

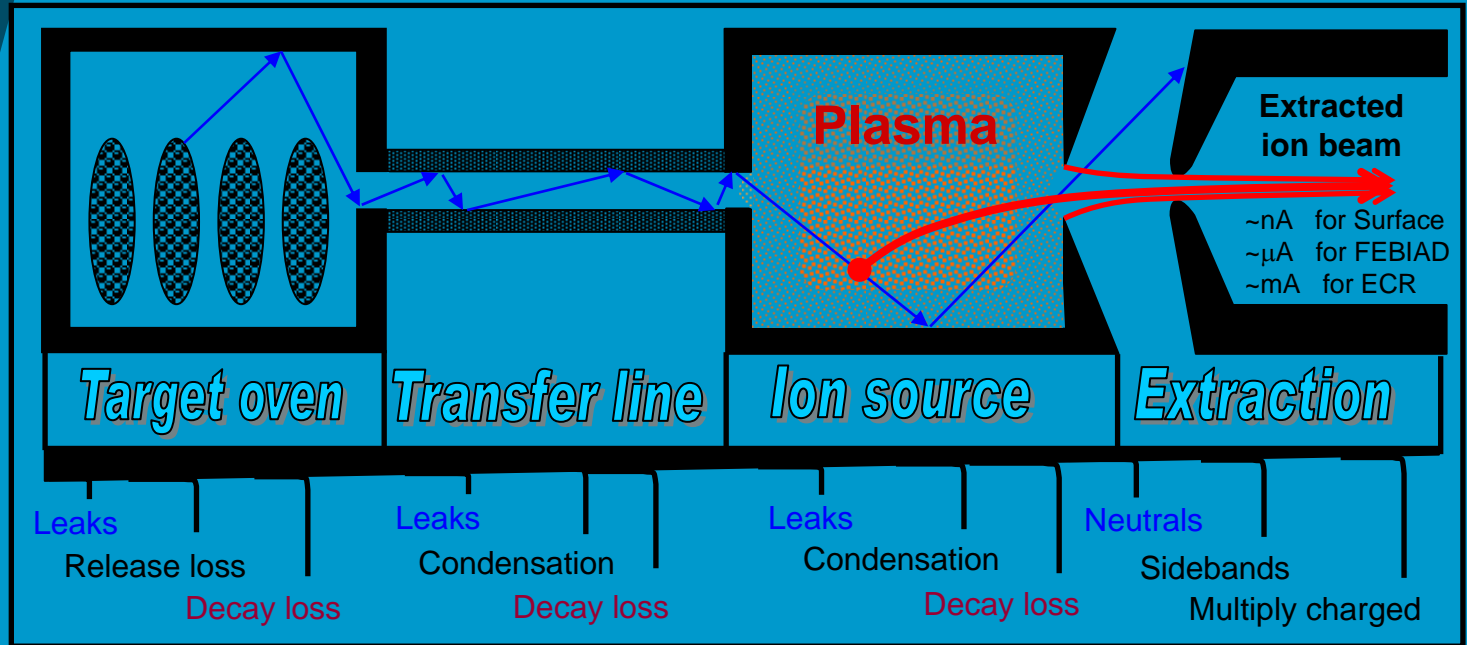
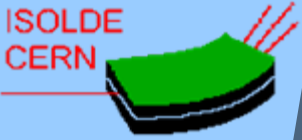
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**FEBIAD ion source *operation modes* for  
tuning its **selectivity**:  
physics processes, numerical simulations  
and experimental data.**

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*CERN, AB-ATB-IF (ISOLDE)*

# Selectivity in a ISOL target-ion source unit

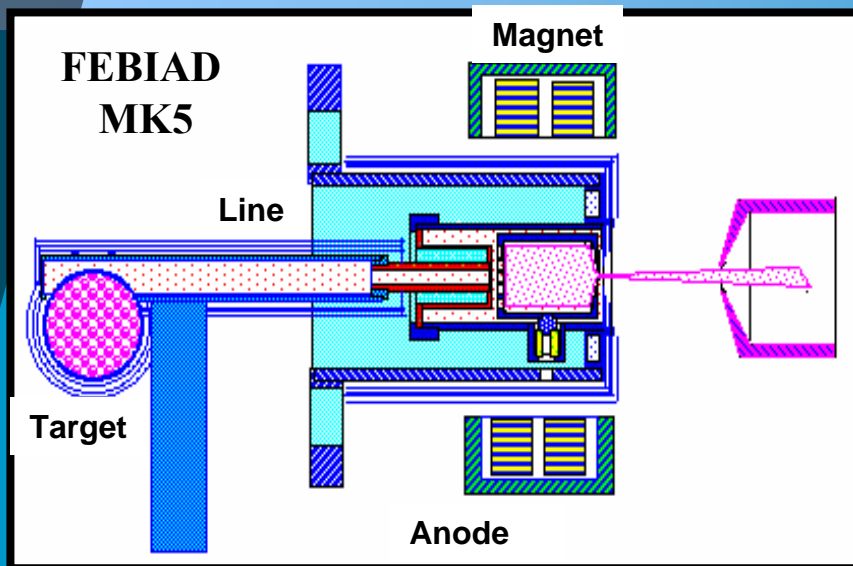


## Selective ISOL steps:

- Neutron converter;
- Thermochromatography in the transfer line;
- Ion source type; RILIS ionization;
- Molecular compounds
- Plasma ion source tuning & customization



# FEBIAD ion sources



→ No limit concerning the ionization potential;

→ Limitation given by the element's volatility:

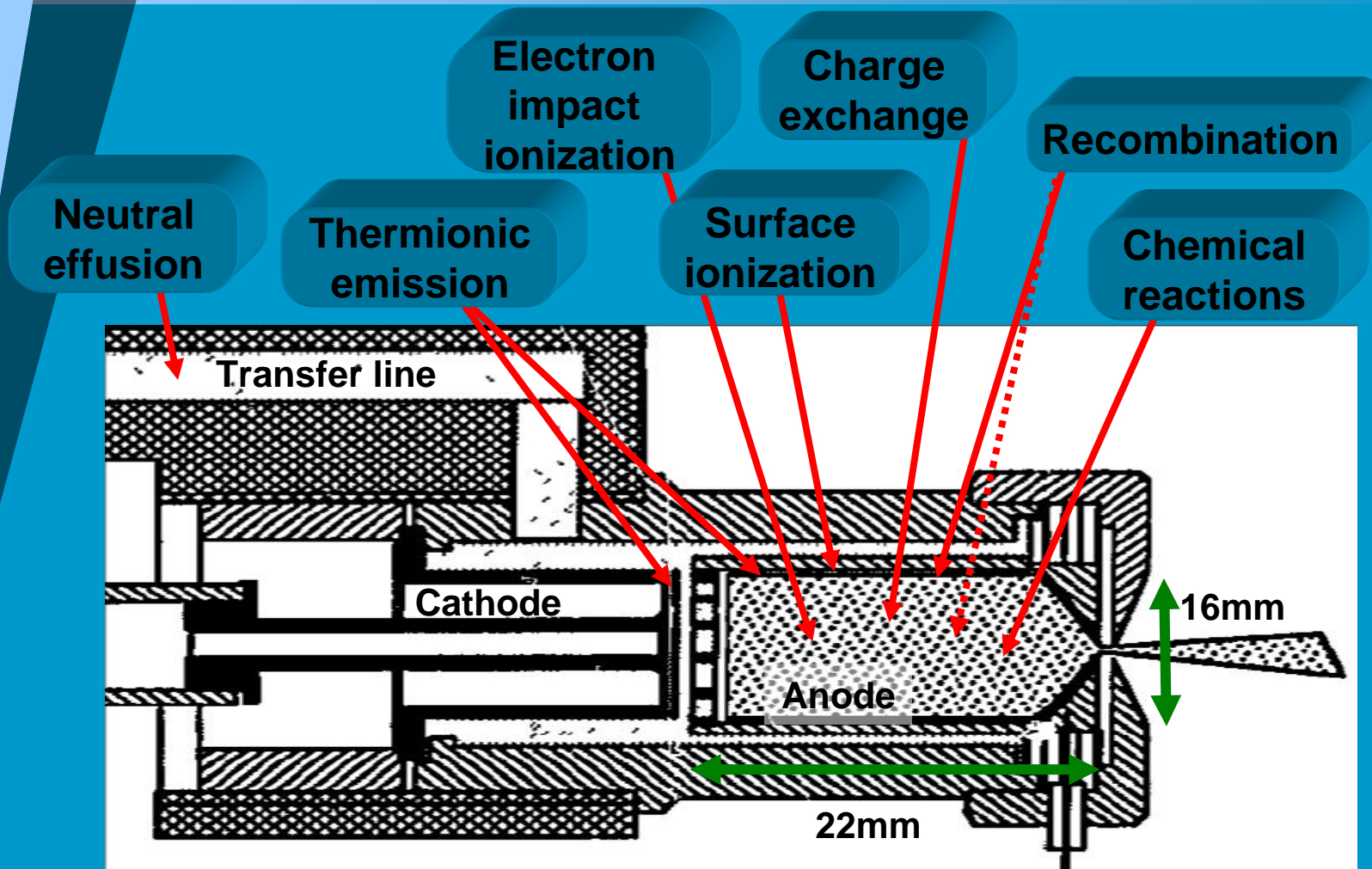
$$\Delta H_a < \sim 6\text{eV}$$

Ionised elements:

Period	Ion source:																					
1	1 H																	2 He				
2	3 Li	4 Be															5 B	6 C	7 N	8 O	9 F	10 Ne
3	11 Na	12 Mg															13 Al	14 Si	15 P	16 S	17 Cl	18 Ar
4	19 K	20 Ca	21 Sc	22 Ti	23 V	24 Cr	25 Mn	26 Fe	27 Co	28 Ni	29 Cu	30 Zn	31 Ga	32 Ge	33 As	34 Se	35 Br	36 Kr				
5	37 Rb	38 Sr	39 Y	40 Zr	41 Nb	42 Mo	43 Tc	44 Ru	45 Rh	46 Pd	47 Ag	48 Cd	49 In	50 Sn	51 Sb	52 Te	53 I	54 Xe				
6	55 Cs	56 Ba	*	71 Lu	72 Hf	73 Ta	74 W	75 Re	76 Os	77 Ir	78 Pt	79 Au	80 Hg	81 Tl	82 Pb	83 Bi	84 Po	85 At	86 Rn			
7	87 Fr	88 Ra	**	103 Lr	104 Rf	105 Db	106 Sg	107 Bh	108 Hs	109 Mt	110 Ds	111 Rg										
			*	57 La	58 Ce	59 Pr	60 Nd	61 Pm	62 Sm	63 Eu	64 Gd	65 Tb	66 Dy	67 Ho	68 Er	69 Tm	70 Yb					
			**	89 Ac	90 Th	91 Pa	92 U	93 Np	94 Pu	95 Am	96 Cm	97 Bk	98 Cf	99 Es	100 Fm	101 Md	102 No					



# The arc discharge plasma



# Selectivity in the arc discharge plasma

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- Making use of the shifts in the **maximum/ minimum/ inflexion regions** for all possible phenomena;
- Selectivity for **different element classes** (*different mass, volatility, ionization mechanism or potential...*)

- **Neutral effusion;**

*Active parameters: temperature, volume*

- **Impact ionization;**

*Active parameters: cross section, secondary electron energy, pressure*

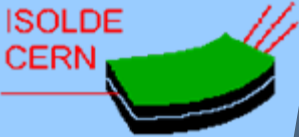
- **Charge exchange;**

*Active parameters: cross section, residence time, pressure*

- **Plasma confinement;**

*Active parameters: magnetic field, temperature, electron beam energy, pressure, source materials*

# Selectivity for different ionization mechanisms



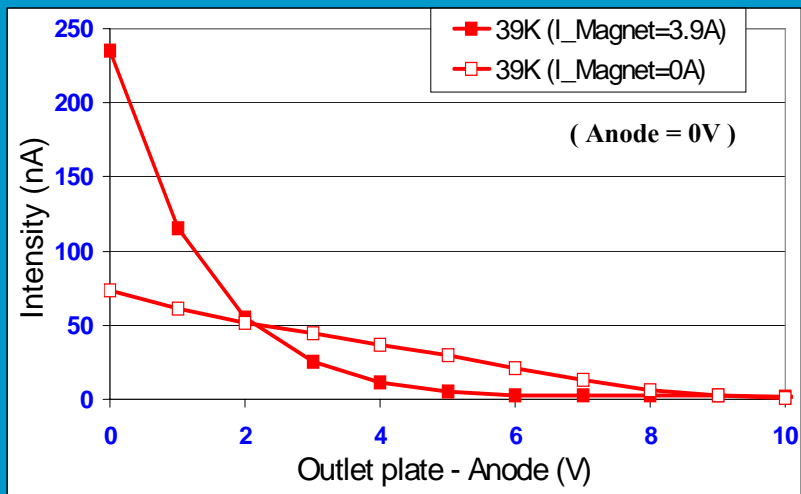
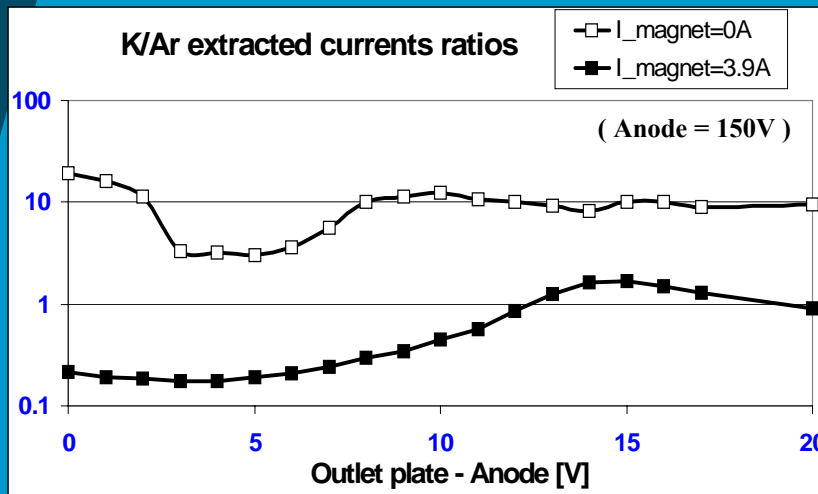
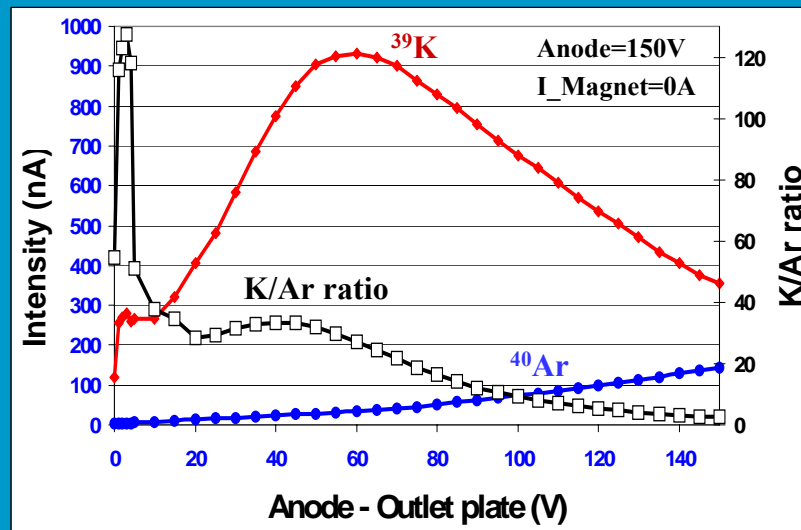
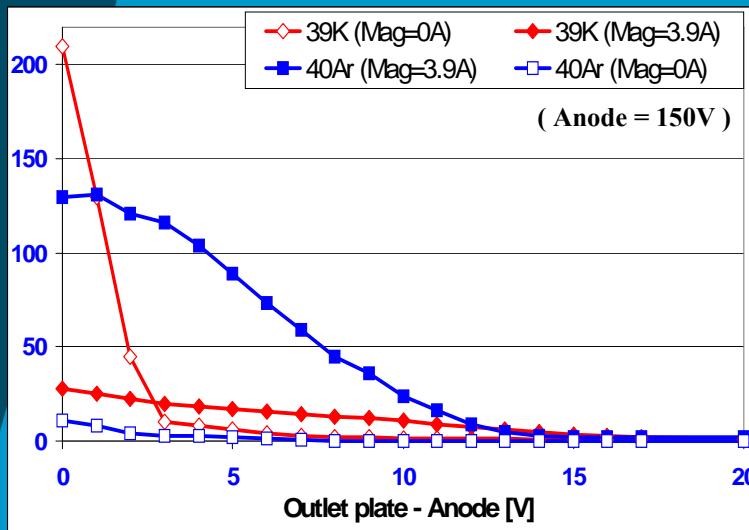
Electron impact ionization

(FEBIAD)

Surface ionization

(RILIS)

Laser ionization

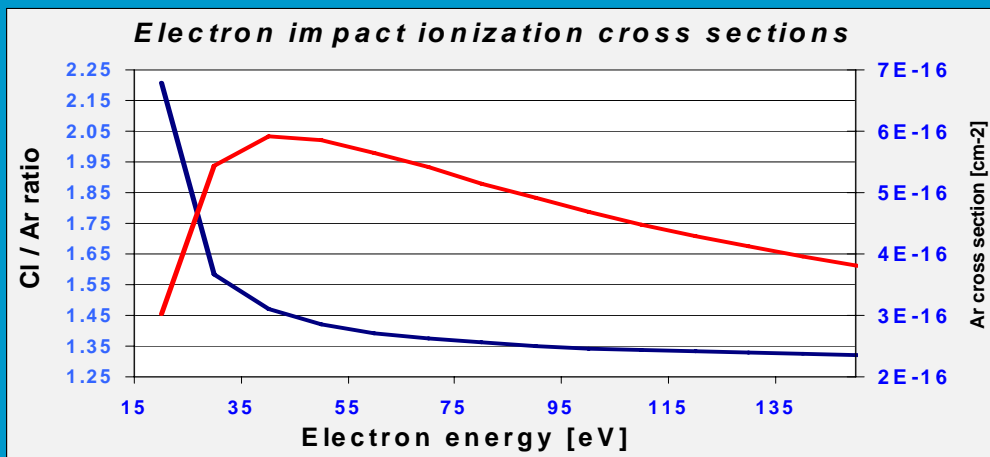


# Selectivity for different ionization potentials

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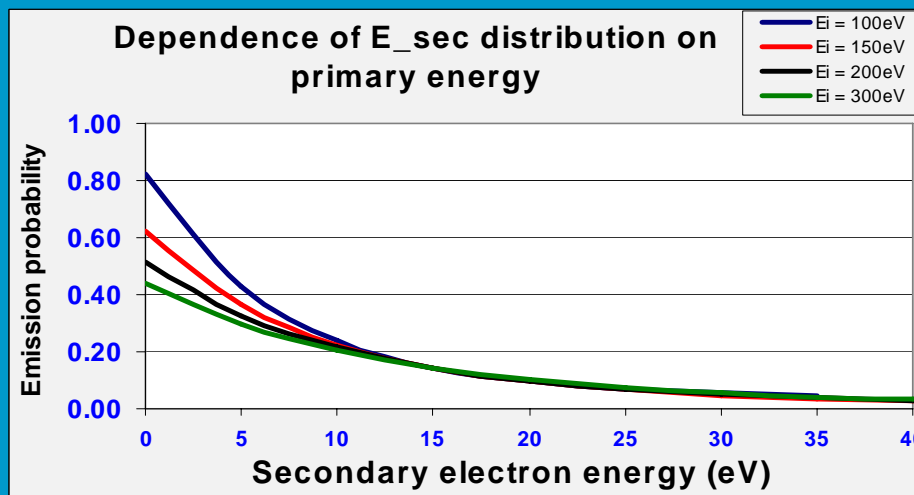
## • Difference in ionization cross section



Ar:  $E_i = 15.76$  eV  
Cl:  $E_i = 12.97$  eV

Primary electrons energy:  
 $0 \div 300$  eV

Secondary electrons energy:





# Selectivity for different ionization potentials

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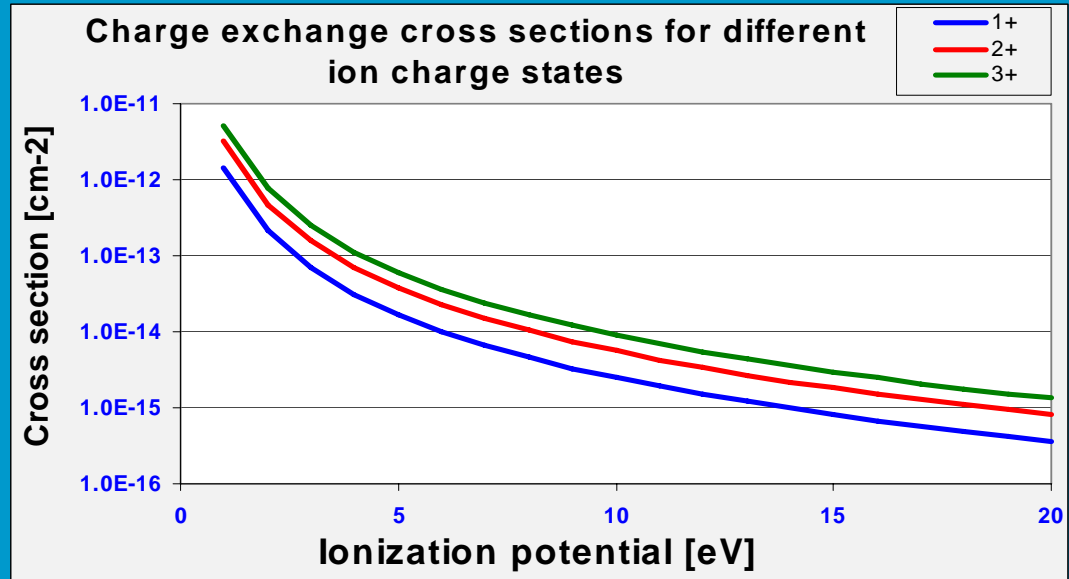


## • Difference in charge exchange cross section

Ar:  $E_i = 15.76$  eV

Cl:  $E_i = 12.97$  eV

=> Cl/Ar cross section  
ratio: 1.7



→ The ionization ratio between two elements can be estimated for the saturated regime (favouring the one with a lower ionization potential);

→ The ionization of the element with a higher ionization potential can be favored by reducing the ion lifetime in the plasma;

→ The dominant element in the ion source establishes a *threshold in the ionization potential* above which the ionization efficiency drops rapidly.



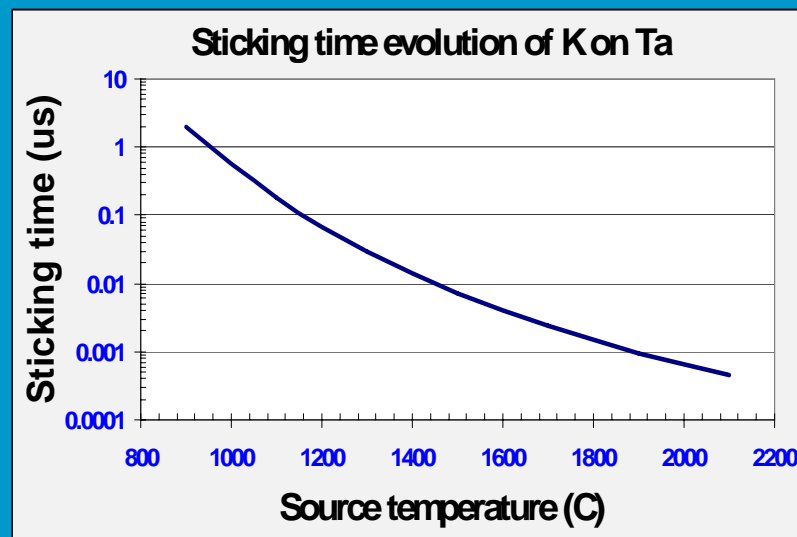
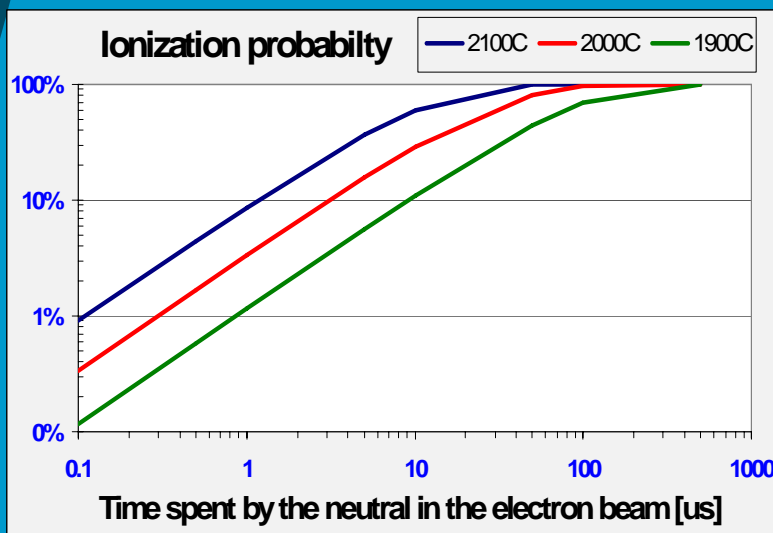
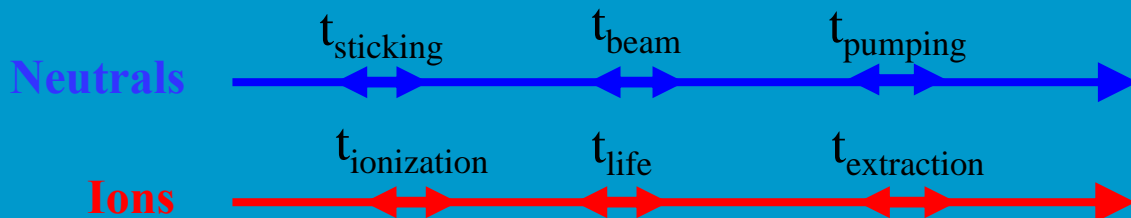


# Selectivity for different volatilities



→ Employed everywhere in the target&ion source unit, but having different weighting factors here

→ Useful for changing the ratio of the **specific times** of a **plasma discharge**:





# Selectivity for different masses

## → NEUTRAL STATE

Higher neutral residence time for higher masses:

$$t_i = k \cdot t_0 = k \cdot \frac{d_0}{v_i} = kd_0 \sqrt{\frac{m_i}{3k_b T}} \quad \Rightarrow \quad t_i \propto \sqrt{m_i}$$

$k$  – number of collisions

$d_0$  – average time between 2 wall collisions

} constants for a given geometry

## → ION STATE

Higher drain rates through plasma boundary for lower masses:

$$j_i = f \cdot n_i \cdot \sqrt{\frac{kT_{el}}{m_i}}$$

$f$  - Bohm factor

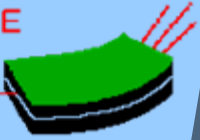
$T_{el}$  – electron temperature

# $^{35}\text{Cl}$ versus $^{35}\text{Ar}$

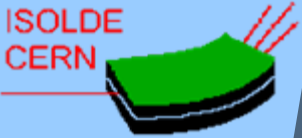
$^{35}\text{Ar}$  run; *Target: CaO; Ion Source: MK7 (GPS, October 2007)*

- Same ionization mechanism
- No mass difference
- Different chemical reactivity
- Different origin & path
- **Small difference in the ionization potential**
- **Different volatility**

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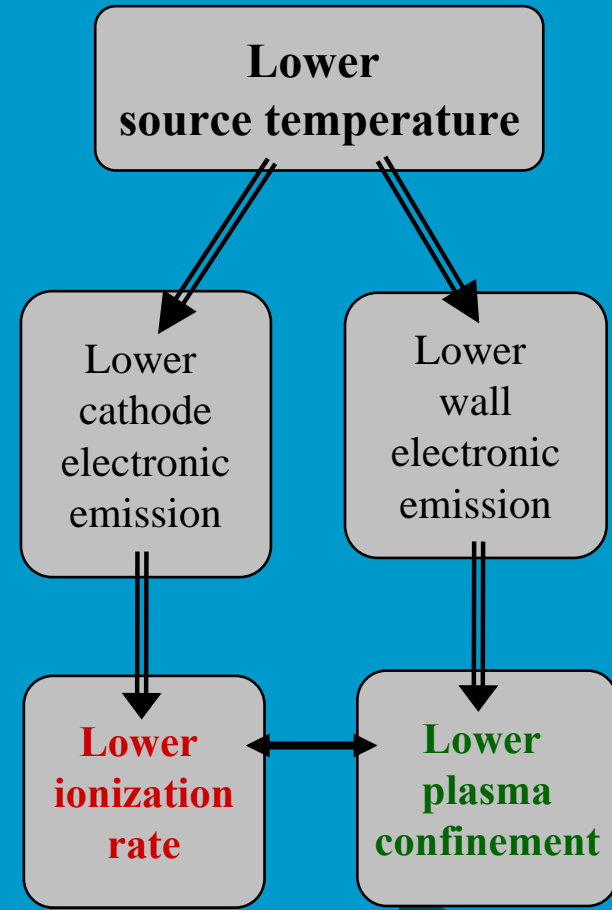
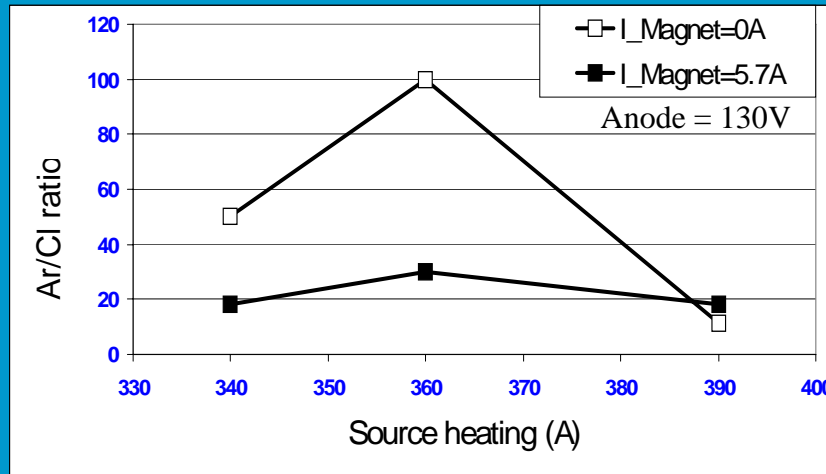
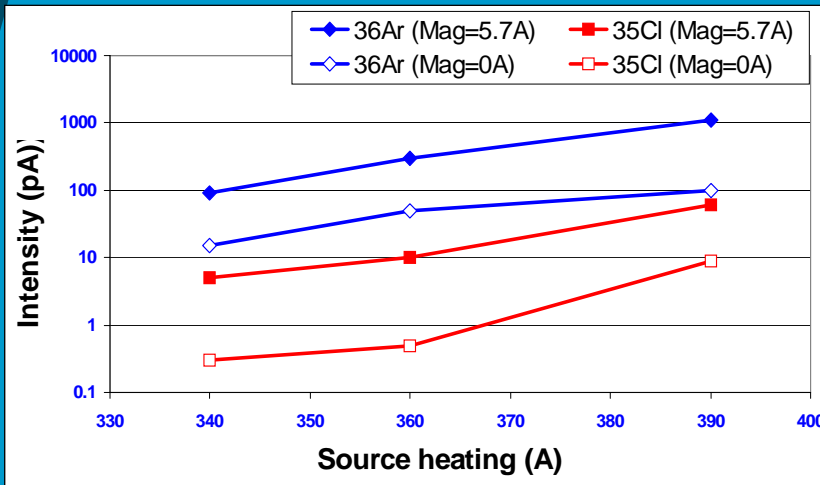
# On-line results



$^{35}\text{Ar}$  run; Target: CaO; Ion Source: MK7 (GPS, October 2007)



## Dependence on source temperature



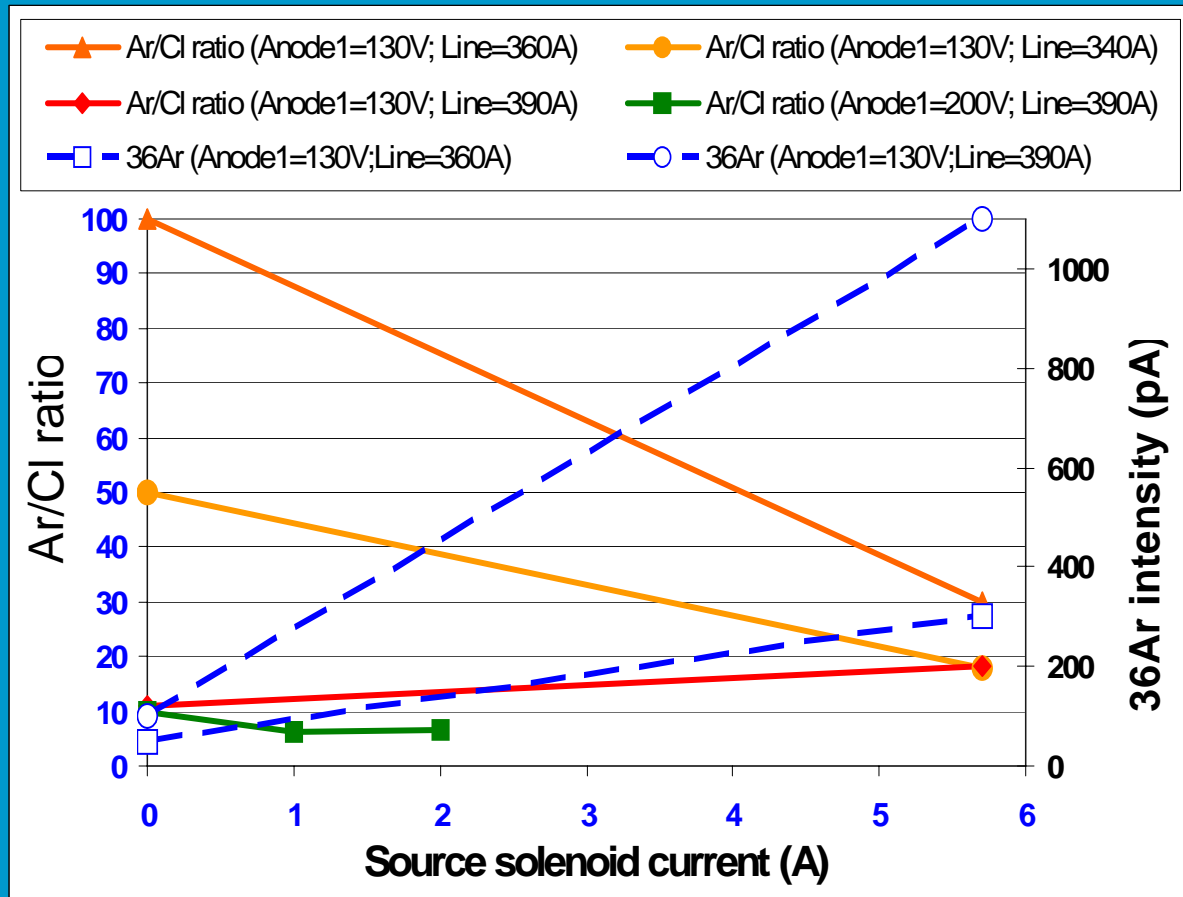
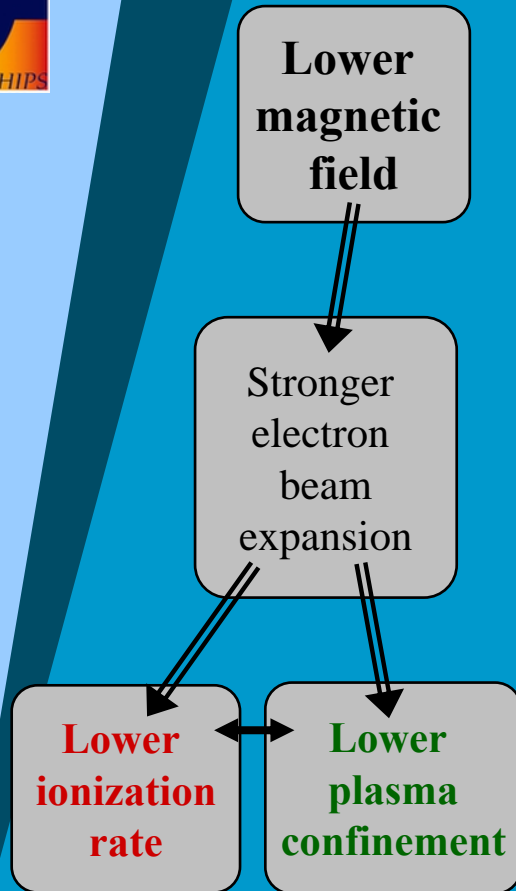
# On-line results

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$^{35}\text{Ar}$  run; Target: CaO; Ion Source: MK7 (GPS, October 2007)

## Dependence on the internal magnetic field

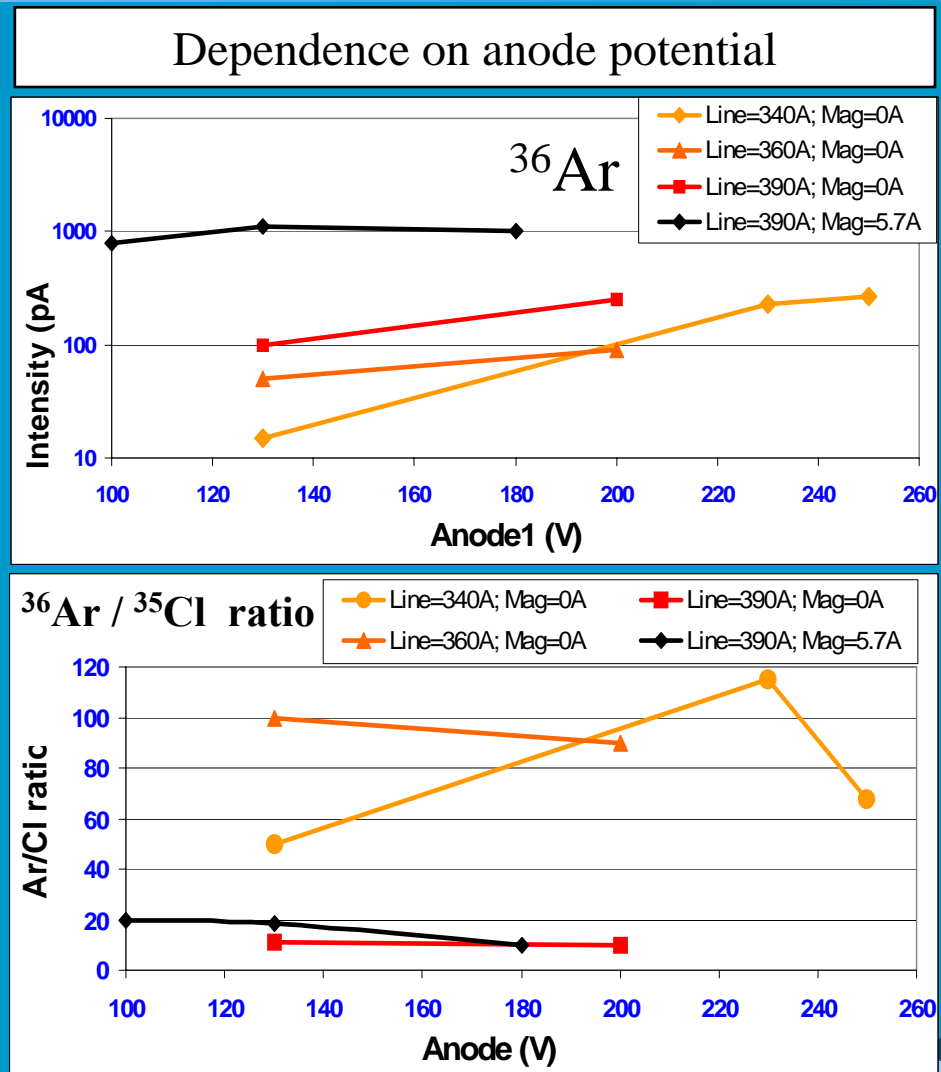
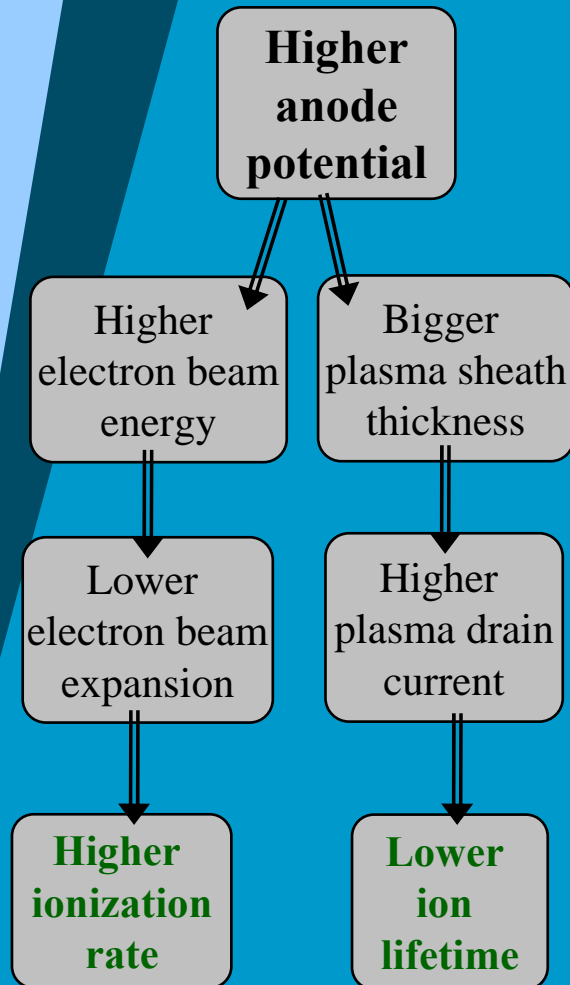


# On-line results

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$^{35}\text{Ar}$  run; Target: CaO; Ion Source: MK7 (GPS, October 2007)





# Ion source customization

## GOALS:

- Reducing the contaminants for specific elements of interest
- Improving the *ionization efficiency* and/or *source emittance* and/or *residence time*
- Optimization of laser ionization
- Adapting the ion source to higher operating pressures

## WAYS:

- Electric field configuration
- Ion source geometry
- Wall materials favouring *electron emission* or *surface ionization*
- Extraction system





# Tools for customization

## THEORY

## SIMULATION

- RIBO (“**R**adioactive **I**on **B**eam **O**ptimizer”)
- CPO (“**C**harged **P**article **O**ptics”)
- VORPAL (“**V**ersatile , **O**bject-oriented, **R**elativistic, **P**lasma Analysis code with **L**asers ”)

## EXPERIMENT

- Off-line separator
- Emittance meter
- Response time measurement system (under testing)
- Beam energy measurement system (under testing)

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*Thank you  
for  
your attention!*