SKY-ZERO @ Boulby Cosmic rays and climate

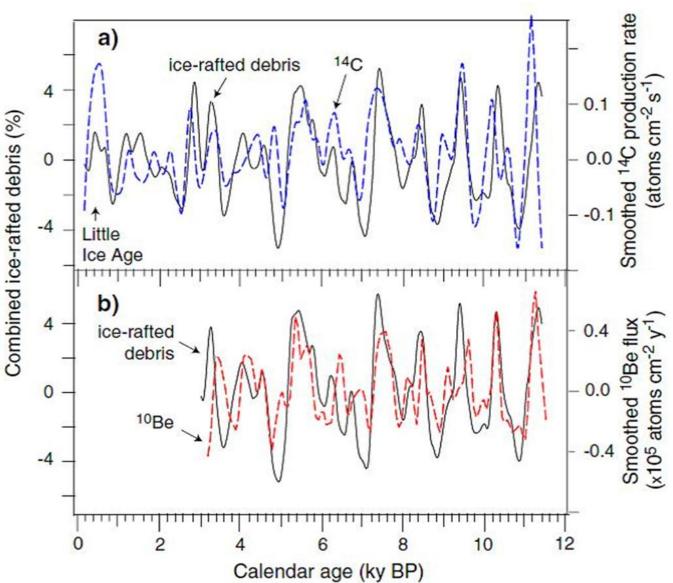
DTU

- what can we learn from the ultra-low radiation environment

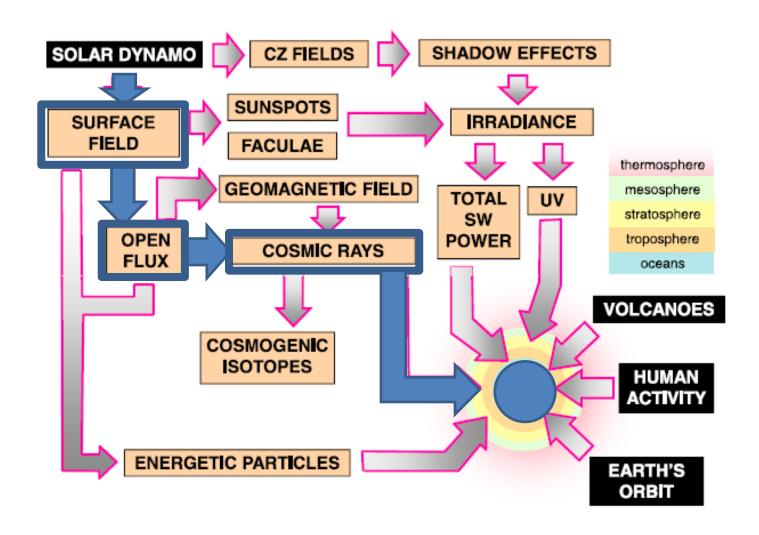


Me

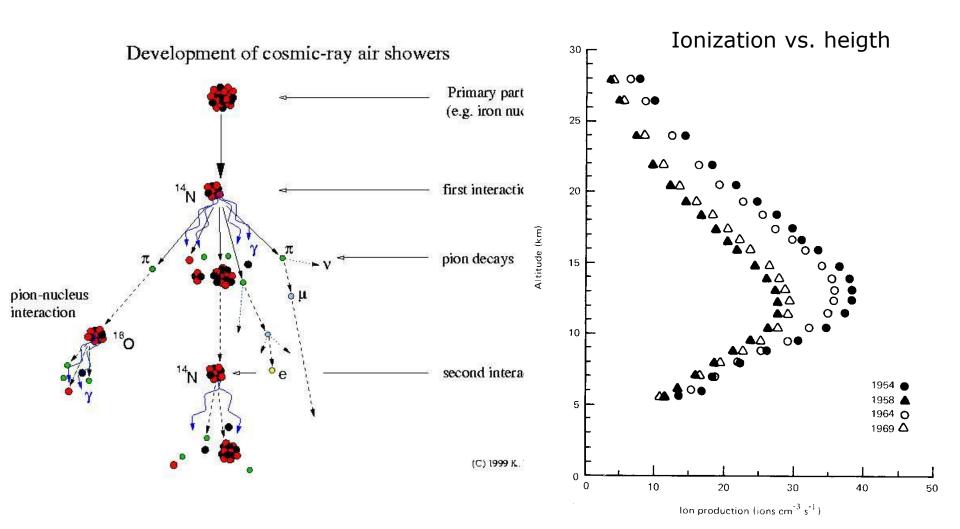
Why is climate changing?



From the Sun to the Earth

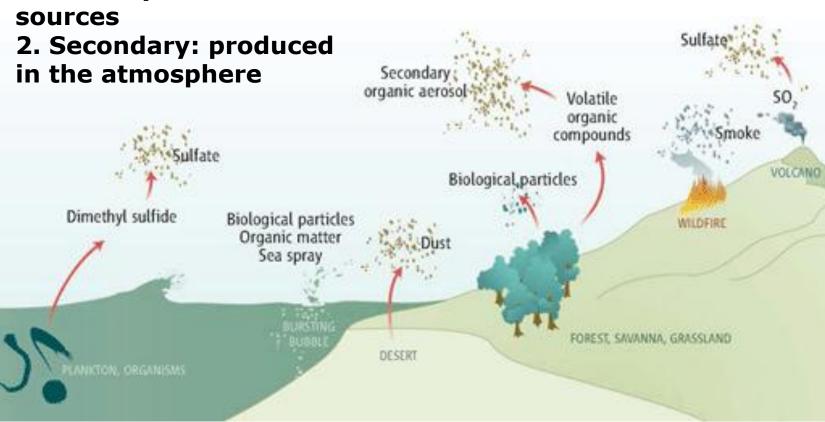


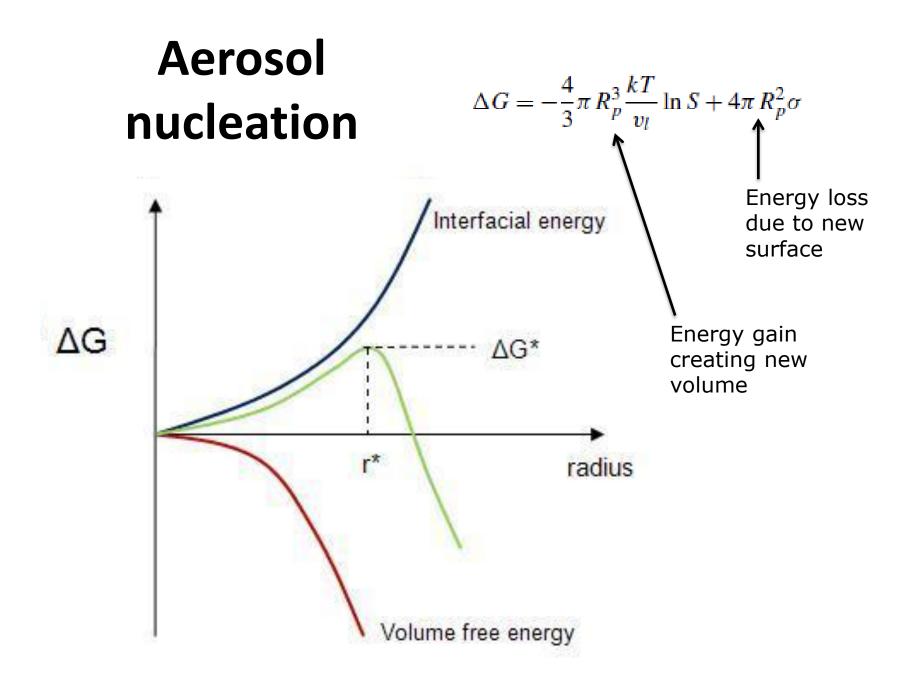
Cosmic rays -> ions



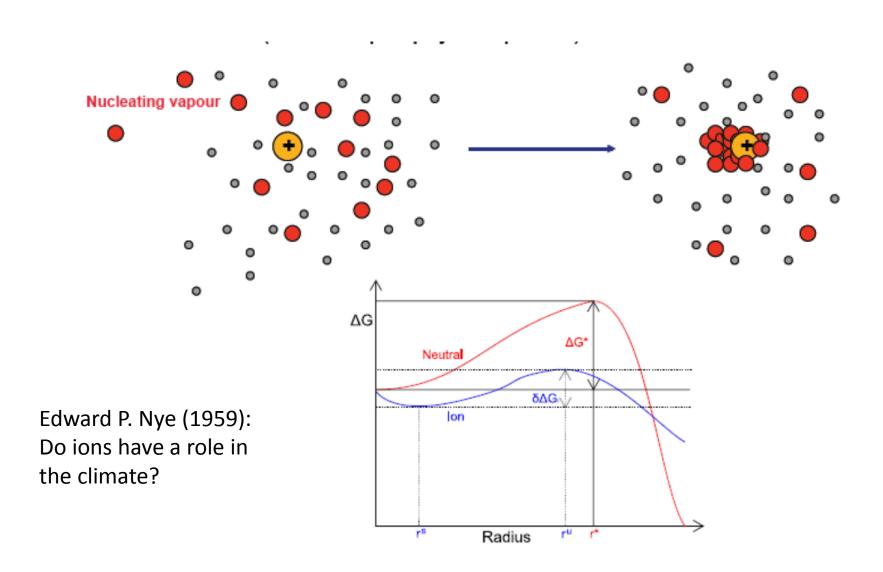
lons -> aerosols

1. Primary: direct

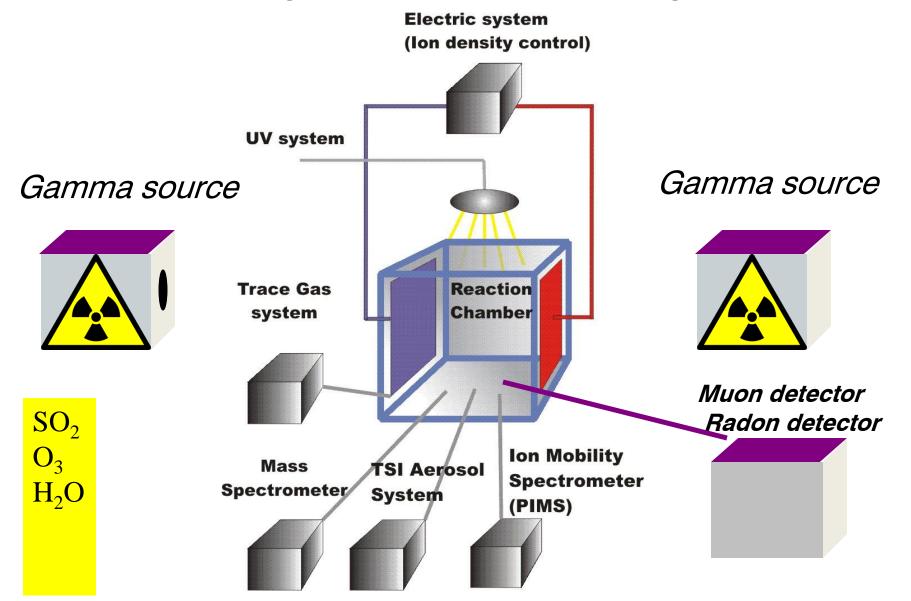




lons -> aerosols



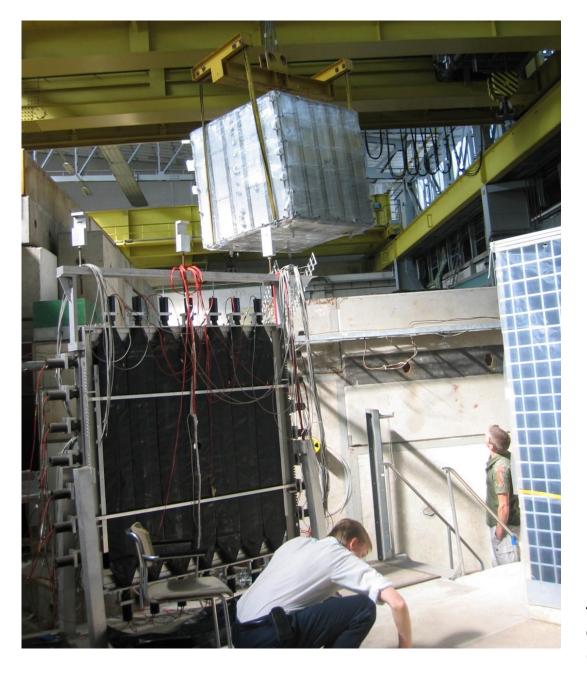
Experimental setup



SKY (Danish for "cloud")



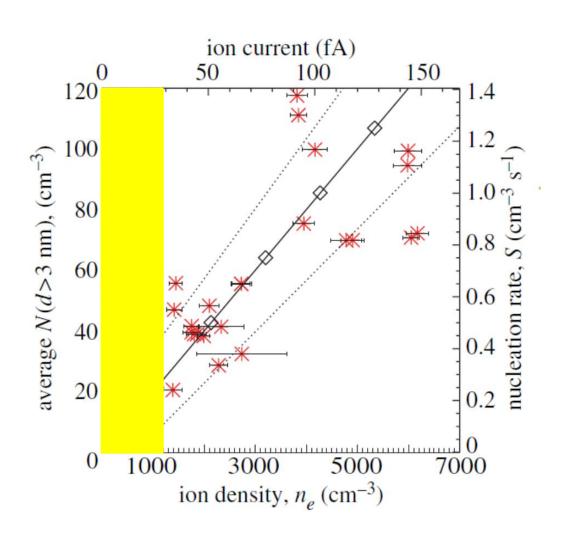
Svensmark *et al.* (2007) Proc.Roy.Soc. A



SKY I-a = pre-CLOUD

J. Duplissy, M.B. Enghoff *et al.* (2009) Atmos. Chem. Phys.

Experimental results



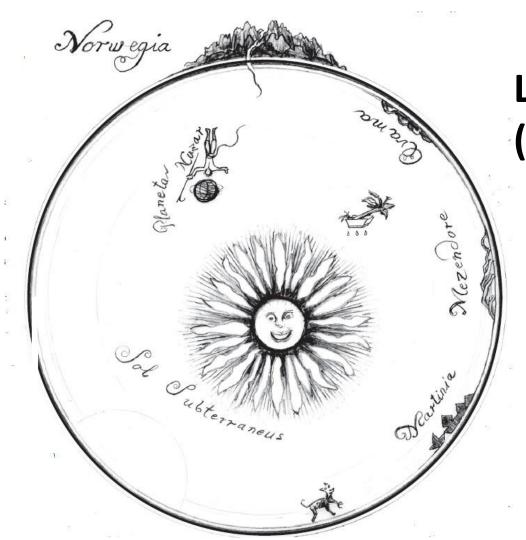
Correlation
between ions and
3 nm aerosols
580 MeV e⁻
662 keV γ
gives same result

Enghoff et al. (2011) Geophys.Res.Lett.

How to stop cosmic rays?



Underground journeys



Ludvig Holberg: (1741) *Niels Klim*

Underground journeys II



Inside the laboratory

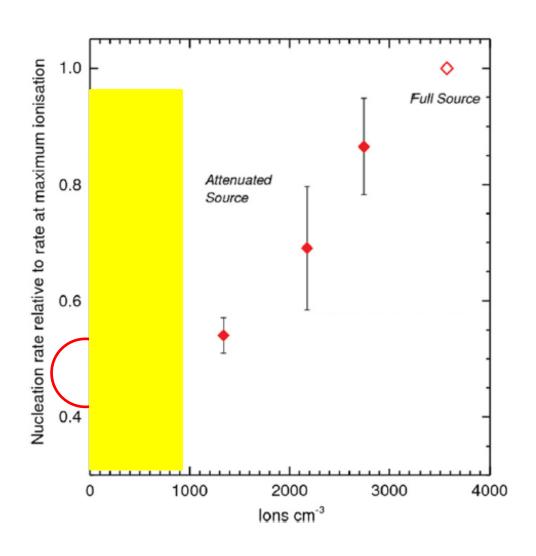








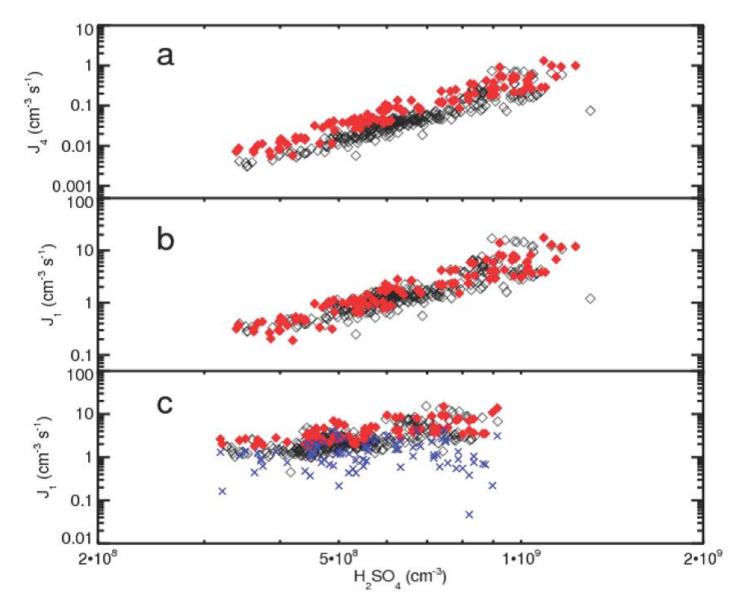
Aerosol nucleation vs. ion density



(Sulfuric acid conc. $5 \times 10^8 \text{ cm}^{-3}$)

Pedersen *et al.* (2012) J. Aerosol Sci.

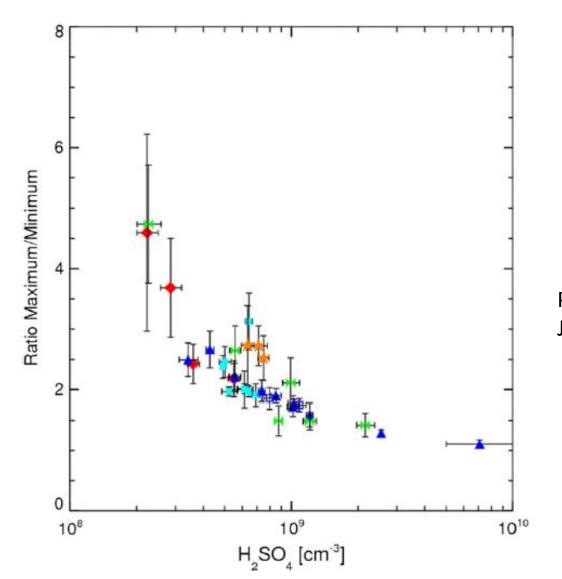
Nucleation rate vs. sulfuric acid concentration



Slope in multi component system = Number of molecules in critical cluster + factor (0-1)

=> 2-3 sulfuric acid molecules in the critical clusters

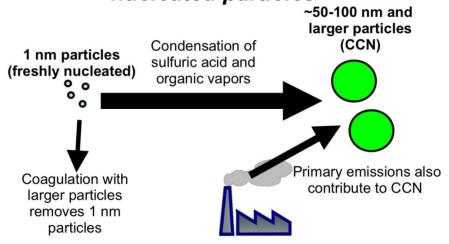
Ion effect vs. [H₂SO₄]



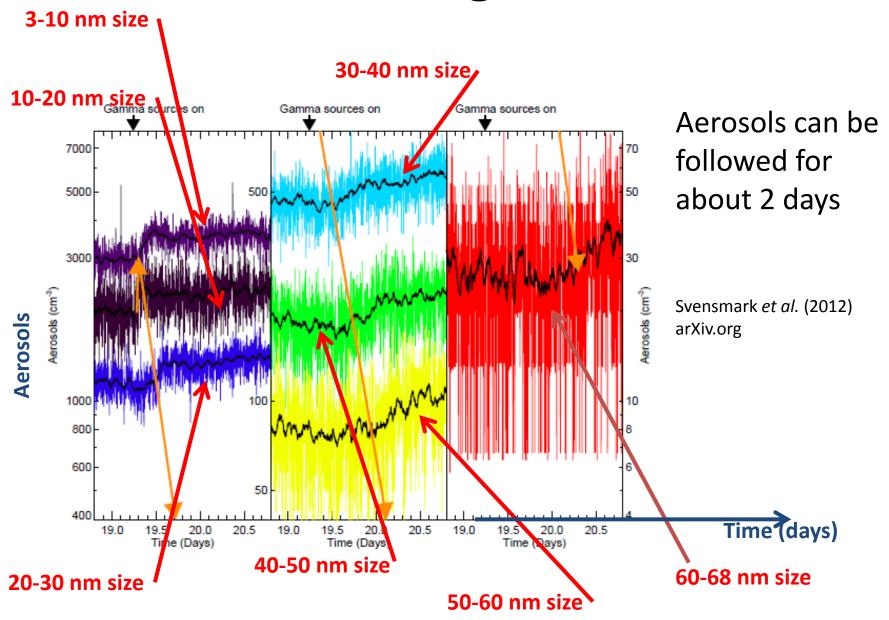
Pedersen *et al.* (2012) J. Aerosol Sci.

Aerosols -> clouds?

Formation of CCN from nucleated particles



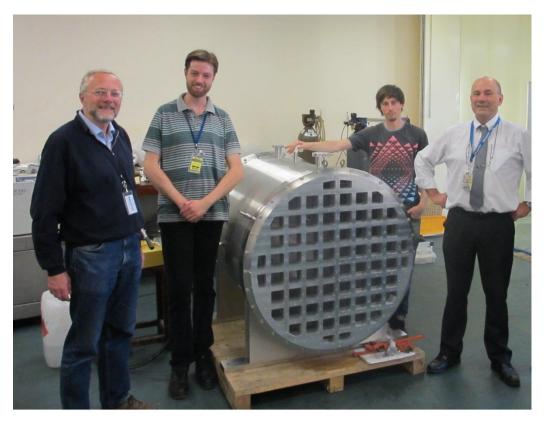
Aerosol growth



Highlights

- Ions do play a role in 3-4 nm nucleation in all experiments
- How the ion is produced is not important
- Underground experiments allows ions to be separated from "neutral" nucleation mechanisms
- Both neutral and ion-induced nucleation takes place in the experiment
- Ions increase aerosol appearance rate at 4 nm due to increase in nucleation rate and faster growth during initial stage
- The critical clusters contain 2-3 sulfuric acid molecules
- 3rd place in weekly pub quiz at the Station Inn in Whitby

Future experiments



Hastings, UK, August 2012

Kurt J. Lesker[®]

Larger chamber (940 L), temperature controlled, better purity, more instruments

UK/Danish collaboration: Manchester, Birmingham, Oxford, Sheffield

STFC grant ST/K002120/1 for Particle Size Magnifier (Prof. Roy Harrison, Birmingham).

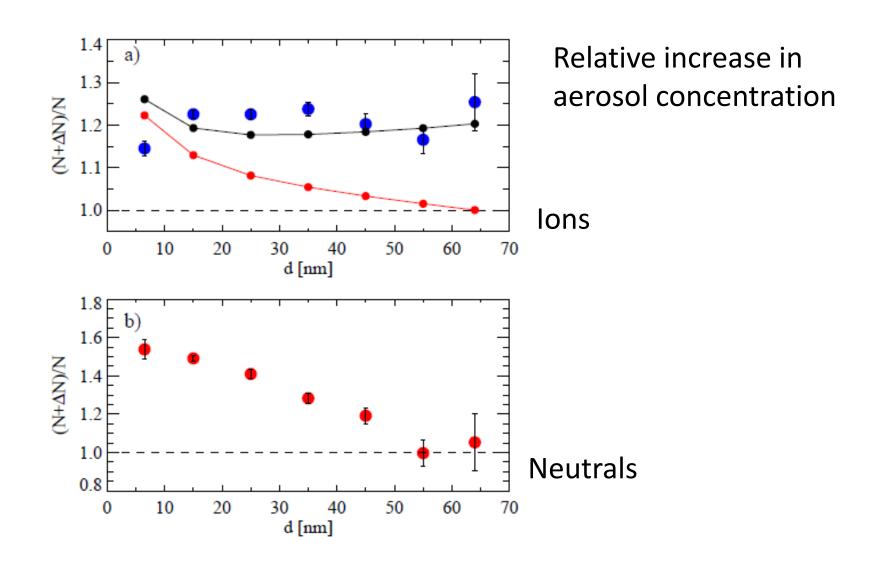
Thank you for your attention!



... we find that ion enhancement of nucleation occurs for all temperatures, humidities and cluster compositions observed so far. Ion-induced nucleation will manifest itself as a steady production of new particles that is difficult to isolate in atmospheric observations because of other sources of variability but is nevertheless taking place and could be quite large when averaged globally over the troposphere.

Kirkby et al. (2011) Nature

Aerosol growth



How to make a cloud?



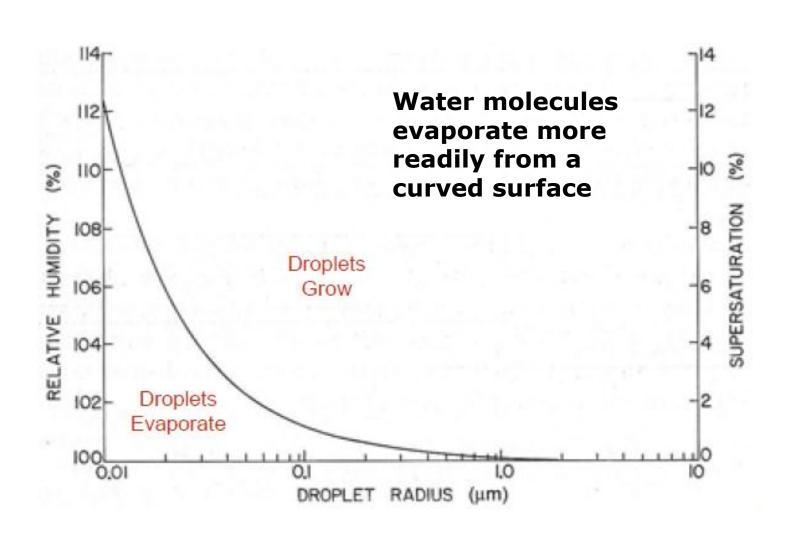
Clouds form when air becomes supersaturated

Rarely exceeds 1% (typically 0.1%)

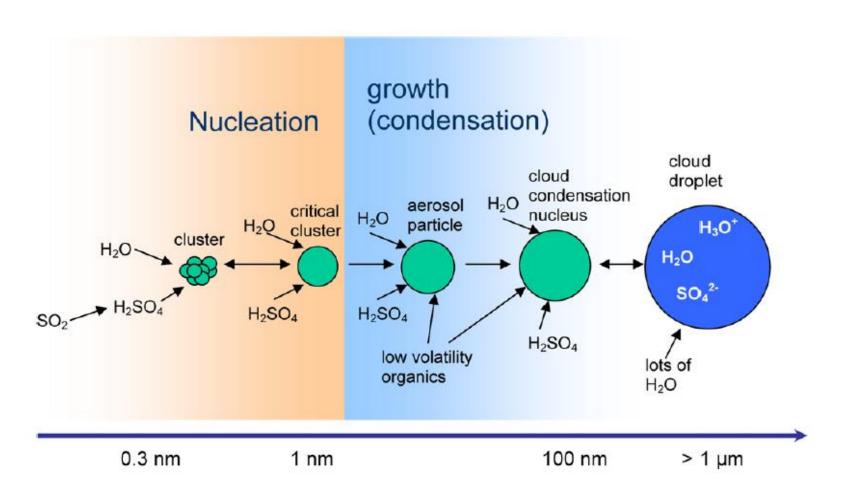
Water vapor condenses on aerosols

No nucleation of pure water

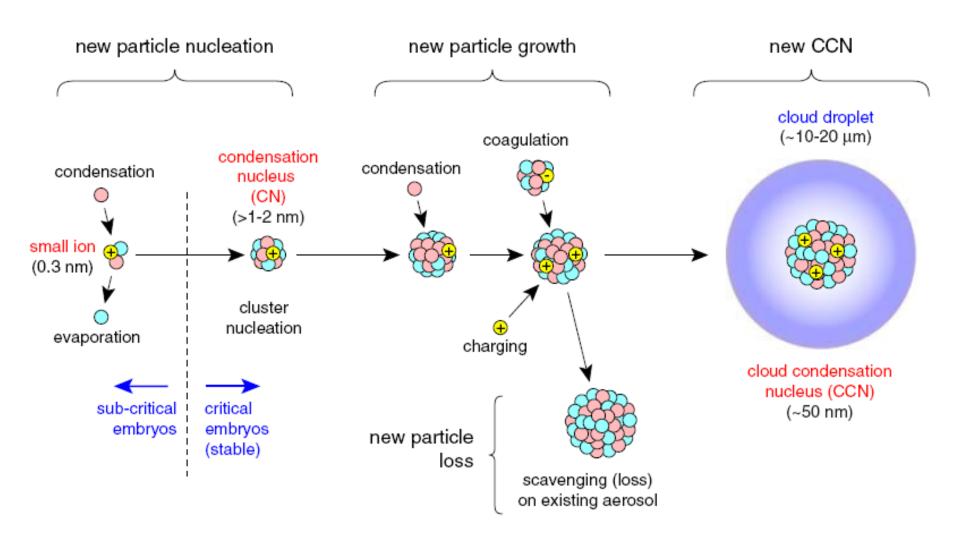
Kelvin curvature effect

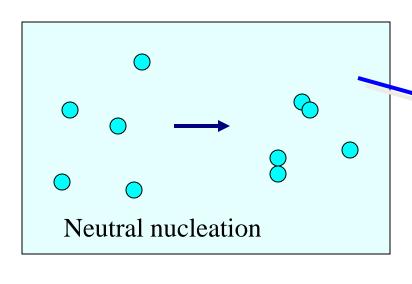


Heterogeneous nucleation (sulfuric acid + water)

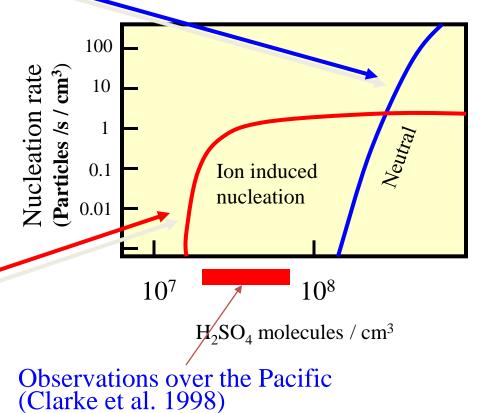


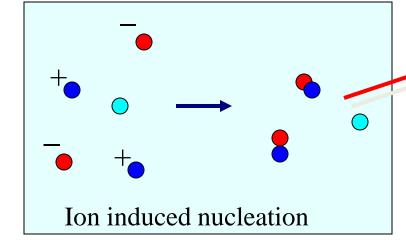
Ion-induced heterogeneous nucleation (sulfuric acid + water)



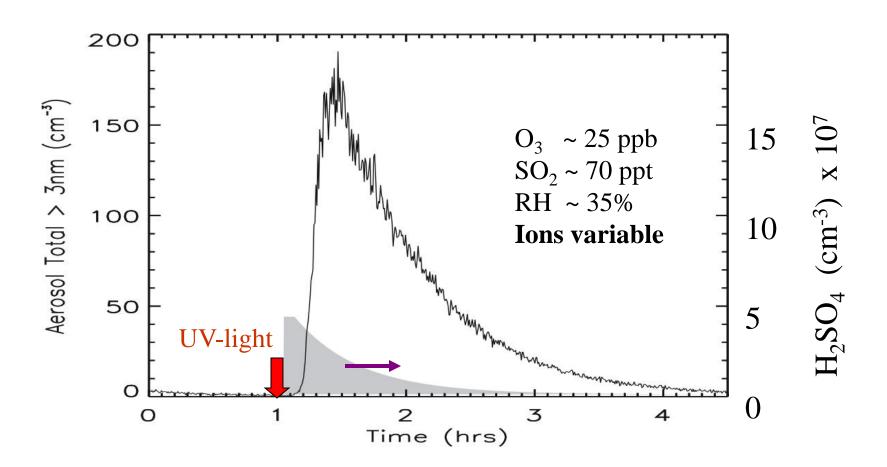


Formation of ultra fine aerosols (~ 3 nm)



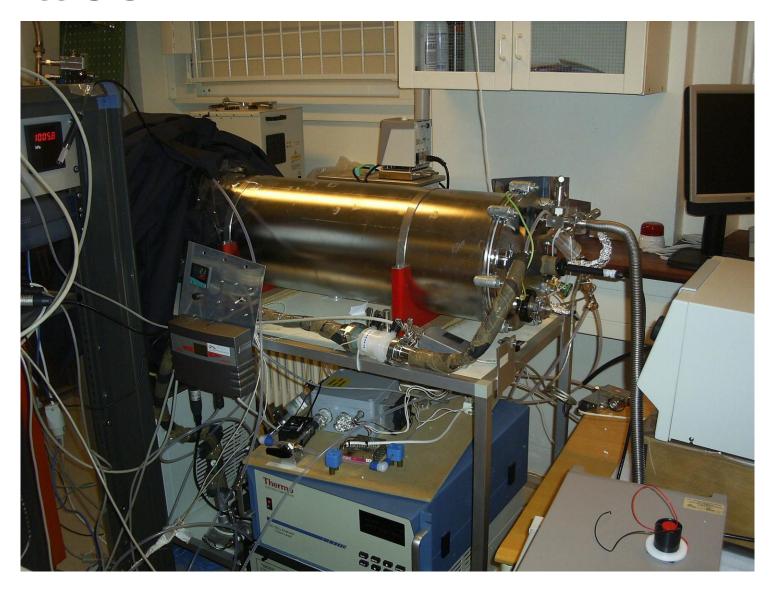


Data



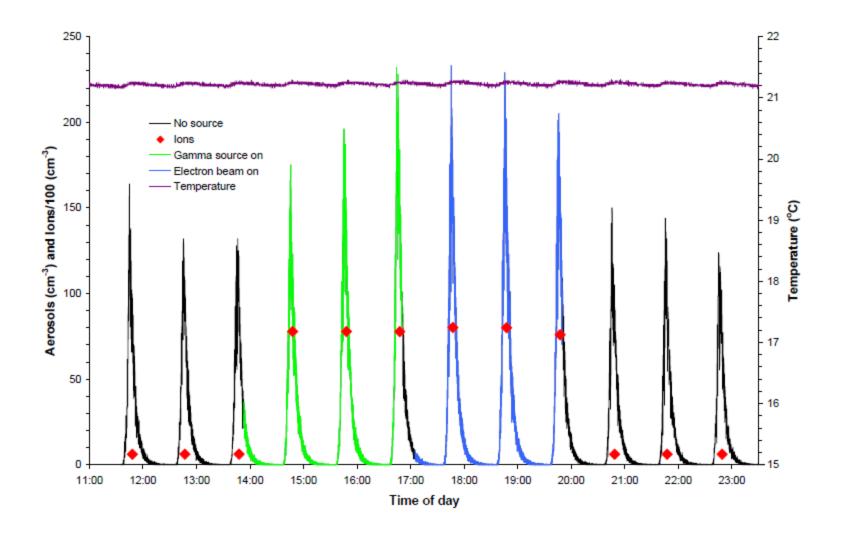
Svensmark et al. Proc.Roy.Soc.A (2007)

Little SKY



580 MeV electrons (University of Aarhus, Denmark)

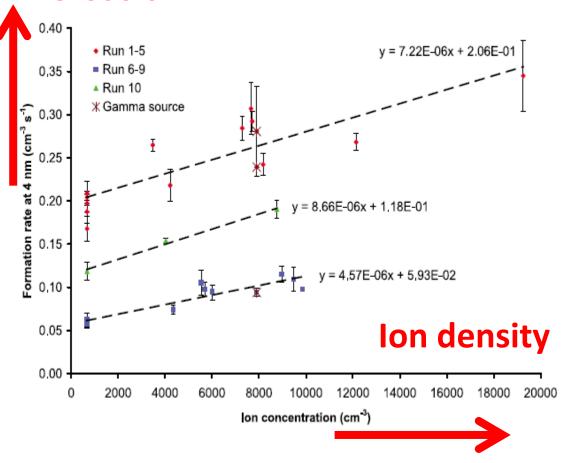




Enghoff et al. GRL (2011)

Main result from SKY in Aarhus

Aerosols > 4 nm



Correlation between ions and aerosols produced at 4 nm (as with SKY).

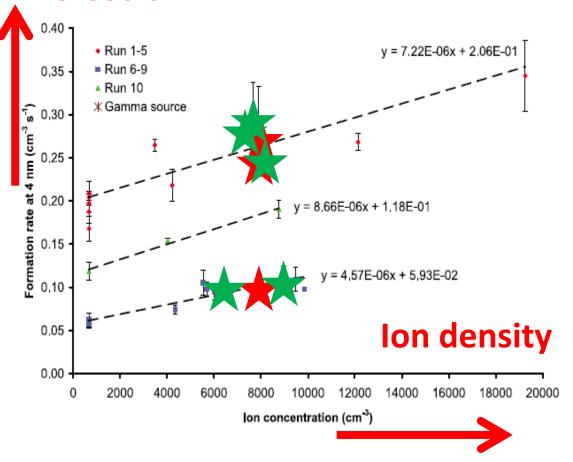
p = 1 atm

T = 25 °C

Enghoff et al. (2011) Geophys.Res.Lett.

Main result from SKY in Aarhus

Aerosols > 4 nm

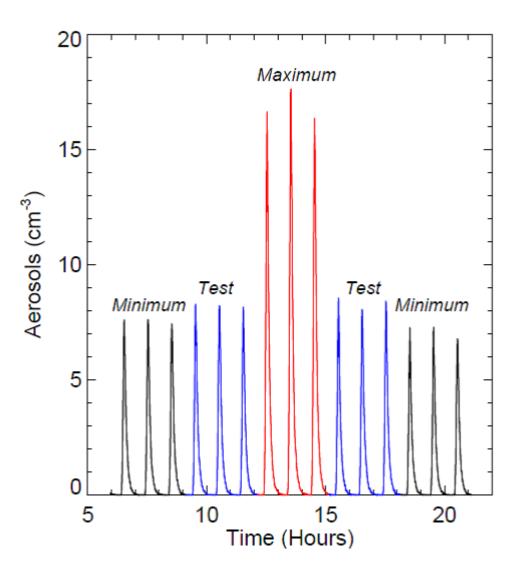


Gamma rays

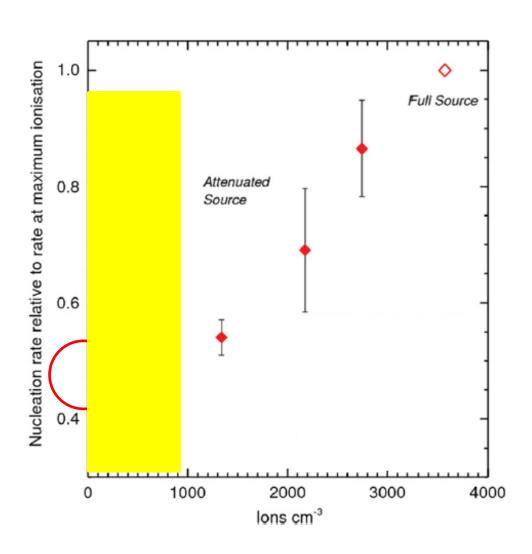
and **Electrons** give same result

(i.e. a beam in not needed)

Enghoff et al. (2011) Geophys.Res.Lett.



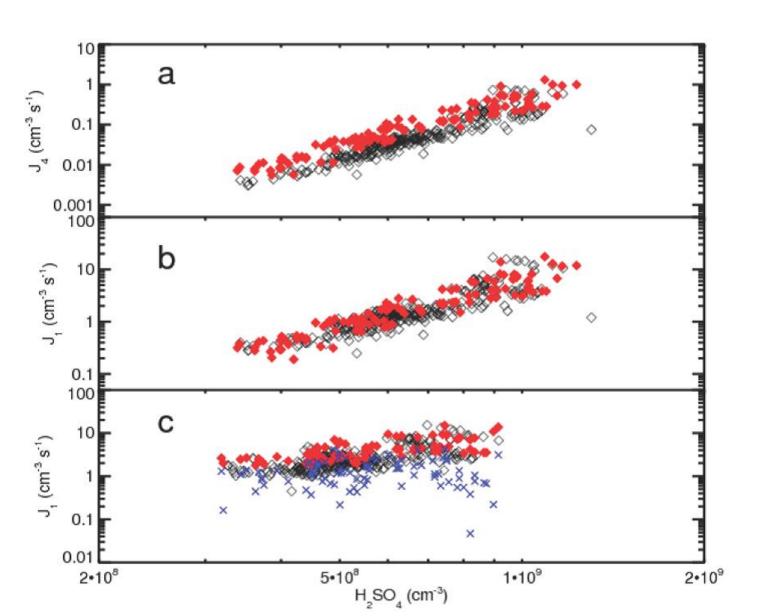
Aerosol nucleation vs. ion density

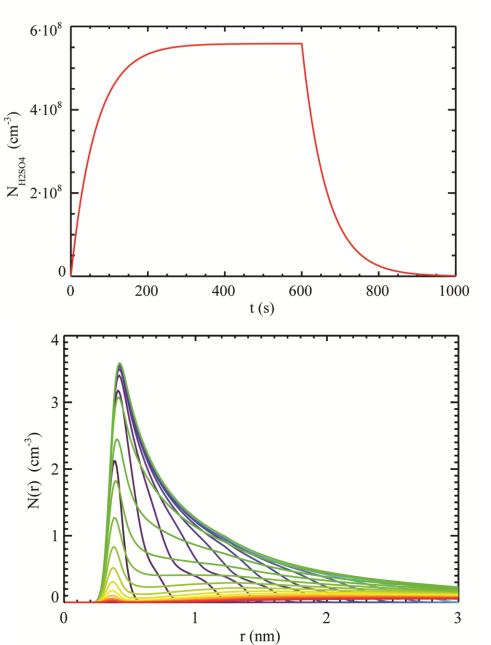


(Sulfuric acid conc. about $5 \times 10^8 \text{ cm}^{-3}$)

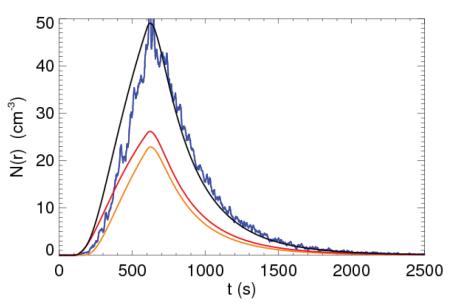
Pedersen et al (2012) J. Aerosol Sci.

Nucleation rate vs. sulfuric acid concentraion

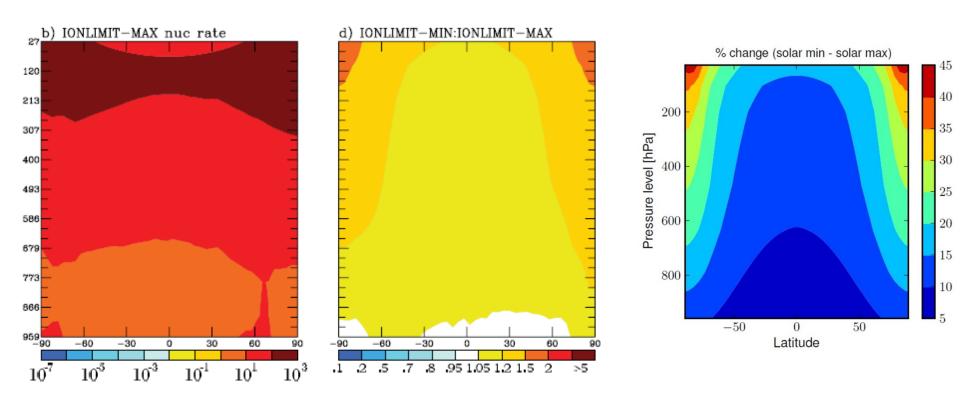




Simulations

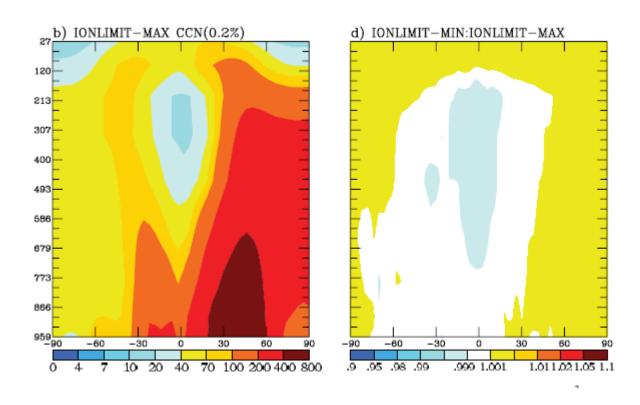


Sun/climate variations?

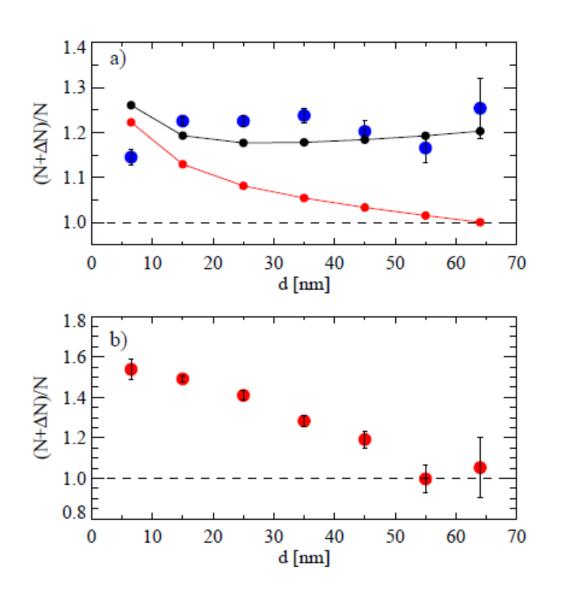


24% increase in global nucleaiton rate Pierce & Adams (2009) GRL

Sun/climate variations?



0.08% increase in CCN Pierce & Adams (2009) GRL



lons

Neutrals

Conclusions

- Ions do play a role in 3-4 nm nucleation in all experiments
- How the ion is produced is not important
- Underground experiments allows ions to be separated from "neutral" nucleation mechanisms
- Both neutral and ion-induced nucleation takes place in the experiment
- Undetected impurities and organic vapors may also participate
- Model simulations show that ions increase aerosol appearance rate at 4 nm due to increase in nucleation rate and faster growth during initial stage
- The model indicates that the critical clusters contain 2-3 sulfuric acid molecules
- Ion effect also visible on larger aerosols (65 nm)

Atmospheric implications?

- SO₂ and chamber pressure & temperature all above atmospheric conditions, so extrapolations are difficult
- If we still extrapolate, ion-induced nucleation may dominate nucleation (at 3-4 nm) in clean air regions of the atmosphere
- Open question: how many of the 3-4 nm aerosols will grow to become activated
- climate effect still unknown

Future work:

New (1,000 L) chamber constructed



