Possible Directions of R&D on the GAGG Crystals



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JSC Fomos-Materials

- Founded in 2001
- Main competence is single crystal growing technology
- Company possesses a full production circle that now includes:
 - Initial charge synthesis
 - Crystal growing process (12 crystal growing Cz pullers),
 - Modern and high productivity equipment for crystal processing (orient cutting, drilling, grinding, polishing and electrode deposition)
 - Testing
- High qualified staff, engineers and PhDs
- ISO 9001 certified
- Wide sales geographic:
- Fruitful and mutually beneficial cooperation with institutions
 of RAS and Ministry of Education NRS-KI, IK RAS, NUST MISiS, IREA
- Private companies RINK, NeoChim etc.







- Company has previous experience in crystal growth of PWO, YAG:Nd, GGG, LaAlO₃ and others
- Company manufactures following crystals
 - Langasite La₃Ga₅SiO₁₄
 - \circ Langatate La₃Ga_{5.5}Ta_{0.5}O₁₄
 - Catangasite $Ca_3TaGa_3Si_2O_{14}$
 - 0 *LN, LT*
 - \circ Isotopic enriched calcium molybdate $^{40}\mathrm{Ca}^{100}\mathrm{MoO}_4$ and others.
- A wide range of products such as:
 - SAW wafers for filters & resonators
 - Piezoelectric sensitive elements for high temperature sensors
 - Scintillation elements for detectors



The IBS/CUP-JSC Fomos-Materials contract for production of ⁴⁰Ca¹⁰⁰MoO₄ SEs for AMoRE-Pilot (Dec. 2014 - June 2017)

Isotopical enrichment	Molybdenum enriched on Mo-100 isotope – no less than 95%						
Isotopical depletion	Calcium depleted on Ca-48 isotope – less than 0,002%						
Radioactive purity	No more than 100 micro-Bq/kg for Bi214 (U238 chain) No more than 500 micro-Bq/kg for Bi211 (U235 chain), <mark>Alpha-activity:</mark> No more than 10 mBq/kg for total alpha-activity of U- and Th-chains.						

RESOURSES in Russia

- Production of Mo-100 exists in Russia *at tenths hundred kg scale*
- Production of Ca-40 isotope exists in Russia (a stock of enriched on Ca-40 and depleted on Ca-48 material after production Ca-48 isotope
- Up to now JSC Fomos-Materials is a unique company to produce isotopic enriched complex oxide crystals with extra-high pure materials
- JSC Fomos-Materials established technology of SE production, fabricated pilot lot of SEs and have a real possibility of SEs production for big-scale experiment

SE10 Impurity Determination and Isotopic Enrichment



Element	Content, ppb
Th	<0,1
U	<0,1

Isotopic composition	Content, atomic %			
⁹² Mo	0,38			
⁹⁴ Mo	0,23			
⁹⁵ Mo	0,4			
⁹⁶ Mo	0,44			
⁹⁷ Mo	0,28			
⁹⁸ Mo	3.17			
¹⁰⁰ Mo	95,1			
Isotopic composition	Content, atomic %	6		
Isotopic composition ⁴⁰ Ca	Content, atomic % 99,949±0,006	6		
Isotopic composition ⁴⁰ Ca ⁴² Ca	Content, atomic % 99,949±0,006 0,019	6		
Isotopic composition ⁴⁰ Ca ⁴² Ca ⁴³ Ca	Content, atomic % 99,949±0,006 0,019 0,002	6		
Isotopic composition ⁴⁰ Ca ⁴² Ca ⁴³ Ca ⁴⁴ Ca	Content, atomic % 99,949±0,006 0,019 0,002 0,028	6		
Isotopic composition ⁴⁰ Ca ⁴² Ca ⁴³ Ca ⁴⁴ Ca ⁴⁶ Ca	Content, atomic % 99,949±0,006 0,019 0,002 0,028 ≤0,001	/0		
Isotopic composition ⁴⁰ Ca ⁴² Ca ⁴³ Ca ⁴⁴ Ca ⁴⁶ Ca ⁴⁸ Ca	Content, atomic % 99,949±0,006 0,019 0,002 0,028 <0,001 0,002	<u>/o</u>		

JSC Fomos-Materials can produce extrapure crystals with impurity content lower than 0.1 ppb

6	Element	Substance %						
	Li	< 0.0001	Cu	< 0.0001	Sn	<0.0001	Yb	< 0.0001
_	Ве	< 0.0005	Zn	< 0.0002	Sb	< 0.0001	Lu	< 0.0001
	В	< 0.001	Ga	< 0.0001	Те	< 0.0002	Hf	< 0.0001
	Na	< 0.002	Ge	< 0.0001	I	< 0.0005	Та	< 0.0001
	Mg	< 0.0003	As	< 0.0001	Cs	< 0.0001	W	0.0007
	Al	< 0.0003	Se	< 0.002	Ba	< 0.0001	Re	< 0.0001
6	Si	< 0.005	Br	< 0.005	La	< 0.0001	Os	< 0.0001
-	Р	< 0.005	Rb	< 0.0001	Ce	< 0.0001	Ir	< 0.0001
	К	< 0.005	Sr	< 0.0001	Pr	< 0.0001	Pt	< 0.0001
	Ca	basement	Y	< 0.0001	Nd	< 0.0001	Au	< 0.0001
	Sc	< 0.0002	Zr	< 0.0001	Sm	< 0.0001	Hg	< 0.0001
	Ti	< 0.0004	Nb	< 0.0001	Eu	< 0.0001	Tl	< 0.0001
	V	< 0.001	Ru	< 0.0001	Gd	< 0.0001	Pb	< 0.0001
	Cr	< 0.001	Rh	< 0.0001	Tb	< 0.0001	Bi	< 0.0001
	Mn	< 0.0001	Pd	< 0.0001	Dy	< 0.0001	Th	< 0.0001
	Fe	< 0.005	Ag	< 0.0001	Но	< 0.0001	U	< 0.0001
	Со	< 0.0001	Cd	< 0.002	Er	< 0.0001		
	Ni	< 0.0001	In	< 0.0001	Tm	< 0.0001		

New Perspective Crystal for Scintillation Application

Crystal*	Den- city	Melting Point	Hydro- scopicity	Main Scint. Decay Time, ns	LY, Ph/MeV	Energy Res. at 662 keV, %
Pr:Lu ₃ Al ₅ O ₁₂	6.71	1980	No	20 (20-40%)	16000-20000	4.5-6.5
Ce:Gd ₃ (Ga,Al) ₅ O ₁₂	6.63	1850	No	88 (90%)	46000-51000	4.9-5.5
PbWO ₄	8.28	1160	No	3-6	200	30-40
CdWO ₄	7.90	1325	No	5000	27000	6.6
Ce:Lu ₂ SiO ₅	7.4	2150	No	35	26000	7.9
Ce:Y ₂ SiO ₅	4.45	2070	No	39	32000	8.1
Ce:Gd ₂ SiO ₅	6.71	1900	No	60	12500	7.8
Bi ₄ Ge ₃ O ₁₂	7.13	1050	No	300	8500	9.0
Ce:LuAlO ₃	8.34	~1900	No	16-20	11400	9.0
Ce,Eu:LiCaAlF ₆	4.88	820	No	1670	40000	
Ce:LaBr ₃	5.30	783	Yes, very high	16(100%)	70000	2.6



Kimura H., et al., Czochralski Growth of $Gd_3(Ga_{1-x}Al_x)_50_{12}$ Single Crystals, Journal of Crystal Growth 74 (1986) 187-190 187

Kamada, K., et al., Composition Engineering in Cerium-Doped (Lu,Gd)₃(Ga,Al)₅O₁₂ Single Crystal Scintillators. Crystal Growth & Design, 2011. 11(10): p. 4484-4490.

*Yoshikawa A., et al., Czochralski Growth and Properties of Scintillating Crystals, Acta Physica Polonica A, 2013., v.124. No.2, p.250-264

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For Wide Commercial Application Need to Decrease or Eliminate Phosphorescence!



1. Effect of codoping on scintillation and optical properties of a Ce-doped Gd₃Ga₃Al₂O₁₂ scintillator / Tyagi M., Meng F., Koschan M., e.a. // Journal of Physics D: Applied Physics. - 2013. - V. 46. - No. 47. - P. 475302.

Impurity Distribution



1. *A. Yoshikawaa, V. Chani and M. Nikl, Czochralski Growth and Properties of Scintillating Crystals) Acta Physica Polonica A 124 (2013) No. 2 250-264

JSC Fomos-Materials can optimize a cost of crystal with guarantee of crystal quality and homogeneity

Cerium Distribution in GGAG

$$C_{Ce_{cryst}} = K_{Ce} C_{0_{melt}} (1-g)^{K_{Ce}-1}$$

Pfann Equation for Normal Crystallization Process



L.Qin, H.Li, Sheng Lu, D.Ding, G.Ren, Growth and characteristics of LYSO(Lu_{2(1-x-y)}Y_{2x}SiO₅:Ce_y) scintillation crystals, J. of Cryst. Growth 281 (2005) 518-524
 A. Yoshikawaa, V. Chani and M. Nikl, Czochralski Growth and Properties of Scintillating Crystals) Acta Physica Polonica A 124 (2013) No. 2 250-264

GGAG:Ce Crystal Growth by Czochralski Method



JSC Fomos-Materials made a lot of crystal growth experiments with optically non-active impurities

Already Grown GGAG Crystals

Up to now JSC Fomos-Materials grew GGAG crystals with following compositions:

 GGAG:Ce with different cerium concentrations •GGAG:Ce with different Ga/Al ratio •GGAG:Ce with codoping of:



listed elements

Magnesium Calcium Strontium Barium Scandium Zirconium Titanium •GGAG:Ce with pair and triple codoping of above





MAGG-MULTIDOPED ALUMINUM GALLIUM GARNET

MAGG is the gadolinium aluminum gallium garnet doped with Ce and pair of codopands is a perfect scintillation material designed to overcome drawbacks of Ce solely doped and Mg or Ca codoped crystals.

Density, g/cm ³	Z _{eff} /photo absorp. coeff., 511 keV,cm ⁻¹	Luminescence maximum, nm	Light yield, ph/MeV	Decay kinetics, ns(%)	Energy resolution, %	Time resolution (CTR), ps
6.68-6.63 depending on Al/Ga ratio	51/0.12	520	38000(RT) 46000(-45°C)	30 (25%), 80 (60%), 100- 200 ns (15%)	6,2(511keV) At -20°C with SiPM 3,6(1270keV) at -20°C with SiPM	170 (-20 to 20°C)

Scintillation kinetics of GAGG:Ce (dark blue) and MAGG (magenta) samples measured at room temperature



Radiation Instruments and New Components LLC Minsk, Belarus

> Noise before scintillation kinetics characterize level of phosphorescence. 1 channels corresponds to 1 ns

Parameters of kinetics of:

GAGG:Ce: τ1= 32ns(36%),τ2=86ns (56%),τ3=354ns (8%); MAGG- τ1= 27ns(34%),τ2=73ns (54%),τ3=200ns (11%).



Amplitude spectra of 137Cs source measured with R329 PMT at different temperatures

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Codoping safe a lightyeld



Change of the energy resolution for a photo-peak with temperature of GAGG:Ce (red dots) and MAGG(black dots)





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Gated (measured in different time gates) light output of sample with temperature



Radiation induced absorption spectra of GAGG and

Radiation Instruments and New Components LLC Minsk, Belarus

MAGG crystals



Ce³⁺ absorption bands and location of the Ce³⁺ luminescence realatively to absorption spectra

Radiation induced absorption spectra of GAGG and MAGG crystals obtained at different stages of the technology development (irradiation with 60Co (1,2MeV)

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Possible Directions of R&D on the GAGG Crystals

For the 1st Stage

- The quality of the crystal (the energy resolution) and the choice of its methods of control
- Optimization of Ce³⁺ and codopands concentrations (light yield and radiation hardness)
- High temperature heat treatment of the grown crystals (temperature, gas medium, duration etc.)
- Development of technological processes for mechanical treatment of the grown crystals (manufacturing of special form crystal elements)
- Product Specification must be determinate
 For the 2nd Stage
- Statistical analysis of the developed complex technology –
- yield, return and irrecoverable loss of the crystal growth process
- Economic analysis initial costs, cost, etc.

As-grown GGAG:Ce Crystal 50 mm in dia.







Full Technological Circle

- Initial Charge Synthesis
- Crystal Growing
- High Temperature Heat Treatment
- Cutting, Grinding, Drilling, Polishing.
- ISO 9001:2008





Examples of Optical Elements Production



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Sensitive elements for HT sensors: pressure and vibration



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MAGG Scintillation Elements Ø16x60 mm



 $\sqrt{R a 0,01}$ (\checkmark)

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