

Rapid Ultralow Pressure Manometry by Light Absorption Spectroscopy

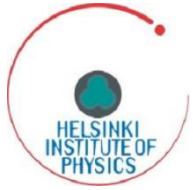
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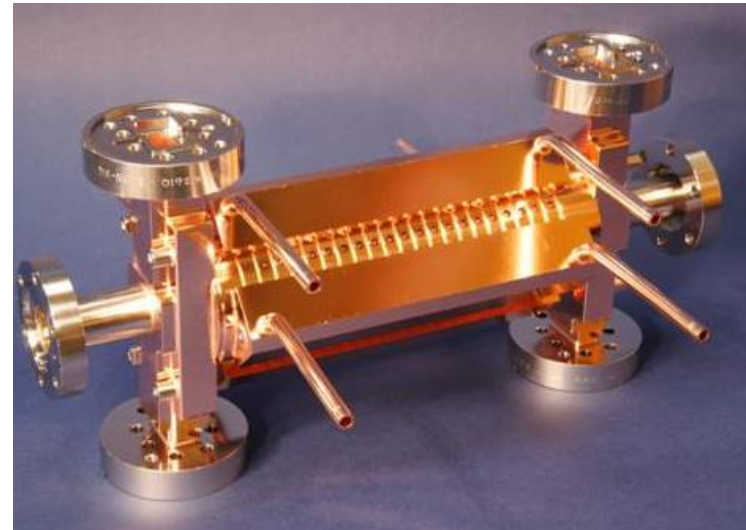
2) Helsinki Institute of Physics



Motivation

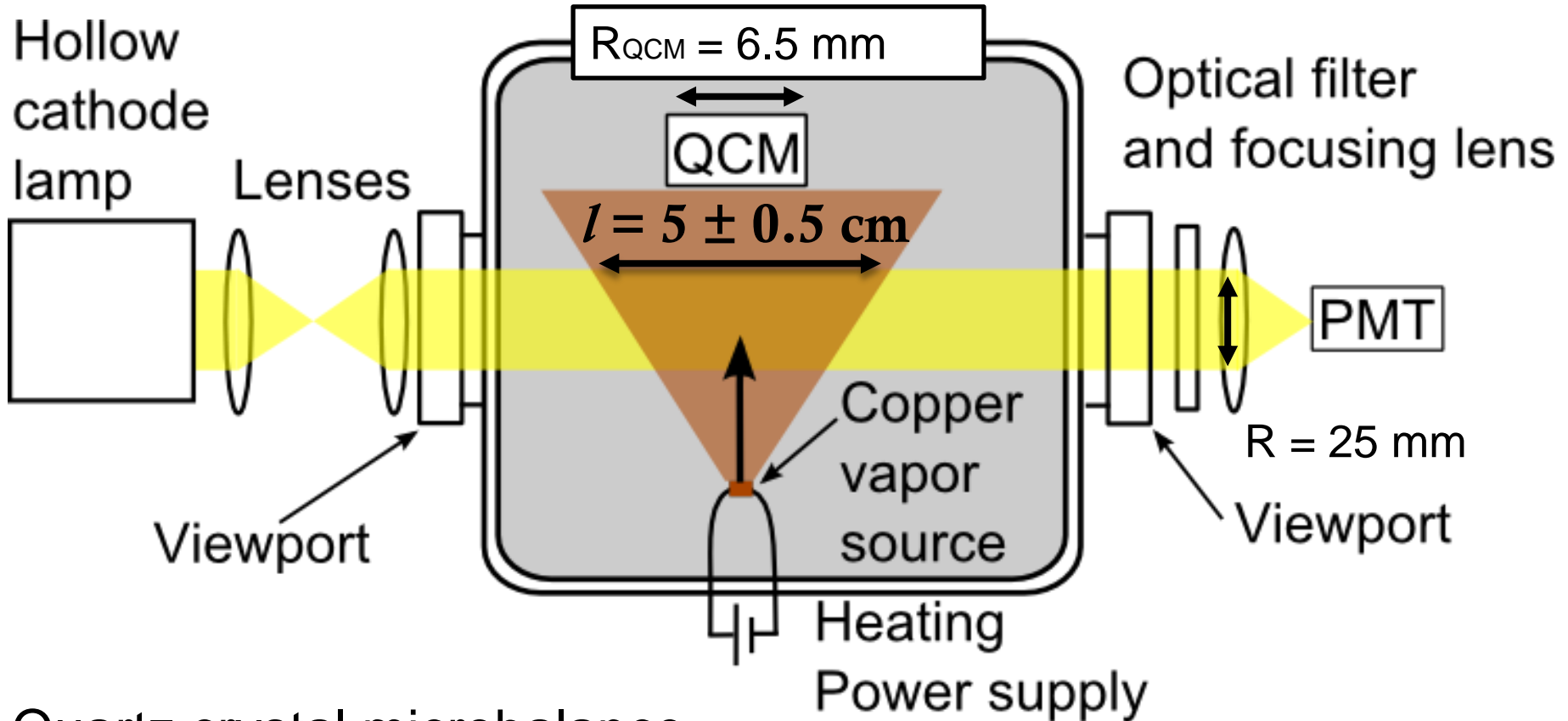
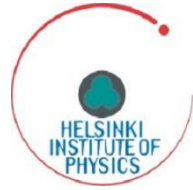


- CLIC main beam vacuum requirements from beam dynamics: H_2O , CO , CO_2 partial pressures $< 3 \cdot 10^{-9}$ mbar
- The RF causes outgassing in CLIC accelerator structures mainly to two mechanism: breakdown and dark current
- Expected pressure rise from simulations: several orders of magnitude on ~ 100 ns time scale. To be verified experimentally.
- **Aim: develop method to measure partial pressure changes of $\sim 10^{-8}$ mbar on ~ 100 ns time scale at a RF test stand ("Xbox"). For the moment focusing on absorption spectroscopy and Cu.**





Test setup

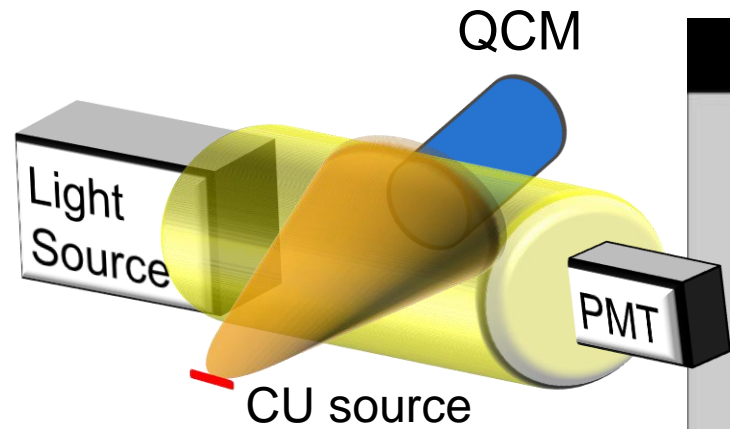
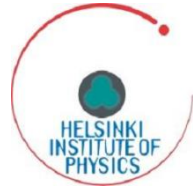


Quartz crystal microbalance (QCM) as reference

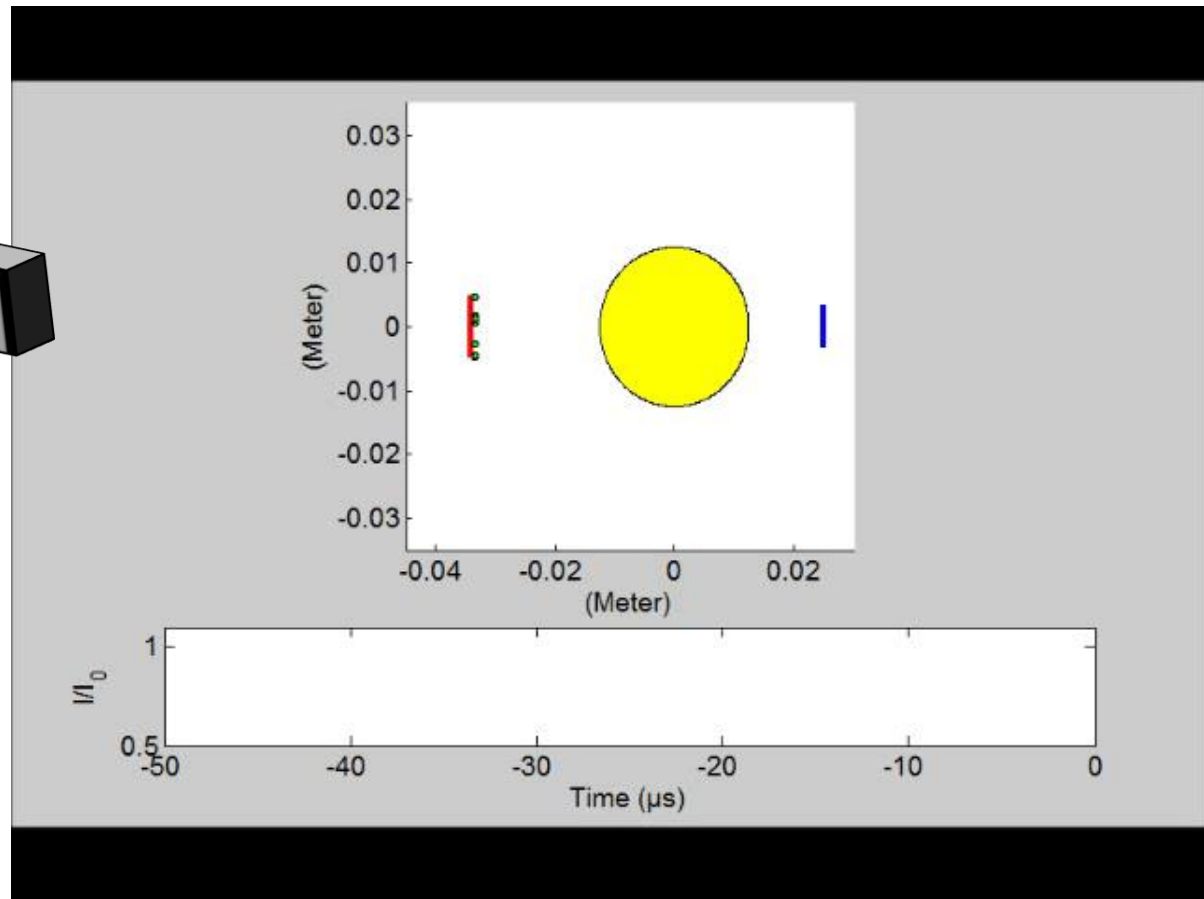
Photo multiplier tube (PMT)



Simulated Visualization

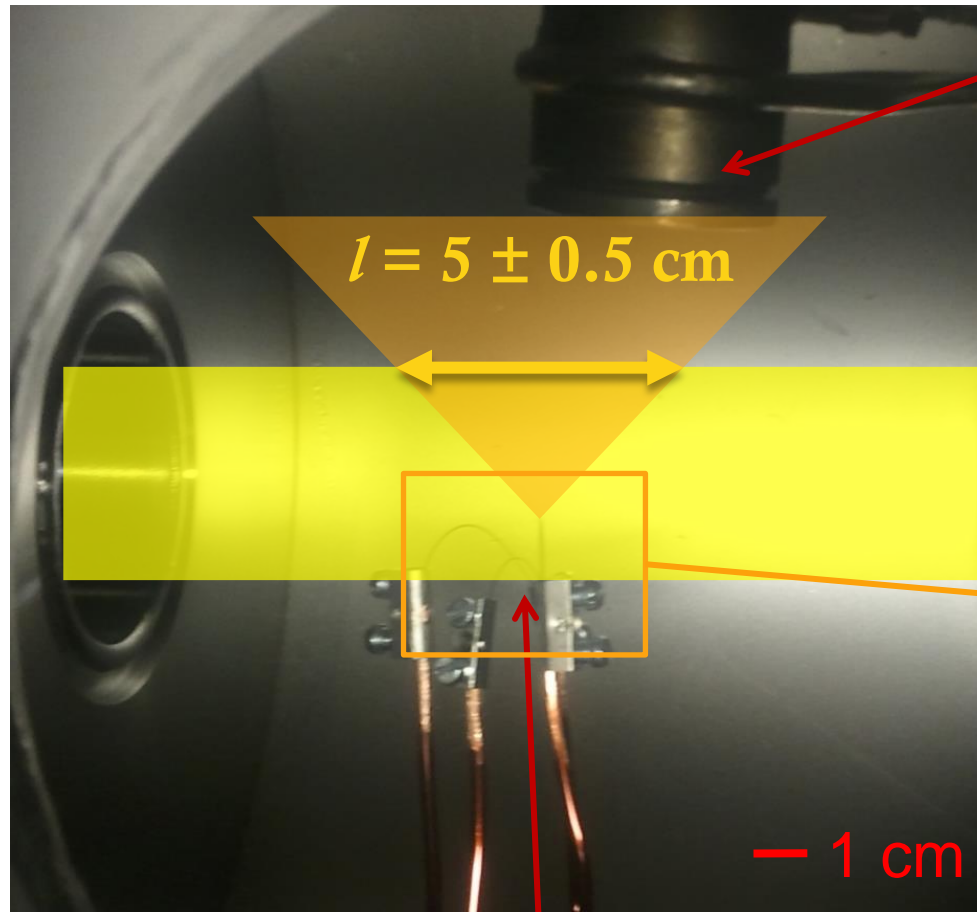
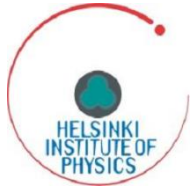


$$dp \propto N_{cu} \propto \frac{I}{I_0}$$





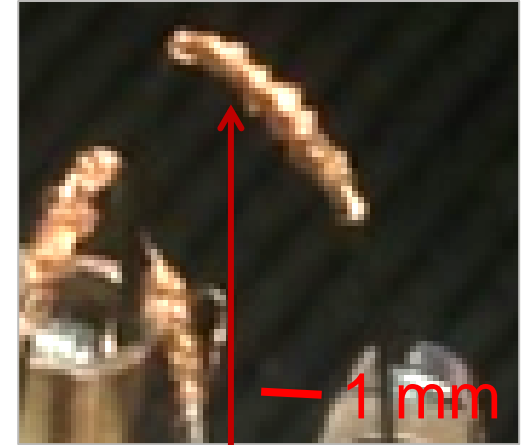
Setup inside vacuum



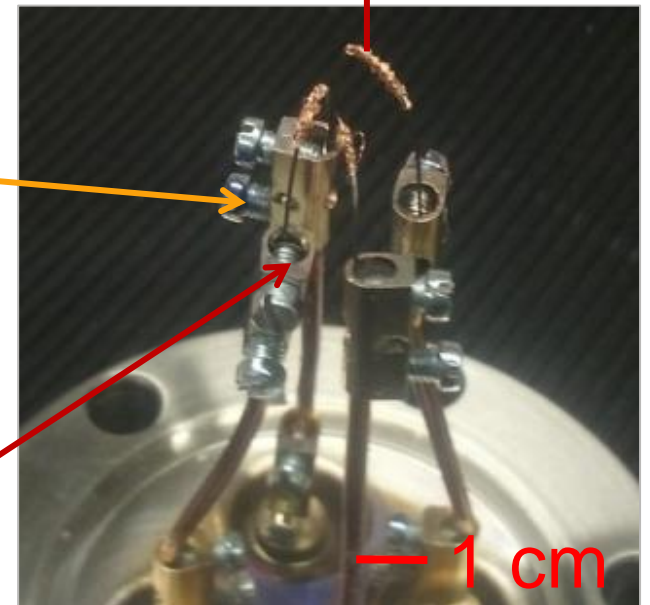
QCM

$l = 5 \pm 0.5 \text{ cm}$

— 1 cm



— 1 mm

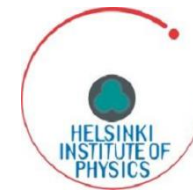


— 1 cm

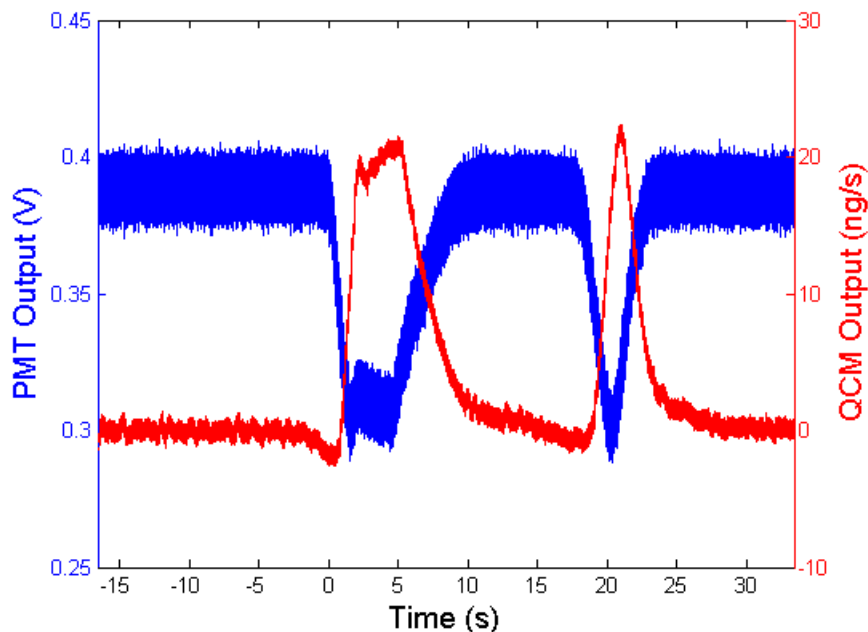
Cu vapour source:
Cu wire around **W** wire



Results

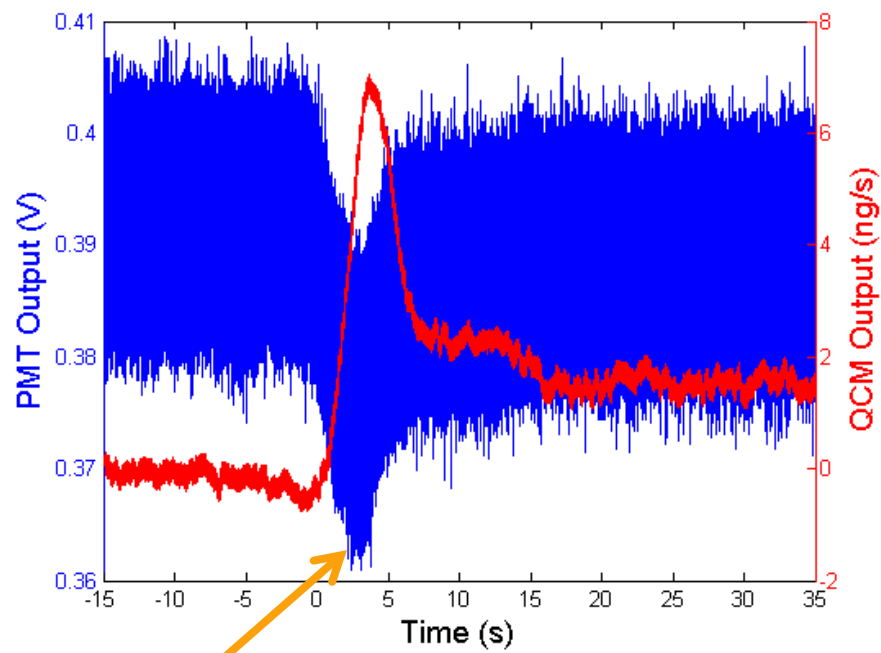


Measurement 1



Sample rate: 100 ks/s
6 kHz low pass filter for PMT
1 Hz low pass filter for QCM

Measurement 2

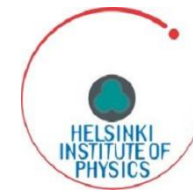


Change in intensity ~5%
 $\approx 2 * 10^{11}$ absorbances
 $\approx 10^{-4}$ mbar

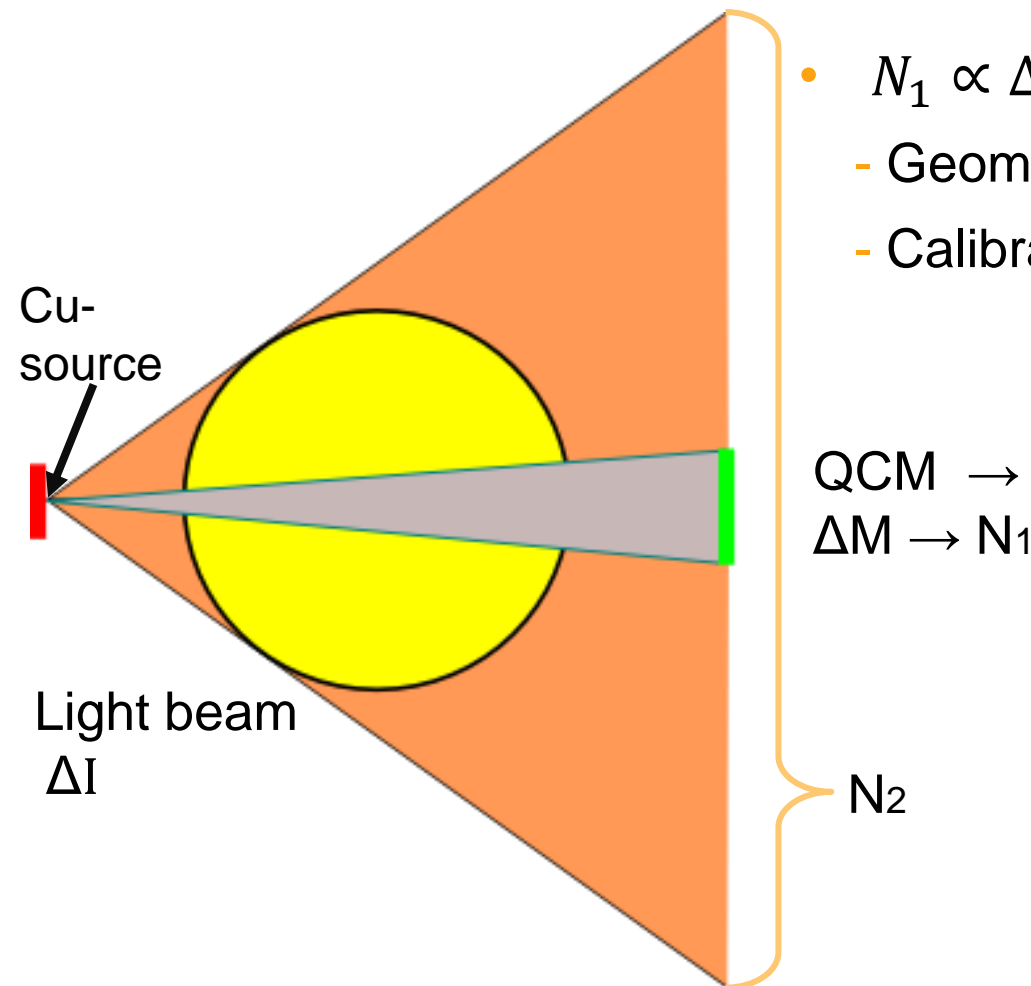


Calibration

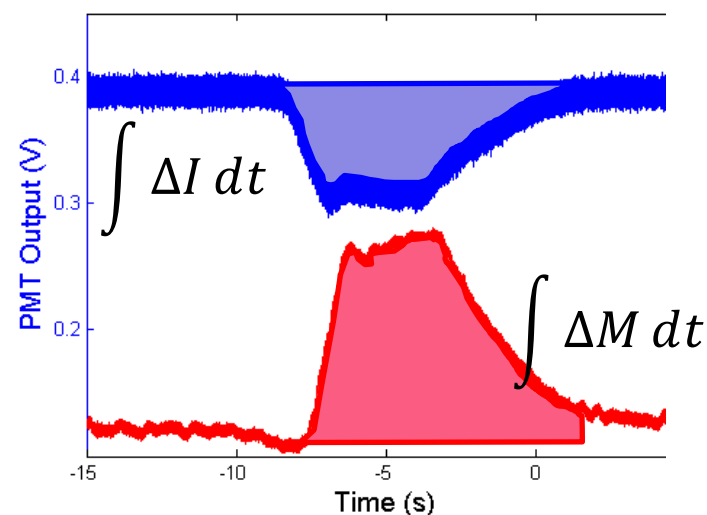
(Relate intensity change to number of atoms)



- $N_1 \propto \Delta M$ atoms was measured by QCM
 - Geometrical correction gives $N_2 \propto N_1 \propto \Delta M$
 - Calibration gives $\Delta I \propto \Delta M \propto N_1 \propto N_2$

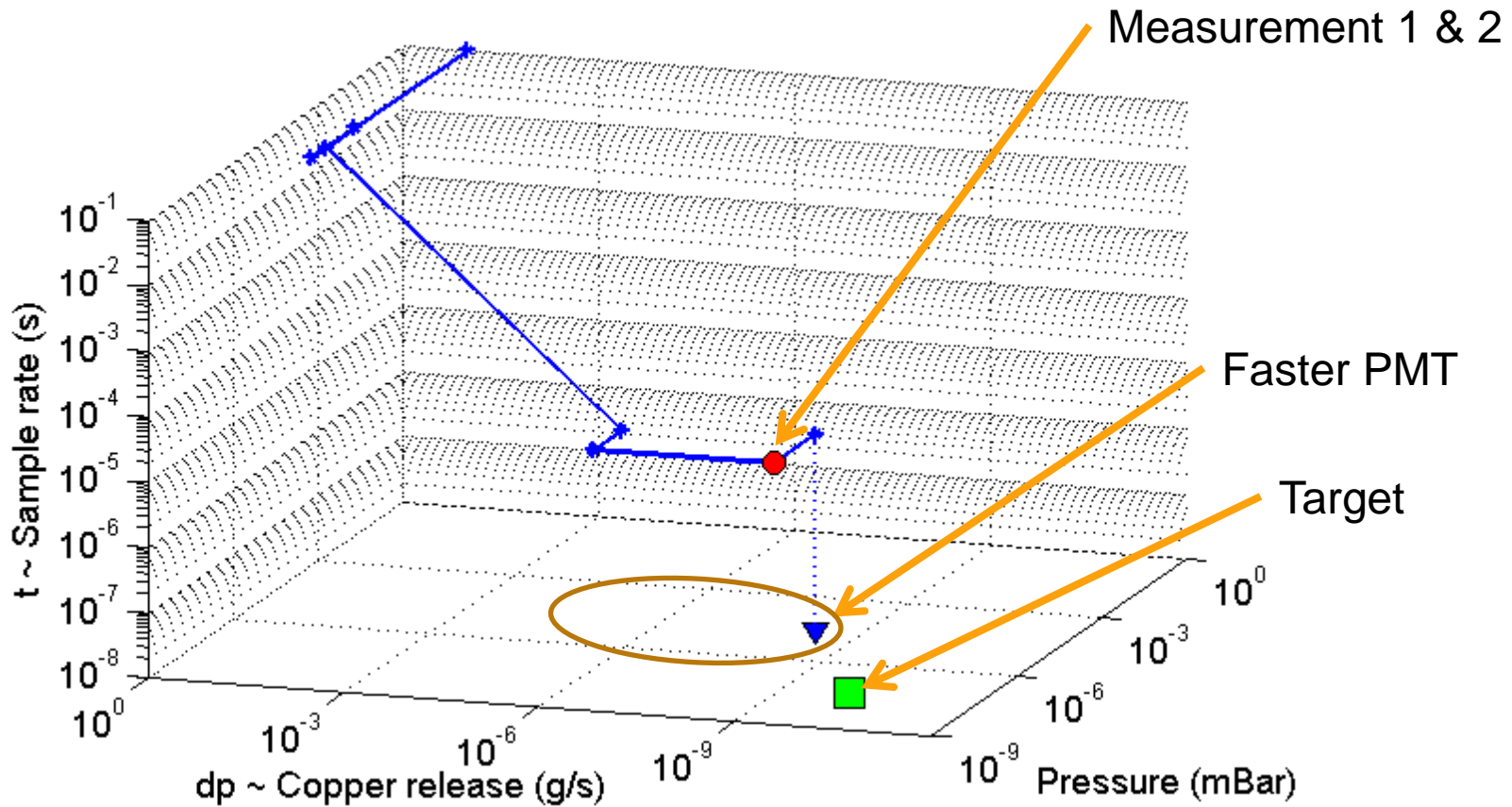
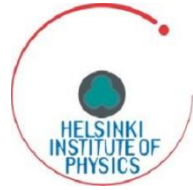


$$\int \Delta I dt \propto \int \Delta M dt = M$$





Progress t – dP – P Diagram



$$1 \text{ ng/s} = 10^{13} \text{ Cu atoms/s} \approx 5 \cdot 10^{-5} \text{ mbar}$$



Conclusion



Achieved:

Measured few ng/s Cu at 10^{-5} mbar base pressure

Challenges:

Faster response time

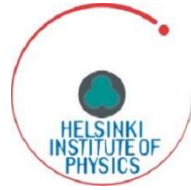
Absolute calibration

Interpretation of measurement:

Can one atom have multiple absorptions?



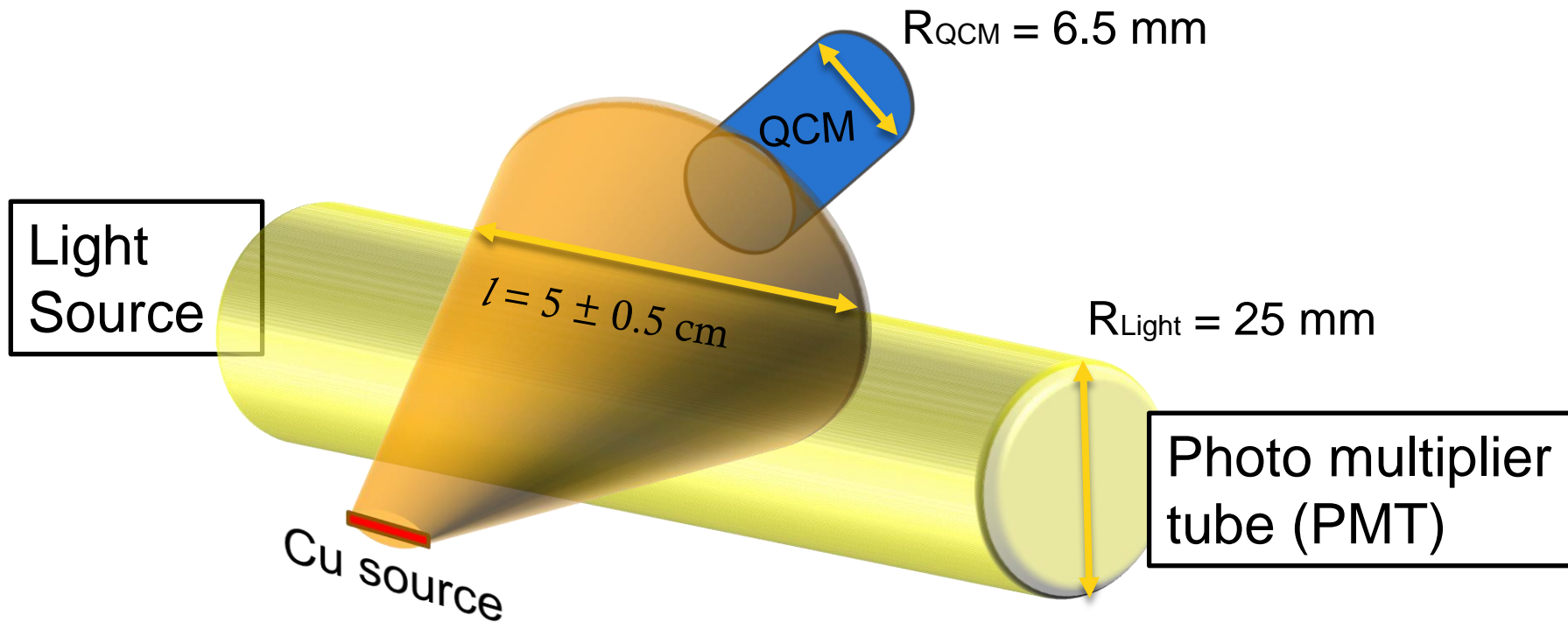
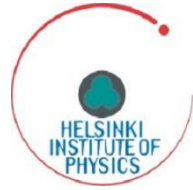
Questions and Comments



Thank You!

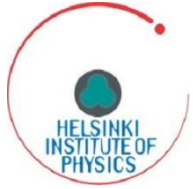


Setup & Cross section dimension

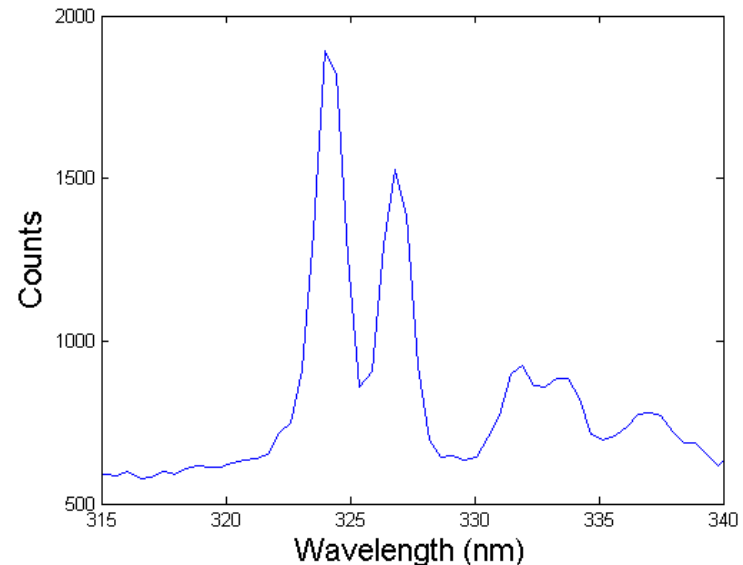




Absorption spectroscopy

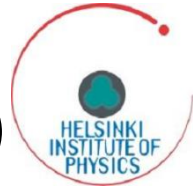


- 324.8 nm and 327.4 nm wavelengths are strongly absorbed in copper vapor
- Light source must be stable
 - Copper hollow cathode lamp is suitable low cost light source
- Light intensity changes is measured with Photomultiplier tube (PMT), bandwidth 20 kHz





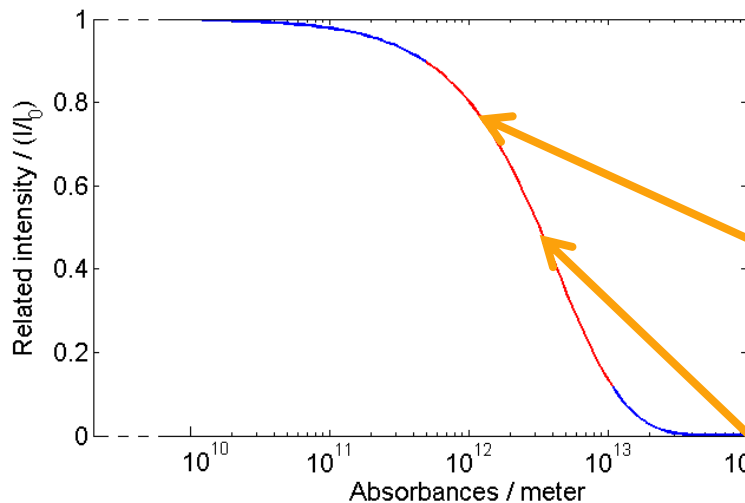
Beer-Lamberts law (Light absorption in copper vapour)



$$I = I_{0_{325}} e^{-\sigma_{325} l N} + I_{0_{327}} e^{-\sigma_{327} l N}$$

$\sigma_{325}, \sigma_{327}$ Cross section for 324.8 nm and 327.4 nm wavelengths

$I_{0_{325}}, I_{0_{327}}$ Emitted intensity of 324.8 nm and 327.4 nm wavelengths



l length of light beam in Cu vapor

N number of absorbing Cu atoms

Current operation area (Signal > |Noise|)

$I = 90\% - 10\%$

$lN = 4.9 * 10^{11} - 1.2 * 10^{13}$ Absorbances/meter

Operation point sensitivity:

$9 * 10^{11}$ Absorbances/meter