

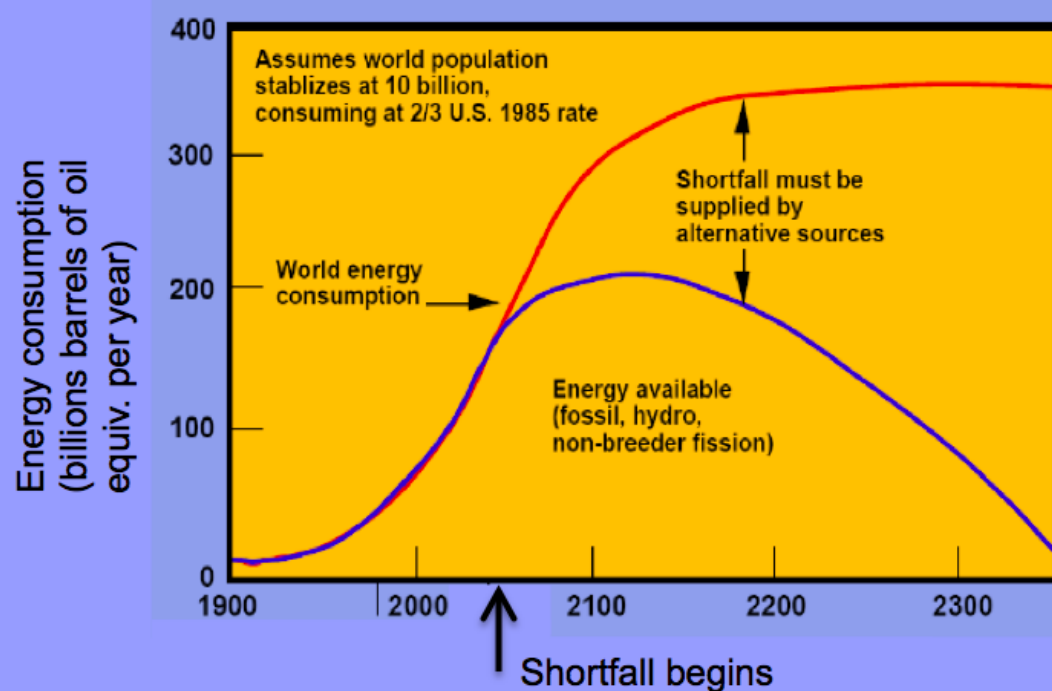


# Surface Analysis Techniques for the Study of Plasma Wall Interactions

**Nicolas Bekris**

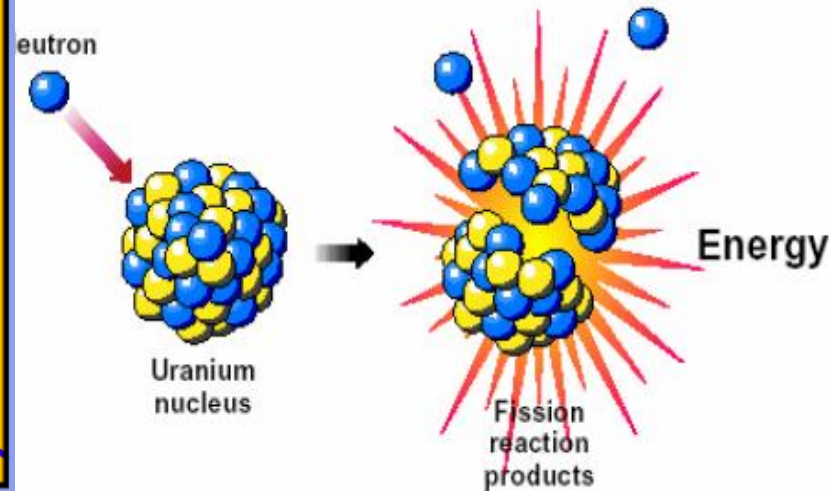


# Why Fusion?

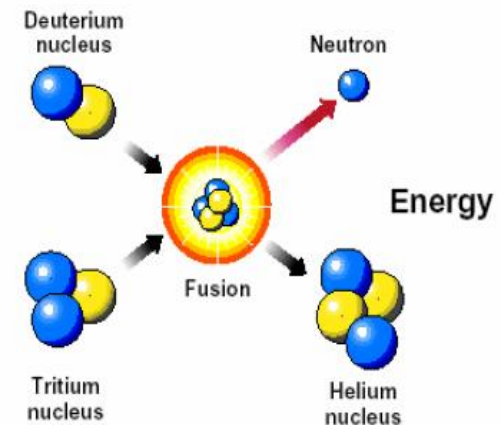


Fusion reaction has the potential to provide energy in the future in a sustainable way exploiting reactions between hydrogen ions

## Fission reaction



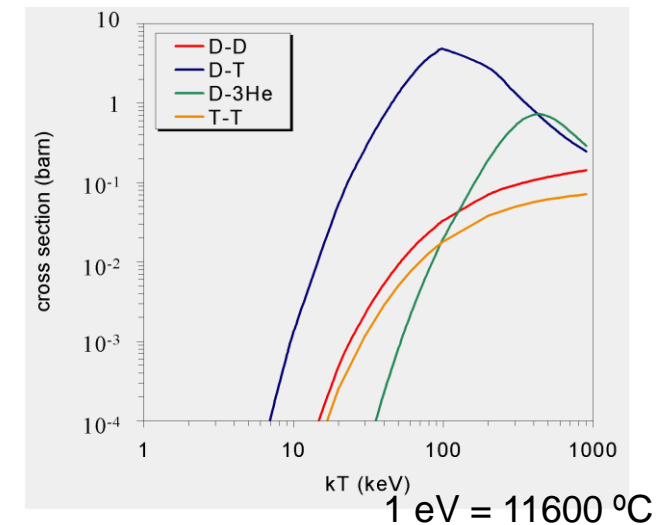
## Fusion reaction



# Fusion Energy Research; Ions, Neutrons & Fusion Products



Examples of Fusion Reactions	Temperatures Needed
$2\text{H} + 2\text{H} \rightarrow {}^3\text{He} + \text{n} + 3.2 \text{ MeV}$	170 Million Degrees
$2\text{H} + 2\text{H} \rightarrow {}^3\text{H} + {}^1\text{H} + 4.0 \text{ MeV}$	170 Million Degrees
$2\text{H} + {}^3\text{H} \rightarrow {}^4\text{He} + \text{n} + 17.6 \text{ MeV}$	80 Million Degrees
$2\text{H} + {}^3\text{He} \rightarrow {}^4\text{He} + {}^1\text{H} + 18.3 \text{ MeV}$	400 Million Degrees
D-T is favoured: - Higher cross-section, Best yield at lowest temperature (ion energy)	



Need *containment* at high temperature and density

In the Sun – Gravitational containment

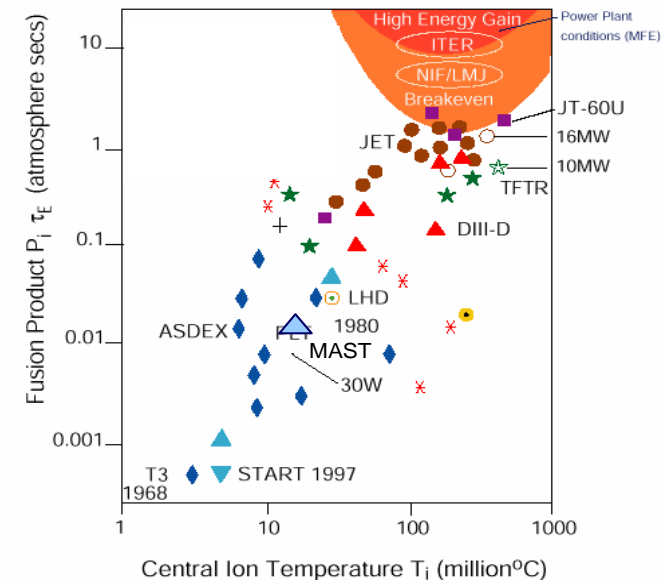
On Earth – Magnetic containment

– ‘Tokamaks’ like JET are now favoured

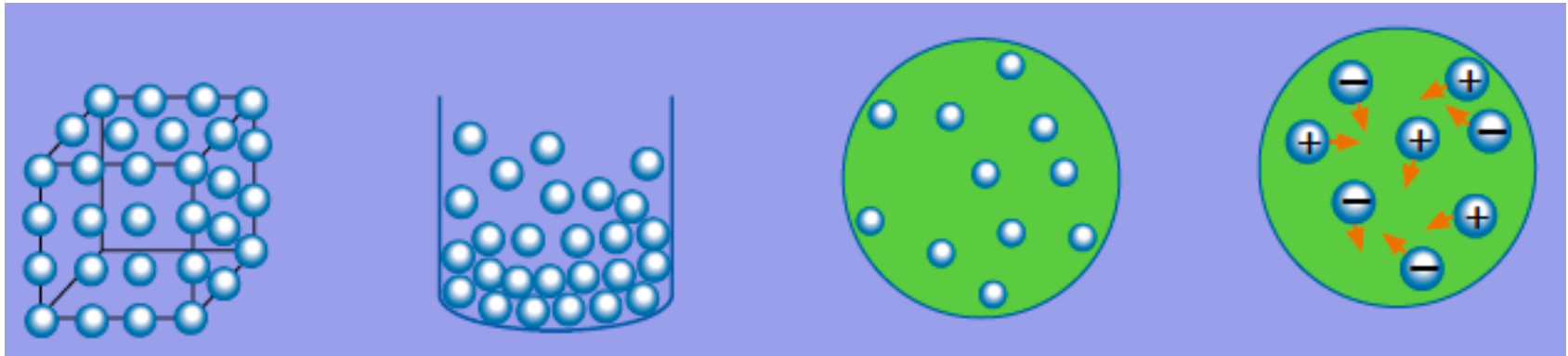
■ **Maximise Fusion Product =  $n_i T_i \tau_E$**

Arising points for Diagnostics

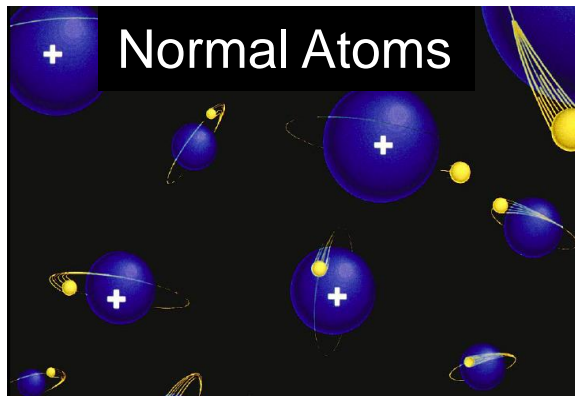
- Neutrons measure the fusion yield – directly
- Same number of p,  ${}^3\text{T}$  ions as n in a D,D plasma
- There are  ${}^3\text{T}$  ions + 14 MeV neutrons, even in D,D plasmas
  - From Triton ‘burnup’
  - And tritium adsorbed in the walls



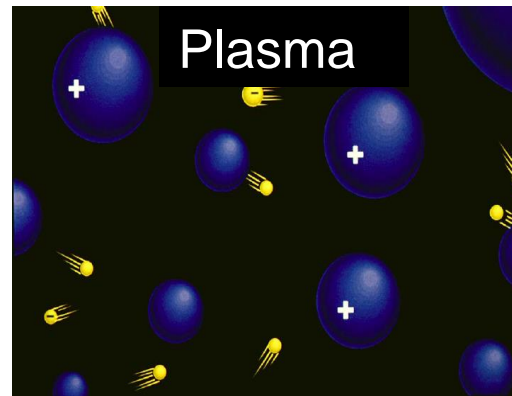
# What is the Plasma?



**Solid**  $\xrightarrow{\text{Heat}}$  **Liquid**  $\xrightarrow{\text{Heat}}$  **Gas**  $\xrightarrow{\text{Heat}}$  **Plasma**

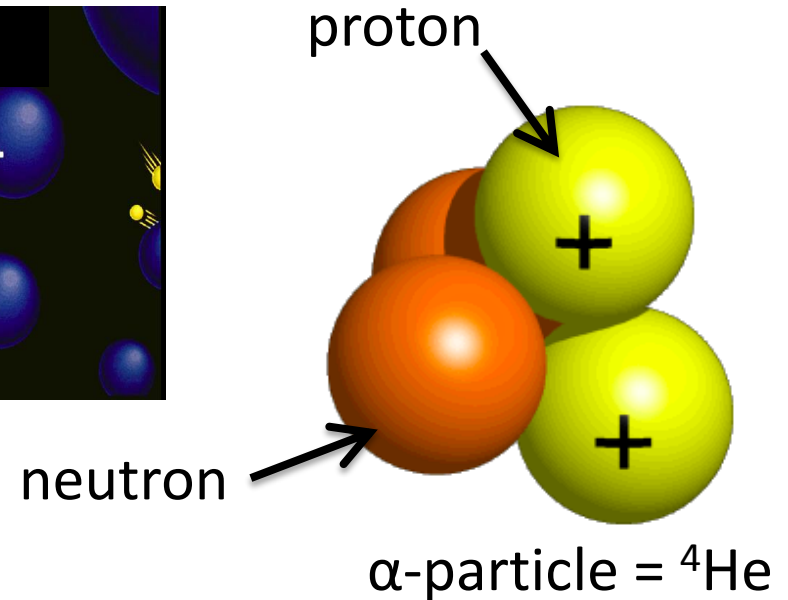


Normal Atoms



Plasma

$2\text{H}^+$ ,  $3\text{H}^+$ ,  $n$ ,  $e^-$  and alphas





A wide-angle, fisheye photograph looking into the central column of the JET tokamak. The image shows the complex, multi-layered structure of the vacuum chamber, with numerous metallic tiles and components arranged in a circular pattern. The lighting is bright, highlighting the metallic surfaces and the intricate details of the machine's interior.

## Joint European Torus

investigates the potential of fusion power as a safe, clean, and virtually limitless energy source for future generations.



## Bulk W

## Saddle coil protection

## Upper Dump Plate

## Restraint Ring Protections

# Mushrooms

## Poloidal Limiters

## Inner Wall Cladding

## Normal NBI Inner Wall GL's

## Re-ionisation Protections

## Inner Wall Guard Limiters

## Normal NBI IW Cladding

## Magnetic covers

## LH + ICRH Protection

**B&C  
tiles**

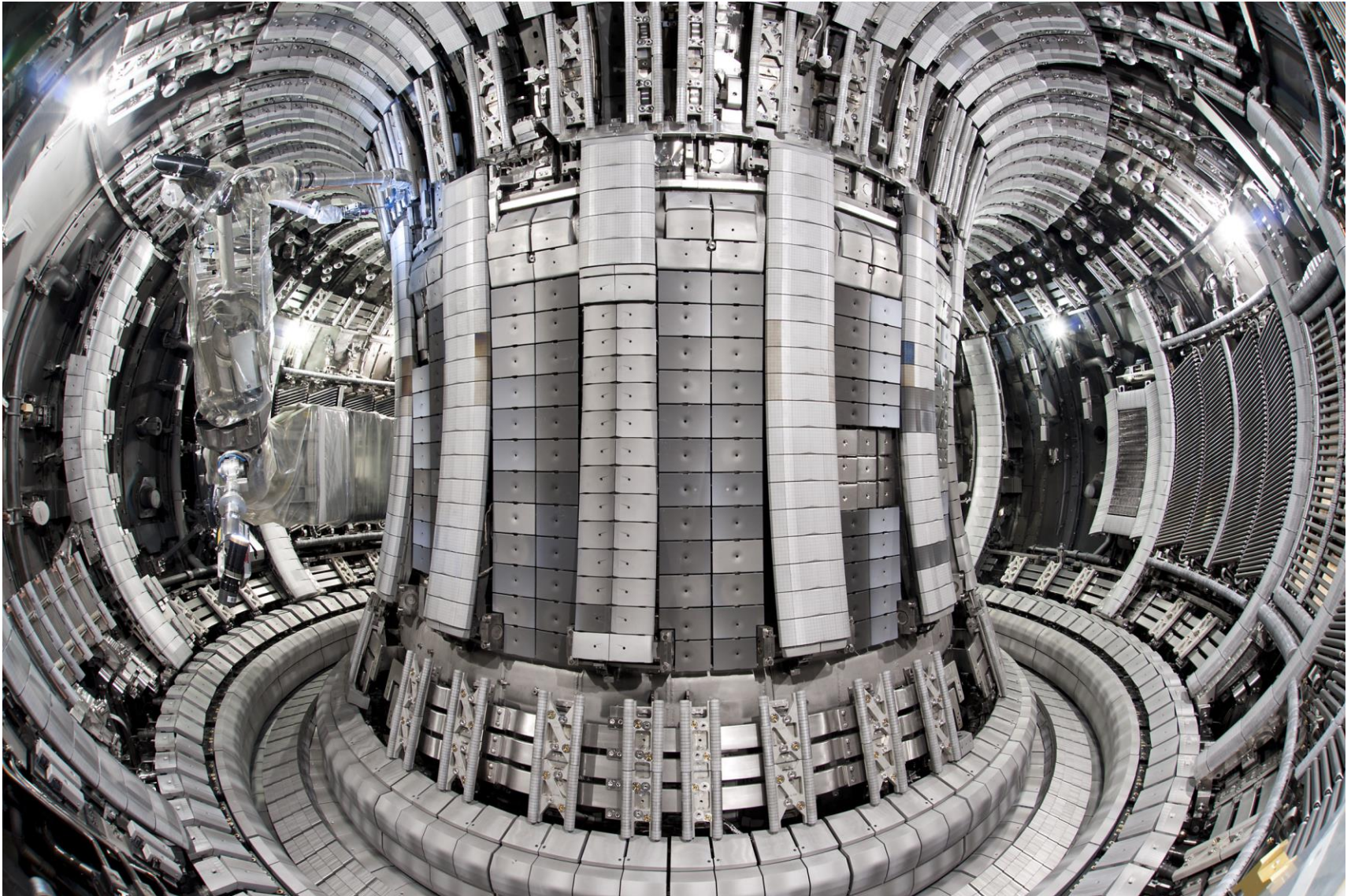
## Saddle Coil Protections

## Divertor

## Bulk W



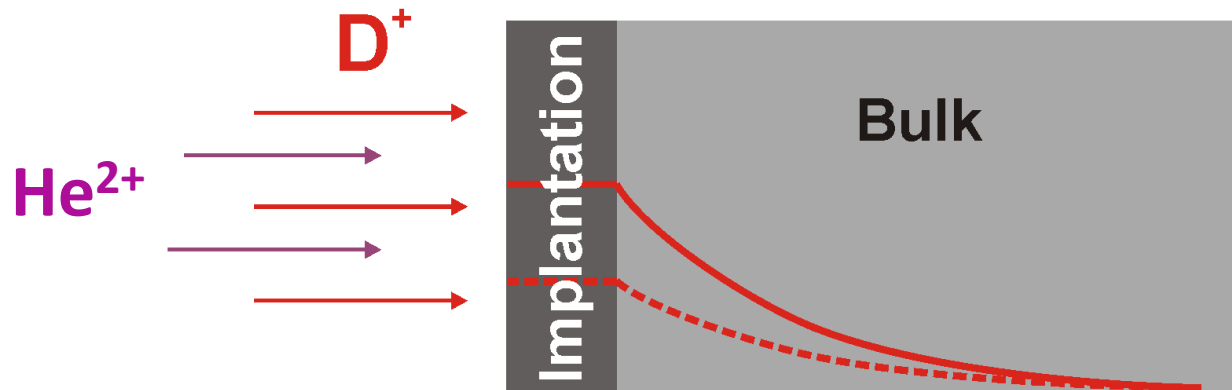
# JET - The ITER-like Wall –May 2011



# Plasma Wall Interaction Consequences



1. Incident deuterium ions saturate the implantation layer (~ up few hundred nm)
2. The level of saturation is defined by the [D] the nature of the FWM, but when saturation is reached then we have a dynamic balance between adsorption and ion-induced desorption
3. From the implantation layer, D 'diffuses' further into the bulk material along the grain surfaces (inter-grain diffusion) or inside the grains (intra-grain diffusion) of the pores. Diffusion is temperature depending and also on some physical parameter (such as weave of the CFC tiles)







**During Fusion reaction tritium leaves the plasma and after interaction with the first wall materials remains in the wall increasing the vessel inventory**

**(in ITER safety limit 700 g tritium inside the machine)**

## **Questions:**

- 1. What is the distribution of the eroded material inside the vacuum vessel and where are the deposition zones ?***
- 2. Where, how and why fuel is stored ?***
- 3. How does the material erosion and transport mechanism influence the fuel inventory ?***

## **Why Care about Fuel Loss ?**

### **Cost & Reactor safety**

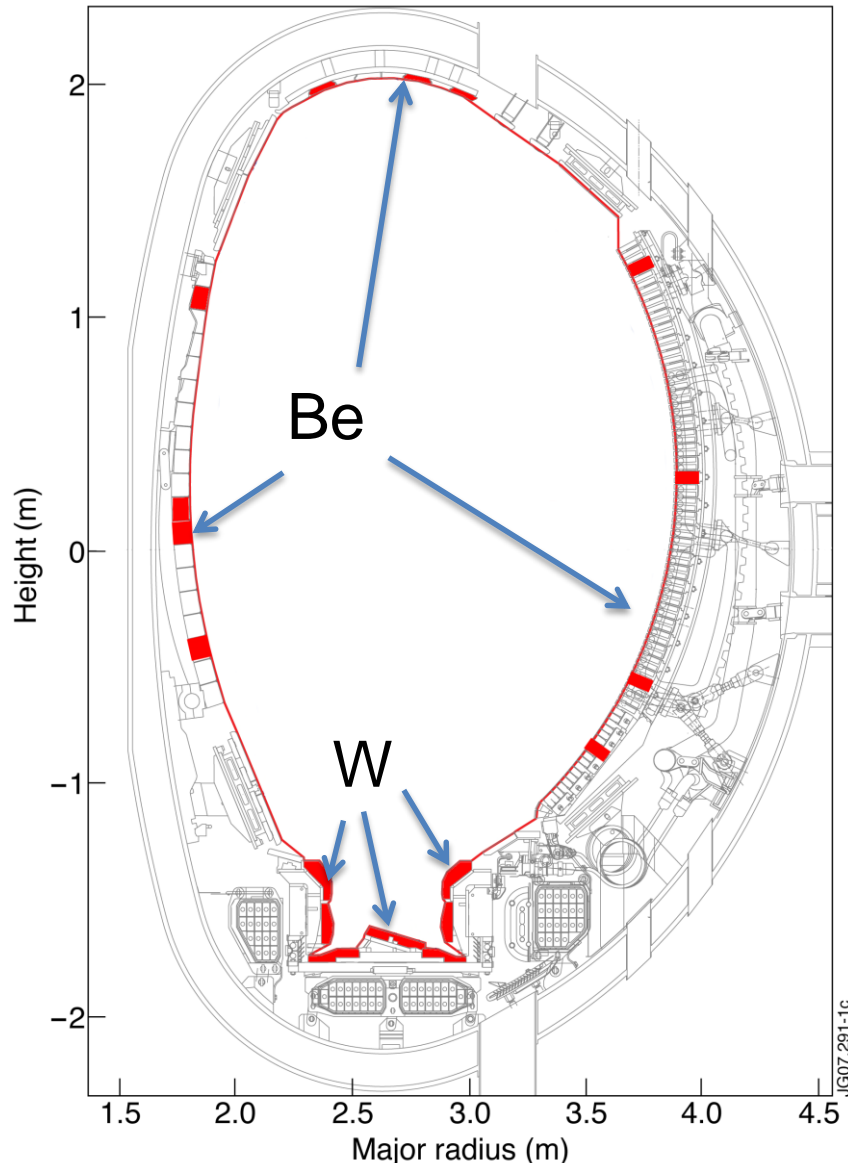
One bottle of 6400L at STP D<sub>2</sub> cost ~£ 5000

**Cost:** Tritium is bought from Canada and cost ~ £ 65 000/g (Au £24/g) if you think that ITER will use about 10-15 **kg** (1-1.2 **kg**/y) tritium during its lifetime operation....

**Safety:** 1 g tritium ~10 000 curies (1Ci= 3.7 E10 decays/s)

Permissible release rate = 1 curie/day

# Post-Mortem Surface Analysis



## Post mortem analysis of tiles

- Poloidal cross section removed from vessel
- Some tiles have marker coatings

### Non-destructive

Ion Beam Analysis;  
(RBS, NRA, PIXE, PIGE)  
& IP (auto-radiography)

### Destructive

SIMS, AMS, FC-LSC



# The Coating Principle

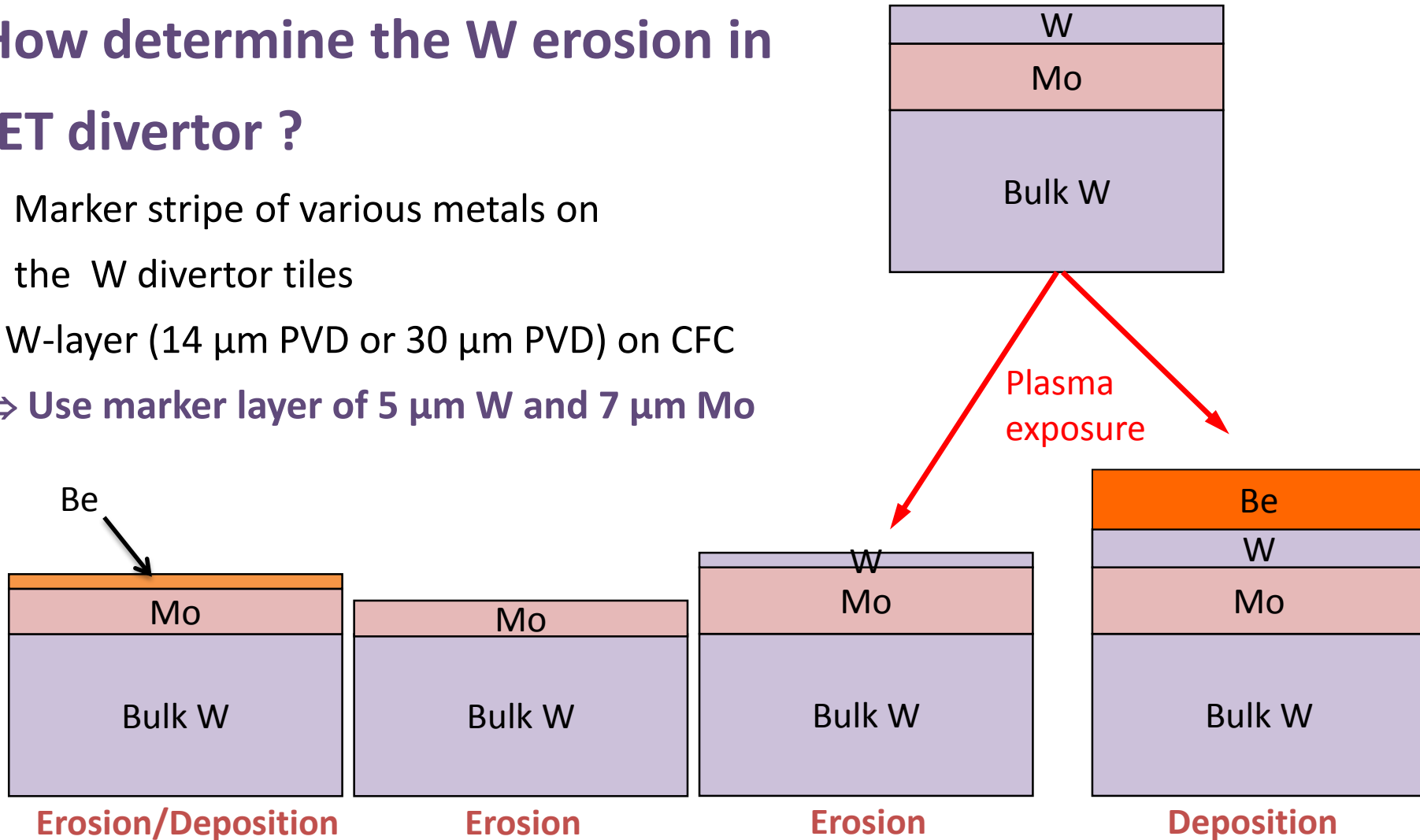


## How determine the W erosion in JET divertor ?

- Marker stripe of various metals on the W divertor tiles

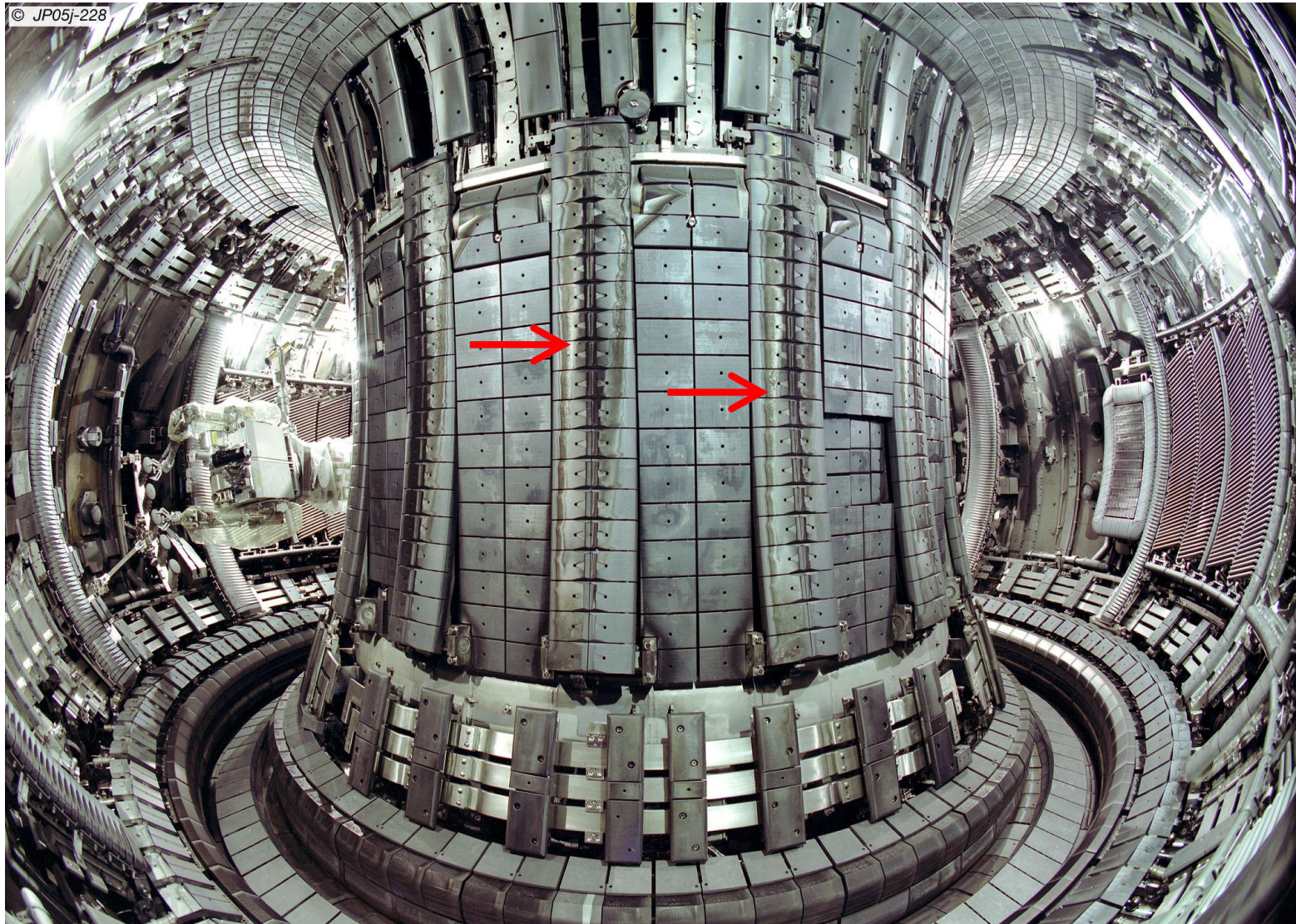
W-layer (14  $\mu\text{m}$  PVD or 30  $\mu\text{m}$  PVD) on CFC

⇒ Use marker layer of 5  $\mu\text{m}$  W and 7  $\mu\text{m}$  Mo



*M. Mayer, JW8-FT-3.41, Tungsten Erosion of JET Divertor. JET Report*

# JET 2009 – Carbon wall





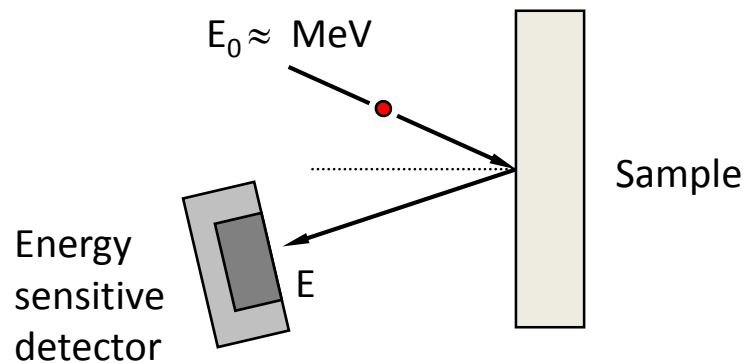
# I. Non-destructive Surface Analysis



## IBA – Ion Beam Analysis

### Group of methods for the near-surface layer analysis of solids

- Elemental composition and depth profiling of individual elements
- Quantitative without reference samples
- No (or small) matrix effects
- Non-destructive
- Analysed depth: 0.5  $\mu\text{m}$  (heavy ions) to 40  $\mu\text{m}$  (using protons)
- High sensitivity  $\approx$  ppm



1. Detection of charged particles
2. Detection of photons

# IBA: 1. Detection of charged particles

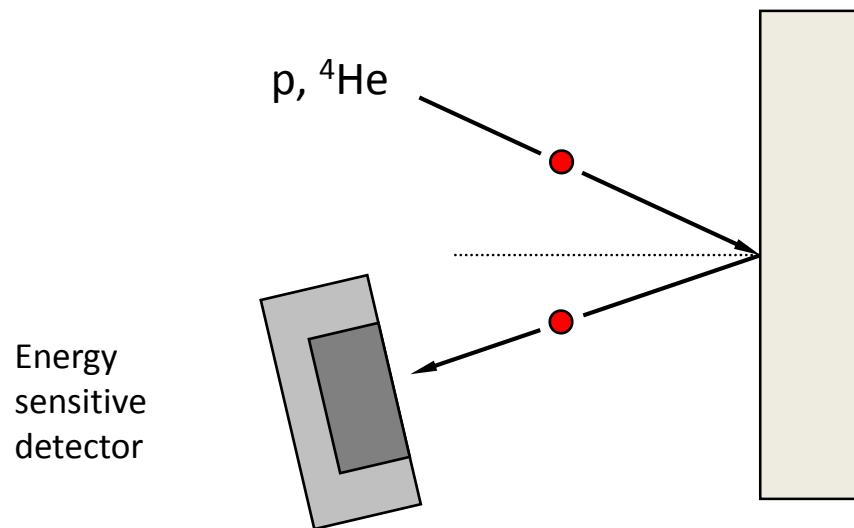


**What is common for all IBA methods: Incident particles have MeV energies**

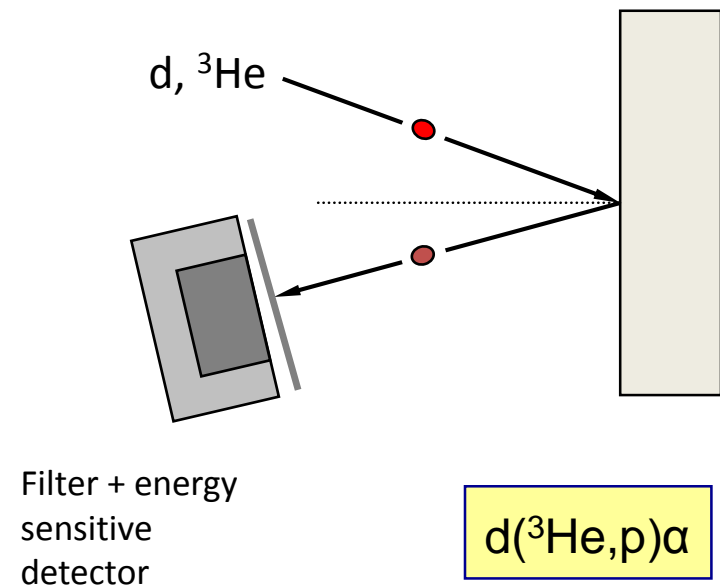
⇒ Methods can often be used in the same experimental setup

⇒ Methods can sometimes be used simultaneously

## Rutherford Backscattering (RBS)



## Nuclear Reaction Analysis (NRA)

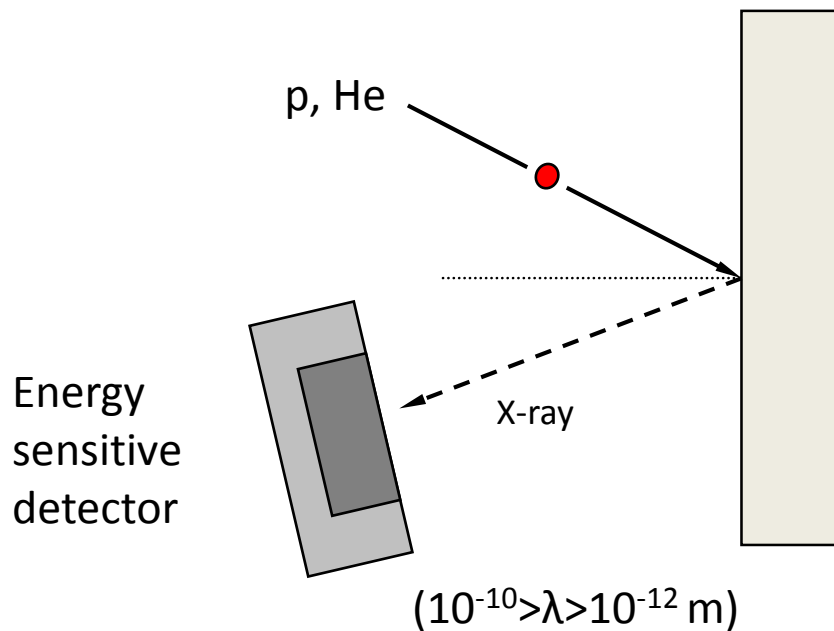




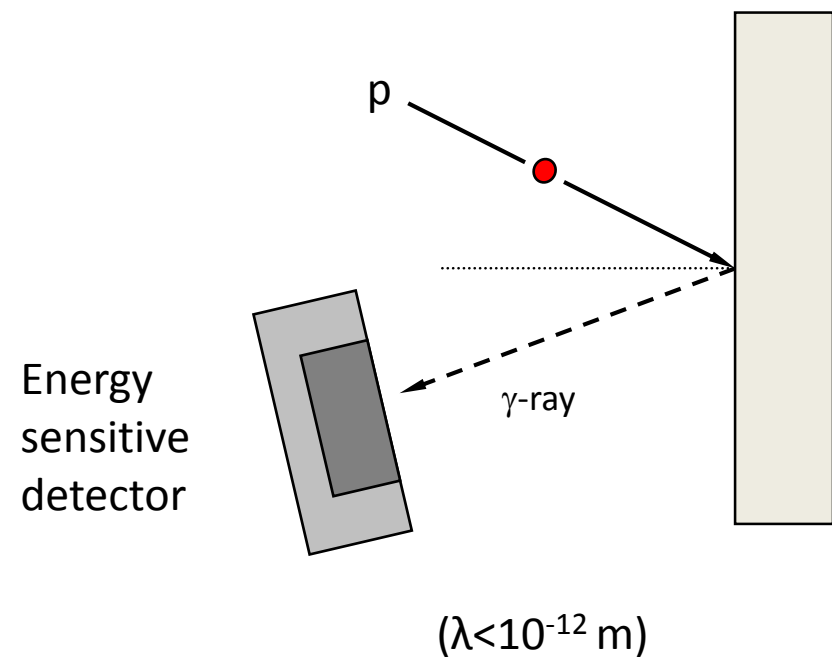
# IBA methods: 2. Detection of photons



## Particle induced X-ray emission (PIXE)



## Particle induced $\gamma$ -ray emission (PIGE)



# The Rutherford Backscattering Spectroscopy



Sir Ernest Rutherford (1871 - 1937)

- 1909-1914: Rutherford's scattering experiments:  $^4\text{He}$  on Au

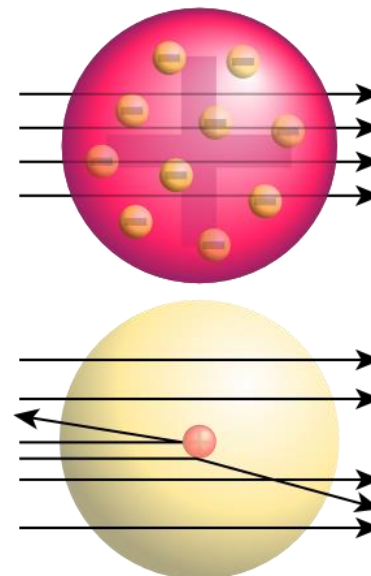
Rutherford supervised a series of experiments carried out by Hans Geiger and Ernest Marsden between 1909 and 1914 studying the scattering of alpha particles through metal foils.

Rutherford suggested that Marsden attempt to measure backscattering from a gold foil sample. According to the then-dominant plum-pudding model of the atom (J.J. Thomson 1904), backscattering of the high-energy positive alpha particles should have been non-existent

At most, only small deflections should have occurred as the alpha particles passed almost unhindered through the foil.

Instead, when Marsden positioned the detector on the same side of the foil as the alpha particle source.....He immediately detected a noticeable backscattered signal. Obviously there were many collisions!!

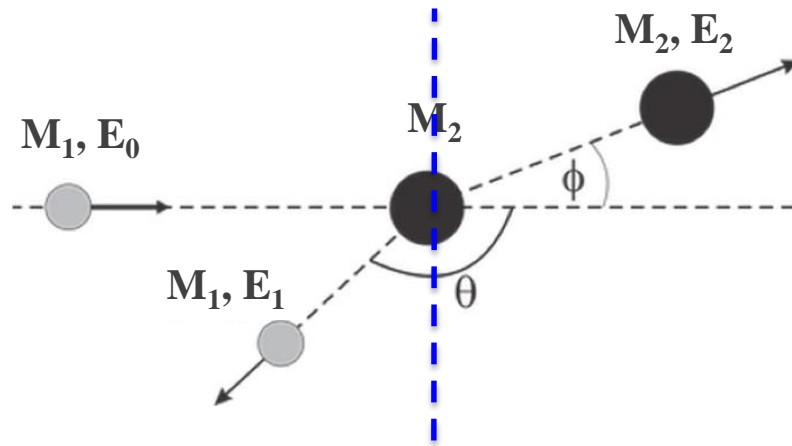
*"It was quite the most incredible event that has ever happened to me in my life. It was almost as incredible as if you fired a 15-inch shell at a piece of tissue-paper and it came back and hit you." (E. Rutherford)*



# RBS: Elastic Scattering Kinematics



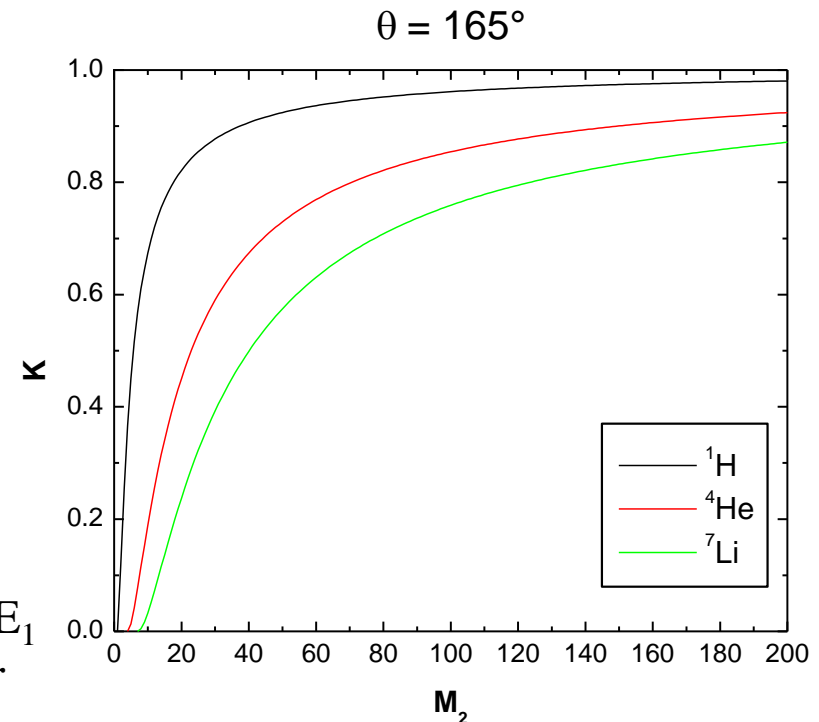
During the collision, energy is transferred from the incident particle to the target atom. The change in energy of the scattered particle depends on both masses. During this elastic scattering the total energy is preserved :  $E_0 = E_1 + E_2$



After the collision, the residual energy  $E_1$  of the scattered particle at angle  $\theta$  can be expressed as:  $E_1 = k E_0$  where  $k$  is known as the *kinematical factor*

$$K = \left[ \frac{(M_2^2 - M_1^2 \sin^2 \theta)^{1/2} + M_1 \cos \theta}{M_1 + M_2} \right]^2$$

⇒ mass resolution decreases for heavier elements



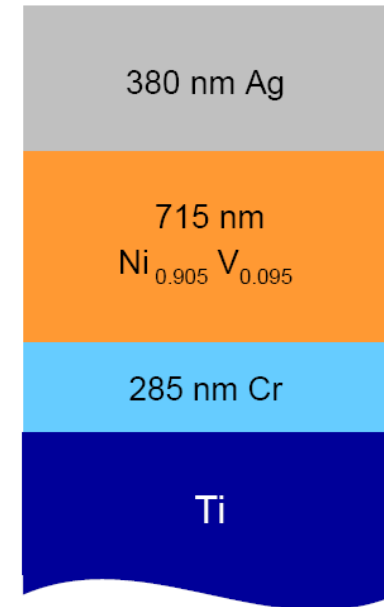
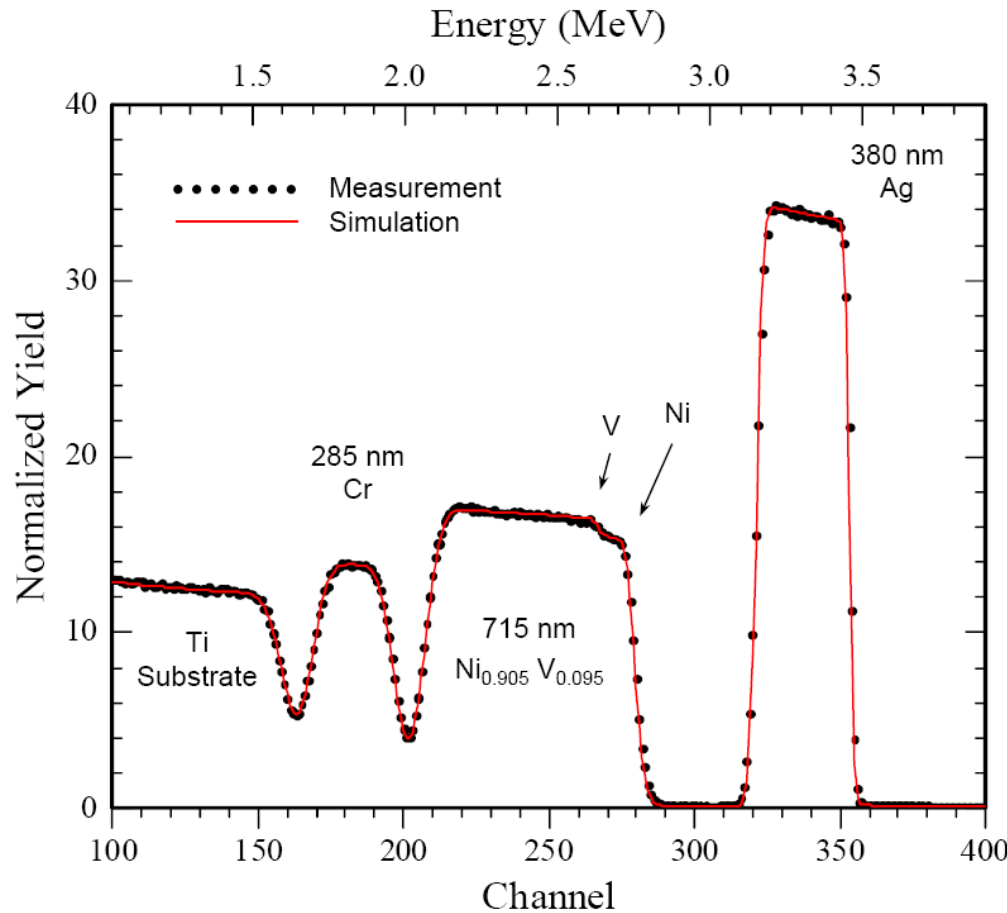
$$\Delta E_1 = E_0 \frac{dK}{dM_2} \Delta M_2$$

$\Delta E_1$ : Energy separation

$\Delta M_2$ : Mass difference



# RBS Typical spectrum



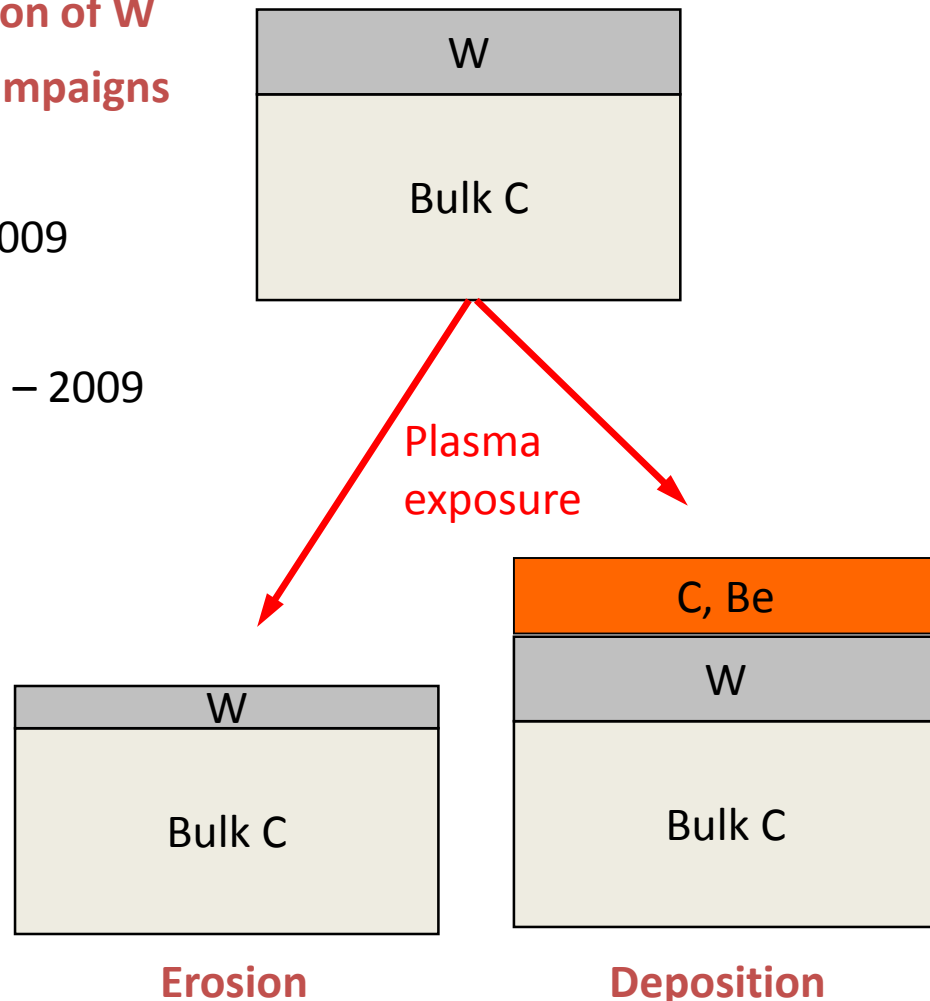
The Ag peak is well separated from the substrate signal, which is due to the higher mass ( $M_2$ ) of Ag (108) compared to the rest of the transition elements (Ti, Cr, Ni, V  $m=47-64$ ). In this elastic scattering process, the He particles ( $M_1$ ) are scattered with a much higher recoil energy than the He from the substrate.

**RBS at JET is used mostly to determine erosion of W and C markers during the last operational campaigns of JET**

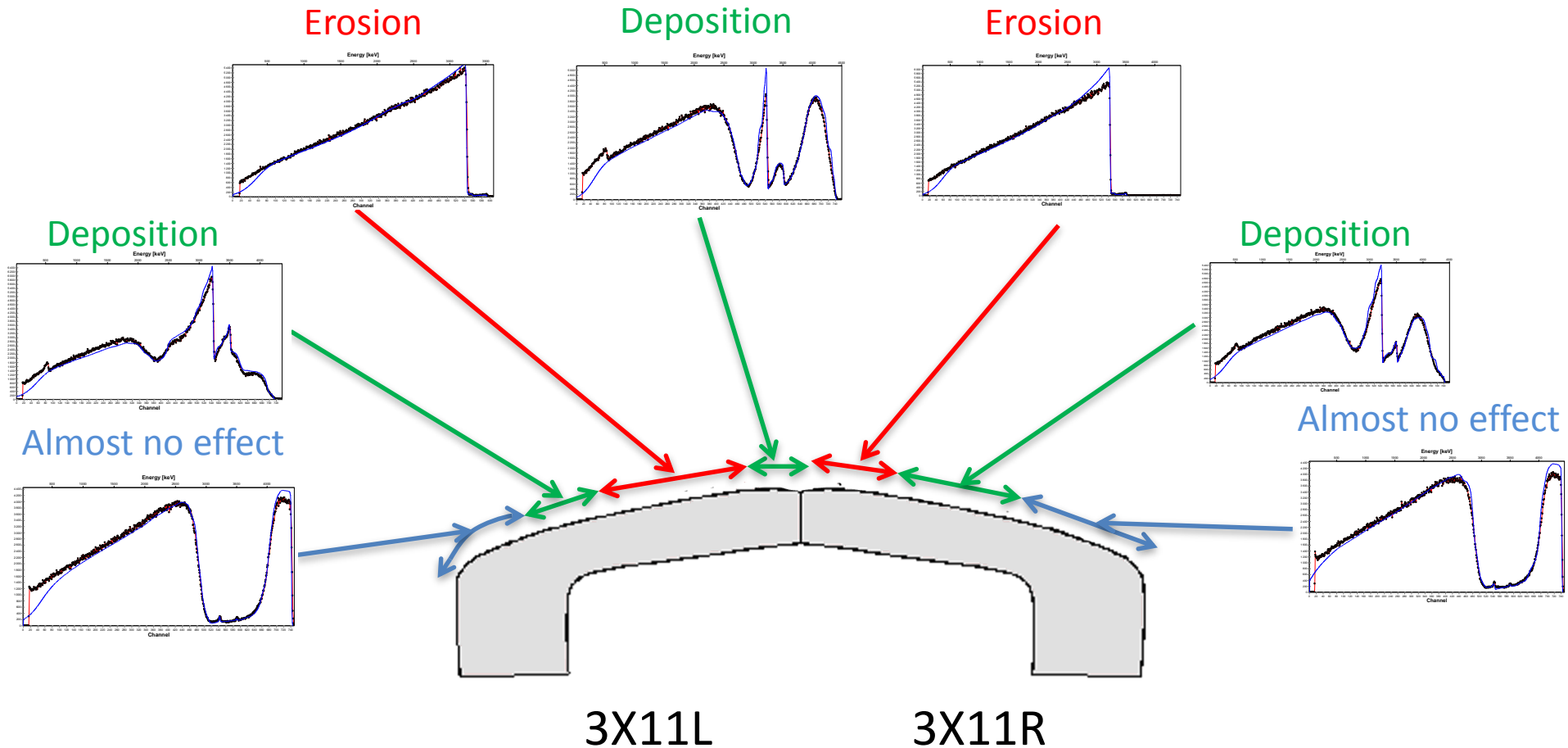
- Divertor tiles LBSRP, 7, 8 exposed 2007 – 2009
- Divertor tiles 6, 7, 8 exposed 2004 – 2009
- Inner wall guard limiter tiles exposed 2004 – 2009

**To determine the erosion taking place at the inner wall**

- 9 sachet samples (2004 – 2009)
- 2 flap units exposed 2004 - 2009



# RBS: Erosion – Deposition map



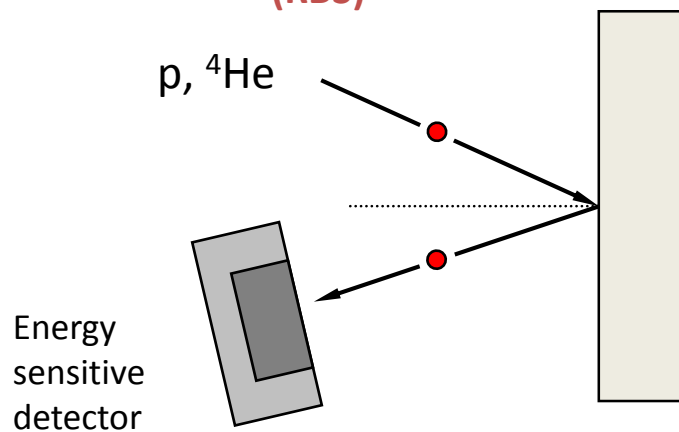
- Total amount of deposited C, extrapolated to the whole limiter: 109 g
- Erosion in erosion-dominated areas  $> 3 \mu\text{m W}$



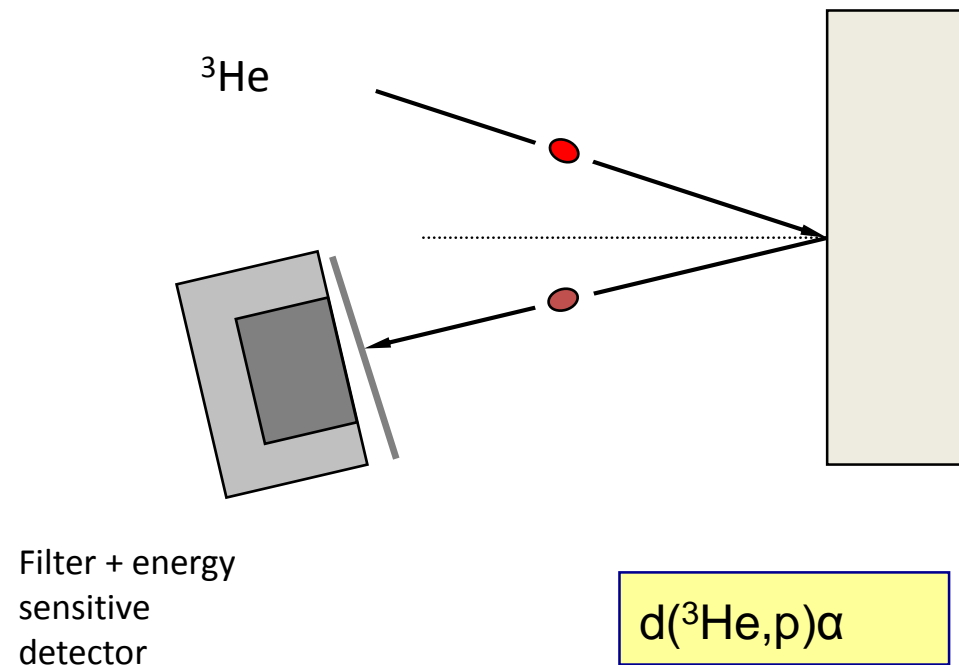
# IBA: 1. Detection of charged particles



## Rutherford Backscattering (RBS)



## Nuclear Reaction Analysis (NRA)



# Nuclear Reaction Analysis

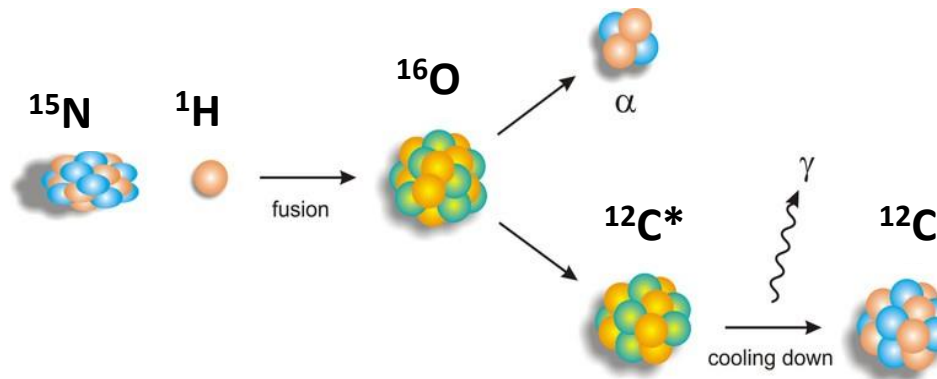


The NRA uses accelerated particles, which initiate a nuclear reaction.

The emitted radiation is characteristic for this reaction and can then be detected while from **the intensity of the radiation we can determine the concentration** of the particular atomic species.

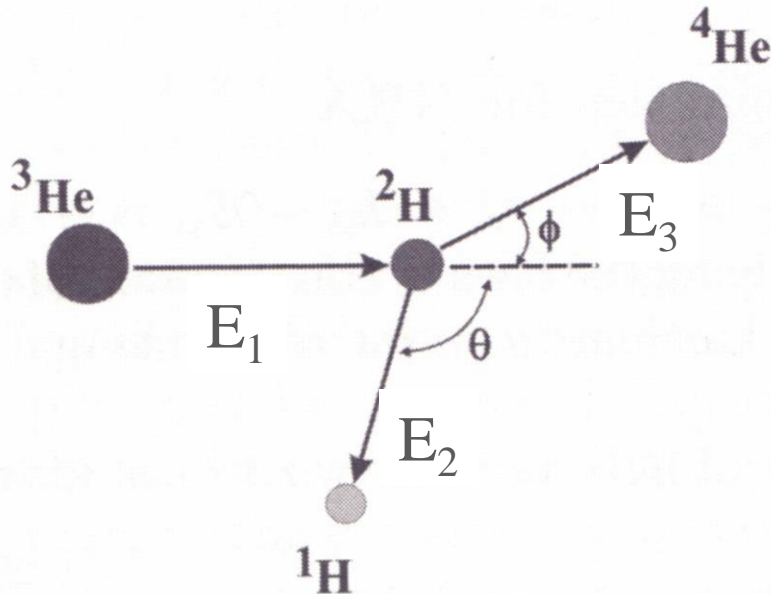
NRA is well suited for the detection of light elements since other methods fail in this field.

The products of the nuclear reaction are charged particles, gamma ray or both



The decay of the  $^{16}\text{O}$  which is in an excited state generates an alpha particle and an excited  $^{12}\text{C}$  isotope. The latter emits a gamma quantum with a well-defined energy of  $E_\gamma = 4.43 \text{ MeV}$  to reach its ground state which is the measured photon characteristic for this nuclear reaction.

For Deuterium detection (for practical reasons) instead of  $^{15}\text{N}$ , we are using  $^3\text{He}$ . The kinetic energy  $E_1$  of  $^3\text{He}$ , must be higher of the Coulomb barrier, then we have a nuclear reaction and production of a scattered p ( $^1\text{H}$ ) and recoiling  $^4\text{He}$ . The excess kinetic energy of the final product is the Q-value of the reaction.



$$Q = (\sum M_r - \sum M_p)c^2 = \sum T_p - \sum T_r$$

The nuclear reactions can be *endothermic* ( $Q < 0$ ) or *exothermic* ( $Q > 0$ ). If  $Q > 0$  any  $E_1$  generates a reaction, however, for the low values of  $E_1$  the cross section is low. For  $Q < 0$  there is a critical energy  $E_1$  above which the reaction becomes possible

*Note. For  $Q = 0$  we have Elastic scattering*

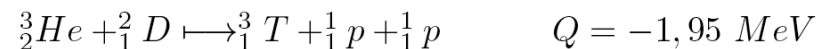
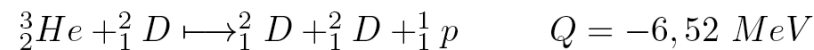
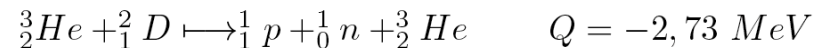
The energy of the fast proton detected depends on the depth of the deuterium atom in the sample



# NRA for Deuterium Measurement

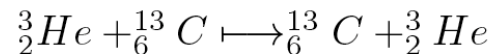
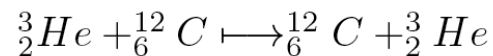
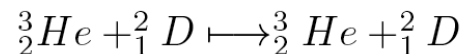


For the Deuterium depth profile measurements a  $^3\text{He}$  ion (+1 or +2) beam is used  
We distinguish 3 cases  $Q>0$ ,  $Q<0$  or  $Q=0$ . However....



Nuclear Reaction

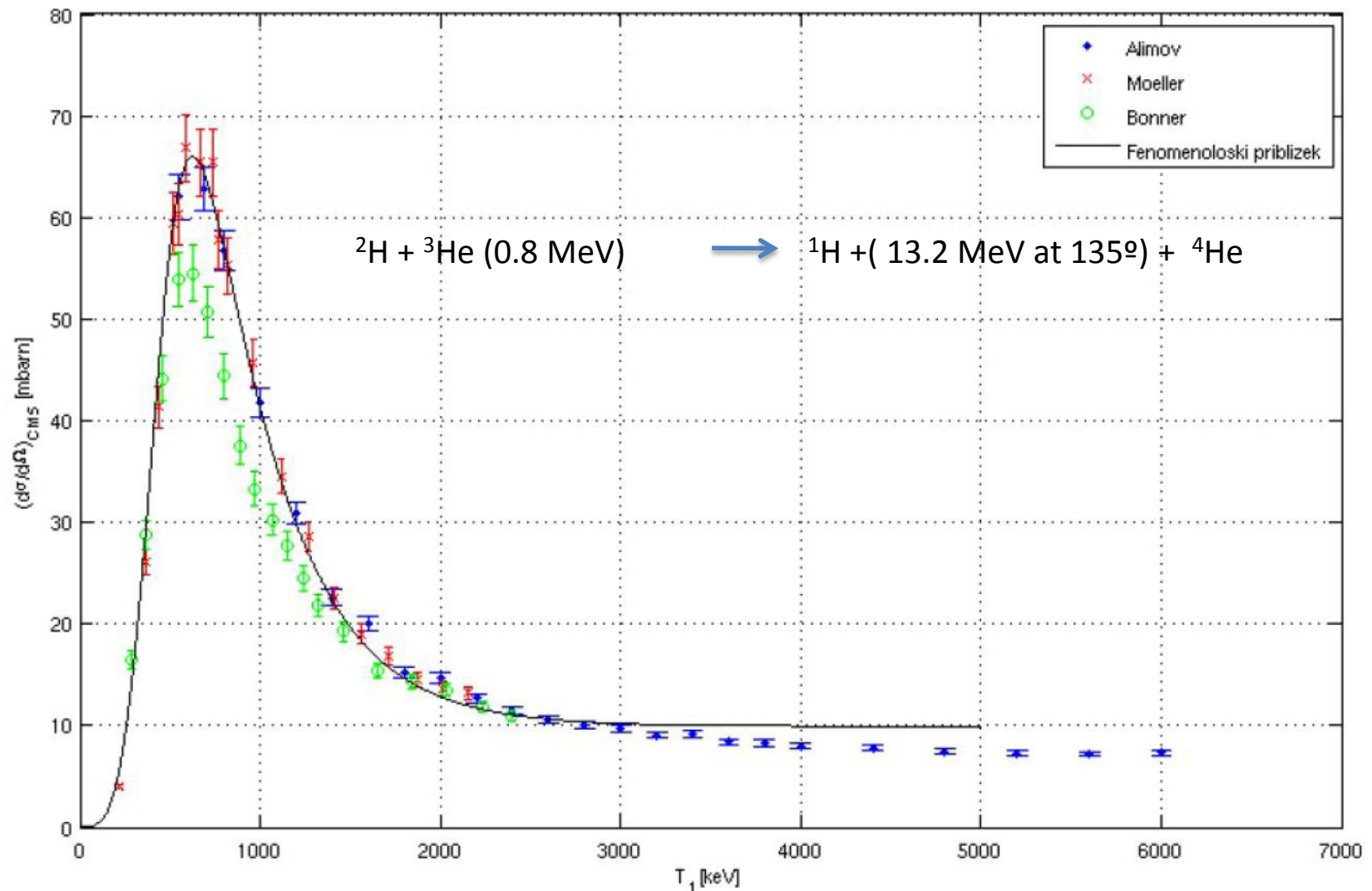
All these reactions produce various products, which can be detected in the detection set-up. However, as  $Q<0$  for most of them we can get rid of the parasitic reactions (2-4).



Elastic Scattering

The Elastic scattering reactions of  $^3\text{He}$  with D are forming a low-energy background. However, we resolve the problem by placing a protective mylar foil in front of the detector. Fast protons are passing through, while the foil stops the scattered helium ions.

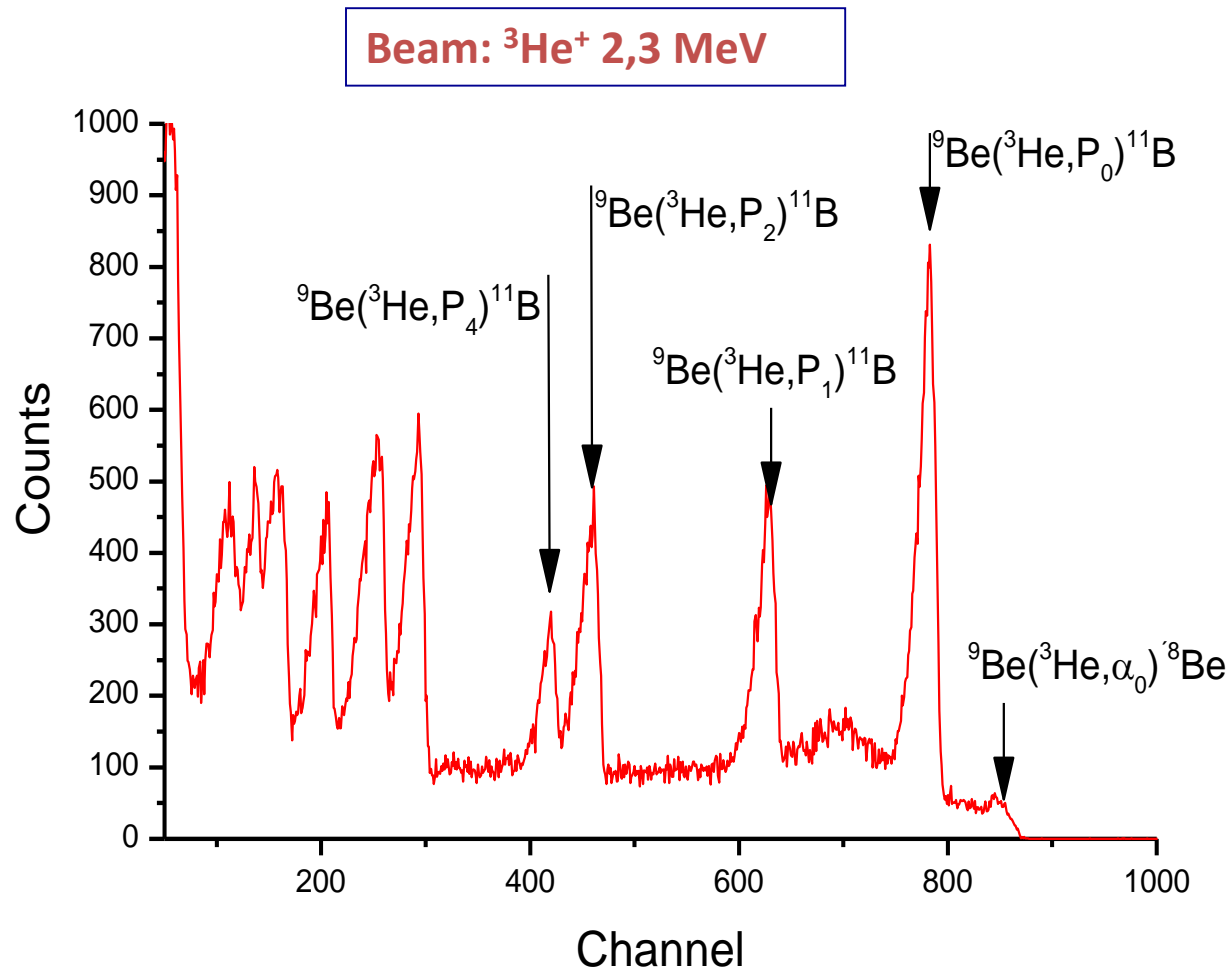
# Cross-Section $d(^3\text{He},p)^4\text{He}$



# NRA spectrum of a Be sample



Several reaction channels are excited with emission of  $^1\text{H}^+$  (identified as Po..P4). The peaks allow the profile of Be in any solid sample down to the ppm level.





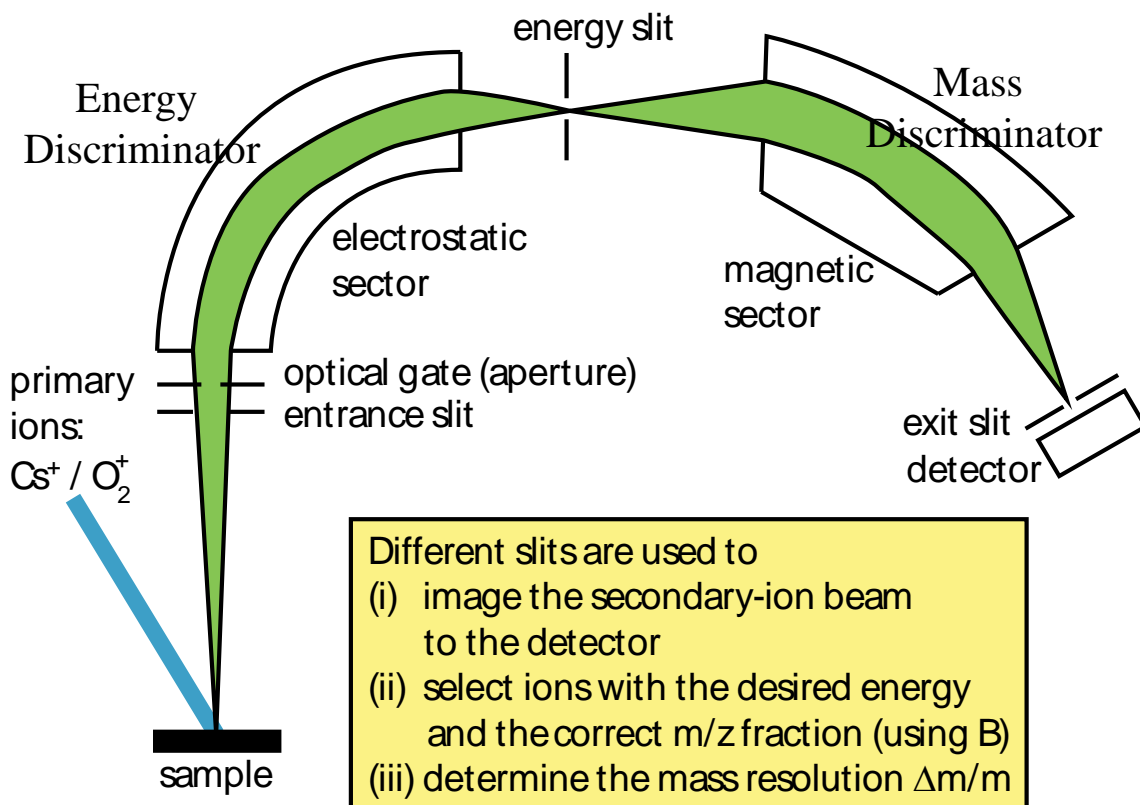
# II. Destructive Surface Analysis: SIMS



SIMS uses an ion beam for sputtering the surface to investigate while we are analysing the masses of the ions of the sputtered particles.

Ion beams are ( $\text{Cs}^+$ ,  $\text{O}_2^+$ ,  $\text{Ar}^+$ ,  $\text{Ga}^+$ ) of energy in the range 1 to 30keV (primary ions)

The sputtered ions (to analyse) are the secondary ions → Secondary Ion Mass Spectrometry (SIMS).



In this configuration (*forward geometry*), multiple ion beams can be measured simultaneously.

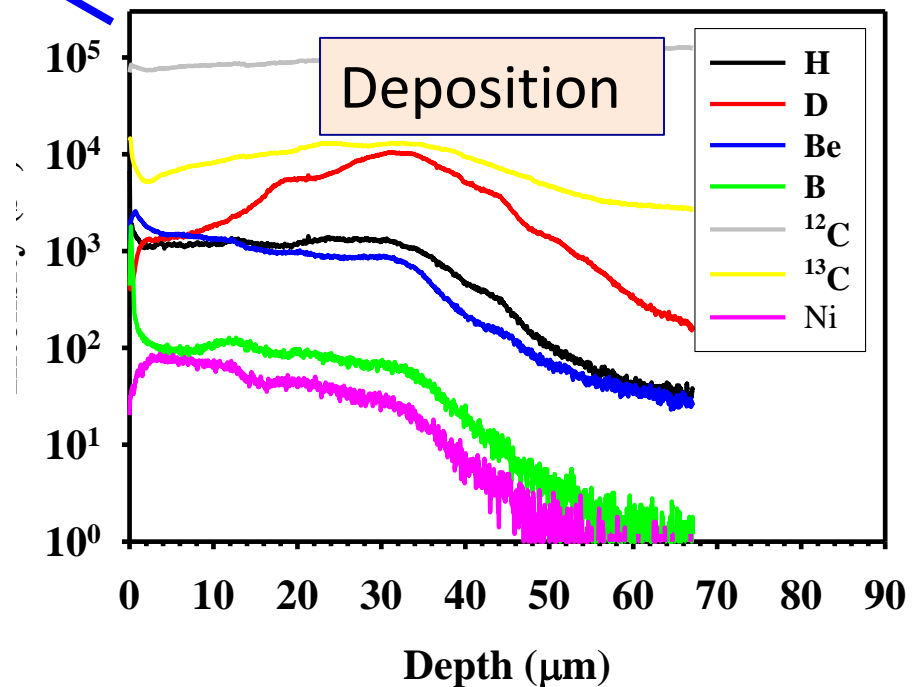
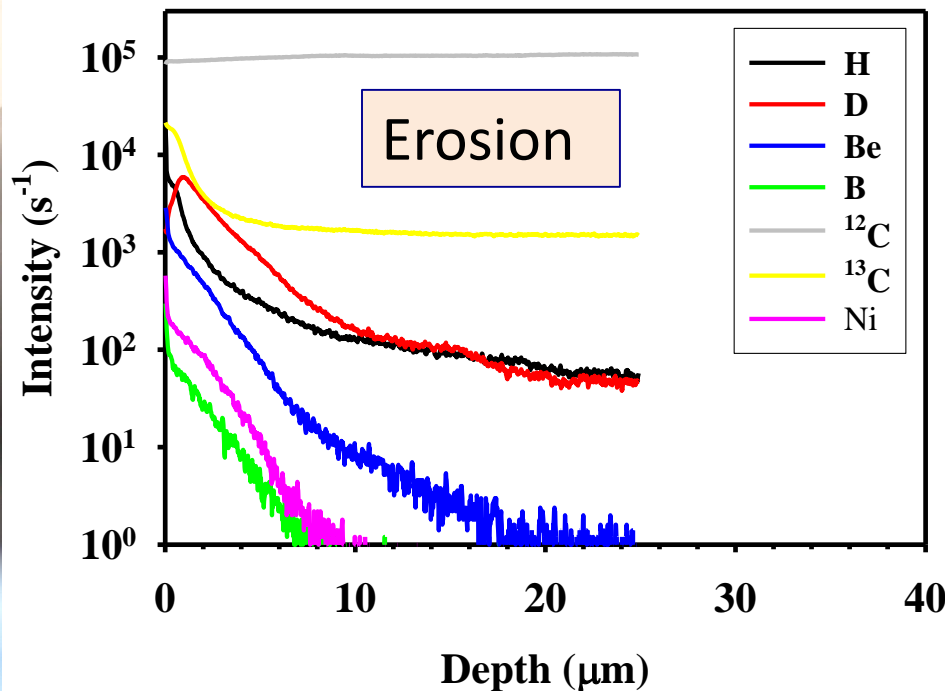
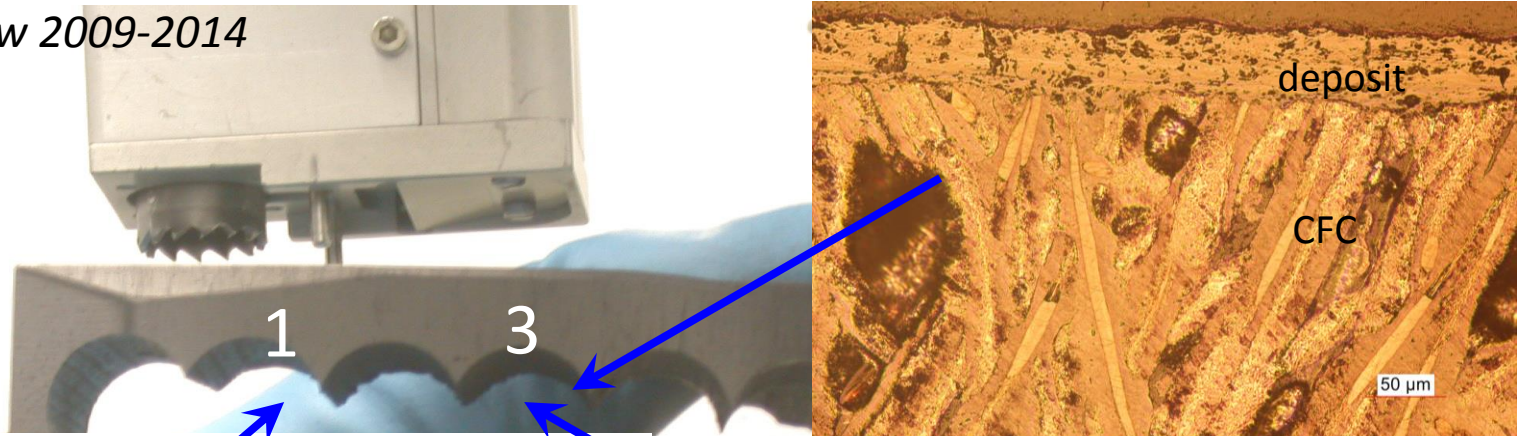
If the magnetic sector precedes the electrostatic sector (*reverse geometry*), then mass resolution is improved at the cost of losing the ability to measure multiple ion beams simultaneously.

Oxygen bombardment increases the yield of positive ions and caesium bombardment increases the yield of negative ions.

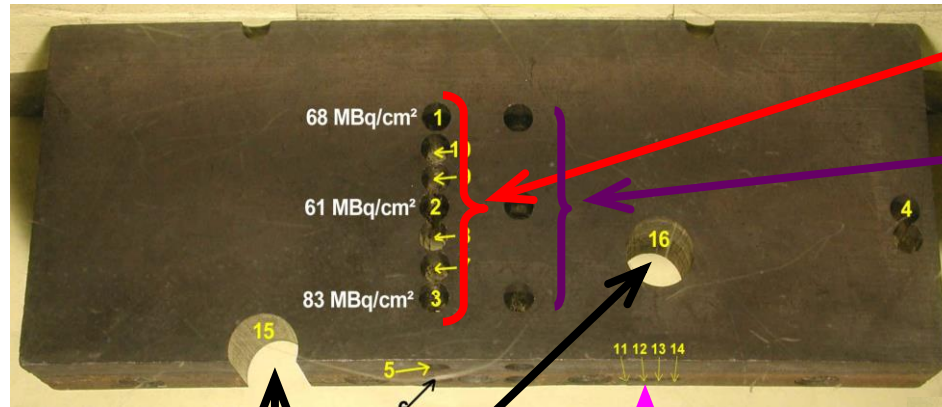
# SIMS Erosion Deposition Profiles



J. Likonen, Review 2009-2014



# Tritium Analysis by Full Combustion



**Combustion**  
(tritium concentration in mm range)

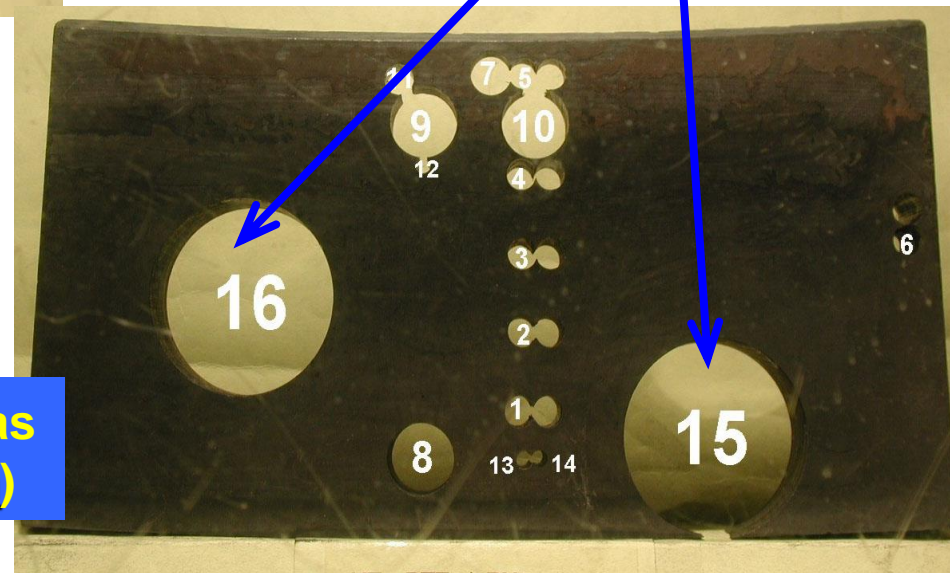
**Ion Beam Analysis (D analysis)**

**Calorimetry and  
BIXS**

**Laser  
(detritiation)**

**AMS**  
(tritium concentration  
in μm range)

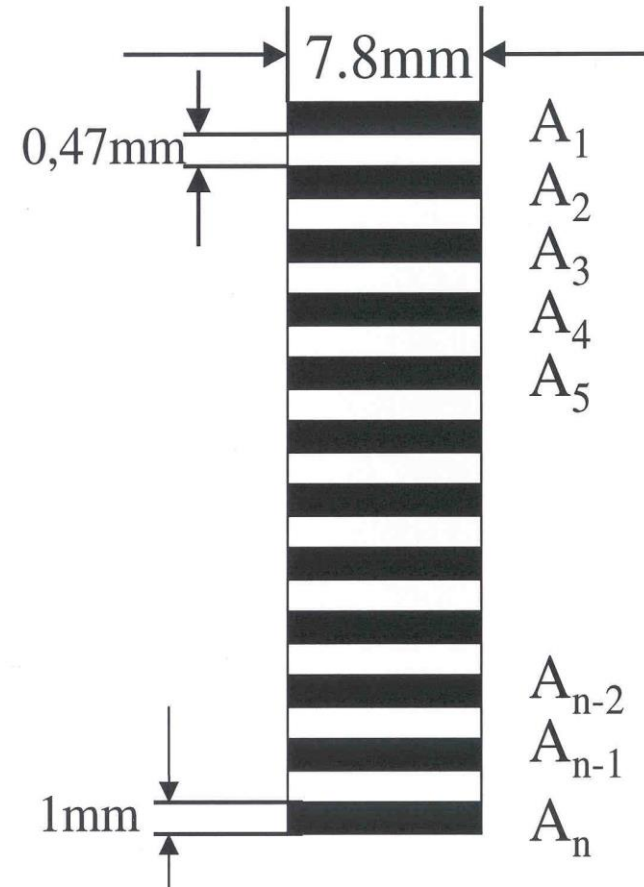
**For autoradiography the whole tile has  
been used (surface tritium distribution)**





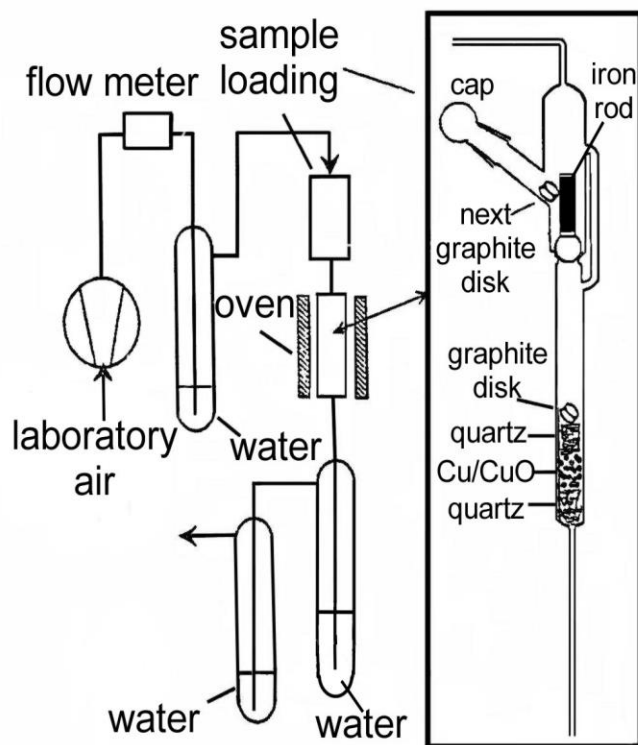


**Graphite (left) and CFC (right) cylinders as well as some typical disks**



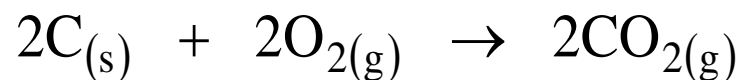
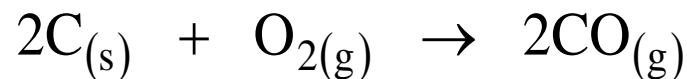
**Specimen designation A1 = Plasma exposed side**

# The Vance Apparatus

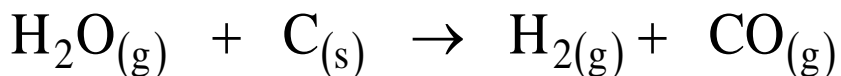
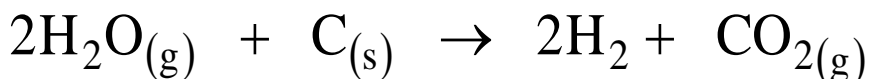


Tritiated Water is analysed by LSA

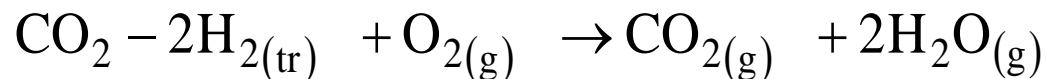
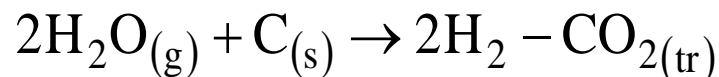
## The reactions



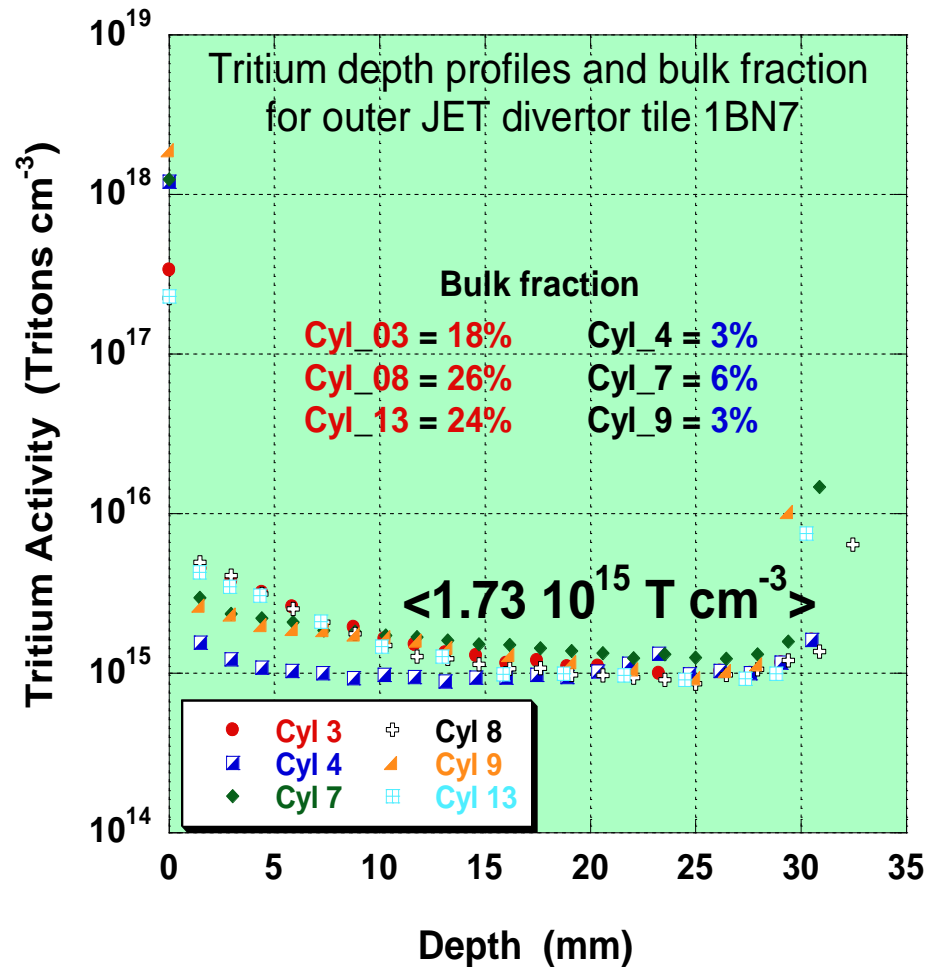
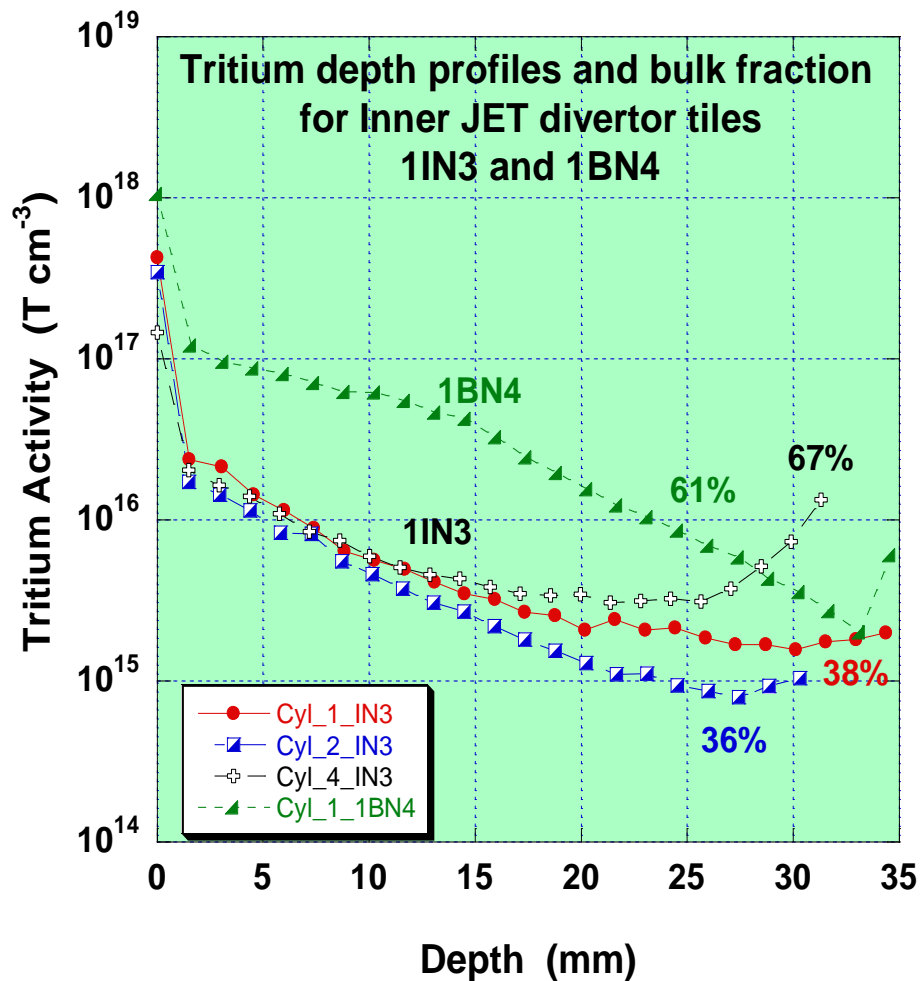
It was established that the presence of water increases significantly the combustion rate.



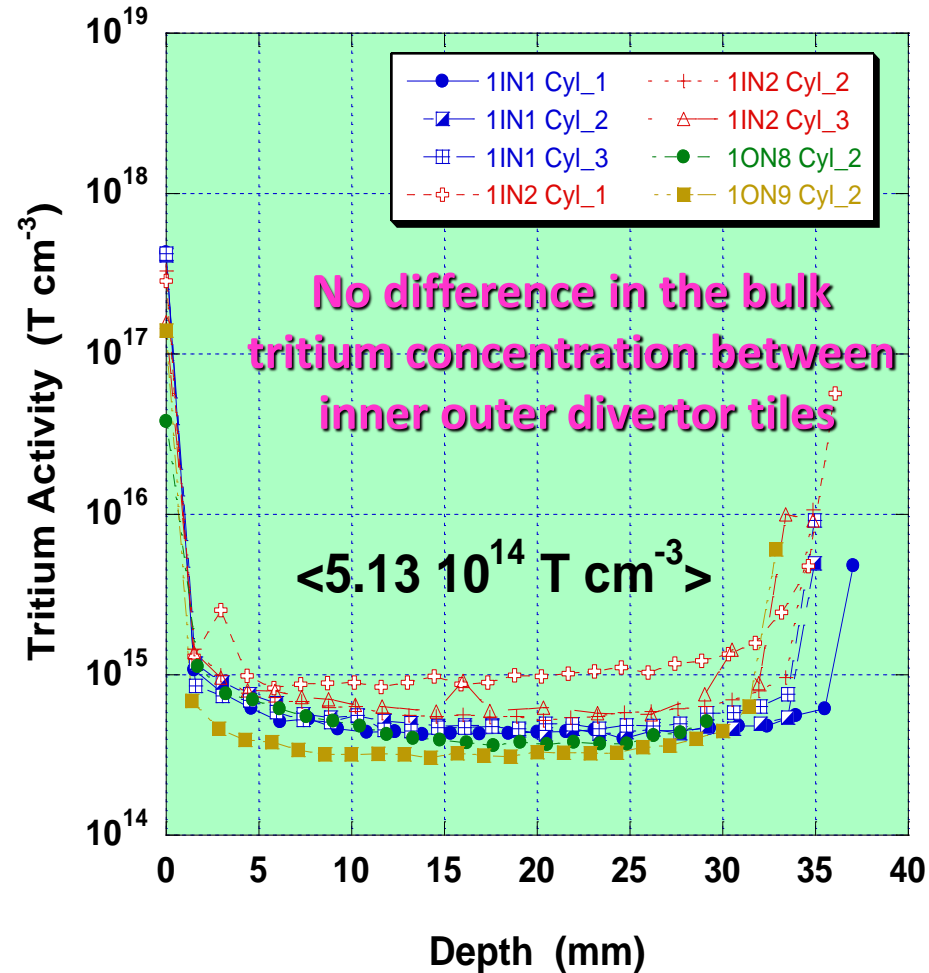
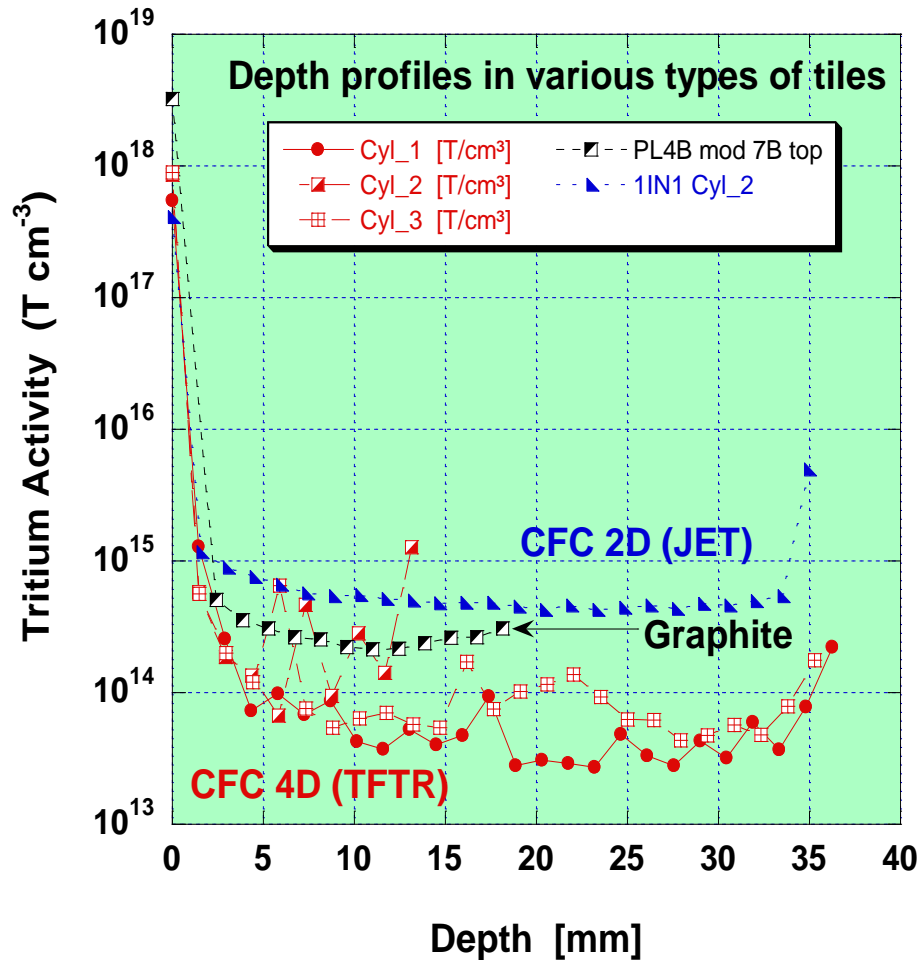
or more probably



# Typical Tritium Profiles for JET Tiles

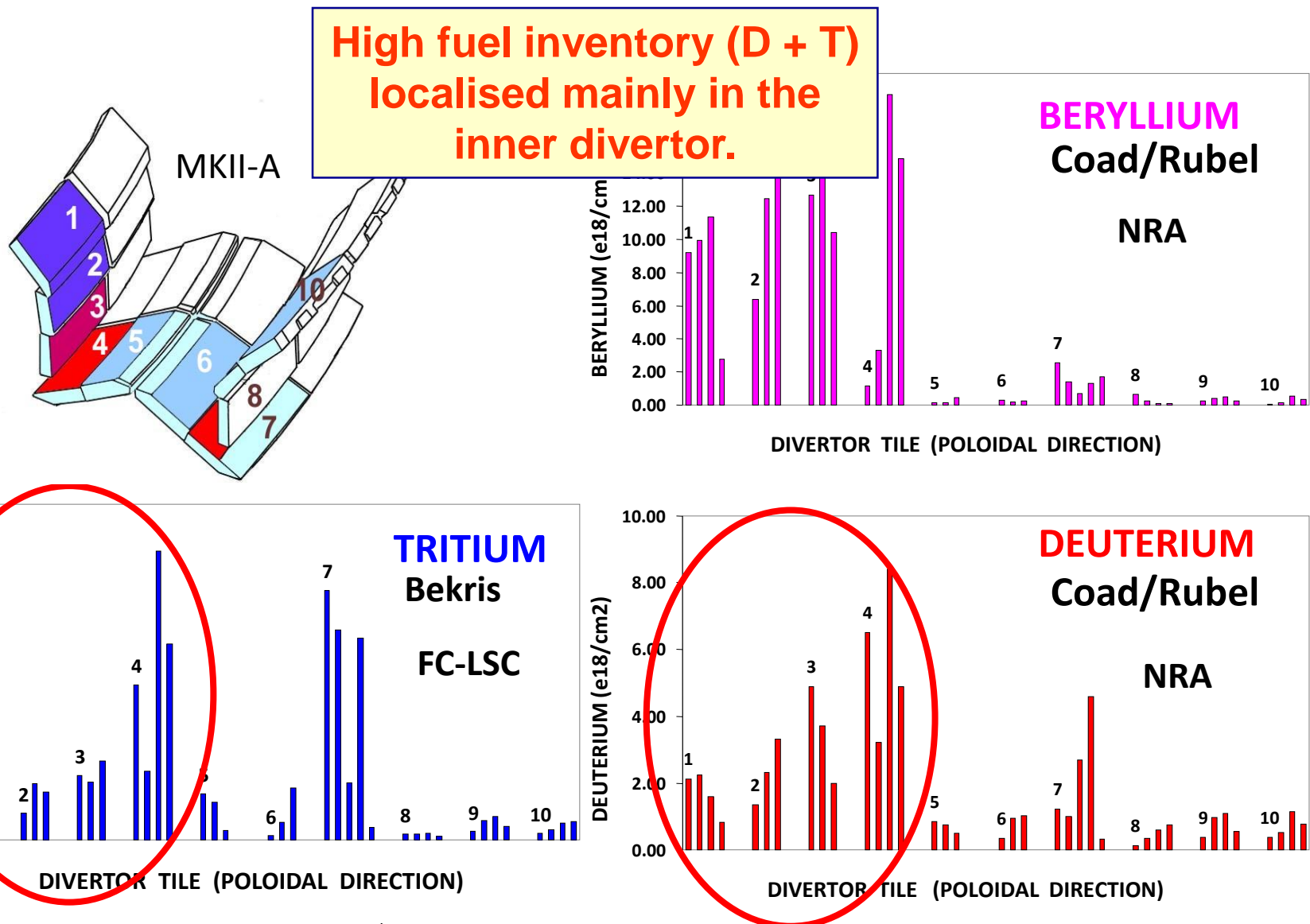


# Tritium profiles in 2D & 4D CFC tiles





# Comparison between the T, D and Be distributions



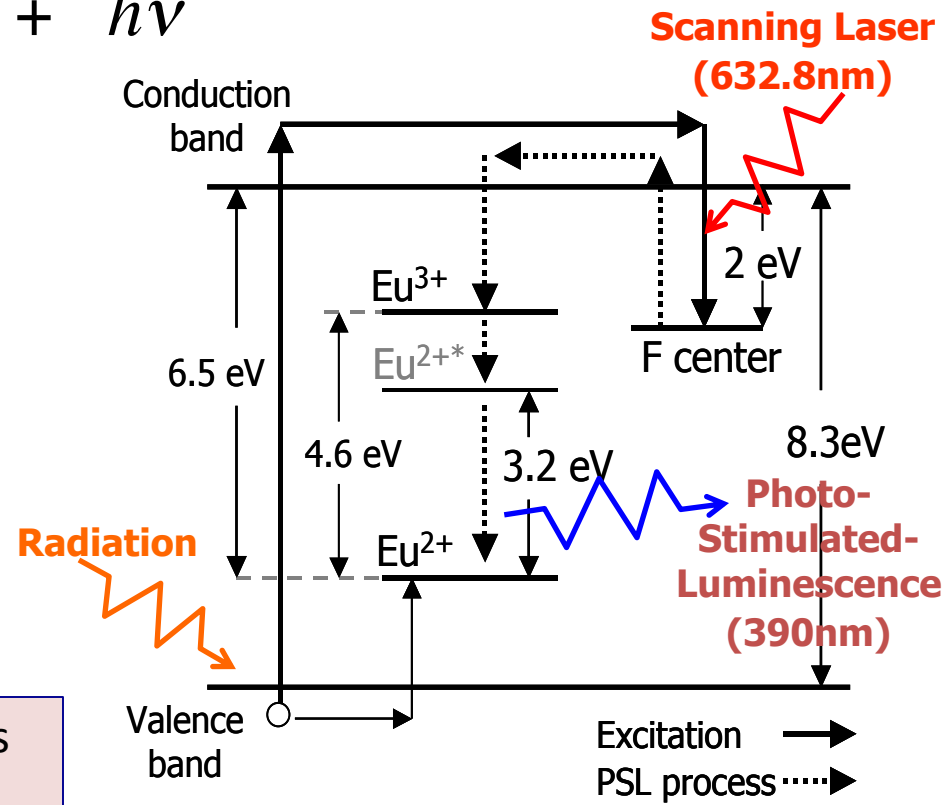
# Imaging Plate (Non-Destructive)



Image plates 2D cartography of [T]. It exploits the photo-stimulable properties of inorganic material. The Fuji image plates used here had BaFBr:Eu<sup>2+</sup> composition. Europium acting as a luminescence centre



Radiation photons or electrons are absorbed by phosphor  
Photoelectron ejected into the conduction band of the phosphor crystals and becomes trapped in a lattice defect created by the absence of halogen ions (F-center)  
These defects were intentionally introduced during the manufacturing process.

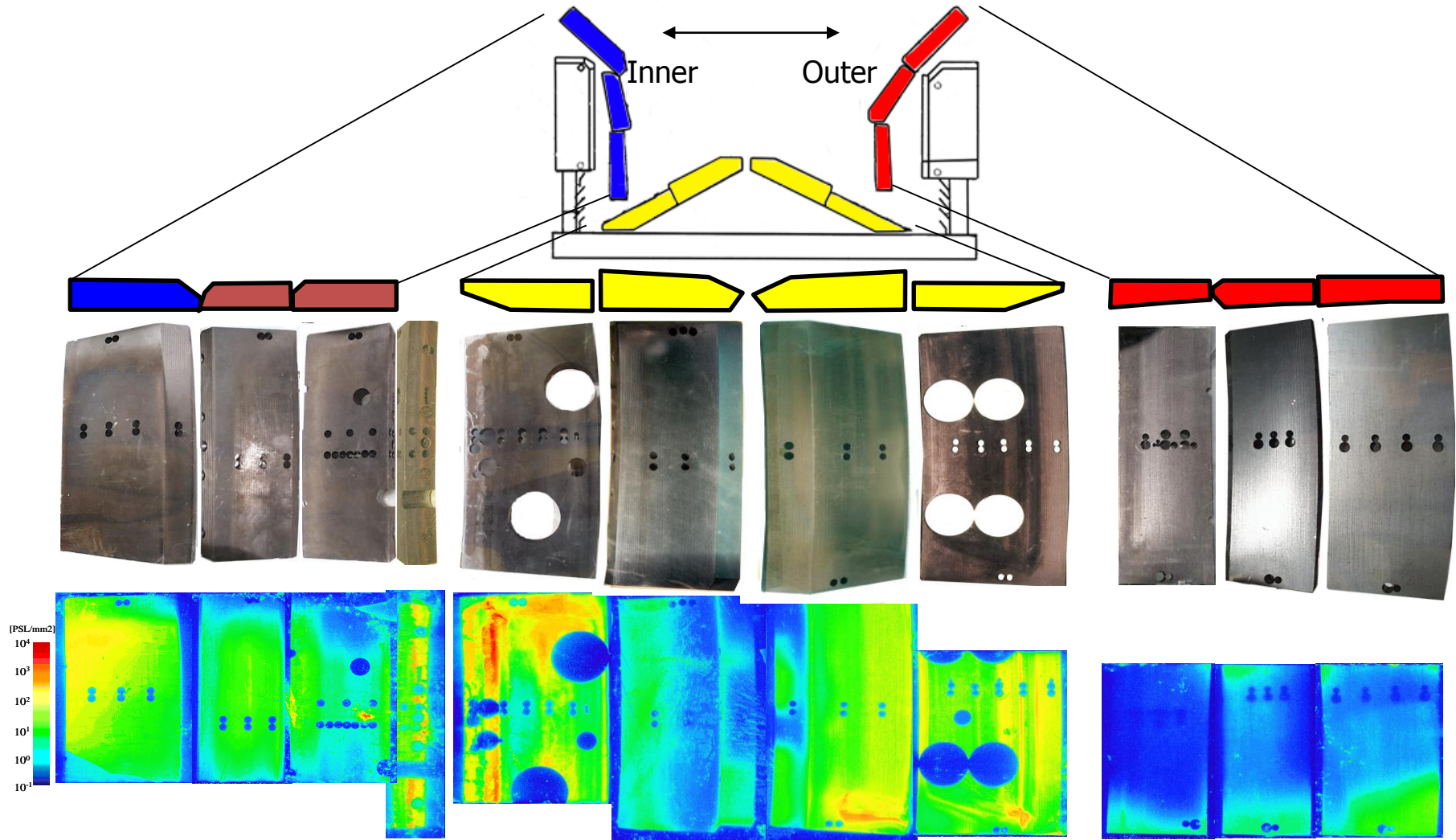


The number of pairs (electron-vacancies) is proportional to the absorbed energy.

# IP Analysis on MK-IIA divertor Tiles



*T. Tanabe et al., J. Nucl. Mater. 345 (2005) 89-95*

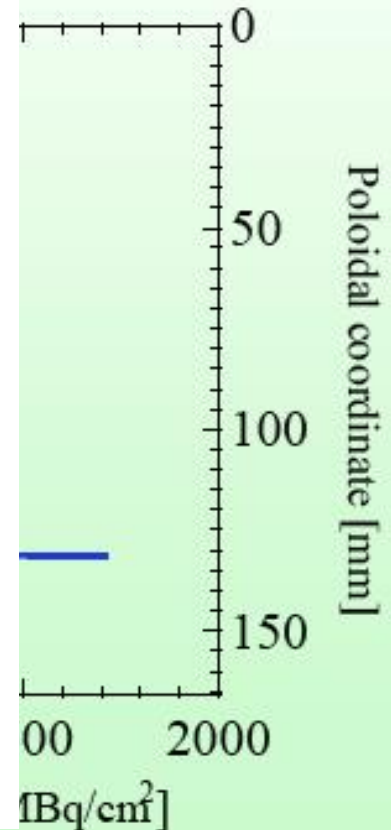
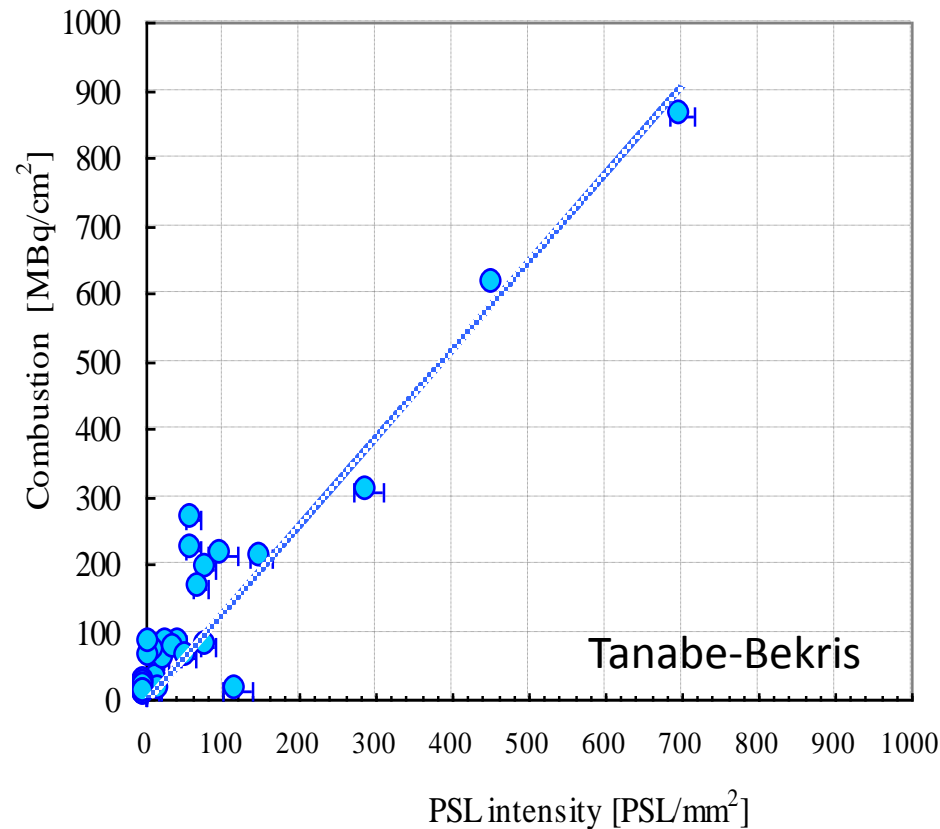
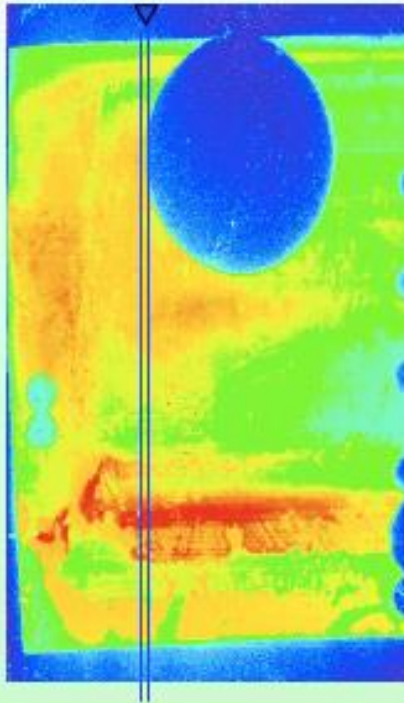


# Converting PSL intensity to T activity



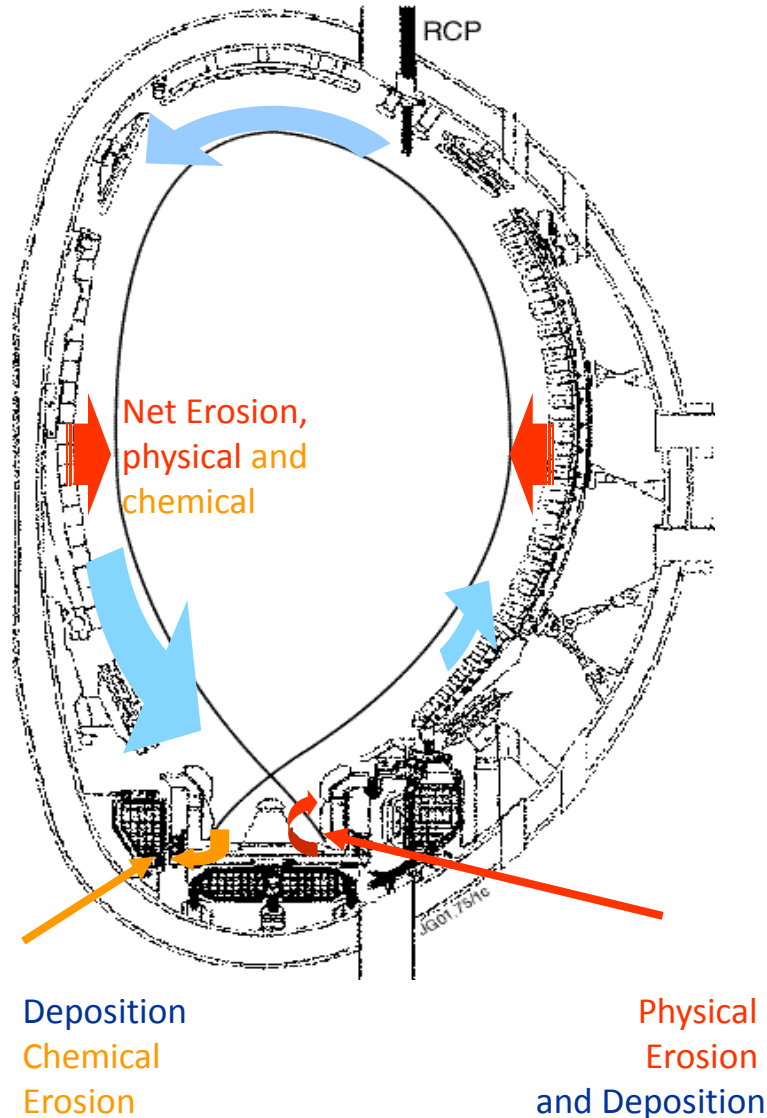
$$\text{Surface Activity [MBq/cm}^2\text{]} = 1.278 \pm 0.048 [\text{PSL/cm}^2\text{}]$$

\*Line profile: ~380 mm





# Conclusions



- Thanks to post-mortem analysis, we could establish that plasma fuel (tritium and deuterium) is co-deposited with the eroded carbon thus retaining fuel.
- The numerous surface analysis using a variety of destructive and non-destructive techniques helped us to establish a mechanism describing the material migration and fuel retention taking place inside the vacuum vessel
- Retention of fuel has safety and operational implications and need to be tackled as

Deposits can lead to:  
Formation of dust and flakes  
Degradation of diagnostics

# Acknowledgments



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*\* See the Appendix of F. Romanelli et al., Proceedings of the 25th IAEA Fusion Energy Conference 2014, Saint Petersburg, Russia*



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# Fusion Technology Contributors



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