The EUROPEAN SCIENTIFIC INSTITUTE

In ARCHAMPS, 7 Km from downtown GENEVA Fifty minutes from Chamonix-Mont-Blanc

organises two schools ESMP : European School of Medical Physics

In partnership with the European Federation of Organisations in Medical Physics (EFOMP)

2012 :15th SESSION of ESMP

Lecture presented in Archamps (Salève Building) by :

Gerd-Jürgen BEYER (CHU-Geneva)



RADIOISOTOPES in **MEDICINE**:

Requirements - Production - Application and Perspectives

History

Gerd-Jürgen BEYER

Prof.Dr.rer.nat.habil. Cyclotron Unit, University Hospital of Geneva, Switzerland GSG-Int. GmbH, Switzerland gerd.beyer@cern.ch gerd.beyer@gsg-int.com



Lecture ESI Week 1 Archamps (France) October 19, 2012 European School of Medical Physics - Archamps



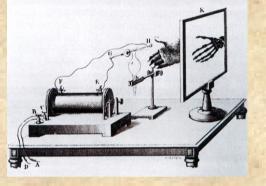
NUCLEAR MEDICINE – HOW IT BEGAN

- 1789 Klaproth
- 1895 Conrad Roentgen
- 1896 Henry Becquerel
- 1898 M.&P.Curie
- 1923 G.Hevesy
- 1932 Lawrence
- 1934 I.&F.Juliot-Curie
- 1938
- Hahn / Strassmann

Uran X-Ray Radioactivity **Po** und **Ra** Tracer Principle Cyclotron Artif.Radiactivity U-Fission

W.C. Roentgen discovers X-rays

Nov.8, 1895



W.C.Roentgens experiment in Würzburg



An early XXth century X-ray tube Radiograph of Mrs.Roentgens hand, the first x-ray image ever taken, 22.Dec.1895, published in The New York Times January 16, 1896



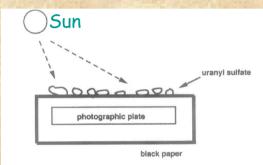
The Académie of Sciences in Paris Monday Meetings in early 1896:

Monday, January 20: Poincaré's Hypothesis:

"... Since Roentgen rays seem to emerge from the fluorescence of the wall of the Crookes tube, other fluorescent substances may emit both visible and invisible X-rays."

Monday, February 24: First Becquerel Note:

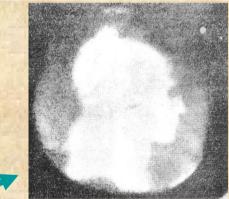
first experiments to verify Poincare's hypothesis: "On radiation emitted by phosphorescence", Comptes Rendus 122 (1896) 420



Monday, March 2: second Becquerel Note: "On the invisible radiation emitted by phosphorescent substances",

Comptes Rendus 122 (1896) 501

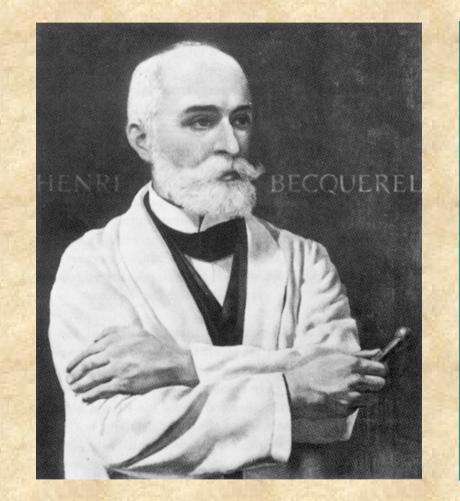
"I had prepared some plates on Wednesday, February 26 and some more on Thursday February 27. Because the sun appeared only intermittently on those days, I had saved the experiments, completely assembled,



and returened the plateholders to the darkness of a drawer, leaving the uranium crystals in place. Sinse the sun did not reappear during the following days, I did develop the plates on March 1, expecting to find very week images. On the contrary, the silhouettes appeared with great intensity..."

1896

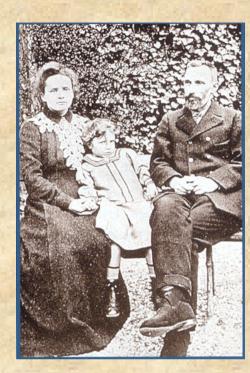
RADIOACTIVITY



First image of potassium uranyldisulfate on **24 February 1896** was the discovery of natural radioactivity

10 - 10 - 96. Sulfah Dull D'uring & d.D. Polici. Popier hois - Cong De Conom Inine -Expens an Alle & 27. at all Come Lifter La b. -Divelifer & Taman.

Antoine Henry Becquerel

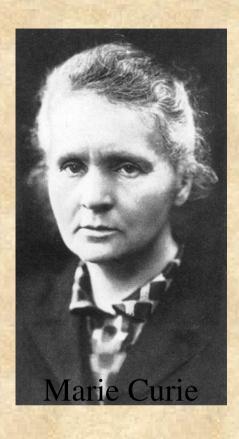


Marie and Pierre Curie with their daughter Irene

RADIOACTIVITY

Marie Curie Pierre Curie (1867 – 1934) (1859 – 1906)

1898	Polonium
	Radium
1903	Nobel Prize
	together with Pierre
	and H.Bequerel
1911	Nobel Prize
	alone



- **1897** Becquerels friend, Pierre Curie, also Prof. of physics in Paris suggested to his young wife, Marie, that she study the phenomena discovered by H.Becquerel for her thesis. She found soon that some components of Uranium minerals were much more radioactive than Uranium itself. "We shall call the mysterious rays 'radioactivity'," she told to her husband Pierre, and the substances that produce the rays "radioelements".
- **1898** Pierre started to join Marie in the study of the mysterious rays. In **July** that year they reported the discovery of **Polonium** (²¹⁰Po) and in **December** they announced the discovery of the **Radium** (²²⁶Ra)

THE TRACER PRINCIPLE 1923

G.V.Hevesy: The Absorption and Translocation of Lead (ThB) by Plants [ThB = ²¹²Pb] Biochem.J. **17**, 439 (1923)

Measurements of the tracer's Radioactivity provided thousand fold increases in sensitivity and accuracy over existing chemical assays. The foundation and basic rationale of much of Hevesy visualized that a radioactive atom might be used as a "representative" tracer of stable atoms of the same element whenever and wherever it accompanied them in biological systems.

1943 Nobel Prize Chemistry



the father of Nuclear Medicine European School of Medical Physics - Archamps

INVENTION of the CYCLOTRON



Ernest O Lawrence and his First cyclotron 1932

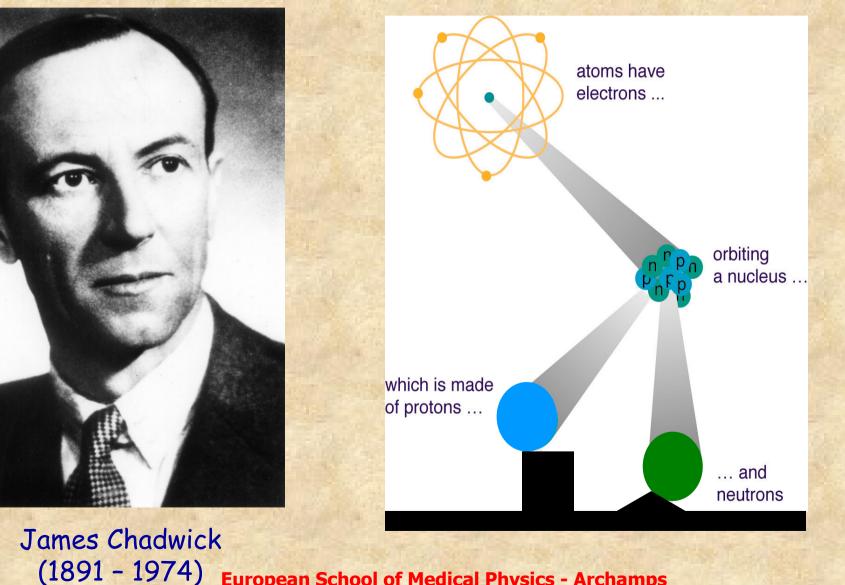
1932

E.O.Lawrence and M.S. Livingston "The production of high speed Light ions without the use of High voltages", A milestone in the production of usable quantities of radionuclides.

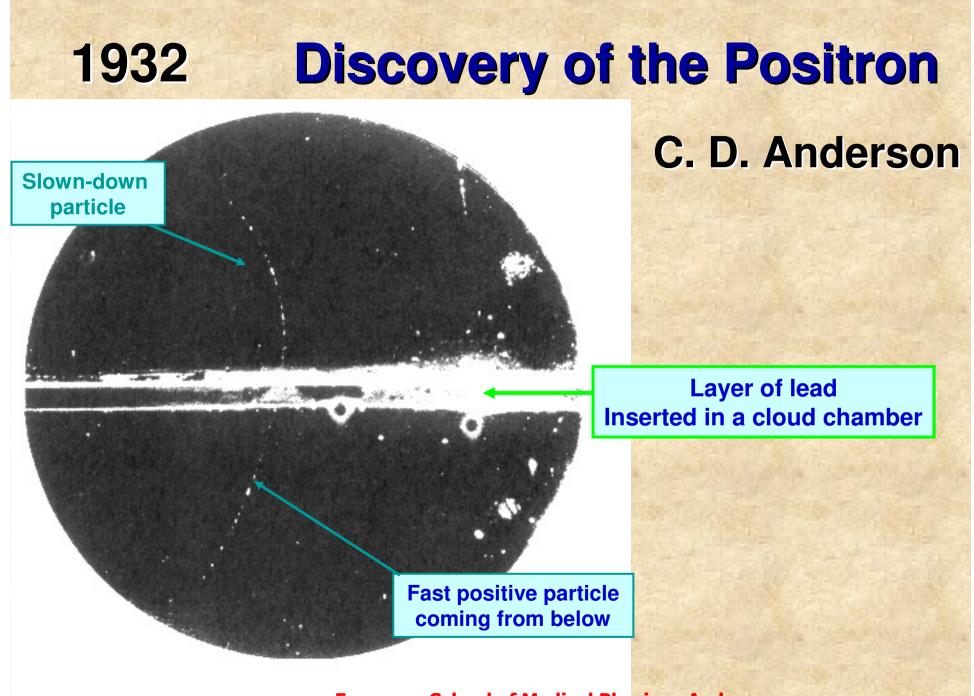
E.O Lawrence and M.S.Livingston with the 27-inch cyclotron at Berkeley 1933, the first cyclotron that produced radioisotopes



Discovery of the neutron



1932



1934 Artificial RADIOACTIVITY Irene and Frédéric Joliot-Curie

1934 Nature, February 101935 Nobel Prize

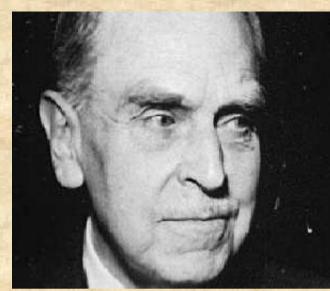
"Our latest experiments have shown a very striking fact: when aluminum foil is irradiated on a polonium preparation, the emission of positrons does not cease immediately when the active preparation is removed. The foil remains radioactive and the emission of radiation decays exponentially as for an ordinary radioelement. We observed the same phenomena with boron and magnesium."

 $^{27}Al(\alpha,n) ^{30}P$ and $^{10}B(\alpha,n) ^{13}N$



The discovery of artificial radioactivity in combination with the cyclotron opened the door to the production of a variety of useful radio-indicators. Practically any element could be bombarded in the cyclotron to generate radioactive isotopes.

1935	Nature 136, 754 O.Chievitz and G.V.Hevesy
	Radioactive indicators in the study of phosphorus metabolism in rats (³² P)
1937	Radiology 28, 178 J.G.Hamilton, R.S.Stone:
	The administration of radio-sodium (²⁴ Na)
1938	Proc.Soc.Exp.Biol.Med. 38, 510 S.Hertz, A.Roberts, R.D.Evans
	Radioactive iodine (1281) – Study of thyroid physiology
1939	Proc.Soc.Exp.Biol.Med. 40, 694, J.H.Lawrence, K.G.Scott:
	Metabolism of phosphorus (³² P) in normal and lymphomatous animals
1940	Am.J.Physiol. 131, 135 J.G.Hamilton, M.H.Soley:
	Studies of iodine metabolism by thyroid in situ
1940	J.Biol.Chem. 134, 543 J.F.Volker, H.C.Hodge, H.J.Wilson
	The adsorption of fluoride (¹⁸ F) by enamel, dentine, bone and
	hydroxyapatite
1945	Am.J.Physiol. 145, 253 C.A.Tobias, J.H.Lawrence, F.Roughton
	The elimination of 11-C-Carbon monoxide from the human body



Otto Hahn, 1944 Nobel Prize

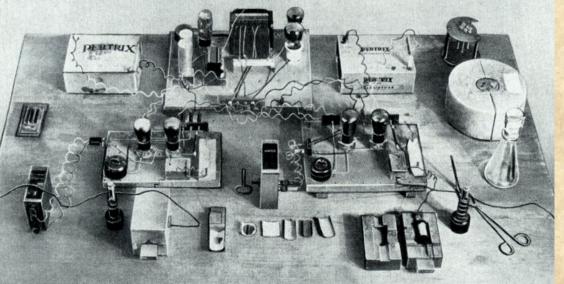
Als Chemiker müßten wir ... statt Ra, Ac und Th die Symbole Ba, La und Ce einsetzen. Als der Physik in gewisser Weise nahestehende Kernchemiker können wir uns zu diesem, allen bisherigen Erfahrungen der Kernphysik widersprechenden Sprung noch nicht entschließen. Es könnten doch vielleicht eine Reihe seltsamer Zufälle unsere Ergebnisse vorgetäuscht haben.

Niels Bor (Jan.1939) Mein Gott, wie haben wir das nur so lange übersehen können

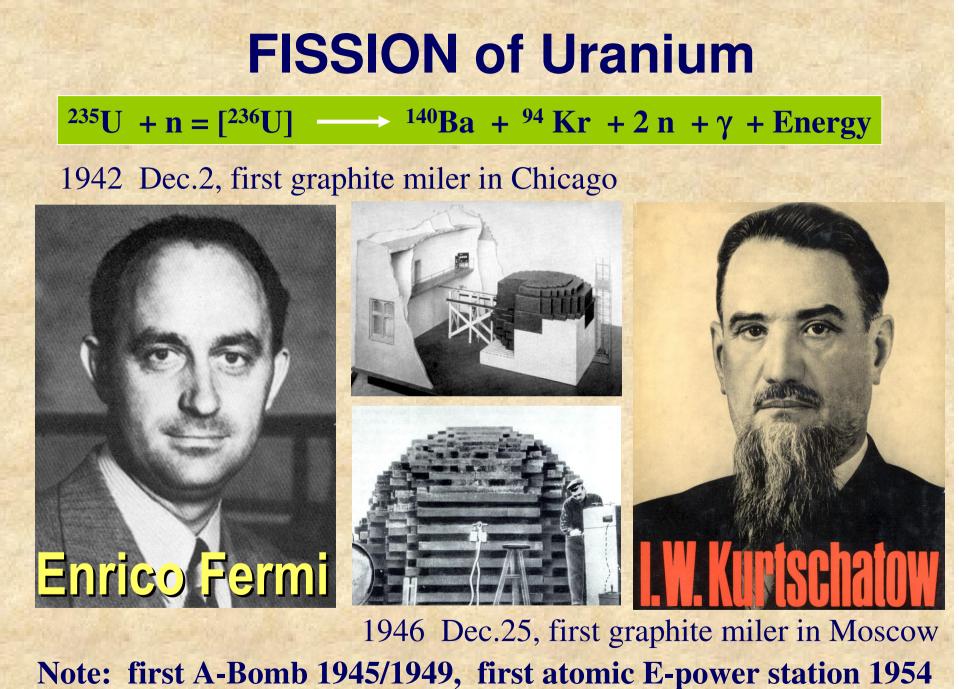
FISSION of Uranium 1938, 17. Dec.

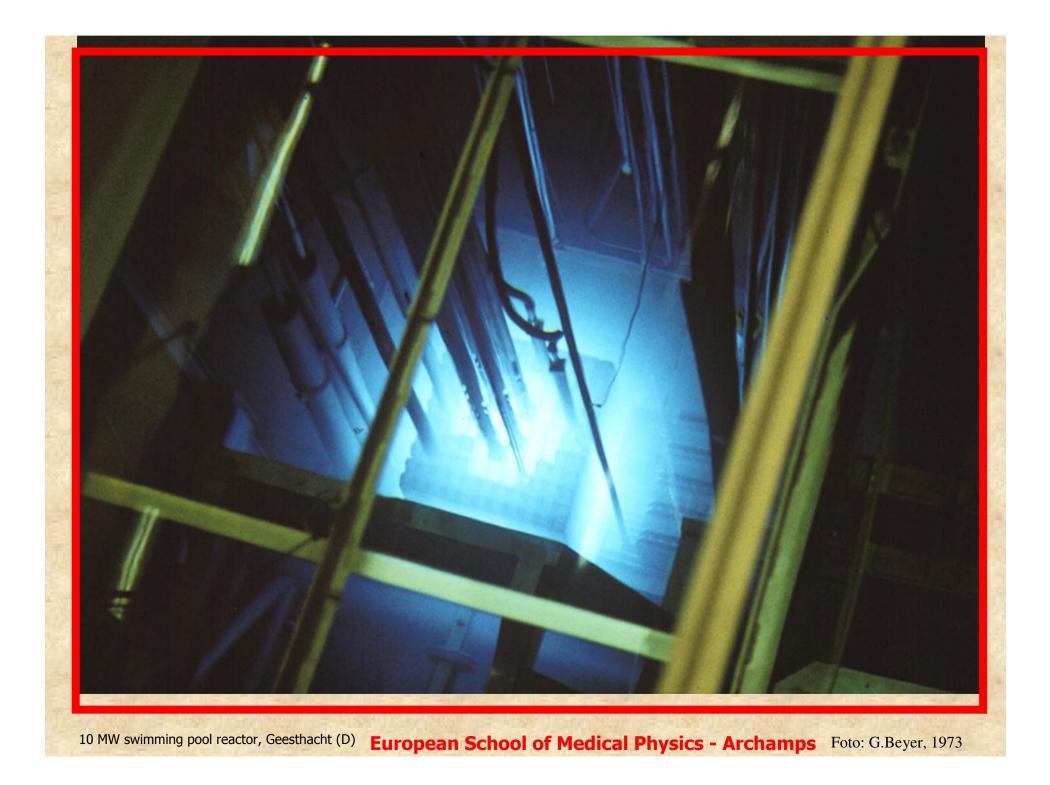
Naturwissenschaften <u>1</u>, (1939) 1 O.Hahn und F.Straßmann Über den Nachweis und das Verhalten der bei der Bestrahlung des Urans mittels Neutronen entstehenden Erdalkalimetalle

Lise Meitner and O.R Frisch described the Explanation and defined the terminus "FISSION"



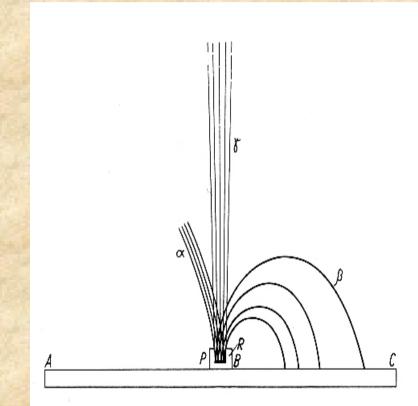
nenLaboratory table of Otto HahnEuropean School of Medical Physics - Archamps



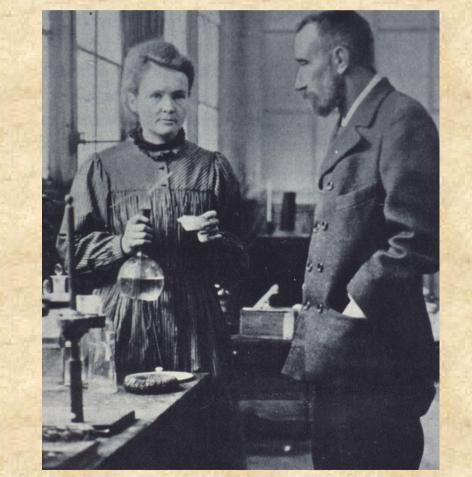


RADIOACTIVITY

Marie Curie Pierre Curie (1867 – 1934) (1859 – 1906)

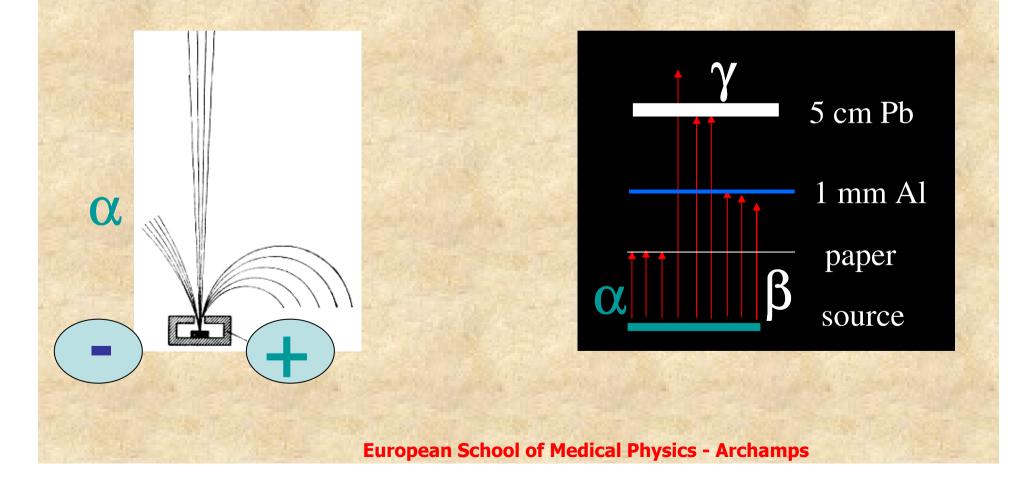


Hundred years ago



RADIOACTIVITY

The radiation characteristics of an isotope determines where and how it can be used in medicine



DECAY MODES

	Energy	Rar	nge in wate	er / Pb
Little and the second se	~ 0.2 – 4 MeV		~ 1 cm	
EC, X-ray f(Z)	Fe: 7 keV Ra: 100 keV		~ 1 mm ~ 4 cm	
α	4 - 8 MeV		– 80 μm	0.01 mm
γ	50 keV 140 keV		~ 3 cm ~ 5 cm	0.01 mm 0.4 mm
	1 000 keV		~ 10 cm	1 cm
conversion electrons	2 000 keV Εγ - Ee		~ 15 cm mm range	1.5 cm
Auger electrons			µm range	
IT exotic decay modes,	like γ spontaneous fission			

1946, June 14 Nuclear Medicine's modern era began

Availability of Radioactive Isotopes,

Announcement from Headquarters, Manhatten Project, Washington D.C.:

Production of tracer and therapeutic radioisotopes has been heralded as one of the greatest peacetime contributions of the uranium chain-pile. This use of the uranium pile will unquestionably be rich in scientific, medical, and technological application.

> On 1.Aug.1946 the Atomic Energy Act passed the congress, releasing radioisotopes from military control.

RADIOISOTOPES in **MEDICINE**:

Requirements - Production - Application and Perspectives

Imaging with Radiotracers

Gerd-Jürgen BEYER

Prof.Dr.rer.nat.habil. Cyclotron Unit, University Hospital of Geneva, Switzerland GSG-Int. GmbH, Switzerland gerd.beyer@cern.ch gerd.beyer@gsg-int.com



Lecture ESI Week 1 Archamps (France) October 19, 2012 European School of Medical Physics - Archamps



ISOTOPES IN MEDICINE

DIAG	NOSIS		THERAPY	Y	
in vitro	in vivo	internal	internal external		rnal
¹⁴ C	⁹⁹ Mo- ^{99m} Tc	systemic	sou	rces	tele radio
³ H 125 others	201 TJ 123J 111 I n 67Ga 81Rb-81mKr 0thers B* emitters for PET 18F, 11C, 13N, 15O 86Y, 124J, 64Cu 68Ge - 68Ga 82Sr - 82Rb 44Ti - 44Sc	 131,90γ 153 Sm, 186 Re 188 W-188 Re 166 Ho, 177 Lu, others 04 dets 05 Ac-213 Bi 25 Ac-213 Bi 211 At, 223 Ra 149 Tb e⁻-emitters: 125 	Sealed s ¹⁹² Ir, ¹⁸² Ta, many othe needles brachyth (¹⁰³ Pd), ¹²⁵ many othe stands ³² P and oth seeds ⁹⁰ Sr or ⁹⁰ Y applicat ¹³⁷ Cs, othe	¹³⁷ Cs ers for nerapy: I rs ners , others OrS	60Co gamma knife ¹³⁷ Cs blood cell irradi- ation

THE TRACER PRINCIPLE 1923

G.V.Hevesy: The Absorption and Translocation of Lead (ThB) by Plants [ThB = ²¹²Pb] Biochem.J. **17**, 439 (1923)

Measurements of the tracer's Radioactivity provided thousand fold increases in sensitivity and accuracy over existing chemical assays. The foundation and basic rationale of much of Hevesy visualized that a radioactive atom might be used as a "representative" tracer of stable atoms of the same element whenever and wherever it accompanied them in biological systems.

1943 Nobel Prize Chemistry



NUCLEAR MEDICINE = in vivo APPLICATION of RADIOTRACERS

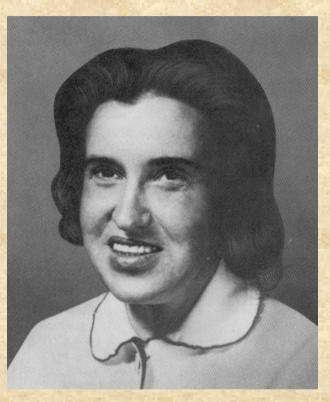
- 1923 First tracer study with ²¹⁰Pb/²¹⁰Bi G.Hevesy
- 1925 ²¹⁴Bi arm-to-arm circulation time, H.Blumgart
- 1935 ³²P renewal of mineral constituents of bone, O.Chieivitz & G.Hevesy
- 1937 dynamics of sodium transport in vivo, J.G.Hamilton
- 1937¹²⁸I, thyroid physiology, R.Hertzs, A.Roberts, R.Evans
- 1938 ¹³¹I discovered by G.T.Seeborg, 1939 first diagnostic use J.G.Hamilton et al.
- 1947 ¹³¹I –Fluorescine, 1950 ¹³¹I –HSA, 1955 ¹³¹I-rose bengale & hippurane, ...
- 1957 ⁹⁹Mo-^{99m}Tc generator (1960 first sale), ¹³³Xe for lung ventilation
- 1969 ⁶⁷Ga accumulation in cancer, C.L.Edwards
- 1970 Instant KIT's for ^{99m}Tc
- 1973 ²⁰¹Tl, ¹²³I, ¹¹¹In, many other isotopes and tracer compounds
- 1978 first ¹⁸FDG PET scan

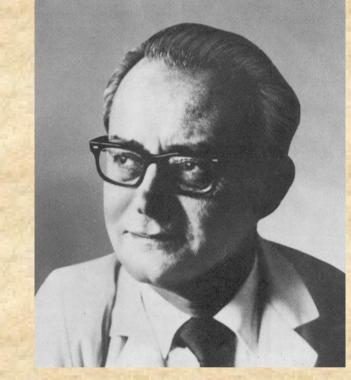
> 30 million individuals receive every year a radiotracer for diagnosis

Application	Requirement	Isotope
DIAGNOSIS in vitro	T _{1/2} = long biogenic behavior	³ H, ¹⁴ C 125
DIAGNOSIS	single photons	99mTc
In vivo	no particles	¹²³], ¹¹¹]r
SPECT	biogenic behavior	
	$T_{\frac{1}{2}} = moderate$	201 T I,
DIAGNOSIS	B ⁺ -decay mode	¹¹ C,
in vivo	biogenic elements	13N, 15O
PET	$T_{\frac{1}{2}} = short$	18

Diagnostic in vitro

RIA



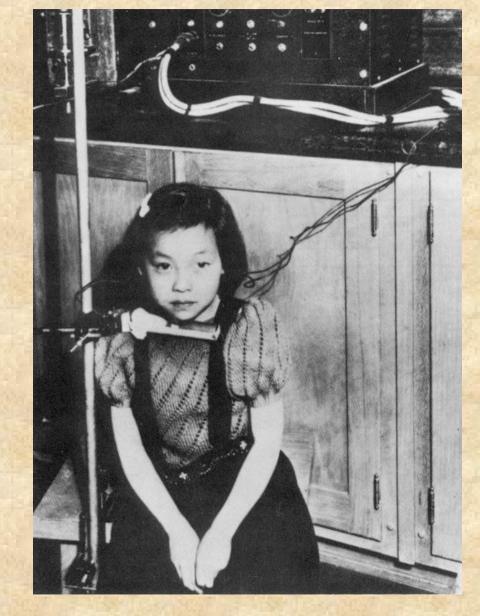


Rosalyn S.YALOW S.A.BERSON Nobel Prize 1977 Introduced the radioimmunoassay (RIA)

assay for insulin based on the principle of competitive binding by antibody of natural and radioactive labeled hormone) European School of Medical Physics - Archamps

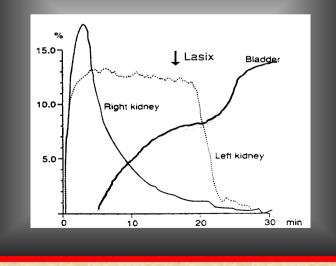
ISOTO	PES in MED	CINE
Application	Requirement	Isotope
DIAGNOSIS In vitro	$T_{1/2} = long$ biogenic behavior	³ H, ¹⁴ C 125J
DIAGNOSIS In vivo SPECT	single photons no particles biogenic behavior $T_{\frac{1}{2}}$ = moderate	99mTc, ¹²³ I, ¹¹¹ In, ²⁰¹ TI,
DIAGNOSIS in vivo PET	B^+ -decay mode biogenic elements $T_{\frac{1}{2}}$ = short	¹¹ C, ¹³ N, ¹⁵ O, ¹⁸ F

Photo published 1942



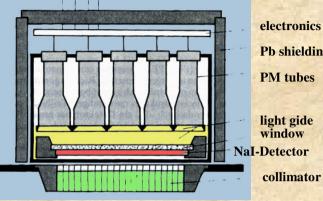
J.G.Hamilton, M.H.Soley: "Studies of iodine metabolism by thyroid in situ" 1940, Am.J.Physiol. <u>131</u>, 135

Kidney Isotope Nephrogram



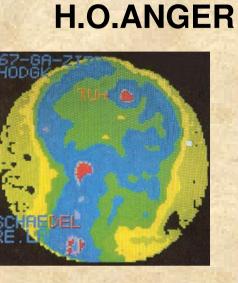


GAMMA CAMERA



electronics Pb shielding PM tubes

Planar scintigram 1958



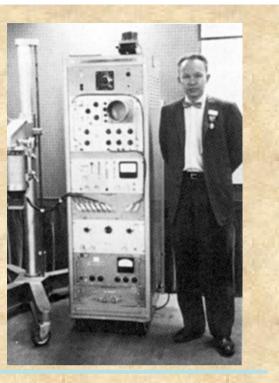


Scan Thyroid normal

B.CASSEN SCANNER

The scanner was designed for 131-I

European School of Medical Physics - Archamps



Pre-amplifier PM-tube **Pb-shielding Nal-Detector** Collimator Object



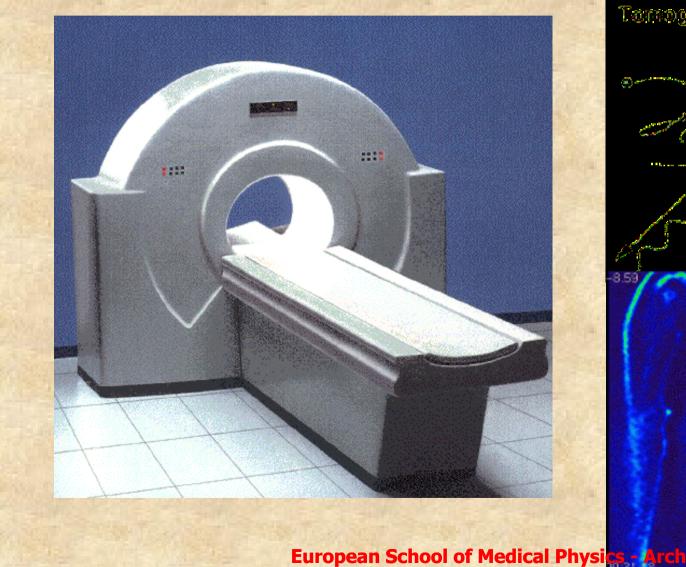


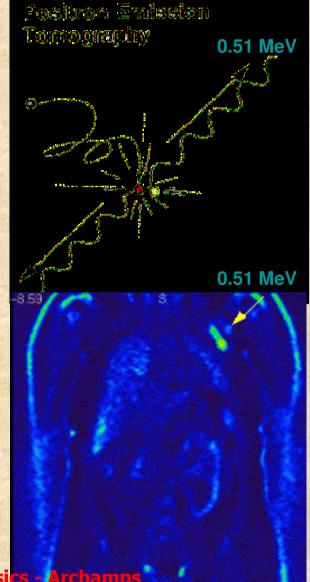
Modern SPET

Cameras (GE Medical Systems)



PET = Positron Emission Tomography







3D whole-body PET

ECAT HR+

ECAT ACCEL

SIEMENS

25 year-old male with Melanoma,50 year-old male with colon CA71 kg, 178 cm, 625 MBq FDG, 45 min p.i.91 kg, 183 cm, 720 MBq FDG, 162 min p.i.



Emission scan time: 54 min Transmission scan time: 18 min

Data courtesy of Kettering Memorial Hospital, Kettering, USA year 2000



Emission scan time: 27 min Transmission scan time: 18 min Data courtesy of NC PET Imaging Center, Sacramento, USA School of Medical Physics - Archamps

PET/CT concepts



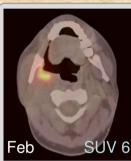


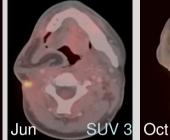
PET/CT today

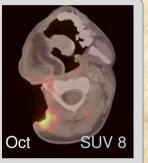


PET/CT 2009: routine application

Oncology







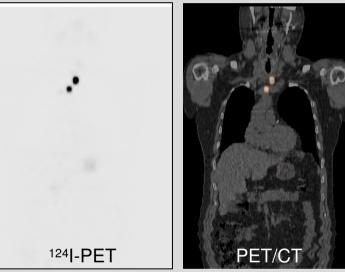
35 y/o M w/ malignancy in mandibula

Feb: ¹⁸FDG-PET/CT identified disease Mar: right mandibulectomy and maxillectomy Jun: PET/CT identified recurrent disease Jun: Extensive surgery Oct: PET/CT showed recurrent disease

UPMC, Pittsburgh

Anatomy CT-based AC

Oncology - Specific tracer



45 y/o M w/ papillary thryorid CA (pT4) 124I-PET/CT

→ mediastinal LN metastasis w/o CT correlate

UH Essen-Duisburg

Anatomy CT-based AC



Structure without function is a corpse, ... function without structure is a ghost.

Stephen Wainwright, Duke Dept Biology



RADIOISOTOPES in **MEDICINE**:

Requirements - Production - Application and Perspectives

Medical Isotope Production

Gerd-Jürgen BEYER

Prof.Dr.rer.nat.habil. Cyclotron Unit, University Hospital of Geneva, Switzerland GSG-Int. GmbH, Switzerland gerd.beyer@cern.ch gerd.beyer@gsg-int.com

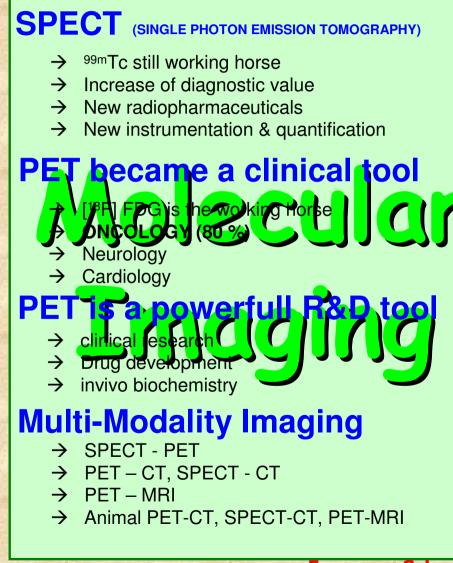
EUROPEAN INSTITUTE ESI Lecture ESI Week 1 Archamps (France) October 19, 2012 European School of Medical Physics - Archamps



NUCLEAR MEDICINE 2009

DIAGNOSIS

THERAPY (RIT)



New Approaches in Radionuclide Therapy

- → free chelates (EDTMP, others)
- bio-selective antibody conjugates (mab) \rightarrow
- bio-specidfic peptide conjugates \rightarrow
- \rightarrow Lyposomes
- \rightarrow Nanoparticles
- \rightarrow others

New Radionuclides for Therapy:

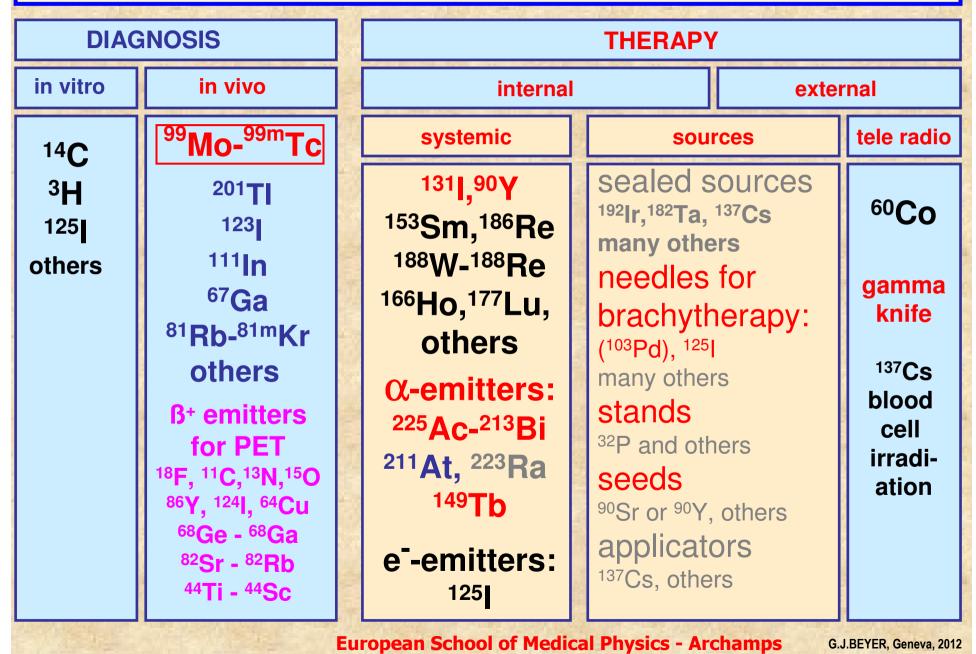
- → Beta-Emitters with different energies
- \rightarrow Alpha emitters
- → Auger electron emitters
- \rightarrow Research isotopes

PET for individual invivo dosimetry

- \rightarrow Quest for metallic positron emitters
- longer lived positron emitters \rightarrow
- \rightarrow PET imaging with "durt" isotopes
- Simmulatne Multi-tracer studies \rightarrow

G.BEYER (ITD, Geneva, 2010)

ISOTOPES IN MEDICINE

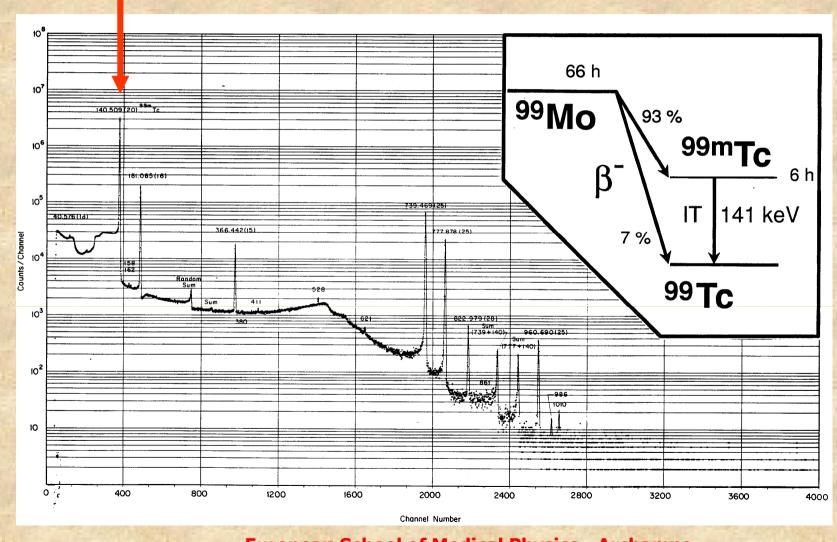


Why 99m-Tc

Three Aspects:

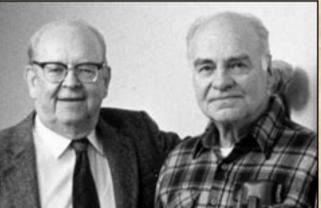
- 1. Nuclear properties
- 2. Generator principle (availability)
- 3. Sn(II) as reducing agent opened the door for KIT Technology

141 keV photons - the strength of ^{99m}Tc



Discovery of ^{99m}Tc generator in 1957 in BNL





^{99m}Tc detected while refining ¹³²I from ¹³²Te → ⁹⁹Mo-^{99m}Tc generator → Stang, Tucker, Greene,

Richards BNL declined to file a patent for this device - ^{99m}Tc generator!

BNL memo in 1958: "We are not aware of a potential market for technetium-99 great enough to encourage one to undertake the risk of patenting in hopes of successful and rewarding licensing."

Slide from Ramamoorthy, IAEA Vienna

Sn(II) as reducing agent

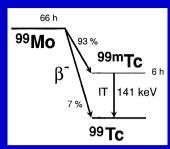
1962: The 99-Mo / 99m-Tc Generator introduced into clincal practice: Harper P.V., G.Autors, K.A.Latrhrop, W siemens, L Weiss: Technetium-99m as a biological tracer, J.Nucl.Med. **3** (1962) 209

1969: Introduction of Sn(II) as reducing agent:

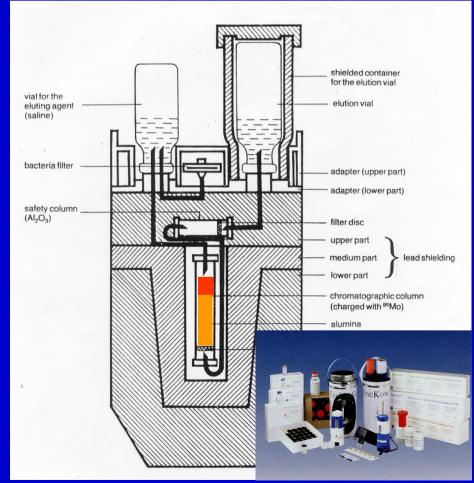
Dreyer R. & Muenze R. (Dresden, Germany); Markierung von Human Serum Albominw with 99m-Tc Wiss.Z.K-Marx Uni Leipzig Nat.Wiss.R. **18** (1969) 629-633 Zur Tc-99m-Markierung von Serumalbumin; Isotopenpraxis **5** (1969) 296

2009: Eckelmann W.C.: JACC cardiovascular Imaging 2, (2009) 364-368:

"...On a practical level, the use of stannous ion was a key development and current radiopharmaceutical kits employ the stannous reduction technique. With the advent of the Mo-99/Tc-99m generator in the 1960s followed by the development of "instant" kits, the use of Tc-99m–labeled compounds expanded rapidly"



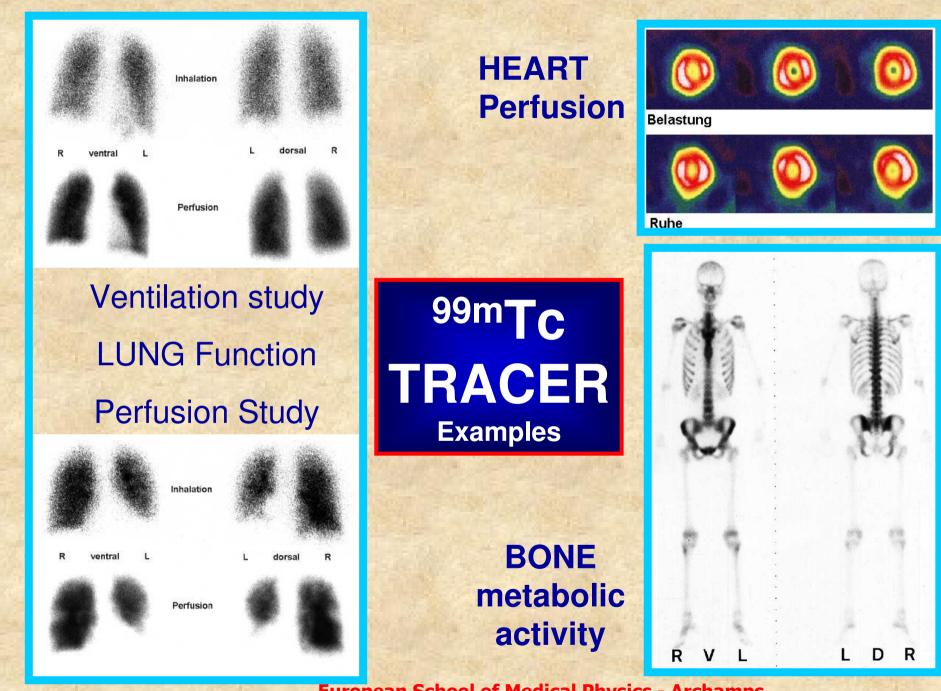
99mTcO₄ 0.9 % NaCl solution

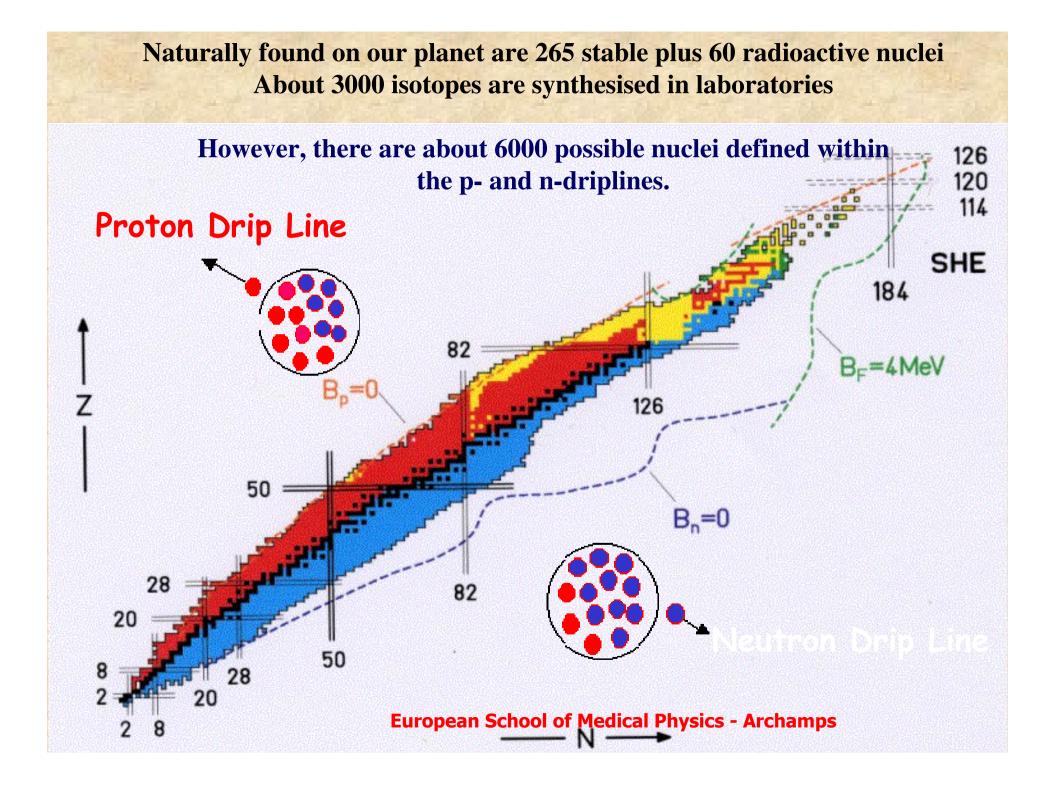


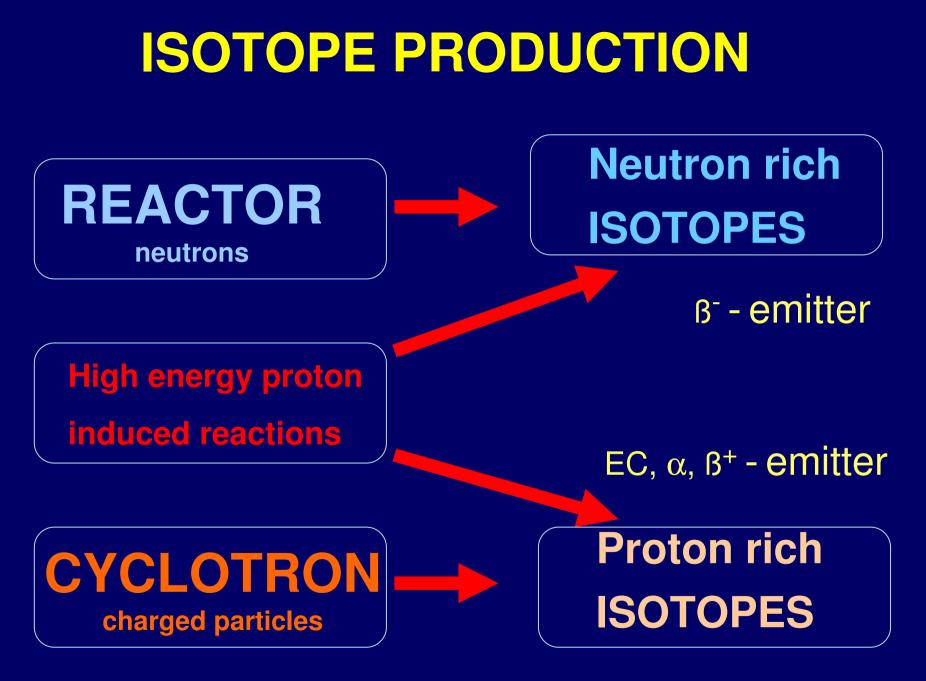


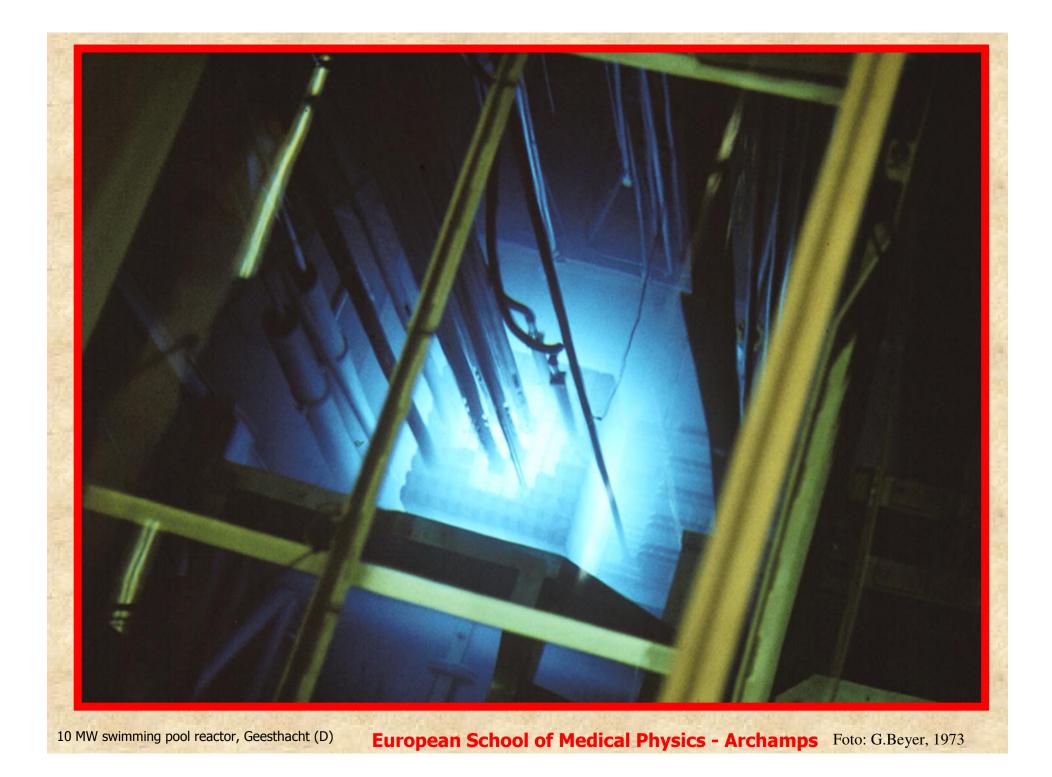
many different 99m C-tracer for imaging of many different organ and tissue functions

European School of Medical Physics - Archamps









⁹⁹Mo: **PRODUCTION ROUTES**

 $\begin{array}{rcl} 98 \, \text{Mo} & (n, \gamma) & 99 \, \text{Mo} \\ \sigma & = & 0.130 \, \text{b} \\ 1 \, g^{\,98} \, \text{Mo}, \, \Phi_{\mu^{\mu}} = 1 \, ^* \, 10^{13} \, \text{cm}^{-2} \text{s}^{-1} \\ 8 \, \text{GBq}^{\,99} \, \text{Mo/g}^{\,98} \, \text{Mo} \, (\text{low specific activity}) \end{array}$

NRU Reactor, Canada

- 15 May 2009: D₂O leak
- stopped till spring 2010+
- license till October 2011



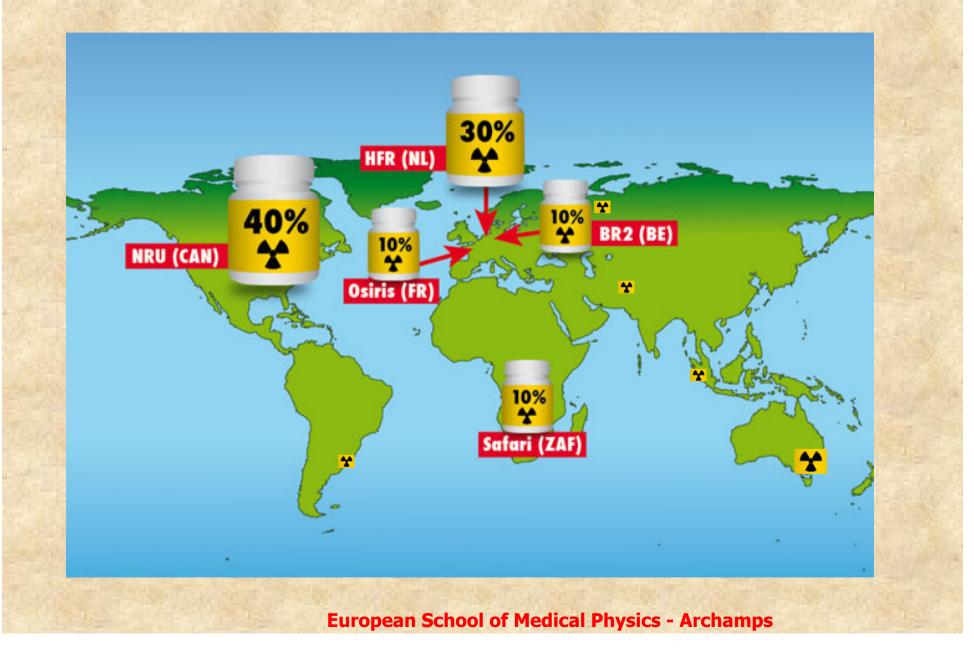
HFR Petten, NL

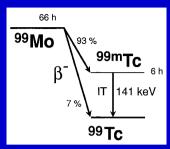
extended maintenance stop from 19 February 2010



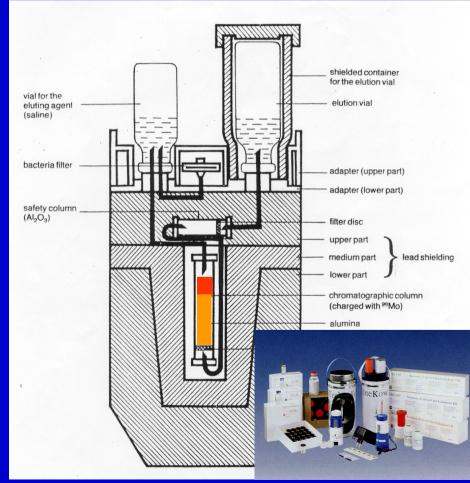


Main ⁹⁹Mo-Production Sites





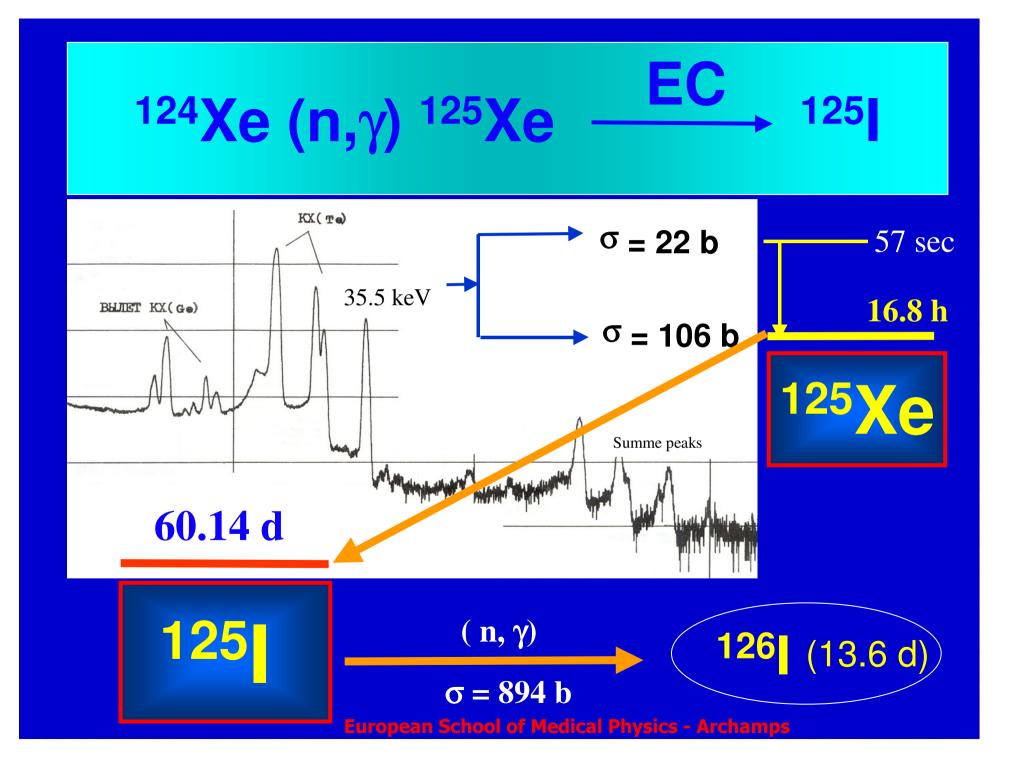
99mTcO₄ 0.9 % NaCl solution





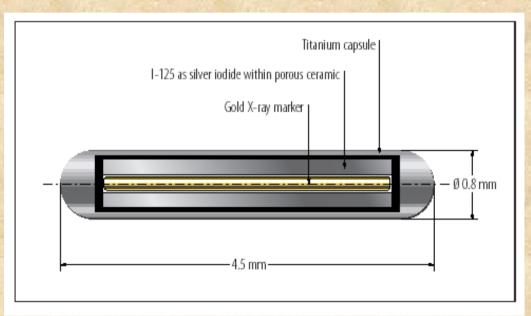
many different 99m C-tracer for imaging of many different organ and tissue functions

European School of Medical Physics - Archamps



IsoSeed J-125

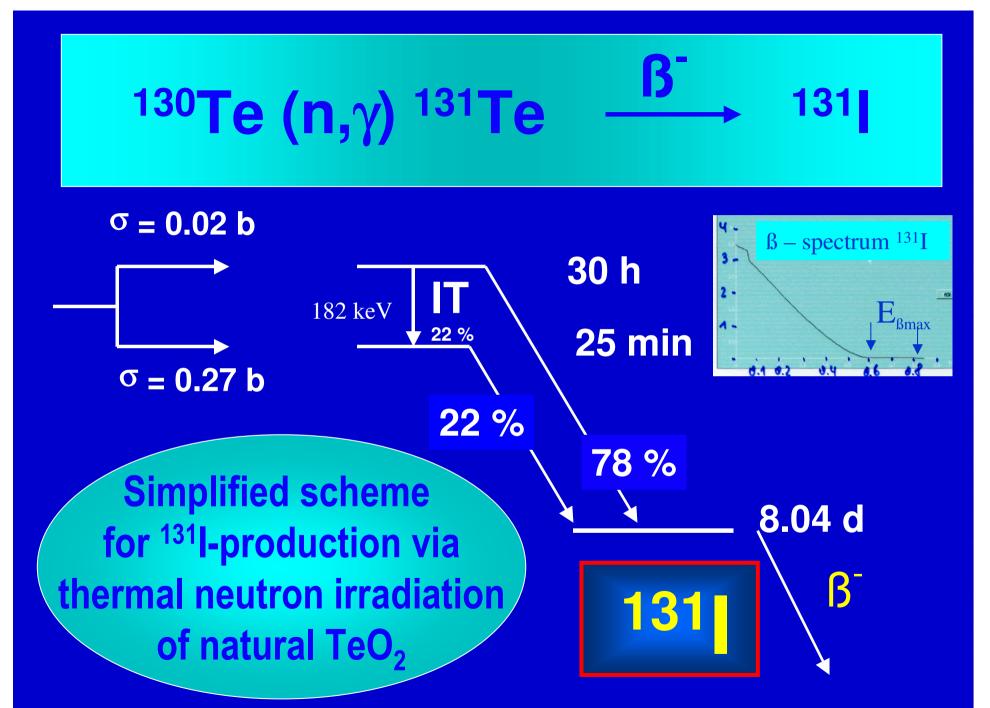
An Eckert & Ziegler Company



The IsoSeed is equipped with a high-density gold marker providing excellent CT visibility. The full-length marker allows easy and precise location of each seed and produces minimal artefacts. This enhances the precision of the post-implant quality control.



European School of Medical Physic



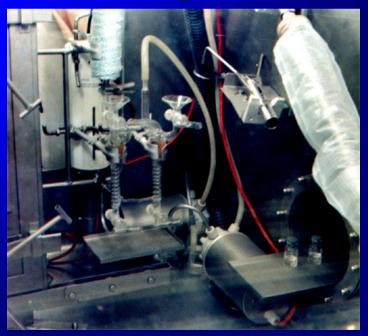


¹³¹I Production Technology

Hot Cell Facility

Target Processing

Iodine Trapping



Problems of medical RI Production with CYCLOTRONS

Short range of the particle beam

Range:

30 MeV p about 1 mm 15 MeV d about 0.3 mm 30 MeV a about 0.1 mm

small target - high thermic energy deposition!

Small cross sections

Low productivuity

Limited and expensive target material

enriched isotopes

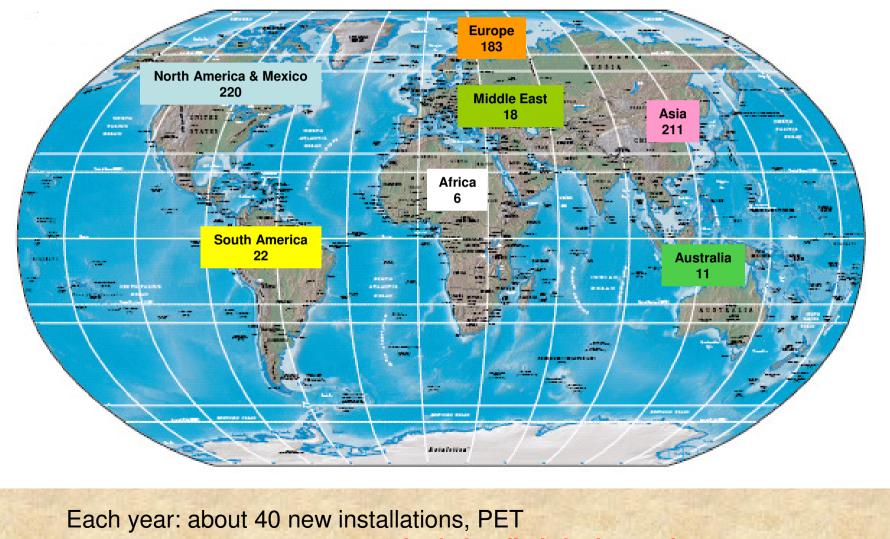
Vacuum

Target window problems, sensitive target material

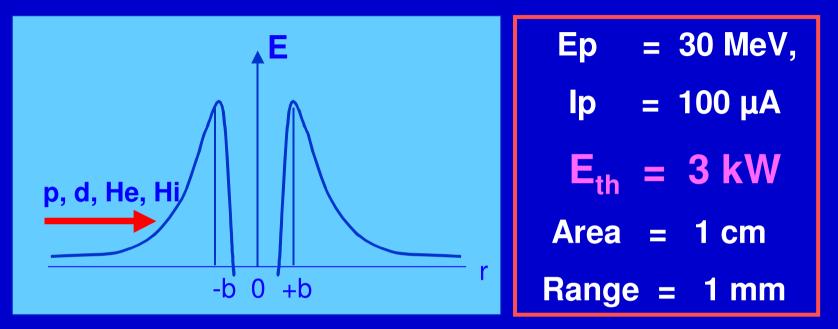
Single target – one isotope

Distribution of cyclotrons for production of PET tracers (2008)

(source: D. Schlyer, BNL/USA, based on inputs of 4 major manufacturers)

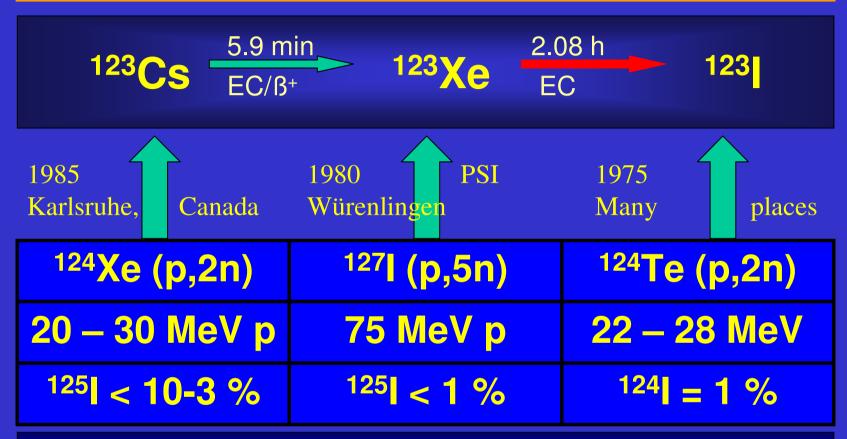


Particle Beam Energy



- at "b" strong nuclear force
- only one particle out of 10⁴.....10⁷ reacts in reality
- full particle beam is stopped inside the target material
- the whole particle energy is transformed into thermic energy (heat)

123-IODINE PRODUCTION ROUTES



ALTERNATIVES:

local ¹²³ I production using PET cyclotrons ¹²³Te (p,n) ¹²³ I 15 MeV p, 150 MBq/μAh Fast, easy, reliable, clean product, suitable for direct labeling,

ISOTOPES for Tracer Synthesis

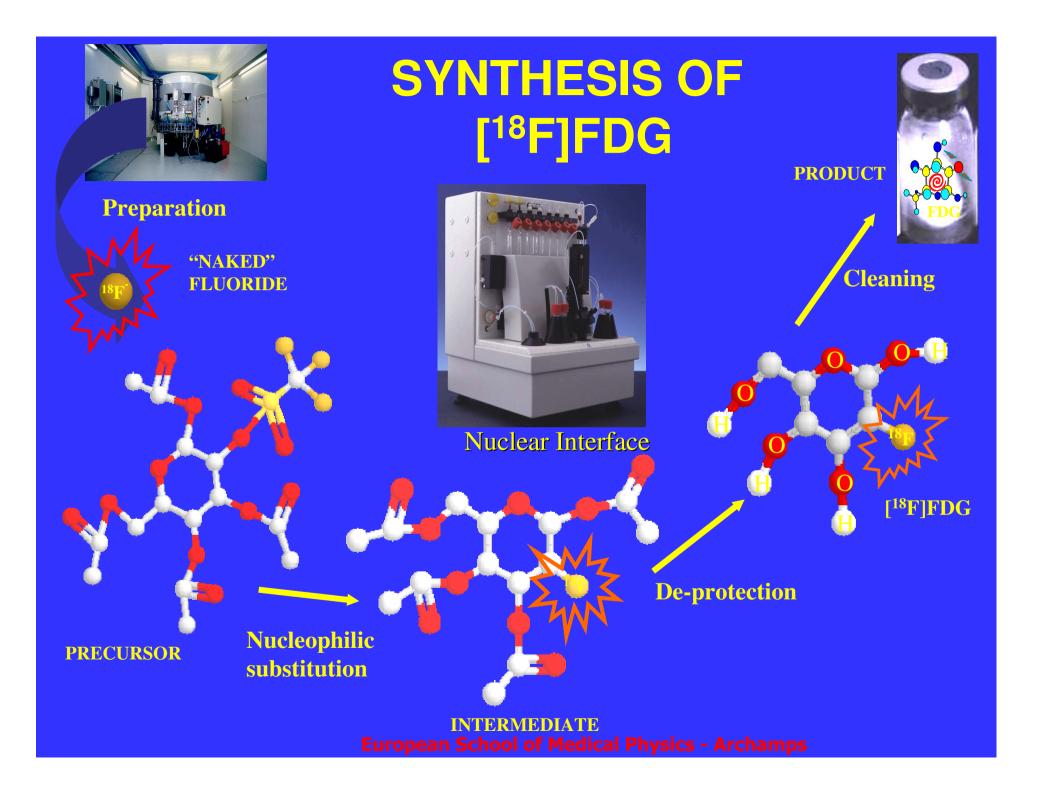
ISOTOPE	T _{1/2}	Reaction	Target	Product
¹¹ C	20 min	¹⁴ N (p,α) ¹¹ C	N ₂	unlimited
¹³ N	10 min	¹⁶ Ο (p,α) ¹³ N	H ₂ O	[¹³ N]NH ₃
¹⁵ O	2 min	¹⁴ N (d,n) ¹⁵ O	H ₂ O	[¹⁵ O]H ₂ O
¹⁸ F	110 min	¹⁸ O (p,n) ¹⁸ F ²⁰ Ne (d, α) ¹⁸ F	[¹⁸ O]H ₂ O ²⁰ Ne	[¹⁸ F]FDG [¹⁸ F]FDOPA

The FIRST FDG SYNTHESIZER

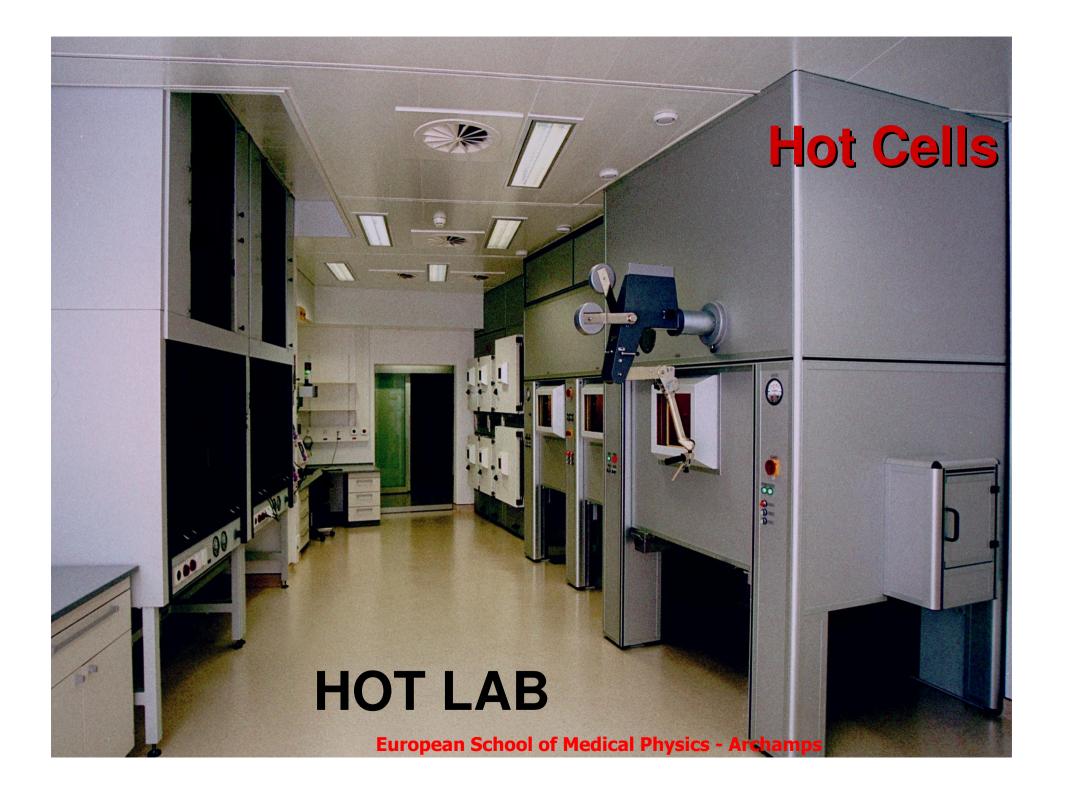


European School of Medical Physics - Archamps

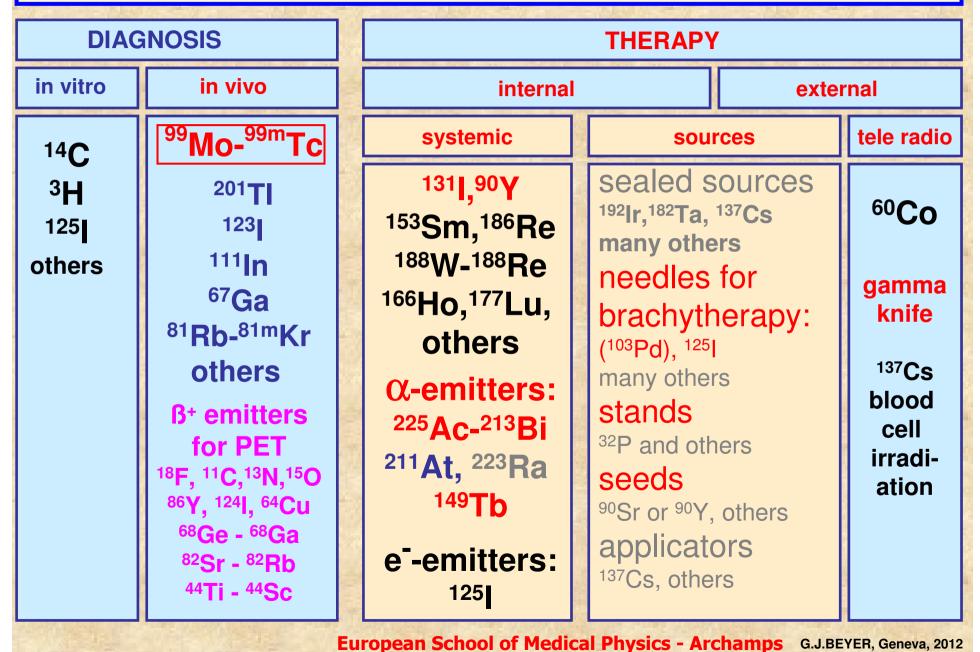
J. Fowler, BNL

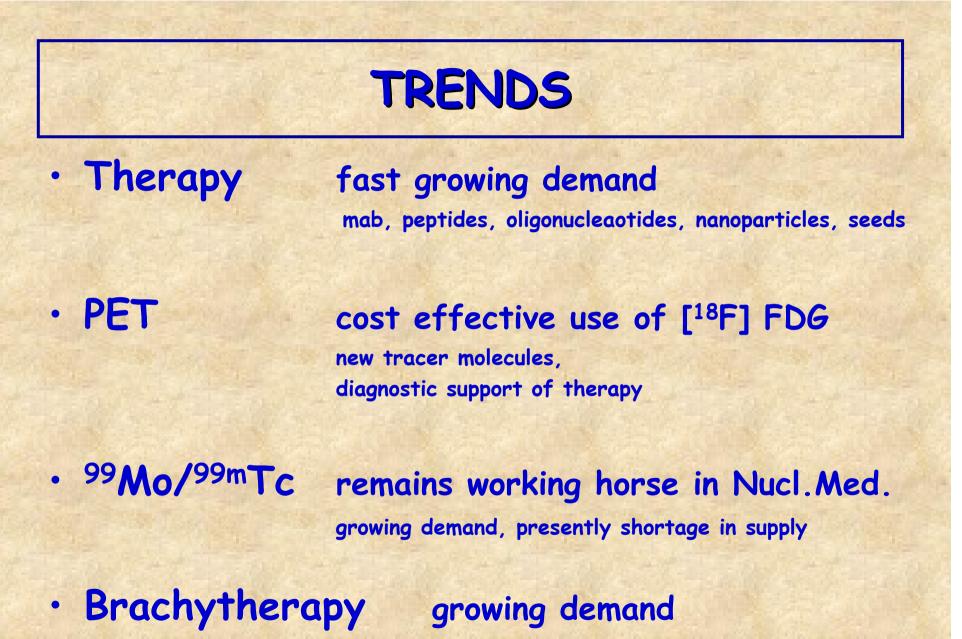






ISOTOPES IN MEDICINE





seeds (¹²⁵I) spheres (⁹⁰Y)