

First observation of new diffusion phenomena in CdTe

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Diffusion in solids

Fick's first law:
$$j(Ag) = -D_{Ag} \frac{\partial [Ag]}{\partial x}$$

Fick's second law:
$$\frac{\partial [Ag]}{\partial t} = -\frac{\partial}{\partial x} j(Ag)$$

Solution for finite source

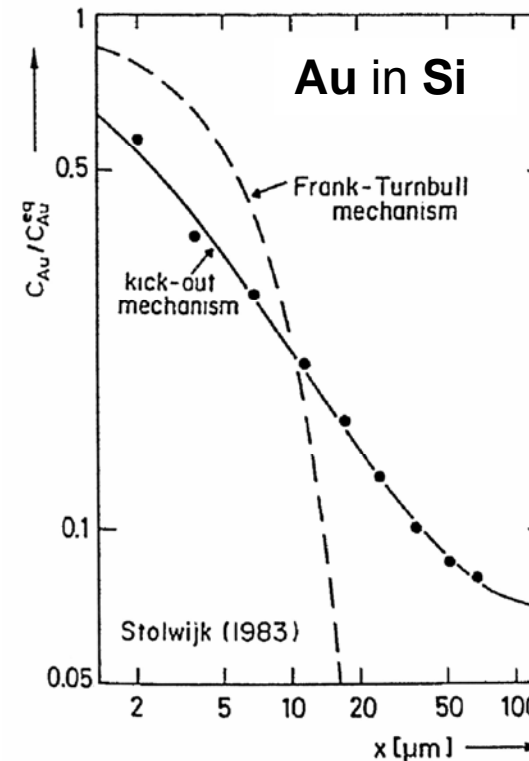
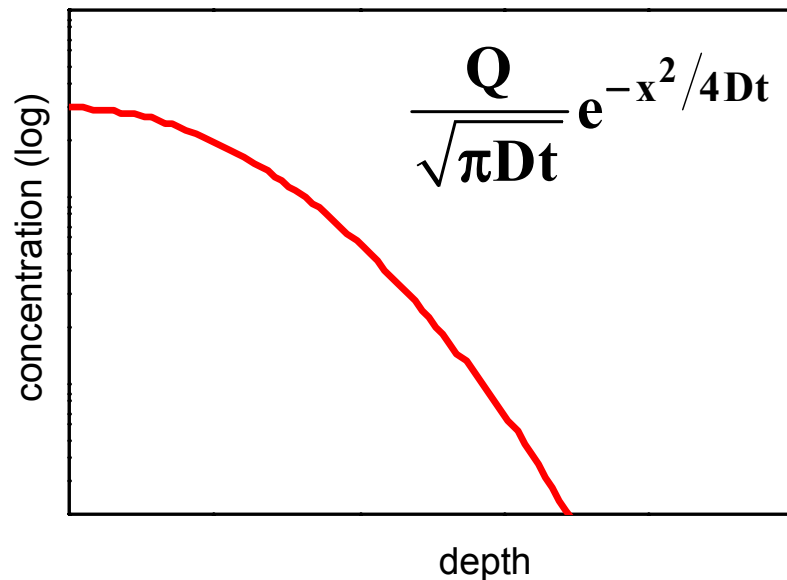
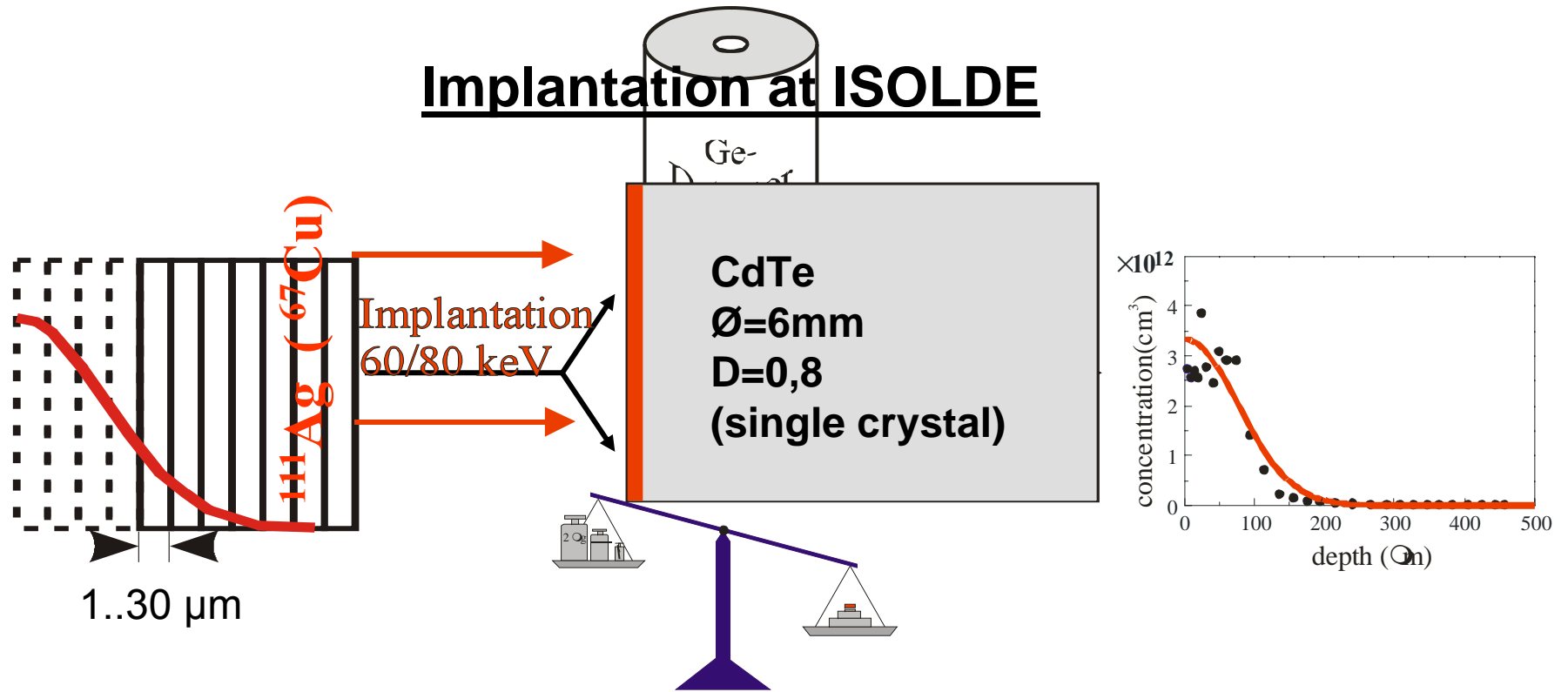


Figure 5-10. Experimental gold concentration profile in dislocation-free silicon (solid circles) compared with predictions of the Frank-Turnbull and the kick-out mechanism (Stolwijk et al., 1983).

N.A. Stolwijk, B. Schuster, J. Hölzl, H. Mehrer and W. Frank, Physica 116 B., 335 – 342, (1983)

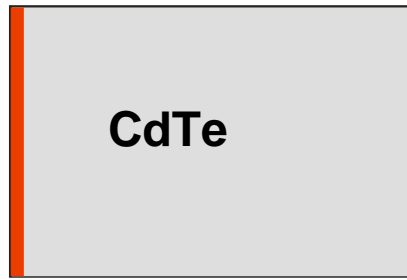
Monotonously decreasing profiles

Experimental details

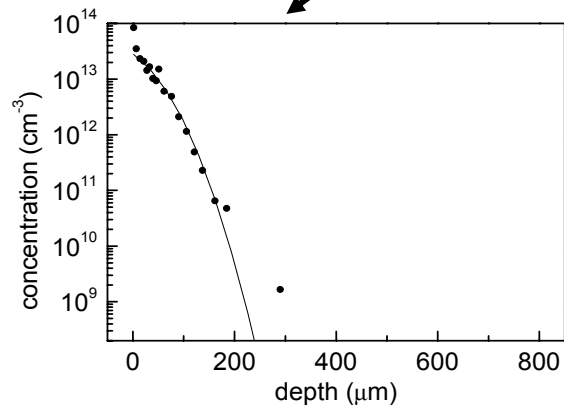


Motivation

¹¹¹Ag

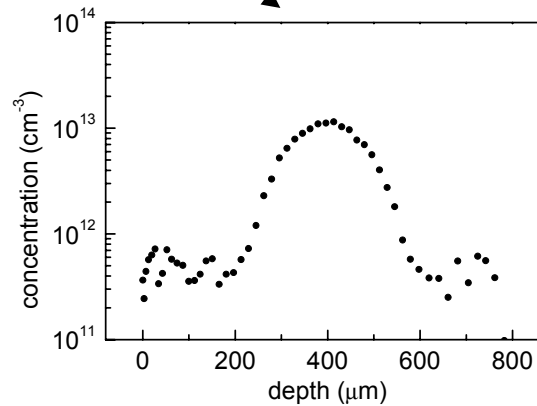


common



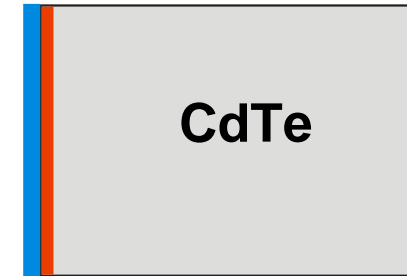
550 K 30 min vac.

unusual



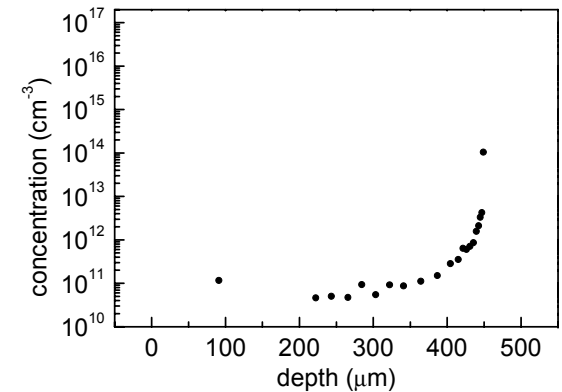
800 K 60 min Cd-atmosphere

¹¹¹Ag



Cu

Cu-film



550 K 30 min vac.

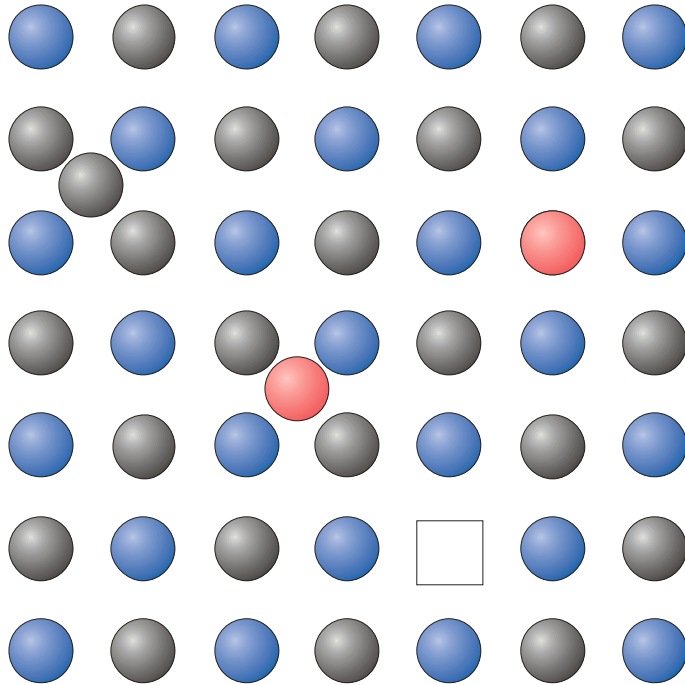
^{*)} Wolf, H., Wagner, F., Wichert, Th., and ISOLDE Collaboration, *Phys. Rev. Lett.* 94, 2005, 125901

Unexpected, new profile forms ^{*)}



Diffusion of Ag under Cd-atmosphere

Defects in CdTe



● Extrinsic defects

Ag: Ag_{Cd}
 Ag_i

● Intrinsic defects

Cd: Cd_i
 V_{Cd}

● Te sub lattice: perfect

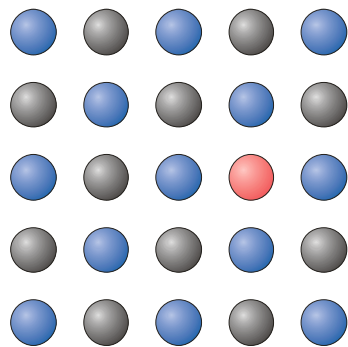
$[Cd_i] - [V_{Cd}]$: *deviation from stoichiometry*

reflected by: $\frac{[Cd_i]}{[V_{Cd}]}$

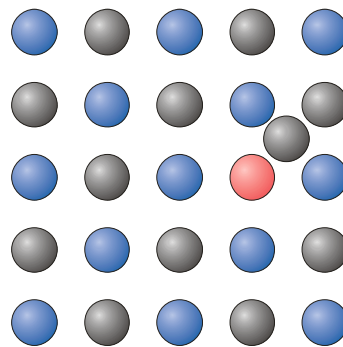
Defect reaction and thermal equilibrium

Ag in CdTe

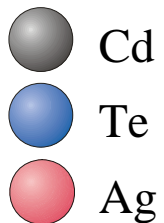
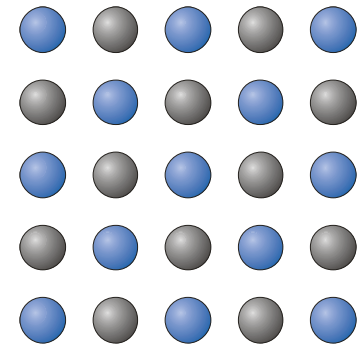
dissociative



kick-out



annihilation

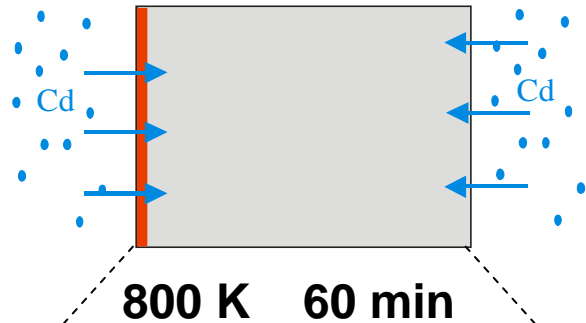


local equilibrium

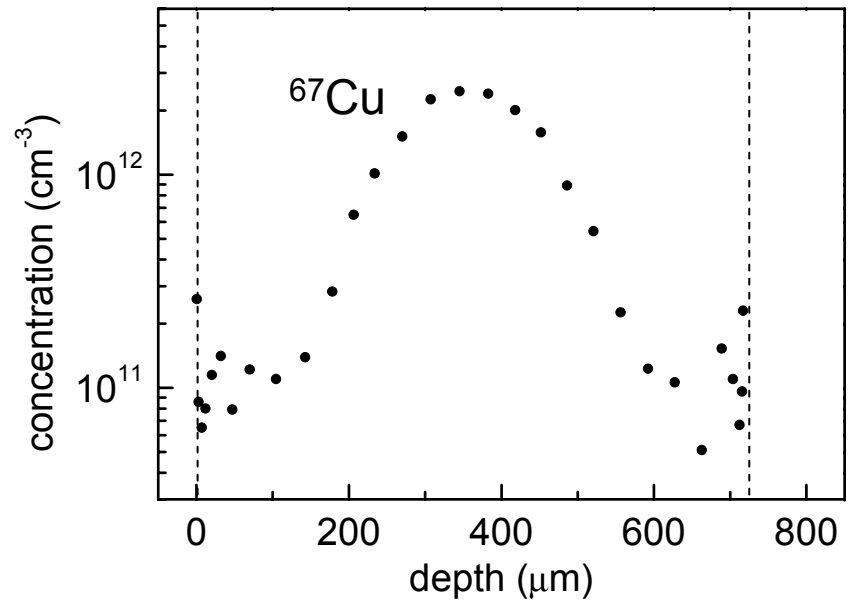
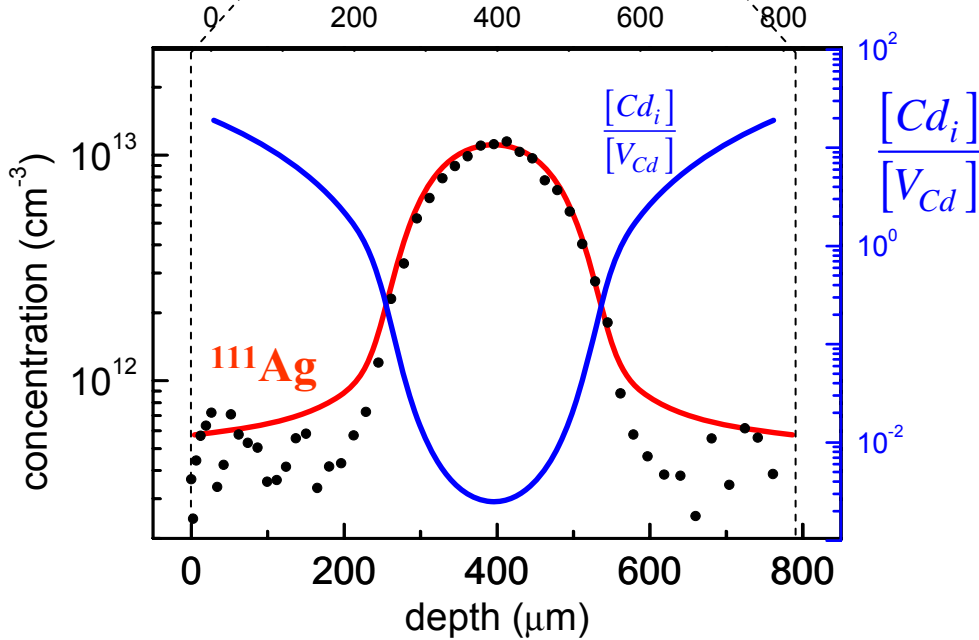
$$\frac{[\text{Ag}_i]}{[\text{Ag}_{\text{Cd}}]} = K \sqrt{\frac{[\text{Cd}_i]}{[\text{V}_{\text{Cd}}]}}$$

Changes in the stoichiometry

Ag in CdTe: Cd-atmosphere



- high mobility of Ag_i (Cu_i)
- low mobility of Ag_{Cd} (Cu_{Cd})



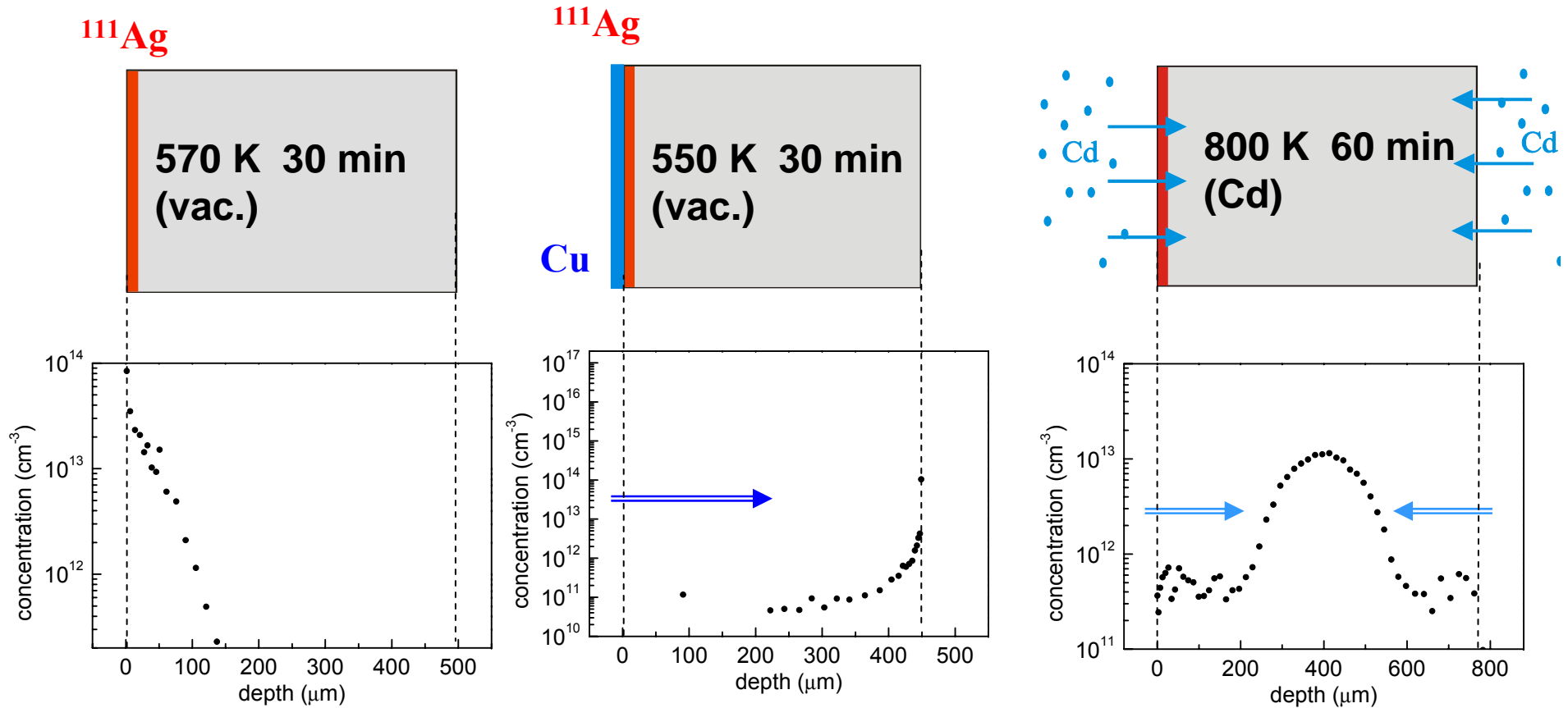
Ag-profile reflects the stoichiometry of the crystal

- ▶ Unusual diffusion profiles observed in CdTe
- ▶ Strong influence of the deviation from stoichiometry
- ▶ Model based on defect reactions



Codiffusion of Ag and Cu

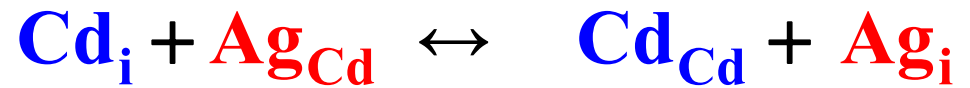
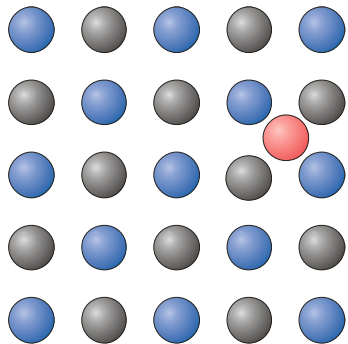
Codiffusion of Ag with Cu in CdTe



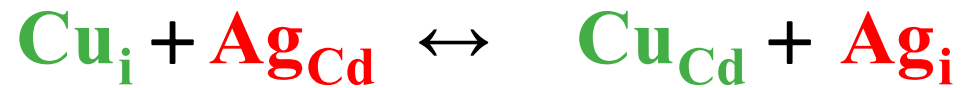
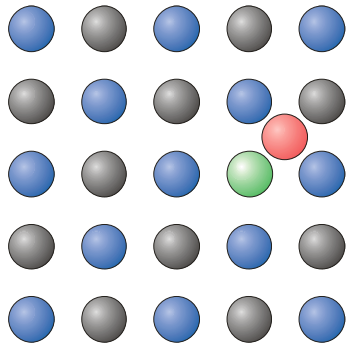
Similarities between codiffusion of Cu and diffusion under Cd-atmosphere

Codiffusion: Important defect reactions

kick out

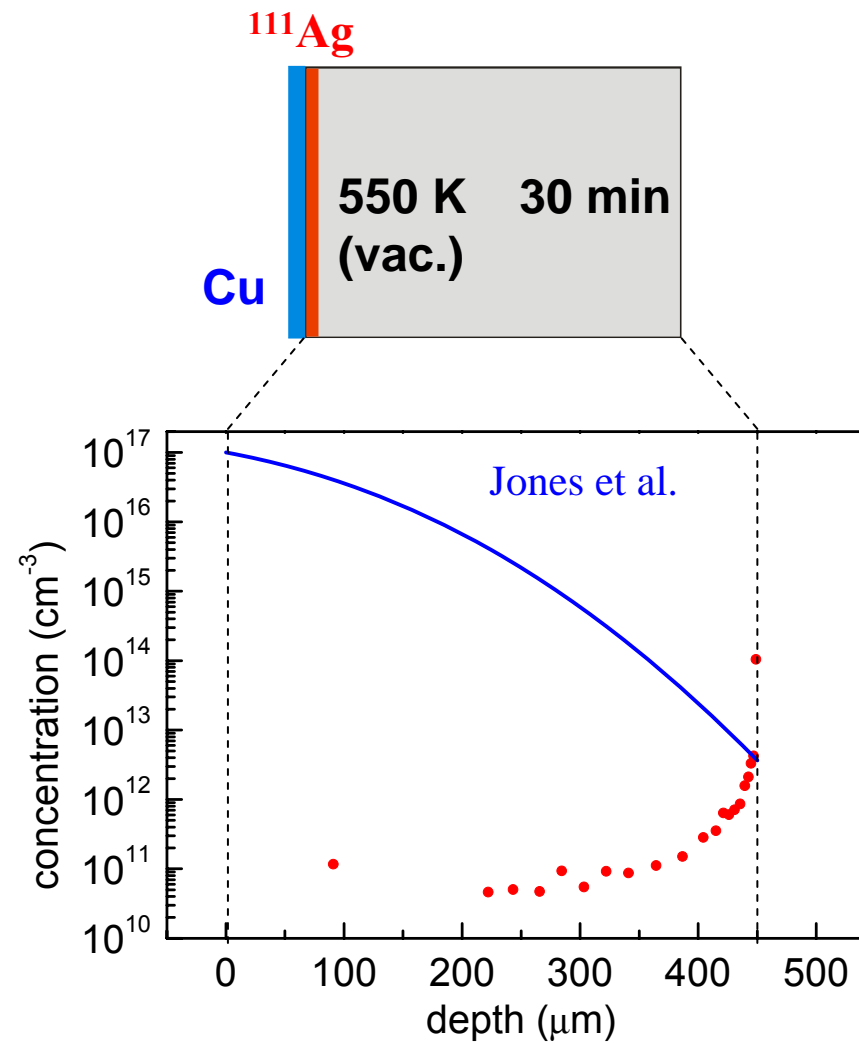


codiffusion



Analogous behavior of Cd_i and Cu_i

Codiffusion of Ag und Cu

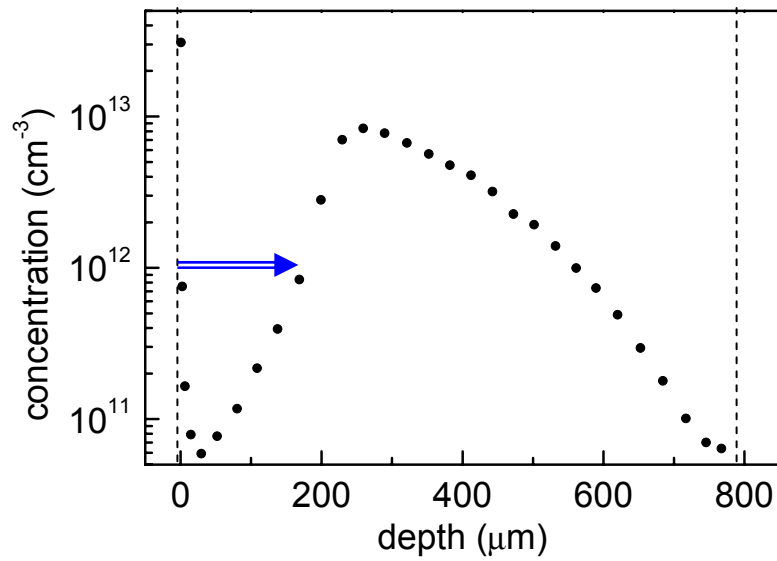
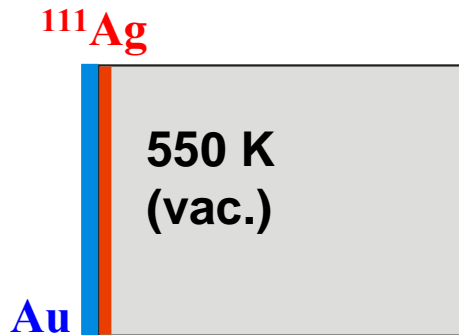


E.D.Jones, N. M. Stewart, J. B. Mullin, J.Crystal Growth 117, 244 (1992)

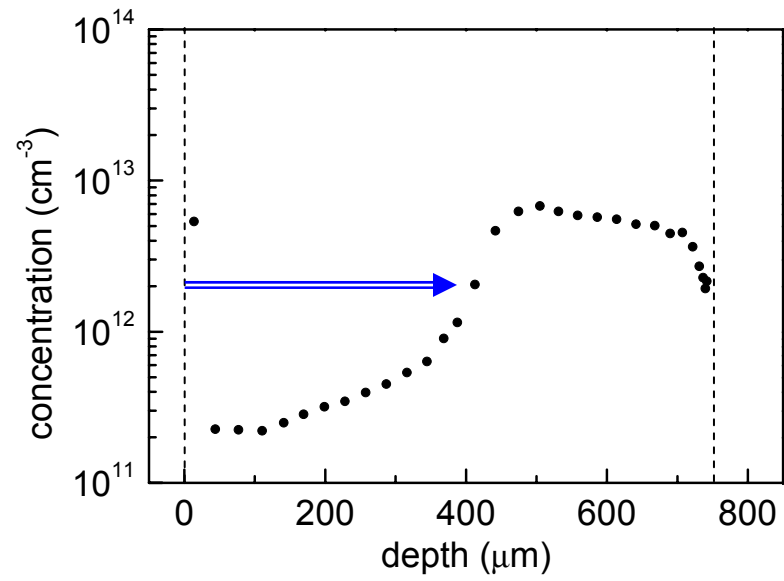
Ag-profile reflects the Cu distribution

Codiffusion of Ag and Au

Codiffusion Ag and Au



30 min



120 min

Time

Similar effect to Cu-codiffusion

Summary

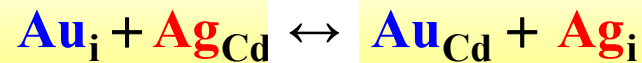
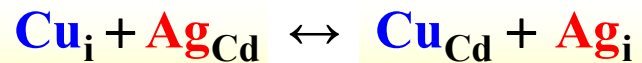
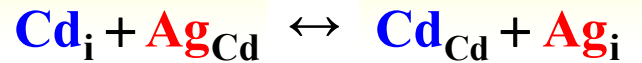
Ag diffusion under Cd atmosphere

- uncommon profile forms
- distribution of Ag reflects the sample stoichiometry

Codiffusion of Ag and Cu or Au

- replacement of Ag by Cu and Au

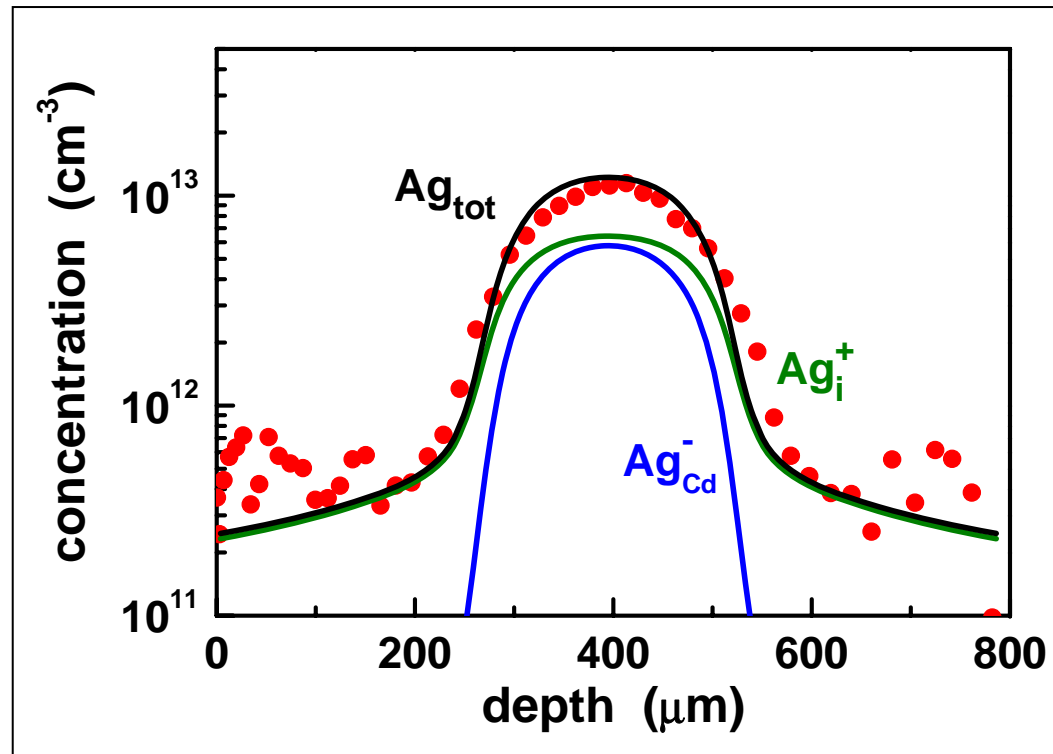
Modell based on defect reaction



State of the art

Charged defects:

- donor: Ag_i^+ Cd_i^+
- acceptor: Ag_{Cd}^- V_{Cd}^-



Outlook: Matrix

Matrix:

- CdTe
- **II-VI Semiconductors**
- III-V semi conductors

