



AGH UNIVERSITY OF SCIENCE
AND TECHNOLOGY

Modelling of Energy Transfer Drops in Ne-CO₂ mixtures

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Systematic gas gain measurements and Penning energy transfer rates in Ne-CO₂ mixtures

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ABSTRACT: In Ne-CO₂ mixtures, excitation energy of Ne atom can be used to ionize CO₂ molecule by the mechanisms called Penning transfers. In the present work, we have measured the gas gain systematically in various Ne-CO₂ mixtures (Ne + 0.6–60 % CO₂) at 0.4, 0.8, 1.2, 1.8 atm. The experimental data have been fitted to investigate the Penning energy transfer rates and the secondary processes playing a role in avalanche formations.

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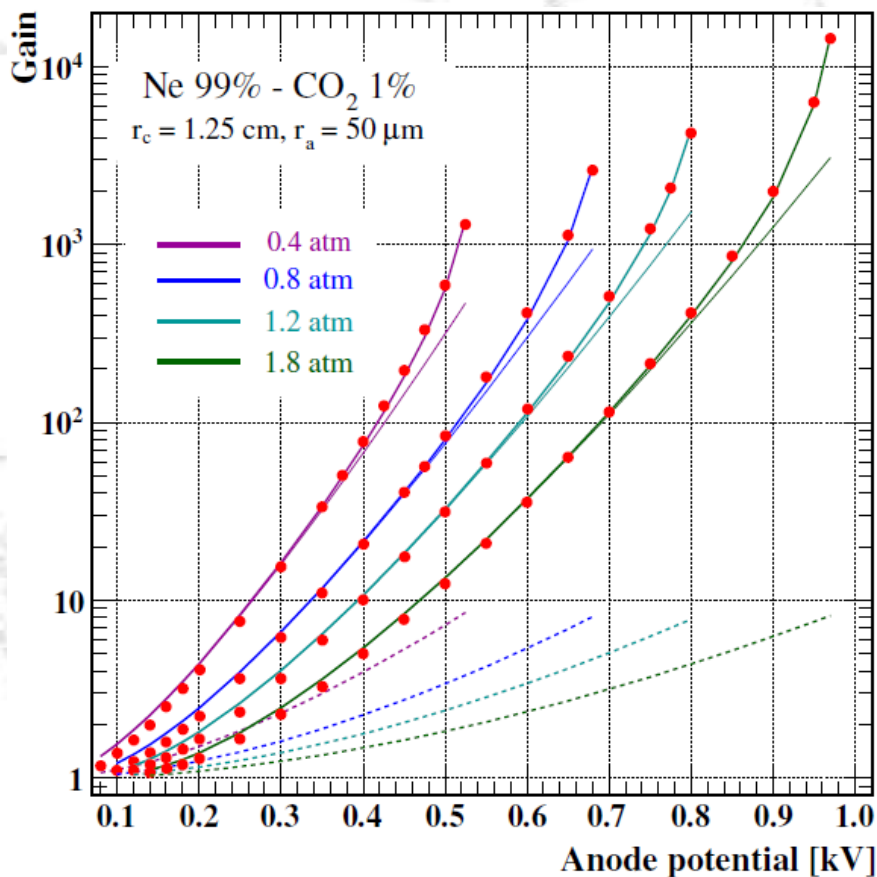
CO₂ percentage

- | | |
|---------|---------|
| 1) 0.6% | 8) 15% |
| 2) 1% | 9) 20% |
| 3) 2% | 10) 30% |
| 4) 4% | 11) 40% |
| 5) 5% | 12) 50% |
| 6) 7% | 13) 60% |
| 7) 10% | 14) 74% |

Pure Ne

Pure CO₂

Ne – CO₂ gas gain calculations



Penning correction

- ❖ $\text{Ne}^* + \text{CO}_2 \rightarrow \text{Ne} + \text{CO}_2^+ + e^-$
- ❖ All of the excited Ne atoms can ionise CO₂

$$\alpha_{\text{Penning}} = \alpha \frac{\sum v_i^{\text{ion}} + \sum r_i v_i^{\text{exc}}}{\sum v_i^{\text{ion}}}$$

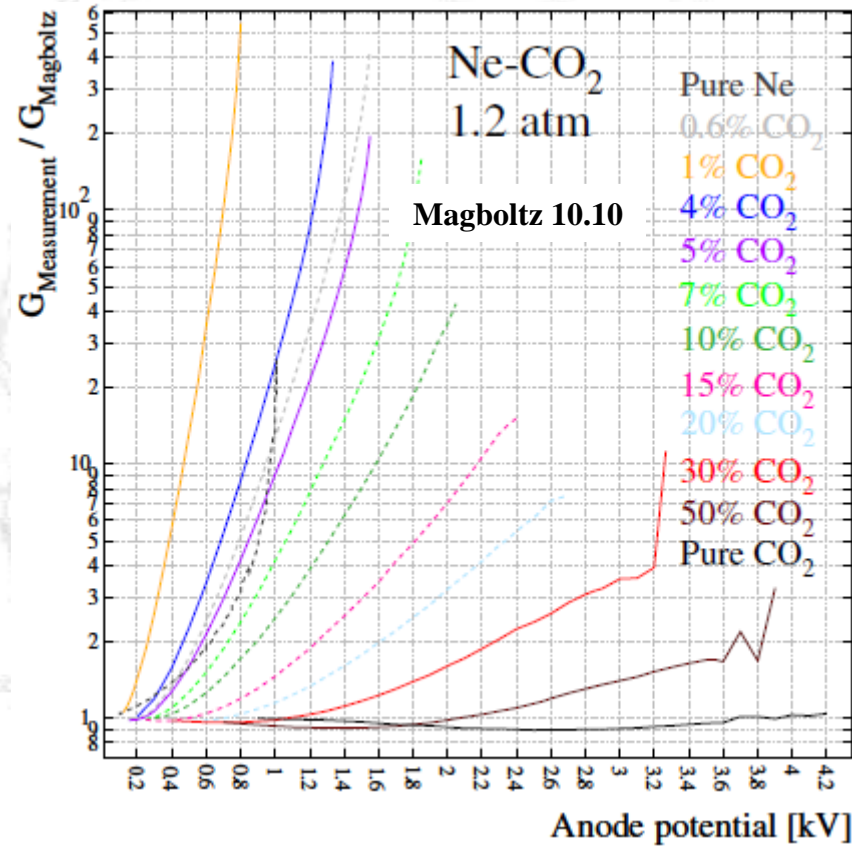
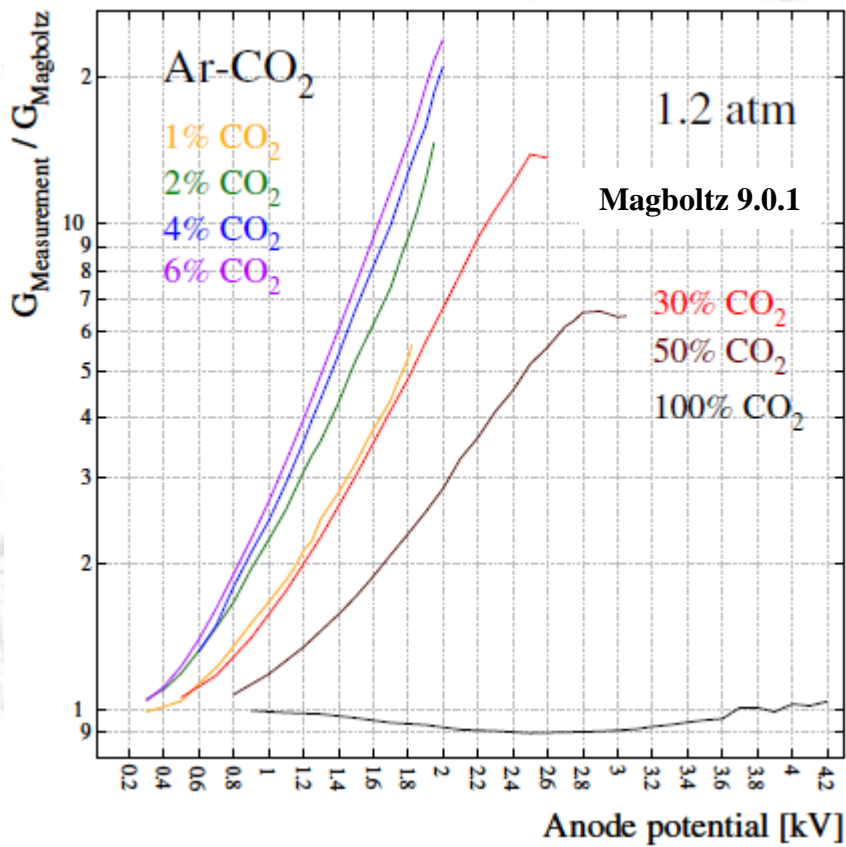
Photon feedback

$$G' = G / (1 - \beta G)$$

Production frequencies of the ionisations and excitations with **Magboltz 10.10**

- ❖ Dashed lines: **without** corrections (Penning, feedback),
- ❖ Thin lines: with Penning, **without** feedback corrections,
- ❖ Thick lines: final fits **with** Penning and feedback corrections.

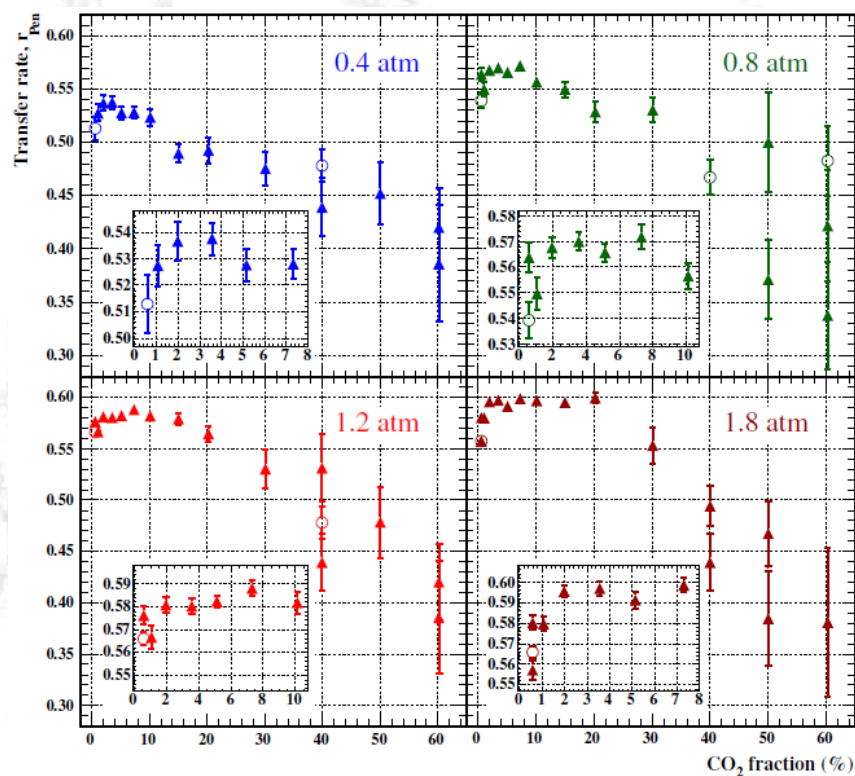
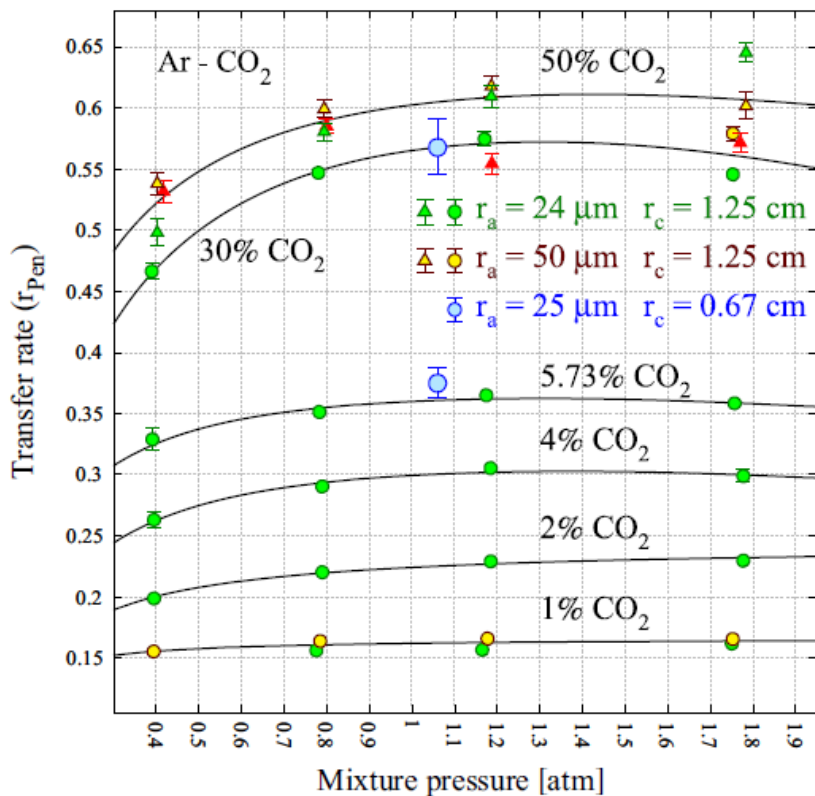
Gain ratio: measured vs calculated without Penning



- ❖ Ratio increases with CO₂ fraction up to 6 %
- ❖ Decrease for 30% and upper CO₂ fractions
- ❖ Approaches unity in pure CO₂
- ❖ No gain data yet in pure Ar

- ❖ Ratio increases from pure neon to 1% CO₂ and becomes smaller at high CO₂ fractions
- ❖ So, almost the same trend for the ratio as seen in Ar – CO₂ mixtures !
- ❖ Better agreement in pure CO₂

Penning rates derived from the gain fits



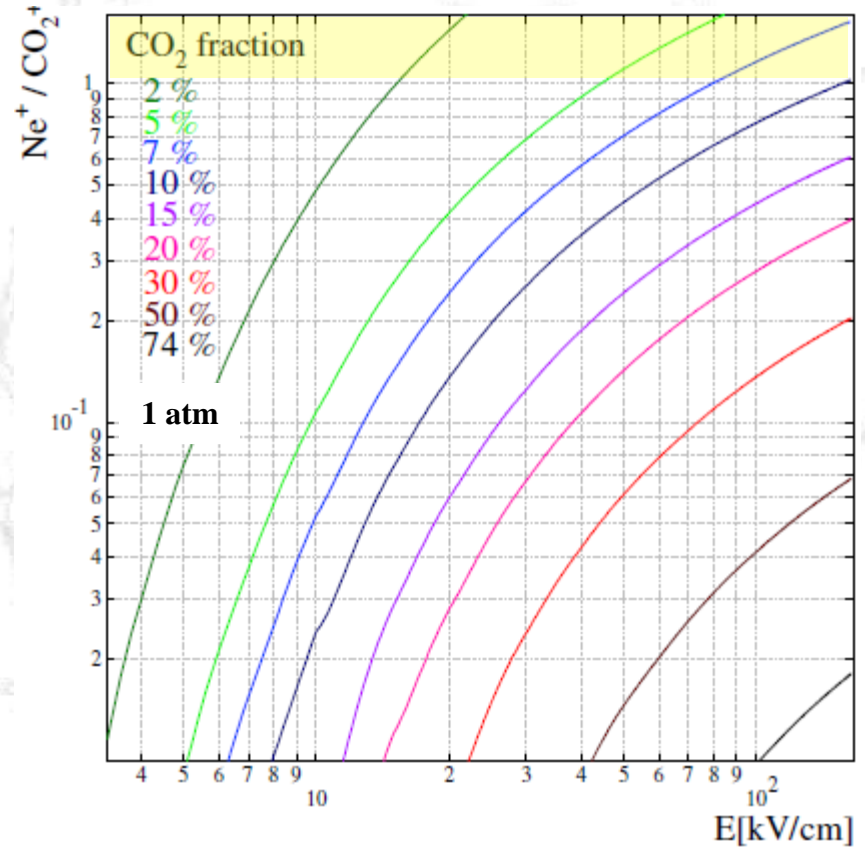
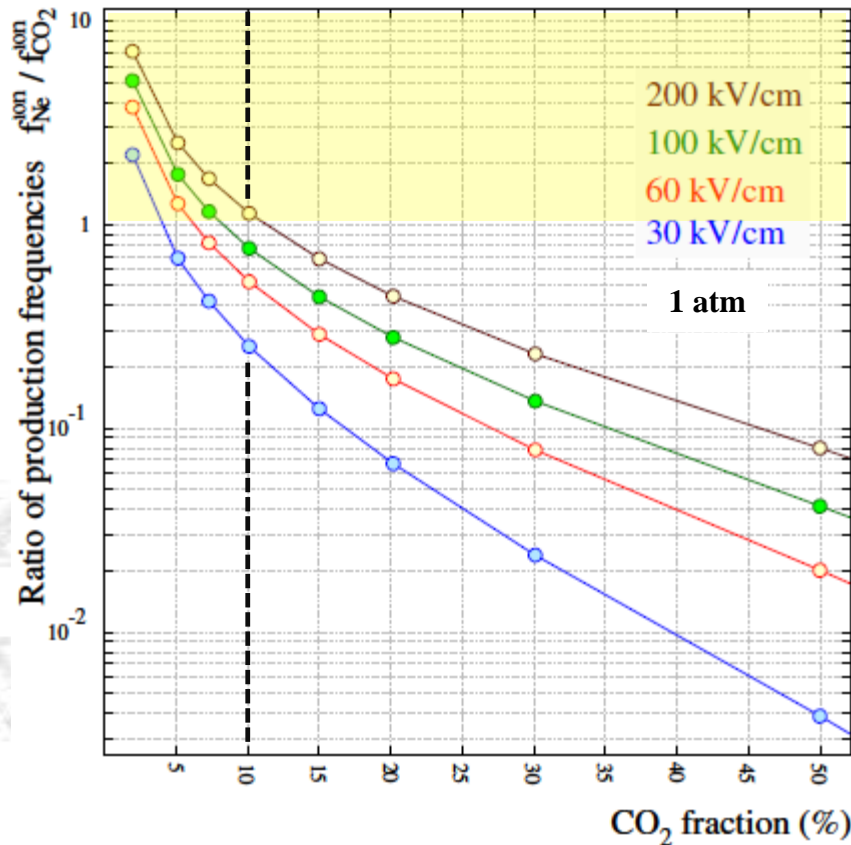
- ❖ Bigger rates at higher CO₂ concentrations and mixture pressures
- ❖ shorter collision time of excited Ar atoms
- ❖ Ar 2p and higher levels are included

Plot: Ö. Şahin, T.Z. Kowalski and R. Veenhof, *High-precision gas gain and energy transfer measurements in Ar-CO₂ mixtures*, [Nucl. Instrum. Meth. A 768 \(2014\) 104](#).

- ❖ The transfer rate increases with pressure
- ❖ shorter collision time
- ❖ **BUT, drops at high CO₂ fractions !!!**
- ❖ Although the gain ratio descends with CO₂, the rate is expected to be high since excited Ne atoms will find more CO₂ molecules to transfer

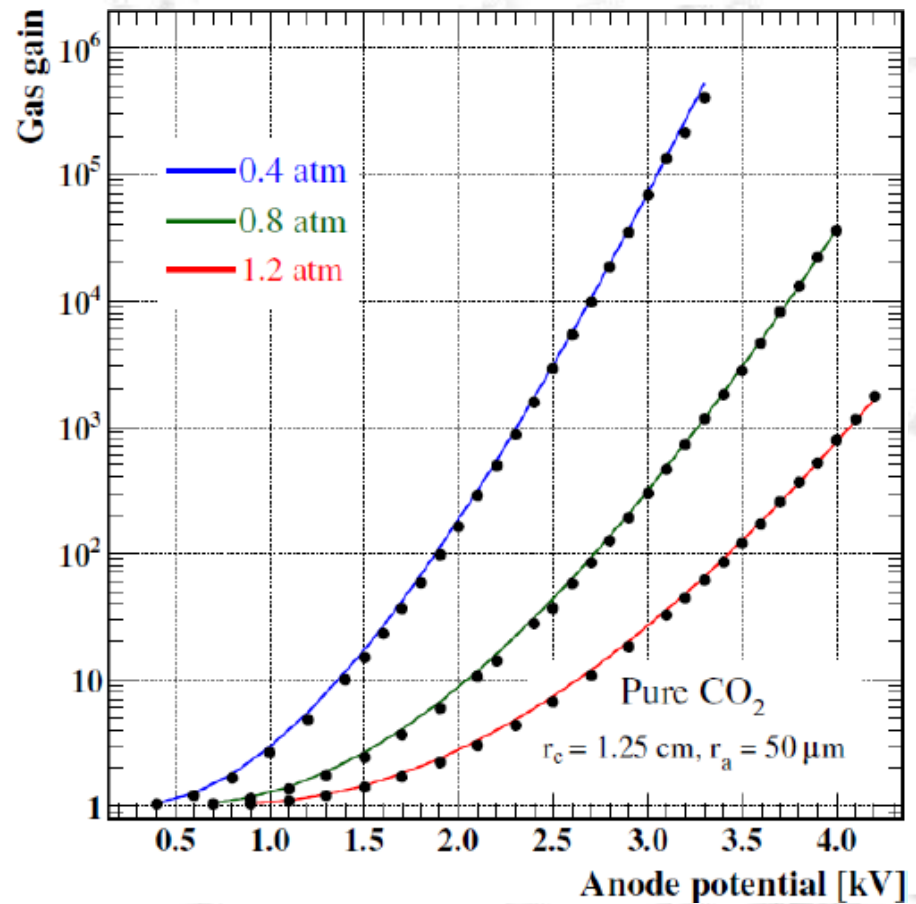
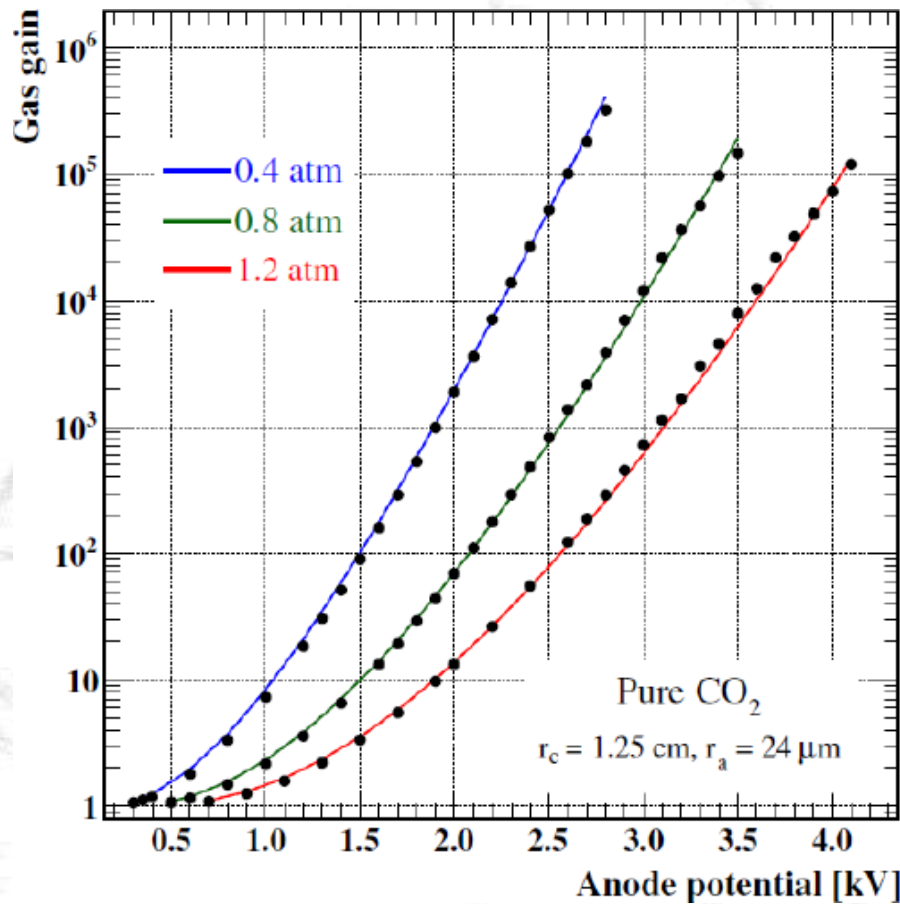
Plot: Ö. Şahin, T.Z. Kowalski and R. Veenhof, *Systematic gas gain measurements and Penning energy transfer rates in Ne - CO₂ mixtures*, [2016 JINST 11 P01003](#).

Production ratios of the direct ionisations



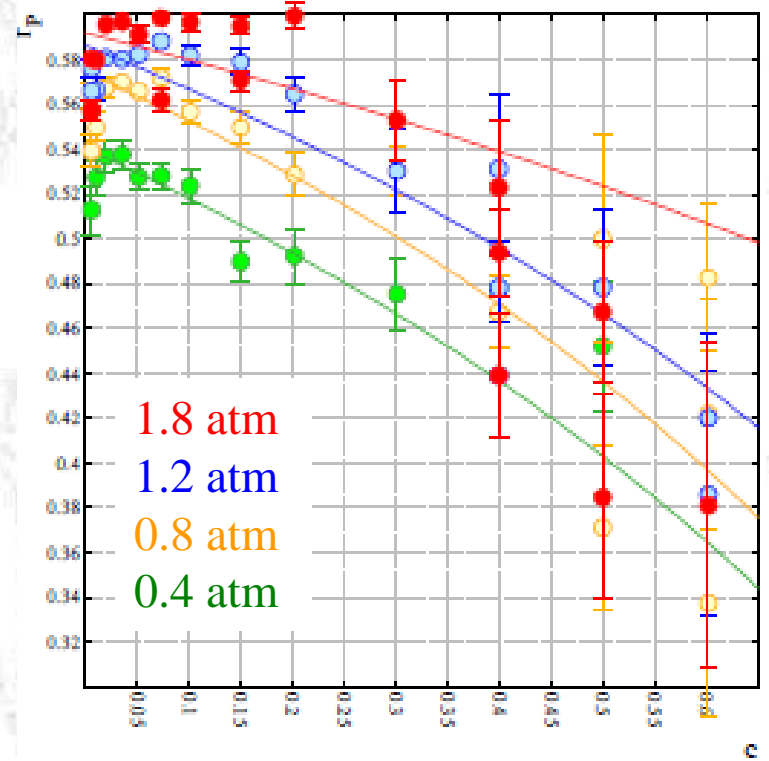
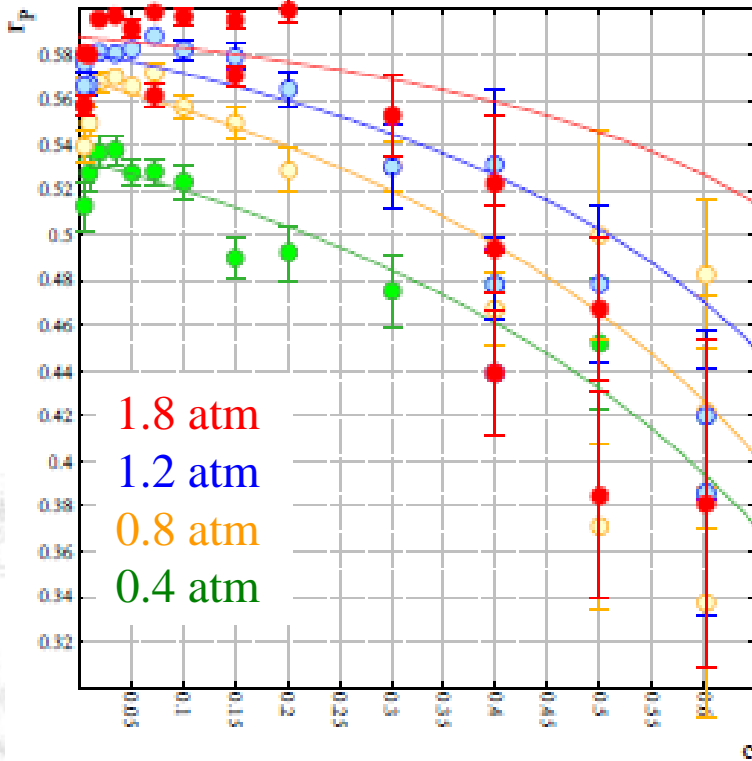
- ❖ Direct ionisation of neon is dominant in the mixtures filled with higher than 90 % Ne,
 - ❖ particularly at high electric fields (ionisation potentials: CO_2^+ 13.78 eV, Ne^+ 21.56 eV)
- ❖ Beyond 10% CO_2 , the largest part of the gain comes from CO_2 ionisations,
- ❖ In 50% and 74% CO_2 mixtures the contribution of Ne^+ downs much lower than 10 %

Are CO₂ ionisation cross sections wrong in Magboltz ???



- ❖ Magboltz calculates the measured gain in pure CO₂ accurately without using any correction factor; direct ionisation cross sections of CO₂ in Magboltz are correct.
- ❖ There should really be other physical processes leading the transfer rate drops

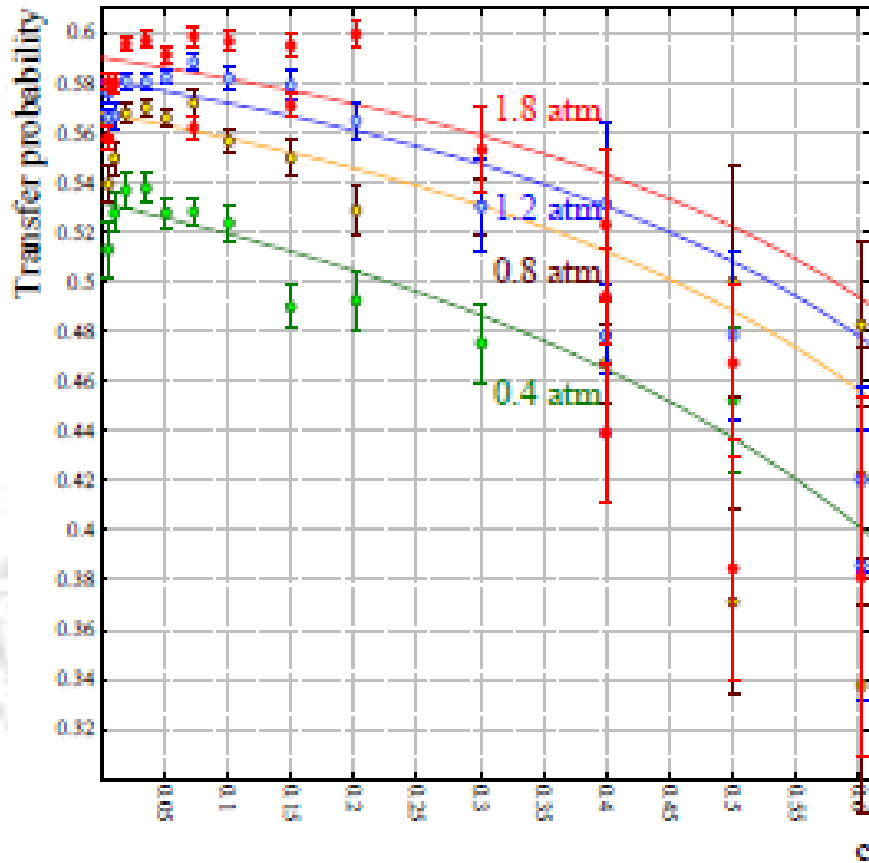
Separate fits for the pressures



- ❖ Hornbeck-Molnar ionisation is assumed to be dominant process ($\text{Ne}^* + \text{Ne} \rightarrow \text{Ne}_2^+ + e^-$)
- ❖ Collisional ionisation of CO_2 takes 0 value
 - ❖ $\text{Ne}^* + \text{CO}_2 \rightarrow \text{Ne} + \text{CO}_2^+ + e^-$
- ❖ Collisional losses of Ne^* to Ne or CO_2 are included
- ❖ Decay is not considered

- ❖ Collisional ionisation of CO_2 is **not** included
- ❖ Collisional losses of Ne^* are **not** included
- ❖ For the loss only decay is included
- ❖ So, Hornbeck-Molnar ionisation (included) is one of the process that can lead to decrease of the transfer rate with increasing CO_2 fraction

Combined fits for the pressures



- ❖ Separate fits do not allow to distinguish between Hornbeck-Molnar and decay processes

- ❖ Combined fit over the 4 pressures is needed:

- ❖ no visible collisional transfer to CO_2
- ❖ there are losses due to Ne and CO_2
- ❖ Hornberck-Molnar is dominant mechanism
- ❖ a decay term is necessary to fit the rates
- ❖ Physically meaningful fit parameters

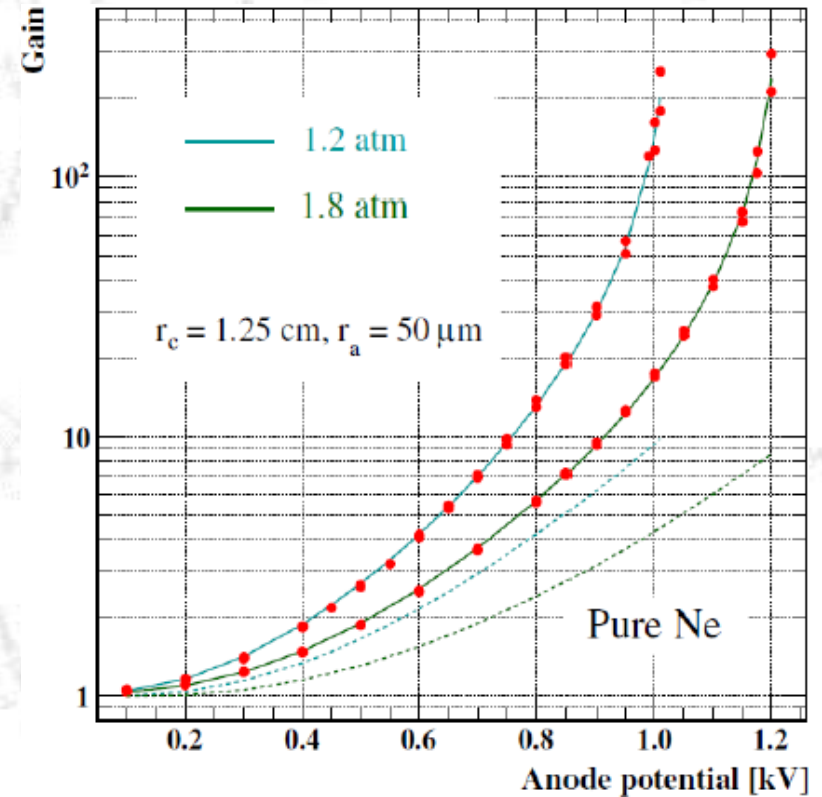
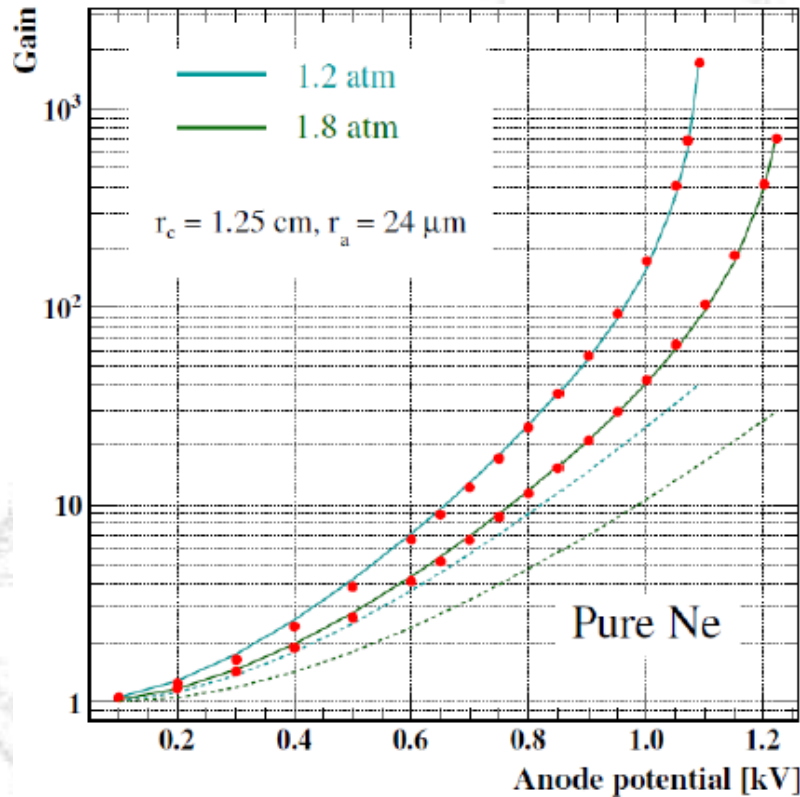
- ❖ There are some articles in literature that supports to above fitting results:

- ❖ Hornbeck-Molnar process could be responsible for the high gain in Ne

- ❖ Experimental data: excited (metastable) neon efficiently is quenched by CO_2

- ❖ Continue to search mechanisms that can be important to explain collisional loss reactions for excited Ne; e.g. ion clustering may lead to drop of the transfer rates

Gain calculations in pure Ne



- ❖ Considering the neon excitation levels above **20 eV** fits the experimental data with the transfer rates varying between $r_p = 40 - 65 \%$
- ❖ Hornbeck-Molnar ionisation threshold is $20.9 \pm 0.2 \text{ eV}$
 - ❖ If the levels above this threshold are used then we get unphysical large values, r_p
- ❖ Contamination of other gases (like O_2 , N_2 etc.) even in purified neon can play an important role and may lead to Penning ionisations

Summary

- ❖ Increase of the rates below 3% CO₂ indicates the typical Penning transfer to CO₂ molecules (excited neon will find more CO₂ to ionise)
- ❖ Hornbeck-Molnar ionisation seem to be dominant mechanism that explains the transfer drops at high CO₂ fractions ($\text{Ne}^* + \text{Ne} \rightarrow \text{Ne}_2^+ + \text{e}^-$)
 - ❖ But why excited neon does not prefer to collisional ionisations at large CO₂ fractions ($\text{Ne}^* + \text{CO}_2 \rightarrow \text{Ne} + \text{CO}_2^+ + \text{e}^-$)
 - ❖ Losses of the excited metastable neon atoms (quenched by CO₂) may have an effect
- ❖ Many-body losses to CO₂ can also be responsible for the drops
 - ❖ Clustering ions should be worked



Thanks and ????