

LaBr₃(Ce): a new generation detector for timing spectroscopy

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Application of scintillation detectors

- Nuclear energy and timing spectroscopy (TDPAC, Positron annihilation spectroscopy)
- High energy physics experiments
- Dark matter search experiment (to detect the recoil spectrum by WIMS)
- Geological exploration
- Medical imaging (PET)

Characteristics of an ideal scintillator

- High Z and density (for high detection efficiency)
- High light output and linearity with energy (for energy spectroscopy)
- Fast response time (for timing spectroscopy)
- Transparent to emitted light (for minimum loss of light)
- Non hygroscopic
- Low afterglow

Properties of common inorganic scintillator

Scintillators	Light yield (photons/keV)	1/e Decay time (ns)	Wavelength of maximum emission (nm)	Refractive index (gm/cm ³)	Density	Z _{eff}	Hygroscopic
LaBr ₃ (Ce)	63	16	380	1.9	5.08	45.22	yes
BaF ₂ (fast component)	1.8	0.6-0.8	220	1.54	4.88	50.96	slightly
BaF ₂ (slow component)	10	630	310	1.50	4.88	50.96	slightly
NaI(Tl)	38	250	415	1.85	3.67	50.6	yes
LSO	32	41	420	1.81	7.1	65.5	no
CsI(Tl)	54	1000	550	1.79	4.51		slightly

The data presented here are taken from Saint-Gobain scintillation detector operating manual.

Temperature dependence of light output for different inorganic scintillators

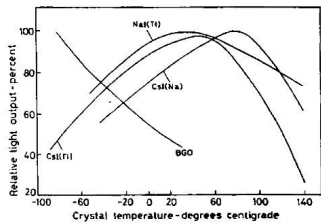


Figure: Response of various inorganic scintillators with temperature

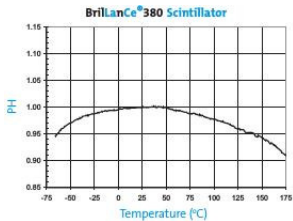


Figure: Response of LaBr₃(Ce) scintillator with temperature

Light emission spectra for different inorganic crystals

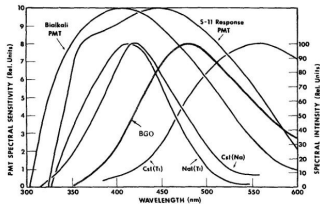


Figure: Emission spectra of several inorganic scintillators

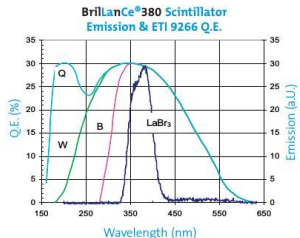


Figure: Scintillation emission spectrum of the BrillLanCe 380 crystal and Quantum Efficiency of a bi-alkali ETI9266 PMT with (B)Borosilicate, (W)UV glass, and (Q)Quartz face plates

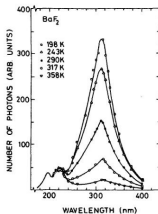
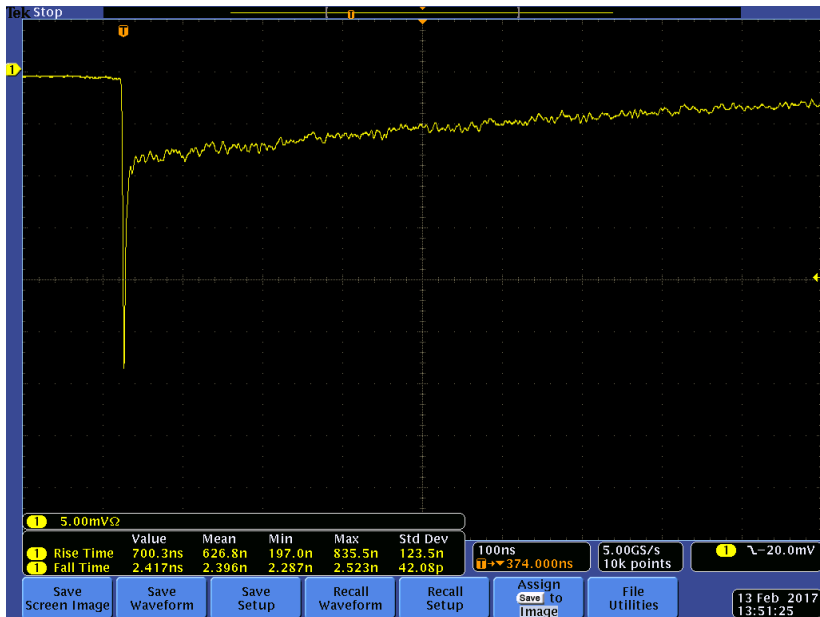
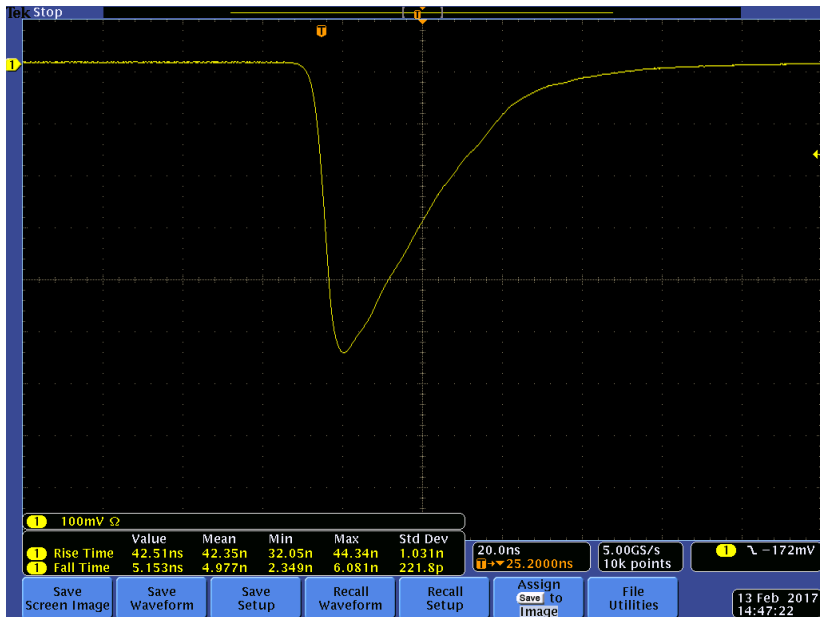


Figure: Scintillation emission spectra from BaF₂ measured at various temperatures

Anode pulse of BaF₂ detector (slow and fast component)



Anode pulse of LaBr₃(Ce) detector



Energy resolutions of $\text{LaBr}_3(\text{Ce})$ and BaF_2 scintillation detectors

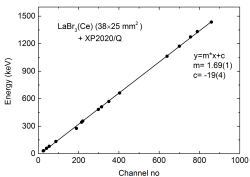


Figure: Energy calibration of $\text{LaBr}_3(\text{Ce})$ scintillation detector

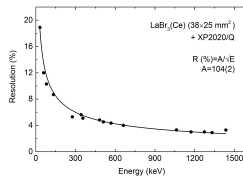


Figure: Energy resolution of $\text{LaBr}_3(\text{Ce})$ with γ -ray energy

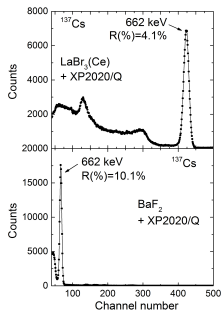


Figure: γ -ray spectra of ^{137}Cs showing relative light output of $\text{LaBr}_3(\text{Ce})$ and BaF_2 . PMT, HV, amplifier settings remain same.

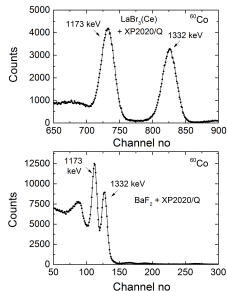


Figure: Difference in energy resolution between $\text{LaBr}_3(\text{Ce})$ and BaF_2 scintillators for ^{60}Co γ -ray source

Prompt time resolutions of LaBr₃(Ce) and BaF₂ scintillators

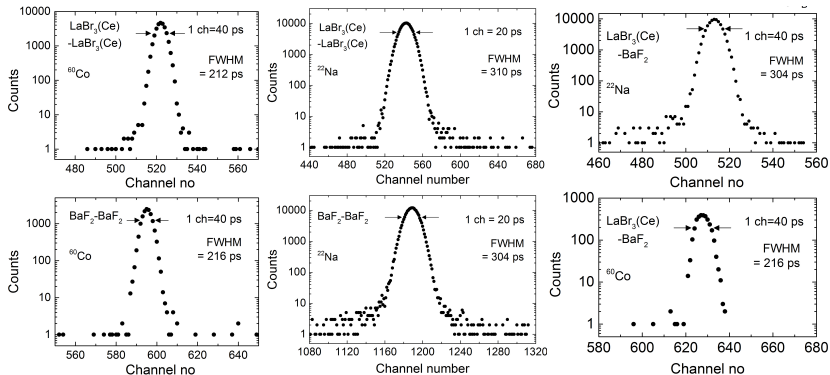


Figure: Prompt time spectra of ⁶⁰Co (left panel) and ²²Na (middle) γ -rays of LaBr₃(Ce)-LaBr₃(Ce) detector combination as well as BaF₂-BaF₂ detector combination. Right panel shows prompt time spectra of ²²Na and ⁶⁰Co γ -rays of LaBr₃(Ce)-BaF₂ detector combination. Crystal sizes : 38 × 25 mm²; PMT : XP2020/Q

- The time differential perturbed angular correlation (TDPAC) technique measures the **effect of perturbations of the γ - γ angular correlation** of the probe nucleus through the **hyperfine interaction**.
- The nuclear moments (**electric quadrupole moment or magnetic dipole moment**) of the intermediate level of probe nucleus interact with the hyperfine fields (**electric field gradients or magnetic field**) present in the investigated sample
- Using this technique, **structural and magnetic properties of crystalline solids** can be studied. Measured electric field gradients by this technique can be compared with the **density functional theory (DFT)** based calculations.

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Principle of PAC technique

Angular correlation of a γ - γ cascade

$$W(\theta) = \sum_{\substack{k=0 \\ \text{even}}}^{k_{\max}} A_k P_k(\cos\theta), \quad \text{Unperturbed}$$

$$W(\theta, t) = \sum_{\substack{k=0 \\ \text{even}}}^{k_{\max}} A_k G_k(t) P_k(\cos\theta), \quad \text{Perturbed}$$

- A_k : Angular correlation coefficient
- $G_k(t)$: Perturbation function

Nuclear quadrupole interaction

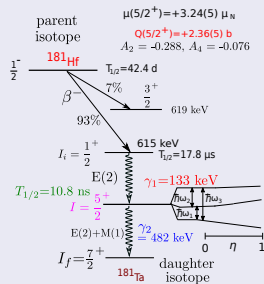


Figure: Decay scheme of the probe ^{181}Hf

Perturbation function

$$G_2(t) = \left[S_{20}(\eta) + \sum_{i=1}^3 S_{2i}(\eta) \cos(\omega_i t) \exp(-\delta \omega_i t) \right]$$

- δ : Frequency distribution width arising from lattice imperfections or chemical inhomogeneities
- ω_j : Transition frequencies between the sublevels of the intermediate state which arise due to hyperfine splitting

Schematic diagram of the four detector PAC spectrometer : $\text{LaBr}_3(\text{Ce})$ - BaF_2 detector setup

D1, D2: $\text{LaBr}_3(\text{Ce})$
D3, D4: BaF_2

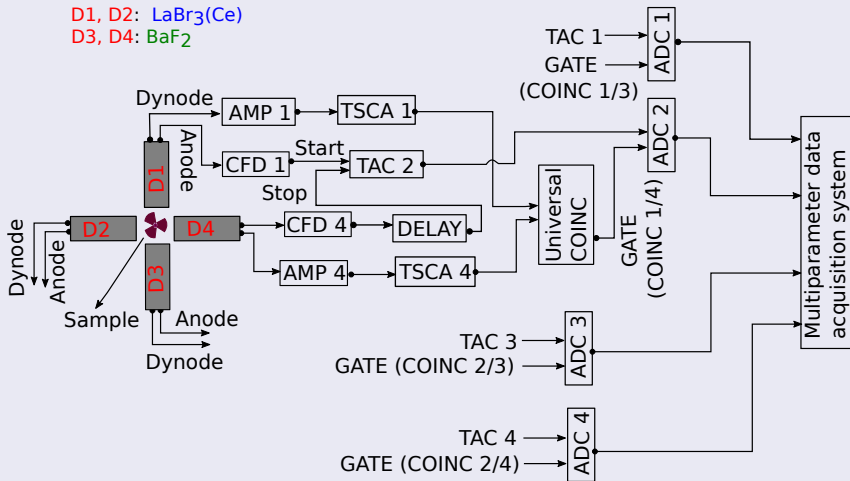
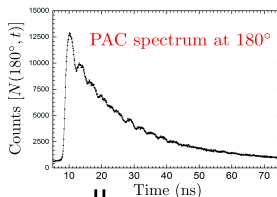
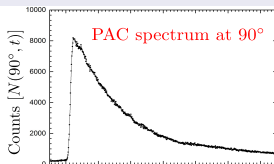
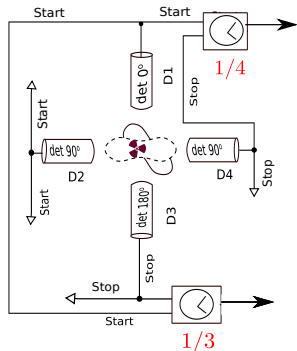


Figure: Slow-fast coincidence set up

Schematic of PAC data reduction

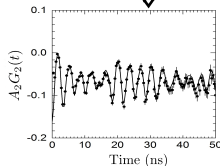


$$W(\theta, t) = 1 + A_2 G_2(t) P_2(\cos \theta)$$

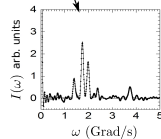
$$N_{ij}(\theta, t) = N_0 \exp(-t/\tau_N) W(\theta, t) + B$$

$$R(t) = \frac{2}{3} \left[\sqrt{\frac{N_{13}(180^\circ, t) N_{24}(180^\circ, t)}{N_{14}(90^\circ, t) N_{23}(90^\circ, t)}} - 1 \right]$$

$$A_2 G_2(t) = \frac{R(t)}{1 + R(t)/2}$$



Fourier cosine transform



PAC spectrum in stoichiometric sample of HfNi_3

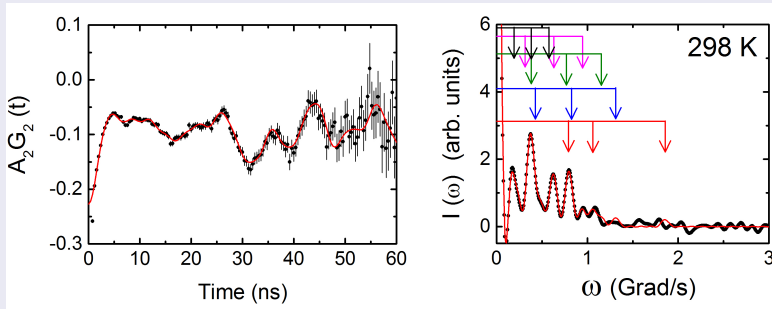


Figure: Figure in the left shows PAC spectrum for the HfNi_3 sample at room temperature and the right one shows the corresponding Fourier cosine transform.

Results of PAC measurements in the Hf-Ni sample

Temperature (K)	Component	ω_Q (Mrad/s)	η	$\delta(\%)$	$f(\%)$	Assignment
298	1	32.0(3)	0	0	32(2)	HfNi ₃
	2	52.6(4)	0	0	23(2)	Hf
	3	94.8(6)	0.67(2)	0	14(2)	Hf ₈ Ni ₂₁
	4	70.6(6)	0.38(3)	0	15(2)	Hf ₂ Ni ₇ ⁽¹⁾
	5	64.3(8)	0	0	16(2)	Hf ₂ Ni ₇ ⁽²⁾

- 1 S. K. Dey, C. C. Dey, S. Saha, J. Belošević-Čavor, *Intermetallics* 84 (2017) 112.
- 2 S. K. Dey, C. C. Dey, S. Saha, *J. Phys. Chem. Solids* 95 (2016) 98.
- 3 P.R.J. Silva, H. Saitovitch, J.T. Cavalcante, M. Forker, *J. Magn. Magn. Mater.* 322 (2010) 1841.

Conclusion

- $\text{LaBr}_3(\text{Ce})$ is found to be best scintillator for energy and timing spectroscopy experiments.
- Due to the superior energy and time resolutions of $\text{LaBr}_3(\text{Ce})$ detector, it has been possible to distinguish weak EFGs corresponding to different phases that present in the stoichiometric sample of HfNi_3 by TDPAC spectroscopy.

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Thank You