

Laser Applications at Accelerators Conference 2015



**Research and development of
photocathodes sensitive to visible laser
beam for photoinjector applications.**

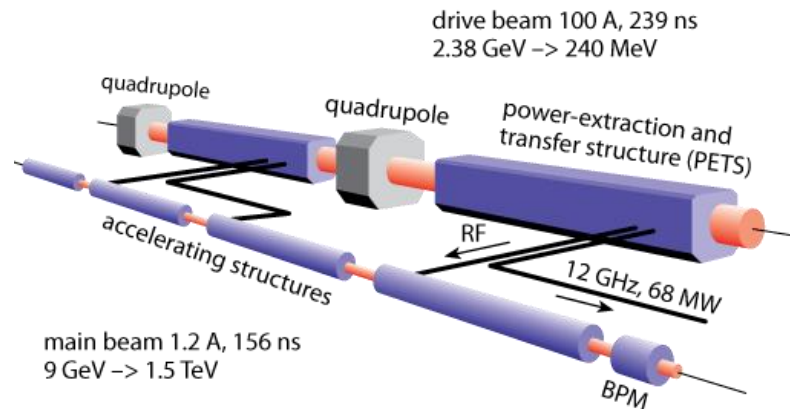


Irene Martini



Mallorca, 25.03.2015

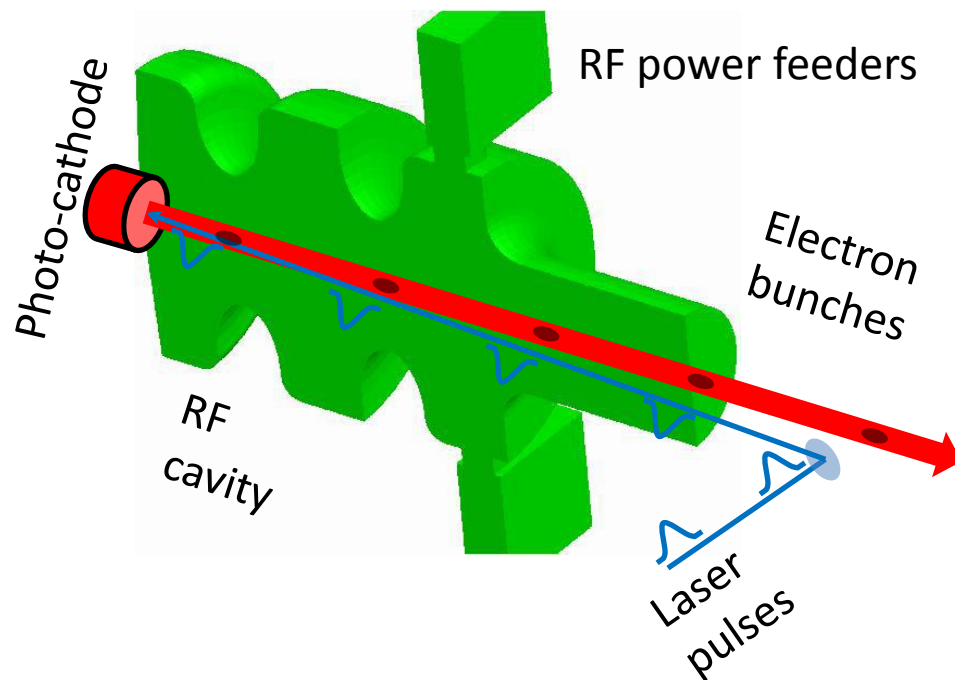
- **CLIC (Compact Linear Collider)** is a study for a future electron-positron room temperature linear collider (~50 km long) at CERN.
- Linear leptons collider for a fine characterization of LHC (Large Hadron Collider) discoveries.
- Two beam-acceleration method: a **drive beam** to supply with the RF power the **main beam** accelerating structures.



<http://cllc-study.org/>

- Drive beam: high peak and average current \Rightarrow challenging requirements to the source of electrons.

- The **PHIN RF photoinjector** is used at CERN for R&D in the framework of feasibility studies of the photoinjector option for the **CLIC drive beam**.



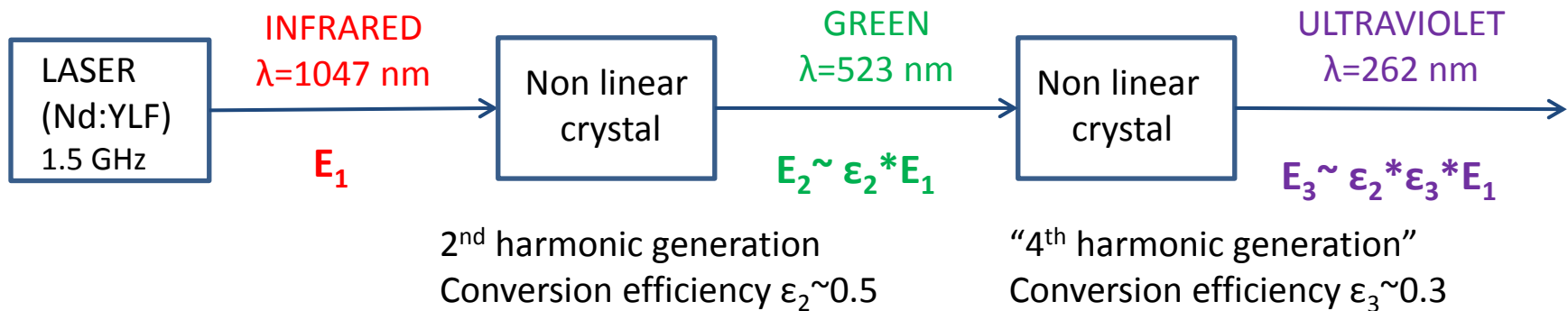
Parameter	CLIC	PHIN
Charge/bunch (nC)	8.4	2.3
Bunch length (ps)	10	10
Bunch rep. rate (GHz)	0.5	1.5
Number of bunches	70000	1800
Train length (μ s)	140	1.2
Charge/train (μ C)	590	4.1
Macro pulse rep. rate (Hz)	50	5
Beam current/train (A)	4.2	3.4
Cathode lifetime (h) at QE>3% (Cs_2Te)	>150	>50
QE>0.5% (Cs_3Sb)		

- Long cathode lifetimes with high bunch charge, long trains and high rep rate.**

- In parallel to the the development of a laser system for the CLIC paramenters, photocathode R&D studies are on-going.
- Photocathodes sensitive to ultra-violet **UV light** (such Cs₂Te) are used in many photoinjector facilities.
- The available laser pulse energy in UV for 0.14 ms long pulse trains is currently limited due to a degradation of the beam quality during the 4th harmonics frequency conversion process. Using **visible laser beams** would overcome this limitation.

NEED of
PHOTOCATHODE with
high QE for green laser
light.

Cs₃Sb



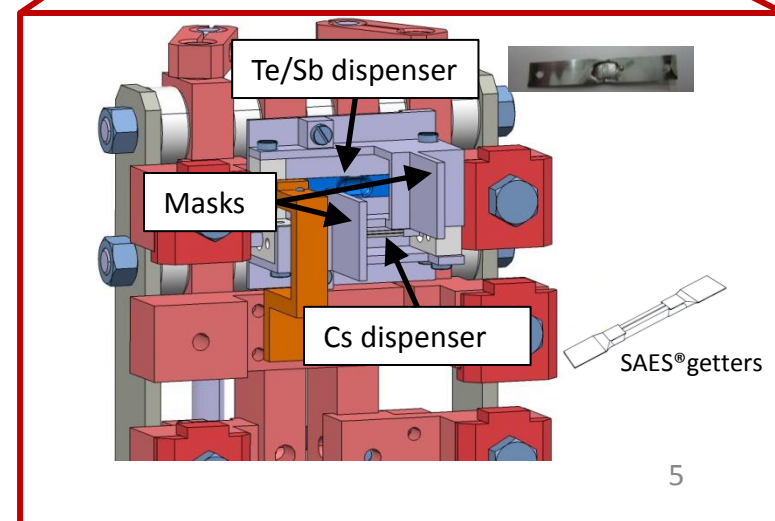
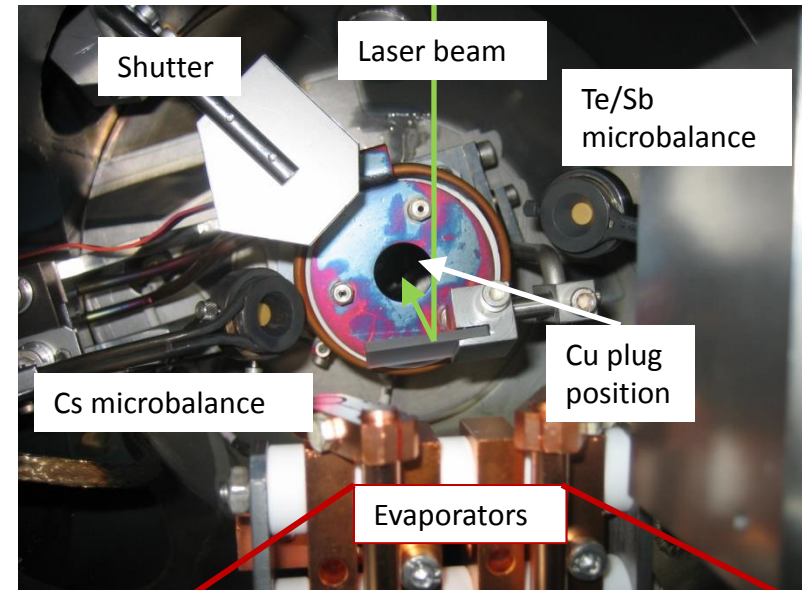
- **CO-DEPOSITION:**

Cs and Te or Sb evaporated at the same time: the metallic elements can mix together in the vapour phase.

- The **Copper** substrate is **heated** up to 125° C.



- **On-line QE measurement** shooting with laser: the evaporators power is controlled in order to reach a maximum in the QE.



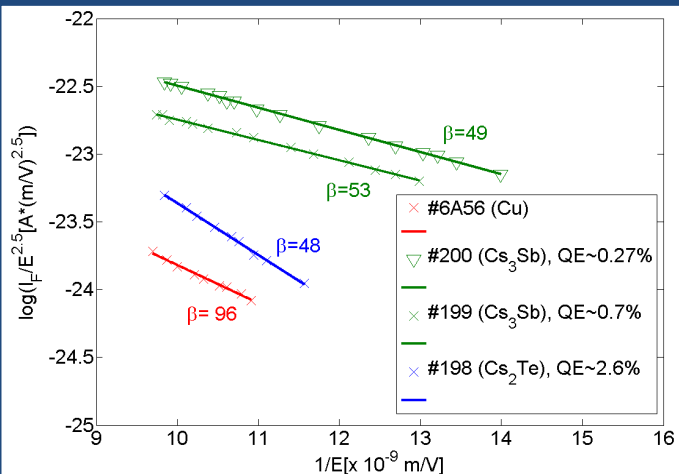
Cathodes feasibility studies:

- Testing the material performance in a real photoinjector setup.
- Understanding the main factor affecting cathode performance (production/operation).
- Improve the cathode performance acting on both the setup or the cathode itself.

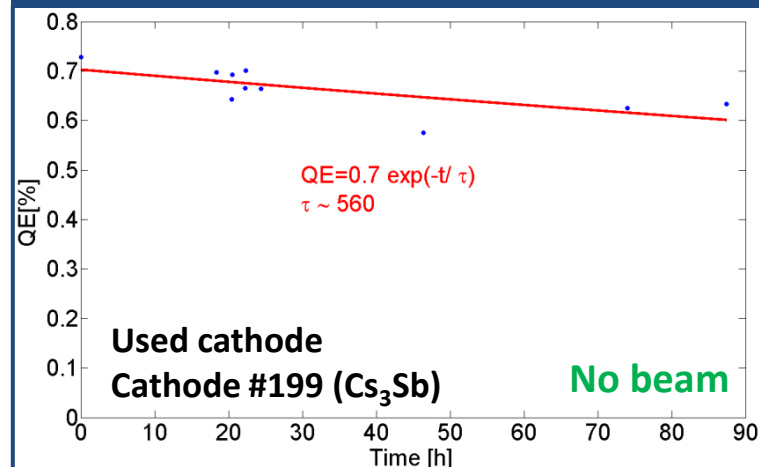
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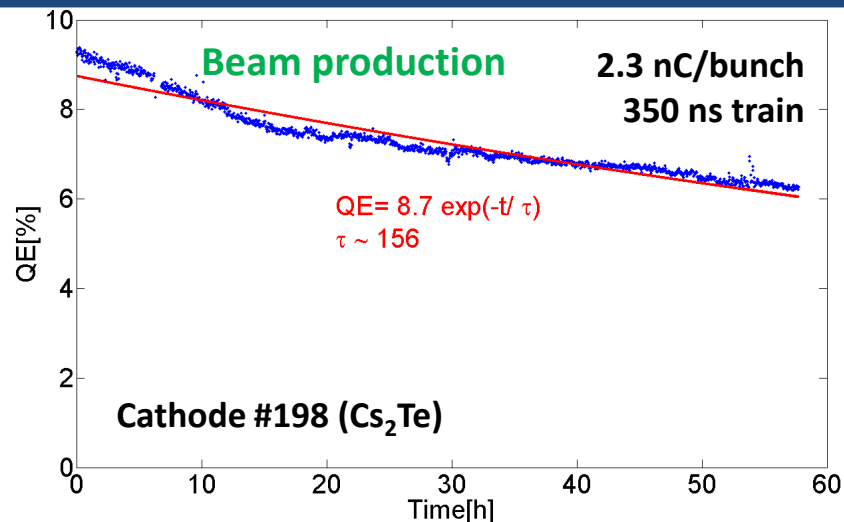
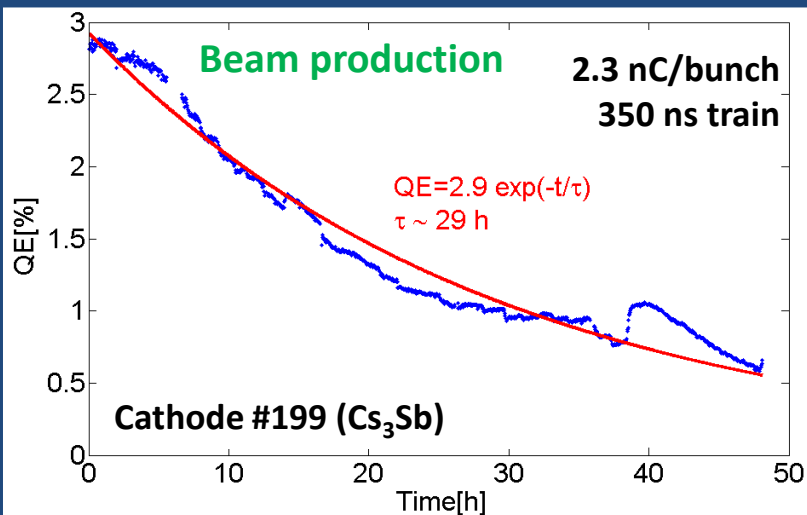
Dark current measurements



RF lifetime measurements



Operational lifetime measurements



Cathodes feasibility studies:

- Testing the material performance in a real photoinjector setup.
- **Understanding the main factor affecting cathode performance.**
- Improve the cathode performance.

CHEMICAL POISONING (Residual Gas Contamination, mainly oxidation)

Static vacuum level (Residual gases)

Electrons (from Beam Losses)
Stimulated Desorption (**Dynamic vacuum level**)

Electrons (from Field Emission)
Stimulated Desorption (**Dynamic vacuum level**)

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RADIATION DAMAGE

Ions (from beam interaction with residual gases) **Back Bombardment**

Independent of beam current

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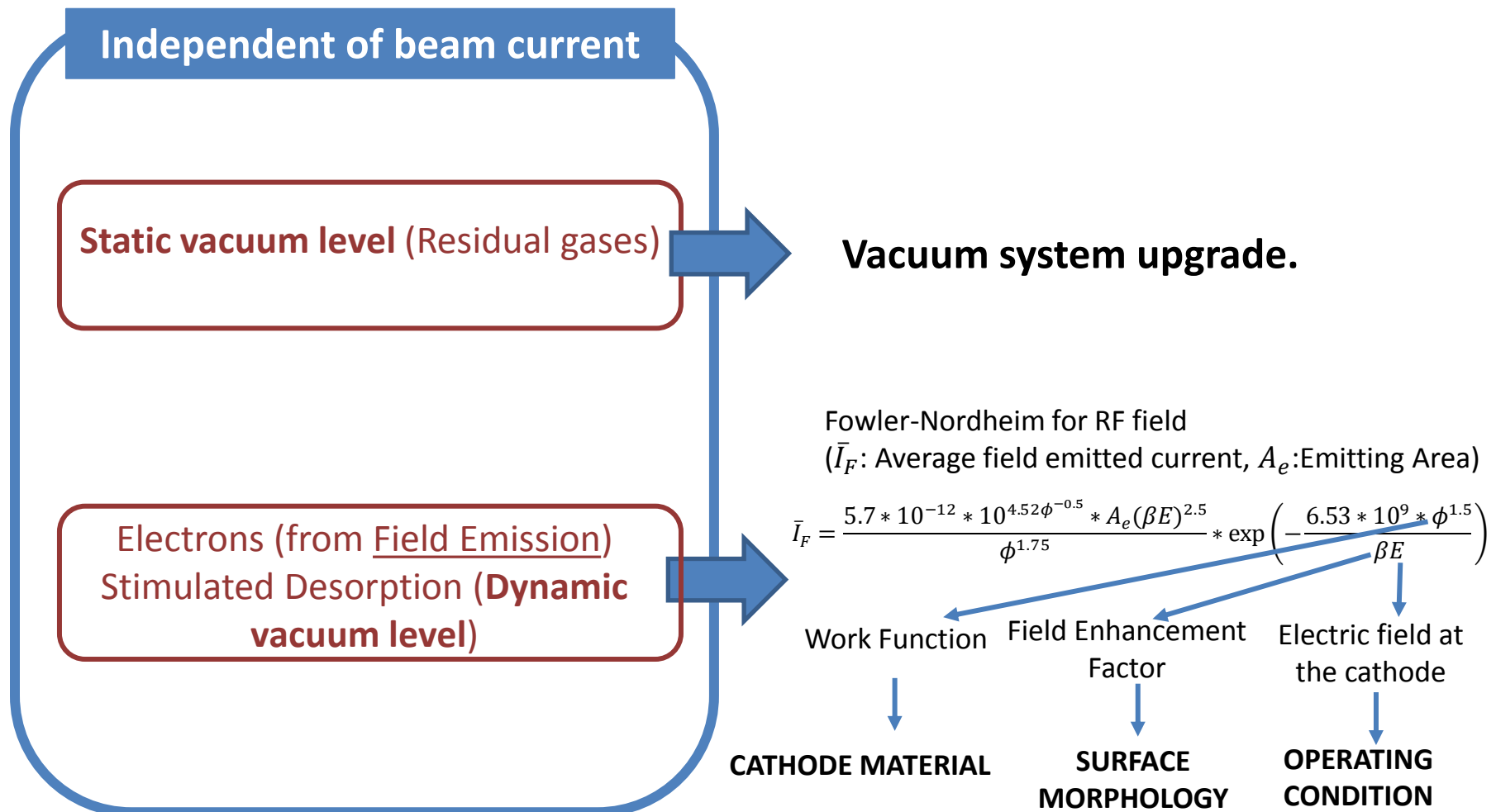
Proportional to beam current

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- **Improve the cathode performance acting on the setup or on the cathode itself.**



Independent of beam current

Static vacuum level (Residual gases)

Electrons (from Field Emission)
Stimulated Desorption (**Dynamic vacuum level**)


Proportional to beam current

Electrons (from Beam Losses)
Stimulated Desorption (**Dynamic vacuum level**)

DEPENDENT ON DESIGN PARAMETERS

Ions (from beam interaction with residual gases) **Back Bombardment**

Cathodes feasibility studies:

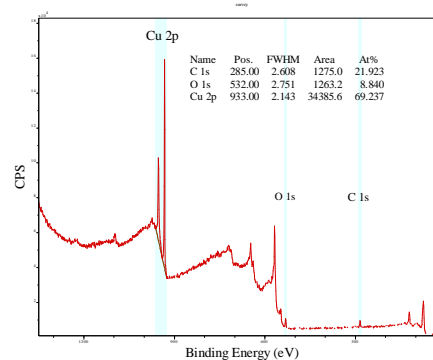
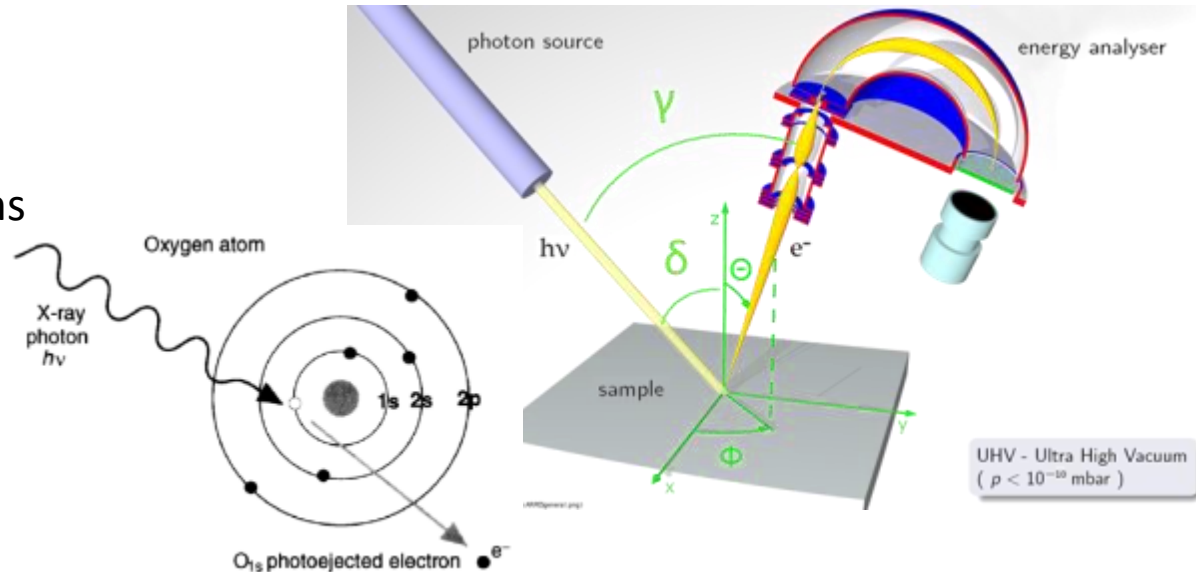
- Testing the material performance in a real photoinjector setup.
- **Understanding the main factor affecting cathode performance.**  **MATERIAL CHARACTERIZATION**
- Improve the cathode performance.

Surface Analysis: X-Rays Photoemission Spectroscopy (XPS)

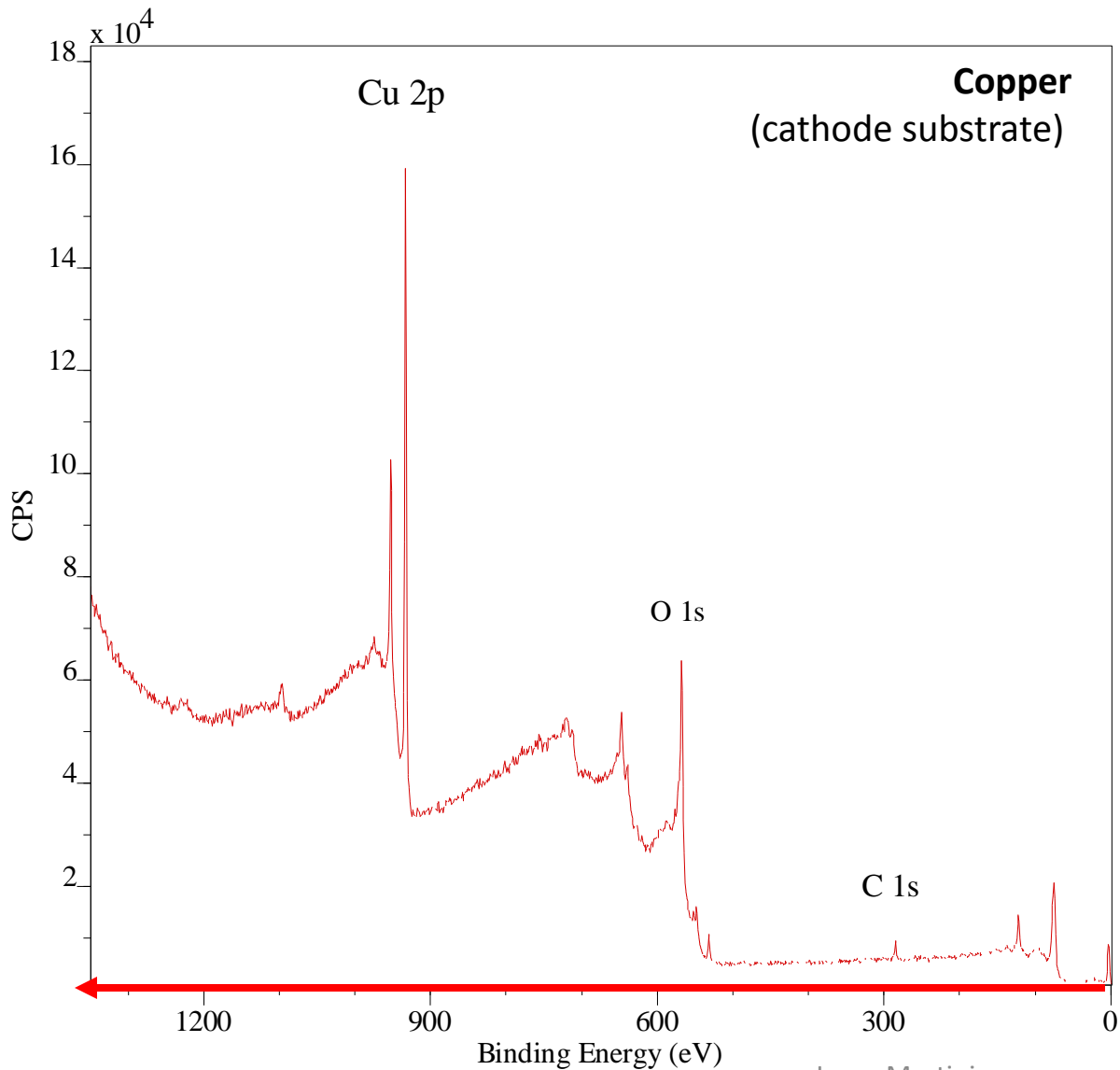
- XPS is a surface analysis technique ($\sim \text{nm}$) able to detect all elements (except H, He) and to distinguish different chemical bonding states.

FUNDAMENTALS OF THE XPS:

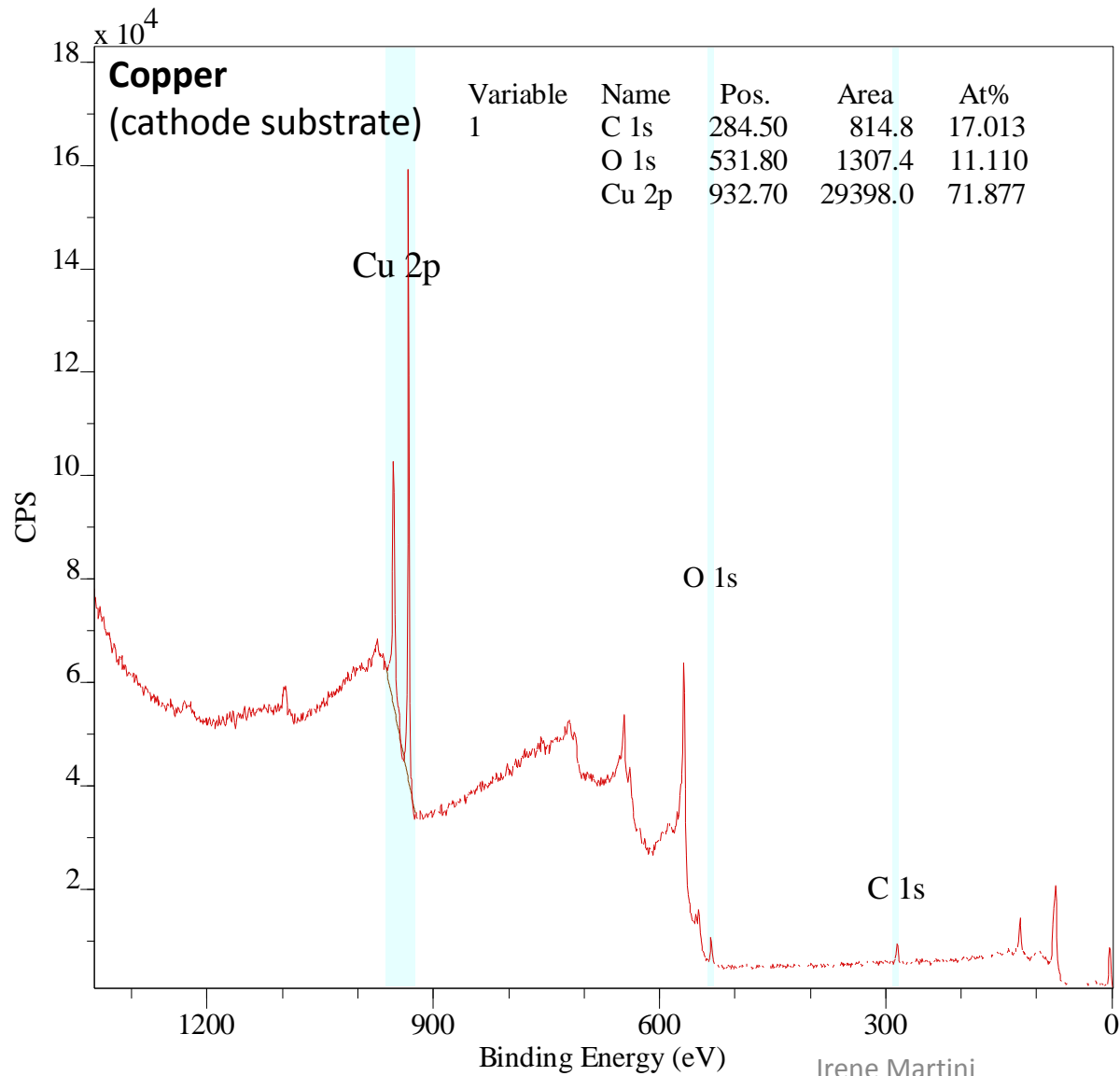
- Photons (from **x-rays gun**) interact with the core electrons of the sample atoms creating ionized states.
- The Kinetic Energy of the subsequently photoemitted electrons is measured thanks to the **Energy Analyser**.
- The measured spectrum is a direct indication of the Binding Energy of the different atomic levels.



The spectrum is the FINGER PRINT of the material in that specific chemical environment!



Qualitative analysis:
**ELEMENTAL
IDENTIFICATION**



Qualitative analysis:
**ELEMENTAL
IDENTIFICATION**

Quantitative analysis:
RELATIVE COMPOSITION
of the surface.

$$I_A = N_A * \sigma_A * \lambda_A * K$$

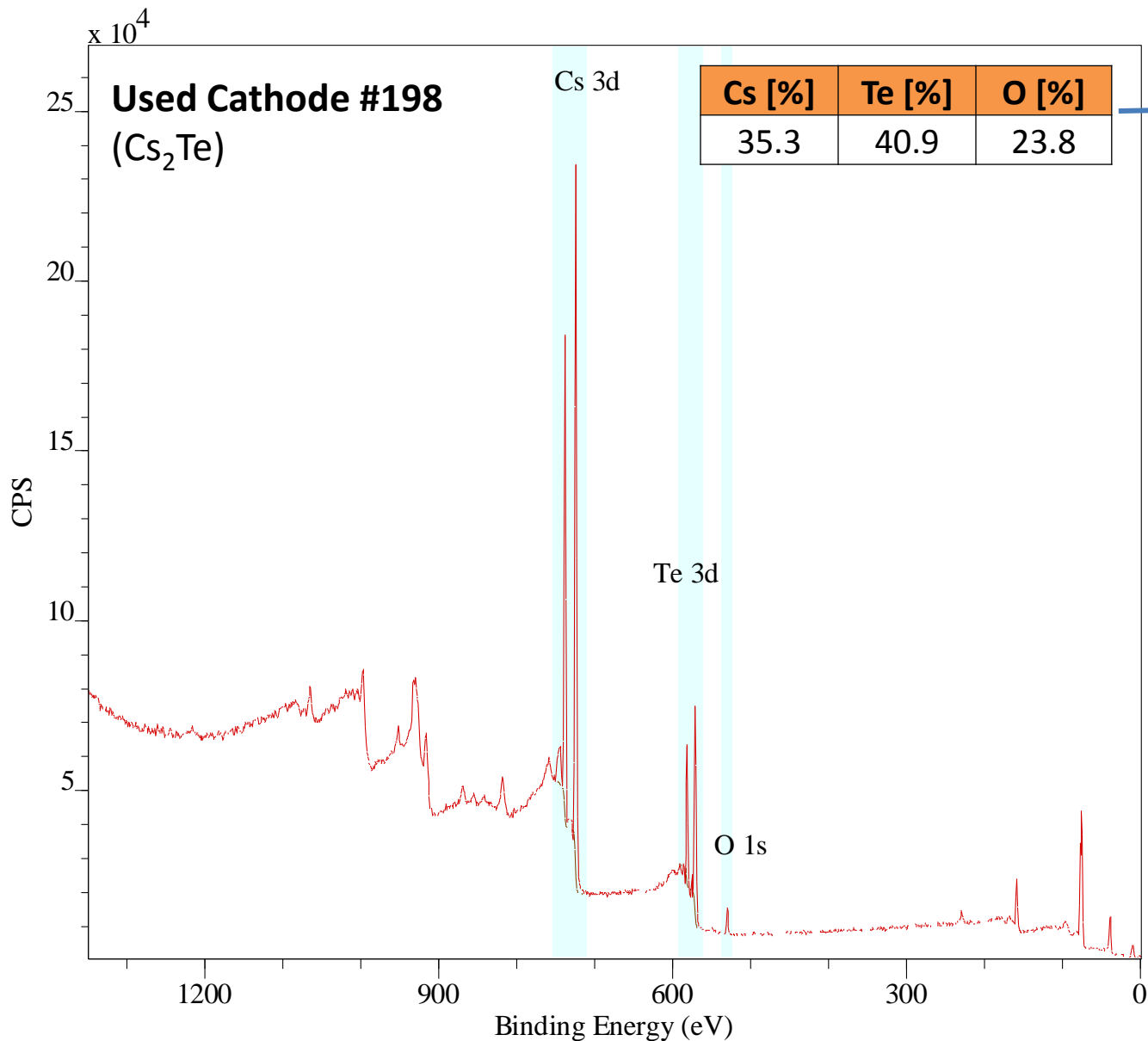
I_A : peak intensity

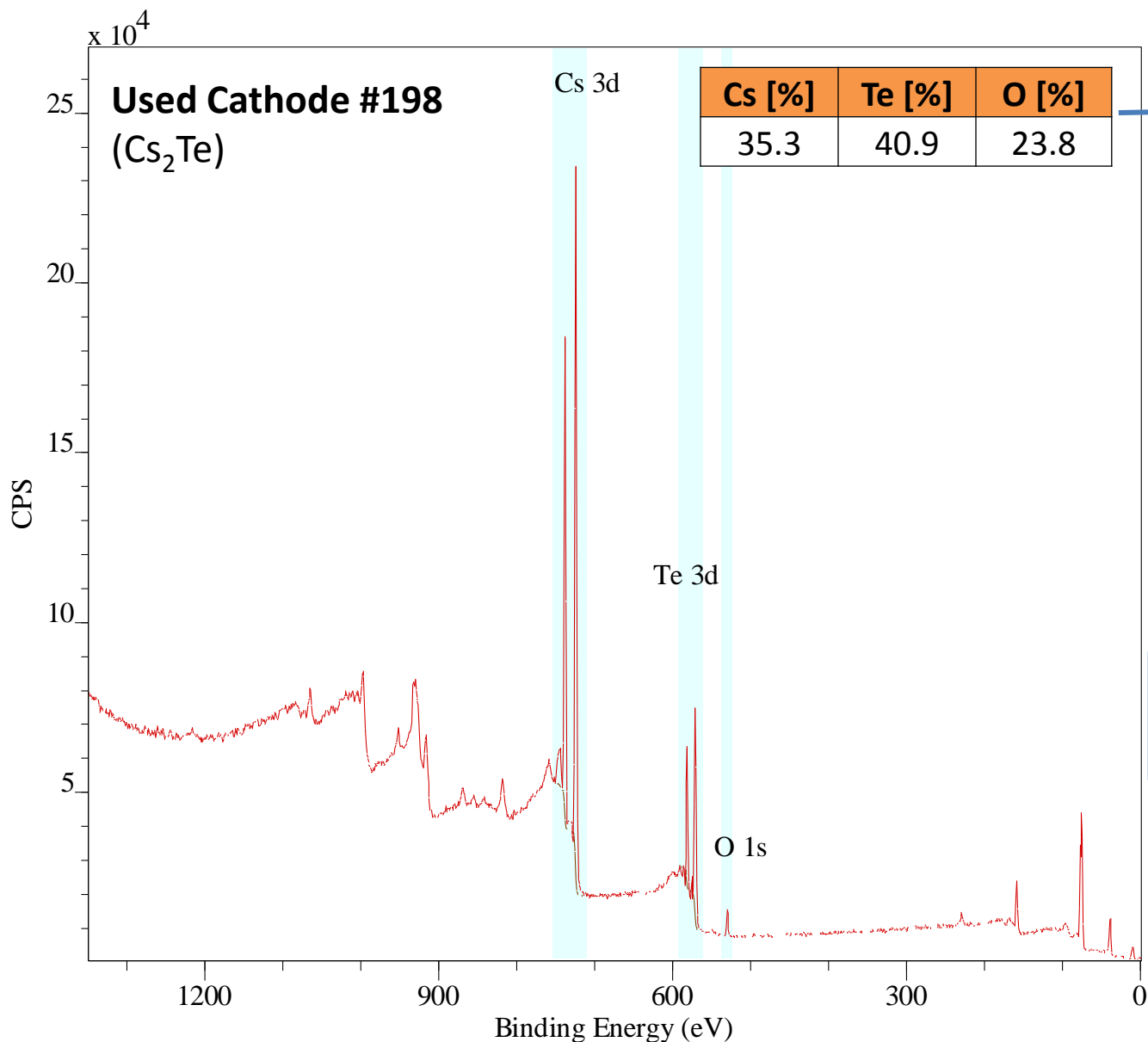
N_A : concentration of element A

σ_A : photoionization cross-section

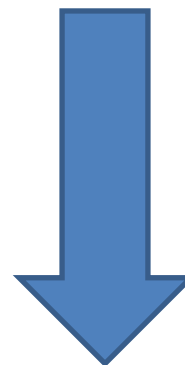
λ_A : mean free path

K: all setup related factors



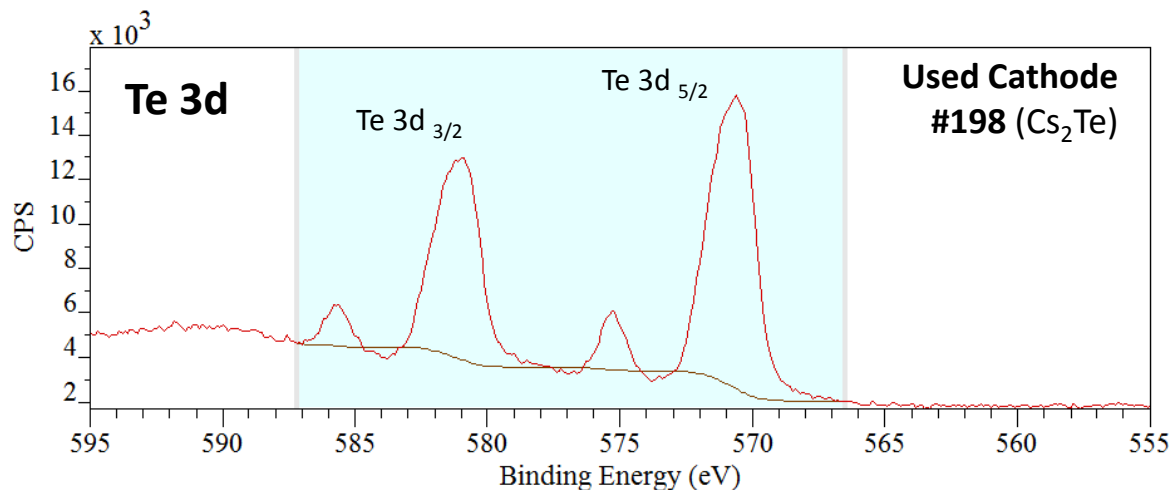


Elemental composition of the surface: lack of Cs compared to the compound Cs_2Te .



CHEMICAL STATES IDENTIFICATION

High Resolution Spectra of the highest peak



CHEMICAL STATES IDENTIFICATION

High Resolution Spectra

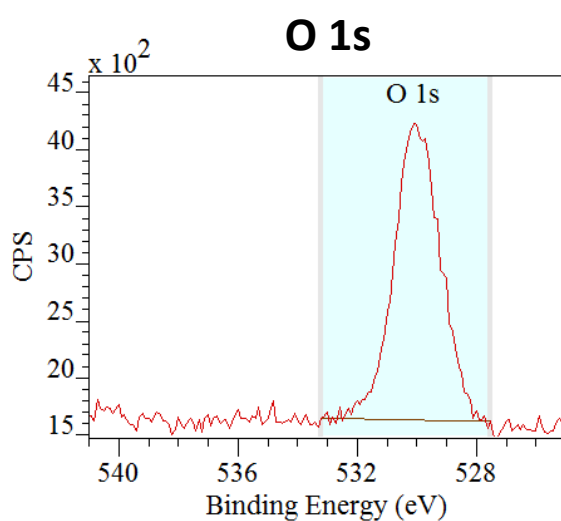
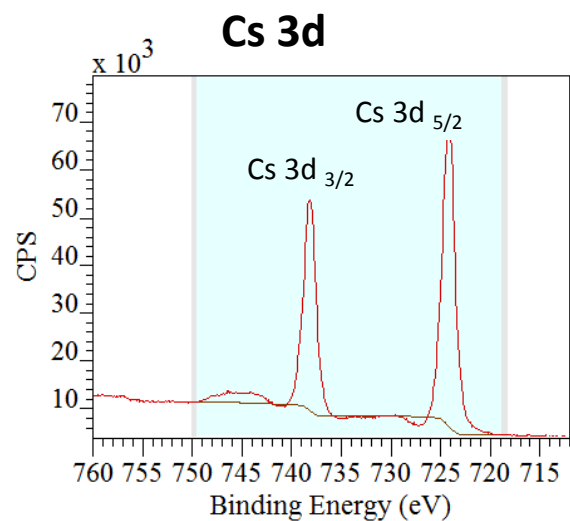
CHEMICAL SHIFT

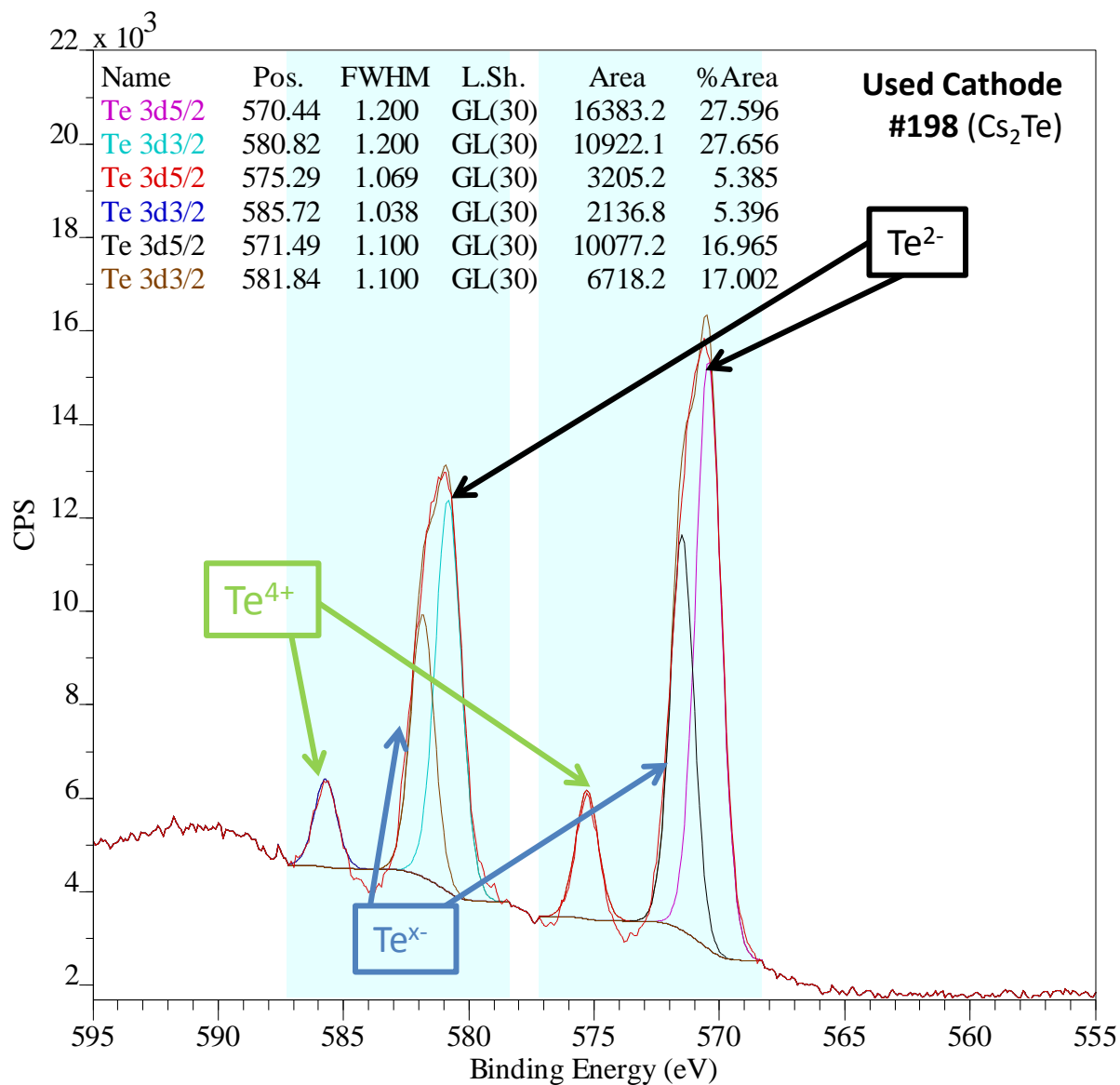
The peak energy position is related to the chemical state of the element.

PEAKS FITTING: different compounds of the same element

Peak position (slightly) and FWHM depends on the measurement setup

LITERATURE +
STANDARD SAMPLES



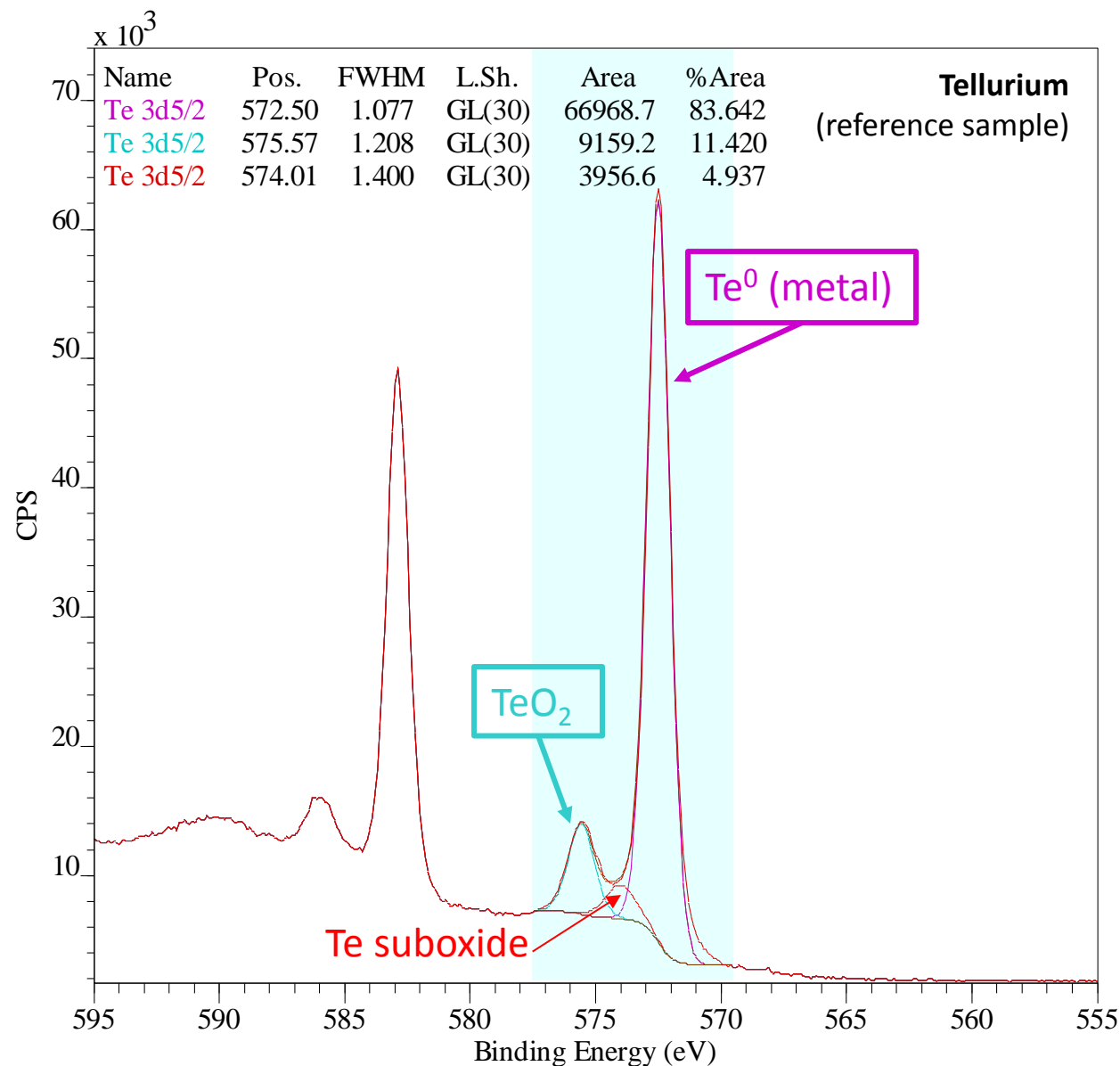


Te 3d region

PEAKS MODEL MUST HAVE PHYSICAL REASON.

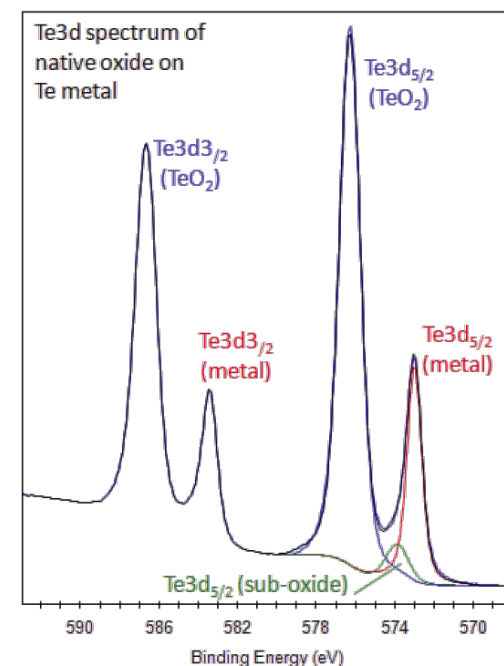
Possible explanation:

- Cs_2Te (Te^{2-})
- Cs_xTe (Te-rich phase)
- TeO_2 (Te^{4+})



Te 3d region

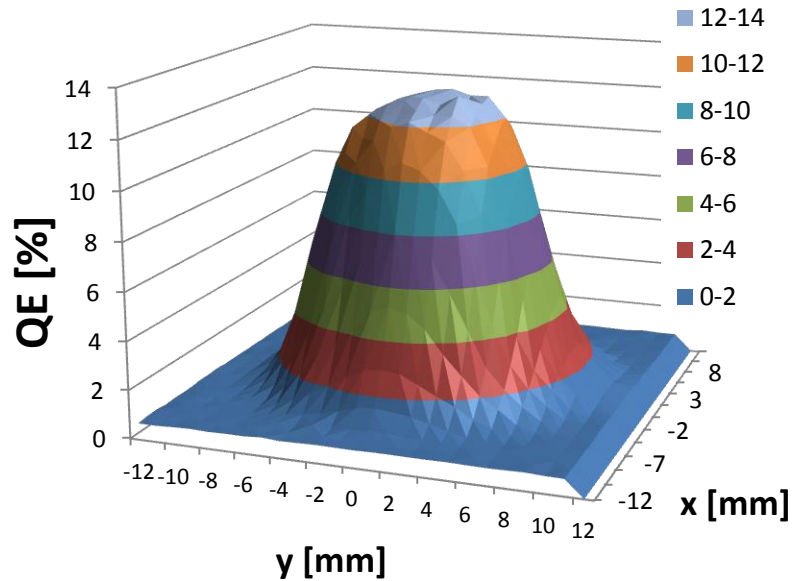
Literature:



<http://xpssimplified.com/elements/tellurium.php>

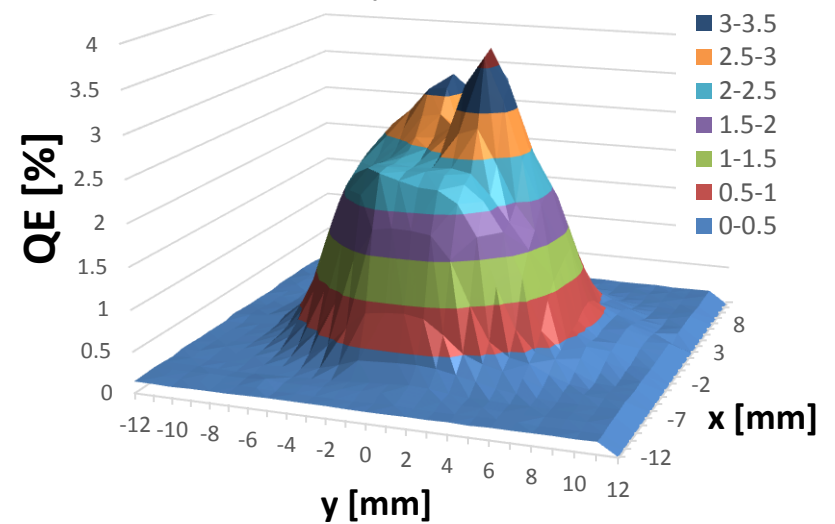
Cathode # 198 (Cs_2Te)

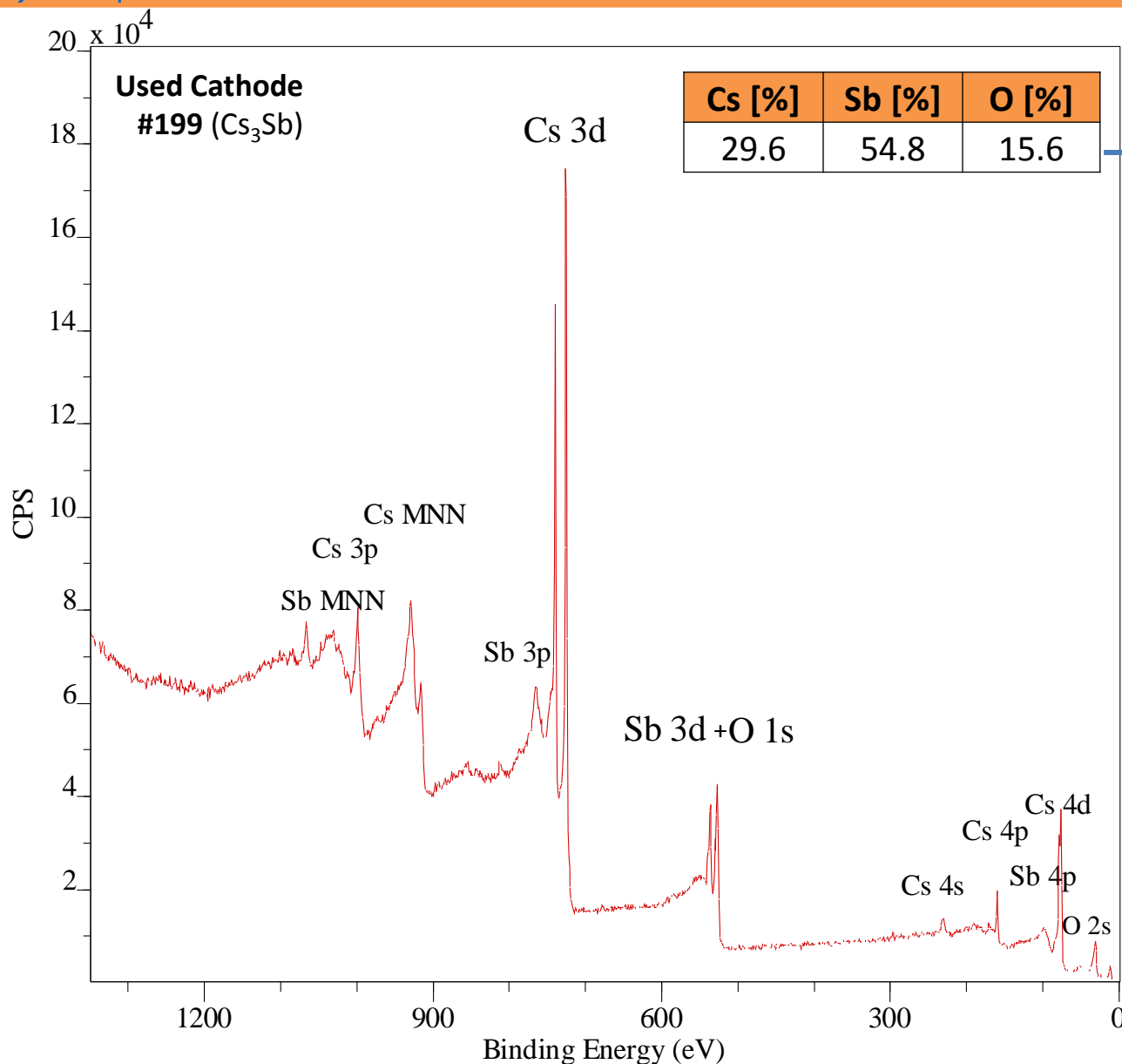
198 NEWLY PRODUCED $\text{QE}_{\text{av}}=14\%$ (DC gun)

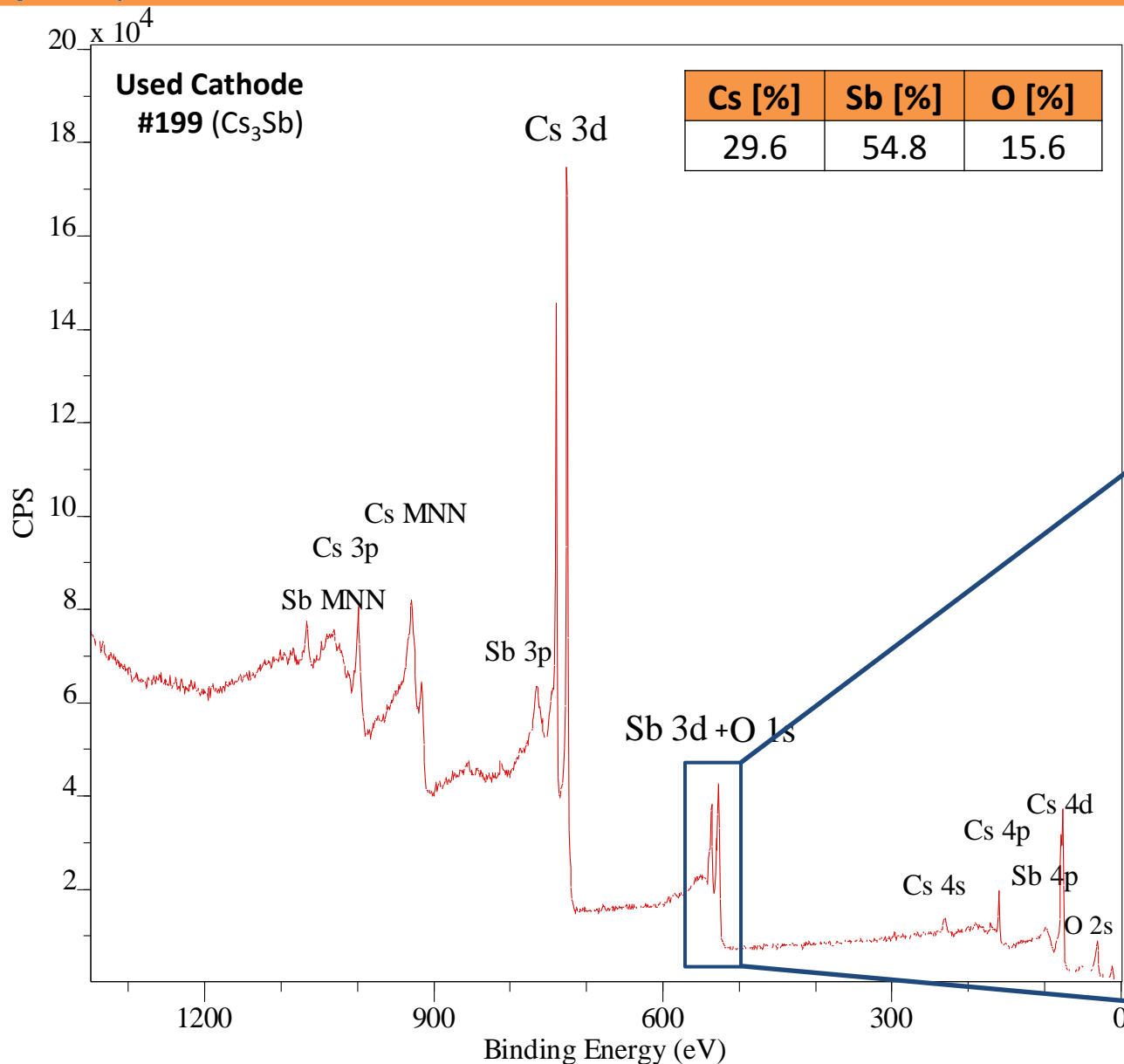


PHIN photoinjector run

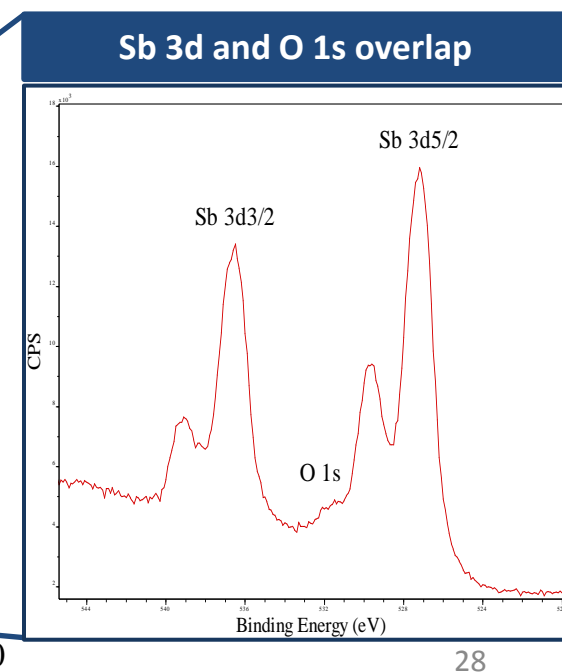
#198 USED $\text{QE}_{\text{av}}=2.7\%$ (DC gun)



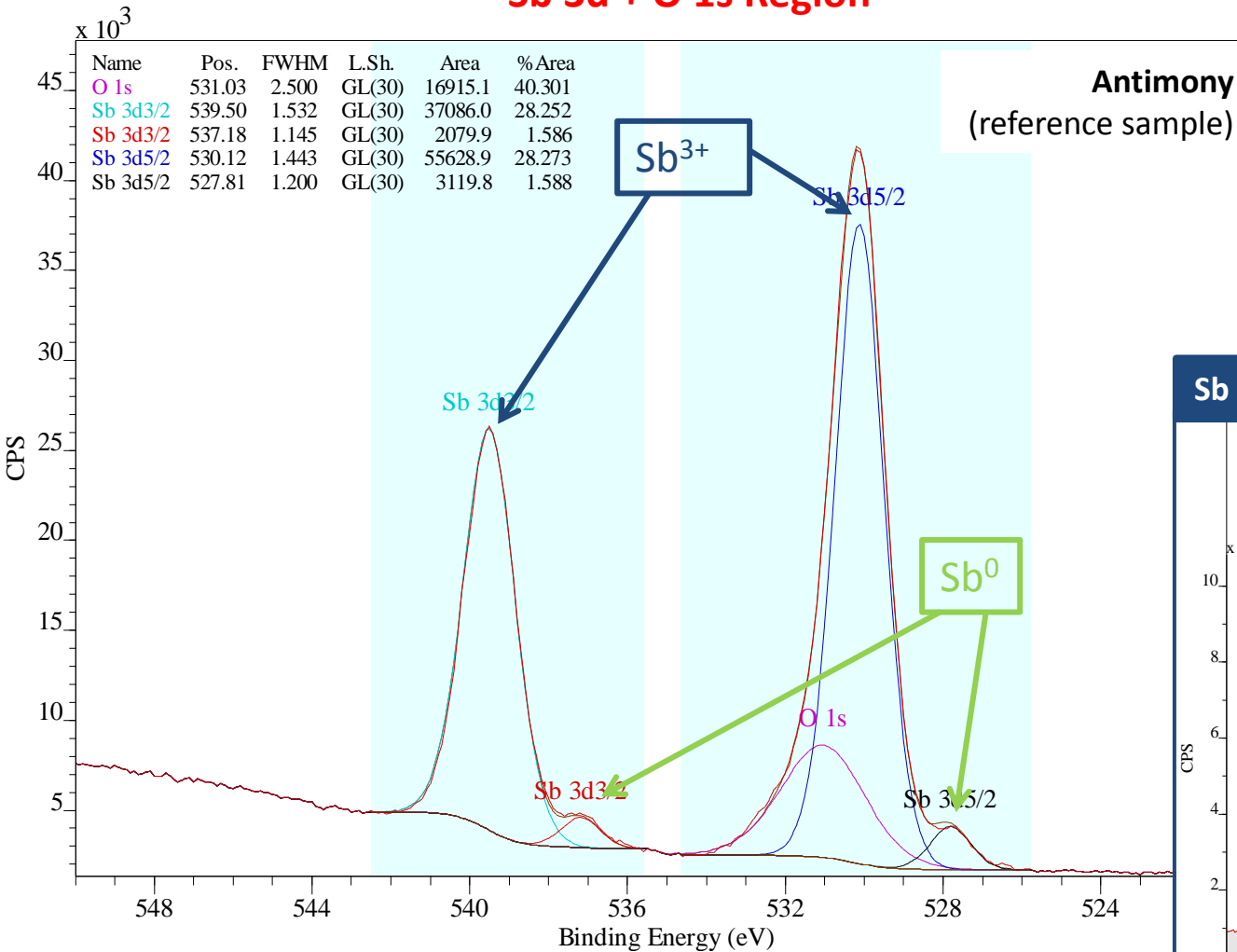




Chemical states identification: O 1s and Sb 3d are overlapping!



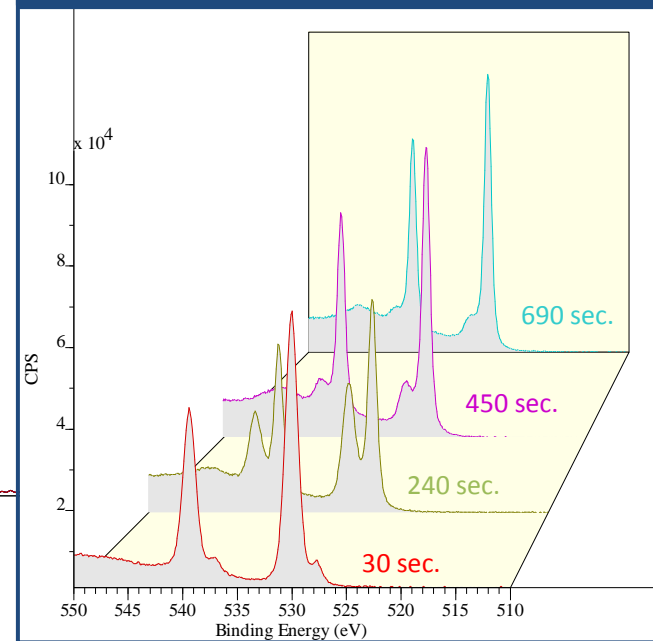
Sb 3d + O 1s Region



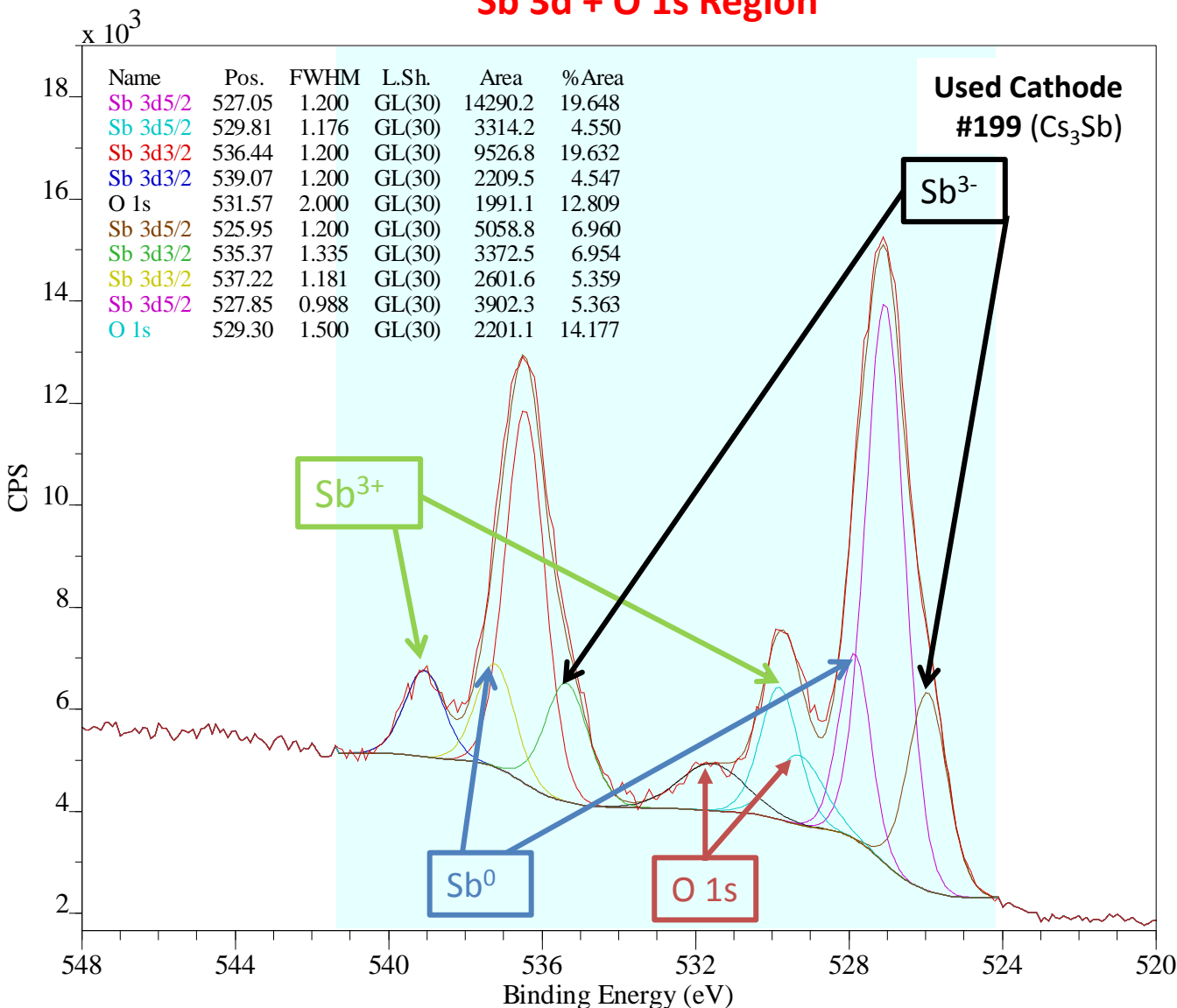
Native oxide layer on top of Sb metallic:

- Sb^{3+} : Sb_2O_3
- Sb^0 : Sb metallic

Sb 3d+O1s during Ar ions sputtering



Sb 3d + O 1s Region



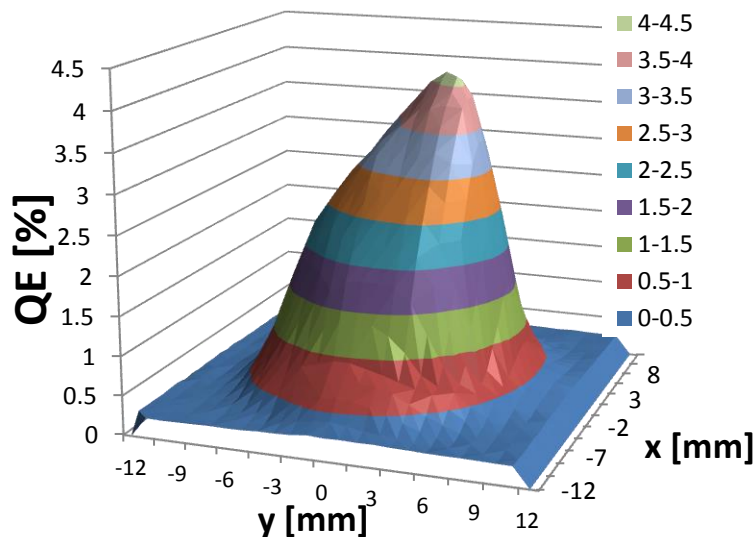
PEAKS MODEL MUST HAVE PHYSICAL REASON.

Possible explanation:

- Sb⁻³: Cs_3Sb
- Cs_xSb (Sb-rich phase)
- Sb⁰ metal
- Sb³⁺: Sb_2O_3
- 2 OXYGEN peaks:
 - Sb_2O_3
 - Cs oxide

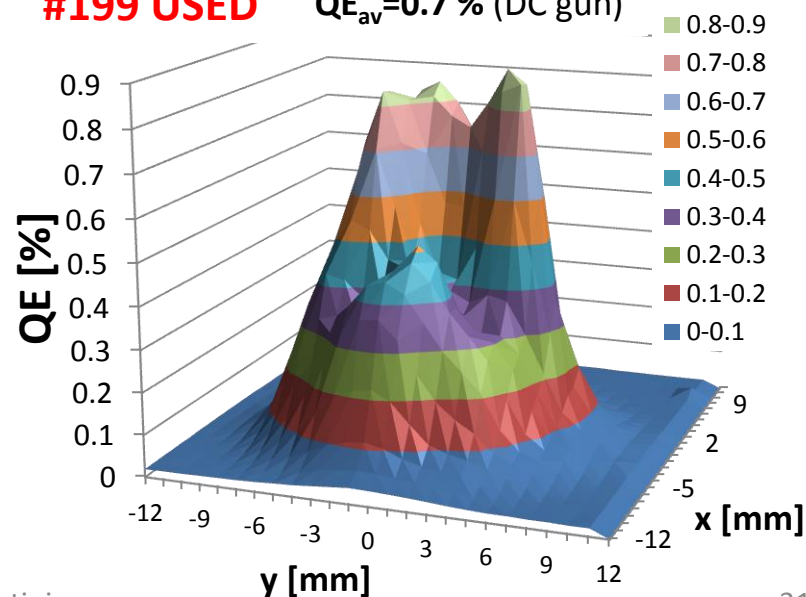
Cathode # 199 (Cs_3Sb)

#199 NEWLY PRODUCED Max $\text{QE}_{\text{av}}=5.2\%$ (DC gun)



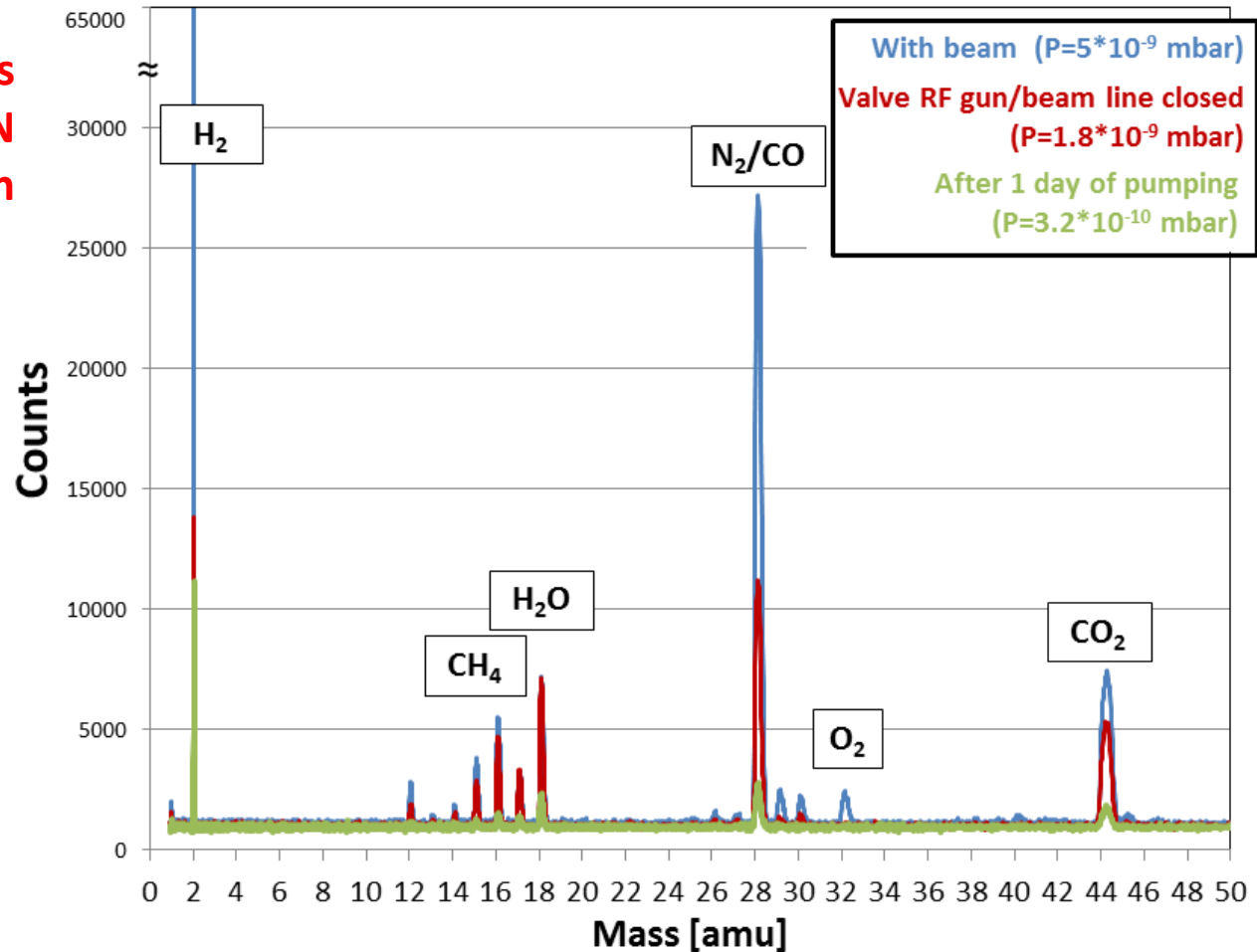
PHIN photoinjector run

#199 USED $\text{QE}_{\text{av}}=0.7\%$ (DC gun)



- Both used cathodes (Cs_3Sb and Cs_2Te) are slightly oxidized, most probably during photoinjector operation (Dynamic Vacuum is worse than Static Vacuum):

**Residual gases
spectrum in PHIN
photoinjector gun**



- **Cathode #198:** more than 50% of Te found on the surface is bound to Cs as Cs_2Te ↔ The measured QE for the used cathode is still good $\text{QE} \sim 3\%$.
- **Cathode #199:** the Sb-rich phase is the main component ↔ The used cathode shows strongly degraded photoemissive properties ($\text{QE} \sim 0.7\%$).
- The low QE phases could be produced during the test in the RF photoinjector (i.e. ions back scattering) or be already there in the deposited layer of the newly produced cathode.



XPS studies on newly produced cathodes are called:

- Comparison with measurement on used cathodes
- Understanding and improving production process

Acknowledgements

Thank you all for your attention!

- Eric Chevallay, Christoph Hessler, Mikhail Martyanov, Valentin Fedosseev (CERN, EN/STI-LP)
- Holger Neupert, Valentin Nistor, Mauro Taborelli (CERN, TE/VSC-SCC)



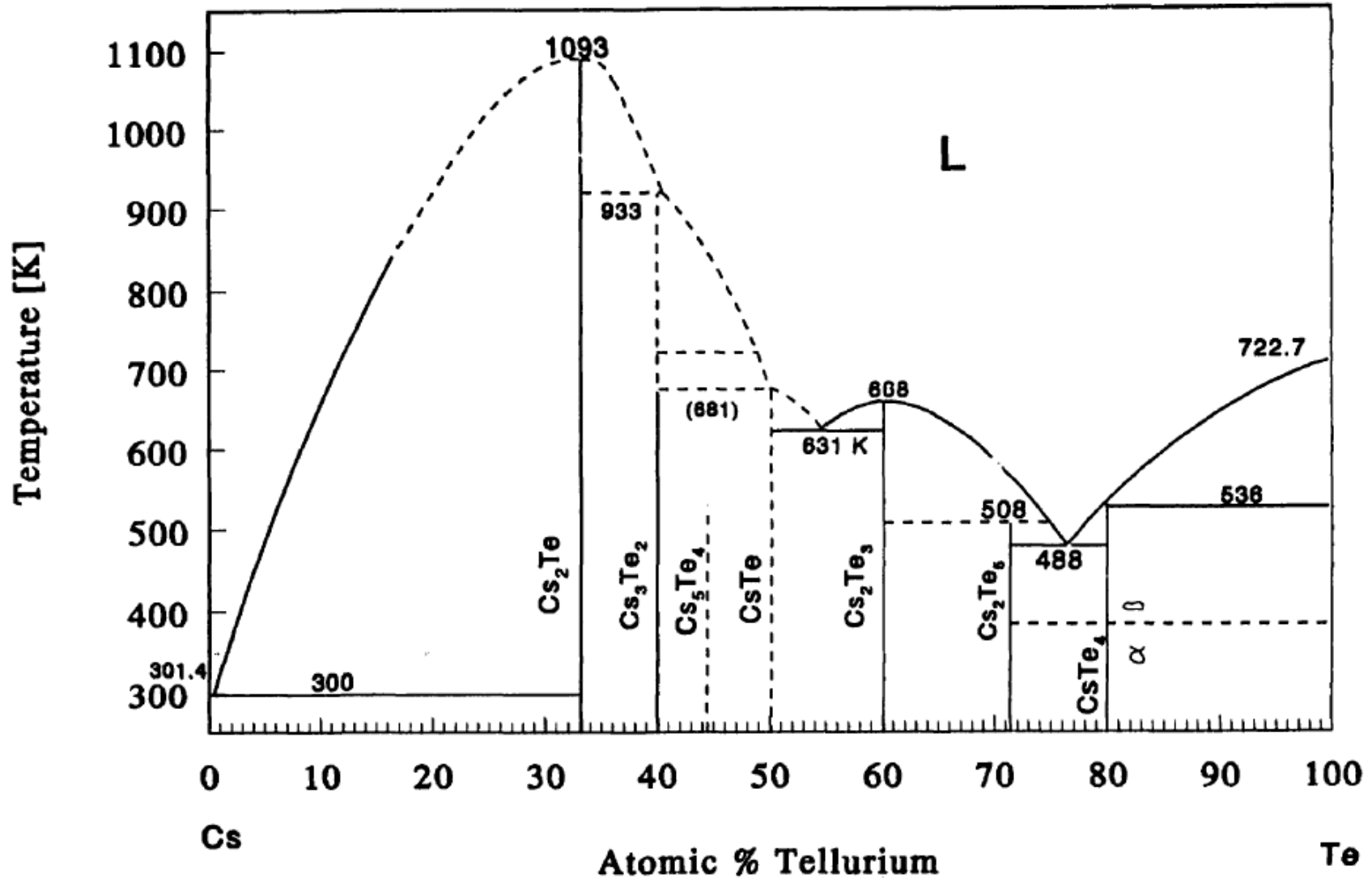
PhD School on Materials
Engineering



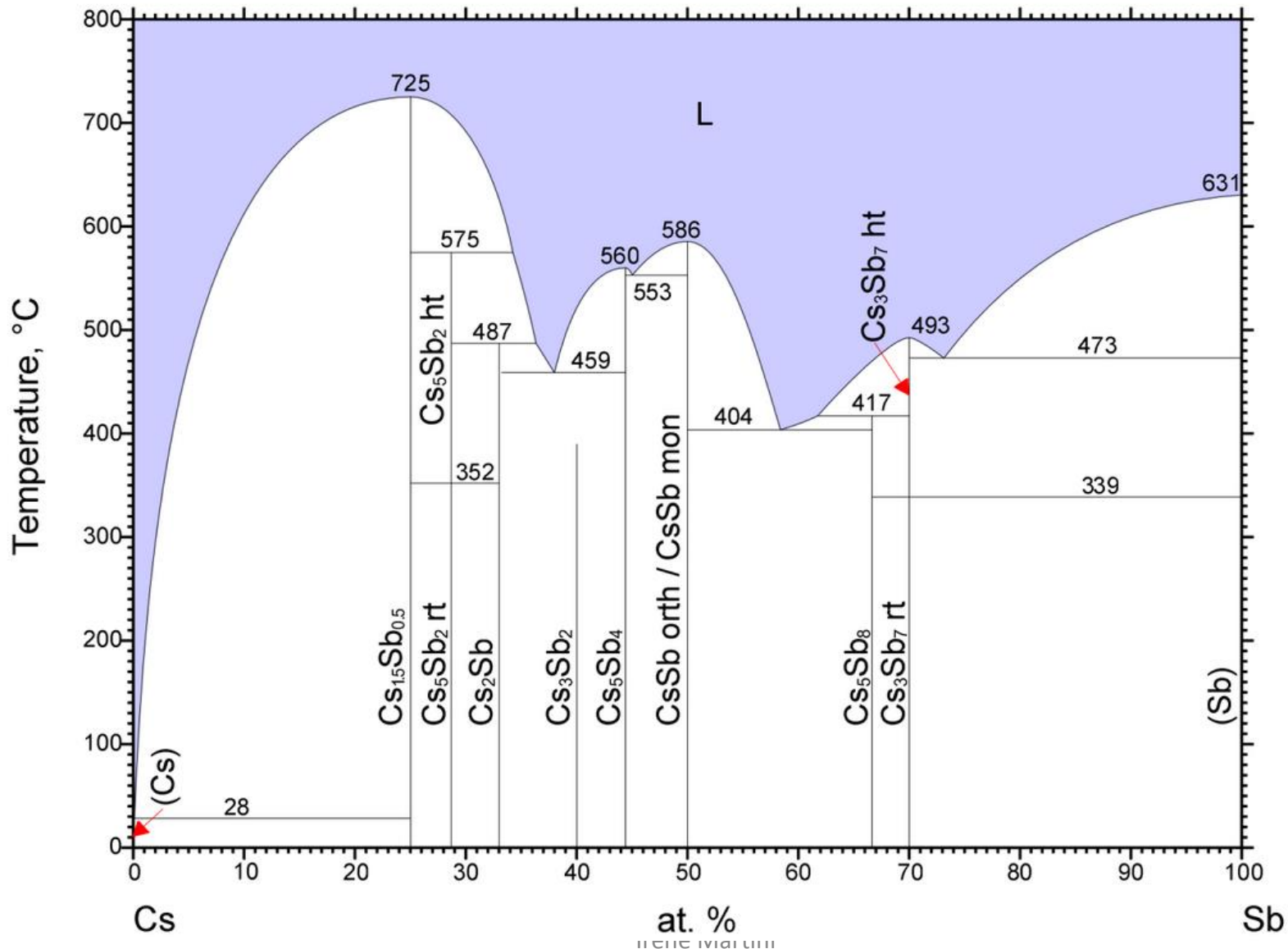
This project has received funding from the European Union's Seventh Framework Programme for research, technological development and demonstration under grant agreement no 289191.

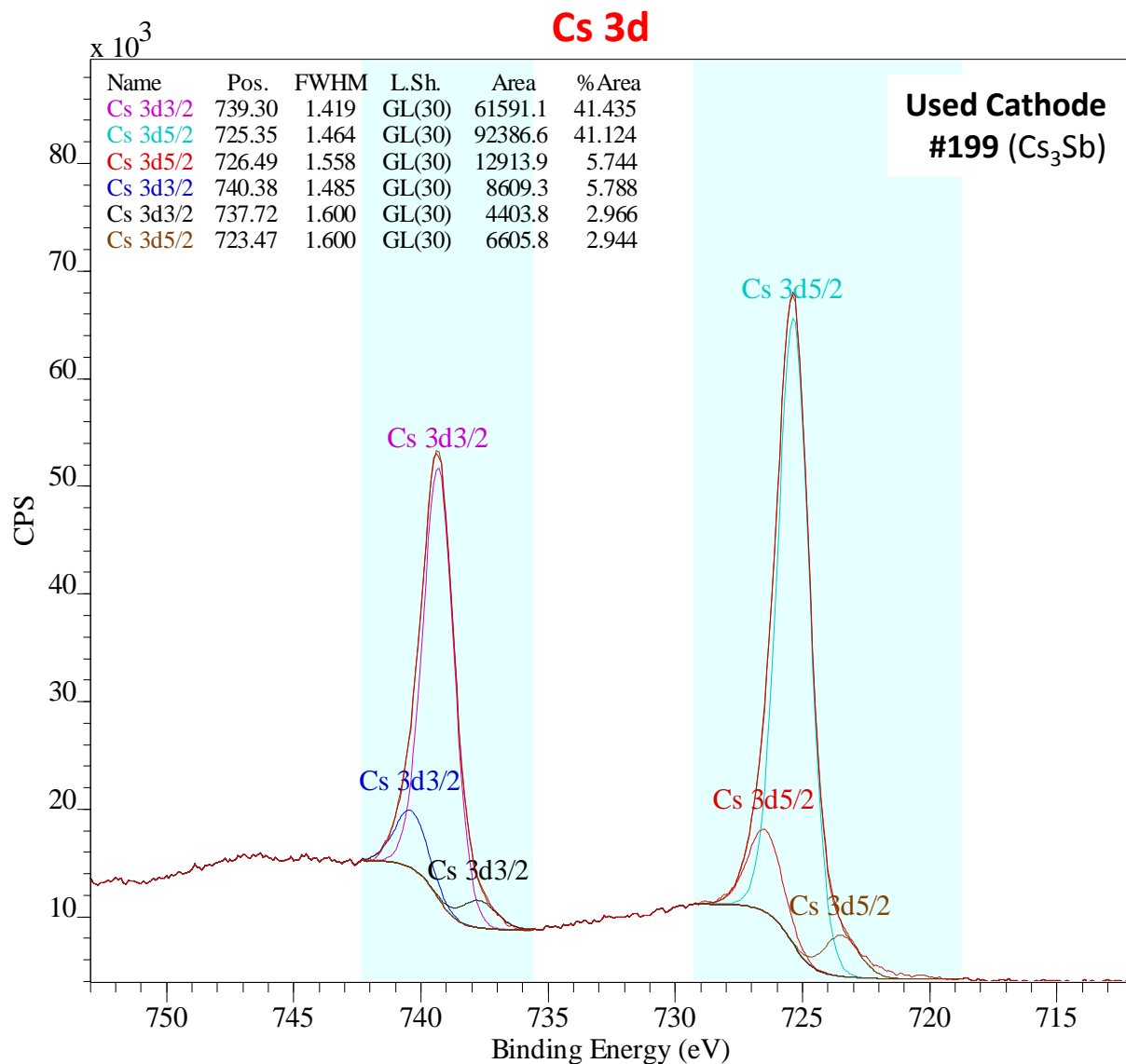
SPARE SLIDES

Cs – Te phase diagram



Cs – Sb phase diagram





Possible explanation (to be consistent with Sb region fitting):

- Cs_3Sb
- Cs_xSb (Sb-rich phase)
- Cs oxide

Surface Analysis: X-rays Photoemission spectroscopy

- Surface analysis of photocathode materials with XPS and their impact on the cathode performance in collaboration with the CERN Vacuum Group has started.



XPS SET UP



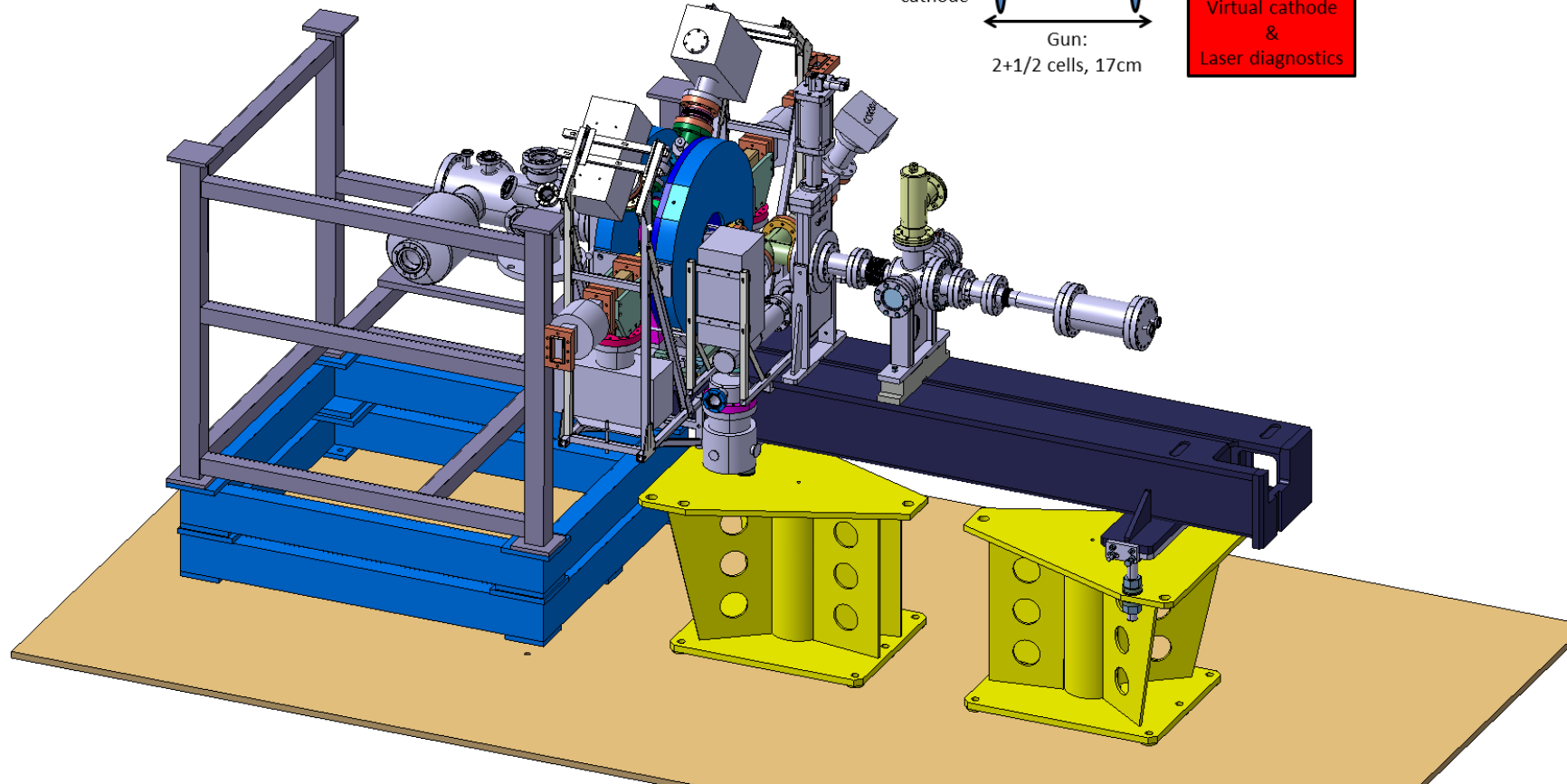
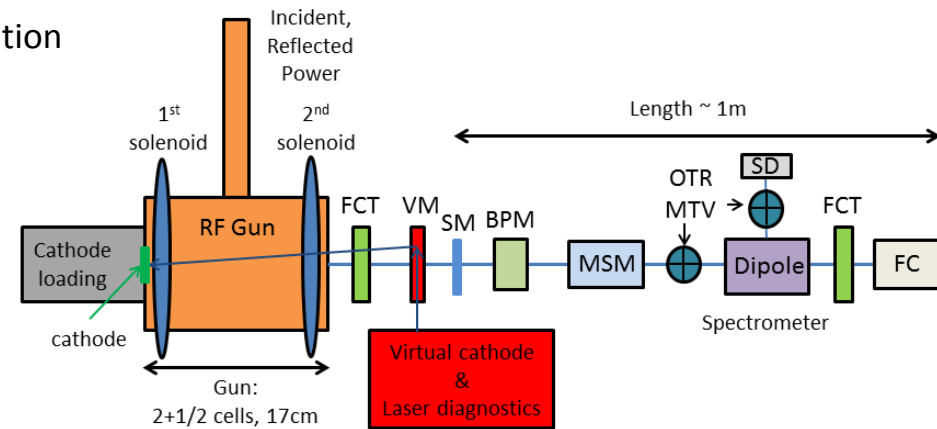
UHV carrier vessel to transfer cathode from production laboratory to the XPS set-up.

2 WEEKS NEEDED EACH TIME FOR 1 CATHODE TRANSFER (included baking of the SAS at the Photoemission Lab and at the XPS set up lab)

PHIN RF photoinjector

FCT: Fast current transformer
VM: Vacuum mirror
SM: Steering magnet
BPM: Beam position monitor
MSM: Multi-slit Mask

OTR: Optical transition radiation screen
MTV: Gated cameras
SD: Segmented dump
FC: Faraday cup



PHOTOEMISSION LABORATORY

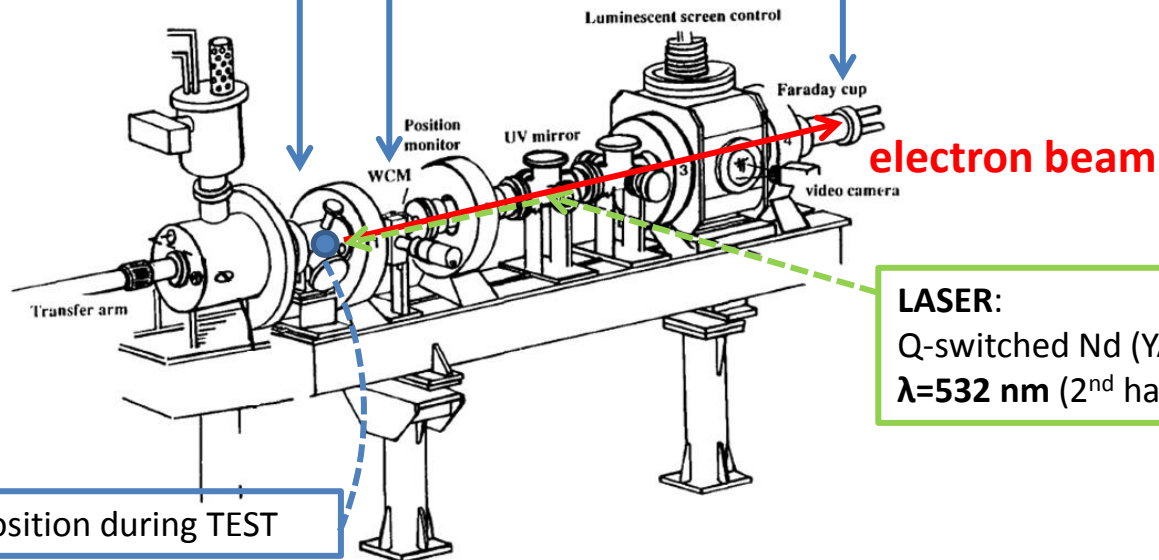
PREPARATION CHAMBER

BEAM LINE

DC GUN:

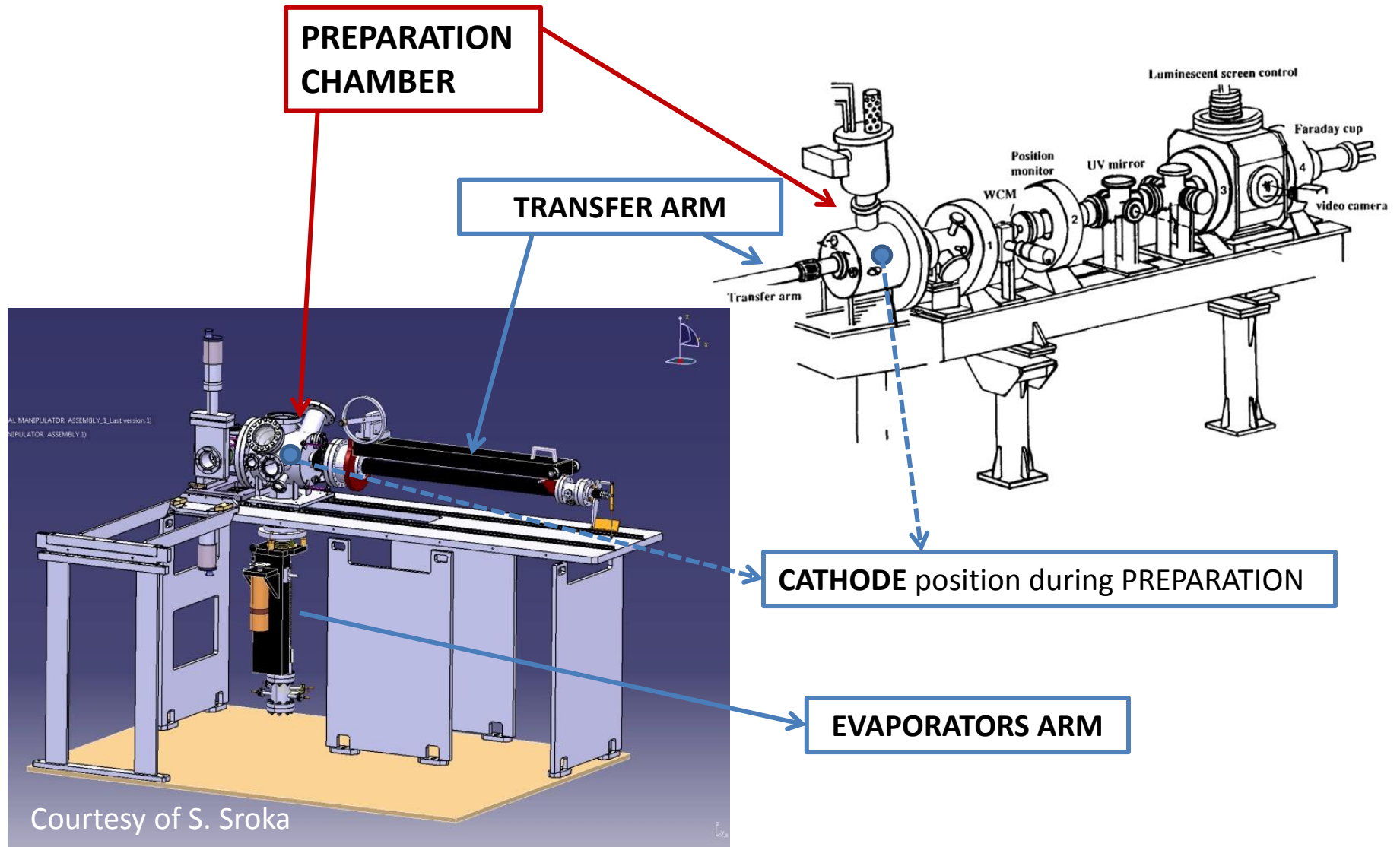
70 kV supplied voltage
(7 MV/m electric field)

Bunch charge measured by:
Wall Current Monitor (WCM)
Faraday Cup (FC)



LASER:

Q-switched Nd (YAG)
 $\lambda=532 \text{ nm}$ (2nd harmonic)



Photocathodes for photoinjectors: Requirements

- High QE (Quantum Efficiency) at a convenient laser wavelength λ

$$QE(\%) = \frac{\# \text{ emitted electrons}}{\# \text{ incident photons}} \sim \frac{Q}{\lambda * E}$$

- Low dark current
- Long lifetime in challenging environment

$$QE(t) = QE_{max} * e^{-\frac{t}{\tau}}, \quad \tau: \frac{1}{e} \text{ lifetime}$$

Photocathodes requirements:

- **High QE at a convenient laser wavelength λ**
- Low dark current
- Long lifetime in challenging environment

Factor Affecting QE

Photoemission process: absorption, electron scattering, escape over surface barrier	Depending on the material (metals, semiconductors,..) and laser
Deposition process	Substrate material, Layer thickness, substrate temperature,..
Vacuum environment	Residual gas composition, Electron stimulated desorption, ..
Operating environment	Space charge,..

Photocathodes requirements:

- High QE at a convenient laser wavelength λ
- **Low dark current**
- Long lifetime in challenging environment

Factor Affecting Dark Current (Field Emission)

$$\bar{I}_F = \frac{5.7 * 10^{-12} * 10^{4.52\phi^{-0.5}} * A_e(\beta E)^{2.5}}{\phi^{1.75}} * \exp\left(-\frac{6.53 * 10^9 * \phi^{1.5}}{\beta E}\right)$$

Fowler-Nordheim for RF field
(\bar{I}_F : Average field emitted current, A_e :Emitting Area)



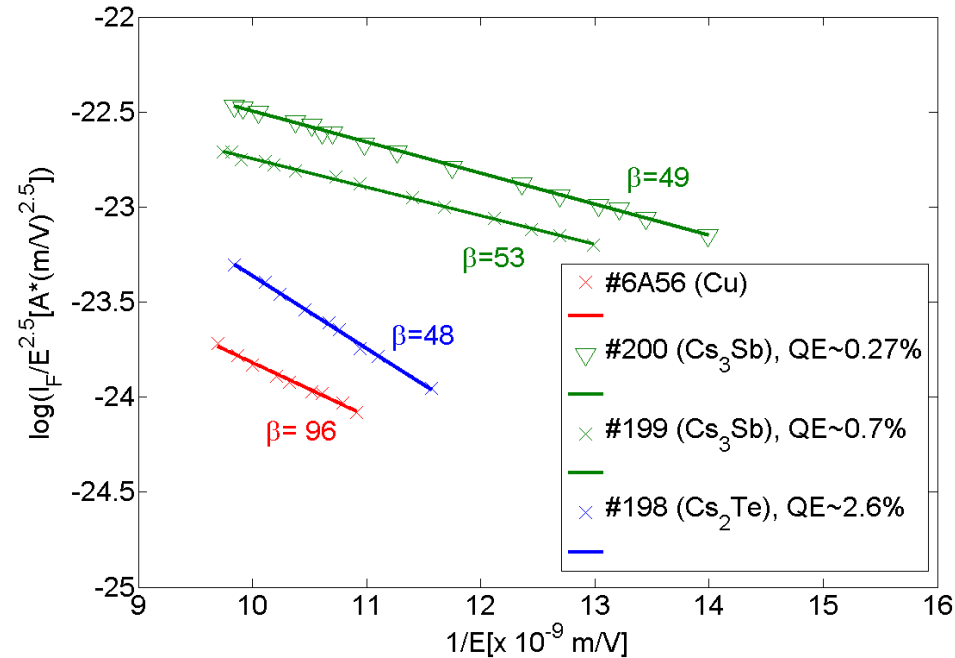
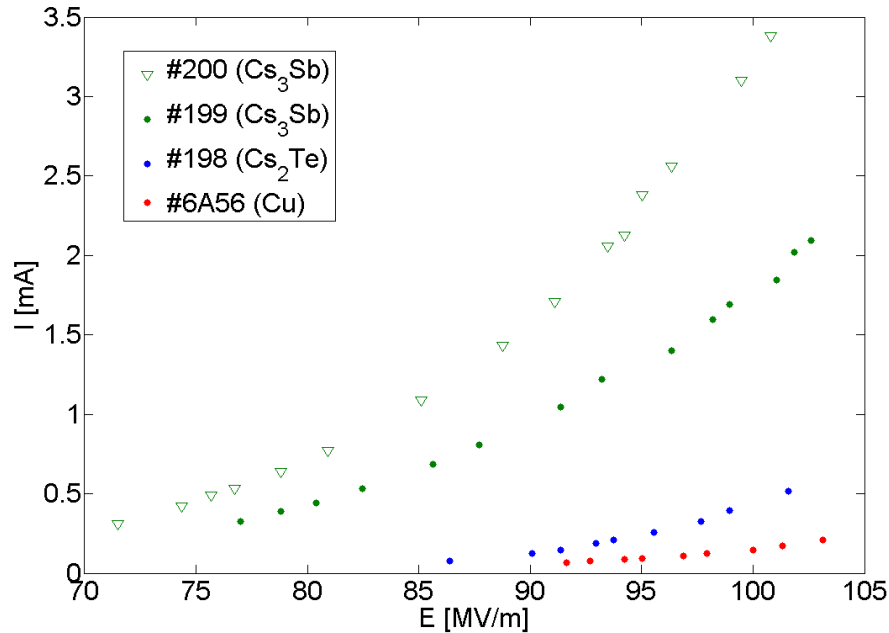
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- High QE at a convenient laser wavelength λ
- Low dark current
- **Long lifetime in challenging environment**

Factor Affecting Lifetime

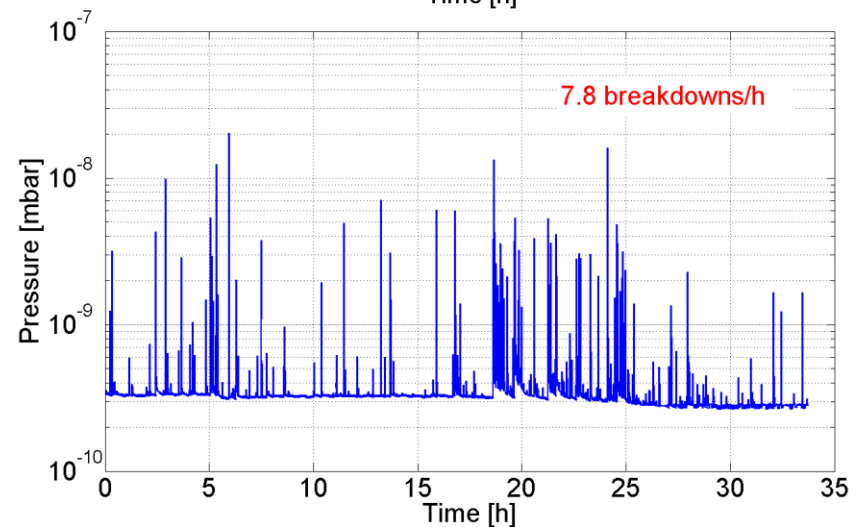
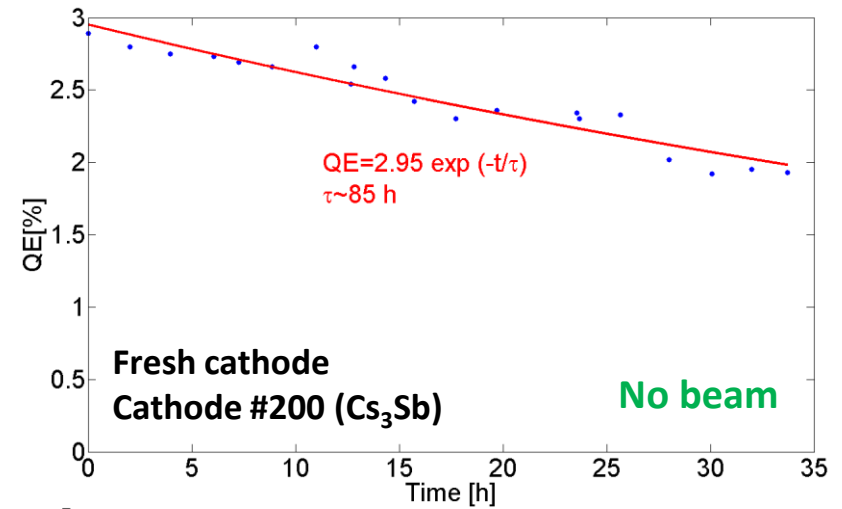
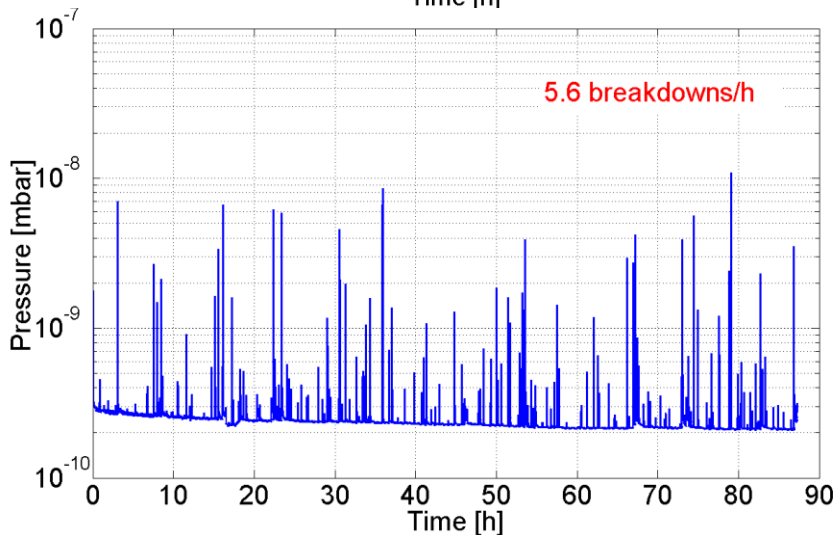
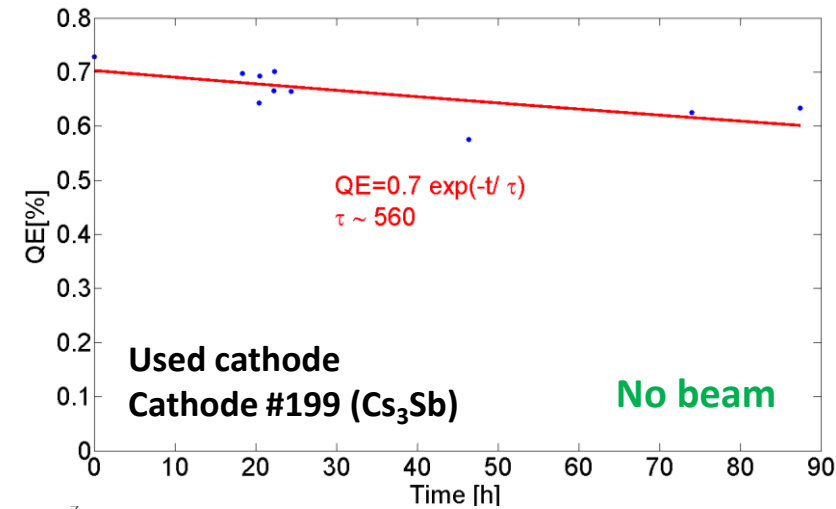
Chemical poisoning	Static and Dynamic Vacuum condition
Radiation damage	Electron and Ion Bombardment on the cathode surface

The dynamic vacuum and the ion back bombardment depend on the beam properties: charge/bunch, rep rate, beam transmission,...

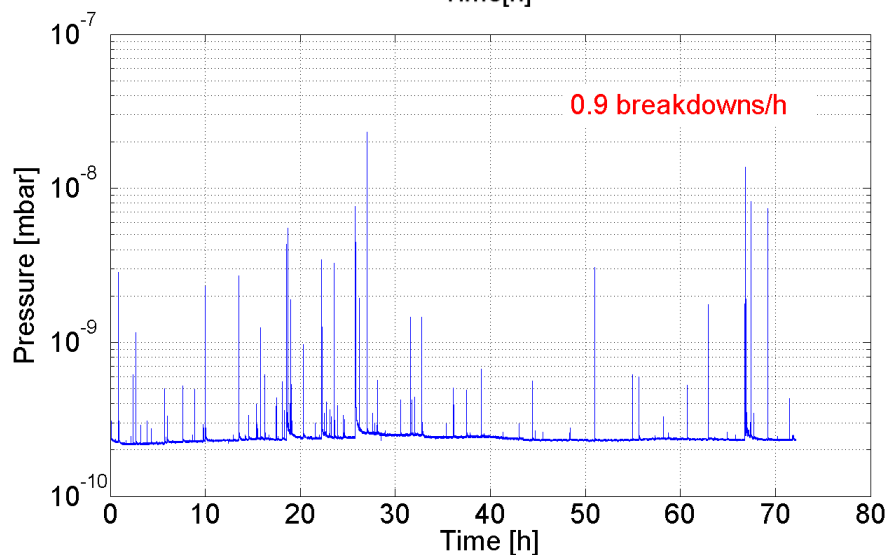
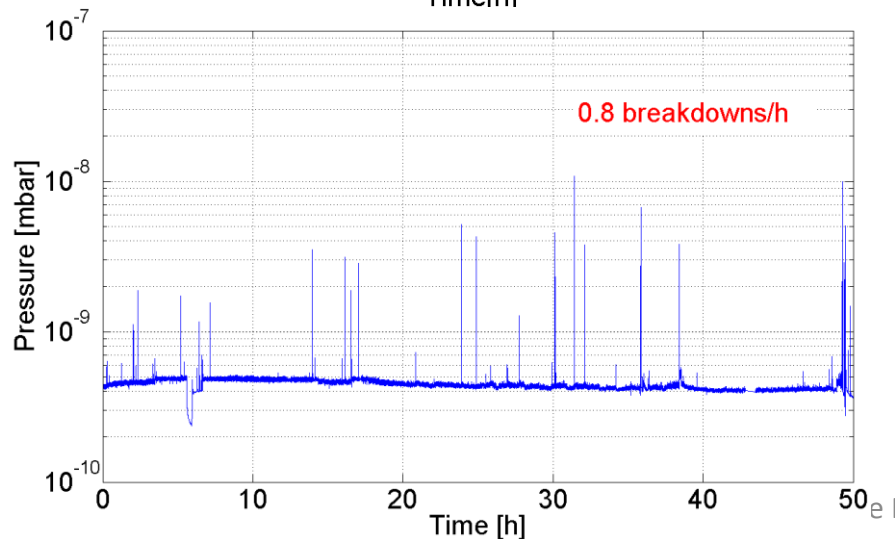
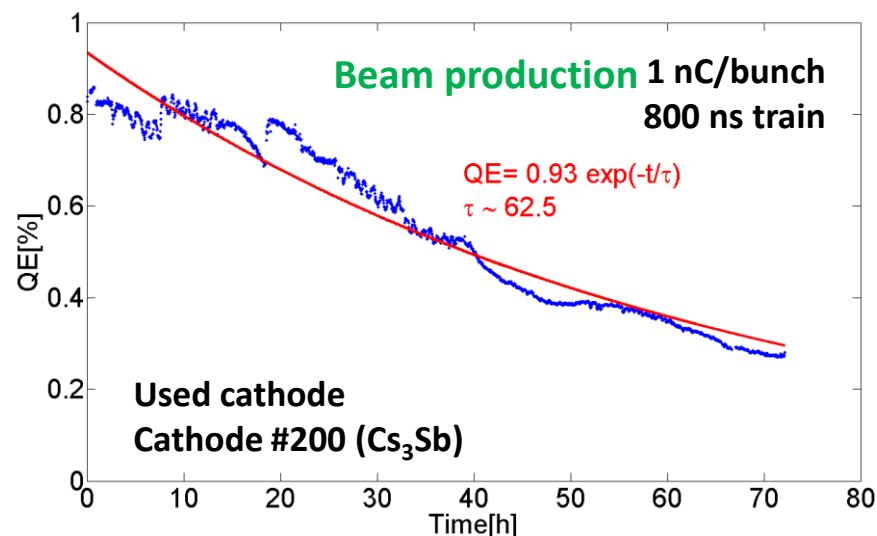
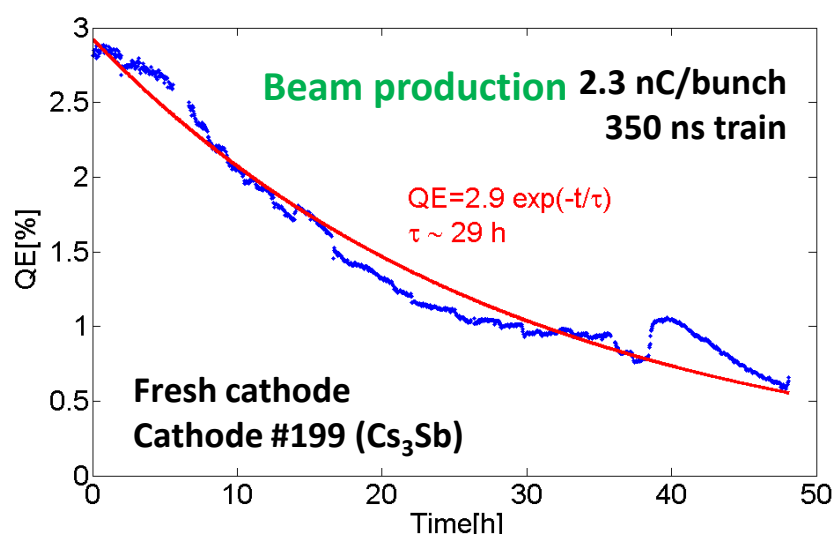


- Non desirable effect: **affecting negatively the beam dynamics** and possibly **originating RF breakdowns** (high voltage discharge inside the gun resulting in pressure sparks).
- Field emission contribution from RF gun walls (copper made) and from the cathode.
- The low dark current measured with copper confirmed the **major contribution is coming for the cathode**.
- Cs_3Sb cathodes ($\phi \sim 2$ eV) produce higher dark current than Cs_2Te ($\phi \sim 3.5$ eV).

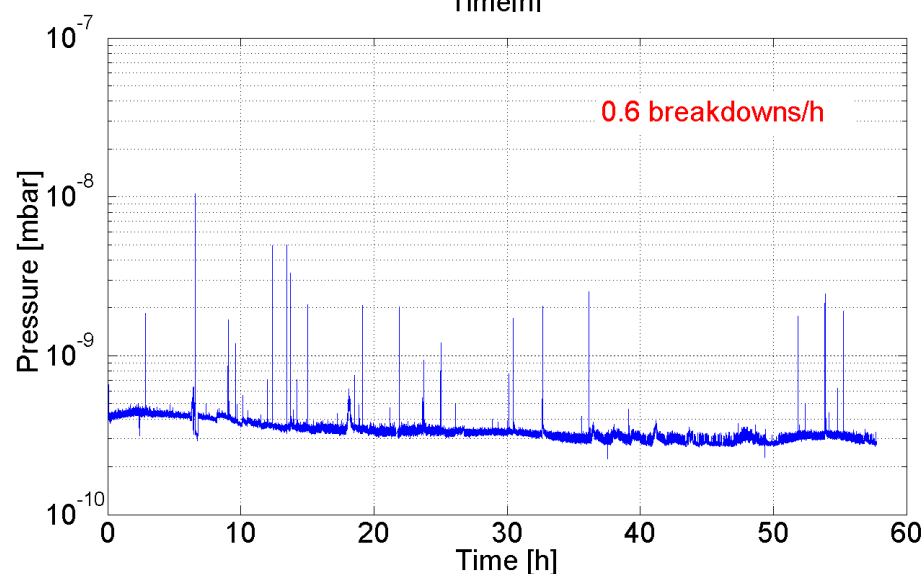
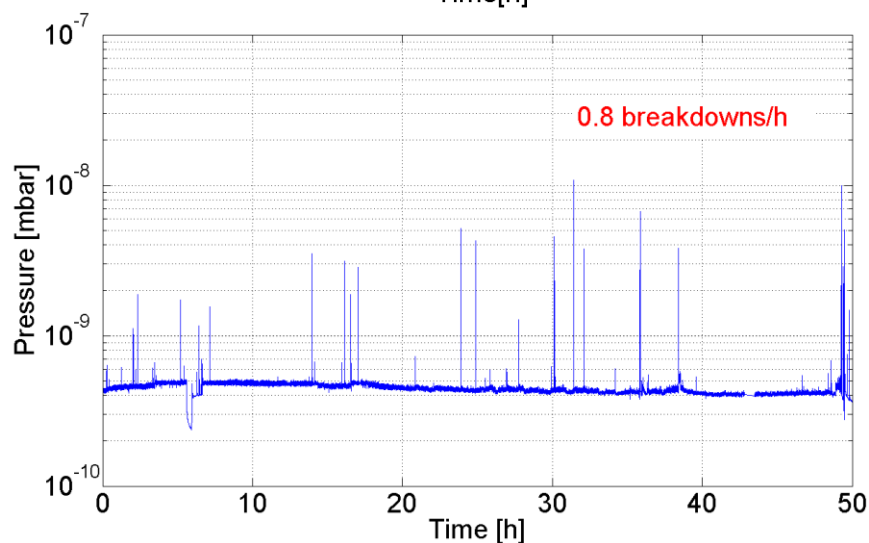
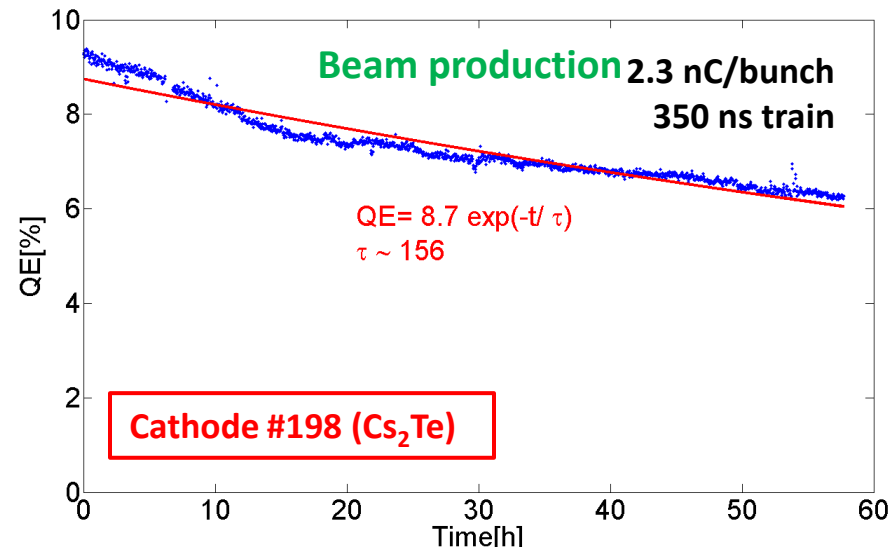
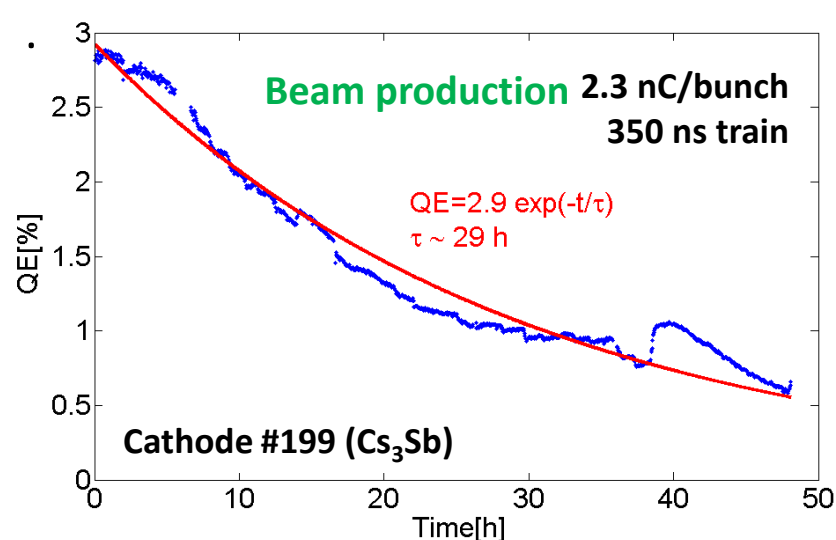
- QE deterioration under the influence of **high electric gradient (100 MV/m)** – no continuous beam production.

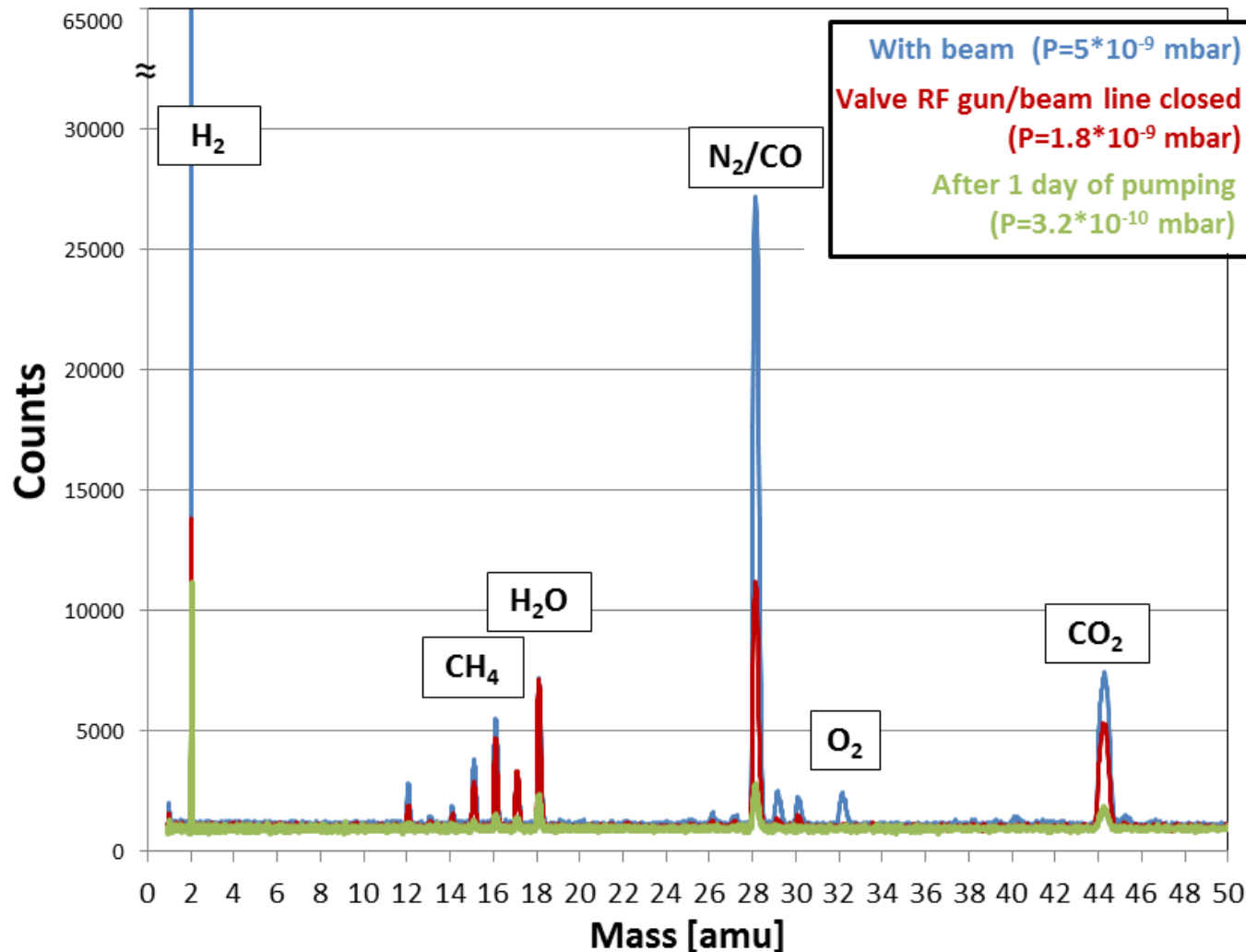


- QE deterioration during beam production.
- The laser power was adjusted for keeping the produced charge constant: 1200 nC/train.

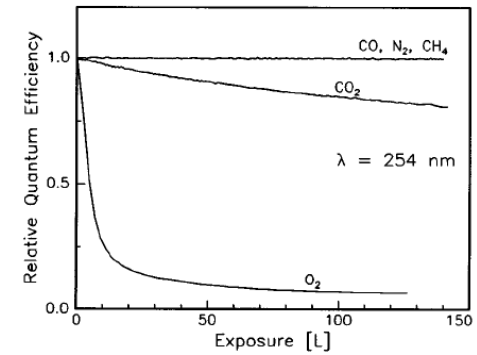


- The laser power was adjusted for keeping the produced charge constant: 1200 nC/train



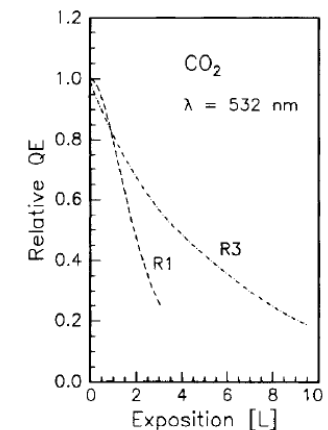


Cs_2Te cathode



A. Di Bona et al., J. Appl. Phys. 80, 3024 (1996)

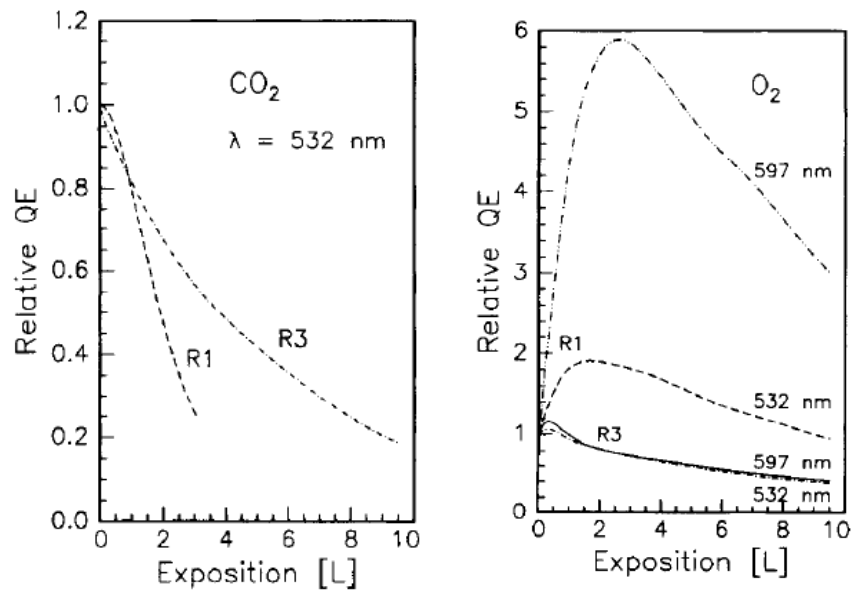
K_2CsSb cathode



A. Di Bona et al., Nucl. Instr. Met. In Phy. Res. A 385 (1997) 385-390

QE drop after exposure to different gas species

K_2CsSb



A. Di Bona et al., Nucl. Instr. Met. In Phy. Res. A 385 (1997) 385-390

K_2CsSb

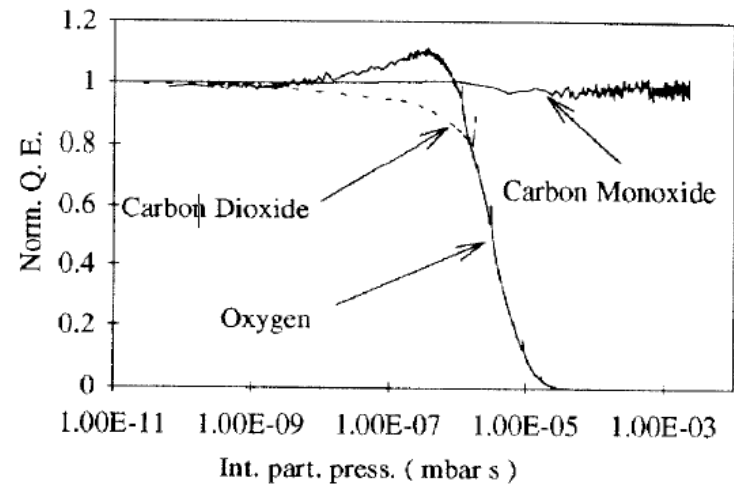


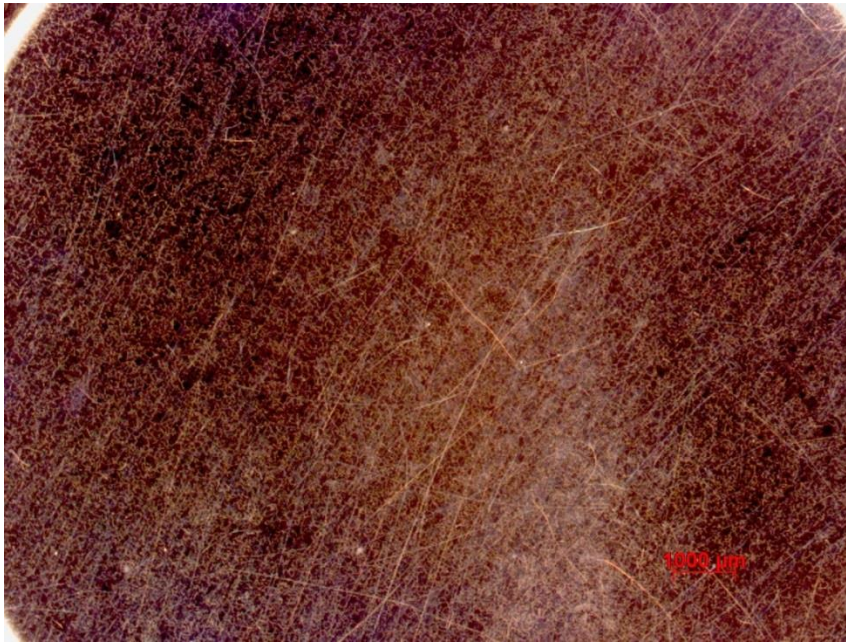
Fig. 6. Poisoning of K₂CsSb cathode with different gases.

P. Michelato et al., "Multialkali Thin Photocathodes for High Brightness guns", Proc. of EPAC '94

Surface morphology

COPPER SUBSTRATE MACHINING

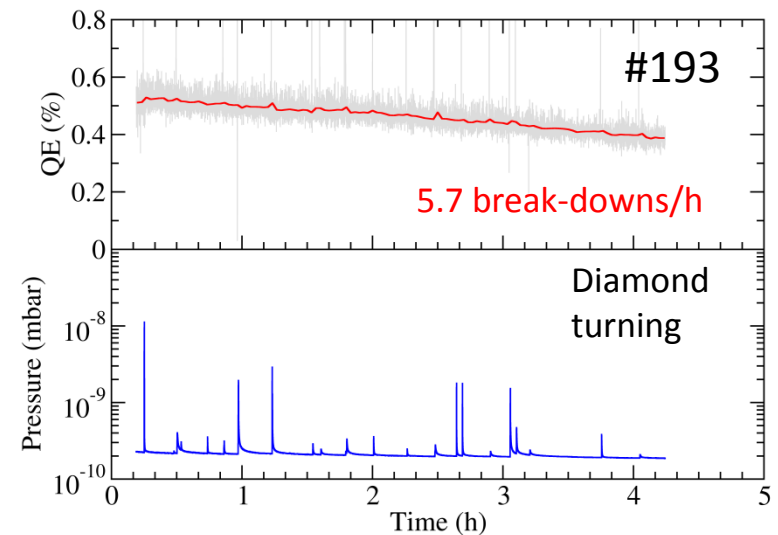
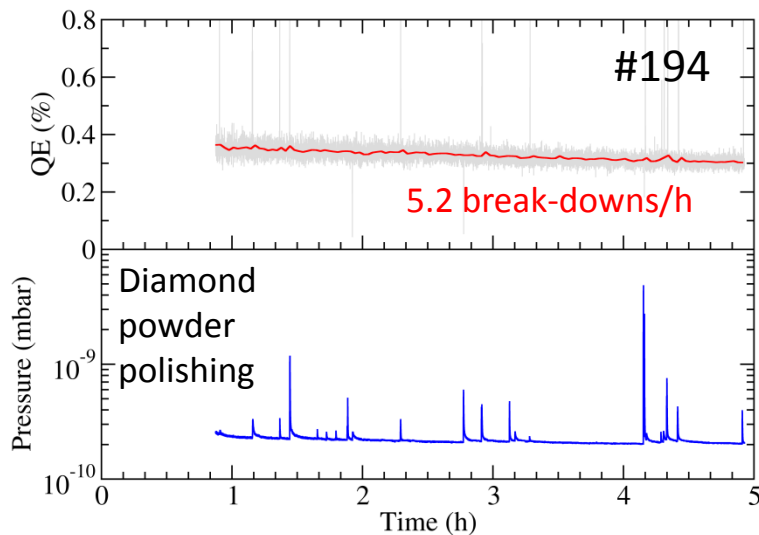
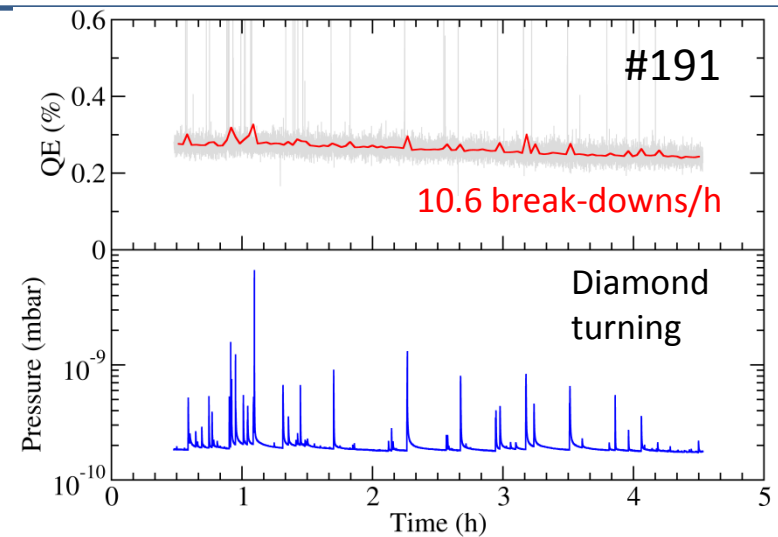
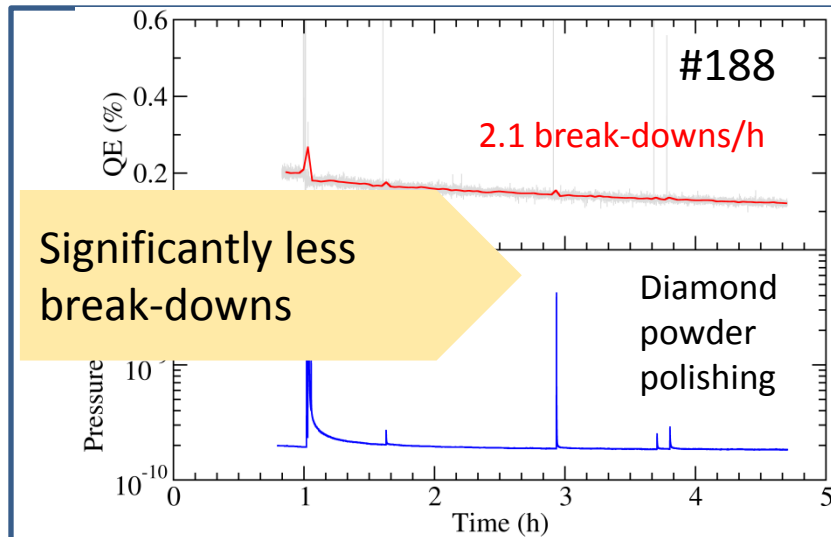
1) Diamond powder polished.



2) Diamond turned.



- Assumption: The thin deposited Cs_3Sb layer (<100 nm) is reproducing the roughness profile of the substrate (determining the β field enhancement factor).
- Future studies in collaboration with Daresbury Laboratory (ASTeC, U.K.) for **reducing roughness** and **improving cleaning** of the copper substrate surface.



MOTIVATION: The roughness and cleanliness of Cu substrate influence the photocathode performance.

Test 2 different cleaning procedures on diamond powder polished Cu (XPS, Roughness):

- Oxygen plasma cleaning+ annealing (250°C,1h)
- CERN chemical treatment + annealing (250°C,1h)

#003	Cu [%]	O [%]	C [%]		#001	Cu [%]	O [%]	C [%]	S _A [Å]
as received	3.9	17.4	78.7		as received	3.8	75.9	20.3	78
O ₂ plasma	32.1	47.5	20.4		O ₂ plasma	13.9	36.5	49.6*	89
annealed	72.8	16.5	10.7		annealed	58.3	16.6	25.1	129

#007	Cu [%]	O [%]	C [%]		#009	Cu [%]	O [%]	C [%]	S _A [Å]
CERN cleaning	18.4	40.4	41.2		CERN cleaning	30,1	27,1	42,8	69
annealed	58.3	22.5	19.2		annealed (x2)	78,3	16,6	5,1	111

*bad handling of sample

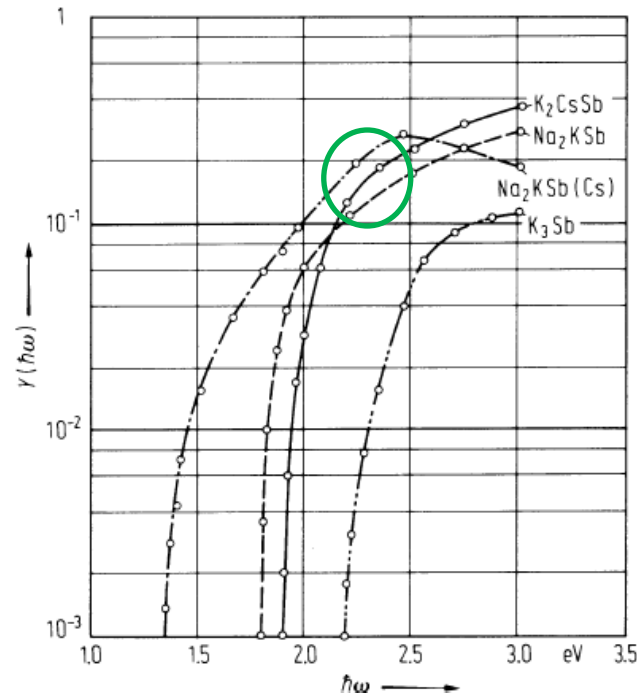
SUMMARY:

- Both cleaning procedure give similar final oxygen content.
- Carbon (from pollution) can be drastically reduced with both procedure (residual C due to embedded diamond particles?)
- Roughness seems to be increased during annealing.

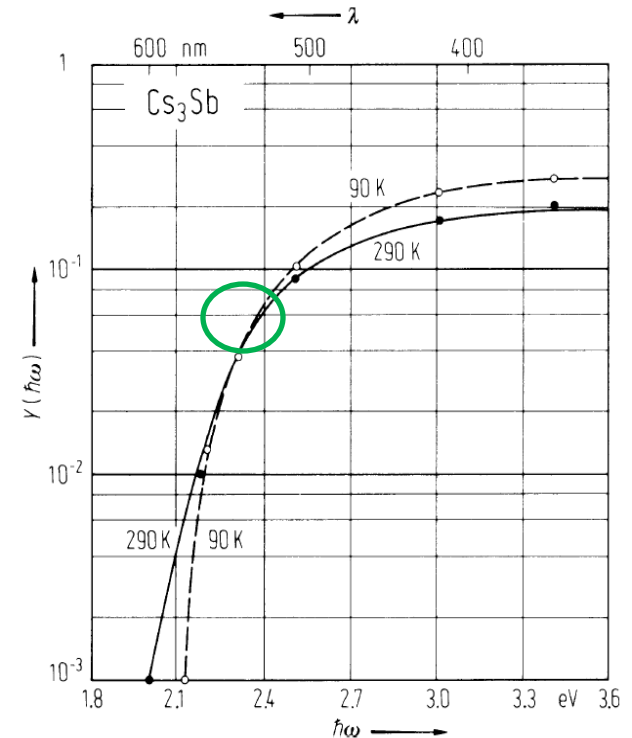
Overview on “green cathodes”

ALKALI-ANTIMONIDE: SPECTRAL RESPONSE

- K_2CsSb
- Cs_3Sb
- K_3Sb
- Na_2KSb
- ...



J. Appl. Phys. 49 (8), August 1987.



A.H.Sommer. Photoemissive Materials.
John Wiley and Sons, Inc., New York, 1986.

Stable for many years in phototubes!
BUT in a photoinjector??

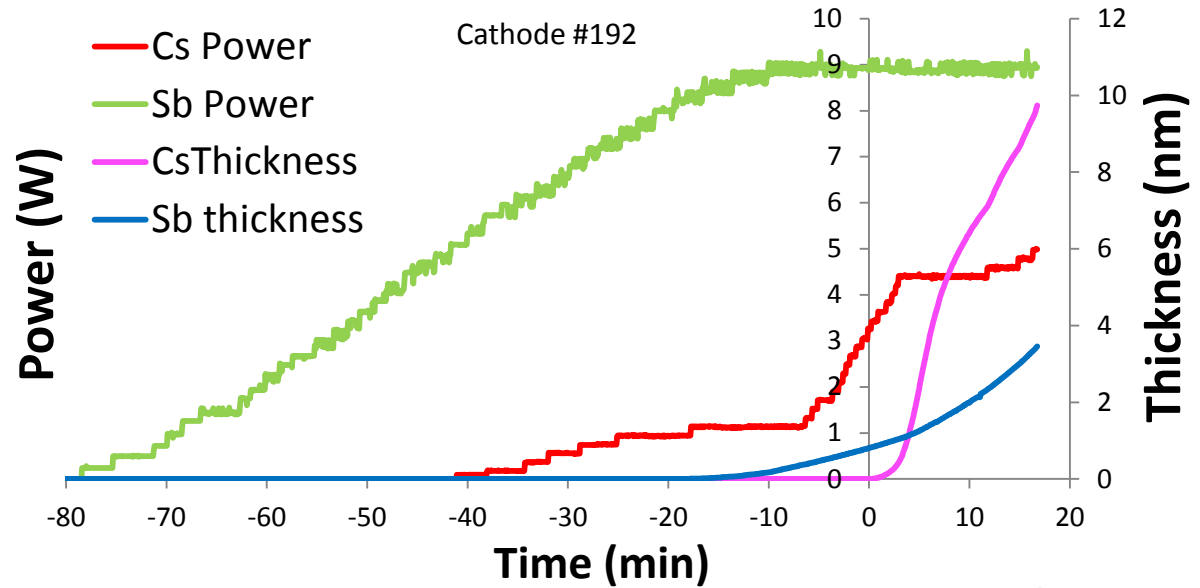


Let's start with **Cs_3Sb** !

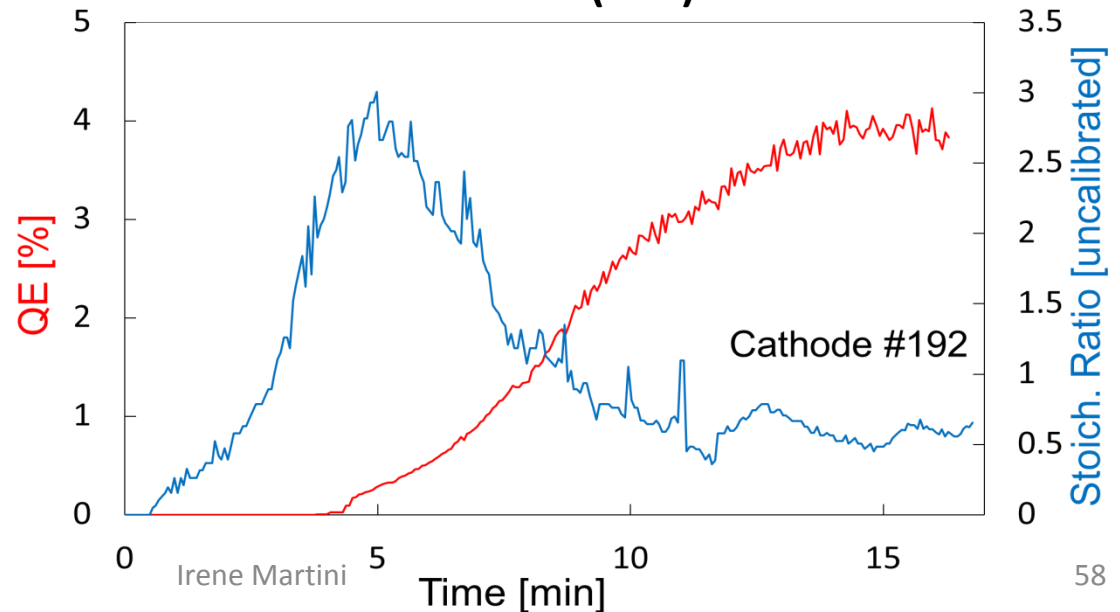
Co-deposition process

- CO-DEPOSITION**

Cs and or Sb evaporated at the same time: the metallic elements can mix together in the vapour phase.

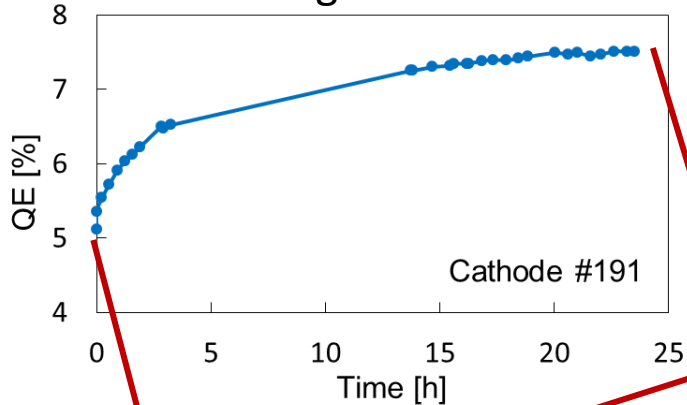


- The evaporators power is tailored in order to reach a maximum value of the QE.

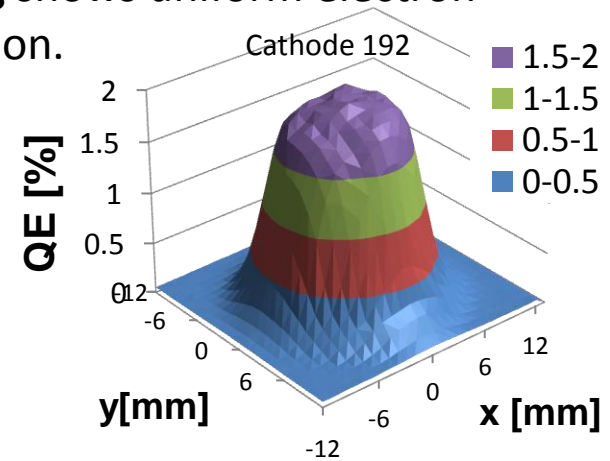


Deposition results

- The QE rise during the first hours of electron beam generation in the DC gun.



- QE mapping shows uniform electron photoemission.



No.	Initial QE (%)	Max QE (%)	Evaporated Cs (nm)*	Evaporated Sb (nm)*	Final stoich. ratio*
178	0.3	0.5	120	18.4	2.9
179	1.4	2.3	156	24.5	1.74
180	0.6	1.0	52	14.4	0.82
187	0.3	0.4	67.6	4.7	4.9
188	1.3	2.2	152	17.8	2.3
189	2.3	4.4	64	15	1
191	5.4	7.5	156	14	1.7
192	2.0	2.7	9.7	3.5	0.66
193	4.2	5.8	10.8	7.6	0.66

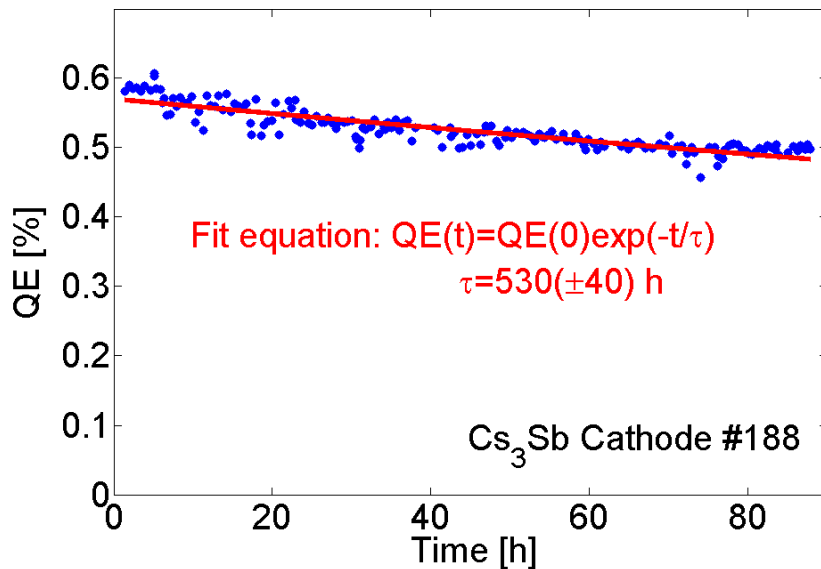
(*)Not calibrated

Irene Martini

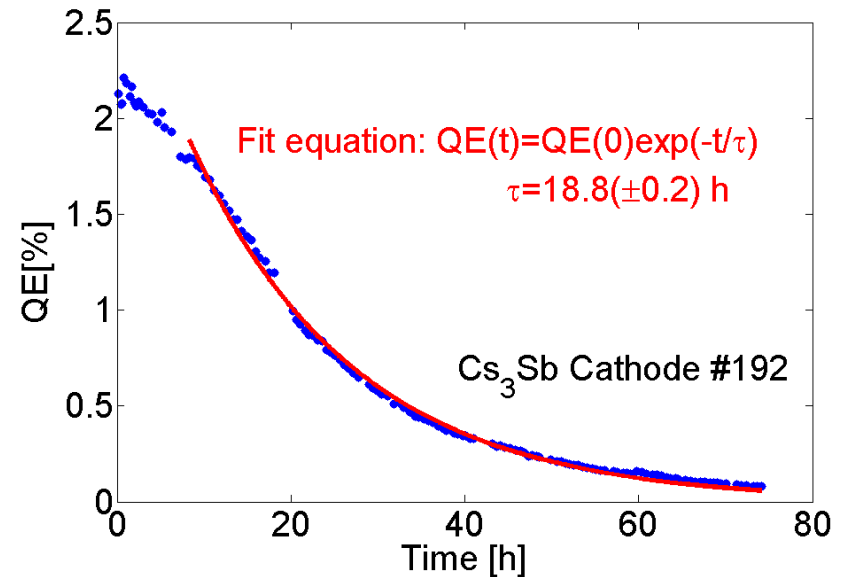
- Excellent QE values of up to 7%.
- No obvious correlation between QE and the final stoichiometric ratio or the evaporated quantity.

Cs₃Sb cathodes lifetime measurements

Low average current measurement (1 μ A)



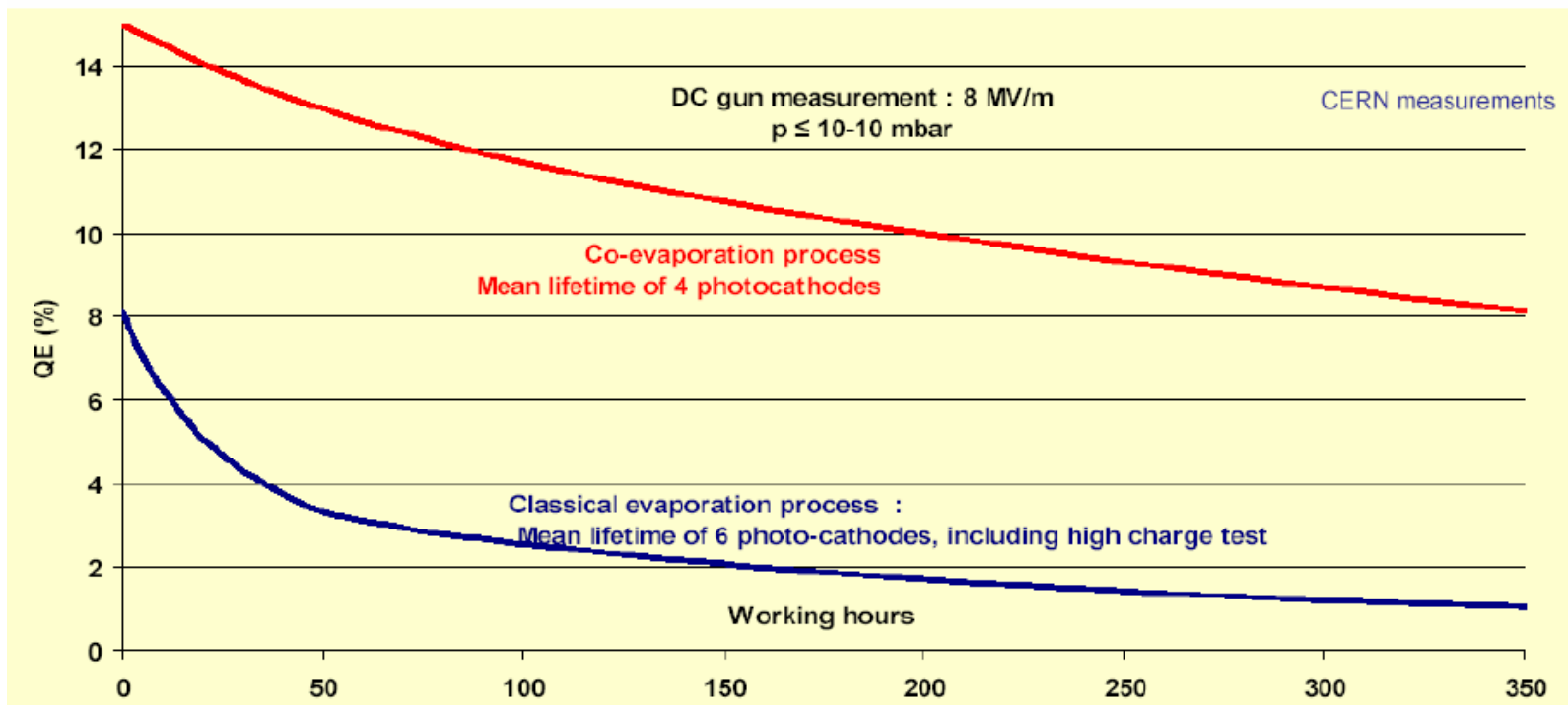
High average current measurement (120 μ A)



- Low current measurements: very promising results!
- High current: faster degradation (chemical poisoning related to desorption induced by high current beam losses?)
- Cathode #192 has thin photoemissive layer \Rightarrow investigate thick photoemissive layer

Co-deposition vs sequentially deposition (Cs_2Te)

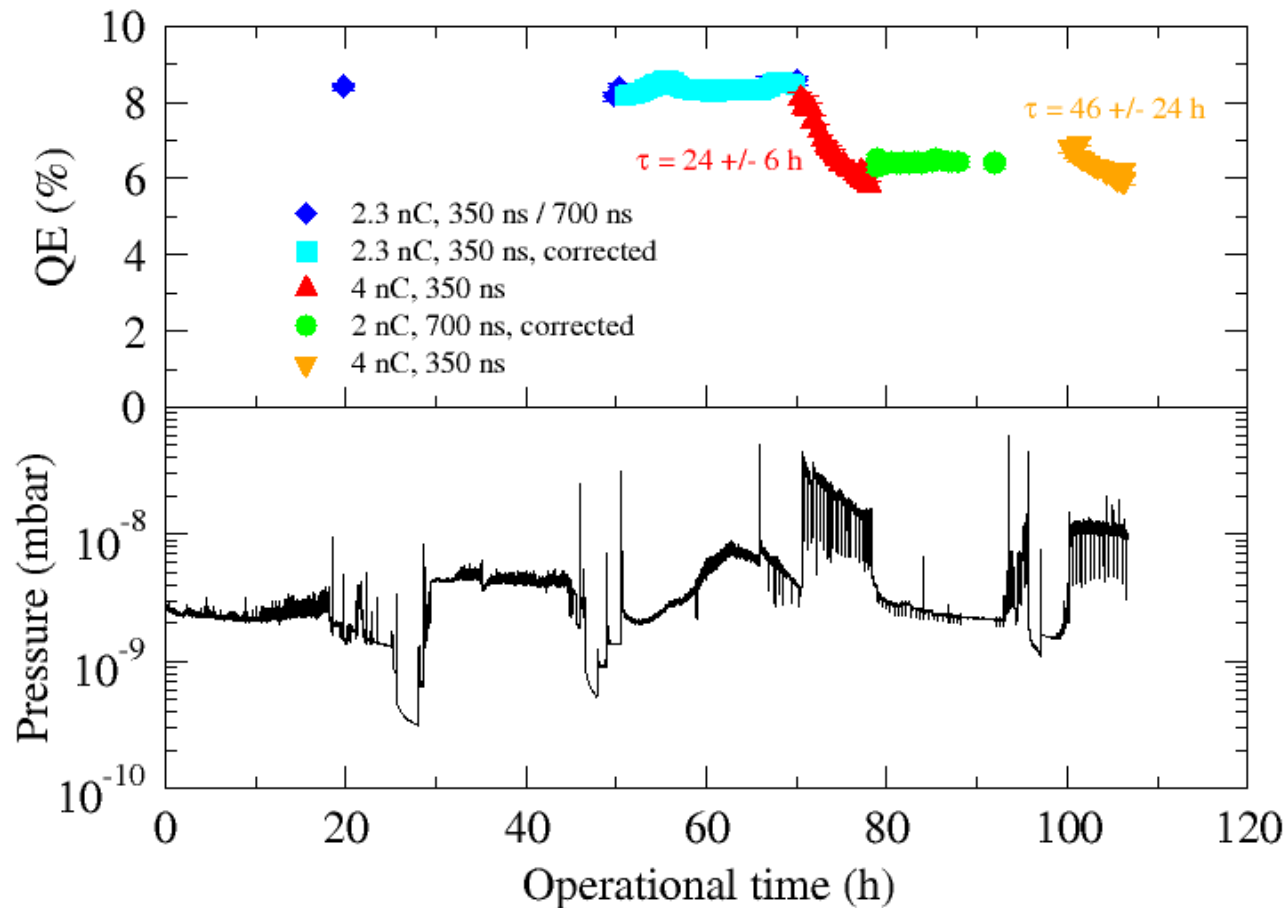
Comparison of lifetime and QE measured on photocathode **produced by co-deposition** and for cathode produced by **sequential deposition**.



R. Xiang et al., “*Report on photocathodes*”, CARE Note-2004-033-PHIN

Correlation between lifetime and vacuum.

Lifetime Studies of Cs_2Te Cathodes in PHIN photoinjector.

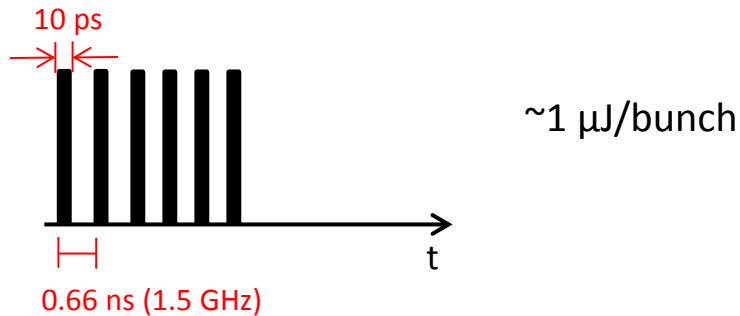


C. Hessler et al., "Lifetime studies of Cs_2Te cathodes at the PHIN RF photoinjector at CERN", Proc. of IPAC '12

PHIN laser time profile

Nominal parameters (Phase coding OFF)

Laser beam



Electron beam

