

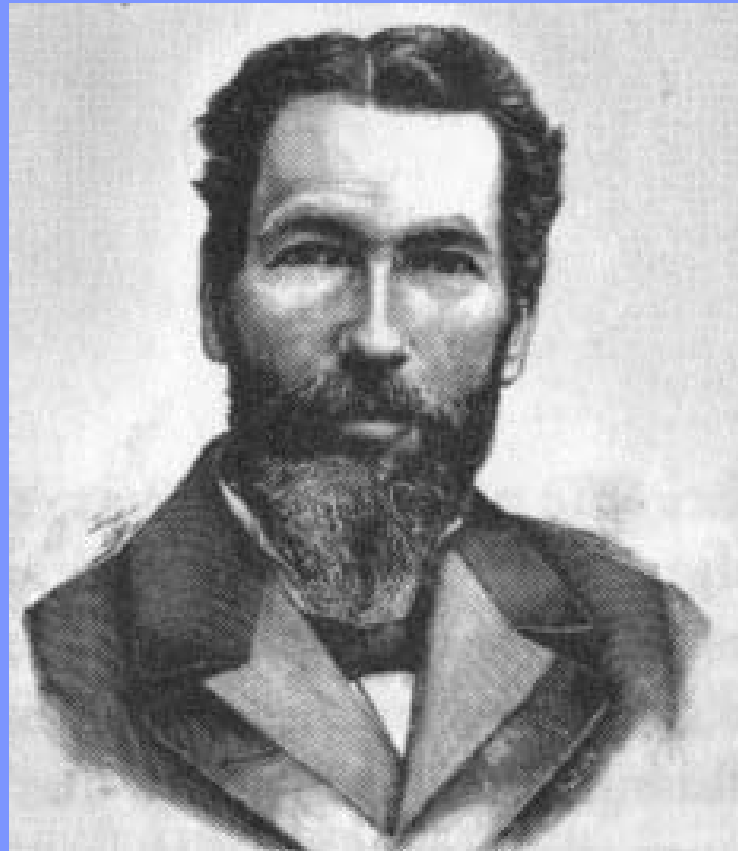


Geneva, December 1877

When *Cryogenic Engineering* was
born



Raoul Pictet



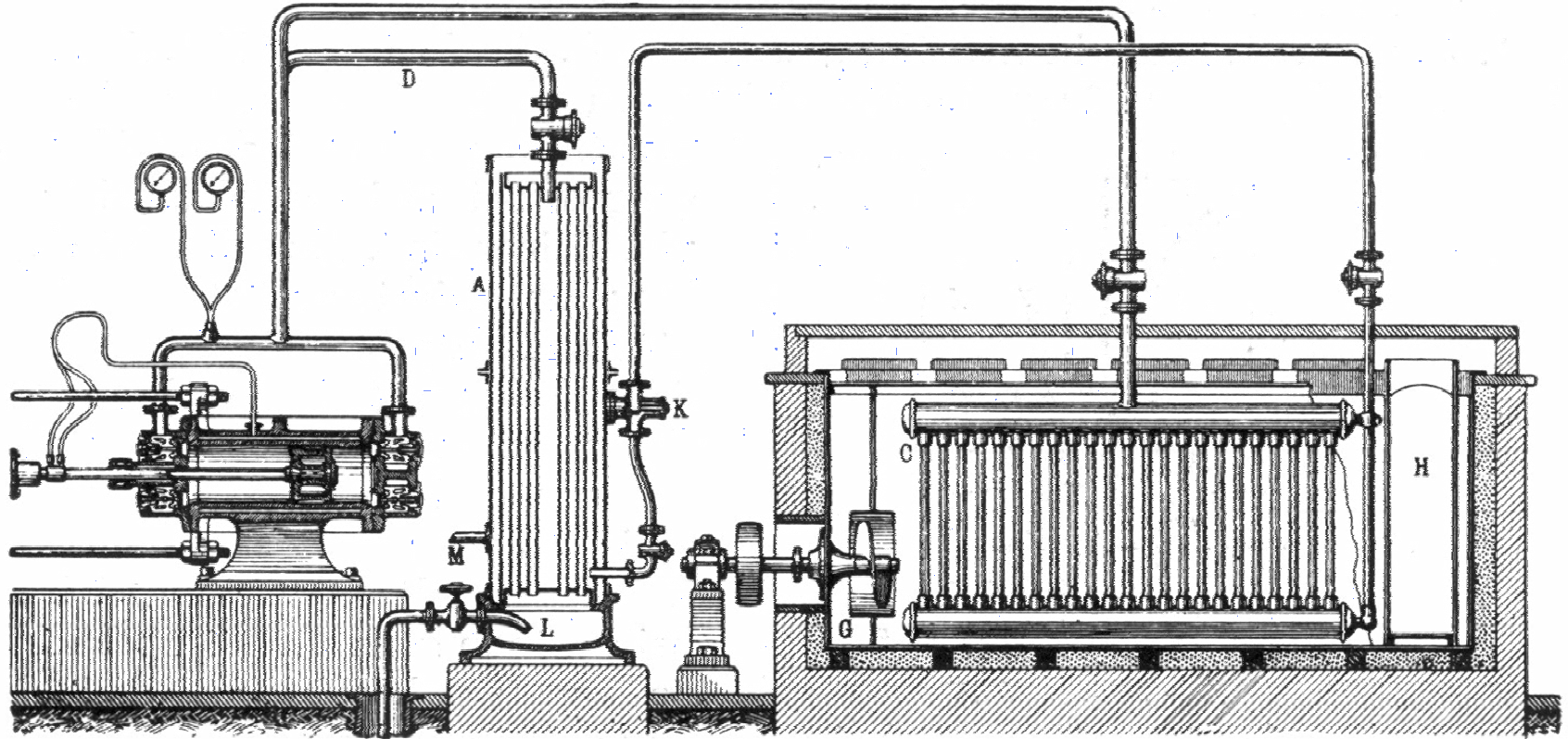
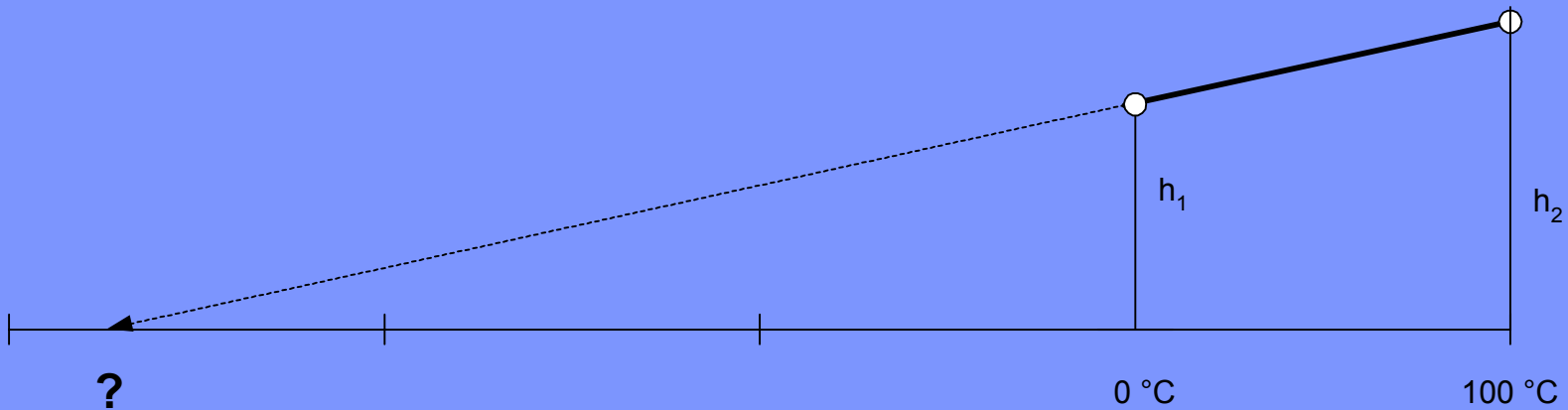
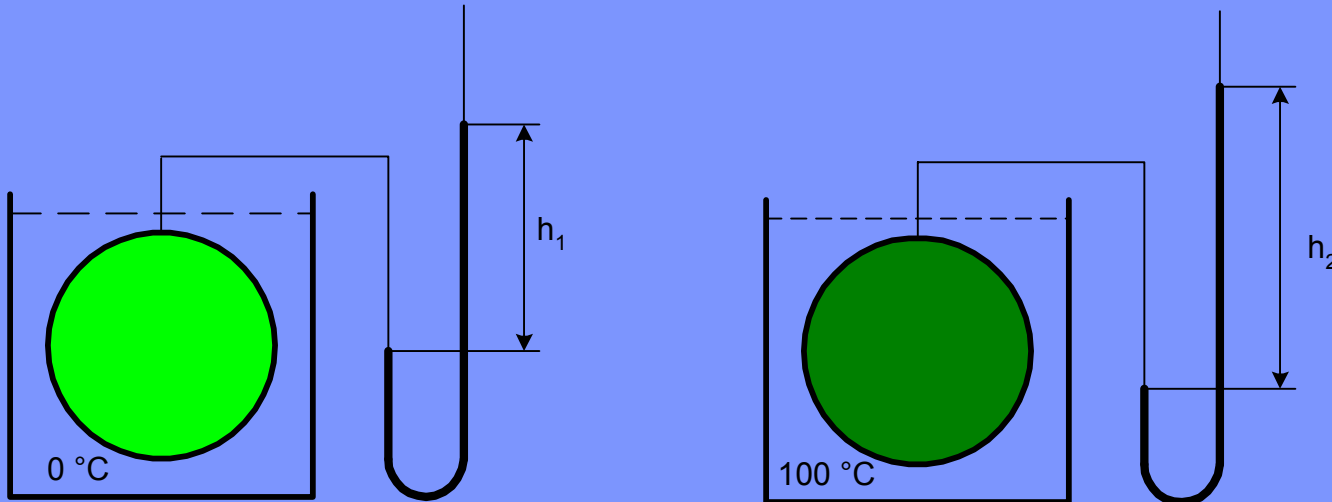


Abb. 44. SO_2 -Eismaschine von RAOUL PICTET.

A Kondensator, *C* Verdampfer, *D* Druckrohr, *G* Rührwerk, *H* Eiszelle, *K* Drosselventil, *L* Wassereintritt in den Kondensator, *M* Wasseraustritt.



Vorhersage des absoluten Nullpunkts
Amontons (17. Jahrh.)



- Fließbild von Pictets Anlage

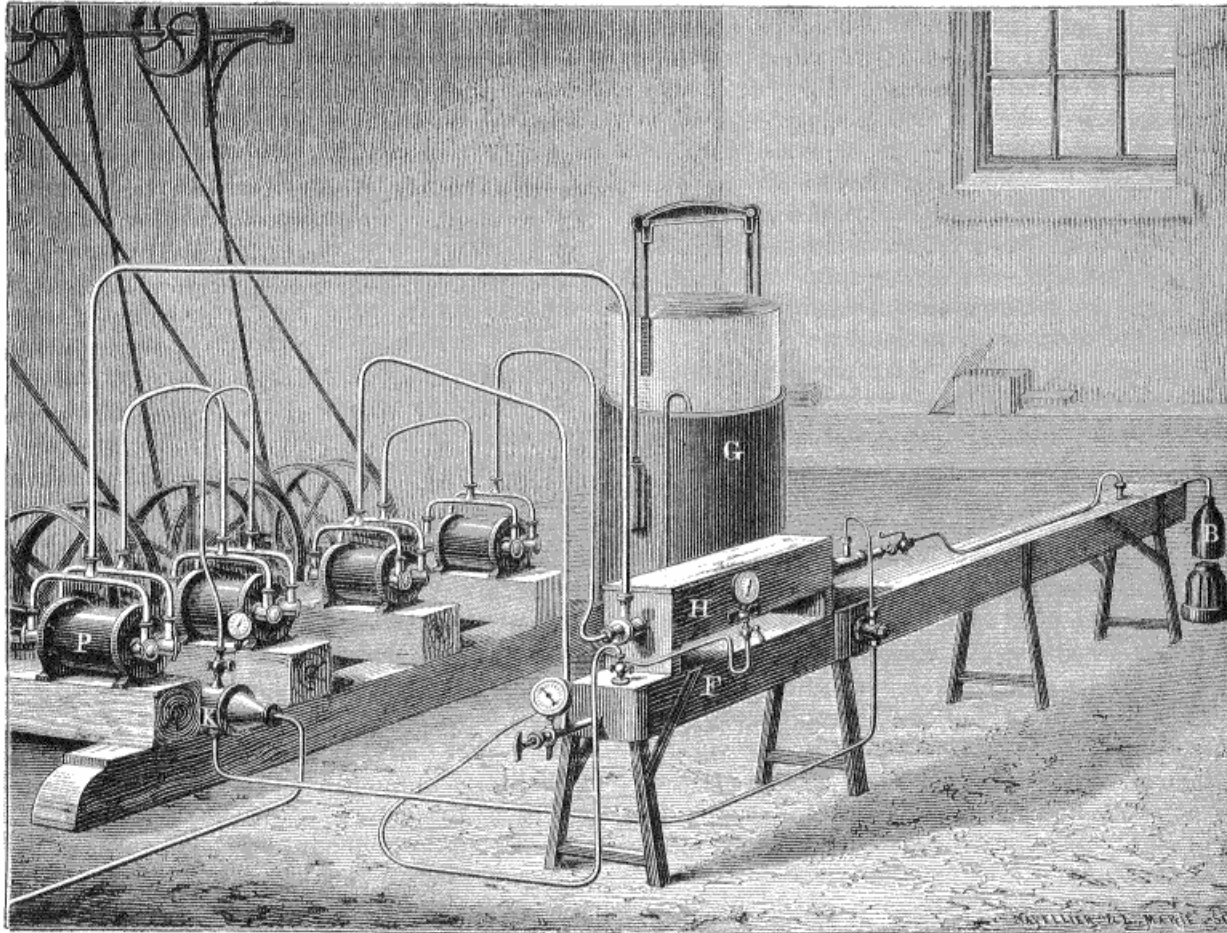
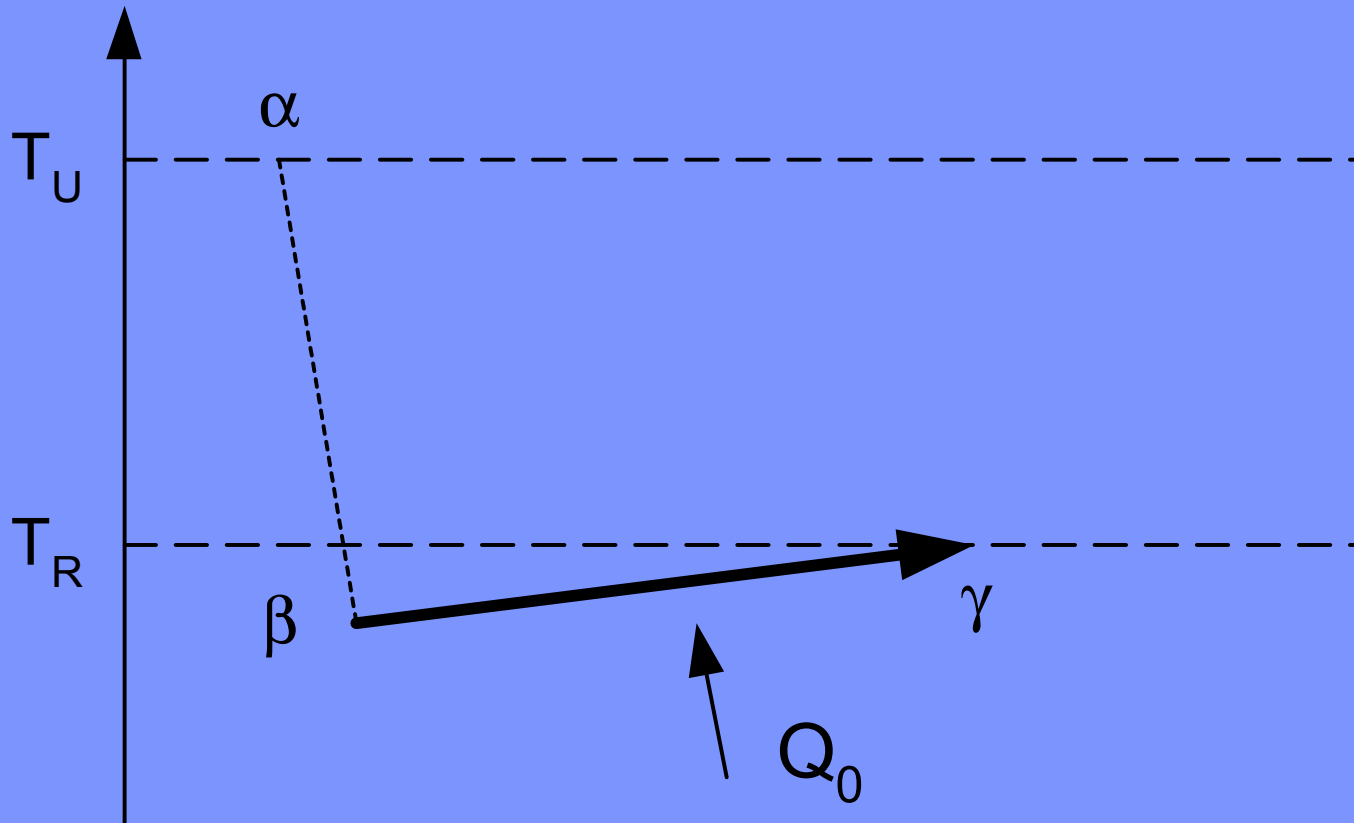


Fig. 1. — Grand appareil de M. Raoul Pictet pour la liquéfaction des gaz. (D'après une photographie.)



Lowering of Temperature and Production of Refrigeration





How can one reduce the temperature of a refrigerant below ambient temperature?

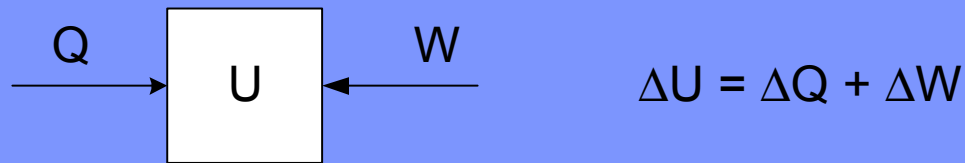
Our forefathers found three methods:

- Heat rejection to an even colder system
- **Throttling (the method Pictet used)**
- Performance of work by the refrigerant

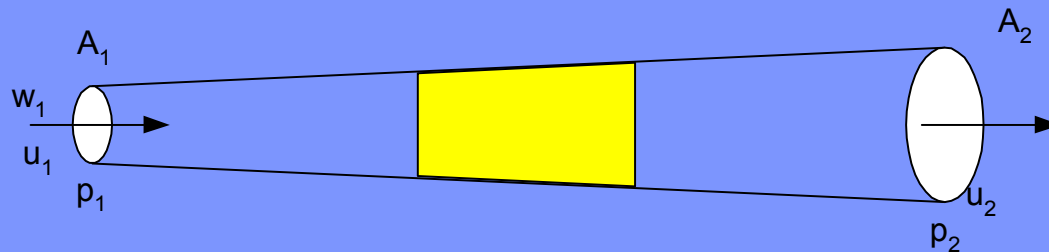


Energy Conservation

Energy conservation for a closed system



Energy conservation for an open system



$$m \cdot u_1 + p_1 \cdot A_1 \cdot w_1 = m (u_1 + p_1 v_1) = m \cdot h_1$$

$$m \cdot u_2 + p_2 \cdot A_2 \cdot w_2 = m (u_2 + p_2 v_2) = m \cdot h_2$$

$$u_2 - u_1 = p_1 v_1 - p_2 v_2$$

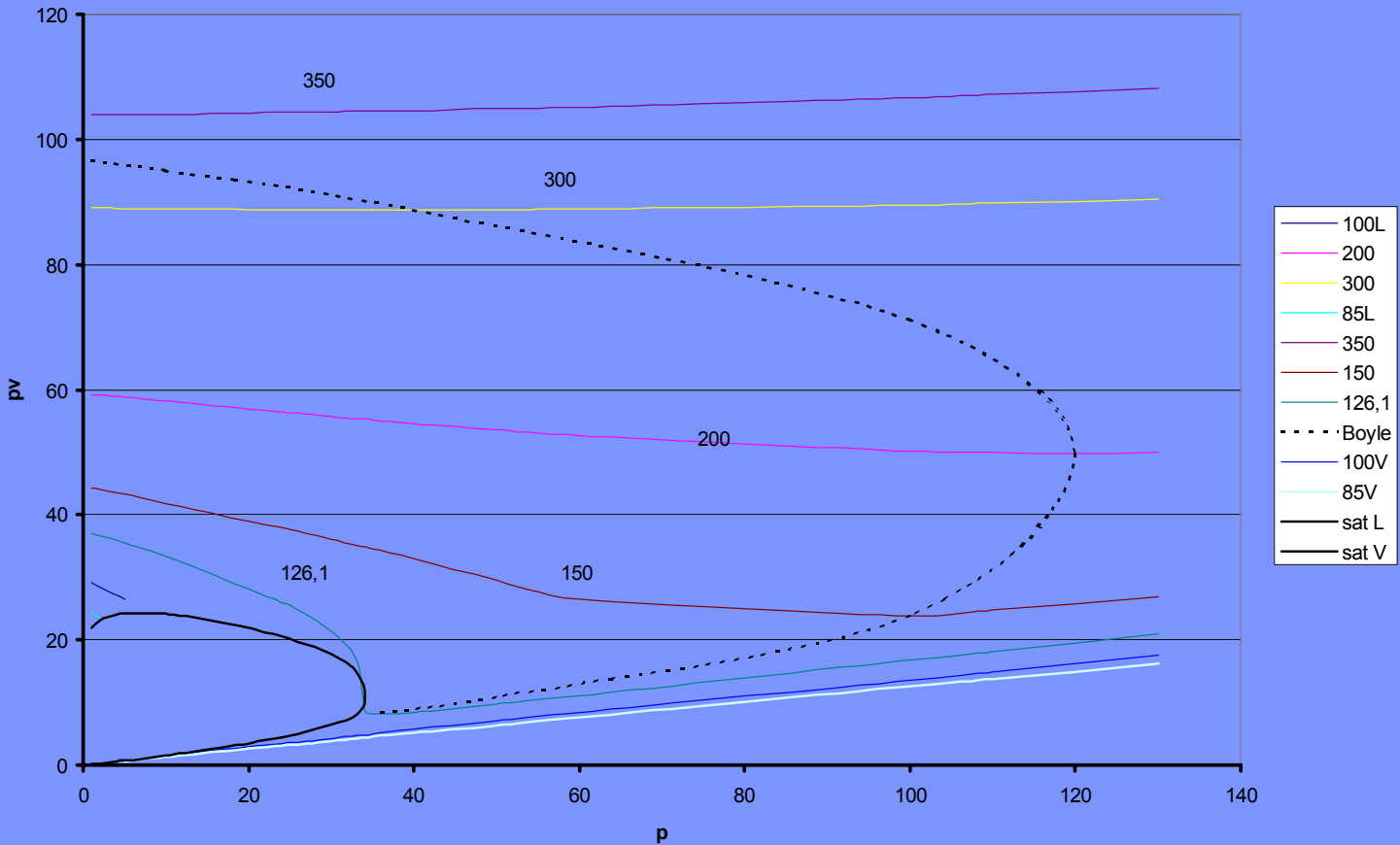


Temperature change of the fluid through throttling depends on the properties

if v increases slower than p decreasing	internal energy is increased
if $p \cdot v = \text{constant}$	internal energy stays constant
if v increases faster than p decreases	internal energy is reduced



$pV - p$ Diagram of Nitrogen



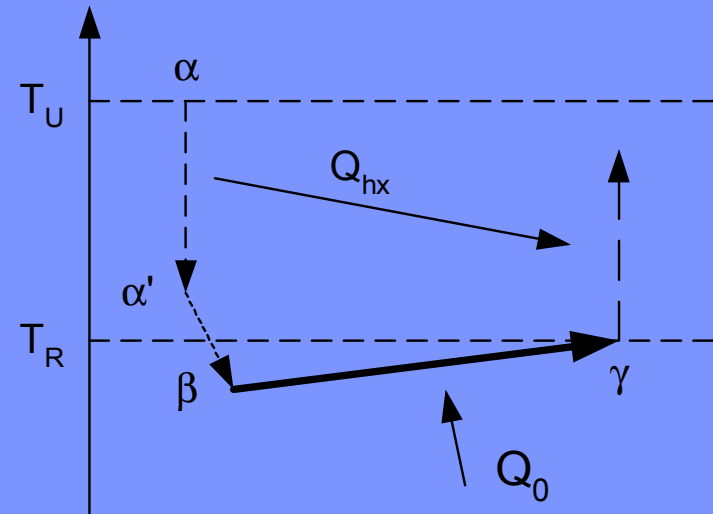
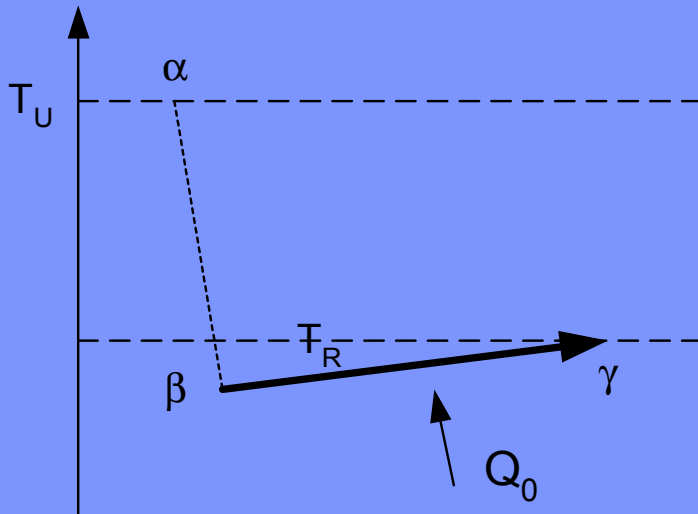


Temperature change due to throttling

- Temperature reduction due to throttling of a saturated liquid is a large effect
- Temperature change of a gas can be positive or negative, depending on the region in the phase diagram
- In any case it is a small effect, initially considered not to be technically usable.
- **But: Remember our assistant!**



Internal heat exchange makes the production of refrigeration much easier



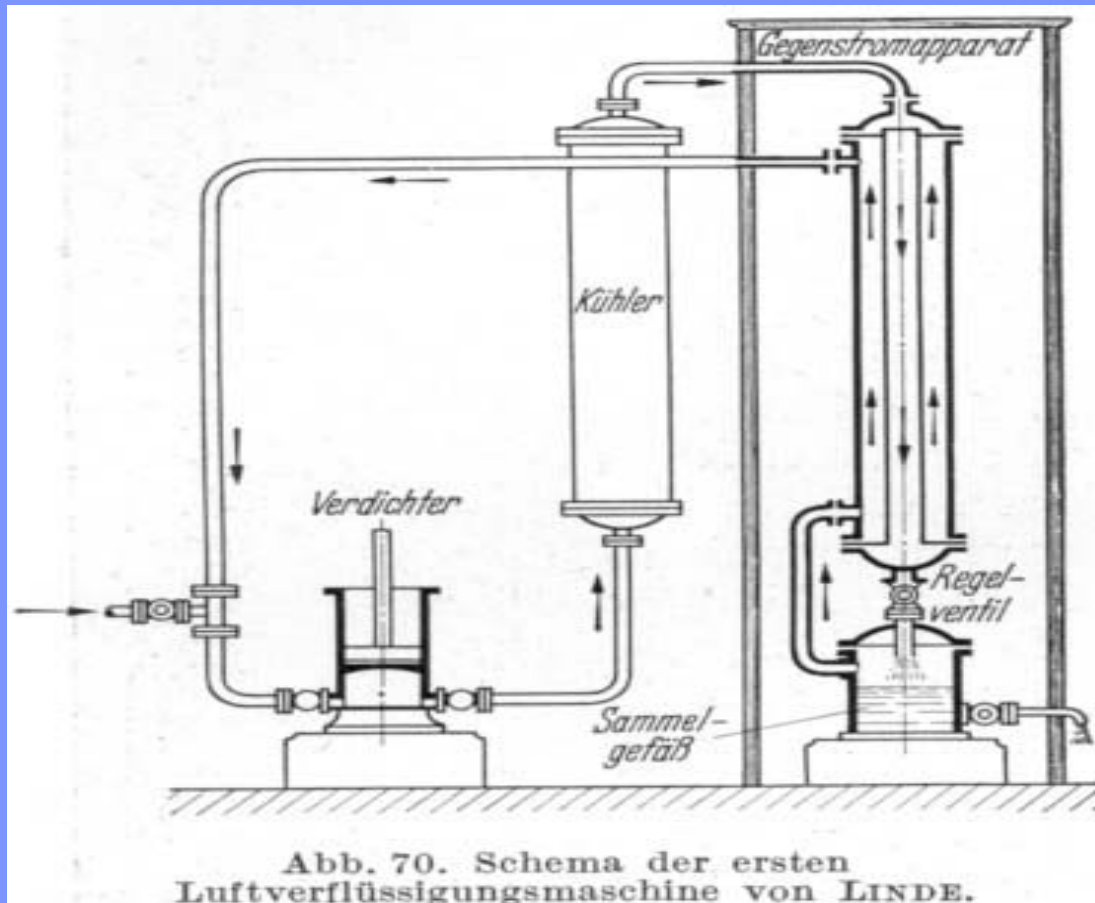


Recover unused Refrigeration by counter-current heat exchange

- Often not all produced refrigeration is used
- Idea: Recover the not-used refrigeration and use it for precooling of the inlet stream before the throttling
- Wilhelm Siemens (later Sir William Siemens) 1858 (long before Pictet!)



Combination of throttle refrigeration with counter-current heat exchange (Linde 1895)





The death and resurrection of throttle refrigeration in cryogenics

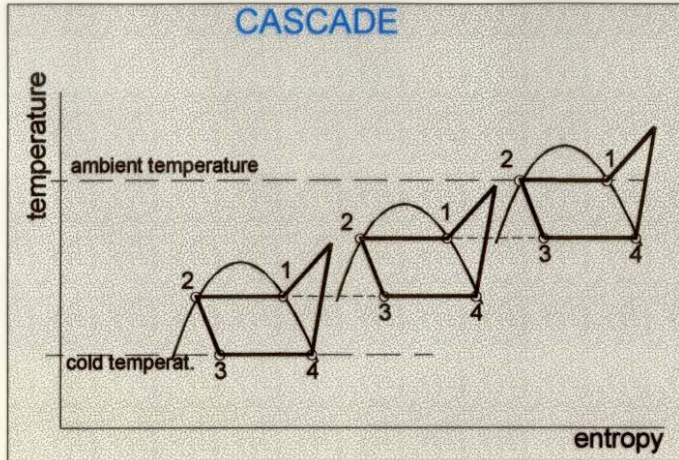
- Linde knew that throttling was simple, but not effective
- Eventually work extracting expansion was to replace the throttling
- But recently throttle refrigeration has made a surprise comeback:
- Mixed Refrigerant JT-systems



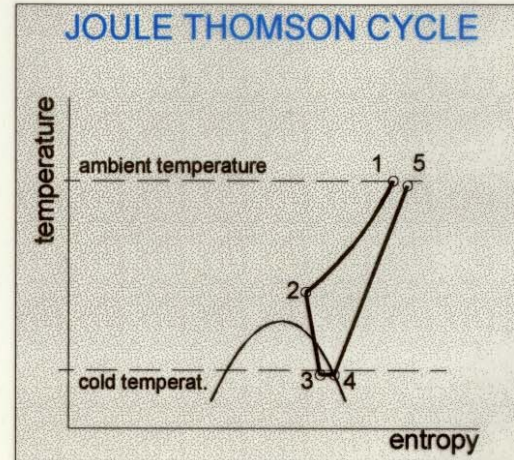
**LOW PRESSURE RATIO
SEVERAL COMPRESSORS**

**HIGH PRESSURE RATIO
ONE COMPRESSOR**

CASCADE

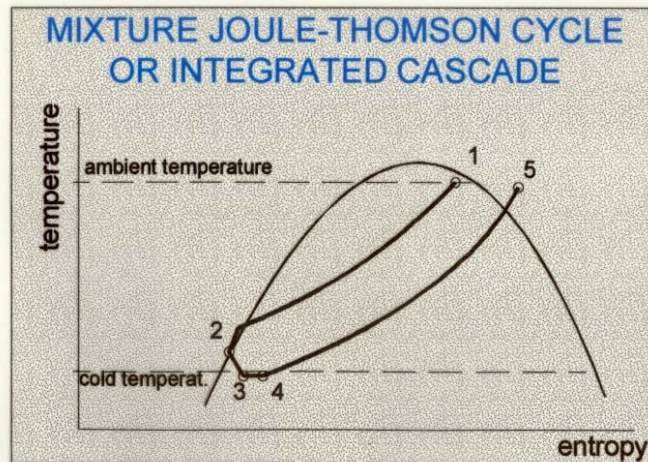


JOULE THOMSON CYCLE

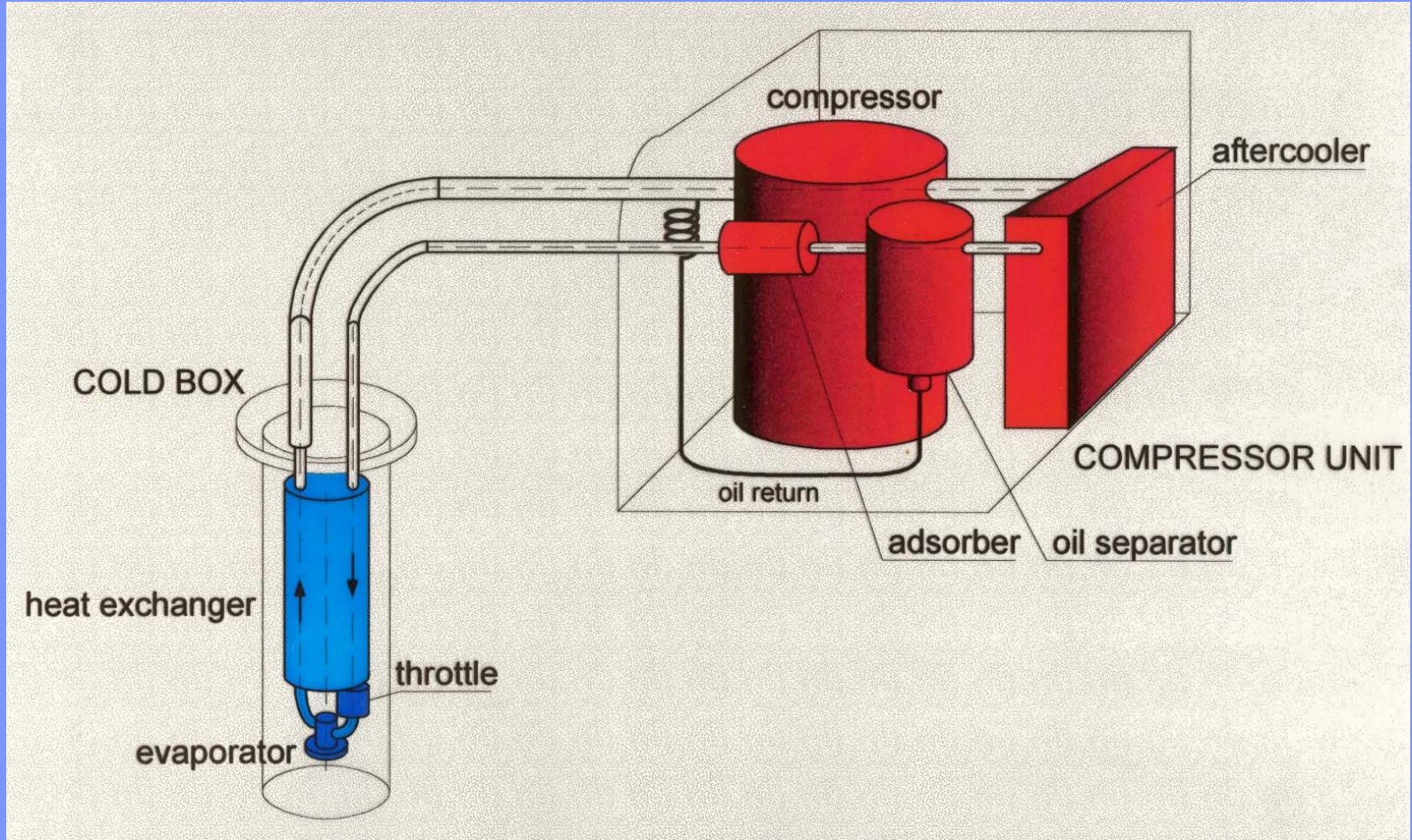


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**MIXTURE JOULE-THOMSON CYCLE
OR INTEGRATED CASCADE**



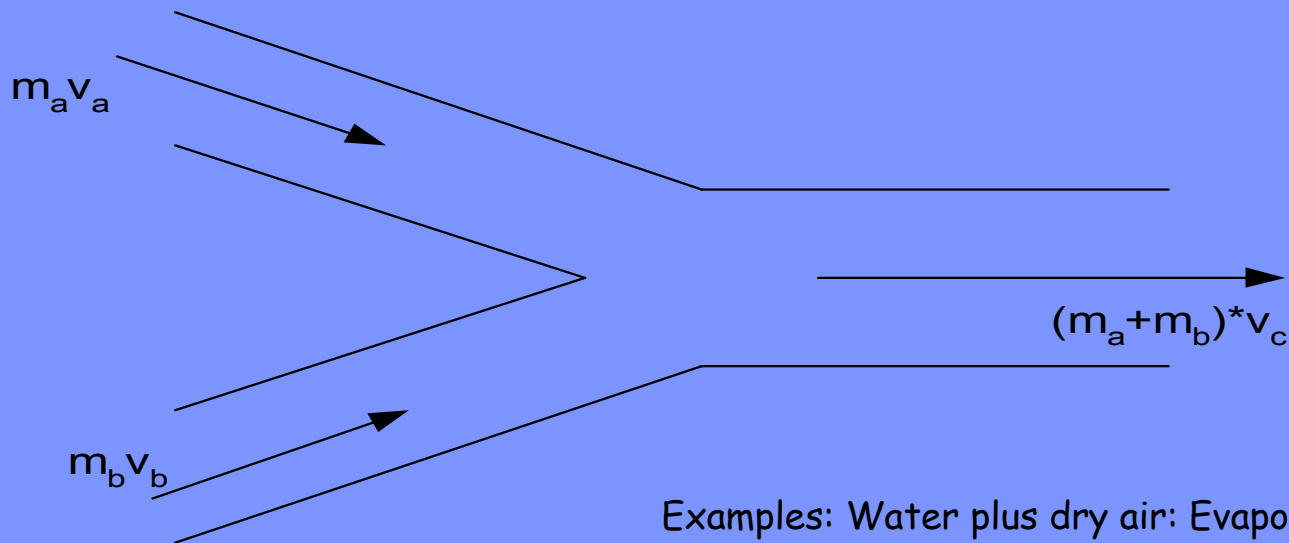
**LOW PRESSURE RATIO
ONE COMPRESSOR**





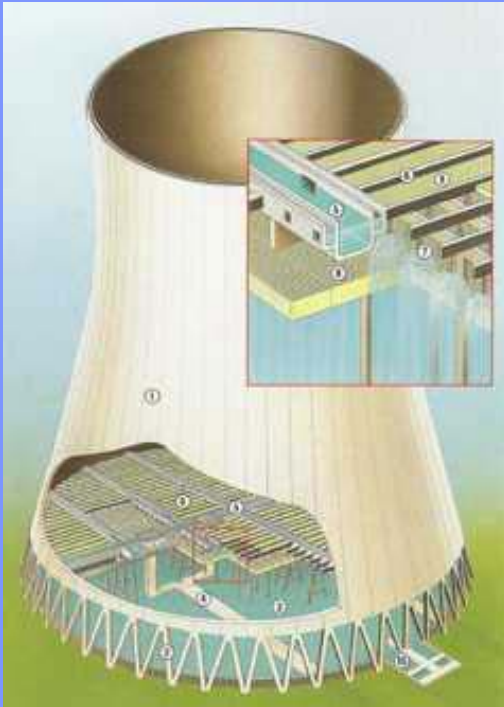
„Mixing“ is Throttling at constant pressure:
If the mixing of two fluids leads to an increase in specific volume, a cooling effect will be observed

Energy conservation during mixing at constant pressure



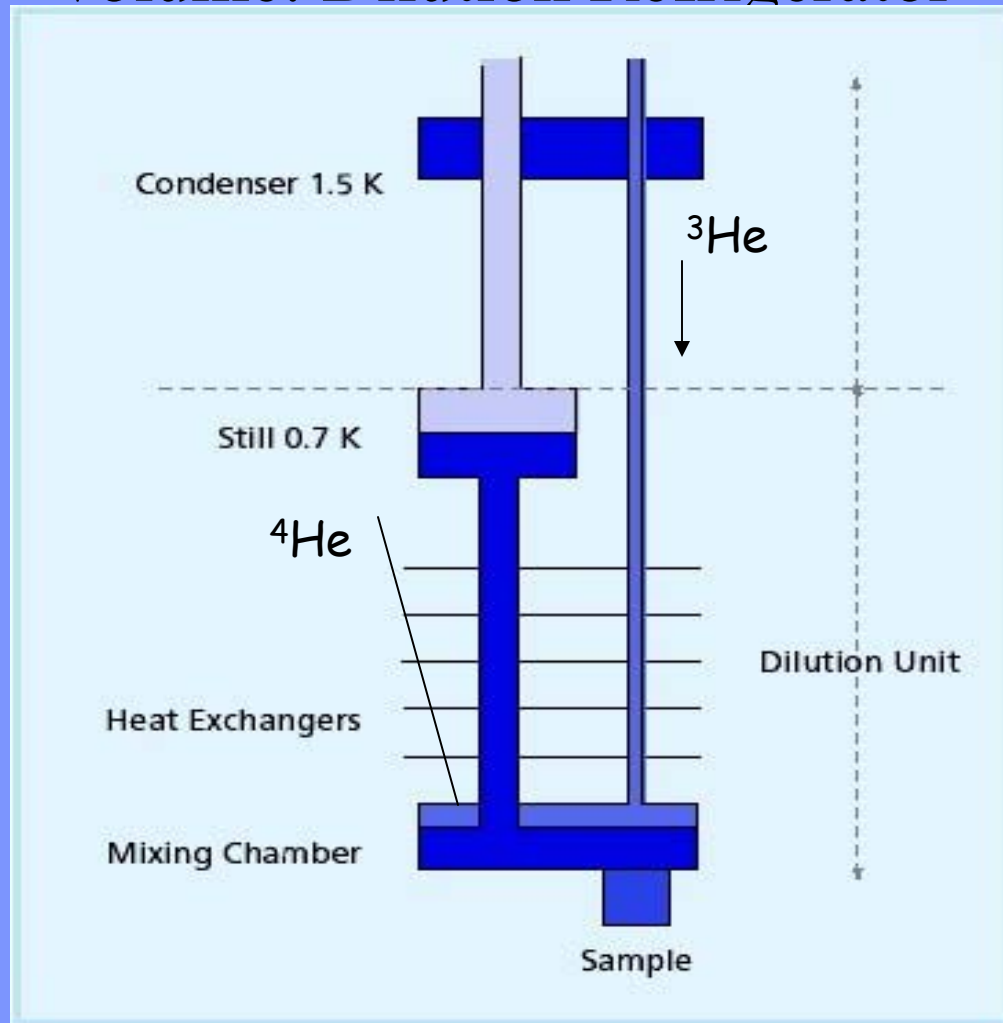


Temperature reduction by mixing at constant pressure and increase in specific volume: Cooling Tower





Mixing at constant pressure with increase in specific volume: Dilution Refrigerator

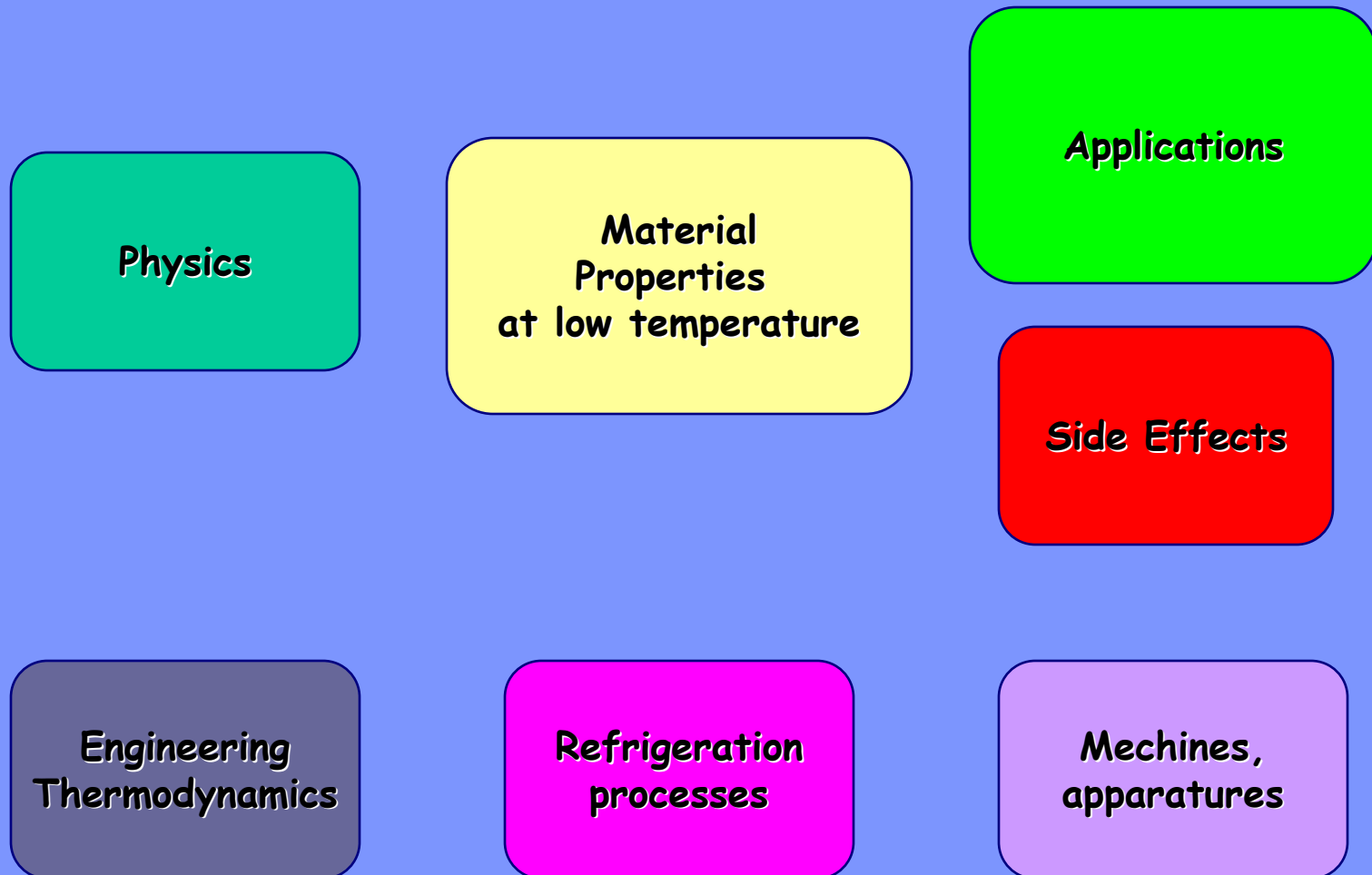




Summary of „Throttle Refrigeration“

- Throttling is a process at constant enthalpy
- Cooling effect depends on on the properties of refrigerant (the volume has to increase faster than the pressure decreases)
- Widely used in near ambient refrigeration
- Outside the two-phase region it is called the Joule-Thomson effect
- Effect can be extended to lower temperatures by recuperator
- Simple device without moving parts in cold section
- But: We can do better: Replace the throttling by a work extracting expander (tomorrow!)







1700	A m o n t o n	Tiefste Temperatur: -240 °C
1800	C h a r l e s , G a y - L u s s a c	Tiefste Temperatur: -273 °C
1842	R . M a y e r	Äquivalenz von Wärme und Arbeit
1857	W . S i e m e n s	Gegenstrom -Wärmeübertrager in Energie-Kreisläufen
1863	A n d r e w s	Kritischer Punkt
1872		Kühleffekt beim Drosseln von realem Gas (Joule-Thomson-Effekt)
1873	V a n d e r W a a l s	Prinzip der korrespondierenden Zustände
1877	C a i l l e t e t , P i c t e t	Flüssige Luft als Nebel
1883	W r ó b l e n s k i , O l s z e w s k i	Ruhig siedende flüssige Luft
1884	W r o b l e n s k i	Flüssiger Wasserstoff als Nebel
1892	D e w a r	Verspiegelte Vakuumisolation
1895	L i n d e , H a m p s o n	Kontinuierliche Luftverflüssigung mit Joule-Thomson Kreislauf
1898	D e w a r	Wasserstoffverflüssigung mit Joule- Thomson Kreislauf
1908	K a m m e r l i n g h O n n e s	Heliumverflüssigung mit Joule- Thomson Kreislauf