

# JUAS Vacuum Systems

## Examination – 14 March 2017

For questions 2, 3, 4 and 5, give explicitly and develop step by step the analytical formulas used for the computations.

The total points for the examination is 25, letting you the possibility to choose the exercise with which you feel more comfortable in order to obtain the highest mark.

### 1. Give the correct answer (there might be several good answers) – 6 points

a) The ultimate vacuum pressure in a system is dominated by:

- a. The amount of molecules initially present in the volume
- b. The cleanliness of the vacuum chamber walls
- c. The temperature of the vacuum vessel

b) For a gas density of  $10^9$  molecules/cm<sup>3</sup> at room temperature, 300 K, the pressure equals:

- a.  $4 \cdot 10^{-12}$  Pa
- b. 760 Torr
- c.  $3 \cdot 10^{-8}$  Torr

c) For a gas density of  $10^{15}$  molecules/m<sup>3</sup> at room temperature, 3 K, the pressure equals:

- a.  $4 \cdot 10^{-14}$  Pa
- b. 0.76 Torr
- c.  $3 \cdot 10^{-10}$  Torr

d) The pressure is:

- a. due to the stress before the exam
- b. proportional to the gas density
- c. defined as the force per unit of area

e) The typical energy of chemisorption is:

- a. in the range of 1 eV
- b. in the range of 0.1 eV
- c. in the range of 100 kJ/mole

f) A residual gas analysis of an unbaked vacuum system will show a spectrum dominated by:

- a. water
- b. hydrogen
- c. helium

## 2. Hot filament ionisation gauge – 2 points

Describe the operating principle of a Bayard-Alpert gauge. Which effect are limiting the reading at low pressure?

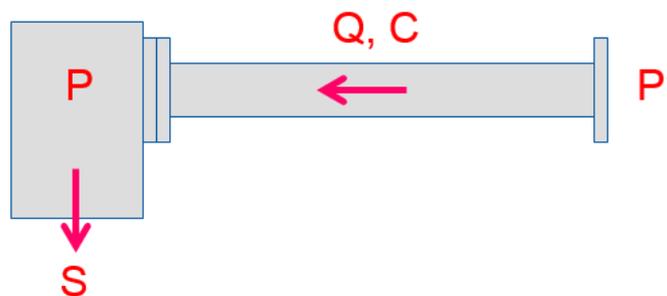
## 3. Desorption of a monolayer of gas – 2 points

Assume a 1 m long vacuum chamber of 10 cm inner diameter onto which a monolayer of gas is adsorbed ( $10^{15}$  molecule/cm<sup>2</sup>). The vacuum system is closed without external pumping, what would be the final pressure if the monolayer of gas is desorbed into the volume held at 300 K?

## 4. A vacuum chamber connected to a turbo-molecular pump – 5 points

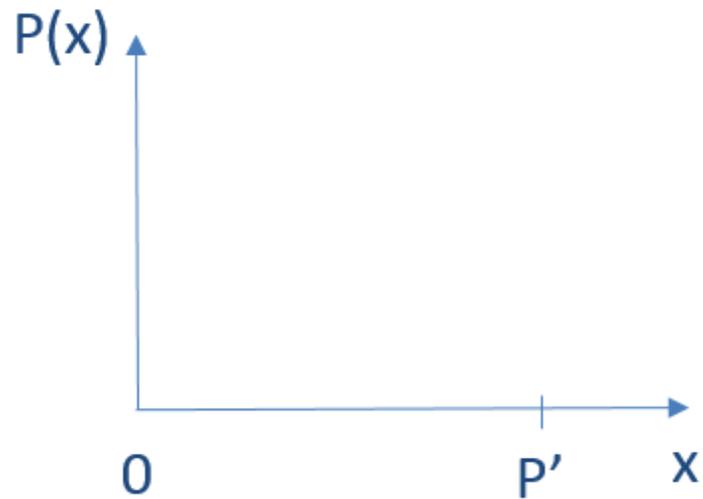
a) Consider a Cu baked vacuum chamber of 5 cm inner diameter and length 4 m. What is the main gas? What is its total outgassing rate?

b) As shown in the sketch below, this vacuum chamber is connected to a Pfeiffer turbomolecular pump HiPace 80 DN63 (see data sheet). What is its pumping speed,  $S$ ? What is the pressure,  $P$ , reached at the level of the pump?



d) Write down the effective pumping speed seen from the pressure gauge located in  $P'$ , what is its value?

e) Draw the pressure profile along the tube, what is the shape of the profile? The pump is placed at  $x=0$ .

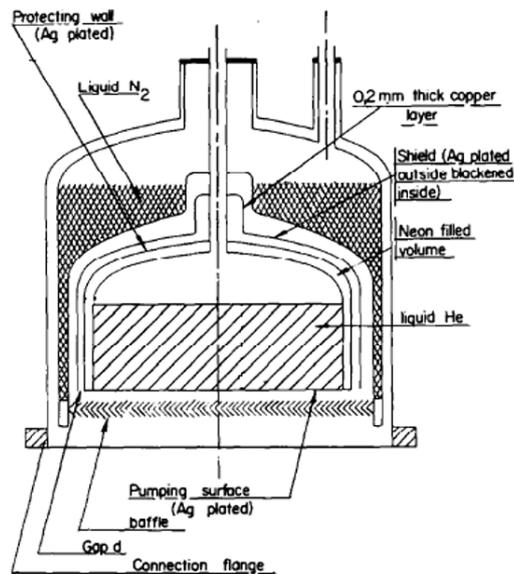


f) Your boss would like to reduce further the pressure in the vacuum system, to do so, he proposes to replace the HP80 turbomolecular pump which cost 1.5 kCHF by a large one, HP300 which has a pumping speed for hydrogen of 220 l/s for a price of 2.5 kCHF. Would you follow his recommendation? Why?

Bearing	Hybrid
Compression ratio for Ar	$> 1 \cdot 10^{11}$
Compression ratio for H <sub>2</sub>	$1.4 \cdot 10^8$
Compression ratio for He	$1.3 \cdot 10^8$
Compression ratio for N <sub>2</sub>	$> 1 \cdot 10^{11}$
Cooling method, optional	Air/Water
Cooling method, standard	Convection
Cooling water consumption	75 l/h
Cooling water temperature	5-25 °C
Electronic drive unit	with TC 110
Flange (in)	DN 63 CF-F
Flange (out)	DN 16 ISO-KF / G 1/4"
Fore-vacuum max. for N <sub>2</sub>	22 hPa   16.5 Torr   22 mbar
Gas throughput at full rotational speed for Ar	0.54 hPa l/s   0.41 Torr l/s   0.54 mbar l/s
Gas throughput at full rotational speed for H <sub>2</sub>	15.3 hPa l/s   11.47 Torr l/s   15.3 mbar l/s
Gas throughput at full rotational speed for He	2.7 hPa l/s   2.02 Torr l/s   2.7 mbar l/s
Gas throughput at full rotational speed for N <sub>2</sub>	1.3 hPa l/s   0.97 Torr l/s   1.3 mbar l/s
Interfaces	RS-485, Remote
Mounting orientation	in any orientation
Operating voltage	24 (± 5 %) V DC
Permissible magnetic field max.	3.3 mT
Protection category	IP54
Pumping speed for Ar	66 l/s
Pumping speed for H <sub>2</sub>	48 l/s
Pumping speed for He	58 l/s
Pumping speed for N <sub>2</sub>	67 l/s
Rotation speed ± 2 %	90,000 rpm   90,000 min <sup>-1</sup>
Rotation speed variable	50 - 100 %
Run-up time	1.7 min
Ultimate pressure according to PNEUROP	$< 5 \cdot 10^{-10}$ hPa   $< 3.75 \cdot 10^{-10}$ Torr   $< 5 \cdot 10^{-10}$ mbar
Venting connection	G 1/8"
Weight	3.8 kg   8.38 lb

## 5. A condensation cryopump – 5 points

A XHV condensation cryopump which operates at 2.3 K is shown in the picture below:



C. Benvenuti *et al.* Vacuum, 29, 11-12, (1974) 591

- The cold surface is a disk of diameter 62 cm. What is the ideal pumping speed (*i.e.* for a sticking coefficient = 1) of H<sub>2</sub> at 300 K?
- A baffle, of chevron type, with a molecular transmission  $\alpha$  is placed in front of the pumping surface. The baffle conductance for H<sub>2</sub> at 300 K equals 36240 l/s. What is the molecular transmission (*i.e.* the ratio of the conductance to the ideal pumping speed)?
- The measured pumping speed of the condensation cryopump is 27 000 l/s, what is the capture factor?
- What is the hydrogen sticking coefficient and the pumping speed of the cold surface?
- A hydrogen gas flux of  $2.7 \cdot 10^{-8}$  mbar.l/s is injected into the condensation cryopump. What is the equilibrium pressure?
- The pump operated at the same constant gas flux during 300 days, what is the total gas load during that period? What is the equilibrium pressure at 2.3 K? What do you conclude?

## 4. Photon stimulated molecular gas desorption in a lumped pumping system – 5 points

In a 3 GeV synchrotron radiation (SR) source, a baked Cu vacuum chamber of 10 cm diameter, placed downstream to bending magnet, is irradiated by synchrotron radiation with 3.75 keV critical energy.

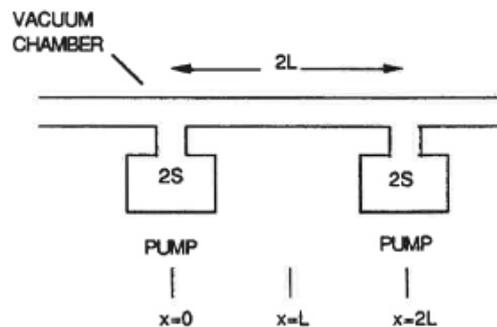
- Write down the formula of critical energy, what does the critical energy means?

b) Write down the formula to compute the linear photon flux for an electron beam (number of photons emitted per meter of trajectory per second). Compute the linear photon flux when a beam current of 100 mA circulates in the ring.

c) Compute the integrated photon dose after 12h of continuous irradiation. What are the typical desorption yield of H<sub>2</sub> at that photon dose (see graph lecture 4, p28)?

d) After 12h of irradiation, what is the gas load due to PSD? Express the PSD gas load in unit of mbar.l/s/m

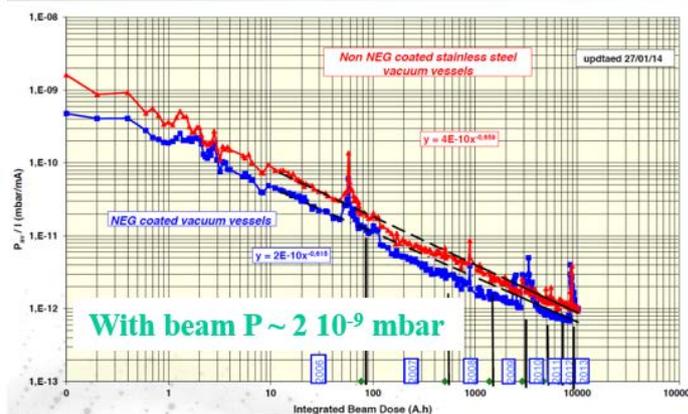
e) The Cu vacuum chamber is part of a “simple machine” where 300 l/s pumps, 2S, are placed every 28 m, 2L, (see lecture 5, p 13). Write down the specific conductance of the tube, c, the gas desorption per unit length, a. Express and compute the average pressure in the simple machine.



f) Assuming the PSD yield scales like  $\eta \approx D^{-1}$ , what is the hydrogen PSD yield after a conditioning period of 12 000h (500 days)? What is the expected average pressure in the “simple machine”? What do you conclude when comparing with p25, lecture 4?

### SOLEIL

Average pressure rise in cell C07 normalized to current Vs. beam dose



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