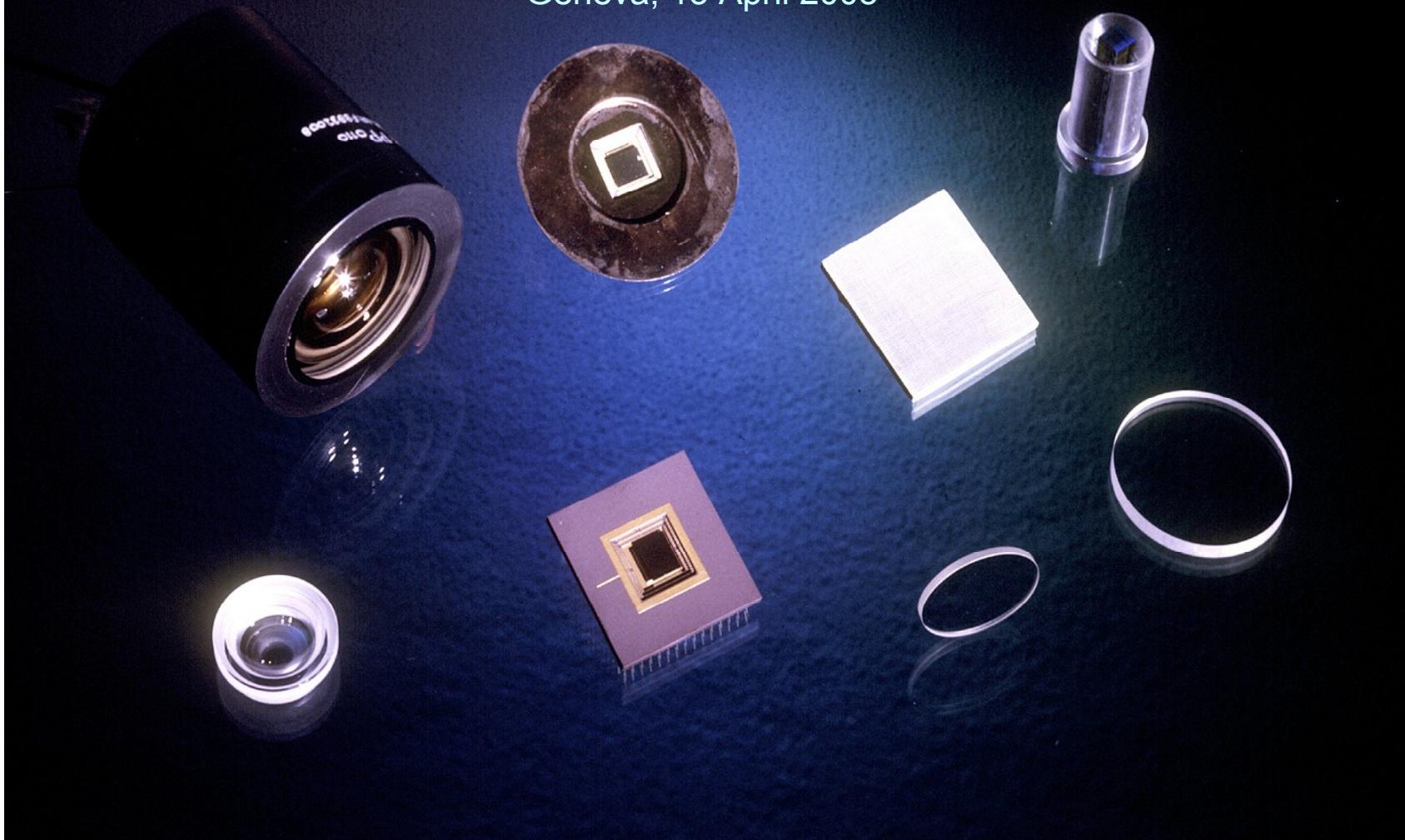


# A short Overview on Scintillators

By C. D'Ambrosio (CERN)

Geneva, 13 April 2005





# Outline

3a Scintillators

**Lecture 1 - Introduction** C. Joram, L. Ropelewski

**Lecture 2 - Tracking Detectors** L. Ropelewski, M. Moll

**Lecture 3 - Scintillation and Photodetection** C. D'Ambrosio, T. Gys

**3a A short overview on scintillators (a personal cut)**

**What are scintillators**

**Inorganic scintillators**

Main properties

Applications

**Organic scintillators**

Scintillation mechanisms

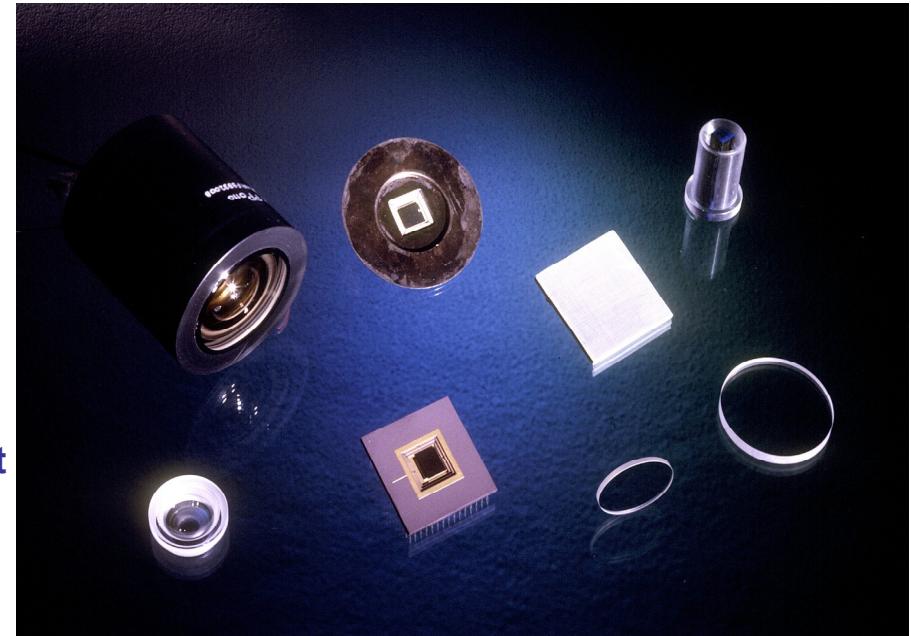
Plastic scintillators and their readout

Scintillating plastic fibres

**3b Photodetectors**

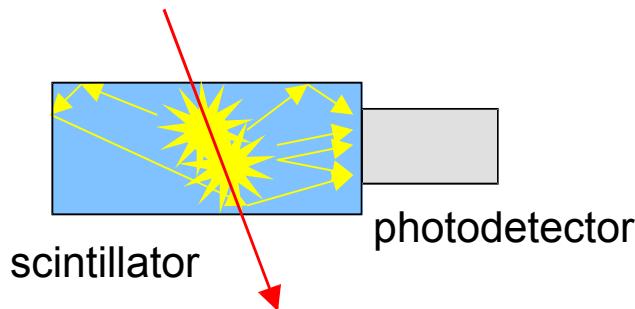
**Lecture 4 - Calorimetry, Particle ID** C. Joram

**Lecture 5 - Particle ID, Detector Systems** C. Joram, C. D'Ambrosio



# Introduction to Scintillators

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Energy deposition by a ionizing particle

→generation  
→transmission  
→detection } of scintillation light

Two categories: Inorganic and organic scintillators

Inorganic  
(crystalline structure)

Up to 40000 photons per MeV  
High Z  
Large variety of Z and  $\rho$   
Undoped and doped  
ns to  $\mu$ s decay times  
Expensive

E.m. calorimetry ( $e, \gamma$ )  
Medical imaging  
Fairly Rad. Hard (100 kGy/year)

Organic  
(plastics or liquid solutions)

Up to 10000 photons per MeV  
Low Z  
 $\rho \sim 1 \text{ gr/cm}^3$   
Doped, large choice of emission wavelength  
ns decay times  
Relatively inexpensive

Tracking, TOF, trigger, veto counters,  
sampling calorimeters.  
Medium Rad. Hard (10 kGy/year)



## Inorganic Scintillators

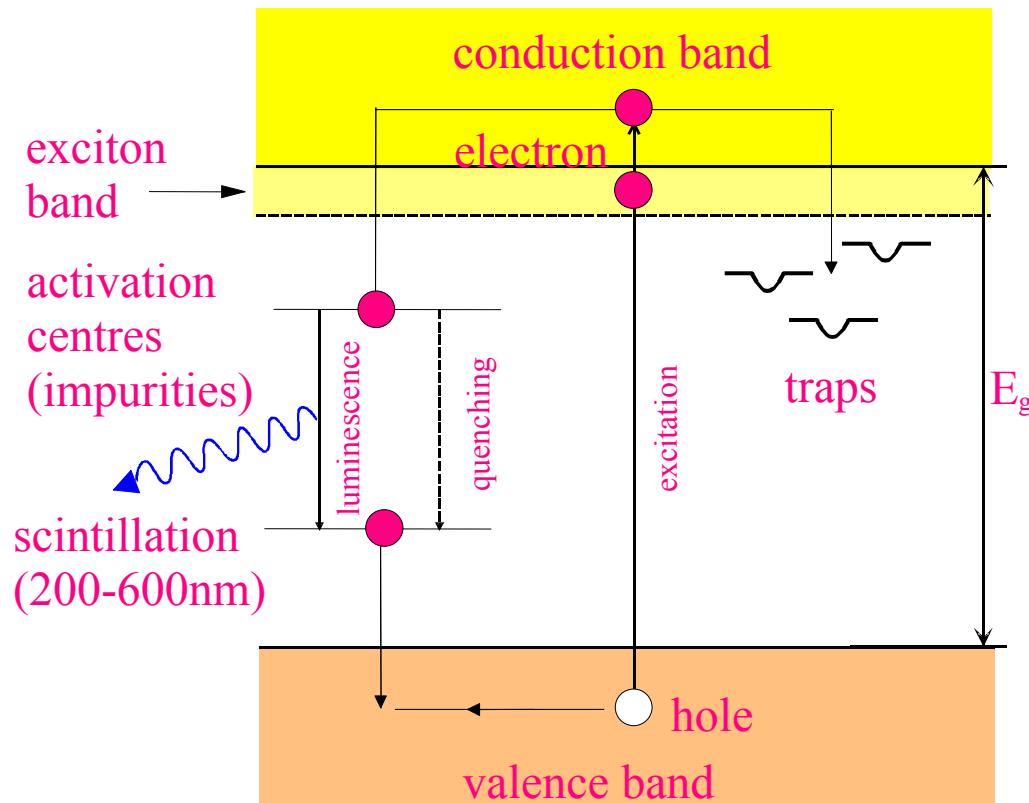
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### Inorganic Scintillators

- Basic scintillation mechanisms in crystals and liquefied noble gases
- Temperature dependence of scintillation yield
- Photon absorption processes in crystals
- Table of common scintillators properties
- Applications:
  - X-ray and Gamma Spectroscopy
  - Imaging

## Basic crystal scintillation mechanism

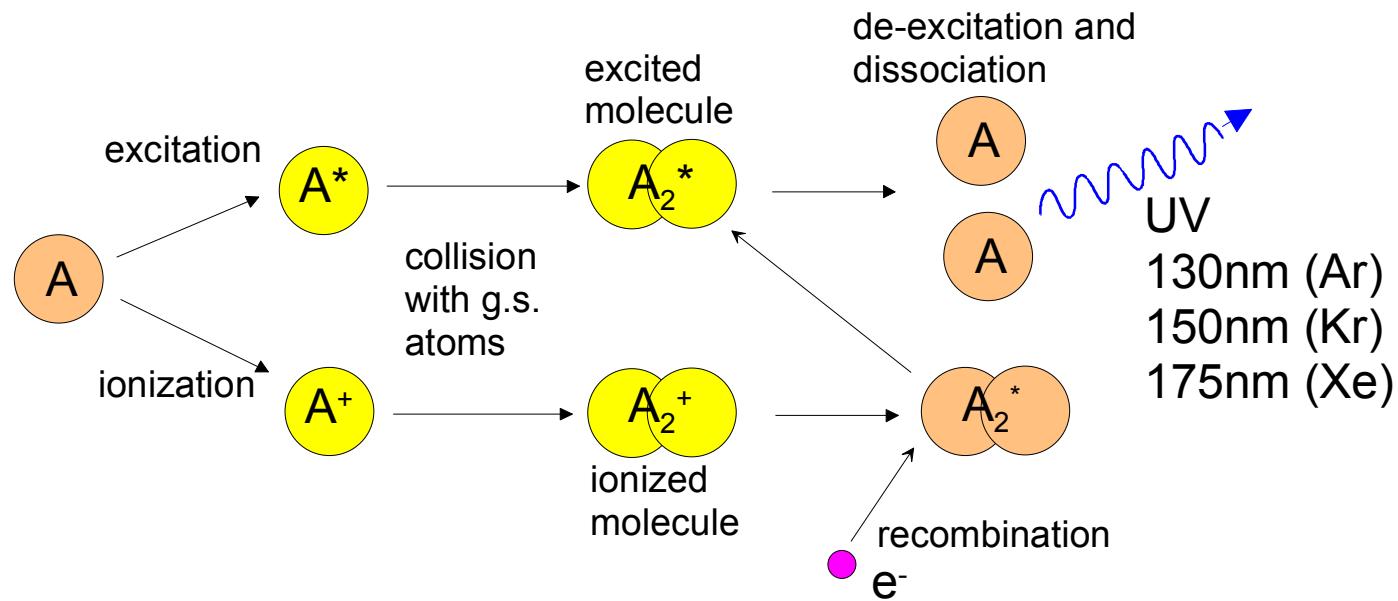
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Warning, sometimes  $\geq 2$  time constants:

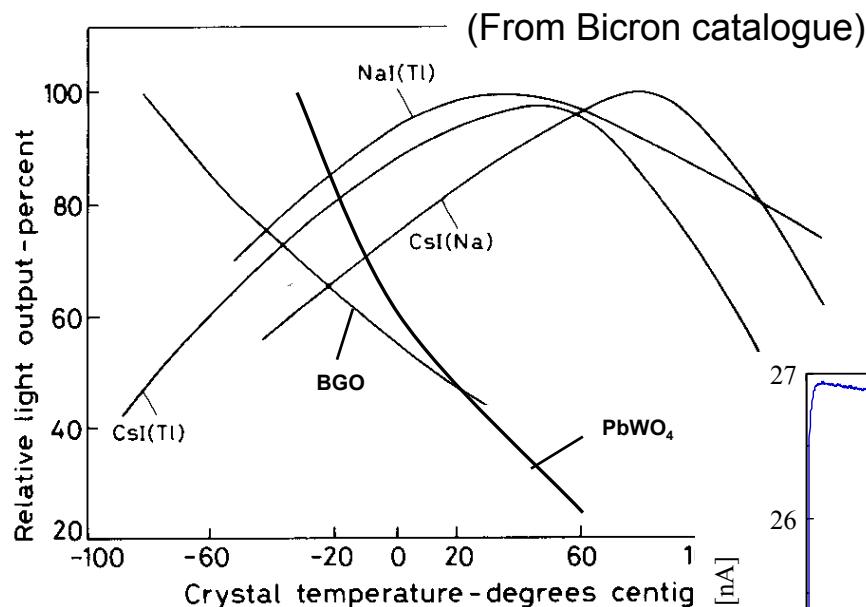
- fast recombination (ns- $\mu$ s) from activation centers
- delayed recombination due to trapping ( $\mu$ s-ms)
- full control of growth, doping and impurities is imperative to optimize light yield, transmission and decay time

## Liquefied noble gases: LAr, LXe, LKr

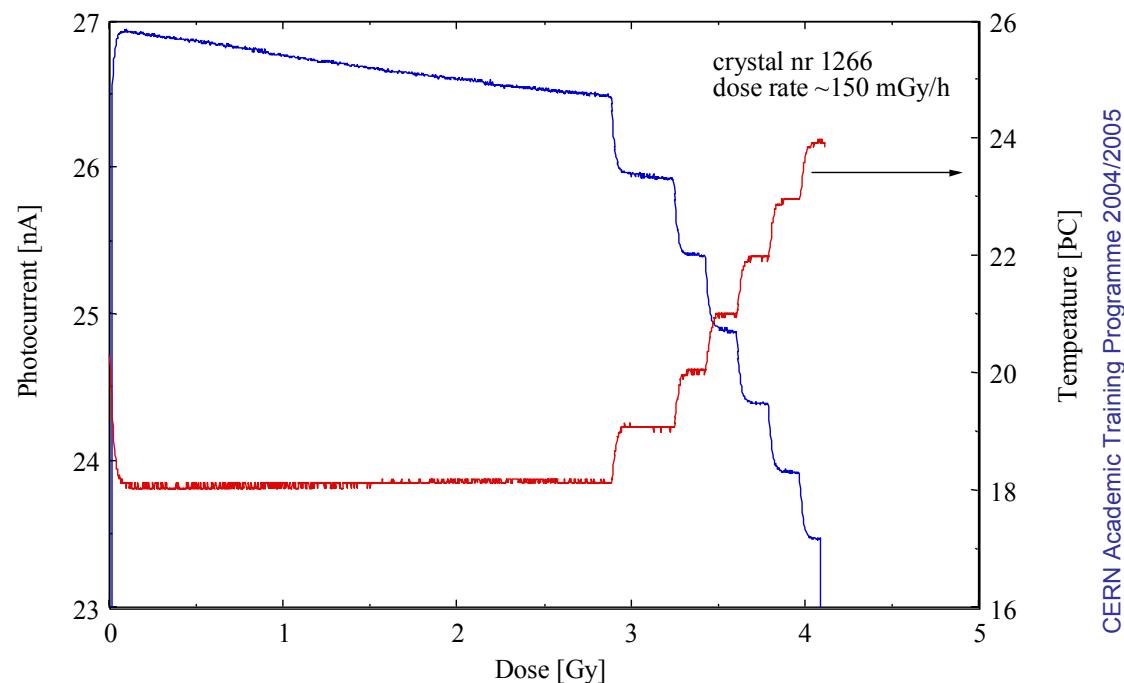


Also here one finds 2 time constants: from a few ns to 1  $\mu$ s.

## Light output of crystals depends on temperature



Low dose irradiation of an old PbWO<sub>4</sub> crystal and check of the temperature dependence of its light yield (1996). The blue curve is the photocurrent generated by the irradiation and the red curve is the temperature of the sample.



The PbWO<sub>4</sub> crystal is used in CMS ECAL and its ~2% light yield decrease per °C asks for temp. control and monitoring

## Photon absorption in crystals

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The intensity  $I$  of a gamma beam traversing a target of thickness  $d$  is

$$I = I_0 e^{-\mu d}$$

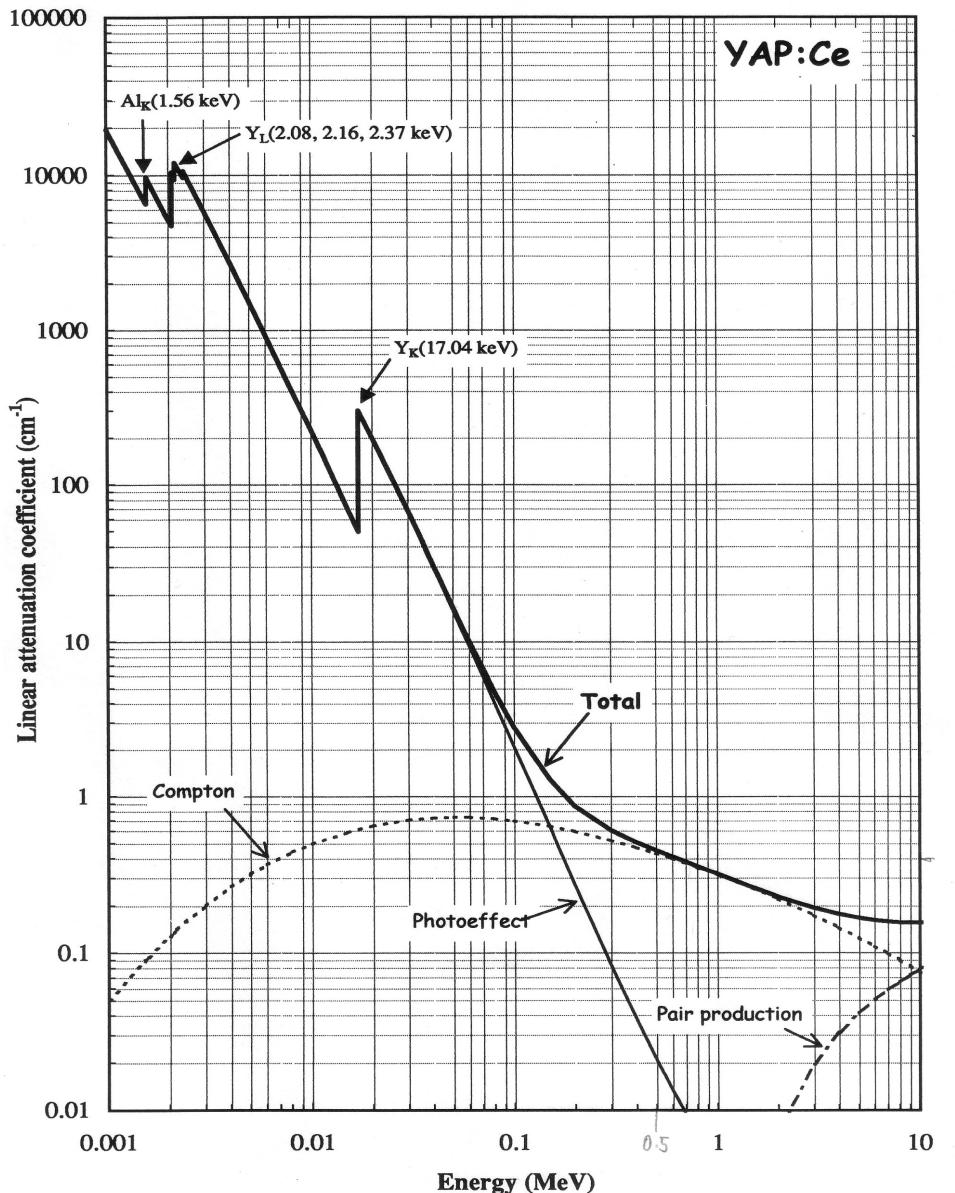
Where  $\mu$  is the sum of three processes taking place in the material:

Photoel. Abs.  $\rightarrow Z^4$  to  $Z^5 \rightarrow E^{-3.5}$  to  $E^{-1}$

Compton scatt.  $\rightarrow Z \rightarrow E^{-1}$

Pair production  $\rightarrow Z^2 \rightarrow \ln E$

(curve will extend to higher energies, see Christian's talk)





## Properties of some crystal scintillators

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Scintillator composition	Density (g/cm <sup>3</sup> )	Index of refraction	Wavelength of max.Em. (nm)	Decay time Constant (μs)	Scinti Pulse height <sup>1)</sup>	Notes
Nal(Tl)	3.67	1.9	410	0.25	100	2)
CsI	4.51	1.8	310	0.01	6	3)
CsI(Tl)	4.51	1.8	565	1.0	45	3)
CaF <sub>2</sub> (Eu)	3.19	1.4	435	0.9	50	
BaF <sub>2</sub>	4.88	1.5	190/220 310	0.0006 0.63	5 15	
BGO	7.13	2.2	480	0.30	10	
CdWO <sub>4</sub>	7.90	2.3	540	5.0	40	
PbWO <sub>4</sub>	8.28	2.1	440	0.020	0.1	
CeF <sub>3</sub>	6.16	1.7	300 340	0.005 0.020	5	
GSO	6.71	1.9	430	0.060	40	
LSO	7	1.8	420	0.040	75	
YAP	5.50	1.9	370	0.030	70	

1) Relative to Nal(Tl) in %; 2) Hygroscopic; 3) Water soluble



## Most common applications of inorganic scintillators

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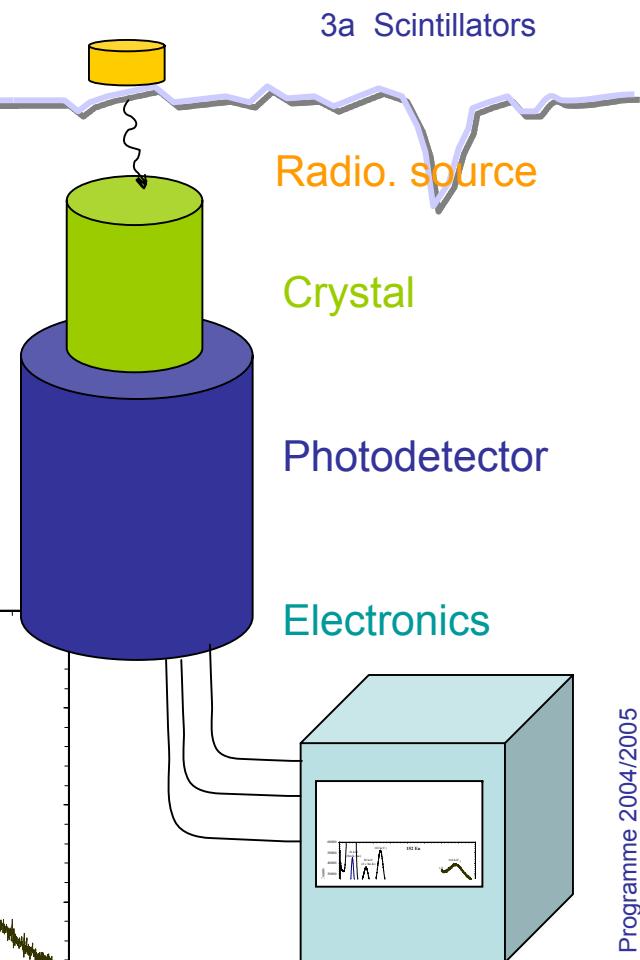
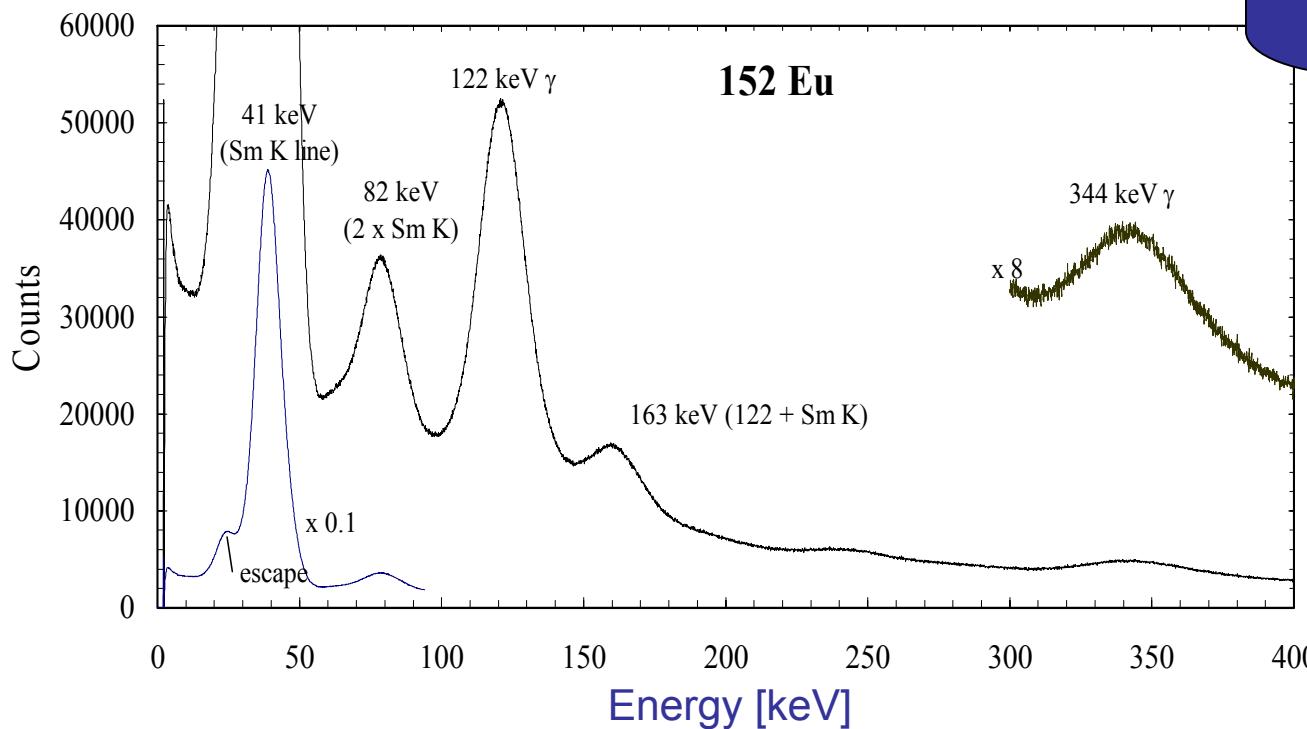
- Calorimetry (for HEP, see Christian's lecture)
- X-ray and gamma spectroscopy
- Imaging
  - Positron Emission Tomography (PET) in medical imaging
  - Gamma Imaging (Anger camera) (see Thierry's talk)
- Monitoring in nuclear plants
- Oil wells, Mining, etc.

## X-ray and gamma spectroscopy

The simple set-up:

each detected gamma provides an amplitude signal, which fills a pulse-height spectrum.

We can **monitor** radiation, **study** atomic or nuclear spectra, **research** and **characterize** new crystals, **test** new photo-detectors, **produce** special probes, etc..

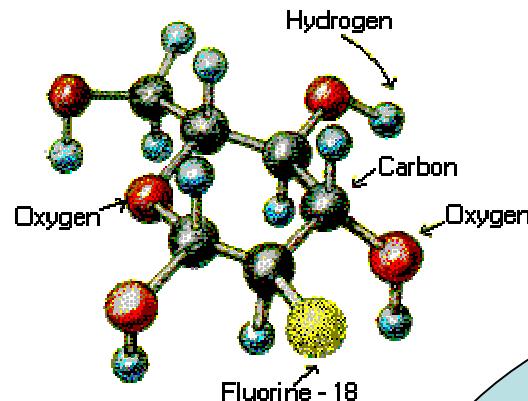


Irradiation of a  
 $\text{YAP}(\text{Ce})$  crystal by  
 $^{152}\text{Eu}$  on a HPD

# Positron Emission Tomography (PET)

<http://www.triumf.ca/welcome/petscan.html>

**2-fluoro-  
2-deoxy-D-glucose  
"FDG"**

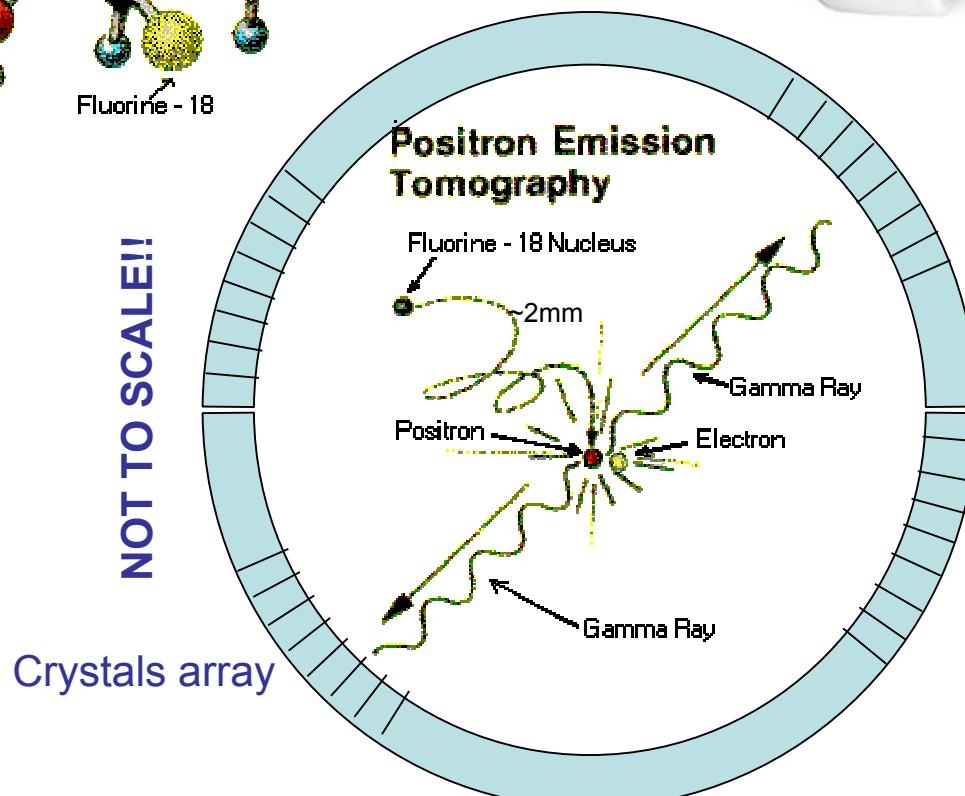


<http://www.medical.philips.com/main/products/pet/products/gemini/clinicalimages/10/index.asp>

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**NOT TO SCALE!!**

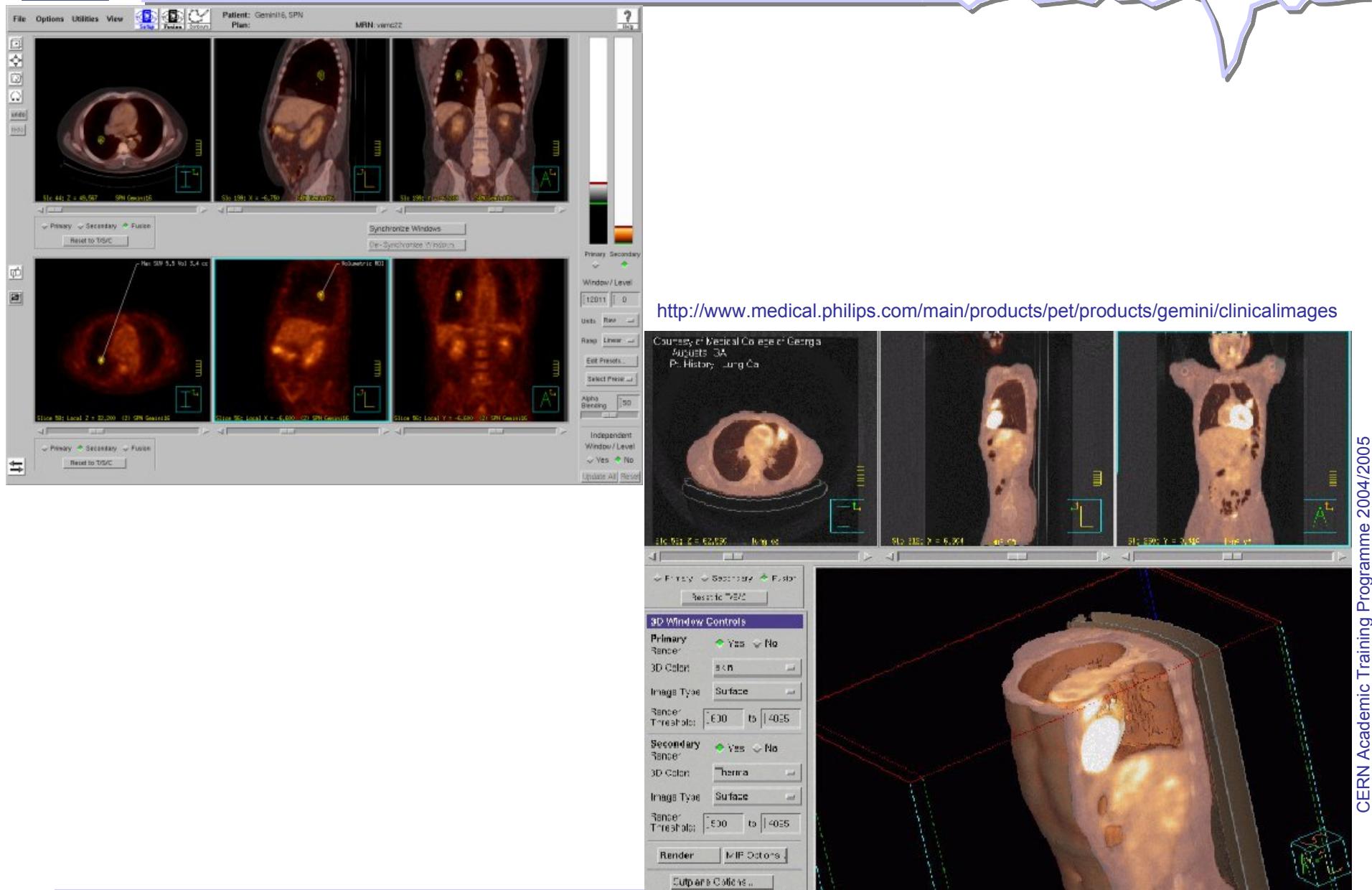


2 x 511 keV energy  
 $\gamma$ - $\gamma$  co-linearity  
 time coincidence  
 reconstruct functional image



## PET images to be found on the web

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<http://www.medical.philips.com/main/products/pet/products/gemini/clinicalimages>



## Main R&D on inorganic scintillators

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- Higher densities for higher Z (improve photoabsorption)
- High light yield (NaI(Tl) light yield still unchallenged)
- Short decay time (improve time resolution)
- Improve light coupling with photon detector
- More radiation hard
- Inexpensive, “easy” to manufacture, reproducible
- Large size, easy handling and “machinable”

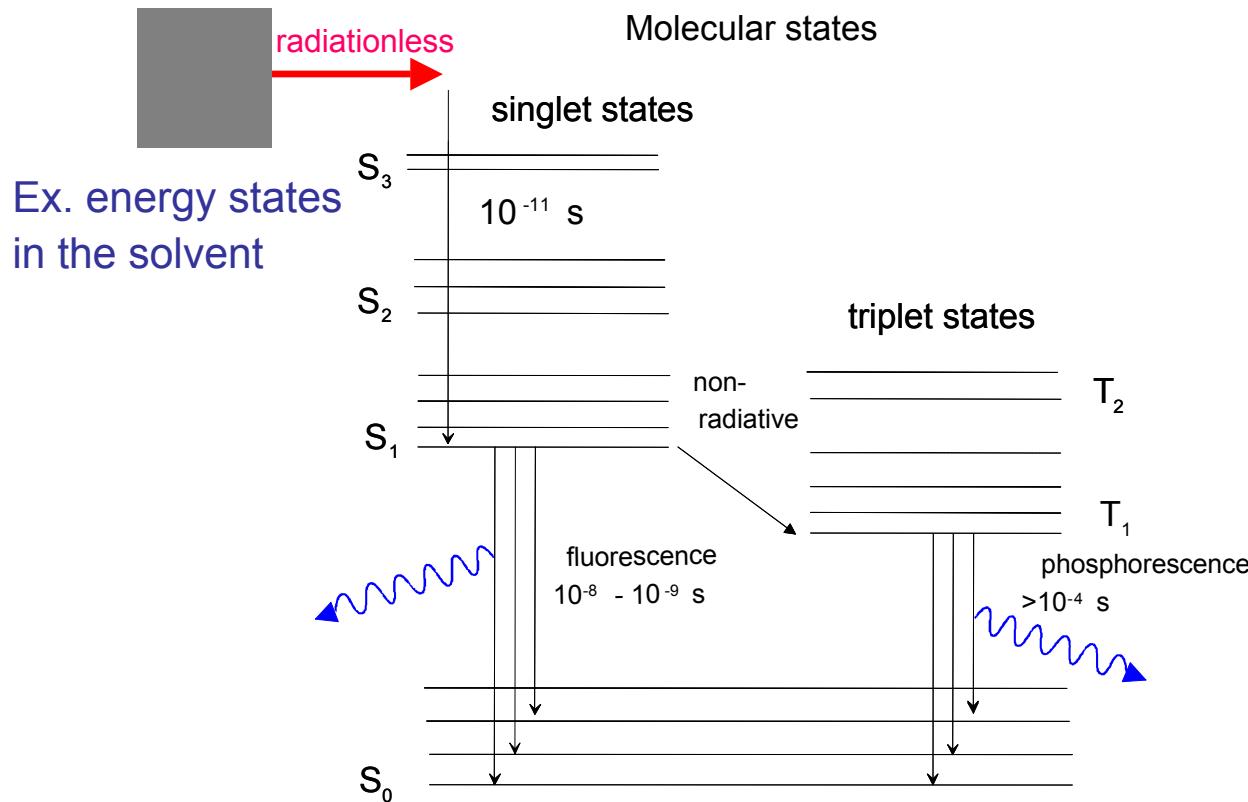


# Organic Scintillators

- Basic scintillation mechanisms in organic scintillators
- Förster energy transfer and self absorption
- One dopant and two dopants scintillators
- Readout of scintillators
- Applications of organic scintillators
- Example: small diameter scintillating fibres and their readout

They usually consist of a solvent + scintillator and a secondary fluor as wavelength shifters.

A traversing ionizing particle releases energy in the solvent. Then, energy flows radiationless\* to the scintillator. Finally, light emitted by the scintillator is absorbed (radiative transfer\*\*) and re-emitted at longer wavelength by the secondary fluor.



A fluor has its absorption and emission spectra shifted. The two peaks difference is called Stokes shift

\*fast and local energy transfer via non-radiative dipole-dipole interactions (Förster transfer).

\*\*~ $1/R^2$  light attenuation

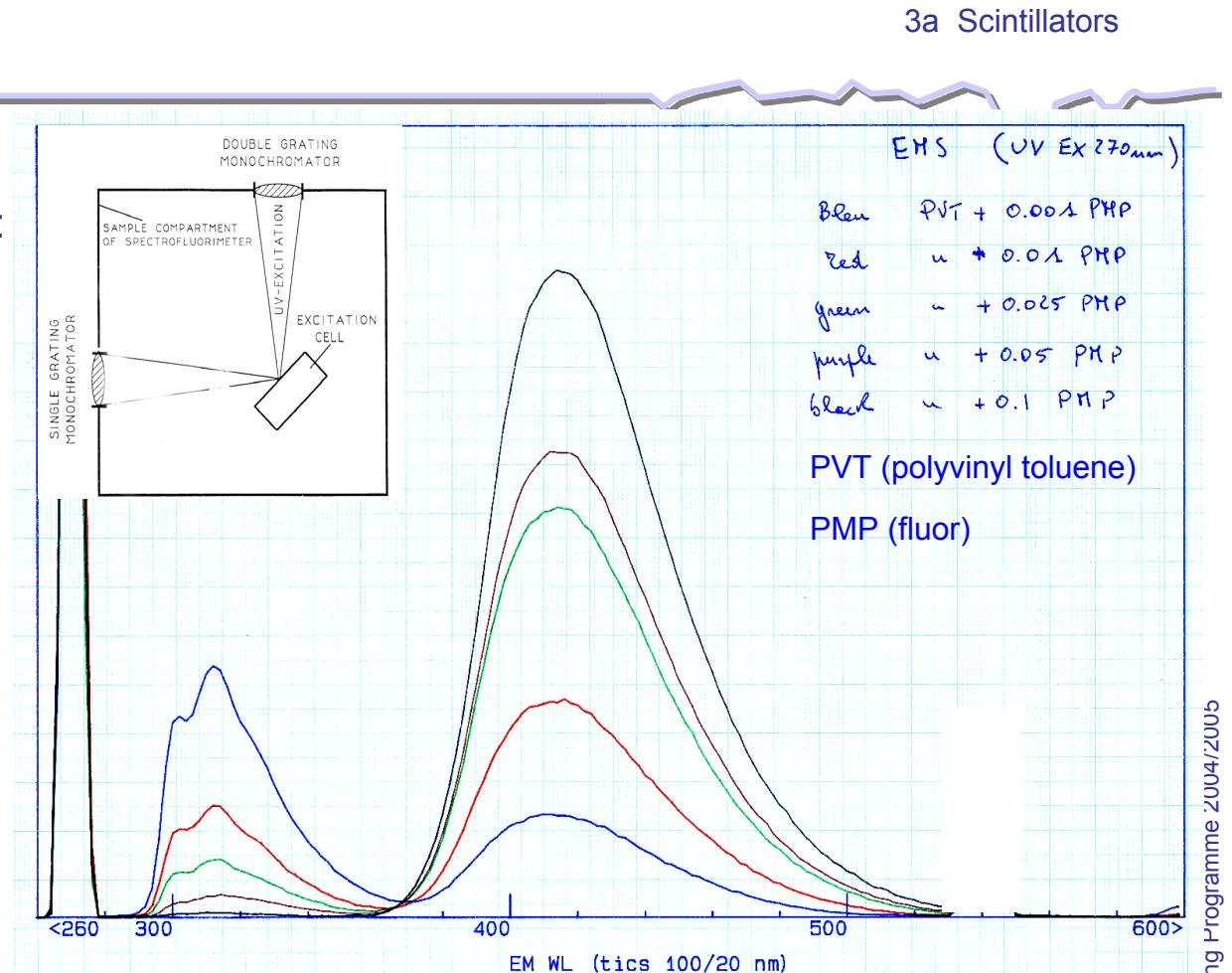
## The Förster transfer

Efficient Förster transfer asks for high concentrations of scintillator:

$$k_{D \rightarrow A} \propto \frac{1}{\tau_D R_{DA}^6} \int \frac{f_D(\nu) \epsilon_A(\nu)}{\nu^4} d\nu$$

where the integral is the overlap between Acceptor abs. and Donor emission spectra and R is the D-A distance

But...



**Self-absorption** increases with doping concentration, that is smaller light yields and shorter attenuation lengths:

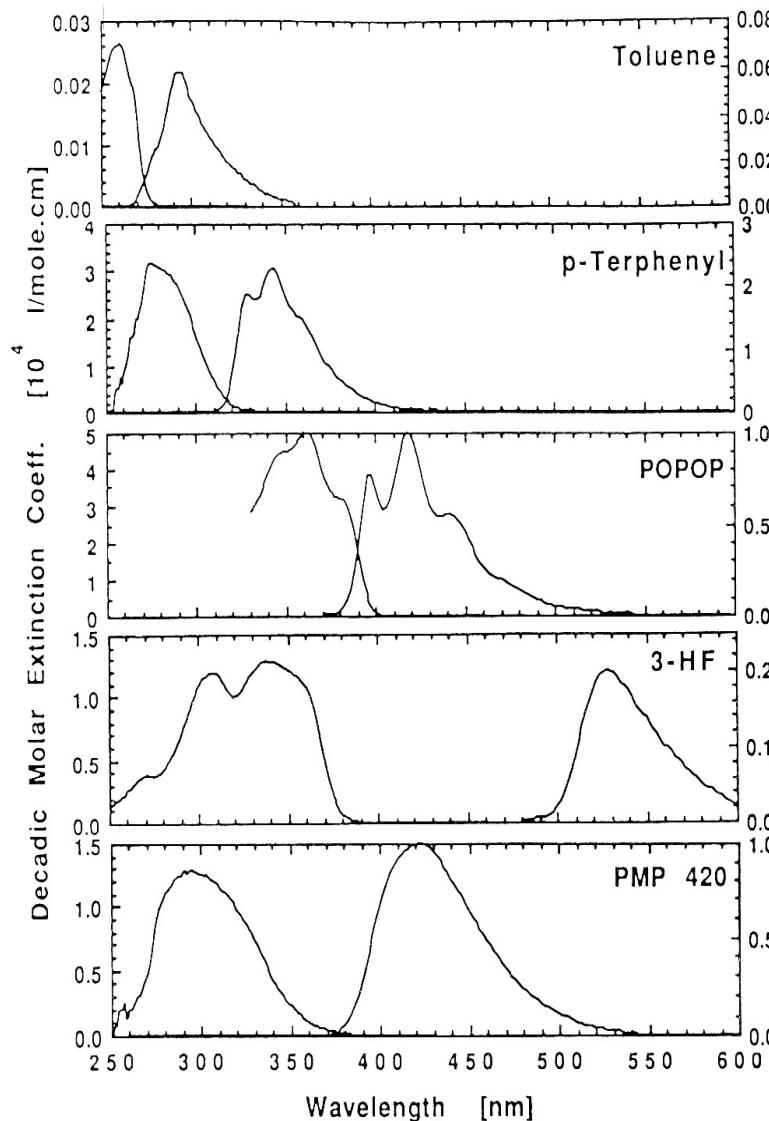
Therefore, add a low concentration wavelength shifter (2 dop. Scheme)

or

Find large Stokes shift dopants (1 dop. Scheme, ex. 3hf, pmp)

## Two / One Dopant scheme

Abs. and emission spectra



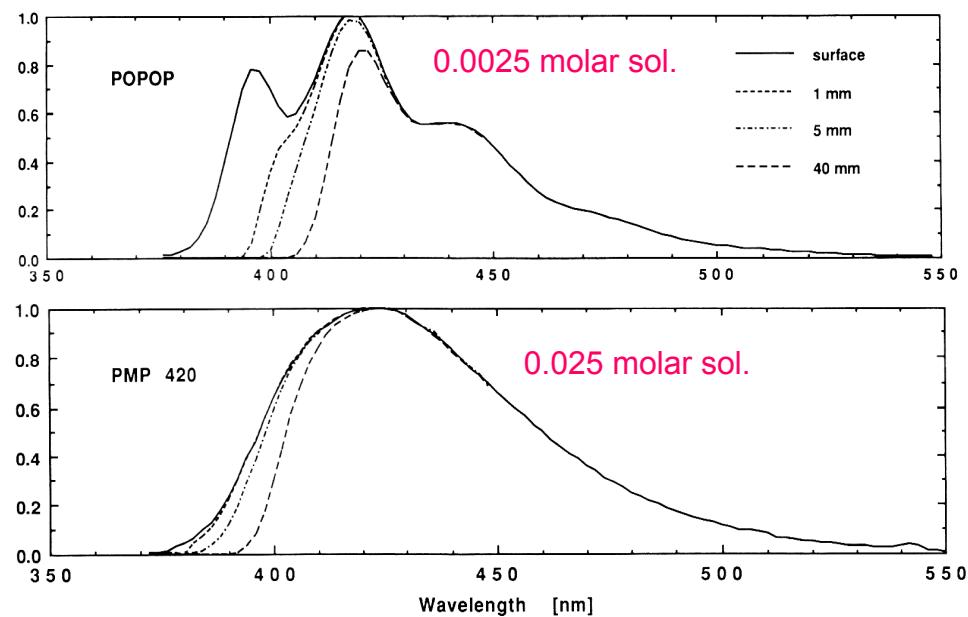
solvent + scintillator + wave shifter

Förster

Radiative

solvent + large stokes shift scintillator

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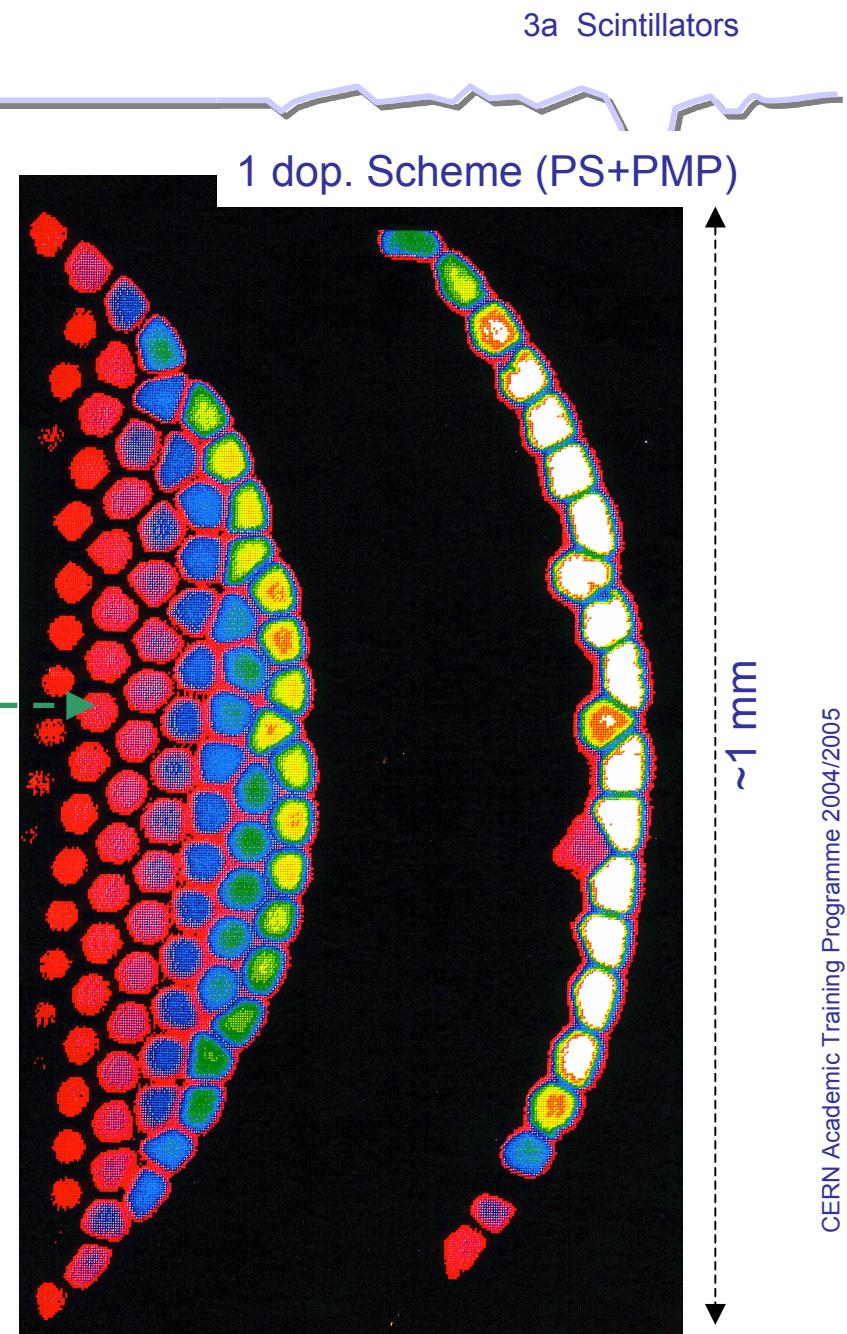


Dopants in toluene: large Stokes shift dopants feature a much smaller self-absorption

## Application of the two types of doping

The **two dopants** scheme is the most common: it is applied to all plastic scintillators, down to  $\sim 1\text{mm}$  fibres.

The **one dopant** scheme is needed to keep the light emission local (only Förster transfer) as it is the case for small diameter fibres





## Scintillators readout

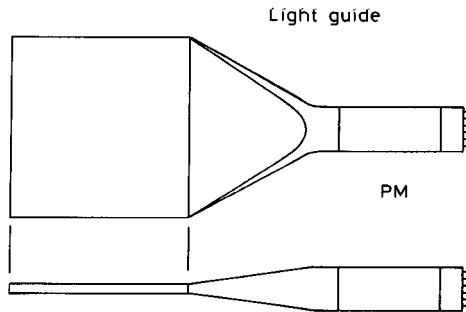
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Readout has to be adapted to geometry, granularity and emission spectrum of scintillator.

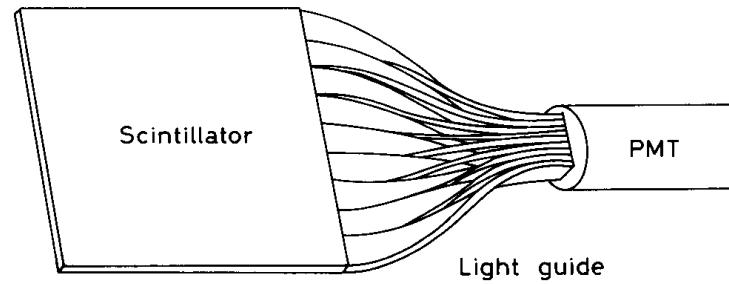
Geometrical adaptation:

- Light guides: transfer by total internal reflection

(+outer reflector)

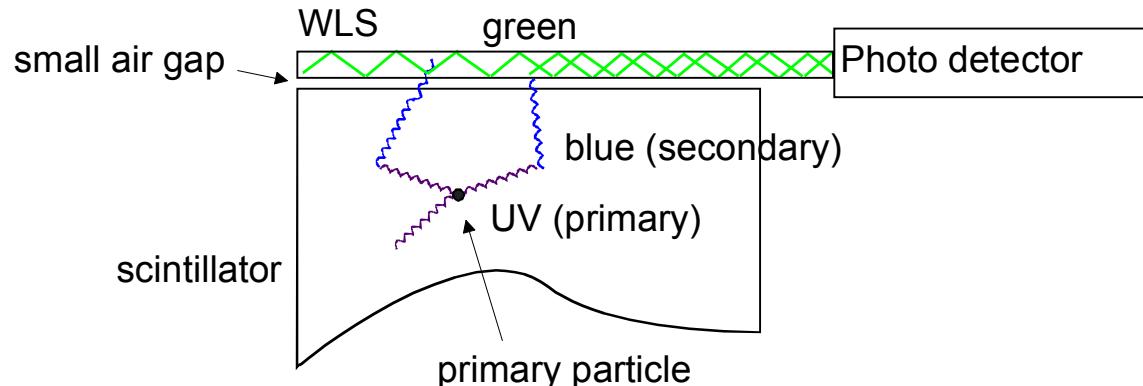


“fish tail”



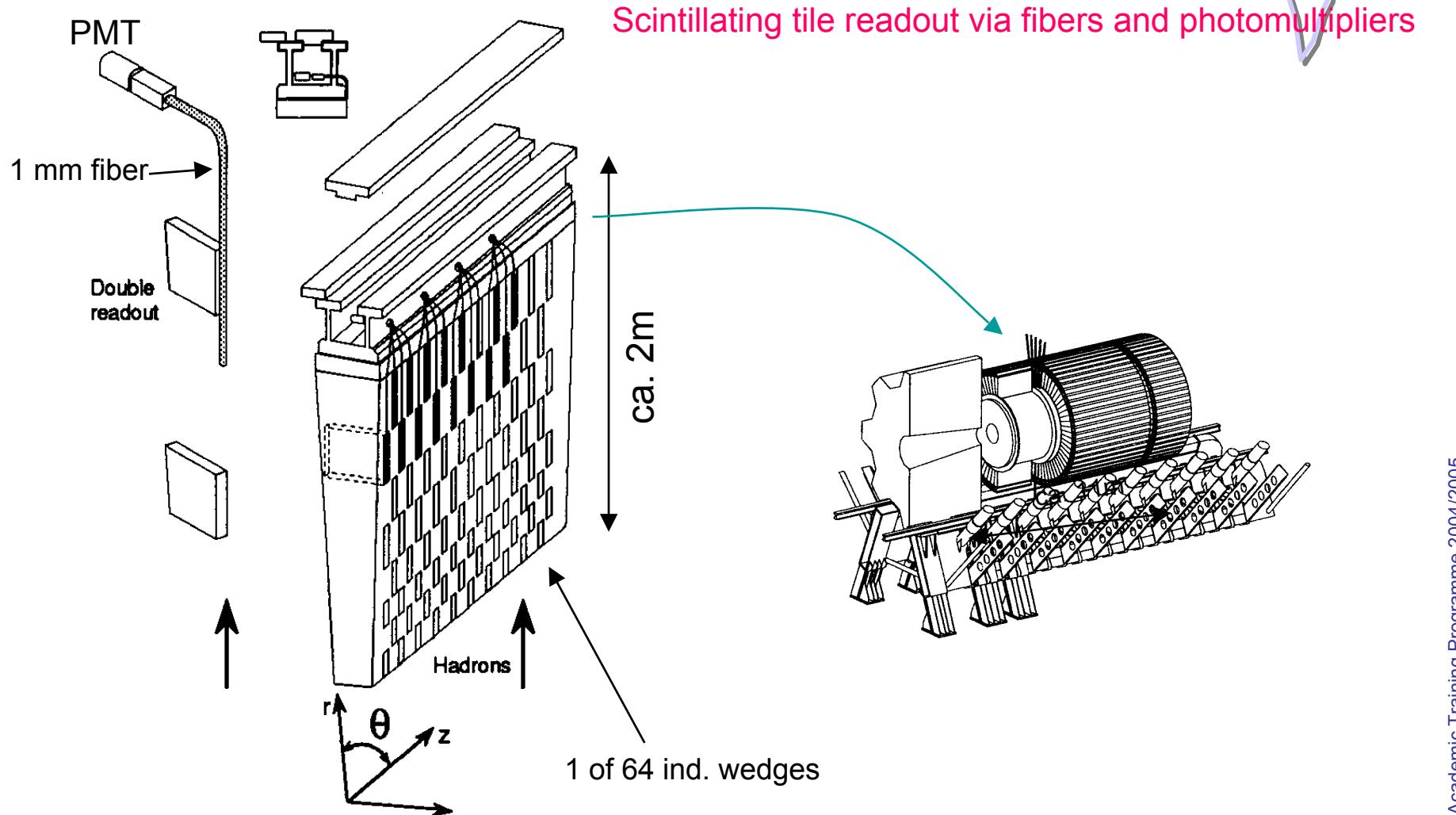
adiabatic

- wavelength shifter (WLS) bars





## ATLAS Hadron Calorimeter



Periodical arrangement of scintillator tiles  
(3 mm thick) in a steel absorber structure

(ATLAS TDR)

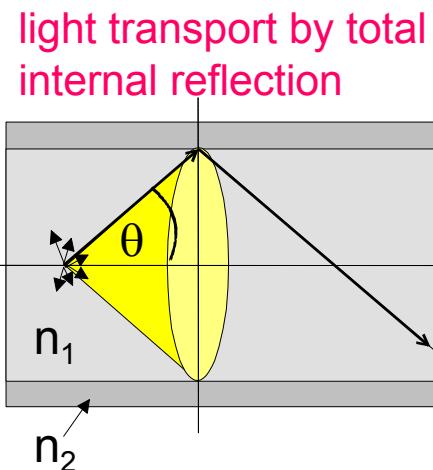
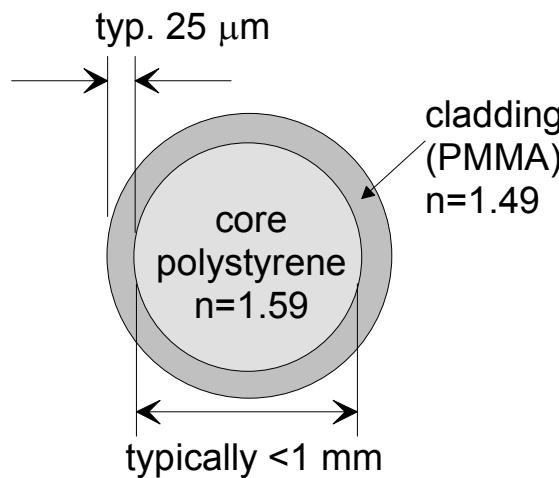
## Most common applications of organic scintillators

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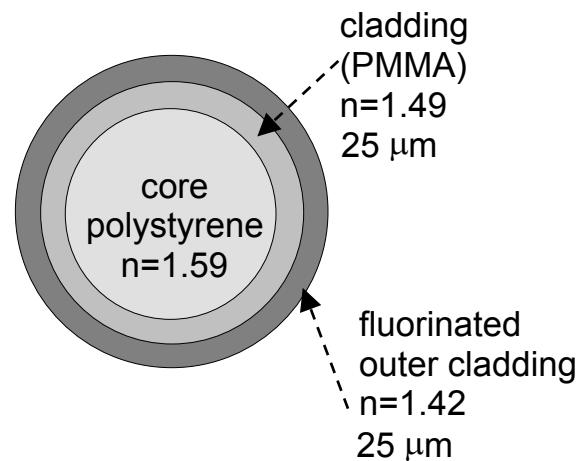
Large volume liquid or solid detectors (in form of tiles): underground experiments, sampling calorimeters (**HCAL** in CMS or **ATLAS**, etc.), counters, light guides.

High precision, small volume active targets and fibre tracking (UA2, D0, CHORUS).

As an example, a **scintillating plastic fibre** working principle:



Double cladding system  
(developed by RD7)



$$\frac{d\Omega}{4\pi} = 0.5 (1 - \cos^2 \theta) = 3\%$$

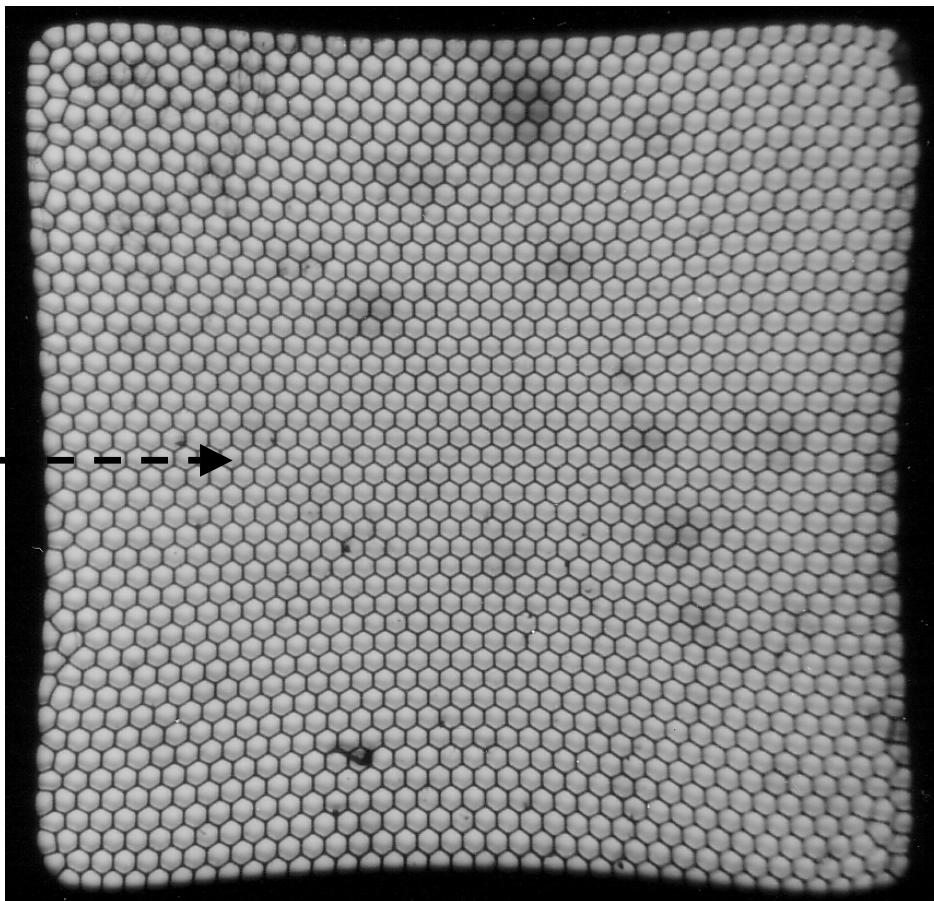
$$\theta \leq \arccos \frac{n_2}{n_1} \approx 69.6^\circ$$

$$\frac{d\Omega}{4\pi} = 0.5 (1 - \cos^2 \theta) \approx 5.3\%$$



## Small diameter scintillating fibres

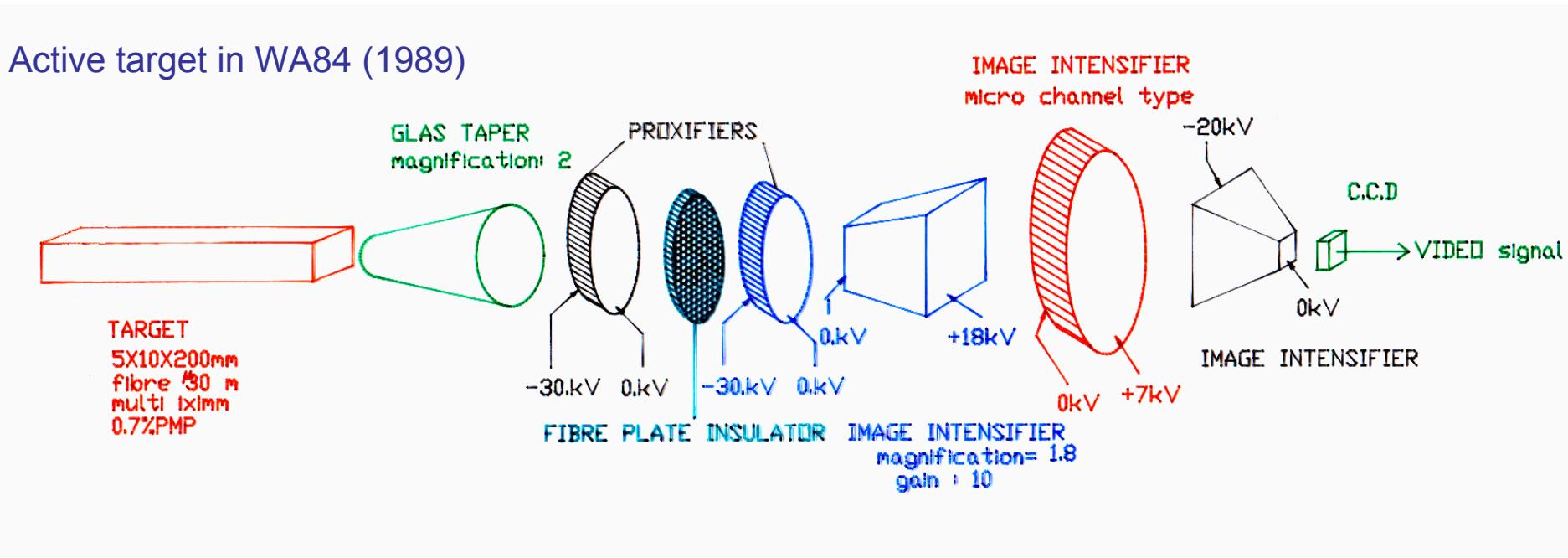
Developed in RD7, they consist of bundles of hexagonal fibres (typ. 60  $\mu\text{m}$  dia., 2.5 mm bundle size)



Images of tracks from 5 GeV/c pions (1989)

Beautiful tracks with only **2.2% of  $X_0$**  and **>20 hits**, but...

**The readout of fibres is a key point for the whole system**



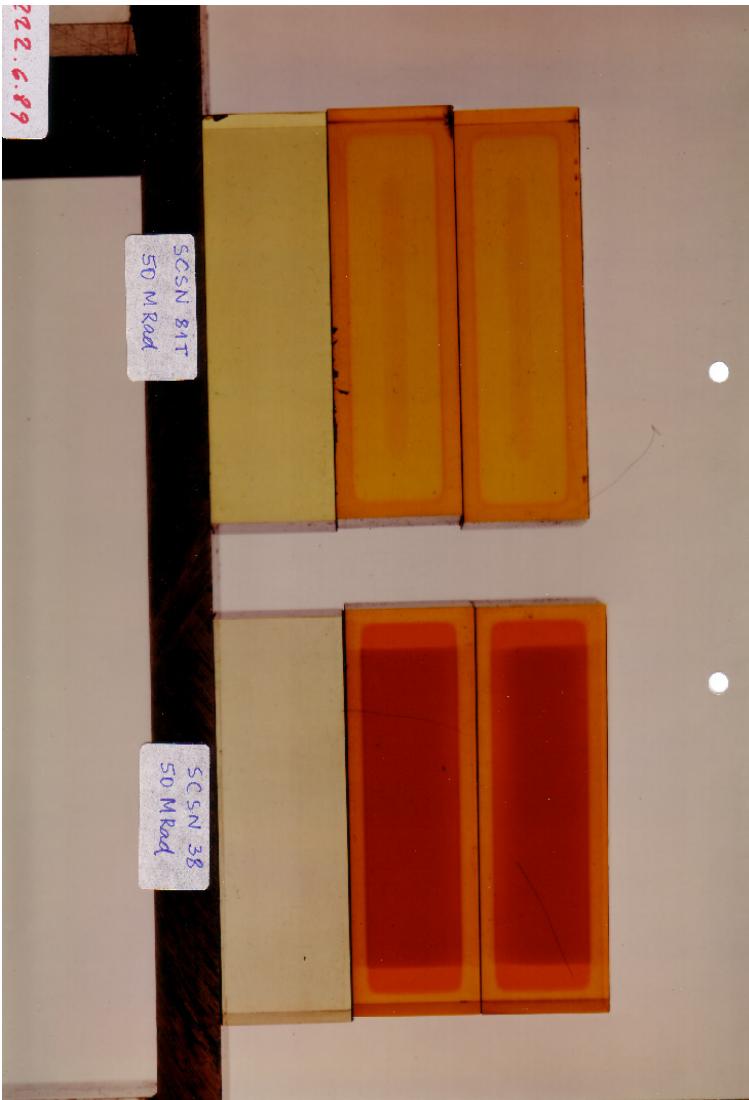
For a tracker, the readout has to match the high granularity of the fibres and bunch crossing time, but stay in the experiment's budget, size, etc.: **a real challenge**.

The **ISPA tube** was especially developed for this (see Thierry's talk)

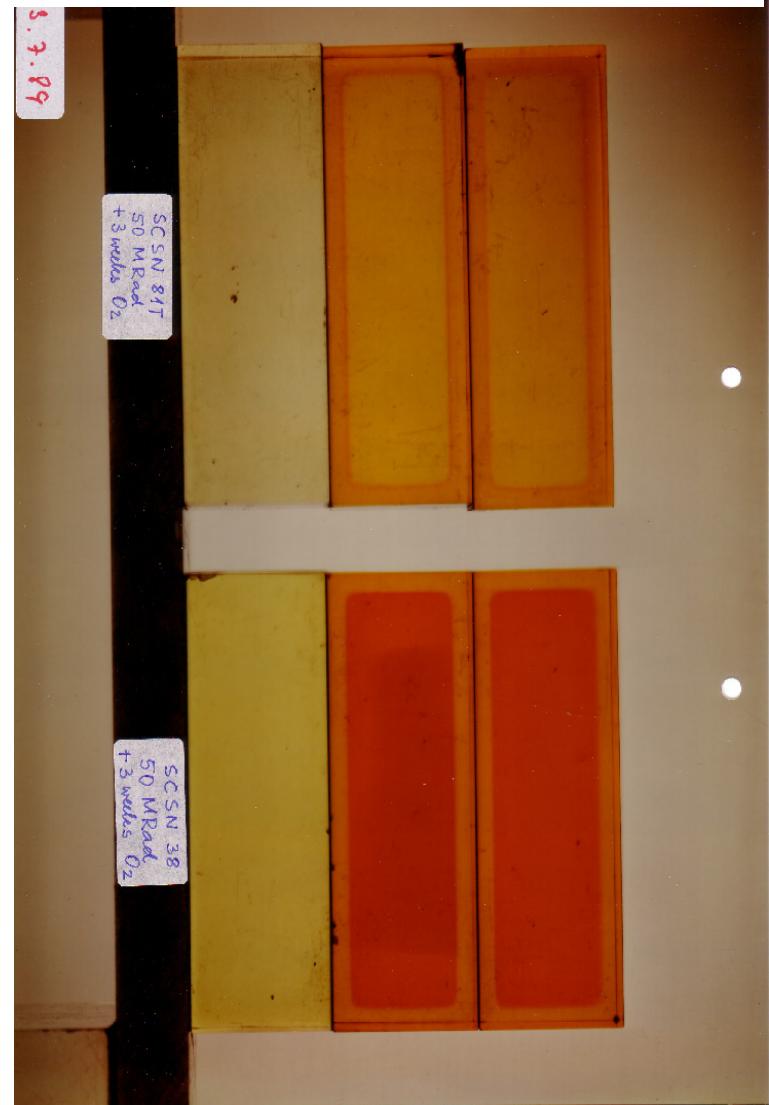


## Radiation Hardness in Plastic

500 kGy irradiation of SCSN81T and SCSN38 from Kuraray



and after 3 weeks recovery in an oxygen-rich atmosphere





## R&D on organic scintillators

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- New dopants with better light yield and larger Stokes shift
- High granularity **readout** of fibres
- Larger attenuation lengths in plastic fibres
- New **radiation hard** plastics to stand 100 kGy/year dose



## References

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