

Synchrotron Radiation — Exercises 2

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1 Applications

Why are synchrotrons important for science?

Mostly Research purposes:

- Biology
- chemistry
- medicine
- Pharmaceutical science
- photolithography

Short wavelengths allows to investigate matter... - High Energy physics

2 Orbit Correction

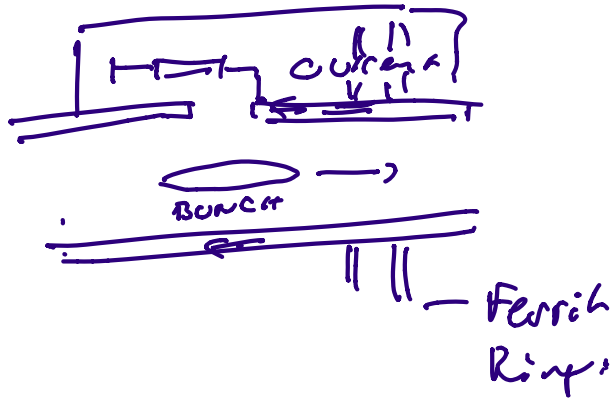
Which devices are used to measure and correct the orbit inside a synchrotron?

- Beam position monitors
- correct the orbit with dipoles
& kickers for fast corrections
- additional coils in quadrupole magnets
- mechanical displacement of quadrupoles

3 Instrumentation

How would you measure the bunch length in a synchrotron?

Wall Current Monitor



Streak camera on synchrotron light from a dipole magnet

4 Detection

What possibilities exist to detect X-Rays?

- CCD
- PHOTODIODES
- SCINTILLATORS
- PHOTOGRAPHIC FILMS
- ionization chambers

5 Monochromators

What dispersive element is used to monochromatize X-Rays? What differences exist to monochromators for visible light?

A crystal can be used to monochromatize X-Rays.

Differences: - wavelengths

- condition on input and output of light
- for visible light we use mechanically fabricated grating
- for X-rays we used grating made of atoms

6 Refractive Index

The passage of electromagnetic radiation can be described classically by an index of refraction. What are the properties of the index of refraction of most materials at X-ray wavelengths?

$$n = 1 - \delta + i\beta$$

$$\delta, \beta \ll 1$$

7 DLSRs

How do longitudinal gradient bends contribute towards the goal of achieving a lower horizontal emittance in a diffraction limited storage ring?

$$\epsilon \propto \int D^2(s) B^3(s) ds$$

c.c. to design the Bending magnets so that largest magnetic fields correspond to lowest dispersion region.

8 Diffraction

Why is diffraction often used in place of imaging when using X-rays?

What is the *phase problem* in X-ray diffraction?

X rays can be used because they have approximately the same order of magnitude in wavelength than the distance between crystal plane \Rightarrow use diffraction

Intensity, ~~polarization~~, ~~angle of incidence~~, phase



9 Crystals

Which of the following are crystalline? More than one answer is possible.

- The glass on the screen of my mobile phone
- The sapphire glass on an expensive watch
- Asbestos
- Icing sugar

10 Absorption and Diffraction

A scientist wants to record a diffraction pattern of a silicon crystal at a photon energy of 8 keV. What is the optimum thickness of the crystal, that maximizes the intensity of the diffracted spot?

Hint: you can find the mass absorption coefficient of silicon on page 1-41 (page 49 in the PDF) of the X-Ray Data Booklet, and the density on page 5-5 (page 153).

$$\mu = 100 \text{ cm}^2/\text{g}$$

$$\rho = 233 \text{ g/cm}^3$$

$$\mu = \frac{1}{\mu \cdot \rho} \text{ cm}$$

$$I = I_0 \cdot \underbrace{e^{-\frac{x}{\mu}}}_{\text{ABSORPTION}} \cdot \underbrace{(\sigma \cdot x)}_{\text{SCATTERING}}$$

$$\downarrow \frac{d}{dx}$$

$$0 = \sigma \cdot e^{-\frac{x}{\mu}} - \frac{1}{\mu} \cdot \sigma x \cdot e^{-\frac{x}{\mu}}$$

$$x = \mu = \frac{1}{233} \text{ cm} = 43 \mu\text{m}$$

