

# **BSCCO 2212 precursor at Nexans:**

## **Current Status and Further Development within EuCARD2**

Mark Rikel

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- Ken Marken (now at DOE), Hanping Miao, Yibing Huang, Seung Hong (OST)
- Christian Scheuerlein, Amalia Balarino, Luca Botura (CERN); Mario Scheel, Marco Di Michiel (ESRF)
- Tengming Shen, Pei Li, Lance Cooley (FNAL)
- Rene Deul, Lisa Koliotassis, Alexis Camus, Simon Krämer, Andreas Klimt, Zemfira Abdoulaeva, Werner Horst (NSC)



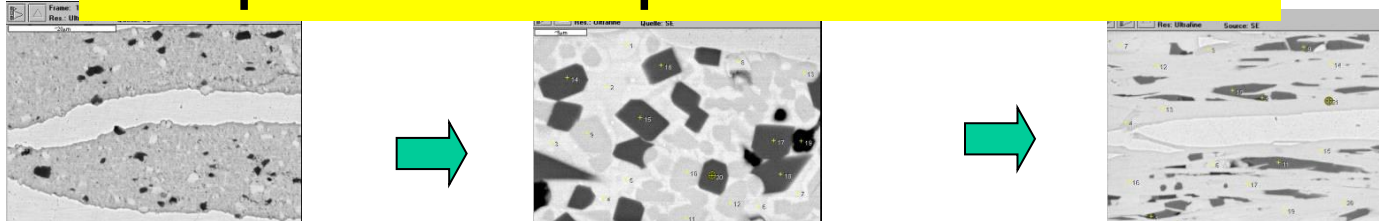
- Requirements on Precursor for Ag-sheathed Bi2212 conductors
- Bi2212 precursor technology at Nexans
  - ◆ Equilibrium Precursor: Development, Current Status, and Quality
- Major Open Issues
  - ◆ Granulated versus Powdered Precursor
  - ◆ Optimum cation composition and optimum O contents
- Work within EUCARD2
  - ◆ Supplying high-quality precursor to OST.
  - ◆ Optimize granulated precursor
    - density homogeneity in granulate; optimum particle size window
  - ◆ Optimizing composition including O contents as variable
  - ◆ Stabilizing the industrial process for optimum composition



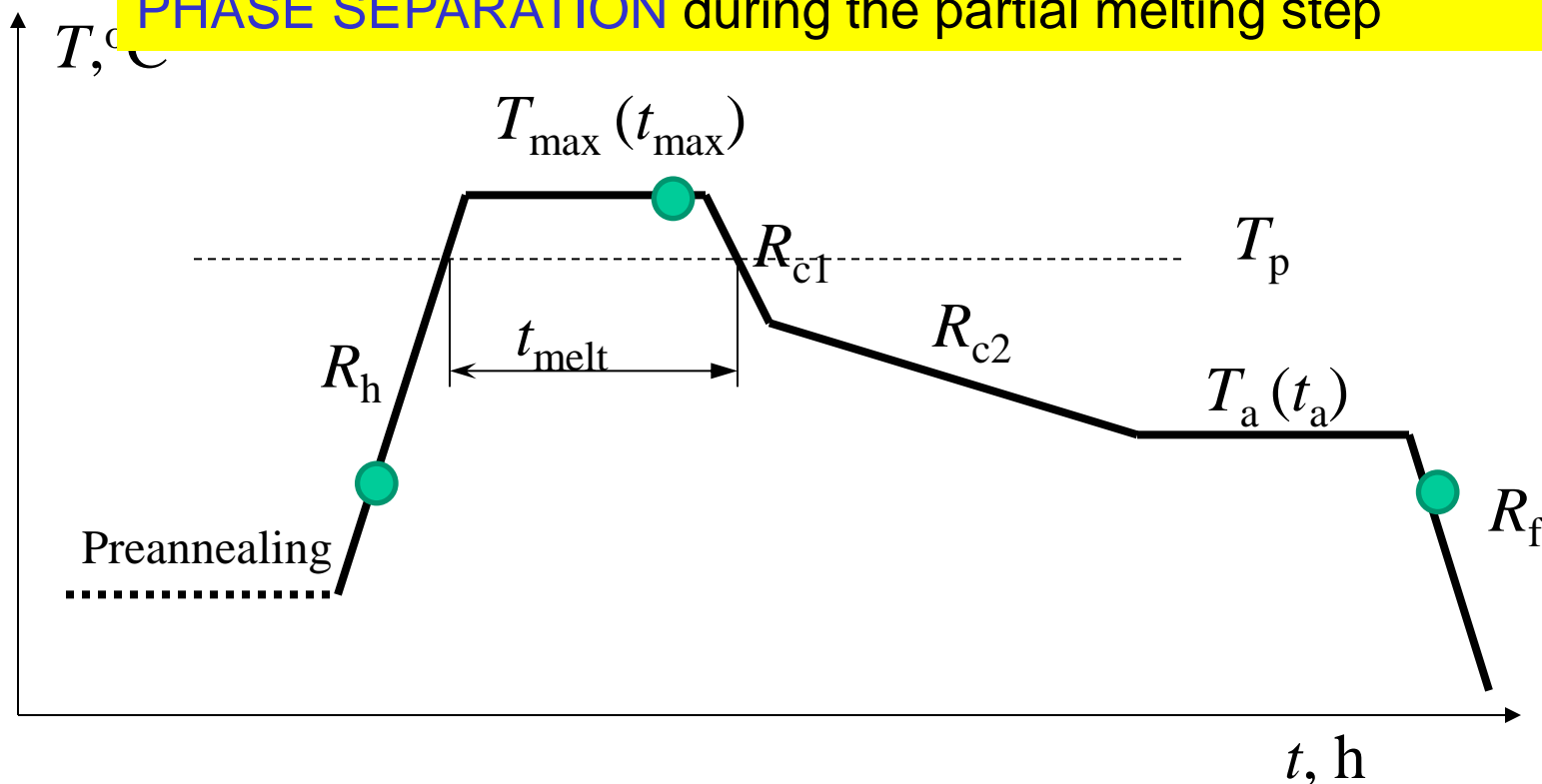
- General for OPIT process
  - ◆ Homogeneity
  - ◆ Size of hard particles much smaller than filament size
  - ◆ Filling / Tapping density to ensure necessary fill factors
  - ◆ Flowability.....
- Specific for Partial Melt Processing of Bi2212
  - ◆ Low C and H<sub>2</sub>O contents to avoid bubbling
  - Equilibration state :



**Final performance depends on the initial state**

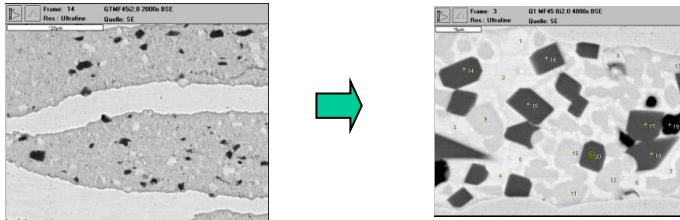


The initial state of precursor predetermines the extent of **PHASE SEPARATION** during the partial melting step



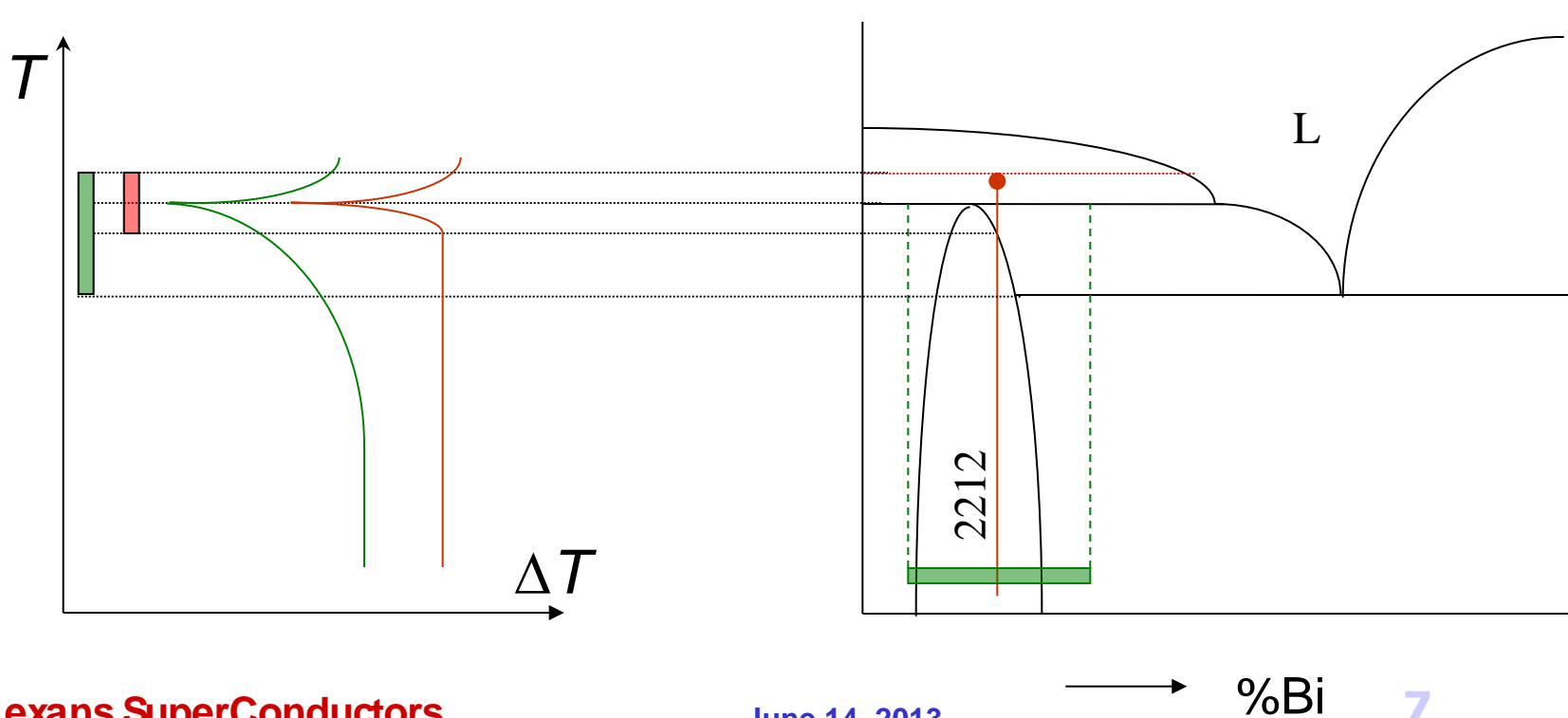
- General for OPIT process
  - ◆ Homogeneity
  - ◆ Size of hard particles much smaller than filament size
  - ◆ Filling / Tapping density
  - ◆ Flowability.....
- Specific for Partial Melt Processing of Bi2212
  - ◆ C contents
  - ◆ Equilibration state that minimizes  
**PHASE SEPARATION** during partial melting
  - ◆ .





Minimize PHASE SEPARATION by

- choosing proper composition;
- making precursor as close to **EQUILIBRIUM** as possible

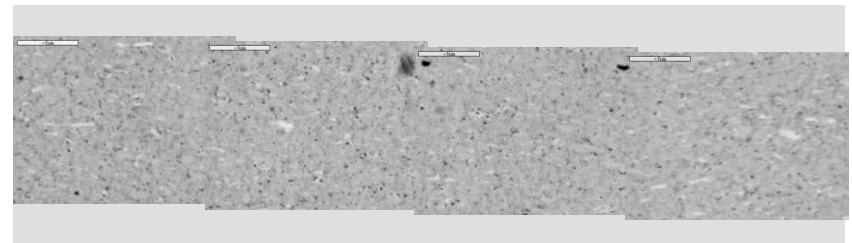
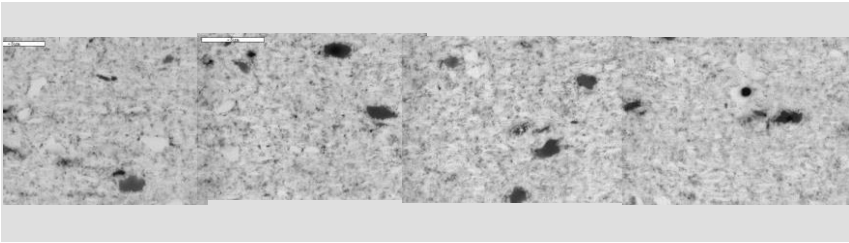


Equilibrium

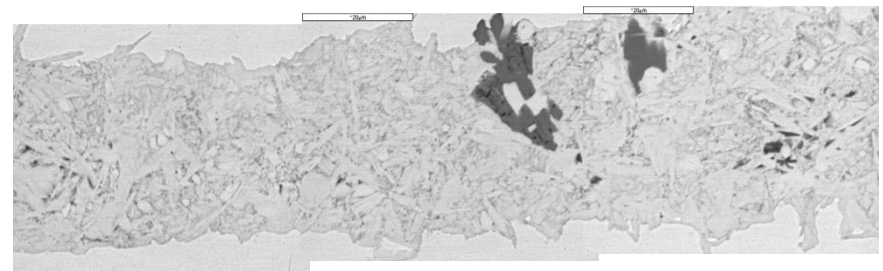
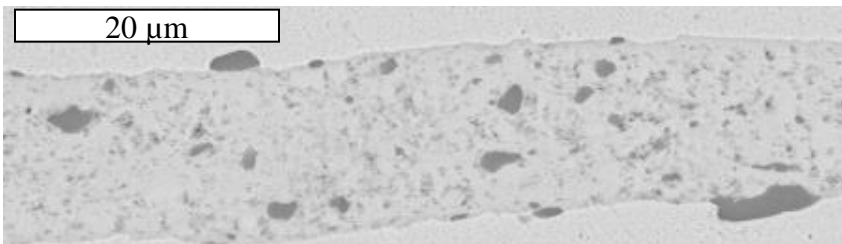
Green Tapes

Non-Equilibrium

looks more uniform in the Green tape



Tapes after heating close to melting (864°C)



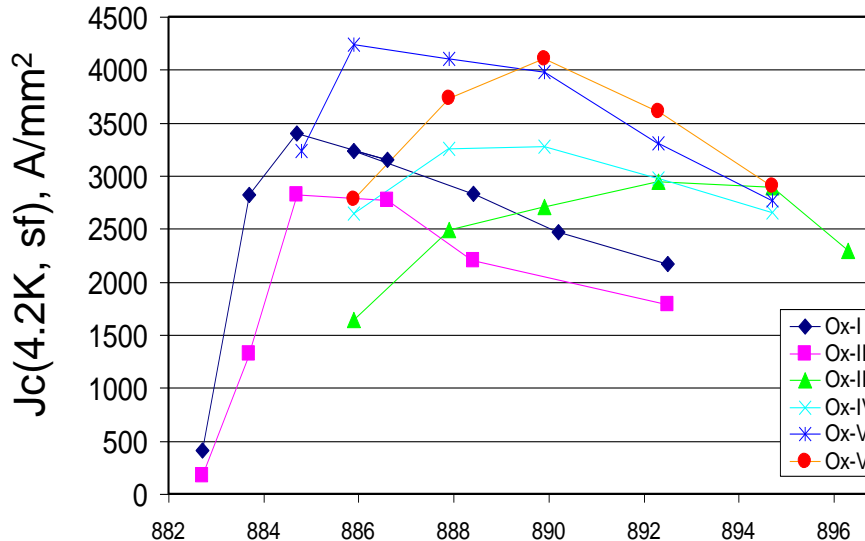
is more uniform close to melting



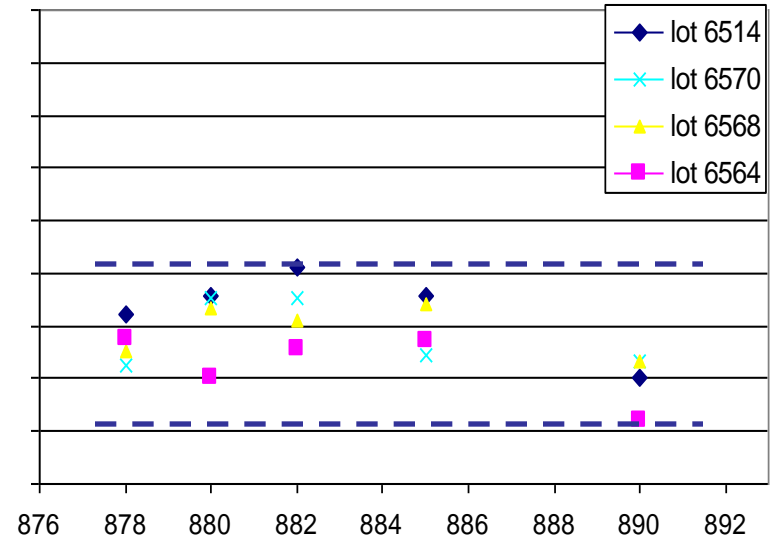


A factor of 3 better performance  
(dip-coated tapes)

Dip Coated Jc optimization in Nexans Equilibrium Lots



non - Equilibrium powder evaluation



Superconducting Technology



NIMS dip-coated Tapes  
Postannealed at 300°C

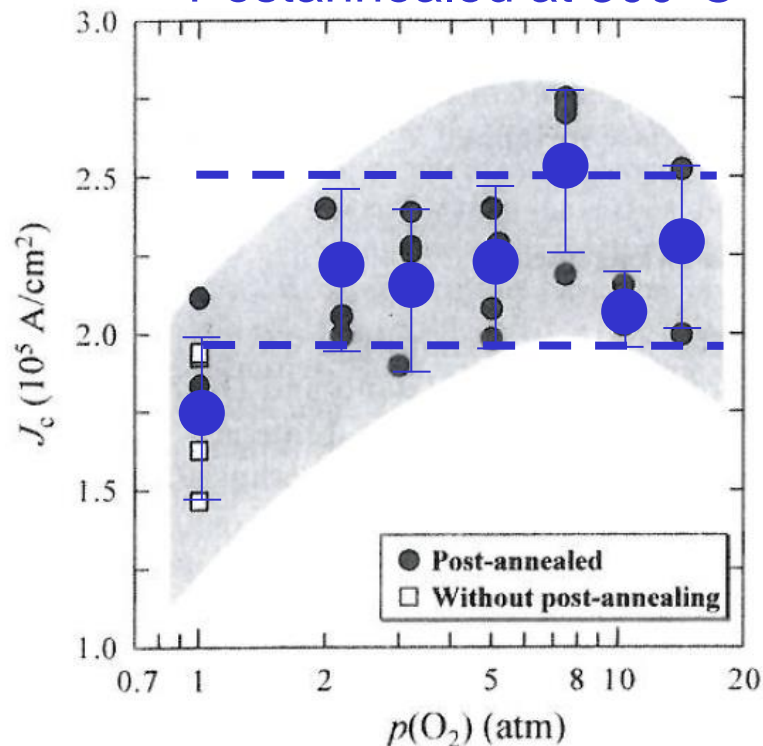
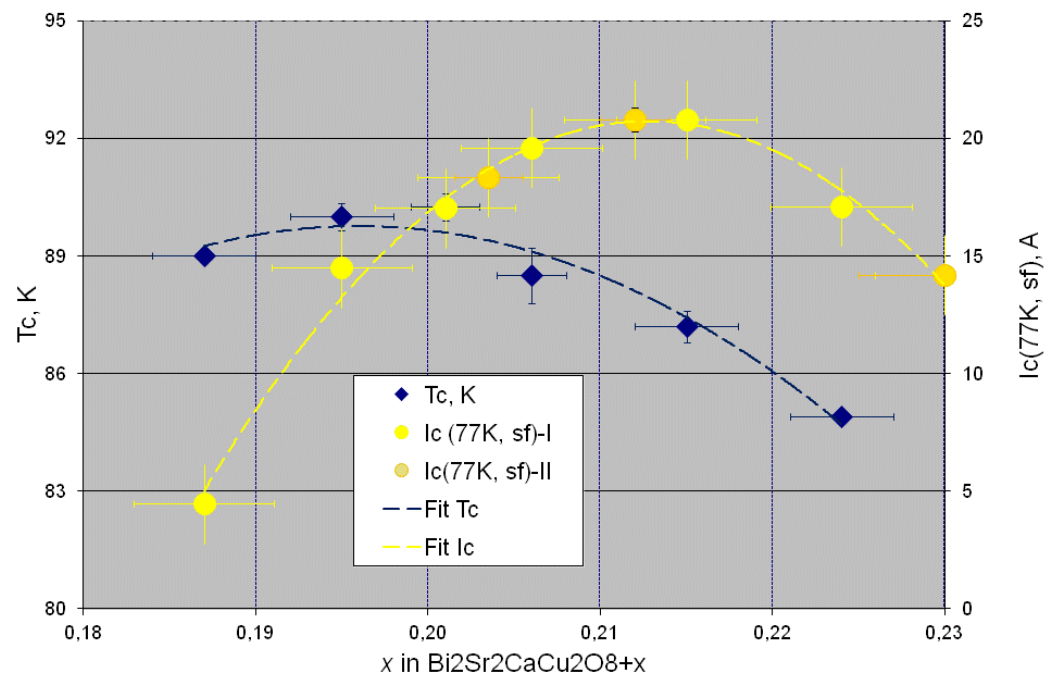


Fig. 2. Correlation between the  $p(\text{O}_2)$  at the post-annealing process and the  $J_c$  value in 10 T at 4.2 K.

Nakane et al (2004), Physica C 412-414, 1163-6

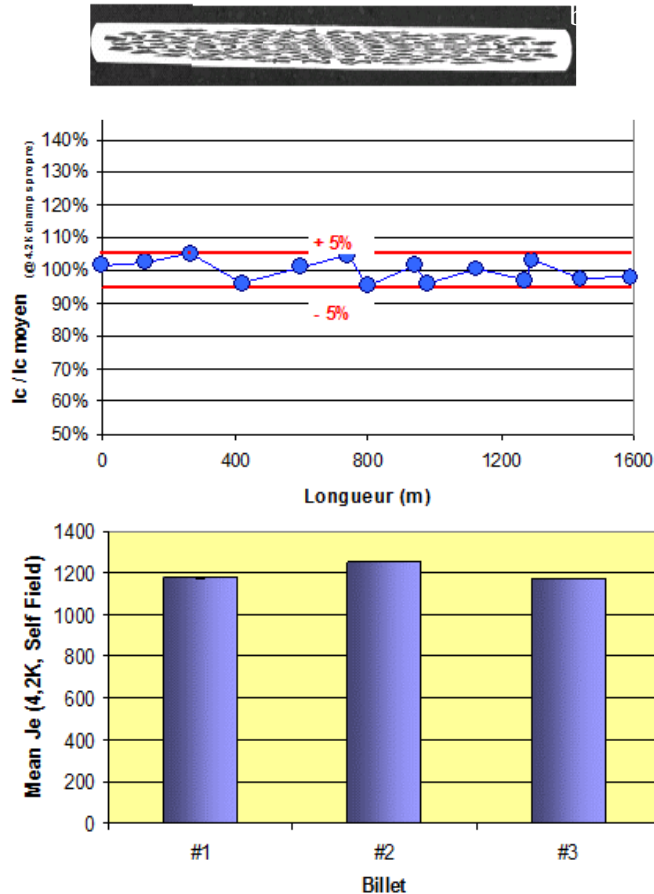
OST wire PMM110106



Rikel et al (2012) ASC, Portland

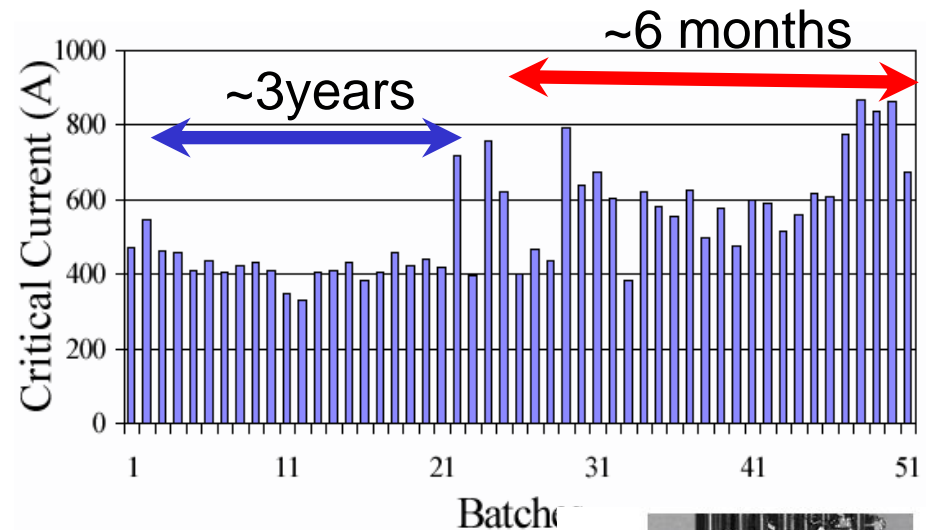


### Nexans experience



C.E.Bruzek et al EUCAS 2003  
ASC 2004

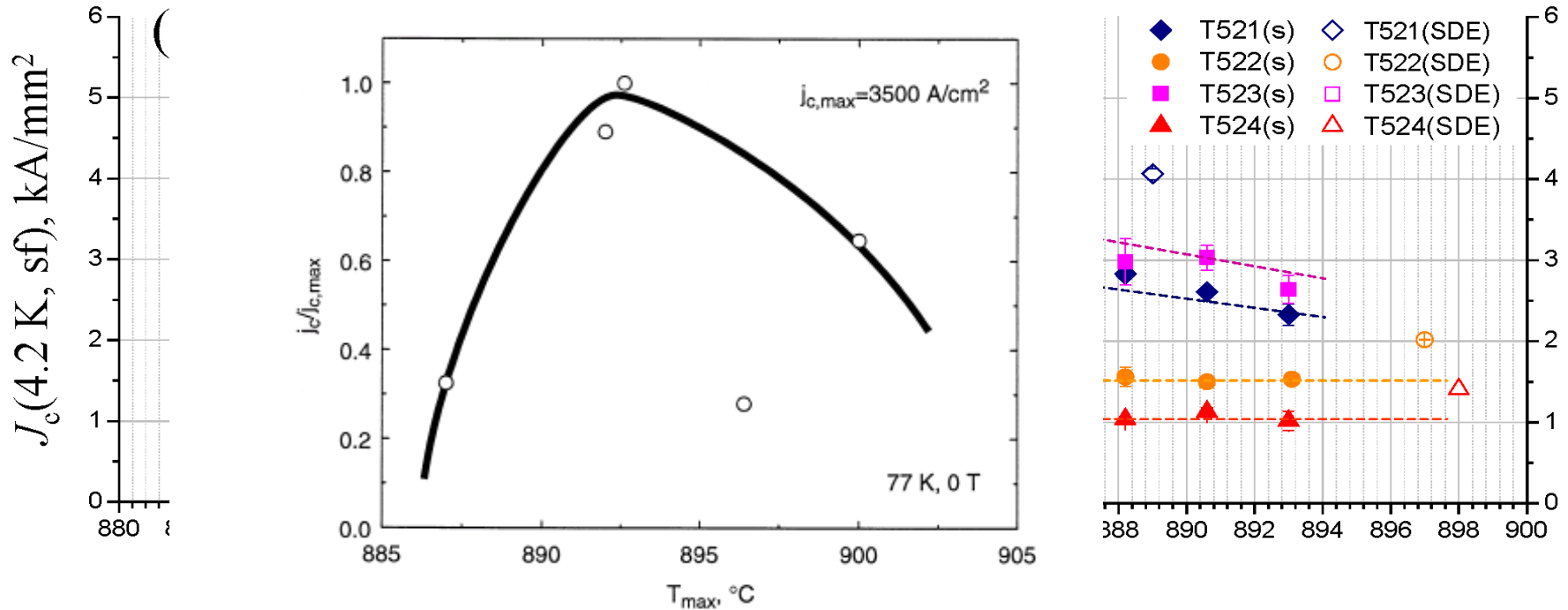
### OST experience



K. R. Marken et al,  
ICMC 2003



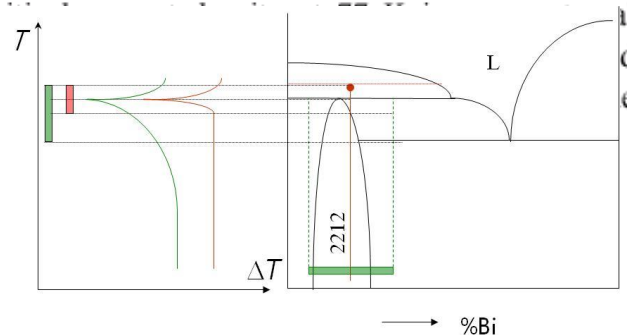
FIGURE 2. The 5 tesla insert magnet which achieved 25 T central field.

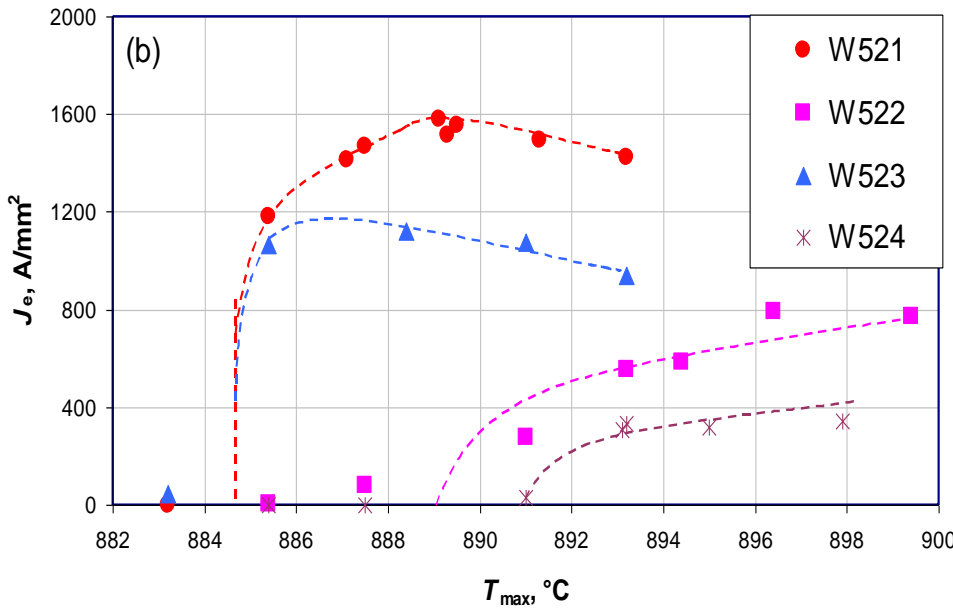


- Very uniform distribution of melting point over macroscopic lengths

Fig. 2. Influence of the maximum processing temperature  $T_{\text{max}}$  on the normalized critical current density  $j_c/j_{c,\text{max}}$  of fully processed samples in the same furnace. The maximum  $j_c$  was measured at  $77 \text{ K, 0 T}$ .

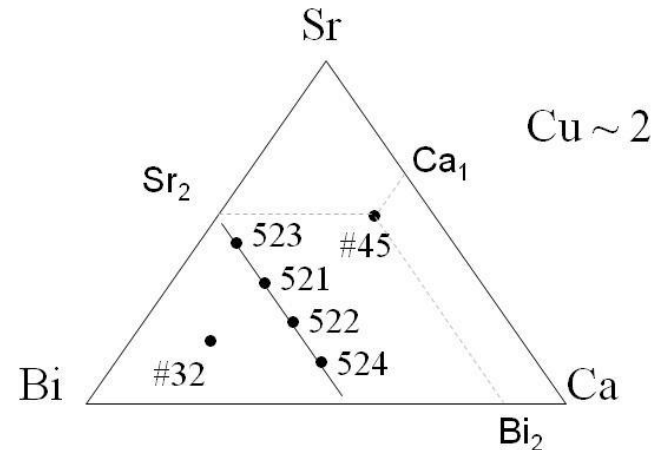
Lang et al





**W521-524** 2.14 : (2.86-x) : x :2.00

**Sr/Ca = 2.25, 2.18, 1.75, 1.34**



H. Miao et al 2006 *Adv Cryo Eng.* **52B**, p. 673, (2006) [Proc. ICMC 2005]

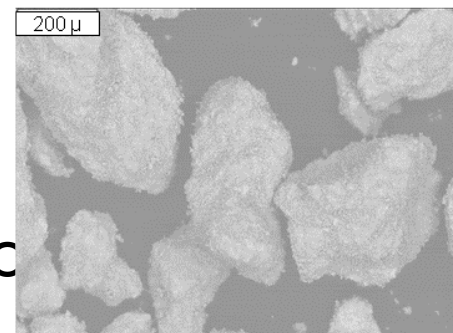
M. Rikel et al 2006 *J Phys: Conf Ser.*, **43** (2006) 51–54 [Proc. EUCAS 2005]

- Why overall composition of Bi2212 has such a strong effect on performance of round wires and tapes ?



- Characterization // Conditioning of Raw Materials
- Mixing oxides
- **Melt Casting** oxide mixtures  
Melt Quenching // Spray drying
- Thermomechanical processing to reach equilibrium and control size of second phases  
Annealing – Jet Milling – Compacting – Annealing – ....
- Final Product:
  - ◆ Powder or **Pressed Rods**
  - Granulate





- Introduced to solve the problem of high C contents  
can be annealed at higher T than powder to decompose SrCC
- Advantages
  - > Lower C contents attainable
  - > Fill factors intermediate between powder and rods
  - > Flowability better than that of powder
  - > Easy filling the billets (no sticking to tools)
  - > Better  $J_c$  achieved by Nexans (France & Korea), Supercon, Supramagnetics, but  
not OST, where  $J_c$  were in average  $\sim 10\%$  worse.



- Equilibrium Rods/Powder (2002-2004)
  - ◆ Nexans SMES project ( $\sim 400$  kg precursor)
  - ◆ OST 25 T magnet project + R&D ( $\sim 50$  kg)
- Collaboration with OST on optimizing precursor composition (2004-2005)
  - ◆  $\text{Bi}_{2.17}\text{Sr}_{1.94}\text{Ca}_{0.89}\text{Cu}_{2.00}$  cation composition (Lot 521) = Nexans Standard
- Granulated Precursor (2005-2006)





Attribute	Target Levels	Test
Cation Composition	Bi : Sr : Ca : Cu 2.16 : 1.95 : 0.89 : 2.00 $\pm 0.03 : 0.03 : 0.015 : 0.03$	XFA
Cation Impurities	H: < 50 ppm Fe: < 50 ppm (?) Zn: < 20 ppm	IR for H ICP
Carbon	20 - 50 ppm for granulate 50-150 ppm for powder	IR Leco
Grain Size d: 10, 50, 90	d10 < 0.5 $\mu\text{m}$ ; d50 < 2 $\mu\text{m}$ ; d90 < 5 $\mu\text{m}$	Laser granulometry
Fill density : g/cm <sup>3</sup>	1.5 (powder) to 1.9-2.5 (granulate)	weight
Melting Temperature, with Ag	Onset in 100% O <sub>2</sub> : 883 $\pm 2^\circ\text{C}$	DTA/TGA
Phase State	Phase composition should correspond to equilibrium state for a given overall composition	2theta XRD Rietveld Refinement
Extent of equilibration	Fraction of intergrowths in 2212 phase < 1%	2theta XRD
Particle size (granulate)	100-500 $\mu\text{m}$ (empirical)	sieving
Storage and shipment	Dry, < 10 <sup>-5</sup> at static P <sub>H<sub>2</sub>O</sub> , P <sub>CO<sub>2</sub></sub>	Monitor & record: sensors



Attribute	Target Levels	Test
Cation Composition	Bi : Sr : Ca : Cu 2.16 : 1.95 : 0.89 : 2.00 $\pm 0.03 : 0.03 : 0.015 : 0.03$ Is this sufficient?	XFA
Cation Impurities	H: < 50 ppm <b>Fe: &lt; 50 ppm</b> (?) Zn: < 20 ppm	IR for H ICP
Carbon	20 - 50 ppm for granulate <b>50-150 ppm</b> for powder	IR Leco
Grain Size d: 10, 50, 90	d10 < 0.5 $\mu\text{m}$ ; d50 < 2 $\mu\text{m}$ ; d90 < 5 $\mu\text{m}$	Laser granulometry
Fill density : g/cm <sup>3</sup>	1.5 $\pm$ 0.2 (powder) to 1.8-2.5 (granulate)	weight
Melting Temperature, with Ag	Onset in 100% O <sub>2</sub> : 883 $\pm$ 2°C <b>How to control low temperature event?</b>	DTA/TGA
Phase State	Phase composition should correspond to equilibrium state for a given overall composition	2theta XRD Rietveld Refinement
Extent of equilibration	Fraction of intergrowths in 2212 phase < <b>1% ?</b>	2theta XRD
Particle size (granulate)	<b>Particle size window</b>	Sieving, <b>OM, SEM</b>
Storage and shipment	Dry, < 10 <sup>-5</sup> at ? static P <sub>H<sub>2</sub>O</sub> , P <sub>CO<sub>2</sub></sub>	Monitor & record: sensors



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  - ◆ Granulated precursor:  
density homogeneity in granulate; optimum particle size window.
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## Why Granulated Precursor shows 10% worse performance? New Data from OST



- Hard Bi2212 particles forming when drawing wires made using granulates 78 and 80 (small powder batches), but not 77 (large batch)

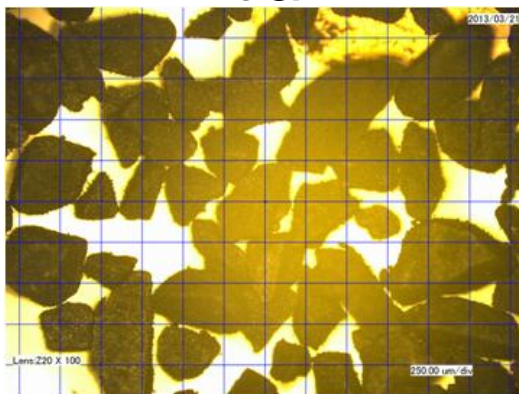


Formation of hard particles may stem from too broad particle size distribution or reflect density inhomogeneity in the granules due to

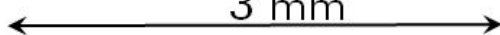
- using different compacting technologies
- multiple compaction steps necessary for the required yield when using small powder batches
- Quick test: Uniformity of granule response to small loads (1 to 50 bar)

82215123121 (200-500  $\mu\text{m}$ )

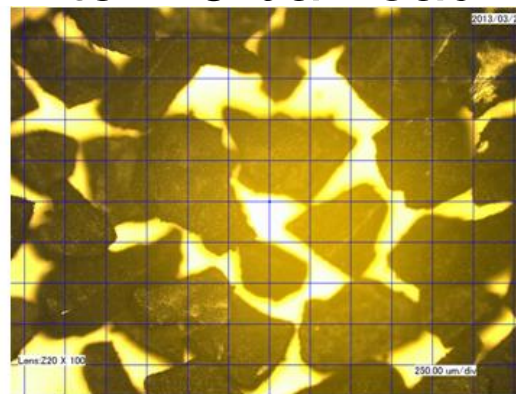
Initial



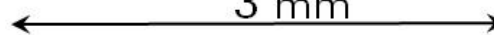
3 mm



After ~3 bar load

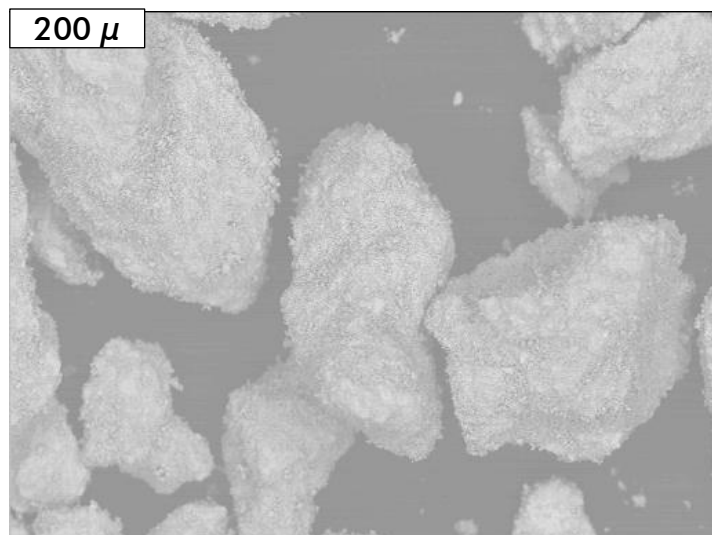


3 mm

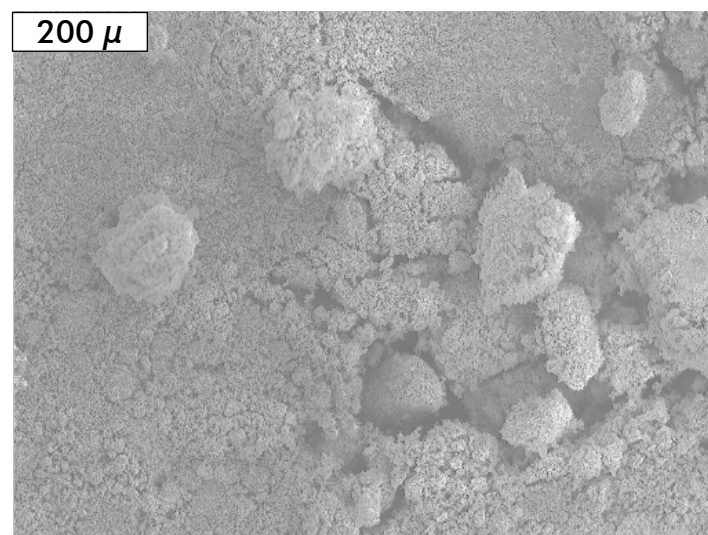


8011322331 (100-500  $\mu\text{m}$ )

Initial



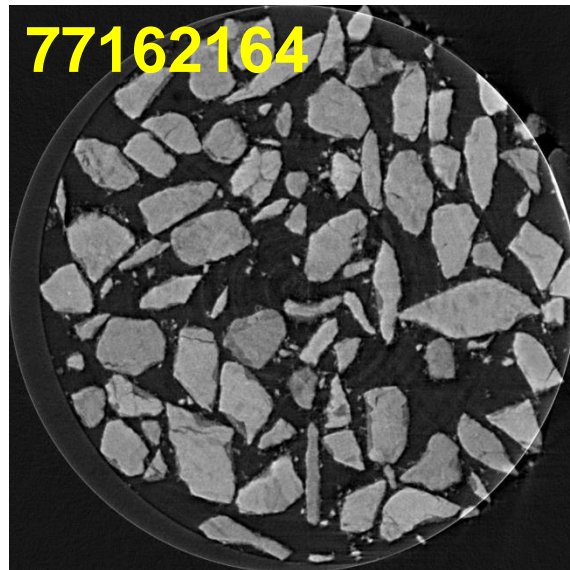
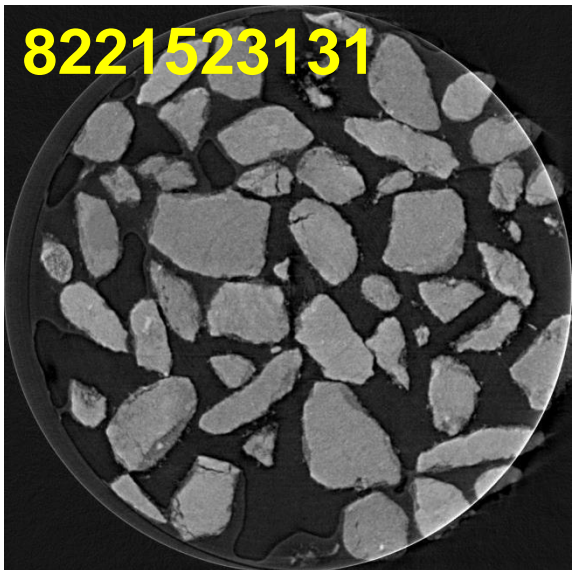
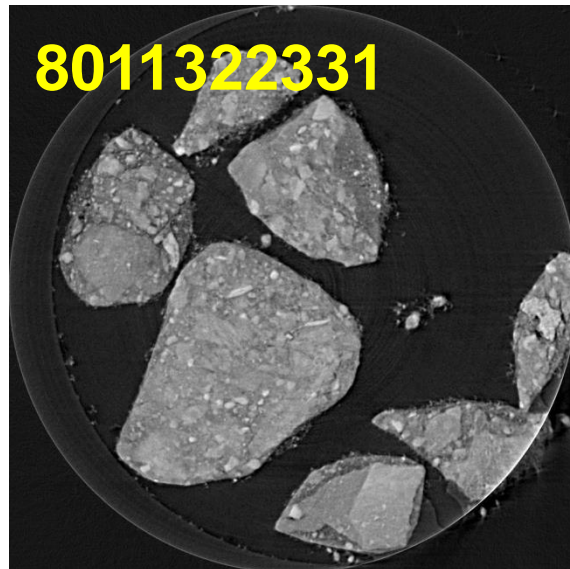
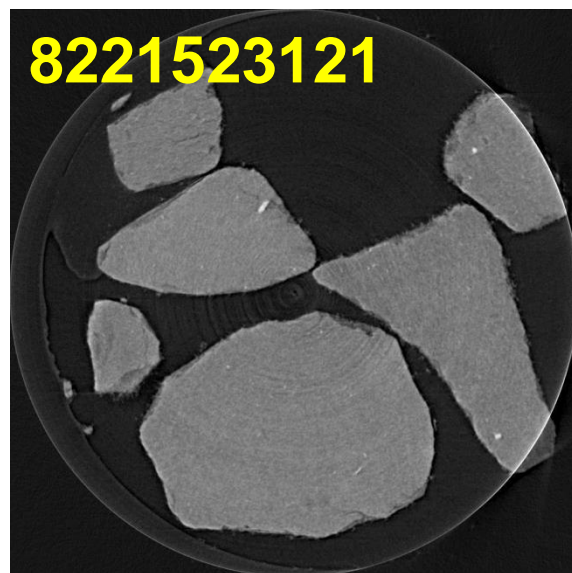
After ~3 bar load



- Old lot 80 seems less homogeneous than new lot 82







- Granulate 82 is much more homogeneous than 80 and even 77
- Inhomogeneity of lot 77 may come not from compacting but from exposure of the 3.5-year old material to environment

C. Scheuerlein et al (2013)





## **Task 1. Production of BSCCO2212 precursor granulate**

### **Subtask 1.1. Production**

10 kg standard (521-like) composition

### **Subtask 1.2. Effect of particle size in the granulate on processing BSCCO2212 round wires**

Goals:

- Optimize the particle size window

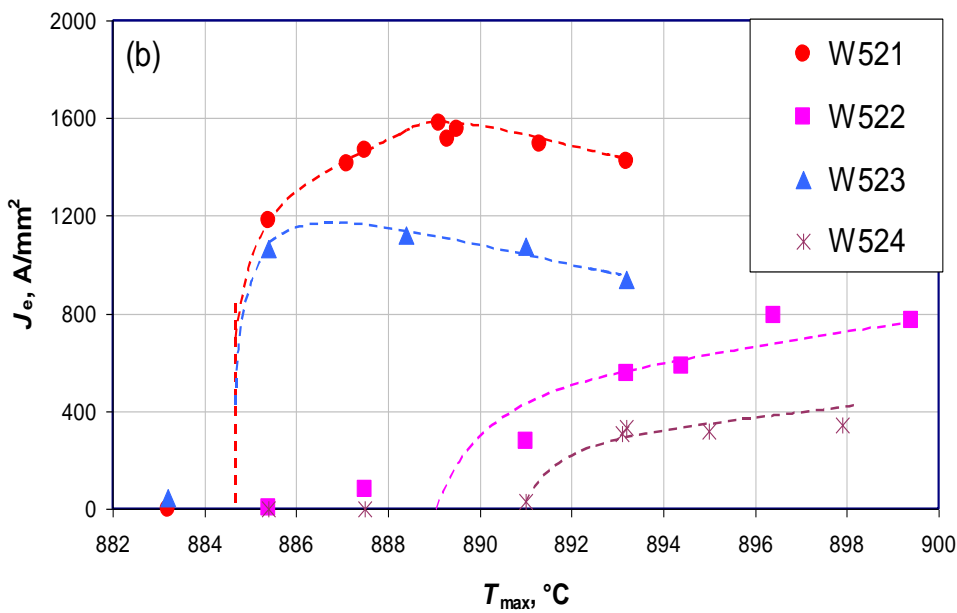
The precursor produced in subtask 1.1 will be classified in fractions with various particle size windows, e.g.,

- powder
- 200-500  $\mu\text{m}$  [standard]
- 100 to 200  $\mu\text{m}$ ,
- 50 to 100  $\mu\text{m}$

## **Task 2. Compositional studies:**

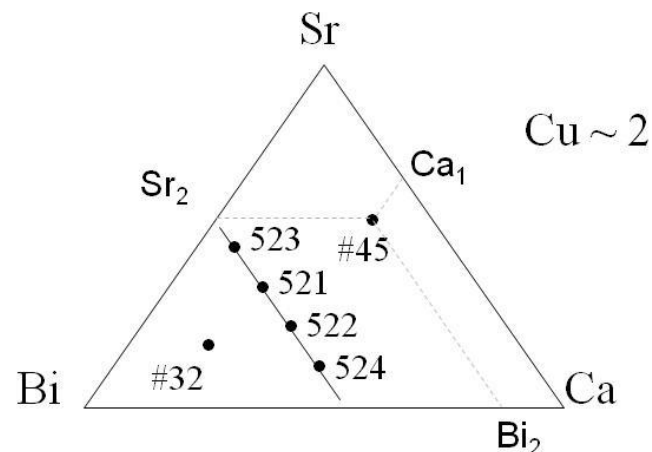
**Subtask 2.1. Preliminary studies of BSCCO2212 bulk with various Sr/Ca ratio and O contents**





**W521-524** 2.14 : (2.86-x) : x :2.00

**Sr/Ca = 2.25, 2.18, 1.75, 1.34**

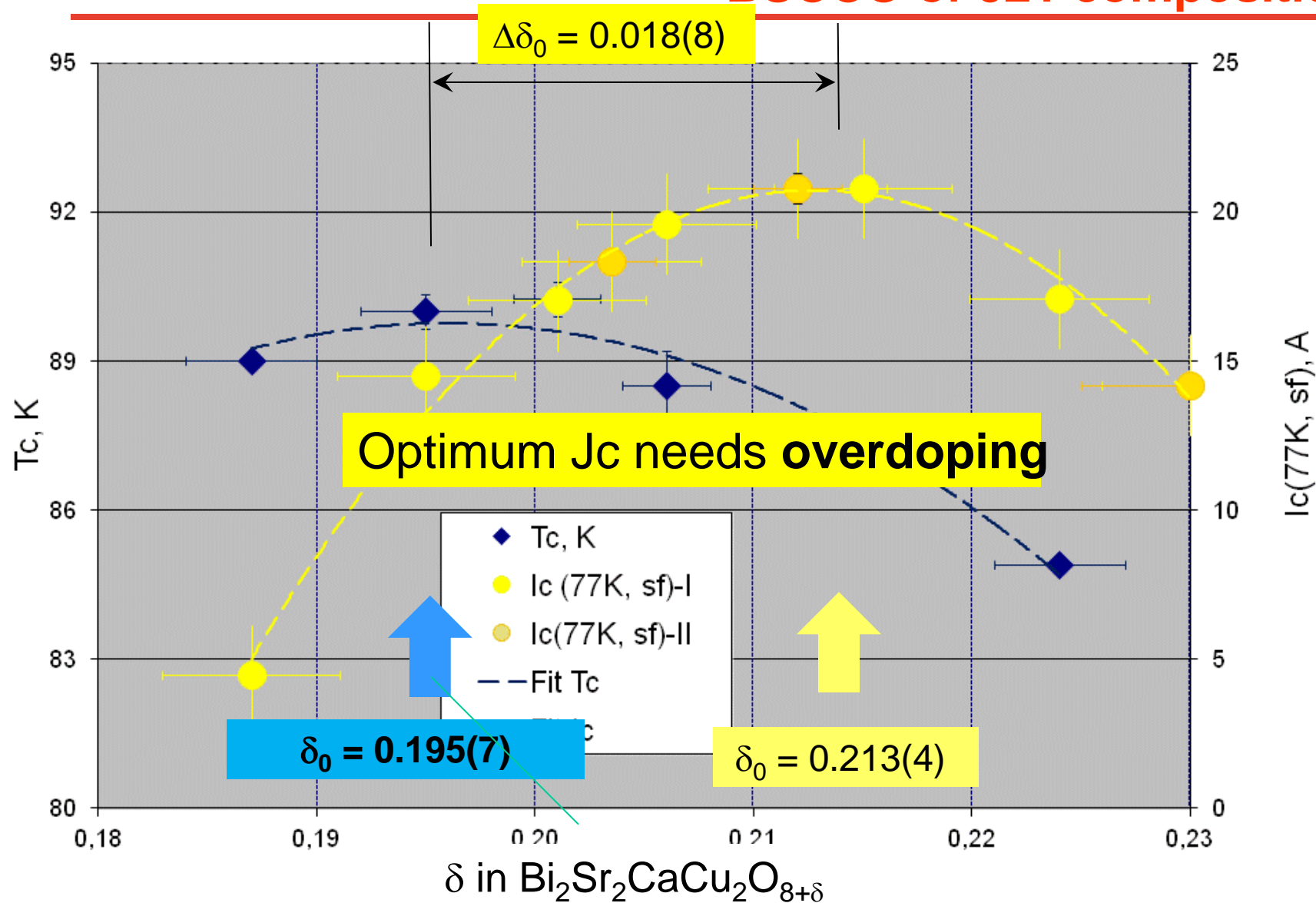


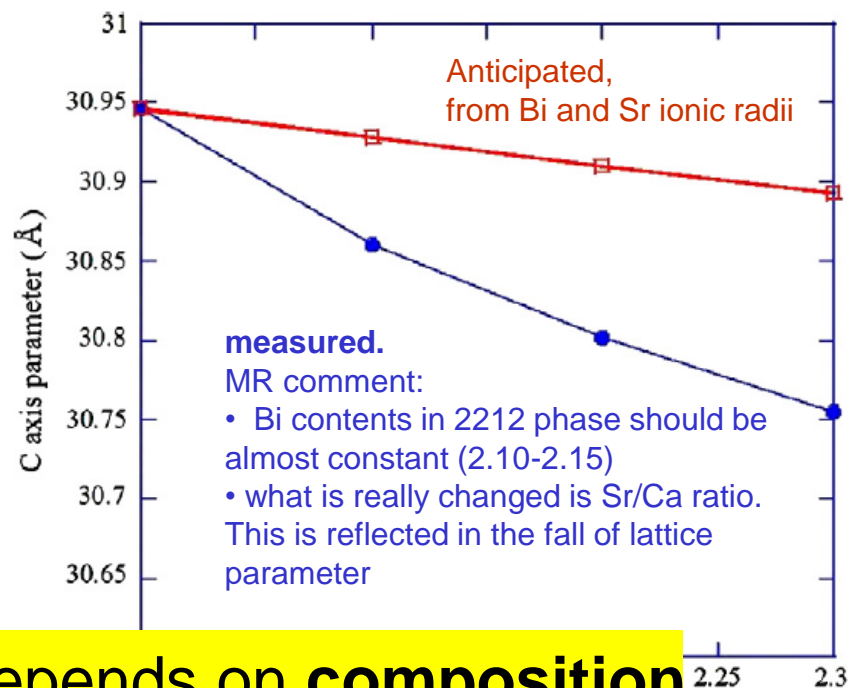
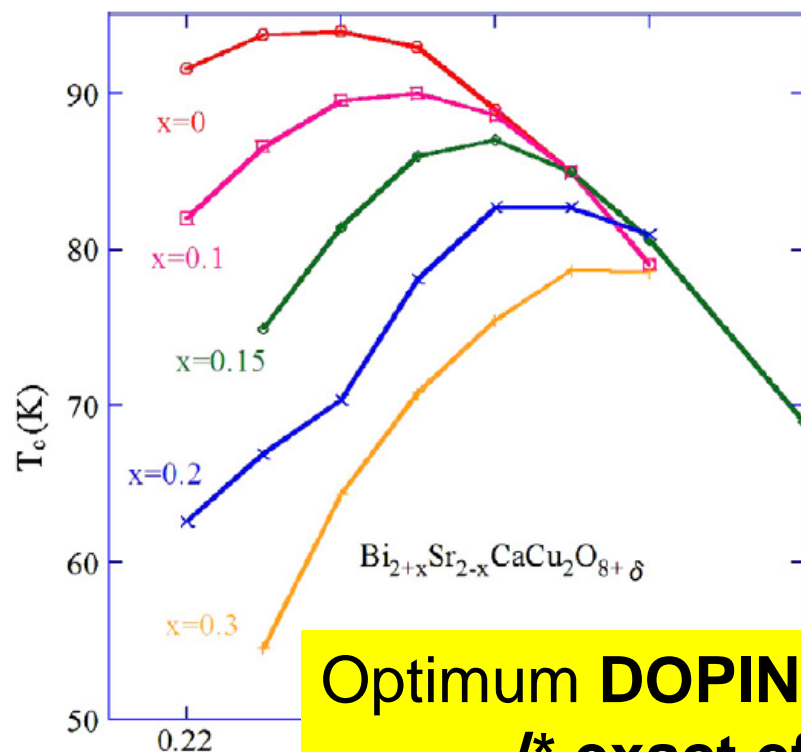
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- Why overall composition of Bi2212 has such a strong effect on performance of round wires and tapes ?



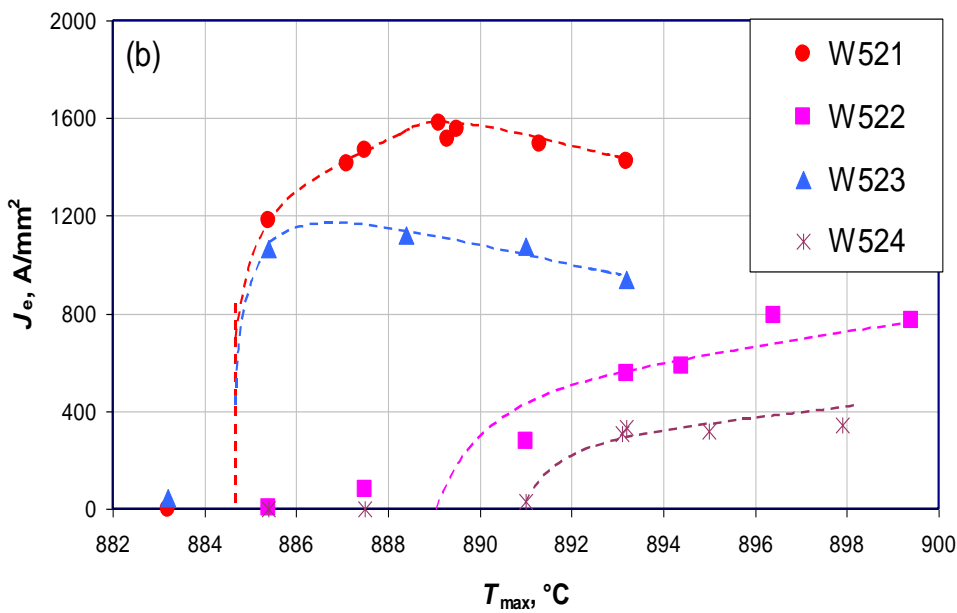




**Optimum DOPING depends on composition**  
**/\* exact effect to be studied\*/**

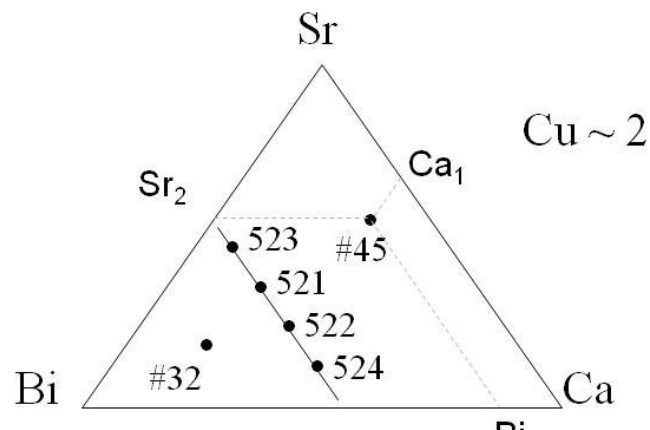
Yamashita et al (2010) studied single crystals grown from powders of  $Bi_{2+x}Sr_{2-x}Ca_1Cu_2$  cation compositions and annealed to have various O contents. They found that the crystals with smaller Sr/Ca ratio have maximum  $T_c$  at stronger overdoping levels. Though the real compositions were not measured, the Sr/Ca ratio in the 2212 phase should scale with that in the overall composition (Rikel et al 2006). Thus, our observation that maximum of  $J_c(\delta)$  in round wires is at higher  $\delta$  than in the bulk may stem from the difference in  $T_c(\delta)$  for bulk (Sr/Ca =  $2.45 \pm 0.02$ ) and round wire (Sr/Ca =  $2.20 \pm 0.03$ ). We should first measure  $T_c$  of the wires.





**W521-524** 2.14 : (2.86-x) : x :2.00

**Sr/Ca = 2.25, 2.18, 1.75, 1.34**



**What is the contribution of O doping level in the  $J_c$  difference ?**

M. Rikel et al 2006 *J Phys: Conf Ser.*, **43** (2006) 51–54 [Proc. EUCAS 2005]

- Why overall composition of Bi2212 has such a strong effect on performance of round wires and tapes ?



- Bulk rods (5 and 8 mm diameter) of  $\text{Bi}_{2.15}\text{Sr}_{2.85-x}\text{Ca}_x\text{Cu}_{2.00}\text{O}_{8+\delta}$  compositions **Melt Cast**

Sr/Ca	$x(*)$	$\delta_{\text{optimum}}$ for	
		$T_c^{(a)}$	$J_c(77\text{ K, sf})$
2.47	0.82	0.190(5)	0.203(3)
2.18	0.90	0.195(7)	0.214(3)
1.76	1.03	0.225	?
1.34	1.22	0.240	?

(\*) will be checked by XFA after homogenization

- ♦ DTA/TGA **done**. Heat Treatment **designed**. Samples **processed**
- ♦ XRD and SEM/EDX **in progress**
- Dip-Coated Tapes 521-524 (from OST and NSC) **sent to FNAL** (Tengming Shen) for **Partial Melt Processing**
- Samples, Bulk Rods & Tapes (if possible) will be **annealed** at NSC to **vary  $\delta$  from 0.180 to 0.250**
- $T_c$  &  $M(T, B)$  will be **studied** to **choose NEW COMPOSITION** for testing in wire



- Supplying high-quality precursor to OST
- Optimizing composition including O contents as variable
- Stabilizing the industrial process for optimum composition



**Thank you**