

## Phase Transition of $\text{LiMn}_{0.85}\text{Cr}_{0.15}\text{PO}_4$ Cathode Material by In-Situ Time-Resolved XANES

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Lithium metal phosphate olivine ( $\text{LiMPO}_4$ ; M= Fe, Mn, Co, Ni) have a great deal of attention as one of the promising cathode material for lithium ion batteries. To date, a considerable number of studies have enhanced the electrochemical behavior of  $\text{LiFePO}_4$  from being barely electrochemically active to having a full capacity at high rates. Based on the success of  $\text{LiFePO}_4$ , an increasing number of research groups have focused their attention on  $\text{LiMnPO}_4$ , which exhibits an obvious advantage over  $\text{LiFePO}_4$  with a redox potential of 4.1 V VS  $\text{Li/Li}^+$ . However, the  $\text{LiMnPO}_4$  kinetics is unusually sluggish due to its intrinsically low ionic and electronic conductivity. Many techniques, including to carbon coating, nano-sized using and aliovalent doping have been done to improve rate capability of this material. The doping of  $\text{LiFePO}_4$  with  $\text{Cr}^{3+}$  has been investigated in several previous studies which show an enhancement in conductivity and rate performance. However, the Cr-associated mechanism during charge/discharge is not yet revealed. Here, we report phase transition investigation of  $\text{LiMn}_{0.85}\text{Cr}_{0.15}\text{PO}_4$  cathode material by in-situ time-resolved XANES.

### Summary

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