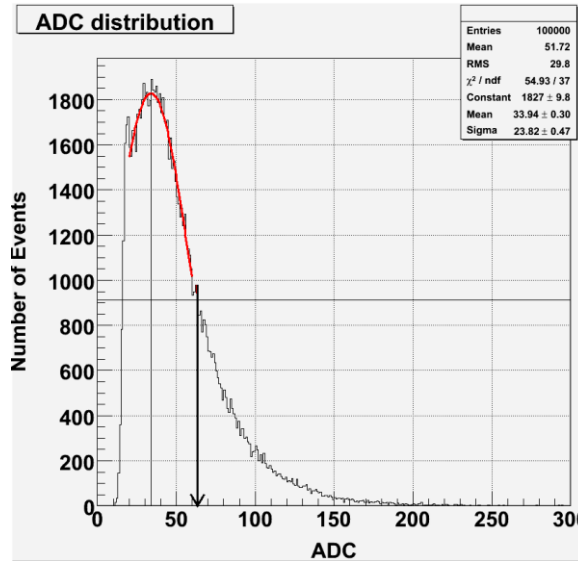
- How much light yield?

# Light yield of 50% detergent+50%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)

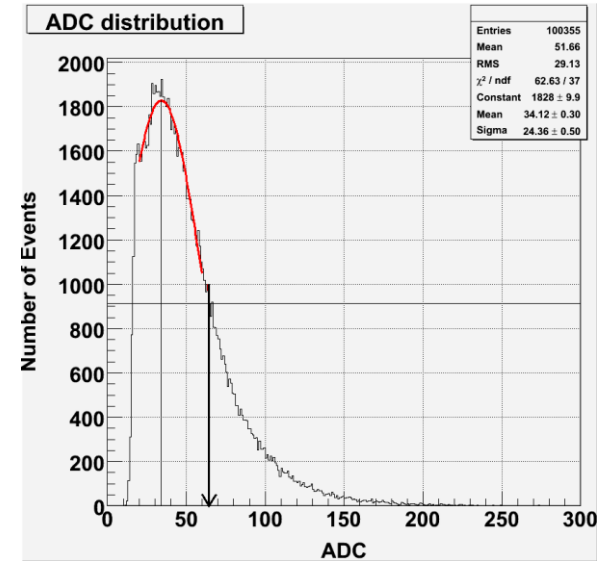


LED calibration

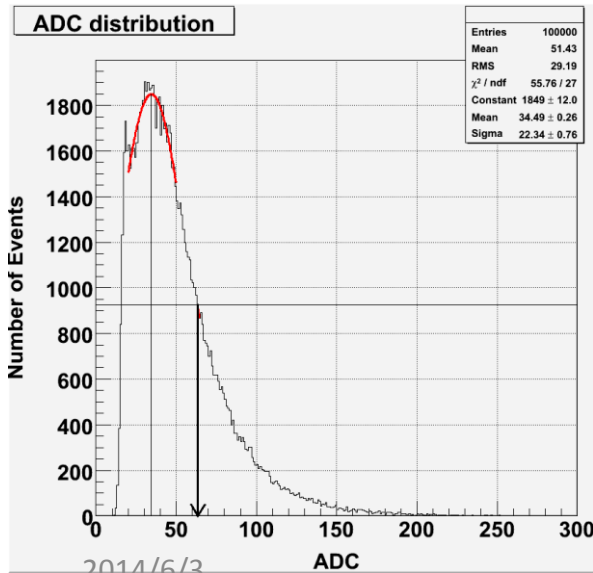
Inclination: 0.0994



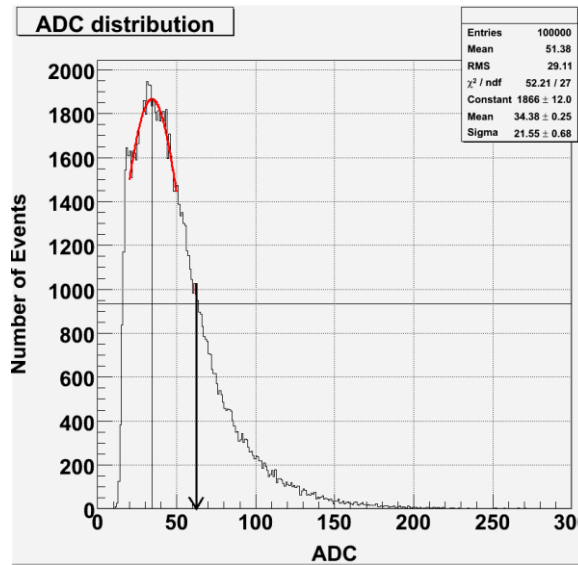
6.42[pe]



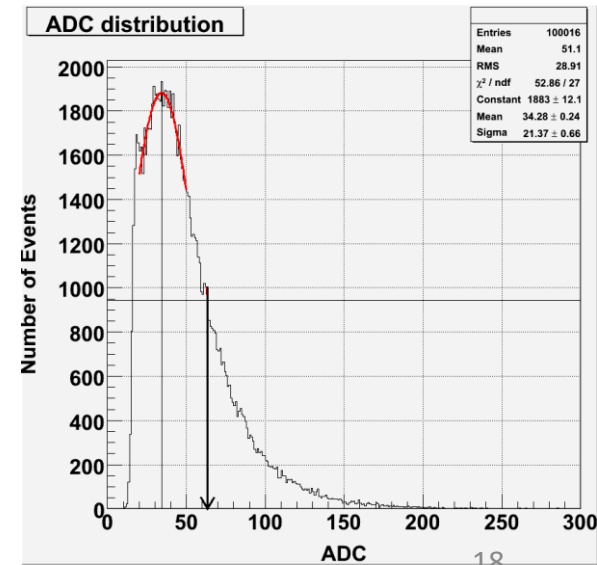
6.53[pe]



6.44[pe]

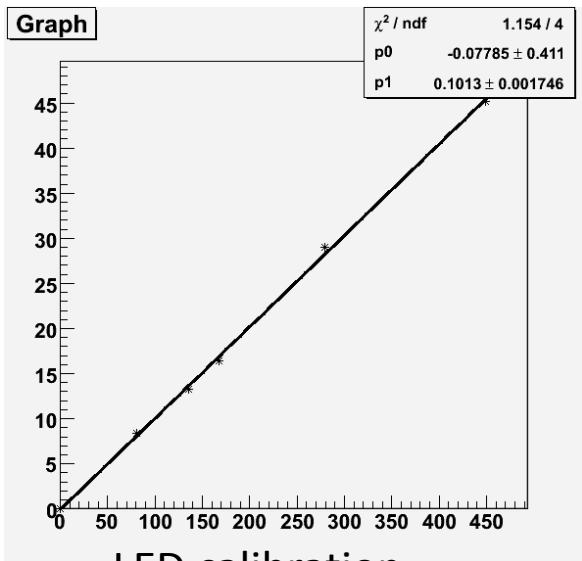


6.32[pe]



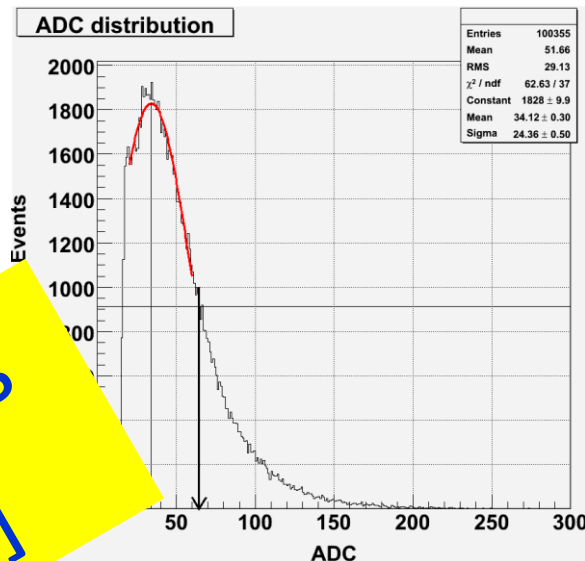
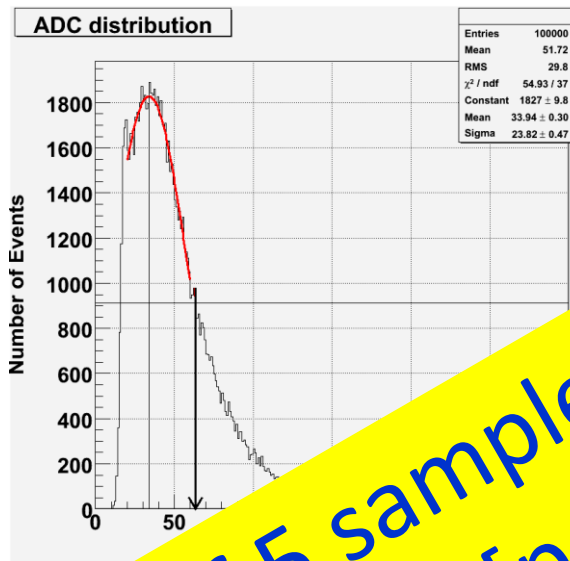
6.43[pe]

# Light yield of 50% detergent+50%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)

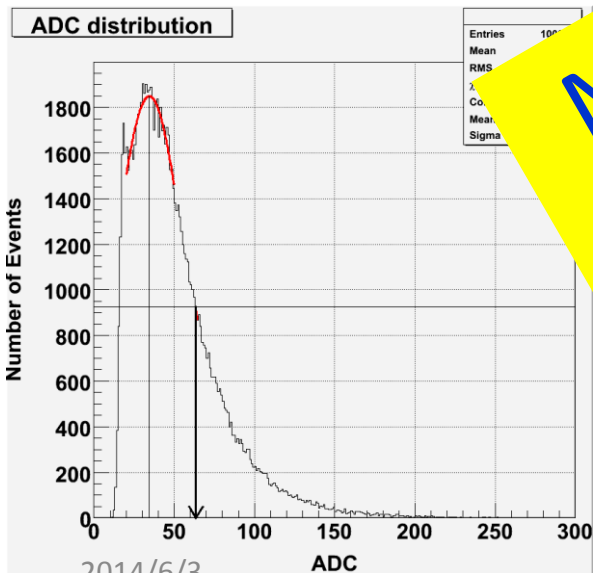


LED calibration

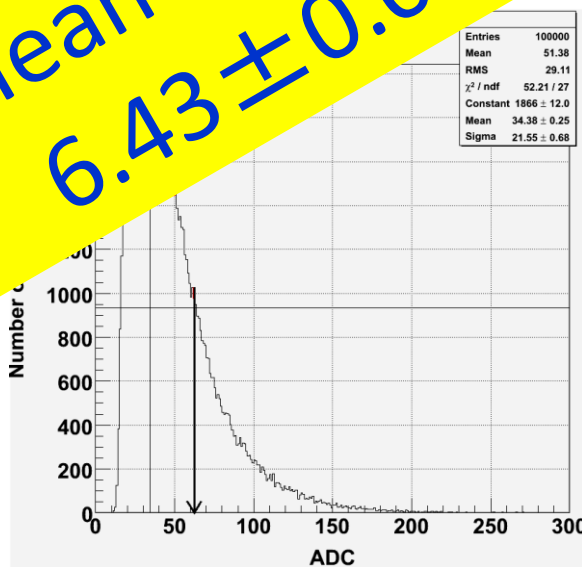
Inclination: 0.0994



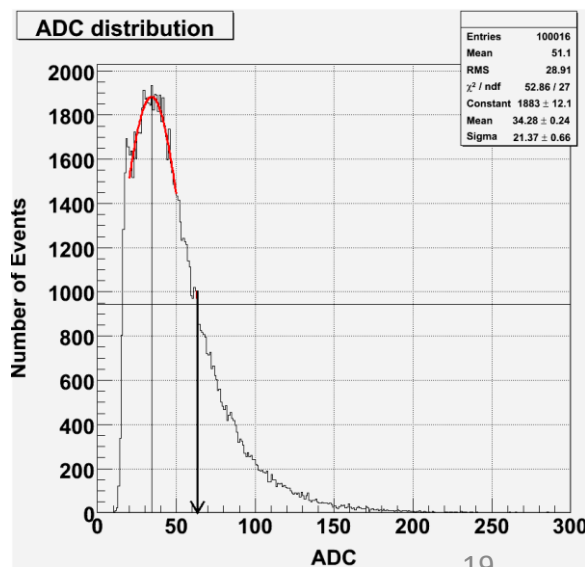
6.53[pe]



6.44[pe]



6.32[pe]

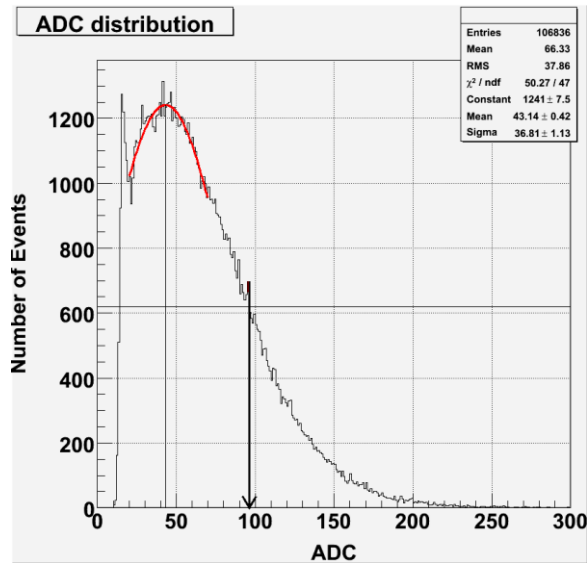


6.43[pe]

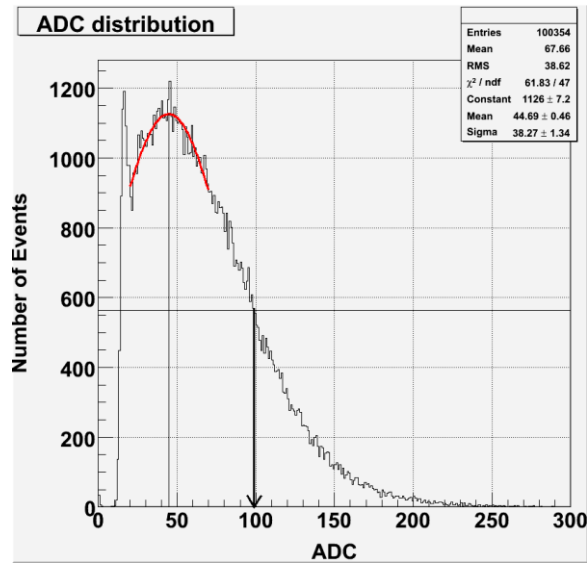
Mean of 5 samples  
6.43 ± 0.07 [pe]

# Light yield of **30%SDS+70%H<sub>2</sub>O+PPO 30g/l(H<sub>2</sub>O)+0.2%Gd**

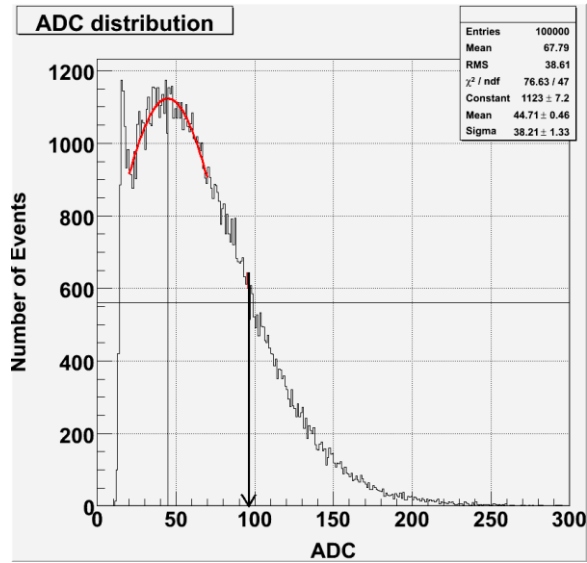
Gd:soluble  
 $9.55 \pm 0.38$ [pe]  
 Same light yield as  
 no Gd one  
 ( $9.99 \pm 0.17$  [pe])



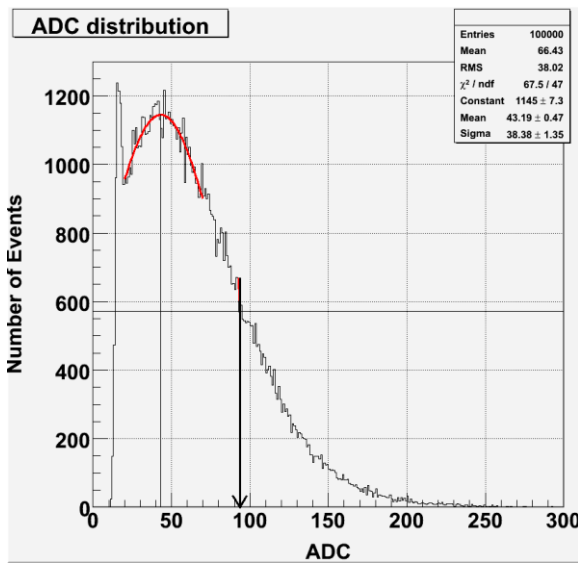
9.01[pe]



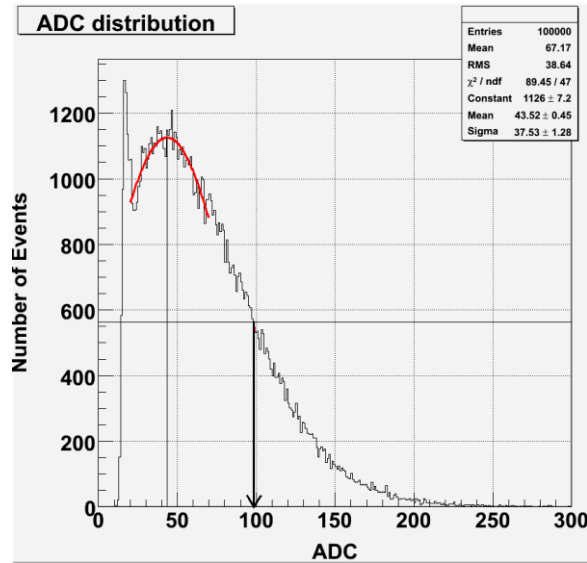
9.83[pe]



2014/6/3 9.55[pe]



9.29[pe]



9.79[pe]

# Summary

- We are developing **water-based liquid scintillator**.
- **SDS** (surfactant) + **water** (solvent) + **PPO** (luminescent agent)  
light yield:  $\sim 10$  [pe] (1/25 of conventional scintillator)  
Aging: no significant or small deterioration for 1 year  
 $\text{Gd}(\text{Gd}_2(\text{SO}_4)_3)$  : soluble, the same light yield ( $\sim 10$  pe)
- **Commercially available detergent** (surfactant) + **water** (solvent)  
light yield :  $\sim 6.5$  [pe]  
 $\text{Gd}(\text{Gd}_2(\text{SO}_4)_3)$  : insoluble

## Future plans:

- Measurement of neutron capture by Gd
- Try the following materials:
  - ① surfactants with benzene ring
  - ② aromatic molecules with hydrophilic group
  - ③ water-soluble luminescent agents



# Light Emission Process of Liquid Scintillator

Organic solvent (ex: PC) is excited by absorbing the radiation energy.

max. absorption: 269nm

mean: 290nm, max: 293nm

**Organic solvent → water or flammable solvent**

First solute (ex: PPO) absorbs the energy from the solvent and emits light.

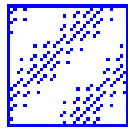
max absorption: 303nm

emission mean: 370nm, max: 364nm

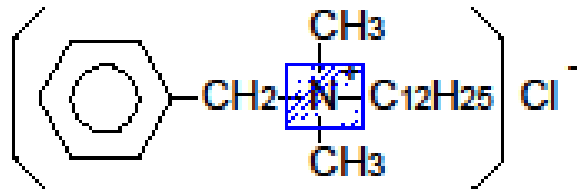
Second solute (ex: Bis-MSB) makes the wavelength longer to prevent reabsorption by the solvent.

max absorption: 347nm

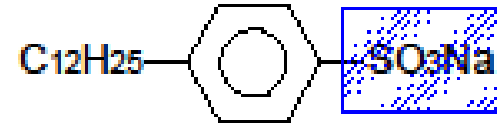
emission mean: 422nm, max: 412nm



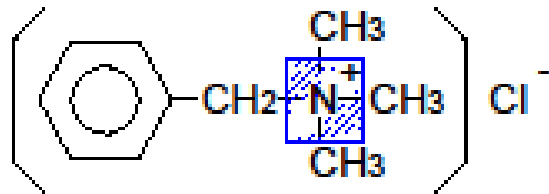
Hydrophilic group



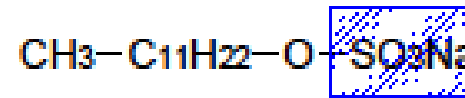
Benzalkonium chloride



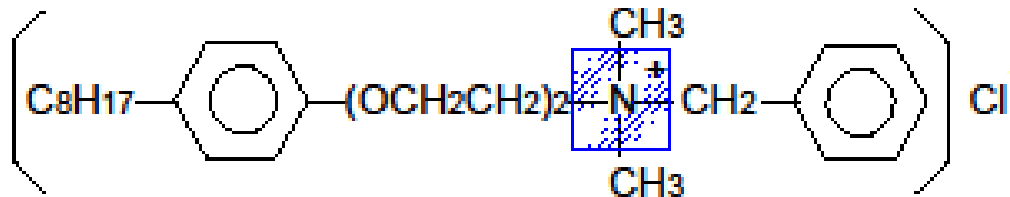
Dodecyl benzene sulfonic acid (ABS)



Benzyl chloride trimethyl ammonium



Sodium dodecyl sulfate (SDS)



Benzethonium chloride

# Surfactants



# Liquid scintillator so far (example)

- KamLAND(1200m<sup>3</sup>): PC(20%)+dodecane(80%)+PPO(0.2%)
- PaloVerde(~10m<sup>3</sup>): PC(36%)+mineral oil(60%)  
+alcohol(4%) + PPO+bis-MSB+BHT(antioxidant)
- Borexino(300m<sup>3</sup>): PC(100%)+PPO(0.2%)
- DCHOOZ (~10m<sup>3</sup>): PXE(20%) + C<sub>12</sub>H<sub>26</sub>(80%) + 0.1% Gd  
+ PPO(6g/ℓ) + Bis-MSB(20mg/ℓ)

Nonvolatile material  
with high flash point

Solvent ~ 20%, diluted solution ~ 80%, and solute ~ several  
times 0.1%

Water

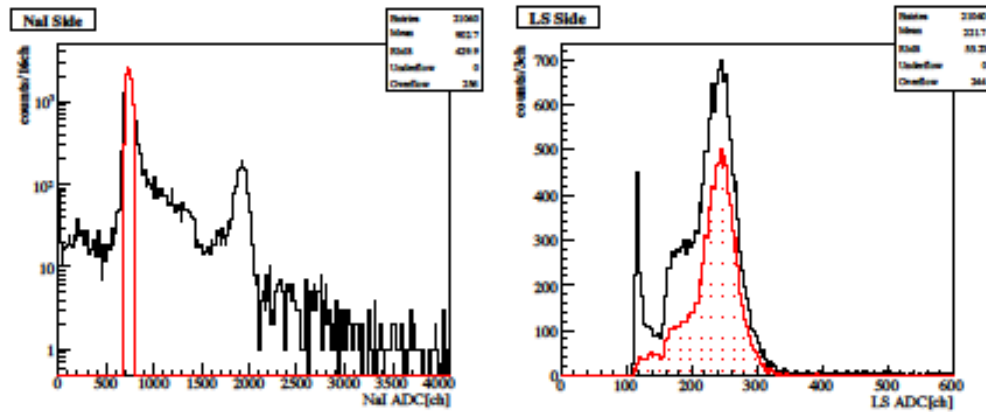


図 5.2: 後方散乱のスペクトル: 左図は NaI のエネルギースペクトルでピンク色の部分でカットした。右図は PMT 側のエネルギースペクトルで NaI のカットによりピンク色の部分が得られた。

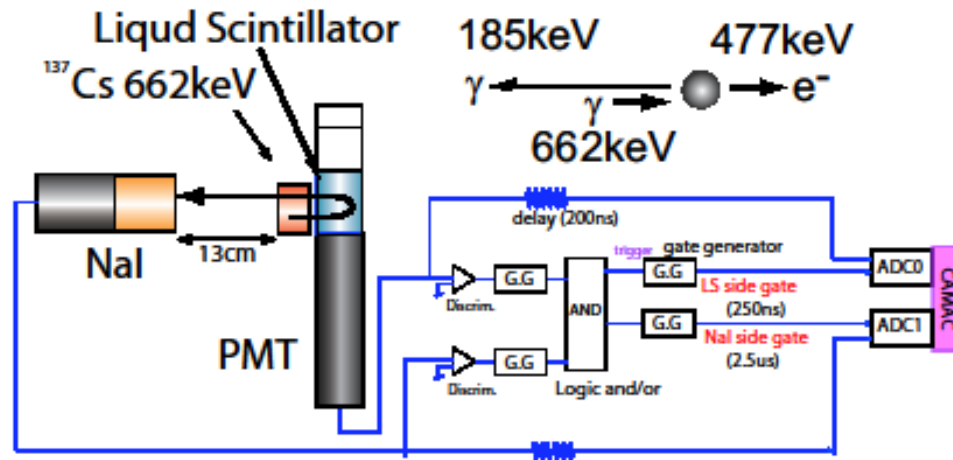


図 5.3: 発光量測定の装置と後方散乱の回路

修士論文: 小川桃世

「KamLANDにおける太陽ニュートリノ観測に向けた液体シンチレーターの純化」  
 東北大学2006年(平成17年度)

# Number of Cherenkov Photons

$$N = 2\pi\alpha lz^2 \left( \frac{1}{\lambda_2} - \frac{1}{\lambda_1} \right) \left( 1 - \frac{1}{n^2 \beta^2} \right)$$

$N$ : number Cherenkov photons

$\alpha$ : fine structure constant

$l$ : path length

$z$ : charge

$\lambda_1, \lambda_2$ : max. and min. wave lengths observed

When  $l=5\text{mm}$ ,  $E = 1.5 \text{ MeV}$ ,  $\lambda_1=400 \text{ nm}$ ,  $\lambda_2=600 \text{ nm}$ ,

$N \approx 70$

Number of pe's obtained = 70 ×

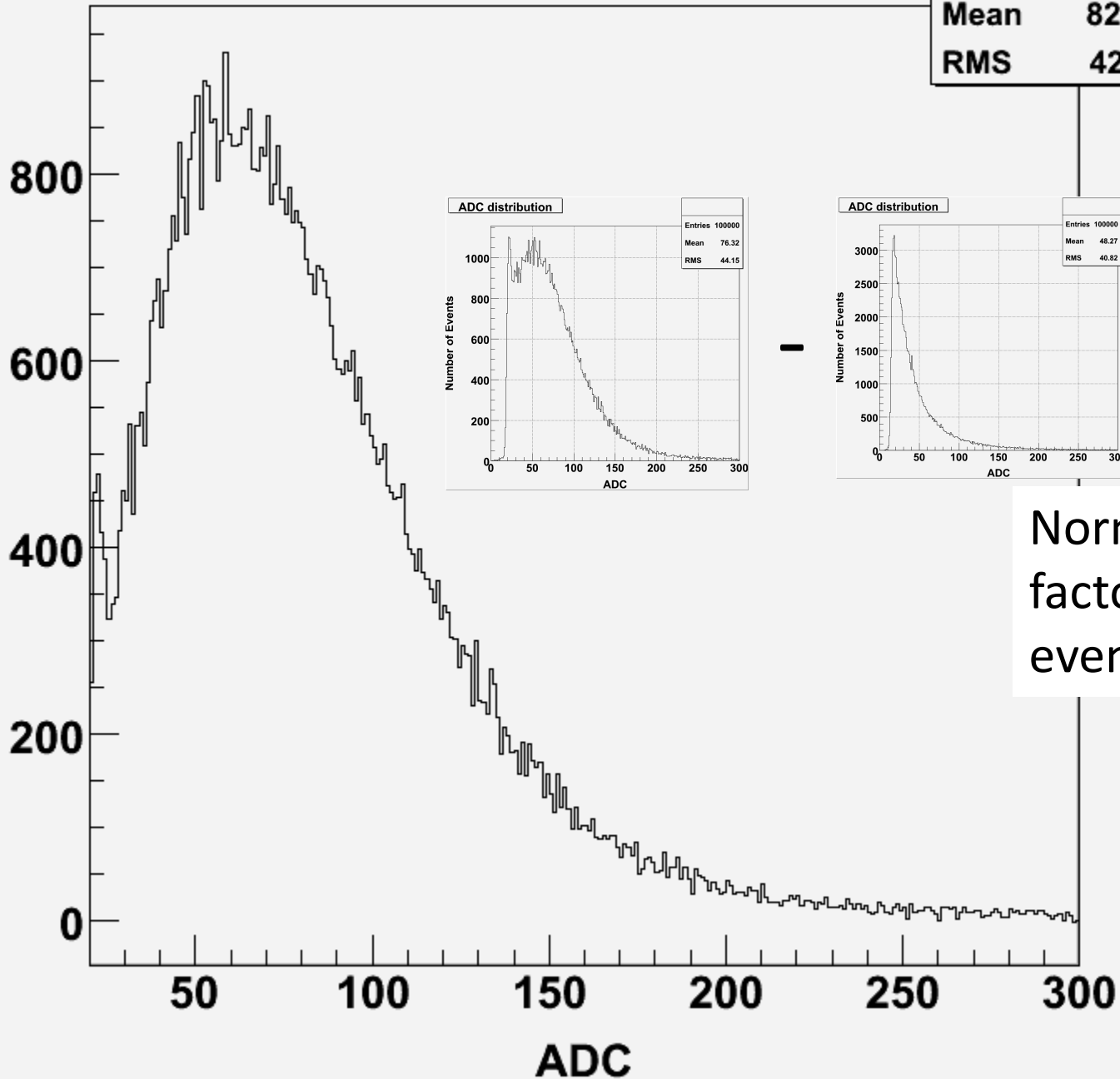
0.2 (QE of PMT) ×

0.5 (acceptance ?)

= 7

# ADC distribution

|         |       |
|---------|-------|
| Entries | 72675 |
| Mean    | 82.48 |
| RMS     | 42.51 |



× 0.25  
↓

Normalization factor to the event rate