R. Mantovan<sup>1</sup>, M. Fanciulli<sup>1</sup>, H.P. Gunnlaugsson<sup>2</sup>, G.Weyer<sup>2</sup>,

R. Sielemann<sup>3</sup>, D. Naidoo<sup>4</sup>, K. Bharuth-Ram<sup>5</sup>, S. Olafsson<sup>6</sup>, G. Langouche<sup>7</sup>, K. Johnston<sup>8</sup>

<sup>1</sup>Laboratorio Nazionale MDM CNR-INFM, Agrate Brianza (MI) 20041, Italy <sup>2</sup>Department of Physics and Astronomy,University of Aarhus, DK-8000 Århus C, Denmark

<sup>3</sup>*Hahn-Meitner Institute, 14109 Berlin, Germany* 

<sup>4</sup>School of Physics, University of the Witwatersrand, WITS 2050, South Africa
<sup>5</sup>School of Physics, University of KwaZulu-Natal, Durban 4041, South Africa
<sup>6</sup>Physics Department, Science Institute University of Iceland, Iceland
<sup>7</sup>Department of Physics, Katholieke Universiteit, Leuven, Belgium
<sup>8</sup>EP Division, CERN, CH-1211 Geneva 23, Switzerland

(CERN-INTC-2006-005; INTC-**P-203**.)





#### Outline

#### Motivation

- <sup>57</sup>Fe Mössbauer spectroscopy at ISOLDE-CERN
- Mössbauer results on  ${}^{57}$ Mn ( $\rightarrow {}^{57}$ Fe) -implanted ZnO
- Dose dependence effects
- An explanation for the observed magnetism in terms of defects
- Conclusions







#### Magnetism in ZnO

[T. Dietl, H. Ohno, F. Matsukura, J. Cibert., and D. Ferrand, Science **287**, 1019 (2000)]



Fig. 3. Computed values of the Curie temperature  $T_{\rm C}$  for various p-type semiconductors containing 5% of Mn and 3.5  $\times$  10<sup>20</sup> holes per cm<sup>3</sup>.



•Magnetic oxides/semiconductors: multifuncional materials for spintronics

•Curie temperature >>300 K is needed





#### Magnetism in ZnO: contradictory experimental findings

Compound	TM content	Substrate	Fabrication method	Growth temperature (°C)	Oxygen pressure (Torr)	Post-annealing	$T_{c}(\mathbf{K})$	Notes
ZnO:Mn	< 0.35	c-Sapphire	PLD	600	$5 \times 10^{-5}$		N/A	
ZnO:Mn	0.36	c-Sapphire	PLD	600	$5 \times 10^{-5}$		N/A	Spin-glass
Zn1_xTMxO		c-Sapphire	PLD	500-600	1 × 10 <sup>-9</sup> to 10 <sup>-6</sup>		N/A	
ZnO:Co	0.02 - 0.5	c-Sapphire	PLD	300-700	1 × 10 <sup>-6</sup> to 10 <sup>-1</sup>			Spin-glass
ZnO:Mn	0.01-0.36	c-Sapphire	PLD	610	$5 \times 10^{-5}$			Paramagnetic
ZnO:(Co, Mn, Cr, or Ni)	0.05-0.25	r-Sapphire	PLD	350-600	$2-4 \times 10^{-5}$		280-300	2 µB/Co
ZnO:Ni	0.01-0.25	c-Sapphire	PLD	300-700	$1 \times 10^{-5}$			Superparamagnetic or ferromagnetic
ZnO:V	0.05-0.15	r-Sapphire	PLD	300	10 <sup>-5</sup> to 10 <sup>-3</sup>		>350	$0.5 \mu_{\rm B} V^{-1}$
ZnO:(Co, Fe)	< 0.15	SiO <sub>2</sub> /Si	Magnetron sputtering	600	$2 \times 10^{-3}$	600 °C, 10 min, 1.0 × 10 <sup>-5</sup> Ton	>300	12-15 emu cm <sup>-3</sup>
ZnO:Co	0.03-0.05	Bulk ZnO	Ion implantation			700 °C, 5 min under O2	>300	Oriented Co precipitates
ZnO:Co	0-0.25	c-Sapphire	Sol-gel	<350		700 °C, 1 min	>350	0.56 μ <sub>B</sub> /Co
ZnO:Mn	0-0.3	c-Sapphire	PLD				>30-45	0.15-0.17 μg/Mn
ZnO:Mn	< 0.04		Sintered pellets	500-700	Air, atmospheric		>425	0.006 emu gm <sup>-1</sup> ,
			-		pressure			single phase
ZnO:Mn	0.02	Fused quartz	PLD	400			>425	0.05 emu gm <sup>-1</sup> , single phase
ZnO:(Fe, Cu)	0-0.1	Solid state reaction	897				5.50	0.75 μ <sub>B</sub> /Fe
ZnO:Co	0.015		PLD	650	$5 \times 10^{-5}$		>300	Ferromagnetic
ZnO:(Co, Al)	0.04 - 0.12	Glass	RF sputtering		1 × 10 <sup>-2</sup> in Ar		>350	0.21 µB/Co
ZnO:Mn	0.04-0.09	c-Sapphire	Reactive sputtering	200-380			>400	3 μ <sub>B</sub> /Co
ZnO:(Mn, Sn)	0-0.3		Implantation			5 min, 700 °C	2.50	Ferromagnetic

#### [S. J. Pearton et al., Semicond. Sci. Technol. **19** (2004) R59–R74.]

•The (3d-) doped and pure ZnO has been reported to be ferromagnetic, paramagnetic, and in the spin glass state

•The origin of the magnetism in ZnO is still a matter of debate in the scientific community





#### Mössbauer spectroscopy at ISOLDE-CERN (IS-443)

- Implantation of ZnO with radioactive  ${}^{57}Mn^+$  (T<sub>1/2</sub> = 1.5 min.), decaying to the  ${}^{57m}$ Fe Mössbauer level (T<sub>1/2</sub> = 100 ns)
- 100% element selective → structural and magnetic information at the Fe sites
- The implantation process induces a large amount of defects (~10<sup>3</sup>/ion)
   → their interaction with the implanted Mn/Fe probe can be studied
- Fluences ~10<sup>10</sup>-10<sup>13 57</sup>Mn<sup>+</sup>/cm<sup>2</sup> to record a Mössbauer spectrum (local concentration ~10<sup>16</sup> cm<sup>-3</sup>)→single isolated Fe impurities (~10<sup>-5</sup> Fe at.%) →truly dilute semiconductors are produced

## study of the magnetism at the atomic scale and role of defects





#### Mössbauer spectroscopy at ISOLDE-CERN (IS-443)







#### Mössbauer spectroscopy at ISOLDE-CERN (IS-443)



•Typically ~5 min to record one spectrum





#### Hyperfine interactions at the <sup>57</sup>Fe nuclear sites





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#### Hyperfine interactions at the <sup>57</sup>Fe nuclear sites



$$E_m = -g\mu_N Hm_I$$

$$\mu = -g\mu_N m_I$$



# Mössbauer sextet means magnetism at the Fe site





#### Mössbauer results: strong magnetic contribution >300 K



[G. Weyer, H.P. Gunnlaugsson, R.

Mantovan, M. Fanciulli, D. Naidoo, K.

Bharuth-Ram, T. Agne, J. Appl. Phys.

Mössbauer component	B <sub>hf</sub> (T)	Isomer shift, IS (mm/s)	Quadrupole splitting, QS (mm/s)
D2	-	0.80(1)	0.3(2)
D3	-	0.55(2)	0.73(3)
Sextet	48.3(2)	0.20(2)	0.13(3)
DistIII	37(2)	0.13(4)	-0.83(5)
Distll	12(3)	0.82(5)	-0.11(8)

no secondary phases
the observed magnetism
must be due to the
interaction of Mn/Fe with the
dense damage cascades
created during implantation



**102**, 113915 (2007)]



#### Mössbauer results: annealing behavior







#### Mössbauer results: time delayed measurements

•Time-dependence of the thermally activated annealing. Principles:

- 1. Implant (t = 0)
- 2. Start measurement
- 3. Measure  $(0 < t < T\frac{1}{2})$
- 4. Measure  $(T\frac{1}{2} < t < 5T\frac{1}{2})$





•The conversion of the sextet→D2 occurs on a minute time scale

•The time and temperature dependence of spectral fractions suggest a scenario in which the association of  $Fe_{Zn}$  with defect complexes takes place in the 300/462 K and the dissociation of the complex is at  $\geq$ 600 K



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#### Dose dependence in Mn/Fe-implanted ZnO

•At the early stage of implantation there is a low magnetic contribution

•The sextet+distIII fractions increases with increasing the implanted dose



The implanted dose is estimated by considering the background Mössbauer counts and a constant fluence

### The dose dependence can be understood by assuming that the implantation-induced defects are responsible for the observed magnetism









Implantation of 3d elements is possibly not needed to observe magnetism in ZnO, but the implantation-induced defects are necessary



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#### TRIM for the <sup>57</sup>Mn and <sup>23</sup>Na implantations in ZnO



The damage cascades after the <sup>57</sup>Mn and <sup>23</sup>Na implantation (E=60 keV) overlap







[G. Weyer, H.P. Gunnlaugsson, R. Mantovan, M. Fanciulli, D. Naidoo, K. Bharuth-Ram, T. Agne, J. Appl. Phys. **102**, 113915 (2007)]



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Defects induced by ion implantation

[G. Weyer, H.P. Gunnlaugsson, R. Mantovan, M. Fanciulli, D. Naidoo, K. Bharuth-Ram, T. Agne, J. Appl. Phys. **102**, 113915 (2007)]

![](_page_16_Picture_4.jpeg)

![](_page_16_Picture_6.jpeg)

![](_page_17_Figure_1.jpeg)

[G. Weyer, H.P. Gunnlaugsson, R. Mantovan, M. Fanciulli, D. Naidoo, K. Bharuth-Ram, T. Agne, J. Appl. Phys. **102**, 113915 (2007)]

![](_page_17_Picture_3.jpeg)

![](_page_17_Picture_5.jpeg)

Assumption:  $V_{Zn}$  are mobile at  $\ge$  300 K

![](_page_18_Figure_2.jpeg)

[G. Weyer, H.P. Gunnlaugsson, R. Mantovan, M. Fanciulli, D. Naidoo, K. Bharuth-Ram, T. Agne, J. Appl. Phys. **102**, 113915 (2007)]

![](_page_18_Picture_4.jpeg)

![](_page_18_Picture_5.jpeg)

![](_page_18_Picture_6.jpeg)

Assumption:  $V_{Zn}$  are mobile at  $\ge$  300 K

![](_page_19_Figure_2.jpeg)

[G. Weyer, H.P. Gunnlaugsson, R. Mantovan, M. Fanciulli, D. Naidoo, K. Bharuth-Ram, T. Agne, J. Appl. Phys. **102**, 113915 (2007)]

![](_page_19_Picture_4.jpeg)

![](_page_19_Picture_6.jpeg)

- •The majority of the Fe atoms are located on Zn sites in a high-spin Fe<sup>3+</sup> state at ≤600 K.
- •The sextet is due to  $Fe_{Zn}$ -O-V<sub>Zn</sub> complexes (V<sub>Zn</sub> mobile at T>RT on a minute time scale) that are stable up to ~600 K.
- •The formation/annealing of the magnetism in ZnO is interpreted as occurring/disappearing upon the association/dissociation of Mn/Fe complexes with the lattice defects created during the implantation process.

![](_page_20_Picture_4.jpeg)

![](_page_20_Picture_5.jpeg)

![](_page_20_Picture_6.jpeg)

- •The majority of the Fe atoms are located on Zn sites in a high-spin Fe<sup>3+</sup> state at  $\leq$ 600 K.
- •The sextet is due to  $Fe_{Zn}$ -O-V<sub>Zn</sub> complexes (V<sub>Zn</sub> mobile at T>RT on a minute time scale) that are stable up to ~600 K.
- •The formation/annealing of the magnetism in ZnO is interpreted as occurring/disappearing upon the association/dissociation of Mn/Fe complexes with the lattice defects created during the implantation process.

![](_page_21_Picture_4.jpeg)

![](_page_21_Picture_5.jpeg)

![](_page_21_Picture_8.jpeg)