Local Probe Studies on Highly Distorted Rare-Earth Manganites (IS390)

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## Outline

→ The Rare Earth's + MnO<sub>3</sub> (RMnO<sub>3</sub>) family and their questions
Ferroelectricity + Ferromagnetism (and Ferroelasticity)
→ Experimental method

Sample preparation and macroscopic characterization

<sup>111m</sup>Cd Perturbed Angular Correlation (PAC):

nanoscopic scale information  $\implies$  probing of local phenomena

 $\rightarrow$  PAC preliminary results

 $\rightarrow$  Summary

#### Multiferroics: Materials showing Ferroelectricity, Ferromagnetism (and Ferroelasticity)



 $BiMnO_2$  ( $Mn^{3+}$ )

48

M. Vopsaroiu et al., Journ. Phys. D: Appl. Phys. 40 (2007) 5027 N. Hill et al., Phys. Rev. B 59 (1998), 8759



**Orthorhombic Perovskites Pnma** 

#### Hexagonal P6<sub>3</sub>/m



Hexagonal RMnO<sub>3</sub> can be stabilized as orthorhombic, with synthesis under pressure





## RMnO<sub>3</sub> versus ionic radius

FE compounds Lu Yb Tm Er Ho Dy Tb Gd Eu Sm Nd Pr 150*±* RMnO<sub>3</sub> Mn3 03 Paramagnetic T<sub>AFM/PM</sub> (K) Mn2-Mn4 A-type AF -type 1.05 1.15 1.2 1.1  $r_R$  (Å) Magnetic Phase Diagram for RMnO<sub>3</sub> for

Orthorhombic structures M. Tachibana et a.,Phys. Rev. B 75 (2007) 144425

#### Influence of ionic radius on macroscopic properties

What happens at atomic scale?

Simple dependence on  $T_{N}$ Orthorhombic: increase with R<sub>i</sub> Hexagonal: decrease with R<sub>i</sub> **Hexagonal Orthorhombic** 140-120 AFM/PM 100 00-0 80 60 0-0 40 0.85 0.90 0.95 1.00 1.05 Lu Yb Tm Er Y Ho Tb Gd Eu Pr Ca La R ionic radius  $(R^{3+})$ 

## Studied Compounds

Compound	Structure	Т <sub>ағм/рм</sub> (К)	T <sub>FE</sub> (K)	Z <sub>Relement</sub>	Ionic radius <sub>R element</sub> (R <sup>3+</sup> ) (nm)
LuMnO <sub>3</sub>	Hexagonal (S.G. <i>P6<sub>3</sub>cm</i> )	90	>750	71	0.860
ErMnO <sub>3</sub>	Hexagonal (S.G.: <i>P6<sub>3</sub>cm</i> )	80	833	68	0.890
YMnO <sub>3</sub>	Hexagonal (S.G.: <i>P6<sub>3</sub>cm</i> )	71	920	39	0.900
HoMnO <sub>3</sub>	Hexagonal (S.G.: <i>P6<sub>3</sub>cm</i> )	76	873	67	0.901
GdMnO <sub>3</sub>	Orthorhombic (S.G.: <i>Pnma</i> )	42	<13	64	0.940
EuMnO <sub>3</sub>	Orthorhombic (S.G.: <i>Pnma</i> )	47	-	63	0.960

# Sample preparation and structural characterization





J.-S. Zhou et al., Phys. Rev. B 74 (2006), 014422 T. Mori et al., Materials Lett. 54 (2002), 238

## Perturbed Angular Correlations PAC



## **Room Temperature Results**



## 573 K Preliminary Results



#### Assymetry parameter vs temperature: EFG<sub>1</sub>





## EuMnO<sub>3</sub>: behavior as function of temperature



Decrease of W with temperature due to an increase of atomic vibrations (typical in other Perovskites)

Measurement@10K: R(t) spectrum different from other temperatures. Hints changes between 10k and 20K. Dielectric constant measurements show a peak below 20K.



#### Summary

#### <u>Hexagonal Manganites (R= Lu, Er, Y, Ho)</u>

Main EFG ( $EFG_1$ ) with assymetry parameter decreasing with bigger R ionic radius which seems independent from temperature.

Orthorhombic Manganites (R= Gd and Eu)

Two EFG's: main EFG (EFG<sub>1</sub>) with assymetry parameter increasing with R ionic radius.

The assymetry parameter evolution as function of the bond angle present the same behavior as  $T_{AFM/PM}$ : correlation between macroscopic and microscopic properties.

#### EuMnO<sub>3</sub>: Behavior as function of temperature

Highly distorted environments ( $\eta \sim 0.5$ ) for all range of temperatures. Frequency evolution following the trend found in other perovskites. Below 20K, changes in R(t) spectra hints an unknown transition (observed as a peak in dielectric measurements). Measurements on temperature in this range are metals for

#### **Future Perspectives**

#### your

Measurements on temperature for all compounds of this series. attention! Use of 3+ isotopes (e.g.  $^{117}Ag^{/117}In^{/117}Cd$ ) allowing the measurement of the same sample at different temperatures.

Synthesis of new multiferroic/magnetoelectric systems (BiMnO<sub>3</sub>, InMnO<sub>3</sub>, PrCdMnO<sub>3</sub>)