



Electron Trains in LXe TPCs

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XENON1T

https://phys.org/news/2017-05xenon1t-sensitive-detector-earthwimp.html

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 10^{3}

 10^{3}

Probing lower S2 energies





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20

CRESST-II

PICO-21

Photoionization



https://arxiv.org/pdf/1311.1088.pdf

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Electron Train Background





Malter effect

Trapped positive ions on the cathode can lower the work function of the metal, making it easier to emit electrons. This may lead to few-electron bursts.

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Long lived excited states



https://arxiv.org/pdf/1711.07025.pdf

My Work



Determining the decay constant(s) of the electron trains



Determining electron efficiency by analyzing data with different electric fields

Conclusion

- Electron trains could be caused by:
 - Photoionization
 - Trapped electrons
 - Malter effect
 - Long lived excited states
- By eliminating these electron trains we can use noble element detectors to probe for sub GeV/c² dark matter particles!

Back up slides

Equations to slide 9

$$\phi_b = \frac{e^2}{8\pi\epsilon_0 z} \frac{\epsilon - 1}{\epsilon + 1}.$$

Schottky barrier – aka work function. As an electron approaches a dielectric boundary that is held at a constant potential, the force due to it's image charge results in an energy barrier.

epislon => dielectric constant of boundary (1.5-2 for xenon)

z => characteristic dimension of the order of the lattice constant (~5 angstrom)

$$\Delta \phi_b = e \left(\frac{eE}{4\pi\epsilon_0 z} \frac{\epsilon - 1}{\epsilon + 1} \right)^{1/2}$$

The external field E does two things with respect to electron emission: it increases the energy of the drifting electrons, and it lowers the height of the barrier by an amount equal to

$$\kappa = \int_{\phi_b - \Delta \phi_b}^{\infty} \varepsilon^{1/2} f_0(\varepsilon) d\varepsilon \bigg/ \int_0^{\infty} \varepsilon^{1/2} f_0(\varepsilon) d\varepsilon,$$

Electron emission efficiency. the factor $\epsilon 1/2$ serves to select electrons whose velocity has a component directed toward the barrier.

f_0 is the electron energy distribution $\epsilon \rightarrow$ energy

$$\kappa_n = 1 - (1 - \kappa)^n.$$

Electron emission efficiency after n tries