Heat transfer during bubble shrinking in saturated He II under microgravity condition

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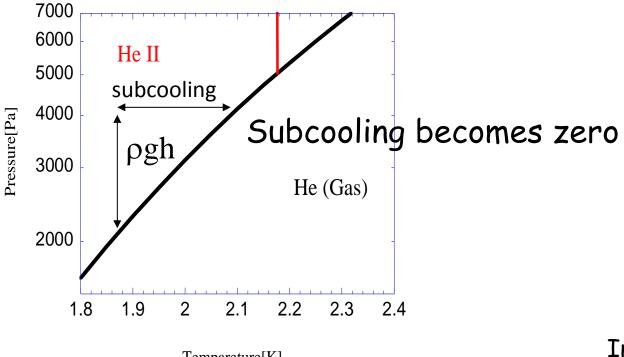
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 :drop tower
 :small cryostat with optical windows
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Background of this " μ -g" experiment

He II is utilized as a Coolant for space applications.





X-ray telescope ASTRO-H @ JAXA



Previous experiments : Critical heat flux for onset of boiling

Gorter-Millink equation for cylindrical cordinate

$$q_{cr} = \left(\frac{(m-1)\varphi}{r} \int_{T_b}^{T_i} \frac{dT}{f(T)}\right)^{1/m}$$

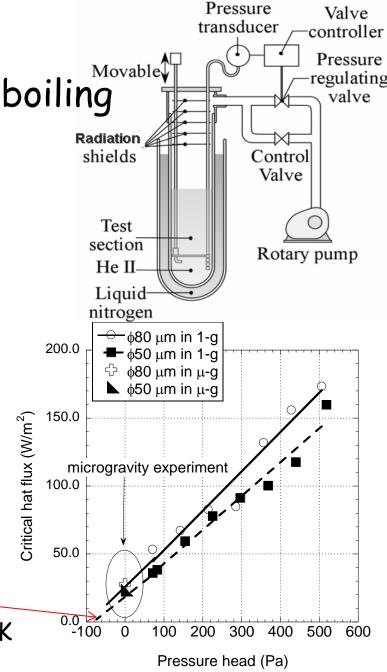
Rewritten using with Clausius-Clapeyron equation

$$q_{cr} = \left(\frac{(m-1)\varphi}{r} \cdot \frac{T_b}{\rho_v L} \cdot \frac{\Delta p_A}{f(T)}\right)^{1/m}$$

$$\Delta p_{\rm A} = \left(\frac{\rho_{\rm v}}{m_{\rm He4}}\right)^2 \cdot a$$

1 /

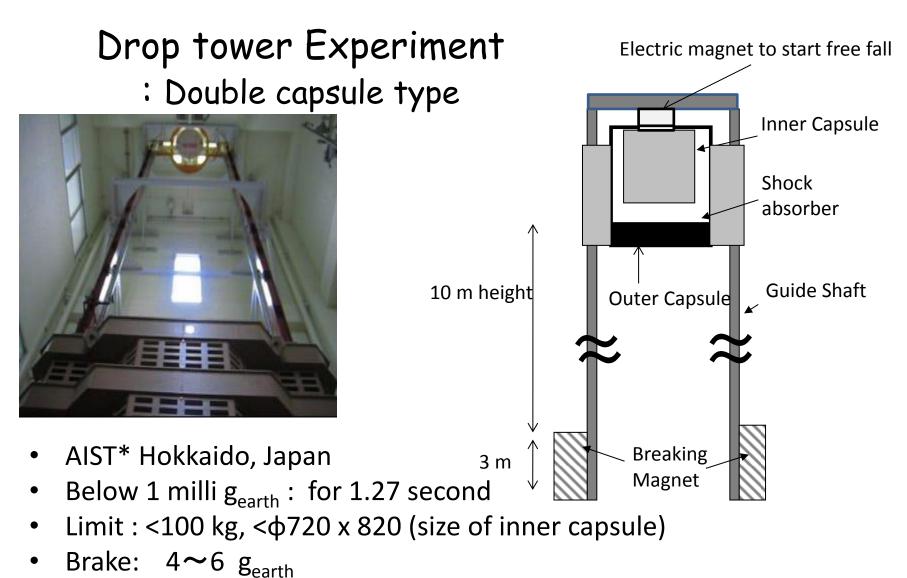
Offset pressure = van der Waals pressure ~80Pa at 1.9K (Not surface tension: ~20 Pa)



Remaining problem is after onset of boiling

How to express the heat transfer $\underline{across vapor-liquid interface}$ under μ -g condition.

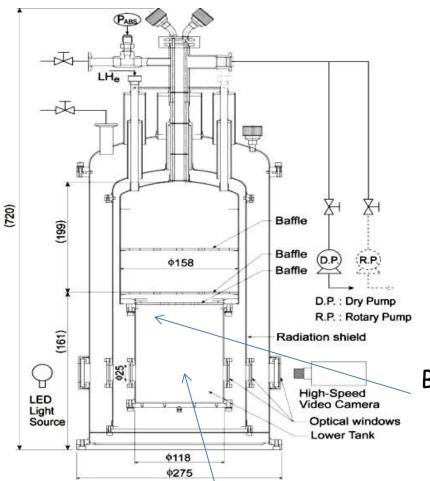
3. Experimental setup



- Turn around: 15 min
- Timing signal synchronized with Electro magnet

* National institute of advanced industrial science and technology, Japan 7

Small cryostat

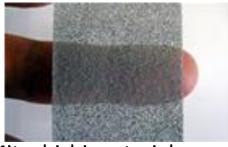


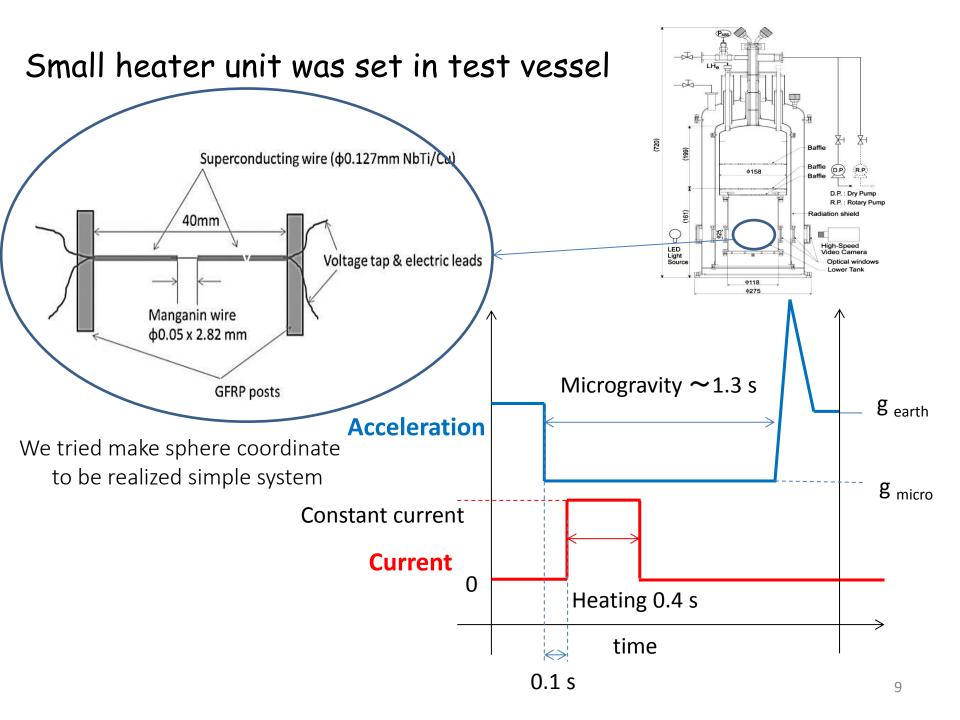
Only Liquid helium



All devices were set in inner capsule (φ720 x 800 mm)

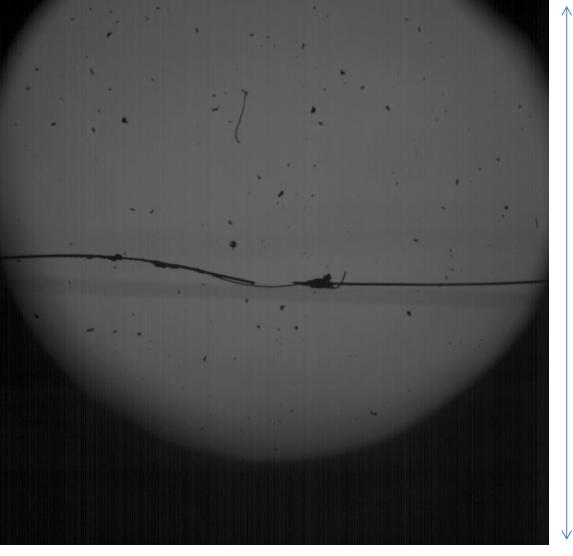
Baffle : porous metal to avoid initial flow due to capillary force





4. Results & Discussion

Visualization results (movie)

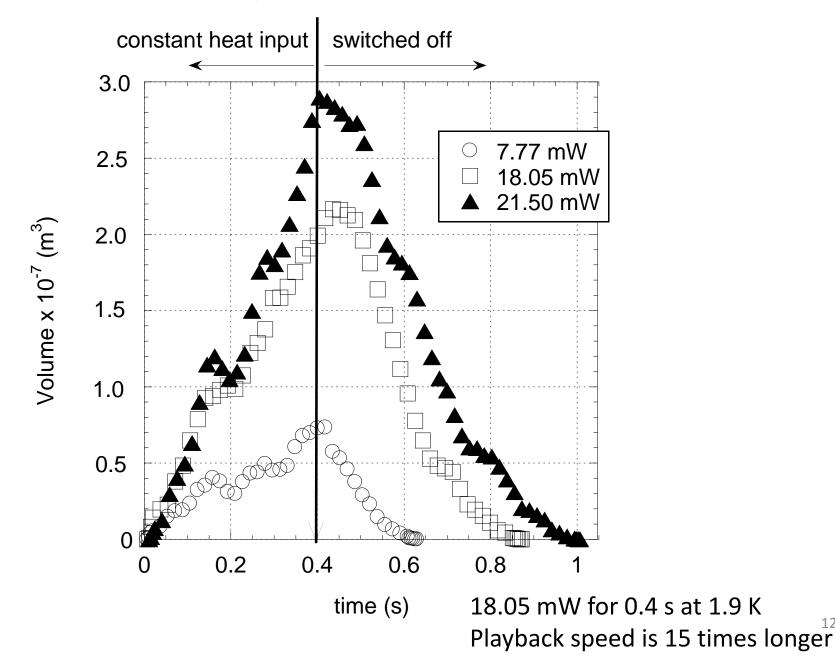


20 mm

18.05 mW for 0.4 s at 1.9 K Playback speed is 15 times longer

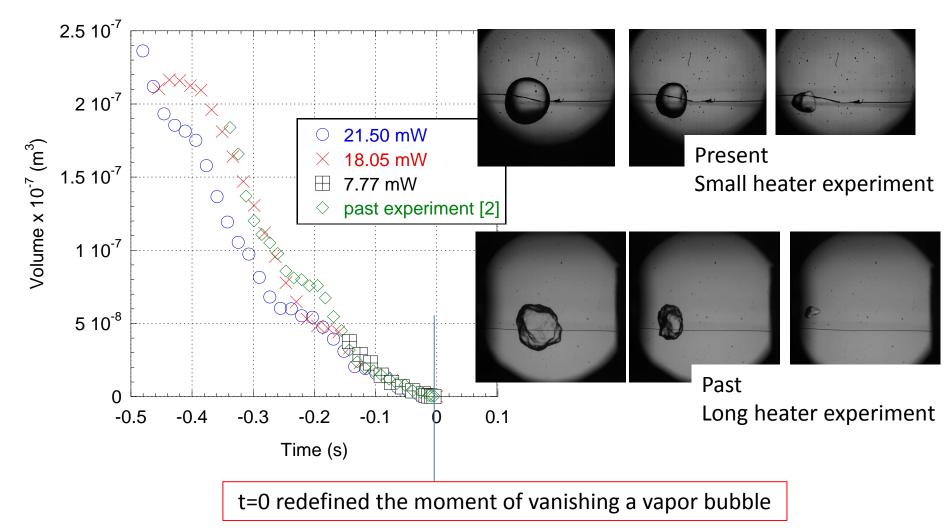
20 mm

Time variation of vapor volume by image analysis



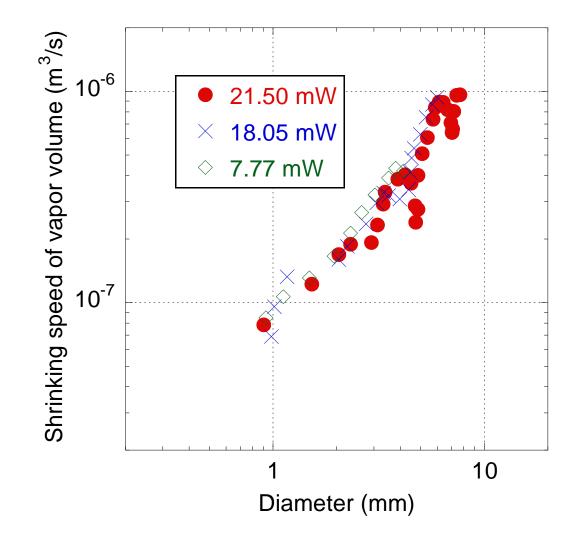
12

Replot of Time variation of vapor volume in shrinking stage



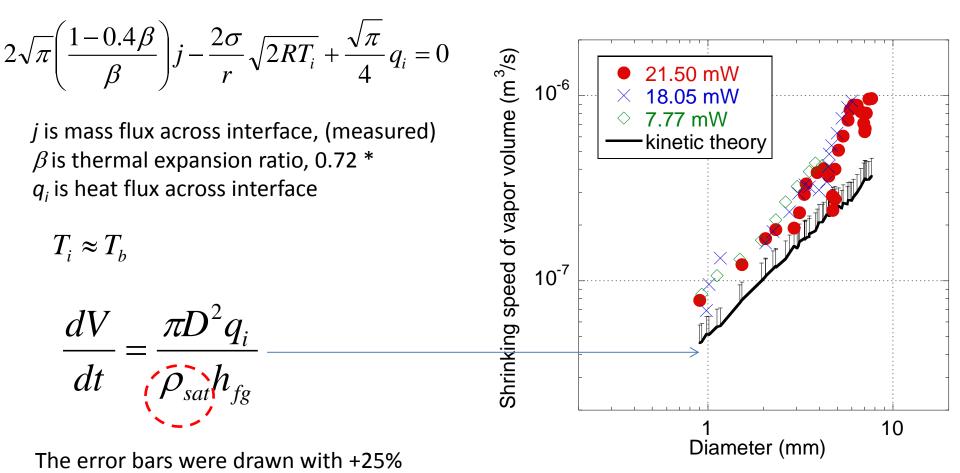
Shrinking speed depends on size dominantly. The vapor motion and gas condition inside a bubble has no significant effect to the bubble shrinking

Derivation dV/dt results from time variation of volume



The shrinking speed is roughly proportional to the diameter

The equation of energy balance on the interface based on kinetic theory

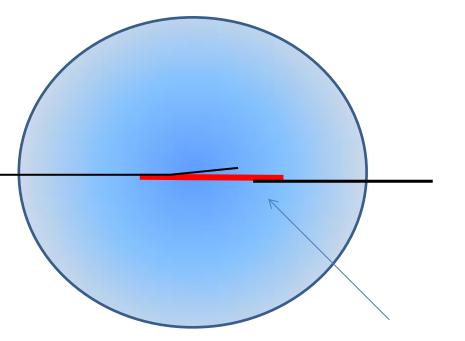


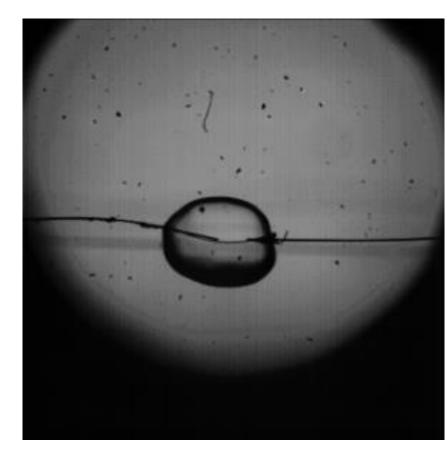
of the maximum ratio between the shadow area of the bubble and peripheral length.

Actually gas density is smaller than ρ_{sat} so that calculation result is smaller than the measured data.

*Murakami M., Furukawa T., Maki M., Fujiyama J., 2002 Experimental Thermal and Fluid Science, Volume 26 p 229-235

Rough Estimation of gas density inside a bubble.



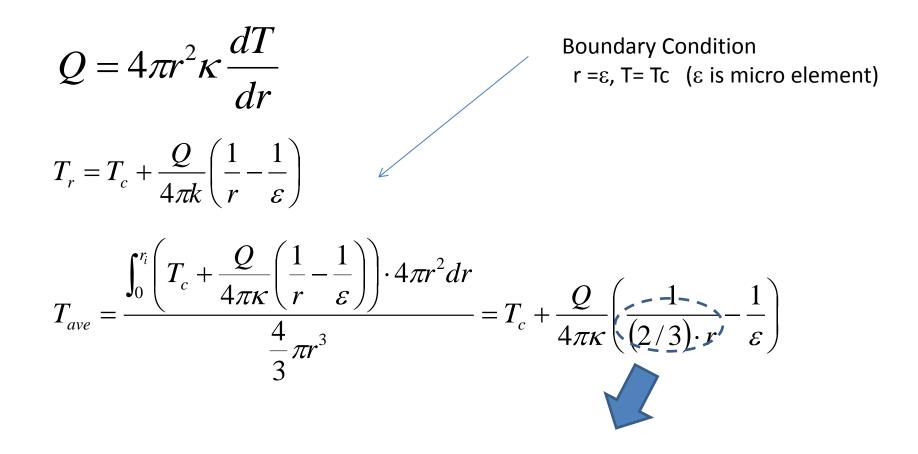


Known values:

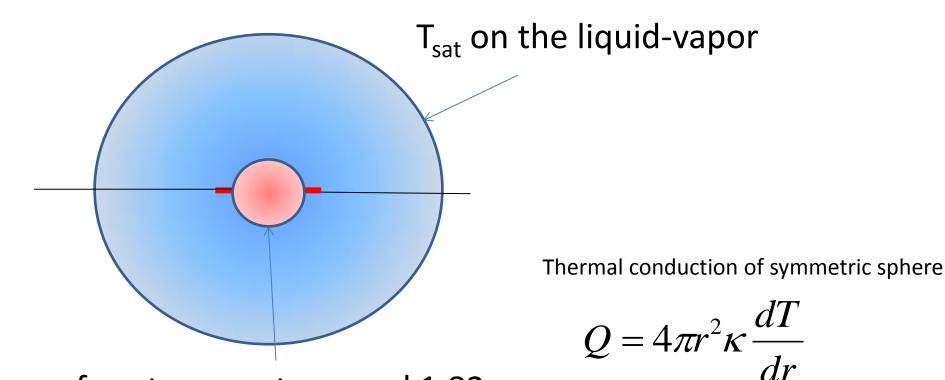
Heater temperature (resistance of Manganin) : 50 ~80 K Bubble size

Temperature on Liquid-vapor interface

Thermal conduction of through a sphere from a point heat source



Heater temperature is represent on the surface temperature of sphere in the 2/3 diameter of heater length.



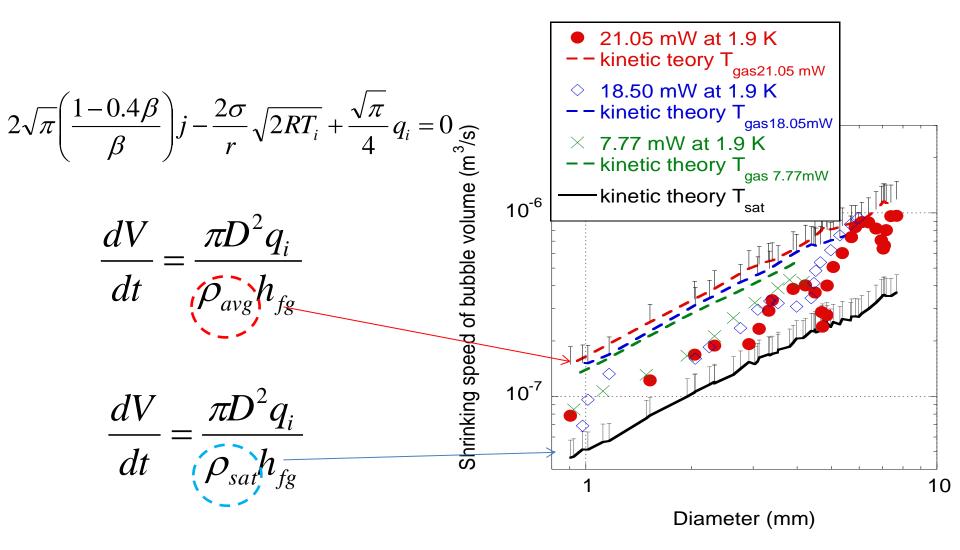
surface temperature on ϕ 1.82

= heater temperature

Effective thermal conductivity can be calculated

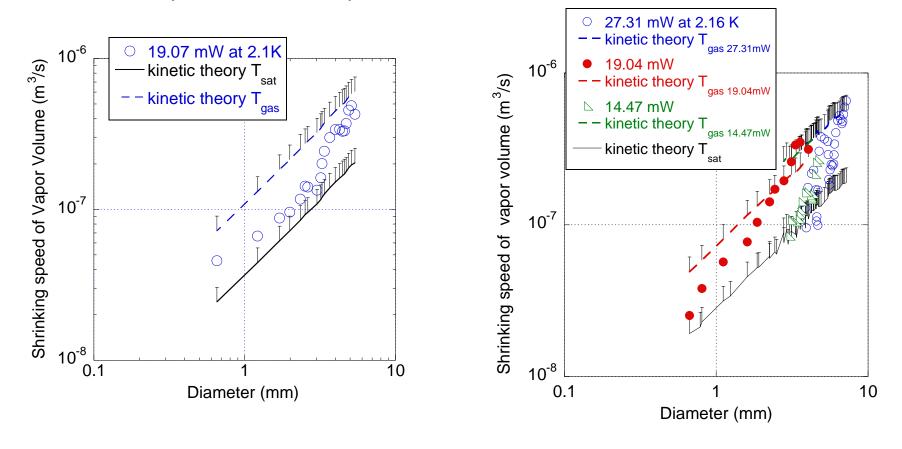
$$\kappa_{avg}(T_{avg}) = \frac{Q(r_i - r_w)}{4\pi r_i r_w (T_w - T_i)}$$

 T_{gas} is estimated by κ-T relation of Helium property the results are between 5~7K 18



Shrinking speed of bubble is related to the heat transport explained by the kinetic theory and the gas constriction

Bath Temperature dependence



2.1 K 2.16 K This phenomenon is not depending on the temperature difference between T_b and T_{λ}.

Kinetic theory is "universal".

But in other liquid this heat transport is hard to be observed because of ΔT in liquid phase,

gas density, and •••

Conclusion

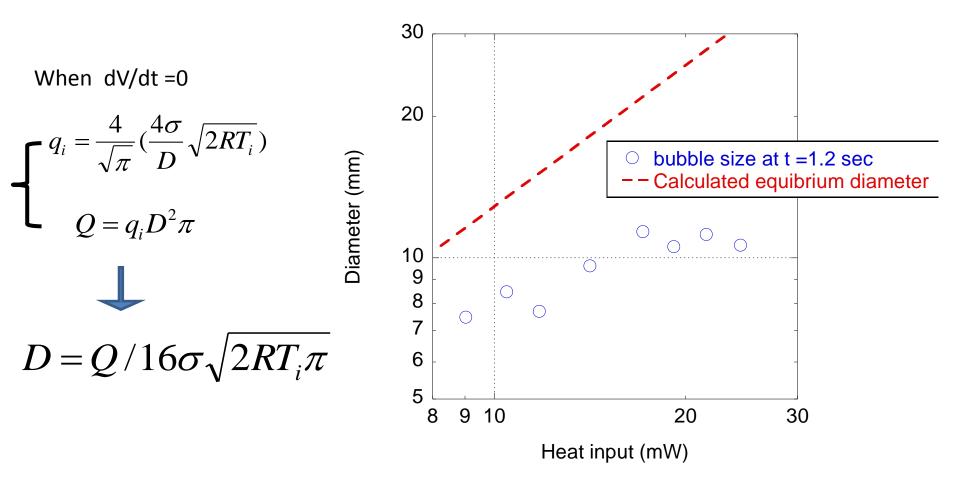
To investigate the heat transfer across the liquid-vapor interface under microgravity condition, a visualization experiment using a single shrinking bubble was carried out with a drop tower. Image analyses of pictures taken by a high-speed video camera reveal that

Findings:

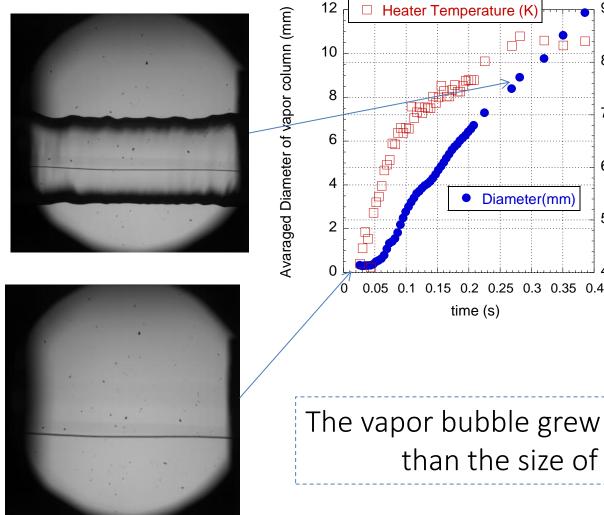
- The shrinking speed of the bubble under the microgravity condition is related to the heat transport across the vapor-liquid boundary as explained by <u>the kinetic theory and gas</u> <u>constriction</u>.
- The pressure difference due to <u>surface tension</u> plays a role for the heat transfer across the liquid-vapor interface in microgravity. The van der Waals pressure plays a role only <u>for the critical heat flux of onset boiling</u>.

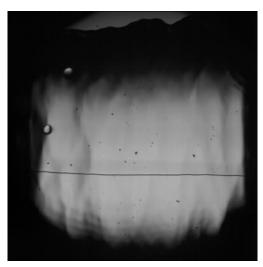
Thank you for your attention

"In addition" Equilibrium diameter can be calculated



Why we don't long wire heater. previous study (Takada S., et al, ICEC24)





Larger than the window

The vapor bubble grew larger than the size of the optical window

90

80

70

60

50

40

Heater Temperature (K)