

High resistivity, nitrogen-enriched FZ Si wafers for particle detectors

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OUTLINE

- Introduction
- Wafer specification
- Supply Chain of Si Wafer @ Topsil
- Resistivity Control
- Nitrogen Control
- Phosphorus Control
- Lifetime Control
- Wafer availability
- Summary

Introduction

- NitroSil project objectives included:
- development of Crystal growth and wafer production technology of a new product: nitrogen enriched high resistivity FZ silicon wafers
- this objective has been achieved
- features of nitrogen doped wafers compared to standard Si:FZ wafers are presented

FZ Si:N wafer parameters

Prime Material

Single Side Polished

Silicon: FZ

Diameter: 4"

Orientation: <100>

Type: n

Thickness: 300 +/- 15 μm

Resistivity: 2200 +/- 200 Ohm-cm

Dislocations: free

N concentration: 1-2E15 at/cm⁻³

O concentration: <1E16 at/cm⁻³

C concentration: <5E15 at/cm⁻³

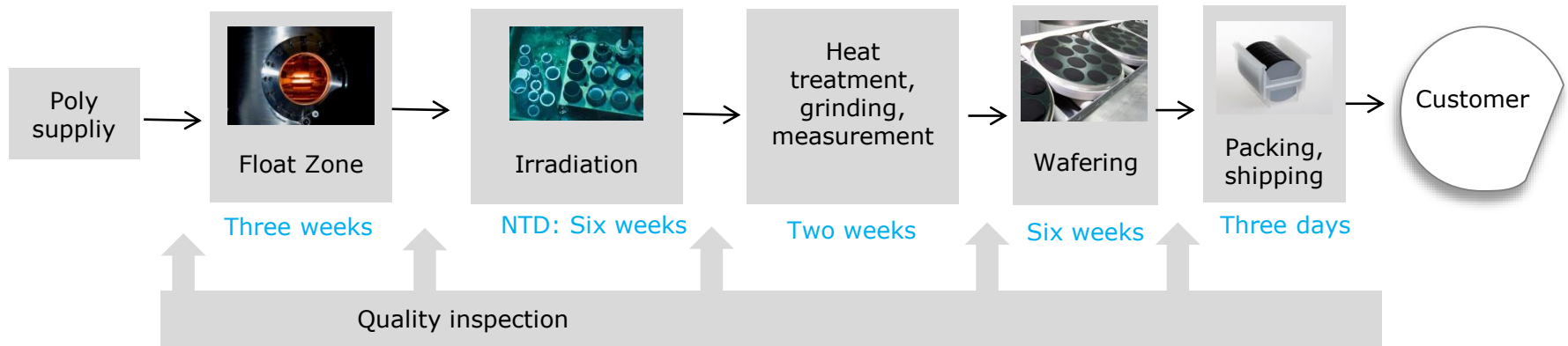
Dopant : Phosphorus ($\sim 2.3\text{E}12$ at/cm⁻³)

Doped with phosphorus in Neutron Transmutation Process



TOPSIL SUPPLY CHAIN IN OVERVIEW

Standard process:



Nitrogen enriched wafers:

- Nitrogen doping during the crystal growth
- Phosphorus doping by Neutron Transmutation (NTD)
- Annealing to remove radiation damage
- Annealing to obtain target resistivity (decomposition of Nitrogen-Oxygen complexes)
- Key is nitrogen control on ingot level, after each step

FTIR Nitrogen Concentration



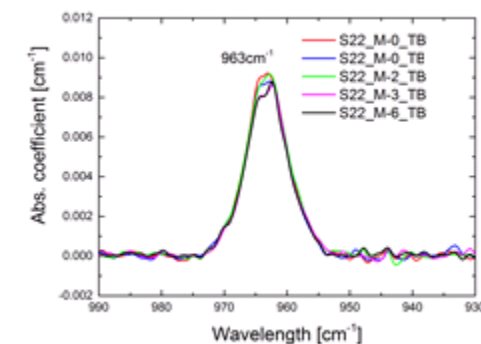
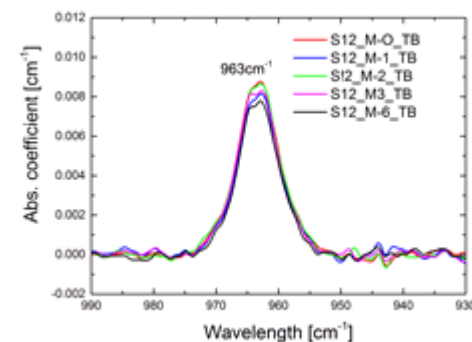
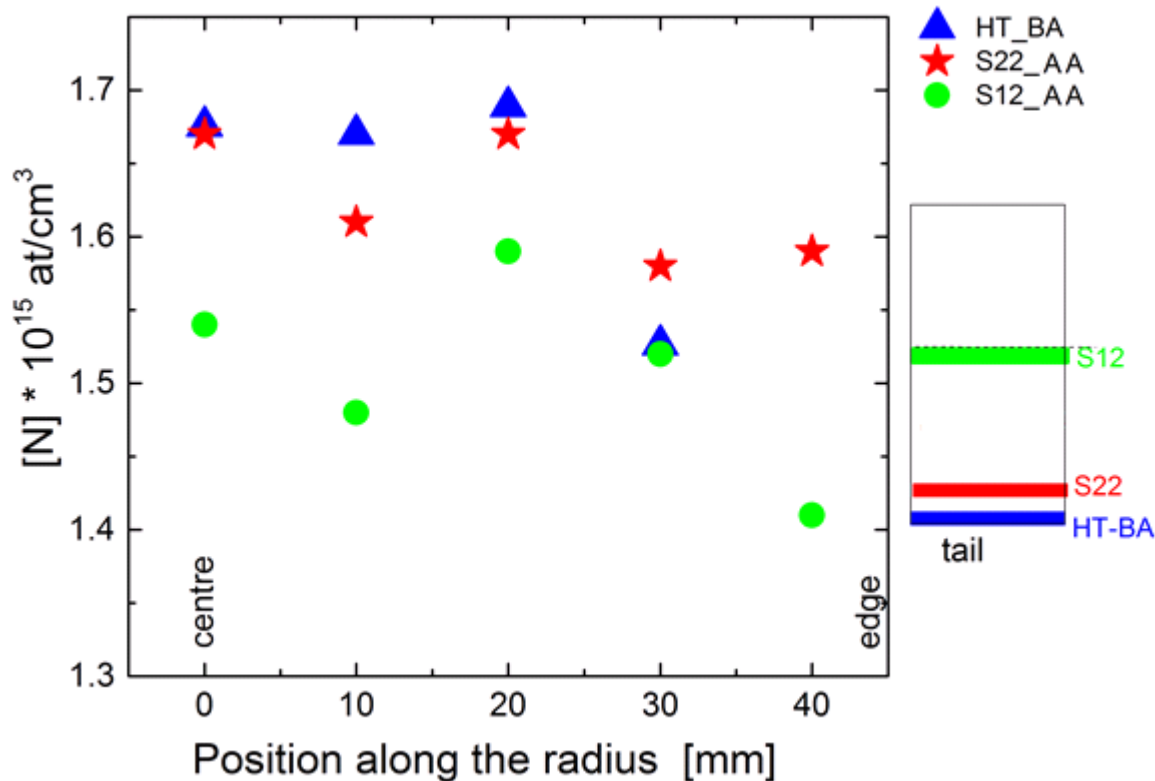
↑ N doping



↑ P doping

HEAT TREATMENT

Lower part of the rod, after annealing removing radiation damage



Fourier Spectrofotometer Bruker Vertex 80v



SIMS Nitrogen Concentration



Evans Lab. (USA)

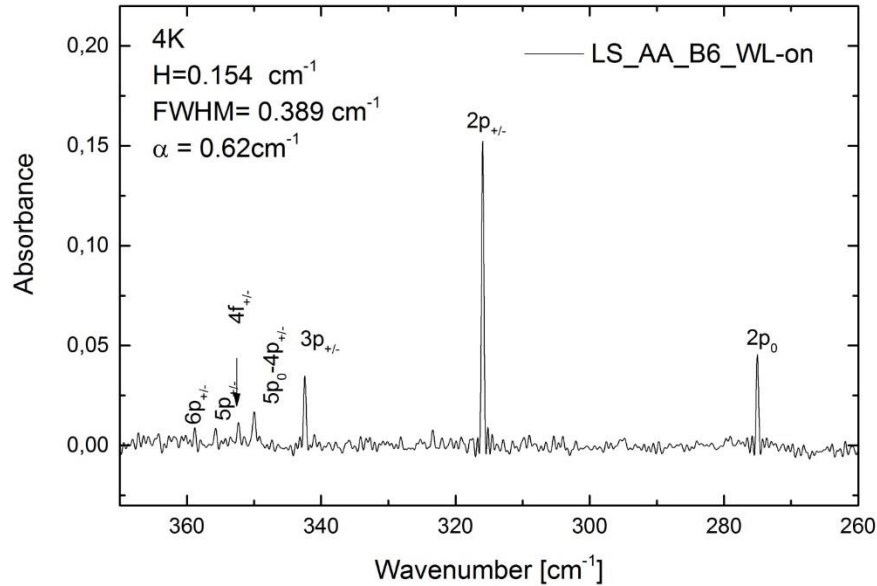
Job# C0FRF644

Average bulk nitrogen concentration

in FZ Si:N wafer: $1.4E15 \text{ cm}^{-3}$

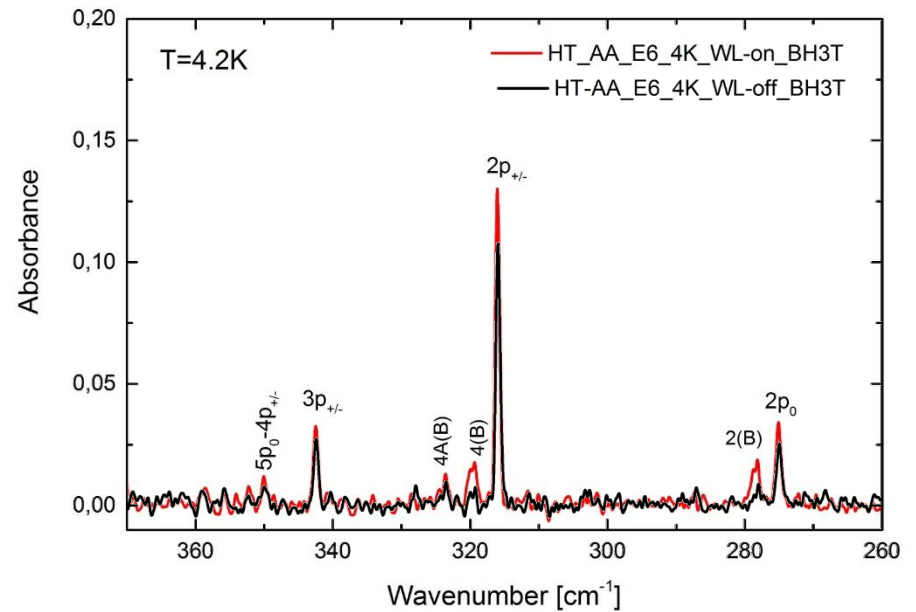
FTIR Phosphorus Concentration

Standard FZ Si



[P] $2.4\text{E}12 \text{ cm}^{-3}$

FZ Si:N

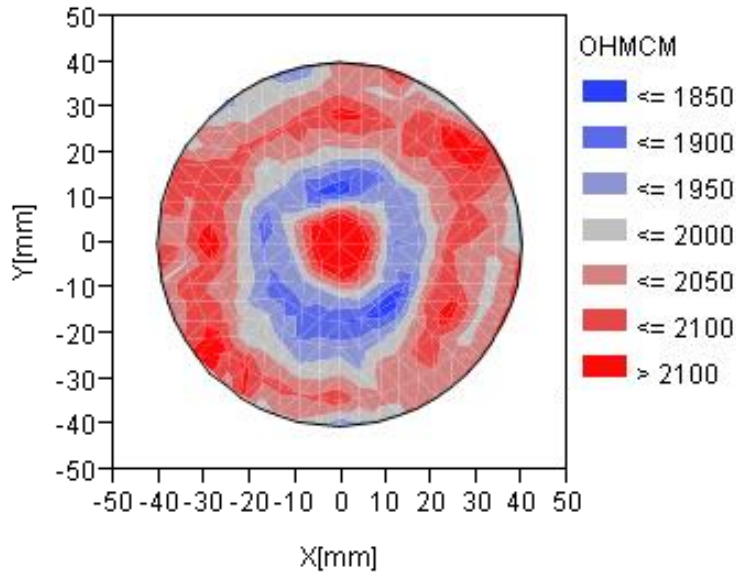


[P] $3.94\text{E}12 \text{ cm}^{-3}$

Resistivity control

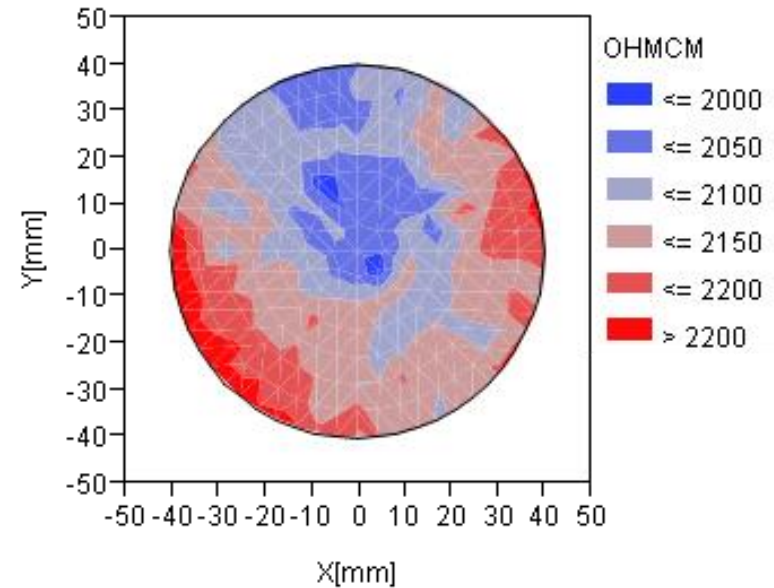
Resistivity distribution on *n*-type high-purity FZ Si wafers (T=300 K)

Standard FZ Si



Nitrogen-lean wafer
[N] $< 1E14 \text{ cm}^{-3}$

FZ Si:N

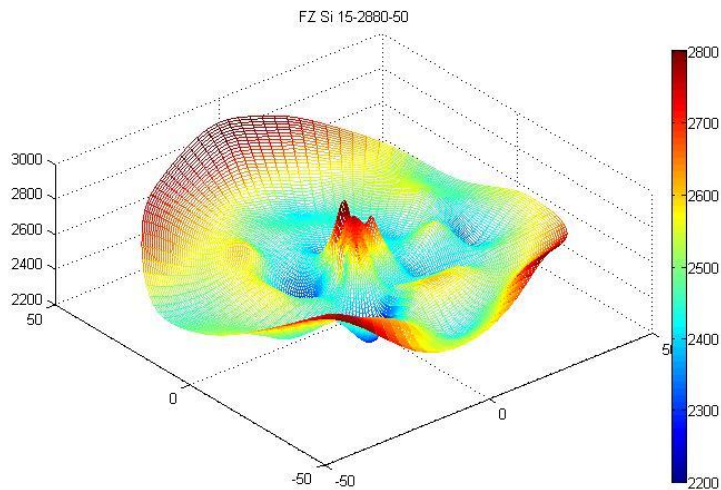


Nitrogen-enriched wafer
[N] $\approx 1.5E15 \text{ cm}^{-3}$

Resistivity control

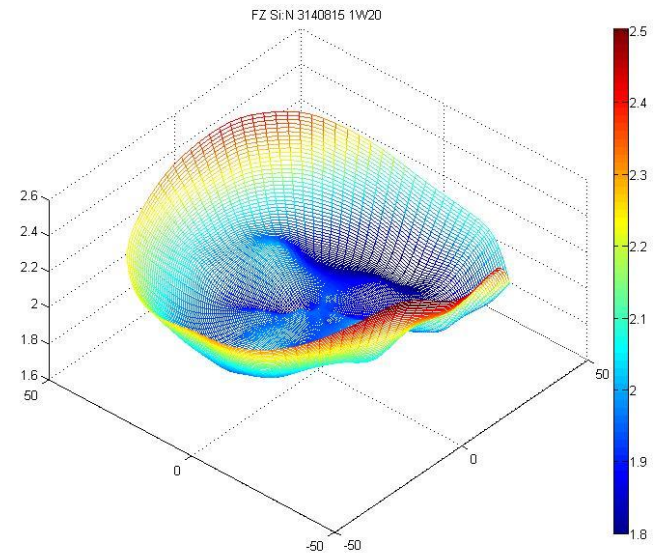
Resistivity distribution on *n*-type high-purity FZ Si wafers (T=300 K)

Standard FZ Si



Nitrogen-lean wafer
[N] < 1E14 cm⁻³

FZ Si:N



Nitrogen-enriched wafer
[N] ≈ 1.5E15 cm⁻³

Lifetime measurements

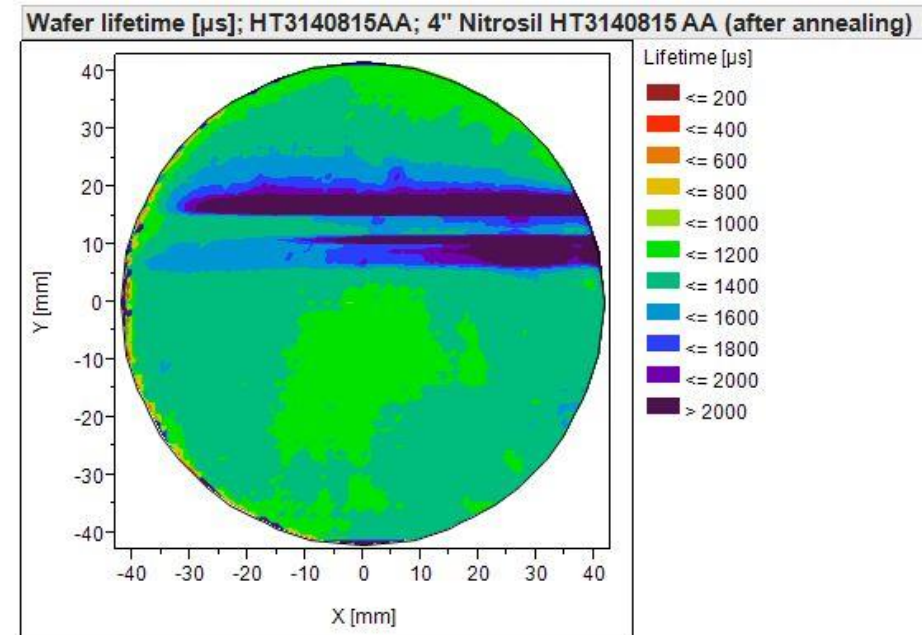
Lifetimes distribution after NTD process and annealing removing radiation damage

Standard FZ Si



$\tau > 1000 \mu\text{s}$

FZ Si:N



$\tau > 1000 \mu\text{s}$

Availability of N-enriched wafers

Availability of N-enriched wafers

- **4" wafers**
 - **2016 – wafers for testing (FOC)**
 - **Q2 2017 – commercially available**

- **6" wafers**
 - **Early stage development product available**
(type n, orientation 100resistivity >5kOhm, nitrogen conc. >1E15 cm⁻³)

Summary

Summary

- The NitroSil project milestone that is fabrication of the first high resistivity, 4", n-type, <100>, nitrogen enriched FZ Si wafers has been achieved.
- Both the nitrogen free and the nitrogen reach wafers are characterized by a uniform resistivity distribution and a high bulk purity: $[O] < 1E16 \text{ cm}^{-3}$, $[C] < 5E15 \text{ cm}^{-3}$.

Acknowledgements

This work has been partially supported by the Polish Centre for Research and Development within the framework of the NitroSil project (ID: 208346) financed by the Program for Applied Research (Contract No. PBS2 / A9 / 26/2014).

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Thank you for attention

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Nitrogen impact on microdefects

- V. Voronkov *et al.* Solid State Phenomena Vols. 131-133 (2008) pp 219-224
- Doping of silicon crystals with nitrogen has an effect on the properties of grown-in microdefects
- In Si crystals vacancies are agglomerated into voids (the voids are revealed by chemical etching as flow pattern effects)
- The void nucleation in FZ crystals occurs at temperature around 1000 °C
- The voids density is at $10^5 - 10^7 \text{ cm}^{-3}$
- The void may include $10^3 - 10^5$ vacancies

Nitrogen atoms in Si lattice

- W. von Ammon *et al.* Journal of Crystal Growth 226 (2001) 19–30
- N-N pairs are stable up to 1270 °C
- Interaction of N-N pairs rather than single nitrogen atoms with vacancies leads to suppression of vacancy aggregation

The following reactions are proposed to be responsible for vacancy annihilation:

- (1) $I+V \rightleftharpoons 0$
- (2) $2N_i \rightleftharpoons N_2$
- (3) $N_s+N_i \rightleftharpoons N_2V$
- (4) $N_2+V \rightleftharpoons N_2V$
- (5) $N_2V+I \rightleftharpoons N_2$

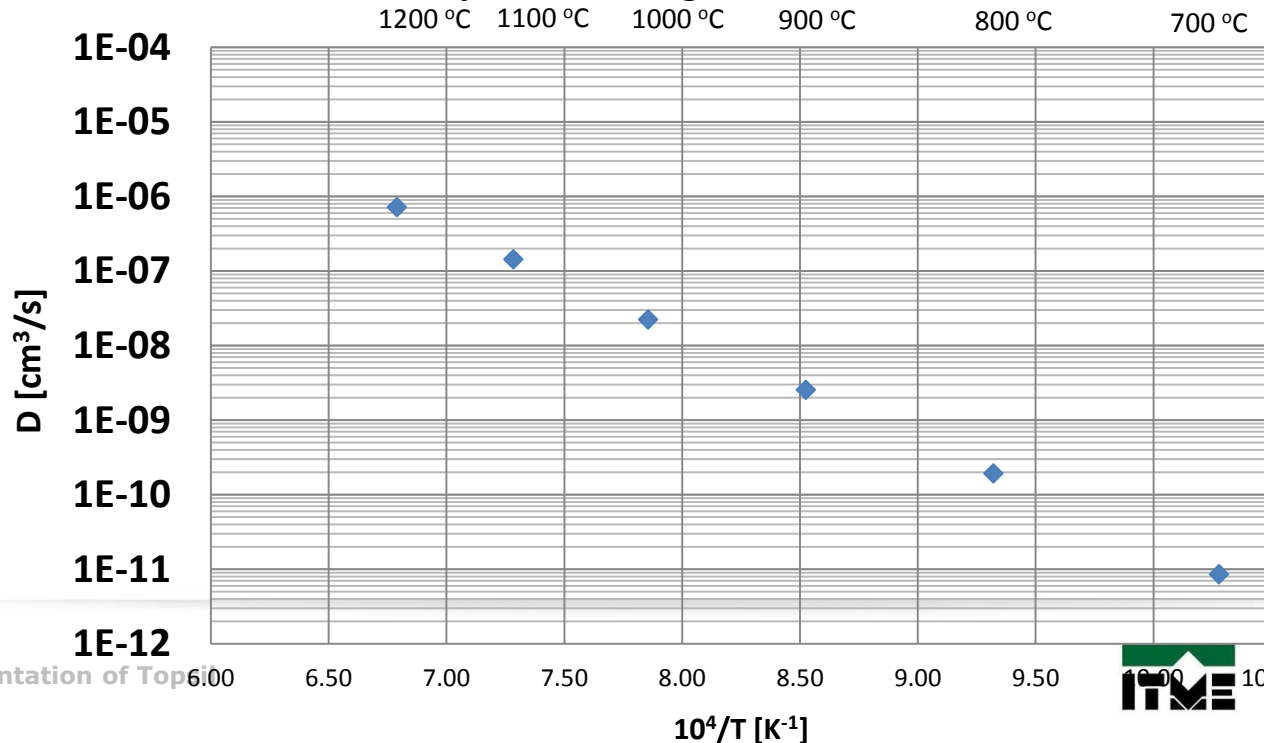
Nitrogen atoms in Si lattice

- T. Itoh and T. Abe Appl. Phys. Lett **53** (1988) p39-41

Diffusion coefficient of nitrogen N-N pairs is given by

$$D = 2.7 \times 10^3 \exp(-2.8 \text{ eV}/kT) \text{ cm}^2/\text{s}$$

**Diffusion coefficient of N-N pairs
in temperature range 700 - 1200 °C**



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