

# Vacuum Systems Lecture 3

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#### Outline

- 1. Cryopumping
- 2. Adsorption isotherms
- 3. Crysorber in cold systems: case of LHC
- 4. Helium leaks in cold systems: case of LHC



# 1. Cryopumping



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#### **Desorption probability**

• The desorption probability, P, is a function of the binding energy, E and the temperature, T (first order desorption, Frenkel 1924). The surface coverage,  $\theta$ , varies like :

$$P = \frac{d\theta}{dt} = -\theta v_0 e^{-\frac{E}{kT}}$$

 $(v_0 \sim 10^{13} \text{ Hz}, \text{ k} = 86.17 \ 10^{-6} \text{ eV/K})$ 

• The desorption process is characterized by the sojourn time,  $\tau$ :

$$\tau = \frac{e^{\frac{E}{kT}}}{\nu_0}$$

• For large E and small T, molecules remains onto the surface : CRYOPUMPING

• For some combination of E and T, the molecule is desorbed (bake out)



#### Sojourn time at cryogenic temperature

• Cryosorption occurs till ~ 100 k

E(eV)	1.9 K	4.2 K	50 K	70 K
0.01	1 10 <sup>6</sup> years	0.1 s	1 ps	0.5 ps
0.02	∞	3 10 <sup>3</sup> years	10 ps	2 ps
0.15	œ	œ	130 s	6 ms
0.21	œ	œ	5 years	130 s
0.3	œ	œ	1 10 <sup>4</sup> years	12 years

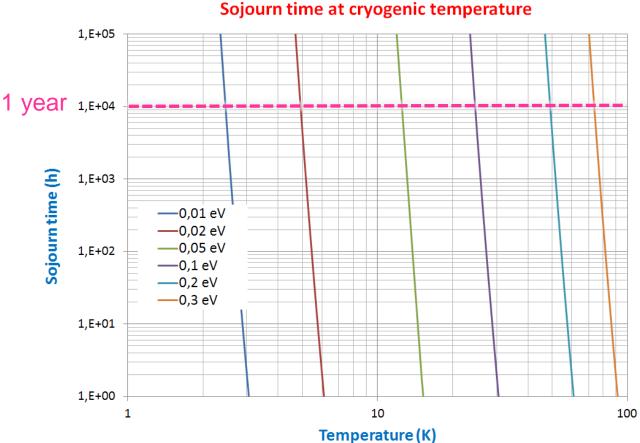
$$\tau = \frac{e^{\frac{E}{kT}}}{\nu_0}$$



#### **Sojourn time - Physisorbed molecules**

• Physisorption occurs:

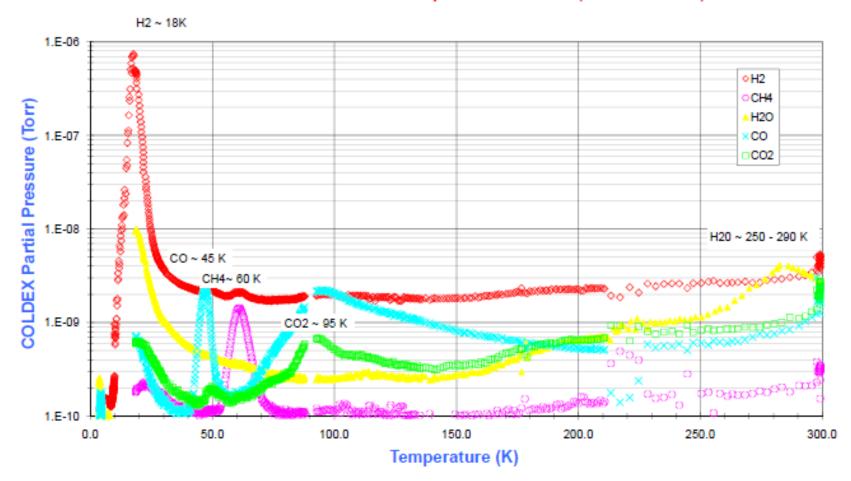
below 20 K for binding energies < 0.1 eV below 50 K for binding energies < 0.2 eV below 70 K for binding energies < 0.3 eV





#### A Natural Warm Up of a St. Steel Cold Bore

COLDEX #14 19-25/3/99, Cu BS. Natural warm up of CB at 2.2 K/h (TBS>20 a 50 K)





## **Cryopumping regimes**

#### **Physisorption**

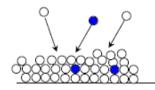
- Sub-monolayer coverage : attractive force (van der Waals) between a gas molecules and a material
- Binding energy for physical adsorption
- $H_2$  from 20 to 85 meV for smooth and porous materials resp.
- 1 h sojourn time at 5.2 K and 26 K for smooth and porous materials resp.

#### Condensation

- For thick gas coverage, only forces between gas molecules
- Energy of vaporisation 9 to 175 meV for  $H_2$  and  $CO_2$  resp.
- 1 h sojourn time at 2.8 K and 53.4 K for  $H_2$  and  $CO_2$  resp.
  - sub-monolayers quantities of gas can be *physisorbed* at their boiling temperature (ex : H<sub>2</sub> boils at 20.3 K and a bake-out above 100 °C removes water)

#### Cryotrapping

• Use of a easily condensable carrier (*e.g.* Ar) to trap molecules with a high vapor pressure gas (*e.g.* He,  $H_2$ )





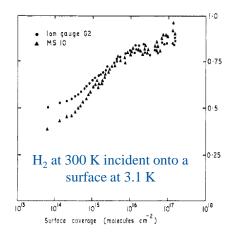
#### Sticking probability/coefficient

• Probability :  $0 < \sigma < 1$ 

v collision rate (molecules.s<sup>-1</sup>.cm<sup>-2</sup>)

$$\sigma = \frac{v_{\text{incident}} - v_{\text{departing}}}{v_{\text{incident}}} = \frac{v_{\text{sticking}}}{v_{\text{incident}}}$$

• Function of gas, surface, surface coverage, temperature of gas and surface temperature

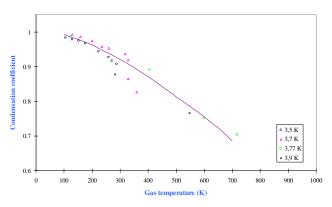


J.N. Chubb et al. J. Phys. D, 1968, vol 1, 361

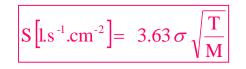
• Pumping speed

$$\mathbf{S} = \frac{1}{4} \sigma \left( 1 - \frac{\mathbf{P}}{\mathbf{P}_{\text{sat}}} \right) \mathbf{A} \mathbf{v} \approx \frac{1}{4} \sigma \mathbf{A} \mathbf{v}$$

*i.e* :  $\sigma$  times the conductance of a surface



J.N. Chubb et al. Vacuum/vol 15/number 10/491-496



•  $H_2$  and CO at 4.2 K :  $S_{H2} = 5.3 \text{ l.s}^{-1}.\text{cm}^{-2}$  $S_{CO} = 1.4 \text{ l.s}^{-1}.\text{cm}^{-2}$ 



#### **Capture factor, C<sub>f</sub>**

• The capture factor Takes into account the geometry of the system :

Vacuum chamber Rodiction boffle K{{{{}} Rodiation shield (77K) vacuum enlosure Second stage Cryopanel (20K) Charcoal cryosorption panel First stope Cold head drive unit Gaseous helium supply from compressor unit

Baffle in a cryopump

 $C_{f} \sim 0.3$ 

R. Haefer. J. Phys. E : Sci. Instrum., Vol 14, 1981, 273-288

Holes in the electron shield of the LHC beam screen

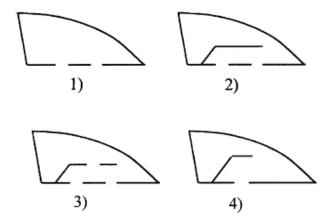


Fig. (1) Two slots in the beam screen, without electron shield, (2) two slots in the beam screen, electron shield without slot, (3) two slots in the beam screen, electron shield with slot, (4) only one slot in the beam screen, electron shield without slot.

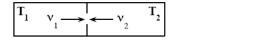
	1			
0.1	0.48 0.68	0.26	0.39	0.43
1	0.68	0.36	0.51	0.57
	7			



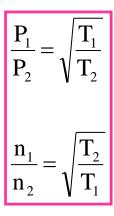
#### **Thermal transpiration**

 $v = \frac{1}{4}nv$ 

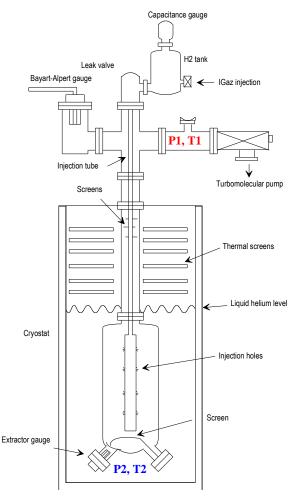
- Vacuum gauges are located at room temperature to reduce heat load
- For small aperture, the collision rate, v, is conserved at the cold / warm transition



• Since the average velocity scales like  $\sqrt{T}$ 

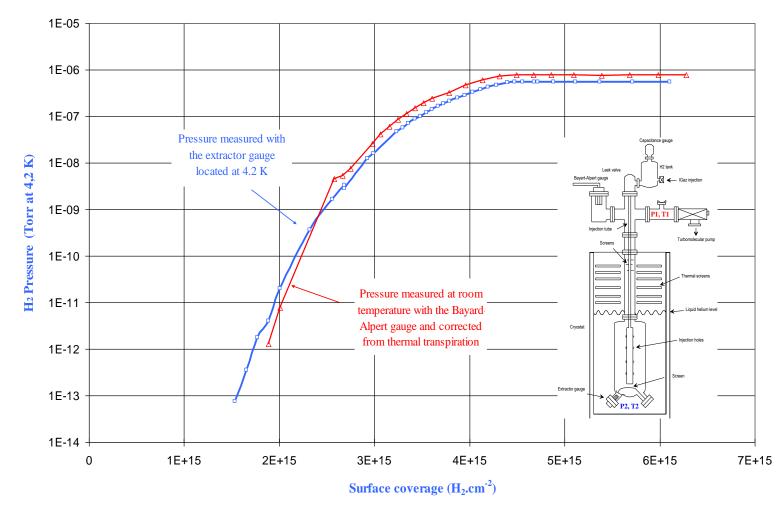


T (K)4.277
$$P_1/P_2$$
82





#### Experimental evidence of thermal transpiration Static conditions



V. Baglin et al. CERN Vacuum Technical Note 1995



# 2. Adsorption isotherms



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#### **Adsorption isotherm**

- Measurement, at constant temperature, of the equilibrium pressure for a given gas coverage, θ
- Varies with:
  - molecular species
  - surface temperature (under 20 K only H<sub>2</sub> and He)
  - surface nature
  - gas composition inside the chamber
- Models :

Henry's law for low surface coverage

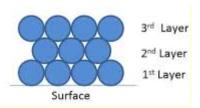
 $\theta = c P$ 

**DRK** (Dubinin, Radushkevich and Kaganer) for metalic, glass and porous substrate. Valid at low pressure. Good prediction with temperature variation

$$\ln(\theta) = \ln(\theta_m) - D\left(kT \ln\left(\frac{P_{sat}}{P}\right)\right)^2$$

BET (Brunauer, Emmet and Teller). Multi-monolayer description

$$\frac{P}{\theta (P - P_{Sat})} = \frac{1}{\alpha \theta_{m}} + \frac{(\alpha - 1)}{\alpha \theta_{m}} \frac{P}{P_{Sat}}$$



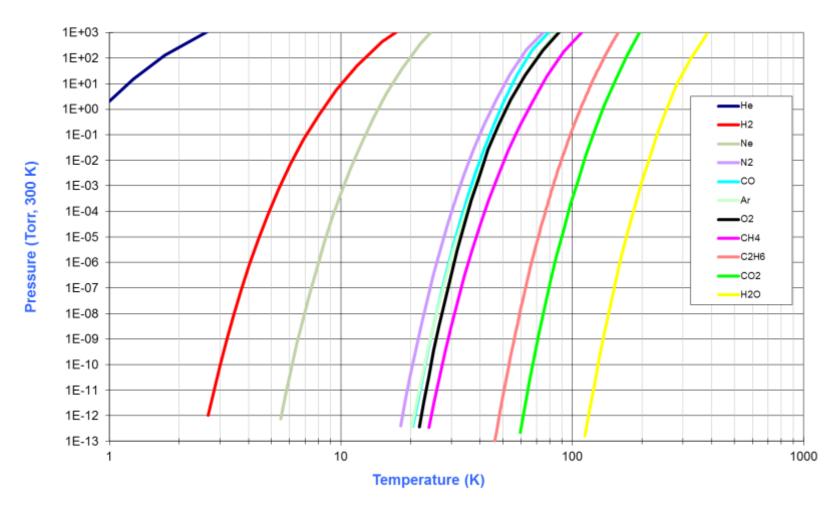


#### **Saturated Vapor Pressure**

• Pressure over liquid or gas phase (many monolayers condensed)

• Follows the Clausius-Clapeyron equation: Log  $P_{sat} = A - B/T$ 

#### Saturated vapour pressure from Honig and Hook (1960) (C2H6 Thibault et al.)





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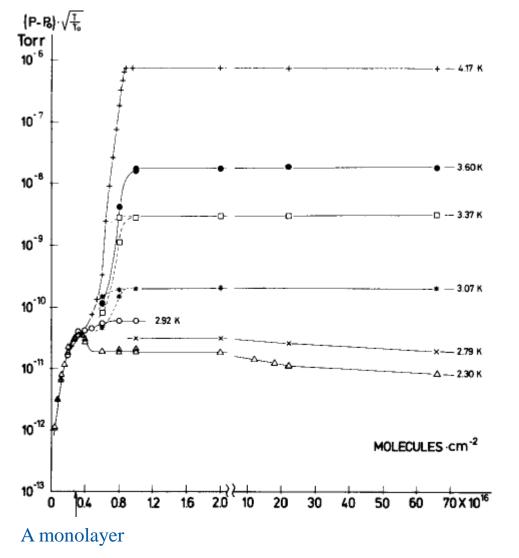
#### Joint Universities Accelerator School, Archamps, February , 2018

## H<sub>2</sub> Adsorption Isotherm on Stainless Steel

• The vapor pressure increases when increasing the adsorption of gas up to a few monolayers (~  $10^{15}$  molecules/cm<sup>2</sup>)

• The vapor pressure saturates when several monolayers of gas are adsorbed

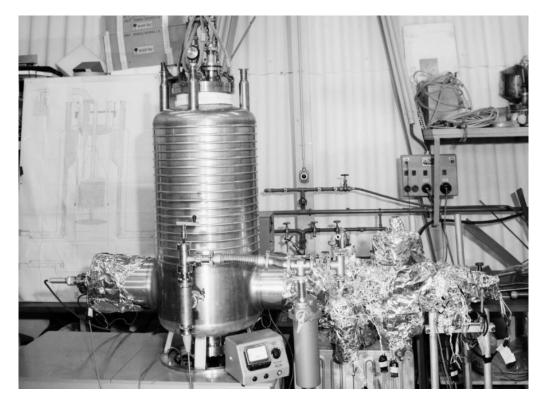
• The pressure level of the saturation is a function of the temperature



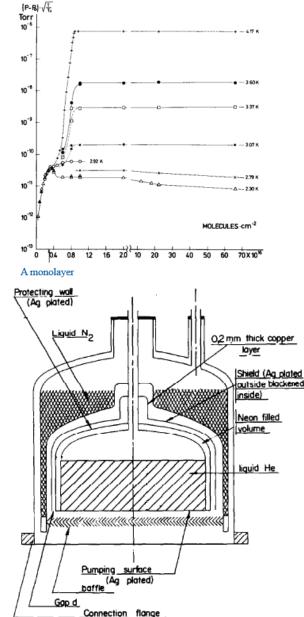
C. Benvenuti, R. Calder, G. Passardi J.Vac.Sci. 13(6), Nov/Dec 1976, 1172-1182

## H<sub>2</sub> adsorption isotherm on stainless steel

- $\bullet$  The condensation cryopumps allows to pump large quantities of  $\rm H_2$
- CERN ISR condensation cryopump operated with liquid He at 2.3 K (50 Torr on the He bath)



#### C. Benvenuti et al. Vacuum, 29, 11-12, (1974) 591





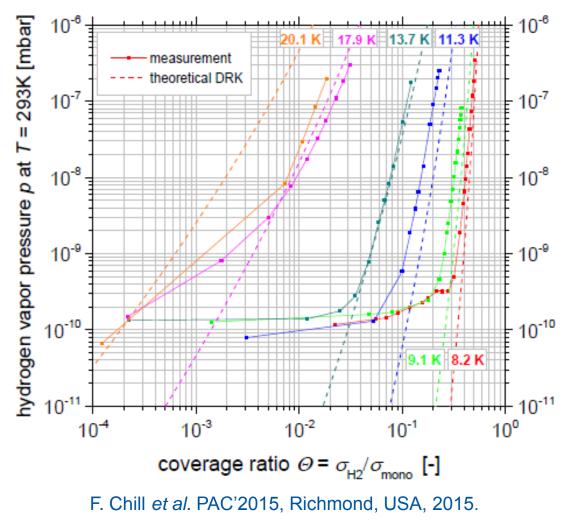
## H<sub>2</sub> adsorption isotherms from 8 to 20 K

• The surface capacity strongly decreases when increasing the surface temperature

- Stainless steel
- DRK description

$$\ln(\theta) = \ln(\theta_m) - D\left(kT\ln\left(\frac{P_{\text{sat}}}{P}\right)\right)^2$$

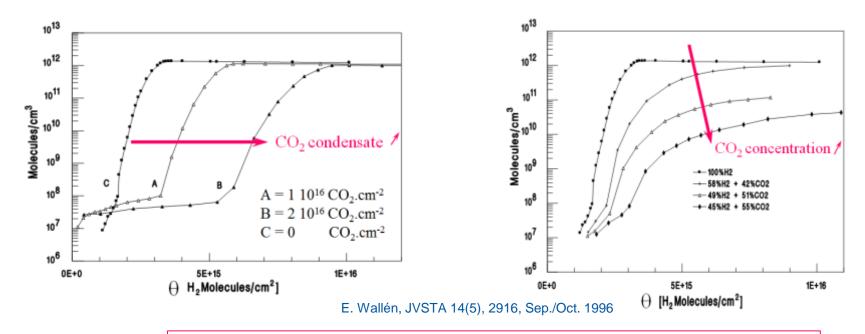
• D = 3125 eV<sup>-2</sup> •  $\Theta_m$  = 7 10<sup>14</sup> H<sub>2</sub>/cm<sup>2</sup>





#### **Vapor Pressure in a Machine**

- Several types of molecules are present in machine vacuum systems
- The adsorption isotherm is affected by the presence of these molecules
- Condensed CO<sub>2</sub> forms a porous layer increasing the hydrogen capacity
- Co-adsorption of CH<sub>4</sub>, CO and CO<sub>2</sub> reduce the vapor pressure of H<sub>2</sub> by cryotrapping



→ Studies with real machine environments are mandatory

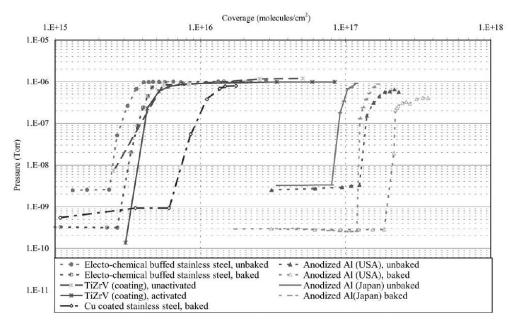


## H<sub>2</sub> Isotherms for Industrial Surfaces

Hydrogen adsorption capacity at 4.2 K

• Identification of two categories of adsorption sites with high energy (pores, defects) and low energy (flat surface).

	Molecules/cm <sup>2</sup> at saturation: $\sigma_{\rm m}$	Molecules/cm <sup>2</sup> at $P_{\rm sat}$ (10 <sup>-6</sup> Torr): $\sigma_{\rm sat}$	Ratio $\sigma_{\rm sat}/\sigma_{\rm m}$
Smooth surfaces			
Copper film unbaked	$6.07 \times 10^{15}$	$1.49 \times 10^{16}$	2.45
Electrochemical buffed stainless-steel unbaked	$2.36 \times 10^{15}$	$4.08 \times 10^{15}$	1.73
Electrochemical buffed stainless-steel baked	$2.68 \times 10^{15}$	$5.22 \times 10^{15}$	1.95
TiZrV film	$3.05 \times 10^{15}$	$6.02 \times 10^{15}$	1.97
Porous surfaces			
Al anodised unbaked (USA)	$1.23 \times 10^{17}$		
Al anodised baked (USA)	$1.80 \times 10^{17}$		
Al anodised (KEK)	$8.1 \times 10^{16}$	$1.18\times10^{17}$	1.46





G. Moulard, B. Jenniger, Y. Saito, Vacuum 60 (2001) 43-60



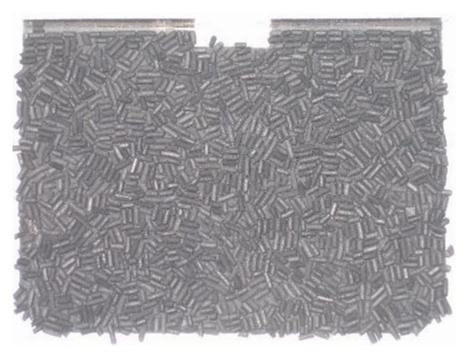
Table 1

#### **Cryosorbing materials**

- Large capacity
- Large pumping speed
- Large temperature working range (up to ~ 30 K)

e.g. Activated Charcoal used for cryopumps

Capacity ~  $10^{22}$  H<sub>2</sub>/g i.e.  $10^{21}$  monolayers (P. Redhead, Physical basis of UHV,1968) Sticking coefficient ~ 30 % at 30 K (T. Satake, Fus. Tech. Vol 6., Sept. 1984) 20 K cryopanels





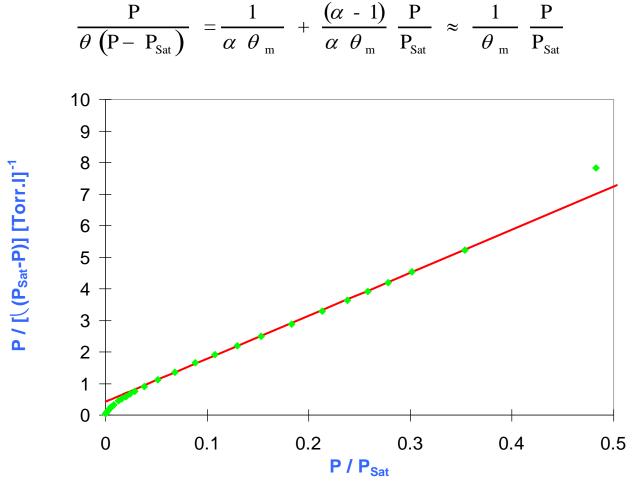
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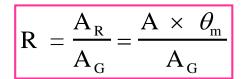
#### **B.E.T surface area**

• Xe is an inert gas which can only be physisorbed on a surface

• Xe adsorption isotherms at 77 K are used to derive the roughness factor of surface using the BET multi-monolayer theory

- Valid for 0.01<P/P<sub>sat</sub><0.3
- BET monolayer =  $\theta_m$
- $\alpha = \exp(\Delta E/kT) >>1$

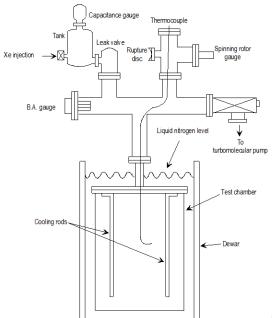




A for Xenon ~ 25  $Å^2$ 



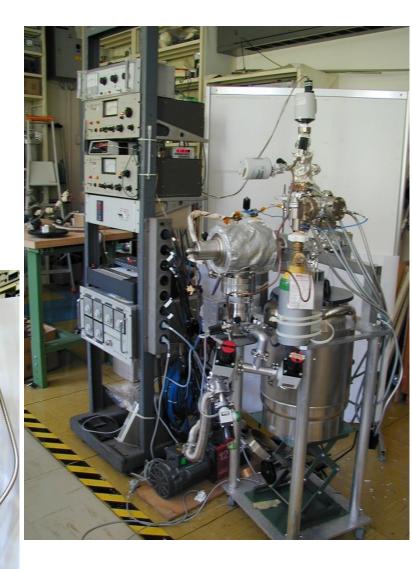
Vacuum, Surfaces & Coatings Group Technology Department





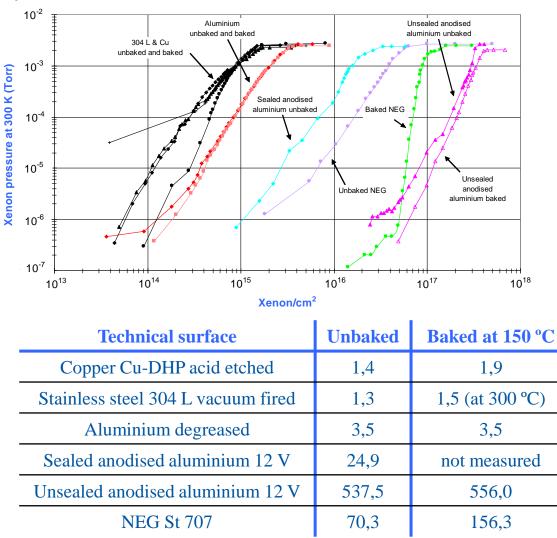
Vacuum, Surfaces & Coatings Group
Technology Department

#### **BET set-up**



#### **Roughness factor**

• Xe saturated vapour pressure ~ 2.6 10<sup>-3</sup> Torr at 77 K



V. Baglin. CERN Vacuum Technical Note 1997

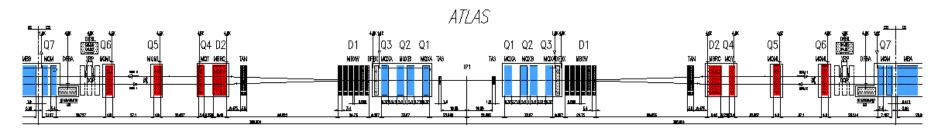


# 3. Cryosorbers in cold systems.Case of the LHC superconducting magnets operating at 4.5 K

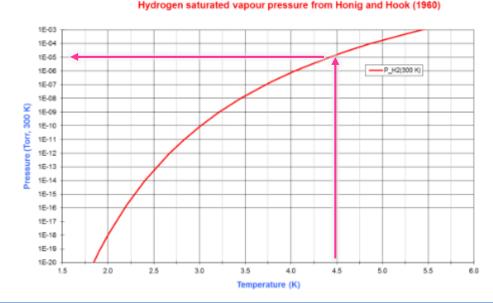


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## LHC Long straight section vacuum system



- Focusing inner triplets located around experiments operate at 1.9 K
- Matching sections operate at 4.5 K
- 1.9 K cold bore (~660 m, arc beam screen technology)
- ~ 4.5 K cold bore (~ 740 m)



Perforated

beam screens

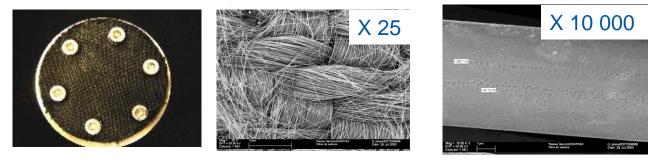
- With a 4.5 K Cold Bore
- Saturated vapour pressure equals 2 10<sup>-5</sup> mbar
- Cryosorbers are needed to provide a porous surface



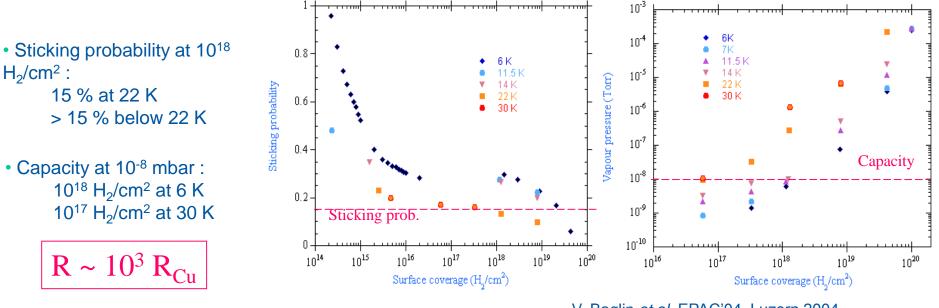
#### H<sub>2</sub> adsorption isotherm on cryosorbers

• Woven carbon fibers are used in LHC as cryosorbers in 4.5 K magnets

• Beam screen operates in the 5-20 K range



V. Anashin et al. Vacuum 75 (2004) 293-299

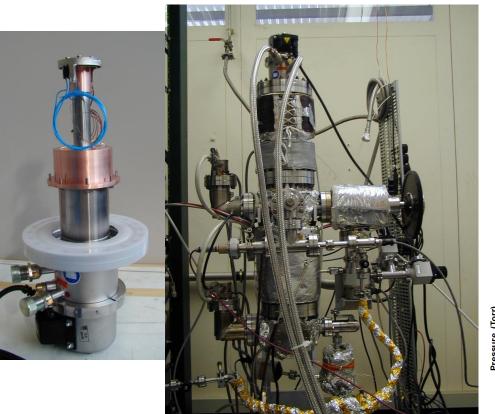


V. Baglin et al. EPAC'04, Luzern 2004.



#### **Fisher-Mommsen set Up**

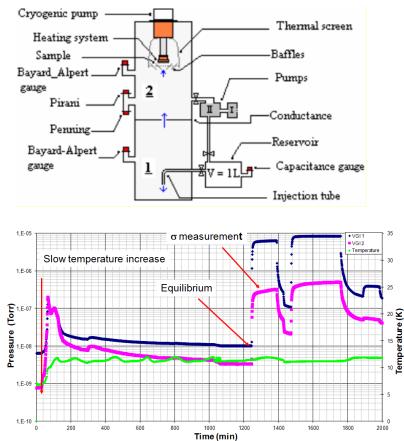
- The pumping speed is measured during the gas injection
- The vapor pressure is measured after injection, when dP/dt = 0



#### $C = 10.5 \text{ l/s for } H_2$

 $\mathbf{S} = \frac{\mathbf{C} \left( \Delta \mathbf{P}_1 - \Delta \mathbf{P}_2 \right)}{\Delta \mathbf{P}_2}$ 

 $\sigma = \frac{S}{S_{max}}$ 

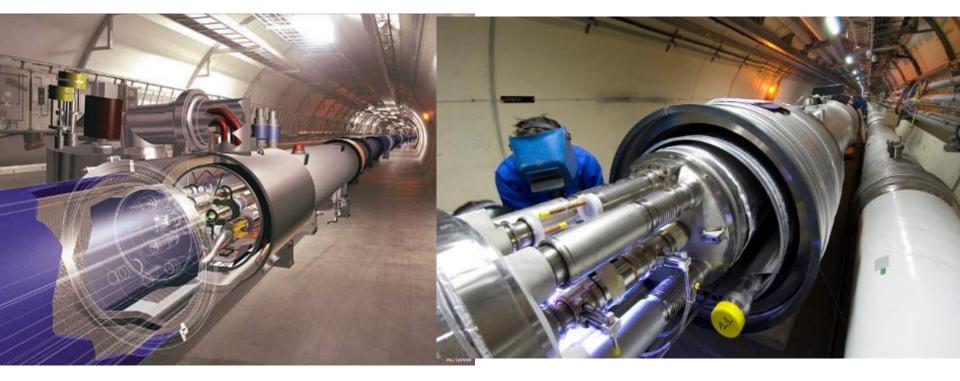




## 4. He leaks in cold systems



#### LHC : Superconducting technology



•Air leak or He leaks could appear in the beam tube during operation : the consequences are risk of magnet quench, pressure bump and radiation dose

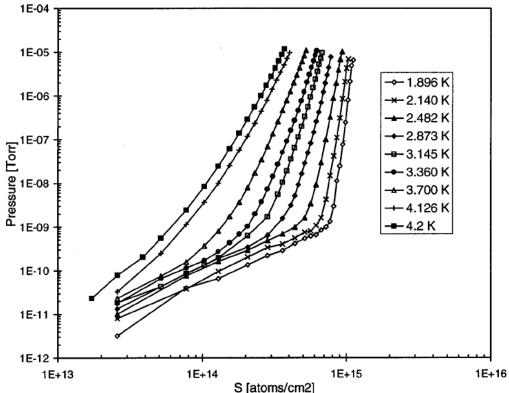


#### He adsorption isotherm from 1.9 to 4.2 K

•Sub-monolayer range

- Approaches Henry's law at low coverage
- The isotherms are well described by the DRK model
- θ<sub>m</sub> ~ 1.3 1015 H2/cm2

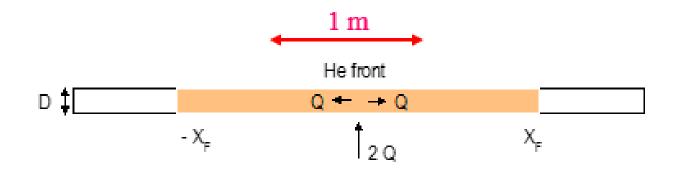
Stainless steel



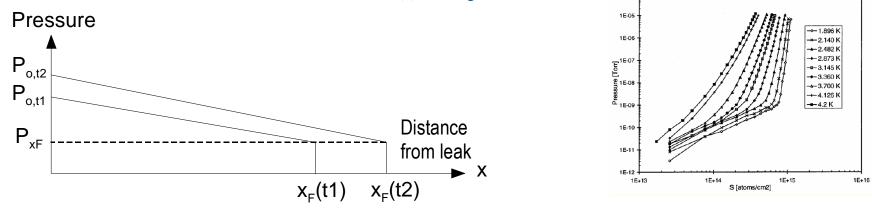
E. Wallén. J.Vac.Sci.A 15(2), Mar/Apr 1997, 265-274.



#### He leaks at 1.9 K



P. Hobson et al. J.Vac.Sci. A. 11(4), Jul/Aug 1993, 1566-1573



E. Wallén. J.Vac.Sci.A 15(2), Mar/Apr 1997, 265-274

33

- A He pressure wave is developed with time along the beam vacuum chamber
- The He wave can span over several tens of meter without being detected
- The local pressure bump gives a local proton loss (risk of quench)

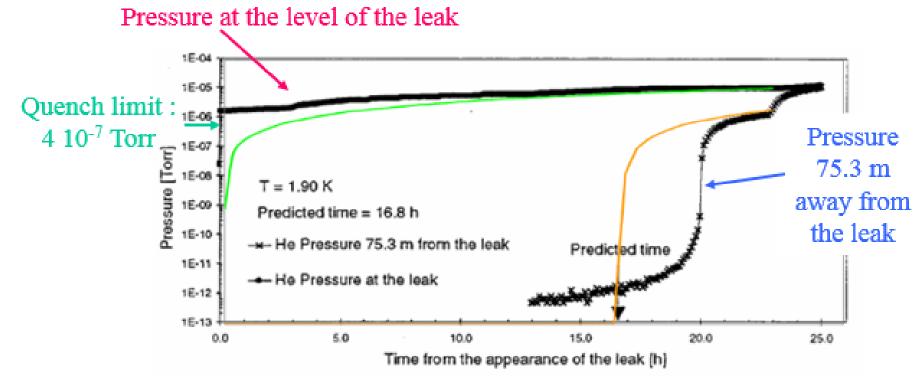


1E-04

#### **Example: LHC test string**

Example : LHC Test string Leak rate 6 10<sup>-5</sup> Torr.l/s Distance 75.3 m

20h to be detected 75 m downstream!



E. Wallén, JVST A 15(6), Nov/Dec 1997

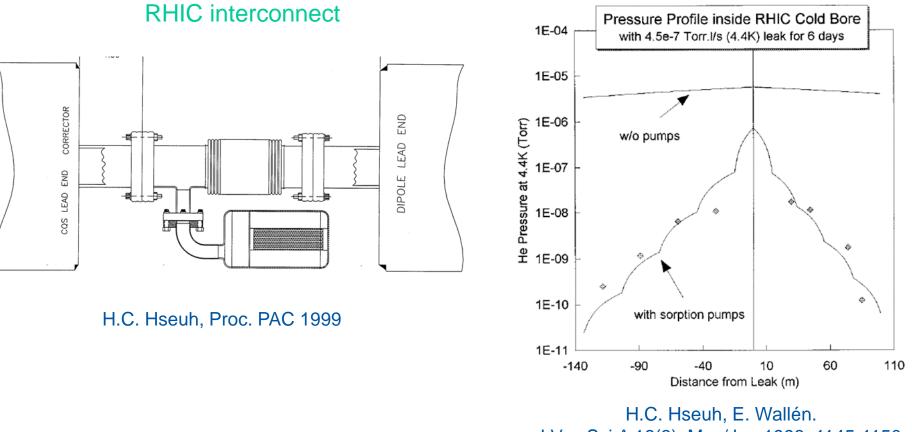


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#### RHIC, Brookhaven, USA

- RHIC's solution: anticipate leaks !!
- RHIC use sorption pumps based on 300 g of activated charcoal.
- They are located every 30 m to mitigate He leaks and to pump H<sub>2</sub>

#### Test in a 480 m long sector at 4.4 K





#### **Lecture 3 summary**

- Gas can be physisorbed for very long period on cryogenic surface
- The sticking coefficient characterise the pumping speed of a surface
- The capture coefficient characterise the pumping speed of a device
- At cryogenic temperature, thermal transpiration correction shall be applied
- •The vapour pressure is the equilibrium pressure as a function of gas coverage
- When saturated (many monolayers), the vapour pressure follows the Clausius Clapeyron law
- Adsorption isotherms vary very much with the conditions
- Some material can be porous so to adsorb many monolayers of gas without reaching the saturated vapour pressure: cryosorbers
- He leak can be difficult to detected at cryogenic temperature



#### **Some References**

- Cern Accelerator School, Vacuum technology, CERN 99-05
- Cern Accelerator School, Vacuum in accelerators, CERN 2007-03
- Cern Accelerator School, Vacuum for particle accelerators, Glumslov, June 2017
- The physical basis of ultra-high vacuum, P.A. Redhead, J.P. Hobson, E.V. Kornelsen. AVS.
- Capture pumping technology, K. Welch, North Holland.
- Cryopumping, theory and practice, R. Haefer, R. Clarendon press

#### **Some Journals Related to Vacuum Technolgy**

- Journal of vacuum science and technology
- Vacuum



#### Thank you for your attention !!!





Complementary information



# 2. Adsorption isotherms



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#### "Anomalous" Saturated Vapor Pressure

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Thermal radiation induced desorption:

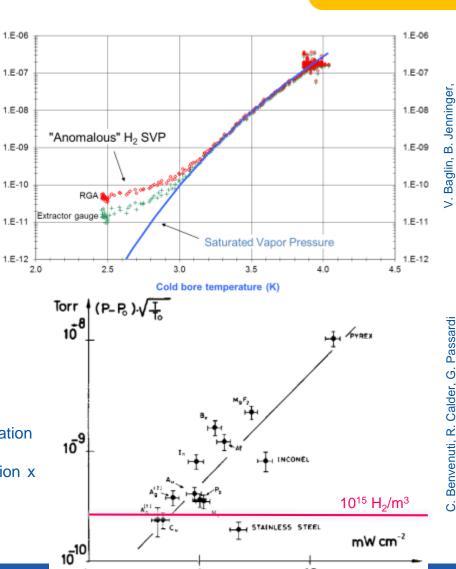
#### • In a "LHC type" mock-up (COLDEX):

 $\cdot$  After condensation of 10 monolayers of H\_2, the pressure follows the Clausius-Clapeyron equation while the cold bore temperature is decreased from 4 to 3 K

• Below 3 K, a deviation is observed due to the thermal radiation coming from the room temperature parts located at the extremities of the 2 m long system.

 $\cdot$  Increasing the beam screen temperature from 20 K to 100 K has no impact on the observed deviation while the cold bore is held at 2.7 K

- <u>Cryopump optimisation:</u>
- 10 monolayers of H<sub>2</sub> is condensed at 2.3 K
- The different cryosurface types are fully exposed to 300 K radiation
- Linear dependence with the absorbed power (incident radiation x substrate emissivity)
- The pressure, measured at 2.3 K, varies from  $10^{-10}$  to  $10^{-8}$  Torr => gas density 5  $10^{14}$  to 5  $10^{16}$  H<sub>2</sub>/m<sup>3</sup>





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1999

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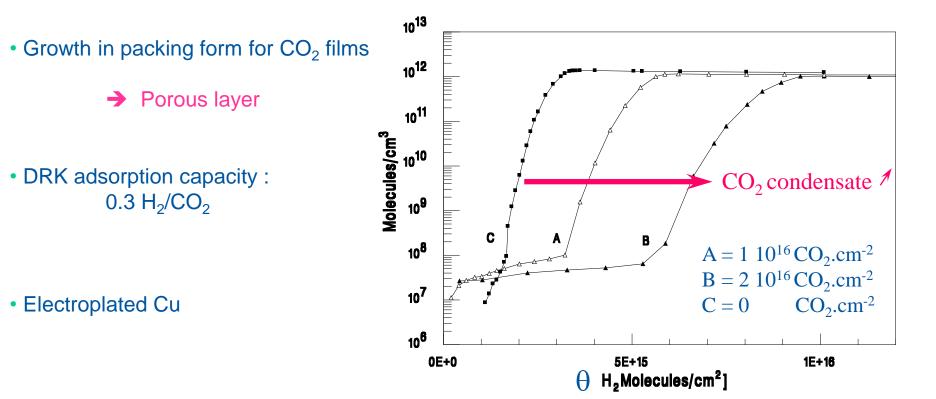
COLDEX Run

Complementary information

10

#### H<sub>2</sub> adsorption isotherm at 4.2 K on CO<sub>2</sub> condensat

Complementary information



E. Wallén, JVSTA 14(5), 2916, Sep./Oct. 1996

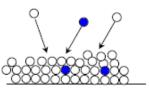


#### H<sub>2</sub> adsorption isotherm at 4.2 K in co-adsorption with CO<sub>2</sub>

Complementary information

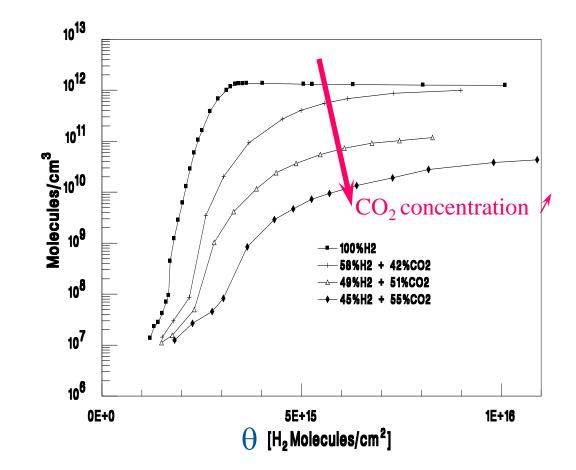
• Reduction of the saturated vapour pressure by orders of magnitude

→ Cryotrapping



Electroplated Cu

- In cryopumps  $CO_2$  is admitted to enhance the pumping of  $H_2$  and He



E. Wallén, JVSTA 14(5), 2916, Sep./Oct. 1996



#### CO<sub>2</sub> adsorption Isotherm at 77 K

### Complementary information

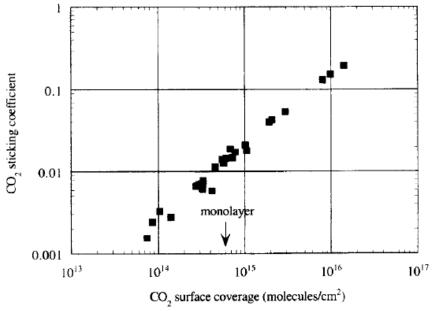


Figure 2. The sticking coefficient for  $CO_2$  at 77 K as a function of the surface coverage.

10-6 10.7 5 min Pressure (Torr) 10-8 12 h 10.9 monolayer Ψ 10.10 1013 1014 1015 10<sup>16</sup> 1012 1011 CO, surface coverage (molecules/cm<sup>2</sup>)

Figure 3. Adsorption isotherm for  $CO_2$  at 77 K as a function of surface coverage. The curves refer to measurements for different waiting times for pressure stabilisation.

V.V Anashin et al, Vacuum 48 (1997) 785-788

- Metallic surface
- Below a monolayer, the equilibrium pressure of the isotherm is obtained after several hours
- Due to the low sticking coefficient and the molecular adsorption by cluster.



# 3. Cryosorbers in cold systems.Case of the LHC superconducting magnets operating at 4.5 K



Vacuum, Surfaces & Coatings Group Technology Department Joint Universities Accelerator School, Archamps, February , 2018

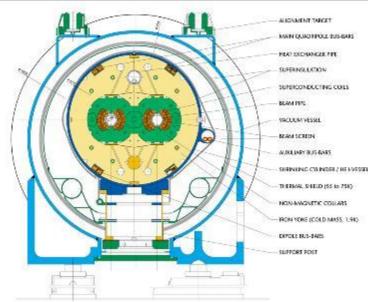
#### LHC dipole vacuum system

Complementary information

- Cold bore (CB) at 1.9 K

Beam screen (BS) at 5-20 K (intercept thermal loads)







CERN AC/00/MM - HE307 - \$2.04 1999



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#### LHC vacuum system principle

- Molecular desorption stimulated by photon, electron and ion bombardment
- Desorbed molecules are pumped on the beam vacuum chamber
- 100 h beam life time (nuclear scattering) equivalent to ~  $10^{15}$  H<sub>2</sub>/m<sup>3</sup> (10<sup>-8</sup> Torr H<sub>2</sub> at 300 K)

#### In cryogenic elements

Dipole cold bore at 1.9 K Cooling tubes Dia. 50/53 mm Dia. 3.7/4.8 mm Molecular physisorption onto cryogenic surfaces Beam screen (weak binding energy) 5 - 20 K Dia. 46.4/48.5 mm Molecules with a low recycling yield are first physisorbed onto the beam screen Electrons Hole  $(CH_4, H_2O, CO, CO_2)$  and then onto the cold bore stripes pumping 0000 36.8 mm Wall Photons pumpingo Desorbed • H<sub>2</sub> is physisorbed onto the cold bore molecules ංශිභිං



Complementary information

#### Required performances (for installation of 200 cm<sup>2</sup>/m):

Operates from 5 to 20 K

coverage

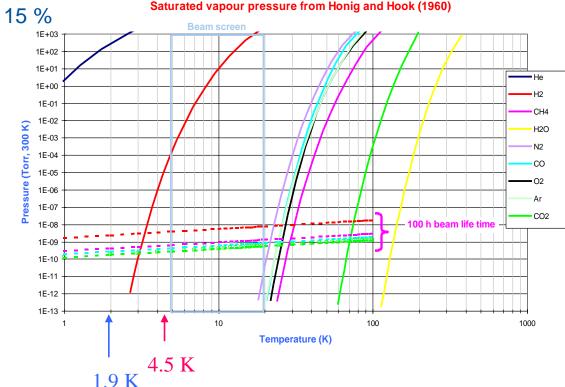
desorption

- Capacity larger than 10<sup>18</sup> H<sub>2</sub>/cm<sup>2</sup>
- Capture coefficient larger than 15 %

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The Design requires > 100 h life time with 4.5 K cold bore and thick surface

Taking into account the gas loads due to the beam stimulated molecular



Complementary information

# LHC Cryosorber Validation By Gas

#### Injection

### Complementary information

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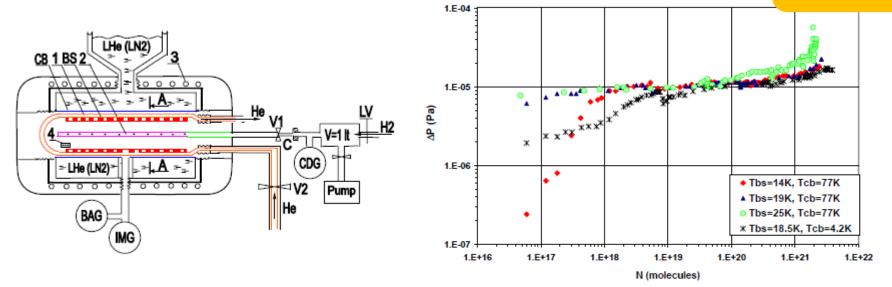


Fig. 3. Dynamic pressure measured by the gauge at room temperature in 1-m long LHC vacuum chamber prototype with a beam screen at different temperatures. Dynamic pressure is normalised to the injecting H<sub>2</sub> flux 10<sup>15</sup> molecule/s.

V.V Anashin et al, Vacuum 75 (2004) 293-299

• The dynamic pressure is below design value (*i.e.*  $10^{-8}$  mbar) for a molecular flux less than  $10^{14}$  H<sub>2</sub>/s

- When  $\eta_{photon} \sim 10^{-4} H_2/ph$ , then maximum photon flux is 10x LHC SR
- When  $\eta_{electron} \sim 10^{-3} H_2/ph$ , the corresponding electron cloud heat load is ~ 1 W/m



#### **Operation of Cryosorbers in LHC**

Complementary information

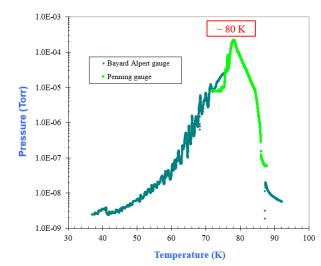
• The cryosorbers installed onto the back of the BS provide the required capacity and pumping speed for H<sub>2</sub>. They are located in cryoelements operating with 4.5 K cold bores

• The cryosorbers installed in D2, D3, D4, Q4, Q5 and Q6 of the LSS require a regeneration during the shutdown for removing the  $H_2$ 

- During normal operation of the LHC machine, a regeneration is not foreseen
- The cryosorber is regenerated at ~ 80 K

• While regenerating, the beam is OFF and the BS should be warmed up to more than 80 K and the CB held at more than 20 K (emptying cold mass)

• While the H<sub>2</sub> is liberated from the cryosorbers, it is pumped by an external pumping system.



• Activation energy Ed = 236 meV

$$\frac{\mathrm{E}_{\mathrm{d}}}{\mathrm{k}\mathrm{T}_{\mathrm{p}}^{2}} = \frac{1}{\tau_{\mathrm{o}}\beta} \exp\left(-\frac{\mathrm{E}_{\mathrm{d}}}{\mathrm{k}\mathrm{T}_{\mathrm{p}}}\right)$$

V. Baglin et al. EPAC'04, Luzern 2004.



## 4. He leaks in cold systems



#### He leak rate with risk of quench

Time to provoke a quench 1 year of operation ~ 150 days 150 **Fime in presence of He leak (days)** Nominal 1/3 of nominal 120 1/10 of nominal Helium leak rate above 90 5 10<sup>-7</sup> Torr.l/s shall be 60 detected to avoid the 30 risk of a quench 0 1.E-07 1.E-06 1.E-05 He leak rate at 300 K (Torr.l/s)

• Lower leak rate :

Require a pumping of the beam tube on the yearly basis (cold bore >~4K)

 Larger leak rate will provoke a magnet quench within : 30 to 100 days beam operation for He leak rate of 10<sup>-6</sup> Torr.l/s A day of beam operation for He leak rate of 10<sup>-5</sup> Torr.l/s



V. Baglin, Vacuum 81 (2007) 803-807

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