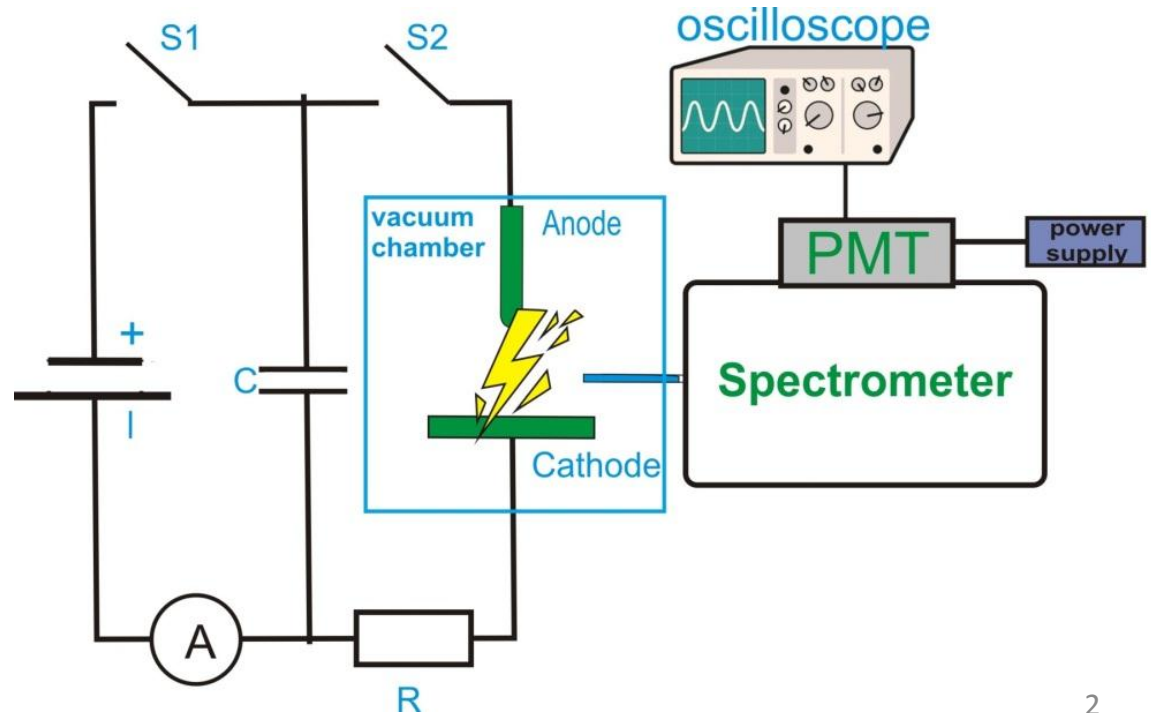


# Optical spectroscopy of breakdowns

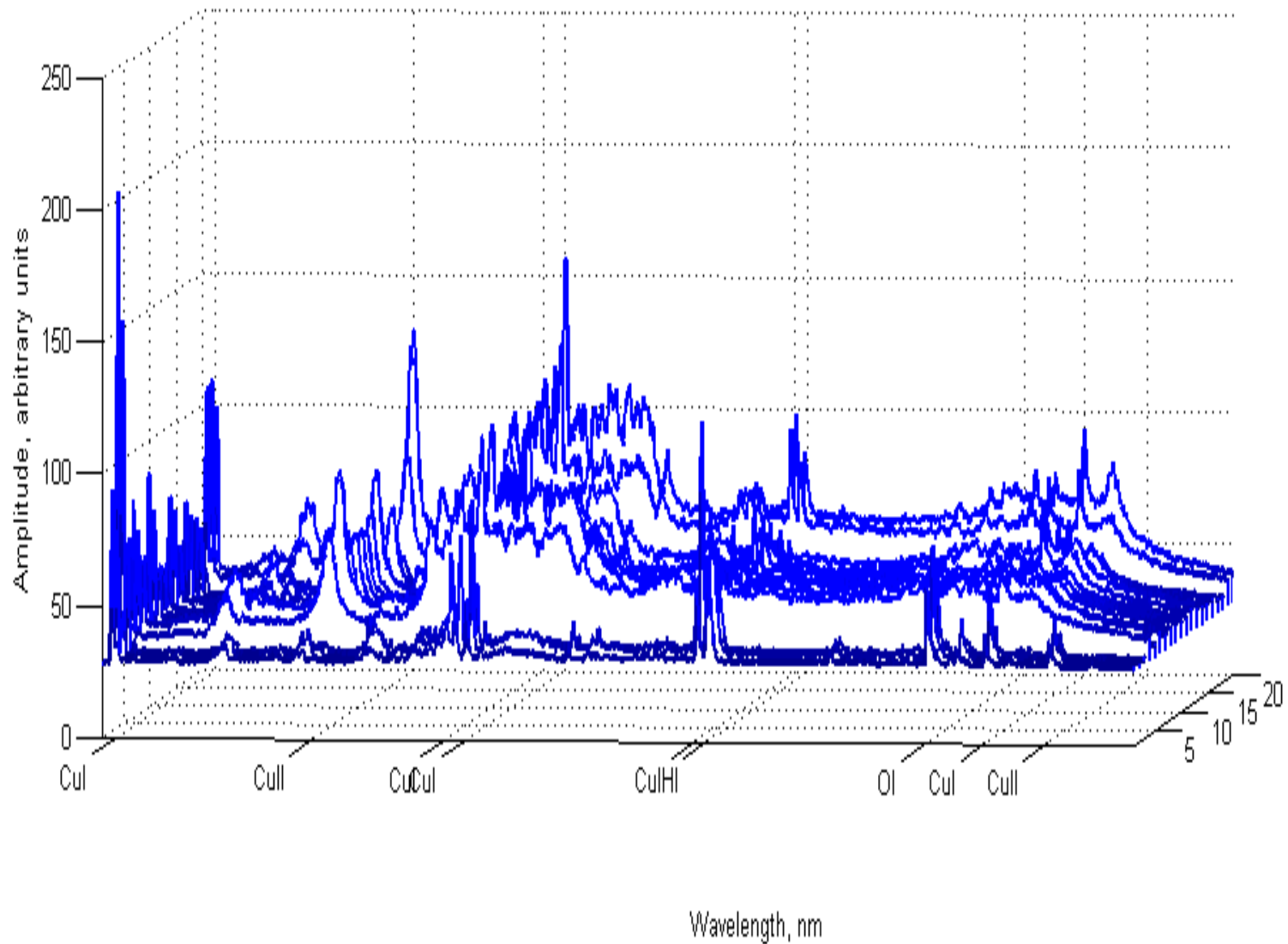
08-10-2013 Profatilova Iaroslava,  
Breakdown meeting

# Ability of DC-breakdowns optical spectroscopy

1. Qualitative analysis (composition and impurities).
2. Plasma diagnostics.



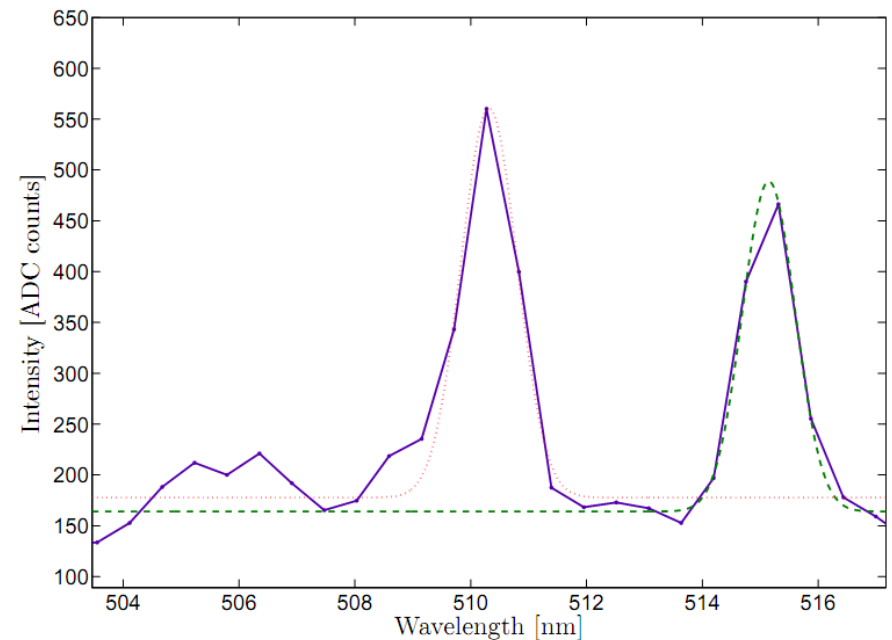
# Qualitative analysis



# Diagnostics of plasma parameters

For plasma in LTE we could use a two-line methods for calculate:

- Electron temperature  $T_e$ ;
- electron density  $n_e$ ;
- plasma conductivity  $\sigma_{dc}$ .



# For Plasma in LTE:

- Maxwell Distribution:  $dn_e = 4\pi n_e \left( \frac{m}{2\pi kT_e} \right)^{\frac{3}{2}} e^{-\frac{mv_e^2}{2kT_e}} v_e^2 dv_e$
- Boltzmann Distribution:  $n_i = n_0 g_i A e^{-\frac{E_i}{kT}}$
- The Saha's Ionization Equation:  $\frac{n_e^2}{n_g - n_e} = A' T_e^{\frac{3}{2}} e^{-\frac{E_i}{kT_e}}$

There is equilibrium in a weak radiation field, where  $T$  from the Maxwell distribution is equal to the temperature  $T$  from Saha ionization equation. This is the so-called plasma in local thermodynamic equilibrium (LTE)

# Boltzmann plot method for T

For plasma in LTE, the energy level populations of the species are given by the Boltzmann distribution law:

$$\ln\left(\frac{I\lambda}{gA}\right) = -\frac{E_u}{k_B T} + \text{const.} = -aE_u + b$$

$$\frac{I_a}{I_b} = \frac{g_a A_a \lambda_b}{g_b A_b \lambda_a} \exp\left[-\left(\frac{E_a - E_b}{kT_e}\right)\right]$$

$$T_e = -\left(\frac{E_a - E_b}{k}\right) \left[\ln \frac{I_a g_b A_b \lambda_a}{I_b g_a A_a \lambda_b}\right]^{-1}$$

For each line, the wavelength  $\lambda$ , the intensity  $I$ , the oscillator's strength  $f$ , the statistical weight of the lower level  $g$  and the energy of the upper level  $E_u$  have each to be measured or be obtained from literature.

# Calculation T

	Wavelength (nm)	Upper level energy (eV)	Lower level energy (eV)	Upper level degeneracy	Transition probability, $s^{-1}$
CuI	465,11	7,74	5,072	8	3,80E+07
CuI	510,55	3,817	1,389	4	2,00E+06
CuI	515,32	6,191	3,786	4	6,00E+07
CuI	521,82	6,192	3,817	6	7,50E+07
CuI	578,21	3,786	1,642	2	1,65E+07
CuII	268,93	13,392	8,783	7	4,10E+07
CuII	271,35	13,432	8,864	5	6,80E+07

$$T_1 (510.55; 515,32) = 7624 \text{ K}$$

$$T_2 (510.55; 521.82) = 7798 \text{ K}$$

$$T_3 (521.85; 578,21) = 12\,013 \text{ K}$$

# Saha-Boltzmann equation method for $n_e$

- The electron density using atom and ion spectral lines emitted from the plasma is determined from the Saha-Boltzmann equation as:

$$n_e = \frac{I_Z^*}{I_{Z+1}^*} 6.04 \times 10^{21} (T)^{3/2} \times \exp\left[\frac{-E_{k,Z+1} + E_{k,Z} - \chi_Z}{k_B T}\right] \text{ cm}^{-3}$$

- where  $I_Z^* = I_Z \lambda_{ki,Z} / g_{k,Z} A_{ki,Z}$  and  $\chi_Z$  is the ionization energy of the species in the ionization stage  $Z$ .



# Plasma conductivity

- RF plasma conductivity

$$\sigma_p = \frac{e^2 n_e}{m(\nu_m + j\omega)}$$

- DC plasma conductivity ( $\omega \ll \nu_m$ )

$$\sigma_{dc} = \frac{e^2 n_e}{m\nu_m}$$

# Thermal Desorption Spectroscopy at IAP NASU

The sample was heated at 300°C for 10 minutes for hydrogen removing from surface. Extraction of hydrogen from the sample was held at 1000°C for 10 minutes. The percentage of hydrogen atoms in the copper samples was determined by chromatograph.



The number of sample	Sample origin	The number of hydrogen atoms in copper, %
№ 1	standard electrochemical copper, IAP NASU	$4 \cdot 10^{-5}$
№ 2	standard electrochemical copper, IAP NASU	$3.4 \cdot 10^{-5}$
№ 3	only etching, CERN	$5.7 \cdot 10^{-5}$
№ 4	brazing coupler, CERN	$2.1 \cdot 10^{-4}$ 10