

# Neutron Sources

Christine Darve

African School of Fundamental Physics and its Applications



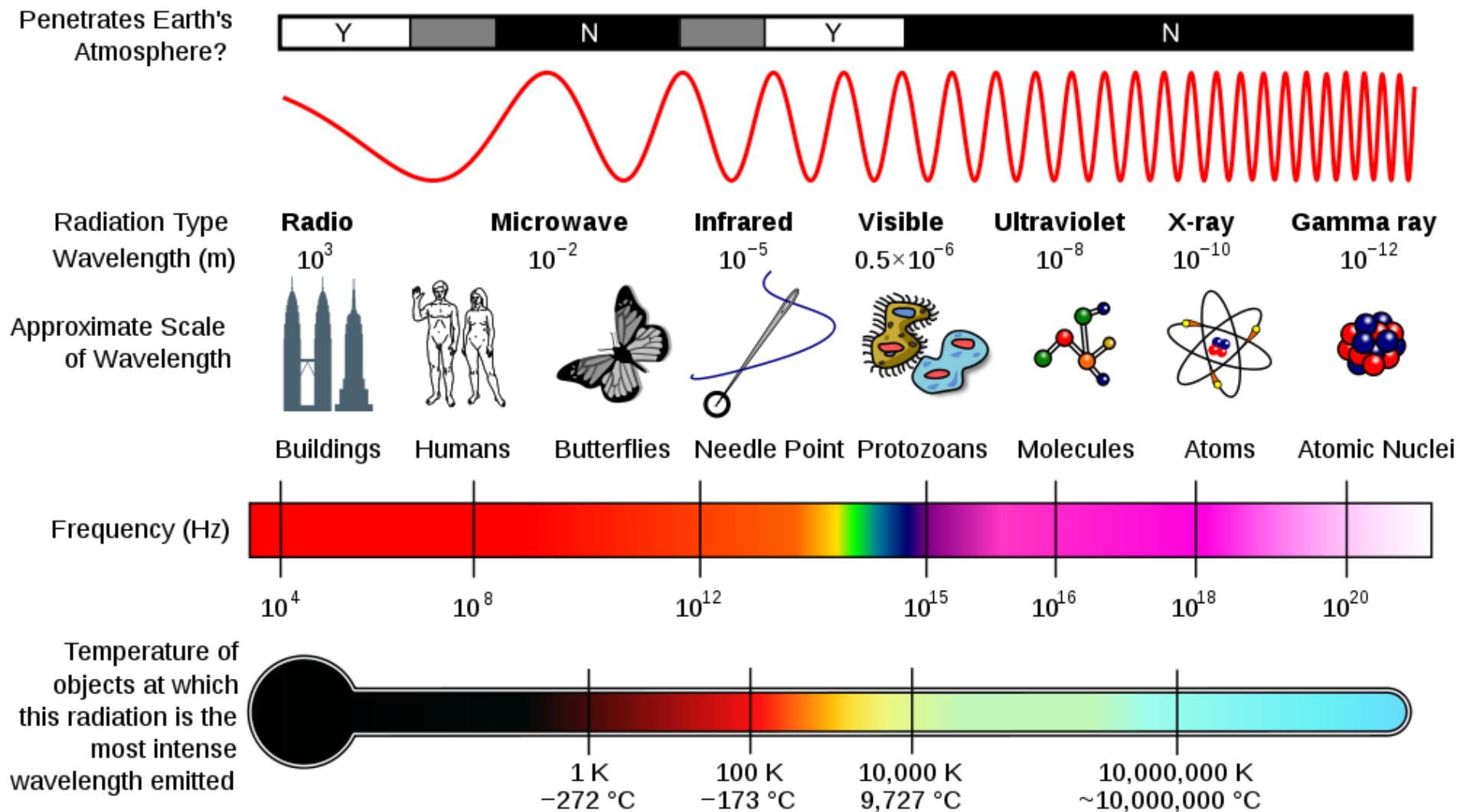
August 20, 2014

- Neutrons properties and their interactions
- How to generate intense neutron beams using high power proton linear accelerator: The example of the ESS

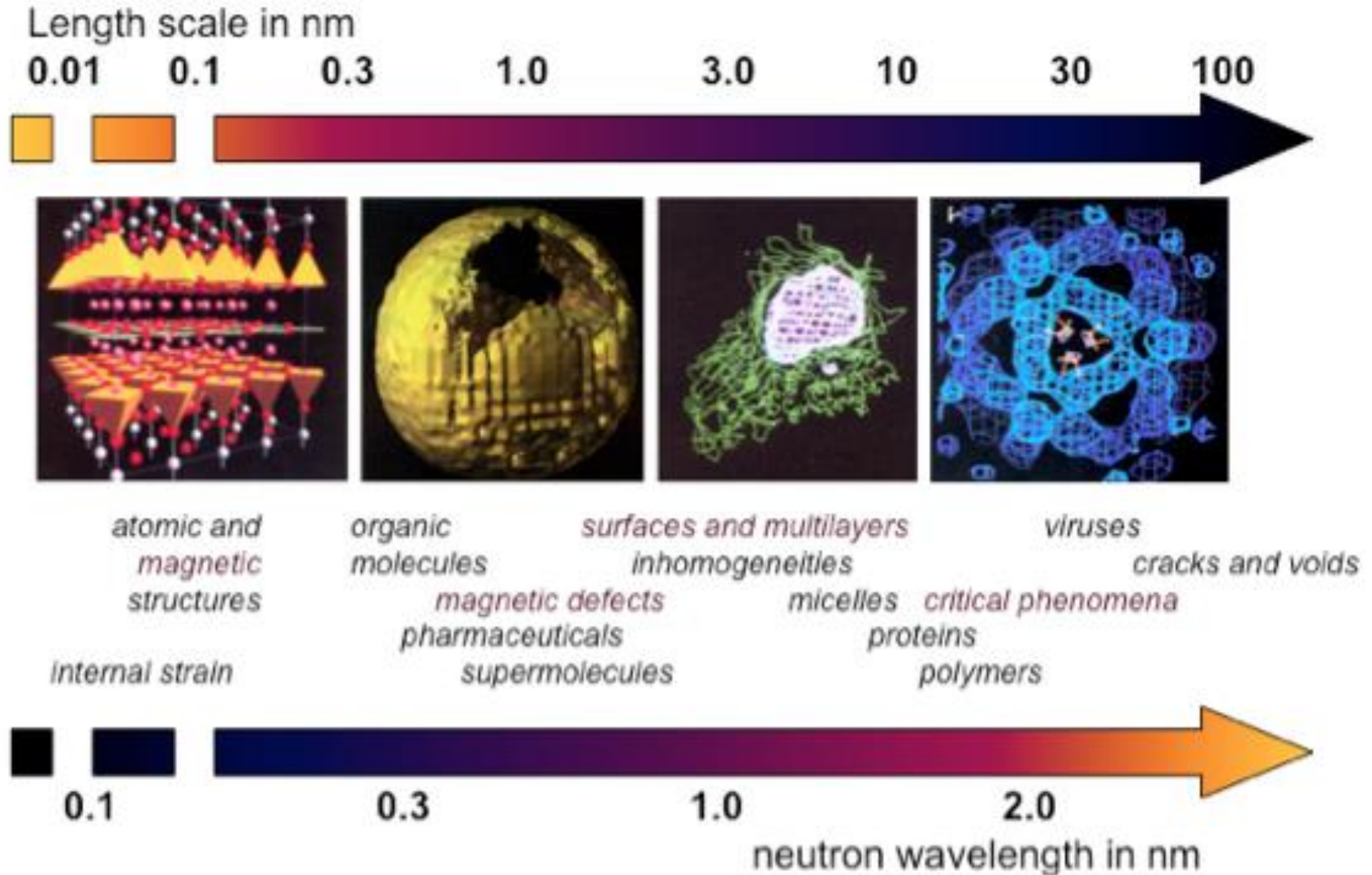
for further reading

- Applications using Neutrons

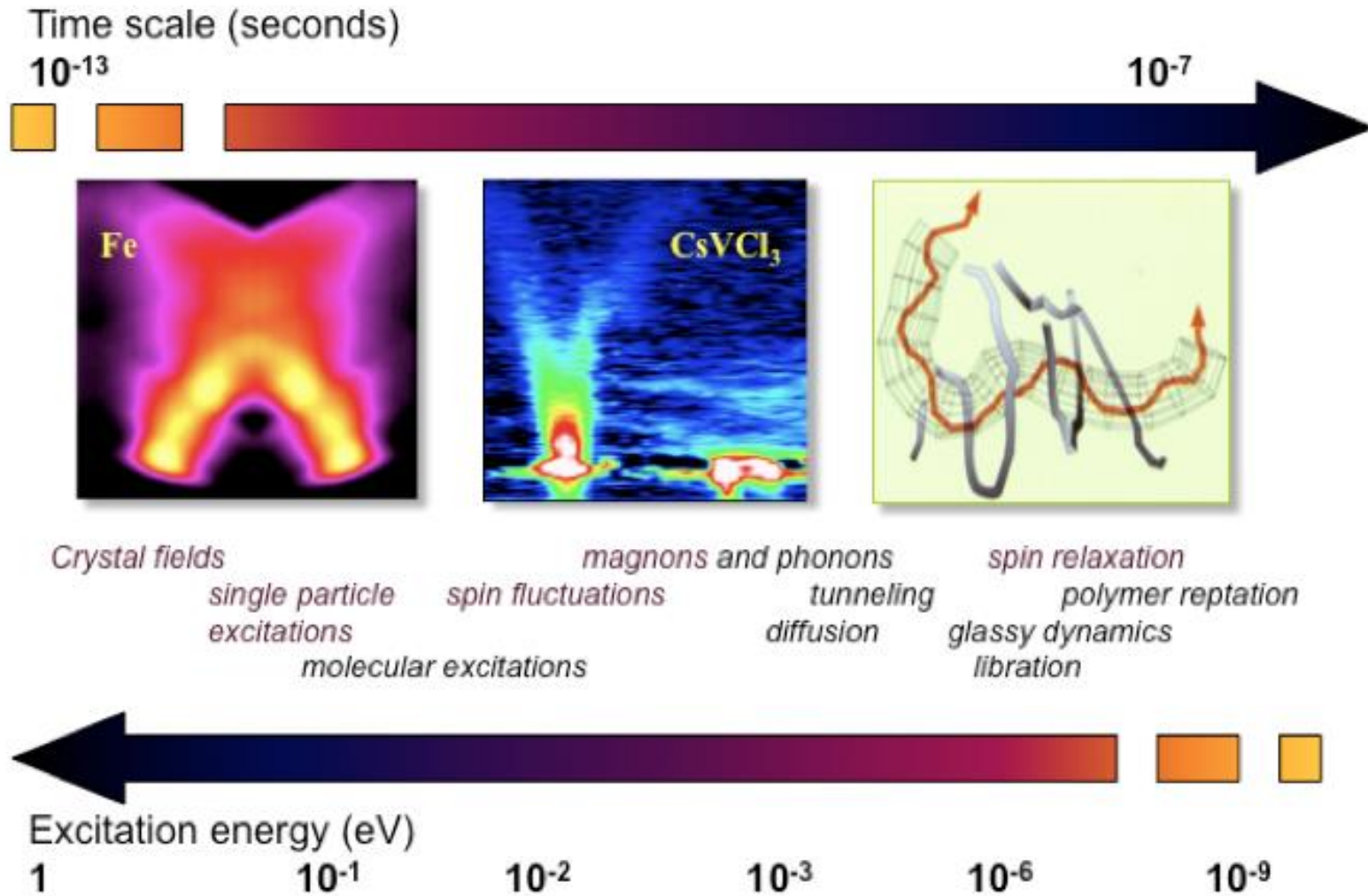
# Electro-magnetic Spectrum



# Neutron Microscope – Length scales

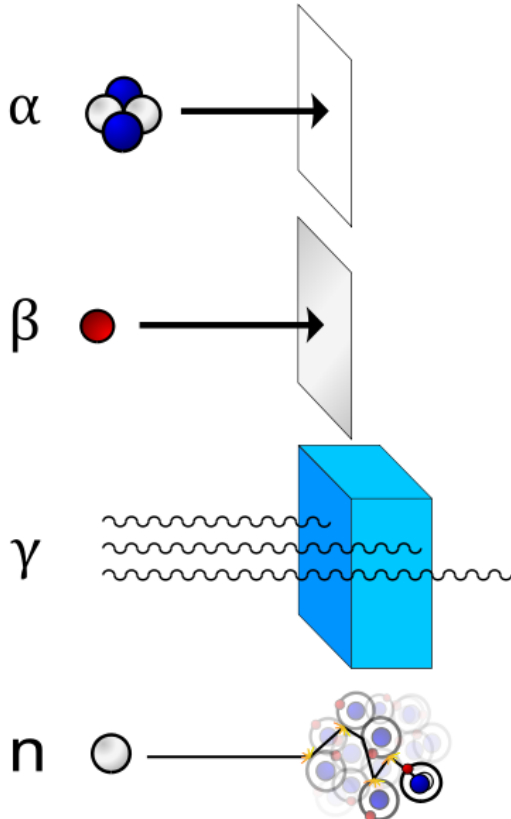


# Neutron Microscope – Time & energy scales



# Ionizing Radiation

Ionizing radiation is radiation composed of particles that individually carry enough energy to liberate an electron from an atom or molecule without raising the bulk material to ionization temperature.

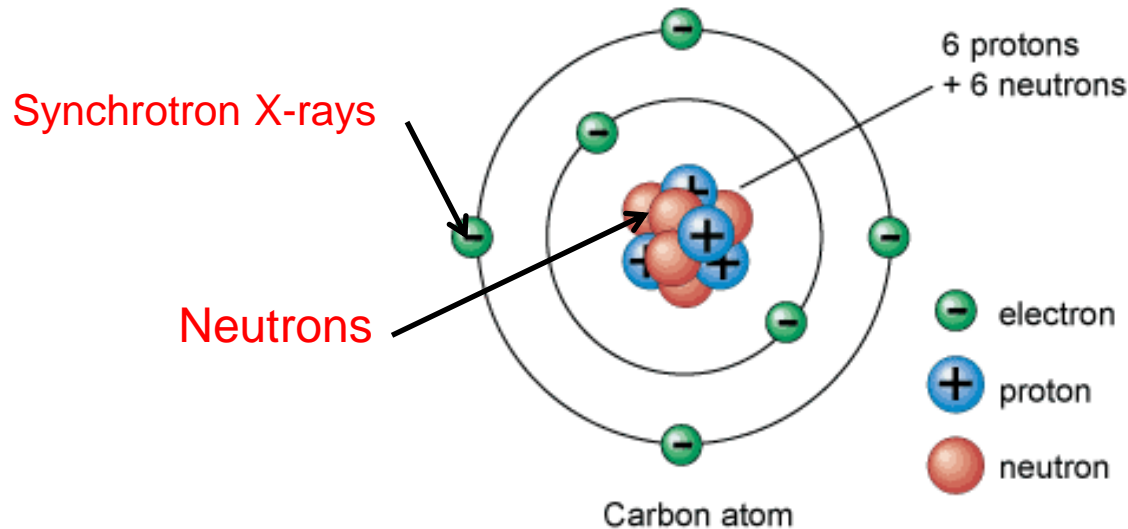


When ionizing radiation is emitted by or absorbed by an atom, it can liberate a particle. Such an event can alter chemical bonds and produce ions, usually in ion-pairs, that are especially chemically reactive.

Note: Neutrons, having zero electrical charge, **do not interact electromagnetically** with electrons, and so they cannot *directly* cause ionization by this mechanism.

→ High precision non-destructive probe ... why ?

# Neutrons Properties



Note: X-rays are emitted by electrons outside the nucleus, while gamma rays are emitted by the nucleus.

$1.675 \times 10^{-27}$ kg	Mass	939.57 MeV	$m_n \approx m_p + 2.5m_e$
	Mean lifetime	15 min	$n \rightarrow p + e^- + \bar{\nu}_e$
	Composition	udd	hadron
	Electric charge	0	high penetration
	Magnetic moment	$-1.04 \mu_B$	feels the nucleus

# Neutron Energy

$$E = k_B T$$

Boltzmann distribution

$$E = k_B T = \frac{1}{2} m v^2 = \frac{h^2}{2m \lambda^2}$$

$$\lambda = \frac{h}{m v}$$

De Broglie

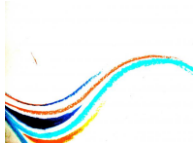
$$E [meV] = 0.0862 T [K] = 5.22 v^2 [km/s] = 81.81 \frac{1}{\lambda^2} [A]$$

Source	Energy	Temperature	Wavelength
cold	0.1-10	1-120	30-3
thermal	5-100	60-1000	4-1
hot	100-500	1000-6000	1-0.4



# Why neutrons?

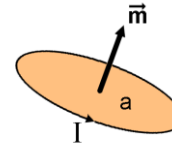
Wave



Particle




Magnetic moment





Neutral





Neutron properties are used to understand the nature of the solid and liquid states of matter, as an analytical tool to aid the development of materials and as a tool to examine curiosity-driven research that spans from cosmology, superconductivity to the dynamics of the molecules of life.


- 

**1. Neutrons see the Nuclei**
- 

**2. Neutrons see Elementary Magnets**
- 

**3. Neutrons see light Atoms next to Heavy Ones**
- 

**4. Neutrons measure the Velocity of Atoms**
- 

**5. Neutrons penetrate deep into Matter**
- 

**6. Neutrons are Elementary Particles**

# Why neutrons?

**Electrically Neutral** – neutrons are non-destructive and can penetrate deep into matter. This makes them an ideal probe for biological materials and samples under extreme conditions of pressure, temperature, magnetic field or within chemical reaction vessels.

**Microscopically Magnetic** – they possess a magnetic dipole moment which makes them sensitive to magnetic fields generated by unpaired electrons in materials. Precise information on the magnetic behavior of materials at atomic level can be collected. In addition, the scattering power of a neutron off an atomic nucleus depends on the orientation of the neutron and the spin of the atomic nuclei in a sample. This makes the neutron a powerful instrument for detecting the nuclear spin order.

**Ångstrom wavelengths** – neutron wavelengths range from 0.1 Å to 1000 Å, making them an ideal probe of atomic and molecular structures, be they single atomic species or complex biopolymers.

# Why neutrons?

**Energies of millielectronvolts** – their energies are of the **same magnitude as the diffusive motion in solids and liquids**, the coherent waves in single crystals (phonons and magnons), and the vibrational modes in molecules. It is easy to detect any exchange of energy between a sample of between 1 microeV (even 1 neV with spin-echo) and 1 eV and an incoming neutron.

**Randomly sensitive** – with neutrons the variation in scattering power from one nucleus to another within a sample varies in a quasi-random manner. This means that lighter atoms are visible despite the presence of heavier atoms, and neighboring atoms may be distinguished from each other. In addition, **contrast** can be varied in certain samples **using isotopic substitution (for example D for H, or one nickel isotope for another)**; specific structural features can thus be highlighted. The neutron is particularly sensitive to hydrogen atoms; it is therefore a powerful probe of hydrogen storage materials, organic molecular materials, and biomolecular samples or polymers.

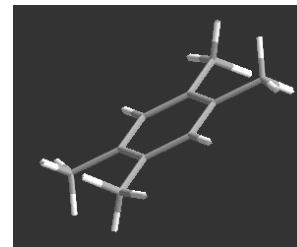
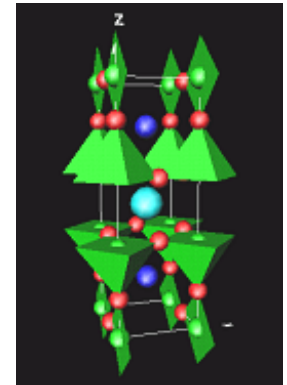
# Why neutrons?

In half a century we have developed neutron scattering science enormously with an effective gain in source performance of **only a factor of 4 !**

- Diffractometers - Measure structures  
– Where atoms and molecules are

1 - 10 Ångström

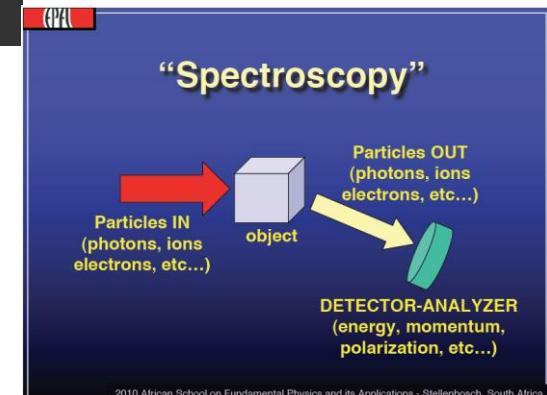
→ To analyze the structure of a material from the scattering pattern produced when a beam of radiation or particles (such as X-rays or neutrons) interacts with it



- Spectrometers - Measure dynamics  
– What atoms and molecules do

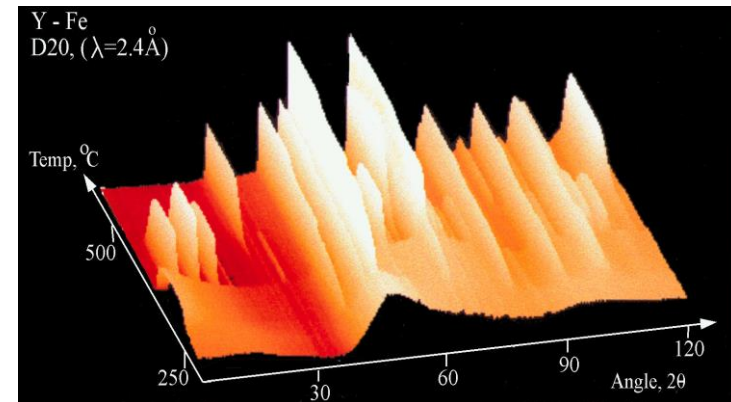
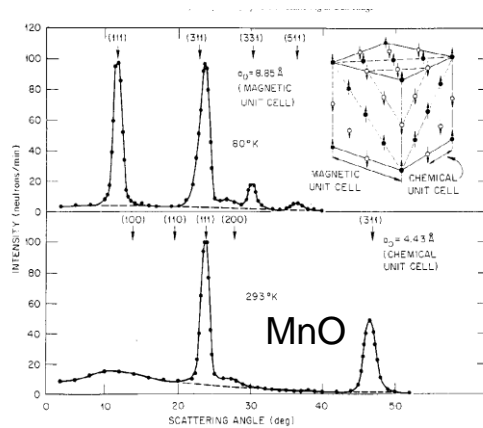
1 - 80 meV

→ To measure properties of light over a specific portion of the electromagnetic spectrum

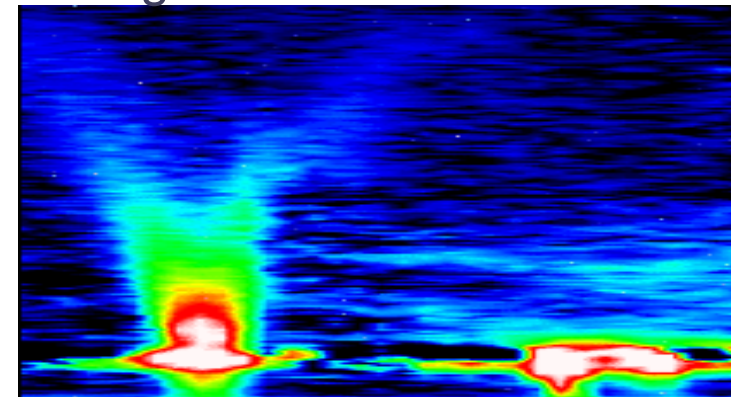
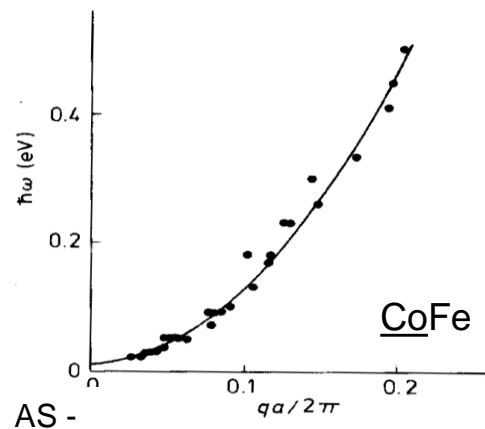


# Why neutrons?

Cliff Shull – Neutron diffraction - showing where atoms are:



Bert Brockhouse – Spectroscopy - showing what atoms do:



*Using nowadays technique..*

# The Nobel Prize in Physics 1994



Clifford G. Shull, MIT, Cambridge, Massachusetts, USA, winner one half of the 1994 Nobel Prize in Physics for development of the neutron diffraction technique.



**S** Shull made use of elastic scattering, i.e. of neutrons which change direction without losing energy when they collide with atoms.

Because of the wave nature of neutrons, a diffraction pattern can be recorded which indicates where in the sample the atoms are situated. Even the placing of light elements such as hydrogen in metallic hydrides, or hydrogen, carbon and oxygen in organic substances can be determined.

The pattern also shows how atomic dipoles are oriented in magnetic materials, since neutrons are affected by magnetic forces. Shull also made use of this phenomenon in his neutron diffraction technique.



Shull's laboratory with P. H. Geiger and P. E. Smith (winners of the 1954 Nobel Prize in Physics).

## Neutrons see more than X-rays

X-rays are scattered by electrons whereas neutrons scatter from nuclei. While X-rays are scattered by electrons, neutrons scatter from nuclei. Neutrons are scattered by nuclei, not by electrons, so they can see through atoms, all kinds of elements are visible.



## Neutrons reveal linear stresses

It tells the beam parallel to an important metal duct pipe. Over the past twenty years, neutron diffraction has been used to study the internal stresses in materials. Neutrons are scattered by nuclei, not by electrons, so they can see through atoms, all kinds of elements are visible.



## Neutrons show what atoms remember

Of these, only neutrons can see the magnetic moments of atoms. Neutrons are scattered by nuclei, not by electrons, so they can see through atoms, all kinds of elements are visible.



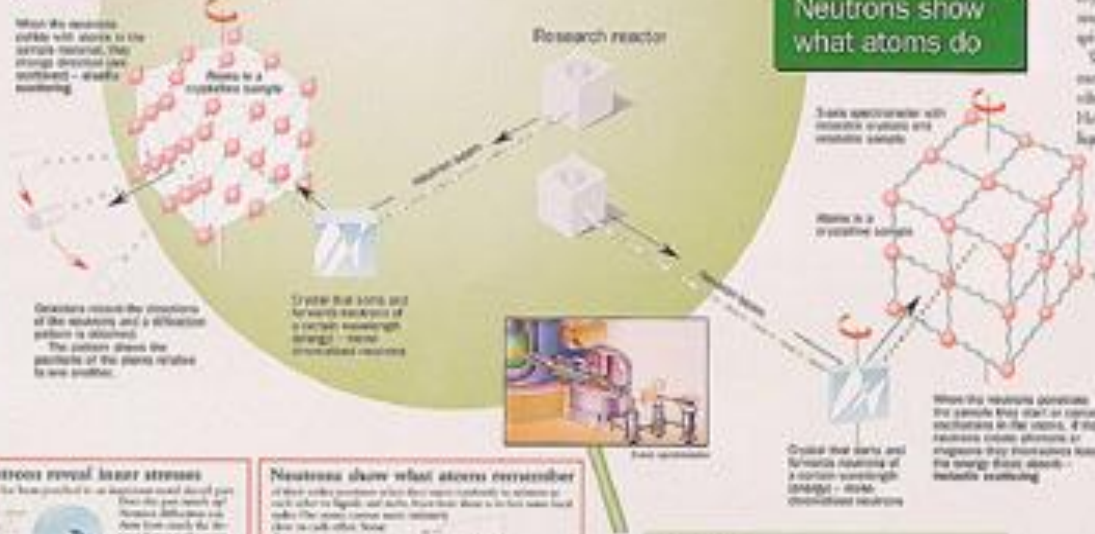
Neutrons behave as particles and as waves

# Neutrons reveal structure and dynamics

Neutrons show where atoms are

Neutrons bounce against atomic nuclei. They also react to the magnetism of the atoms.

Neutrons show what atoms do



Bertram N. Brockhouse, McMaster University, Hamilton, Ontario, Canada, winner one half of the 1994 Nobel Prize in Physics for the development of neutron spectroscopy.



**B** Brockhouse made use of inelastic scattering, i.e. of neutrons, which change both direction and energy when they collide with atoms. They then start in certain atomic conditions in crystals and sound waves in liquids and solids. Phonons can also interact with spin waves in magnets.

With his 3-axis spectrometer Brockhouse measured energies of phonons (atomic vibrations) and magnons (magnetic waves). He also studied how atomic vibrations in liquids change with time.

Further reading:



Neutrons show us : where atoms are (left, Bragg scattering) and what they do (right, Spectroscopy).  
 Energies of neutrons are same scale as excitations. 1 Aneutron 6orders of magnitude less than 1 A Synch.

# Scientific challenges



## Solid State Physics

*Dynamics of superlattices, wires and dots, molecular magnets, quantum phase transitions*

## Liquids and Glasses

*Solvent structures, influence of molecular structures on protein folding*

## Fundamental Physics

*Left and right handedness of the universe, neutron decay, ultracold neutrons*

## Soft Condensed Matter

*Time resolution, molecular rheology, structures and dynamics*

## Biology and Biotechnology

*Hydrogen and water, membranes, biosensors, functions*

## Materials Science and Engineering

*Real time investigations with realistic dimensions under real conditions*

## Chemical Structure, Kinetics and Dynamics

*Thin films, pharmaceuticals, supramolecules - structures and functionality*

## Earth and Environmental Science, Cultural Heritage

*Extreme temperatures and pressures simulating the mantle*

# Fields of interest

A wide range of length and timescales.

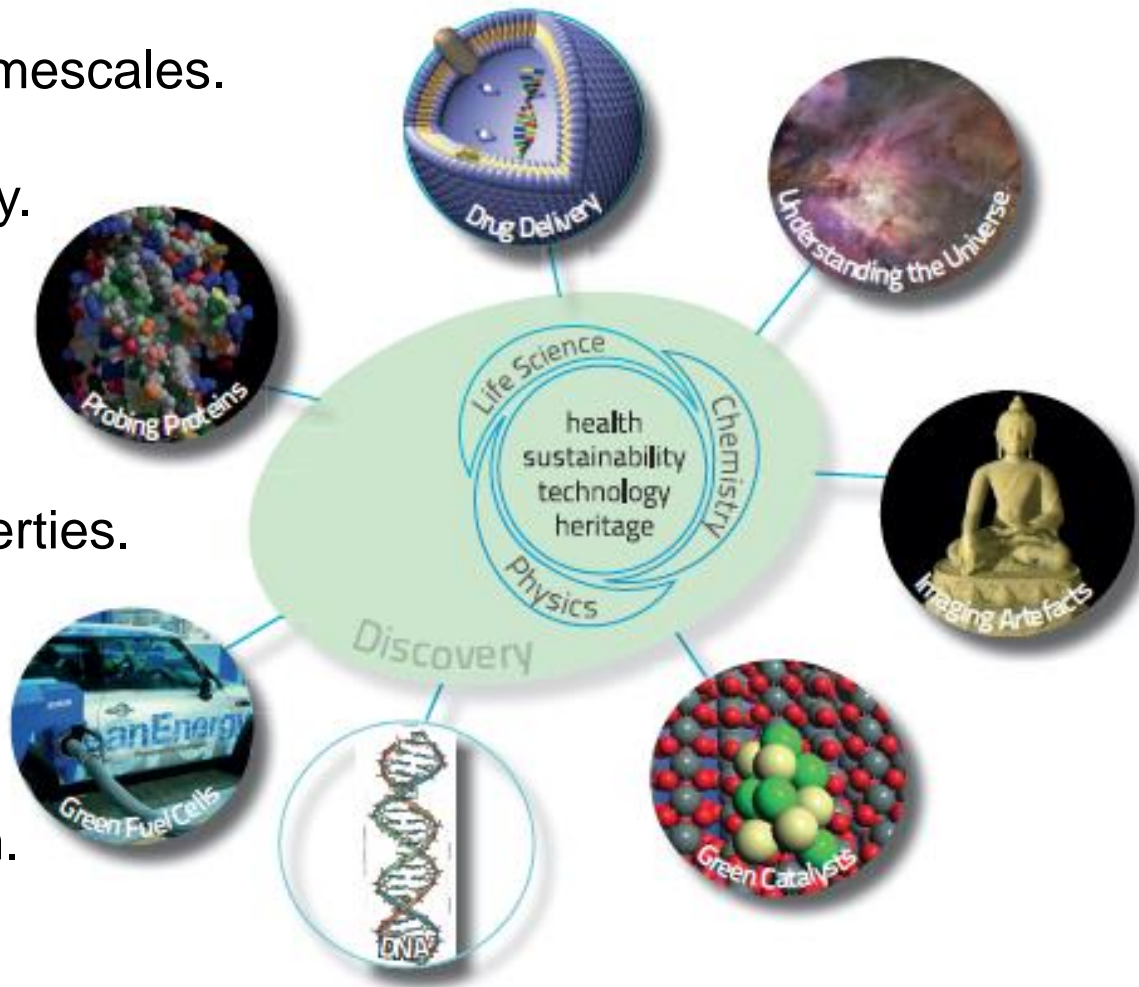
High sensitivity and selectivity.

Deep penetration.

A probe of fundamental properties.

A precise tool

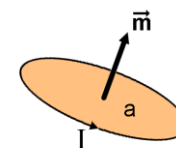
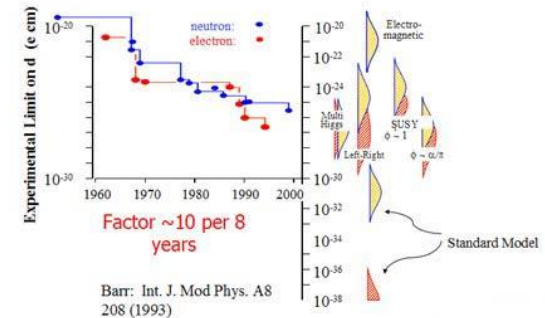
An ideal probe for magnetism.





# The visions

- Room Temperature Super Conductors
- Sterile neutrinos
- Hydrogen storage substrate
- Neutron electric dipole moment
- Efficient membrane for fuel cells
- Flexible and highly efficient solar cells
- Carbon nano-tubes for controlled drug release
- Self healing materials – smart materials
- Spin-state as a storage of data ( $10^{23}$  gain in capacity)
- CO<sub>2</sub> sequestration

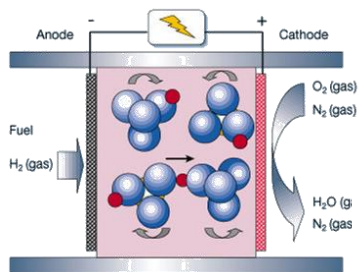
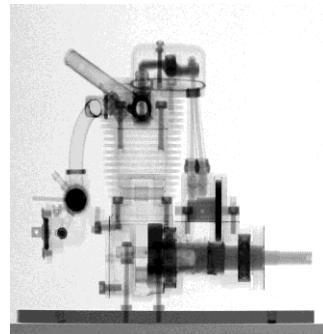
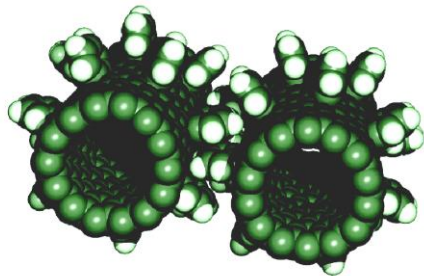
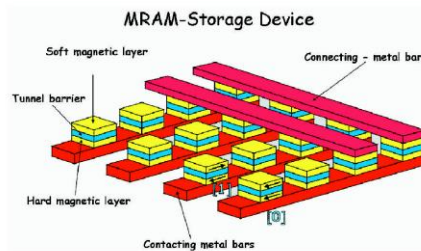
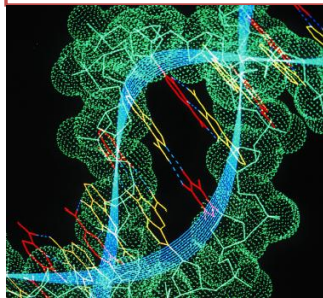


# Multi-science with neutrons

Materials science  
Energy Technology

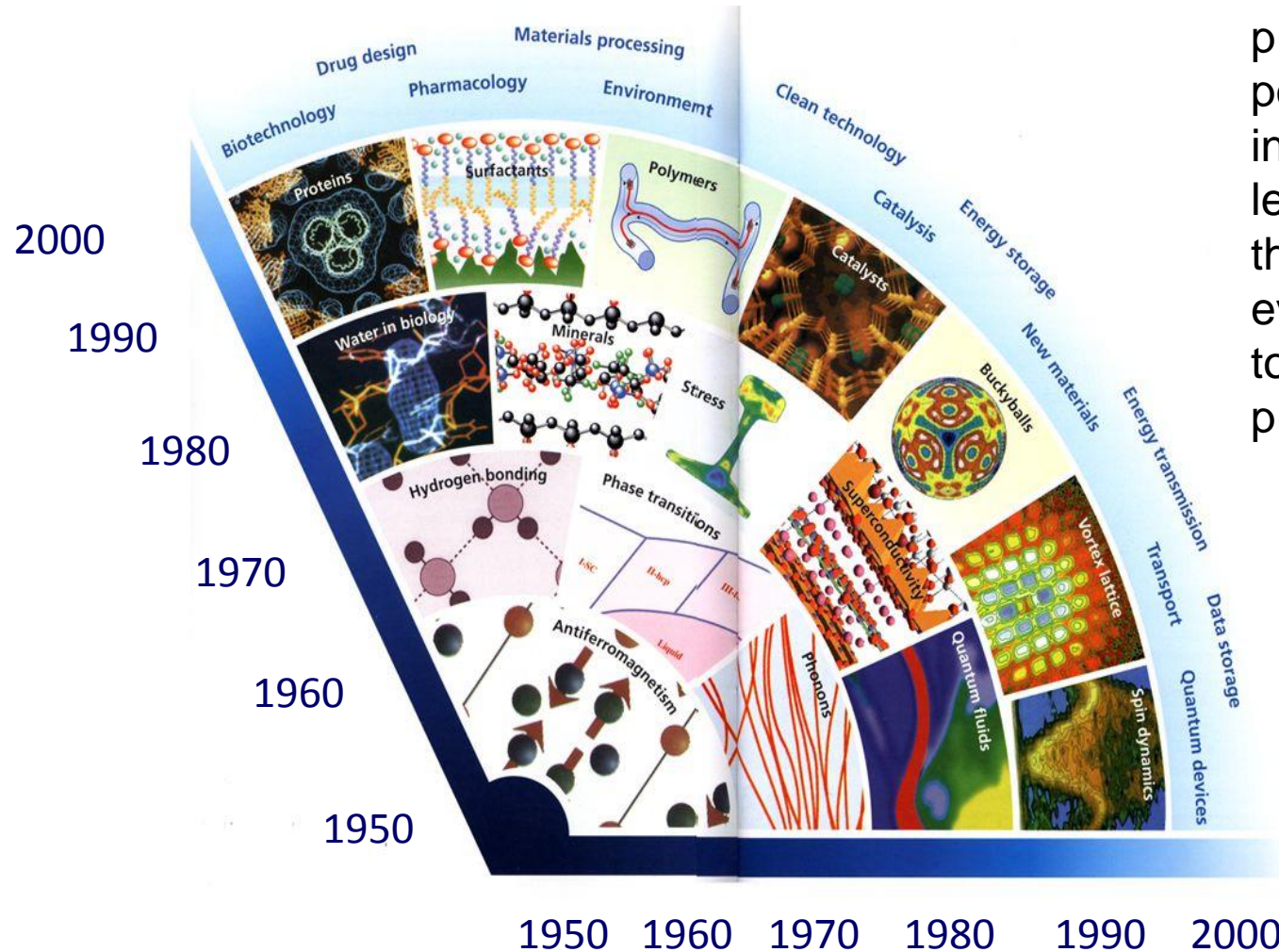
Bio-technology  
Hardware for IT

Nano science  
Engineering science



- Neutrons can provide unique and information on almost all materials.
- Information on both structure and dynamics simultaneously. "Where are the atoms and what are they doing?"
- 6000 primary users in Europe today and 6000 secondary users. Access based on peer review.
- Science with neutrons is limited by the intensity of today's sources

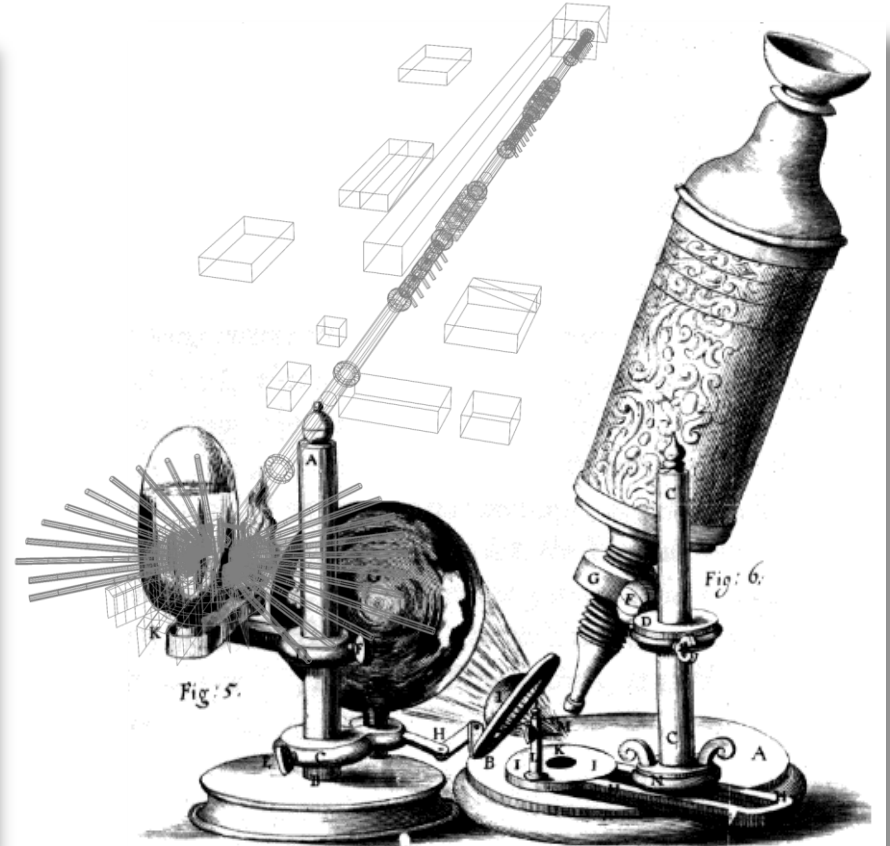
# Neutron scattering - an expanding field



Because of their unique properties, neutrons are a powerful tool for investigating Nature at all levels, from testing theories about the evolution of the Universe to elucidating the complex processes of life.

# X-Ray and Neutron beam

# Complementarity between X-rays and Neutrons



Consider ESS/MaxIV equal in terms of functionality

# Complementarity between X-rays & Neutrons

## Neutrons

- Particle beam (neutral subatomic particle)
- Interactions with the nuclei and the magnetic moment of unpaired electrons (in the sample)
- Scattered by all elements, also the light ones like the hydrogen isotopes
- Deep penetration depth (bulk studies of samples)
- Less intense beam measuring larger samples

## Synchrotron radiation

- Light beam (electromagnetic wave)
- Interactions with the electrons surrounding the nuclei (in the sample)
- Mainly scattered by heavy elements
- Small penetration depth (surface studies of samples)
- Very intense beam measuring small or ultra-dilute samples

# Complementarity between X-rays & Neutrons

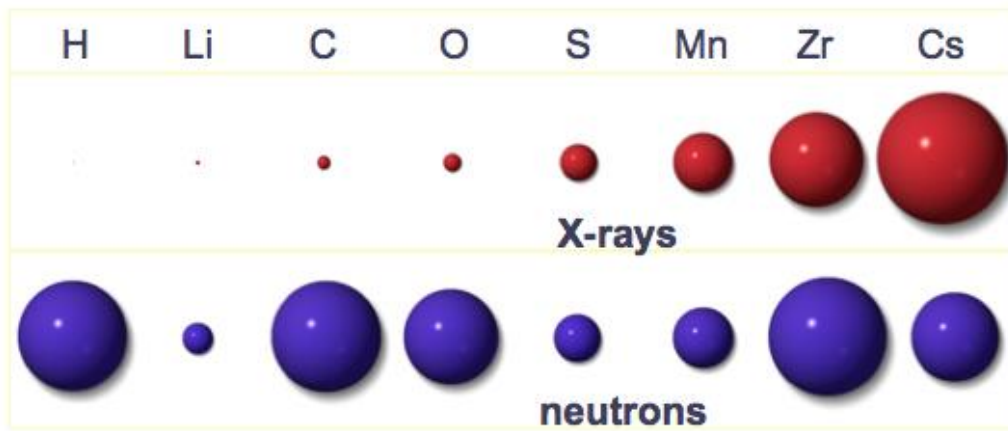
## Neutrons Applications

- Magnetic structures & excitations
- Organic structures using the H-D isotope effect
- Bulk studies (strains, excitations)
- Low-energy spectroscopy e.g. molecular vibrations

## SR Applications

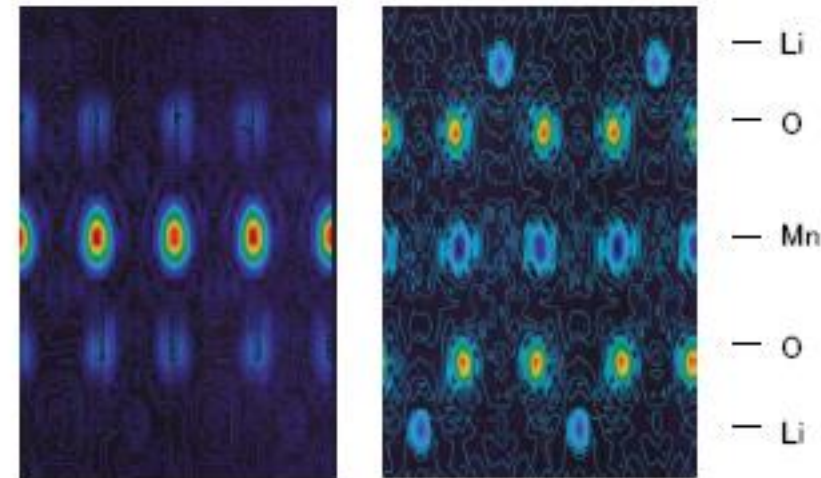
- Protein-crystal structures
- Fast chemical reactions
- Surface studies (defects, corrosion)
- High-energy spectroscopy e.g. measurements of electron energy-levels

# Complementarity between X-rays & Neutrons



X-ray interact with electrons  
 → X-rays see high-Z atoms

Neutron interact with nuclei  
 → Neutrons see low-Z atoms

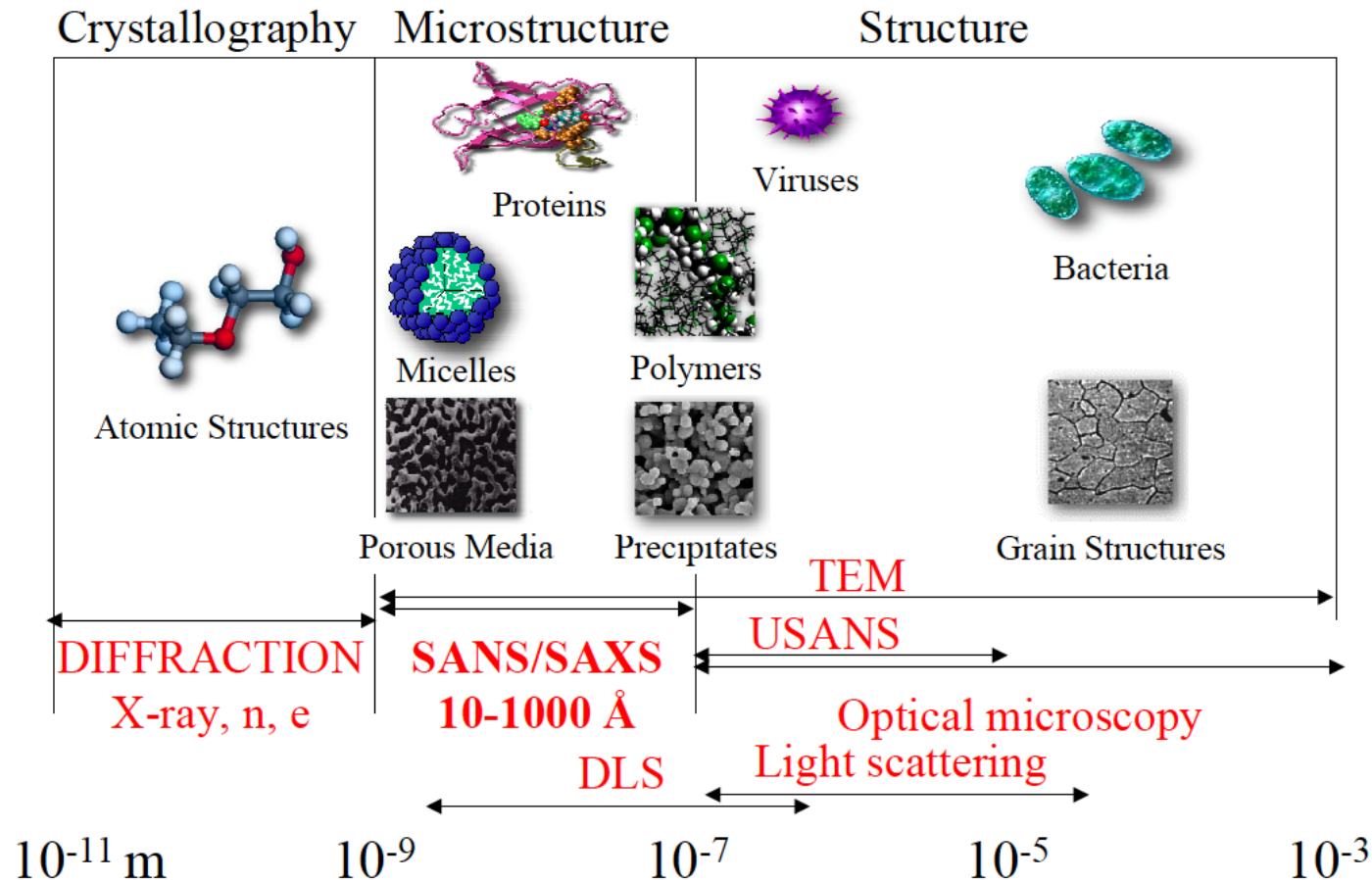


Material for Li-battery seen by  
 X rays (left) and  
 Neutrons (right)

**T. Kamiyama, et al.**



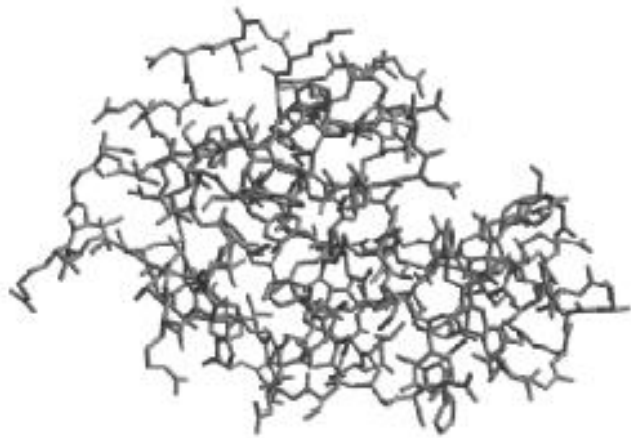
# Small Angle Neutron Scattering



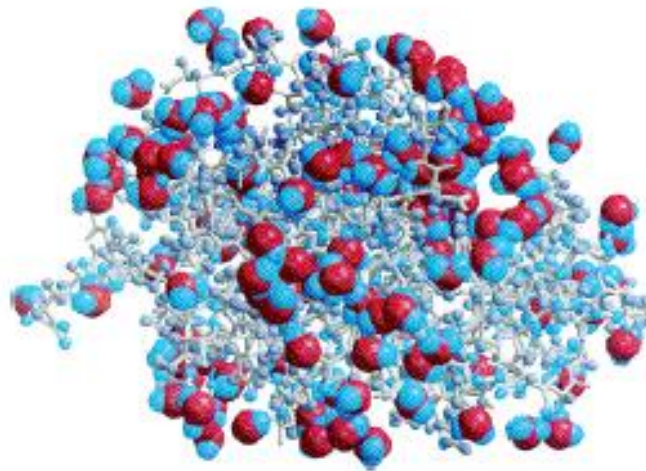
Advantages of SANS over SAXS are its sensitivity to light elements, the possibility of isotope labeling, and the strong scattering by magnetic moments.

# Complementarity between X-rays & Neutrons

Hen Egg-White Lysozyme



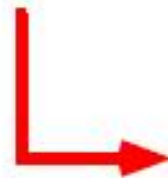
X-rays



Water molecules  
Observed with  
neutrons

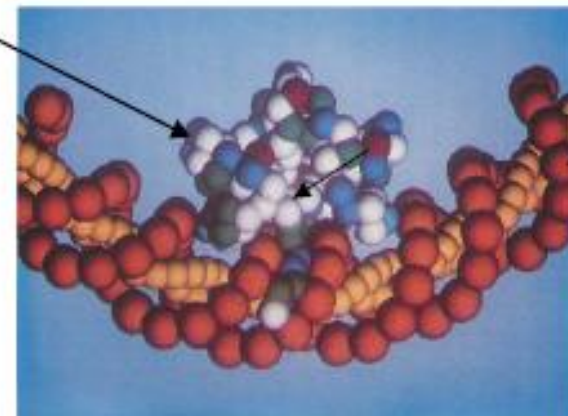
**N. Niimura, et al.**

Neutrons



From structure to function

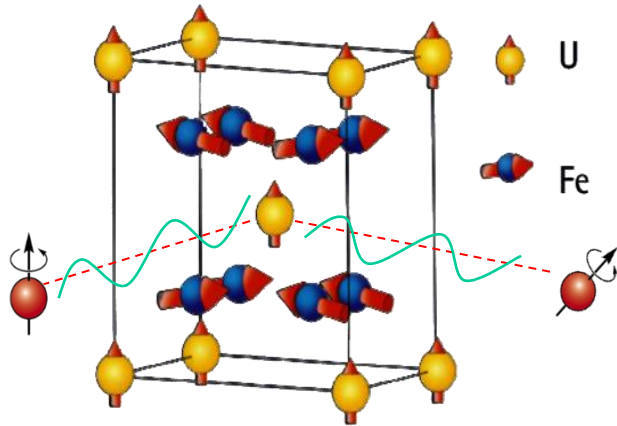
Protein



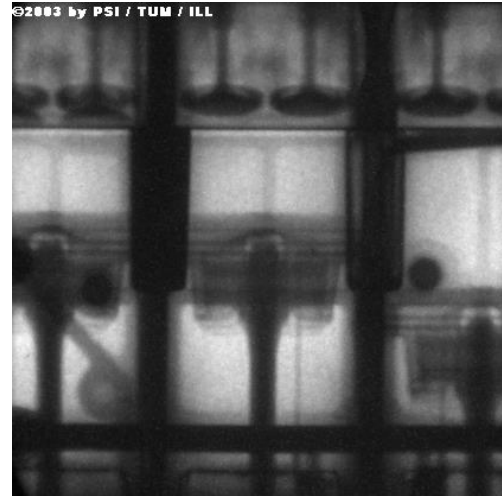
DNA

A protein  
molecule  
moving along  
the DNA chain

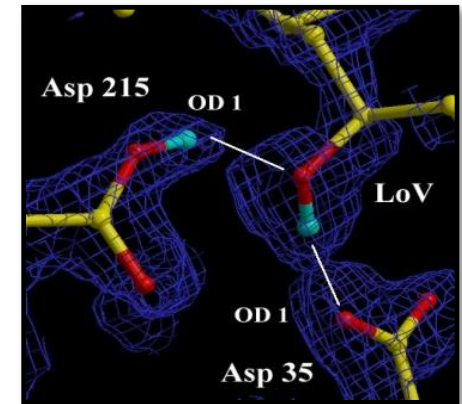
# Complementarity between X-rays & Neutrons



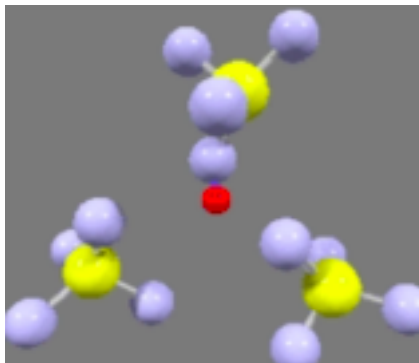
..see magnetic atoms



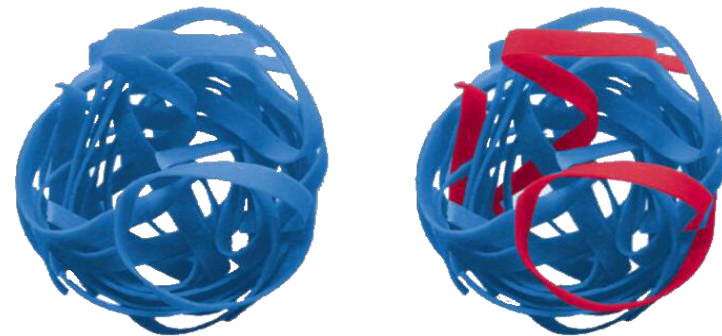
..see inside materials



..see light atoms



..see atoms move



..see isotopes

# Neutron beam and X-Ray for Medical Applications

**iThemba Particle Therapy Centre  
[iTPTC]**

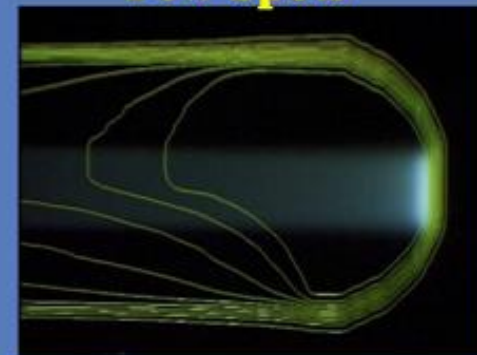


## Spot Scanning Principle

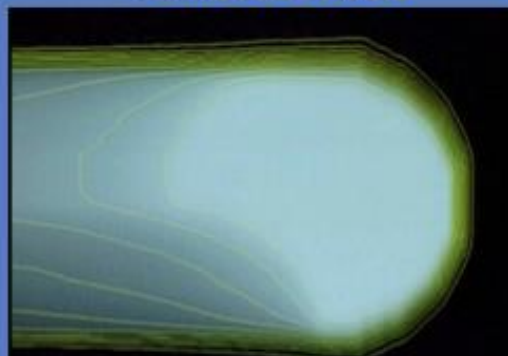
**Single Spot**



**Few Spots**

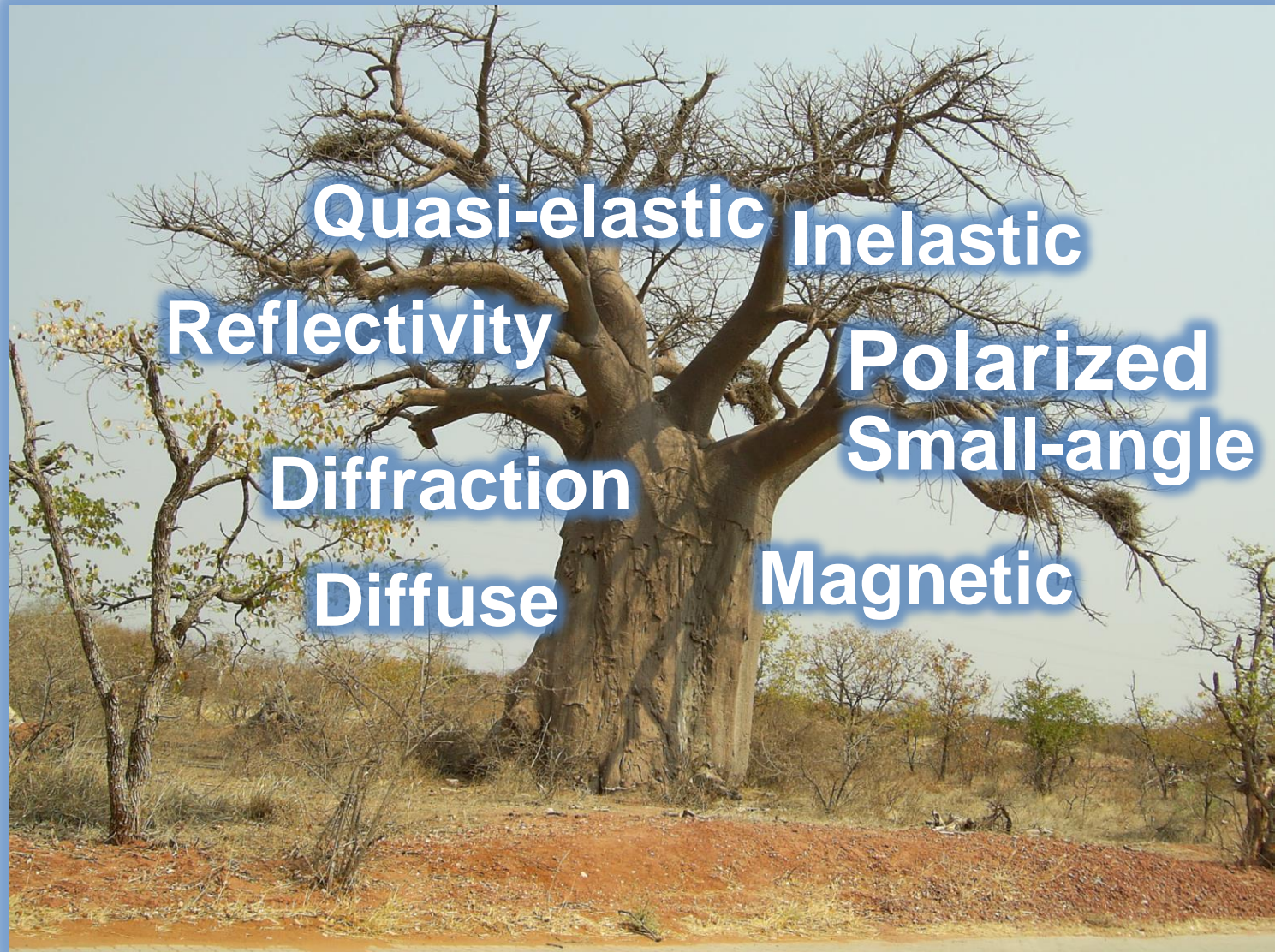


**Final Dose  
Distribution**



# Scattering and Diffraction

# Neutron Scattering Techniques



Quasi-elastic Inelastic

Reflectivity

Polarized

Diffraction

Small-angle

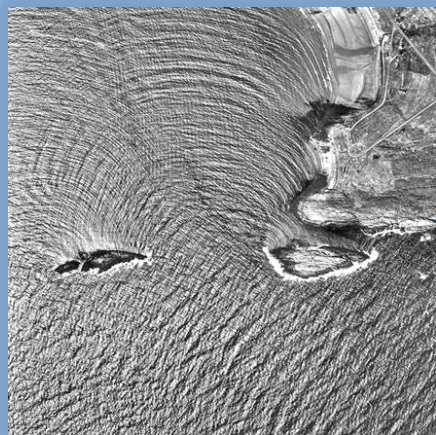
Diffuse

Magnetic

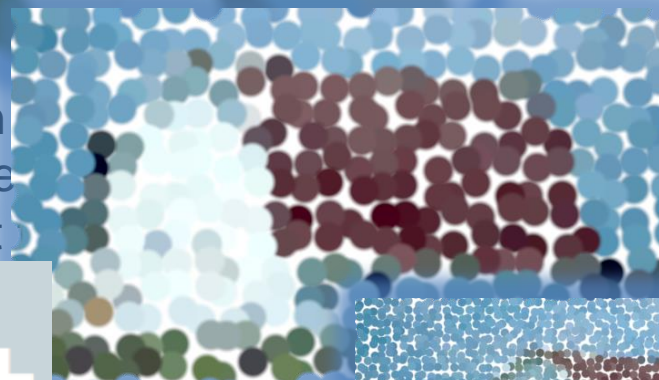
# Scattering – coherence and incoherence

$$S_{coh} = 4p\langle b \rangle^2$$

$$S_{incoh} = 4p\left(\langle b^2 \rangle - \langle b \rangle^2\right)$$



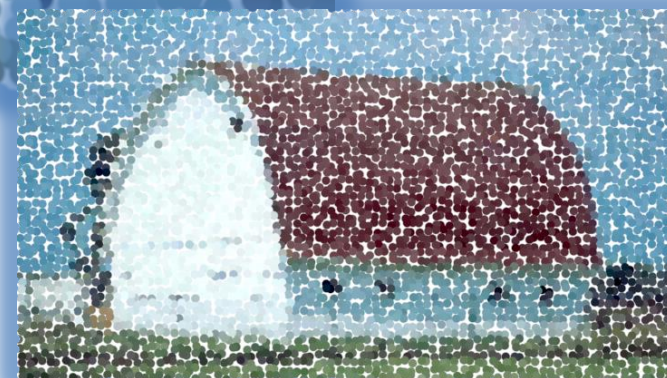
Definition  
One depends  
are, what



scattering units.  
t (where atoms

## WHY NEUTRON SCATTERING IS USEFUL

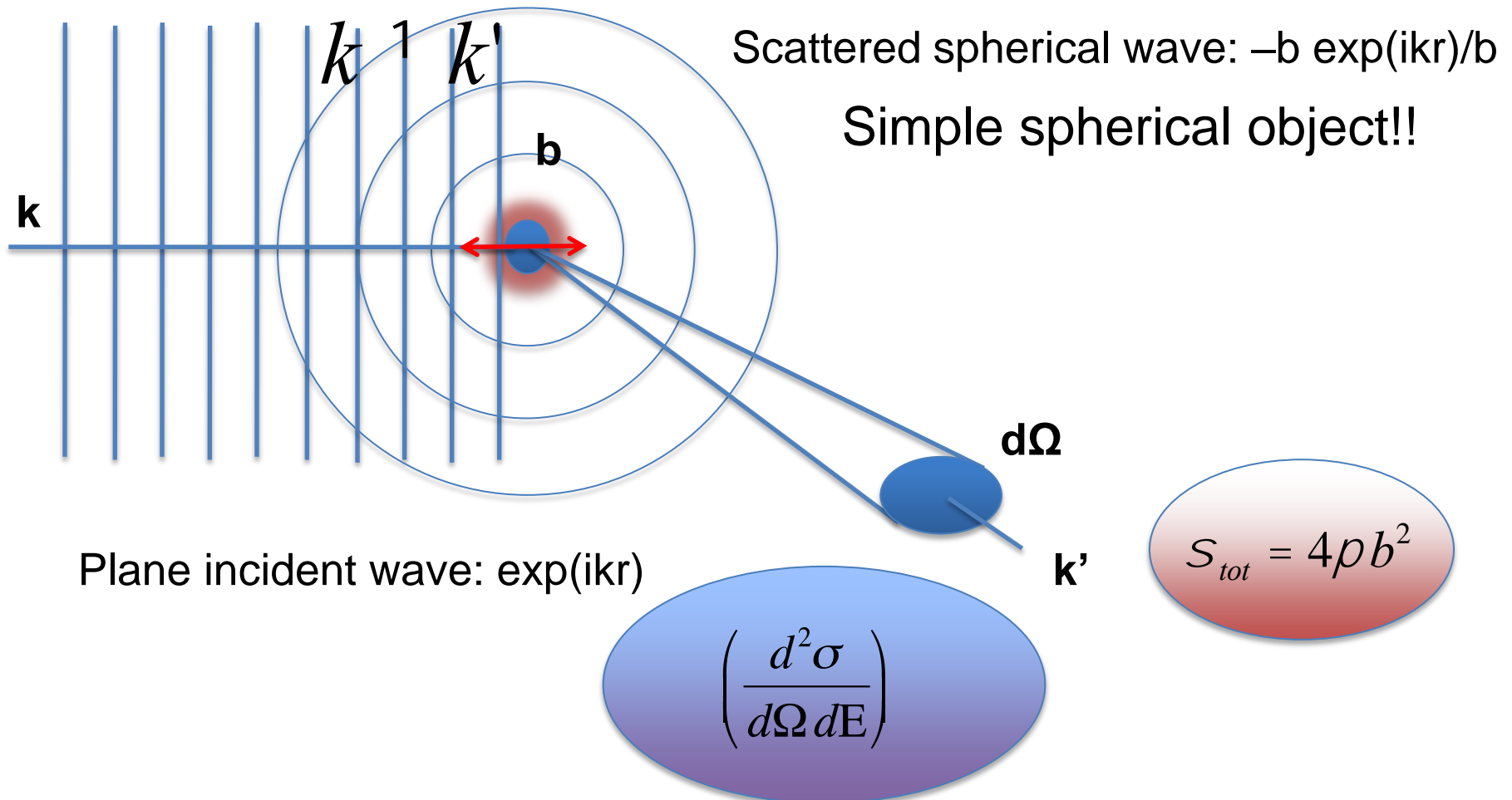
When used as a probe for small samples of materials, neutron beams have the power to reveal what is invisible using other radiations. Neutrons can appear to behave either as particles or as waves or as microscopic magnetic dipoles, and it is these specific properties which enable them to uncover information which is often impossible to access using other techniques.



Isotopes and spin?

# Inelastic Scattering of plane wave

$\sigma_{\text{tot}}$  = number of neutrons scattered in all directions per sec/incident flux

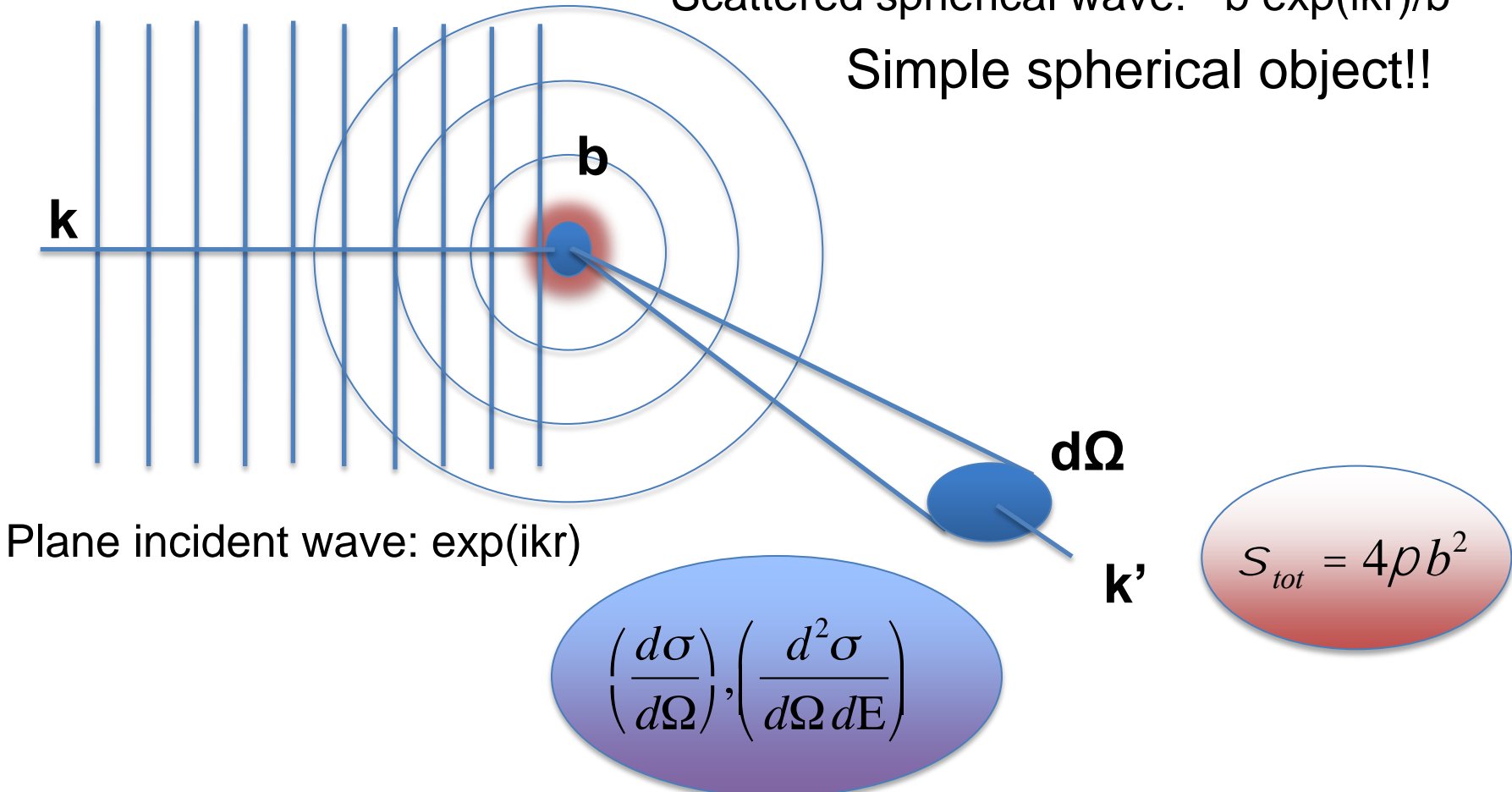




# Diffraction = Coherent Elastic Scattering of plane wave

$\sigma_{tot}$  = number of neutrons scattered in all directions per sec/incident flux

Scattered spherical wave:  $-b \exp(ikr)/b$   
 Simple spherical object!!



# Scattering

- Cross section: 
$$d\sigma = \frac{(\# \text{ particles scattered into solid angle } \Delta\Omega/\text{s})}{(\# \text{ particles incident/sec})(\# \text{ scattering centers/area})}$$

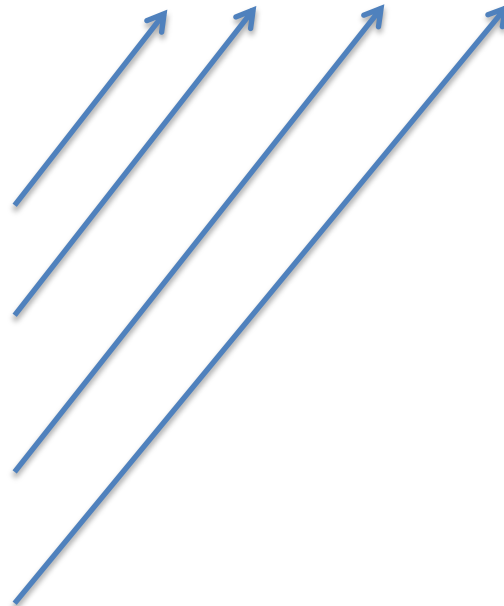
$$\frac{d\sigma(Q)}{d\Omega} = N_p V_p \bar{\rho} P(Q) S(Q)$$

Volume fraction

Contrast

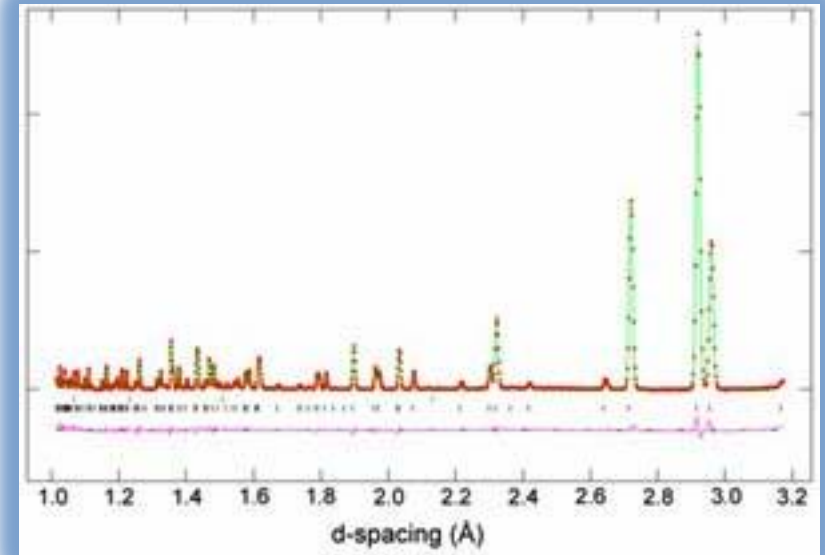
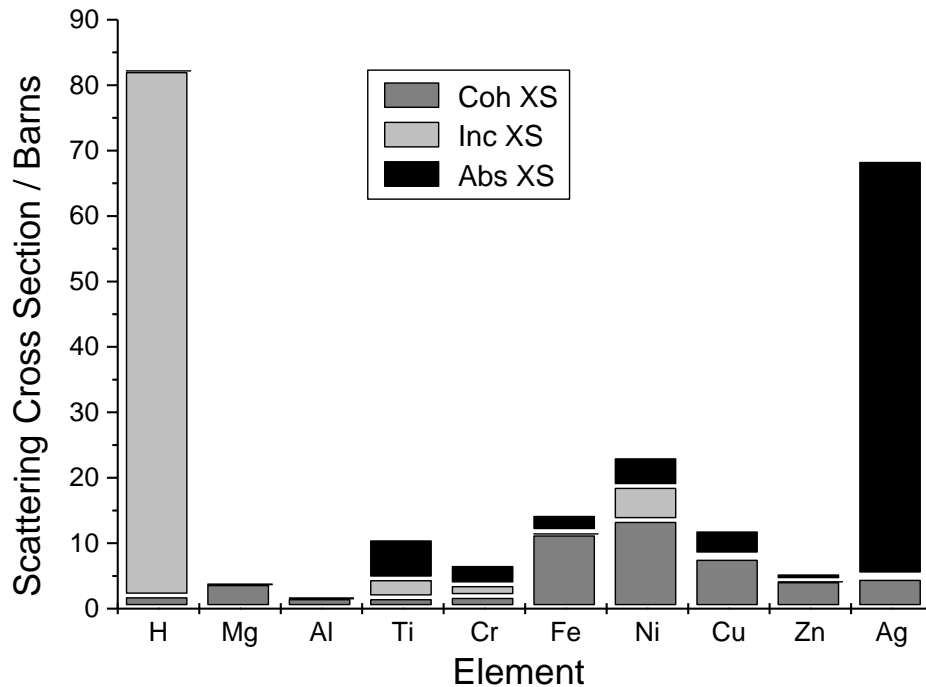
Shape

Interaction



# Scattering

Coherent XS~ Signal  
Incoherent XS~ Background  
Absorption XS~ 1/Intensity



The neutron diffraction pattern of sulfuric acid tetrahydrate at 4.2 K as seen by the 90 degree detectors on HRPD, and fitted with the existing X-ray derived structural model for the deuterated species.

# Neutron cross sections

**NIST Center for Neutron Research**

**NIST**  
National Institute of  
Standards and Technology

Home
ICP
Experiments
UserProposal
Instruments
SiteMap

## Neutron scattering lengths and cross sections

Neutron scattering lengths and cross sections							
Isotope	conc	Coh b	Inc b	Coh xs	Inc xs	Scatt xs	Abs xs
Fe	---	9.45	---	11.22	0.4	11.62	2.56
54Fe	5.8	4.2	0	2.2			
56Fe	91.7	9.94	0	12.4			
57Fe	2.2	2.3	---	0.66			
58Fe	0.3	15.(7.)	0	28			

				He
C	N	O	F	Ne
Si	P	S	Cl	Ar

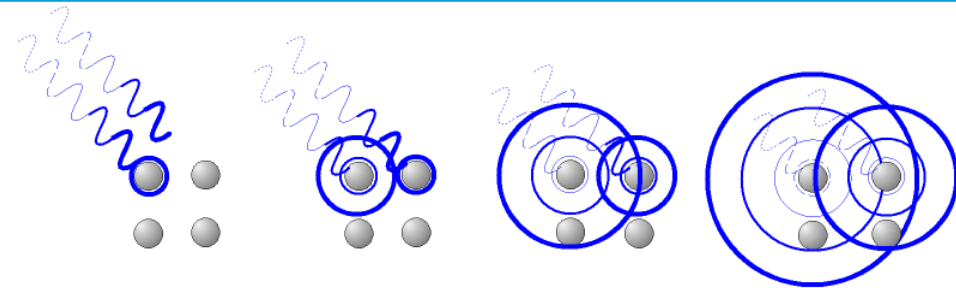
Neutron scattering lengths and cross sections							
Isotope	conc	Coh b	Inc b	Coh xs	Inc xs	Scatt xs	Abs xs
Gd	---	6.5-13.82i	---	29.3	151.(2.)	180.(2.)	49700.(125.)
152Gd	0.2	10.(3.)	0	13.(8.)	0	13.(8.)	735.(20.)
154Gd	2.1	10.(3.)	0	13.(8.)	0	13.(8.)	85.(12.)
155Gd	14.8	6.0-17.0i	(+/-)5.(5.)-13.16i	40.8	25.(6.)	66.(6.)	61100.(400.)
156Gd	20.6	6.3	0	5	0	5	1.5(1.2)
157Gd	15.7	-1.14-71.9i	(+/-)5.(5.)-55.8i	650.(4.)	394.(7.)	1044.(8.)	259000.(700.)
158Gd	24.8	9.(2.)	0	10.(5.)	0	10.(5.)	2.2
160Gd	21.8	9.15	0	10.52	0	10.52	0.77

**NOTE:** The above are only thermal neutron dependent cross sections please go to

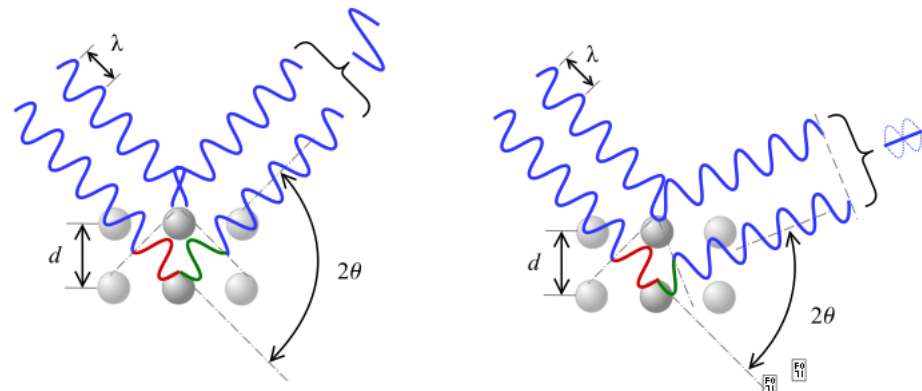
Select the element, and you will get a list of the feature section of neutron scattering lengths and cross sections. *Journal of Applied Physics*, 3, 1992, pp. 29-37.

# Diffraction - Bragg

Diffraction of X-rays or neutrons by polycrystalline samples is one of the most important, powerful and widely used analytical techniques available to materials scientists. For most crystalline substances of technological importance, the bulk properties of a powder or a polycrystalline solid, averaged throughout the sample, are required; in general a single-crystal data, even if they can be obtained, are usually of little interest except for determination of the crystal structure or for studying some other fundamental physical property. By *J Ian Langford and Daniel Louer*



X-rays interact with the atoms in a crystal.

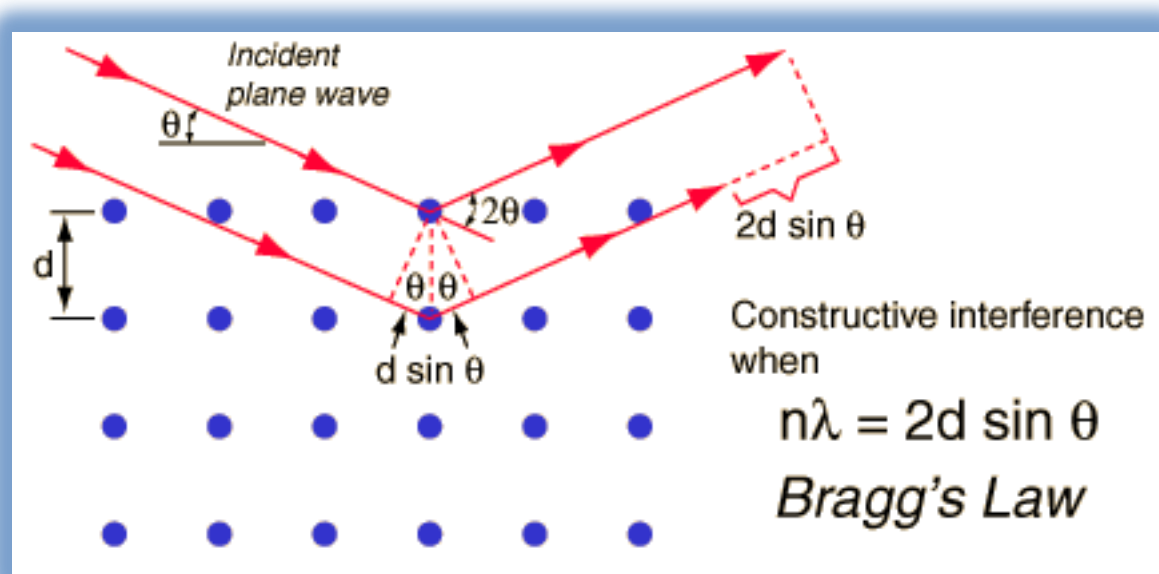


According to the  $2\theta$  deviation, the phase shift causes constructive (left figure) or destructive (right figure) interferences.

# Diffraction - Bragg

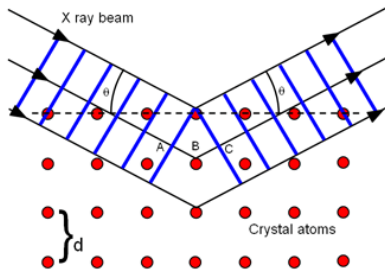
- Bragg / Laue scattering
- Coherent elastic scattering

Two beams with identical wavelength and phase approach a crystalline solid and are scattered off two different atoms within it. The lower beam traverses an extra length of  $2d \sin \theta$ . **Constructive interference** occurs when this length is equal to an integer multiple of the wavelength of the radiation.



# Diffraction - Bragg

## Using the grains as internal strain gauges



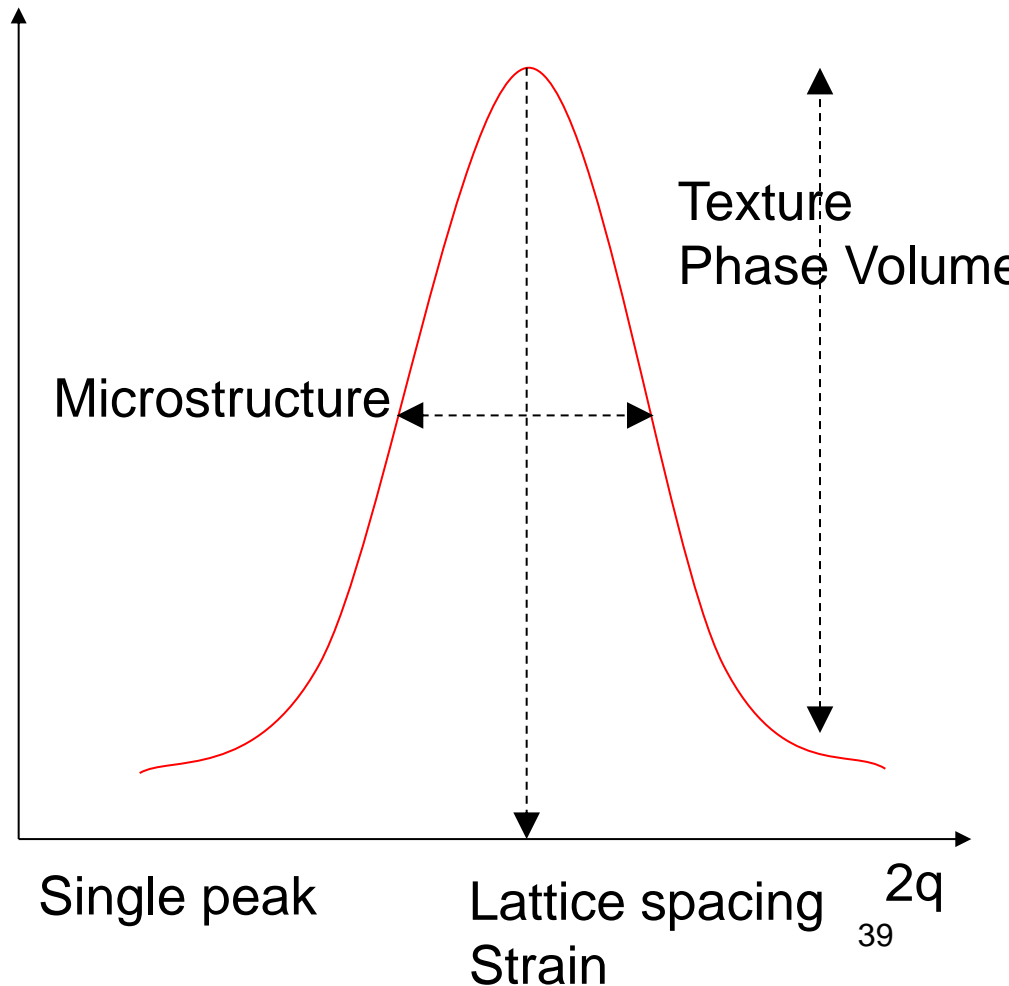
$$\lambda = 2d \sin\theta$$

$$\varepsilon = -\cot(\theta) (\theta - \theta_0)$$

Two ways to measure  $d$ :

- keep  $\lambda$  fixed and measure  $\theta$
- constant wavelength
- keep  $\theta$  fixed and measure  $\lambda$
- time-of-flight

Braggs law gives information about position of a diffraction peak from a type of lattice planes. Rietveld approach calculated height, position and width of diffraction peaks from first principles, i.e. Temperature, composition, vacancies, etc.

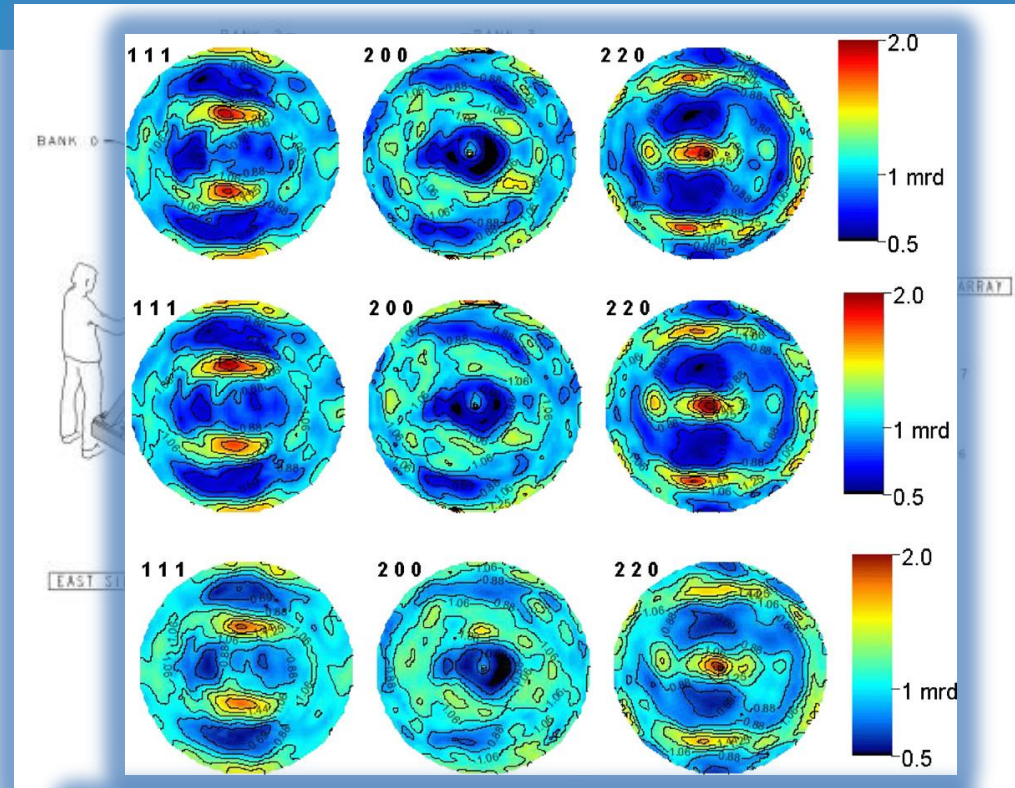


# Diffraction: Texture

- ISIS: GEM instrument
- Near  $4\pi$  coverage

Example:

- Cold rolled copper, simulating manufacturing process for archeometry
- 2mm thick disks
- $20 \times 20 \text{ mm}^2$  beam
- 2 min counting times

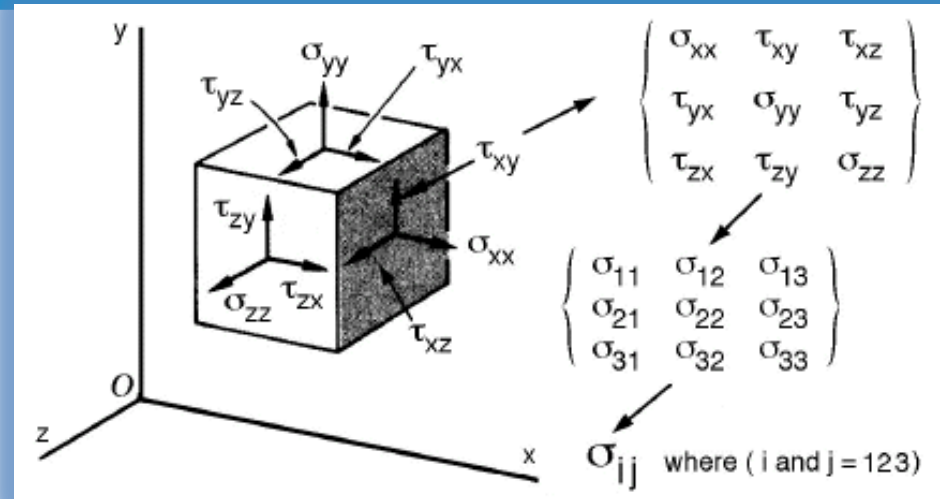
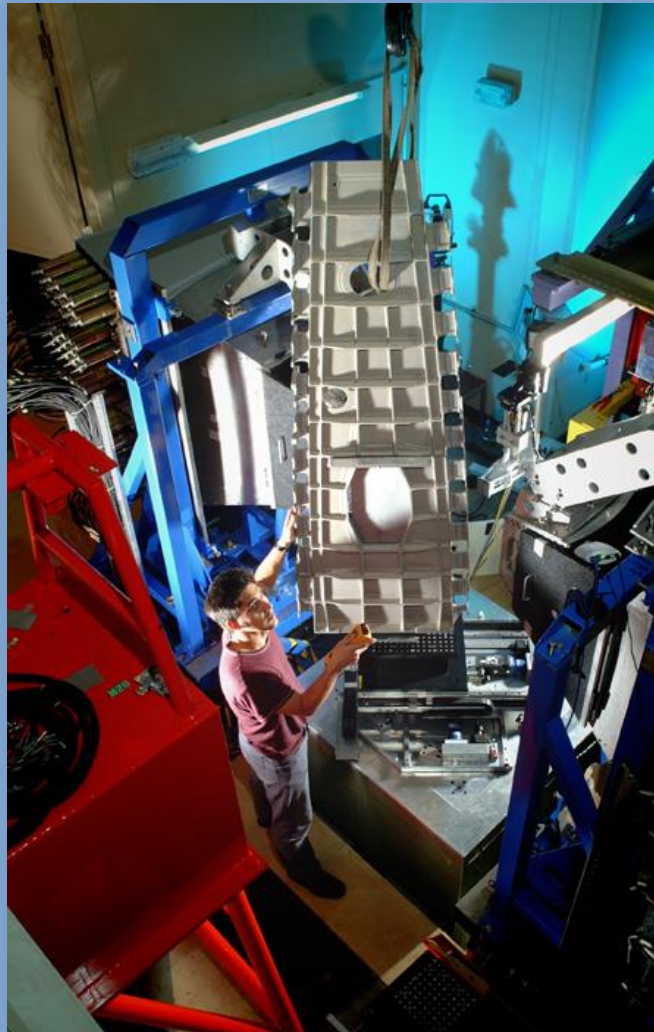


$$\frac{P_{hkl}}{P_0} = \frac{\lambda^3 l}{4\pi r} \frac{h\rho'}{\rho} \frac{e^{-\mu_h \sec\theta}}{\sin^2 2\theta} j_{hkl} N^2 F_{hkl}^2, \quad (15)$$

Intensity of diffraction peak



# Diffraction: Stress and Strain



## Applications:

- Residual stresses
- Fatigue/Structural Integrity
- Welds
- Alloy development
- Microstructure/Texture
- Phase transformation
- In-situ experiments

# Example - Diffraction

PHYSICAL REVIEW

VOLUME 73, NUMBER 8

APRIL 15, 1948

## The Diffraction of Neutrons by Crystalline Powders

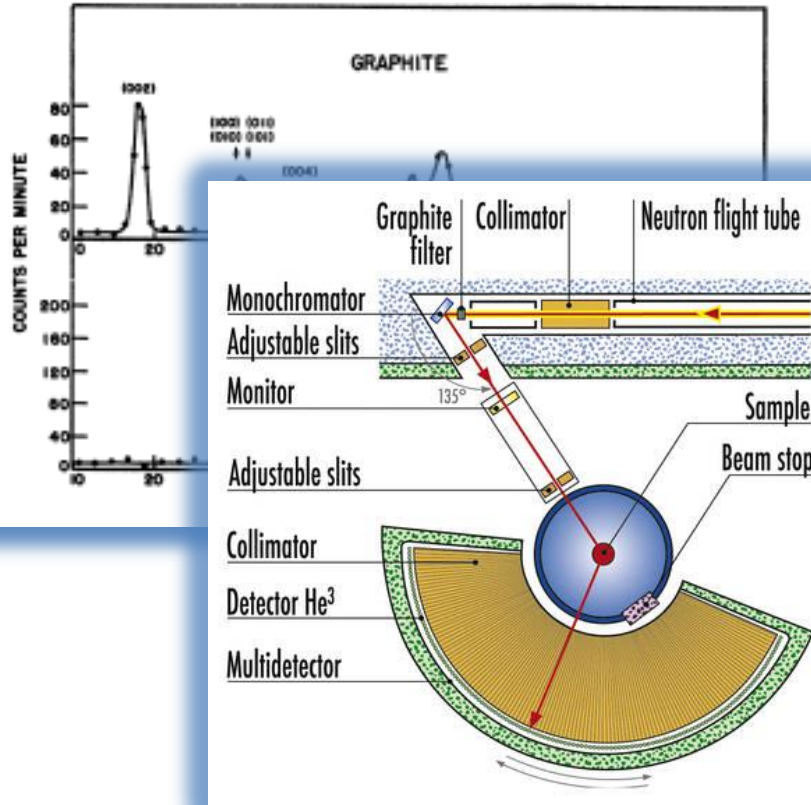


Fig. 3. Powder diffraction patterns for diamond and graphite. The major part of the diffuse scattering in these patterns arises from multiple scattering in the samples.

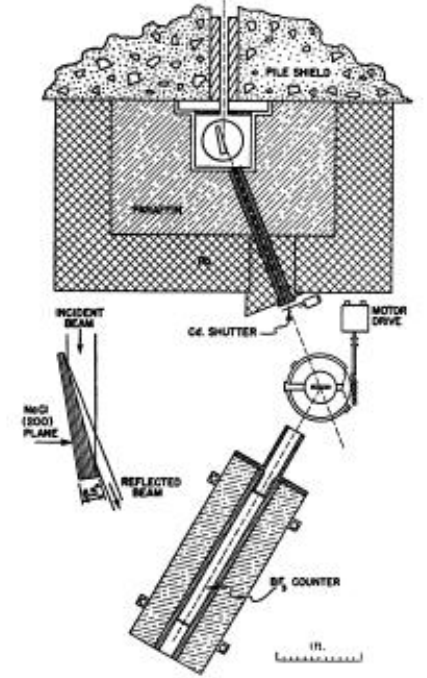


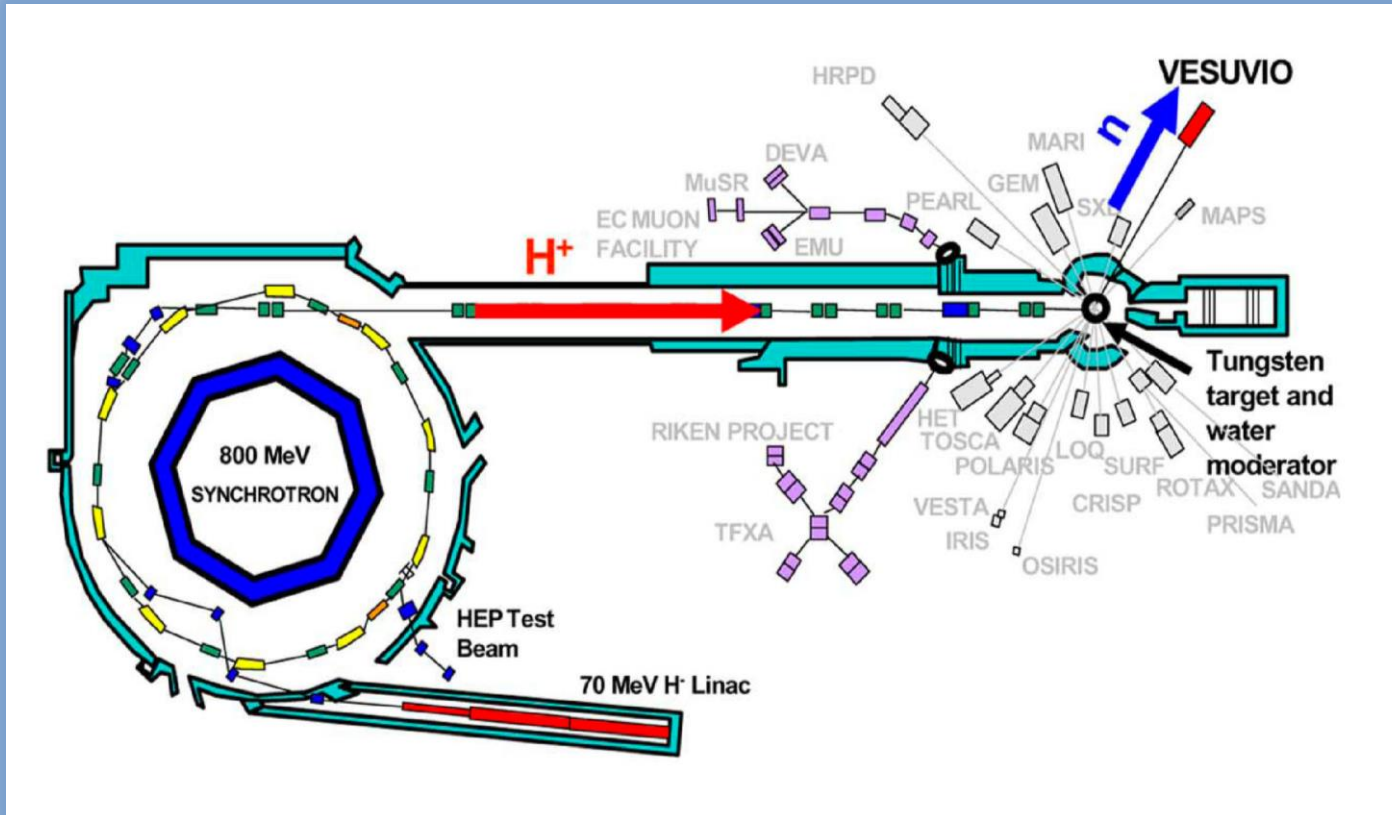
Fig. 1. Arrangement of apparatus, showing the monochromating crystal (detailed in left center) collimating slits, shielding, second spectrometer with location of powder specimen and counter.

$$\frac{P_{hkl}}{P_0} = \frac{\lambda^3 l}{4\pi r} \frac{h\rho'}{\rho} \frac{e^{-\mu h \sec\theta}}{\sin^2 2\theta} j_{hkl} N^2 F_{hkl}^2, \quad (15)$$

Intensity of diffraction peak

D2B at the ILL, Grenoble, 50 years later

# Time of Flight



$$\lambda = \frac{h}{p} = \frac{h}{mv} = \frac{ht}{mL}$$

De Broglie  
+

$$\lambda = 2d \sin\theta$$

Bragg  
=

Time-of-Flight:

$$d = \frac{h}{2mL \sin\theta} t$$

Fast, short-wavelength neutrons arrive earlier at detector!

Diffraction in ToF mode at pulsed source. Flight time is prope to wavelength

# Ice

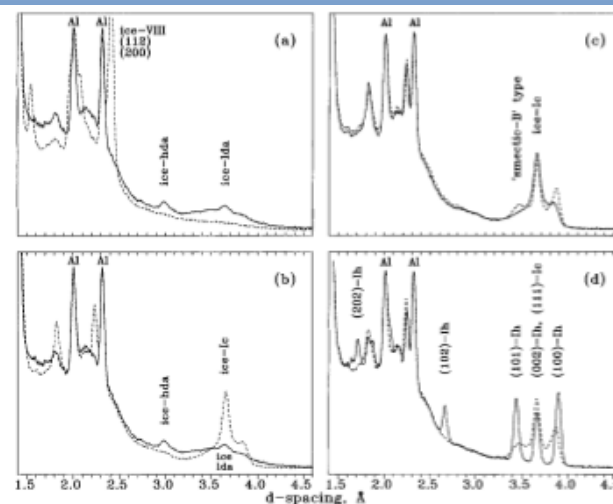
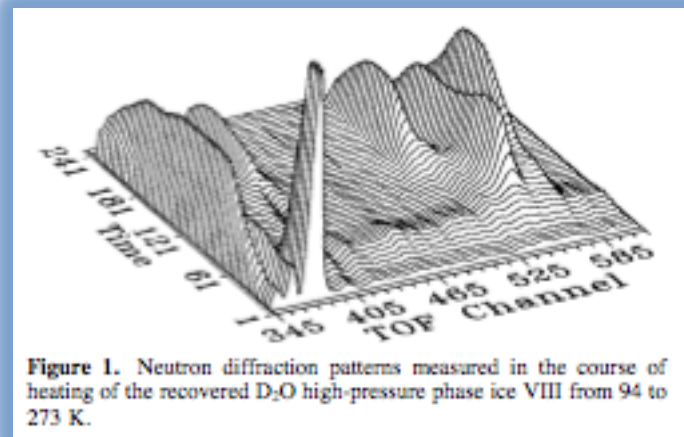
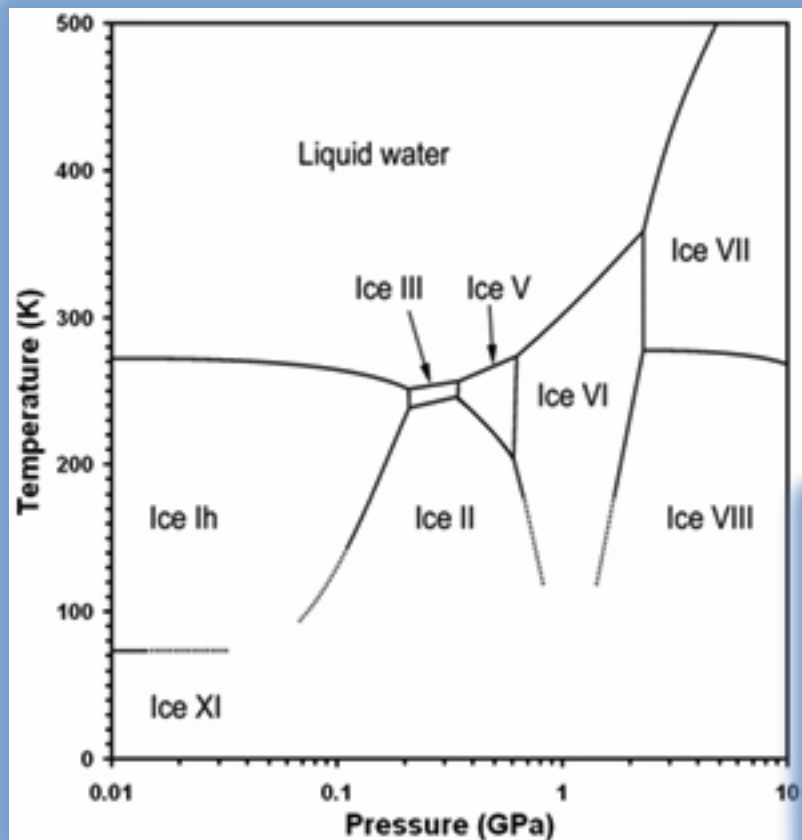
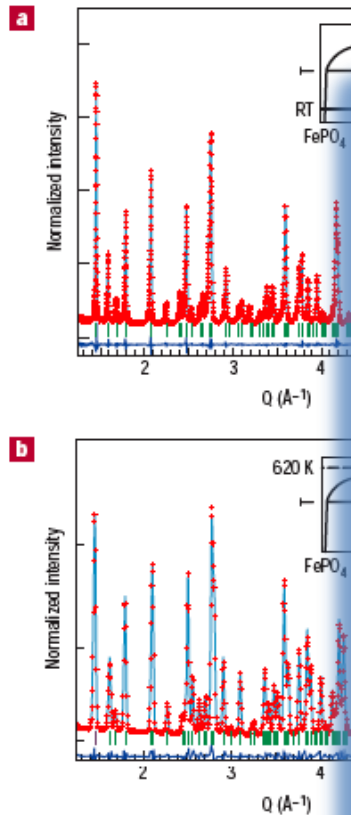
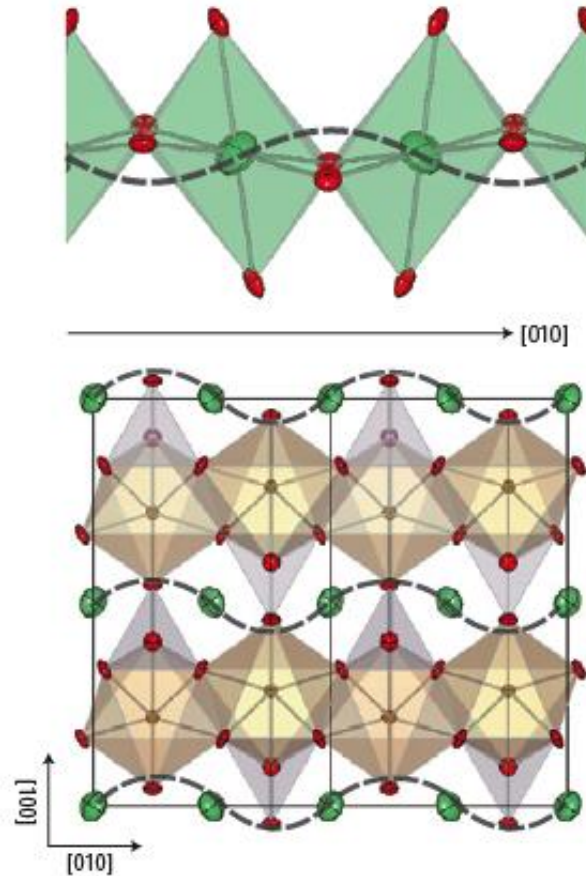


Figure 2. Neutron diffraction patterns of D<sub>2</sub>O ices before and after phase transitions: (a) from ice VIII (dashed line) to a mixture of hda + lda ices (solid line) at 125 K; (b) from the mixture of hda + lda ices (solid line) to ice Ic (dashed line) at 160 K; (c) from ice Ic (solid line) to the hexagonal phase, "smectic B" type structure (dashed line), at 190 K; (d) from "smectic B" type ice (dashed line) to ice Ih (solid line) at 230 K. The diffraction patterns also contain peaks from the aluminum sample can and cryostat.

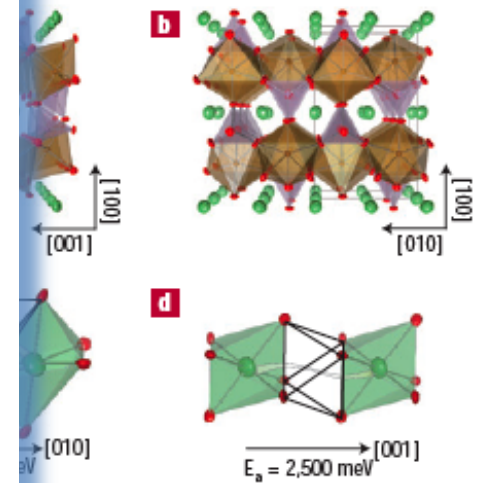
# Batteries



**Figure 2** Neutron diffraction patterns measured at the  $\text{FePO}_4\text{-LiFePO}_4$  binary phase diagram. **a,b**, Rietveld fit and time-of-flight neutron diffraction profile measured for  $\text{LiFePO}_4$  at room temperature (a) and the angle-dispersive neutron diffraction profile measured at 620 K (b). Two different neutron diffractometers were used for each measurement as explained in the Supplementary Information. The data points are plotted using the common scale  $Q = 4\pi \sin \theta / \lambda$  in the  $Q$  range for VEGA and HERMES for comparison. Specific composition and temperature are given in the inset phase diagram. Observed intensity  $I_{\text{obs}}$  are represented by red plus signs and the green curve  $I_{\text{fit}}$  represents the Rietveld fit. The blue curve at the bottom represents the residual difference  $I_{\text{res}} = I_{\text{obs}} - I_{\text{fit}}$ . The refined parameters are summarized in Supplementary Information. An impurity phase was identified, and the crystal structure was determined with the space group  $Pnma$ .



**Figure 3** Anisotropic harmonic lithium vibration in  $\text{LiFePO}_4$ , shown as green thermal ellipsoids and the expected diffusion path. The ellipsoids were refined with 95% probability by Rietveld analysis for room-temperature neutron diffraction data. Expected curved one-dimensional continuous chains of lithium motion are drawn as dashed lines to show how the motions of Li atoms evolve from vibrations to diffusion.



**Figure 4** Lithium migration pathways in  $\text{LiFePO}_4$  and possible lithium pathways. **a,b**, The structure is projected along the  $[010]$  (a) and  $[001]$  (b) directions. The migration pathways are parallel to these directions. The structures were refined with parameters obtained through this work and are given in the Supplementary Information, Table S1. The structure can be described as a layered orthorhombic close-packed oxygen sub-array, in which Li, Fe and P occupy tetrahedral sites to form (1) corner-sharing  $\text{FeO}_6$  octahedra that form a distorted two-dimensional square lattice perpendicular to the  $c$  axis, and (2)  $\text{LiO}_6$  octahedra aligned in parallel chains along the  $c$  axis, with the  $\text{LiO}_6$  groups connecting neighbouring planes or arrays. The thermal ellipsoids indicate Li, Fe, P and O atoms, respectively. The migration pathways: **c**, along the  $[010]$  direction through the tetrahedral sites; and **d**, along the  $[001]$  direction through the octahedral sites. One-dimensional diffusion along the  $[010]$  direction was studied by the computational method<sup>15,16</sup>.

ing stresses...

# Fuel Cells

Electric Circuit  
(40% - 60% efficiency)



Available online at [www.sciencedirect.com](http://www.sciencedirect.com)



Solid State Ionics 177 (2006) 2357–2362

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IONICS**

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Proton co



Table 1

Summary of results obtained from Rietveld analysis of neutron powder diffraction data for  $\text{BaZr}_{1-x}\text{In}_x\text{O}_{3-0.5x}$ , collected at 10K on the NPD diffractometer

	$x=0.0$ (as-prepared)	$x=0.0$ (deuterated)	$x=0.25$ (as-prepared)	$x=0.25$ (deuterated)	$x=0.50$ (as-prepared)	$x=0.50$ (deuterated)
Space group	Pm-3m	Pm-3m	Pm-3m	Pm-3m	Pm-3m	Pm-3m
$a$ (Å)	4.1879(1)	4.1880(1)	4.1916(1)	4.1983(1)	4.1942(2)	4.2260(4)
Thermal parameters, $B_{\text{iso}}$ (Å <sup>2</sup> )						
Ba on 1(b) 1/2,1/2,1/2	0.09(3)	0.08(2)	0.10(3)	0.21(4)	0.48(6)	0.97(10)
Zr on 1(a) 0,0,0	0.14(2)	0.11(2)	0.19(3)	0.23(3)	0.43(5)	0.37(9)
In on 1(a) 0,0,0	–	–	0.19(3)	0.23(3)	0.43(5)	0.37(9)
O on 3(d) 1/2,0,0	0.24(2)	0.26(2)	0.63(2)	0.56(2)	1.16(4)	1.17(5)
Oxygen site occupancy	2.98(1)	2.99(1)	2.82(1)	2.90(2)	2.68(2)	2.98(3)
$\chi^2$	1.96	1.99	1.85	1.64	1.72	3.16
Weighted $R_{\text{wp}}$	4.17%	5.16%	4.7%	4.74%	5.28%	6.62%
Bragg $R_{\text{B}}$	3.97%	3.86%	4.63%	4.40%	8.16%	6.62%
Fitted parameters	16	16	16	16	16	16

$2\theta$  (°)

Fig. 2. Low temperature (10K) neutron powder diffraction patterns of as-prepared and respective deuterated (marked with D)  $\text{BaZr}_{1-x}\text{In}_x\text{O}_{3-\delta}$  ( $x=0.00, 0.25$  and  $0.50$ ) samples.

Ap  
str  
(Water transport) on macro scale.

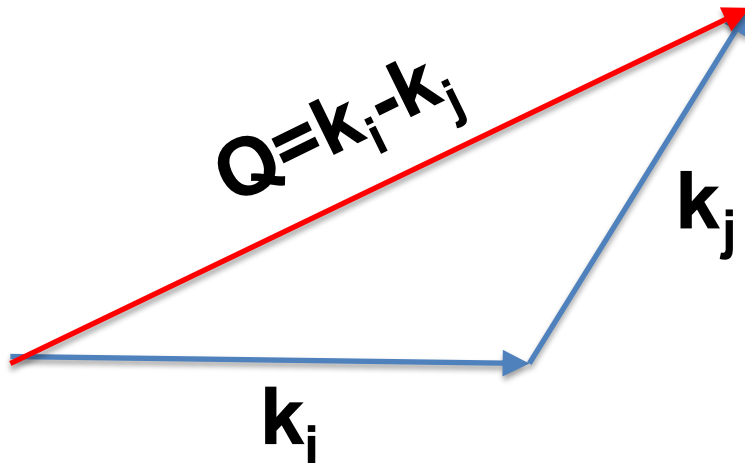
The conductivity of the  
number of protons in  
proton conduction is  
© 2006 Elsevier B.V.

# Small Angle Scattering

- Scattering Vector

$$\sin \theta = Q\lambda/4\pi = \lambda/2d$$

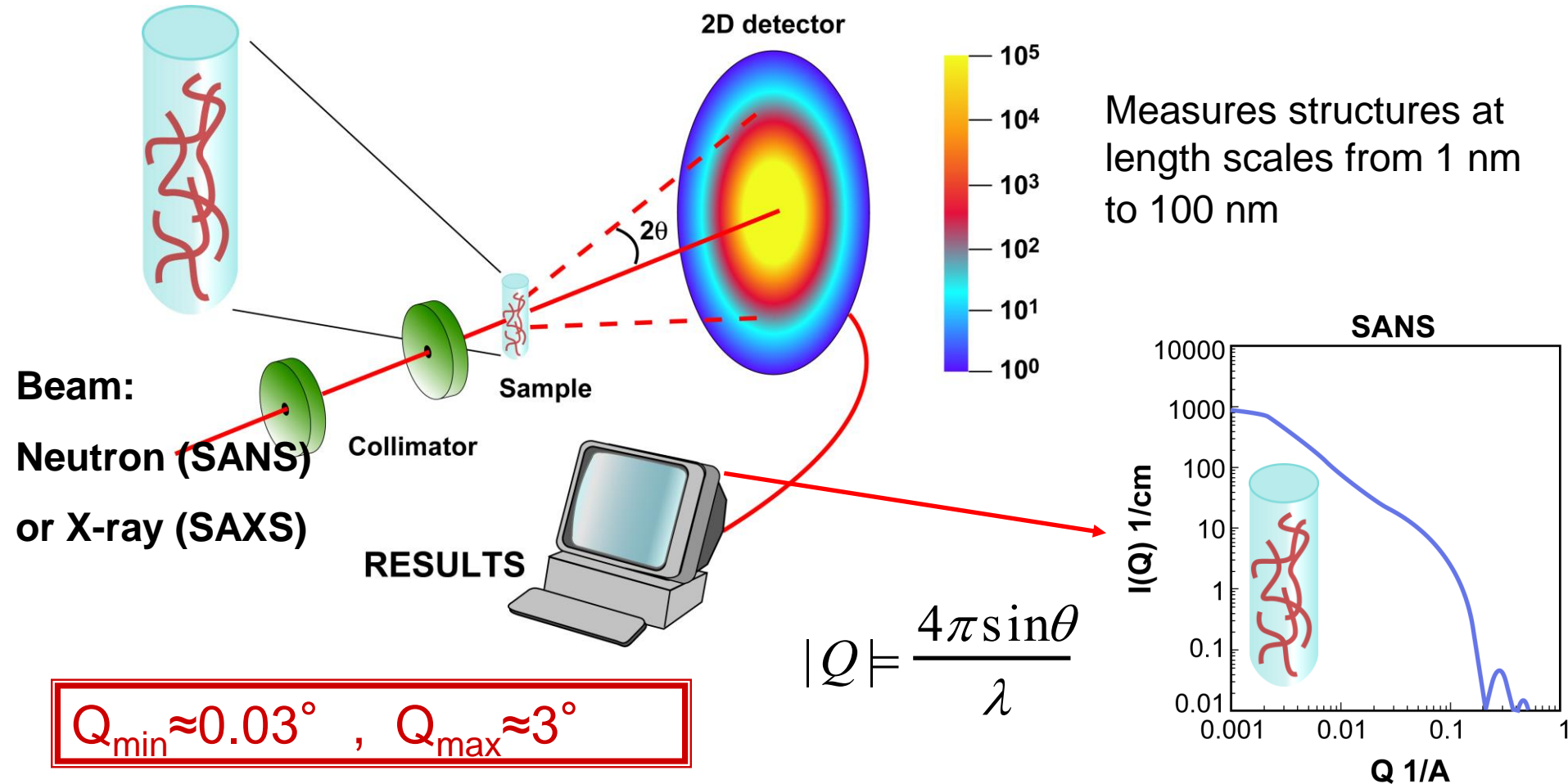
$$\rightarrow d = 2\pi/Q$$



Small angle  $\rightarrow$  small  $Q \rightarrow$  large distances  
 long  $\lambda$  (cold neutrons) = 6 Å,  $d = 200$  Å,  
 $2\theta = 1.7^\circ$

$\rightarrow$  The larger the object, the smaller the angle

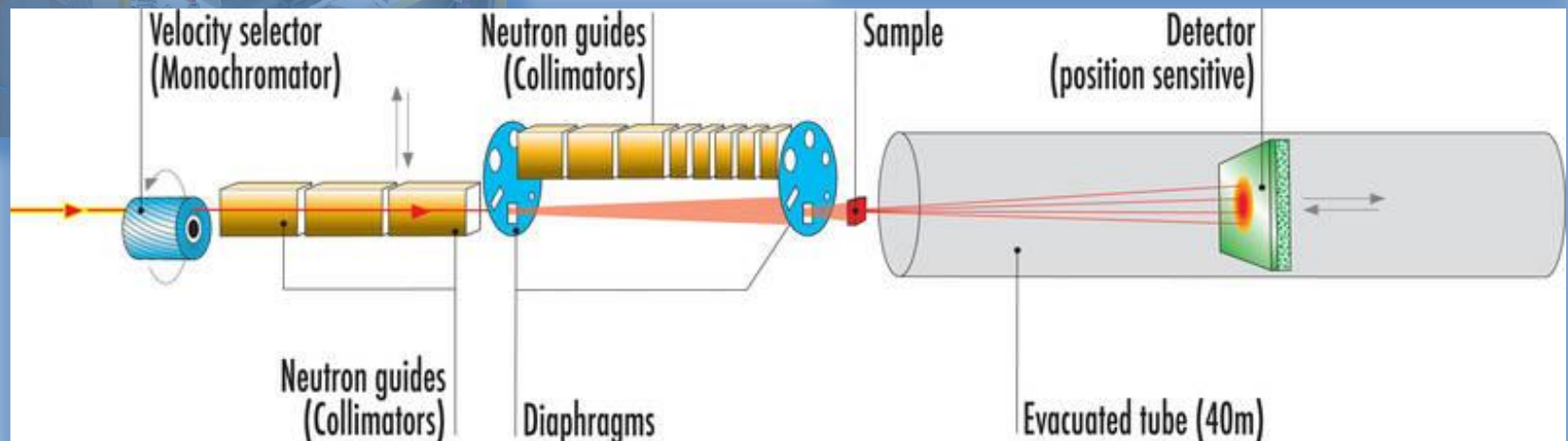
# SANS: Experimental Setup



In a standard crystallography experiment, theta\_max is typically 45 degrees



# Small Angle Instruments



# Scattering

- Cross section: 
$$d\sigma = \frac{(\# \text{ particles scattered into solid angle } \Delta\Omega/\text{s})}{(\# \text{ particles incident/sec})(\# \text{ scattering centers/area})}$$

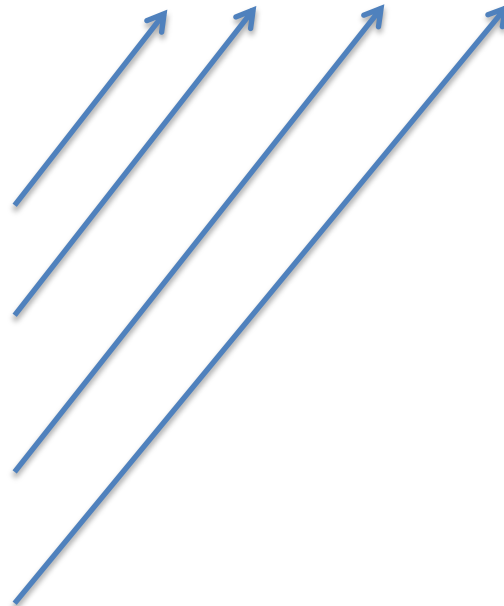
$$\frac{d\sigma(Q)}{d\Omega} = N_p V_p \bar{\rho} P(Q) S(Q)$$

Volume fraction

Contrast

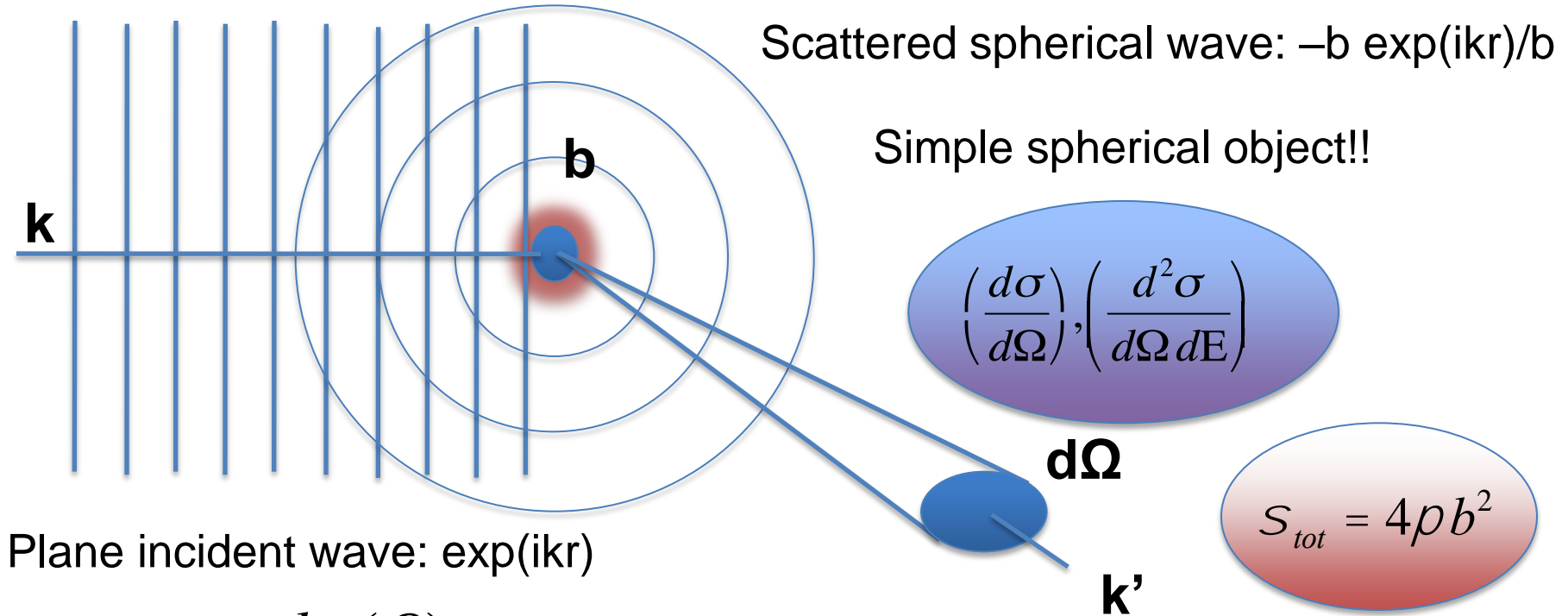
Shape

Interaction



# SANS: Scattering of plane wave

$\sigma_{\text{tot}}$  = number of neutrons scattered in all directions per sec/incident flux

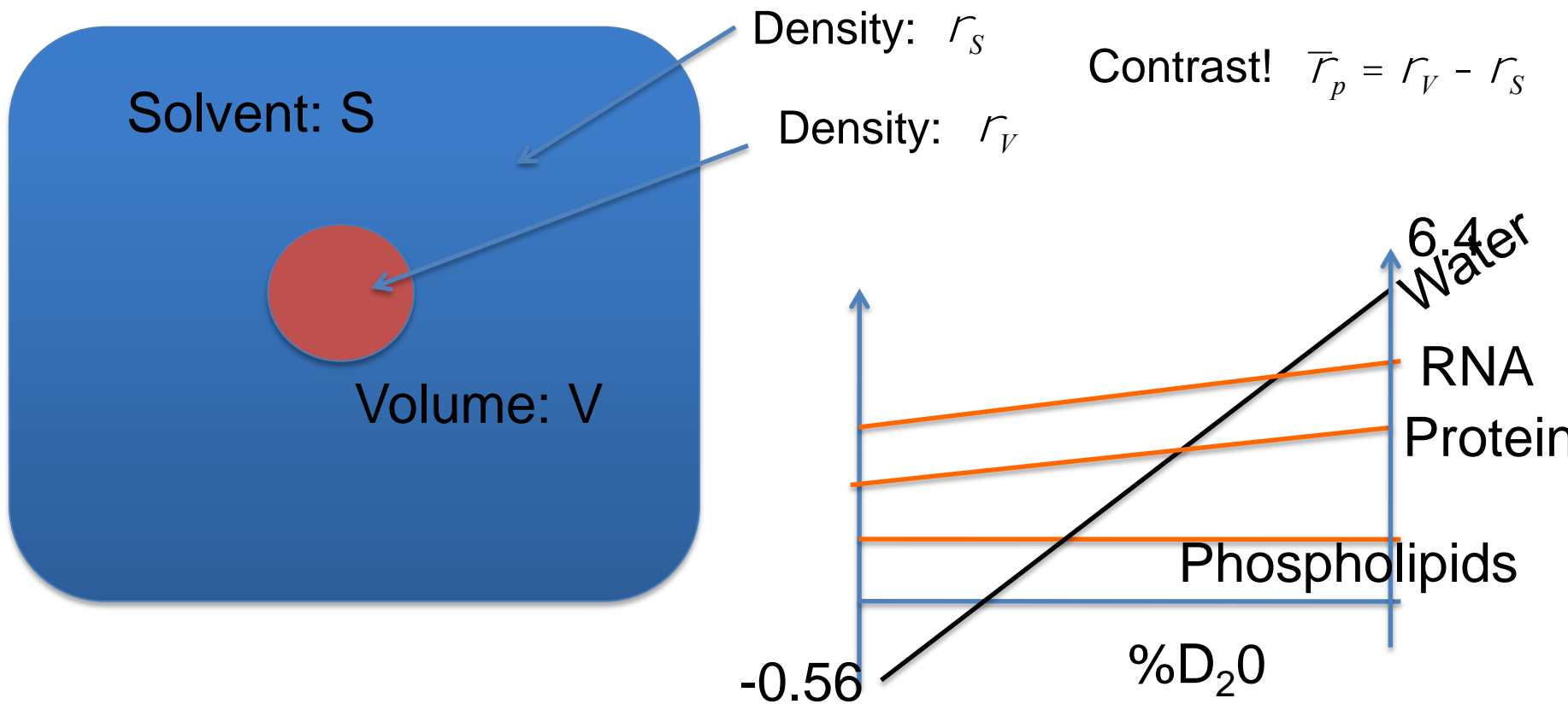


$$\frac{d\sigma(Q)}{d\Omega} = N_p V_p \bar{\rho} P(Q) S(Q)$$

Volume fraction

Contrast

# SANS: Particles: contrast!

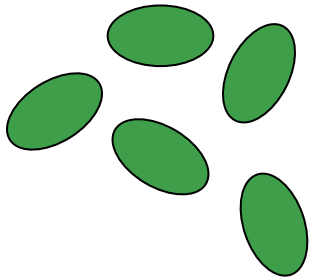


# SANS: quick example

## Protein based drugs:

- Typically proteins in solution to be injected
- Long shelf life (up to 2 years)
- Control of release profile is desirable

**Fast action:** Monomeric and dimeric insulin

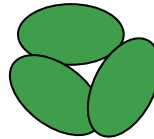


10 nm

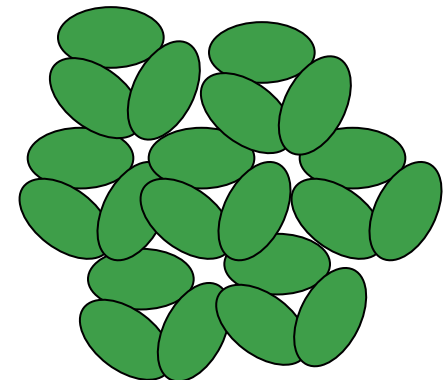


Knowledge and control of solution properties of the proteins are crucial

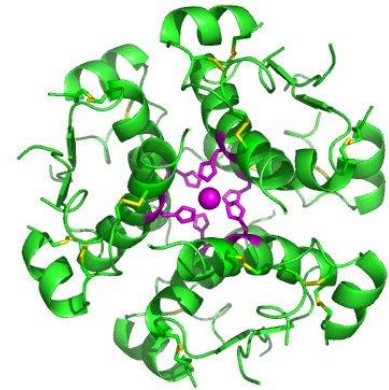
**Medium action:** Hexameric insulin



**Slow action:** Large complexes of hexameric insulin

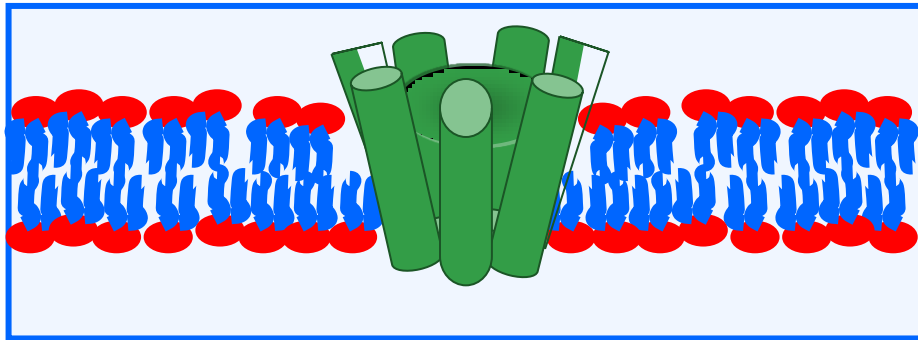


**Insulin Hexamer**

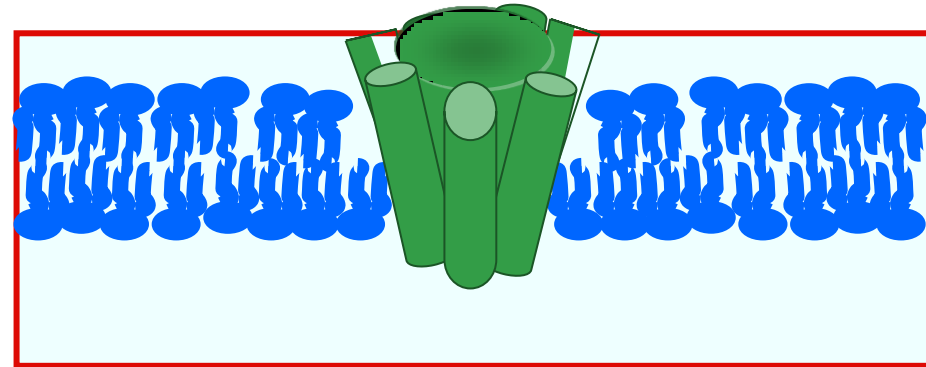


# SANS versus SAXS

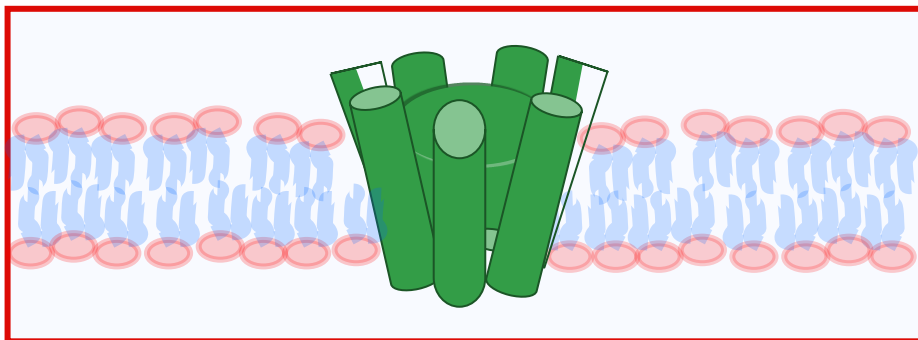
## SAXS contrast



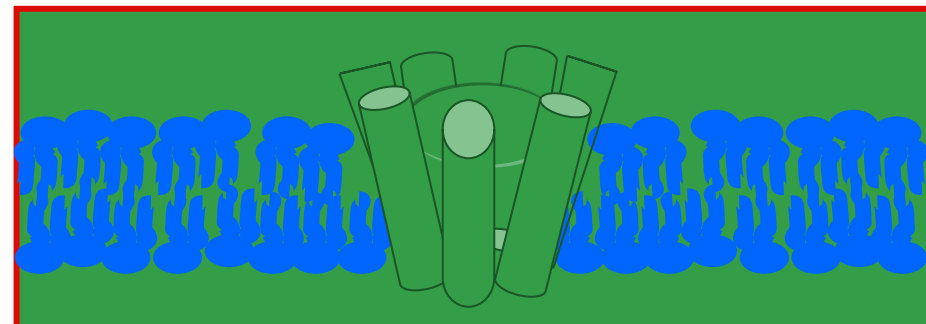
## SANS contrast 1



## SANS contrast 2



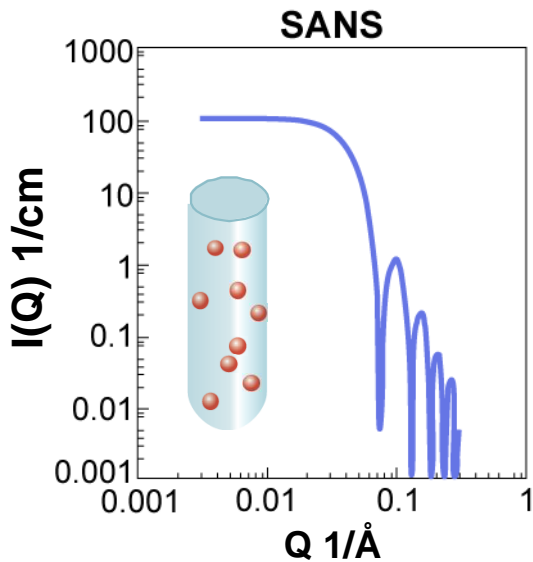
## SANS contrast 3



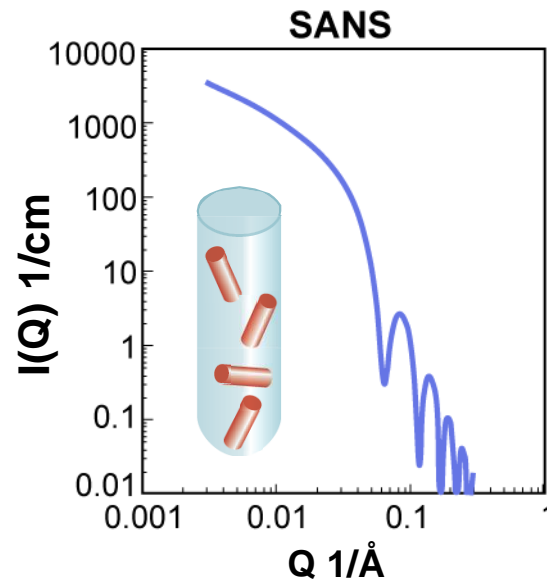
SANS gives the possibility of **not** seeing everything at the same time....

# SANS: Different Shapes

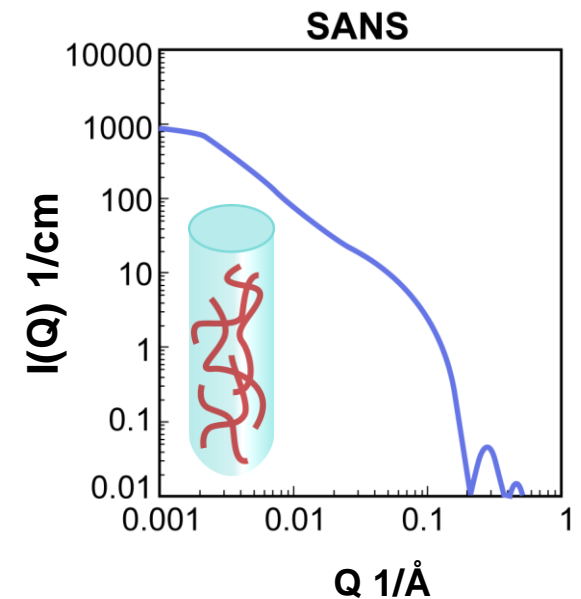
- Spheres:  
 $R = 60 \text{ \AA}$



- Rods:  $R = 60 \text{ \AA}$   
 $L = 1200 \text{ \AA}$

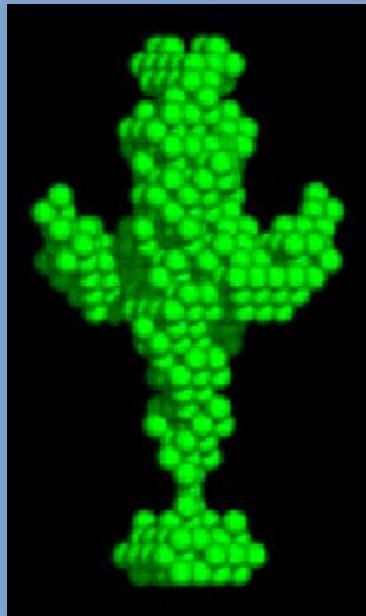


- Worms:  $R = 18 \text{ \AA}$   
 $L = 5000 \text{ \AA}$ ,  
Kuhn Length =  $300 \text{ \AA}$

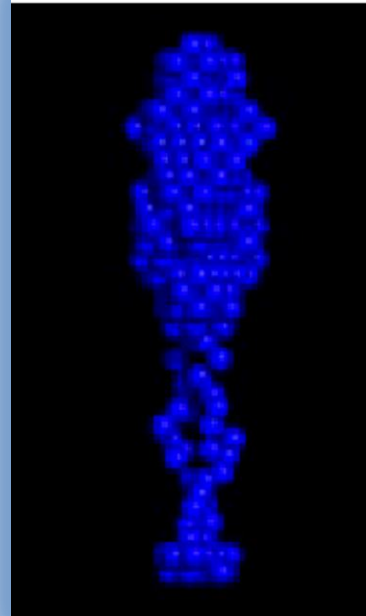


# SANS: Selective Deuteration

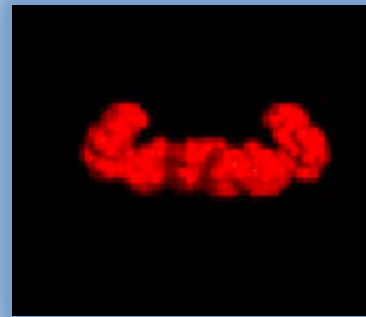
sensitivity and selectivity  
isotopic substitution/contrast variation



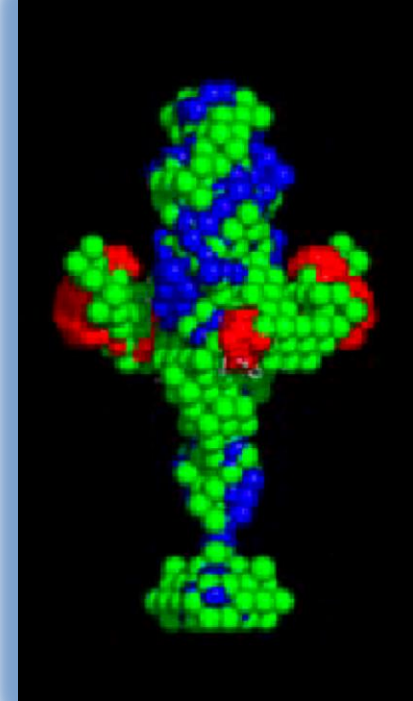
+



+



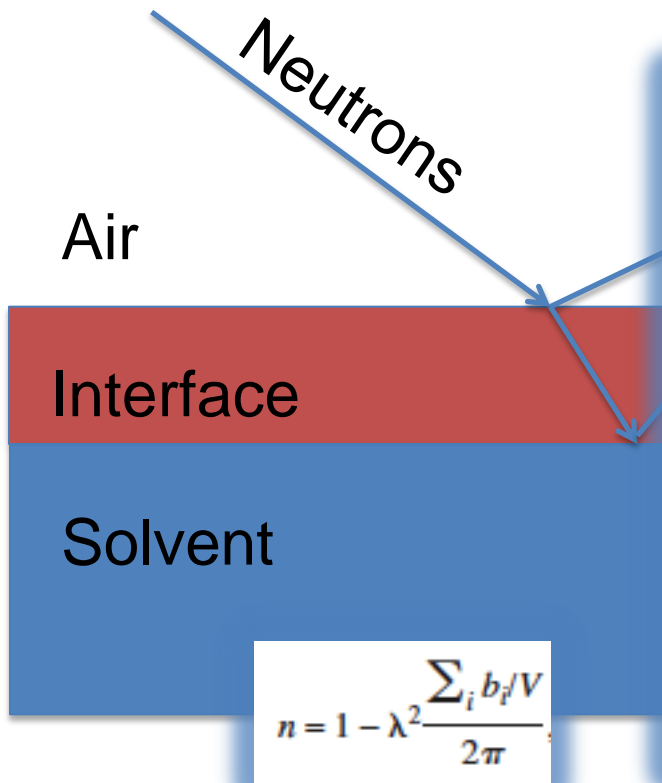
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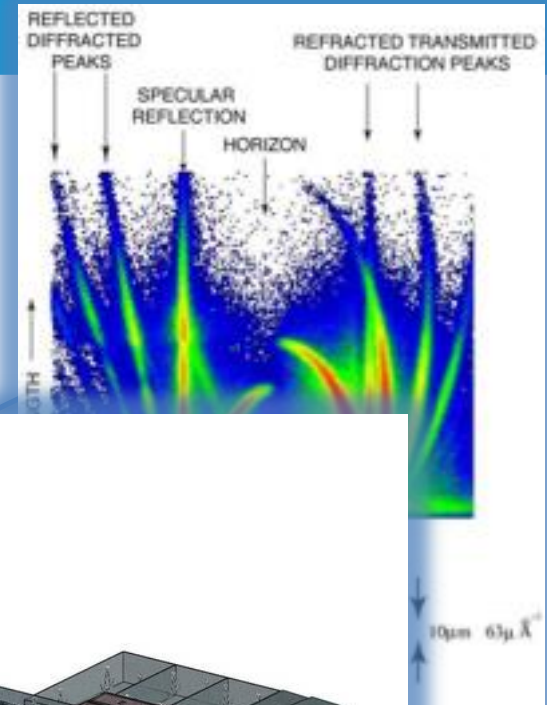
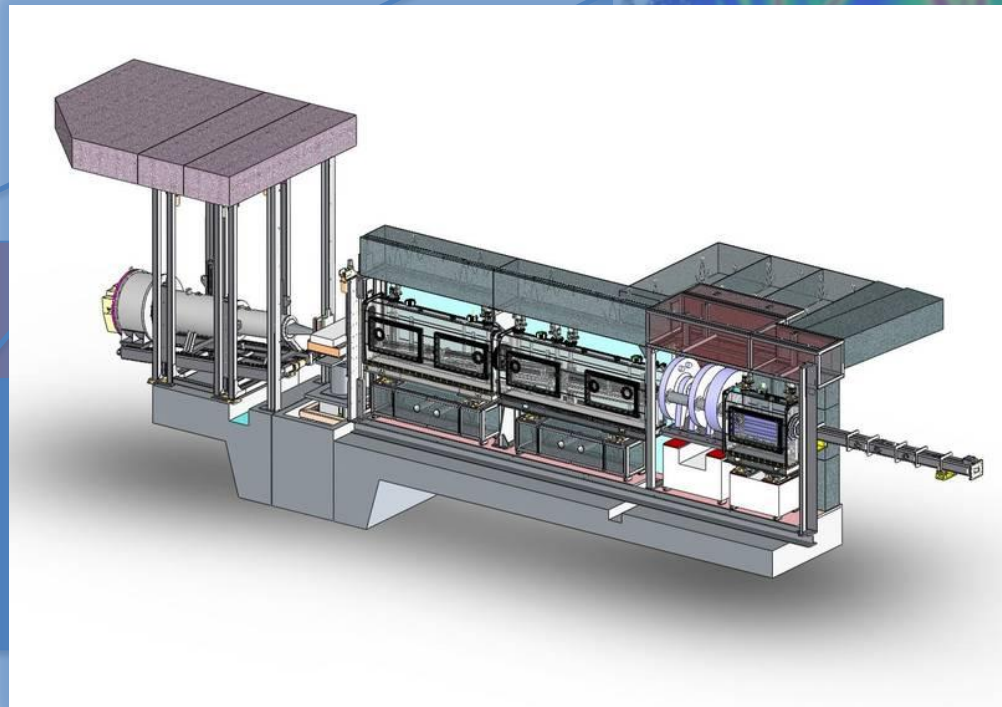


# Neutron Reflectivity

- Basic principle



$$n = 1 - \lambda^2 \frac{\sum_i b_i / V}{2\pi}$$



10  $\mu\text{m}$  63  $\mu\text{\AA}$

## Neutron reflectometry to investigate the delivery of lipids and DNA to interfaces (Review)

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Marité Cárdenas

*Biomedical Technology Laboratory, Faculty of Health and Society, Malmö University, SE-205 06 Malmö, Sweden*

Per Linse

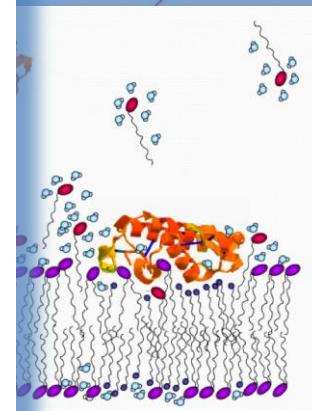
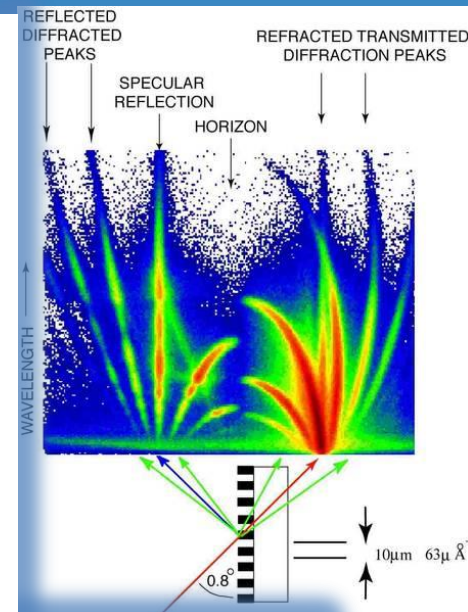
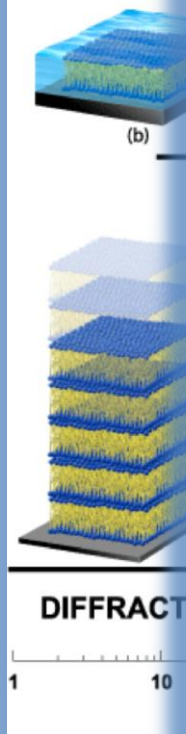
*Physical Chemistry 1, Lund University, Box 124, SE-221 00 Lund, Sweden*

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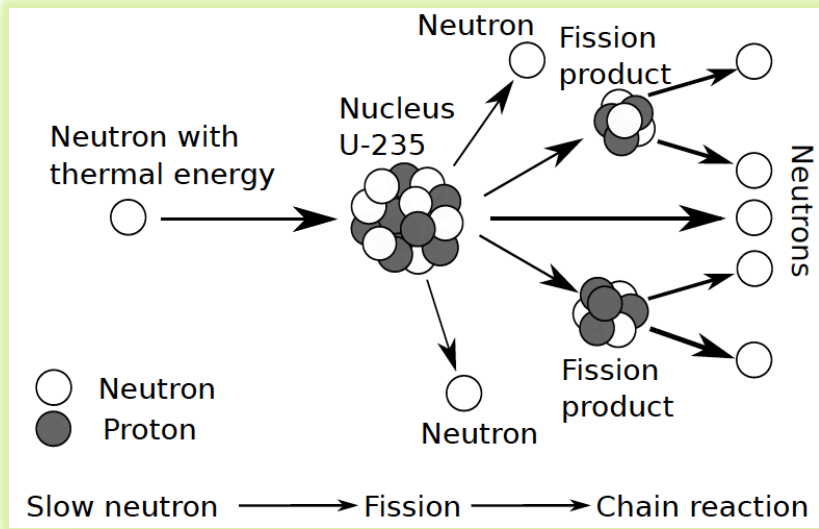
(Received 6 May 2008; accepted 1 August 2008; published 19 December 2008)

The application of scattering methods in the study of biological and biomedical problems is a field of research that is currently experiencing fast growth. In particular, neutron reflectometry (NR) is a technique that is becoming progressively more widespread, as indicated by the current commissioning of several new reflectometers worldwide. NR is valuable for the characterization of biomolecules at interfaces due to its capability to provide quantitative structural and compositional information on relevant molecular length scales. Recent years have seen an increasing number of applications of NR to problems related to drug and gene delivery. We start our review by summarizing the experimental methodology of the technique with reference to the description of biological liquid interfaces. Various methods for the interpretation of data are then discussed, including a new approach based on the lattice mean-field theory to help characterize stimulus-responsive surfaces relevant to drug delivery function. Recent progress in the subject area is reviewed in terms of NR studies relevant to the delivery of lipids and DNA to surfaces. Lastly, we discuss two case studies to exemplify practical features of NR that are exploited in combination with complementary techniques. The first case concerns the interactions of lipid-based cubic phase nanoparticles with model membranes (a drug delivery application), and the second case concerns DNA compaction at surfaces and in the bulk solution (a gene delivery application). © 2008 American Vacuum Society. [DOI: 10.1116/1.2976448]



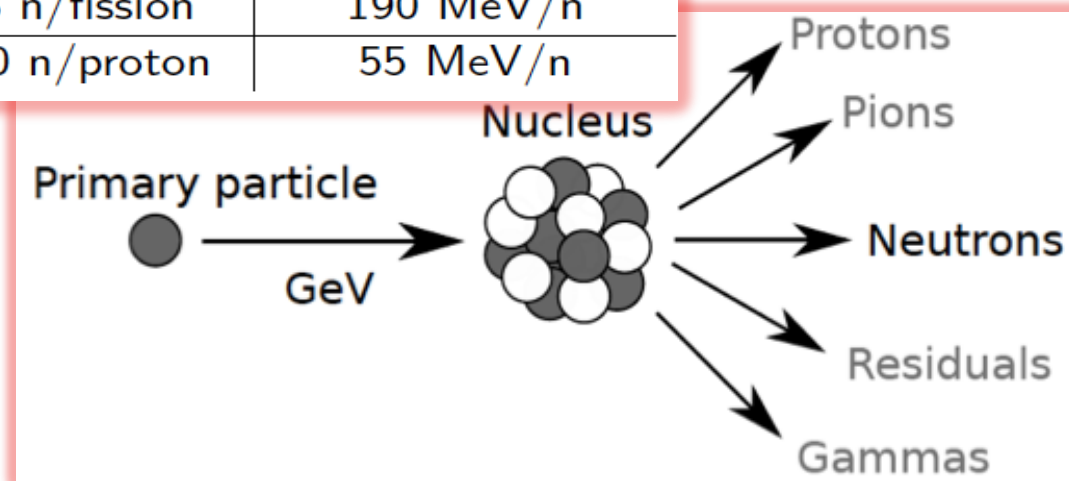
# Fission and Spallation

# Fission and Spallation

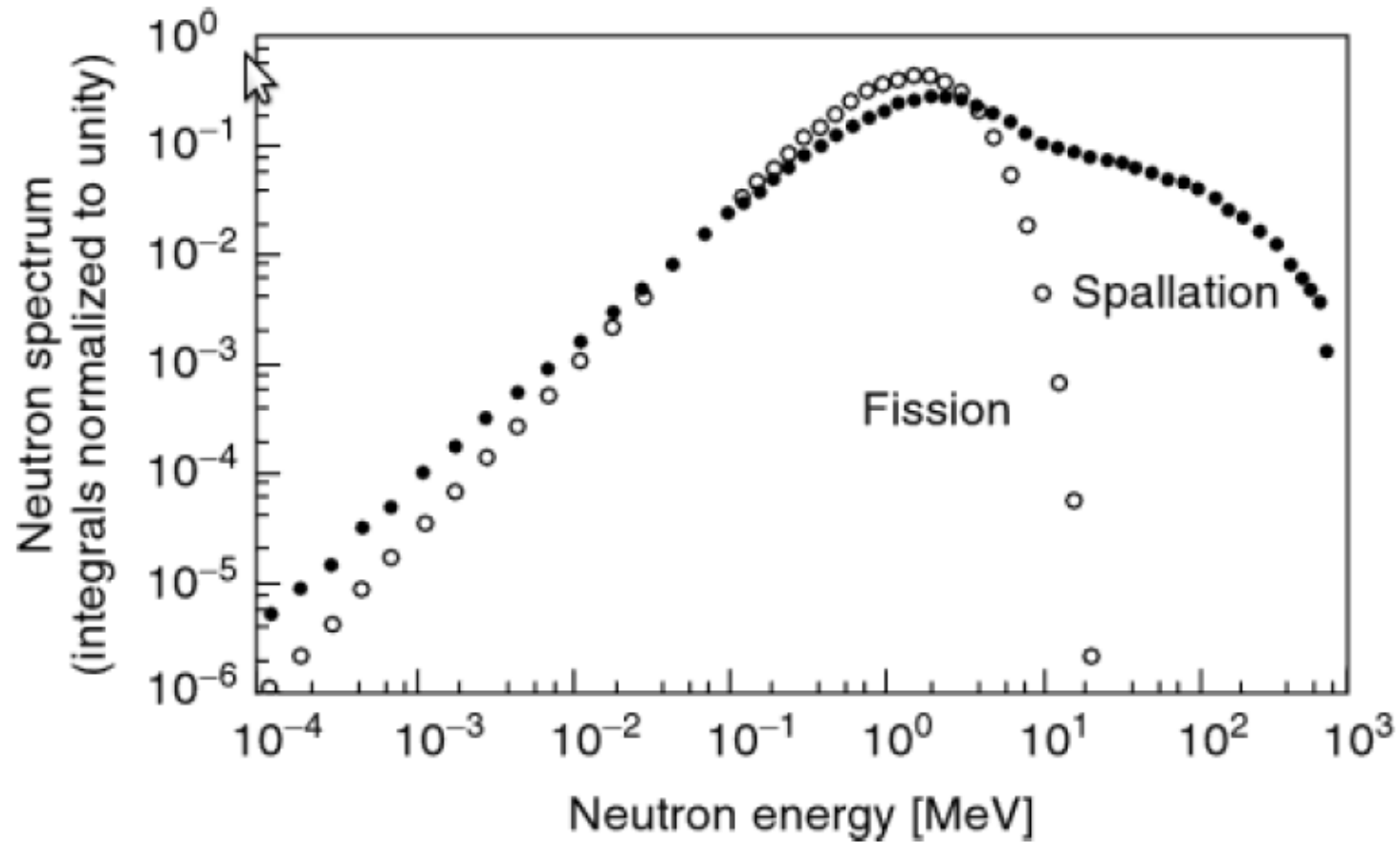


Spallation is a non-elastic nuclear interaction induced by a high-energy particle producing numerous secondary particles

Process	Reaction	Neutron yield	Energy deposition
Fission	$^{235}\text{U}(n,f)$	3 n/fission	190 MeV/n
Spallation	$p\ 1\ \text{GeV} \rightarrow \text{Hg}$	30 n/proton	55 MeV/n



# Fission and Spallation



## Energy efficiency is key for high intensity neutron beam production

### Fast neutrons produced / joule **heat deposited** in target station

Fission reactors:  $\sim 10^9$  (in  $\sim 50$  liter volume)

Spallation:  $\sim 10^{10}$  (in  $\sim 2$  liter volume)

Fusion:  $\sim 1.5 \times 10^{10}$  (in  $\sim 2$  liter volume)  
(but neutron slowing down efficiency reduced by  $\sim 20$  times)

Photo neutrons:  $\sim 10^9$  (in  $\sim 0.01$  liter volume)

Nuclear reaction (p, Be):  $\sim 10^8$  (in  $\sim 0.001$  liter volume)

Laser induced fusion:  $\sim 10^4$  (in  $\sim 10^{-9}$  liter volume)

**Spallation: most favourable for the foreseeable future**

# Spallation Process



## **Spallation Neutron Yield** (i.e. multiplicity of emitted neutrons)

determines the requirement in terms of the accelerator power (current and energy of incident proton beam).

## **Spallation Neutron Spectrum** (i.e. energy distribution of emitted neutrons)

determines the damage and activation of the structural materials (design of the beam window and spallation target)

## **Spallation Product Distributions**

determines the radiotoxicity of the residues (radioprotection requirements).

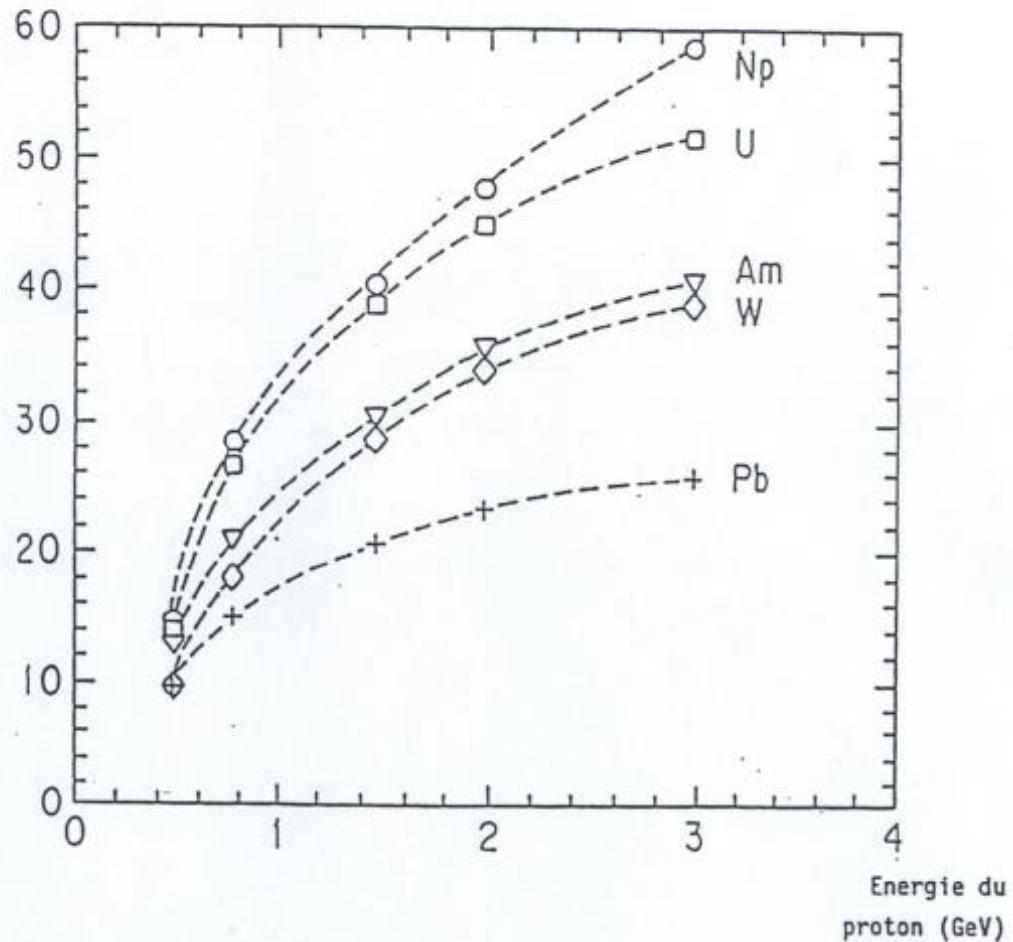
## **Energy Deposition**

determines the thermal-hydraulic requirements (cooling capabilities and nature of the spallation target).

**→ Sub-critical Reaction**

# Spallation : choice of target materials

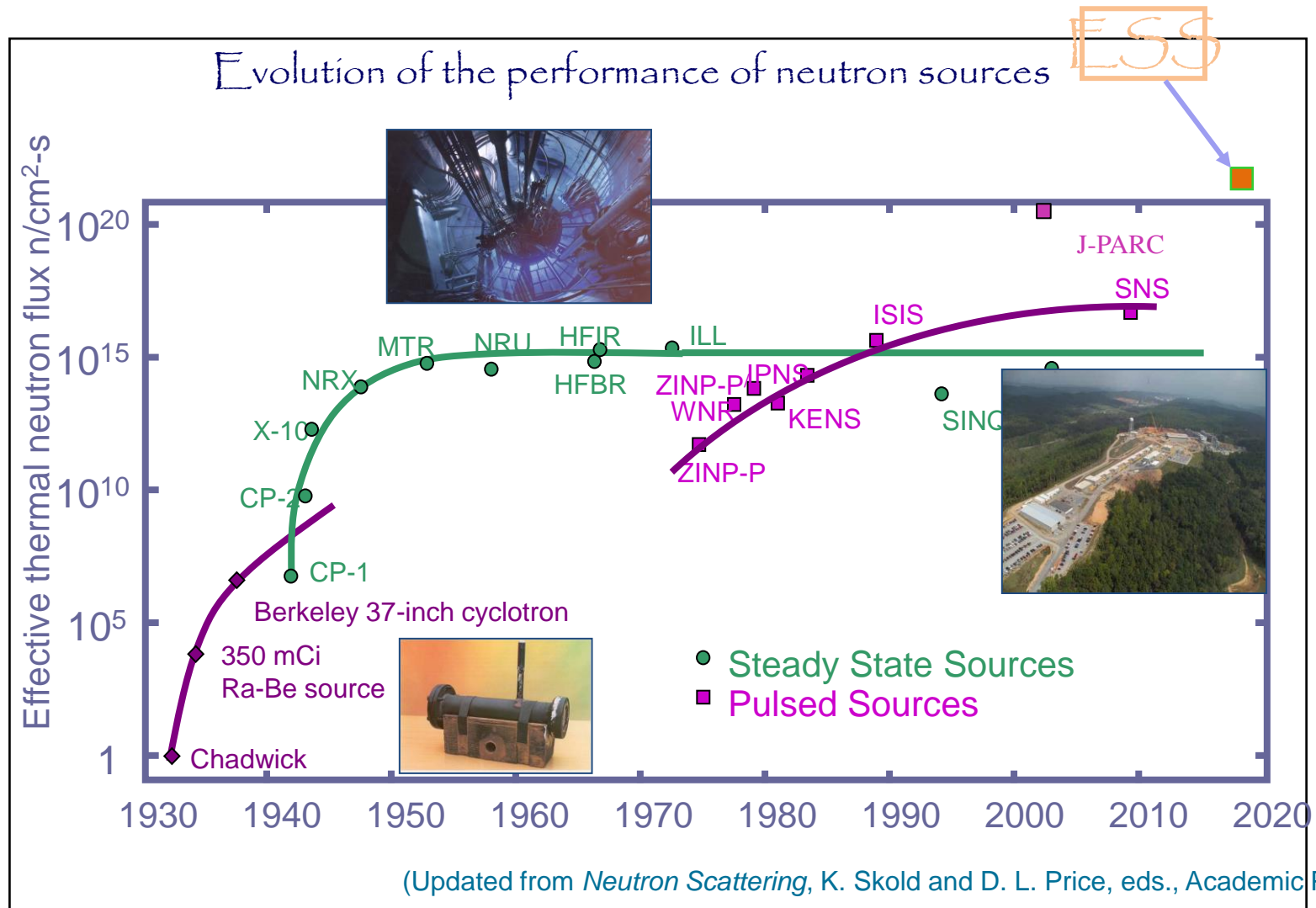
Nombre de neutrons par proton



$$Y(E,A)=0.1 [A+20] [E (\text{GeV}) -0.12] \quad n/p$$



# High time average and peak flux



- Neutrons properties and their interactions
- How to generate intense neutron beams using high power proton linear accelerator: The example of the ESS

for further reading

- Applications using Neutrons

# ESS: Materials, Life Science and Society



## EU Horizon 2020 – strategy

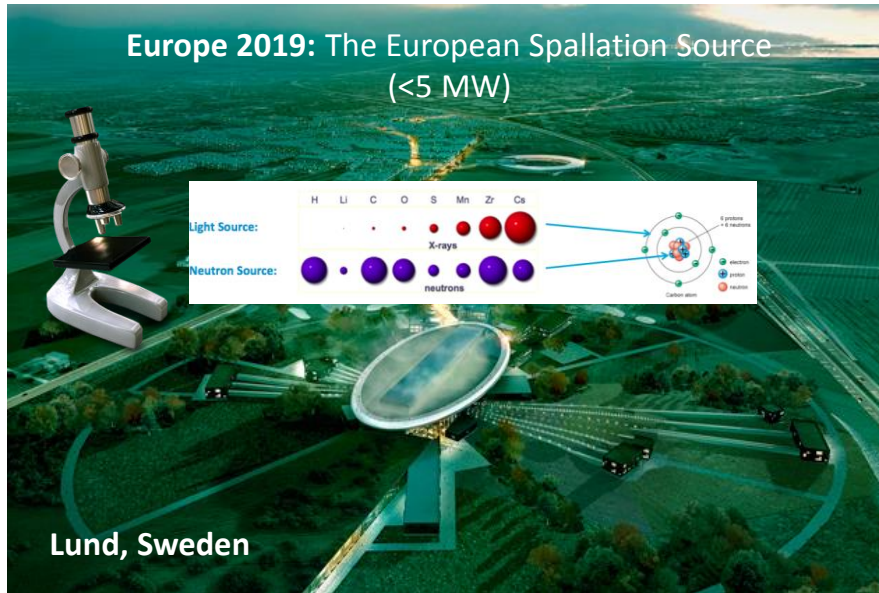
The structure which the EC proposed consists of three basic priorities:

1. **Excellent Science**
2. **Industrial Leadership**
3. **Societal Challenges**

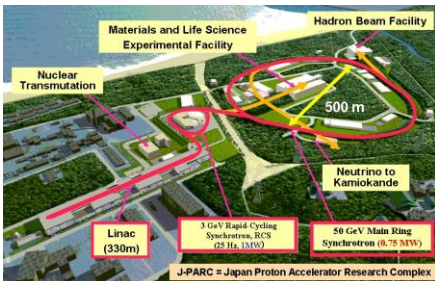


# Spallation Sources

## Philosophie de “Pré-vert”: Greenfield



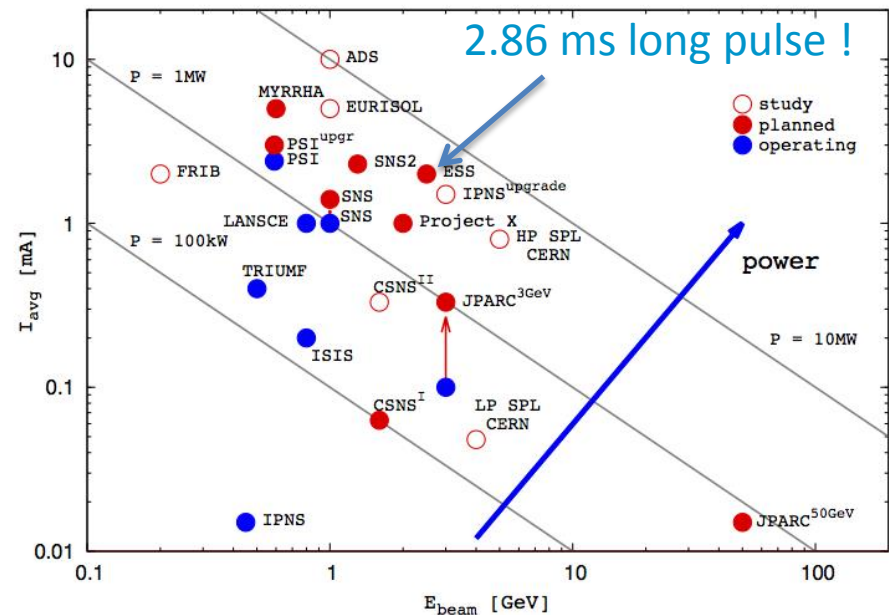
- Will bring new insights to the grand challenges of science and innovation
- Collaborative project: more than 17 countries
- 2014: Start of construction phase of the world's most powerful linear proton accelerator
- 2019: Provide the world's most advanced tools for studying materials with neutrons (~ 450 employees; > 2500 users / year)



Japan 2008:  
J-PARC (<1MW)



USA 2006:  
SNS (<1.4 MW)  
1 GeV, 26 mA in linac, 627 ns long pulse,  
60 Hz



# Helicopter view of ESS

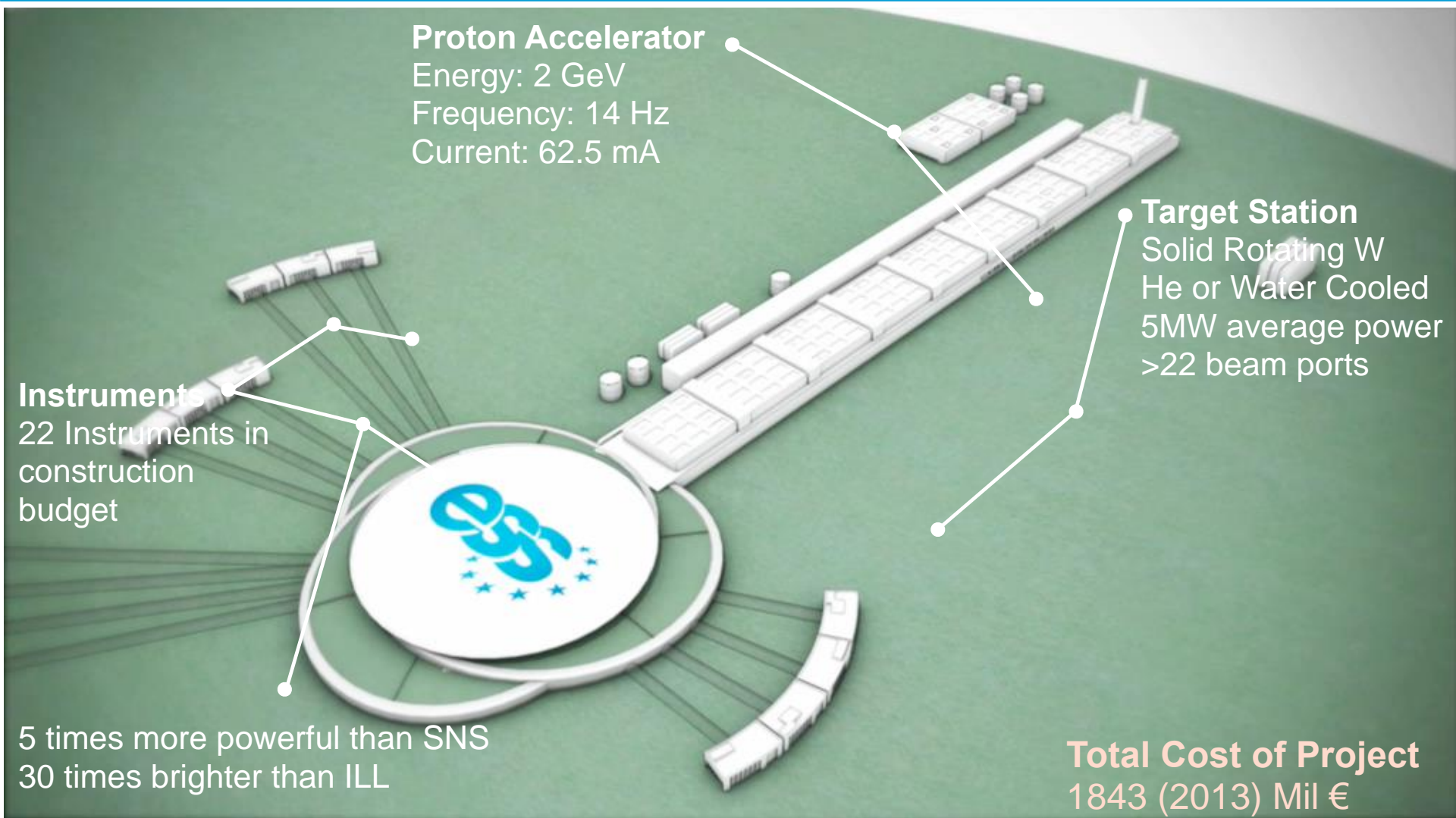
**Proton Accelerator**  
Energy: 2 GeV  
Frequency: 14 Hz  
Current: 62.5 mA

**Target Station**  
Solid Rotating W  
He or Water Cooled  
5MW average power  
>22 beam ports

**Instruments**  
22 Instruments in  
construction  
budget

5 times more powerful than SNS  
30 times brighter than ILL

**Total Cost of Project**  
1843 (2013) Mil €

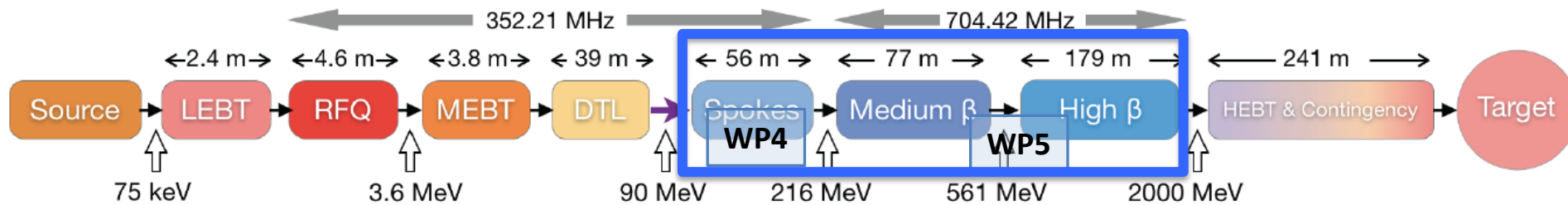


# Collaborative projects

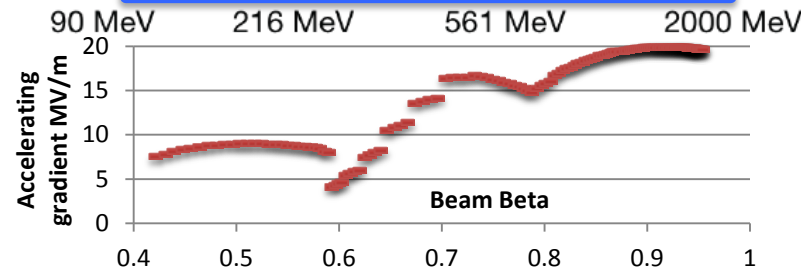
- ESS is an emerging research laboratory with (still) very limited capacity in-house
- Two possibilities:
  - Limit the scope of the project so that it can be done with in-house resources
  - Work in a collaboration where the scope of the project can be set by the total capacity (distributed) of the partners
- The accelerator part of the project well suited for this as this community has a strong tradition of open collaboration (e.g. E-XFEL, FAIR, LHC, European Commission Framework Programs and design studies)
- To keep cost down and to optimize schedule this requires that investments in required infrastructure is done at the partner with best capacity to deliver

# Linac redesign to meet ESS cost objective

Optimus+\_2013\_10\_31



Beam power (MW)	5
Beam current (mA)	62.5
Linac energy (GeV)	2
Beam pulse length (ms)	2.86
Repetition rate (Hz)	14



	Num. of CMs	Num. of cavities
Spoke	13	26
Medium $\beta$ (6-cell)	9	36
High $\beta$ (5-cell)	21	84

Style	Spoke	Medium- $\beta$	High- $\beta$
Freq. (MHz)	352.21	704.42	704.42
Cavity #	26	36	84
Velocity range	0.42 to 0.58	0.58 to 0.78	0.78 to 0.95
Nom. Acc. Voltage (MV)	5.74	14.3	18.2
Loaded quality factor	$2.85 \times 10^5$	$8 \times 10^5$	$7.6 \times 10^5$

[SRF2013 – “The ESS Superconducting Linear Accelerator”, C. Darve, M. Eshraqi, M. Lindroos, D. McGinnis, S. Molloy, P. Bosland and S. Bousson]



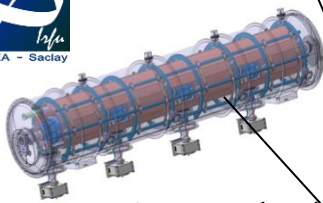
# Accelerator collaborations for ESS “start-up”




**IPN**  
INSTITUT DE PHYSIQUE NUCLEAIRE  
ORSAY





WP4 : Sebastien Bousson






CEA - Saclay



WP5 : Pierre Bosland

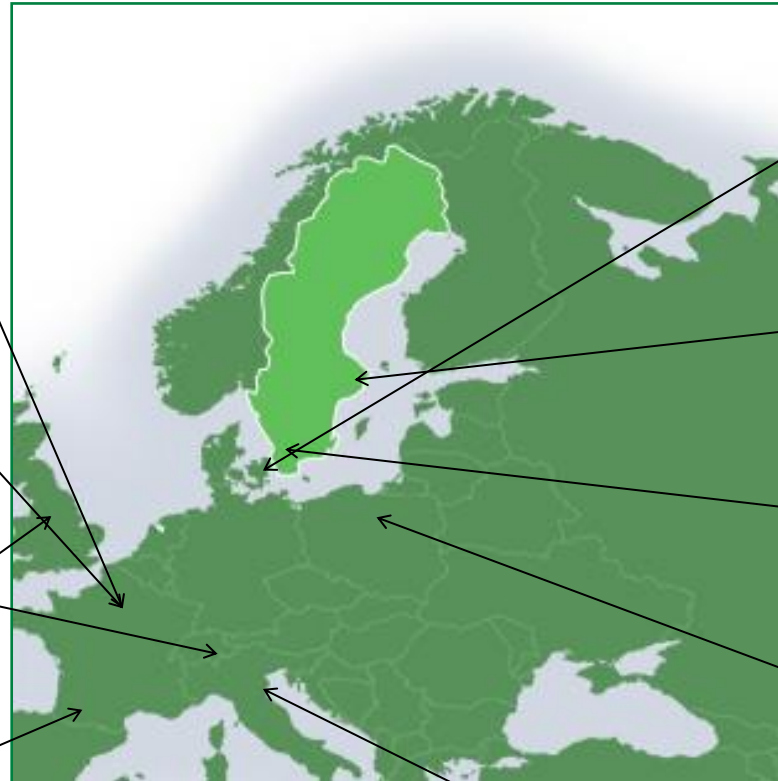
CERN



Roger Barlow

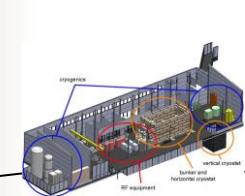





Ibon Bustinduy

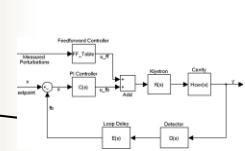







Søren Pape Møller

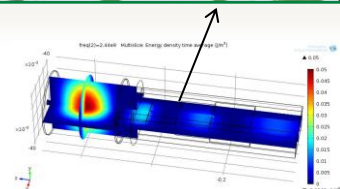
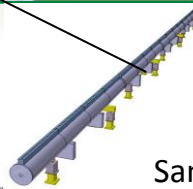





Roger Ruber

Anders J Johansson

The National Center for Nuclear Research, Swierk

Santo Gammino





# Cavity Cryomodule Technology Demonstrators

## One full scale 704 MHz medium and high-beta cavity cryomodules

A staged approach towards the series industrialization and the ESS Linac tunnel installation

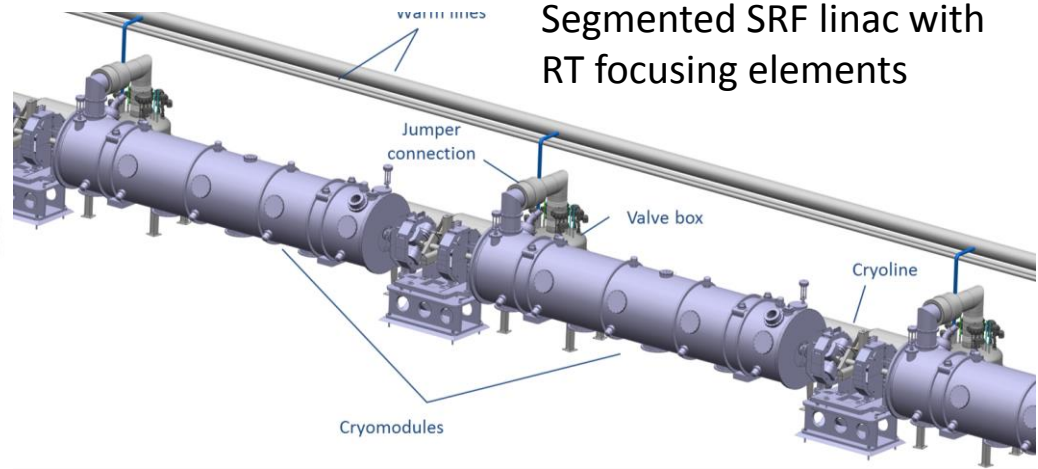
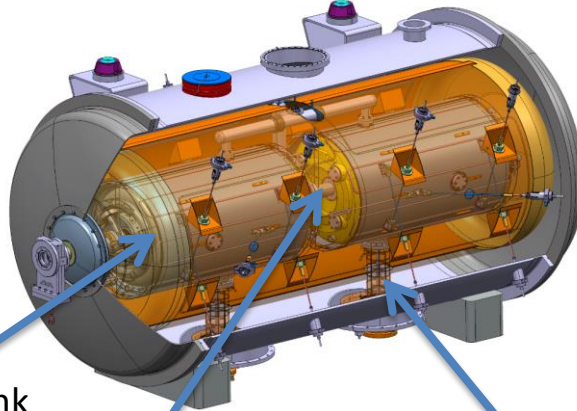
- Validate designs and construction capability of SRF components
- Prepare the industrialization process by validating component life-cycles (incl. assembling process, QA)
- Validate component performances (incl. RF, mechanical, thermal)
- Develop ESS 704 MHz SRF linac operating procedures
- Validate control command strategy (Control box, PLC, EPICS, LLRF)
- Validate ESS integration and interfaces with RF, cryogenics, vacuum and control systems
- Train people in ESS SRF Technology and build an ESS collaboration



→ Similar process for the spoke cryomodules

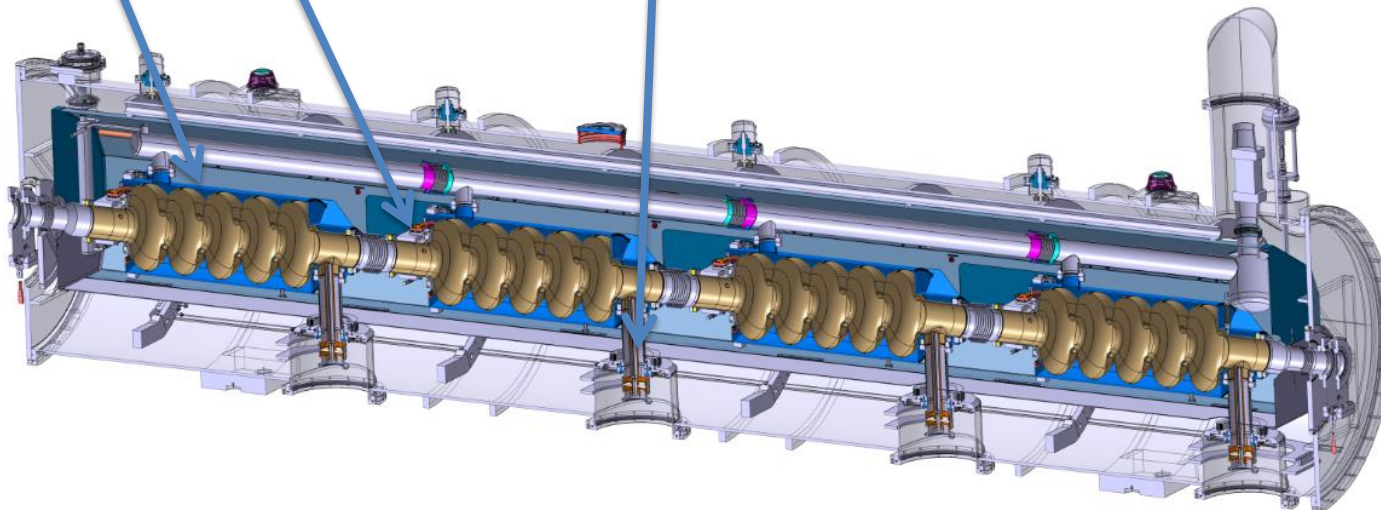
# ESS SRF Cavity Cryomodules

## 2 Spoke Cavities per Cryomodule



## Segmented SRF linac with RT focusing elements

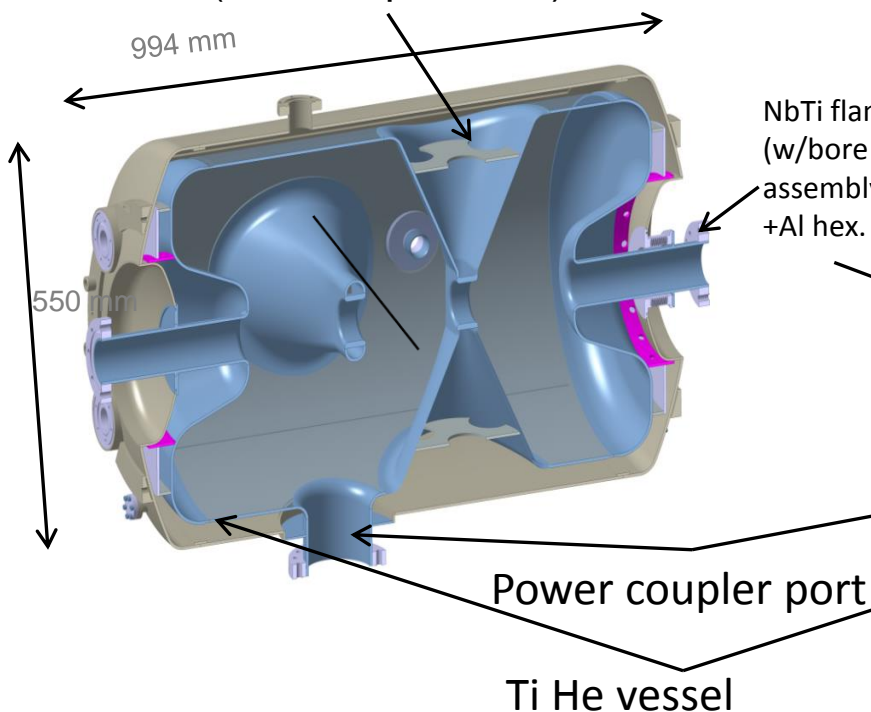
## 4 Elliptical Cavities per Cryomodule



# SRF Cavities Development

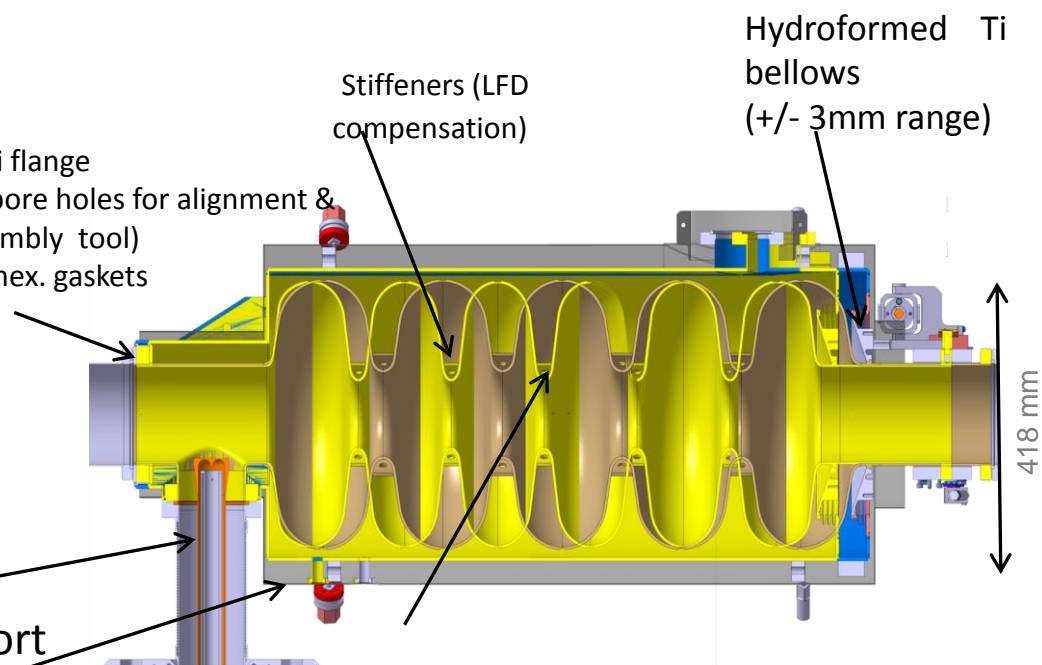
## Spoke cavity

→ Stiffeners on the Spoke bars  
(vacuum pressure)



## Elliptical cavities

→ No HOM power couplers



### Medium beta:

- 6 cells – beta=0.67
- Length 1259,40mm

### High beta:

- 5 cells – beta=0.86
- Length 1316,91mm

# ESS Requirements and RF Parameters



EUROPEAN  
SPALLATION  
SOURCE

## Spoke cavities

Frequency (MHz)	352,2
Optimum beta	0,50
Operating temperature (K)	2
Nominal Accelerating gradient (MV/m)	9
Lacc ( $\beta_{opt} \times nb \text{ gaps} \times \lambda/2$ ) (m)	0,639
Bpk (mT)	79 (max)
Epk (MV/m)	39 (max)
Bpk/Eacc (mT/MV/m)	<8,75
Epk/Eacc	<4,38
Beam tube diameter (mm)	50
RF peak power (kW)	335
G ( $\Omega$ )	130
Max R/Q (W)	427
Q <sub>ext</sub>	2,85 10 <sup>5</sup>
Q <sub>0</sub> at nominal gardient	1,5 10 <sup>9</sup>

## Elliptical cavities

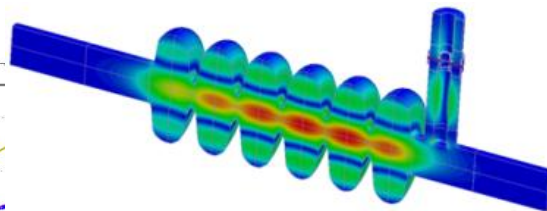
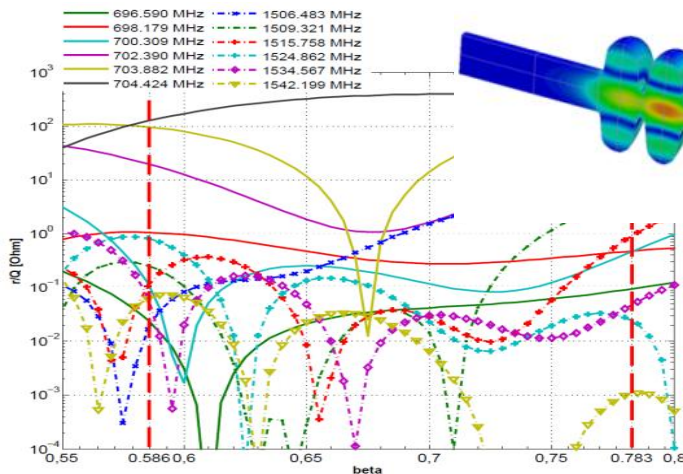
	Medium	High
Geometrical beta	0.67	0.86
Frequency (MHz)	704.42	
Number of cells	6	5
Operating temperature (K)	2	
Epk max (MV/m)	45	45
Nominal Accelerating gradient (MV/m)	16.7	19.9
Q <sub>0</sub> at nominal gradient	> 5e9	
Q <sub>ext</sub>	7.5 10 <sup>5</sup>	7.6 10 <sup>5</sup>
Iris diameter (mm)	94	120
Cell to cell coupling k (%)	1.22	1.8
p,5p/6 (or 4p/5) mode sep. (MHz)	0.54	1.2
Epk/Eacc	2.36	2.2
Bpk/Eacc (mT/(MV/m))	4.79	4.3
Maximum. r/Q (W)	394	477
Optimum $\beta$	0.705	0.92
G ( $\Omega$ )	196.63	241
RF peak power (kW)	1100	

# Medium- $\beta$ Elliptical Cavities

$K_L$  reduction using compensation rings for medium and high-beta



Nominal wall thickness [mm]	3.6
Cavity stiffness $K_{cav}$ [kN/mm]	2.59
Tuning sensitivity $Df/Dz$ [kHz/mm]	197
$K_L$ with fixed ends [Hz/(MV/m) <sup>2</sup> ]	-0.36
$K_L$ with free ends [Hz/(MV/m) <sup>2</sup> ]	-8.9
Pressure sensitivity $K_p$ [Hz/mbar] (fixed ends)	4.85

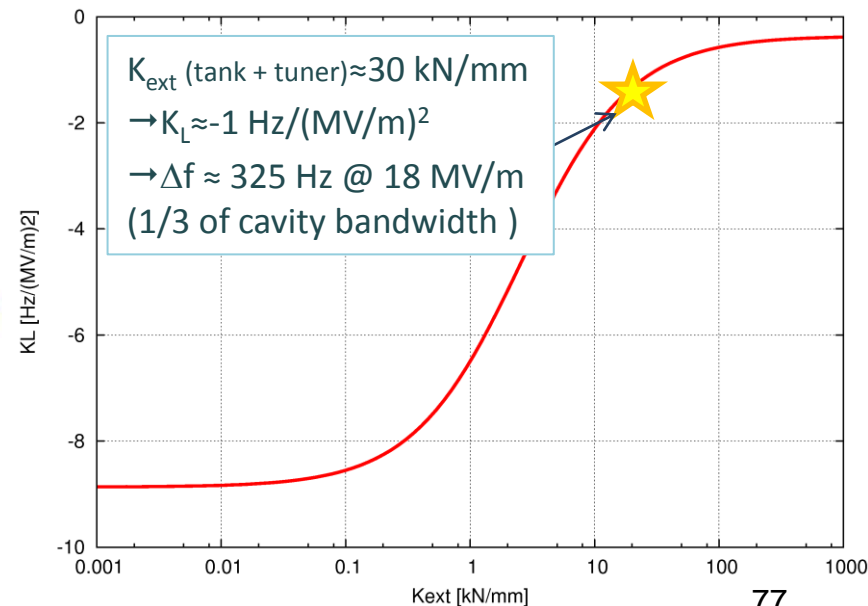


## RF/mechanical design

Lorentz detuning

$$K_L = \Delta f / E_{acc}^2$$

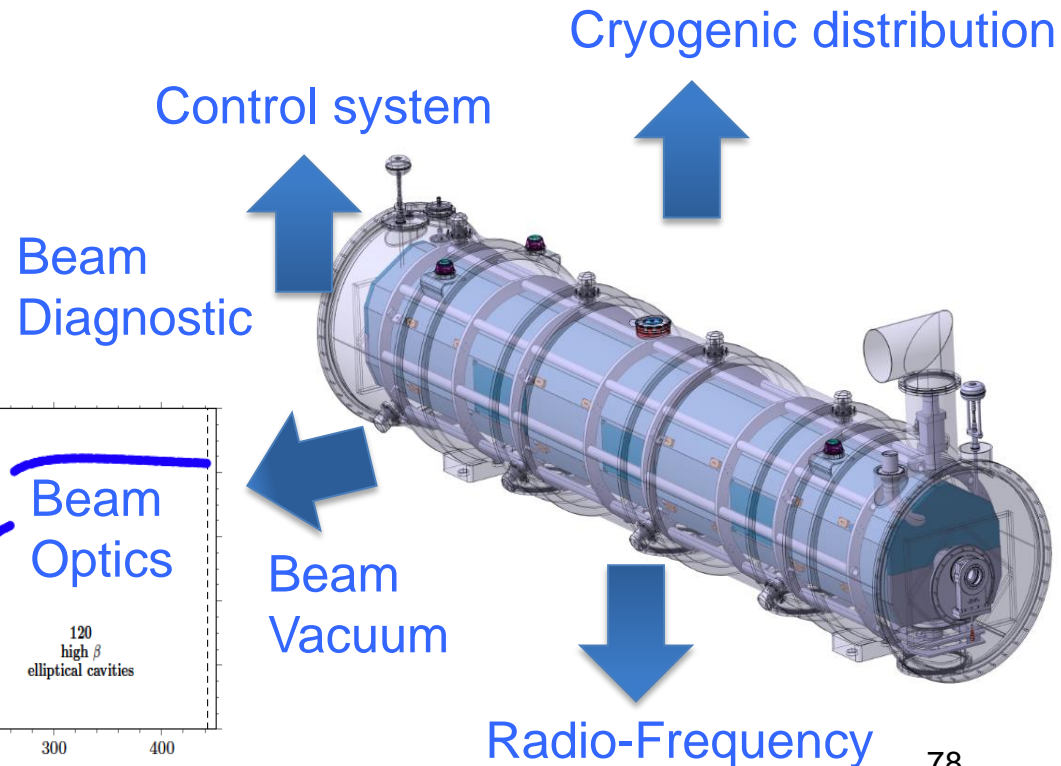
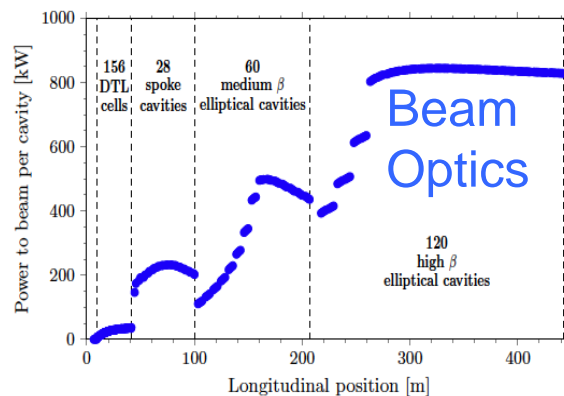
$$K_L = K_{L\infty} + \frac{\Delta f \vec{F}_\infty \cdot \vec{u}_z / E_{acc}^2}{\Delta Z (K_{ext} + K_{cav})}$$



# Cryomodule Interfaces

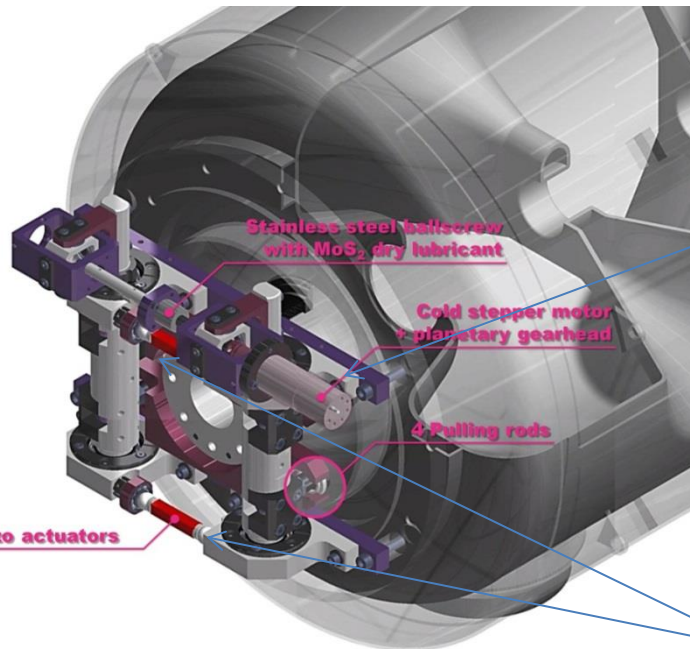
- Most AD internal Work Packages (beam optics, RF, cryo, vacuum, test stands, electrical, cooling, installation)
- External WPs cryomodule, cavity and designers and potential In-Kind collaborators
- Control command (Control Box, PLC, LLRF, MPS, EPICS)
- Data-logging ICS teams
- ESS ES&H
- Conventional Facility
- ESS system engineer, QA
- Survey experts
- Transport

Previous Linac version for comparison →

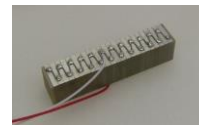


# Cold Tuning System

## Spoke CTS



Stepper motor and planetary gearbox (1/100e) at cold and in vacuum



2 piezo stacks

### Slow tuner

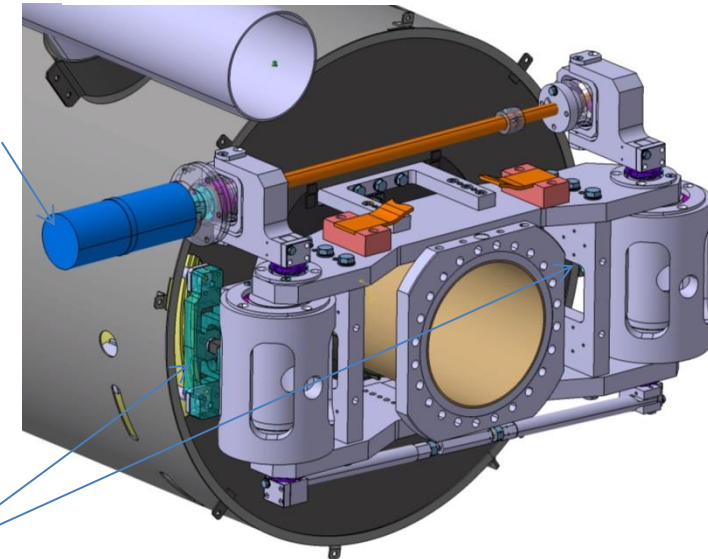
Main purpose : Compensation of large frequency shifts with a low speed

Actuator used : Stepper motor

&

## Elliptical CTS

Type V ; 5-cell prototype  
+/- 3 mm range on cavity

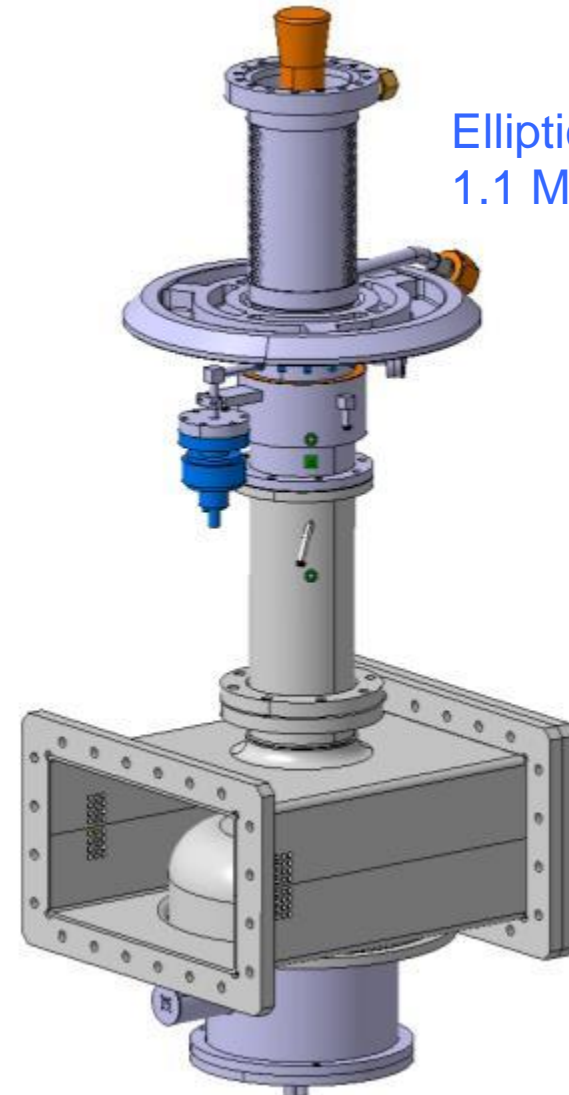
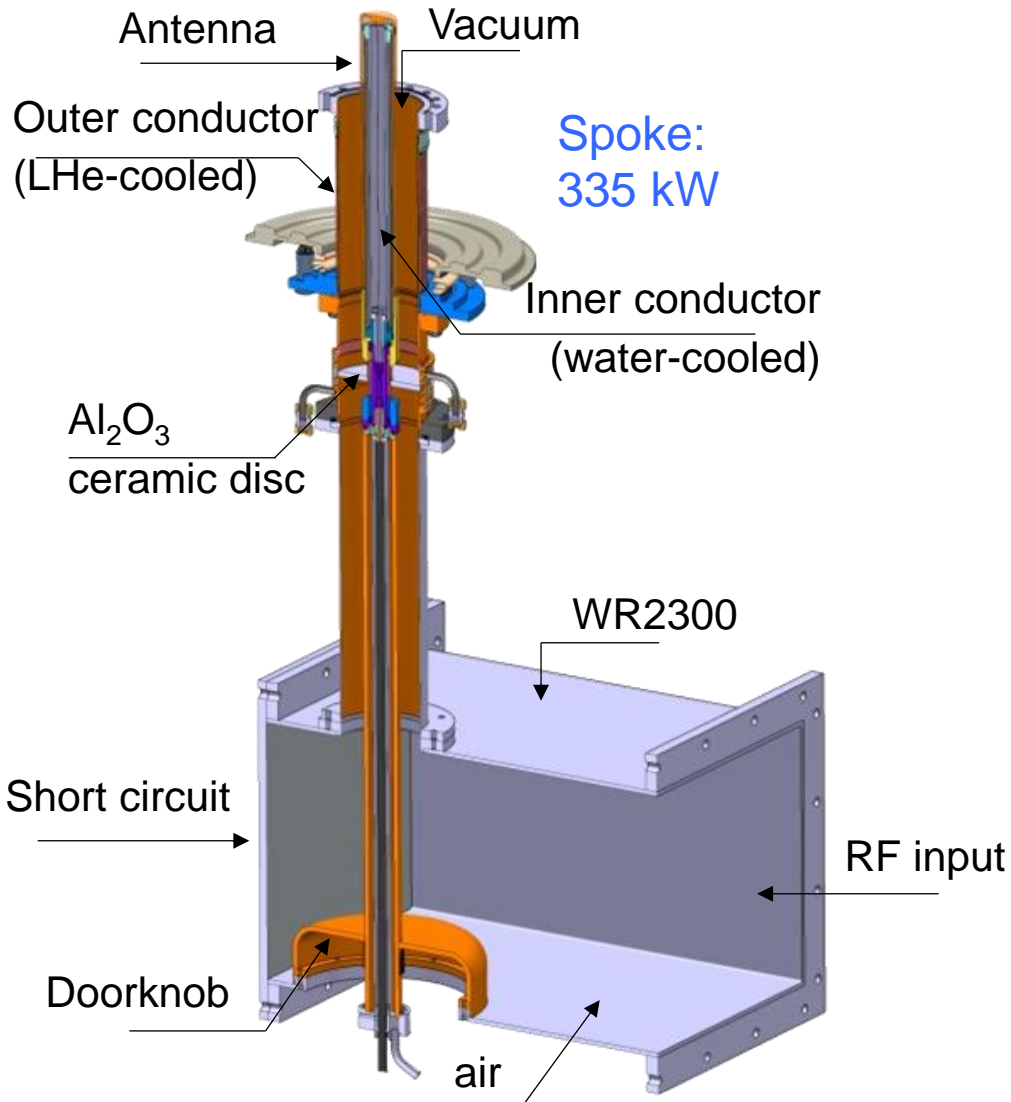


### Fast tuner

Main purpose : Compensation of small frequency shifts with a high speed

Actuator used : Piezoelectric actuators

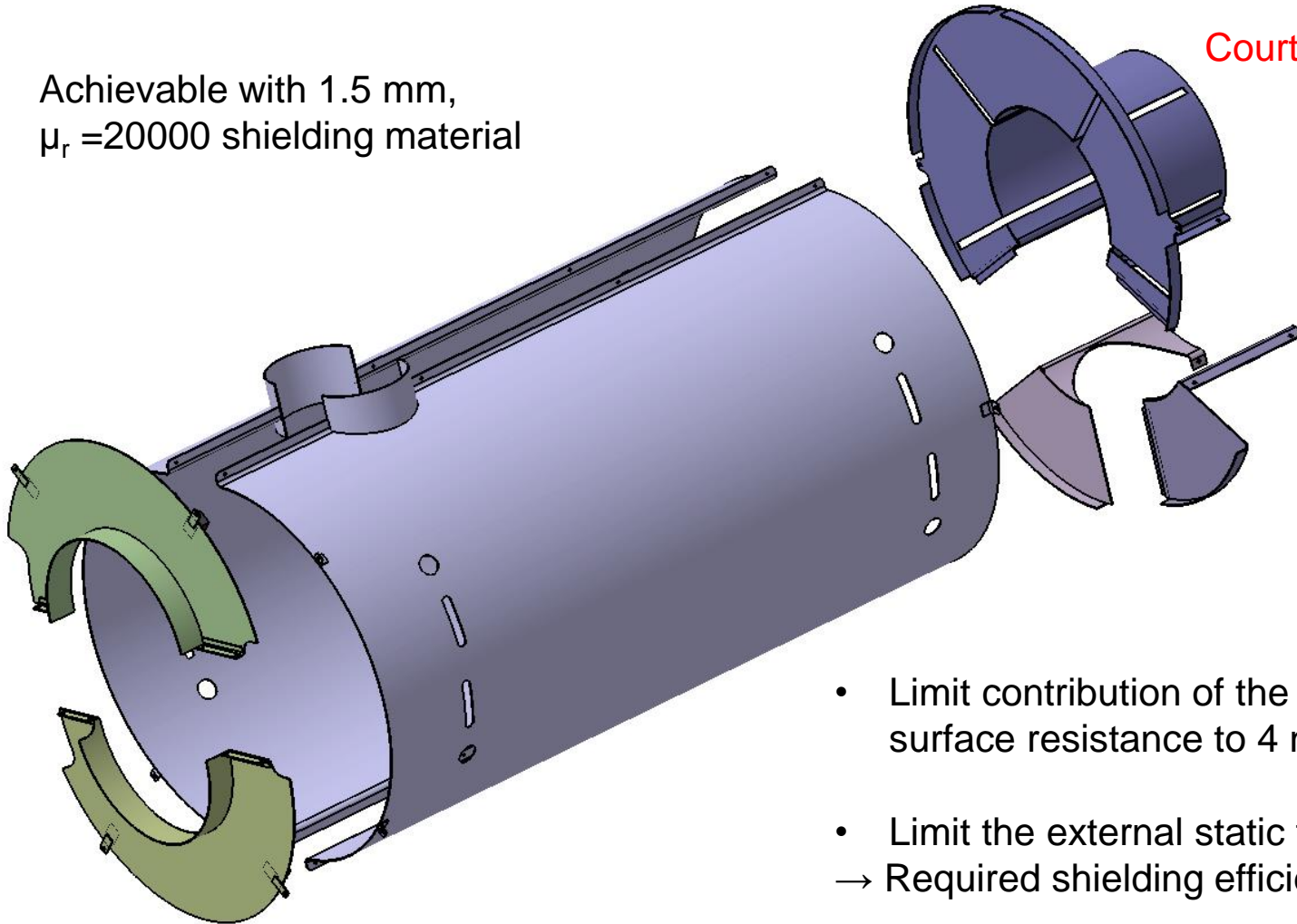
# Fundamental Power Coupler





# Magnetic shield, e.g. Elliptical cavity

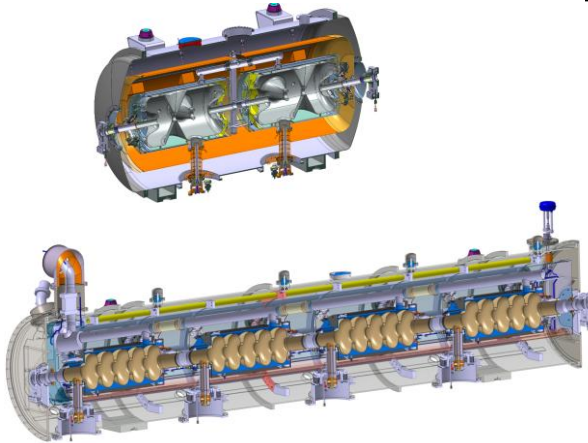
Achievable with 1.5 mm,  
 $\mu_r = 20000$  shielding material



Courtesy of J. Plouin/ CEA

- Limit contribution of the trapped flux to the surface resistance to  $4 \text{ n}\Omega$
- Limit the external static field to  $B_{\text{ext}} = 14 \text{ mG}$ .  
→ Required shielding efficiency equal to 35.

# Cryomodule Heat Load Distribution



## Per cryomodule

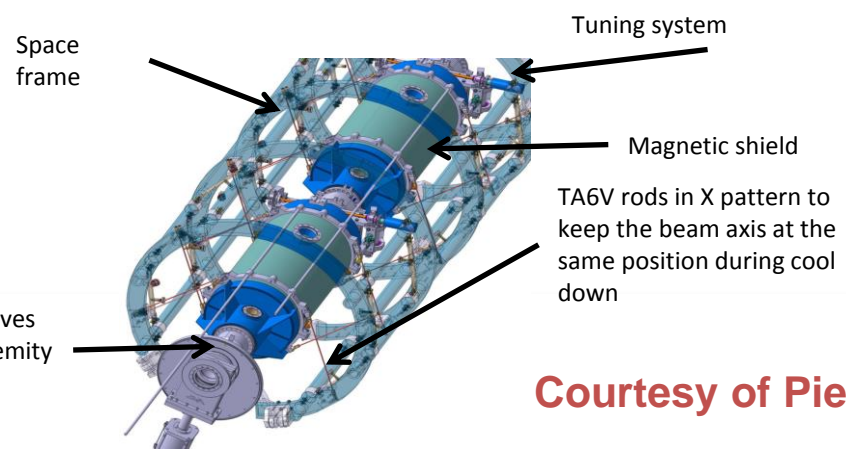
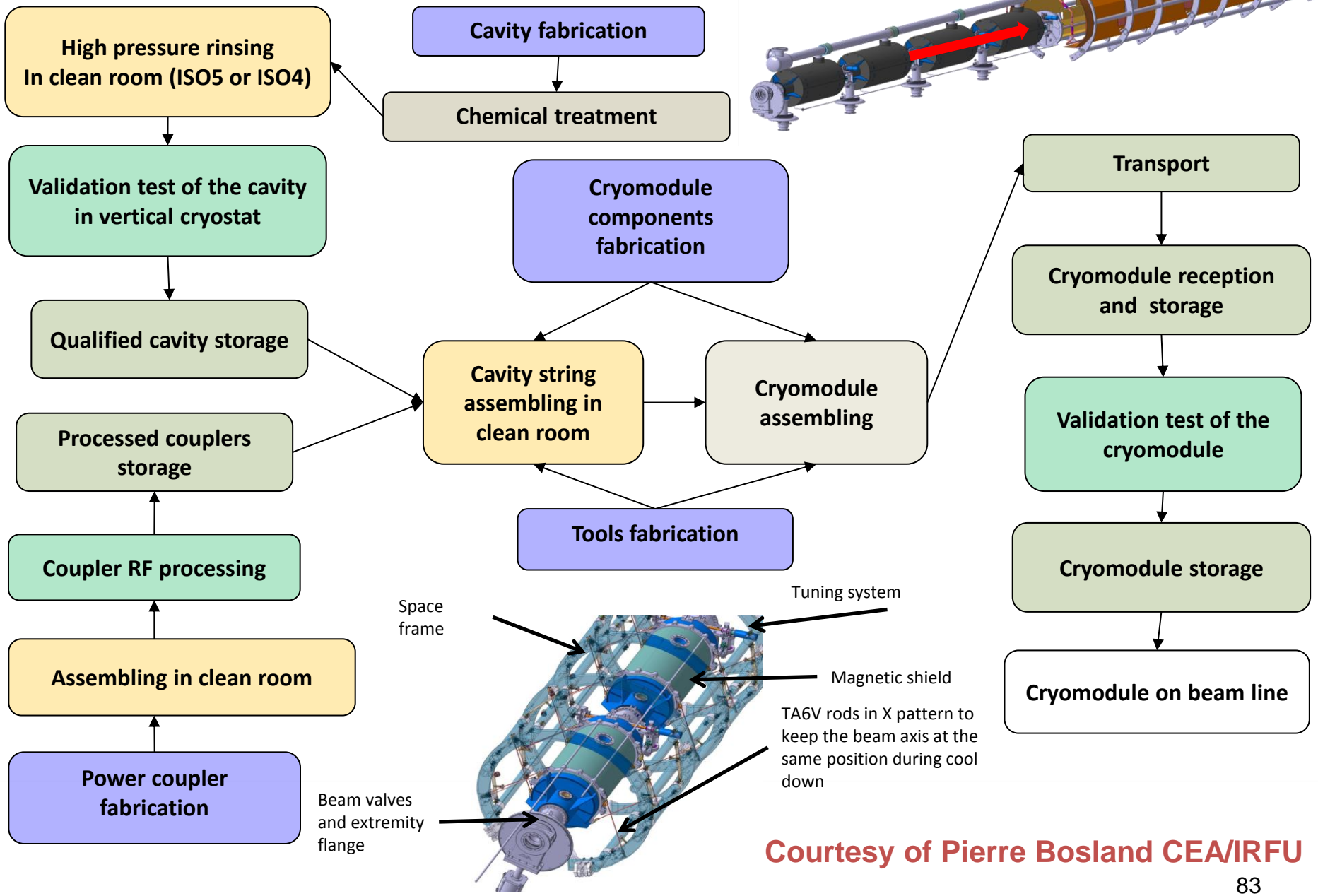
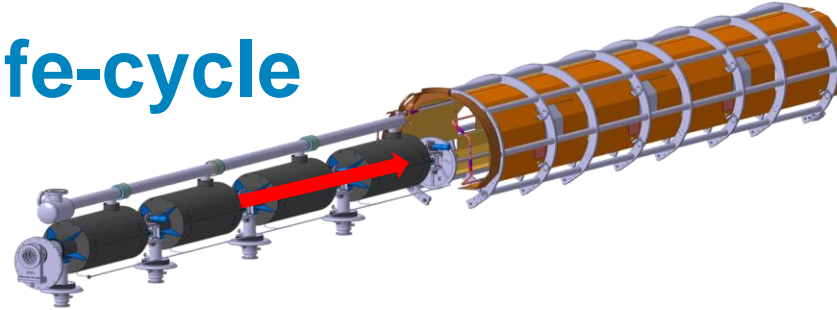
	Watts to 2 K							4.5 K Liquefaction (g/s)	Watts to ~50 K
	Static				Dynamic			Total	Total
	Others	Valves	Coupler	<b>Total</b>	Beam	Cavity	<b>Total</b>		
1 Spoke	3.3	0.2	3.5	<b>7</b>	1.5	5.0	<b>6.5</b>	0.092	30
1 MB	6.3	0.2	6.8	<b>13.3</b>	3.3	20	<b>23.3</b>	0.092	46.5
1 HB	6.3	0.2	6.8	<b>13.3</b>	3.3	24.4	<b>27.7</b>	0.092	46.5

## Sum for the Linac cryoplant (incl. 14 extra HB for contingency space)

	Number of CMs	Watts to 2K			4.5K Liquefaction (g/s)	Watts to ~ 50K
		Static	Dynamic	Total	Total	Total
Spoke	13	91	84.5	175.5	1.196	390
Medium beta	9	119.7	209.7	329.4	0.828	418.5
High beta	35	472.5	627.9	1100.4	3.22	1627.5
<b>Total</b>	<b>57</b>	<b>683.2</b>	<b>922.1</b>	<b>1605.3</b>	<b>5.244</b>	<b>2436</b>

Heat load due to the beam losses deposit a maximum of 0.5 W/m to the Spoke, medium and high-beta cavities sections 2 K temperature levels.

# Cryomodule life-cycle



Courtesy of Pierre Bosland CEA/IRFU

# Elliptical Cavity Preparation

High beta cavity fabrication (Zanon and RI)



Vertical Electropolishing system @ CEA



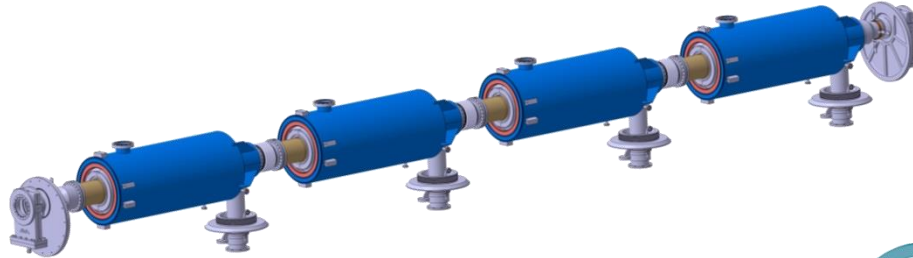
Study of the tooling in progress @ CEA

Example of the tooling for the assembling of the coupler on the cavity in clean room



# Elliptical Assembly Procedure

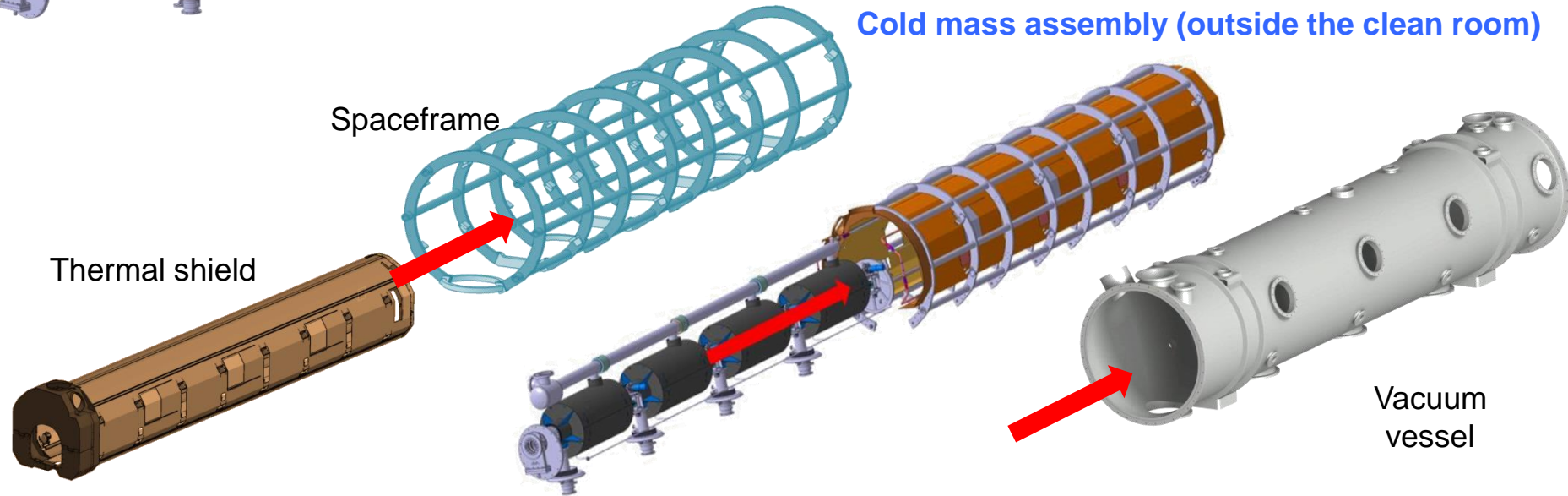
## Cavity string assembly in clean room



Build on existing knowledge (SNS, XFEL)

- Develop Training and “Fabrication file”
- Pre-industrialization
- Industrialization

## Cold mass assembly (outside the clean room)



**Design concept of the tooling:** most of parts will be used for both types of elliptical cryomodules

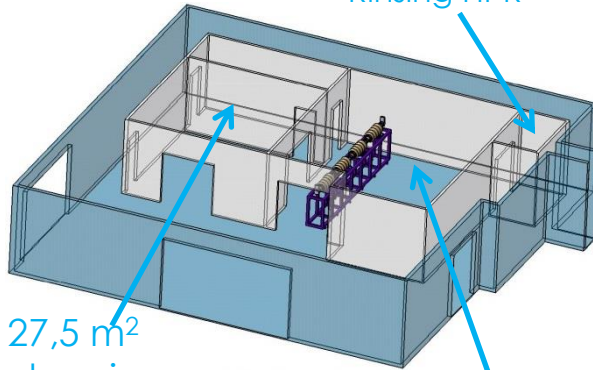
# Infrastructure in Saclay

Clean room for the M-ECCTD  
(and H-ECCTD)

High Pressure  
Rinsing HPR

ISO 7 27,5 m<sup>2</sup>  
Water cleaning

ISO 5  
52,69 m<sup>2</sup>



The clean room inauguration  
→ May 13th 2014

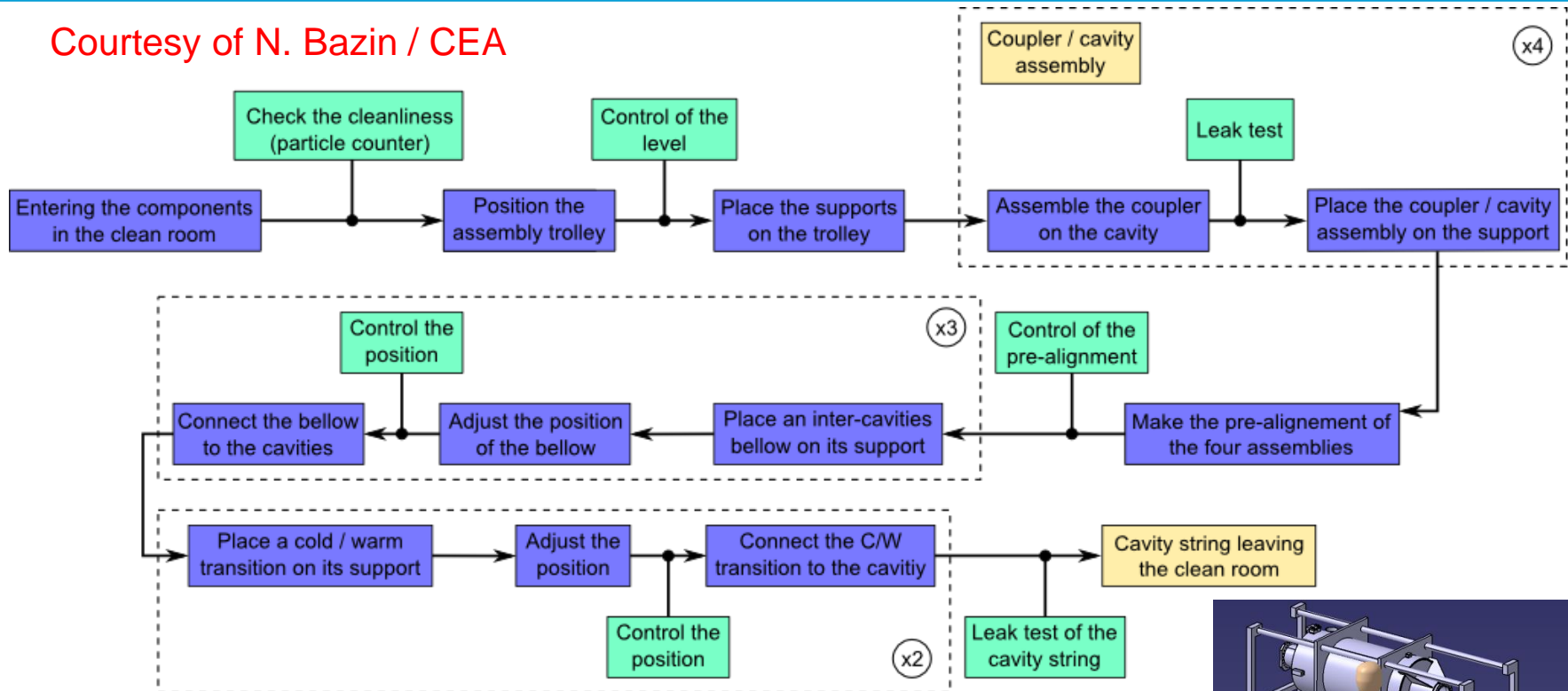
Possible IKC for the assembly by industry at Saclay  
(XFEL cryomodules assembly)

- Uses the current infrastructure at Saclay
- Benefits from the experience of the XFEL cryomodule assembly (ALSYOM)



# Assembly of elliptical cryomodules

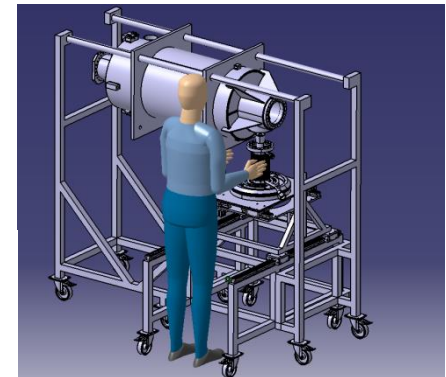
Courtesy of N. Bazin / CEA



□ Detailed procedures will be defined for every phases

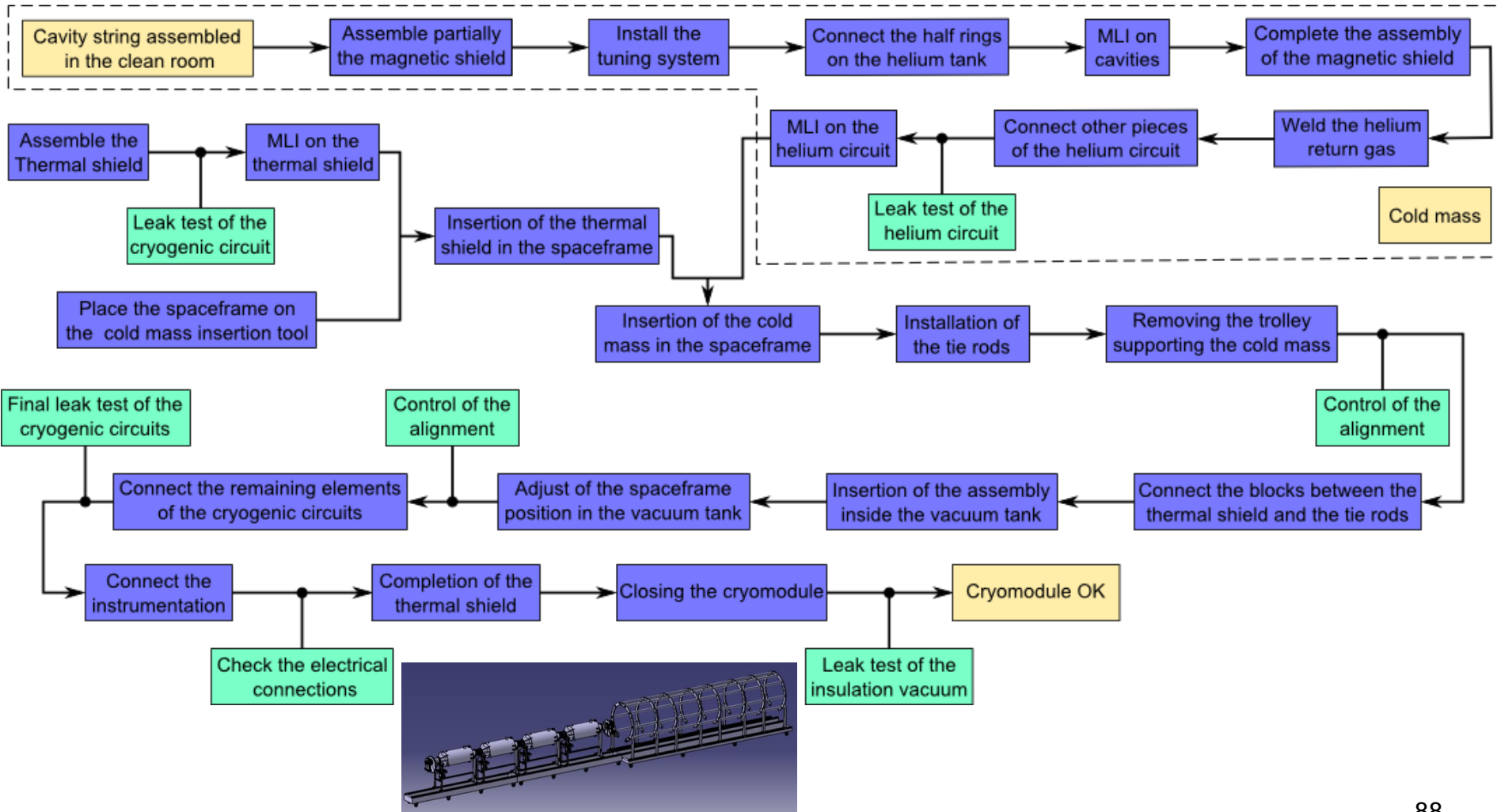
- Components and tools
- Operations
- Controls and tests

Tooling for elliptical cavities:



# Assembly: outside clean room

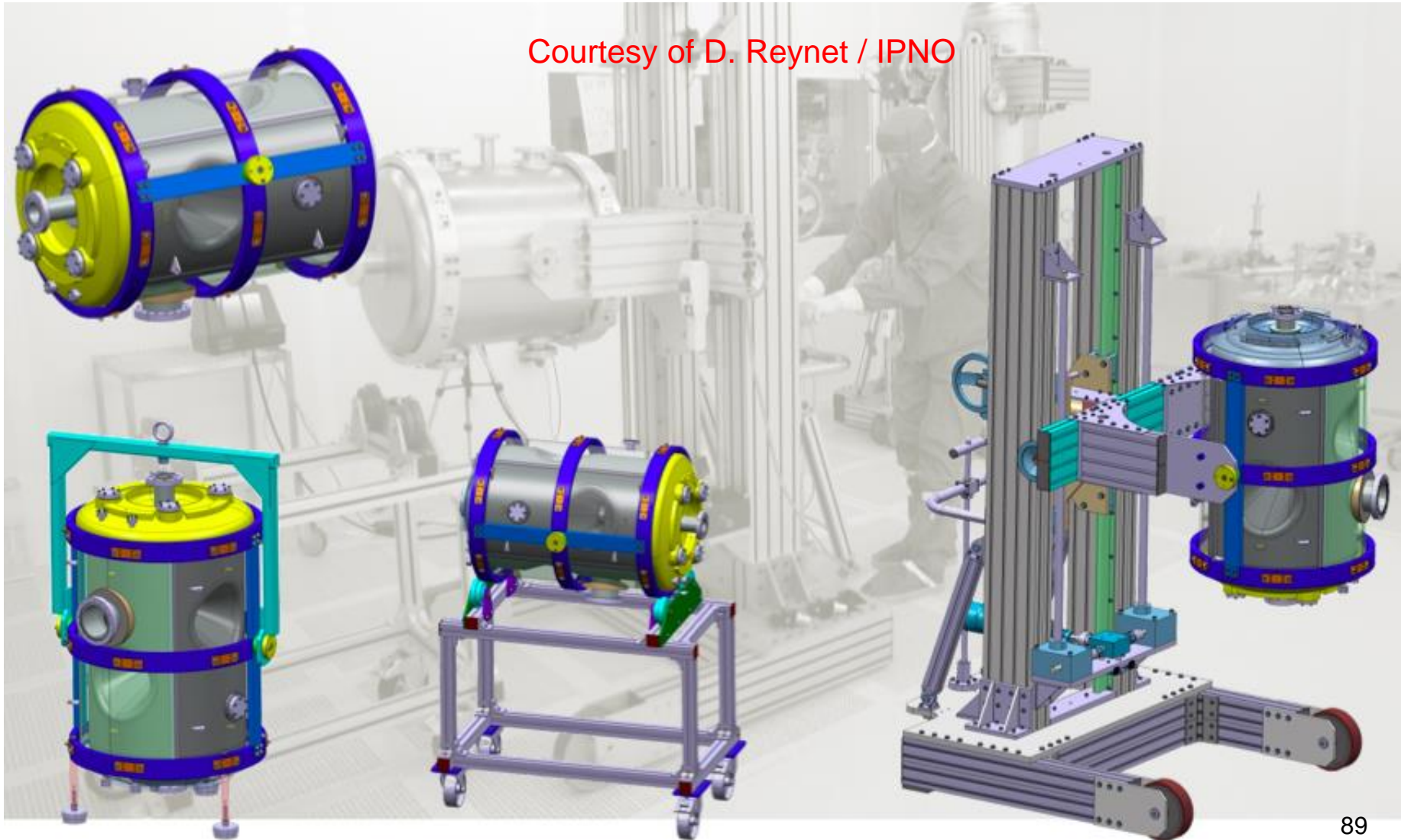
Courtesy of N. Bazin / CEA



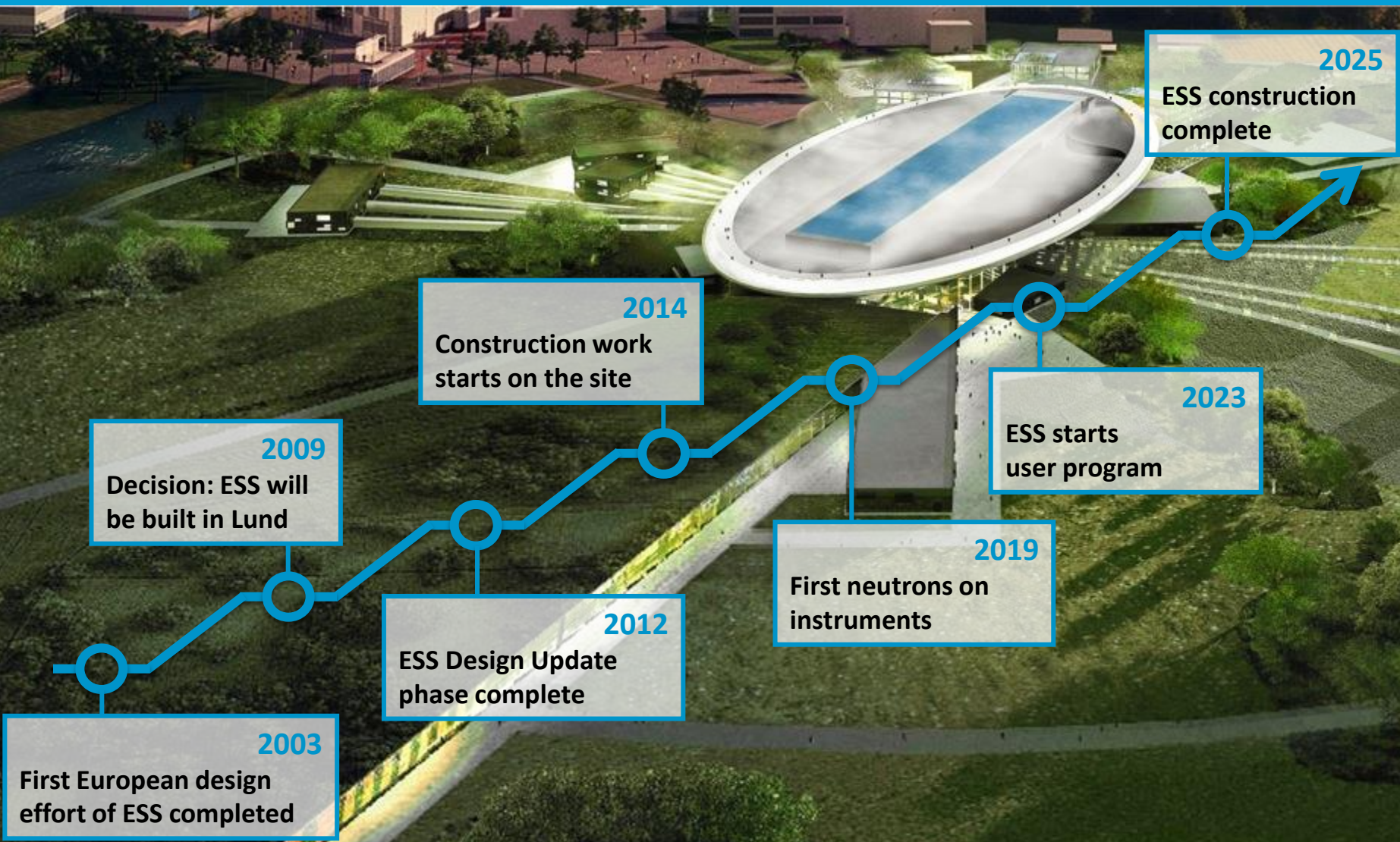


# Spoke assembling in clean room/IPNO

Courtesy of D. Reynet / IPNO



# Road to realizing the world's leading facility for research using neutrons



# EXTRA SLIDES

# Secondary particle produced at J-PARC

