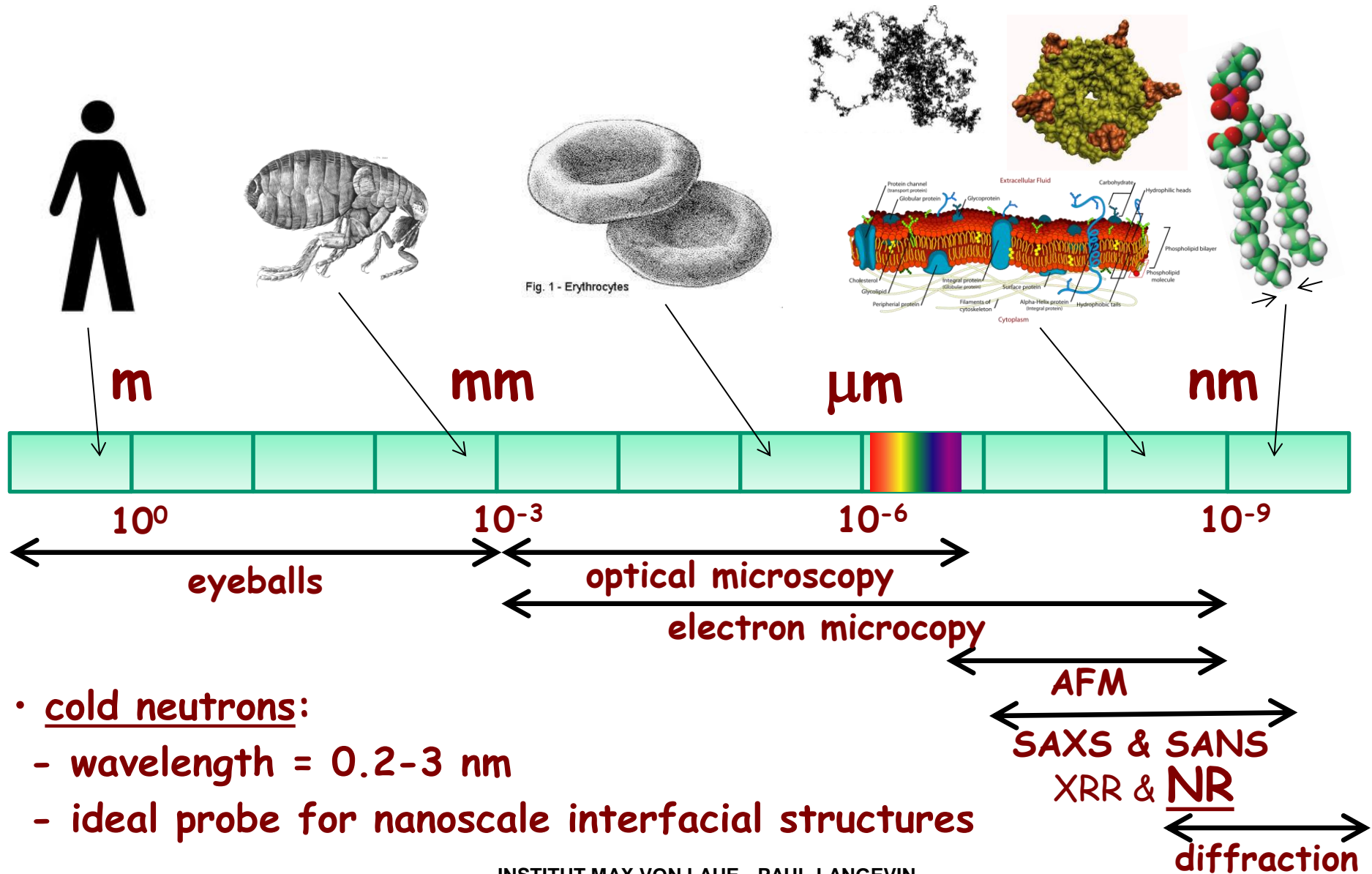


wavelengths for nanoscale interfacial analysis

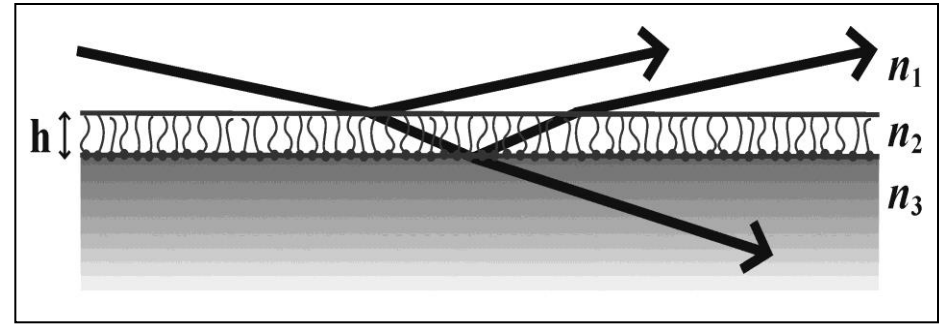


- cold neutrons:

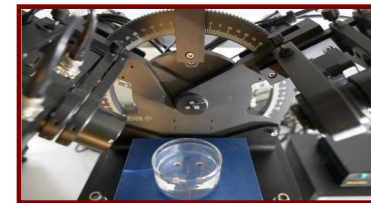
- wavelength = 0.2-3 nm

- ideal probe for nanoscale interfacial structures

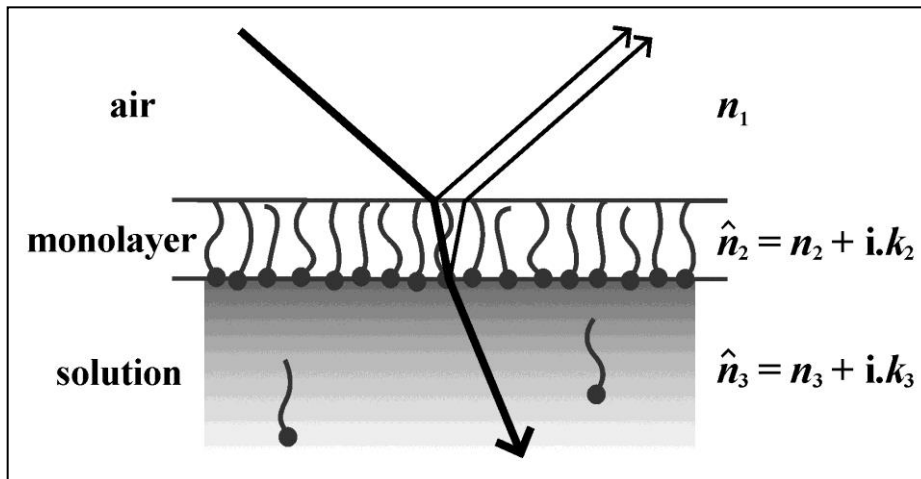
- neutron reflectometry:
 - contrast from isotopic labelling
 - specular \rightarrow structure & composition
 - off-specular \rightarrow lateral morphology



- ellipsometry:
 - precision, sensitivity & fast kinetics
 - calibration to real physical parameters



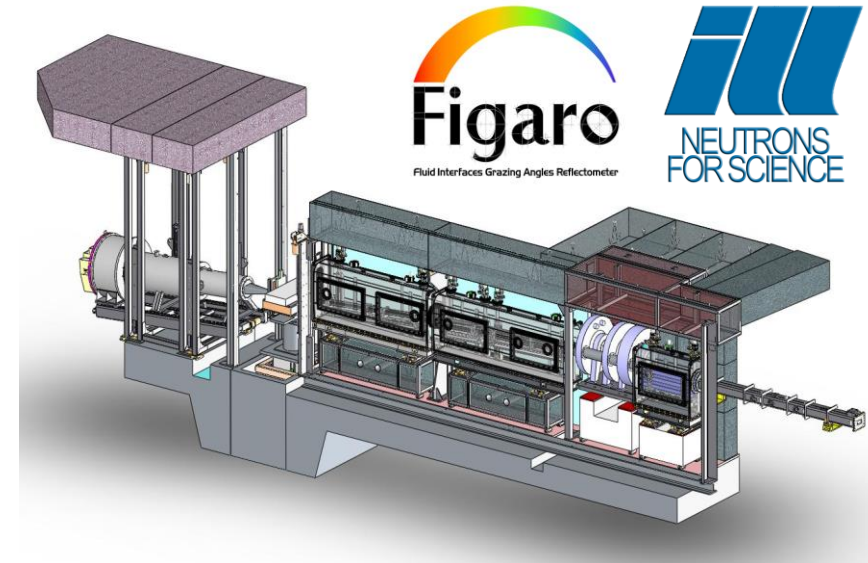
Nucleus	$\rho_s / 10^{-5} \text{ \AA}$
^1H	-3.7406
^2H	6.671
C	6.6460
N	9.36
O	5.803



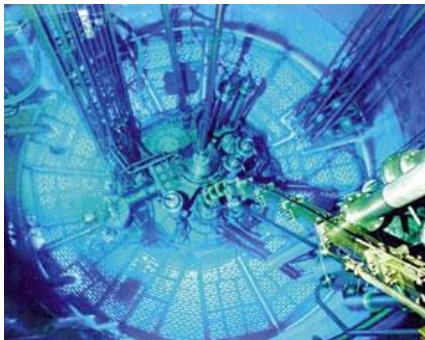
- other techniques:
 - surface tensiometry
 - Brewster angle microscopy
 - infrared reflectometry *etc.*

• figaro stats:

- world-leading instrument from 2009
- flux at sample: $4 \times 10^2 \text{ n mm}^{-2} \text{ s}^{-1}$
- large beam size: 0.5-5 mm x 40 mm
- flexible resolution: 1-10% dq/q
- dozens of publications: www.ill.eu



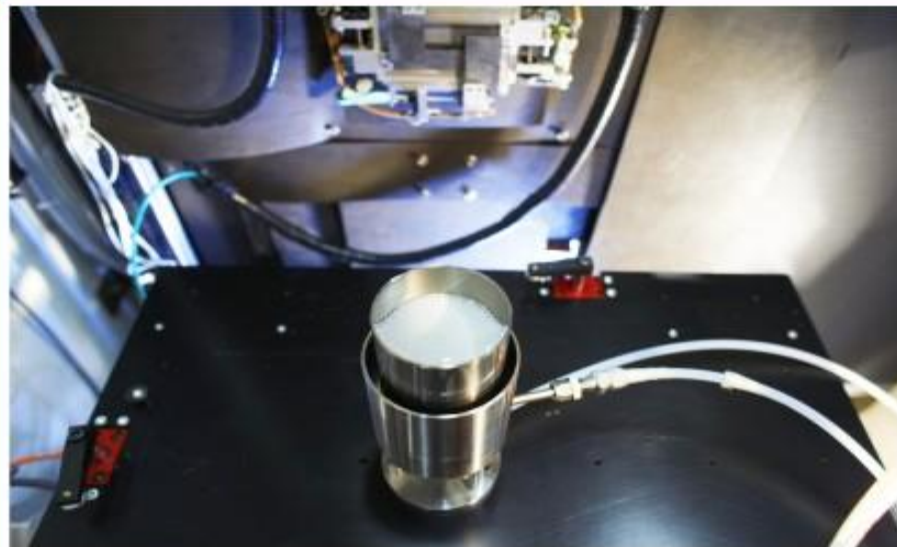
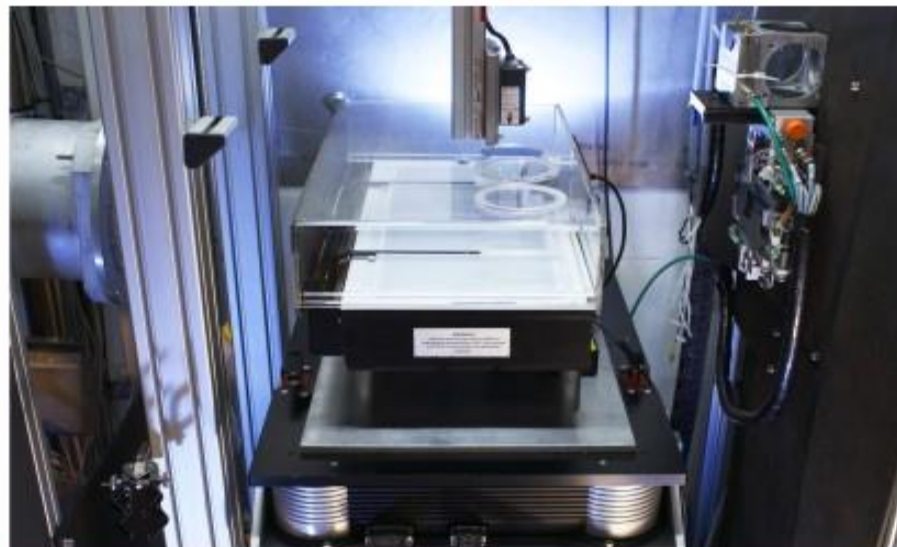
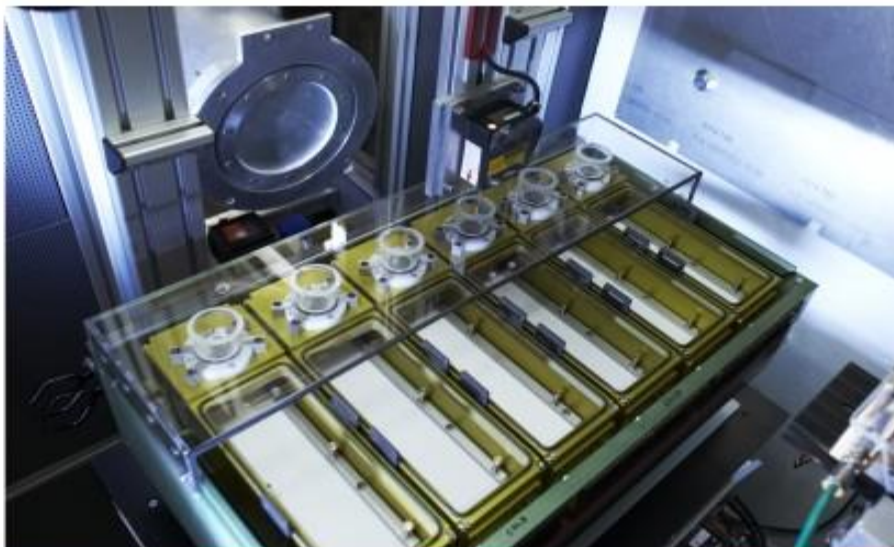
ILL's powerful & stable reactor

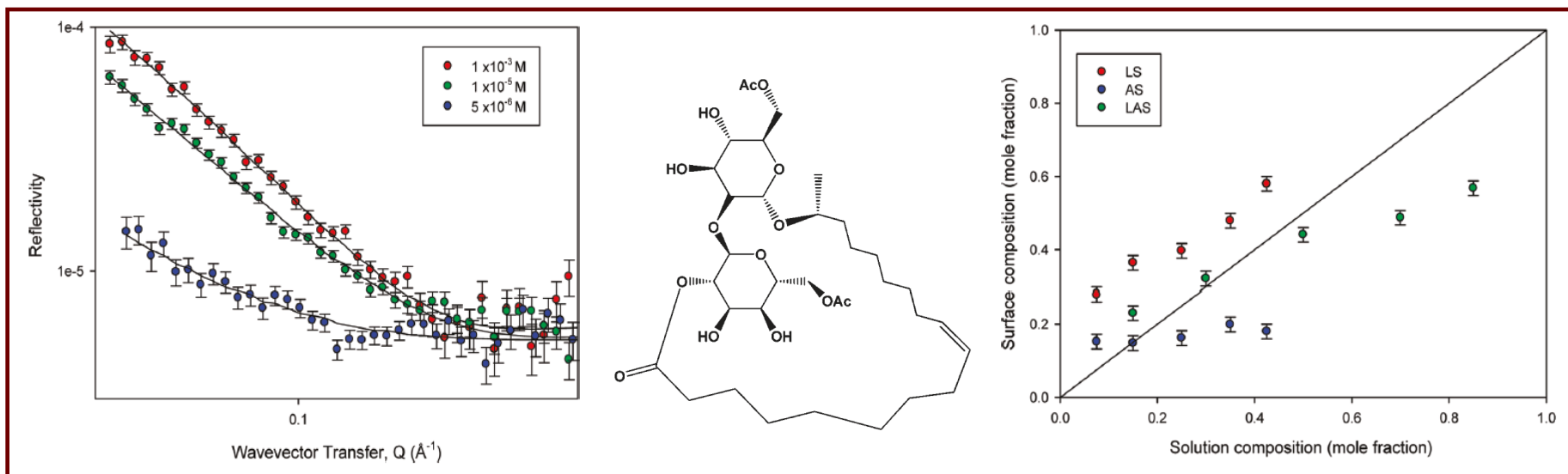


beautifully set in Grenoble

• figaro examples:

- polymer film structure
- nanoparticle interactions
- dna & protein studies
- solvent drying in glues & paints
- rheology of polymer blends
- detergency & formulation mechanisms





Langmuir ARTICLE
pubs.acs.org/Langmuir

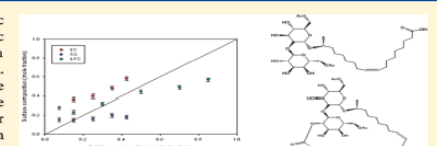
Adsorption of Sophorolipid Biosurfactants on Their Own and Mixed with Sodium Dodecyl Benzene Sulfonate, at the Air/Water Interface

Minglei Chen,[†] Chuchuan Dong,[†] Jeff Penfold,^{*,†,‡} Robert K. Thomas,[†] Thomas J. P. Smyth,[§] Amedea Perfumo,[§] Roger Marchant,[§] Ibrahim M. Banat,[§] Paul Stevenson,^{||} Alyn Parry,^{||} Ian Tucker,^{||} and Richard A. Campbell[⊥]

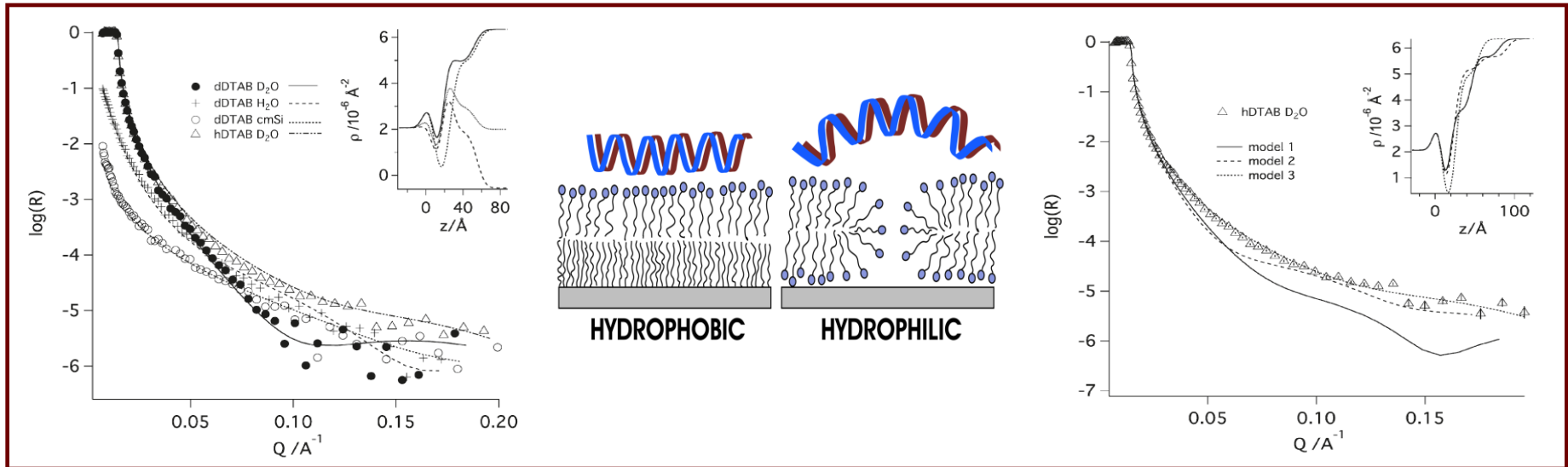
[†]Physical and Theoretical Chemistry Laboratory, University of Oxford, South Parks Road, Oxford, United Kingdom
[‡]ISIS, STFC, Rutherford Appleton Laboratory, Chilton, Didcot, OXON, United Kingdom
[§]School of Biomedical Sciences, University of Ulster, Coleraine, Northern Ireland
^{||}Unilever Research and Development Laboratory, Port Sunlight, Wirral, United Kingdom
[⊥]Institut Laue Langevin, 6 rue Jules Horowitz, F-38042 Grenoble, Cedex 09, France

Supporting Information

ABSTRACT: The adsorption of the lactonic (LS) and acidic (AS) forms of sophorolipid and their mixtures with the anionic surfactant sodium dodecyl benzene sulfonate (LAS) has been measured at the air/water interface by neutron reflectivity, NR. The AS and LS sophorolipids adsorb with Langmuir-like adsorption isotherms. The more hydrophobic LS is more surface active than the AS, with a lower critical micellar concentration, CMC, and stronger surface adsorption, with an area/molecule $\sim 70 \text{ \AA}^2$ compared with 85 \AA^2 for the AS. The



- drive to produce surfactants by microorganisms cf. petroleum
- surface composition of three-component mixtures deconvoluted
- biosurfactants dominated surface over conventional surfactants



- study of the interactions of dna with membranes (gene delivery)
- measurements resolved dna structure & surface excess
- dna coverage determined by that of self-assembled surfactant

Langmuir

ARTICLE

pubs.acs.org/Langmuir

Structure of DNA–Cationic Surfactant Complexes at Hydrophobically Modified and Hydrophilic Silica Surfaces as Revealed by Neutron Reflectometry

Marité Cárdenas,^{*,†} Hanna Wacklin,^{*,§} Richard A. Campbell,^{*,||} and Tommy Nylander^{||}

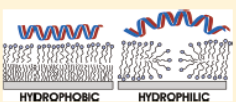
^{*}Nanoscience Center and Institute of Chemistry, Copenhagen University, Universitetsparken 5, DK-2100 Copenhagen E, Denmark

[†]Institut Laue-Langevin, 6 rue Jules Horowitz, BP 156, 38042 Grenoble, France

[§]European Spallation Source ESS AB, P.O. Box 176, 221 00 Lund, Sweden

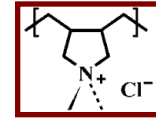
^{||}Physical Chemistry, Department of Chemistry, Lund University, P.O. Box 124, SE-22100 Lund, Sweden

ABSTRACT: In this article, we discuss the structure and composition of mixed DNA–cationic surfactant adsorption layers on both hydrophobic and hydrophilic solid surfaces. We have focused on the effects of the bulk concentrations, the surfactant chain length, and the type of solid surface on the interfacial layer structure (the location, coverage, and conformation of the DNA and surfactant molecules). Neutron reflectometry is the technique of choice for revealing the surface layer structure by means of selective deuteration. We start by studying the interfacial complexation of DNA with dodecyltrimethylammonium bromide (DTAB) and hexadecyltrimethylammonium bromide (CTAB) on hydrophobic surfaces, where we show that DNA molecules are located on top of a self-assembled surfactant monolayer, with the thickness of the DNA layer and the surfactant–DNA ratio determined by the surface coverage of the underlying cationic layer. The surface coverages of surfactant and DNA are determined by the bulk concentration of the surfactant relative to its critical micelle concentration (cmc). The structure of the interfacial layer is not affected by the choice of cationic surfactant studied.



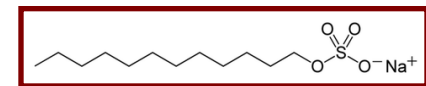
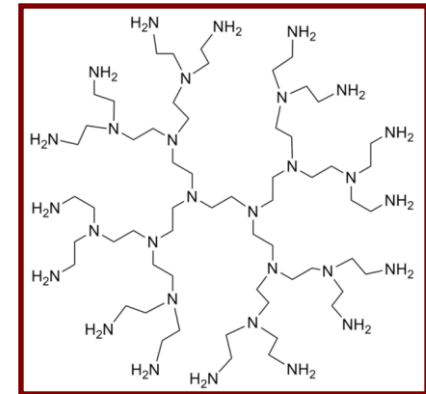
- cationic polyelectrolytes:

- poly(diallyldimethylammonium chloride) [Pdadmac]
- poly(ethylene imine) [PEI]
- used in detergents, adhesives, CO₂ extraction etc.
- PEI has high charge density at low pH due to protonation



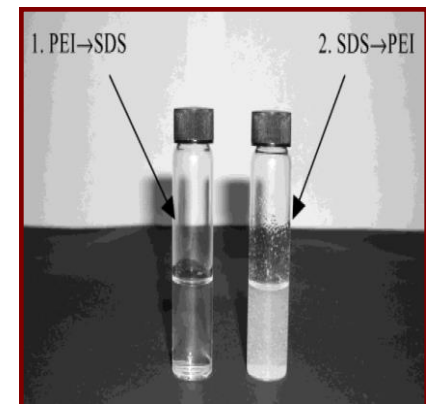
- anionic surfactants:

- sodium dodecyl sulfate [SDS]
- ubiquitous in shampoos & cleaning products
- binds to oppositely charged polyelectrolytes

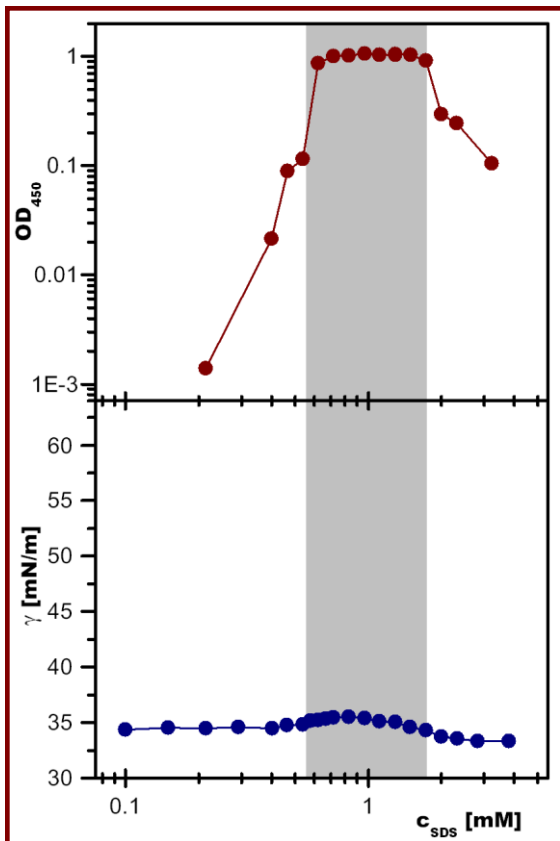


- polyelectrolyte/surfactant mixtures:

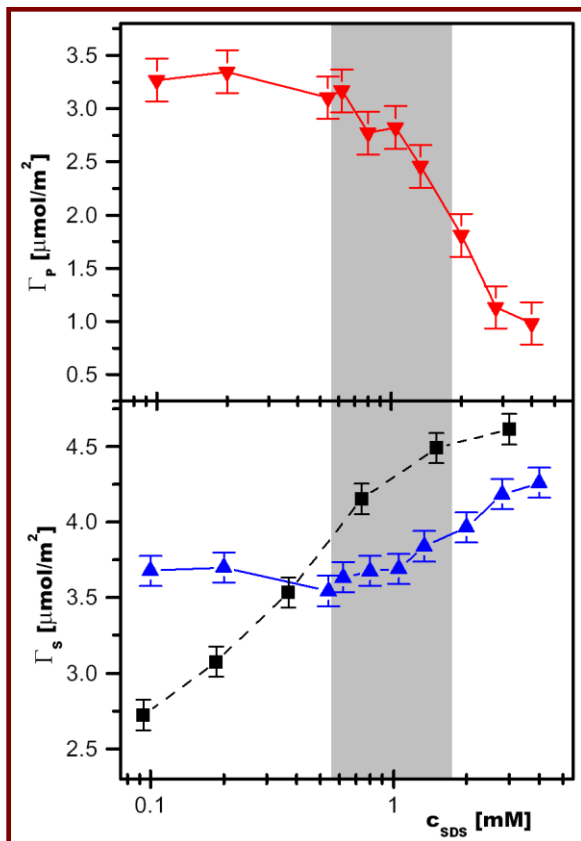
- used in household products (cleaning, food etc.)
- behavior interpreted by equilibrium self-assembly
- bulk non-equilibrium effects are also important



(1) fresh-mixed

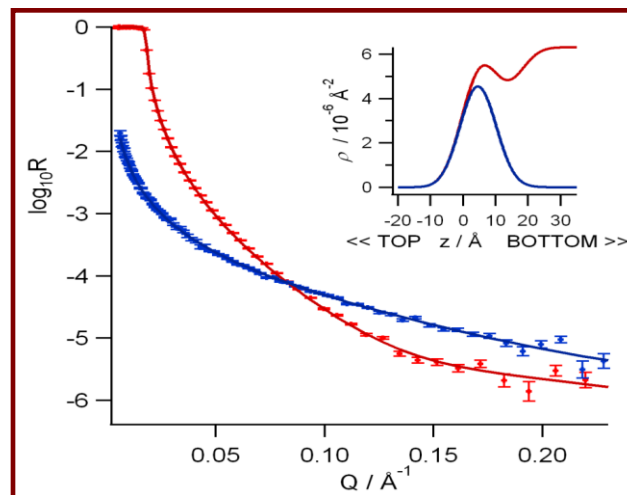


(2) aged-settled



(3) aged-redispersed

[Pdadmac] = 100 ppm
[SDS] = varied
[NaCl] = 100 mM

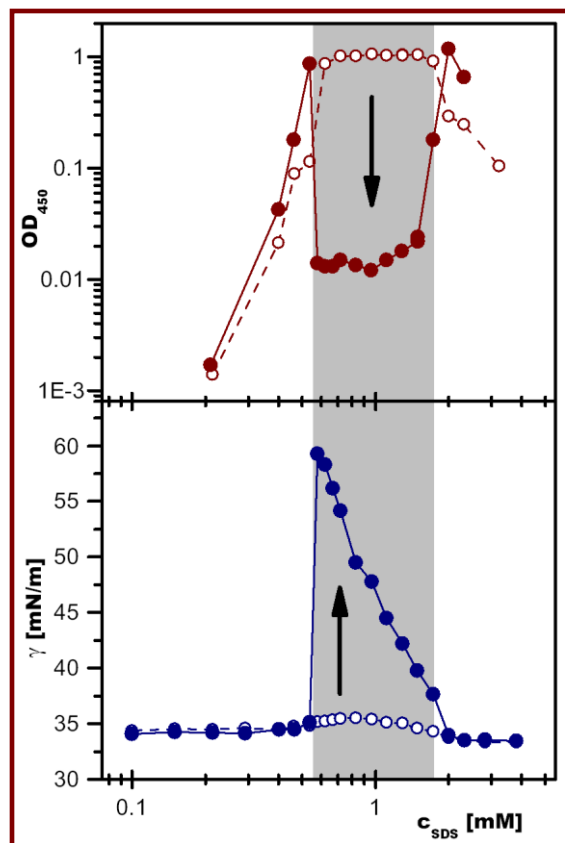


Campbell, Yanez, Angus-Smyth, Nylander & Varga, J Phys Chem B, 2011, 115, 15202

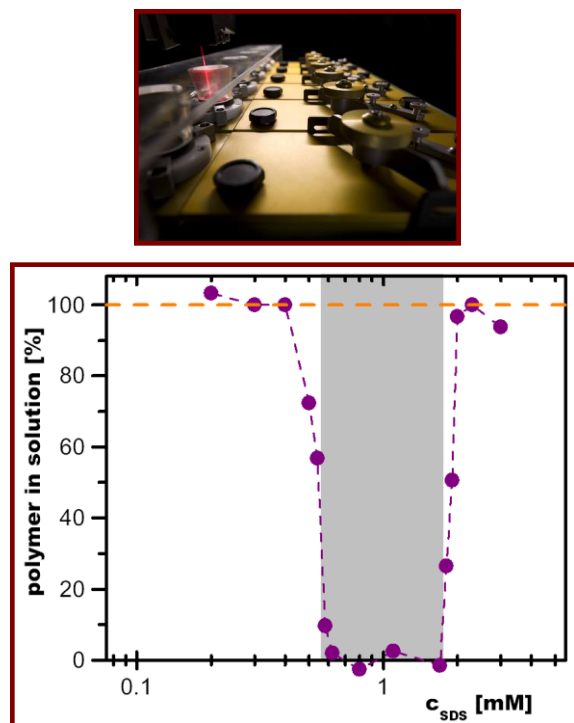
Campbell, Angus-Smyth, Yanez, Tonigold, Nylander & Varga, J Phys Chem Lett, 2010, 1, 3021

example 1: static air/liquid measurements of complex systems

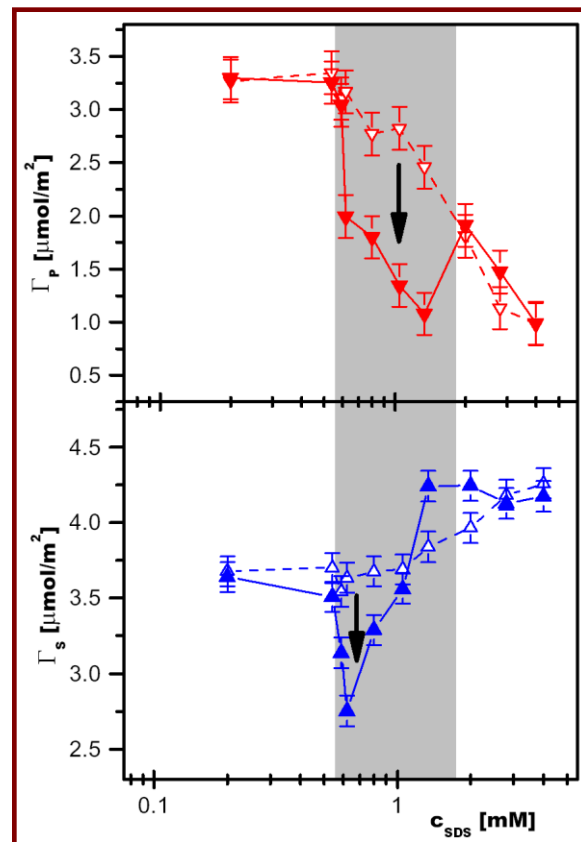
(1) fresh-mixed



(2) aged-settled



(3) aged-redispersed

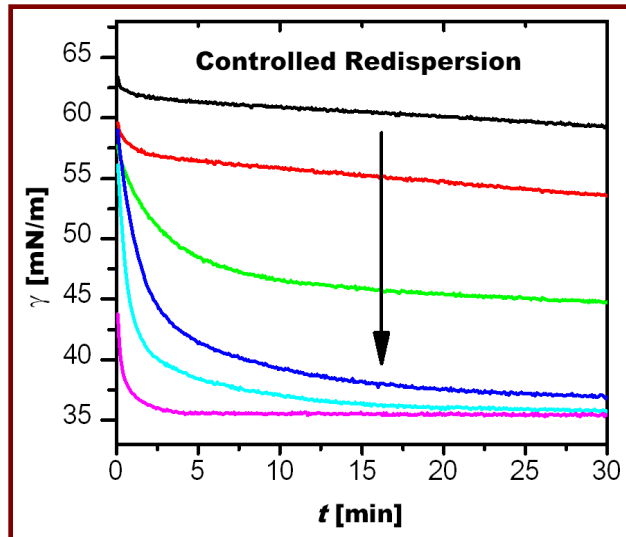


Campbell, Yanez, Angus-Smyth, Nylander & Varga, *J Phys Chem B*, 2011, 115, 15202

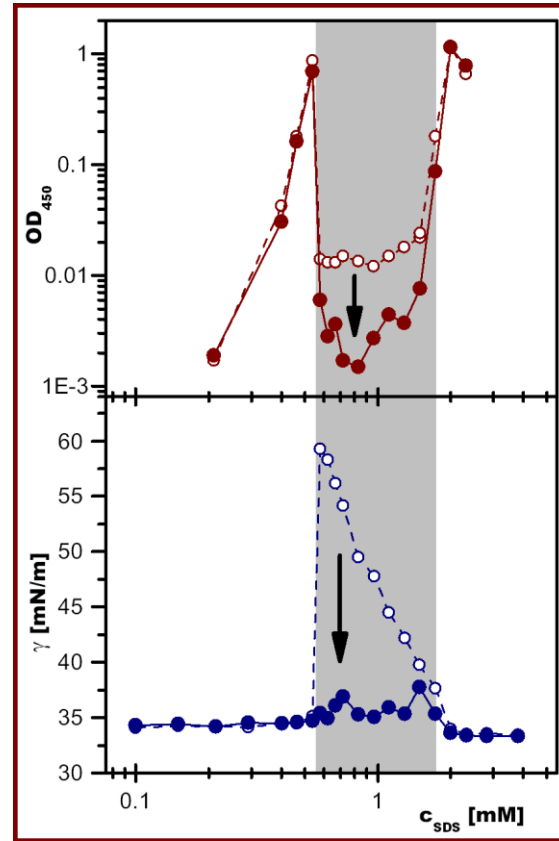
Campbell, Angus-Smyth, Yanez, Tonigold, Nylander & Varga, *J Phys Chem Lett*, 2010, 1, 3021

example 1: static air/liquid measurements of complex systems

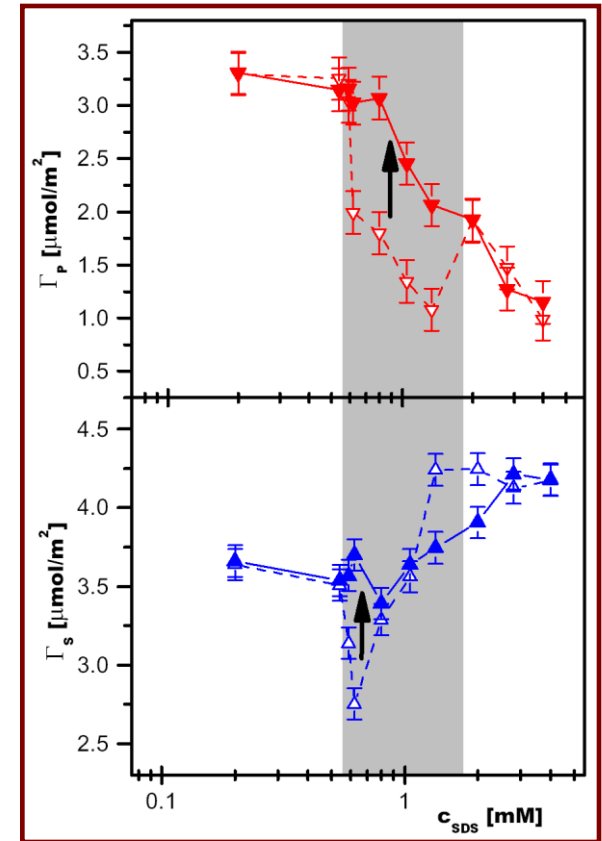
(1) fresh-mixed



(2) aged-settled



(3) aged-redispersed

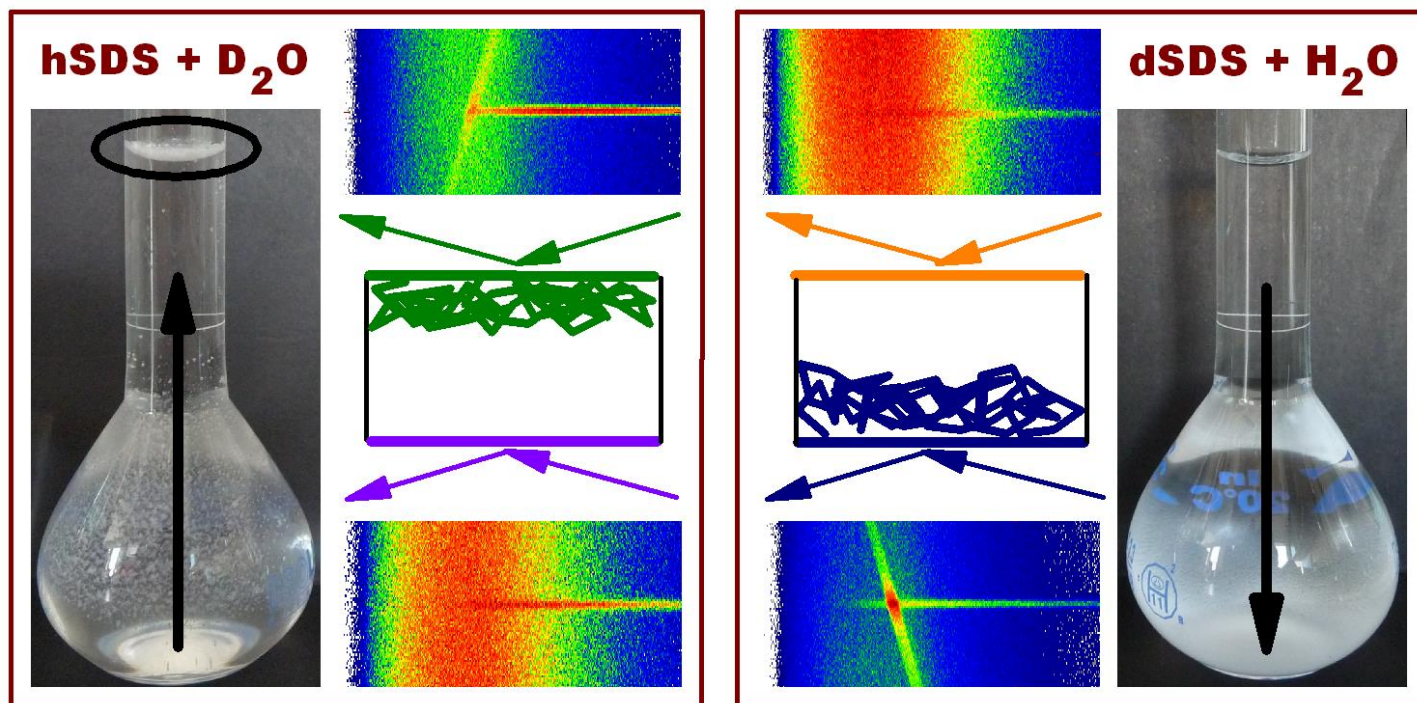


Campbell, Yanez, Angus-Smyth, Nylander & Varga, *J Phys Chem B*, 2011, 115, 15202

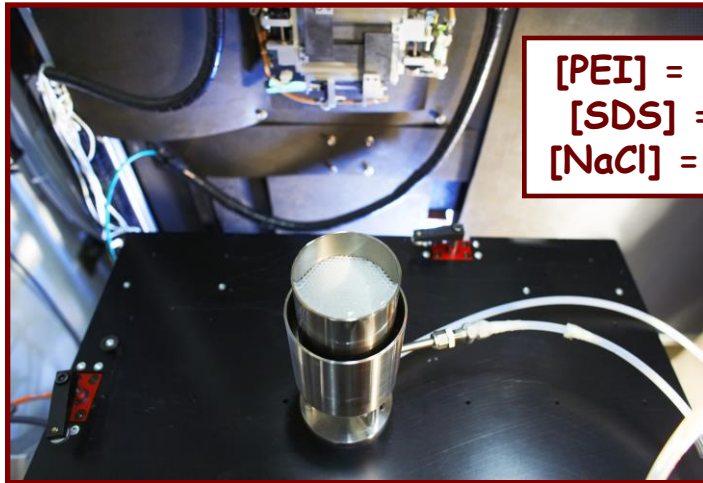
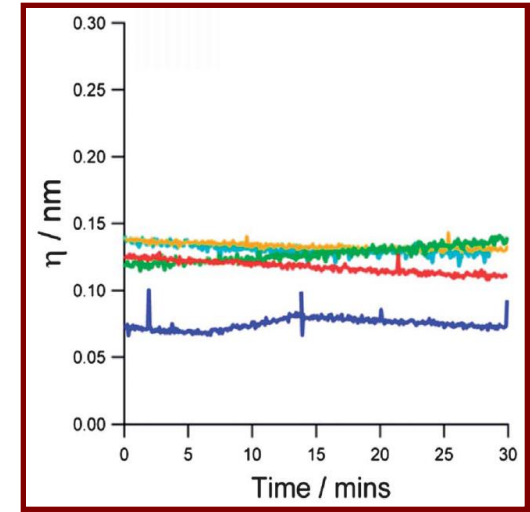
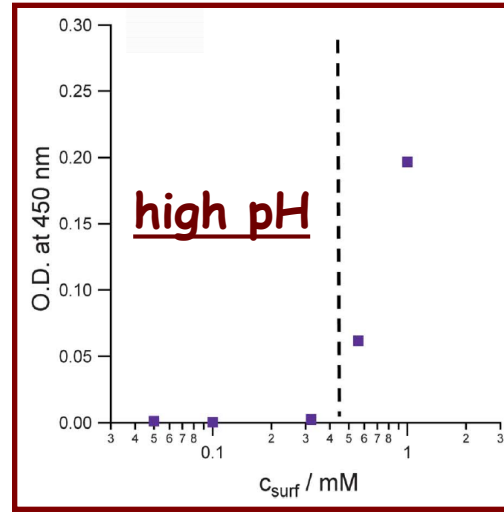
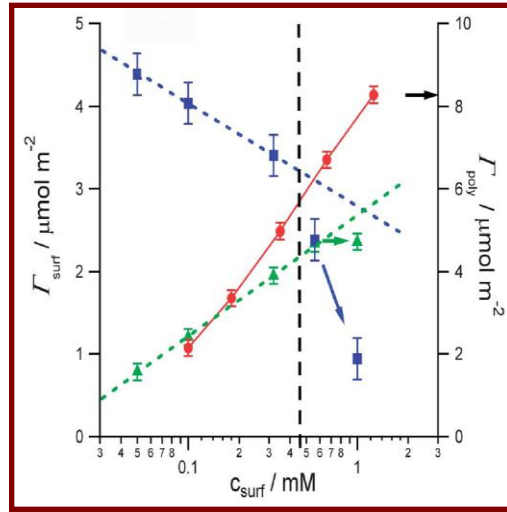
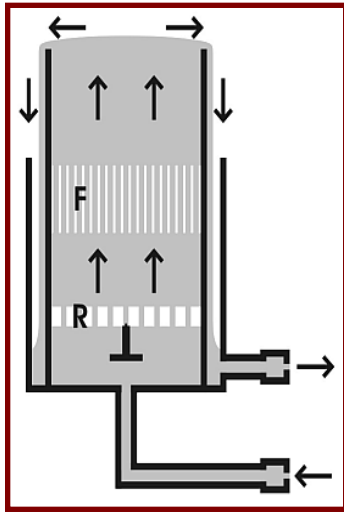
Campbell, Angus-Smyth, Yanez, Tonigold, Nylander & Varga, *J Phys Chem Lett*, 2010, 1, 3021



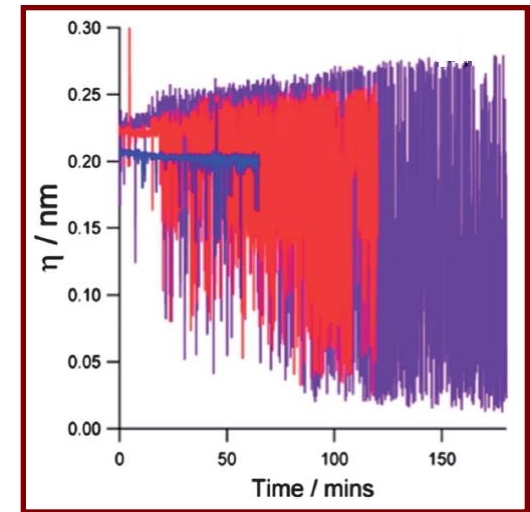
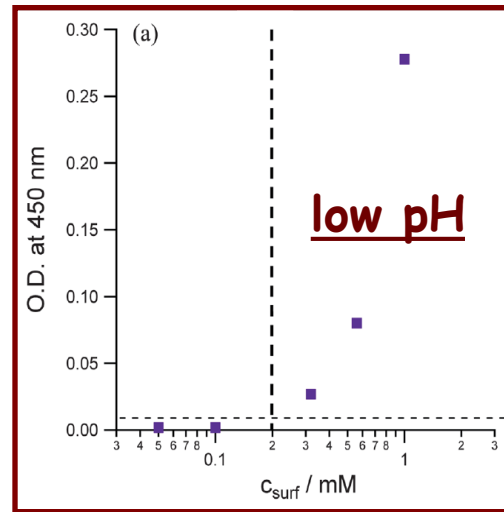
[Pdadmac] = 100 ppm
[SDS] = 0.82 mM
[NaCl] = 100 mM



Campbell, Yanez, Angus-Smyth, Nylander & Varga, *J. Phys. Chem. B*, 2012, 116, 7981

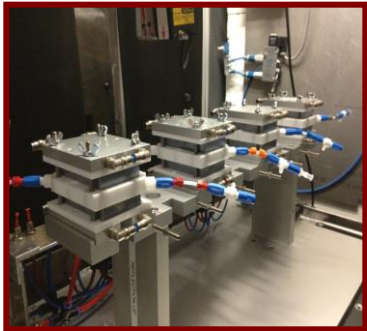


[PEI] = 100 ppm
[SDS] = varied
[NaCl] = 100 mM



Angus-Smyth, Varga, Bain & Campbell, *Soft Matter*, 2013, 9, 6103

- static air/liquid measurements:
 - non-equilibrium aggregates can be controlled to tune the surface properties
- solid/liquid/solid measurements:
 - surface properties from self-assembly modified by phase separation/gravity
- dynamic air/liquid measurements:
 - dynamic surface properties are seemingly uncorrelated with the static ones



- new platform for unique interfacial characterization at the ILL:
 - emerging as a center for resolving the surface behavior of complex mixtures